# Journal of Materials Chemistry A





Cite this: J. Mater. Chem. A, 2014, 2, 14896

A universal method to form the equivalent ohmic contact for efficient solution-processed organic tandem solar cells<sup>†</sup>

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The highly transparent, conductive and robust intermediate layer (IML) is the primary challenge for constructing efficient organic tandem solar cells. In this work, we demonstrate an easy but generic approach to realize the fully functional, solution-processed IMLs. In detail, solution-processed silvernanowires are packed at low concentration between hole- and electron-transporting layers to convert an otherwise rectifying interface into an ohmic interface. The IMLs are proven to be of ohmic nature under applied bias, despite the unipolar charge selectivity of the single layers. Ohmic recombination within IMLs is further proven in organic tandem solar cells fabricated by doctor-blading under ambient conditions. The tandem solar cells based on PCDTBT:[70]PCBM as the bottom cell and pDPP5T-2:[60] PCBM as the top cell give a power conversion efficiency of 7.25%, which is among the highest values for solution-processed organic tandem solar cells fabricated by using a roll-to-roll compatible deposition method in air.

Received 23rd June 2014 Accepted 11th July 2014 DOI: 10.1039/c4ta03182b

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### Introduction

The rapid growth in the field of organic photovoltaics (OPV) has attracted more and more attention from worldwide researchers during the last decade due to their large-scale, low-cost and easy manufacturing properties.1-6 It has been reported that the power conversion efficiency (PCE) of OPV devices has already surpassed the 10% milestone, which shows enormous promise and potential for commercial applications.<sup>7,8</sup> The PCE limitations of OPV devices are mainly due to the narrow absorption spectra of the donor materials, resulting in decreased short circuit current density  $(I_{sc})$  and the thermalization losses, resulting in decreased open circuit voltage  $(V_{\rm oc})$ .<sup>9,10</sup> The reduced J<sub>sc</sub> can be addressed by ternary or multicomponent donoracceptor systems, which were successfully shown to broaden the absorption of organic semiconductor composites.<sup>11-13</sup> The thermalization losses are more complex to tackle. Hummelen et al. recently suggested the design of organic semiconductors with high dielectric constant to reduce losses related to the exciton binding energy.<sup>14</sup> Another approach is the tandem

concept, which stacks two or more cells with complementary absorption spectra in series or parallel connection. This concept addresses both losses: the  $J_{sc}$  related absorption losses as well as the  $V_{oc}$  related thermalization losses. In the last few years, a number of organic tandem solar cells with high efficiencies were reported,<sup>8,15–22</sup> and the efficiency roadmap for tandem cells is clearly pointing towards the predicted 15%.<sup>23</sup>

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To realize high performance organic tandem solar cells, an efficient and reliable intermediate layer (IML) is required, which is typically designed from a series-connected sequence of a hole-transporting layer (HTL) and an electron-transporting layer (ETL).<sup>22</sup> The performance of an organic tandem solar cell is strongly dependent on the quality and functionality of its IML. Ideally, the IML should be highly transparent, conductive and robust enough to protect the underlying semiconductor layer and to form a quasi ohmic contact between HTL and ETL.<sup>24</sup> The majority of novel active layer materials are sensitive to humidity and oxygen at elevated temperatures and the microstructure of the semiconductor layer might be negatively influenced by the high temperature processing.<sup>25</sup> Therefore the fabrication of efficient organic tandem solar cells including the IML should avoid high temperature steps.

