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Published on: 01 Aug 2019 - Journal of Biobased Materials and Bioenergy (American Scientific Publishers)

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A novel amidase signature family amidase from the

marine actinomycete Salinispora arenicola CNS-205

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Abstract

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We cloned a new gene from the amidase signature (AS) family, 23 designated am, from the marine actinomycete Salinispora arenicola 24 CNS-205. As indicated by bioinformatics analysis and site-directed 25 mutagenesis, the AM protein belonged to the AS family. AM was 26 expressed, purified, and characterised in *Escherichia coli* BL21 (DE3), 27 and the AM molecular mass was determined to be 51 kDa. The optimal 28 temperature and pH were 40 °C and pH 8.0, respectively. AM exhibited a 29 wide substrate spectrum and showed amidase, aryl acylamidase, and acyl 30 transferase activities. AM had high activity towards aromatic and 31 aliphatic amides. The AM substrate specificity for anilides was very 32 narrow; only propanil could be used as an effective substrate. The 33 extensive substrate range of AM indicates it may have broad potential 34 applications in biosynthetic processes and biodegradation. 35

Keywords AM·amidase signature (AS) family·Aryl acylamidase·Acyl

transferase activity Propanil

Introduction

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Carboxylic acid amides can be hydrolysed by amidases (EC 3.5.1.4),
forming carboxylic acids and ammonia. Most amidases also produce
hydroxamic acids through their acyltransferase activity (Asano et al.
1982; Fournand et al. 1998). Amidases are very important for chemical
industrial synthesis and for control of environmental pollution.

Amidases can be divided into two categories (Chebrou et al. 1996; 44 Fournand and Arnaud 2001). The first category is the nitrilase 45 superfamily, which is characterised by a cysteine residue and includes 46 aliphatic amidases. The second category is the amidase signature (AS) 47 family, which has a conserved GGSS signature in the amino acid 48 sequence (Mayaux et al. 1990; Chebrou et al. 1996). Amidases are 49 extensively present in bacteria, archaea and eukaryotes (d'Abusco et al. 50 2001; Galadari et al. 2006; Neu et al. 2007; Ohtaki et al. 2010; Politi et al. 51 2009). 52 The marine actinomycete Salinispora arenicola CNS-205 produces 53 many bioactive natural products, including saliniketals A and B, which 54 55 were originally isolated by Fenical and co-workers in 2006 (Fenical et al. 2006). Genome sequencing of the strain S. arenicola CNS-205 identified 56 a gene encoding a putative amidase, named AM, which belongs to the AS 57 family. The amidase activity of AM was confirmed, and its catalytic 58 parameters and optimal conditions were determined. This enzyme was 59 shown to have an abnormally wide substrate spectrum and activities. 60

Materials and methods

62 Chemicals

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The chemicals used in this paper were graded as analytical reagents and purchased from J&K Scientific Company (Beijing, China).

Bacterial strains and culture conditions

66 S. arenicola CNS-205 was cultured in liquid ISP2S (0.4% yeast extract,

1% malt extract, 0.4% glucose, and 7% sea salt; pH 7.3) in an incubator

with rotation at 28 °C and 220 rpm and harvested after 2–3 days to obtain

the genomic DNA.

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Cloning of the am gene

71 The genomic DNA of *S. arenicola* CNS-205 was extracted. Genes

encoding potential AS family amidases were identified through BLASTP

analysis (http://www.ncbi.nlm.nih.gov/blast). The primers used for the

74 polymerase chain reaction of the am ORF were am-F (5'-

75 GGGCATATGGCGGTGCAGGACATCA-3') and am-R (5'-

76 CAGGAATTCCAGTTTCGTCATGCCC-3'). The *NdeI* and *EcoRI* sites

(underlined) were used to clone am into the protein expression vector

78 pET-28a(+) (Novagen).

