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Infrared absorption cross-sections in HITRAN2016 and beyond: Expansion for climate, environment, and atmospheric applications

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Abstract

Spectroscopic analysis of hydrocarbons, halocarbons and related species is required in atmospheric applications such as climate-change research, tracking of pollution and biomass burning, and remote sensing. Critically evaluated line-by-line spectroscopic data are needed for such analyses. To address this need we report the large addition of spectra to the HITRAN2016 database. This extension increases the number of molecules in the database by almost a factor of 6 from that in HITRAN2012. Spectra from the Pacific Northwest National Laboratory database (Appl. Spectrosc. 2004;58:1452–61) and the Hodnebrog et al. compilation (Rev Geophys 2013;51:300–78) have been integrated into HITRAN2016. Spectra from high-resolution laboratory observations, predominantly from Fourier transform spectrometers, are now provided

1 for 328 compounds in the IR spectral range. Spectra for atmospherically-important compounds
2 organized by chemical category are presented. The HITRAN infrastructure (www.hitran.org),
3 HITRAN Application Programming Interface (HAPI), and best practices for searching and
4 downloading data are discussed.

5 **1. Introduction**

6 Human activities result in the emission of a diverse range of gases into the atmosphere,
7 including hydrocarbons, halocarbons, acids, alcohols, amines, sulfur- and nitrogen-containing
8 species. Emission of radiatively-potent compounds into the atmosphere can cause significant
9 changes in the climate [1]. In addition to water vapor, carbon dioxide, methane, and nitrous
10 oxide (see the line-by-line section of the HITRAN database [2]), halogen-containing organic
11 compounds contribute significantly to the atmospheric radiative budget. Since the middle of the
12 twentieth century, much attention has been paid to hydrocarbons, halocarbons and related
13 species. The radiative forcing of climate change from increased atmospheric levels of halogen-
14 containing organic compounds over the period 1750-2011 is approximately 20% of that from
15 increased levels of carbon dioxide over the same period [3]. Emissions of reactive volatile
16 organic compounds (VOCs) are implicated in photochemical ozone formation and local air
17 quality degradation.

18 Halogen-containing organic compounds are used in a broad range of production and industrial
19 activities, including fire suppression, air conditioning, refrigeration, heat transfer, electronics,
20 and semiconductor manufacture. The spectroscopic signatures of these compounds have been
21 suggested as potential indicators of “intelligent” life on exoplanets [4]. In addition, chlorine- and
22 bromine-containing compounds can participate in chemistry leading to stratospheric ozone
23 depletion. Some halogen-containing compounds (such as saturated perfluorocarbons (PFCs))
24 have extremely long lifetimes - thousands of years - due to very strong C-F bonds [5] with very
25 high global warming potentials. Significant efforts have been expended to develop substitute
26 compounds for chlorofluorocarbons (CFCs), hydrochlorofluorocarbons (HCFCs), and
27 hydrofluorocarbons (HFCs), ideally with much less global warming and ozone depletion
28 potentials (GWPs and ODPs). Detailed knowledge of the spectroscopy of halogen-containing
29 organic compounds is paramount for monitoring their atmospheric abundances and thus required
30 to assess their contributions to radiative forcing of climate change. Many of these same species
31 are powerful greenhouse gases, and assessments of this impact also require detailed
32 spectroscopic knowledge.

1 To control emission of pollutant gases, many ongoing efforts are being made which originated
 2 in the 1980s in the framework of the United Nations (UN) environment programs and
 3 agreements. The HITRAN2016 database edition includes many of the halocarbons and related
 4 gases controlled under United Nations agreements. These agreements include the Montreal
 5 Protocol on Substances that Deplete the Ozone layer (see <http://ozone.unep.org>) and its
 6 amendments, and the Framework Convention on Climate Change (UNFCCC) (see
 7 <https://unfccc.int/process>). A major use of cross-section data in HITRAN is thus for the
 8 calculation of the effects on climate of such gases as well as for monitoring emissions and
 9 concentrations via remote sensing. **Table 1** summarizes the evolution of the inclusion of families
 10 of halocarbons (including the halons, which are brominated halocarbons) and related molecules
 11 within these agreements and emphasizes the relevance of the compounds included here. In
 12 contrast to the Kyoto Protocol, the UNFCCC Paris Agreement is a “bottom-up” agreement in
 13 which each country decides on its emission ambitions through so-called Nationally Determined
 14 Contributions (NDCs) – many of these NDCs contain references to the inclusion of HFCs and
 15 related gases within their emission targets; however as there is no overarching requirement for
 16 their inclusion, the Paris Agreement is not included in **Table 1**.

17 **Table 1.** Overview of inclusion of halocarbons in United Nations protocols and agreements.

Process (focus)	Protocol, Adjustment or Amendment (Year)	Number of CFCs	Number of Halons	Number of HCFCs	Number of HFCs	Number of other related substances
Montreal (ozone)	Montreal 1987	5	3			
Montreal (ozone)	London 1990	15	3	34		2
Montreal (ozone)	Copenhagen 1992	15	37	40		3
Montreal (ozone)	Beijing 1999	15	38	40		
Montreal (climate)	Kigali 2016				18	
UNFCCC (climate)	Kyoto 1997				13	8
UNFCCC (climate)	Doha 2012				13	9

18 Note: For more details of those falling within the Montreal Protocol on Substances that Deplete
 19 the Ozone Layer see <http://ozone.unep.org>. For those falling within the Framework Convention
 20 on Climate Change (UNFCCC), see <https://unfccc.int/process>.

21 The HITRAN database [2] is the most commonly used source of reference molecular
 22 spectroscopic data for molecules of atmospheric interest. A major extension of the number of
 23 large complex molecules represented in HITRAN, as well as of the spectral range, temperature,
 24 and pressure coverage for many of them is described in the present article. The compounds
 25 discussed in the present paper are mostly considered in terms of their Earth climate and
 26 environmental relevance but are also important in astrophysics, theoretical and fundamental
 27 physics, biology, and metrology. We do not consider these applications in any detail in this

1 paper; detailed information on the relevance of emissions of such gases to radiative forcing of
2 climate change, and metrics such as the Global Warming Potential, and the methods by which
3 these are calculated, can be found in [1].

4 The HITRAN2016 database provides absorption cross-sections for more than 328
5 hydrocarbons, halocarbons and related compounds (approximately six times more than the
6 HITRAN2012 edition [6]) as a supplement to the discrete molecular transitions in the traditional
7 part of HITRAN. For some molecules (e.g. ethane and methyl bromide), HITRAN provides both
8 line-by-line parameterization and experimental cross-sections. This can arise when the cross-
9 sections have some information that is absent in the line-by-line part of HITRAN, or for cases
10 when the line-by-line data are not precise enough. The IR spectral data, which comprise the
11 focus of the present paper, are mostly air-, self-, or nitrogen-broadened measured with Fourier
12 transform spectrometers using a range of resolutions at a variety of temperatures. In the
13 following sections of this paper we provide a general overview of the cross-sections in the
14 HITRAN database, followed by a description of the spectral data separated by categories.
15 Finally, we give a more detailed outline of the corresponding part of the HITRAN*Online* web
16 interface, which provides tools for searching and plotting the molecular cross-sections.

17 **2. Formalism and history for the HITRAN cross-sections**

18 The HITRAN spectroscopic database [2] was designed to provide a machine-readable
19 compilation of basic spectroscopic parameters that could be used in radiative-transfer computer
20 codes. The acronym HITRAN stands for **H**igh resolution **T**ransmission; the high resolution
21 implies that the database is intended to enable the user to simulate spectra in as fine detail as
22 possible. The HITRAN project began in the 1960s and culminated in the first line list in 1973
23 [7]. The effort was supported by the US Air Force and initially had the objective to cover the
24 main absorbing gases in the terrestrial atmosphere (H_2O , CO_2 , O_3 , N_2O , CO , CH_4 , and O_2) in the
25 spectral interval covering approximately 1 to 10 micrometers. The scope of the database at the
26 time was driven by sensor technology and the speed of computer calculations.

27 The main goal of HITRAN, including its cross-section part, is to provide fundamental
28 parameters for radiative transfer in gaseous media using the Lambert-Beer law of absorption,
29 namely

$$30 \quad \tau(\nu) = e^{-k(\nu)L}, \quad (1)$$

31 where $\tau(\nu)$ is the spectral transmittance at wavenumber ν , $k(\nu)$ is a normalized absorption
32 coefficient in cm^{-1} , and L is an increment of path through the media in cm. The absorption cross-
33 section is in units of $\text{cm}^2/\text{molecule}$ and is often expressed by

$$\sigma(\nu, T, p) = - \frac{\ln \tau(\nu)}{\rho L} \quad (2)$$

where ρ is the volume density (molecule/cm³). The absorption coefficient and absorption cross-sections can be calculated from the parameters in the line-by-line portion of the data base (see Appendix of the HITRAN1996 paper [8]). The cross-section portion provides experimental cross-sections as a function of wavenumber recorded at different thermodynamic conditions. A powerful tool for calculating cross-sections and other parameters from the line-by-line data is now provided with the database and is called the HITRAN Application Programming Interface (HAPI) [9]. It is worth noting that in Eq. (2) the Napierian logarithm is used, while many experimental spectra obtained using manufacturer software employ base 10 logarithms. The conversion is straightforward and one just needs to multiply those experimental cross-section by $\ln(10)$ ¹.

Over time, the uses of HITRAN have caused the database to expand significantly. They now include astrophysical, medical, pollution monitoring, and many other applications. The expansion has been in:

(1) Spectral range which now covers the microwave through ultraviolet, corresponding to pure-rotation, rovibrational, and electronic transitions;

(2) Additional molecules and their important isotopologues. As of the current 2016 edition [2] there are 49 molecules with an associated 126 isotopologues represented in the high-resolution (line-by-line) portion of the database. This expansion has been driven by the increase in the number of applications of HITRAN;

(3) Increase in the number of parameters for each transition. This increase in parameters has been driven by both the sophistication in transmission and radiance models as well as the improvement in observational and archiving techniques, for example, greater signal-to-noise, higher resolution, superior photometric accuracy, and more efficient data handling. Table 1 of Ref. [2] shows the current list of parameters.

Many bands of molecular spectra are currently intractable to theoretical treatment. This situation is particularly true for large polyatomic molecules, for which the following problems can arise simultaneously: they have dense spectra with many low-lying vibrational modes (thus a complete representation in a line list requires a vast number of hot bands and combination bands), they have many resonances that are difficult to characterize, and finally, they have complicated line-shape phenomena, including line-mixing. The solution that has been taken is to provide the user with absorption cross-sections, σ (cm²/molecule), that have been measured in controlled laboratory experiments. To be useful, these cross-section spectra should be given at

¹ The same conversion rule applies to the intensity parameters of the line-by-line part of HITRAN

1 many pressure and temperatures so that the radiative-transfer codes can interpolate to obtain
2 reasonable simulations. They should also be obtained for different broadening gases, for
3 example self, nitrogen, and oxygen. Note that applying these cross-sections in a radiative-
4 transfer code, while perhaps not a satisfying physical approach, nonetheless can achieve quite
5 satisfactory results since the cross-sections may possess within them many of the missing
6 ingredients of the line-by-line approach. Thereby, the fidelity of a simulation using absorption
7 cross-sections can be quite good if they are obtained with accurate measurements (that is, well-
8 determined amounts of the gas and perturber in question, adequate resolution, good calibration)
9 and have a large representative range of temperature-pressure measurements. To achieve better
10 modeling accuracy, it is also important that the cross-sections do not possess such artifacts as
11 mixture contaminations, channel fringes, zero level offsets, etc. An interesting approach for
12 interpolating cross-sectional data obtained at different temperatures and pressures was proposed
13 by Toon (<https://mark4sun.jpl.nasa.gov/pseudo.html>). This approach, known as “pseudo-lines”,
14 fits multiple cross-sectional data to a “pseudo” line list where the “lines”, although in HITRAN
15 format, are simply a fitting tool. In other words, the end product is a HITRAN-formatted “line
16 list” that allows experimental cross-sections at any temperature and pressure within the range
17 probed by experiment to be reproduced.

18 The absorption cross-section component of the HITRAN compilation was introduced in the
19 mid-1980s [10] with a major dedicated update carried out for the 1992 edition by Massie and
20 Goldman [11]. High-resolution spectra taken by Fourier-Transform spectrometers (FTS) were
21 already available that had been digitized into convenient forms [12]. Many more laboratories
22 began acquiring lists of infrared cross-sections taken by FTS, primarily of polyatomic molecules
23 in the long wavelength region of the spectrum. These efforts include the Pacific Northwest
24 National Laboratory (PNNL) [13], the State University New York at Stony Brook (hereafter
25 referred to as SUNY) under the support of the US Dept of Energy Atmospheric Radiation
26 Measurement (ARM) program [14,15], the Rutherford Appleton Laboratory (RAL) in the UK
27 [16], the US National Institute of Standards and Technology [17], and the Université Libre de
28 Bruxelles, Belgium[18].

29 In addition to the infrared cross-sections, the majority of ultraviolet data in HITRAN are in
30 the form of cross-sections. In this paper, however, we will focus on the revision and expansion of
31 the IR cross-sections.

32 **Table 2** illustrates the evolution of the HITRAN cross-section data from the 1986 to 2016
33 version. For comparison purposes, only molecules from HITRAN2012 are given in this
34 summary (as we discuss in this paper HITRAN2016 has substantially more molecules). The “o”

1 sign means that the data have not been changed in the corresponding database issue. The “+”
 2 sign signifies a change for the given molecule. A change in an edition can mean several different
 3 things: (a) new data completely replacing old data; (b) new data added in a new spectral region,
 4 with old data kept in other regions, (c) new data added and old data relegated to a supplemental
 5 folder since they were deemed to be good but in some way duplicated the new data.

6 HNO₃ was added in the 1986 edition with the idea that as high-resolution line-by-line data
 7 (line-by-line parameters) were eventually obtained, the corresponding cross-sections would be
 8 removed from the database. However, this protocol seems to be inconsistent, since for other
 9 molecules, such as SF₆ and ClONO₂, there are both cross-sections as well as line-by-line
 10 parameters, at least for some spectral regions. The idea here was that there are a significant
 11 number of missing hot bands from the line-by-line part, as well as other unsolved spectroscopic
 12 phenomena. Therefore, cross-sections for these latter two molecules were retained in the current
 13 HITRAN issue.

14 Note that the whole idea of archiving cross-sections in HITRAN is counter to the overall
 15 HITRAN philosophy of providing basic parameters that one can use to calculate radiative
 16 transfer in diverse environments based on good physics. On the other hand, the philosophy of
 17 HITRAN also promotes providing the user with the "best" tools for these calculations. Since
 18 molecules with extremely congested lines, with low-lying fundamentals, with extremely difficult
 19 line-shape properties, etc, are not yet treatable in a satisfactory way, the experimental cross-
 20 sections will remain in HITRAN for the foreseeable future.

21 **Table 2.** Evolution of IR cross-section data sets in the HITRAN versions since 1986 until 2016
 22 (only molecules from HITRAN2012 are explicitly shown here). The “o” sign means that the data
 23 have not been changed in the corresponding database issue. The sign “+” signifies a change for
 24 the given molecule.

Molecule Chemical formula/ common name	HITRAN edition								
	1986	1992	1996	2000	2004	2008	2012	2016	
C ₂ H ₆ / Ethane							+	+	
CH ₃ OH / Methanol							+	o	
CH ₃ CH ₂ CH ₃ / C ₃ H ₈ / Propane							+	+	
CH ₃ COCH ₃ / Acetone							+	+	
CH ₃ CHO / Acetaldehyde							+	+	
CH ₃ CClF ₂ / HCFC-142b							+	+	
BrONO ₂ / Bromine nitrate							+	+	
ClOOCl / Chlorine peroxide							+	o	
CH ₃ CN / Acetonitrile / Methyl cyanide						+	+	+	
CH ₃ C(O)OONO ₂ / PAN						+	+	o	
C ₆ H ₆ / Benzene						+	o	o	
SF ₅ CF ₃ / Trifluoromethyl sulfur pentafluoride				+	+	o	o	+	
C ₂ F ₆ / Hexafluoroethane / CFC-116				+	o	o	o	+	
CHCl ₂ CF ₃ / HCFC-123				+	o	o	o	+	
CHClFCF ₃ / HCFC-124				+	o	o	o	+	

CH ₃ CCl ₂ F / HCFC-141b									
CH ₃ CCIF ₂ / HCFC-142b									
CHCl ₂ CF ₂ CF ₃ / HCFC-225ca									
CCIF ₂ CF ₂ CHCIF / HCFC-225cb									
CH ₂ F ₂ / HFC-32									
CHF ₂ CF ₃ / HFC-125									
CHF ₂ CHF ₂ / HFC-134									
CFH ₂ CF ₃ / HFC-134a									
CF ₃ CH ₃ / HFC-143a									
CH ₃ CHF ₂ / HFC-152a									
SF ₆ / Sulfur hexafluoride									
CHCIF ₂ / HCFC-22									
CCIF ₃ / CFC-13									
C ₂ ClF ₅ / CFC-115									
ClONO ₂ / Chlorine nitrate									
CCl ₃ F / CFC-11									
CCl ₂ F ₂ / CFC-12									
N ₂ O ₅ / Dinitrogen pentoxide									
CF ₄ / CFC-14									
C ₂ Cl ₂ F ₃ / CFC-113									
C ₂ Cl ₂ F ₄ / CFC-114									
CCl ₄ / Carbon tetrachloride									
HNO ₄ / Peroxynitric acid									
CHCl ₂ F / HCFC-21									

1

2 The earliest editions of HITRAN presented data in ASCII; there was also a real concern about

3 minimize storage space in that chosen format. This legacy has been inherited to some extent.

4 When the cross-sections were added to the HITRAN compilation, the traditional line-by-line

5 portion was composed of records (transitions) of 100 characters in length. Although the

6 transitions were in text format, they were given the file extension ".par". It was decided to cast

7 the cross-sections also in records of 100 characters, although each record would contain ten

8 values (fields) of the cross-sections. These cross-sections would be ordered from lowest to

9 highest wavenumber of the set. The values would be given at equal wavenumber intervals. For

10 each molecule, it was thus necessary to have metadata not only to name the molecule, but also

11 give the lowest and highest wavenumber, the temperature and pressure of the set, the number of

12 individual cross-sections, and associated laboratory details such as the resolution of the

13 experiment and the diluent gas if any. The metadata are given in a 100-character header before

14 each group of temperature-pressure sets. The molecules are each contained in a file with the

15 extension ".xsc". Individual files will thus have several embedded headers depending on how

16 many temperature-pressure sets exist. **Table 3** illustrates the metadata in the header.

17 **Table 3.** Metadata header for cross-section data.