Although many efforts have been made to understand the recombination properties of IMLs, utilizing commonly used buffer layers to form the quasi ohmic contact in a general way is still a problem for solution-processed tandem solar cells. Today's reference materials for the IMLs are poly(3,4-ethyl-enedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS) for the hole injection and either titanium oxide (TiOx)<sup>26</sup> or zinc oxide

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<sup>†</sup> Electronic supplementary information (ESI) available: A universal method to form the equivalent ohmic contact for efficient solution-processed organic tandem solar cells. See DOI: 10.1039/c4ta03182b

(ZnO)<sup>8,15-20</sup> for the electron injection. Solution-processed p-type metal oxides, such as vanadium pentoxide (V<sub>2</sub>O<sub>5</sub>),<sup>27,28</sup> molybdenum trioxide  $(MOO_3)^{29,30}$  and tungsten trioxide  $(WO_3)^{31,32}$ were reported as substitutions for the widely used PEDOT:PSS, owing to the comparable device performances but enhanced environmental lifetime for single-junction solar cells.33,34 However, the commonly used n- and p-type interface layers, such as PEDOT:PSS and intrinsic metal oxides, do not necessarily form a quasi ohmic contact, especially for the direct contact between metal oxides and metal oxides. Although several groups reported the necessity to thermally-evaporate an ultra-thin metal layer (Au, Ca, or Al) between the HTL and ETL,35-37 the combination of solution processing with vacuum processing is unattractive for up-scaling and low-cost fabrication. Moreover, the mandatory oxygen plasma or ozone pretreatment activation, specifically relevant for many current p-type metal oxides like MoO<sub>3</sub> or NiO, is incompatible with tandem processing.29,38

Nowadays, solution-processed IMLs reported by several research groups were mainly based on highly conductive PEDOT:PSS, which were exclusively modified to certain n-type interface layers, such as ZnO, to form the quasi ohmic contact.<sup>8,15-20</sup> Solution-processed ZnO nanoparticles (ZnO-np) are not well defined in terms of their electrical and semiconducting properties (density of states and density of charge carriers) and may differ for various processes and routes.<sup>24,39</sup> Moreover, the chemical nature and the density of the ligand groups terminating the surface of ZnO-np, which is essential for contact/interface formation, is very difficult to assess and not known for most systems.

Low-temperature solution-processed metallic nanowires were recently reported as a promising transparent electrode and a potential substitute for the sputtered ITO and the thermally-evaporated top electrode.40-43 Metallic nanowire electrodes impress with high conductivity and high transparency, coupled with ease of manufacture. In this manuscript we demonstrate an easy but generic approach to fully solutionprocess efficient IMLs at fairly low temperatures ( $\leq 80$  °C). In detail, a solution-processed thin silver-nanostructure layer, coated from highly diluted silver-nanowire (AgNW) solution between the p-type and n-type charge transporting layers, is able to form an ohmic recombination contact between otherwise non-ohmic semiconductors. The generic approach of this concept is verified by two very different HTL/ETL recombination layers, on the one hand a PEDOT:PSS/ZnO-np layer and on the other hand a WO<sub>3</sub>/ZnO-np layer. It is again worthwhile to highlight that the fully functional solution-processed IMLs have been reported by several research groups for efficient tandem solar cells, however, the specific interface material requirements as well as the strict processing conditions restricted the reproduction of the reported promising results. Therefore, the universal approach demonstrated in this contribution indicates a way of forming equivalent ohmic contact between arbitrary qualified n- and p-type interface layers for efficient IMLs, which significantly reduces the workload of screening qualified interface materials for solution-processed organic tandem solar cells.

## Results and discussion

The charge recombination property of IMLs is investigated in the single-junction OPV devices, as illustrated in Fig. 1(a), which gives immediate feedback on the ohmic nature of the IMLs.<sup>24,39</sup> To match the requirements of the IML for efficient tandem solar cells, the holes and electrons that are selectively extracted by HTL and ETL should efficiently recombine at the interface. In this case, the single-junction devices with the recombination layers are supposed to show the same performance as devices with a single interface layer.<sup>24</sup> Fig. 1(b) and (c) represent the transmission spectra of the thin layers and chemical structures



**Fig. 1** (a) Architecture of the single-junction solar cells with a solution-processed intermediate layer; (b) transmission spectra of holeand electron-transporting layers. AgNW was diluted in IPA at a volume-ratio of 1 : 5 for AgNW 1 or 1 : 10 for AgNW 2; (c) chemical structures of active materials used in this work.

