

Electronic Supplementary Information

A highly emissive inorganic hexamolybdenum cluster complex as a handy precursor for the preparation of new luminescent materials

Olga Efremova,^{1a} Michael A. Shestopalov,^{1b} Natalya A. Chirtsova,^{bc} Anton I. Smolentsev,^b Yuri V. Mironov,^b Noboru Kitamura,^d Konstantin A. Brylev^{*bc} and Andrew Sutherland^{*a}

¹ Equally contributed to the work reported.

^a *Chemical Engineering and Applied Chemistry, Aston University, Aston Triangle, Birmingham, B4 7ET, UK; Fax: +44 (0)121 204 3679; Tel: +44 (0)121 204 3425; E-mail: a.j.sutherland@aston.ac.uk*

^b *Nikolaev Institute of Inorganic Chemistry SB RAS, 3 Acad. Lavrentiev Ave., 630090 Novosibirsk, Russia; Fax: +7(383) 330-94-89; Tel: +7(383) 330-92-53; E-mail: kbrylev@gmail.com*

^c *Novosibirsk State University, 2 Pirogova Str., 630090 Novosibirsk, Russia*

^d *Department of Chemistry, Faculty of Science, Hokkaido University, 060-0810 Sapporo, Japan*

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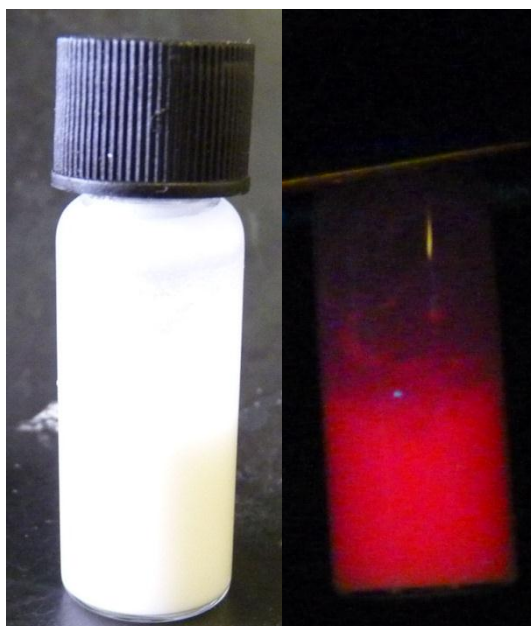


Fig. S1 A suspension of $\{\text{Mo}_6\text{I}_8\}@PS\text{-SH}$ (**2**) microspheres in water under a) day-light illumination and b) UV-light illumination.

