

A dynamic model of oceanic sulfur (DMOS) applied to the Sargasso Sea: Simulating the dimethylsulfide (DMS) summer paradox

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[1] A new one-dimensional model of DMSP/DMS dynamics (DMOS) is developed and applied to the Sargasso Sea in order to explain what drives the observed dimethylsulfide (DMS) summer paradox: a summer DMS concentration maximum concurrent with a minimum in the biomass of phytoplankton, the producers of the DMS precursor dimethylsulfoniopropionate (DMSP). Several mechanisms have been postulated to explain this mismatch: a succession in phytoplankton species composition towards higher relative abundances of DMSP producers in summer; inhibition of bacterial DMS consumption by ultraviolet radiation (UVR); and direct DMS production by phytoplankton due to UVR-induced oxidative stress. None of these hypothetical mechanisms, except for the first one, has been tested with a dynamic model. We have coupled a new sulfur cycle model that incorporates the latest knowledge on DMSP/DMS dynamics to a preexisting nitrogen/carbon-based ecological model that explicitly simulates the microbial-loop. This allows the role of bacteria in DMS production and consumption to be represented and quantified. The main improvements of DMOS with respect to previous DMSP/DMS models are the explicit inclusion of: solar-radiation inhibition of bacterial sulfur uptakes; DMS exudation by phytoplankton caused by solar-radiation-induced stress; and uptake of dissolved DMSP by phytoplankton. We have conducted a series of modeling experiments where some of the DMOS sulfur paths are turned “off” or “on,” and the results on chlorophyll-a, bacteria, DMS, and DMSP (particulate and dissolved) concentrations have been compared with climatological data of these same variables. The simulated rate of sulfur cycling processes are also compared with the scarce data available from previous works. All processes seem to play a role in driving DMS seasonality. Among them, however, solar-radiation-induced DMS exudation by phytoplankton stands out as the process without which the model is unable to produce realistic DMS simulations and reproduce the DMS summer paradox.

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1. Introduction

[2] The oceanic sulfur cycle, believed to be an important part of the Earth biogeochemical system because of its potential for climate regulation has received considerable

attention in the last two decades. However, owing to the complexity of the cycle, in which the whole microbial food web is involved [Simó, 2001], some important features regarding its seasonal dynamics remain largely unexplained. Phytoplankton are the primary producers of dimethylsulfoniopropionate (DMSP), the biochemical precursor of dimethylsulfide (DMS), a volatile compound that is ubiquitous in the global surface ocean. Emission of oceanic DMS to the atmosphere [Bates *et al.*, 1992; Kettle and Andreae, 2000] is thought to contribute to non-sea-salt sulfate (nss-SO₄) production and cloud condensation nuclei (CCN) formation [Charlson *et al.*, 1987; Andreae and Crutzen, 1997; Vallina *et al.*, 2007]. The amount of atmospheric CCN is linked to cloud albedo and therefore to the Earth radiative budget [Twomey, 1974; Albrecht, 1989; Kaufman *et al.*, 2002]. In this regard, a negative feedback between oceanic DMS

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production and Earth albedo has been postulated [Charlson *et al.*, 1987].

[3] Intracellular DMSP (also called particulate DMSP or DMSPp) is released to the water as dissolved DMSP (DMSPd) during phytoplankton cell lysis by natural (non-grazing) mortality, zooplankton grazing and virus attacks [Groene, 1995; Yoch, 2002; Steinke *et al.*, 2002a]. However, the amount of DMSPp varies among phytoplankton groups [Keller *et al.*, 1989; Keller and Korjef-Bellows, 1996] as well as with the physiological state of the cells within each group [Keller and Korjef-Bellows, 1996; Stefels, 2000; Sunda *et al.*, 2002; Bucciarelli and Sunda, 2003; Slezak and Herndl, 2003]. DMSP may also be exuded by phytoplankton living cells as an overflow of energy [Groene, 1995; Stefels, 2000]. The conversion of DMSP to DMS is mediated by DMSP-lyase, an enzyme that has been found in DMSP-producing phytoplankton groups as well as in numerous groups of DMSP-consuming bacteria [Groene, 1995; Yoch, 2002; Zubkov *et al.*, 2002; Niki *et al.*, 2000; Wolfe *et al.*, 2002; Steinke *et al.*, 2002b]. Until recently it was believed that the majority of DMS production was due to zooplankton grazing on phytoplankton and bacterial activity on DMSPd [Levasseur *et al.*, 1996; Dacey *et al.*, 1998; González *et al.*, 1999]. However, recent studies suggest that the role of phytoplankton DMS production has been overlooked [Simó and Pedrós-Alió, 1999; Niki *et al.*, 2000; Wolfe *et al.*, 2002; Sunda *et al.*, 2002; Toole and Siegel, 2004; Toole *et al.*, 2006]. Under conditions of high UV radiation stress or severe nutrient limitation it seems that phytoplankton may be responsible of an important fraction of the total DMS production [Stefels and van Leeuwe, 1998; Wolfe *et al.*, 2002; Sunda *et al.*, 2002]. Most groups of oceanic bacteria are able to undertake DMSPd consumption (from which only a small fraction is cleaved to DMS plus acrylate, the rest being demethylated to other forms of sulfur [Groene, 1995; Yoch, 2002; Kiene and Linn, 2000]). DMS is consumed as a carbon source mostly by some methylotrophic bacteria [Kiene and Bates, 1990; Kiene, 1992, 1993; Bates *et al.*, 1994; Wolfe *et al.*, 1999; Simó *et al.*, 2000; Yoch, 2002; Zubkov *et al.*, 2004; Vila-Costa *et al.*, 2006a] or converted to dimethylsulfoxide (DMSO) with energy gain by unknown bacteria [Vila-Costa *et al.*, 2006a; del Valle *et al.*, 2007]. The other major sinks of DMS are photolysis by UV (a process mediated by photosynthesizer substances) [Brimblecombe and Shooter, 1986; Brugger *et al.*, 1998; Toole *et al.*, 2003; Kieber *et al.*, 1996] and emission to the atmosphere [Kettle and Andreae, 2000]. Also, it has been recently discovered that non DMSP-producing phytoplankton are also able to take up DMSPd, potentially reducing the amount of DMSPd available for bacteria degradation and its conversion to DMS [Vila-Costa *et al.*, 2006b].

[4] Both DMSP and DMS are an integral part of the dissolved organic matter (DOM) pool. The oceanic cycles of DOM and organic sulfur are therefore thought to be tightly coupled [Vézina, 2004]. DMSPd appears to be the main source of sulfur (S) for bacteria [Kiene *et al.*, 1999; Kiene and Linn, 2000; Yoch, 2002; Zubkov *et al.*, 2001, 2002], although it is also a source of carbon (C) [Yoch *et al.*, 1997; Zubkov *et al.*, 2001; Yoch, 2002]. On the other hand, DMS is mainly a source of carbon and energy, sulfate and DMSO being the primary fate of sulfur from bacterial consumption of DMS [Vila-Costa *et al.*, 2006a; del Valle *et al.*,

et al., 2007]. DMS dynamics are therefore regulated by a complex interplay of biotic and abiotic processes where phytoplankton, zooplankton and bacteria are believed to have a prominent role.

[5] With the aim at gaining a better understanding on the processes governing the oceanic sulfur cycle, several dynamic (i.e. mechanistic) models of DMSP/DMS have been developed in the last decade or so [Gabric *et al.*, 1993; Lawrence, 1993; van den Berg *et al.*, 1996; Laroche *et al.*, 1999; Jodwalis *et al.*, 2000; Archer *et al.*, 2002; Lefevre *et al.*, 2002; Chu *et al.*, 2004]. These models usually consist of two submodels: a nitrogen based one (N-cycle) characterizing the ecosystem, and a sulfur based one (S-cycle) of the DMSP/DMS dynamics [Vézina, 2004]. These two submodels are coupled but without feedbacks between them: the N-cycle affects the S-cycle, but not viceversa [Vézina, 2004]. Most of these models do not, however, include a characterization of the microbial loop, that is, an explicit representation of bacteria, bacterivory, and the DOM cycle. This is due to the fact that the first ecosystem models did not give sufficient relevance to the microbial loop.

[6] However, in recent years, bacteria and DOM dynamics have gained relative importance in ecosystem models and it has been shown that their inclusion is fundamental in order to obtain realistic simulations of the seasonal cycles of the model state variables [Spitz *et al.*, 2001]. In the Sargasso Sea, for example, most of the carbon cycling is through the microbial loop [Steinberg *et al.*, 2001, and references therein]. The DMSP/DMS model of Archer *et al.* [2002] (which is based on the ERSEM ecosystem model [Baretta *et al.*, 1995]) is the only ecosystem model to additionally incorporate bacteria and DOM dynamics. Although the model of Gabric *et al.* [1993] incorporated bacteria as part of the N-cycle, bacteria were not explicitly represented in the S-cycle. Rather, bacterial effects on DMSP and DMS concentrations were parameterized as constant rates, independently of their evolution in the N-cycle. This is probably due to the fact that the N-cycle of Gabric's model (which is based on Moloney *et al.* [1986]) does not include DOM dynamics, such that predicted bacteria display "catastrophic behavior", being close to zero for much of the time [Cropp, 2002; Cropp *et al.*, 2004]. Variations in bacterial sulfur demand have been suggested to affect DMS production, so that if DMSPd is in excess of bacterial requirements for sulfur, a larger proportion of the DMSPd taken up could be converted to DMS [Kiene *et al.*, 1999]. Therefore, Cropp [2002] recommends that a priority for the next generation of DMSP/DMS models is the inclusion of a realistic microbial loop. Similar conclusions were reported by other authors [Lefevre *et al.*, 2002; Le Clainche *et al.*, 2004; Vézina, 2004].

[7] Another significant problem with most of the current DMSP/DMS models is the difficulty of decoupling DMS dynamics from that of phytoplankton. It has been observed that DMS peaks in summer at tropical, subtropical and low temperate latitudes, a time when chlorophyll-a (CHL, a common proxy for phytoplankton biomass) is at its annual minimum [Simó and Pedrós-Alió, 1999; Uher *et al.*, 2000; Toole and Siegel, 2004; Vallina *et al.*, 2006; Vila-Costa *et al.*, 2008]. This finding has been dubbed the "DMS summer paradox" [Simó and Pedrós-Alió, 1999]. Several mechanisms have been proposed to explain it, such as a succession

in phytoplankton species composition towards high DMSPp producers, inhibition of bacterial DMS consumption by high UV, and a higher (direct) production of DMS from phytoplankton cells due to UV stress [Simó and Pedrós-Alió, 1999; Sunda et al., 2002; Toole and Siegel, 2004]. Most of these potential explanations have not as yet been tested in models. The model of Lefevre et al. [2002] and the parameterization used by Gabric et al. [2005] are the sole examples of including a variable sulfur to nitrogen (S:N) phytoplankton internal quota as a function of light in order to account for a shift in species composition and/or a change in phytoplankton physiological state. This allowed a higher degree of decoupling between DMS and CHL in these models. It was, however, shown by Le Clainche et al. [2004] for the Sargasso Sea (using the same model as Lefevre et al. [2002] but coupled to a dynamic turbulent scheme) that the seasonality of modeled DMS was lower than that of DMS observations and the summer maximum was underestimated.

[8] In this work we present a Dynamic Model of Oceanic Sulfur (DMOS) which is based on a modified version of the ecosystem model (N/C-cycles) developed by Anderson and Pondaven [2003] (hereafter AP'03). It has an explicit representation of bacteria and DOM dynamics, and is adapted for the Sargasso Sea using data collected during the Bermuda Atlantic Time-series Study (BATS) [Steinberg et al., 2001]. We have coupled a new S-cycle to it that incorporates the latest knowledge of DMSP/DMS dynamics. Model complexity was progressively increased in order to test several of the hypotheses generally used to explain the DMS summer paradox. The performance of the model in simulating annual cycles of concentrations, fluxes and turnover rates of the sulfur variables is analyzed.

2. Data and Methodology

2.1. Sargasso Sea Data

[9] The Sargasso Sea is located in the subtropical West North Atlantic and represents an oligotrophic open ocean region where the DMS summer paradox is readily observable [Dacey et al., 1998; Toole and Siegel, 2004]. Phytoplankton seasonality is regulated by physical processes which drive the deep nutrient entrainment in the upper layers during winter and spring, followed by nutrient depletion in summer due to a strong stratification of the water column [Goericke, 1998; DuRand et al., 2001; Steinberg et al., 2001]. The dominant phytoplankton groups are prokaryotic picophytoplankton (Prochlorococcus and Synechococcus) and eukaryotic phytoplankton (Prymnesiophytes, Pelagophytes) [Goericke, 1998; DuRand et al., 2001; Steinberg et al., 2001]. Diatoms are not a dominant group, although rare episodic blooms have been observed [Steinberg et al., 2001]. Dinoflagellates are also represented as a low percentage of the phytoplankton community [Goericke, 1998; Steinberg et al., 2001]. The mixed layer depth (MLD) has a marked seasonal cycle with values from ≈ 200 m to less than 10 m in summer [Steinberg et al., 2001; Spitz et al., 2001] and the sea surface temperature (SST) varies from $\approx 20^\circ$ in winter to $\approx 28^\circ$ in summer [Steinberg et al., 2001].

[10] During the Bermuda Atlantic Time-series Study, station BATS (31.75°N , 64.17°W) was sampled for verti-

cally resolved profiles of CHL and bacteria (along with many other physical and biological variables) approximately monthly from 1989. Data are available at the BATS database (<http://bats.bbsr.edu/>). Hydrostation-S (32.17°N , 64.50°W) has been sampled for vertically resolved profiles of DMSPp, DMSPd and DMS approximately biweekly from 1992 to 1994 [Dacey et al., 1998; Toole and Siegel, 2004]. Using these depth resolved time series we have constructed two-dimensional (2-D; time and depth) climatologies of CHL and bacteria (more than 10 years of data, from 1989 to 2000) as well as DMSPp, DMSPd and DMS (three years of data, from 1992 to 1994). The methodology used for building the climatology was as follows. All measured profiles were merged by month. Then, for each month, a 6th degree polynomial regression was used to fit the cloud of data, obtaining a single depth-resolved profile per month. Finally the resulting monthly profiles were interpolated in time, generating 2-D (time, depth) plots with a resolution of $1 \text{ day} \times 1 \text{ m}$. To be consistent, model results were also interpolated in depth and averaged in time to obtain the same $1 \text{ day} \times 1 \text{ m}$ resolution (see section 3. Results and Discussion). DMSPd did not display a clear seasonal pattern over the sampling period [Dacey et al., 1998] and therefore the obtained climatology has to be viewed with some caution. On the other hand, DMSPp and DMS showed a much clearer seasonal cycle.