AM expression and purification

80 Escherichia coli BL21 (DE3) carrying pET28a(+)-am was grown in

Luria-Bertani medium with 100 µg ml⁻¹ ampicillin at 37 °C (1 l medium

was inoculated with 1% inoculum from a 20 ml overnight culture). The

cultures were induced with 0.1 mM isopropyl-β-D-thiogalactopyranoside

at OD_{600} =0.6 and incubated for 20 h at 16 °C. Then, the cultures were

centrifuged at 5,000 rpm for 10 min at 4 °C. The cell pellet was

resuspended in prechilled binding buffer (20 mM Tris-HCl, pH 8.0, 0.5

mM NaCl, and 5 mM imidazole) and lysed by sonication on ice (60%

amplitude, 4 s on and 10 s off). The supernatant was harvested after 88 centrifugation at 15,000 rpm for 30 min at 4 °C. Then, the supernatant 89 was loaded onto a His-Bind Ni resin column pre-equilibrated with 90 binding buffer (GE Healthcare). An imidazole step gradient was used. 91 and the His₆-AM protein (50, 100, 200, 400, and 800 mM) was recovered 92 in elution buffer (20 mM Tris-HCl, pH 8.0, and 0.5 mM NaCl). Then, 93 SDS-PAGE and HPLC-MS (Thermo) were used to analyse the fractions 94 with His₆-AM. The supernatant containing His₆-AM was further purified 95 with PD-10 desalting columns (GE Healthcare) after it was concentrated 96 through a VIVASPIN concentrator. The protein concentration was 97 quantified with a NANODROP 2000c (Thermo), and a Thermo Hypersil 98 GOLD C4 column (1.9 µ, 100×2.1 mm) was used. The recombinant 99 proteins were assessed with HPLC-ESI-HRMS and eluted with a gradient 100 of 0.1% formic acid (A) and CH₃CN-containing 0.1% formic acid (B). 101 The elution program was 2% B for 3 min, 2 to 20% B for 1 min, 20 to 102 70% B for 16 min, 70 to 90% B for 1 min, 90% B for 4 min, 90 to 2% B 103 for 1 min, and 2% B for 4 min at a flow rate of 0.2 ml min⁻¹. A Orbitrap 104 mass spectrometer (Thermo) was used in positive ion mode, with 105 scanning from m/z 300 to 2,000. Xcalibur software (v.1.1; Thermo 106 Finnigan) was used for analysis, and the data were processed and 107 deconvoluted. 108

Amidase assays

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The purified AM was resuspended in buffer (pH 8.0, 20 mM Tris-HCl, 10% glycerol and 100 mM NaCl). The assays were performed with 1 μ g purified AM in 20 mM Tris-HCl, pH 8.0, 100 mM NaCl at 35 °C. The phenol-hypochlorite ammonia method (Weatherburn 1967) was used to assess the amidase activity, which yielded ammonia. The amount of enzyme catalysing the release of 1 μ mol NH₃/min was defined as one unit of enzyme activity. The Hanes-Woolf method was used to estimate the $K_{\rm m}$ and $V_{\rm max}$, and the $k_{\rm cat}$ and $k_{\rm cat}/K_{\rm m}$ values were determined, indicating a molecular mass of 51 kDa. Control reactions were performed without AM.

Aryl acylamidase activity assay

The aryl anilide pesticides propanil, butachlor and acetochlor were assessed as substrates to determine the aryl acylamidase activity of AM. The aryl acylamidase activity was verified following the method of Shen et al. (2012) as follows: 1 μ g of His₆-AM was added to 0.2 mM anilide in 1 ml of 20 mM Tris-HCl, pH 8.0, 100 mM NaCl and incubated at 35 °C. Addition of HCl changed the pH to 3, and the sample was extracted with ethyl acetate, terminating the reaction. This organic layer was dried and re-dissolved in methanol. Reverse-phase HPLC (Shimadzu LC-20 AD, Waters 2998 photodiode array detector) with a Thermo C18 cartridge (particle size 3 μ ; 2.1×150 mm) and 250 nm detection wavelength was used to recognise the reaction products, with 2:3 0.1% formic

acid/methanol (isocratic elution mode) for 20 min at a flow rate of 0.2 ml

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Hydroxylamine-acyl transferase activity assay