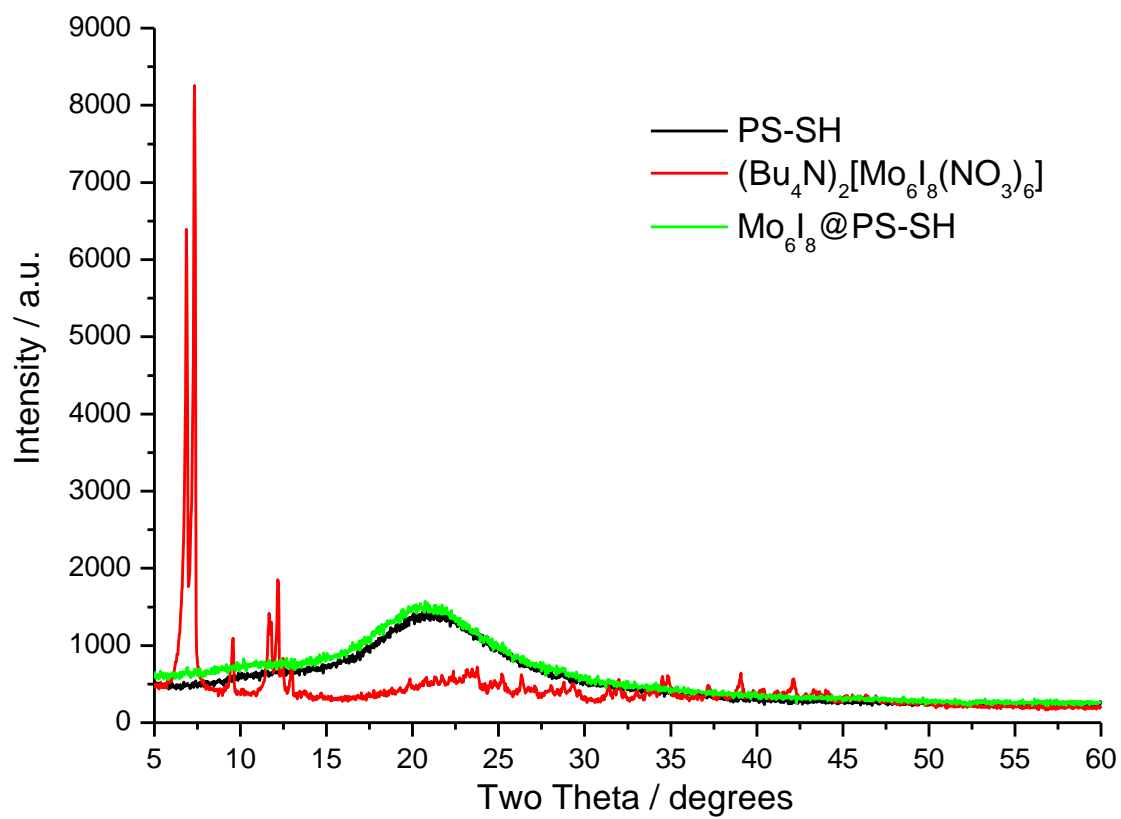


Fig. S2 The powder X-Ray diffraction patterns of **1**, **2** and untreated PS-SH beads.

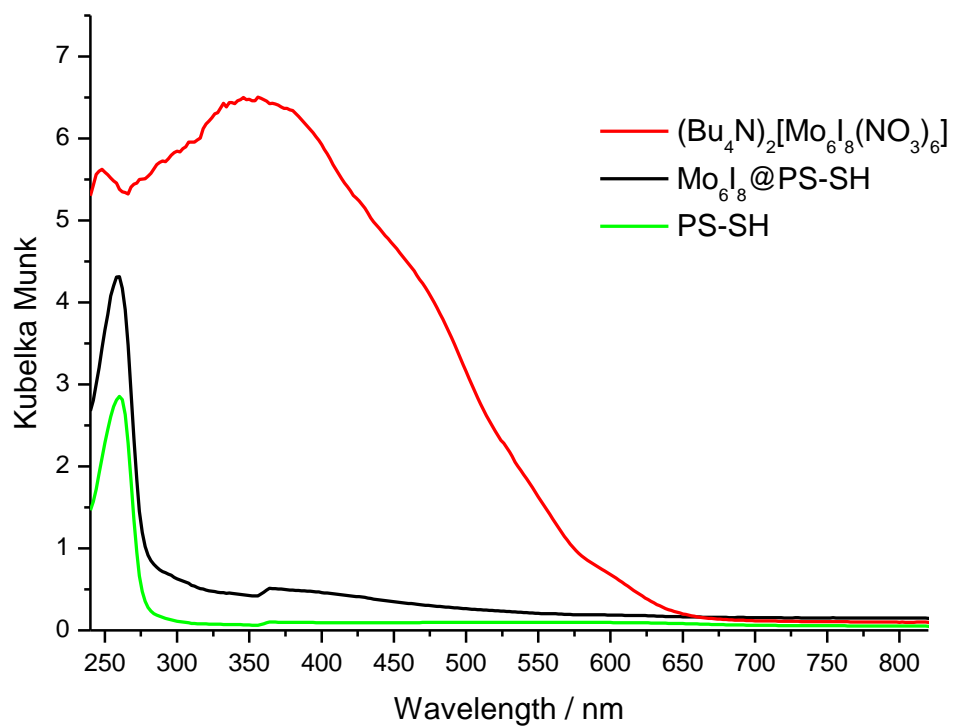


Fig. S3 Diffuse reflectance spectra of **1**, **2** and pure PS-SH beads. The Kubelka–Munk function, $F(R_\infty) = (1 - R_\infty)^2 / 2R_\infty$, is used as the equivalent of absorbance.

