

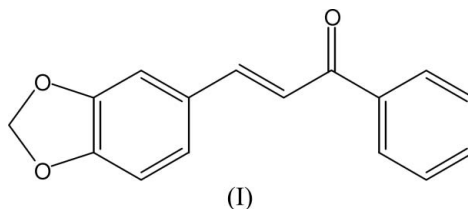
3-(1,3-Benzodioxol-5-yl)-1-phenylprop-2-en-1-one

H. S. Yathirajan,^a B. K. Sarojini,^b
B. Narayana,^c S. Bindya^d and
Michael Bolte^{e*}^aDepartment of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, ^bDepartment of Chemistry, P. A. College of Engineering, Nadupadavu, Mangalore 574 153, India, ^cDepartment of Chemistry, Mangalore University, Mangalagangotri 574 199, India, ^dDepartment of Chemistry, Sri Jayachamarajendra College of Engineering, Manasagangotri, Mysore 570 006, India, and ^eInstitut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Marie-Curie-Strasse 11, 60439 Frankfurt/Main, GermanyCorrespondence e-mail:
bolte@chemie.uni-frankfurt.de

Key indicators

Single-crystal X-ray study
T = 173 K
Mean $\sigma(C-C)$ = 0.001 Å
R factor = 0.034
wR factor = 0.092
Data-to-parameter ratio = 16.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.In the title biologically active compound, C₁₆H₁₂O₃, the central C=C double bond is *trans* configured. The molecule consists of two essentially planar parts which are twisted by 26.89 (5)° with respect to each other.Received 14 July 2006
Accepted 17 July 2006

Comment

Chalcones possess a broad spectrum of biological activities, including antibacterial, antihelmintic, amoebicidal, anti-ulcer, antiviral, insecticidal, antiprotzoal, anticancer, cytotoxic and immunosuppressive activities (Dimmock *et al.*, 1999). Certain chalcone derivatives were reported to inhibit the polymerization of tubulin to form microtubules and were therefore antimitotic agents which can be used as anti-inflammatory agents. Chalcone derivatives were also reported to inhibit the destruction of myelin sheath in the central nervous system of multiple sclerosis patients and were thus useful in controlling the progressive nature of the disease (Edwards *et al.*, 1989). Chalcones can be easily obtained from the aldol condensation of aromatic aldehydes and aromatic ketones. This class of compounds presents interesting biological properties, such as cytotoxicity (Lawrence *et al.*, 2001), antiherpes activity and antitumour activity and may be useful for the chemotherapy of leishmaniasis (Pandey *et al.*, 2005), among others. A comparison of the supramolecular structures of 1-(6-amino-1,3-benzodioxol-5-yl)-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one and 1-(6-amino-1,3-benzodioxol-5-yl)-3-[4-(*N,N*-dimethylamino)phenyl]prop-2-en-1-one has been described previously (Low *et al.*, 2002). The crystal structure of (1,3-benzodioxol-5-ylmethyl)ammonium 2-methoxy-5-[(1*E*)-3-oxo-3-phenylprop-1-en-1-yl]benzenesulfonate monohydrate (da Silva *et al.*, 2006) has recently been reported. In view of the importance of the title compound, (I), its crystal structure is reported here.The molecular structure of the title compound is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Crystallographic Database, Version 5.27, November 2005 updated May 2006; Mogul Version 1.1; Allen, 2002). The central C=C double bond is *trans* configured. The molecule consists of two planar segments which are twisted by 26.89 (5)° with respect to each other. One of these contains the

1,2-(methylenedioxy)benzene group which is coplanar with the propenone group (r.m.s. deviation for all non H atoms 0.049 Å), and the other segment is the phenyl ring. All torsion angles are close to 0 or 180°; only those about the C10—C11 bond differ significantly from planarity (Table 1). The crystal packing shows a herringbone pattern (Fig. 2) and reveals one weak C—H···O contact (Table 2).

Experimental

The title compound was synthesized according to the method reported in the literature (Vogel, 1989) with a yield of 75%. The compound was purified by recrystallization from ethanol. The crystal growth was performed in acetone solvent by slow evaporation (m.p. 365 K). Analysis found (calculated) for C₁₆H₁₂O₃: C 76.20 (76.18), H 4.75 (4.79)%.

