Supporting Information

A Free Phosphaborene Stable at Room Temperature

Jiancheng Li, Zhihao Lu, Liu Leo Liu*

Department of Chemistry and Guangdong Provincial Key Laboratory of Catalysis, Southern University of Science and Technology, Shenzhen 518055 (China)

Table of Contents

| I. | Experimental Section | .S2 |
|-----|------------------------------|------|
| II. | NMR Spectra of new compounds | .S9 |
| III | X-ray crystallographic data | .S34 |
| IV | Computational details | S36 |
| v. | References | .S46 |

I. Experimental Section

All manipulations were carried out in a nitrogen-filled glovebox or under an atmosphere of dry nitrogen using standard Schlenk techniques, unless otherwise stated. Toluene, nhexane and tetrahydrofuran (THF) were purified by LiAlH₄ and stored over molecular sieves. C₆D₆ was dried by sodium/potassium alloy. NMR spectra were acquired on a Bruker Avance 400 (¹H: 400 MHz, ¹³C: 101 MHz) or 600 (¹H: 600 MHz, ¹³C: 151 MHz) NMR spectrometer at 298 K. ¹H, ¹³C{¹H} spectra were referenced to internal C₆H₆. Data are presented as follows: chemical shift, multiplicity (s = singlet, d = doublet, t =triplet, q = quartet, sept=septet, m = multiplet and/or multiple resonances), coupling constant in hertz (Hz), integration, attribution. High resolution mass spectrometry (HRMS) was performed with a Thermo Fisher Scientific Q-Exactive MS System. Crystal data were collected on a Bruker D8 VENTURE diffractometer with graphite monochromated Cu Ka ($\lambda = 1.54178$). Data reduction, scaling and absorption corrections were performed using SAINT (Bruker, V8.38A, 2013). The structure was solved with the XT structure solution program using the Intrinsic Phasing solution method and by using Olex2 as the graphical interface. The frames were integrated with the Bruker SAINT software package using a narrow-frame algorithm. Data were corrected for absorption effects using the empirical multi-scan method (SADABS). The model was refined with the ShelXL program using Least Squares minimization. All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were included in structure factor calculations. All hydrogen atoms were assigned to idealized geometric positions. Commercial reagents were purchased from Energy Chemical, J&K, or TCI Chemical Co. and used as received. NHB-Br,^[S1] NaPH₂,^[S2] and NHC=NSiMes^[S3] were prepared according to the procedure described in the literature.



Scheme S1-1. Synthesis of 1 and 2.

1) Synthesis of 1 To the solid mixture of NHB-Br (2.25g, 4.8 mmol) and NaPH₂ (269 mg, 4.8 mmol) at -50 °C pre-cooled THF (20 mL) was added under stirring. The reaction solution was then allowed to warm to room temperature and stirred for further 8 hours. The volatiles were removed through evaporation under vacuum. Then the residues were extracted with *n*-hexane (20 mL), after filtration through a pad of celite, the volatiles were removed to give a colorless solid 1 (1.85 g, 91%). ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 1.28$ (d, ³*J*_{HH} = 6.9 Hz, 12 H, CH*Me*₂), 1.35 (d, ³*J*_{HH} = 6.9 Hz, 12 H, CH*Me*₂), 1.29 (d, ¹*J*_{PH} = 202.6 Hz, 2 H, PH₂), 3.51 (s, 4 H, NCH₂CH₂N),

3.53 (sept, ${}^{3}J_{\text{HH}} = 6.9$ Hz, 4 H, *CH*Me₂), 7.13-7.21 (m, 6 H, Ar-*H*). ${}^{13}\text{C}\{{}^{1}\text{H}\}$ NMR (150.9 MHz, C₆D₆, 298 K, ppm): $\delta = 25.0$, 25.0, 25.2, (CH*Me*₂), 28.7 (CHMe₂), 53.9 (NCH₂CH₂N), 124.3 (Ar-C), 127.5 (Ar-C), 139.6 (Ar-C), 147.7 (Ar-C). ${}^{31}\text{P}\{{}^{1}\text{H}\}$ NMR (243 MHz, 298 K, C₆D₆, ppm): $\delta = -246.8$ (t, ${}^{1}J_{\text{PH}} = 202.6$ Hz, *P*H₂). ${}^{11}\text{B}\{{}^{1}\text{H}\}$ NMR (192.6 MHz, 298 K, C₆D₆, ppm): $\delta = 35.1$ (br). HRMS(m/z): [M+H]⁺ calcd. for C₂₆H₄₁N₂BP: 423.30938, found: 423.30949.

2) Synthesis of 2 To a solid mixture of 1 (1.85 g, 4.4 mmol) and benzyl potassium (Bnk) (568.9 mg, 4.4 mmol) at -30 °C pre-cooled THF (20 mL) was added under stirring. The red suspension was then allowed to warm to room temperature and stirred for further one hour to give an orange-red clear solution. Then it was cooled to -30 °C again, and to which slightly excess chlorotrimethylsilane (Me₃SiCl) (0.64 mL, 5 mmol) was dropwise added via syringe. The reaction solution was then allowed to warm to room temperature and stirred for further 3 hours. The volatiles were removed through evaporation under vacuum and the residues were then extracted with *n*-hexane (20 mL). After filtration through a pad of celite, the clear colorless solution was cooled to -50 °C and to which tert-butyllithium (tBuLi) (1.6 M in n-hexane, 2.7 mL, 4.4 mmol) was added drop-by-drop. The reaction solution was then allowed to warm to room temperature and stirred for further 8 hours, resulting in a white suspension. The precipitate was collected by filtration, washed with cold *n*-hexane (5 mL) and dried under vacuum to give 2 as white solid (1.47 g, 67%). ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = -0.07$ (br, 9 H, SiMe₃), 1.25 (d, ³J_{HH} = 6.9 Hz, 12 H, CHMe₂), 1.50 (d, ³J_{HH} = 6.9 Hz, 12 H, CHMe₂), 3.61 (s, 4 H, NCH₂CH₂N), 3.71 (sept, ${}^{3}J_{HH}$ = 6.9 Hz, 4 H, CHMe₂), 7.17-7.23 (m, 6 H, Ar-H). ¹³C{¹H} NMR (150.9 MHz, C₆D₆, 298 K, ppm): δ = 6.7 (SiMe₃), 24.8, 26.3, (CHMe₂), 28.7 (CHMe₂), 53.7 (NCH₂CH₂N), 124.8 (Ar-C), 126.7 (Ar-*C*), 143.6 (Ar-*C*), 148.5 (Ar-*C*). ³¹P{¹H} NMR (243 MHz, 298 K, C₆D₆, ppm): $\delta = -286.4$. ¹¹B{¹H} NMR (192.6 MHz, 298 K, C₆D₆, ppm): $\delta = 38.3$ (br). HRMS(m/z): [M-Li]⁻ calcd. for C₂₉H₄₇N₂BPSi: 493.33447, found: 493.33469.



Scheme S1-2. Synthesis of 3

Synthesis of 3 To a colorless solution of NHC=NSiMe₃ (952 mg, 2 mmol) in toluene (20 mL) at -30 °C boron tribromide (1.0 M in methylene chloride, 2 ml, 2mmol) was added under stirring. The reaction solution was allowed to warm to room temperature and stirred for further 8 hours. After filtration through a pad of celite, the volatiles were removed through evaporation under vacuum. The residues were then washed with *n*-hexane (2 mL) and dried under vacuum to give a colorless solid **3** (974 mg, 85%). ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 1.09$ (d, ³*J*_{HH} = 6.9 Hz, 12 H, CH*Me*₂), 1.44

(d, ${}^{3}J_{HH} = 6.9$ Hz, 12 H, CH*Me*₂), 2.99 (sept, ${}^{3}J_{HH} = 6.9$ Hz, 4 H, C*H*Me₂), 6.01 (s, 2 H, *H*C=C*H*), 7.06-7.19 (m, 6 H, Ar-*H*). ${}^{13}C\{{}^{1}H\}$ NMR (150.9 MHz, C₆D₆, 298 K, ppm): $\delta = 23.6, 25.0, (CHMe_{2}), 29.2 (CHMe_{2}), 116.2 (HC=CH), 124.5 (Ar-C), 130.7 (Ar-C), 131.7 (Ar-C), 147.2 (Ar-C). <math>{}^{11}B\{{}^{1}H\}$ NMR (192.6 MHz, 298 K, C₆D₆, ppm): $\delta = 12.7$ (br). HRMS(m/z): [M+H]⁺ calcd. for C₂₇H₃₇N₃B⁷⁹Br₂: 572.1442, found: 572.1419; C₂₇H₃₇N₃B⁷⁹Br⁸¹Br: 574.1421, found: 574.1401; C₂₇H₃₇N₃B⁸¹Br₂: 576.1401, found: 576.1379.