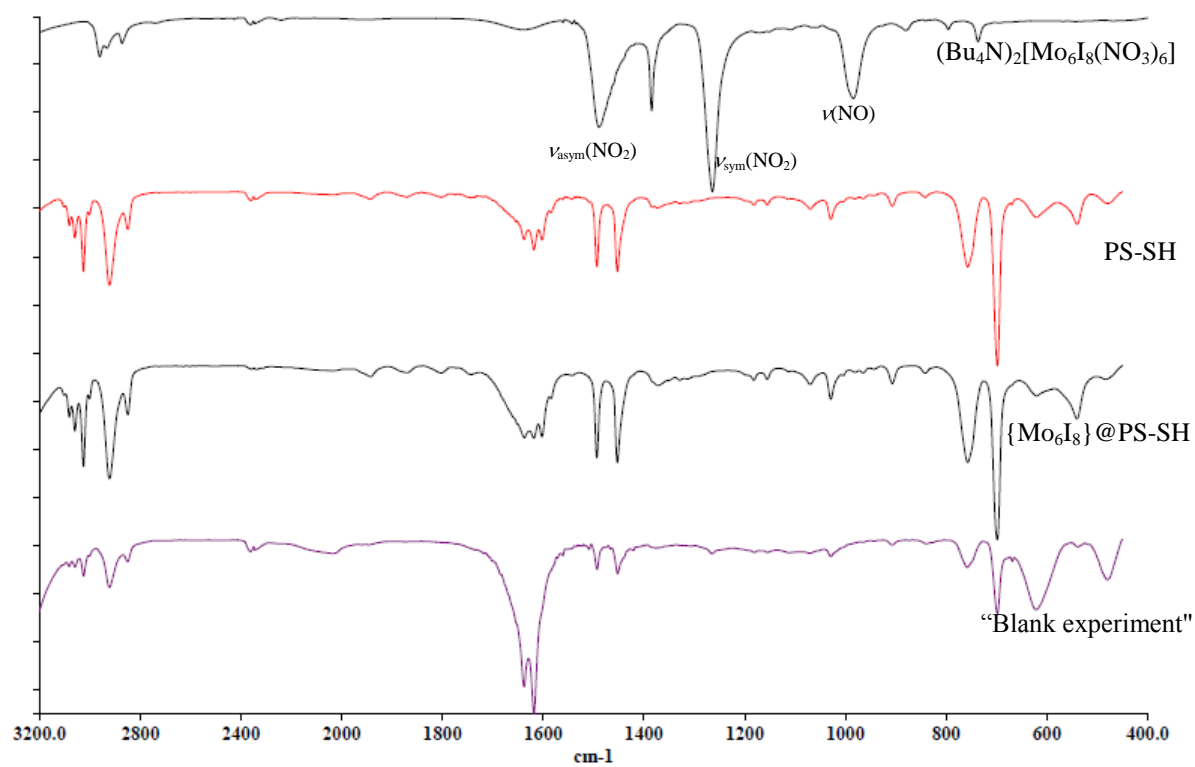


Fig. S4 The FT-IR spectra of **1**, **2**, pure PS-SH beads and PS beads after blank experiment.

