Supplementary

A site energy distribution function from Toth isotherm for adsorption of gases on heterogeneous surfaces

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## Derivation of site energy distribution function based on a Toth isothem (Type 1) by condensation approximation method

The Toth isotherm involving three parameters is given by: <sup>1-2</sup>

$$q_e = \frac{q_m p}{\left(b + p^t\right)^{1/t}} \tag{1}$$

where,  $q_e$  is the amount adsorbed at equilibrium,  $q_m$  is the maximum adsorption capacity, b is a constant related to the binding affinity and is specific to adsorbate-adsorbent pairs and t is the Toth isotherm exponent related to surface heterogeneity, usually less than or equal to unity. The t equal to unity suggests a homogeneous adsorption process.

For convenience of determining the site energy distribution, let us replace  $q_m$  by a.

The site energy distribution function,  $f(E^*)$  according to a *condensation approximation* method is given by <sup>3</sup>

 $f(E^*) = -dq_e(E^*)/dE^*$  (2)

The site energy distribution of an adsorbent material based on Toth isotherm can be obtained by differentiating local isotherm applied with respect to  $E^*$ 

The equilibrium pressure in eq (1) can be related to the energy of adsorption using the Polanyi potential theory given by <sup>3,4</sup>

$$p = p_s \exp\left(-\frac{E^*}{RT}\right) = p_s \exp\left(-\frac{E - E_s}{RT}\right)$$
(3)

The *E*<sub>s</sub> reference state for *E* represents the lowest physically realizable sorption energy, and its magnitude depends only on the

solute and is independent of the adsorbent <sup>4</sup>.

substituting eq (3) in (1) we get

$$amount.adsorbed = \frac{ap_s \exp\left(-\frac{E^*}{RT}\right)}{\left(b + \left(p_s \exp\left(-\frac{E^*}{RT}\right)\right)^t\right)^{1/t}}$$
(4)

for convenience, let us rewrite,  $-E^* = x$ ,  $p = p_s$  and RT = r, thus eq (4) can be written as:

$$amount.adsorbed = \frac{ap \exp\left(-\frac{x}{r}\right)}{\left(b + \left(p \exp\left(-\frac{x}{r}\right)\right)^{t}\right)^{1/t}}$$
(5)

substituting eq(5) in (4), the site energy distribution function can be written as:

$$f(x) = \frac{d}{dx} \left( pa \exp\left(-\frac{x}{r}\right) \left( b + \left( p \exp\left(-\frac{x}{r}\right) \right)^t \right)^{-1/t} \right)$$
(6)

Let 
$$u = \exp(-x/r)$$
 and  $v = \left(b + \left(p \exp\left(-\frac{x}{r}\right)\right)^t\right)^{-1/t}$ , then from the product rule:  $\frac{d(uv)}{dx} = \frac{du}{dx}v + u\frac{dv}{dx}$ , we get

$$= pa\left(b + \left(p\exp\left(-\frac{x}{r}\right)\right)^{t}\right)^{-1/t} \frac{d}{dx}\left(\exp\left(-\frac{x}{r}\right)\right) + \exp\left(-\frac{x}{r}\right) \frac{d}{dx}\left(\left(b + \left(p\exp\left(-\frac{x}{r}\right)\right)^{t}\right)^{-1/t}\right)$$
(7)

if u = -x/r, then applying the chain rule of differentiation,  $\frac{de^u}{dx} = e^u \frac{du}{dx}$ , and applying the rule:  $d(x^n) = nx^{n-1}$ , we get

$$= pa\left(\exp\left(-\frac{x}{r}\right)\frac{d}{dx}\left(b + \left(p\exp\left(-\frac{x}{r}\right)\right)^{t}\right)\right)^{-1/t} - \frac{\exp\left(-\frac{x}{r}\right)\left(\left(b + \left(p\exp\left(-\frac{x}{r}\right)\right)^{t}\right)^{-1/t}\right)}{r}\right)$$