The acvl transfer activity was detected as described by Fournand et al. 135 (1998). All experiments were performed at 35 °C for 10 min, and the 136 reaction system was as follows: 1 µg of AM, hydroxylamine 137 hydrochloride (100 mM, pH 7.0), and amide or anilide (1~25 mM) in 138 buffer (20 mM Tris-HCl, pH 8.0, 100 mM NaCl). An acidic solution of 139 FeCl₃ (0.133 M in 0.68 M HCl) was adopted to terminate the reactions. 140 The supernatant was initially centrifuged at 12,000 rpm for 10 min and 141 subsequently collected, and the hydroxamate concentration was measured 142 143 at λ =500 nm. The blank control experiments were performed without AM. The optical density of the experimental groups (marked A1) was 144 determined. Additionally, the control group (marked A2) was assessed to 145 determine the concentration (C) of hydroxamate C=(A1-A2)/\(\epsilon\)L (A refers 146 to the optical density, ε denotes the coefficient of molar extinction, and L 147 indicates the layer thickness). Different substrates have different ε values: 148 propionamide, 1,029 M⁻¹ cm⁻¹; hydroxamate derivative of acetamide, 996 149 M^{-1} cm⁻¹; propanil, 1,029 M^{-1} cm⁻¹; isobutyramide, 1,016 M^{-1} cm⁻¹. The 150 control groups did not have AM. One unit of enzyme activity was defined 151 as the amount of enzyme required to catalyse the formation or hydrolysis 152 of 1 µmol of substrate or product every minute. 153

Effects of temperature and pH on enzyme activity

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For determination of the optimal temperature, the experiments were 155 performed in buffer (20 mM Tris-HCl, pH 8.0, 100 mM NaCl) with 20 156 mM of the substrate benzamide. Thermo stability was detected by pre-157 incubation of the protein for 1 h at different temperatures. Then, the 158 residual activity was tested at 35 °C. For determination of the optimal pH, 159 the experiments were performed in various buffers as follows: 0.1 M 160 sodium acetate buffer (pH 4.0, 4.5, 5.0, 5.5, and 6.0), 0.1 M potassiumphosphate buffer (pH 6.0, 6.5, 7.0, 7.5, 7.8, and 8.5), and 0.1 M Tris-HCl 162 buffer (pH 7.5, 8.0, 8.5, 9.0, 9.5, and 10.0). For the pH stability detection, 163 the experiments were performed in buffers with pH values that ranged 164 from 4.0 to 10.0 with incubation at 25 °C for 1 h; then, the residual 165 protein activity was tested at 35 °C. 166

Impact of metal ions and other reagents

The impact of metal ions (Ni²⁺, Ba²⁺, Zn²⁺ and Ca²⁺) and chemical agents 168 (1,10-phenanthroline, EDTA, SDS and PMSF) on the amidase activity 169 was detected. The samples were preincubated for 10 min at 35 °C with 170 benzamide as a substrate, and then, the amidase activity was determined 171

as described previously.

Site-directed mutagenesis of am

Site-directed mutagenesis primer pairs (Table 1) were designed to produce mutated am with a QuikChange site-directed mutagenesis kit (Stratagene). The recombinant plasmid pET-28a(+)-am served as a template in the mutagenesis reactions. The PCR products were purified by agarose gel electrophoresis, and the DNA bands with the appropriate sizes were eluted from the gel pieces. Then, the plasmid DNA was digested with *Dpn*I and transformed into competent *E. coli* BL21 (DE3) cells. Three mutant (K84A, S158A, and S182A) plasmids were constructed with this technique and were verified through DNA sequencing. The mutant proteins were expressed, purified, and analysed as described above.

Nucleotide sequence accession number

The nucleotide sequence of *am* was stored in GenBank with a sequence

187 ID of CP000850.1.

Results

Sequence analysis of the am gene

The amino acid sequence of AM was compared to sequences of known amidases accessible in the GenBank database. The comparisons showed that AM shared 29–45% identity with several enzymes, including an amidase from *Rhodococcus* sp. N771 (45% identity), an amidase from *Thermus thermophilus* HB8 (31% identity), the aspartyl/glutamyl-tRNA amidotransferase subunit A from *T. thermophilus* HB8 (33% identity), and the aspartyl/glutamyl-tRNA amidotransferase subunit A from *Thermotoga maritima* MSB8 (28% identity). Additionally, we confirmed

the presence of Ser-Ser-Lys, the highly conserved catalytic triad of the AS family, in the enzyme amino acid sequence (Fig 1).