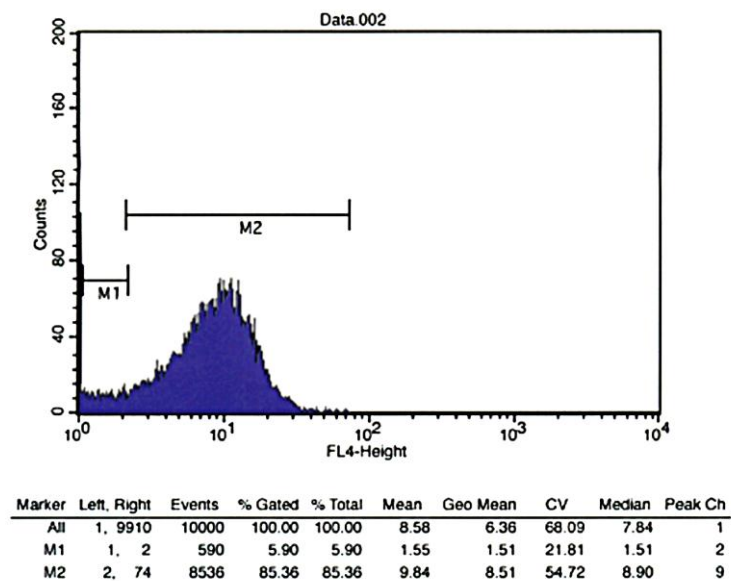
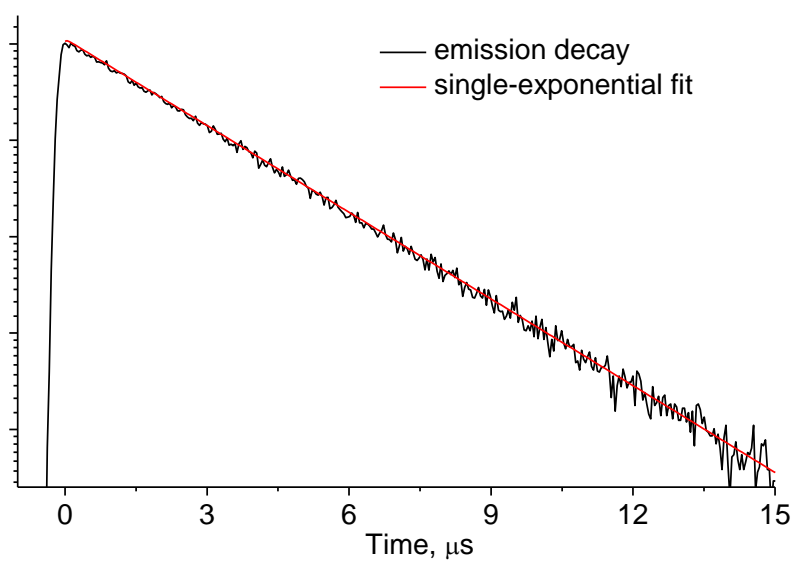
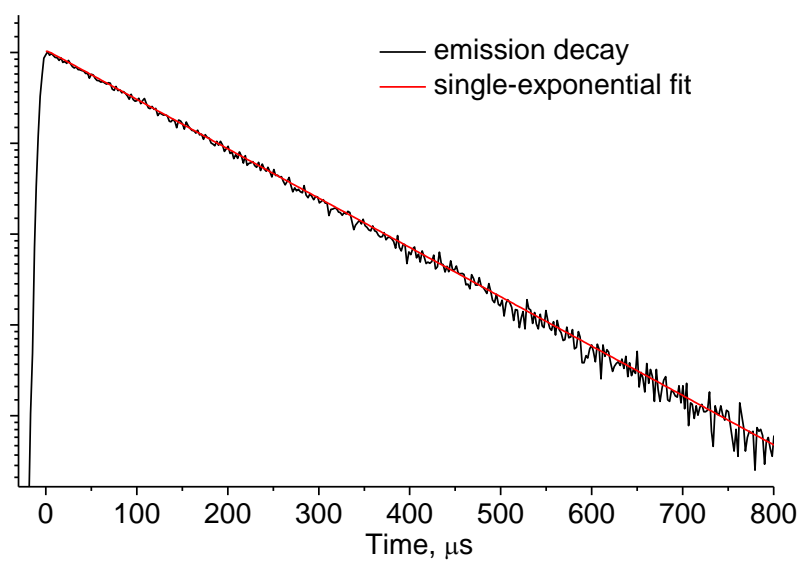


Fig. S5 Flow cytometric data for a suspension of $\{\text{Mo}_6\text{I}_8\}@\text{PS-SH}$ (2) in water.



a)



b)

Fig. S6 The emission decay profiles of solutions of **1** in (a) aerated and (b) deaerated acetone.