Scheme S1-3. Synthesis of 4

Synthesis of 4 To a solution of 2 (500 mg, 1 mmol) in toluene (5 mL) at -50 °C a precooled (-50 °C) solution of 3 (573 mg, 1 mmol) was added with stirring. The reaction solution was then allowed to warm to room temperature and stirred for further two hours. After filtration through a pad of celite, the volatiles were removed through evaporation under vacuum. The residues were then washed with *n*-hexane (2 mL) and dried under vacuum to give a colorless solid 4 (809 mg, 82%). ¹H NMR (400 MHz, C_6D_6 , 298 K, ppm): $\delta = -0.07$ (d, ${}^{3}J_{PH} = 5.5$ Hz, 9 H, SiMe₃), 1.05 (d, ${}^{3}J_{HH} = 6.8$ Hz, 12 H, CHMe₂), 1.26 (d, ³J_{HH} = 6.8 Hz, 12 H, CHMe₂), 1.33 (d, ³J_{HH} = 6.8 Hz, 12 H, CHMe₂), 1.34 (d, ${}^{3}J_{HH} = 6.5$ Hz, 12 H, CHMe₂), 3.10 (sept, ${}^{3}J_{HH} = 6.8$ Hz, 4 H, CHMe₂), 3.59 (s, 4 H, NCH₂CH₂N), 3.74 (sept, ${}^{3}J_{HH} = 6.8$ Hz, 4 H, CHMe₂), 6.00 (s, 2 H, HC=CH), 7.05-7.23 (m, 12 H, Ar-H). ¹³C{¹H} NMR (150.9 MHz, C₆D₆, 298 K, ppm): $\delta = 3.3$ (d, ²J_{PC} = 13.0 Hz, SiMe₃), 23.2, 24.3, 25.9, 27.5, (CHMe₂), 28.7 (CHMe₂), 28.8 (CHMe₂), 54.7 (NCH2CH2N), 116.3 (HC=CH), 124.1 (Ar-C), 124.3 (Ar-C), 126.3 (Ar-C), 128.3 (Ar-C), 130.1 (Ar-C), 133.4 (Ar-C), 142.2 (Ar-C), 130.1 (Ar-C), 147.0 (Ar-C), 147.4 (NH*C*=N). ${}^{31}P{}^{1}H{}$ NMR (243 MHz, 298 K, C₆D₆, ppm): $\delta = -193.7$. ${}^{11}B{}^{1}H{}$ NMR (192.6 MHz, 298 K, C₆D₆, ppm): $\delta = 34.4$ (br, NHB), 31.0 (br, PSiMe₃BBr). HRMS(m/z): [M+H]⁺ calcd. for C₅₆H₈₄N₅B₂⁷⁹BrPSi: 986.5598, found: 986.5621; C₅₆H₈₄N₅B₂⁸¹BrPSi: 988.5577, found: 988.5602. X-ray quality single-crystals of **4** were obtained by recrystallization from its saturated solution in *n*-hexane at -35 °C.



Scheme S1-4. Synthesis of 5

Synthesis of 5 A solution of 4 (49.3 mg, 0.05 mmol) in C₆D₆ (0.5 mL) was heated at 65 °C for 4 hours. The reaction solution was then allowed to cool to room temperature. After filtration through a pad of celite, the volatiles were removed through evaporation under vacuum. The residues were then washed with cold *n*-hexane (0.2 mL) and dried under vacuum to give a colorless solid 5 (36.7 mg, 88%). ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 1.02$ (d, ${}^{3}J_{HH} = 6.8$ Hz, 12 H, CHMe₂), 1.31 (d, ${}^{3}J_{HH} = 6.8$ Hz, 12 H, $CHMe_2$), 1.34 (d, ${}^{3}J_{HH} = 6.8$ Hz, 12 H, $CHMe_2$), 1.44 (d, ${}^{3}J_{HH} = 6.8$ Hz, 12 H, $CHMe_2$), 2.59 (sept, ${}^{3}J_{HH} = 6.8$ Hz, 4 H, CHMe₂), 3.53 (s, 4 H, NCH₂CH₂N), 3.64 (sept, ${}^{3}J_{HH} =$ 6.8 Hz, 4 H, CHMe₂), 5.70 (s, 2 H, HC=CH), 6.85-7.16 (m, 12 H, Ar-H). ¹³C{¹H} NMR $(150.9 \text{ MHz}, C_6D_6, 298 \text{ K}, \text{ppm}): \delta = 23.9, 24.2, 25.3, (CHMe_2), 28.6 (CHMe_2), 29.1$ (CHMe₂), 53.8 (NCH₂CH₂N), 116.1 (HC=CH), 126.2 (Ar-C), 128.3 (Ar-C), 130.9 (Ar-*C*), 131.8 (Ar-*C*), 141.5 (Ar-*C*), 142.3 (Ar-*C*), 147.0 (Ar-*C*), 148.0 (NH*C*=N). ³¹P{¹H} NMR (243 MHz, 298 K, C₆D₆, ppm): δ = -291.9. ¹¹B{¹H} NMR (192.6 MHz, 298 K, C_6D_6 , ppm): $\delta = 38.6$ (br, NHB), 53.0 (br, P=B). HRMS(m/z): $[M+H]^+$ calcd. for C₅₃H₇₅N₅B₂P: 834.5941, found: 834.5928. X-ray quality single-crystals of 5 were obtained by recrystallization from its saturated toluene solution layered with *n*-hexane at -35 °C.



Scheme S1-5. Reactivity of 5 towards aldehyde.

Synthesis of 6 To a solution of 5 (41.7 mg, 0.05 mmol) in toluene (1 mL) at -50 $^{\circ}$ C a pre-cooled (-50 $^{\circ}$ C) solution of *p*-methyl benzaldehyde (6 mg, 0.05 mmol) was added in toluene (1 mL) with stirring. The reaction solution was then allowed to warm to room s5

temperature and stirred for further 12 hours. After filtration through a pad of celite, the volatiles were removed through evaporation under vacuum. The residues were then washed with *n*-hexane (0.2 mL) and dried under vacuum to give a colorless solid 6(42.9 mg, 90%). ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 0.91$ (d, ³J_{HH} = 6.9 Hz, 6 H, CHMe₂), 1.08 (d, ${}^{3}J_{HH} = 6.9$ Hz, 6 H, CHMe₂), 1.12 (d, ${}^{3}J_{HH} = 6.6$ Hz, 6 H, CHMe₂), 1.12 (d, ${}^{3}J_{\text{HH}} = 6.6 \text{ Hz}, 6 \text{ H}, \text{CH}Me_{2}$), 1.23 (d, ${}^{3}J_{\text{HH}} = 6.9 \text{ Hz}, 6 \text{ H}, \text{CH}Me_{2}$), 1.25 (d, ${}^{3}J_{\text{HH}}$ = 6.9 Hz, 6 H, CHMe₂), 1.27 (d, ${}^{3}J_{HH}$ = 6.8 Hz, 6 H, CHMe₂), 1.37 (d, ${}^{3}J_{HH}$ = 6.8 Hz, 6 H, CHMe₂), 2.20 (s, 3 H, Ar-Me), 2.96 (sept, ${}^{3}J_{HH} = 6.8$ Hz, 2 H, CHMe₂), 3.01 (sept, ${}^{3}J_{\text{HH}} = 6.9 \text{ Hz}, 2 \text{ H}, CHMe_{2}), 3.40 \text{ (m, 2 H, NCH}_{2}CH_{2}N), 3.47 \text{ (sept, } {}^{3}J_{\text{HH}} = 6.9 \text{ Hz}, 2$ H, CHMe₂), 3.56 (m, 2 H, NCH₂CH₂N), 3.66 (sept, ${}^{3}J_{HH} = 6.9$ Hz, 4 H, CHMe₂), 4.84 (d, ${}^{2}J_{PH} = 3.1$ Hz, 1 H, PCHO), 5.85 (s, 2 H, HC=CH), 6.17 (d, ${}^{3}J_{HH} = 7.7$ Hz, 2 H, MeC₆*H*₄CHO), 6.83 (d, ${}^{3}J_{HH}$ = 7.7 Hz, 2 H, MeC₆*H*₄CHO), 7.04-7.26 (m, 12 H, Ar-*H*). $^{13}C{^{1}H}$ NMR (150.9 MHz, C₆D₆, 298 K, ppm): $\delta = 21.9, 24.1, 24.3, 24.6, 24.6, 24.8,$ 25.3, 25.4, 25.6, (CHMe2), 27.2 (Ar-Me), 29.1 (CHMe2), 29.3 (CHMe2), 29.4 (CHMe2), 29.5 (CHMe₂), 55.2 (NCH₂CH₂N), 68.9 (d, ¹J_{PC}=6.3 Hz), 115.7 (HC=CH), 124.7 (Ar-C), 124.9 (Ar-C), 125.0 (Ar-C), 125.3 (Ar-C), 127.3 (Ar-C), 128.9 (Ar-C), 129.0 (Ar-C), 129.1 (Ar-C), 130.2 (Ar-C), 134.5 (Ar-C), 16.2 (Ar-C), 141.8 (Ar-C), 148.1 (Ar-C), 148.1 (Ar-C), 148.3 (Ar-C), 148.5 (NHC=N). ³¹P{¹H} NMR (243 MHz, 298 K, C₆D₆, ppm): $\delta = -89.6$. ¹¹B{¹H} NMR (192.6 MHz, 298 K, C₆D₆, ppm): $\delta = 33.9$ (br, NH*B*), P=B not observed. HRMS(m/z): $[M+H]^+$ calcd. for C₆₁H₈₃ON₅B₂P: 954.6516, found: 954.6529. X-ray quality single-crystals of 6 were obtained by recrystallization from its saturated solution in toluene at -35 °C.



Scheme S1-6. Reactivity of 5 towards ketone.