of active materials used in this work. PEDOT:PSS AI4083 was purchased from Heraeus and diluted in isopropyl alcohol (IPA) at a volume-ratio of 1:5 before processing. ZnO-np were synthesized from zinc acetate44 and dissolved in ethanol at 2 wt%. WO<sub>3</sub> nanoparticles were synthesized from flame pyrolysis<sup>31,32</sup> and dissolved at 2.5 wt% in ethanol. The AgNW ink was prepared from a water based master solution and diluted in IPA at a volume-ratio of 1:5 (AgNW 1) or 1:10 (AgNW 2). The AgNW 1 and AgNW 2 are highly transparent, as depicted in Fig. 1(b) inset. In the configuration measured, and after correction of the substrate, transmission values of over 99% are observed in a wavelength range from 350 to 600 nm. A full analysis of the optical properties of AgNW films is reported elsewhere.40 The metal oxides WO3 and ZnO-np absorb in the blue regime, while PEDOT:PSS is more absorbing in the infrared regime. Overall transmission of the charge extraction layers is >90%.

Poly(3-hexylthiophene):[6,6]-phenyl-C61 butyric acid methyl ester (P3HT:PCBM) with a thickness of 100 nm was used as an active layer. The I-V characteristics of the corresponding OPV devices are summarized in Fig. 2 and Table 1. As depicted in Fig. 2(a), we find significant limitations for the PEDOT:PSS/ ZnO-np interface. Most obvious is the rather low injection under forward bias, resulting in a low fill factor (FF). Consequently, this recombination layer is not expected to properly work in a tandem configuration. Nevertheless, all these problems can be overcome by employing the solution-processed Ag nanostructure-based thin layer, which completely lifts the recombination restrictions at the HTL/ETL interface. The organic solar cells incorporating the AgNW-based recombination layer (Devices C and D) exhibit comparable performance with the reference (Device A). As shown in Fig. 2(c), the singlejunction solar cells comprising an IML of WO<sub>3</sub>/ZnO-np



Fig. 2 *J–V* characteristics of single-junction solar cells employing different intermediate layers: (a) and (c) under AM1.5 illumination; (b) and (d) in the dark. Device A: ZnO-np layer; Device B: PEDOT:PSS/ZnO-np; Device C: PEDOT:PSS/AgNW 1/ZnO-np; Device D: PEDOT:PSS/AgNW 2/ZnO-np; Device F: WO<sub>3</sub>/AgNW 1/ZnO-np; Device F: WO<sub>3</sub>/AgNW 1/ZnO-np; Device G: WO<sub>3</sub>/AgNW 2/ZnO-np.

(Device E) suffer from the same deficiencies as the PEDOT:PSS/ ZnO-np devices (Device B), namely low rectification from a high series resistance. Similarly, the performance of the devices utilizing the WO<sub>3</sub>/ZnO-np IML was again significantly improved by inserting AgNW at the interface. In the case of WO<sub>3</sub>, a more distinct difference was observed for AgNW 1 versus AgNW 2. Device F, employing the WO<sub>3</sub>/AgNW 1/ZnO-np IML suffers from a significantly higher shunt than Device G with the WO<sub>3</sub>/AgNW 2/ZnO-np interface. Furthermore, in comparison with the reference devices, the performances of the single-junction solar cells with the HTL/AgNW/ETL injection layer were not affected by the optical losses in the IMLs. In contrast, the single-junction solar cells employing the AgNW-based IML exhibit even slightly higher  $J_{sc}$  compared with the reference device employing a pristine ZnO-np layer, which may be caused by either small variation in the active layer thickness or by a change in the morphological properties of the ZnO-np layer.46,47 Summarizing the single-junction devices, it is most relevant to notice that the series resistances  $(R_s)$  of OPV devices, as shown in Table 1, were significantly reduced by inserting AgNW 2 at the interface of the IMLs, while the leakage current was still in the same range as that of reference devices, indicating that the recombination property of the IML was significantly improved by inserting this ultrathin AgNW layer. The nanoparticles, degraded from nanowires, serve even more efficiently as recombination centers at the interface of HTL/ETL compared with nanowires due to their better shunt. Moreover, if several nanowires overlap each other, as shown in Fig. 2(b), this AgNW layer may not be fully covered by overlying 100 nm ZnO-np resulting in high leakage current in the OPV device, which can be observed in the J-V characteristic of Device F.