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Fig 1. Comparison of the amino acid sequences of the AM and homologous proteins. Sequence alignment of the amino acid sequence of AM showing the high homology with the AS amidase (3A1K A) from *Rhodococcus* Sp. N771, the *Thermus* HB8 amidase (YP 145063.1), the Т. thermophilus thermophilus HB8 aspartyl/glutamyl-tRNA amidotransferase subunit A (YP 143839.1), and the Thermotoga maritima MSB8 aspartyl/glutamyl-tRNA amidotransferase subunit A (NP 229077.1). The alignment was produced by ClustalW. Multiple alignments were generated with BioEdit. The dark grey under the sequence indicates the residues of the Ser-Ser-Lys catalytic triad.

We demonstrated that this motif was the catalytic site through site-directed mutagenesis (Table 1). Thus, we showed that AM of *S. arenicola* CNS-205 contains the highly conserved catalytic triad Ser-Ser-Lys and is a member of the AS family.

Table 1. Oligonucleotides used for site-directed mutagenesis

AM gene allele	Mutagenic primer sets
	GTG CCG GTG GCG GTC GCG GAG AAC ACC GCT GTG
K84A	G
K04A	(K)
	C CAC AGC GGT GTT CTC CGC GAC CGC CAC CGG CAC
C150 A	CGT ACC CCC GGT GGC GCG TCC GGT GGA TCG GCC G
S158A	(S)

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C GGC CGA TCC ACC GGA CGC GCC ACC GGG GGT

ACG AAC GAC GGT CTC GGA GCG ATC CGG ATC CCG GCA S182A (S) C TGC CGG GAT CCG GAT CGC TCC GAG ACC GTC GTT The boldface type indicates the mutated amino acids. **Expression and purification of AM** The fusion protein His₆-AM was overexpressed in E. coli BL21 (DE3). The purity of the purified fusion protein was greater than 90%. The SDS-PAGE results indicated that the molecular mass of the major band was 51.2 kDa (Fig 2), which conformed to mass of the deduced protein sequence. Fig 2. SDS-PAGE of AM-1 and its mutants. Lanes 1, protein molecular weight marker; 2, wild-type AM-1; 3, K84A; 4, S158A; and 5, S182A To obtain the molecular mass of the proteins, we analysed the His₆-AM fusion protein and three mutants (K84A, S158A, and S182A) by HPLC-ESI-HRMS. The molecular weights of the AM wild-type, K84A, S158A, and S182A were 51.037, 50.980, 51.021, and 51.021 kDa, which were consistent with the predicted values (Fig 3).

Fig 3. HPLC-ESI-HRMS analysis of AM. WT (a), K84A (b), S158A (c) and S182A 233 (d). Extra minor peaks marked by asterisks denote glycosylation (+178 Da) of the N-234 terminal His₆-tag added during the expression of the recombinant protein in E. coli 235 (Geoghegan et al. 1999) 236 237 Effects of temperature and pH on AM activity and stability 238 To determine the optimal temperature, we assessed the amidase activity at 239 a temperature range from 15 to 65 °C with benzamide as the substrate. 240 The AM activity peaked at 40 °C, and it showed an excessively wide 241 peak (Fig 4a). More than 50% of the residual activity was observed at 242 temperatures from 30 to 50 °C. Thermo stability tests indicated moderate 243 loss of amidase activity within 1 h up to approximately 45 °C (Fig 4b). 244 Only 9% of the activity was observed at 55 °C after 1 h, and the amidase 245 activity was lost completely after 1 h at 60 °C. 246 The optimal pH for AM was determined in the buffers described in the 247 Materials and methods. Figure 4c indicates that AM was highly active 248 between pH 7.5 and 8.5. AM showed low activity below pH 4.5 or above 249 pH 10.0. For the pH stability test, AM was preincubated for 1 h at 250 different pH values, and the results indicated that more than 60% residual 251 activity was observed between pH 4.5 and 10.0 (Fig 4d). 252 253