Quantity	Field length	Type	Comment
----------	-----------------	------	---------

Molecule	20	Character	Chemical formula (right adjusted)
Minimum wavenumber, ν_{\min}	10	Real	Start of range (cm^{-1})
Maximum wavenumber, ν_{\max}	10	Real	End of range (cm^{-1})
Number of cross-sections in set	7	Integer	Number of cross-sections in set
Temperature, T	7	Real	Temperature (K) of set
Pressure, P	6	Real	Pressure of set in Torr
Maximum cross-section value in set, σ_{\max}	10	Real	Useful for scaling plots ($\text{cm}^2/\text{molecule}$)
Instrument resolution	5	Real	See note
Common name	15	Character	Familiar name of molecule
Not currently used	4		Reserved for future use
Broadener	3	Character	Air, N_2 , or (if left blank) self-broadened
Reference	3	Integer	Index pointing to source of data

1 Note: Most cross-sections have been taken from Fourier transform spectrometer (FTS) measurements. In
2 that case the resolution is given in cm^{-1} . There are some cross-sections taken from grating spectrometer
3 measurements in the UV. In those cases, the instrument resolution is given in milli-angströms in the form
4 $\text{xxx m}\text{\AA}$, where xxx are up to three digits.

5 Several remarks should be made concerning the format of the cross-sections. It was decided
6 early on to not have any negative absorption values. Some of the experiments contained small
7 negative values, usually within experimental uncertainties, and these were zeroed out, partly in
8 response to the preferences of some radiative-transfer codes. Subsequently, a small percentage
9 of users objected to this procedure, because the negative absorption values were a useful
10 indicator of systematic errors in the measurements. Setting the negative values to zero creates a
11 bias when used in retrievals since the average cross sections will be positive when they should be
12 zero instead. This may affect retrieval of the species in question as well as other species having
13 absorption in that spectral range. For this reason, we now preserve the original data in a
14 supplemental folder in addition to the "rectified" files.

15 Another aspect of the original spectra requires an extra step when mapping to the HITRAN
16 format. If the cross-sections obtained cover a very wide spectral range, and there are obvious
17 regions between bands with basically no data, we occasionally break these spectra into multiple
18 sets, thereby avoiding many records of zero values. Thus, for the case of two distinct spectral
19 regions, the data in HITRAN would appear in two sets in the molecule file with different headers
20 within the file. This is especially true for the older files, when disk space was more of an issue.

21 In most cases the total number of cross-sections in a set is not a multiple of 10. The last
22 record of a file will be extended with zero values; this should cause no problem since the header
23 gives the number of cross-sections in the set, N . The equal wavenumber interval is thus simply
24 $\Delta\nu = (\nu_{\max} - \nu_{\min})/N$.

25 As new cross-section data are validated and added to HITRAN, another dilemma arises. The
26 older data may have slightly different temperature-pressure sets than the new. Mixing them with
27 the newer data might cause discontinuities. In this case the data is validated using several criteria
28 (spectral coverage, source-to-source comparison, stated accuracies, agreement with third-party

1 data etc...). If the old data are deemed to be fairly reliable, they are not discarded, but rather also
2 relegated to the supplemental folder. The situation is less of a problem if the older data do not
3 overlap the newer with respect to the wavenumber coverage; in that case they are simply
4 retained.

5 As mentioned above, the cross-section values, σ , are presented at a uniform wavenumber grid.
6 This only presents a problem for some ultraviolet sets where they were measured by a grating
7 spectrometer and hence obtained at equal wavelength spacing. We have resampled them onto a
8 uniform wavenumber grid where they reside in the principal folder, but now retain the original
9 values in the supplemental folder.

10 Finally, it should be mentioned that the HITRAN and HITEMP [19] databases do not adhere
11 to SI units. For instance, the line centers are given in wavenumbers (cm^{-1}). Not only are
12 centimeters used throughout, but atmosphere (atm) and Torr are used for pressure, the latter in
13 the headers of the cross-sections. The choice of units is in harmony with many standard
14 radiative-transfer codes.

16 3. Description of the cross-section data molecule by molecule

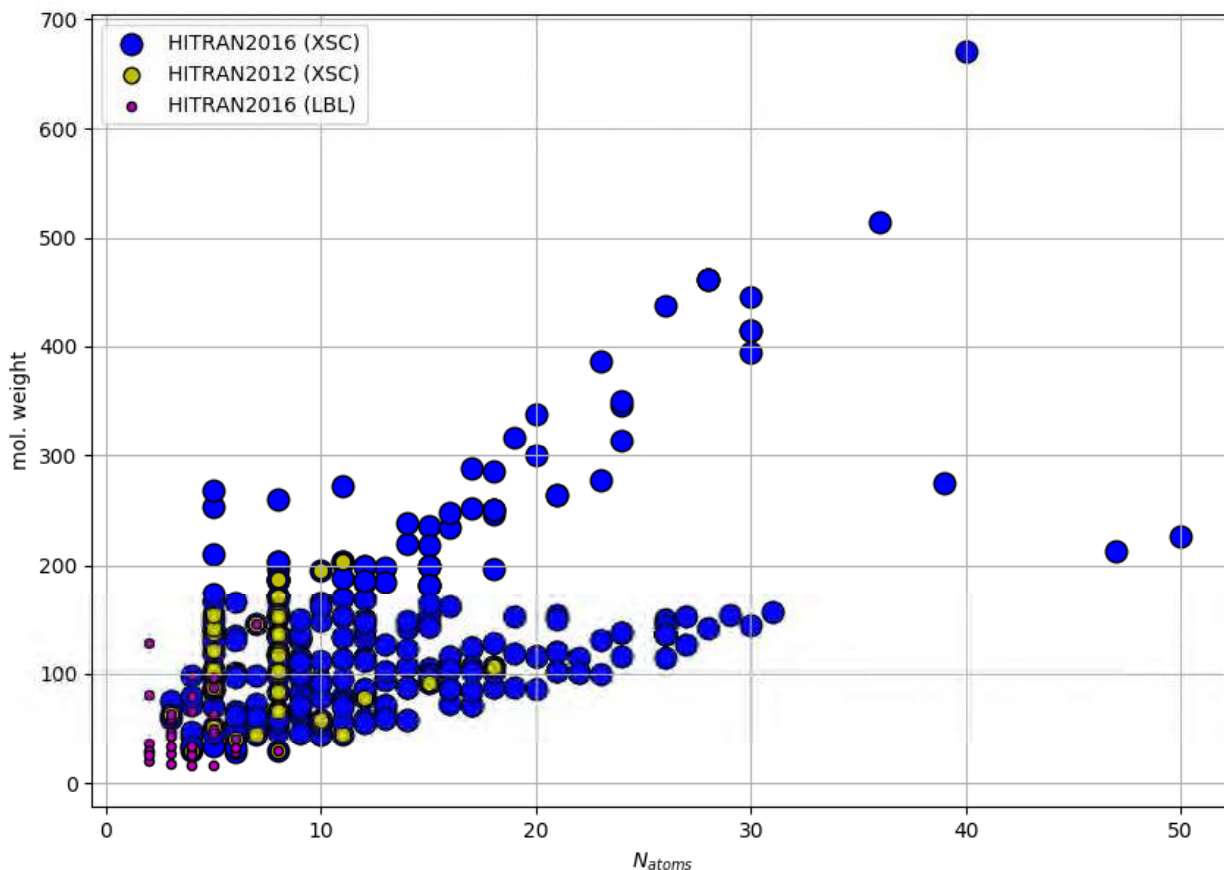
17 In the HITRAN2016 release, the spectral data for more than 320 molecules have been either
18 added, or updated. To assist the user in searching molecules in the HITRAN database, the gases
19 are organized into several categories. In this section we give a description of each category, and
20 characterize and compare the spectra for important molecules from each category. In addition, in
21 HITRAN*Online* supplemental folder (<https://hitran.org/suppl/xsec/>) we provide summary tables
22 (**Table S1**, **Table S2**, and **Table S3**) containing metadata for sources, molecules, and cross-
23 sections. These tables can be found using the link in the Supplemental tab at the cross-section
24 search page in the HITRAN web site. More details on the general usage of the HITRAN*Online*
25 web-interface can be found in Ref. [20], and more specifically to cross-sections in Section 4.

26 When there were different data sources with similar coverage of broadening conditions,
27 spectral intervals, temperature, pressure, etc., we compared the data to decide if the
28 corresponding spectra should be included in the database, and if so, should they be available via
29 the HITRAN*Online* web interface or instead relegated to the supplemental folder. The latter is
30 used either for storing the original spectra with non-zeroed negative absorption (if any), or for
31 storing the data sources which were “worse” by some measure (e.g., stated accuracy, spectral
32 coverage, number of temperature and pressure sets), but nevertheless have complementary
33 information which could be useful for some applications. Here we put the word “worse” in

1 quotes since in many cases it is impossible to state that one dataset is generally superior to
2 another. This arises, for example, when one dataset has more temperature and pressure sets while
3 another has better resolution. However, since the HITRAN database strives to facilitate the
4 user's choice and provide spectroscopic data for many applications, we made remarks on the
5 differences of such data, and specified which cross-sections were made available via the
6 HITRAN*online* web interface (further "included in HITRAN*online*"), which were relegated to
7 the supplemental folder, and which were not included at all.

8 **Figure 1** shows the molecules in the cross-section part of the HITRAN2016 issue (large
9 circles in blue) and those from the HITRAN2012 cross-section part (medium circles in olive).
10 The molecules from the line-by-line part of the HITRAN2016 database are in magenta (small
11 circles). The plot displays the molecular weights versus number of atoms. It can be seen that the
12 range of sizes of molecules in the database has been greatly expanded with inclusion of species
13 such as halogenated aldehydes, alcohols, ethers, and long-chain hydrocarbons. The molecule
14 with the largest molecular weight in HITRAN2016 is PFTBA, perfluorotributylamine,
15 (measured by Godin et al. [21]) while the molecule with the largest number of atoms is
16 hexadecane (measured by PNNL [13,22]).

17



18

1 **Figure 1.** Overview of the molecular weights vs number of atoms for molecules from the
2 HITRAN database. All molecules from the cross-section part (“XSC”) of HITRAN2016 are
3 highlighted in blue, while those from HITRAN2012 are in olive. The molecules from the line-
4 by-line part (“LBL”) of HITRAN2016 are in magenta.

5 In the following, we give a brief summary of the data sources that provided either a large
6 number of molecules, or a detailed set of temperatures and pressures.

7 Hodnebrog et al. [1] compiled absorption cross-sections of more than 200 halocarbons and
8 related compounds from experimental and ab initio studies and calculated radiative efficiencies,
9 global warming potentials, and global temperature potentials of these gases. These climate
10 metrics were adopted in the Intergovernmental Panel on Climate Change Fifth Assessment
11 Report [3]. The Hodnebrog et al [1] compilation focused on available room-temperature
12 measurements (as these were the most widely available) although selected lower-temperature
13 measurements were included for some gases.

14 In the PNNL DOE (Department of Energy) database, Sharpe and co-workers [13] recorded
15 the broadband quantitative reference spectra of 462 molecules for applications in tropospheric
16 monitoring or laboratory studies near standard atmospheric pressure and ambient temperature .
17 Each molecule has a single spectrum (600 to 6500 cm^{-1}) at both 298.15 and 323.15 K and, if the
18 vapor pressure was sufficient, also at 278.15 K. The spectrum was derived from a series of
19 single measurements using a variety of analyte partial pressures (the range often covering more
20 than two orders of magnitude) in 760 Torr of N_2 diluent. The analyte/ N_2 mixtures were generated
21 by either of two methods. The first method used a static system (Sharpe et al. [13]) with the
22 passive vapor pressure of liquid or an aliquot of the gas added followed by addition of the N_2
23 ballast; quantitation was achieved using the ideal gas law. The second method (Johnson et al.
24 [23]) used a disseminator system where the liquid analyte was dispensed at the tip of a syringe
25 and flash-vaporized into a flowing stream of N_2 gas; quantitation was achieved using total mass
26 calculations from the syringe pump rate in $\mu\text{l}/\text{min}$ and the N_2 gas flow rate. In both cases spectra
27 were recorded using a Bruker IFS 66v/S spectrometer at 0.11 cm^{-1} resolution and boxcar
28 apodization; protocols and instrument details can be found in Johnson et al. [23]

29 While the technical aspects of the data were identical, a subsequent project produced the
30 PNNL SERDP (Strategic Environmental Research and Development Program) database of IR
31 spectra (<https://datahub.pnnl.gov/projects/serdp>). Although smaller in quantity (55 molecules),
32 the PNNL SERDP database (Johnson et al. [22]) was focused on VOCs and oxygenated-VOCs
33 (OVOCs) associated with forest fires and biomass burning.

1 New absorption cross-sections, measured by Harrison et al. [24–28], have been derived for
2 the halogen species HFC-23, HFC-134a, CFC-12, HCFC-22, and CCl₄, and improve upon those
3 in previous versions of HITRAN. Multiple air-broadened spectra were recorded in a 26 cm path
4 length sample cell by FTS over a range of atmospherically relevant temperatures and pressures.
5 These studies provide spectra with extended wavenumber, temperature, and pressure ranges, and
6 having high signal to noise ratio. The spectra are compared with those from previous HITRAN
7 issues – see the corresponding references by Harrison (and coauthors). These datasets are
8 principally intended for use in infrared limb sounding by instruments such as the ACE-FTS and
9 MIPAS.

10 Fourier transform infrared measurements of bromine nitrate by Birk and Wagner [29] have
11 been performed in the spectral region 675–1400 cm⁻¹ at 0.014 cm⁻¹ spectral resolution.
12 Absorption cross-sections were derived from 38 spectra covering the temperature range from
13 203 to 296 K and air pressure range from 0 to 190 mbar (where 0 means there is only the
14 absorber in the gas) applying a global least squares fitting approach. Bromine nitrate was
15 generated in a complicated chemical process from HNO₃ → N₂O₅ → ClONO₂ → BrONO₂,
16 where ClONO₂ required the production of Cl₂O from HgO+Cl₂. The measurements of pure and
17 air-broadened bromine nitrate were carried out in a sealed-off 22 cm absorption path cell which
18 was mounted inside the sample compartment of the spectrometer and cooled by flowing
19 methanol through a glass double jacket and the flanges. Decomposition was low, especially at
20 lower temperatures, since chemically inert materials were used: Pyrex glass, PFA-coated steel,
21 Teflon, and AgCl as windows.

22 Rodriguez et al. [30] recorded experimental absorption spectra of allyl trifluoroacetate
23 (ATFA) and vinyl trifluoroacetate (VTFA). Cross-sections were measured at 298 K and no
24 dependence on total pressure was observed. These data were then used to calculate the radiative
25 efficiencies and global warming potentials of these gases using the methodology described by
26 Hodnebrog et al. [1]

27 Rodriguez et al. [31] measured absorption spectra of hydrofluoroether HFE-7300 and HFE-
28 7500 at 298 K and 25-150 Torr total pressure. No dependence on total pressure was observed for
29 both compounds. The cross-sections were then adopted for the calculation of the radiative
30 efficiencies along with the global warming potentials using the methodology described by
31 Hodnebrog et al. [1].

32 For the CFCs, HCFCs, and fluorinated species from the list below, measured by Le Bris et al.
33 [21,32–39], the experimental acquisitions of the absorption spectra were performed on pure

1 vapor samples using a Bomem DA8 spectrometer with a globar source, a KBr beamsplitter and a
2 MCT (mercury cadmium telluride) detector. Depending on the compound, the spectral range
3 covers all the fundamental modes from the mid-500 cm^{-1} up. Two types of acquisitions have
4 been carried on:

- 5 • Cross-sections in a low-temperature range and at a resolution of 0.02 cm^{-1} (0.05 cm^{-1} for
6 CFC-113) for temperature-dependent cross-sections of molecules that are relatively
7 abundant in the atmosphere, and currently or possibly detectable by remote sensing
8 techniques (CFC-113 [32], HCFC-142b [33], HCFC-141b [34], HFC-143a [35]).
- 9 • Cross-sections at room temperature and above and at a resolution of 0.1 cm^{-1} to calculate
10 the radiative efficiency and global warming potential of new or minor atmospheric
11 compounds (PFTBA [21], HFIP [36], PFPO [37], PFD [38], HFC-43-10mee [39])

12 For each temperature-pressure set, hundreds of scans were co-added to create an average
13 absorption spectrum. Using the method presented by Sharpe et al. [13], optical depths
14 corresponding to too optically thin or thick are eliminated. The composite spectrum for a given
15 temperature is obtained by calculating the cross-section at zero partial pressure from a linear or
16 quadratic fit of the optical depth versus the partial pressure for each wavenumber.

17 Uncertainties are given using the 95% confidence interval (two standard deviations). The
18 processed experimental data were validated by a comparison with theoretical calculations
19 performed using density functional theory and ab-initio methods. In recent papers from this
20 group [36,38,39], the theoretical calculations have also been used to extend the spectral range of
21 the cross-sections to the lower fundamental modes. Synthetic spectra in the $0\text{-}550 \text{ cm}^{-1}$ spectral
22 region were created after calibration of the theoretical wavenumbers and integrated intensities of
23 the modes with respect to the experimental data in the $550\text{-}3500 \text{ cm}^{-1}$ spectral range.

1 **Table 4.** Summary of the data sources.