Synthesis of 7 To a solution of 5 (41.7 mg, 0.05 mmol) in toluene (1 mL) at -50 °C a pre-cooled (-50 °C) solution of *p*-fluoroacetophenone (6.9 mg, 0.05 mmol) was added in toluene (1 mL) with stirring. The reaction solution was then allowed to warm to room temperature and stirred for further four days. After filtration through a pad of celite, the solution was concentrated to 0.3 mL and kept in a refrigerator at -35 °C overnight to obtain colorless crystals of 7 which were collected by filtration and dried under vacuum. (32.9 mg, 68%). ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 0.71$ (d, ¹*J*_{PH} = 212.3 Hz, 1 H, P*H*), 1.01 (d, ³*J*_{HH} = 6.8 Hz, 12 H, CH*Me*₂), 1.04 (d, ³*J*_{HH} = 6.8 Hz, 24 H, CH*Me*₂), 1.17 (d, ³*J*_{HH} = 6.8 Hz, 12 H, CH*Me*₂), 3.01 (sept, ³*J*_{HH} = 6.8 Hz, 4 H, C*H*Me₂), 3.58 (s, 4 H, NCH₂C*H*₂N), 3.67 (sept, ³*J*_{HH} = 6.8 Hz, 4 H, C*H*Me₂), 4.77 (s, 1 H, C=C*H*₂), 4.81

(s, 1 H, C=C*H*₂), 5.97 (s, 2 H, *H*C=C*H*), 6.82-7.60 (m, 16 H, Ar-*H* and C₆*H*₄). ¹³C {¹H} NMR (150.9 MHz, C₆D₆, 298 K, ppm): δ = 23.0, 24.6, 25.9, 25.9, 28.2, 28.6, (CH*Me*₂ and CHMe₂), 54.4 (NCH₂CH₂N), 94.8 (C=CH₂), 114.7 (d, ³*J*_{PC} = 21.0 Hz), 115.8 (H*C*=CH), 124.1 (Ar-C), 124.2 (Ar-C), 126.6 (Ar-C), 127.6 (Ar-C), 129.8 (Ar-C), 133.4 (Ar-C), 141.4 (Ar-C), 147.4 (Ar-C), 148.1 (NH*C*=N), 155.1, 162.9 (d, ¹*J*_{FC} = 244.9 Hz, Ar-C-F). ³¹P {¹H} NMR (243 MHz, 298 K, C₆D₆, ppm): δ = -233.4 (d, ¹*J*_{PH} = 212.3 Hz). ¹⁹F {¹H} NMR (564.7 MHz, 298 K, C₆D₆, ppm): δ = -115.5. ¹¹B {¹H} NMR (192.6 MHz, 298 K, C₆D₆, ppm): δ = 37.1 (br, NH*B*), 27.5 (br, P-*B*-O). HRMS(m/z): [M+H]⁺ calcd. for C₆₁H₈₂ON₅B₂FP: 972.6422, found: 972.6416.



Scheme S1-7. Reactivity of 5 towards carbon disulfide.

Synthesis of 8 and 9 To a solution of 5 (41.7 mg, 0.05 mmol) in toluene (1 mL) at room temperature excess carbon disulfide (10 µL) was added. The reaction solution was stood without disturbance for overnight. After filtration through a pad of celite, the volatiles were removed through evaporation under vacuum. The residues were then extracted with *n*-hexane and then the solution was concentrated to 0.5 mL and kept in a refrigerator at -35 °C overnight to obtain colorless crystals of 8 (16.2 mg, 73%) which were collected by filtration and dried under vacuum. The filtrate was further concentrated to 0.2 mL and kept in a refrigerator at -35 °C overnight toobtain another small amounts of colorless crystals of 8. The mother solution was vacuumed to give orange residues which was further extracted with *n*-pentane (0.5 mL), and the resulting solution was concentrated to 0.1 mL and kept in a refrigerator at -35 °C 2 days to obtained light yellow crystals of 9 (17.8 mg, 66%). For 8: ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 1.15$ (d, ${}^{3}J_{\text{HH}} = 6.6$ Hz, 12 H, CHMe₂), 1.34 (d, ${}^{3}J_{\text{HH}} = 6.6$ Hz, 12 H, CHMe₂), 3.03 (sept, ${}^{3}J_{HH} = 6.6$ Hz, 4 H, CHMe₂), 5.91 (s, 2 H, HC=CH), 7.07-7.24 (m, 6 H, Ar-H). ${}^{13}C{}^{1}H{}$ NMR (150.9 MHz, C₆D₆, 298 K, ppm): $\delta = 23.9, 24.5, (CHMe_2),$ 29.1 (CHMe2), 114.8 (HC=CH), 124.1 (Ar-C), 129.9 (Ar-C), 133.1 (Ar-C), 143.3 (Ar-*C*), 147.7 (NH*C*=N). ¹¹B{¹H} NMR (192.6 MHz, 298 K, C₆D₆, ppm): δ = 34.9 (br, =NBS). HRMS(m/z): $[M+H]^+$ calcd. for C₅₄H₇₃N₆B₂S₂: 891.5519, found: 891.5520.; For 9: ¹H NMR (600 MHz, C₆D₆, 298 K, ppm): $\delta = 1.19$ (d, ³J_{HH} = 7.0 Hz, 12 H, CHMe₂), 1.21 (d, ${}^{3}J_{HH} = 7.0$ Hz, 12 H, CHMe₂), 3.41 (sept, ${}^{3}J_{HH} = 7.0$ Hz, 4 H, CHMe₂), 3.47 (s, 4 H, NCH₂CH₂N), 7.06-7.19 (m, 6 H, Ar-H). ¹³C {¹H} NMR (150.9 MHz, C₆D₆, 298 K, ppm): $\delta = 24.1, 25.7, (CHMe_2), 28.7 (CHMe_2), 53.7 (NCH_2CH_2N), 124.8 (Ar-$ C), 128.3 (Ar-C), 128.4 (Ar-C), 138.8 (Ar-C), 147.4 (Ar-C), 224.1 (d, ²J_{PC}=4.8 Hz, *C*=S). ³¹P{¹H} NMR (243 MHz, 298 K, C₆D₆, ppm): δ = 309.9. ¹¹B{¹H} NMR (192.6

MHz, 298 K, C₆D₆, ppm): δ = 28.0 (br). HRMS(m/z): [M+H]⁺ calcd. for C₂₈H₃₉N₂BPS₃: 541.2101, found: 541.2098.



Figure S2-2. ${}^{13}C{}^{1}H$ NMR Spectrum of 1 (150.9 MHz, C₆D₆, 298 K).



Figure S2-4. ³¹P NMR Spectrum of 1 (243 MHz, C₆D₆, 298 K).



-35.2

Figure S2-5b. ¹¹B{¹H} NMR Spectrum of 1 (baseline corrected) (192.6 MHz, C₆D₆, 298 K).



S12



Figure S2-9a. ${}^{11}B{}^{1}H$ NMR Spectrum of 2 (192.6 MHz, C₆D₆, 298 K).



Figure S2-10. ¹H NMR Spectrum of 3 (600 MHz, C₆D₆, 298 K).

-38.3



Figure S2-12a. ¹¹B{¹H} NMR Spectrum of 3 (192.6 MHz, C₆D₆, 298 K).



S16



Figure S2-15. ${}^{31}P{}^{1}H$ NMR Spectrum of 4 (243 MHz, C₆D₆, 298 K).



Figure S2-16a. ¹¹B $\{^{1}H\}$ NMR Spectrum of **4** (192.6 MHz, *n*-hexane, 298 K) (* marks signals attributed to the N-heteroboryl of **5** generated from spontaneous ClSiMe₃-elimination of **4** during the data collection).





Figure S2-16b. ¹¹B{¹H} NMR Spectrum of **4** (baseline corrected) (192.6 MHz, *n*-hexane, 298 K) (* marks signals attributed to the N-heteroboryl of **5** generated from spontaneous ClSiMe₃-elimination of **4** during the data collection).





S19



Figure S2-20a. $^{11}B\{^{1}H\}$ NMR Spectrum of 5 (192.6 MHz, C₆D₆, 298 K).



Figure S2-20c. Overlay ¹¹B{¹H} NMR Spectra of **5** (C₆D₆, in blue-green) and blank sample (pure C₆D₆, in red) (baseline corrected) (192.6 MHz, 298 K).





--89.6



-33.9

Figure S2-25. ¹H NMR Spectrum of 7 (600 MHz, C₆D₆, 298 K).



Figure S2-27. ³¹P{¹H} NMR Spectrum of 7 (243 MHz, C₆D₆, 298 K).





-37.1 -27.7

Figure S2-30b. ¹¹B{¹H} NMR Spectrum of 7 (baseline corrected) (192.6 MHz, C₆D₆, 298 K).



S28



-34.9



Figure S2-35. $^{13}\rm{C}\{^{1}\rm{H}\}$ NMR Spectrum of 9 (150.9 MHz, C₆D₆, 298 K).



S31



Figure S2-37b. ¹¹B $\{^{1}H\}$ NMR Spectrum of 9 (baseline corrected) (192.6 MHz, C₆D₆, 298 K).



Figure S2-38. Monitoring ³¹P NMR spectrum of the spontaneous transformation of **4** to **5** at room temperature. The conversion was determined by the relative integral area.