2.2. Model Description

[11] The AP'03 ecosystem model includes a detailed characterization of the microbial loop. It incorporates a complex description of the DOM cycle and explicitly includes heterotrophic bacteria as a state variable. The treatment of DOM includes dual currencies, nitrogen (DON) and carbon (DOC). Since DMSP and DMS are part of the DOM, and therefore they share many of their processes (such as bacterial uptake and degradation), a DMSP/DMS model including a detailed microbial loop is fundamental [Vézina, 2004]. We thus coupled our S-cycle model to the AP'03 N/C-cycles, calling this coupled ecosystem-DMSP/DMS model "DMOS" (Dynamic Model of Oceanic Sulfur). The new S-cycle model contains important improvements like the explicit representation of bacterial activity in the sulfur cycle (including for the first time UV inhibition of bacterial sulfur uptake), a time-varying DMS exudation term from phytoplankton (due to UV stress) as well as an uptake of DMSPd for phytoplankton. In a manner similar to Archer et al., 2002, we include nonlinear kinetics for sulfur uptake (other models use linear relationships [Vézina, 2004]).

[12] One of the advantages of including bacteria explicitly in DMSP/DMS models is that it is then possible to evaluate the relative contributions of phytoplankton and bacteria to the DMS production, this being one of the important unanswered questions concerning the biogeochemistry of DMS [Yoch, 2002]. Further, the DMS-yield of the whole food web (total DMS production/total DMSP consumption), which is a very sensitive parameter in DMSP/DMS models [Lefevre et al., 2002; Vézina, 2004; Le Clainche et al., 2004; Cropp et al., 2004] is not prescribed as an a-priori parameter: it is now an output of the model (see section 3. Results and Discussion). The full set of DMOS equations is described in Appendix A. Model

Table 1. List of DMOS Parameters

Parameter	Symbol	Value	Unit	Source
Phyto. max. specific growth rate	μ_P^{\max}	3.7	$[d^{-1}]$	Spitz <i>et al.</i> [2001]
Phyto. saturating irradiance	I_s	60	$[W\ m^{-2}]$	Cropp <i>et al.</i> [2004]
Phyto. half-sat. for NO_3^- uptake	k_P^N	0.15	$[mmolN\ m^{-3}]$	Anderson and Pondaven [2003]
Phyto. half-sat. for NH_4^+ uptake	k_P^A	0.05	$[mmolN\ m^{-3}]$	Anderson and Pondaven [2003]
Phyto. NH_4^+ inhibition parameter	ψ	1.5	$[mmolN^{-1}]$	Anderson and Pondaven [2003]
Phyto. leakage fraction	γ_1	0.05	$[adim]$	Anderson and Pondaven [2003]
Phyto. DOC exudation parameter	γ_2	0.34	$[adim]$	Anderson and Pondaven [2003]
Phyto. specific mortality rate	m_P	0.045	$[d^{-1}]$	Anderson and Pondaven [2003]
Phyto. mortality losses to DOM	ε	0.34	$[adim]$	Anderson and Pondaven [2003]
Phyto. sinking rate	$ w_p^- $	0.05	$[m\ d^{-1}]$	This study
Phyto. C:N ratio	$\theta_{P:C:n}$	6.625	$[mmolC\ mmolN^{-1}]$	Anderson and Pondaven [2003]
Phyto. $CaCO_3$:C ratio	θ_{Ca}	0.10	$[mmolC\ mmolC^{-1}]$	Anderson and Pondaven [2003]
Phyto. max. CHL:C ratio	θ_{chl}^{\max}	0.041	$[mgCHL\ mgC^{-1}]$	Spitz <i>et al.</i> [2001]
Phyto. initial slope of P-I curve	α_{chl}	1.0	$[mgC\ mgCHL^{-1}\ (W\ m^{-2})^{-1}\ d^{-1}]$	Spitz <i>et al.</i> [2001]
Phyto. molecular weight of Carbon	C_{mw}	12	$[mgC\ mmolC^{-1}]$	Spitz <i>et al.</i> [2001]
Phyto. min. S:N internal quota	$\theta_{P:S:n}^{\min}$	0.044	$[mmolS\ mmolN^{-1}]$	Lefevre <i>et al.</i> [2002]
Phyto. max. S:N internal quota	$\theta_{P:S:n}^{\max}$	0.220	$[mmolS\ mmolN^{-1}]$	Lefevre <i>et al.</i> [2002]
Phyto. max. DMS specific exudation rate	γ_s^{\max}	0.25	$[d^{-1}]$	This study
Phyto. fraction of DMSPd consumers	α_P	0.1	$[adim]$	This study
Phyto. free DMSP-lyase activity	f	0.01	$[d^{-1}]$	This study
Zoo. max. specific ingestion rate	g	3.2	$[d^{-1}]$	This study
Zoo. N assim. efficiency	β_n	0.75	$[adim]$	Anderson and Pondaven [2003]
Zoo. C assim. efficiency	β_c	0.65	$[adim]$	Anderson and Pondaven [2003]
Zoo. C net growth efficiency	ω_Z	0.8	$[adim]$	Anderson and Pondaven [2003]
Zoo. half-sat. const. for N ingestion	k_g	0.75	$[mmolN\ m^{-3}]$	Anderson and Pondaven [2003]
Zoo. grazing preference upon Phyto.	p_P	1/3	$[adim]$	Anderson and Pondaven [2003]
Zoo. grazing preference upon Bact.	p_B	1/3	$[adim]$	Anderson and Pondaven [2003]
Zoo. grazing preference upon Det.	p_D	1/3	$[adim]$	Anderson and Pondaven [2003]
Zoo. C:N ratio	$\theta_{Z:C:n}$	5.5	$[mmolC\ mmolN^{-1}]$	Anderson and Pondaven [2003]
Zoo. messy feeding losses to DOM	ϕ	0.23	$[adim]$	Anderson and Pondaven [2003]
Zoo. max. specific mortality rate	m_Z	0.3	$[d^{-1}]$	Anderson and Pondaven [2003]
Zoo. half-sat. const. for mortality	k_Z	0.2	$[mmolN\ m^{-3}]$	Anderson and Pondaven [2003]
Zoo. mortality fraction going to DOM	Ω_{dom}	0.38	$[adim]$	Anderson and Pondaven [2003]
Zoo. mortality fraction going to NH_4^+	Ω_A	0.33	$[adim]$	Anderson and Pondaven [2003]
Zoo. mortality fraction going to Detritus-N	Ω_{Dn}	0.29	$[adim]$	Anderson and Pondaven [2003]
Zoo. mortality fraction going to Detritus-C	Ω_{Dc}	0.46	$[adim]$	Anderson and Pondaven [2003]
Zoo. mortality fraction going DIC	Ω_{DIC}	0.16	$[adim]$	Anderson and Pondaven [2003]
Zoo. DMSPp ingestion: fraction converted to DMSPd	α_1	0.7	$[adim]$	Simó [2004]
Bact. max. Lc/ NH_4^+ and sulfur uptake	μ_B^{\max}	13.3	$[d^{-1}]$	Anderson and Pondaven [2003]
Bact. max. Sc hydrolysis	μ_{Sc}	4	$[d^{-1}]$	Anderson and Pondaven [2003]
Bact. half-sat. const. for NH_4^+ uptake	k_A	0.5	$[mmolN\ m^{-3}]$	Anderson and Pondaven [2003]
Bact. half-sat. const. for Lc uptake	k_{Lc}	25	$[mmolC\ m^{-3}]$	Anderson and Pondaven [2003]
Bact. half-sat. const. for Sc hydrolysis	k_{Sc}	417	$[mmolC\ m^{-3}]$	Anderson and Pondaven [2003]
Bact. max. inhibition by UV of nutrient uptake	ϕ_{inhib}^{\max}	0.75	$[adim]$	This study
Bact. specific nitrification rate	v	0.03	$[d^{-1}]$	Anderson and Pondaven [2003]
Bact. C gross growth efficiency	ω_B	0.17	$[adim]$	Anderson and Pondaven [2003]
Bact. specific mortality rate	m_B	0.04	$[d^{-1}]$	Anderson and Pondaven [2003]
Bact. C:N ratio	$\theta_{B:C:n}$	5.1	$[mmolC\ mmolN^{-1}]$	Anderson and Pondaven [2003]
Bact. S:C ratio	$\theta_{B:S:c}$	1/250	$[mmolS\ mmolC^{-1}]$	del Valle <i>et al.</i> [2007]
Bact. fraction of DMS consumers	α_B	0.20	$[adim]$	This study
Bact. half-sat. const. for DMSPd uptake	k_{DMSPd}	0.01	$[mmolS\ m^{-3}]$	This study
Bact. half-sat. const. for DMS uptake	k_{DMS}	0.01	$[mmolS\ m^{-3}]$	This study
Bact. DMSPd excess uptake: fraction converted to DMS	α_2	0.1	$[adim]$	Kiene and Linn [2000]
Labile fraction of DOM produced	δ_1	0.7	$[adim]$	Anderson and Pondaven [2003]
Labile fraction of Phyto. extra-DOC exudation	δ_2	0.4	$[adim]$	Anderson and Pondaven [2003]
Detrital-N breakdown rate	m_{Dn}	0.055	$[d^{-1}]$	Anderson and Pondaven [2003]
Detrital-C breakdown rate	m_{Dc}	0.04	$[d^{-1}]$	Anderson and Pondaven [2003]
Detrital- $CaCO_3$ dissolution rate	m_{Dh}	0.05	$[d^{-1}]$	Anderson and Pondaven [2003]
Detrital sinking rate	$ w_{Di}^- $	0.05	$[m\ d^{-1}]$	This study
Irradiance max.	I_{\max}	150	$[W\ m^{-2}]$	This study
Irradiance min.	I_{\min}	45	$[W\ m^{-2}]$	This study
Irradiance threshold	I^*	25	$[W\ m^{-2}]$	Lefevre <i>et al.</i> [2002]
Irradiance attenuation due to water	k_w	0.04	$[m^{-1}]$	Popova <i>et al.</i> [2002]
Irradiance attenuation due to Phyto. self-shedding	k_p	0.03	$[m^2\ mmolN^{-1}]$	Popova <i>et al.</i> [2002]
Max. specific DMS photolysis rate	k_{photo}^{\max}	0.15	$[d^{-1}]$	Bailey <i>et al.</i> [2008]
Max. turbulent diffusion	kz_{\max}	250	$[m^2\ d^{-1}]$	Cropp <i>et al.</i> [2004]
Min. turbulent diffusion	kz_{\min}	1	$[m^2\ d^{-1}]$	Cropp <i>et al.</i> [2004]
Max. sea temperature (changes each day)	st_{\max}	SST	$[^{\circ}C]$	This study
Min. sea temperature	st_{\min}	19	$[^{\circ}C]$	This study
Steepness of the pycnocline	r	-20	$[adim]$	Cropp <i>et al.</i> [2004]

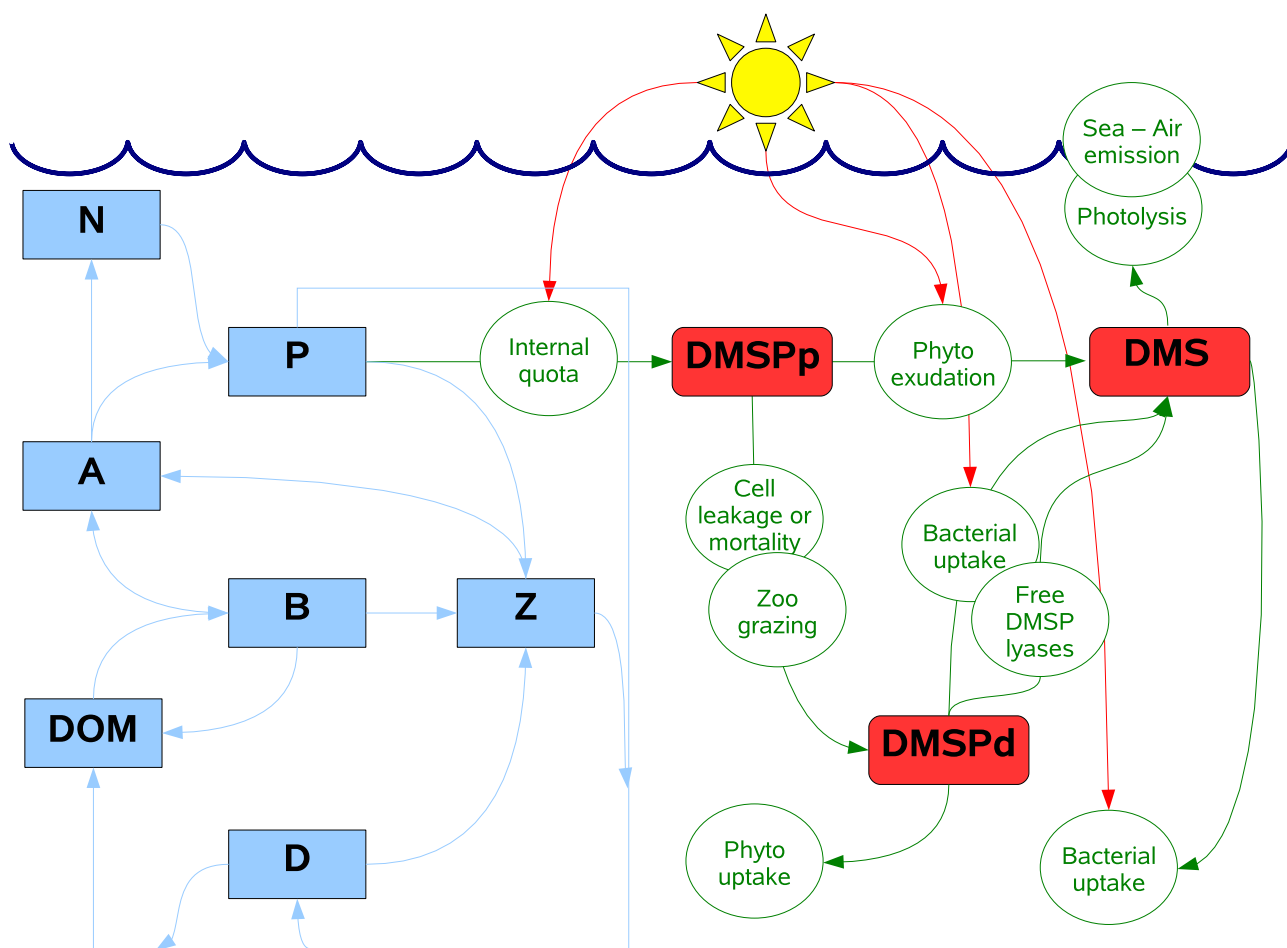


Figure 1. Schematic diagram of DMOS model. (left) Ecosystem submodel: (N/C-cycles; mmol m^{-3}) A, ammonium; N, nitrates; P, phytoplankton; B, bacteria; Z, zooplankton; DOM, dissolved organic matter (can be either nitrogen based or carbon based, and labile or semilabile); D, detritus (can be either nitrogen based or carbon based). Note that the modeled cycling of dissolved organic matter and detritus has been purposely simplified in this diagram as a generic DOM and D pools for clarity; a more detailed scheme of the ecosystem submodel can be found in the work of *Anderson and Pondaven* [2003]. (right) DMSP/DMS submodel (S-cycle; mmol m^{-3}): DMSPp, particulated dimethylsulfoniopropionate; DMSPd, dissolved dimethylsulfoniopropionate; DMS, dimethylsulfide. The red lines coming from the Sun refer to the S-cycle processes directly affected by solar radiation in DMOS model that has been tested by the five modeling experiments performed.

parameters are listed in Table 1. An schematic diagram of DMOS model is shown in Figure 1.