Source	Molecule(s)	Uncertainty	Apodization	Spectrometer	Resolution (cm ⁻¹)	Type	Facility
Sharpe et al. (2004) [13], Johnson et al. (2010) [22]	~230 molecules	2.1%	Boxcar	Bruker IFS 66v/S	0.112	composite	PNNL, USA
Clerbaux et al. (1993) [18]	8 molecules	1-4% ^b	Norton-Beer strong	Bruker IFS 120HR	0.03	100 co-added scans	Université Libre de Bruxelles, Belgium
Di Lonardo et al. (2000) [40]	HFC-125	1.32% ^c	Boxcar / Norton-Beer strong	Bomem DA8	0.03	256 co-added scans	University of Bologna, Italy
Smith et al. (1998) [16]	HFC-143a, HFC-134	5%	Norton-Beer strong	Bruker IFS 120HR	0.03	100 co-added scans x 8	RAL, STFC, UK
Smith et al. (1996) [41]	HFC-32	4%	Norton-Beer strong	Bruker IFS 120HR	0.03	100 co-added scans	RAL, STFC, UK
Harrison et al. (2010) [42]	Ethane	4%	Norton-Beer medium	Bruker IFS 125/HR	0.015	Spectra were recorded in blocks of 5 or 10 scans and subsequently co-added	RAL, STFC, UK
Etminan et al. (2014) [43]	HCFC-133a, CFC-113a	2.7%		Bruker IFS 125 HR	0.01		RAL, STFC, UK
Massie et al. (1985) [12]	HCFC-21	10-25%	Triangular		0.03		University of Denver, USA
McDaniel et al. (1991) [44], Massie et al. (1991) [45]	CFC-113*, CFC-13, CFC-115, CFC-114	10-15% ^d		Bomem DA3.01	0.03 ^e	85 co-added scans	National Center for Atmospheric Research, Boulder, USA
Zou et al. (2004) [46]	PFC-116	0.3%		Bruker IFS-120HR	0.01/0.03		SUNY, USA
Sellevag et al. (2004) [47]	2,2,2-trifluoroacetaldehyde, 3,3,3-Trifluoropropanal	~2% ^a	Blackman-Harris 3-Term	Bruker IFS 113v	1.0	32 co-added scans	University of Oslo, Norway

Young et al. (2009) [48]	HFC-329p	5%		Mattson Instruments Sirius 100	0.5		Ford, USA
Le Bris and Graham (2015) [35]	HFC-143a	5%	Boxcar	Bomem DA8	0.02	composite	University of Toronto, Canada
Le Bris et al. (2012) [34]	HCFC-141b	<4%	Boxcar	Bomem DA8	0.02	composite	University of Toronto, Canada
Le Bris et al. (2011) [32]	CFC-113	3%-5% ^f	Boxcar	Bomem DA8	0.05	composite	University of Toronto, Canada
Le Bris and Strong (2010) [117]	HCFC-142b	3%-4% ^f	Boxcar	Bomem DA8	0.02	composite	University of Toronto, Canada
Le Bris et al. (2018) [39]	HFC-43-10-mee	3%-6% ^f	Boxcar	Bomem DA8	0.1	composite	University of Toronto, Canada
Le Bris et al. (2017) [38]	Perfluorodecalin	4% (cis), 7% (trans)	Boxcar	Bomem DA8	0.1	composite	University of Toronto, Canada
Wagner and Birk (2016) [29]	Bromine nitrate	1.8% ν_3 , 4.2% ν_2	Boxcar	Bruker IFS 125 HR	0.014	composite	German Aerospace Center
Harrison (2013) [24]	HFC-23	3%	Norton-Beer weak	Bruker IFS 125HR	0.015	Up to 100 co-added scans	RAL, UK
Harrison (2015) [25]	HFC-134a	3%	Norton-Beer weak	Bruker IFS 125HR	0.015	Up to 100 co-added scans	RAL, UK
Harrison et al. (2016) [28]	CCl ₄	3%	Boxcar	Bruker IFS 125HR	0.01-0.03	100 co-added scans	RAL, UK
Harrison (2015) [26]	CFC-12	3%	Boxcar	Bruker IFS 125HR	0.01-0.03	100 co-added scans	RAL, UK
Harrison (2016) [27]	HCFC-22	3%	Boxcar	Bruker IFS 125HR	0.01-0.03	100 co-added scans	RAL, UK
Reed and Hodges (2015) [49]	Ethane	0.41-0.96%					NIST, USA
D'Anna et al. (2005) [50]	perfluoro-2-methyl-3-pentanone		Blackman-Harris 3-term		1.0		University of Oslo, Norway
Nielsen et al.	HFC-1234yf	5%		Mattson	0.5		Ford, USA

(2007) [51]						Instruments Sirius 100			
Nilsson et al. (2009) [52]	(Z)-HFC-1234ze	5%				Mattson Instruments Sirius 100	0.5		Ford, USA
Bravo et al. (2010) [53]	C8F18, PFC-51-14, PFC-41-12*, PFC-31-10*, PFC-218 *	5%	Norton-Beer weak			Bruker IFS 125 HR	0.01	100 co-added scans	RAL, UK.
Osterstrom et al. (2012) [54]	HFE-263fb1	5%				Mattson Instruments Sirius 100	0.5		Ford, USA
Oyaro et al. (2004) [55]	HFE-356mmz1, HFE-365mcf3, CHF2CF2CH2OCH 3, HFE-356mff2*		Blackman- Harris 3-term			Bruker IFS 113v	1.0		University of Oslo, Norway
Oyaro et al. (2005) [56]	9 molecules								University of Oslo, Norway
Bravo et al. (2010) [57]	HFE-449s1, HFE-569sf2, C3F7CH2OH, HFE-347mcc3	4 - 21%	Norton-Beer weak			Bruker IFS 125 HR	0.03	50 co-added scans	RAL, UK
Andersen et al. (2012) [58]	HFC-1243zf, C6F13CH=CH2, C8F17CH=CH2, HFC-1345zfc, C4F9CH=CH2	5%				Mattson Instruments Sirius 100	0.5		Ford, USA
Sellevag et al. (2004) [59]	CHF2CH2OH, CH2FCH2OH, CF3CH2OH		Boxcar			Bruker IFS 113v	1.0		University of Oslo, Norway
Myhre et al. (1999) [60]	HFE-236ca12, HFE-338pcc13		4P			Bruker IFS 113v	1.0		University of Oslo, Norway
Sihra et al. (2001) [61]	~48 molecules	5%				Mattson Instruments	0.5		Ford, USA

				Sirius 100		
Andersen et al. (2012) [62]	Halon-2311, HCFE-235ca2	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Wallington et al. (1994) [63]	HFC-152	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Wallington et al. (2000) [64]	CH ₂ ClOH, CCl ₃ OH, CHCl ₂ OH	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Wallington et al. (2004) [65]	HFE-338mec3, HFE-329me3	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Wallington et al. (2009) [66]	HFE-43-10pccc124	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Waterland et al. (2005) [67]	CF ₃ (CF ₂) ₄ CH ₂ CH ₂ OH, CF ₃ (CF ₂) ₆ CH ₂ CH ₂ OH, CF ₃ (CF ₂) ₈ CH ₂ CH ₂ OH, CF ₃ CH ₂ CH ₂ OH	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Young et al. (2006) [68]	PFPMIIE	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Young et al. (2009) [69]	CF ₃ CF ₂ CF=CF ₂	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Gohar et al. (2004) [70]	HFC-227ea, HFC-32, HFC-23 [†] , HFC-134a [†]	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Mashino et al. (2000) [71]	HFE-216	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	
Inoue et al. (2008) [72]	HFC-365mfc	5%	Mattson Instruments Sirius 100	0.5	Ford, USA	

Hurley et al. (2007) [73]	(E)-HFC-1225ye, (Z)-HFC-1225ye	5%		Mattson Instruments Sirius 100	0.5		Ford, USA
Hashikawa et al. (2004) [74]	C2F5CHO, C3F7C HO, C4F9CHO, CF3CHO [†]	5%		Mattson Instruments Sirius 100	0.5		Ford, USA
Sellevag et al. (2007) [75]	CHF2CF2CH2OH, CF3CHF2CH2OH, CF3CF2CF2CH2OH [†] , CF3CH2CH2OH [†] , CF3CF2CH2OH [†]	5%	Blackman-Harris 3 term	Bruker IFS 113v	1.0	32 co-added scans	University of Oslo, Norway
Andersen et al. (2010) [76]	Sevoflurane, HCFE-235da2 [†] , HFE-236ea2 [†]	5%		Mattson Instruments Sirius 100	0.5		Ford, USA
Andersen et al. (2008) [77]	HCFO-1233zd	5%		Mattson Instruments Sirius 100	0.5		Ford, USA
Andersen et al. (2004) [78]	HG ⁻ -01, HG ⁻ -02, HG ⁻ -03	5%		Mattson Instruments Sirius 100	0.5		Ford, USA
Ryan and Nielsen (2010) [79]	HCFE-235da2, Sevoflurane [†]	5%	Boxcar	Bruker IFS 66v	1.0	512 co-added scans	Abbott Laboratories, Limited, Saint-Laurent, Québec, Canada
Acerboni et al. (2001) [5]	PFC-1114, CF2=CF=CF2, PFC-1216 [†]	10%	Boxcar	Bruker IFS 113v	1.0		University of Oslo, Norway
Godin et al. (2017) [37]	PFPO	4%	Boxcar	Bomem DA8	0.1	Composite	University of Toronto, Canada
Rodrigues et al. (2016) [30]	Vinyl trifluoroacetate Allyl trifluoroacetate	5%		Bruker VERTEX 80V	0.03	64 co-added scans	University of Castilla La Mancha, Spain
Etmnan et al. (2014) [43]	HFC-133a, CFC-113a				0.01		RAL, STFC, UK
Davis et al. (2016)	CFC-114a, CFC-	4%			1.0	100 co-added scans	ESRL, NOAA, USA

[80]	113a [†] , CFC-112a [†] , CFC-112 [†]								
Godin et al. (2017) [36]	HFIP	<3%	Boxcar	Bomem DA8	0.1	Composite	University of Toronto, Canada		
Godin et al. (2016) [21]	PFTBA	10%	Boxcar	Bomem DA8	0.1	Composite	University of Toronto, Canada		
Rodriguez et al. (2014) [31]	HFE-7300 HFE-7500	5%		Bruker VERTEX 80V	1	64 co-added scans	University of Castilla La Mancha, Spain		

- 1 a) Separate uncertainties for spectral features are given [47]
- 2 b) 1-2% for strong features, 3-4% for weaker features [18]
- 3 c) Uncertainty is given for integrated intensities [40]
- 4 d) 10% for integrated intensities and 15% for band peaks [44]
- 5 e) For all gases except CFC-113; for CFC-113 measurements were taken at 1.0 cm⁻¹ resolution [44]
- 6 f) Uncertainty varies with the temperature
- 7 ♦ Supplemental folder

1 3.1. Alcohols, ethers and other oxygenated hydrocarbons

2 The compounds from this category are used in industrial applications, as solvents, in organic
3 synthesis, electrical systems, production of drugs, resins, explosives, fragrances, as well as in the
4 food and textile industries. Spectroscopic monitoring of molecules from this category can be
5 used in tracking the effluents of biomass burning [13,22], including from forest and rural fires.
6 Oxygenated molecules are typically reactive towards hydroxyl radicals, have relatively short
7 atmospheric lifetimes, and small global warming potentials.

8 The HITRAN2012 [6] edition contained only three molecules from this category, namely
9 acetaldehyde, formaldehyde, and acetone; these molecules are found in the Earth's atmosphere
10 [81–84]. The infrared data for these molecules were already included either in the line-by-line or
11 in the cross-sections parts of the previous HITRAN issue. The cross-sections part from
12 HITRAN2012 included 31 temperature and pressure sets for acetone [83–85] (including scaled
13 cross-sections), 16 sets for acetaldehyde [82], and approximately 44600 transitions for
14 formaldehyde for the region of the strongest IR absorption (6 μm) [6]. These data have not been
15 changed in the HITRAN2016 edition [2].

16 The latest HITRAN2016 update [2] contains over seventy new compounds in this category.
17 The summary of new molecules is shown in [Table 5](#). The new data can be divided into two
18 sources: the original Department of Energy (PNNL-DOE) data [13], and more recently published
19 measurements of compounds relative to the biomass burning, performed by the same laboratory
20 under the same conditions, but providing a new data set sponsored by the Strategic
21 Environmental Research and Development Program (PNNL-SERDP) [22].

22 New data for acetaldehyde, formaldehyde, and formic acid were added in HITRAN2016,
23 which thus combines information available with HITRAN2012 with new data. PNNL data for
24 acetaldehyde span a wider wavenumber range than the previous HITRAN data from the work by
25 Tereszchuk and Bernath [82], and have stronger absorption features in the low-wavenumber
26 range near 1500 cm^{-1} (see [Figure 2](#)) comparing to the 2400-3400 cm^{-1} range from the
27 HITRAN2012 release. The HITRAN2012 spectral coverage corresponds to Tereszchuk and
28 Bernath (blue color in [Figure 2](#)). This data was made available through HITRAN*online* in
29 addition to spectra from Tereszchuk and Bernath. Spectra for formaldehyde and formic acid
30 from PNNL are in a good agreement with the absorption cross-sections generated from the line-
31 by-line HITRAN data. Comparisons of PNNL with HITRAN for both these molecules were
32 previously published in Ref. [86]. These spectra are available in the supplemental folder of the
33 database compilation.

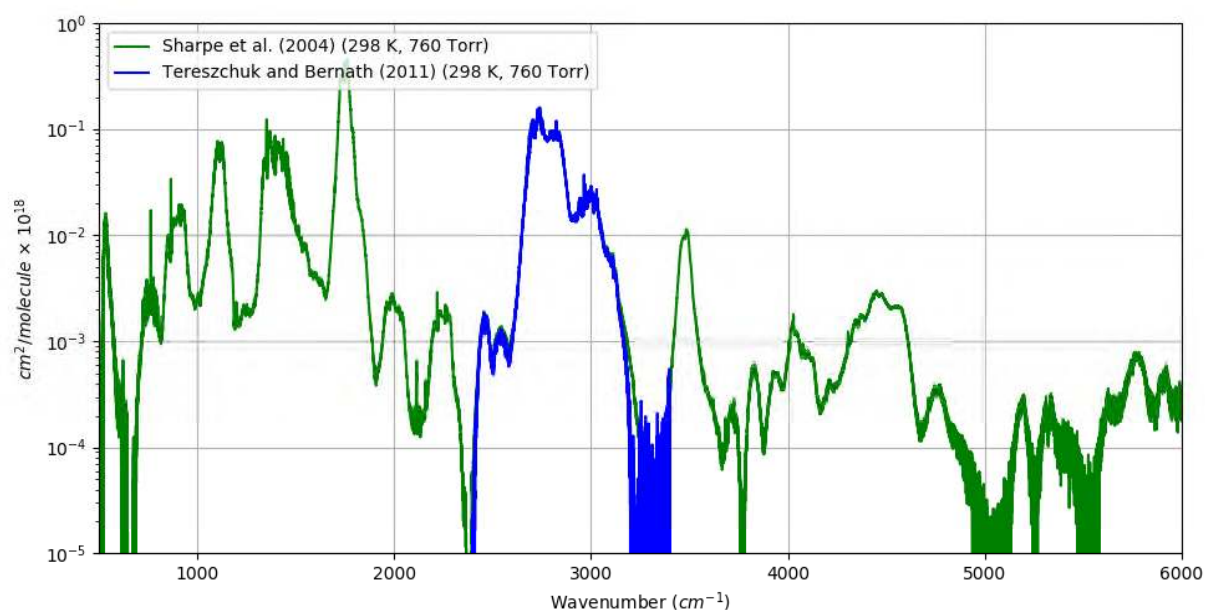
1 For acetone, the PNNL data do not provide new strong features, but do display a saturated
2 band at the lower edge of the PNNL spectral range. However, the strongest feature centered at
3 1750 cm^{-1} in HITRAN2012 (scaled data from Refs. [83] and [85]) was cut-off at 1780 cm^{-1} ,
4 while PNNL, although having lower resolution and thermodynamic range, provides this feature
5 uncut. For this reason, the PNNL data were included into the supplemental folder (as a post-
6 release [2] update) and will be considered for including to the HITRAN*Online* interface in the
7 future.

8 For around half of the molecules from Refs. [13,22], the original PNNL data were overwritten
9 by the cross-sections supplied in Ref. [22], some of them re-measured, such as crotonaldehyde
10 ($\sim 7\%$ difference in integrated intensities at room temperature between Refs. [13] and [22], acetol
11 ($\sim 6.7\%$ difference), and methyl propyl ketone (1.3% difference). In **Table 5**, spectra for all
12 molecules are given for 760 Torr pressure, and for a maximum of three temperatures, which are
13 specified for each molecule separately. All spectra are nitrogen-broadened, with additional self-
14 broadened data for only glycoaldehyde due to its rapid dimerization upon nitrogen pressure-
15 broadening (in the table the corresponding entry is marked as “*neat*”). For methyl propyl ketone,
16 new sets (298.15 K and 323.15 K) were added from Ref. [22] to the 278.15 K set from the
17 original PNNL database.

18 For four compounds, high-quality data are present in the NIST quantitative infrared database
19 [17]. We used these data to validate the spectra added to the HITRAN database. **Figure 3** shows
20 comparisons between PNNL and NIST measurements for ethyl alcohol, ethyl acrylate,
21 methylethyl ketone, and ethyl acetate (at the same thermodynamic conditions). PNNL (green)
22 spectra are overlapped by NIST (blue). It is interesting to note that the NIST data have uniformly
23 larger absorption, having noticeable intensity differences in the stronger bands, which possibly
24 can be attributed to the intensity calibration. Quantitatively, the integrated intensities for these
25 room-temperature spectra are in good agreement, and their relative residuals (3.1% for ethyl
26 alcohol, 2.6% for ethyl acrylate, 0.7% for methylethyl ketone, and 0.35% for ethyl acetate) are
27 within the combined measurement uncertainties of the PNNL and NIST measurements (see
28 **Table 4**). Integration was performed over the $500\text{-}4000\text{ cm}^{-1}$ wavenumber range.

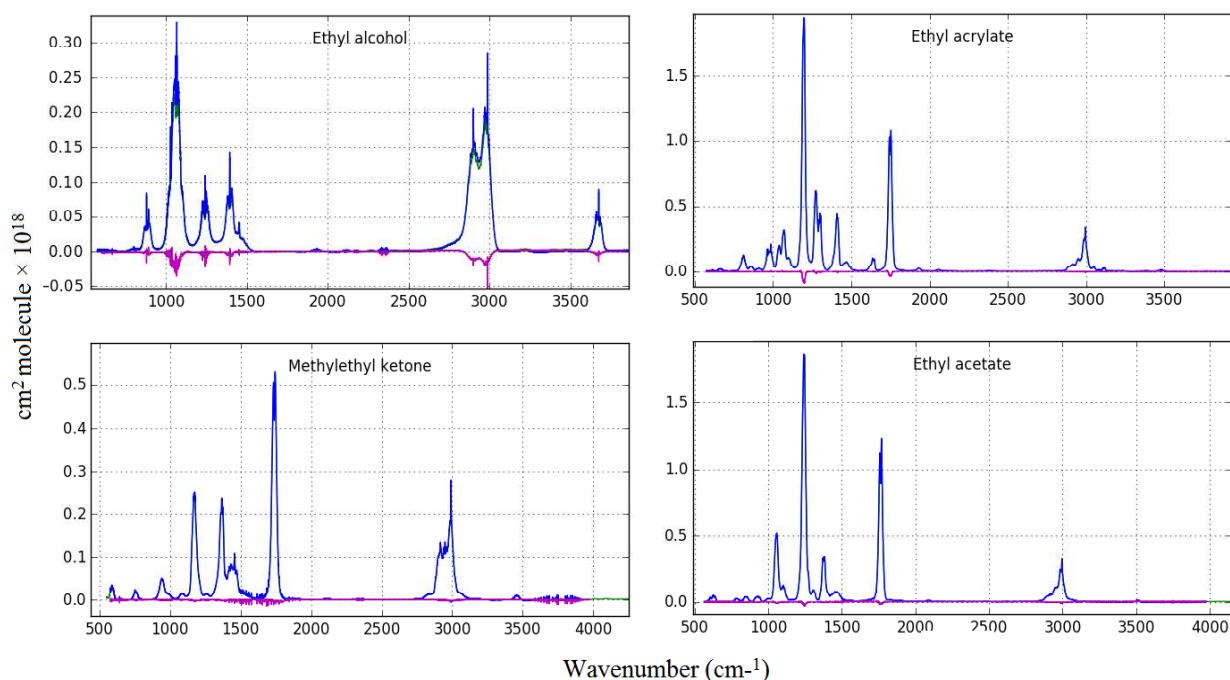
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Figure 2. Comparison of the spectral coverage for the acetaldehyde (acetylaldehyde) data from PNNL [13] which are new to HITRAN and those from Tereszchuk and Bernath [82] represented in the HITRAN2012 [6] database. The HITRAN2016 set is a combination of the spectra from these two sources.



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Figure 3. Comparison of PNNL [13] (green) and NIST [17] (blue) spectra. Residuals (PNNL-NIST) are displayed on the same scale in magenta. Intensity differences are apparent in the stronger bands.

10 **Table 5.** PNNL data for alcohols, ethers and other oxygenated hydrocarbons.