Figure S2-39. UV-Vis spectrum of 5 in toluene (10^{-5} mol/L) .

| | 4 | 5 | 6 |
|---|-----------------------------------|--------------------------------|--|
| Empirical formula | C56H83B2BrN5PSi | $C_{53}H_{74}B_2N_5P$ | C ₆₂ H ₈₂ BN ₅ OP |
| formula weight | 986.86 | 833.76 | 955.10 |
| crystal system | Monoclinic | Triclinic | Monoclinic |
| space group | P 1 21/n 1 | P-1 | P 1 21/c 1 |
| a/Å | 11.3609(4) | 11.6492(4) | 18.0357(6) |
| <i>b</i> /Å | 29.6600(12) | 12.9718(4) | 16.6539(6) |
| c/Å | 17.1112(6) | 18.7856(7) | 18.7110(6) |
| α/deg | 90 | 94.196(2) | 90 |
| β/deg | 100.873(2) | 94.871(2) | 90.877(2) |
| γ/deg | 90 | 115.345(2) | 90 |
| $V/Å^3$ | 5662.4(4) | 2537.23(16) | 5619.5(3) |
| Ζ | 4 | 2 | 4 |
| $\rho_{\rm calcd}/{\rm g\cdot cm^{-3}}$ | 1.158 | 1.091 | 1.129 |
| μ/mm^{-1} | 1.744 | 0.760 | 0.763 |
| F(000) | 2112.0 | 904.0 | 2068 |
| crystal size/mm ³ | 0.26 x 0.18 x 0.15 | 0.25 x 0.18 x 0.15 | 0.24 x 0.18 x 0.15 |
| θ range/deg | 2.980 - 68.403 | 2.378 - 72.078 | 2.450 - 68.465 |
| index ranges | -13<=h<=13 | -14<=h<=14 | -21<=h<=20 |
| 5 | -28<=k<=35 | -15<=k<=15 | -16<=k<=19 |
| | -18<=l<=20 | -23<=l<=23 | -21<=l<=22 |
| collected data | 53573 | 41106 | 70495 |
| unique data | 10358 | 9695 | 10292 |
| 1 | $R_{\rm int} = 0.0622$ | $R_{\rm int} = 0.0325$ | $R_{\rm int} = 0.\ 1100$ |
| completeness to θ | 99.8 % | 98.2 % | 99.9 % |
| data/restraints/parameters | 10364 / 0 / 614 | 9695 / 14 / 560 | 10292/0/ 648 |
| GOF on F^2 | 1.046 | 1.085 | 1.055 |
| final R indices $[I > 2 (I)]$ | $R_1 = 0.0364$ | $R_1 = 0.0408$ | $R_1 = 0.0688$ |
| R indices (all data) | $WK_2 = 0.0780$ $R_1 = 0.0465$ | WK2 = 0.1013 $R_1 = 0.0449$ | $WK_2 = 0.1599$ $R_1 = 0.0928$ |
| r mulices (all uala) | $wR_2 = 0.0819$ | $wR_2 = 0.1041$ | $wR_2 = 0.1724$ |
| Largest diff peak/hole (e·Å-3) | 0.345 / -0.402 | 0.435 / -0.495 | 0.470/ -0.464 |

III.X-ray crystallographic data

| 7 | 0 | |
|-----------------------------------|--|---|
| / | δ | 9 |
| $C_{67}H_{95}B_2FN_5OP_2$ | $C_{45}H_{72}B_2N_6S_2$ | $C_{28}H_{38}BN_2PS_3$ |
| 1058.06 | 890.91 | 540.56 |
| Orthorhombic | Triclinic | Monoclinic |
| P2(1)2(1)2(1) | P-1 | P 1 21/c 1 |
| 12.9560(6) | 10.3831(5) | 26.3039(11) |
| 21.9701(9) | 10.7408(5) | 11.7737(5) |
| 22.1424(10) | 24.9975(11) | 20.0625(9) |
| 90 | 84.124(3) | 90 |
| 90 | 85.546(3) | 108.695(2) |
| 90 | 72.073(3) | 90 |
| 6302.7(5) | 2635.4(2) | 5885.4(4) |
| 4 | 2 | 8 |
| 1.115 | 1.123 | 1.220 |
| 0.744 | 1.211 | 1.945 |
| 2296 | 960 | 2304 |
| 0.11 x 0.10 x 0.09 | 0.20 x 0.20 x 0.10 | 0.23 x 0.15 x 0.06 |
| 2.833 - 69.080 | 3.559 - 68.375 | 1.542 - 53.960 |
| -15<=h<=15 | -11<=h<=11 | -31<=h<=31 |
| -26<=k<=26 | -12<=k<=12 | -14<=k<=13 |
| -26<=l<=26 | -30<=l<=30 | -24<=1<=24 |
| 104625 | 56539 | 62894 |
| 11717 | 9364 | 10769 |
| $R_{\rm int} = 0.1174$ | $R_{\rm int} = 0.0853$ | $R_{\rm int} = 0.0762$ |
| 100 % | 97.2 % | 99.9 % |
| 11717 / 198 / 712 | 9364 / 0 / 593 | 10769 / 0 / 647 |
| 1.050 | 1.038 | 1.042 |
| $R_1 = 0.0721$ | $R_1 = 0.0549$ | $R_1 = 0.0422$ |
| wR2 = 0.1891 $P_{1} = 0.0878$ | wR2 = 0.1270 $P_{1} = 0.0725$ | wR2 = 0.0887 $P_1 = 0.0655$ |
| $n_1 = 0.0878$ $wR_2 = 0.2046$ | $wR_2 = 0.1352$ | $m_1 = 0.0035$ $w_{R_2} = 0.0968$ |
| 0.610 / -0.465 | 0.673 / -0.437 | 0.260 / -0.351 |
| | $C_{67}H_{95}B_{2}FN_{5}OP_{2}$ 1058.06 Orthorhombic $P2(1)2(1)2(1)$ $12.9560(6)$ $21.9701(9)$ $22.1424(10)$ 90 90 90 90 6302.7(5) 4 1.115 0.744 2296 0.11 x 0.10 x 0.09 2.833 - 69.080 -15<=h<=15 -26<=k<=26 -26<=l<=26 104625 11717 $R_{int} = 0.1174$ 100 % 11717 / 198 / 712 1.050 $R_{1} = 0.0721$ wR2 = 0.1891 $R_{1} = 0.0878$ wR2 = 0.2046 0.610 / -0.465 $T_{i} = 0.724 = 0.0721$ | $C_{67}H_{95}B_2FN_5OP_2$ $C_{45}H_{72}B_2N_6S_2$ 1058.06890.91OrthorhombicTriclinic $P2(I)2(I)2(I)$ $P-I$ 12.9560(6)10.3831(5)21.9701(9)10.7408(5)22.1424(10)24.9975(11)9084.124(3)9085.546(3)9072.073(3)6302.7(5)2635.4(2)421.1151.1230.7441.21122969600.11 x 0.10 x 0.090.20 x 0.20 x 0.102.833 - 69.0803.559 - 68.375-15<=h<=15 |

 ${}^{a}R_{1} = \sum (||F_{o}| - |F_{c}||) / \sum |F_{o}|, wR_{2} = [\sum w(F_{o}^{2} - F_{c}^{2})^{2} / \sum w(F_{o}^{2})]^{1/2}, \text{GOF} = [\sum w(F_{o}^{2} - F_{c}^{2})^{2} / (N_{o} - N_{p})]^{1/2}.$

IV. Computational details

Geometry optimizations were carried out with the Gaussian 16 package^{S4} with the M06-2X functional.^{S5} The def2-SVP basis set was used for all the atoms. Frequency calculations at the same level of theory were performed to identify the number of imaginary frequencies (zero for local minimum) and provide the Gibbs free energies. All the energies reported in the paper correspond to the reference state of 1 mol/L, 298K. Natural bond orbital (NBO) and natural resonance theory (NRT) calculations were carried out using NBO 7.0 program^{S6} at the M06-2X/def2-SVP level of theory. These NRT calculations were done for a model of 5 in which Dipp groups were replaced by hydrogen atoms, but at the geometry corresponding to 5. Optimized structures were visualized by the CYLview,^{S7} Chemcraft^{S8} or IBOview program.^{S9} EDA-NOCV and ELF calculations were carried out using Amsterdam Modeling Suite (ADF/2019.304)^{S10} at the BP86/TZP level of theory. Intrinsic bond orbitals (IBOs) were carried out using ORCA program at the M06-2X/def2-SVP level of theory.^{S9}



Figure S4-1. Depiction of selected IBOs of 5. (a) $P-B(1) \sigma$ -bonding orbital. (b) $N-B(2) \sigma$ -bonding orbital.



Figure S4-2. Frontier molecular orbitals of 5.



Figure S4-3. Four predominant resonance structures (weight > 7.0%) and their weights for the simplified model of **5** (Dipp groups were replaced with H).



Figure S4-4. ELF plot of 5 in the P(1)–B(2)–N(1) plane.



Figure S4-5. Selected NBOs of 5 for the second-order perturbation theory analysis. Energies are given in kcal/mol.

| łł | nartree | eV | kcal/mol | kJ/mol | |
|--|---|---|--|--|---|
| Pauli Repulsion Kinetic (Delta T^0): Delta V^Pauli Coulomb: Delta V^Pauli LDA-XC: Delta V^Pauli GGA-Exchange: Delta V^Pauli GGA-Correlation | 2.4248350509 -1.2911421 -0.4331464 0.08464 : -0.03464 | 982034 187406291 90493037 411534622 466387566 | 65.9831 -35.1338 -11.7865 208 2.303 501 -0.931 | 1521.61 -810.20 -271.80 32 53.11 19 -21.49 | 6366.40 -3389.89 -1137.23 222.23 -89.91 |
| Total Pauli Repulsion: (Total Pauli Repulsion = Delta E^Pauli in BB paper) | 0.750940887 | 788314 | 20.4341 | 471.22 | 1971.60 |
| Steric Interaction Pauli Repulsion (Delta E^Pauli) Electrostatic Interaction: (Electrostatic Interaction = Delta V_elstat in the BB paper) | : 0.750940 -0.391164927 | 088778831 7472801 | 4 20.434 ⁻ -10.6441 | 1 471.22 -245.46 | 1971.60 -1027.00 |
| Total Steric Interaction: (Total Steric Interaction = Delta E^0 in the BB paper) | 0.359775960 | 315513 | 9.7900 | 225.76 | 944.59 |
| Orbital Interactions A: -0.499 | 995271932120 | 6 -13.0 | 6056 -31 | 3.75 -1312 | 2.74 |
| Total Orbital Interactions: | -0.49999527 | 1932127 | -13.6056 | -313.75 | -1312.74 |
| Alternative Decomposition Orb. Kinetic: -2.22 Coulomb: 1. XC: 0.192 | Int. 62650575467 533497027922 77275769249 | 18 -60 2097 4 5.2 | 0.5798 -13 41.7286 2456 120 | 397.00 -58 962.28 4 9.97 506.1 | 45.06 026.20 12 |
| Total Orbital Interactions: | -0.49999527 | 1932127 | -13.6056 | -313.75 | -1312.74 |
| Residu (E=Steric+OrbInt+Res): | 0.00000 | 000676163 | .000 | 0.00 | 0.00 |
| Total Bonding Energy: | -0.14021924 | 14000222 | -3.8156 | -87.99 | -368.15 |

