Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## H. S. Yathirajan, ${ }^{\text {a }}$ Anil N.

 Mayekar, ${ }^{\text {a }}$ B.K. Sarojini, ${ }^{\text {b }}$ B. Narayana ${ }^{c}$ and Michael Bolte ${ }^{d}$ *${ }^{\text {a }}$ Department of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, ${ }^{\mathbf{b}}$ Department of Chemistry, P. A. College of Engineering, Nadupadavu, Mangalore 574 153, India, 'Department of Chemistry, Mangalore University, Mangalagangotri 574 199, India, and dnstitut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Max-von-Laue-Strasse 7, 60438 Frankfurt/Main, Germany

Correspondence e-mail:
bolte@chemie.uni-frankfurt.de

## Key indicators

Single-crystal X-ray study
$T=173 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.040$
$w R$ factor $=0.097$
Data-to-parameter ratio $=7.6$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## (2E)-3-(6-Methoxy-2-naphthyl)-1-phenyl-prop-2-en-1-one

The geometric parameters of the title compound, $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{2}$, are in the usual ranges. The central double bond is trans configured. Its two C atoms are slightly twisted out of the naphthyl plane. The dihedral angle between the aromatic groups is $14.09(8)^{\circ}$.

## Comment

Reviews on the bioactivities of varieties of chalcones are given by Dimmock et al. (1999) and Go et al. (2005). Recently, it has been noted that, among many organic compounds reported for their second harmonic generation, chalcone derivatives are known for their excellent blue light transmittance and good crystallizability (Fichou et al., 1988; Goto et al., 1991; Uchida et al., 1998; Zhao et al., 2000; Sarojini et al., 2006). The crystal structures of 3-(4-chlorophenyl)-1-(2-naphthyl)prop-2-enone (Shanmuga Sundara Raj et al., 1997), 1-(2-naphthalenyl)-3-(3-nitrophenyl)-2-propen-1-one (Shanmuga Sundara Raj et al., 1998), 3-(6-methoxy-2-naphthyl)-1-(2-naphthyl)prop-2-en-1one (Yathirajan, Sarojini, Bindya et al., 2006) and 3-(6-meth-oxy-2-naphthyl)-1-(2-thienyl)prop-2-en-1-one (Yathirajan, Narayana et al., 2006) have been reported. The crystal structures of 1-(2,4-dichloro-5-fluorophenyl)-3-(3,4-dimethoxy-phenyl)prop-2-en-1-one (Yathirajan, Sarojini, Narayana et al., 2006) and (2E)-1-(2,4-dichlorophenyl)-3-[4-(methylsulfanyl) phenyl]prop-2-en-1-one (Butcher et al., 2007) have also been reported. In continuation of our broad programme on chalcones, the present paper reports the crystal structure of a newly synthesized chalcone.


A perspective view of the title compound, (I), is shown in Fig. 1. Bond lengths and angles can be regarded as normal [Cambridge Crystallographic Database, Version 5.28, November 2006 (Allen, 2002); Mogul Version 1.1 (Bruno et al., 2004)]. The carbonyl group is twisted by $21.0(3)^{\circ}$ out of the plane of the phenyl ring. The torsion angle between the carbonyl group and the C atoms of the double bond is $-15.9(4)^{\circ}$. The torsion angle between the the C atoms of the double bond and the adjacent naphthyl residue ( $\mathrm{C} 2-\mathrm{C} 3-$ $\mathrm{C} 21-\mathrm{C} 30)$ is $-10.4(3)^{\circ}$. The two aromatic residues are not coplanar [dihedral angle 14.9 (8) ${ }^{\circ}$ ].

There are non-conventional hydrogen bonds of the type $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ in the structure that link the molecules into chains

Received 22 January 2007
Accepted 26 January 2007
lying along the $b$ axis. In addition to these intermolecular interactions, intramolecular interactions $\mathrm{C} 3-\mathrm{H} 3 \cdots \mathrm{O} 1$ are also present (Table 1).