The synthesis of AgNW is typically based on a polyol process, which requires the presence of a polymeric binder like poly-(vinylpyrrolidone) (PVP). PVP as well as other polymeric binders are known to environmentally stabilize the AgNW by efficiently cladding them, and further play an essential role as a matrix in the film formation properties.45 The morphologies of AgNWbased IMLs were studied by atomic force microscopy (AFM), as shown in Fig. 3(a)-(i). We observed that the formation of the doctor-bladed AgNW layer is strongly affected by the underlying layer. The formation of AgNW coated on top of WO3 nanoparticles is comparable to that on glass. In contrast, the formation of AgNW will be strongly affected by the underlying PEDOT:PSS layer. The water-IPA-based AgNW solution most likely partially dissolves the PEDOT:PSS layer in the interface, resulting in a mixture of nanowires and nanoparticles. Fig. 3(j) depicts the height value along the line in Fig. 3(b), indicating that the thickness of the polymeric binder is around 10 nm and the diameters of silver nanowires are  $\sim$ 30 nm.

This picture changes completely for highly diluted AgNW solution. Increasing the dilution with IPA to 1 : 10 vol% results in a nanoparticular coating, which originates from completely degraded AgNWs. Fig. 3(k) depicts the height distributions calculated from the whole area ( $10 \times 10 \ \mu m^2$ ) of Fig. 3(b) and (e). The curves present that the height distributions of nanowires (AgNW 1: full square) and nanoparticles (PEDOT:PSS/AgNW 1: full circle) are between 10–60 nm and 30–80 nm