Figure S4-6. EDA-NOCV results of 5 ([NHB-P:](singlet) + [NHC=NB:](singlet)).

| | hartr | ee | eV | kcal/mol | kJ/mol | | |
|--|---|--|---|--|---|---|---|
| Pauli Repulsion Kinetic (Delta T^0): Delta V^Pauli Coulomb Delta V^Pauli LDA-XC: Delta V^Pauli GGA-Excl Delta V^Pauli GGA-Cor | 1.29 -0 nange: relation: | 990422518 0.6886114 .32111122 0.07822 -0.03235 | 83337 1555097 1214903 5528640 4721909 | 35.3722 9 -18.7 -8.73 964 2 607 -0 | 815.70 '381 -4. 79 -20' 2.1286).8804 | 0 3 32.11 1.50 49.09 -20.30 | 412.90 -1807.95 -843.08 205.38 -84.95 |
| Total Pauli Repulsion: (Total Pauli Repulsion = Delta E^Pauli in BB pap | 0.3 er) | 360523951 | 148811 | 9.1445 | 5 210.8 | 8 | 882.31 |
| Steric Interaction Pauli Repulsion (Delta E Electrostatic Interaction (Electrostatic Interaction Delta V_elstat in the BB | ^Pauli): : -0.2 = paper) | 0.3360523 298823934 | 3951488 447951 | 11 9.1 -6.255 | 445 2 [.] 4 -144. | 10.88 25 | 882.31 -603.56 |
| Total Steric Interaction: (Total Steric Interaction Delta E^0 in the BB pap | 0.1 = er) | 061700017 | 700859 | 2.8890 | 66.62 | 2 2 | 78.75 |
| Orbital Interactions A: | -0.3125581 | 87183628 | -8. | 5051 | -196.13 | -820.62 | <u>!</u> |
| Total Orbital Interaction | s: -0. | 312558187 | 7183629 | -8.505 | 51 -196 | .13 | -820.62 |
| Alternative Decomposition Kinetic: Coulomb: XC: | on Orb.Int. -1.090626 0.6295 0.1484798 | 586632222 588860293 318845416 | 29 -2 185 5 4.0 | 9.6775 17.1320 0403 | -684.38 395.07 93.17 | -2863 165 389.83 | .44 2.99 |
| Total Orbital Interaction | s: -0. | 312558187 | 7183629 | -8.505 | 51 -196 | .13 | -820.62 |
| Residu (E=Steric+OrbIn | t+Res): | -0.00000 | 1034824 | 439 0 | 0.0000 | 0.00 | 0.00 |
| Total Bonding Energy: | -0. | 206389220 | 0307209 | -5.61 | 61 -129 | 9.51 | -541.87 |

Figure S4-7. EDA-NOCV results of 5 ([NHB-P](triplet) + [NHC=NB](triplet)).

| | 5 | |
|--------------------------------------|-------------------------------------|------------------------------------|
| Energy Term | [NHB -P:](S) + [NHC=NB:](S) | [NHB -P:](T)+ [NHC=NB:](T) |
| ΔE_{int} (kcal/mol) | -87.99 | -129.51 |
| ΔE_{Pauli} (kcal/mol) | 471.22 | 210.88 |
| $\Delta E_{elstat}^{[a]}$ (kcal/mol) | -245.46(43.9%) | -144.25 (42.4%) |
| $\Delta E_{orb}^{[a]}$ (kcal/mol) | -313.75 (56.1%) | -196.13 (57.6%) |

 Table S4-1. Results of EDA analysis of compound 5 using different electronic states as interacting fragments.

^[a]The values in parentheses are the percentage contributions to the total attractive interactions $(\Delta E_{elstat} + \Delta E_{orb})$.

Cartesian Coordinates:

Compound **5**:

| Р | -2.71205000 | 5.19015400 | 14.03214300 |
|--------|--------------------------|-------------|-------------|
| N | -4.92736000 | 8.86168300 | 14.54148100 |
| N | -3.58555700 | 10.50251100 | 14.01944800 |
| N | -2.71817800 | 8.28178700 | 13.83389900 |
| N | -0.5/566600 | 5.192/4900 | 13.91955000 |
| N C | 0.32022300 5.42064200 | 5.30302300 | 13.03943900 |
| C | -3.42004200 | 6 64036600 | 13 20313300 |
| C | -3.65419400 | 9.13505700 | 14.10678600 |
| č | -1.40720100 | 2.11284500 | 14.31408400 |
| С | -5.84942700 | 6.74435500 | 13.74643700 |
| С | 0.90270600 | 7.62604000 | 14.09529200 |
| С | -2.55160000 | 11.23956500 | 13.35664300 |
| C | 0.05810800 | 6.96759900 | 11.88091300 |
| C | -5.41269/00 | 7.101/8400 | 16.15124500 |
| н | 1.55102800 | 6 23935700 | 15.49850800 |
| C | -1.60257600 | 11.92816200 | 14.13268100 |
| č | -1.83263100 | 1.19288200 | 13.33010900 |
| С | 0.12601700 | 8.30771500 | 11.48538000 |
| Н | -0.15813300 | 8.58218400 | 10.46646200 |
| С | -2.59169900 | 11.30654400 | 11.95569800 |
| C | -4.80037300 | 11.06032300 | 14.40619500 |
| H C | -4.95464100 | 12.13418400 | 14.39106300 |
| Ч | 1 22732500 | 2 155/1300 | 13.98/94900 |
| Н | 1.22732300 | 2 93711000 | 15 04057100 |
| C | -5.62659900 | 10.04392300 | 14.73833700 |
| Н | -6.65043500 | 10.03991600 | 15.09826700 |
| С | -6.35391800 | 5.48099900 | 14.05549700 |
| Н | -6.68164700 | 4.81965600 | 13.25420200 |
| C | 0.95773700 | 8.95043900 | 13.65582400 |
| П | 1.3310/100 | 9./31//800 | 14.31930100 |
| C | -1 66423700 | 11 86154500 | 15.65075400 |
| Ĥ | -2.72994100 | 11.91361800 | 15.92731000 |
| С | 1.39433900 | 4.35165400 | 13.38207600 |
| Н | 2.35459000 | 4.66687400 | 13.82028600 |
| H | 1.54643200 | 4.24256200 | 12.29011200 |
| C | 0.55733600 | 9.29307300 | 12.36591500 |
| С | -0.35128000 | 5 90051200 | 12.04390300 |
| Н | -0.44604500 | 4.94592700 | 11.41382600 |
| C | -4.79683000 | 7.91881700 | 17.27361700 |
| Н | -4.46828400 | 8.88339800 | 16.85813800 |
| С | -1.44273100 | 1.40698700 | 11.87609200 |
| H | -0.44215300 | 1.86564700 | 11.87470100 |
| C | -2.61408400 | 0.88156500 | 16.00335400 |
| Г | -2.92039100 | 5.82606000 | 16 40777400 |
| Н | -5.92317000 | 5.44181300 | 17.42994900 |
| C | 2.33721900 | 8.18596800 | 16.12478200 |
| Н | 3.19045600 | 8.37613700 | 15.45757300 |
| Н | 2.72132700 | 7.76988000 | 17.06736200 |
| H | 1.87064100 | 9.15378100 | 16.36550100 |
| C | -2.64994500 | 0.12851800 | 13.71478100 |
| Г | -2.98888100 | -0.39210000 | 12.90929100 |
| Н | -1.58775300 | 12.12392000 | 10.22954300 |
| C | -0.63434200 | 12.66963100 | 13.45201300 |
| Н | 0.11810900 | 13.22317800 | 14.01374400 |
| С | -6.40211300 | 5.03117700 | 15.37297400 |
| Н | -6.78553400 | 4.03290100 | 15.58831800 |
| C H | -3.67628200 | 10.60846000 | 11.15063300 |
| п | -4.4/066600 | 10.29691300 | 11.84/18/00 |
| H | 0.13830100 | 13.31004900 | 11.54554700 |
| Ċ | -5.71475500 | 7.22626900 | 12.30895500 |
| Н | -4.76050200 | 7.77441500 | 12.24385900 |
| С | -4.32107000 | 11.54223400 | 10.12371500 |
| Н | -3.60762100 | 11.83186200 | 9.33785800 |
| Н | -4.69977000 | 12.46027200 | 10.59557500 |