Let u=-x/r, then by applying the chain rule, we get

$$= pa\left(p\left(-\exp\left(-\frac{2x}{r}\right)\left(p\exp\left(-\frac{x}{r}\right)\right)^{t-1}\right)\left(b+\left(p\exp\left(-\frac{x}{r}\right)\right)t\right)^{-\frac{1}{t}}\frac{d}{dx}\left(-\frac{x}{r}\right)\right) - \frac{\exp\left(-\frac{x}{r}\right)\left(\left(b+\left(p\exp\left(-\frac{x}{r}\right)\right)^{t}\right)^{-1/t}\right)}{r}\right)^{-\frac{1}{t}}\frac{d}{dx}\left(-\frac{x}{r}\right)}{r}\right)$$

Applying the rule:  $d(x^n) = nx^{n-1}$ , the above equation can be further simplified to

$$f(E^*) = -bpa \exp\left(-\frac{x}{r}\right) \left(b + \left(p \exp\left(-\frac{x}{r}\right)\right)^t\right)^{-\frac{t+1}{t}}$$
(8)

Replacing the parameters:  $x = E^*$ ,  $p = p_s$ , r = RT, and a = q, the site energy distribution function following a Toth isotherm explained in eq (1) is given by:

$$f(E^*) = -bp_s q \exp\left(-\frac{E^*}{RT}\right) \left(b + \left(p_s \exp\left(-\frac{E^*}{r}\right)\right)^t\right)^{-\frac{t+1}{t}}$$
(9)

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## Derivation of site energy distribution function based on a Toth isothem (Type 2) by condensation approximation method

The Toth isotherm involving three parameters is given by: <sup>1-2</sup>

$$q_e = \frac{q_m bp}{\left(1 + \left(bp\right)^t\right)^{1/t}} \tag{1}$$

where,  $q_m$  is the maximum adsorption capacity, b is a constant related to the binding affinity and is specific to adsorbate-adsorbent pairs and t is the Toth isotherm exponent related to surface heterogeneity, usually less than or equal to unity. The t equal to unity suggests a homogeneous adsorption process. For convenience, let us replace  $q_m$  by q.

The site energy distribution function,  $f(E^*)$  is given by <sup>3</sup>

$$f(E^*) = -dq_e(E^*)/dE^*$$
 (2)

The site energy distribution of an adsorbent material based on <u>Toth</u> isotherm can be obtained by differentiating local isotherm applied with respect to  $E^*$ 

The equilibrium pressure in eq (1) can be related to the energy of adsorption using the Polanyi potential theory given by <sup>3-4</sup>

$$p = p_s \exp\left(-\frac{E^*}{RT}\right) = p_s \exp\left(-\frac{E - E_s}{RT}\right)$$
(3)

substituting eq (3) in (1) we get

$$amount.adsorbed = \frac{q_m b p_s \exp\left(-\frac{E^*}{RT}\right)}{\left(1 + \left(b p_s \exp\left(-\frac{E^*}{RT}\right)\right)^t\right)^{1/t}}$$
(4)

for convenience, let us rewrite,  $-E^* = x$ ,  $p_s = p$ ,  $q_m = a$  and RT = r, thus eq (4) can be written as:

$$amount.adsorbed = \frac{abp \exp\left(-\frac{x}{r}\right)}{\left(1 + \left(bp \exp\left(-\frac{x}{r}\right)\right)^t\right)^{1/t}}$$
(5)

substituting eq(5) in (4), the site energy distribution function can be written as:

$$f(x) = \frac{d}{dx} \left( bpa \exp\left(-\frac{x}{r}\right) \left( \left( bp \exp\left(-\frac{x}{r}\right) \right)^t + 1 \right)^{-1/t} \right)$$
(6)

let 
$$u = \exp(-x/r)$$
 and  $v = \left( \left( bp \exp\left(-\frac{x}{r}\right) \right)^t + 1 \right)^{-1/t}$ , then from the product rule:  $\frac{d(uv)}{dx} = \frac{du}{dx}v + u\frac{dv}{dx}$ , we get

$$= bpa\left[\left(\left(bp\exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-1/t} \frac{d}{dx}\left(\exp\left(-\frac{x}{r}\right)\right) + \exp\left(-\frac{x}{r}\right) \frac{d}{dx}\left(\left(bp\exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-1/t}\right)\right]$$
(7)