Molecule	CAS	Range, cm ⁻¹	T, K
PNNL – SERDP DATA / Johnson et al. [22]			
1-Penten-3-ol	616-25-1	550-7300	298.15, 323.15
2,3-Butanedione	431-03-8	580-6500	278.15, 298.15, 323.15

2,3-Dimethylfuran	14920-89-9	510-6500	278.15, 298.15, 323.15
2,5-Dimethylfuran	625-86-5	550-7000	278.15, 298.15, 323.15
2,6-Dimethoxyphenol	91-10-1	580-6500	323.15
2-Hexanol	626-93-7	520-7250	298.15, 323.15
2-Methylfuran	534-22-5	550-6500	278.15, 298.15, 323.15
2-Nonanone	821-55-6	550-6500	298.15, 323.15
2-Pentylfuran	3777-69-3	550-7300	298.15, 323.15
3,3-Demethyl-2-pentanol	19781-24-9	520-7250	298.15, 323.15
3-Methoxyphenol	150-19-6	550-6500	323.15, 333.15
3-methylfuran	930-27-8	550-6500	278.15, 298.15, 323.15
3-Pentanol	584-02-1	570-7250	298.15, 323.15
4-Methyl-1-pentanol	626-89-1	520-7250	298.15, 323.15
4-Methylvaleric acid	646-07-1	500-7300	323.15, 338.15
4-Penten-1-ol	821-09-0	520-7400	298.15, 323.15
5-nonanol	623-93-8	580-7200	298.15, 323.15
Acetol	116-09-6	435-6500	298.15, 323.15
Acrylic acid	79-10-7	525-6500	323.15
Cineole	470-82-6	520-7300	323.15
Diacetone alcohol	123-42-2	520-7000	298.15, 323.15
Ethyl benzoate	93-89-0	600-6500	323.15
Furfural	98-01-1	550-6500	298.15, 323.15
Geraniol	106-24-1	600-6500	323.15
Glycoaldehyde	141-46-8	570-6500	298.15, 323.15
Glycoaldehyde (neat) †	141-46-8	570-6500	298.15, 332.15
Glyoxal	107-22-2	600-6500	278.15, 298.15, 323.15
Guaiacol	90-05-1	520-7250	298.15, 323.15
Hexyl acetate	142-92-7	580-6500	298.15, 323.15
Isobutyric acid	79-31-2	530-7250	337.15
Isopentyl acetate	123-92-2	550-6500	298.15, 323.15
Isovaleraldehyde	590-86-3	500-6500	278.15, 298.15, 323.15
Limonene oxide	1195-92-2	510-6500	298.15, 323.15
Menthol	89-78-1	580-6500	323.15
Methyl 2-methylbutyrate	868-57-5	520-6500	298.15, 323.15
Methyl acetoacetate	105-45-3	520-6500	298.15, 323.15
Methyl vinyl ether	107-25-5	550-6500	278.15, 298.15, 323.15
Methylglyoxal	78-98-8	550-6500	298.15, 323.15
Octanoic acid	124-07-2	550-6500	298.15, 323.15
Phenol	108-95-2	550-7300	298.15, 323.15
Propylene carbonate	108-32-7	600-6500	298.15, 323.15
trans-Crotonaldehyde	123-73-9	500-6500	278.15, 298.15, 323.15
Valeraldehyde	110-62-3	600-6500	278.15, 298.15, 323.15
Valeric acid	109-52-4	520-7100	340.15
Methyl propyl ketone ‡	107-87-9	520-6500	278.15, 298.15, 323.15
PNNL – DOE DATA / Sharpe et al. [13]			
Acetic acid	64-19-7	550-7100	298.15, 323.15
Acetylaldehyde	75-07-0	510-6500	278.15, 298.15, 323.15
Acrolein	107-02-8	540-6500	278.15, 298.15, 323.15
Benzaldehyde	100-52-7	600-6500	278.15, 298.15, 323.15
Benzyl alcohol	100-51-6	550-7400	298.15, 323.15

Dimethyl ether	115-10-6	600-6500	278.15, 298.15, 323.15
Ethyl acetate	141-78-6	565-6500	278.15, 298.15, 323.15
Ethyl acrylate	140-88-5	580-6500	278.15, 298.15, 323.15
Ethyl alcohol	64-17-5	600-7500	278.15, 298.15, 323.15
Ethyl formate	109-94-4	600-6500	278.15, 298.15, 323.15
EVE	109-92-2	570-6500	278.15, 298.15, 323.15
Formic acid, dimer	14523-98-9	540-7200	278.15, 298.15
Furan	110-00-9	550-6500	278.15, 298.15, 323.15
Furfuryl alcohol	98-00-0	550-7300	298.15, 323.15
Isophorone	78-59-1	600-6500	298.15, 323.15
m-Cresol	108-39-4	570-6500	298.15, 323.15
Methyl butyl ether	628-28-4	600-6500	278.15, 298.15, 323.15
Methyl isoamyl ketone	110-12-3	550-6500	298.15, 323.15
Methyl methacrylate	80-62-6	550-6500	278.15, 298.15, 323.15
Methyl nitrite	624-91-9	520-6500	278.15, 298.15, 323.15
Methyl pivalate	598-98-1	550-6500	278.15, 298.15, 323.15
Methyl propionate	554-12-1	530-6500	278.15, 298.15, 323.15
Methyl salicylate	119-36-8	550-6500	298.15, 323.15
Methylethyl ketone	78-93-3	550-6500	278.15, 298.15, 323.15
Nitrobenzene	98-95-3	550-6500	298.15, 323.15
Propionaldehyde	123-38-6	520-6500	278.15, 298.15, 323.15

1 All spectra are broadened by nitrogen and recorded at 0.112 cm⁻¹ resolution.

2 † Self-broadened spectra, provided in Johnson et al. [22], in addition to the air-broadened data

3 ‡ Two cross-sections for Methyl propyl ketone from Sharpe et al. [13] were superseded by the later work
4 of Johnson et al. [22].

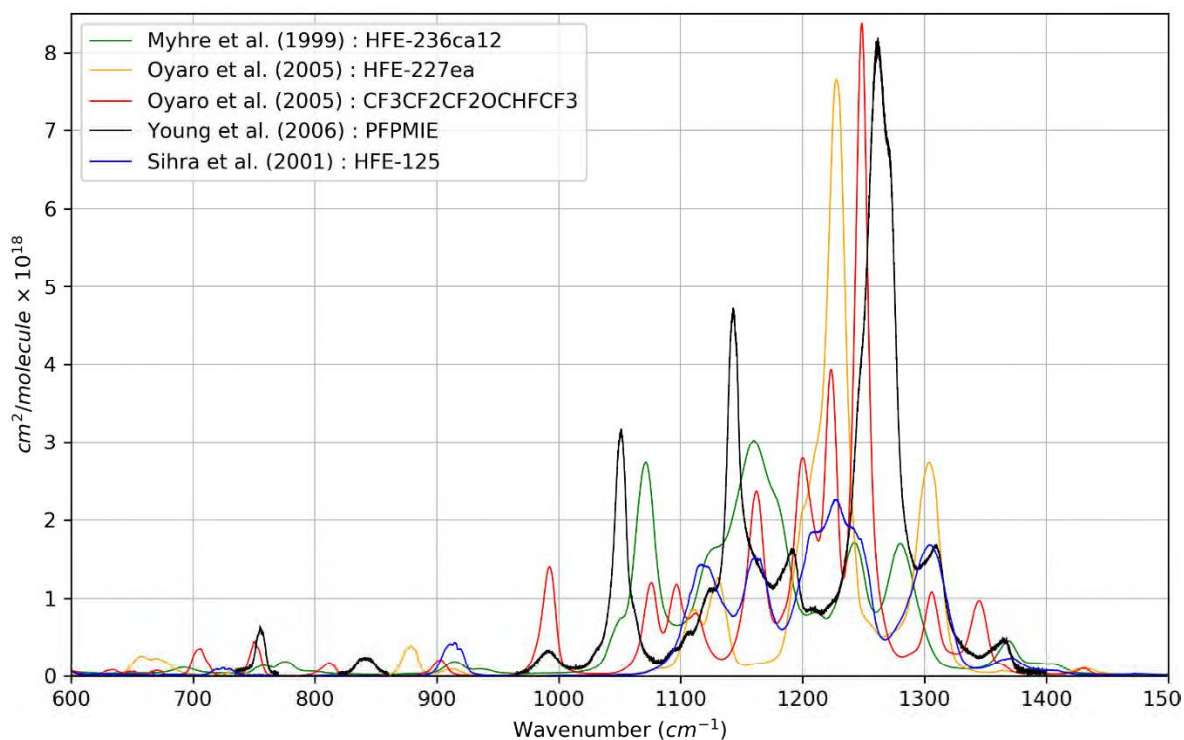
5 **3.2. Halogenated Alcohols and Ethers**

6 A significant number of halogenated alcohols and ethers have been added to the HITRAN
7 database. In this section the fluorinated and chlorinated compounds are presented. Stretching
8 modes of the C-F and C-O bonds occur between 1000 and 1360 cm⁻¹ depending on the
9 molecule; fluorinated compounds absorb in the atmospheric window region and are of interest
10 for climate science. The molecules have a broad range of lifetimes [87] from days to more than a
11 century in the case of CHF₂OCF₃ and a broad range of uses such as in anesthetics and heat
12 transfer agents. Although some of species are powerful greenhouse gases, on a molecule-per-
13 molecule basis, and potentially ozone depleting substances (ODSs) when they contain chlorine,
14 current emissions and atmospheric abundances are small, so far leading to negligible effects.
15 These compounds have GWPs that are lower than for the corresponding HFCs and HCFCs, thus
16 making halogenated alcohols and ethers potential HFC and HCFC substitutes [88]. In this
17 subsection, we give an overview for the compounds in HITRAN. Many of the spectra were
18 measured at room temperature, either air- or self-broadened, and were previously compiled in the
19 review on halocarbons and related compounds by Hodnebrog and coauthors [1]. Other data were
20 obtained from the PNNL database [13] and measurements published by Godin et al. [36,37] and
21 Rodriguez et al. [30,31]. In this section we consider the category of all compounds

1 corresponding to halogenated alcohols and ethers divided into the subcategories of hydrofluoro
2 ethers (HFEs), fluorinated polyethers (HFPEs and PFPEs), fluorinated alcohols, and fluorinated
3 aldehydes, ketones, and acids.

4 3.2.1. Hydrofluoroethers (HFEs)

5 Hydrofluoroethers (HFEs) are a potential replacement for chlorofluorocarbons (CFCs),
6 hydrofluorocarbons (HFCs), hydrochlorofluorocarbons (HCFCs), and halons. HFEs have higher
7 reaction rates with the atmospheric oxidants such as OH radicals, which reduces their
8 atmospheric lifetimes relative to CFCs [30,31,57,65]. HFEs do not contain chlorine and thus do
9 not take part in the chlorine-based catalytic destruction of atmospheric ozone [54,89]. This
10 makes HFEs an attractive replacement to CFCs and their derivatives in various applications,
11 such as, for instance, production of carrier fluids for lubricants, cleaning solvents for electronic
12 components, fire suppression agents, heat transfer fluids (e.g. in air conditioners and
13 refrigeration), and anesthetic agents [30,54,55,57,62,76,79]. Due to the C-F and C-O bonds,
14 fluorinated ethers and polyethers absorb in the atmospheric window and contribute to radiative
15 forcing of climate change [66,68]. **Figure 4** shows the spectra of the five compounds (HFEs and
16 PFPEs) having 100-year GWPs higher than 5000 [1]



19
20 **Figure 4.** Spectra of five fluorinated ethers and polyethers in HITRAN2016 having the largest
21 100-year GWP values among halogenated alcohols and ethers category: HFE-236ca12 (HG-10)
22 [60], HFE-227ea [56], Fluoroether E-1 ($\text{CF}_3\text{CF}_2\text{CF}_2\text{OCHF}_2\text{CF}_3$) [56], PPFMIE [68], and HFE-

1 125 [61]. Corresponding 100-year GWPs are 5350, 6450, 6490, 9710, and 12400, respectively
2 [3].

3 3.2.2. Fluorinated polyethers (HFPEs and PFPEs)

4 Fluorinated polyethers (FPEs) contain multiple oxygen atoms creating a chain of hydrocarbon
5 groups. Similar to HFEs, FPEs have been used as CFC replacements as heat transfer fluids,
6 refrigeration working fluids, solvents, fire suppressants, and heat exchange agents [60,66,68,78].
7 As with HFEs, FPEs do not contain chlorine and thus do not contribute to chlorine-based
8 catalytic destruction of atmospheric ozone [66].

9 The HITRAN2016 edition includes two groups of fluorinated polyethers,
10 hydrofluoropolyethers (HFPEs) and perfluoropolyethers (PFPEs). The presence of multiple ether
11 groups and C-H bonds in HFPEs generally limits the atmospheric lifetimes and GWPs of these
12 substances mainly by reactions with atmospheric OH radicals [66,78]. A subgroup of HFPEs, H-
13 Galdens, with the general formula $\text{CHF}_2\text{O}(\text{CF}_2\text{O})_m(\text{CF}_2\text{O})_n\text{CF}_2\text{H}$ have low reactivity towards OH
14 radicals, long atmospheric lifetimes, and high GWPs [1].

15 Unlike HFPEs, PFPEs have all hydrogen atoms substituted by fluorine. This makes these
16 compounds unreactive in the lower atmosphere due to the lack of the C-H bonds. The absence of
17 sites for OH reaction leads to long atmospheric lifetimes and hence high GWPs. The lifetimes of
18 PFPEs are limited by photolysis in the upper atmosphere [68].

19 The HITRAN2016 update includes self-broadened [31,55–57], air-broadened [54,61,65,71]
20 and nitrogen-broadened spectra [89] for fluorinated ethers, polyethers, and esters measured at
21 room temperature (295–298 K). For these molecules, as well as for the rest of the subgroups of
22 halogenated alcohols and ethers, the full list of molecules made available through the
23 HITRAN*Online* interface is presented in **Table 6** and **Table 7**. **Table 6** contains data provided in
24 Hodnebrog et al. [1] with original sources given in the table, while **Table 7** lists data from Godin
25 et al. [36,37], Rodriguez et al. [30,31], and the PNNL database [13].

26 For a number of molecules, comparisons were made where possible to decide which data
27 should be provided via the HITRAN*Online* interface, and which should be put into the
28 supplemental folder. **Figure 5** shows spectra from different sources for molecules from HFE,
29 HFPE, and PFPE subcategories. Spectra for HFE-263fb1 were published by Osterstrom et al.
30 [54] and Oyaro et al. [56]. Agreement in spectral band shapes and integrated band intensities is
31 very good (~0.15%). According to Ref. [56], these compounds did not show “discernible
32 dependency in their absorption cross section by the addition of N_2 to atmospheric pressure”.
33 Spectra from Ref. [54] were included in the HITRAN*Online* web interface because of their

1 higher resolution. For HFE-356mff2, spectra from Sihra et al. [61] and Oyaró et al. [55] had
2 been compared. There is good agreement between the band shapes and the difference in the
3 reported integrated band intensities is $\sim 3.3\%$ which lies well within the combined experimental
4 uncertainties (see **Table 4**). Due to higher resolution, the data from Ref. [61] were included to
5 the HITRAN*Online* interface. Spectra for HFE-245fa2 were measured in Refs. [61] and [56] with
6 a good agreement in their integrated intensities (1.65%). The spectra with higher resolution,
7 provided in Ref. [61], were included in HITRAN*Online*. For HFE-7200 and HFE-7000 (HFE-
8 347mcc3), the spectra published by Bravo et al. [57] were included in HITRAN*Online*. **Figure 5**
9 displays comparisons of spectra from this source with Ref. [61] and Ninomyya et al [89] for
10 HFE-7200 and HFE-347mcc3 respectively. In both cases, the data from Ref. [57] had higher
11 resolution (0.1 cm^{-1} for HFE-7200 and 0.03 cm^{-1} for HFE-7000). The data provided in Ref. [57]
12 are self-broadened; no significant differences between self- and air-broadened spectra for this
13 molecule were reported [57]. Since the data from this reference show non-linearity at the edges
14 of the spectra (as can be seen by the low-resolution residual on the plots) it was decided to
15 relegate the other data [61,89] on these molecules to the supplemental folder.

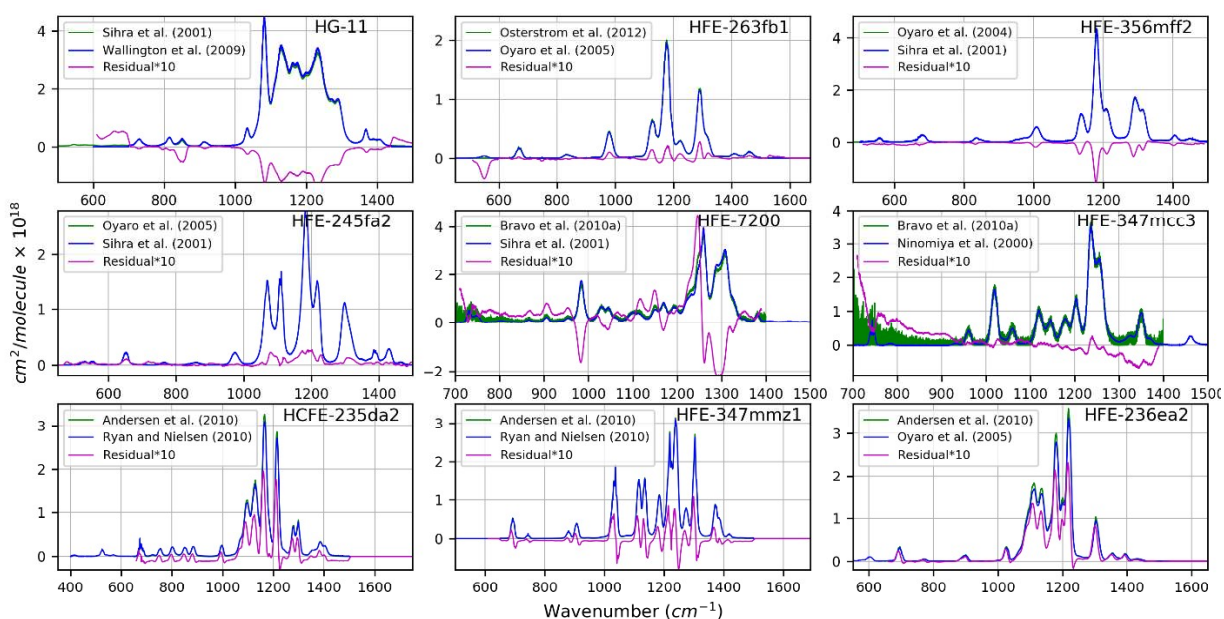
16 For general anesthetic gases, the HITRAN2016 edition includes self- and air-broadened
17 spectra of HCFE-235ca2 (enflurane), HCFE-235da2 (isoflurane), HFE-236ea2 (desflurane), and
18 HFE-347mmz1 (sevoflurane) previously published by Ryan and Nielsen [79] and Andersen et al.
19 [62,76]. **Figure 5** shows spectra for isoflurane, sevoflurane, and desflurane. Measurements for
20 isoflurane were published by Sihra et al. [61], Andersen et al. [76], and Ryan and Nielsen [79].
21 The total integrated absorption reported in these three studies agrees within 5.7%. For the
22 strongest band, the data from Ref. [76] have the largest absorption, followed by Ref. [79], then
23 by Sihra et al. [61]. Spectra from Ref. [76] do not include small features at 500 cm^{-1} . All three
24 sources have the same uncertainty (see **Table 4**), and the corresponding spectra agree well within
25 the stated uncertainty. Spectra from Ref. [79] were included in the HITRAN*Online* web interface
26 since it roughly represents the mean value for the band shapes of the two other spectra, and gives
27 the same GWP value that was obtained in Hodnebrog et al. [1] using the data from Ref. [76].
28 Spectra from Refs. [76] and [61] were placed in the supplemental folder of the HITRAN
29 database. For sevoflurane, there is excellent agreement in band shapes and integrated intensities
30 ($\sim 0.3\%$) in the spectra published by Andersen et al. [76] and Ryan and Nielsen [79]; the spectra
31 from Ref. [76] were included in the HITRAN*Online* web interface due to their better resolution.
32 Desflurane spectra considered in the comparison were published by Andersen et al. [76] and
33 Oyaró et al. (2005) [56]. The integrated intensities differ by 6.6%, which is within the combined
34 uncertainties. The cross-section values from Ref. [76] are uniformly larger than those from Ref.

