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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=173 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.037$
$w R$ factor $=0.087$
Data-to-parameter ratio $=13.7$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 2-(2-Bromo-5-methoxyphenyl)-6-phenyl-1,3-thiazolo[3,2-b][1,2,4]triazole

The molecule of the title compound, $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{BrN}_{3} \mathrm{OS}$, is essentially planar. Geometric parameters are in the normal ranges. There are two intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds.

## Comment

The reaction of $\alpha$-haloketones with 2 -substituted 5-mercapto-$1,2,4$-triazoles may result in the formation of either the $2,5-$ disubstituted-thiazolo[3,2-b]-s-triazole or the 3,5-disubstituted thiazolo[3,2-b]-1,2,4-triazole or both (Berk et al., 2001; Potts \& Husain, 1971). The synthesis of thiazolo[3,2-b]-1,2,4-triazoles and the isomeric thiazolo[2,3-c]-1,2,4-triazoles, and their diuretic, antibacterial and antifungal activities, have been studied by Jag Mohan \& Kiran (1988). In the present study, 5-(2-bromo-5-methoxyphenyl)-4H-1,2,4-triazole-3-thiol, (1), was refluxed with phenacyl bromide, (2), in ethanol to obtain 2-(2-bromo-5-methoxyphenyl)-6-aryl-1,3-thiazolo[3,2-b]$[1,2,4]$ triazole, (3) (see scheme). We present here the crystal structure of (3).


(2)
(1)

(3)

A perspective view of compound (3) is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 5.27, November 2005, updated August 2006; MOGUL, Version 1.1; Allen, 2002). The molecule is essentially planar (r.m.s. deviation for all non-H atoms is $0.059 \AA$ ).

The molecular conformation of (3) is stabilized by two intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds (Table 1).

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## Experimental

For the synthesis of compound (3), 5-(2-bromo-5-methoxyphenyl)$4 H-1,2,4$-triazole-3-thiol $(2.86 \mathrm{~g}, 0.01 \mathrm{~mol})$ and the appropriate 2 -bromo-1-phenylethanone ( $2 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) were refluxed in ethanol for 4 h . The progress of the reaction was monitored by thin-layer chromatography. After completion of the reaction, the reaction mixture was cooled and the precipitated solid was filtered off. The solid obtained was recrystallized from a methanol-acetone solvent mixture (1:1). The compound was obtained as creamish crystals in $48 \%$ yield (m.p. 483 K ). Analysis for $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{BrN}_{3} \mathrm{OS}$, found (calculated): C 52.80 (52.86), H 3.07 (3.13), N 10.72 (10.88) \%. Spectroscopic data: IR (KBr, $v, \mathrm{~cm}-1): 3118$ and $3070(-\mathrm{CH}), 1475(-\mathrm{C}=\mathrm{N}-), 734(\mathrm{C}-\mathrm{Br})$.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{17} \mathrm{H}_{12} \mathrm{BrN}_{3} \mathrm{OS} \\
& M_{r}=386.27 \\
& \text { Orthorhombic, } P b c a \\
& a=14.9271(13) \AA \\
& b=10.9247(13) \AA \\
& c=18.8582(16) \AA \\
& V=3075.3(5) \AA \\
& \AA
\end{aligned}
$$

## $Z=8$

$D_{x}=1.669 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=2.82 \mathrm{~mm}^{-1}$
$T=173$ (2) K
Plate, colourless
$0.32 \times 0.16 \times 0.08 \mathrm{~mm}$

## Data collection

| Stoe IPDS II two-circle | 10330 measured reflections |
| :--- | :--- |
| $\quad$ diffractometer | 2883 independent reflections |
| $\omega$ scans | 2189 reflections with $I>2 \sigma(I)$ |
| Absorption correction: multi-scan | $R_{\text {int }}=0.064$ |
| (MULABS; Spek, 2003; Blessing, | $\theta_{\max }=25.6^{\circ}$ |
| $1995)$ |  |
| $T_{\min }=0.466, T_{\max }=0.806$ |  |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.037$
$w R\left(F^{2}\right)=0.087$
$S=0.98$
2883 reflections
210 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0505 P)^{2}\right] \\
& \quad \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.55 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.80 \mathrm{e}^{-3} \\
& \text { Extinction correction: } S H E L X L 97 \\
& \quad \text { (Sheldrick, 1997) } \\
& \text { Extinction coefficient: } 0.0046 \text { (4) }
\end{aligned}
$$



Figure 1
The molecular structure of compound (3), with the atom numbering; displacement ellipsoids are drawn at the $50 \%$ probability level.

H atoms were found in a difference map but they were subsequently refined using a riding model, with $\mathrm{C}-\mathrm{H}=0.95 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$ or $\mathrm{C}-\mathrm{H}=0.98 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}\left(\mathrm{C}_{\text {methyl }}\right)$. The methyl group was allowed to rotate but not to tip.

Data collection: $X$-AREA (Stoe \& Cie, 2001); cell refinement: $X-A R E A$; data reduction: $X-A R E A$; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97.

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Table 1
Hydrogen-bond geometry ( $\mathrm{A}^{\circ}{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| C26-H26 $\cdots \mathrm{N} 8$ | 0.95 | 2.40 | $2.788(4)$ | 104 |
| C12-H12 N 6 | 0.95 | 2.35 | $3.040(4)$ | 129 |


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