Fig 4. Effects of pH and temperature on AM activity and stability. a Determination of the optimal pH value. The reactions were performed at 35 °C for 10 min in buffers with varying pH values. b pH stability. The assays were performed in 20 mM Tris-HCl, 100 mM NaCl buffer (pH 8.0) at 35 °C for 10 min after pretreatment of the purified enzyme at 25 °C for 1 h in 0.1 M buffer (pH 4.0-10.0). c Determination of the optimal temperature. The activity was measured in 20 mM Tris-HCl, 100 mM NaCl, pH 8.0, at 15-65 °C for 10 min. d Thermal stability. The reactions were performed under optimal conditions after incubation of AM at the indicated temperature for 1 h.

Impact of metal ions and other reagents

The majority of the metal ions, including Ba²⁺, Ca²⁺, Zn²⁺, and Ni²⁺, in the assays exerted no noticeable effect on the amidase activity of AM (Table 2).

Table 2. Effects of metal ions and inhibitors on the amidase activity of AM

Reagent	Relative activity	Reagent	Relative activity
(1 mM)	(%)	(1 mM)	(%)
No addition	100	PMSF	10.13 ± 0.60
Ba ²⁺	94.30 ± 2.59	1,10-Phenanthroline	82.46 ± 0.73
Ca ²⁺	88.60 ± 2.13	EDTA	77.64 ± 0.92
$\mathbb{Z}n^{2+}$	85.09 ± 3.87	SDS	51.78 ± 0.27
Ni ²⁺	96.93 ± 1.90		

The activity was strongly inactivated by PMSF, which is an inhibitor of

serine hydrolases. However, the chelating agents EDTA and 1,10-phenanthroline (10 mM) resulted in only a 20–30% inhibition of AM hydrolysis, demonstrating that these chemicals did not chelate a possible divalent cation(s) required for the activity of the enzyme. The surfactant SDS showed a 48.22% inhibition of AM activity.

Substrate spectrum

To determine the substrate specificity of AM, we assessed whether the purified AM could hydrolyse different aromatic and aliphatic amides. The results showed that AM had high activity towards the aromatic and aliphatic amides, including acetamide, propionamide, propanil, benzeneacetamide and benzamide (Table 3).

Table 3. Substrate spectrum of AM

Substrate	Relative activity a	Substrate	Relative activity ^a
Substrate	(%)	Substrate	(%)
Acetamide	103.79 ± 0.46	Propanil	95.91 ± 0.42
Benzamide	100	Propanamide	106.36 ± 0.38
Benzeneacetami	107.75 ± 0.48	Isobutyrami	18.02 ± 0.72
d	107.73 ± 0.48	de	18.02 ± 0.72
Nicotinamide	56.42 ± 1.29	Acetochlor	0
Pyrazinamide	43.01 ± 2.76	Butachlor	0

^a The activity with benzamide was defined as 100%. All measurements were performed in triplicate

The aromatic amides, including nicotinamide and pyrazinamide, with substitutions of one or two carbons in the ring by a nitrogen, had a negative impact on the activity. The anilide substrate range of AM was very narrow, and the protein could not hydrolyse butachlor and acetochlor. Only propanil was a good substrate for AM.

Kinetic parameters for AM were estimated by the Hanes-Woolf method.

The $K_{\rm m}$ values for acetamide and propionamide were 3.36±0.17 mM and

 $3.33\pm0.08 \text{ mM}$ (Table 4).