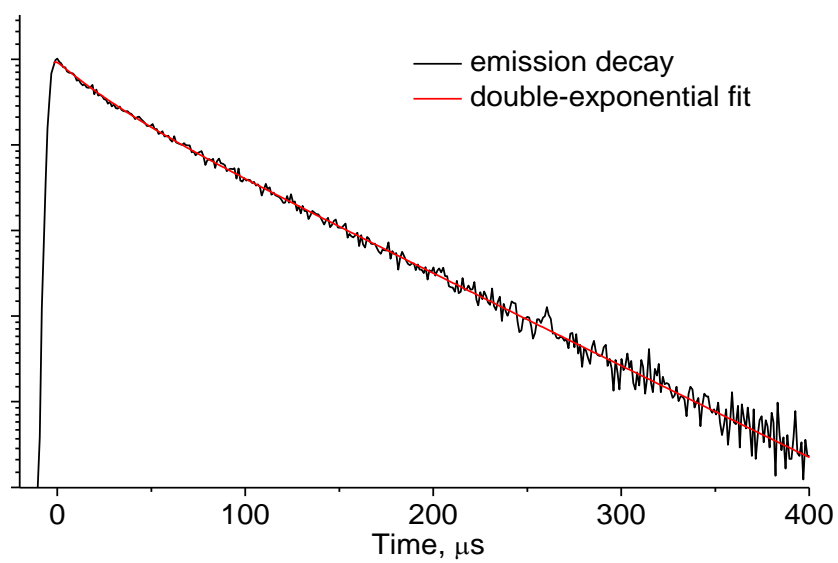
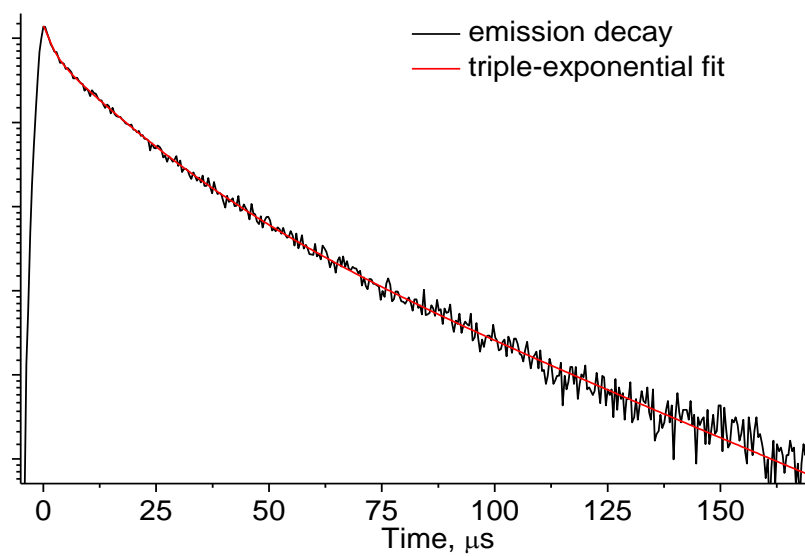
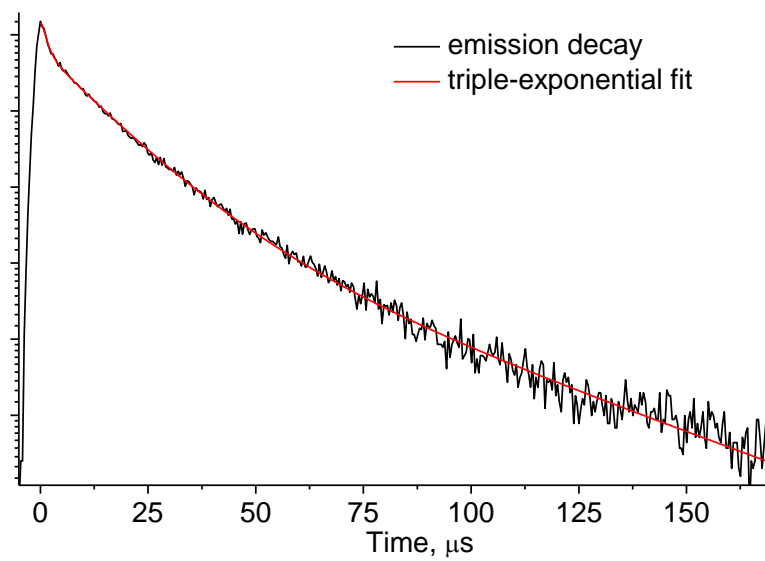