		0.000	11.[\0]	PCE [%]	$R_{\rm s} \left[\Omega \ \rm cm^{-}\right]$	$R_{\rm p} [k\Omega  {\rm cm}^2]$
nO-np	0.58	-8.32	61	2.92	0.82	23.95
EDOT:PSS/ZnO-np	0.46	-7.21	37	1.24	4.61	2.32
EDOT:PSS/AgNW 1/ZnO-np	0.56	-9.28	52	2.72	0.84	5.66
EDOT:PSS/AgNW 2/ZnO-np	0.56	-9.42	51	2.72	0.93	12.52
/O <sub>3</sub> /ZnO-np	0.56	-4.20	30	0.70	22.10	57.31
/O <sub>3</sub> /AgNw 1/ZnO-np	0.56	-7.82	46	2.03	4.25	0.43
O <sub>3</sub> /AgNW 2/ZnO-np	0.58	-9.24	58	3.10	3.01	16.71
	C-np EDOT:PSS/ZnO-np EDOT:PSS/AgNW 1/ZnO-np EDOT:PSS/AgNW 2/ZnO-np 'O <sub>3</sub> /ZnO-np 'O <sub>3</sub> /AgNW 1/ZnO-np 'O <sub>3</sub> /AgNW 2/ZnO-np	IO-Inp       0.38         EDOT:PSS/ZnO-np       0.46         EDOT:PSS/AgNW 1/ZnO-np       0.56         CO_3/ZnO-np       0.56         O_3/AgNW 1/ZnO-np       0.56         O_3/AgNW 2/ZnO-np       0.56         O_3/AgNW 2/ZnO-np       0.58	DOTP       0.38       -0.32         EDOT:PSS/AgNW 1/ZnO-np       0.46       -7.21         EDOT:PSS/AgNW 1/ZnO-np       0.56       -9.28         EDOT:PSS/AgNW 2/ZnO-np       0.56       -9.42         O <sub>3</sub> /ZnO-np       0.56       -4.20         O <sub>3</sub> /AgNW 1/ZnO-np       0.56       -7.82         O <sub>3</sub> /AgNW 2/ZnO-np       0.58       -9.24	DOTP       0.38       -8.32       61         EDOT:PSS/ZnO-np       0.46       -7.21       37         EDOT:PSS/AgNW 1/ZnO-np       0.56       -9.28       52         EDOT:PSS/AgNW 2/ZnO-np       0.56       -9.42       51         'O <sub>3</sub> /ZnO-np       0.56       -4.20       30         'O <sub>3</sub> /AgNW 1/ZnO-np       0.56       -7.82       46         'O <sub>3</sub> /AgNW 2/ZnO-np       0.58       -9.24       58	DOTIP       0.58       -8.32       61       2.92         EDOT:PSS/AgNW 1/ZnO-np       0.46       -7.21       37       1.24         EDOT:PSS/AgNW 1/ZnO-np       0.56       -9.28       52       2.72         EDOT:PSS/AgNW 2/ZnO-np       0.56       -9.42       51       2.72         O <sub>3</sub> /ZnO-np       0.56       -4.20       30       0.70         O <sub>3</sub> /AgNW 1/ZnO-np       0.56       -7.82       46       2.03         O <sub>3</sub> /AgNW 2/ZnO-np       0.58       -9.24       58       3.10	DO-Trp Dispersive       0.58       -8.52       61       2.92       0.82         EDOT:PSS/AgNW 1/ZnO-np       0.46       -7.21       37       1.24       4.61         EDOT:PSS/AgNW 1/ZnO-np       0.56       -9.28       52       2.72       0.84         EDOT:PSS/AgNW 2/ZnO-np       0.56       -9.42       51       2.72       0.93         O <sub>3</sub> /ZnO-np       0.56       -4.20       30       0.70       22.10         'O <sub>3</sub> /AgNW 1/ZnO-np       0.56       -7.82       46       2.03       4.25         'O <sub>3</sub> /AgNW 2/ZnO-np       0.58       -9.24       58       3.10       3.01

Table 1Photovoltaic parameters of single-junction solar cells employing different interlayers.  $R_s$  and  $R_p$  were calculated from the correspondingJ-V characteristics in the dark at 0 and 2 V, respectively. The photovoltaic parameter distributions of devices B–G are summarized in Fig. S1<sup>+</sup>



Fig. 3 AFM images of different interface layers (scale bar 2  $\mu$ m): (a) bare glass; (b) AgNW 1 (1 : 5 vol% in IPA) on glass; (c) AgNW 2 (1 : 10 vol% in IPA) on glass; (d) 50 nm thick PEDOT:PSS on glass; (e) AgNW 1 on PEDOT:PSS; (f) AgNW 2 on PEDOT:PSS; (g) 60 nm thick WO<sub>3</sub> on glass; (h) AgNW 1 on WO<sub>3</sub>; (i) AgNW 2 on WO<sub>3</sub>. (j) Profile analysis extracted from (b), (k) and (l) height distributions calculated from (b), (e) and (g)–(i).

respectively, indicating that the maximal height of AgNW above the polymer film is ~50 nm, which is supposed to be easily covered by the subsequent 100 nm ZnO-np. In contrast, the formation of AgNW on top of the WO<sub>3</sub> layer is almost the same as that on glass. The roughness (RMS) of the WO<sub>3</sub> layer and WO<sub>3</sub>/AgNW 2 were measured to be 6.5 and 8 nm on average, respectively. The increase in roughness after blading AgNW 2 on top of the WO<sub>3</sub> layer shows great accordance to the roughness of the AgNW 2 layer on the glass substrate (2 nm in average). Additionally, the height distributions of Fig. 3(g)–(i) are calculated from the whole scanning area (10 × 10 µm<sup>2</sup>) and depicted in Fig. 3(l). After blading the AgNWs thin layer on top of WO<sub>3</sub> the mean value of the height distributions increased from ~56 nm to ~80 nm being in great accordance with the diameters of AgNWs, as shown in Fig. 3(j).