2.3. N/C-Cycles

[13] The model contains single state variables for phytoplankton (equation (A1)), zooplankton (equation (A2)) and heterotrophic bacteria (equation (A3)), two nutrient pools (nitrate (equation (A4)) and ammonium (equation (A5))), labile and semilabile DON and DOC (equations (A6 – A9)), detritus (equations (A10–A12)), dissolved inorganic carbon (DIC, equation (A13)) and alkalinity (equation (A14)). CHL (equation (A15)) is calculated for phytoplankton N at each time step based upon *Geider et al.* [1997] as in *Spitz et al.* [2001], permitting comparison with field data. Phytoplankton primary production (equation (A19)) is controlled by light (equation (A22)) [*Walsh et al.*, 2001] and temperature

(equation (A21)) [*Eppley*, 1972], affecting the specific growth rate (equation (A20)), as well as by nutrient availability (equation (A25)) [*Spitz et al.*, 2001]. Phytoplankton losses are due to zooplankton grazing (equation (A29)), natural mortality (equation (A55)) and vertical sinking (equation (A80)). Zooplankton graze upon phytoplankton, bacteria and soft detritus (equations (A29–A32)). Zooplankton production (equation (A37) or equation (A39)), ammonium excretion (equation (A38) or equation (A40)), and respiration (equation (A41)) are calculated according to a stoichiometric model [*Anderson and Hessen*, 1995; *Anderson and Pondaven*, 2003]. Zooplankton mortality is assumed to occur in the form of a quadratic Michaelis-Menten equation (equation (A56)). This is a classical way of parameterizing both natural mortality and grazing by higher predators (which are not explicitly modeled). Bacteria production, excretion and respiration (equations (A47–

A52)) are calculated from elemental stoichiometry [Anderson, 1992; Anderson and Williams, 1998; Anderson and Pondaven, 2003]. Labile DOC and DON are the primary growth substrates, with ammonium supplementing DON when the ratio DOC/DON (C:N ratio of DOM) is high. Uptake of labile DOC and DON and the maximum potential uptake of ammonium are described in equations (A43)–(A45). Bacteria either take up or regenerate ammonium at any one time depending on the availability of DOC and DON (equation (A47)), an upper limit of ammonium uptake being given by equation (A44). The fraction of DOC taken up not used for balanced (C:N) growth is respired (equation (A49) or equation (A52)). Bacteria loss terms are zooplankton grazing (equation (A30)) and natural mortality (equation (A57)).

[14] The main sink for nutrients is phytoplankton uptake (equation (A19)), ammonium also being lost to the nitrate pool via nitrification at constant rate (see second term on equation (A5)). The DOM pools are produced by phytoplankton leakage, excretion and exudation, zooplankton messy feeding, phytoplankton and bacterial natural mortality, and noncarbonate detrital breakdown (equations (A58–A59)). The semilabile DOM pool is converted to labile DOM due to the action of exoenzymes by bacterial (equations (A53–A54)). Phytoplankton exudation of DOC is directly proportional to primary production (see equation (A28)). Noncarbonate (or soft) detritus (equations (A10–A11)) arises from zooplankton egestion as well as phytoplankton and zooplankton mortality, and is lost by zooplankton grazing (equations (A31–A32)), breakdown, and vertical sinking (equation (A80)). The carbonate (or hard) detritus (equation (A12)) originates from the contribution of carbonate-forming (e.g. organisms such as coccolithophores to primary production). This is parameterized assuming a constant CaCO_3 :C ratio for phytoplankton (see Table 1). The carbonate fraction of total detritus is variable (for soft tissue and carbonate production, and also as additional carbon fixed as DOC), and returned by zooplankton and bacteria respiration as well as zooplankton mortality. Other return pathways, such as breakdown of carbonate detritus (equation (A60)), occur via cycling of DOC. Exchange of CO_2 with the atmosphere can also be estimated (F_{atm} term in equation (A13)) although it is not necessary for our purposes. Parameterization of alkalinity (equation (A14)) is performed according to the stoichiometry described by Broecker and Peng [1982].

2.4. S-Cycle

[15] DMSPp (equation (A16)) production by phytoplankton is modeled by using a sulfur/nitrogen (S:N) internal quota ($\theta_{P_{S:n}}$ parameter, see Table 1). $\theta_{P_{S:n}}$ is allowed to vary as function of light intensity following Lefevre et al. [2002] (equations (A63–A64)). Since the model has only one generic phytoplankton group, this method permits an implicit simulation of a shift in species composition towards high DMSPp producers in summer and/or a shift in phytoplankton physiological state [Lefevre et al., 2002], one of the proposed explanations of the DMS summer paradox. DMSPp is released to the water as DMSPd (equation (A17)) due to phytoplankton leakage and natural mortality as well by zooplankton grazing. It is assumed that 70% of

the grazed DMSPp is recovered in the dissolved phase [Levasseur et al., 2004; Simó, 2004]. The DMSPd losses are bacterial (equation (A68)) and phytoplankton uptake (equation (A67)) as well as cleavage to DMS by free DMSP-lyases (as in Archer et al. [2002]). Bacterial uptake of DMSPd is a well known sink for DMSPd since this compound is a major source of reduced sulfur for apparently most of marine bacteria [Kiene and Linn, 2000; Yoch, 2002; Zubkov et al., 2002; Vila et al., 2004; Vila-Costa et al., 2007]. A close seasonal correlation between DMSP assimilation by bacteria and bacterial heterotrophic production (measured as leucine incorporation) has been observed recently in a Mediterranean coastal site [Vila-Costa, 2006]. We therefore assumed that DMSPd consumption is proportional to the total bacterial community in the model. On the other hand, various phytoplankton take up DMSPd such as diatoms, Synechococcus and Prochlorococcus, these all being low or non DMSP producers [Vila-Costa et al., 2006b; Malmstrom et al., 2004]. With current uncertainties regarding what fraction of total phytoplankton biomass is able to take up DMSPd and at what rates, we considered that this process is carried out by 10% of the phytoplankton (see parameter α_P in Table 1).

[16] In the model, DMS (equation (A18)) production has 3 sources: cleavage from DMSPd by bacteria (equations (A71–A72)) and free DMSP-lyases (3rd term in equation (A18)) as well as direct exudation by phytoplankton (equation (A65)). The total amount of sulfur required by bacteria for balanced (C:S) growth (also called bacterial sulfur demand) is given by equation (A70). If the DMSPd taken up is in excess of bacterial sulfur demand (no S-limitation), a fraction (α_2 , see Table 1) of this sulfur excess is cleaved to DMS (equation (A71)), the remainder being converted to other forms of sulfur (e.g. sulfates via the methanethiol pathway) [Kiene, 1996; Kiene and Linn, 2000; Kiene et al., 2000]. It has been observed experimentally that bacterial production of methanethiol dominates over DMS production [Kiene, 1996; Kiene and Linn, 2000; Zubkov et al., 2002]. Bacterial DMS-yield rarely goes beyond 10%. Recent works found a range between 2 and 12% (Slezak, personal communication). Similar values were reported by Kiene and Linn [2000] and [Zubkov et al., [2002] (6–12%). Therefore α_2 is assumed to be small (10%) [Kiene and Linn, 2000; Niki et al., 2000]. On the other hand, if the DMSPd taken up by bacteria is lower than the requirement for balanced (C:S) growth (S-limitation), DMS is not produced (equation (A72)) because sulfur is fixed exclusively into proteins.

[17] Phytoplankton direct exudation of DMS has been assumed to be constant and very small in some models [e.g., Gabric et al., 1993; Chu et al., 2004], with many models not even including it as a process. Recent field research have however suggested that phytoplankton is likely an important source of DMS, mainly under high UV stress [Toole and Siegel, 2004; Toole et al., 2006; Vila-Costa et al., 2008]. In support of these findings, the work of Sunda et al. [2002] showed increases up to 3500% in the amount of DMS per unit cell volume in phytoplankton cultures exposed to high doses of UV-A. They proposed that DMS acts as an efficient hydroxyl radical scavenger, i.e. as an intracellular antioxidant under conditions of high UV exposure. In the model, therefore, direct exudation of DMS was made

a function of light intensity (first fraction in equation (A66)), although the level of phytoplankton activity is also taken into account (second fraction in equation (A66)).

[18] There is only one biological loss for DMS, namely its uptake by bacteria (equation (A69)). The complete phylogeny of marine DMS-consuming bacteria is not known, but a recent study has shown that the use of DMS as a C source seems mostly restricted to some methylotrophic bacteria [Vila-Costa *et al.*, 2006a], yet DMS consumption as a source of energy by unknown bacteria (conversion to DMSO without use of the C) might be more common [del Valle *et al.*, 2007]. Therefore DMS consumption seems to be not as widespread a process among bacterioplankton as DMSPd utilisation [Vila-Costa *et al.*, 2006a]. Thus, we assumed that only a fraction (20%, see parameter α_B in Table 1) of the generic pool of modeled bacteria acts as a sink for DMS. The other two sinks for DMS in the model are photolysis and emission to the atmosphere. Photolysis is assumed to be solely a function of light intensity (equation (A73)) although in reality it is a process mediated by chromophoric DOM (or CDOM) [Brimblecombe and Shooter, 1986; Brugger *et al.*, 1998; Toole *et al.*, 2003], which is not modeled in the current version of DMOS. DMS emission to the atmosphere (equation (A75)) is parameterized with the gas transfer model of Nightingale *et al.* [2000] (equations (A76)–(A78)) using climatological (1992–1994) surface wind speed (U , m s^{-1}) from the NCEP/NCAR Reanalysis Project (provided by the NOAA-CIRES Climate Diagnostics Center). Monthly data were interpolated in time and smoothed to generate daily values.

[19] Regarding the bacterial uptake of nutrients (either labile DON/DOC or NH_4^+) and sulfur (either DMSPd or DMS), the model includes a light inhibition parameter that influences the specific rate of bacterial uptake (equation (A45)). This parameter accounts for the well known effect of UVR upon bacterial heterotrophic activity and bacterial DMSP/DMS consumption [Herndl *et al.*, 1993; Slezak *et al.*, 2001; Toole *et al.*, 2006]. We assume a maximum inhibition of bacterial uptake of 75% (see $\phi_{\text{inhib}}^{\text{max}}$ in Table 1) [Slezak *et al.*, 2001; Toole *et al.*, 2006].

2.5. Physical Frame and Forcings

[20] The biogeochemical model is embedded in a one-dimensional (1-D) vertical physical frame. The model therefore neglects horizontal transport processes and takes into account only vertical processes, i.e. advection (equation (A80)) and diffusion (equation (A81)), which are considered the main driving forces of ecosystem dynamics in the upper ocean [Eigenheer *et al.*, 1996; Denman and Peña, 1999]. As in the models of Lefevre *et al.* [2002] and Cropp *et al.* [2004], vertical mixing in the current version of DMOS is parameterized using a prescribed turbulent diffusion coefficient (k_z , Table 1) following the approach used by Cropp *et al.* [2004]. Coefficient k_z is generated using a sigmoid equation (equation (A82)) and climatological MLD data [Levitus, 1982]. As a result the maximum diffusion ($k_{z_{\text{max}}}$) occurs in the upper mixed layer (UML) and the minimum diffusion ($k_{z_{\text{min}}}$) occurs below the UML. In between, k_z decreases from $k_{z_{\text{max}}}$ to $k_{z_{\text{min}}}$ as dictated by the parameter r (Table 1) which defines the steepness of the pycnocline [Cropp *et al.*, 2004]. The same sigmoid function (equation (A82)) is used to generate the

vertical temperature profiles, the only difference being that the maximum value for temperature is the sea surface temperature (SST) which varies seasonally, instead of being constant as for diffusion. SST data for the Sargasso Sea were obtained from a climatology (1971–2000, NOAA-CIRES Climate Diagnostics Center). Monthly data were interpolated in time and smoothed to generate daily values.

[21] Light in the model (I_z , W m^{-2}) is defined as daily averaged photosynthetic available radiation (PAR). Daily values of PAR at the surface (I_0) of the Sargasso Sea were obtained after interpolating in time and smoothing a SeaWiFS climatology (from years 2002 to 2004). Light decays with depth (z , in metres) following an exponential function (equation (A23)) that depends on water and phytoplankton. In the current version of DMOS we wanted to explore if the observed DMS seasonality could be simulated through the inclusion of light-driving processes, with the assumption that UVR is the forcing behind these processes (e.g. bacterial inhibition, phytoplankton stress), yet bacterial inhibition of sulfur uptakes has been also described to occur under PAR [Slezak *et al.*, 2001]. PAR seasonality can be used as a proxy for UVR seasonality, UVR being a constant fraction of PAR. Given that all the parameterizations used to account for UVR-driven processes are based on the term $\frac{I_z}{I_{\text{max}}}$, the constant fraction cancels out and we are just left with a nondimensional term representing UVR that varies between 0 and 1. While UVR attenuates faster in the water column than PAR, the UVR driving processes affecting DMS production may operate at higher depths. It has been described that organisms need some time for recovering after being exposed to high UVR doses [Toole *et al.*, 2006]. Therefore when they escape from the UV zone (i.e. by sinking and/or turbulent diffusion) they may keep a “memory” of the stress deeper in the water column. Nevertheless, exploring the use of an explicit wavelength-resolved UVR formulation, with the inclusion of CDOM as a state variable, is desirable and will be object of future research.