Table 4. Kinetic parameters for AM amidase reaction

Substrate	K (mM)	$V_{ m max}$	l. (a-1)	$k_{\rm cat}/K_{ m m}$						
Substrate	$K_{\rm m}$ (mM)	(mM min ⁻¹)	$k_{\rm cat}$ (s ⁻¹)	$(mM^{-1} s^{-1})$						
Acetamide	3.36±0.17	0.491±0.001	63.89±0.20	19.06±0.94						
Nicotinamide	6.20±0.04	0.293 ± 0.006	38.19±0.75	6.16±0.12						
Propionamide	3.33±0.08	0.503 ± 0.001	65.54±0.08	19.69±0.43						
Benzamide	3.49±0.12	0.508 ± 0.005	66.15±0.66	18.95±0.72						

^a All substrates were tested at 25-200 mM, except for benzamide, which was insoluble above 50 mM

The anilide substrate range of AM was very narrow, and the protein could not hydrolyse acetochlor and butachlor. Only propanil was a good substrate for AM, but the $K_{\rm m}$ value could not be determined due to the low solubility of the compound.

Site-directed mutagenesis

The potential catalytic active site residues of AM were replaced by the QuikChange site-directed mutagenesis kit. The mutants were overexpressed in *E. coli* BL21 (DE3) cells and further purified as described above. The results indicated that the AM K84, S158 and S182 mutants had no activity with benzamide as a substrate. These results suggest that AM is a member of the AS family and utilises the highly

conserved catalytic triad Ser-Ser-Lys.

Discussion

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An isolate of S. arenicola CNS-205 identified by Fenical (Fenical et al. 2006) and co-workers in 2006 encodes a putative amidase. Sequence alignments of the primary AM sequence indicated that AM had a high similarity with the AS family and showed that AM contains the central GGSS signature, which is a typical characteristic of the AS family. The point mutation results also indicated that no hydrolytic activity could be detected in the K84A, S158A, and S182A mutants. These findings indicated that AM belongs to the AS family. The effects of different metal ions and chemical reagents on AM activity were different. AM activity was affected by reducing agents, such as PMSF, and the results revealed that serine was the active site of the amidase. This result was consistent with the crucial role of Ser¹⁸³, as revealed by site-directed mutagenesis experiments. A metal chelating agent (EDTA) did not impact the activity, indicating that a possible divalent cation(s) required for enzyme activity was not chelated by these chemicals. Analysis of the substrate specificity of AM showed that the enzyme had high activity against short-chain aliphatic amide substrates (acetamide, isobutyramide and propanamide), which are typical substrates of the AS family. Interestingly, AM also hydrolyses ring amide substrates, such as

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aromatic and heterocyclic amides. The hydrolytic product of nicotinamide is nicotinic acid, a water-soluble B-complex vitamin, which has been extensively applied in treatment of schizophrenia, autoimmune diseases, hypercholesterolemia, diabetes and osteoarthritis. Benzoic acid, the hydrolytic product of benzamide, has antifungal activity and is extensively used as a preservative in production of processed and convenience foods. AM also had aryl acylamidase activity against aniline substrates, including propanil (a commercial amide-containing pesticide), which was hydrolysed to produce 3,4-dichloroaniline. However, acetochlor and butachlor, which are structurally analogous to propanil, were not substrates for AM, indicating the anilide substrate range of AM was very narrow. Propanil, an acyl anilide herbicide, can contaminate the soil environment, and AM, by attacking the amide bonds in propanil, can reduce its concentrations in soil. Thus, AM may have potential applications in bioremediation. In addition to the amidase and aryl acylamidase activities, acyl transferase activity is an important characteristic as it produces hydroxamic acids (Fournand et al. 1998). In this study, AM from S. arenicola CNS-205 had acyl transferase activity on anilide substrates, including propanil. The extensive substrate specificity range and acyl

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transferase activity indicate that AM has broad potential applications in biosynthesis processes and biodegradation. Overall, a new amidase gene, AM, was cloned from S. arenicola CNS-205, and the amidase, arvl acvlamidase, and acvl transferase activities of the enzyme were verified. These activities indicate that AM has a broad substrate spectrum. AM may be a potential agent for environmental remediation and for the biosynthesis of novel amides by virtue of these characteristics, as well as the broad pH tolerance of the enzyme. References Asano Y, Tachibana M, Tani Y, Yamada H, Purification and characterization of amidase which participates in nitrile degradation. Agric Biol Chem. 1982; 46:1175–1181. Chebrou H, Bigey F, Arnaud A, Galzy P, Study of the amidase signature group. Biochim Biophys Acta. 1996; 1298:285–293. Fournand D, Arnaud A, Aliphatic and enantioselective amidases: from hydrolysis to acyl transfer activity. J Appl Microbiol. 2001; 91:381–393. Fournand D, Bigey F, Arnaud A, Acyl transfer activity of an amidase from *Rhodococcus* sp. strain R312: formation of a wide range of hydroxamic acids. Appl Environ Microbiol. 1998; 64: 2844-2852. Geoghegan KF, Dixon HB, Rosner PJ, Hoth LR, Lanzetti AJ, Borzilleri KA, Marr ES, Pezzullo LH, Martin LB, LeMotte PK, McColl AS,