1 [56], with the difference being more pronounced in the absorption peaks. Although having
2 similar integrated cross-sections, the data from Oyaro et al. are more complete in terms of the
3 spectral range and contain minor features. Spectra from Ref. [56] have been made available via
4 HITRAN*Online* because of their greater stated accuracy (2%). The spectra from Ref. [76] were
5 added to the supplemental folder.

6 For H-Galdens (HFPEs), the HITRAN2016 edition includes the following data published in
7 different years: HG'-01, HG'-02, and HG'-03 from the work by Andersen et al. [78], HG-01 and
8 HG-10 from Myhre et al. [60], and HG-11 from Wallington et al. [66]. In **Figure 5**, the
9 comparison for HG-11 is shown. The spectrum from Ref. [66] has been made available through
10 the HITRAN*Online* web interface because it supersedes that published by Sihra et al. [61].

11 For PFPEs, the database update includes only PPFMIE spectra, which was initially published
12 by Young et al [68] and, like many other molecules, is provided in HITRAN format in Ref. [1].
13 For fluorinated esters, the recently measured self-broadened data published by Rodriguez et al.
14 [30] have been included for allyl 2,2,2-trifluoroacetate (ATFA) and vinyl 2,2,2-trifluoroacetate
15 (VTFA).

16



17

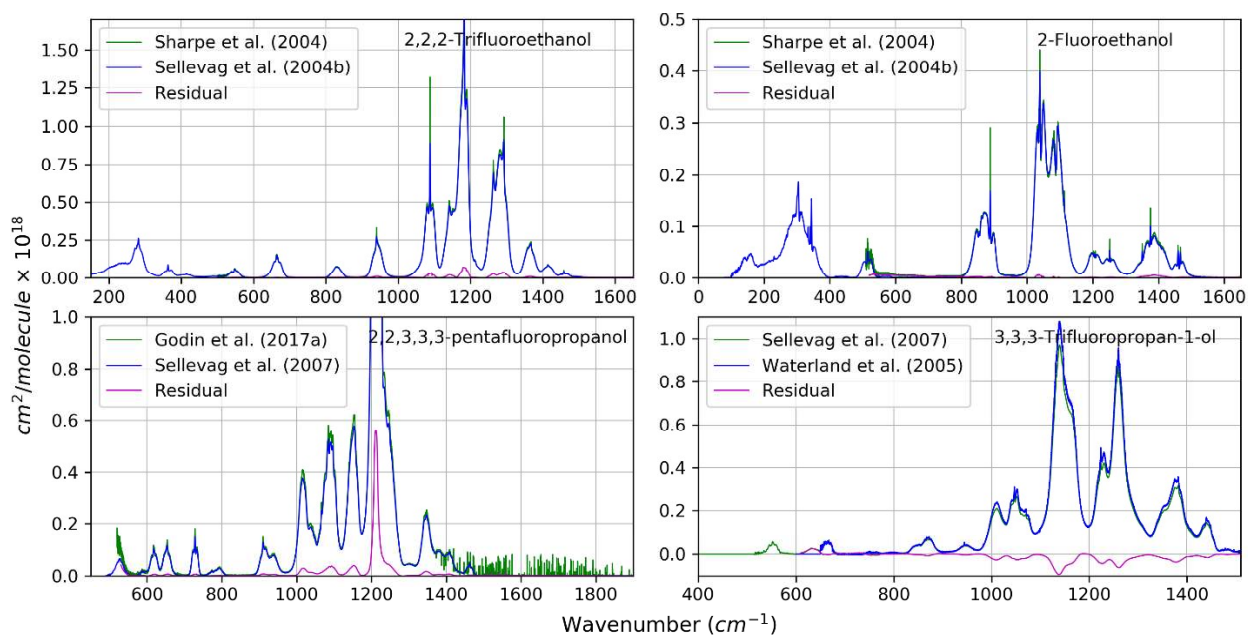
18 **Figure 5.** Spectra for HFEs, HFPEs, and anesthetics. The residuals (magenta derived from the
19 green minus blue lines in each frame) are multiplied by 10. Due to the fact that in all cases the
20 spectral resolutions for the raw spectra are different, the wavenumber-dependent residuals were
21 calculated on spectra with the final resolution degraded to 10 cm⁻¹ to show the systematic
22 discrepancies. The data made available through the HITRAN*Online* web interface are outlined in
23 **Table 6** for each molecule.

1 3.2.3. Fluorinated alcohols

2 Fluorinated alcohols are used in electronic and industrial applications, and in organic
3 synthesis [37,47]. Due to the presence of the C-F bond, these compounds absorb in the
4 atmospheric window region, however because their atmospheric lifetimes are relatively short
5 they have small global warming potentials [1]. The compounds from this subclass, which are
6 included in the HITRAN2016 edition, include partially fluorinated alcohols and fluorotelomer
7 alcohols (FTOHs).

8 For partially-fluorinated alcohols, the HITRAN2016 edition includes the following updates:
9 the spectra recently published by Godin et al. for 2,2,3,3,3-pentafluoropropan-1-ol / *PFPO* [37]
10 and 1,1,1,3,3,3-hexafluoropropan-2-ol / *HFIP* [36]. Spectra for both molecules are self-
11 broadened, recorded with 0.1 cm⁻¹ resolution, and have a strong absorption in the 500-1500 cm⁻¹
12 spectral range. A brief description of these cross-sections is given in **Table 7**. **Figure 6** shows a
13 comparison of the room-temperature spectra for 2,2,3,3,3-pentafluoropropanol published by
14 Godin et al [37] and Sellevag et al. [75]. Older data published in Ref. [75] were measured with
15 lower resolution. The data from Godin et al. [37] were included in the HITRAN*Online* interface,
16 while the data from Ref. [75] were placed in the supplemental folder. Note that the latter data
17 contain the uncut spectral features below 600 cm⁻¹. The integrated absorption between those
18 cross-sections differs by 8%, which can possibly be attributed to the intensity calibration. For
19 2,2,2-trifluoroethanol (CF₃CH₂OH) and 2-fluoroethanol (CH₂FCH₂OH), room-temperature
20 spectra were compared between two sources, namely the PNNL database [13] and Sellevag et al.
21 [59]. These comparisons are also presented in **Figure 6**. Data from PNNL have been included in
22 the HITRAN*Online* interface since they are better resolved and provide more temperature
23 coverage. Due to the additional features at the lower wavenumber range, spectra from Sellevag
24 et al. have also been included in HITRAN*Online*. For 2,2,3,3,4,4,4-heptafluorobutan-1-ol, a
25 comparison was made between the room-temperature spectra published by Bravo et al. [57] and
26 Sellevag et al. [75] with an integrated absorption cross-section difference of 6.1%, which lies
27 within the combined uncertainties (see **Table 4**). Data from Ref. [57] have been included in
28 HITRAN*Online* for this molecule. Three more molecules (2,2-difluoroethanol, 2,2,3,3-
29 tetrafluoro-1-propanol, and 2,2,3,4,4,4-hexafluoro-1-butanol) from Refs. [59] and [75] have been
30 included in HITRAN*Online*. Cross-sections for fluorotelomer alcohols (1H,1H,2H,2H-
31 Perfluorononan-1-ol, 1H,1H,2H,2H-perfluoroundecan-1-ol, 3,3,4,4,5,5,6,6,7,7,7-
32 undecafluoroheptan-1-ol, and 3,3,3-Trifluoropropan-1-ol) were included in HITRAN*Online* from
33 the work by Waterland et al. [67]. For one of these molecules, 3,3,3-trifluoropropan-1-ol
34 (CF₃CH₂CH₂OH), comparison with the data from Ref. [75] is displayed in **Figure 6**. Data from
35 Ref. [75] have been added to the supplemental folder since it has the uncut absorption features

1 below 650 cm^{-1} . Finally, three molecules from the chloroethenes subclass (chloromethanol,
2 dichloromethanol, trichloromethanol) published by Wallington et al. [64] have been included in
3 HITRAN*Online*.



4
5 **Figure 6.** Spectra for the fluorinated alcohols. The residual (magenta derived from the green
6 minus blue lines in each frame) is given on the same scale as the spectra. The spectral resolutions
7 for the raw spectra are different, and wavenumber-dependent residuals were calculated using
8 spectra with resolution degraded to 10 cm^{-1} to show the systematic discrepancies. The data made
9 available via the HITRAN*Online* web interface are outlined in [Table 6](#) and [Table 7](#) for each
10 molecule.

11 3.2.4. Fluorinated aldehydes, ketones, and acids

12 A number of fluorinated aldehydes, ketones and acids were added to the HITRAN database.
13 Aldehydes are the degradation products of alcohols and HFCs [74]. Three perfluoroalkyl
14 aldehydes (pentafluoropropionaldehyde, heptafluorobutyraldehyde, and nonafluoropentanal)
15 were added from the work by Hashikawa et al. [74]. For 2,2,2-trifluoroacetaldehyde (CF_3CHO),
16 spectra were measured both in Refs. [74] and Sellevag et al. [47]. In the latter reference the
17 uncertainties are given for each band separately in Table 1 of their paper, and the overall
18 uncertainty for intensity is less than two percent. In Hashikawa et al. [74], the uncertainty is 5%
19 over the whole range. According to the figures, the integrated intensity of the spectrum from
20 Hashikawa et al. is lower by 16%, and has a partly negative baseline, possibly due to
21 miscalibration. The agreement between these two spectra is good and lies within the combined
22 uncertainties. The data from Ref. [47] have been included in HITRAN*Online*, while spectra from
23 Ref. [74] have been put in the supplemental folder, as well as the raw data from Ref. [47]
24 containing some negative values. In addition, the spectra from Ref. [47] for 3,3,3-Trifluoro-
25 propanal have been included in HITRAN*Online*. Finally, spectra for two ketones, chloroacetone

1 and perfluoro-2-methyl-3-pentanone, and trifluoroacetic acid, have been included in
 2 HITRAN*Online* from Refs. [13], [50], and [13] respectively. The brief outline of the
 3 experimental conditions for all molecules described in this section can be found in **Table 6** and
 4 **Table 7**.

5 **Table 6.** Spectra from the review on halocarbons and related compounds (Hodnebrog et al. [1])
 6 for halogenated alcohols and ethers.

Molecule	CAS	T (K)	P (Torr)	Broadener	Resolution (cm ⁻¹)	Source
HG-01	73287-23-7	296	700	air	0.25	
HG-02	485399-46-0	296	700	air	0.25	Andersen et al. (2004) [78]
HG-03	485399-48-2	296	700	air	0.25	
HFE-347mmz1	28523-86-6	298	700	air	0.25	Andersen et al. (2010) [76]
HCFE-235ca2	13838-16-9	298	700	air	0.25	Andersen et al. (2012) [62]
2,2,3,3,4,4,4- Heptafluorobutan-1-ol	375-01-9	298			0.1	
HFE-347mcc3	375-03-1	298			0.03	Bravo et al. (2010) [57]
HFE-449s1		298			0.1	
HFE-7200		298			0.1	
perfluoro-2-methyl-3- pentanone	756-13-8	298		air	1	D'Anna et al. (2005) [50]
Heptafluorobutyraldehyde	375-02-0	296	700	air	0.5	
Nonafluoropentanal	375-53-1	296	700	air	0.5	Hashikawa et al. (2004) [74]
Pentafluoropropionaldehyde	422-06-0	296	700	air	0.5	
HFE-216	1187-93-5	296	700	air	0.25	Mashino et al. (2000) [71]
HG-01	188690-78-0	296		air	1	Myhre et al. (1999) [60]
HG-10	78522-47-1	296		air	1	
HFE-263fb1	460-43-5	296	700	air	0.25	Osterstrom et al. (2012) [54]
1,1,2,2-Tetrafluoro-3- methoxy-propane	60598-17-6	298			1	
HFE-356mmz1	13171-18-1	298			1	Oyaro et al. (2004) [55]
HFE-365mcf3	378-16-5	298			1	
1,1,2-Trifluoro-2- (trifluoromethoxy)-ethane	84011-06-3	298			1	
1-Ethoxy-1,1,2,3,3,3- hexafluoropropane	380-34-7	298			1	
Fluoroether E-1	3330-15-2	298			1	Oyaro et al. (2005) [56]
HFE-227ea	2356-62-9	298			1	
HFE-236ea2	57041-67-5	298			1	
HFE-263m1	690-22-2	298			1	
HCFE-235da2	26675-46-7	298			1	Ryan and Nielsen (2010) [79]
2,2,2-trifluoroacetaldehyde	75-90-1	298			1	
3,3,3-Trifluoro-propanal	460-40-2	298			1	Sellevag et al. (2004) [47]
2,2,2-Trifluoroethanol	75-89-8	293			1	
2,2-Difluoroethanol	359-13-7	293			1	Sellevag et al. (2004) [59]
2-Fluoroethanol	371-62-0	293			1	
2,2,3,3-Tetrafluoro-1-propanol	76-37-9	298				
2,2,3,4,4,4-Hexafluoro-1- butanol	382-31-0	298				Sellevag et al. (2007) [75]
HFE-125	3822-68-2	296	700	air	0.5	
HFE-143a	421-14-7	296	700	air	0.5	
HFE-245fa2	1885-48-9	296	700	air	0.5	
HFE-356mff2	333-36-8	296	700	air	0.5	Sihra et al. (2001) [61]
i-HFE-7100	163702-08-7	296	700	air	0.5	
i-HFE-7200	163702-06-5	296	700	air	0.5	
n-HFE-7100	163702-07-6	296	700	air	0.5	
Chloromethanol	62-51-1	295	700	N2	0.25	
Dichloromethanol	35911-93-4	295	700	N2	0.25	Wallington et al. (2000) [64]
Trichloromethanol	35911-92-3	295	700	N2	0.25	
HFE-329me3	428454-68-6	296	700	air	0.25	
HFE-338mec3	56860-85-6	296	700	air	0.25	Wallington et al. (2004) [65]

HG-11		296	700	air	0.25	Wallington et al. (2009) [66]
1H,1H,2H,2H-Perfluorononan-1-ol	755-02-2	296	700	air	0.5	
1H,1H,2H,2H-Perfluoroundecan-1-ol	87017-97-8	296	700	air	0.5	Waterland et al. (2005) [67]
3,3,4,4,5,5,6,6,7,7,7-Undecafluoroheptan-1-ol	185689-57-0	296	700	air	0.5	
2,2,3,3,3-pentafluoropropanol	2240-88-2	296	700	air	0.5	
PFPME	1309353-34-1	296	700	air	0.25	Young et al. (2006) [68]

1 **Table 7.** Spectra from the PNNL database (Sharpe et al. [13]) and more recently published
2 sources for halogenated alcohols and ethers.

Molecule	CAS	T (K)	P (Torr)	Broadener	Resolution (cm ⁻¹)	Source
Trifluoroacetic acid	76-05-1	298.15	760.0	N ₂	0.112	
2-Fluoroethanol	371-62-0	298.15, 323.15	760.0	N ₂	0.112	
bis(2-Chloroethyl) ether	111-44-4	298.15, 323.15	760.0	N ₂	0.112	
2,2,2-Trifluoroethanol	75-89-8	278.15, 298.15, 323.15	760.0	N ₂	0.112	Sharpe et al. [13]
Ethyl 2,2,2-trifluoroacetate	383-63-1	278.15, 298.15, 323.15	760.0	N ₂	0.112	
Chloroacetone	78-95-5	278.15, 298.15, 323.15	760.0	N ₂	0.112	
HFE-7300	132182-92-4	298.00			1	Rodriguez et al. [31]
HFE-7500	297730-93-9	298.00			1	
Allyl 2,2,2-trifluoroacetate	383-67-5	298.00			0.03	Rodriguez et al. [30]
Vinyl 2,2,2-trifluoroacetate	433-28-3	298.00			0.03	
1,1,1,3,3,3-Hexafluoropropan-2-ol	920-66-1	300.40, 303.50, 312.10, 326.80, 335.60, 343.10, 353.80, 362.40			0.1	Godin et al. [36] †
2,2,3,3,3-pentafluoropropanol	422-05-9	297.80, 303.60, 311.90, 317.50, 324.40, 334.20, 343.10, 352.80, 362.20			0.1	Godin et al. [37] †

3 † Detailed information on the experimental conditions can be found in the corresponding
4 references.

5 3.3. Hydrofluorocarbons

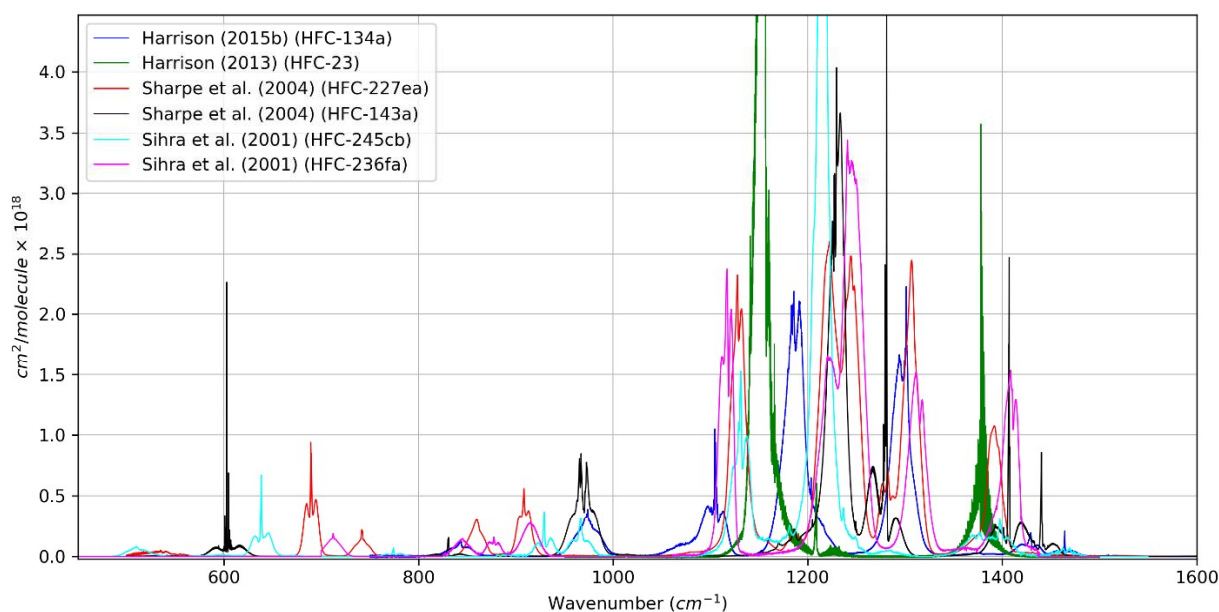
6 Hydrofluorocarbons (HFCs) are widely used as replacement for ozone depleting substances
7 (ODS) such as CFCs and HCFCs in industrial and domestic applications and are included in the
8 Kyoto Protocol and the Kigali Agreement. HFCs are used in fire suppression, heat transfer,
9 refrigeration, air conditioning, foam blowing, solvent production, and cleaning applications
10 [72,90,91]. The absence of Cl or Br atoms means that HFCs do not contribute to stratospheric
11 ozone depletion [52,69]. However, because of their strong absorption in the 8-13 μm
12 atmospheric window, some members of this class of compounds are potent greenhouse gases
13 [1,91,92]. The tropospheric lifetimes for HFCs are determined mostly by the reaction with OH

1 radicals [52,72] and range from days to centuries depending on the reactivity of the HFC. We
2 describe the infrared spectra for two groups of HFCs, namely saturated and unsaturated
3 hydrofluorocarbons.