[22] The model domain is from 0 to 200 m with a vertical resolution of 2.5 m. Initial conditions (in mmol m^{-3}) are constant profiles for all variables: 0.1 for phytoplankton, zooplankton and bacteria; 2.0 for nitrates; 0.5 for ammonium and labile DON; 24 for labile DOC; 2100 for TIC; 2375 for alkalinity; 0.16 for chlorophyll-a; 0.013 for DMSPp; 0.1 for DMSPd and DMS; and zero for the remaining variables. The boundary conditions are zero-flux in order to conserve mass (with the exception of sulfur since DMS emission in the upper top grid is allowed). A mass-conservative ecosystem model is desirable so that the biotic pools do not eventually run out of nutrients [Spitz *et al.*, 2001; Cropp *et al.*, 2004]. An accumulation of dying phytoplankton and detritus in the bottom boundary due to vertical sinking occurs, which implies that DOM increases and finally that NH_4^+ and NO_3^- (from NH_4^+ nitrification) also accumulates. This simulates the observed presence of high NO_3^- levels deep in the water column at BATS [Steinberg *et al.*, 2001]. During winter, the MLD reaches the bottom and the strong mixing carry some of this nutrients pool to the UML, generating the vernal phytoplankton bloom (see section 3. Results and Discussion). In order to reach an equilibrium state, the model was ran for 10 years (with a time step of 0.005 days) prior to the analysis of the results. Previous tests

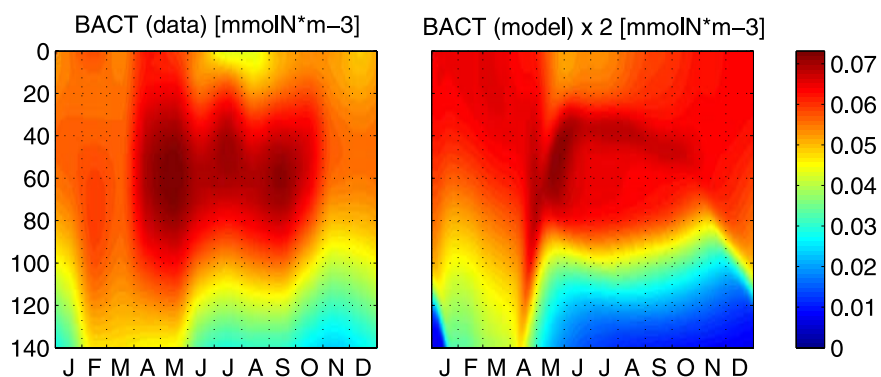


Figure 2. Two-dimensional (time, depth) plots of bacteria (mmolN m^{-3}) from climatological in situ data (left) and DMOS model results (right). Conversion from bacterial counts (10^8 cells kg^{-1} , BATS data) to mmolN m^{-3} was done using a conversion factor of 0.0118 ($\text{mmolN m}^{-3}/10^8$ cells kg^{-1}), which was obtained assuming that bacterial cells have 7.2 fgC cell^{-1} [Gundersen *et al.*, 2002] and a C:N molar ratio of 5.1 (Table 1). Note that simulated bacteria are scaled up by a factor of 2 for the sake of visual comparison against data.

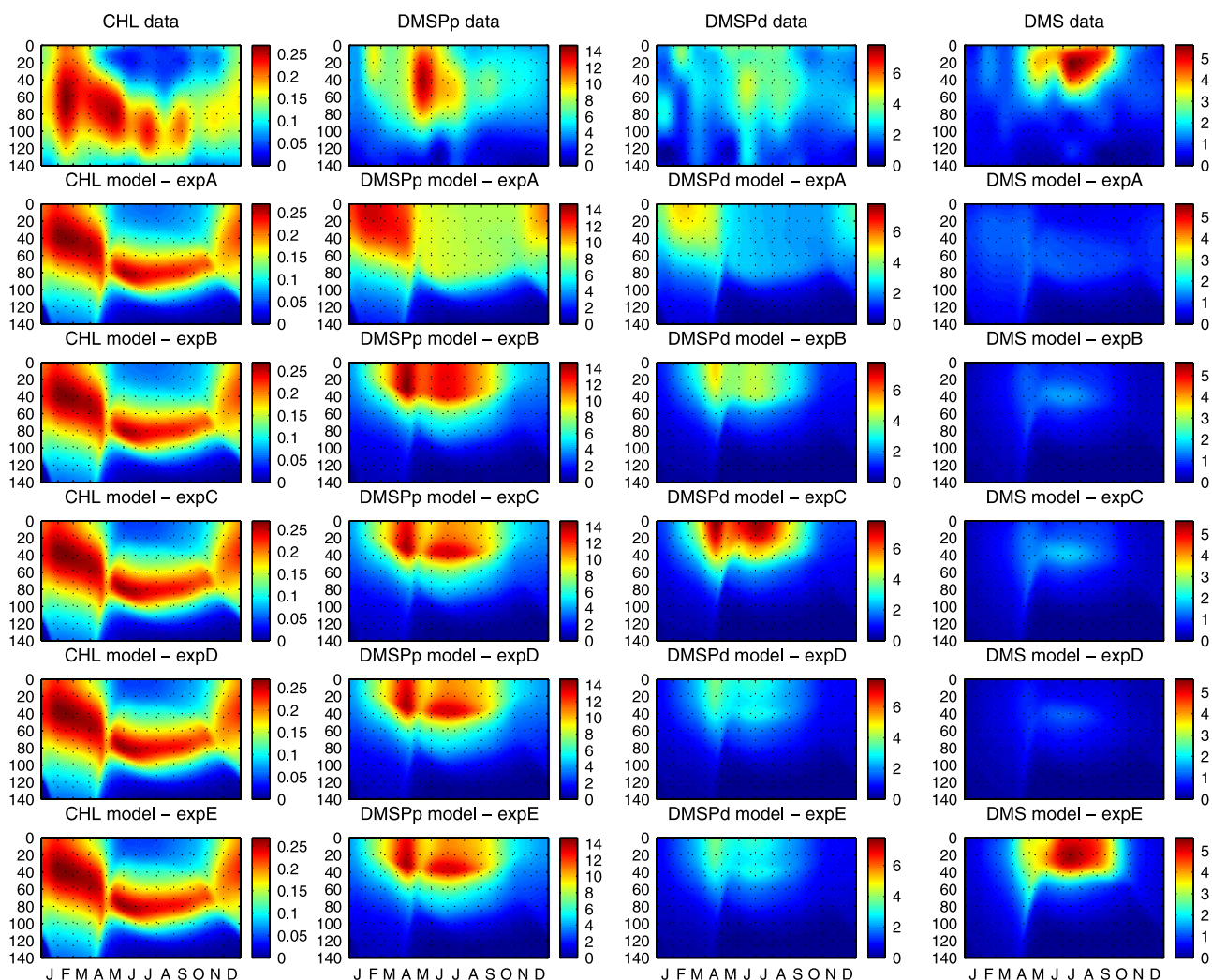


Figure 3. Two-dimensional (time, depth) plots of chlorophyll-a, DMSP particulate, DMSP dissolved and DMS from climatological in situ data (upper-row panels) and DMOS model results for several modeling experiments (see Table 2): experiment A (second-row panels), experiment B (third-row panels), experiment C (fourth-row panels), experiment D (fifth-row panels), experiment E (sixth-row panels). Units: chlorophyll-a (mg m^{-3}), DMSPp-DMSPd-DMS ($\mu\text{molS m}^{-3}$).

Table 2. Model Experiments

Process Affecting S-Cycle	ExpA	ExpB	ExpC	ExpD	ExpE
Phyto. S:N ratio shift by UV:	NO	YES	YES	YES	YES
Bact. inhibition by UV:	NO	NO	YES	YES	YES
Phyto. DMSPd consumption:	NO	NO	NO	YES	YES
Phyto. DMS production by UV:	NO	NO	NO	NO	YES

without seasonal forcings showed that the model reaches a “stable node” equilibrium, indicating that the model does not have unwanted internal dynamics (e.g. oscillations) and that the seasonal changes are driven by the seasonal forcings.

3. Results and Discussion

[23] The model successfully reproduces bacteria (Figure 2) and CHL (Figure 3) distributions, capturing the winter/spring phytoplankton bloom in surface and the deep chlorophyll maximum (DCM) during summer months (Figure 3) as well as the development of a subsurface maximum (40–60 m depth) of bacteria from late spring to early fall (Figure 2). Modeled values of bacteria concentrations are, however, about a half those of the data. This is because the model only simulates active bacteria, while the data includes both active and nonactive bacteria [Anderson and Pondaven, 2003]. The vernal phytoplankton bloom is triggered by the entrainment of deep nutrients. In contrast, nutrients are depleted in summer leading to lowest phytoplankton biomass in surface waters. Deeper in the water column, the presence of higher concentrations of nutrients along with light in sufficient quantity for primary production results in the formation of a DCM. Bacteria distributions mainly result from the interplay of DOM release by phytoplankton and the inhibition of bacterial DOM uptake by high solar radiation doses during summer.

3.1. Model Experiments

[24] In order to gain some insight into the processes that are most relevant for explaining the observed seasonality of DMS in the Sargasso Sea, and therefore what drives the DMS summer paradox, we conducted several model experiments turning “off” or “on” various DMOS sulfur paths. This exercise was undertaken in a sequence of steps, starting from the simplest characterization of the S-cycle and increasing complexity in a stepwise fashion until realistic simulations were obtained. The first scenario excluded all of the processes usually cited in the literature to explain the DMS summer paradox, namely shift in the S:N ratio of phytoplankton, inhibition of bacterial uptake by UV light, phytoplankton uptake of DMSPd, and phytoplankton exudation of DMS under UV stress. Each of these processes was then added one after the other (see Table 2). Simulations for each of the experiments (A, B, C, D, E; Table 2) are compared with observations for the Sargasso Sea observations (0–140 m) in Figure 3.

3.1.1. Experiment A

[25] In this experiment there is no seasonal increase in the S:N ratio of phytoplankton (therefore $\theta_{P:S:n}$ is constant and equal to $0.13 \text{ mmolS mmolN}^{-1}$, the middle value between $\theta_{P:S:n}^{\min}$ and $\theta_{P:S:n}^{\max}$), UV induces neither bacterial inhibition of semilabile-DOM/ NH_4^+ and sulfur uptake (DMSPd and

DMS) nor phytoplankton stress-driven DMS production, and phytoplankton does not take up DMSPd. Modeled DMSPp, DMSPd and DMS closely follow the predicted CHL distribution, all displaying maximum values in winter/spring and minima in summer/fall (Figure 3). The main differences between DMSPp and CHL are due to the variability of CHL as a response of the levels of light intensity. This constancy between sulfur species and CHL is not observed in the data. Further, DMSPp maximum values are slightly underestimated while DMS values are highly underestimated. We must conclude that this experiment is not capturing at all the main processes controlling oceanic sulfur dynamics.

3.1.2. Experiment B

[26] In contrast to the previous experiment, a variable S:N internal quota in phytoplankton was now added in order to parameterize a seasonal change in species composition towards high DMSPp producers and/or a change in phytoplankton physiological state due to higher UV doses [Sunda *et al.*, 2002; Slezak and Herndl, 2003]. Results are shown in Figure 3. The simulations for DMSPp are improved in comparison to the first experiment. However, the modeled DMSPp maximum in spring takes place earlier than observed (by about one month, similar to what was observed by [Le Clainche *et al.*, 2004]) and DMSPd distributions correlate too closely with DMSPp, a feature not observed in the field [Dacey *et al.*, 1998]. DMSPp simulations are also clearly overestimated during summer. Therefore, the parameterization of the S:N ratio as a function of light, although better than using a constant value, is far from being perfect. There is a need to explore other ways of modeling DMSPp concentrations, e.g., by including in the model several phytoplankton groups with specific S:N internal quotas. On the other hand, DMS values are again severely underestimated and, although a summer maximum is now predicted, it is deeper in the water column and smaller in magnitude than seen in the observations. Furthermore, predicted DMS in surface waters does not display the summer maximum seen in the observations, but rather shows a spring maximum. It therefore appears that inclusion of a variable the S:N ratio of phytoplankton in the model can not on its own account for the observed seasonality of DMS nor explain the DMS summer paradox.

3.1.3. Experiment C

[27] Next, the inhibition of bacterial uptake of nutrients (semilabile-DOM/ NH_4^+) and sulfur (DMSPd and DMS) by solar radiation was added to the model [Herndl *et al.*, 1993; Slezak *et al.*, 2001; Toole *et al.*, 2006]. This fact has been also cited as a potential explanation for the DMS summer paradox since a reduction in a major sink may cause DMS to accumulate [Simó and Pedrós-Alió, 1999; Simó, 2001, 2004]. Model results (see Figure 3) indicate that inhibition of bacterial sulfur uptake by UV could partly explain the

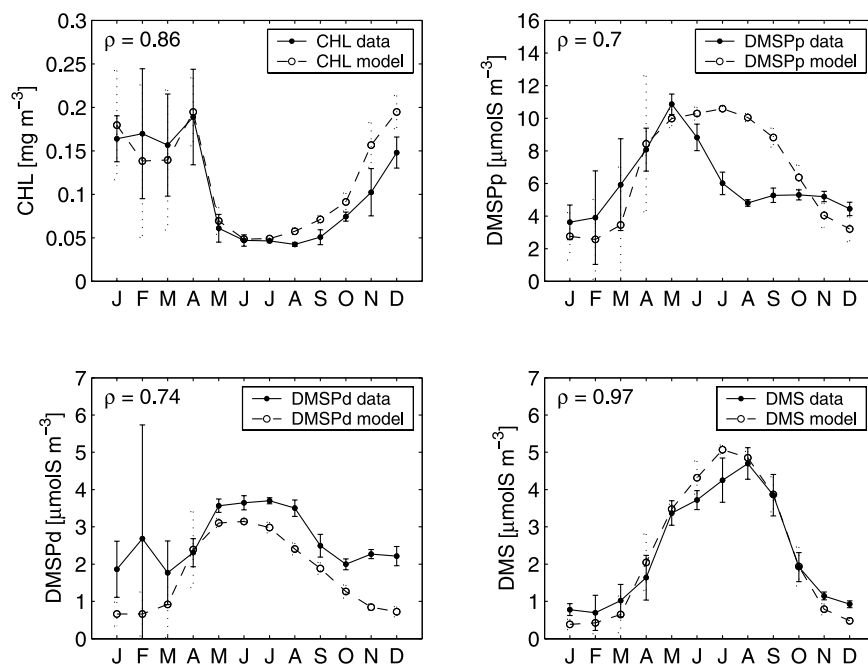


Figure 4. Monthly upper mixed-layer averaged values of chlorophyll-a, DMSP particulate, DMSP dissolved and DMS for climatological in situ data (solid line) and DMOS model results (dashed line) for Experiment E. Units: chlorophyll-a (mg m^{-3}), DMSPp-DMSPd-DMS ($\mu\text{molS m}^{-3}$).

cause of the deep summer maximum of DMS, although the predicted maximum is slightly deeper than in the observations and DMS values remain underestimated. However, DMS accumulation occurs in conjunction with an unrealistically large accumulation of DMSPd. This overestimate in modeled DMSPd may be due to the absence in the model of phytoplankton uptake, a new sink for DMSPd that has been recently discovered experimentally [Vila-Costa *et al.*, 2006b]. The DMSPd accumulation implies a large contribution to DMS from free DMSP-lyases (1% of the DMSPd pool is converted into DMS each day). We therefore have repeated experiments B and C, but without free DMSP-lyases activity to evaluate if inhibition of bacterial sulfur uptake by UV can produce this DMS accumulation. Results (not shown) displayed a very weak increase of the deep DMS maximum in summer (much weaker than the increase observed from experiment B to experiment C, see Figure 3), not enough to be the origin of the DMS summer paradox. This result is a consequence of the fact that, although bacterial DMS uptake is reduced by the UV-induced inhibition (which tends to increase DMS), there is also a reduction of bacterial DMS production (from the associated inhibition of DMSPd uptake). The net balance of these two opposite effects is almost zero.