Fig. S7 The emission decay profile of a powdered sample of **1**



a)



b)

Fig. S8 The emission decay profiles of a) a powdered sample of **2** and b) a suspension of **2** in water.

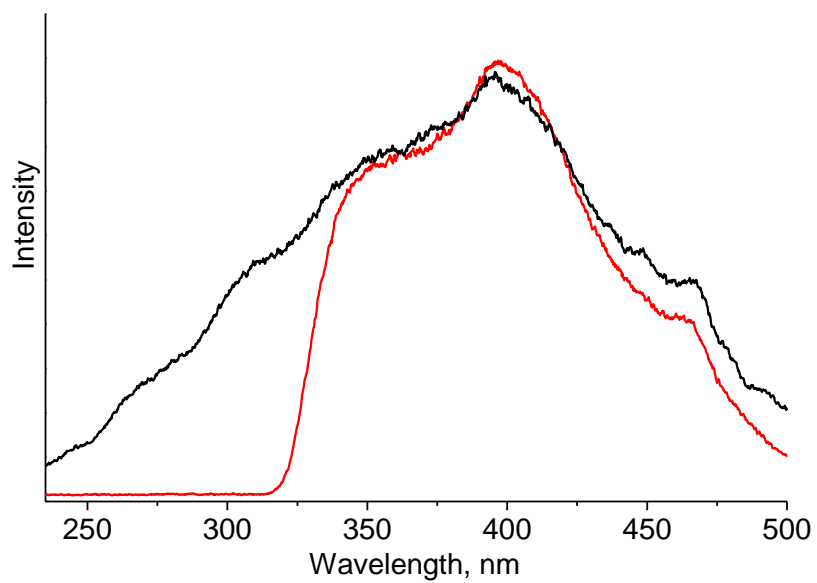


Fig. S9 Excitation spectra of **1** in an acetone solution (red line) and **2** as a suspension in water (black line).

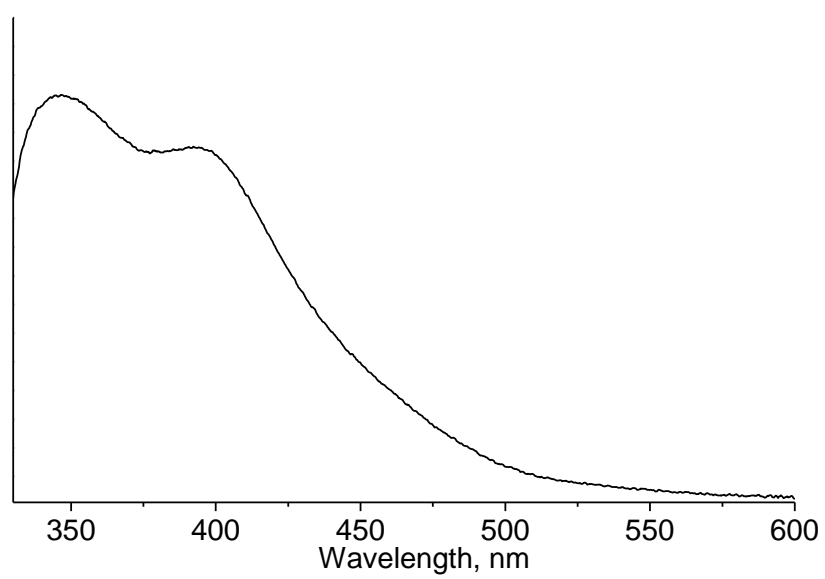


Fig. S10 UV-Vis spectrum of an acetone solution of **1**