Crystal data

C ₁₆ H ₁₂ O ₃	Z = 8
M _r = 252.26	D _x = 1.366 Mg m ⁻³
Orthorhombic, <i>Pbca</i>	Mo Kα radiation
a = 11.1234 (5) Å	μ = 0.09 mm ⁻¹
b = 7.7504 (4) Å	T = 173 (2) K
c = 28.4607 (11) Å	Plate, light yellow
V = 2453.62 (19) Å ³	0.38 × 0.21 × 0.19 mm

Data collection

Stoe IPDS-II two-circle diffractometer	2813 independent reflections
ω scans	2590 reflections with I > 2σ(I)
Absorption correction: none	R _{int} = 0.049
31874 measured reflections	θ _{max} = 27.6°

Refinement

Refinement on F ²	w = 1/[σ ² (F _o ²) + (0.0462P) ² + 0.6049P]
R[F ² > 2σ(F ²)] = 0.035	where P = (F _o ² + 2F _c ²)/3
wR(F ²) = 0.092	(Δ/σ) _{max} = 0.001
S = 1.04	Δρ _{max} = 0.22 e Å ⁻³
2813 reflections	Δρ _{min} = -0.15 e Å ⁻³
173 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: 0.033 (2)

Table 1

Selected torsion angles (°).

C6—C8—C9—C10	-178.11 (9)	O10—C10—C11—C12	-25.96 (14)
O10—C10—C11—C16	151.40 (11)	C9—C10—C11—C12	154.85 (9)
C9—C10—C11—C16	-27.79 (14)		

Table 2

Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C1—H1A···O10 ⁱ	0.99	2.59	3.0397 (14)	107

Symmetry code: (i) x - ½, -y + ½, -z + 1.

H atoms were found in a difference map, but placed geometrically and allowed to ride on their parent C atoms at distances of 0.95 and 0.99 Å for aromatic and methylene groups, respectively, and with U_{iso}(H) = 1.2U_{eq}(C).

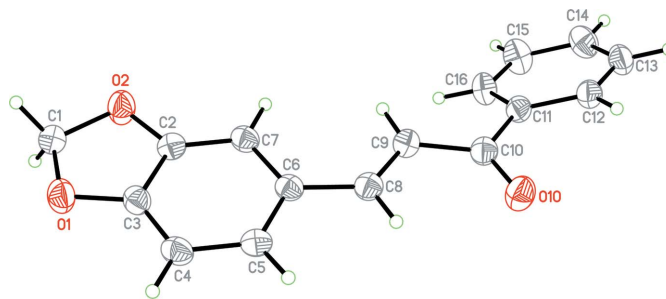


Figure 1

The molecular structure of (I), with the atom numbering; displacement ellipsoids are at the 50% probability level.

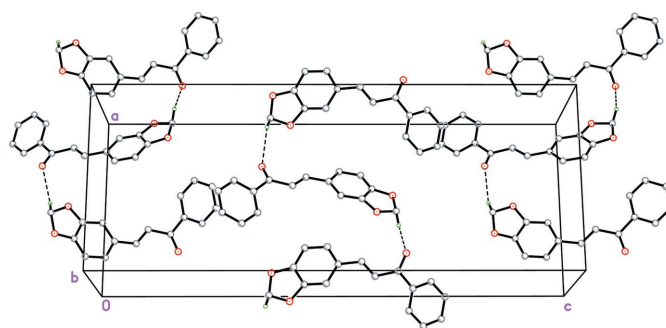


Figure 2

Packing diagram of the title compound, with a view approximately along the *b* axis. Weak C—H···O hydrogen bonds are shown as dashed lines, and H atoms not involved in these interactions have been omitted.

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP in SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

BKS thanks AICTE, Government of India, for financial assistance through the Career Award for Young Teacher's Scheme.

References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
- Dimmock, J. R., Elias, D. W., Beazely, M. A. & Kandepu, N. M. (1999). *Curr. Med. Chem.* **6**, 1125–1149.
- Edwards, M. L., Sunkara, S. P. & Stemerick, D. M. (1989). US Patent No. 4 863 968.
- Lawrence, N. J., Rennison, D., McGown, A. T., Ducki, S., Gul, L. A., Hadfield, J. A. & Khan, N. (2001). *J. Comb. Chem.* **3**, 421–426.
- Low, J. N., Cobo, J., Noguera, M., Sánchez, A., Albornoz, A. & Abonia, R. (2002). *Acta Cryst.* **C58**, o42–o45.
- Pandey, S., Suryawanshi, S. N., Gupta, S. & Srivastava, V. M. L. (2005). *Eur. J. Med. Chem.* **40**, 751–756.
- Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Silva, L. E. da, Andrighetti-Fröhner, C. R., Nunes, R. J., Simões, C. M. O. & Foro, S. (2006). *Acta Cryst.* **E62**, o2785–o2787.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Stoe & Cie (2001). *X-AREA*. Stoe & Cie, Darmstadt, Germany.
- Vogel, A. I. (1989). *Vogel's Textbook of Practical Organic Chemistry*, edited by B. S. Furniss, A. J. Hannaford, P. W. G. Smith & A. R. Tatchell, 5th ed. London: Longman.