| Н | -5.16297600 | 11.03442800 | 9.63107300 |
|--------|-------------|-------------|--------------|
| C | -3.04193400 | -0.02670100 | 15.04319000 |
| Н | -3.68210100 | -0.86339100 | 15.32840200 |
| C | -1.12394000 | 10.52241200 | 16.16403300 |
| Н | -0.03820800 | 10.46433000 | 15.99618400 |
| Н | -1.57586700 | 9.66684900 | 15.64250900 |
| Н | -1.30731900 | 10.41832100 | 17.24437100 |
| C | -1.70896000 | 6.18348500 | 10.22982700 |
| Н | -1.68189800 | 7.10224800 | 9.62308700 |
| Н | -2.49065300 | 6.28808600 | 10.99690700 |
| Н | -1.99194500 | 5.35357600 | 9.56460800 |
| С | -1.31277000 | 2.91025200 | 16.74790600 |
| Н | -0.73956200 | 3.71486300 | 16.26429300 |
| С | -5.80608600 | 8.21364800 | 18.38588100 |
| Н | -6.16141200 | 7.28324200 | 18.85394800 |
| Н | -6.68368900 | 8.75224900 | 17.99935100 |
| Н | -5.34285600 | 8.82717900 | 19.17238500 |
| С | -1.36669000 | 0.11156200 | 11.06887700 |
| Н | -2.36412900 | -0.32723700 | 10.91370900 |
| Н | -0.73699400 | -0.64075400 | 11.56586400 |
| Н | -0.94432100 | 0.31249100 | 10.07357800 |
| C | 0.73290200 | 5,73307200 | 9.80398100 |
| н | 0 47364500 | 4 91475200 | 9 11511800 |
| Н | 1.71206700 | 5.51217100 | 10.25330100 |
| Н | 0.83946400 | 6 65470400 | 9 21052000 |
| B | -0.91236700 | 4 58754100 | 13 85503200 |
| Č | -0.95516000 | 13 02642800 | 16 33757700 |
| н | -1 13881600 | 12 98796200 | 17 42047400 |
| Н | -1 30607700 | 13 99877300 | 15 96279600 |
| Н | 0 13418600 | 12 97389300 | 16 19106200 |
| Ĉ | -3.55252400 | 7.20504600 | 17.81383500 |
| H | -3.03023900 | 7.83889100 | 18.54638200 |
| Н | -2.85659700 | 6.95446200 | 16.99859400 |
| Н | -3.82909100 | 6.26422600 | 18.31387500 |
| C | 0.12295500 | 7.02964900 | 16.42260900 |
| H | -0.40969700 | 7.98284200 | 16.55783300 |
| Н | 0.44684600 | 6.67608800 | 17.41423100 |
| Н | -0.58594000 | 6.30021300 | 16.00642800 |
| B | -2.55453500 | 6.95652700 | 13.89416200 |
| Č | -2.39826400 | 2.40429100 | 11.20857200 |
| H | -2.08670800 | 2.60373000 | 10.17102200 |
| Н | -2 42972100 | 3 35794600 | 11 75604000 |
| Н | -3 42042800 | 1 99430900 | 11 18829500 |
| Ĉ | -3.12800100 | 9.34173100 | 10.49068900 |
| H | -3.93115600 | 8,79445500 | 9.97220900 |
| Н | -2.67415500 | 8.67685000 | 11.23859500 |
| Н | -2.35590900 | 9.59683700 | 9.74643700 |
| C | -5 62426100 | 6.07652900 | 11 30615300 |
| H | -4.85977700 | 5.34760500 | 11.61426800 |
| Н | -5 35131300 | 6 46814000 | 10 31 520900 |
| Н | -6.58842300 | 5.55590100 | 11.19868700 |
| Ĉ | -2.48225500 | 3.56799000 | 17.48646500 |
| н | -3.08652900 | 2 81891600 | 18 02270600 |
| Н | -3 13166400 | 4 10852300 | 16 78154700 |
| Н | -2.10283400 | 4.28617900 | 18.23008800 |
| C | -6.84215900 | 8.19406600 | 11.92946100 |
| н Н | -7.81681700 | 7.68492300 | 11.97995300 |
| н | -6 70155000 | 8 56065700 | 10 90083100 |
| Н | -6 87866000 | 9.06662800 | 12 59773200 |
| C | -0 37972000 | 2 19055400 | 17 72858300 |
| й | 0.01144500 | 2.19033400 | 18 47857600 |
| Н | 0 47233100 | 1 73106200 | 17 20657400 |
| н | -0.91433100 | 1 39017500 | 18 26373700 |
| | 5.71 155100 | 1.59017500 | 10.20070700 |

Compound 5₂ dimer:

| Р | 0.00323800 | -1.31100300 | -0.41907400 |
|---|-------------|-------------|-------------|
| Ν | -0.62995500 | -4.23373100 | -1.12599900 |
| Ν | -5.10596800 | 0.59981200 | 0.36765300 |
| Ν | 0.51632600 | -4.09080200 | 0.92518100 |
| Ν | -4.17559600 | 0.33813100 | 2.37099500 |
| Ν | -2.71628000 | 0.07710900 | 0.46266600 |

| C | -3 84983400 | 0 31043600 | 0 99054700 |
|---------|-------------|-------------|-------------|
| C C | 1 88561600 | 4 58086200 | 2 27262700 |
| C | -1.88501000 | -4.38080200 | -3.27202700 |
| С | -0./0381300 | -4.24266300 | -2.55827900 |
| С | 1.45178800 | -4.03446100 | 2.01941700 |
| С | -3.10236100 | -5.25423000 | -2.64361600 |
| Н | -3.19488600 | -4.90065900 | -1.60450300 |
| C | -5 54968400 | 0.63070700 | -0 99460000 |
| C C | 4.04235400 | 1 82007000 | 2 47526500 |
| C | -4.04233400 | -1.83097000 | 5.47550500 |
| C | -6.09/88900 | 0.73159300 | 1.3383/500 |
| Н | -7.12991600 | 0.88794900 | 1.06028100 |
| С | 3.32393800 | -4.37892300 | 0.36047000 |
| Н | 2.54572200 | -4.84279400 | -0.25667000 |
| С | 0 48277800 | -4 02743300 | -3 31807400 |
| C C | 2 68548400 | 0.46227200 | 2 46558000 |
| C | -3.08348400 | -0.40327300 | 3.40338900 |
| C | -5.26288900 | -0.451/8000 | -1.8366/500 |
| С | -6.35849100 | 1.70678900 | -1.45102200 |
| С | 2.80097400 | -4.39005000 | 1.77789500 |
| С | -3.21441000 | 0.18980100 | 4.61969900 |
| С | -0.89954200 | -5,47729400 | -0.39513800 |
| н | -0.76101200 | -6 35301700 | -1 03734100 |
| 11 | -0.70101200 | -0.33301700 | -1.03/34100 |
| П | -1.93292500 | -5.50322100 | -0.008/1800 |
| С | -5.54313900 | 0.59033800 | 2.54006200 |
| Н | -5.98803600 | 0.58480800 | 3.52831300 |
| С | -0.44533400 | -3.58592300 | 3.62448800 |
| Н | -1.02066300 | -3.81558500 | 2.71917100 |
| C | -3 93141300 | -2 53032100 | 4 67689300 |
| U U | 4 22286800 | 2.55052100 | 4.07009500 |
| п | -4.23280800 | -5.57752100 | 4./1000900 |
| C | 3.62100700 | -4./3411500 | 2.85313700 |
| Н | 4.64277400 | -5.06277200 | 2.66496100 |
| С | -1.91375600 | -4.39124200 | -4.66190800 |
| Н | -2.83398100 | -4.59856400 | -5.20696600 |
| С | 0.09945800 | -5 48940200 | 0 75453600 |
| ч | 0.35158700 | 5 86616100 | 1 68672100 |
| 11 | -0.33138700 | -5.80010100 | 1.00072100 |
| П | 0.97991400 | -0.12451400 | 0.54360100 |
| С | 1.84684200 | -4.27820900 | 4.399/2300 |
| Н | 1.47229500 | -4.23262500 | 5.42361700 |
| С | -3.46177900 | -1.90723300 | 5.82858100 |
| Н | -3.38175100 | -2.47023700 | 6.76002300 |
| C | -5 68997100 | -0.41276000 | -3 16825600 |
| ч | 5 43305900 | 1 24423400 | 3 82831300 |
| 11 C | -3.43303900 | -1.24423400 | -5.82851500 |
| C | -2.89368900 | 1.6/991900 | 4.03/30300 |
| H | -2.38935200 | 1.91364400 | 3.68820300 |
| С | 0.98136800 | -3.97558000 | 3.34161700 |
| С | -4.41905600 | -4.97303900 | -3.37908300 |
| Н | -4.56340400 | -3.90945800 | -3.59739600 |
| Н | -5.26816900 | -5.32044700 | -2.77429800 |
| ч | 4 45733900 | 5 52537300 | 4 32010400 |
| | -4.45/55500 | -5.52557500 | -4.52910400 |
| C III | -0./8301800 | 2.91080200 | -0.01133200 |
| Н | -6.1/234/00 | 2.93435800 | 0.30303200 |
| С | -6.45458600 | 0.64160400 | -3.64256100 |
| Н | -6.79380200 | 0.65906500 | -4.67954800 |
| С | -6.80228700 | 1.67212100 | -2.77379300 |
| Н | -7.41913400 | 2,49209900 | -3.14516900 |
| C | 3 09258800 | 0 56013000 | 5 70224500 |
| U U | -3.09238800 | -0.30913900 | 6 70218400 |
| п | -2.72324300 | -0.09834300 | 0.70218400 |
| С | 3.15048900 | -4.68553000 | 4.16180900 |
| Н | 3.80213700 | -4.96813200 | 4.99116600 |
| С | 1.84380100 | -4.31179500 | -2.70343500 |
| Н | 1.94164500 | -3.74044400 | -1.77375000 |
| С | 0.40071000 | -3.81733800 | -4.69550100 |
| н | 1 30571900 | -3 59145600 | -5 25966900 |
| C | 1.04521400 | 1 26046700 | 4 70291500 |
| | -1.04321400 | -4.30940/00 | 4.79201300 |
| Н | -0.80895500 | -5.44151800 | 4./1548/00 |
| Н | -2.13512000 | -4.25801700 | 4.81158700 |
| Н | -0.66911100 | -4.00707600 | 5.76255100 |
| С | -0.80831600 | -3.94278600 | -5.36779200 |
| Н | -0.87106300 | -3.76170900 | -6,44207600 |
| С | -4.55781600 | -1.69130800 | -1.34709200 |
| н | _4 23175600 | -1 52908400 | -0 32167300 |
| 11 | | 1.52700400 | 0.5210/500 |