To investigate the functionality of the AgNW-based IML in a real tandem structure, P3HT:PCBM-based tandem solar

cells employing different AgNW-based IMLs were constructed. The J-V characteristics of corresponding tandem solar cells are summarized in Fig. S2 and Table S1<sup>†</sup>. By inserting the AgNW between HTL and ETL of the IML, the Voc and FF values of tandem solar cells are significantly improved. The prominent improvements reveal that the AgNW-based IMLs are on the one hand robust enough to protect the underlying active layer from the diffusion of solvents during solution processing of upper layers. On the other hand, these IMLs are also efficient enough to collect the charge carriers selectively from subcells. It is worthwhile to notice that in our case the combination of PEDOT:PSS/ZnO-np is not robust enough to serve as an eligible IML in the tandem structure. The polymer-matrix of AgNW is not soluble in the nonpolar solvents. Thus, the stability and reliability of the IML can be enhanced by inserting the AgNW thin layer. Similar improvements were also observed in the tandem devices employing the WO<sub>3</sub>/AgNW 2/ZnO-np intermediate layer. Additionally, tandem solar cells based on the PEDOT:PSS/ AgNW 2/ZnO-np IML shows a series resistance ( $R_s$ ) of 1.93  $\Omega$  $cm^2$ , which is only slightly higher than the sum- $R_s$  of two sub-cells (1.75  $\Omega$  cm<sup>2</sup>), indicating that there is almost no extra losses at the interface of this IML. A cross-sectional TEM image of the P3HT:PCBM-based tandem solar cell incorporating the PEDOT:PSS/AgNW/ZnO-np IML is shown in Fig. 4.



**Fig. 4** Cross-sectional TEM image of a P3HT:PCBM-based tandem solar cell incorporating the PEDOT:PSS/AgNW/ZnO-np IML.

To demonstrate the advantages of the AgNW-based IML on the tandem concept, tandem solar cells based on active layer materials with complementary absorption spectra were constructed. As illustrated in Fig. 5(a), the combination of PEDOT:PSS/AgNW 2/ZnO-np was employed as the IML for tandem solar cells. Poly[N-9"-hepta-decanyl-2,7-carbazole-alt-5,5-(4',7'-di-2-thienyl-2',1',3'-benzothiadiazole)] (PCDTBT)48 and a low bandgap diketopyrrolopyrrole-quinquethiophene alternating copolymer (pDPP5T-2)49,50 were employed as donor materials. The I-V characteristics of tandem solar cells based on the PCDTBT:[70]PCBM bottom cell and the pDPP5T-2:[60] PCBM top cell, and corresponding reference single cells are summarized in Fig. 5(b) and Table 2. The champion tandem solar cell achieved a  $V_{\rm oc}$  of 1.44 V along with a  $J_{\rm sc}$  of 8.64 mA  $cm^{-2}$  and a FF of 58% resulting in a PCE of 7.25%, while the corresponding reference solar cells obtained a PCE of 5.55% and 5.27% for bottom and top cells respectively. An improvement of >30% in PCE was achieved by incorporating both active layers into a tandem structure, indicating on the one hand the advantage of the tandem concept and on the other hand the promising functionality and reliability of the AgNW based IML. The photovoltaic parameters distribution of 6 tandem solar cells (from one substrate) shown in Fig. S3<sup>†</sup> indicates that Ag nanoparticles were homogeneously deposited and served identically as recombination centres at the interface of HTL/ETL.