3.1.4. Experiment D

[28] The next process added to the model was consumption of DMSPd by phytoplankton. It has been reported recently that some groups of phytoplankton (diatoms and cyanobacteria) are able to take up DMSPd [Vila-Costa *et al.*, 2006b]. This process may help explain the low values of DMSPd observed in the field and the strong decoupling of DMSPd to either DMSPp or DMS [Dacey *et al.*, 1998]. The resulting simulations (see Figure 3) show an improvement in predicted DMSPd concentrations, although spring values

are still overestimated relative to the observations. On the other hand, modeled DMS is totally unsatisfactory, with values clearly underestimated and showing a very weak deep summer maximum.

3.1.5. Experiment E

[29] The next addition to the model was a direct exudation term of DMS from phytoplankton cells as a response to UV-induced stress [Sunda *et al.*, 2002; Toole and Siegel, 2004]. Direct DMS production by phytoplankton has been reported previously in the literature [Vairavamurthy *et al.*, 1985; Niki *et al.*, 2000, and references therein; Wolfe *et al.*, 2002]. This DMS production, therefore, is not routed through the DMSPd pool but comes directly from the DMSPp (in the model it is assumed that the amount of DMS exuded from the cells is immediately replaced by newly produced DMSPp). The results of this experiment, which incorporates all the mechanisms thought to be important in the sulfur cycle of the ocean, show a clear improvement over those of the previous experiments and also of existing DMSP/DMS models. Simulated DMS now shows good agreement with the observations (see Figure 3), with a summer maximum of about $5 \mu\text{molS m}^{-3}$ at around 20 m depth and winter minima of about $0.5 \mu\text{molS m}^{-3}$. The predicted summer DMS maximum occurs below the surface because of the high DMS photolysis and ventilation rates occurring in the upper layers. The results suggest that phytoplankton DMS exudation may be an important factor contributing to the high summer to winter ratio of DMS concentrations observed in the field as well as explaining the strong decoupling between CHL and DMS.

3.2. Further Analysis of Experiment E

[30] Results from the series of experiments described above indicate that the various mechanisms involved in

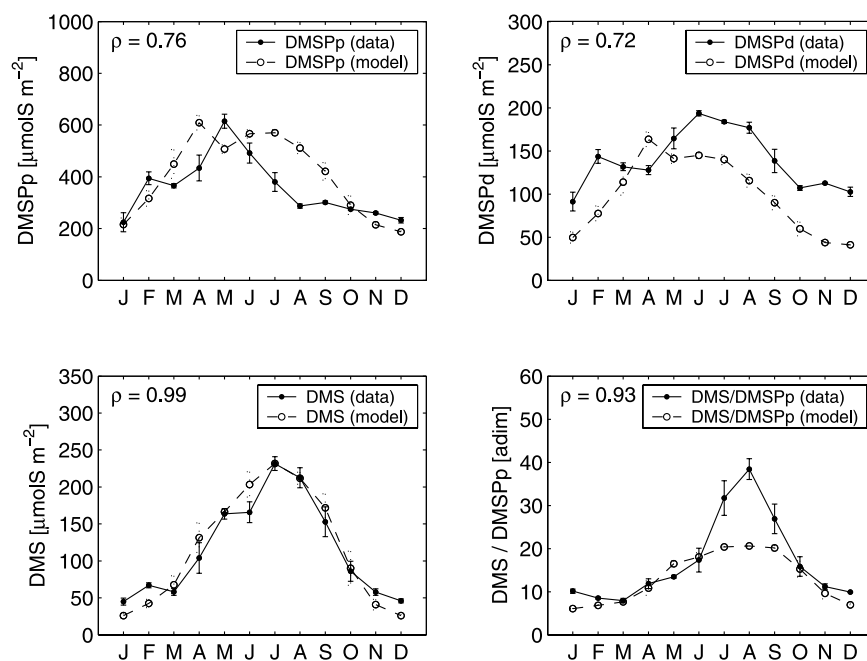


Figure 5. Monthly upper 50 m-integrated values of DMSP particulate, DMSP dissolved, DMS and DMSPp/DMS ratio for climatological in situ data (solid line) and DMOS model results (dashed line) for Experiment E. Units: DMSPp-DMSPd-DMS ($\mu\text{molS m}^{-2}$).

the S-cycle (seasonal variations in the internal S:N quota, bacterial inhibition of sulfur uptake, exudation of DMS as well as DMSPd uptake by phytoplankton) all can play a role in driving the DMS seasonality. The direct exudation of DMS from phytoplankton cells (as a response of high UV doses) does however appear to be a major one: without this process the model is unable to realistically reproduce the observed DMS cycle and thereby explain the summer paradox.

[31] Monthly UML averages of CHL, DMSPp, DMSPd and DMS (data and model) are shown in Figure 4, along with their associated (Spearman) correlation coefficients. In general, the model is able to realistically capture both the seasonality and the absolute magnitude of these four variables. The main discrepancies between the model and the observations are that model summer DMSPp values are clearly overestimated (by as much as a factor of two in August) and that DMSPd values are slightly underestimated in the model. In this regard, however, experimentalists believe that past and current DMSPd measurements are likely to be overestimates due to filtration artifacts [Kiene and Slezak, 2006].

[32] Examination of the 50-m depth integrated values for DMSPp, DMSPd, DMS and the DMS:DMSPp ratio (Figure 5) leads to similar conclusions. The model does a reasonably good job at reproducing the sulfur variables, although the predicted DMSPp maximum occurs one month too early and predicted DMSPp concentrations are generally too high. On the other hand, DMSPd values are underestimated in summer. Both modeled DMS and data display maxima in July, showing good agreement both in magnitude and seasonality. Due to the predictions of DMSPp being too high in August, the modeled DMS/DMSPp ratio is, however, markedly underestimated for this month.

[33] The vertically resolved seasonal cycles of predicted sulfur fluxes in the model are shown in Figure 6. Bacterial uptake of DMSPd (Figure 6a) is highest in spring and summer (e.g., more than $\approx 1.5 \mu\text{molS m}^{-3} \text{d}^{-1}$ occurred at depths between 20 m and 60 m) coincident with high modeled bacterial biomass (see Figure 2) and DMSPd concentrations (see Figure 3). Uptake rates decline thereafter until an annual minimum is reached in winter. Phytoplankton uptake of DMSPd (Figure 6b) displays a similar seasonality, with an annual maximum in spring ($\approx 1.5 \mu\text{molS m}^{-3} \text{d}^{-1}$ in April) and a secondary maximum during summer ($\approx 1.0 \mu\text{molS m}^{-3} \text{d}^{-1}$) for surface waters (<30 m). Bacterial uptake of DMSPd is in general higher than that of phytoplankton, and has a broader and deeper distribution (values higher than $1 \mu\text{molS m}^{-3} \text{d}^{-1}$ reach depths of about 80 m while phytoplankton uptake does not occur deeper than 40 m).

[34] Bacterial uptake of DMS (Figure 6c) is highest in summer at depths between 20 m and 40 m, reaching values of $0.5 \mu\text{molS m}^{-3} \text{d}^{-1}$. In July 2004, *del Valle et al.* [2007] reported a maximum DMS consumption rate of $\approx 0.8 \mu\text{molS m}^{-3} \text{d}^{-1}$ around 40 m depth. In surface waters in summer, bacterial uptake rates of both DMSPd and DMS (Figures 6a and 6c) are reduced by almost a half due to UV-induced bacterial inhibition. Minimum bacterial DMS uptake occurs during winter due to the very low DMS concentrations. Relatively low bacterial DMS consumption rates ($0.3 \mu\text{molS m}^{-3} \text{d}^{-1}$) were measured in the Sargasso Sea in May at the DCM by *Levasseur et al.* [2004]. Similar rates ($0.2\text{--}0.3 \mu\text{molS m}^{-3} \text{d}^{-1}$, see Figure 6c) are predicted in the model in May between 60–80 m (the depth of modeled DCM, see Figure 3).

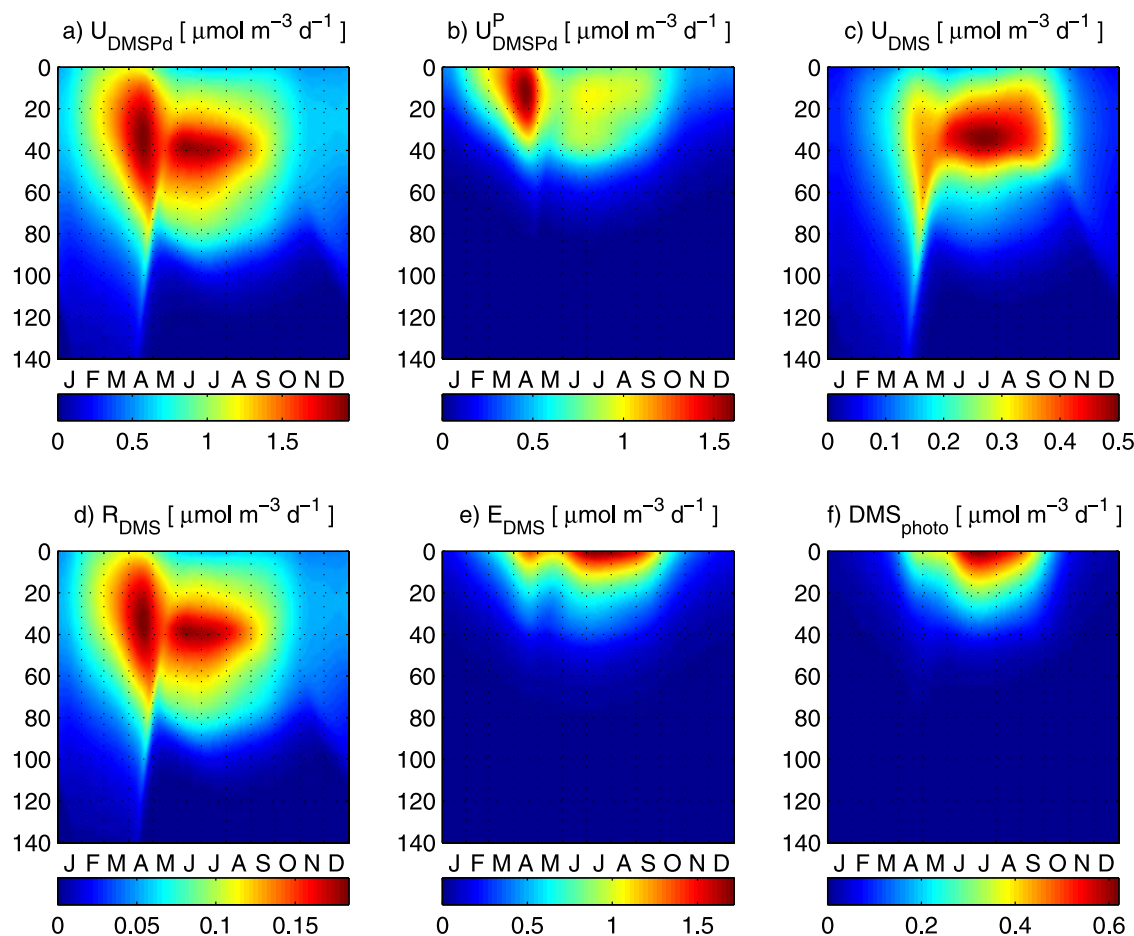


Figure 6. Two-dimensional (time, depth) plots of DMOS model sulfur fluxes ($\mu\text{molS m}^{-3} \text{d}^{-1}$) for Experiment E: (a) Bacterial uptake of DMSP dissolved. (b) Phytoplankton uptake of DMSP dissolved. (c) Bacterial uptake of DMS. (d) Bacterial production of DMS. (e) Phytoplankton production of DMS. (f) Photolysis of DMS.

[35] The predicted bacterial production of DMS by enzymatic cleavage of DMSPd (Figure 6d) follows the DMSPd uptake (Figure 6a) and represents approximately 8–9% of the former, in agreement with estimates for the Northern Sea (6–12% [see Zubkov *et al.*, 2002]). Maximum values (higher than $\approx 0.15 \mu\text{molS m}^{-3} \text{d}^{-1}$) therefore are obtained in spring and summer at depths between 20 m and 60 m. On the other hand, DMS production from phytoplankton exudation (Figure 6e) displays its highest values in summer at the very surface, reaching rates of more than $1.5 \mu\text{molS m}^{-3} \text{d}^{-1}$ in July. Since this process is light-driven, the values increase exponentially towards the surface. The sum of phytoplankton exudation and bacterial DMS production gives a higher gross biological production in summer (up to $1.8 \mu\text{molS m}^{-3} \text{d}^{-1}$, not shown) in the UML. In spring, the modeled gross biological production of DMS at the surface is about $1.4\text{--}1.6 \mu\text{molS m}^{-3} \text{d}^{-1}$, a value in good agreement with in situ data from the Sargasso Sea ($1.8 \mu\text{molS m}^{-3} \text{d}^{-1}$, [Levasseur *et al.*, 2004]). In a separate short-term study, where a mass balance assumption was applied to measured rates, light-mediated biological DMS production rates as high as $8 \mu\text{molS m}^{-3} \text{d}^{-1}$ were estimated by Toole *et al.* [2006]. There is such a scarcity

of in situ measurements addressing this issue that future field work is needed. DMS photolysis (Figure 6f) also reaches a maximum in summer (up to $0.6 \mu\text{molS m}^{-3} \text{d}^{-1}$ in July) as a consequence of maximum DMS concentrations and light exposure. Toole *et al.* [2003] estimated maximum UML-integrated DMS photolysis rates in summer of about $10\text{--}15 \mu\text{molS m}^{-2} \text{d}^{-1}$. These estimates were based on a summer MLD of ≈ 20 m. For this MLD value, our integrated DMS photolysis rates in summer are very similar ($10\text{--}12 \mu\text{molS m}^{-2} \text{d}^{-1}$).

[36] Model sulfur fluxes averaged over the UML are shown in Figure 7. Bacterial and phytoplankton uptake of DMSPd are similar in magnitude (Figures 7a and 7b, respectively) as are bacterial DMS consumption and DMS photolysis (Figures 7c and 7f). On the other hand, predicted phytoplankton DMS exudation can be an order of magnitude higher (e.g. in summer) than bacterial production (Figures 7e and 7d, respectively).