| C | 4 52976000 | 2 5 4 2 9 6 4 0 0 | 2 22420400 |
|--------|--------------|-------------------|-------------|
| C . | -4.32870000 | -2.54580400 | 2.22439400 |
| H | -4.95553500 | -1.79030000 | 1.54/64600 |
| В | 0.02944300 | -3.29811500 | -0.18084000 |
| C | -0 50546900 | -2 07060700 | 3 83053900 |
| | 0.00040000 | 1 7020(500 | 4.74522200 |
| Н | 0.04080200 | -1./9206500 | 4./4552200 |
| Н | -1.53811800 | -1.72491200 | 3.92782300 |
| Н | -0.03834800 | -1.54428100 | 2.98256000 |
| C . | 2 24401800 | 2 21400200 | 1 51214000 |
| C | -3.34401800 | -5.21499800 | 1.31314000 |
| Н | -2.52446400 | -2.50303100 | 1.32291700 |
| Н | -3.65720400 | -3.64218000 | 0.54854000 |
| ч | 2 05708400 | 4 02582700 | 2 12766200 |
| 11 | -2.93708400 | -4.03383700 | 2.13/00200 |
| C | 3.46875400 | -2.92308800 | -0.09274300 |
| Н | 2.52252200 | -2.37670300 | 0.00878400 |
| н | 3 79097500 | -2 85310700 | -1 14088800 |
| 11 | 1.10010200 | -2.03310700 | -1.14000000 |
| Н | 4.19819300 | -2.38/28/00 | 0.53156600 |
| С | -2.93396200 | -6.78588500 | -2.65200400 |
| Н | -2.88049500 | -7.13602400 | -3.69397400 |
| 11 | 2 80227200 | 7 26800500 | 2 17744000 |
| п | -3.80227300 | -7.20800300 | -2.1//44000 |
| Н | -2.02931000 | -7.13646300 | -2.14686300 |
| С | -6.57811400 | 4.24893400 | -1.33691300 |
| н | 5 57318600 | 4 3 4 3 4 7 1 0 0 | 1 76483600 |
| 11 | -3.37318000 | 4.34347100 | -1./0483000 |
| H | -6.73530900 | 5.08048600 | -0.63438/00 |
| Н | -7.30216700 | 4.37499100 | -2.15507600 |
| C | -1 96766700 | 2 06057000 | 5 79202400 |
| e u | 2.502(2700 | 2.00057000 | 6.75410000 |
| H | -2.50262/00 | 2.03362000 | 6.75410800 |
| Н | -1.60039100 | 3.08585500 | 5.64721000 |
| Н | -1.09985500 | 1.39062400 | 5.86152300 |
| D | 1 26542400 | 0.01447100 | 0.12971400 |
| Б | -1.30342400 | 0.01447100 | 0.128/1400 |
| C | 1.90512100 | -5.81995600 | -2.39952500 |
| Н | 1.14987500 | -6.12972300 | -1.66933900 |
| н | 2 89425100 | -6 10459100 | -2.01062200 |
| 11 | 2.07425100 | -0.10+39100 | -2.01002200 |
| H | 1.72910500 | -6.38412300 | -3.32831000 |
| С | -4.15653700 | 2.55322400 | 4.72881800 |
| н | -4 76819100 | 2 51101200 | 3 82035900 |
| 11 | 2 96590200 | 2 60260000 | 4 89292100 |
| п | -3.80380300 | 5.00509900 | 4.88282100 |
| Н | -4.77488100 | 2.24659800 | 5.58775400 |
| С | -5.64016100 | -3.55814100 | 2.49975200 |
| н | 5 27384000 | 4 42184300 | 3 07/20/00 |
| 11 | -3.27384000 | -4.42184300 | 3.07429400 |
| H | -6.03186100 | -3.94545800 | 1.54835600 |
| Н | -6.47157500 | -3.10091100 | 3.05548500 |
| С | 4 60750700 | -5 17422000 | 0 13624200 |
| U U | 5.47520000 | 4.71(24(00 | 0.13621200 |
| Н | 5.4/520000 | -4./1034000 | 0.03301200 |
| Н | 4.83750100 | -5.21132300 | -0.93954500 |
| Н | 4.51281200 | -6.20879100 | 0.49825900 |
| C | 8 26001200 | 2 82207400 | 0.22427200 |
| C II | -0.20701200 | 2.02207400 | -0.22427200 |
| H | -8.89056200 | 2.88185400 | -1.13058800 |
| Н | -8.54679400 | 3.66146600 | 0.43007700 |
| Н | -8.54105700 | 1.88783600 | 0.28393200 |
| C | 3 0/1578500 | 3 08030700 | 3 58638600 |
| | 3.04100000 | -5.70757/00 | -5.50050000 |
| H | 3.04100000 | -4.60198800 | -4.50081100 |
| Н | 3.96618600 | -4.23896200 | -3.03880900 |
| Н | 3.09558500 | -2.93334000 | -3.87311700 |
| C . | 5 57410500 | 2.92965200 | 1 20720200 |
| C | -5.5/410500 | -2.83803300 | -1.29/29300 |
| Н | -6.36333200 | -2.60111000 | -0.56598900 |
| Н | -5.10076900 | -3.78521300 | -0.99759900 |
| н | 6.06504400 | 2 0088/300 | 2 26675700 |
| | -0.00304400 | -2.9988-1900 | -2.20075700 |
| C | -3.28064500 | -1.94438400 | -2.144/8800 |
| Н | -3.46596800 | -2.23172000 | -3.18975900 |
| Н | -2.67895700 | -2.73150000 | -1.66919600 |
| U | 2 66000000 | 1 02478500 | 2 16550000 |
| п | -2.000999900 | -1.034/8500 | -2.10559900 |
| Р | -0.01667400 | 1.47287200 | 0.24739200 |
| Ν | -0.07453000 | 2.86501000 | -2.54535100 |
| N | 4 83221100 | 1 421 77800 | -0.45754000 |
| 1 Y | T.05221100 | 1.721//000 | -0 |
| IN | 0.25142000 | 4.26824700 | -0./2954100 |
| Ν | 4.09032600 | 1.15239300 | 1.60180400 |
| Ν | 2.72788600 | 0.24591100 | -0.17296800 |
| с. | 2 75500000 | 0.2(000000 | 0.26244500 |
| C . | 3./3398600 | 0.80808200 | 0.20344500 |
| С | 0.22054500 | 0.75475600 | -3.79817800 |
| | | | 2 40265700 |

| С | 0 14321900 | 4 90037800 | 0 54301200 |
|---------|-------------|-------------|-------------|
| C C | 1 70758000 | 0.73919000 | -3 51704100 |
| U U | 1.20082200 | 1.24774200 | -5.51704100 |
| н ~ | 1.89083200 | 1.24//4200 | -2.55846600 |
| С | 5.19672000 | 1.37143300 | -1.84507900 |
| С | 2.51745400 | 0.88327900 | 3.46026600 |
| С | 5.70509000 | 2.07983100 | 0.41231400 |
| Н | 6.56495700 | 2.61982500 | 0.02954500 |
| C | -2 38925000 | 4 49677600 | 0.54515000 |
| U U | 2.12(80200 | 4.0(200200 | 0.34313000 |
| п ~ | -2.12089200 | 4.00390300 | -0.42///900 |
| С | -1.90330800 | 1.96276400 | -3.86650200 |
| С | 3.63226600 | 0.41773300 | 2.74611300 |
| С | 5.21526800 | 2.56946700 | -2.57958700 |
| С | 5.76695500 | 0.19186300 | -2.36927100 |
| С | -1.11171600 | 5.00122000 | 1.18539800 |
| C | 4 39255600 | -0.69087600 | 3 18046100 |
| C C | 0.15260100 | 4 20008700 | 2 10815000 |
| | 0.15500100 | 4.20098700 | -5.10815000 |
| H | -0.60226400 | 4.46049100 | -3.86236600 |
| H | 1.12343600 | 4.25059600 | -3.62298000 |
| С | 5.25241700 | 1.91394100 | 1.66246700 |
| Н | 5.63537500 | 2.28373000 | 2.60627200 |
| С | 2.59887700 | 5.59263000 | 0.38652800 |
| Н | 2,49974200 | 4.96317600 | -0.50642100 |
| C | 2 08466300 | 0 13236300 | 4 55730700 |
| е н | 1 20569400 | 0.45870300 | 5 11/05800 |
| II C | 1.20309400 | 5 (2(02400 | 2 42959200 |
| | -1.19300000 | 5.05092400 | 2.42636200 |
| H ~ | -2.16144/00 | 5.69/88800 | 2.93231900 |
| С | -0.35892900 | -0.25139700 | -4.58005900 |
| Н | 0.23005900 | -1.12648100 | -4.85408700 |
| С | 0.10878900 | 5.14293200 | -1.89593800 |
| Н | 0.90587800 | 5.90345700 | -1.93466500 |
| Н | -0.85476500 | 5.68642500 | -1.83752800 |
| С | 1.14422200 | 6.16507200 | 2.35427400 |
| Н | 2.01903000 | 6.64712500 | 2,79777400 |
| C | 2 76507400 | -1 01400000 | 4 94975100 |
| U U | 2.70507400 | 1.50072600 | 5 70845800 |
| II C | 2.40083100 | -1.39972000 | 3.79643600 |
| C . | 5./4553400 | 2.551/6600 | -3.8/541000 |
| Н | 5.75744400 | 3.4/491/00 | -4.45821000 |
| С | 5.75986800 | -1.01431700 | 2.59028400 |
| Н | 5.78712200 | -0.62730700 | 1.55971900 |
| С | 1.27182200 | 5.52464900 | 1.11762100 |
| С | 2.29558700 | -0.66394100 | -3.43300800 |
| Н | 1.69410200 | -1.31333000 | -2.77839200 |
| Н | 3.32130300 | -0.62140300 | -3.03620300 |
| н | 2 33979900 | -1 11966000 | -4 43528700 |
| C C | 5.00264700 | 1.01757500 | 1 48540600 |
| | 5.99504700 | -1.01/3/300 | -1.48340000 |
| н ~ | 5.13859500 | -1.08/84400 | -0./98/4000 |
| С | 6.23445000 | 1.38067700 | -4.4344/000 |
| Н | 6.62651600 | 1.37975900 | -5.45258500 |
| С | 6.24286300 | 0.20929700 | -3.68201800 |
| Н | 6.66255900 | -0.69952200 | -4.11285300 |
| С | 3.92316600 | -1.40405700 | 4.28513000 |
| Н | 4.47296800 | -2.27668200 | 4.63710700 |
| С | -0.07435600 | 6.20855900 | 3.02193900 |
| н | -0 15502800 | 6 70417500 | 3 99076600 |
| C | -2 79941500 | 3 05363500 | -3 31340100 |
| U U | -2.79941500 | 2.02005600 | 2 06476600 |
| п | -2.20433100 | 3.93903000 | -3.004/0000 |
| C | -2.43444800 | 0.96238600 | -4.68419300 |
| Н | -3.46777800 | 1.03649800 | -5.02/39900 |
| С | 2.88734400 | 7.03268200 | -0.05458100 |
| Н | 2.05958500 | 7.44789400 | -0.64804400 |
| Н | 3.80557500 | 7.08243100 | -0.65955100 |
| Н | 3.02817900 | 7.68688900 | 0.81998000 |
| С | -1.67273100 | -0.15319300 | -5.02129900 |
| Н | -2,10262000 | -0.95319100 | -5.62785800 |
| C | 4 70353800 | 3 88347700 | -2 02366000 |
| с н | A 20002100 | 3 77750700 | 0.02155500 |
| C | 4.37672100 | 2.12230/00 | -0.20133300 |
| | 1.95510500 | 2.232/8000 | 3.1/933300 |
| п | 2.00241500 | 2.44844800 | 2.09859000 |
| В | 0.03029800 | 2.89231100 | -1.11266200 |