Fig. 5 (a) Device architecture of tandem solar cells employing the AgNW-based IML; (b) J-V characteristics of the tandem solar cell based on the PCDTBT:[70]PCBM bottom cell and the pDPP5T-2:[60] PCBM top cell, and corresponding reference single cells.

Table 2Photovoltaic parameters of a tandem solar cell based on thePCDTBT:[70]PCBM bottom cell and the pDPP5T-2:[60]PCBM top cell,and corresponding reference single cells. A distribution of photovol-taic parameters over 6 tandem solar cells is shown in Fig. S3†

$V_{ m oc} \left[ { m V}  ight]$		$J_{ m sc}  [ m mA \ cm^{-2}]$	FF [%]	PCE [%]	
Ref. bottom cell	0.87	-10.66	60	5.55	
Ref. top cell	0.55	-14.37	68	5.27	
Tandem cell	1.44 (1.46)	-8.64 (-8.55)	58 (58)	7.25 (7.24)	

### Experimental

### Materials

P3HT ( $M_w = 65.5 \text{ kg mol}^{-1}$ ) and PCDTBT ( $M_w = 127 \text{ kg mol}^{-1}$ ) were purchased from Merck and St-Jean Photochemicals Inc., respectively. [60]PCBM (99.5%) and [70]PCBM (99%) were purchased from Solenne BV. PEDOT:PSS (Clevios AI4083) was purchased from Heraeus. Silver-nanowire ink and pDPP5T-2 ( $M_w = 47 \text{ kg mol}^{-1}$ ) were supplied by Cambrios Technology Corporation and BASF, respectively. WO<sub>3</sub> nanoparticles suspension (product no. 4035) was provided by Nanograde Llc. ZnO nanoparticles were synthesized in our lab according to previous publications.<sup>44</sup>

#### Fabrication of single-junction devices

All the devices were fabricated by doctor-blading under ambient conditions with an architecture shown in Fig. 1(a). Pre-structured ITO-coated glass substrates were cleaned in sequence with acetone and isopropyl alcohol (IPA) for 10 minutes. After drying, the substrates were coated with 50 nm thick PEDOT:PSS layer or with 60 nm thick WO<sub>3</sub> layer (filtered through a 0.2 µm filter before use) and annealed on a hot plate at 80 °C for 5 min. Afterwards, the AgNW layer (diluted in IPA at 1:5 vol% for AgNW 1 and 1 : 10 vol% for AgNW 2) and ~100 nm thick ZnOnp layer were bladed subsequently and dried at 80 °C for 5 min again. An active layer with a thickness of  $\sim 100$  nm was bladed from a chlorobenzene solution of P3HT and PCBM with a mixture ratio of 1:1 wt%. Then, a diluted solution of PEDOT:PSS (1:5 vol% in IPA) was bladed on top of the active layer. The whole stack was annealed on a hot plate at 140 °C for 5 min after evaporation of a 100 nm thick Ag layer to form the top electrode. For reference, single-junction solar cells based on the ZnO-np buffer layer were constructed under the same conditions with an architecture of ITO/ZnO-np/P3HT:PCBM/ PEDOT:PSS/Ag.

#### Fabrication of tandem solar cells

Organic tandem solar cells were fabricated by doctor-blading under ambient conditions with an architecture shown in Fig. 5(a). Pre-cleaned ITO coated glass substrates were coated with ~50 nm thick ZnO-np and dried at 80 °C for 5 min. PCDTBT:[70]PCBM (1 : 4 wt% dissolved in dichlorobenzene at a total concentration of 20 mg mL<sup>-1</sup>) with a thickness of ~80 nm was bladed on top of ZnO-np. After that, the IML of PEDOT:PSS/ AgNW 2/ZnO-np was bladed on top of the active layer under the same conditions as used for single-junction devices. The second active layer of pDPP5T-2:[60]PCBM (1 : 2 wt% dissolved in a solvent mixture of 95 vol% chloroform and 5 vol% dichlorobenzene at a total concentration of 18 mg mL<sup>-1</sup>) with a thickness of ~100 nm was deposited. Afterwards, the whole stack was transferred into a nitrogen-filled glovebox. A 15 nm thick MoOx and a 100 nm thick Ag were thermally evaporated in sequence to form the top electrode.