4 3.3.1. Saturated HFCs

5 The saturated HFCs, i.e. those containing only single interatomic bonds, have been widely
6 used since the early 1990s after emissions of CFCs were controlled by the Montreal Protocol
7 [93]. The most abundant HFC in the atmosphere is HFC-134a, with a global-mean surface
8 concentration of approximately 100 ppt [94], with several others with concentrations exceeding 1
9 ppt, followed by HFC-23 (about 30 ppt) which is mostly emitted as a by-product of HCFC-22
10 production [3]. The 2011 radiative forcing due to HFCs is about 0.024 W m^{-2} , about 1.3% of the
11 forcing due to CO_2 [3].

12 The HITRAN2016 database release includes cross-sections for HFCs which have been
13 introduced in HITRAN for the first time, namely HFC-161, HFC-152, HFC-41, HFC-272ca,
14 HFC-143, HFC-365mfc, HFC-245fa, HFC-236cb, HFC-43-10mee, HFC-329p, HFC-227ca,
15 HFC-227ea, HFC-245cb, HFC-236fa, HFC-23. Cross-sections for these molecules have been
16 provided by Harrison et al. [24][25], Hodnebrog et al. [1], Le Bris et al. [39], and Sharpe et al.
17 [13]. The comparison overview of the saturated HFCs having the largest GWP values is shown
18 in **Figure 7**.



19

20 **Figure 7.** Spectra for HFC-134a, HFC-23, HFC-227ea, HFC-143a, HFC-245cb, and HFC-
21 236fa.

22

3.3.1.1. HFC-152a

HFC-152a (1,1-difluoroethane, CH_3CHF_2) is a saturated compound with a small GWP which is used in refrigeration and aerosol products. The HITRAN2012 edition included data from Clerbaux et al. [18]. These data were measured at a temperature range of 253-287 K and “no dependence of these cross-sections on pressure was found” [18]. The spectra from this reference are retained in HITRAN2016 with no changes. Additional spectra from different sources have been added to the HITRAN2016 issue for temperatures of 253 K, 278 K, 296 K, and 323 K. For 253 K, data provided by Highwood and Shine [90] have been included in HITRAN*Online*. These self-broadened spectra were originally measured by Vander Auwera at 0.03 cm^{-1} resolution and described in Ref. [95]. According to our comparisons, the spectra from Refs. [90] and [95] are somewhat better resolved compared to the old data from Clerbaux et al., although the nominal resolution is the same for both sources. For temperatures near 253 K, spectra from Refs [90,95] have smaller integral intensity with the discrepancies of 4.3% for strongest band near 1140 cm^{-1} , 5.4% and 6.5% for weaker bands at 950 and 1450 cm^{-1} respectively, compared to Ref. [18]. At temperatures near 270 K, excellent agreement can be observed between the data from PNNL and Clerbaux et al. ($<0.8\%$). At temperatures near 296 K, data from Clerbaux et al. [18], Sihra et al. [61], and Sharpe et al. (PNNL) [13] have been compared. Data from Clerbaux et al. have more resolved spectral features, while the resolution of spectra from PNNL and Sihra et al. are comparable. PNNL data have a wider spectral range and contain weak (but visible on a linear scale) features at lower and higher wavenumbers. Data from Clerbaux et al., having the intensity uncertainty of 3%, agree well with Sihra et al. (0.1%-1.6%) and with PNNL (4%) in terms of integrated intensities. PNNL spectra, having wider spectral coverage than Sihra et al., smaller uncertainty, and comparable resolution, have been made available via the HITRAN*Online* interface. Data from Sihra et al. have been relegated to the supplemental folder. For 323K, only data from PNNL are present.

3.3.1.2. HFC-125

HFC-125 (pentafluoroethane, C_2HF_5) is a fire suppression agent and refrigerant. While it has no ozone depleting effect, it has a significant global warming potential. HITRAN2016 inherited high-resolution data from HITRAN2012 provided by Di Lonardo and Masciarelli [40]. This dataset contains both self- and air-broadened spectra at a range of temperatures between 200 K and 295 K, and pressures between 150 Torr and 600 Torr (for air-broadened spectra). It is important to note that one self-broadened spectrum from Di Lonardo and Masciarelli has a large shift on a wavenumber scale, which possibly can be explained by errors in the data format. This spectrum (293K) was removed from the HITRAN*Online* web interface and placed into the supplemental folder. The rest of the data from Di Lonardo and Masciarelli are available in the

1 HITRAN*Nonline* interface. Additional data published by Young et al. [48], Sihra et al. [61], and
2 Sharpe et al. (PNNL) [13], spanning temperatures from 253 K to 323 K, have been considered
3 for incorporation into the database. The 253K spectra from Highwood and Shine [90] was
4 omitted for this molecule because it was originally measured and provided to the HITRAN
5 database by Di Lonardo and Masciarelli [40]. PNNL data agree well with Di Lonardo and
6 Masciarelli at all available temperatures. The intensities integrated between 1000-1400 cm^{-1}
7 agree well within the combined experimental uncertainties (2.3% for 273-278K and 2.6% for
8 293-298K) for these sources. At 293-298K, in addition to spectra from these sources, data from
9 Sihra et al. and Young et al. have been included in the comparison, as well as spectra from the
10 NIST database [17]. Comparison shows that data from Young et al. disagree with most of the
11 other published spectra, having integrated intensity residuals of 9.2%, 5.67%, 6.9%, and 9.03%
12 compared to Di Lonardo and Masciarelli, Sihra et al., PNNL, and NIST respectively. The latter
13 sources have internal agreement of not worse than 3.9% for integrated intensities in the 1000-
14 1400 cm^{-1} interval. For this reason, and assuming that data from Sihra et al. does not have
15 features additional to those from PNNL, the data from Young et al. and Sihra et al. have been
16 omitted from HITRAN.

17 **3.3.1.3. HFC-134**

18 HFC-134 (1,1,2,2-tetrafluoroethane, CHF_2CHF_2) is not currently in common usage,
19 although it has RE of 0.19 $\text{W m}^{-2} \text{ppb}^{-1}$ which is comparable to HFC-23 [1]. The HITRAN2016
20 issue includes self- and air-broadened spectra provided by Smith et al. [16] for a range of
21 temperatures from 203 K to 297 K. This set of data is unchanged from HITRAN2012. Besides
22 these data, the new spectra from PNNL containing absorption for higher temperatures have been
23 included in HITRAN*Nonline*. The data provided by Hodnebrog et al. (originating from Sihra et al.)
24 [1,61] were measured under conditions similar to PNNL, but with slightly lower resolution.
25 These data contain additional band features at lower wavenumbers ($<600 \text{ cm}^{-1}$) absent in the
26 PNNL spectra; thus they have been added to the supplemental folder.

27 **3.3.1.4. HFC-143a**

28 HFC-143a (1,1,1-trifluoroethane, CH_3CF_3) is used as a substitute for CFCs and HCFC in
29 refrigeration [35] and has one of the largest 100-yr GWP values (4800 [3]) in the HFC category.
30 HITRAN includes the data provided by Smith et al. [16] available via HITRAN*Nonline*, and Di
31 Lonardo and Masciarelli [40] in supplemental folder. These data were inherited by
32 HITRAN2016 from previous editions. Other data for this molecule were provided by Sihra et al.
33 [61], Sharpe et al. (PNNL) [13], and more recently by Le Bris and Graham [35]. The full dataset
34 can be divided into two parts. The first part contains the self-broadened spectra from Refs.

1 [16,35,40], and the second part the air- and nitrogen-broadened ones from Refs. [13,16,40,61].
2 Measurements performed in Smith et al. and Di Lonardo and Masciarelli demonstrate the strong
3 dependence of the spectral line shape on the presence of the broadener. For this reason, both self-
4 and nitrogen-broadened spectra have been included in HITRAN*Online*. For the nitrogen-
5 broadened data, the composite PNNL measurements for 278.15 K, 298.15 K, and 323.15 K have
6 been included in HITRAN*Online*. The similar data from Sihra et al., when compared with PNNL,
7 did not reveal any additional spectral features. For this reason the spectra from Sihra et al. have
8 been omitted. The recent self-broadened composite measurements performed by Le Bris and
9 Graham have been included in HITRAN*Online*. In Ref. [35] a review and comparison of the data
10 in HITRAN2012 on this molecule was presented, raising a problem of the wavenumber shift of
11 the spectra provided by Smith et al. The integrated intensities in Smith et al. agree well with the
12 ones from Le Bris and Graham, while the spectra from Di Lonardo and Masciarelli
13 (supplemental folder) have somewhat lower intensities. It should be noted that the PNNL data
14 agree with Le Bris et al. within the experimental accuracy (see Fig.3 in Ref. [35]). We
15 recommend using the recent data from Le Bris et al. for the corresponding experimental
16 conditions. Since the data from Smith et al. span a wider temperature and pressure range than Le
17 Bris et al., these spectra were kept in the database. Further corrections of the wavenumber shift
18 for data from Smith et al. should be made in future versions of the database.

19 **3.3.1.5. HFC-227ea**

20 HFC-227ea (1,1,1,2,3,3,3-heptafluoropropane, $\text{CF}_3\text{CHF}_2\text{CF}_3$) is a saturated HFC with a 100-yr
21 GWP of 3350 [3] and is used in fire suppression. This molecule has been introduced in HITRAN
22 for the first time. Three sources of spectra have been considered for this molecule, namely Gohar
23 et al. [70], Sihra et al. [61], and Sharpe et al. [13]. The data from Sihra et al. were omitted
24 because extended versions of the same data were provided in Gohar et al. These data were
25 measured at 296 K and 700 Torr, broadened by air. Another set of data have been measured by
26 PNNL (Sharpe et al.) and incorporated cross-sections for three temperatures (278.15, 298.15,
27 323.15 K). Although the difference of absorption integrated on a wide spectral range (800-1600
28 cm^{-1}) is noticeable, this difference is within the combined uncertainties of the sources (see **Table**
29 **4**) and is approximately 7.4%. The PNNL spectra have been included in HITRAN*Online*. The
30 spectra from Gohar et al. include two minor features which are either absent (near 450 cm^{-1}), or
31 distorted (near 500 cm^{-1}) in the PNNL spectra; for this reason, the data from Gohar et al. was
32 also made available via HITRAN*Online* interface.

1 *3.3.1.6. HFC-41*

2 HFC-41 (fluoromethane, CH₃F) is a flammable gas at room temperature and pressure. For this
3 compound, the considered spectra are from two sources: PNNL [13] and Sihra et al. [61]. The
4 discrepancy in integrated intensities is around 12%, which is greater than the combined
5 uncertainty of both sources (~7%). This probably reflects calibration problems, or to differences
6 between experimental conditions for the spectra of the same temperature (700 Torr and air
7 broadening in Sihra et al., and 760 Torr and nitrogen broadening for PNNL). Spectra from Sihra
8 et al. have been put in the supplemental folder, while the PNNL data are included in
9 HITRAN*Online*. Both data sets should be considered by the user until the differences are
10 resolved.

11 *3.3.1.7. HFC-32*

12 HFC-32 (difluoromethane, CH₂F₂) is a replacement for CFCs and HCFCs [41] and for higher-
13 GWP HFCs in air-conditioning units. It has a 100-yr GWP of 677 [3]. For HFC-32, a wide range
14 of temperature and pressure sets was provided to previous HITRAN editions by Smith et al. [41].
15 These sets include air- and self-broadened cross-sections for temperatures between 203K and
16 297K, and pressures up to 760 Torr. The cross-sections were measured with 4% accuracy [41].
17 These sets were retained in the HITRAN2016 version with no changes. More recent data were
18 considered from the following sources: Highwood and Shine [90], Gohar et al. [70], Sihra et al.
19 [61] and Sharpe et al. [13]. Three cross-sections from Highwood and Shine, provided in
20 Hodnebrog et al. [1], are originally from two sources: Beukes and Nicolaisen [96] for 450-630
21 cm⁻¹, and Smith et al. [41] for the rest of the wavenumber range. Only data from Beukes and
22 Nicolaisen were added to HITRAN. The spectra from Sihra et al. were omitted because data
23 from Gohar et al. are an extended version [70] of those from Sihra et al. The spectra containing
24 additional features absent in Smith et al., namely PNNL (near 1850-3620 cm⁻¹), Gohar et al.
25 (near 490-570 cm⁻¹), and Highwood and Shine (near 400-600 cm⁻¹), have been included in
26 HITRAN*Online*. At 253K, the data from Gohar et al. agree well with Smith et al. (5% for
27 intensity integrated in 1000-1250 cm⁻¹ range). Near 280K, only data from PNNL and Smith et al.
28 are present, which also have a good agreement (5% for the strong band in 1000-1200 cm⁻¹
29 range). Near 296 K, besides the spectra from Smith et al., the data from Gohar et al. were added
30 since they contained additional features in the lower wavenumber range (near 500 cm⁻¹). The
31 PNNL spectra were also added for the same reason. These data have been included in
32 HITRAN*Online*. It is also worth noting that PNNL HFC-32 spectra cover temperatures up to
33 323K, as is the case for most gases that PNNL measured. The user should use the PNNL datasets
34 to obtain more reliable extrapolations to temperatures higher than 300K.

1 *3.3.1.8. HFC-134a*

2 The atmospheric abundance of 1,1,1,2-tetrafluoroethane (HFC-134a, CH₂FCF₃) has increased
3 rapidly since the late 1990s and continues to rise. HFC-134a has mostly replaced CFC-12 in
4 several refrigeration and air-conditioning applications. With an atmospheric lifetime of 13.4
5 years and a 100-yr GWP of 1300 [3], HFC-134a is currently the most abundant HFC in the
6 atmosphere.

7 The retrieval of HFC-134a profiles from ACE-FTS (Atmospheric Chemistry Experiment –
8 Fourier transform spectrometer) [97] spectra was the first space-based measurements [98] for
9 any HFC. The retrievals employed a window from 1180 to 1200 cm⁻¹ using a set of cross-
10 sections from Nemtchinov and Varanasi [99]. A new set of high-resolution IR absorption cross-
11 sections have been determined by Harrison [25] over the spectral range 750–1600 cm⁻¹ using a
12 high-resolution FTIR spectrometer (Bruker IFS 125HR) and a 26-cm-pathlength cell. Spectra of
13 HFC-134a/dry synthetic air mixtures were recorded at 0.015 cm⁻¹ resolution (calculated as
14 0.9/MOPD²) at a number of temperatures and pressures (22–761 Torr and 191–296 K)
15 appropriate for atmospheric conditions.

16 This new dataset is now included in HITRAN*Online*. The spectra from Nemtchinov and
17 Varanasi are now available in the HITRAN supplemental folder. Besides the data from Harrison
18 et al. [25], datasets from two sources, namely Highwood and Shine [90] (using data from Beukes
19 and Nicolaisen [96] between 50 and 650 cm⁻¹ and Newnham et al. [100] between 650 and 1600
20 cm⁻¹) and PNNL [13] have been included in HITRAN*Online* for the low and high temperature
21 range respectively. These datasets are complementary to those from Harrison et al. in terms of
22 the wavenumber range since they contain additional features below 700 cm⁻¹ and above 1600
23 cm⁻¹.

24 *3.3.1.9. HFC-23*

25 Trifluoromethane (HFC-23, CHF₃) is a long-lived species with 222 years lifetime. This gas is
26 emitted primarily as a by-product in HCFC-22 production. It has a large GWP of 12400 [3], and
27 prior to the early 2000s it was the dominant HFC in the atmosphere, later overtaken by HFC-
28 134a. Nevertheless, the atmospheric abundance of HFC-23 continues to rise, and it is being
29 closely monitored by the international community [101].

30 Previously the only available cross-sections were those derived from spectra recorded by
31 Chung [102]. However, these data were not included in the HITRAN database. This dataset was
32 used to create a set of HFC-23 pseudo-lines, which were utilized in the first remote-sensing

² Maximum Optical Path Difference

1 measurements of this molecule taken by the balloon-borne MkIV interferometer and the ACE-
2 FTS. These cross-sections suffer from a number of problems: they do not cover an appropriate
3 range of atmospheric pressures and do not extend as low as 195 K. Furthermore, some
4 measurements are under-resolved, while others appear to suffer from erroneously recorded
5 absorber amounts, and there are inconsistencies in the wavenumber calibration. These problems
6 all contributed to a large bias in the retrieved HFC-23 profiles.

7 A new HFC-23 cross-section dataset from Harrison [24] over the range 950–1500 cm^{-1} has
8 been included in HITRAN2016. It was determined from spectra of HFC-23/dry synthetic air
9 mixtures recorded using a high-resolution FTIR spectrometer (Bruker IFS 125HR) and a 26-cm-
10 pathlength cell. Spectra were recorded at 0.015 cm^{-1} resolution (calculated as 0.9/MOPD) at a
11 number of temperatures and pressures (23–762 Torr and 188–294 K) appropriate for atmospheric
12 conditions.

13 Data for this molecule provided by Highwood and Shine [90] (from the measurements of
14 McPheat and Duxbury [103]) were recorded at 253K. It is worth noting that there was a fairly
15 large discrepancy for the integrated band intensities ($\sim 20\%$ for the 1000-1600 cm^{-1} range)
16 between Refs. [24] and [90], which lies outside of the combined experimental uncertainty.
17 Considering this, most of the data from Ref. [90] were omitted, except the low-wavenumber
18 band which has been placed in the supplemental folder. Further investigations should be made
19 before elevating these data to the HITRAN*online* web interface.

20 Other considered data contain fewer temperature and pressure sets and are either for room
21 temperature (Gohar et al. [70], Sihra et al. [61]) or for 5-50°C (PNNL [13]). Data from Sihra et
22 al. coincide with those from Gohar et al. except having a slightly narrower wavenumber range.
23 The spectrum from Gohar et al. contains a weak feature near 500 cm^{-1} which is absent in all
24 other sources; it has been placed in the supplemental folder. The PNNL data have been included
25 in HITRAN*online* to improve the higher-temperature extrapolations (near 350K).