Let, u = -x/r, then applying the chain rule of differentiation  $\frac{de^u}{dx} = e^u \frac{du}{dx}$ , we get

$$= bpa\left(\exp\left(-\frac{x}{r}\right)\left(\left(bp\exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-1/t} \frac{d}{dx}\left(-\frac{x}{r}\right) + \exp\left(-\frac{x}{r}\right)\frac{d}{dx}\left(\left(\left(bp\exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-1/t}\right)\right)\right)$$
(8)

Since,  $d(x^n) = nx^{n-1}$ , the above equation can be modified as

$$= bpa \left( \exp\left(-\frac{x}{r}\right) \frac{d}{dx} \left( \left( \left( bp \exp\left(-\frac{x}{r}\right) \right)^{t} + 1 \right)^{-1/t} \right) + \frac{\exp\left(-\frac{x}{r}\right) \left( \left( \left( bp \exp\left(-\frac{x}{r}\right) \right)^{t} \right) + 1 \right)^{-1/t} \right)}{r} \right) \right)$$
(9)

For convenience, let  $u = \left(bp \exp\left(-\frac{x}{r}\right)\right)^t + 1$  and n = -1/t, then using the chain rule of differentiation,  $\frac{du^n}{dx} = nu^{n-1}\frac{du}{dx}$ , and simplifying the

resulting expression, we get

$$= bpa \left( -\frac{\exp\left(-\frac{x}{r}\right) \left( \left(bp \exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-\frac{1}{t}}}{t} \frac{d}{dx} \left( \left(bp \exp\left(-\frac{x}{r}\right)\right)^{t} \right) - \frac{\exp\left(-\frac{x}{r}\right) \left( \left(bp \exp\left(-\frac{x}{r}\right)\right)^{t} \right) + 1\right)^{-1/t}}{r} \right)$$
(10)

if 
$$u = bp\exp(-x/r)$$
 and  $n = t$ , then using the chain rule,  $\frac{du^n}{dx} = nu^{n-1}\frac{du}{dx}$ , we get

$$= bpa \left( -bp \exp\left(-\frac{x}{r}\right) \left(bp \exp\left(-\frac{x}{r}\right)\right)^{t-1} \left(\left(bp \exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-\frac{1}{t}} \frac{d}{dx} \exp\left(-\frac{x}{r}\right) - \frac{\exp\left(-\frac{x}{r}\right) \left(\left(bp \exp\left(-\frac{x}{r}\right)\right)^{t}\right) + 1\right)^{-1/t}}{r} \right)$$
(11)

if u = -x/r, the applying the chain rule  $\frac{de^u}{dx} = e^u \frac{du}{dx}$ , we get

$$= bpa \left( -\frac{bp \exp\left(-\frac{2x}{r}\right) \left(bp \exp\left(-\frac{x}{r}\right)\right)^{t-1} \left(\left(bp \exp\left(-\frac{x}{r}\right)\right)^{t} + 1\right)^{-\frac{1}{t}-1}}{r} \exp\left(-\frac{x}{r}\right) \left(\left(bp \exp\left(-\frac{x}{r}\right)\right)^{t}\right) + 1\right)^{-1/t}} \frac{d}{dx} (x) - \frac{\exp\left(-\frac{x}{r}\right) \left(\left(bp \exp\left(-\frac{x}{r}\right)\right)^{t}\right) + 1\right)^{-1/t}}{r} \right)$$
(12)

Since,  $d(x^n) = nx^{n-1}$ , the above equation can be simplified to

$$f(x) = -\frac{bpa \exp\left(-\frac{x}{r}\right) \left(\left(bp \exp\left(-\frac{x}{r}\right)\right)^t + 1\right)^{-\frac{t+1}{t}}}{r}$$
(13)

Replacing the parameters:  $x = E^*$ ,  $p = p_s$ , r = RT, and  $a = q_m$  the site energy distribution function following a Toth isotherm explained in eq (1) is given by:

$$f(E^*) = -\frac{bp_s q_m \exp\left(-\frac{E^*}{RT}\right) \left(\left(bp_s \exp\left(-\frac{E^*}{RT}\right)\right)^t + 1\right)^{-\frac{t+1}{t}}}{RT}$$
(14)

## References

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