[37] Turnover rates of DMSPd and DMS due to bacterial consumption, phytoplankton DMSPd uptake, and DMS photolysis are shown in Figure 8. Turnover rate (or rate constant, d^{-1}) is the process rate ($\mu\text{molS m}^{-3} \text{d}^{-1}$) divided by the concentration ($\mu\text{molS m}^{-3}$). Maximum turnover rates, thus lower turnover times ($1/\text{turnover rates}$, d), due

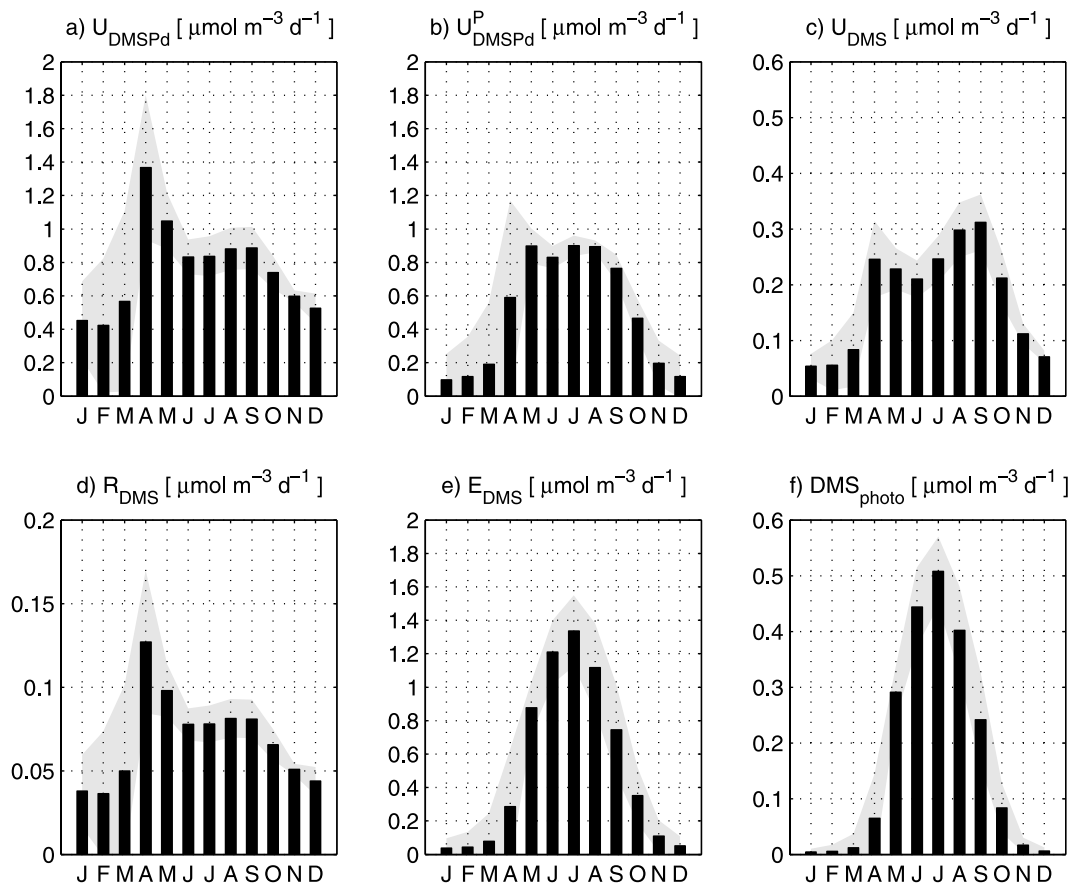


Figure 7. Monthly upper mixed-layer averaged values of DMOS model sulfur fluxes ($\mu\text{mol S m}^{-3} \text{d}^{-1}$) for Experiment E: (a) Bacterial uptake of DMSP dissolved. (b) Phytoplankton uptake of DMSP dissolved. (c) Bacterial uptake of DMS. (d) Bacterial production of DMS. (e) Phytoplankton production of DMS. (f) Photolysis of DMS. The gray shadow area represents the standard deviation of the averages.

to bacterial uptake are obtained at depth (between 20–60 m in winter and 60–80 m in summer) both for DMSPd and DMS (Figures 8a and 8c, respectively). Bacterial inhibition of sulfur uptake by UV at the surface is observed in summer. Turnover rates of DMS are about 20% of those for DMSPd because only this percentage of bacteria was assumed to be DMS consumers (see section 2.2. Model Description). Thus, minimum turnover times for bacterial consumption of DMSPd and DMS are estimated as ≈ 1.5 and ≈ 6 days, respectively. Bacterial turnover rates of DMSPd and DMS have also been observed to be highest in subsurface waters of the North Sea during a coccolithophore bloom (although they were nevertheless about an order of magnitude higher than our results) [Zubkov *et al.*, 2001, 2002]. For the Sargasso Sea, DMSPd and DMS turnover times have been estimated to vary between 0.4–2.8 days and <0.5 –8 days respectively [Ledyard and Dacey, 1996].

[38] Predicted rates of phytoplankton DMSPd turnover (Figure 8b) are of the same order of magnitude (yet slightly lower) than those of bacterial uptake (Figure 8a). Their vertical distribution is however very different because phytoplankton uptake depends on primary production and hence is only significant in irradiated waters (<40 m). Maximum values are present in spring due to the highest primary production. Total DMSPd turnover rates (bacterial

and phytoplankton uptakes plus free DMSP-lyases activity) for the upper 50 m are in the range of 0.5 – 1.0 d^{-1} (not shown), in good agreement with the values reported by Ledyard and Dacey [1996] for the Sargasso Sea (0.7 – 1.5 d^{-1} , see Table 2 in Kiene and Linn [2000]). Maximum turnover rates of DMS photolysis (Figure 8d) are about 75% of the maximum rates of bacterial consumption (Figure 8c). Highest photolysis rates are, however, mainly concentrated in the upper 20 m in summer, a layer and period where bacterial consumption is low. Photolysis then dominates the loss of DMS in the upper ocean, while bacterial consumption dominates at greater depths [Toole *et al.*, 2006; Kieber *et al.*, 1996].

[39] Turnover rates averaged for the UML are shown in Figure 9. UML DMS turnover rates are between 0.05 and 0.15 d^{-1} . For a 60 m mixed water column in the equatorial Pacific, Kieber *et al.* [1996] obtained bacterial DMS turnover rates of 0.04 – 0.66 d^{-1} . In the Sargasso Sea, bacterial DMS consumption rates in September were estimated to be $\approx 0.5 \mu\text{mol S m}^{-3} \text{d}^{-1}$ [Lefevre *et al.*, 2002, and references therein]. For that month UML averaged DMS concentration from the data is about $4.0 \mu\text{mol S m}^{-3}$ (see Figure 4), giving a specific rate of 0.12 d^{-1} , similar in magnitude to the model results ($\approx 0.08 \text{ d}^{-1}$, see September in Figure 9c). An interesting feature emerging from the model results is that

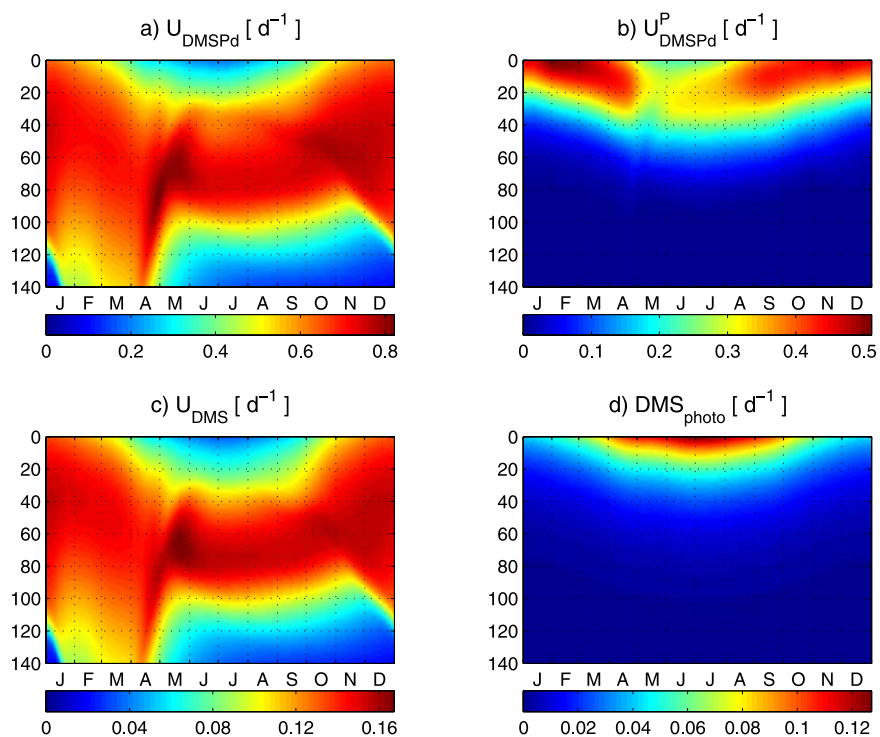


Figure 8. Two-dimensional (time, depth) plots of DMOS model sulfur turnover rates (d⁻¹) for Experiment E: (a) Bacterial uptake of DMSP dissolved. (b) Phytoplankton uptake of DMSP dissolved. (c) Bacterial uptake of DMS. (d) Photolysis of DMS.

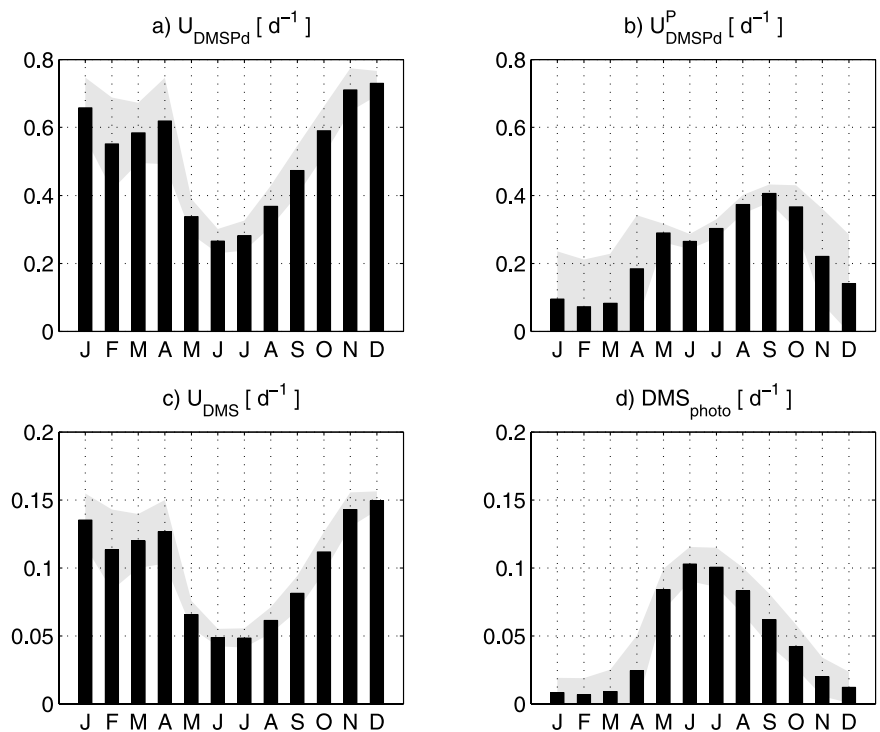


Figure 9. Monthly upper mixed-layer averaged values of DMOS model sulfur turnover rates (d⁻¹) for Experiment E: (a) Bacterial uptake of DMSP dissolved. (b) Phytoplankton uptake of DMSP dissolved. (c) Bacterial uptake of DMS. (d) Photolysis of DMS. The gray shadow area represents the standard deviation of the averages.

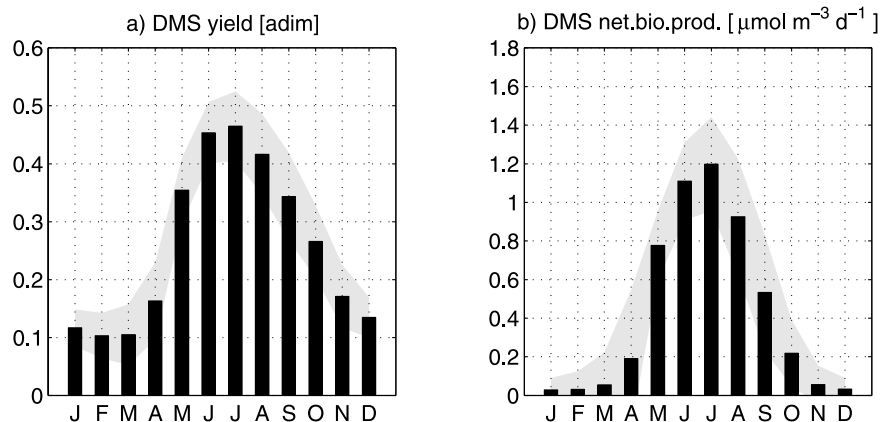


Figure 10. Monthly upper mixed-layer averaged values of: (a) DMS yield (DMS production/DMSP consumption). (b) Net biological DMS production (DMS production - bacterial uptake of DMS). The gray shadow area represents the standard deviation of the averages.

bacterial DMS consumption is the dominant loss term in winter and spring (Figure 9c), while in summer it is DMS photolysis (Figure 9d), as postulated previously [Toole *et al.*, 2006].

[40] We calculated the DMS-yield (equation (A79)) for the whole food web (total DMS production divided by total DMSP consumption) from the model sulfur fluxes, as well as the net biological production of DMS (total DMS production minus bacterial DMS consumption) averaged over the UML (see Figure 10). The DMS-yield (Figure 10a) displays a clear seasonal pattern with higher values during summer (up to $\approx 45\%$) and lower values in winter ($\approx 10\%$). For both variables the annual maximum is observed in July. These values are in the range reported in the literature [Simó and Pedrós-Alió, 1999]. Net biological production of DMS (Figure 10b) follows the same seasonality, increasing from almost zero to $\approx 1.2 \mu\text{molS m}^{-3} \text{d}^{-1}$ in summer. Very similar values and seasonality of net biological production of DMS were estimated by Toole and Siegel [2004], based on the same set of data, when using a mass-balance model (from zero in winter to $1.0\text{--}1.5 \mu\text{molS m}^{-3} \text{d}^{-1}$ in summer).

[41] Since DMS fluxes to the atmosphere are believed to have a role in regulating climate [Charlson *et al.*, 1987; Andreae and Crutzen, 1997], we plotted them along with UML DMS emission rates (DMS flux divided by the MLD) and surface DMS concentrations (upper top cell grid) (see Figure 11). The highest values of the three variables occur in summer (June, July, August). In particular, DMS flux to the atmosphere shows a clear maximum in July ($\approx 12 \mu\text{molS m}^{-2} \text{d}^{-1}$). The UML is exposed to very high doses of solar radiation during this month (because of the high solar incident radiation and very shallow MLD). If the proposed antioxidant function of DMS [Sunda *et al.*, 2002] as well as the impact of DMS on CCN and Earth albedo are shown to be important, then this higher DMS flux to the atmosphere in summer could act as a negative feedback between the ocean's ecosystems and the amount of solar radiation reaching the ocean's surface [Charlson *et al.*, 1987; Vallina and Simó, 2007; Vallina *et al.*, 2007]. Maximum DMS emissions from the UML (June/July; $\approx 1.0 \mu\text{molS m}^{-3} \text{d}^{-1}$) are about a factor of two and four higher than the summer DMS losses

by photolysis or bacterial consumption respectively (≈ 0.5 and $\approx 0.25 \mu\text{molS m}^{-3} \text{d}^{-1}$; see Figures 7c and 7f).