26 **3.3.1.10. HFC-43-10mee**

27 HFC-43-10mee is used as a cleaning solvent in electronics. It is sold as a mixture of two
28 diastereomers (erythro and threo) which have the same estimated lifetime of 16.1 years. The
29 global mean tropospheric abundance was 0.211 ± 0.046 ppt in 2012 [104]. Its 100-year GWP is
30 1410 [39].

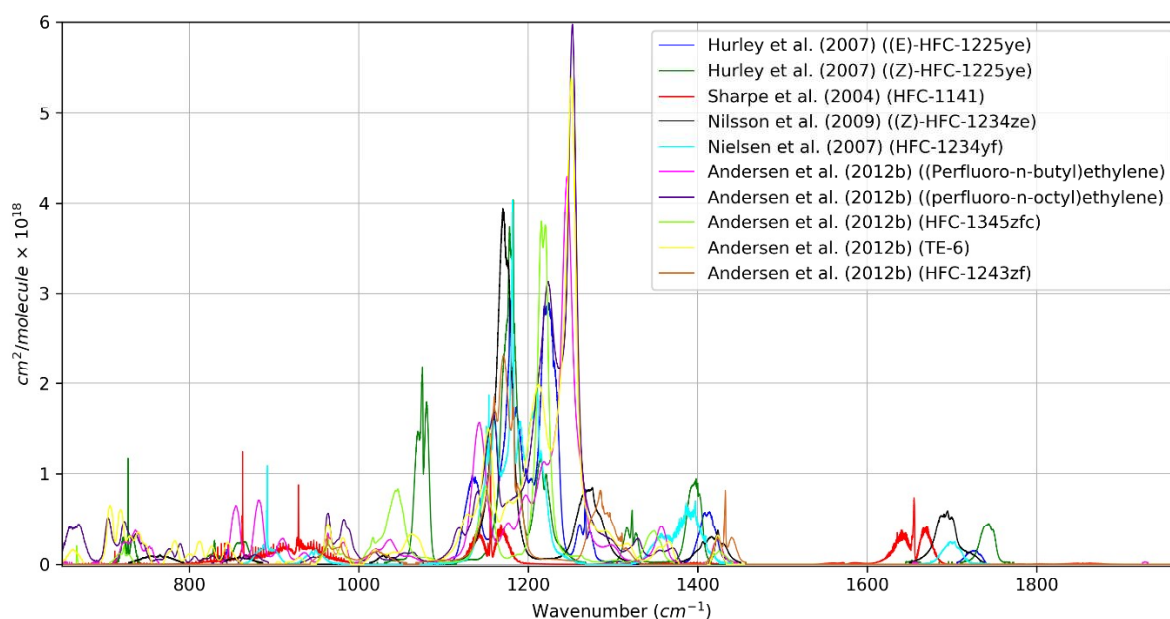
31 Data from the PNNL database [13] have been included in HITRAN16. Another data set has
32 recently been published by Le Bris et al. [39]. The pure vapor spectra of Le Bris et al. overlap
33 well with the PNNL broadened spectra. The two sets of data agree very well at around 300K

1 with a difference of integrated intensities of only 1%. The difference is 6% at 320 K but is
2 within the combined experimental uncertainties. The PNNL and Le Bris et al. data have been
3 included in HITRAN*Online* (Ref. [39] as a post-release update).

4 3.3.2. Unsaturated HFCs

5 Hydrofluoroolefins (HFOs, fluorinated alkenes, unsaturated HFCs) are of industrial and
6 scientific interest as next-generation replacements for saturated HFCs in the applications
7 mentioned above [73,91,105][106]. The unsaturated HFCs (HFOs) contain at least one carbon-
8 carbon double bond, which makes them more reactive towards OH radicals and O₃ than their
9 saturated analogues [105] leading to short atmospheric lifetimes and low GWPs. The
10 atmospheric fate of hydrofluoroalkenes is mainly reaction with OH radicals with reaction with
11 ozone playing a minor role; chlorine-initiated degradation may also contribute in coastal regions
12 [92].

13 Spectra for around twelve fluoroolefins have been added to HITRAN, including data from
14 Refs. [51,52,58,73] which are either air- or nitrogen broadened. These data were provided by
15 Hodnebrog et al. [1]. The HITRAN2016 edition also includes three different temperature spectra
16 from PNNL [13] for vinyl fluoride (HFC-1141) and more recent helium-broadened data from
17 Refs. [92] and [91]. **Figure 8** shows the overview for those spectra. These molecules are (E)-
18 HFC-1225ye, (Perfluoro-n-butyl)ethylene, (perfluoro-n-octyl)ethylene, (Z)-HFC-1225ye, (Z)-
19 HFC-1234ze, HFC-1132a, HFC-1141, HFC-1234yf, HFC-1243zf, HFC-1345zfc, TE-6, (E)-
20 HFC-1234ze (see the **Table 8**). All these species have short lifetimes (two weeks or less) and
21 thus extremely small GWPs. Lifetimes and GWPs can be found in Ref. [3] for all the molecules
22 described in this subsection.



23

1 **Figure 8.** Spectra for unsaturated HFCs included in HITRAN2016. All spectra were recorded at
 2 296K in 700 Torr total pressure except PNNL which were recorded at 760 Torr pressure and
 3 three temperatures (see **Table 9**).

4 A summary of spectra (for saturated and unsaturated HFCs) available through the
 5 HITRANonline web interface is given in **Table 8** for the data from Hodnebrog et al. [1] (and the
 6 references within), in **Table 9** for the PNNL data, and in **Table 10** for the spectra recently
 7 published by Harrison, Le Bris, and coauthors.

8 **Table 8.** Spectra from the review on halocarbons and related compounds (Hodnebrog et al. [1])
 9 for hydrofluorocarbons.

Molecule	CAS	Range, cm ⁻¹	T, K	Res, cm ⁻¹
<i>Andersen et al. (2012)</i> [58]				
(Perfluoro-n-butyl)ethylene	19430-93-4	650-1999	298	0.25
(perfluoro-n-octyl)ethylene	21652-58-4	650-1999	298	0.25
HFC-1243zf	677-21-4	650-1999	298	0.25
HFC-1345zfc	374-27-6	650-1999	298	0.25
TE-6	25291-17-2	650-1999	298	0.25
<i>Gohar et al. (2004)</i> [70]				
HFC-227ea	431-89-0	400-1550	296	0.5
HFC-32	75-10-5	400-1550	296	0.5
<i>Beukes and Nicolaisen (2000)</i> [96]				
HFC-134a †	811-97-2	75-590	253	
<i>Newnham et al. (1996)</i> [100]				
HFC-134a †	811-97-2	600-1540	253	
<i>Vander Auwera (2000)</i> [95]				
HFC-152a †	75-37-6	830-1500	253	0.03
<i>McPheat and Duxbury (2000)</i> [103]				
HFC-32 †	75-10-5	450-630	253	0.03
<i>Hurley et al. (2007)</i> [73]				
(E)-HFC-1225ye	5595-10-8	650-2000	296	0.25
(Z)-HFC-1225ye	5528-43-8	650-2000	296	0.25
<i>Inoue et al. (2008)</i> [72]				
HFC-365mfc	406-58-6	600-2000	296	0.25
<i>Nielsen et al. (2007)</i> [51]				
HFC-1234yf	754-12-1	700-2000	296	0.25
(Z)-HFC-1234ze	29118-25-0	600-2000	296	0.25
<i>Sihra et al. (2001)</i> [61]				
HFC-143	430-66-0	350-1550	296	0.5
HFC-161	353-36-6	350-1600	296	0.5
HFC-227ca	2252-84-8	400-1550	296	0.5
HFC-236cb	677-56-5	300-1550	296	0.5
HFC-236fa	690-39-1	350-1500	296	0.5
HFC-245cb	1814-88-6	400-1550	296	0.5
HFC-245fa	460-73-1	640-1500	296	0.5
HFC-272ca	420-45-1	300-1500	296	0.5
<i>Wallington et al. (1994)</i> [63]				
HFC-152	624-72-6	700-3100	296	0.25
<i>Young et al. (2009)</i> [48]				
HFC-329p	375-17-7	500-4000	296	0.25

10 † Self-broadened data for the molecules from Highwood and Shine [40]. Other data in this table
 11 were measured at 700 Torr total pressure broadened by air.

12 **Table 9.** PNNL-DOE data for hydrofluorocarbons.

Molecule	CAS	Range, cm ⁻¹
HFC-1141	75-02-5	500-6500
HFC-125	354-33-6	510-6500

HFC-134	359-35-3	570-6500
HFC-143a	420-46-2	500-6500
HFC-152a	75-37-6	525-6500
HFC-227ea	431-89-0	500-6500
HFC-32	75-10-5	510-6500
HFC-41	593-53-3	600-6500
HFC-43-10mee	138495-42-8	500-6500

1 Spectra from Sharpe et al. [13] broadened by N₂. Spectra were recorded at 278.15 K, 298.15 K,
2 and 323.15 K, 760 Torr total pressure, and 0.112 cm⁻¹ resolution.

3 **Table 10.** More recent data for hydrofluorocarbons.

Molecule	CAS	Range, cm ⁻¹	T, K	p, Torr	Source
HFC-23	75-46-7	950-1500	187.6-294.4	23-762.4	Harrison (2013) [24]
HFC-134a	811-97-2	750-1600	190.9-295.8	22.4-760.8	Harrison (2015) [25]
HFC-143a	420-46-2	550-3500	263-283		Le Bris and Graham (2015) [35]†
HFC-43-10mee	138495-42-8	550-3500	305-340		Le Bris et al. (2018) [39]

4 † Self-broadened data from Le Bris and Graham [35], measured with 0.03 cm⁻¹ resolution.

5 Spectra published by Harrison [24,25] were recorded with air as a bath gas with 0.015 cm⁻¹
6 resolution. Full lists of cross-sections for these molecules can be found in the corresponding
7 references.

8 3.4. Bromocarbons, Hydrobromocarbons, and Halons

9 This class of gases has been widely used in fire extinguishing equipment and, in the case of
10 CH₃Br, fumigants in agriculture. Lifetimes range from months for the hydrobromocarbons,
11 which react readily with OH, to several decades for fully-halogenated species such as CBrF₃
12 (Halon 1301). The presence of bromine (and in some cases chlorine too) makes them potentially
13 potent ODSs. Bromine atoms are regarded as 60 times more powerful than chlorine atoms in
14 calculations of the EESC³ (ODGI⁴ [107]). Halon production and consumption has been banned
15 under the Montreal Protocol. They also contribute to the greenhouse effect because of absorption
16 in the thermal infrared. The most abundant bromine containing gas in the atmosphere is CH₃Br
17 whose concentration peaked at nearly 10 ppt in the late-1990s but has since fallen to
18 approximately 7 ppt, believed to be near its natural background. H-1211 and H-1301 have
19 abundances of around 2.5 ppt, appear to have peaked, and have started falling. The radiative
20 forcing associated with these species is less than 0.1% that of CO₂ [3] but their contribution to
21 ozone depletion is substantial; when measured by the EESC it is about 25% of the total mid-
22 latitude EESC [107].

23 Bromocarbons, hydrobromocarbons, and halons are a category which is relatively new to
24 HITRAN. Despite the presence of some bromine-containing species in the line-by-line section of
25 the HITRAN database, the cross-section part of the database did not contain data on such species

³ Equivalent Effective Stratospheric Chlorine

⁴ Ozone Depleting Gas Index

1 until HITRAN2012 [6]. The atmospherically-important molecule – bromine nitrate – was first
2 introduced in the HITRAN2012 issue. In HITRAN2016, we completely updated the set of
3 absorption cross-sections for this molecule with the recent high-quality spectra measured by
4 Wagner and Birk [29]. In addition, we introduced into the cross-section part ten more molecules
5 measured by PNNL [13] and Ford Motor Company (hereafter referred to as Ford) [61,62]. For
6 methyl bromide, the data have been added to the supplemental folder for the reason of the
7 presence of the high-quality spectroscopic transitions in the line-by-line part [2].

8 In **Table 11**, the list of bromocarbons, hydrobromocarbons, and halons available in
9 HITRAN*online* is given. Comparisons between different sources were made for all molecules
10 except six, for which the data were only available from one source: halon-2311 [62], halon-1201
11 [61], halon-1202, bromochloromethane, tribromomethane, and 1,2-dibromoethane [13]. We
12 briefly outline the comparisons made for the rest of the molecules in this section.

13 **3.4.1. Halon-1211**

14 The lifetime of halon-1211 (bromochlorodifluoromethane, CBrClF₂) is 16 years, and the 100-
15 year GWP is 1750 [3]. The PNNL [13] spectra were measured at 278.15 K, 298.15 K, and
16 323.15 K and 760 Torr pressure broadened by nitrogen. Additional air-broadened data are
17 available in Sihra et al. [61] at 296 K and 760 Torr. The agreement is excellent between the
18 corresponding datasets at room conditions both for line shape and integrated intensity. CBrClF₂
19 has four distinct bands: a weak feature near 650 cm⁻¹ due to the C-Br stretch, one isolated feature
20 at 800-950 cm⁻¹ due to the C-Cl stretch, and two partially overlapped features at 1000-1250 cm⁻¹
21 due to the C-F functional group. The integrated intensity difference in the 800-950 cm⁻¹ and
22 1000-1250 cm⁻¹ ranges are 3.3% and 6.2% respectively. The room temperature set from Sihra et
23 al. contains no additional features and is thus omitted. The PNNL data have been included in
24 HITRAN*online*.

25 **3.4.2. Halon-1301**

26 Halon-1301 (bromotrifluoromethane, CBrF₃) spectra from PNNL [13] are available at three
27 temperatures along with the single temperature and pressure set from Sihra et al. [61]. The
28 spectrum has three isolated features: a C-Br stretch feature near 700 cm⁻¹ and other features at
29 1050-1150 cm⁻¹ and 1150-1300 cm⁻¹. The agreement between spectra from PNNL and Sihra et
30 al. is good and is within the combined experimental uncertainties (8% for 1050-1150 cm⁻¹ and
31 7% for 1150-1300 cm⁻¹). As for halon-1211, the room temperature set from Sihra et al. contains
32 no additional features and thus is omitted, while the PNNL data have been included in
33 HITRAN*online*.

3.4.3. Halon-2402

For halon-2402 (1,2-dibromotetrafluoroethane, $\text{CBrF}_2\text{CBrF}_2$), excellent agreement is found between PNNL data (278.15, 298.15, and 323.15 K) and Sihra et al. [61]: integrated intensity differences are 1.3% for the 700-950 cm^{-1} range and 1.8% for the 950-1300 cm^{-1} range. The band shapes for all bands are in a good agreement as well. As for Halon-1301 and Halon-1211, the PNNL data were included in HITRAN*Online*. However, since the spectrum from Sihra et al. contains fine-grained features which are probably due to a higher resolution, it has been placed in the supplementary folder rather than omitted completely.

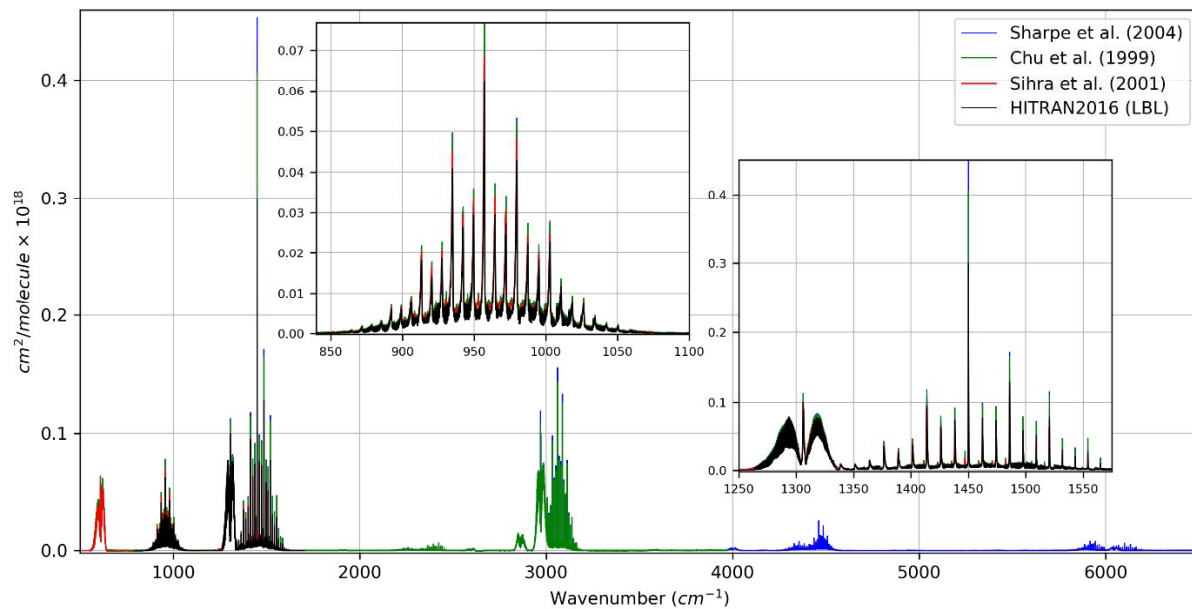
3.4.4. Methylene bromide

Available spectra for methylene bromide (dibromomethane, CH_2Br_2) are from PNNL [13] and Sihra et al [61]. For two major isolated features at 650 cm^{-1} and 1200 cm^{-1} , the integrated intensities differ by 4.8% and 3.8% respectively. PNNL spectra have a much wider range and a number of weaker, but noticeable features above 1500 cm^{-1} . Data from Sihra et al. do not reveal additional spectral features, and some details of the spectra are missing due to lower resolution. The data from PNNL have been included in HITRAN*Online*, while the data from Sihra et al. have been omitted.