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Salinispora arenicola CNS-205	73	-	-		-	-	- L	. P	L	A G	٧	P	Ø A	V	K	E۱	I T	A	V	A C	L	P	T١	1 V	1 G	S	A	A A	R	S	P١	/ A	106	
Rhodococcus Sp. N771	79	S	11	PP	T	S	DO	٩V	L	TC	R	R	/ A		K	DI	I۷	T	٧	A	٧	P	MI	1 N	I G	S	R	Γ۷	E	G	F	T P	118	1
Thermus thermophilus HB8	63							P	L	HG	L	P	T	٧	ĸ	DL	.F	P	V	K	30	P	TI	R	4 G	I	K	AP	L	P	Pl	- P	95	
Thermus thermophilus HB8	57		•						L	AG	L	VΝ	/ A	٧	K	DI	11	A	T	R	L	R	T	T	A G	S	RI	L	E	N	F١	P	88	
Thermotoga maritima MSB8	57	•	•	•		•			F	WC	1	P	A	1	K	D	11	L	T	L	M	R	T	TC	CA	S	R	L	E	N	YE	S	88	
Salinispora arenicola CNS-205	107	E	A	DH	IE	٧	V	R	L	RG	A	G	A V		A	GV	/ T	R	М	PE	L	G	LI	NO	3 1	Т	DI) .	A	T	A١	/ T	145	5
Rhodococcus Sp. N771	119	S	R	D A	T	٧	V	R	L	LA	A	G	A T	٧	A	GH	(A	٧	C	E	L	C	F	SC	3 5	S	F	ΓP	A	S	GF	V	158	1
Thermus thermophilus HB8	95		E	E	R	A	V	R	L	RE	A	G	AL	L	F	AH	T	N	M	H	1	A	L	G I	ı	G	EI	IP	W	T	GF	V	134	1
Thermus thermophilus HB8	89	P	Y	EA	T	A	V I	R	L	KA	L	G	AL	٧	L	G H	T	N	L	DE	F	G	M	3 5	S	T	EH	15	A	F	FF	T	128	1
Thermotoga maritima MSB8	89	V	F	D A	T	٧	V	K	M	KE	A	G	V	٧	۷	GH	(A	N	L	DE	F	A	M	3 5	S	T	EF	2 5	A	F	FF	T	128	
Salinispora arenicola CNS-205	146	R	N	PV	E	L	G	T	P	GG	S	S (GG	S	A	A A	V	A	A	GL	. v	P	1	A	A h	N	D	3 L	G	S	I	3 1	185	5
Rhodococcus Sp. N771	159	R	N	PV	/D	R	Q	E	A	GG	S	S	GG	S	A	A L	V.	A	N	GE	V	D	F	A I	G	G	D (20	G	S		₹ []	198	1
Thermus thermophilus HB8	135	R	N	A V	D	P	S	Q	A	GG	S	S	GG	S	A	VA	V	A	L	GI	G	L	A	SL	G	T	D	C	G	S	ij,	₹ []	174	
Thermus thermophilus HB8	129	K	N	PF	D	P	D	٧	P	GG	S	SC	GG	S	A	A A	L	A	A	DL	. A	P	L	A L	G	S	D	F	G	S	۷	Q	168	1
Thermotoga maritima MSB8	129	R	N	PV	/D	LI	E	V	P	GG	S	S	GG	S	A	A A	V	S	A	GI	IV	٧	A	A L	G	S	D	G	G	S	۷	Q	168	
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