3.3. Sensitivity Analysis

[42] In order to evaluate the sensitivity of model results to model parameters values we carried out a sensitivity analysis (SA) by increasing/decreasing each parameter by 50% in each run and comparing the results to the control simulation (experiment E, Table 1 with the exception of parameters δ_1 , α_1 and ϕ_{inhib}^{\max} that were lowered to 0.5 in the reference control to allow an increase of 50%). The SA index was taken from Le Clainche *et al.* [2004] and is defined as follows:

$$S_k = \frac{X_{k_{\max}} - X_{k_{\min}}}{X_{k_{\text{control}}}} * 100 \quad (1)$$

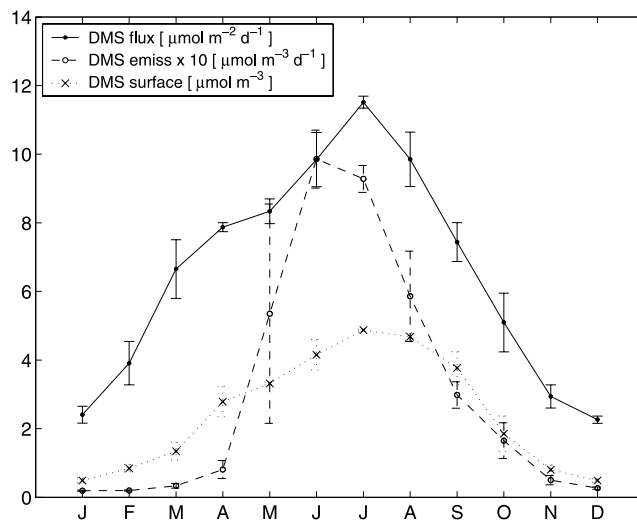


Figure 11. Monthly averaged values and standard deviation of surface DMS ($\mu\text{molS m}^{-3}$, dotted line), DMS ventilation flux ($\mu\text{molS m}^{-2} \text{d}^{-1}$, solid line), upper mixed layer DMS ventilation (defined as DMS flux/MLD) ($10^* \mu\text{molS m}^{-3} \text{d}^{-1}$, dashed line).

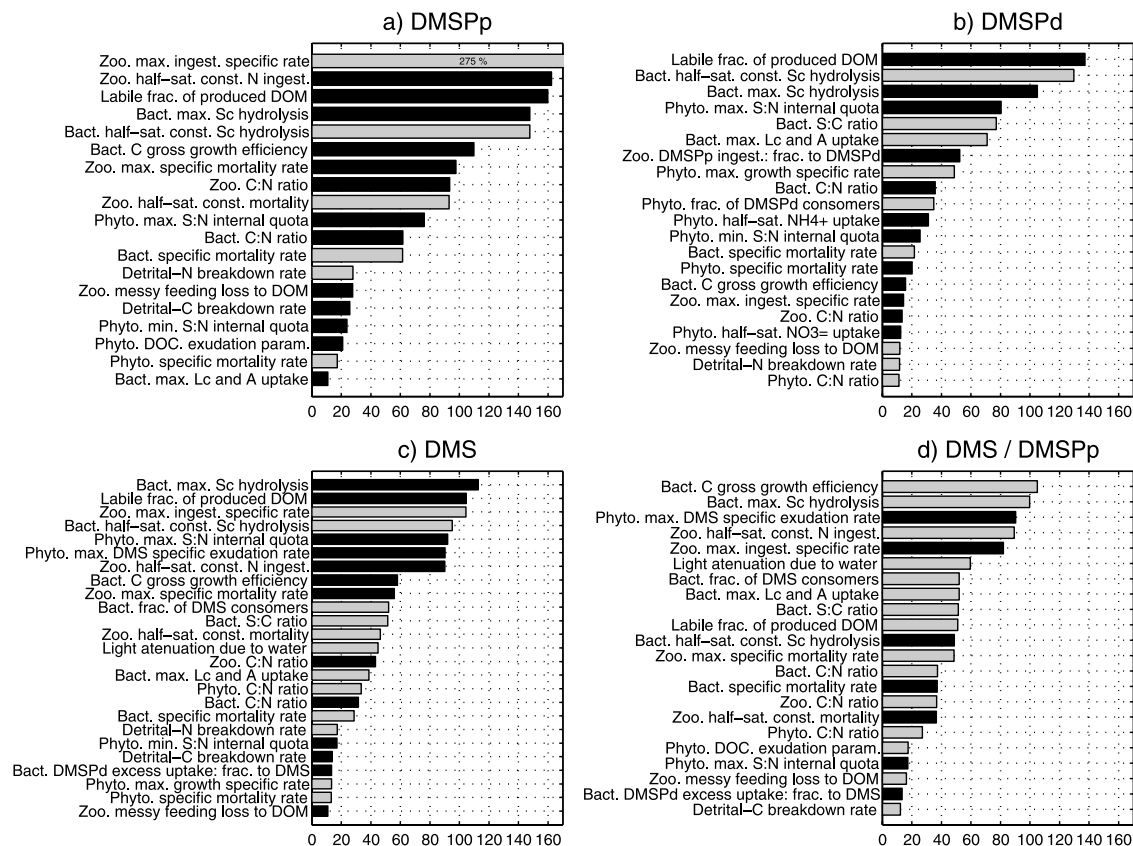


Figure 12. Sensitivity analysis indices of the DMOS model to changing parameters by $\pm 50\%$. Black bars indicate that an increase of the parameter causes an increase of the variable, while grey bars indicate that an increase of the parameter causes a decrease of the variable.

where $X_{kcontrol}$, X_{kmax} and X_{kmin} are the annual budgets of DMSPp, DMSPd and DMS integrated over the upper 50m obtained for the 3 simulations: control (reference value of the parameter k), 50% increase in the parameter k ($k_{max} = 1.5k$), and 50% decrease in the parameter k ($k_{min} = 0.5k$) [Le Clainche et al., 2004]. Only those parameters which gave SA indices greater than 10% are plotted in Figure 12. Black bars indicate that an increase of the parameter results in an increase in the state variable, while grey bars indicate that an increase in the parameter produced a decrease in the state variable.

[43] For DMSPp (Figure 12a) we observe that parameters related to zooplankton grazing have the largest effects, mainly the maximum zooplankton specific ingestion rate. This is not surprising since they directly affect the phytoplankton biomass. Another important parameter is the labile fraction of DOM produced. Other SA tests (not shown) revealed that this parameter is a key one for the nutrient pools due to the bacterial microbial loop; an increase of DOM concentrations is associated to a rise of NO_3^- and NH_4^+ in the model, and thus to an increase of modeled phytoplankton. Bacterial hydrolysis of semilabile DOM and gross growth efficiency are also very important because they affect the amount of labile DOM (and then again the nutrient pools and phytoplankton biomass). Parameters related to zooplankton mortality are associated to DMSPp through the levels of grazing activity upon phytoplankton, while the zooplankton C:N ratio affects the DOC pool (and again the microbial loop). Interestingly, the maximum S:N

internal quota of phytoplankton is not the most important parameter contrary to the results of Lefevre et al. [2002] and Le Clainche et al. [2004]. This difference is in part attributable to the fact that in the SA carried out by these authors they increased/reduced by 50% the minimum and maximum S:N internal ratios at the same time, while we have increased/reduced them separately. Nevertheless, these results suggest that bottom-up (nutrient availability, regulated by the microbial loop) and top-down (zooplankton grazing) processes that control phytoplankton biomass are more important for DMSPp concentrations than the internal sulfur quota.

[44] For DMSPd (Figure 12b), the 10 most sensitive parameters are related to the microbial loop, except for the maximum phytoplankton internal S:N quota (4th position), the bacterial S:C ratio (5th position), the fraction of ingested DMSPp by zoo that is recovered as DMSPd (7th) and the phytoplankton maximum specific growth rate (8th).

[45] Regarding DMS (Figure 12c), we observe a set of seven parameters that have a consistently high influence (from $\approx 90\%$ to from $\approx 110\%$), clearly larger than the rest ($< 60\%$). With the exception of the maximum phytoplankton S:N internal quota (5th position) and the maximum phytoplankton DMS exudation specific rate (6th position), all were ranked already in the top five most sensitive parameters for DMSPp (Figure 12a). They are related to the bottom-up and top-down processes controlling phytoplankton biomass previously cited. The parameter for the fraction

of bacteria that is DMS consumers is ranked on the 10th position, which is significant since at present this is an uncertain parameter. It is followed by the bacterial S:C ratio. A higher S:C ratio implies higher sulfur requirements of bacteria (lower DMS production from DMSPd uptake) at the same time than higher DMS uptake. There is large variation in published values for the S:C molar ratio ranging from $\approx 1/50$ to $\approx 1/250$ [Zubkov *et al.*, 2002, and references therein]. Further research is needed in order to better constrain this parameter. As expected, the light attenuation coefficient is also quite important (13th position) because any increase reduces the amount of DMS exudation by phytoplankton. On the other hand, increasing it also reduces the DMS photolysis. However, since DMS exudation by phytoplankton was about 3 times higher than DMS photolysis (see Figure 6), the reduction of the source term dominates.

[46] In order to evaluate which parameters affect DMS concentrations in a way not directly related to changes in phytoplankton biomass (i.e., changes in DMSPp) the SA index was also calculated for the ratio DMS/DMSPp (Figure 12d). As expected the DMS/DMSPp ratio is very sensitive to the maximum phytoplankton DMS exudation rate. Significant increases of the DMS/DMSPp ratio are also observed for higher zooplankton grazing rates. On the other hand, parameters that increase bacteria concentrations (e.g., carbon gross growth efficiency, maximum DOM/NH₄⁺ uptake, the labile fraction of DOM produced, the phytoplankton DOC exudation parameter, etc.) are associated with a decrease in the ratio. An increase of the light attenuation coefficient also produces a decrease of the DMS/DMSPp ratio.

4. Conclusions

[47] We have presented a state-of-the-art model of the oceanic sulfur cycle which includes the various processes currently thought to be important in DMSP/DMS dynamics and which resolves explicitly DOM and bacteria dynamics within the ecosystem. Sensitivity analyses have shown that parameters related to the microbial loop have a great impact on the N/C/S-cycles, in agreement with previous conclusions reached by both modeling and experimental studies [Spitz *et al.*, 2001; Simó *et al.*, 2000]. The model is able to reproduce the seasonal DMS summer paradox observed in the Sargasso Sea and highlights that bacterial consumption of DMSPd to give DMS may not be the main process in the overall DMS budget. Field studies have also shown that DMS concentrations were not controlled by DMSPd uptake by bacteria [Dacey *et al.*, 1998; Zubkov *et al.*, 2002]. Rather it seems that the key process determining DMS concentrations in the upper ocean may be direct exudation from phytoplankton cells under high UV conditions, this providing an explanation for the strong seasonal decoupling observed between chlorophyll-a and DMS over most of the ocean's surface [Vallina *et al.*, 2006; Vallina and Simó, 2007]. This mechanism is missing in all current models of the sulfur cycle, with the exception of the one presented here.

[48] Our model results suggest that DMS production by phytoplankton, despite being only one among several processes that are relevant (such as light-induced increases in the S:N ratio of phytoplankton and light-induced inhibition of bacterial sulfur uptake), is a major contributor to the

DMS summer paradox. This has been previously proposed from the analysis of field data in the Sargasso Sea [Toole and Siegel, 2004]. The fact that phytoplankton can directly produce DMS has been previously reported in the literature [Vairavamurthy *et al.*, 1985; Niki *et al.*, 2000, and references therein; Wolfe *et al.*, 2002] and there is increasing experimental evidence in support of this claim [Toole *et al.*, 2006]. The implication is that changes in UV levels due to shoaling of the MLD, as might occur in Global Warming scenarios, could have an impact on the oceanic DMS production, and therefore on its potential effect upon Earth climate through CCN formation. Global DMSP/DMS models should incorporate the light-mediated processes affecting DMS production included in DMOS if better estimates of surface DMS concentrations, and specially of its seasonality, are to be achieved.

Appendix A

A1. Model Equations

[49] Phytoplankton [mmolN m⁻³] equation

$$\frac{\partial P}{\partial t} = (1 - \gamma_1)F_P - G_P - M_P - S(P) + D(P) \quad (A1)$$

Zooplankton [mmolN m⁻³] equation

$$\frac{\partial Z}{\partial t} = F_Z - M_Z + D(Z) \quad (A2)$$

Bacteria [mmolN m⁻³] equation

$$\frac{\partial B}{\partial t} = F_B - G_B - M_B + D(B) \quad (A3)$$

Nitrates [mmolN m⁻³] equation

$$\frac{\partial N}{\partial t} = -F_P^N + \nu A + D(N) \quad (A4)$$

Ammonium [mmolN m⁻³] equation

$$\frac{\partial A}{\partial t} = -F_P^A - \nu A + E_B + E_Z + \Omega_A M_Z + D(Z) \quad (A5)$$

Labile DON [mmolN m⁻³] equation

$$\begin{aligned} \frac{\partial L_n}{\partial t} = & \gamma_1 F_P + \delta_1 [\phi G_n + \varepsilon M_P + M_B + M_{D_n} + \Omega_{dom} M_Z] + U_{S_n} \\ & - U_{L_n} + D(L_n) \end{aligned} \quad (A6)$$

Labile DOC [mmolC m⁻³] equation

$$\begin{aligned} \frac{\partial L_c}{\partial t} = & \gamma_1 \theta_{P_{cn}} F_P + \gamma_1 E_{doc} + \delta_2 (1 - \gamma_1) E_{doc} \\ & + \delta_1 [\phi G_c + \varepsilon \theta_{P_{cn}} M_P + \theta_{B_{cn}} M_B + M_{D_c} + \Omega_{dom} \theta_{Z_{cn}} M_Z] \\ & + U_{S_c} - \theta_{B_{cn}} F_B - R_B + D(L_c) \end{aligned} \quad (A7)$$

Semilabile DON [mmolN m⁻³] equation

$$\frac{\partial S_n}{\partial t} = (1 - \delta_1)[\phi G_n + \varepsilon M_P + M_B + M_{D_n} + \Omega_{dom} M_Z] - U_{S_n} + D(S_n) \quad (\text{A8})$$

Semilabile DOC [mmolC m⁻³] equation

$$\frac{\partial S_c}{\partial t} = (1 - \delta_2)(1 - \gamma_1)E_{doc} + (1 - \delta_1) \cdot [\phi G_c + \varepsilon \theta_{P_{cn}} M_P + \theta_{B_{cn}} M_B + M_{D_c} + \Omega_{dom} \theta_{Z_{cn}} M_Z] - U_{S_c} + D(S_c) \quad (\text{A9})$$

Detrital nitrogen [mmolN m⁻³] equation

$$\frac{\partial D_n}{\partial t} = (1 - \beta_n)(1 - \phi)G_n + (1 - \varepsilon)M_P + \Omega_{D_n} M_Z - G_{D_n} - M_{D_n} - S(D_n) + D(D_n) \quad (\text{A10})$$

Detrital carbon [mmolC m⁻³] equation

$$\frac{\partial D_c}{\partial t} = (1 - \beta_c)(1 - \phi)G_c + (1 - \varepsilon)\theta_{P_{cn}} M_P + \Omega_{D_c} \theta_{Z_{cn}} M_Z - G_{D_c} - M_{D_c} - S(D_c) + D(D_c) \quad (\text{A11})$$