## Experimental

5 ml of $40 \% \mathrm{KOH}$ solution was added to a thoroughly stirred solution of acetophenone ( $1.2 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) and 6-methoxy-2-napthaldehyde $(1.86 \mathrm{~g}, 0.01 \mathrm{~mol})$ in 25 ml of methanol. The mixture was stirred overnight and filtered. The solid obtained was recrystallized from acetone-toluene ( $1: 1$ ) mixture (m.p. 421-423 K). Analysis for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{2}$ found (calculated): C 83.18 (83.31), H 5.50 (5.59) \%.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{2} \\
& M_{r}=288.33 \\
& \text { Orthorhombic, } P \mathrm{Pc} a 2_{1} \\
& a=14.5275(17) \AA \AA \\
& b=17.0930(15) \AA \\
& c=5.9950(5) \AA \\
& V=1488.7(3) \AA^{3}
\end{aligned}
$$

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.286 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation }^{\prime} \\
& \mu=0.08 \mathrm{~mm}^{-1} \\
& T=173(2) \mathrm{K} \\
& \text { Thick plate, light yellow } \\
& 0.27 \times 0.24 \times 0.13 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Stoe IPDS II two-circle
$\quad$ diffractometer
$\omega$ scans
Absorption correction: none
7578 measured reflections

1535 independent reflections
1367 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.089$
$\theta_{\text {max }}=25.6^{\circ}$
7578 measured reflections

## Refinement

Refinement on $F^{2}$

$$
w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0648 P)^{2}\right]
$$

$$
\text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3
$$

$(\Delta / \sigma)_{\max }<0.001$
$\Delta \rho_{\text {max }}=0.21 \mathrm{e}^{\mathrm{m}}{ }^{-3}$
$\Delta \rho_{\min }=-0.21 \mathrm{e}^{-3}$
Extinction correction: SHELXL97
Extinction coefficient: 0.029 (5)


Figure 1
The molecular structure of the title compound with the atom numbering; displacement ellipsoids are at the $50 \%$ probability level.

Data collection: $X$-AREA (Stoe \& Cie, 2001); cell refinement: $X$-AREA; data reduction: $X$-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PLATON.

ANM thanks the University of Mysore for research facilities.

## References

Allen, F. H. (2002). Acta Cryst. B58, 380-388.
Bruno, I. J., Cole, J. C., Kessler, M., Luo Jie, Motherwell, W. D. S., Purkis, L. H., Smith, B. R., Taylor, R., Cooper, R. I., Harris, S. E. \& Orpen, A. G. (2004). J. Chem. Inf. Comput. Sci. 44, 2133-2144.
Butcher, R. J., Yathirajan, H. S., Narayana, B., Mithun, A. \& Sarojini, B. K. (2007). Acta Cryst. E63, o30-o32.

Dimmock, J. R., Elias, D. W., Beazely, M. A. \& Kandepu, N. M. (1999). Curr. Med. Chem. 6, 1125-1149.
Fichou, D., Watanabe, T., Takeda, T., Miyata, S., Goto, Y. \& Nakayama, M. (1988). Jpn. J. Appl. Phys. 27, 429-430.

Go, M. L., Wu, X. \& Liu, X. L. (2005). Curr. Med. Chem. 12, 483-499.
Goto, Y., Hayashi, A., Kimura, Y. \& Nakayama, M. (1991). J. Cryst. Growth, 108, 688-698.
Sarojini, B. K., Narayana, B., Ashalatha, B. V., Indira, J. \& Lobo, K. G. (2006). J. Cryst. Growth, 295, 54-59.

Shanmuga Sundara Raj, S., Ponnuswamy, M. N., Shanmugam, G. \& Nanjundan, S. (1997). Acta Cryst. C53, 917-918.
Shanmuga Sundara Raj, S., Ponnuswamy, M. N., Shanmugam, G. \& Nanjundan, S. (1998). Acta Cryst. C54, 541-542.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
Stoe \& Cie (2001). X-AREA. Stoe \& Cie, Darmstadt, Germany.
Uchida, T., Kozawa, K., Sakai, T., Aoki, M., Yoguchi, H., Abduryim, A. \& Watanabe, Y. (1998). Mol. Cryst. Liq. Cryst. 315, 135-140.
Yathirajan, H. S., Narayana, B., Ashalatha, B., Sarojini, B. K. \& Bolte, M. (2006). Acta Cryst. E62, o4440-o4441.

Yathirajan, H. S., Sarojini, B. K., Bindya, S., Narayana, B. \& Bolte, M. (2006). Acta Cryst. E62, o4046-o4047.
Yathirajan, H. S., Sarojini, B. K., Narayana, B., Bindya, S. \& Bolte, M. (2006). Acta Cryst. E62, o3631-o3632.
Zhao, B., Lu, W.-Q., Zhou, Z.-H. \& Wu, Y. (2000). J. Mater. Chem. 10, 15131517.


[^0]:    (C) 2007 International Union of Crystallography All rights reserved