### Characterisations

The active area of OPV devices was defined by the top electrode with a value of 10.4 mm<sup>2</sup>. The J-V characteristics were measured using a source measurement unit from a BoTest through a mask with an opening of 10.4 mm<sup>2</sup>. Illumination was provided by a solar simulator (Oriel Sol 1A, from Newport) with AM1.5G spectra at 100 mW cm $^{-2}$ . The optical investigations of thin films were carried out by a UV-VIS-NIR spectrometer (Lambda 950, from Perkin Elmer). The thicknesses of films were measured by a profilometer (Tencor Alpha Step D 100). The morphologies of thin films were characterized by AFM (Veeco Model D3100, tapping mode) from the samples prepared on microscopic slides. Samples for cross-sectional TEM were fabricated on a conductive silicon wafer and prepared by focused ion-beam (FIB) sectioning. For FIB sectioning a FEI Strata 235 dual-beam instrument was employed. TEM imaging was performed with an aberration-corrected FEI Titan 80-300. To minimize electronbeam induced damaging of sensitive organic layers the high tension was reduced to 80 kV.41

### Conclusions

In conclusion, we demonstrated a universal method to fabricate an efficient low-temperature solution-processed IML, in which a solution-processed AgNW-layer was introduced to improve the recombination properties at the interface of the HTL/ETL. Due to the limitation of recombination efficiency, the combination of PEDOT:PSS/ZnO-np and WO<sub>3</sub>/ZnO-np cannot be directly utilized as efficient IMLs for organic tandem solar cells. By inserting an ultra-thin solution-processed AgNW layer, these IMLs showed a similar functionality with the commonly used single buffer layer in single-junction solar cells, indicating that the equivalent ohmic contact was formed between HTL and ETL. With the improvement of recombination properties, single-junction organic solar cells utilizing the combinations of PEDOT:PSS/ AgNW/ZnO-np as well as WO<sub>3</sub>/AgNW/ZnO-np as IMLs gave PCEs of 2.72 and 3.10% respectively, while the corresponding cells without AgNW showed PCEs of only 1.24 and 0.70% respectively. Furthermore, the AgNW-based IMLs were investigated under the same conditions for P3HT:PCBM-based tandem solar cells, suggesting that the combination of PEDOT:PSS/AgNW 2/ZnO-np can serve as an efficient and robust IML for efficient tandem solar cells. To further verify its functionality and reliability, tandem solar cells based on the PCDTBT:[70]PCBM bottom cell and the pDPP5T-2:[60]PCBM top cell were constructed. The tandem solar cell achieved a  $V_{\rm oc}$  of 1.44 V along with a  $J_{\rm sc}$  of 8.64 mA cm<sup>-2</sup> and a FF of 58% resulting in a PCE of 7.25%.

## Acknowledgements

The authors thank Dr. Michael Salinas and Prof. Marcus Halik at the Institute of Polymer Materials (LSP) for the access to AFM measurements, and Christel Dieker and Prof. Erdmann Spiecker at the Center of Nanoanalysis and Electron Microscopy (CENEM) for the help with cross-sectional TEM measurements. Cambrios Technology Corporation and Nanograde are acknowledged for providing the silver-nanowire ink and the WO<sub>3</sub> nanoparticles suspension, respectively. The authors thank the support of the Cluster of Excellence "Engineering of Advanced Materials" at FAU, which is funded by the German Research Foundation (DFG) within the framework of its "Excellence Initiative", and the support from "Synthetic Carbon Allotropes" (SFB 953) project and "Solar Technologies Go Hybrid" (SolTech) project.

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