Detrital CaCO₃ [mmolC m⁻³] equation

$$\frac{\partial D_h}{\partial t} = \theta_{Ca} \theta_{P_{cn}} (G_P + M_P) - M_{D_h} - S(D_h) + D(D_h) \quad (\text{A12})$$

Dissolved inorganic carbon [mmolC m⁻³] equation

$$\frac{\partial DIC}{\partial t} = -(1 + \theta_{Ca})\theta_{P_{cn}} F_P - E_{doc} + R_B + R_Z + M_{D_h} + \Omega_{DIC} \theta_{Z_{cn}} M_Z + F_{atm} + D(DIC) \quad (\text{A13})$$

Alkalinity [mmolC m⁻³] equation

$$\frac{\partial ALK}{\partial t} = (Q_N - Q_A)F_P - 2\theta_{Ca} \theta_{P_{cn}} F_P + E_B + E_Z + 2M_{D_h} + \Omega_A M_Z - \nu A + D(ALK) \quad (\text{A14})$$

CHL [mg m⁻³] equation

$$\frac{\partial CHL}{\partial t} = (\rho_{chl} C_{mw} \theta_{P_{cn}}) F_P - (G_P + M_P)(CHL/P) - S(CHL) + D(CHL) \quad (\text{A15})$$

DMSPp [mmolS m⁻³] equation

$$\frac{\partial DMSPp}{\partial t} = \theta_{P_{sn}} \frac{\partial P}{\partial t} = \theta_{P_{sn}} [(1 - \gamma_1)F_P - G_P - M_P - S(P) + D(P)] \quad (\text{A16})$$

DMSPd [mmolS m⁻³] equation

$$\frac{\partial DMSPd}{\partial t} = \theta_{P_{sn}} [\gamma_1 F_P + \alpha_1 G_P + M_P] - U_{DMSPd} - U_{DMSPd}^P - fDMSPd + D(DMSPd) \quad (\text{A17})$$

DMS [mmolS m⁻³] equation

$$\frac{\partial DMS}{\partial t} = R_{DMS} + E_{DMS} + fDMSPd - U_{DMS} - DMS_{photo} - DMS_{emiss} (*) + D(DMS) \quad (\text{A18})$$

[50] (*) This term is applied only to the top model cell.

A2. N/C-Cycles Model Terms[51] Phytoplankton production (F_P)

$$F_P = F_P^N + F_P^A = JQ_N P + JQ_A P = JQP \quad (\text{A19})$$

$$J = \mu_P R \quad (\text{A20})$$

$$\mu_P = \mu_P^{\max} e^{(0.063(T - T_{\max}))} \quad (\text{A21})$$

$$R = \frac{I_z}{I_s} e^{(1 - \frac{I_z}{I_s})} \leq 1 \quad (\text{A22})$$

$$I_z = I_0 e^{-(k_w + k_p \bar{P})z} \quad (\text{A23})$$

$$\bar{P} = \frac{1}{z} \int_0^z P dz \quad (\text{A24})$$

$$Q = Q_N + Q_A \leq 1 \quad (\text{A25})$$

$$Q_N = \frac{\frac{N}{k_p} e^{(-\psi A)}}{1 + \frac{N}{k_p} + \frac{A}{k_p^2}} \quad (\text{A26})$$

$$Q_A = \frac{\frac{A}{k_p}}{1 + \frac{N}{k_p} + \frac{A}{k_p^2}} \quad (\text{A27})$$

Phytoplankton extra-DOC exudation

$$E_{doc} = \gamma_2 \theta_{P_{cn}} F_P \quad (\text{A28})$$

Zooplankton grazing on phytoplankton (G_P), bacteria (G_B), detrital nitrogen (G_{D_n}) and detrital carbon (G_{D_c})

$$G_P = \frac{gZ_P P^2}{k_g (p_P P + p_B B + p_D D_n) + p_P P^2 + p_B B^2 + p_D D_n^2} \quad (\text{A29})$$

Bacterial maximum potential uptake of ammonium

$$G_B = \frac{gZp_B B^2}{k_g(p_P P + p_B B + p_D D_n) + p_P P^2 + p_B B^2 + p_D D_n^2} \quad (\text{A30})$$

$$U_A^* = \mu_B B \frac{A}{k_A + A} \quad (\text{A44})$$

$$G_{D_n} = \frac{gZp_D D_n^2}{k_g(p_P P + p_B B + p_D D_n) + p_P P^2 + p_B B^2 + p_D D_n^2} \quad (\text{A31})$$

$$\mu_B = \mu_B^{\max}(1 - \phi_{inhib}) \quad (\text{A45})$$

$$G_{D_c} = \left(\frac{D_c}{D_n}\right) G_{D_n} \quad (\text{A32})$$

$$\phi_{inhib} = \phi_{inhib}^{\max} \frac{I_z}{I_{\max}} \quad (\text{A46})$$

$$G_n = G_P + G_B + G_{D_n} \quad (\text{A33})$$

Bacterial production (F_B), ammonium excretion or uptake (E_B) and respiration (R_B)

$$E_B = U_{L_n} - \left(\frac{\omega_B}{\theta_{B_{cn}}}\right) U_{L_c} \quad (\text{A47})$$

$$G_c = \theta_{P_{cn}} G_P + \theta_{B_{cn}} G_B + G_{D_c} \quad (\text{A34})$$

Zooplankton production (F_Z) and excretion (E_Z)

$$\theta_f^* = \frac{\beta_n \theta_{Z_{cn}}}{\beta_c \omega_Z} \quad (\text{A35})$$

if $E_B > 0$ (ammonium excretion)
 if $E_B < 0$ (ammonium uptake)
 if $U_A^* \geq -E_B$ (C-limitation):

$$F_B = U_{L_n} - E_B = \left(\frac{\omega_B}{\theta_{B_{cn}}}\right) U_{L_c} \quad (\text{A48})$$

$$\theta_f = \frac{(1 - \phi) G_c}{(1 - \phi) G_n} \quad (\text{A36})$$

$$R_B = (1 - \omega_B) U_{L_c} \quad (\text{A49})$$

if $\theta_f > \theta_f^*$ (N-limitation):

$$F_Z = \beta_n (1 - \phi) G_n \quad (\text{A37})$$

if $U_A^* < -E_B$ (N-limitation):

$$E_B = -U_A^* \quad (\text{A50})$$

$$E_Z = 0 \quad (\text{A38})$$

$$F_B = U_{L_n} - E_B = U_{L_n} + U_A^* \quad (\text{A51})$$

if $\theta_f < \theta_f^*$ (C-limitation):

$$F_Z = \frac{\beta_c \omega_Z}{\theta_{Z_{cn}}} (1 - \phi) G_c \quad (\text{A39})$$

$$R_B = \left(\frac{1}{\omega_B - 1}\right) \theta_{B_{cn}} F_B \quad (\text{A52})$$

$$E_Z = \left(\frac{\beta_n}{\theta_f} - \frac{\beta_n}{\theta_f^*}\right) (1 - \phi) G_c \quad (\text{A40})$$

Bacterial hydrolysis of semilabile DOC (U_{S_c}) and semilabile DON (U_{S_n})

$$U_{S_c} = \mu_{S_c} \theta_{B_{cn}} B \frac{S_c}{k_{S_c} + S_c} \quad (\text{A53})$$

Zooplankton respiration

$$R_Z = \beta_c (1 - \phi) G_c - \theta_{Z_{cn}} F_Z \quad (\text{A41})$$

$$U_{S_n} = \left(\frac{S_n}{S_c}\right) U_{S_c} \quad (\text{A54})$$

Bacterial uptake of labile DOC (U_{L_c}), labile DON (U_{L_n})

$$U_{L_c} = \mu_B \theta_{B_{cn}} B \frac{L_c}{k_{L_c} + L_c} \quad (\text{A42})$$

Mortality of phytoplankton (M_P), zooplankton (M_Z) and bacteria (M_B)

$$M_P = m_P P \quad (\text{A55})$$

$$U_{L_n} = \left(\frac{L_n}{L_c}\right) U_{L_c} \quad (\text{A43})$$

$$M_Z = \frac{m_Z Z^2}{k_Z + Z} \quad (\text{A56})$$

$$M_B = m_B B \quad (\text{A57})$$

Breakdown of detrital nitrogen (M_{Dn}), detrital carbon (M_{Dc}) and detrital $CaCO_3$ (M_{Dh}) if $U_{DMSPd}^* \geq U_{DMSPd}$ (S-limitation):

$$M_{Dn} = m_{Dn} D_n \quad (A58) \quad R_{DMS} = 0 \quad (A72)$$

$$M_{Dc} = m_{Dc} D_c \quad (A59)$$

$$M_{Dh} = m_{Dh} D_h \quad (A60)$$

Chlorophyll-a production

$$\rho_{chl} = \frac{\theta_{chl}^m J_Q}{\alpha_{chl} \theta_{chl} I_z} \quad (A61)$$

$$\theta_{chl} = \frac{CHL}{C_{mw} \theta_{P.cn} P} \quad (A62)$$

DMS photolysis

$$DMS_{photo} = k_{photo} DMS \quad (A73)$$

$$k_{photo} = k_{photo}^{\max} \frac{I_z}{I_{\max}} \quad (A74)$$

DMS emission to the atmosphere

$$DMS_{emiss} = k_{emiss} DMS \quad (A75)$$

$$k_{emiss} = \frac{k_v \frac{24}{100}}{\Delta z} \quad (A76)$$

A3. S-Cycle Model Terms

[52] S:N phytoplankton internal quota

$$\theta_{P.sn} = \theta_{P.sn}^{\min} + \left(\theta_{P.sn}^0 - \theta_{P.sn}^{\min} \right) \frac{\min(I_z, I^*)}{I^*} \quad (A63)$$

$$\theta_{P.sn}^0 = \theta_{P.sn}^{\max} - \left(\theta_{P.sn}^{\max} - \theta_{P.sn}^{\min} \right) \frac{I_{\max} - I_0}{I_{\max} - I_{\min}} \quad (A64)$$

$$k_v = (0.24U^2 + 0.061U) \left(\frac{Sc}{600} \right)^{(-0.5)} \quad (A77)$$

$$Sc = 2674 - 147.12(SST) + 3.726(SST^2) - 0.038(SST^3) \quad (A78)$$

Phytoplankton DMS exudation

$$E_{DMS} = \gamma_s DMSPp \quad (A65)$$

$$\gamma_s = \gamma_s^{\max} \frac{I_z}{I_{\max}} \frac{\mu_P}{\mu_P^{\max}} \quad (A66)$$

DMS yield

$$DMS_{yield} = \frac{DMS_{prod.}}{DMS_{cons.}} = \frac{E_{DMS} + R_{DMS} + fDMSPd}{E_{DMS} + U_{DMSPd} + U_{DMSPd}^P + fDMSPd} \quad (A79)$$

Phytoplankton DMSPd uptake

$$U_{DMSPd}^P = \frac{DMSPd}{A} \alpha_P F_P^A + \frac{DMSPd}{N} \alpha_P F_P^N \quad (A67)$$

Bacterial DMSPd and DMS uptake

$$U_{DMSPd} = \mu_B \theta_{B.sc} \theta_{B.cn} B \frac{DMSPd}{k_{DMSPd} + DMSPd} \quad (A68)$$

$$U_{DMS} = \mu_B \theta_{B.sc} \theta_{B.cn} \alpha_B B \frac{DMS}{k_{DMS} + DMS} \quad (A69)$$

Bacterial sulfur demand

$$U_{DMSPd}^* = \theta_{B.sc} \theta_{B.cn} F_B \quad (A70)$$

Bacterial DMS production

if $U_{DMSPd}^* < U_{DMSPd}$ (no S-limitation):

$$R_{DMS} = \alpha_2 (U_{DMSPd} - U_{DMSPd}^*) \quad (A71)$$

A4. Advection (Sinking) and Diffusion Terms

[53] Phytoplankton and Detritus sinking

$$S(X_i) = |\bar{w}_i| \frac{\partial X_i}{\partial z} \quad (A80)$$

for $X_i = P, D_n, D_c, D_h$ and CHL . Turbulent diffusion

$$D(X_i) = \frac{\partial}{\partial z} \left(k_z \frac{\partial X_i}{\partial z} \right) \quad (A81)$$

for $X_i = P, Z, B, N, A, L_n, L_c, S_n, S_c, D_n, D_c, D_h, DIC, ALK, CHL, DMSPp, DMSPd$ and DMS .

A5. Turbulent Diffusion and Temperature Profiles

[54]

$$D^* = \frac{D_{\max}^* + D_{\min}^* e^{r(D_{\min}^* - D_{\max}^*) \frac{2(z^* - MLD^*)}{H^*}}}{1 + e^{r(D_{\min}^* - D_{\max}^*) \frac{2(z^* - MLD^*)}{H^*}}} \quad (A82)$$

where the asterisk symbol denotes normalized variables (values between 0 and 1):

$$X^* = X/H \quad (\text{A83})$$

$$Y^* = Y/D_{\max} \quad (\text{A84})$$

for $X = z, MLD, H$, and $Y = D, D_{\min}, D_{\max}$. H is the model vertical domain (200 m) and D can be either diffusion (kz) or sea temperature (st).

A6. Numerical Scheme for Solving the 1-D Model

[55] The general form of the model equations is:

$$\frac{\partial X_i}{\partial t} = J_i - |\bar{w}_i| \frac{\partial X_i}{\partial z} + \frac{\partial}{\partial z} \left(kz \frac{\partial X_i}{\partial z} \right) = f(X_i) \quad (\text{A85})$$

where J_i are the biological source/sink terms for each variable i as defined above, $|\bar{w}_i| \frac{\partial X_i}{\partial z}$ is the vertical sinking (only applies to phytoplankton and detritus), and $\frac{\partial}{\partial z} (kz \frac{\partial X_i}{\partial z})$ is the vertical turbulent diffusion. The numerical scheme is implemented using a finite difference approximation. At each time step and for each vertical grid point j , all three terms are calculated independently to obtain $f(X_i)$. The sinking term is calculated using a first-order upwind discretization:

$$\frac{\partial X_i^j}{\partial z} = \frac{X_i^{j-1} - X_i^j}{\Delta z} \quad (\text{A86})$$

while the diffusion term is calculated using a second-order centered discretization:

$$\frac{\partial}{\partial z} \left(kz \frac{\partial X_i}{\partial z} \right) = \frac{kz^{j+1/2} (X_i^{j+1} - X_i^j) - kz^{j-1/2} (X_i^j - X_i^{j-1})}{(\Delta z)^2} \quad (\text{A87})$$

where $kz^{j+1/2} \equiv kz(z_{j+1/2})$. Finally, for each vertical grid point we then advance the solution in time from X_i^t to X_i^{t+1} using a forward Euler method [Press *et al.*, 1992]:

$$X_i^{t+1} = X_i^t + f(X_i^t) \Delta t \quad (\text{A88})$$

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