| С | 3.75254600 | 5.03332700 | 1.21673900 |
|--------|-------------|--------------------------|-------------|
| Н | 3.86112300 | 5.56410700 | 2.17517200 |
| Н | 4.70612400 | 5.13156700 | 0.67500100 |
| Н | 3.59183400 | 3.96770200 | 1.43127800 |
| С | 0.47366200 | 2.39213700 | 3.59246500 |
| Н | -0.12463100 | 1.53069100 | 3.26147200 |
| Н | 0.04203600 | 3.29586900 | 3.13616500 |
| Н | 0.39627600 | 2.49189900 | 4.68421900 |
| С | -3.04388400 | 3.39892500 | 1.38299500 |
| Н | -2.34939900 | 2.55858500 | 1.53106400 |
| Н | -3.94807100 | 3.01352700 | 0.88354300 |
| Н | -3.34282500 | 3.78546300 | 2.37012700 |
| С | 2.40430200 | 1.52581300 | -4.63540600 |
| Н | 2.20824000 | 1.04215500 | -5.60550600 |
| Н | 3.49027900 | 1.55323600 | -4.47498000 |
| Н | 2.02879600 | 2.55741300 | -4.69854200 |
| С | 6.08607000 | -2.32654300 | -2.26401800 |
| Н | 5.24571200 | -2.43906300 | -2.96229700 |
| Н | 6.07310500 | -3.17750100 | -1.56727500 |
| Н | 7.02332000 | -2.38789900 | -2.83778500 |
| С | 6.08878600 | -2.50409900 | 2.53965500 |
| Н | 6.15409300 | -2.93833300 | 3.54882200 |
| Н | 7.06790900 | -2.65466100 | 2.06007900 |
| Н | 5.33455100 | -3.06630900 | 1.98399900 |
| В | 1.34209900 | 0.08862200 | -0.10919600 |
| С | -3.90899900 | 3.49808900 | -4.26379800 |
| Н | -3.50409100 | 3.78979400 | -5.24435300 |
| Н | -4.44468000 | 4.36302800 | -3.84433000 |
| Н | -4.65115600 | 2.70144400 | -4.41886700 |
| С | 6.86219700 | -0.31887800 | 3.40647300 |
| Н | 6.68434400 | 0.75376200 | 3.53946900 |
| Н | 7.83993300 | -0.44887300 | 2.91767100 |
| Н | 6.91761000 | -0.77087300 | 4.40896500 |
| С | 2.77541100 | 3.28918400 | 3.93943600 |
| Н | 2.74490500 | 3.07290700 | 5.01885200 |
| Н | 2.37399400 | 4.29921500 | 3.77479500 |
| Н | 3.82840100 | 3.28040800 | 3.62402100 |
| С | -3.36500500 | 5.64988700 | 0.29015400 |
| Н | -3.69537500 | 6.10843500 | 1.23514600 |
| H | -4.26050300 | 5.28357400 | -0.23278500 |
| Н | -2.90432200 | 6.43902100 | -0.32226600 |
| C | 7.27476400 | -0.82108900 | -0.66063400 |
| H | 8.14400000 | -0.73846300 | -1.33119200 |
| H | 7.43440500 | -1.68411300 | 0.00352300 |
| H | 7.23547400 | 0.08459400 | -0.03958700 |
| C | -3.35624300 | 2.48146100 | -2.01/06600 |
| H | -3.91881300 | 1.58415300 | -2.27788600 |
| H | -4.01448/00 | 3.18216800 | -1.48398300 |
| н | -2.55190200 | 2.17123000 | -1.53448300 |
| U U | 5.78805500 | 4.96351900 | -2.03/81000 |
| п | 0.08482/00 | 4.04438000 | -1.48594000 |
| п | 5.41212800 | 5.88/1/000 | -1.3/413900 |
| п | 0.09/15/00 | 5.210/0900 | -3.00449600 |
| с н | 3.45/9/100 | 4.31084200 | -2./9449400 |
| н | 3.04/42000 | 4.33320000 5 30457000 | -3.0/040000 |
| н | 2 65603000 | 3 58006600 | -2.4/102300 |
| 11 | 2.05005900 | 5.50090000 | -2.01400900 |

VI. References

[S1] Segawa, Y.; Suzuki, Y.; Yamashita, M.; Nozaki, K. J. Am. Chem. Soc. 2008, 130, 16069–16079.

[S2] Schreiber, R. E.; Goicoechea, J. M. *Angew.Chem., Int.Ed.* 2021, *60*, 3759–3767.
[S3] a) Tamm, M.; Randoll, S.; Herdtweck, E.; Kieigrewe, N.; Kehr, G.; Erker, G.; Rieger, B. *Dalton. Trans.* 2006, *3*, 459-467. b) Liu, B.-C.; Ge, N.; Zhai, Y.-Q.; Zhang, T.; Ding, Y.-S.; Zheng, Y.-Z. *Chem. Commun.* 2019, *55*, 9355-9358.

[S4] Gaussian 16, Revision A.03, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; Li, X.; Caricato, M.; Marenich, A. V.; Bloino, J.; Janesko, B. G.; Gomperts, R.; Mennucci, B.; Hratchian, H. P.; Ortiz, J. V.; Izmaylov, A. F.; Sonnenberg, J. L.; Williams-Young, D.; Ding, F.; Lipparini, F.; Egidi, F.; Goings, J.; Peng, B.; Petrone, A.; Henderson, T.; Ranasinghe, D.; Zakrzewski, V. G.; Gao, J.; Rega, N.; Zheng, G.; Liang, W.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Throssell, K.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M. J.; Heyd, J. J.; Brothers, E. N.; Kudin, K. N.; Staroverov, V. N.; Keith, T. A.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Millam, J. M.; Klene, M.; Adamo, C.; Cammi, R.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Farkas, O.; Foresman, J. B.; Fox, D. J. Gaussian, Inc., Wallingford CT, **2016**.

[S5] Zhao, Y.; Truhlar, D. G.; Theor. Chem. Acc. 2008, 120, 215-241.

[S6] Glendening, E. D; Badenhoop, J. K.; Reed, A. E.; Carpenter, J. E.; Bohmann, J. A.; Morales, C. M.; Karafiloglou, P.; Landis, C. R.; Weinhold, F. NBO 7.0, University of Wisconsin: Madison, WI, **2018**.

[S7] Legault, C. Y. CYLview, 1.0b; Université de Sherbrooke: Sherbrooke, Quebec, Canada, **2009**; www.cylview.org.

[S8] Andrienko, G. A. ChemCraft, http://www.chemcraftprog.com.

[S9] Knizia, G. J. Chem. Theory Comput. 2013, 9, 4834-4843.

[S10] te Velde, G.; Bickelhaupt,F. M.; Baerends, E. J.; Fonseca Guerra, C.; van Gisbergen, S. J. A.; Snijders, J. G.;Ziegler, T.; *J. Comput. Chem.* **2001**, *22*, 931-967. ADF 2019.304, SCM, Theoretical Chemistry, Vrije Universiteit, Amsterdam, The Netherlands, http://www.scm.com