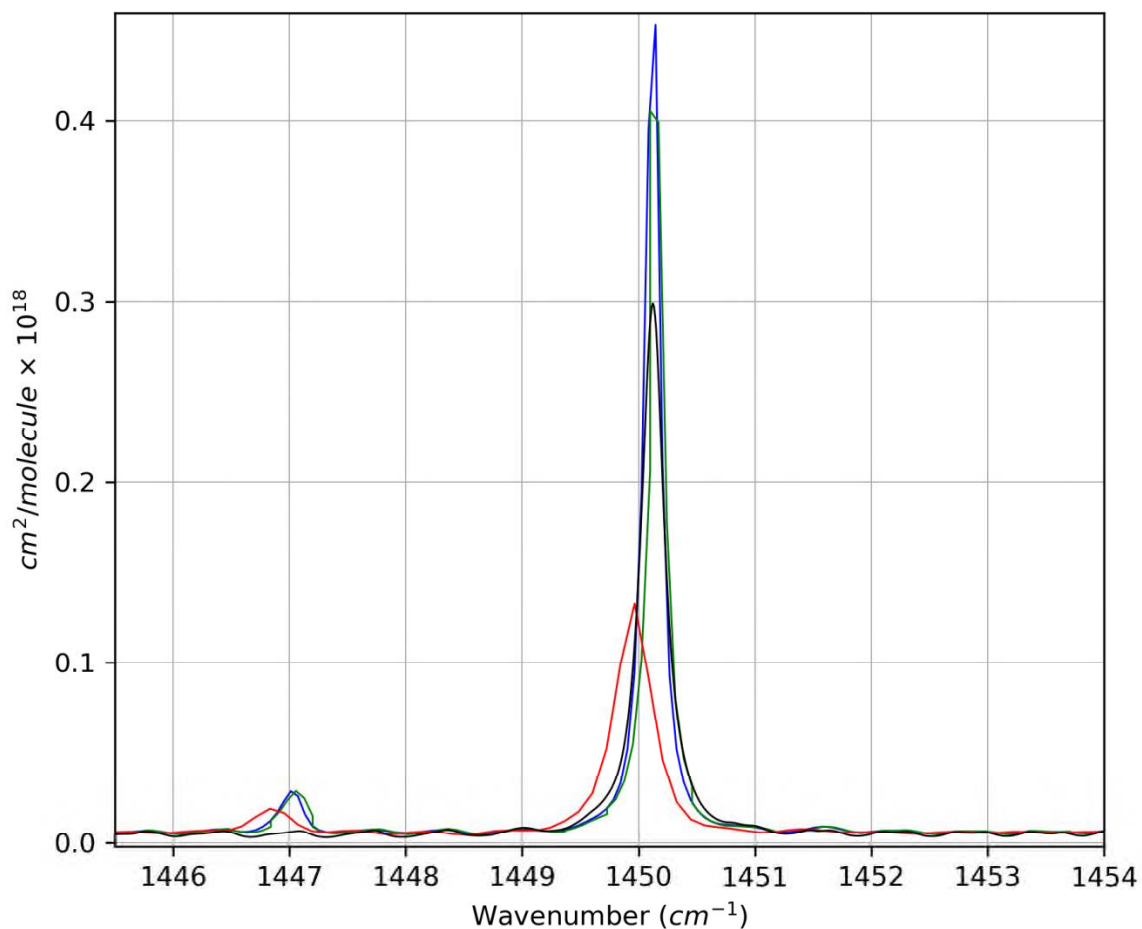
3.4.5. Methyl bromide

For methyl bromide (bromomethane, CH_3Br), the HITRAN2016 line-by-line section includes 36911 transitions for the two most abundant isotopologues (^{79}Br and ^{81}Br) in the 794-1706 cm^{-1} range. In our earlier work (Kochanov et al. [86]) we had commented on 10-20% discrepancies (in terms of integrated band intensities) between cross-sections generated from line-by-line and the PNNL database. It is interesting to check if the same holds when compared with experimental cross-sections from other sources. The absorption cross-sections have been published for this molecule by Sharpe et al. (PNNL) [13], Chu et al. (NIST) [17], and Sihra et al. [61]. The data with the widest spectral range are supplied by PNNL and cover three temperature ranges (278.15, 298.15, 323.15K) and have features up to 6500 cm^{-1} . Three features in the atmospheric window have been measured in PNNL, NIST, and Ford (Ref. [61]). Integrated intensities for these features reported in these three sources agree well within 5%. The HAPI-generated absorption coefficient for the HITRAN line-by-line data disagrees with all the three sources by 12-17% over the 850-1100 cm^{-1} range and by 9-14% over the 1240-1620 cm^{-1} range. The additional absorption in Refs. [13,17,61] reflect possible impurities in the sample, missing hot bands, line-mixing effect in the Q-branches, or differences in the isotopic abundances in the samples. For this reason, we put the PNNL cross-sections in the supplemental folder. **Figure 9** shows the overview for the spectra comparison for this molecule. In this figure, all sources cover the atmospheric window region, while PNNL and NIST data contain additional features above

- 1 2000 cm^{-1} . The spectrum generated using HAPI from the line-by-line data is highlighted in cyan.
- 2 The data from Sihra et al. were omitted because the wavenumber scale is shifted in this spectrum
- 3 due to possible calibration problems (**Figure 10**). The intensity discrepancies between the line-
- 4 by-line data and raw cross-sections will hopefully be resolved in future HITRAN issues.



5
6 **Figure 9.** Absorption spectra for methyl bromide. The spectral features at 1000 cm^{-1} and 1500
7 cm^{-1} are zoomed in in the insets.



8

1 **Figure 10.** Absorption spectra comparison for methyl bromide. Data from Sihra et al. [61] in red
2 vs. PNNL [13] (in blue), NIST [17] (in green) and HAPI-generated spectra from the line-by-line
3 HITRAN2016 line-by-line data [2] (in cyan). The wavenumber shift in spectra from Sihra et al.
4 is approximately 0.1 cm^{-1} .

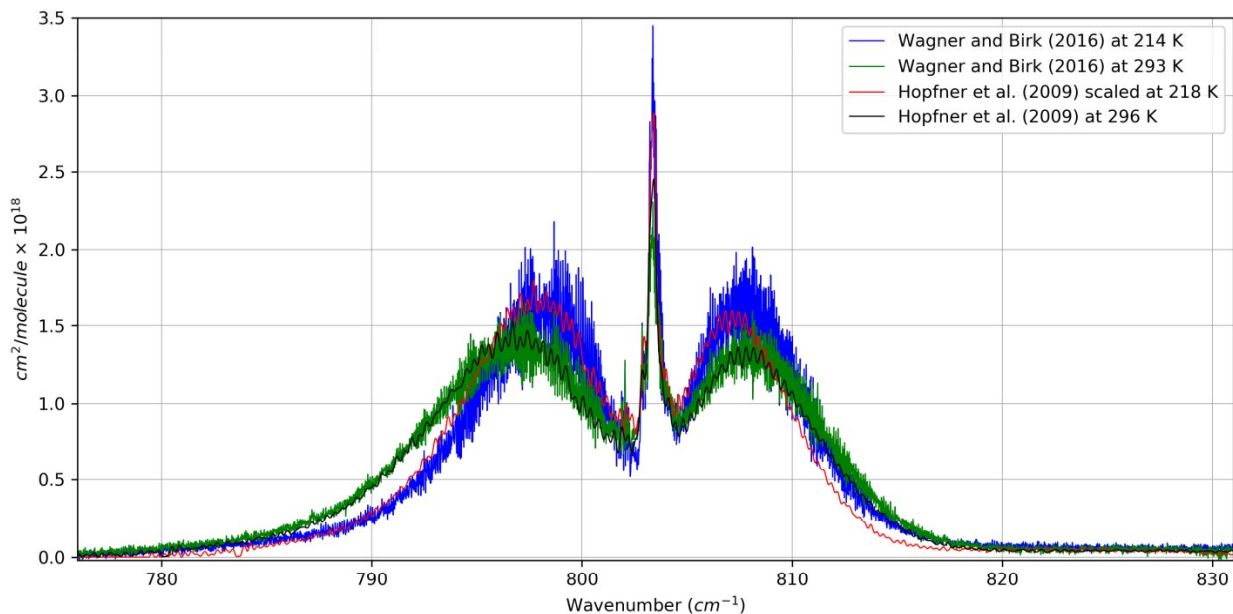
6 3.4.6. Bromine nitrate

7 Despite its low abundance, stratospheric bromine has a considerable impact on stratospheric
8 ozone [108,109]. The abundance of the reservoir species bromine nitrate, BrONO_2 , is required to
9 determine the inorganic stratospheric bromine budget, gain insight into bromine partitioning in
10 the stratosphere, and validate photochemical models. BrONO_2 has been detected in the
11 stratosphere by MIPAS/ENVISAT [110] and the MIPAS-B2 balloon [111] instrument using the
12 strong infrared ν_3 band near 800 cm^{-1} . Retrieval accuracy was limited to 20% by the insufficient
13 absorption cross-section data [112] because the only absorption cross-section measurements
14 available for pure BrONO_2 were at ambient temperature.

15 New temperature and air pressure dependent absorption cross-sections of the ν_3 ($765\text{-}850 \text{ cm}^{-1}$)
16 and ν_2 ($1230\text{-}1320 \text{ cm}^{-1}$) bands have been measured by Wagner and Birk [29] at a resolution
17 of $0.5 \text{ MOPD} = 0.014 \text{ cm}^{-1}$. The measurements consisted of more than 30 transmittance spectra
18 covering the temperature range 204 to 293 K and air pressure range 0 to 180 mbar. The spectra
19 were baseline corrected, contaminant spectra were removed, and temperature and pressure
20 dependent absorption cross-sections were fitted as a two-dimensional polynomial in $\ln(\text{pressure})$
21 and temperature. The overall scalar uncertainties for the ν_3 and ν_2 absorption cross-sections are
22 about 2% and 4%, respectively. The polynomial coefficients together with software to convert to
23 absorption cross-sections at desired temperatures and pressures are available in HITRAN2016 in
24 the supplementary folder. Additionally, 33 absorption spectra for each band, ν_3 and ν_2 ,
25 calculated for the approximate pressures and temperatures of the laboratory measurements, have
26 been included in HITRAN*Online*. Details of measurements, analysis, statistical and systematic
27 uncertainties are given in Ref. [29].

28 HITRAN2012 had zero-pressure spectra with lower resolution for this molecule only in the
29 region of the ν_3 band. The 296K cross-sections were previously provided by Höpfner et al. [110],
30 and were scaled to obtain an estimation of the lower-temperature spectra at 218K. Here we give
31 a comparison of these spectra with the new data from Wagner and Birk (**Figure 11**). The
32 differences in integrated intensities are 5% both for 218K and 293 K, and there are systematic
33 differences in the band shapes of these spectra. The data from Ref. [110] have been removed
34 from the web interface and placed to the supplemental folder. Having nearly the same
35 temperature and pressure conditions, the data from Wagner and Birk have sharper line features

1 originating from the higher spectral resolution. The spectral set for bromine nitrate from
2 HITRAN2012 previously available through HITRAN*Online* has thus been replaced by the data
3 from Wagner and Birk.



4
5 **Figure 11.** Self-broadened ν_3 band of bromine nitrate measured by Wagner and Birk [29] and
6 Höpfner et al. [110] for 214-218K and 293-296K.

7

1 **Table 11.** Summary of the spectra for bromocarbons, hydrobromocarbons, and halons.

Molecule	CAS	Range, cm^{-1}
<i>Sharpe et al. (2004) [13], measured at 5, 25, and 50C, 760Torr, N_2-broadened, 0.112 cm^{-1} resolution</i>		
Bromochloromethane	74-97-5	550-7100
Tribromomethane	75-25-2	600-6500
1,2-Dibromoethane	106-93-4	570-6500
Halon-1202	75-61-6	575-6500
Halon-1211	353-59-3	600-6500
Halon-1301	75-63-8	510-6500
Halon-2402	124-73-2	550-6500
Methylene bromide	74-95-3	540-7100
<i>Sihra et al. [61], measured at 296K, 700Torr, air-broadened, 0.5 cm^{-1} resolution</i>		
Halon-1201	1511-62-2	500-1550
<i>Andersen et al. [62], measured at 298K, 700Torr, air-broadened, 0.25 cm^{-1} resolution</i>		
Halon-2311	151-67-7	600-1499
<i>Wagner and Birk [29]†, measured with 0.014 cm^{-1} resolution</i>		
Bromine Nitrate	40423-14-1	765-1320

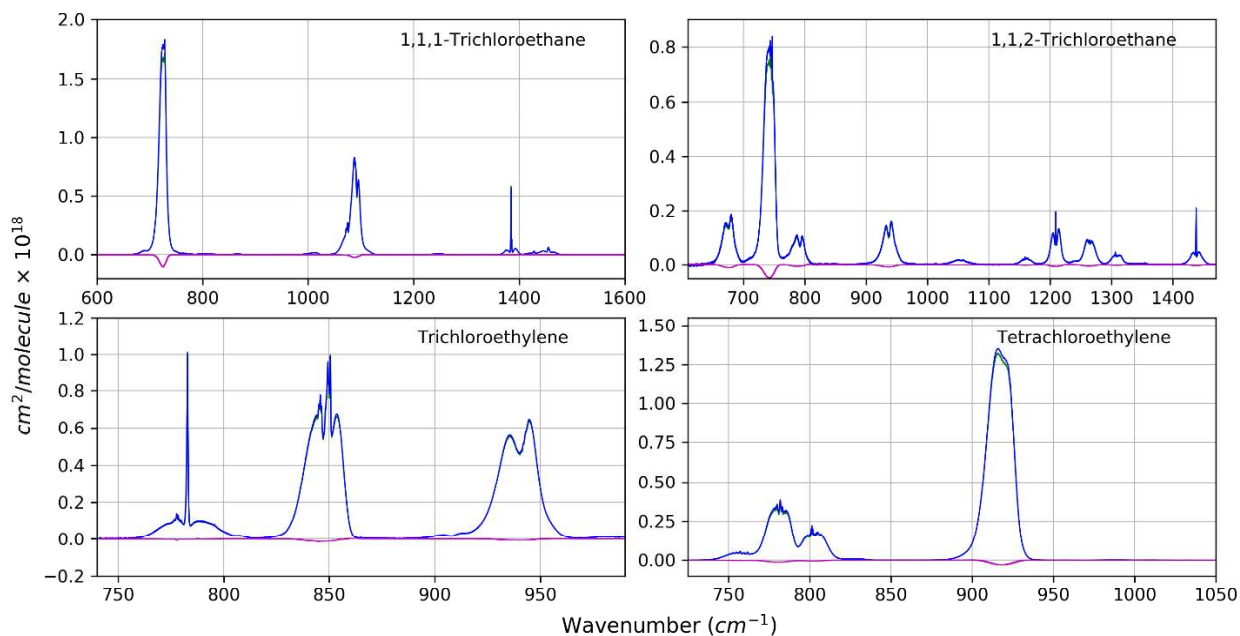
2 † Spectra from Ref. [49] were measured at 204-293 K temperatures and 0-141 Torr pressure
 3 ranges. The full list of cross-sections can be found in the corresponding reference.

4 3.5. Chlorocarbons and Hydrochlorocarbons

5 The chlorocarbons and hydrochlorocarbons category is relatively new to HITRAN. In pre-
 6 2016 HITRAN issues, among molecules from this category only carbon tetrachloride (CCl_4) was
 7 present in the cross-section part, and methyl chloride (CH_3Cl) in the line-by-line part. In the
 8 HITRAN2016 [2] issue, we have included a number of chloroalkanes, chloroalkenes, and
 9 chlorine-containing aromatic hydrocarbons for the first time. **Table 12** and **Table 13** represent a
 10 summary of the molecules from this category. Among them one can outline different isomers of
 11 chloromethane, chloroethene, chlorobenzene, chloropropane, and vinyl chloride. A large number
 12 of the spectra for these molecules has been provided by the PNNL database [13]. Also, for 1,2-
 13 dichloroethane and chloroform, a high-resolution spectrum was provided in the Hodnebrog
 14 review [1] by Highwood and Shine [90] (sourced from the laboratory measurements of Refs.
 15 [95] and [103] respectively). Finally, recently published spectra from Harrison et al. [28] and
 16 Wallington et al. [113] have also been adapted.

17 Most of the gases in this class, the hydrochlorocarbons, have lifetimes ranging from months to
 18 a few years, except CCl_4 , whose lifetime is 26 years. Because of the presence of chlorine, these
 19 ODSs are controlled under the Montreal Protocol, and concentration of methyl chloroform
 20 (CH_3CCl_3) has undergone a dramatic decrease from an early 1990s peak in excess of 130 ppt to
 21 current values of a few ppt. CCl_4 is currently the most abundant of this class of molecules
 22 (around 80 ppt). Despite its concentration falling from an early 1990s peak in excess of 100 ppt
 23 [94], CCl_4 is an important contributor to radiative forcing (0.015 W m^{-2} in 2011 or about 1% of
 24 that due to CO_2) [3] and contributes about 25% to the total mid-latitude EESC [107]. More
 25 detailed discussion of CCl_4 is given below.

1 **Figure 12** displays spectra for 1,1,1-trichloroethane, 1,1,2-trichloroethane, trichloroethylene,
2 and tetrachloroethylene from PNNL and NIST [17] which show a good agreement. As in
3 previous sections, we use the data from NIST to cross-validate the spectral datasets. In this
4 section, we give a more detailed outline of comparisons for several molecules of the major
5 atmospheric and remote sensing importance in this category.



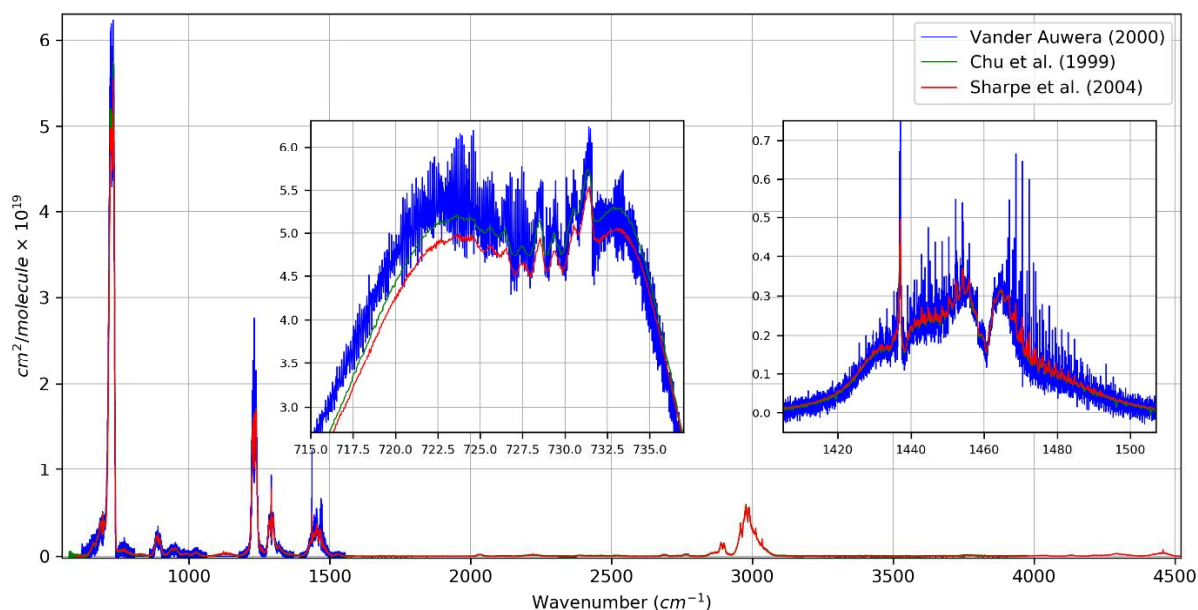
7
8 **Figure 12.** Absorption from NIST [17] in blue and PNNL [13] in green. The residual (magenta)
9 between PNNL and NIST is given in the same scale.

10

11 3.5.1 1,2-Dichloroethane

12 1,2-Dichloroethane (CH₂ClCH₂Cl) is a halocarbon species which is often used in chemical
13 synthesis. For this molecule, spectra from PNNL [13], NIST [17], and Vander Auwera [95] have
14 been compared. It should be noted that only PNNL contains multiple-temperature spectra for this
15 molecule. Since one of the temperature and pressure sets are nearly the same for all sources, we
16 have presented the comparison of the band shapes in **Figure 13**. The resolution and broadener
17 for datasets from PNNL and NIST are the same (see **Table 4**), whereas the resolution for the
18 data from Vander Auwera [95] is 0.03 cm⁻¹. The spectra from Ref. [95] have more resolved band
19 structure than the rest of the sources, although some negative absorption values exist at the edges
20 of the spectral features. Spectra from PNNL have additional noticeable features in the mid- and
21 near-IR spectral range which is covered neither by NIST, nor by the data from Vander Auwera
22 (2000). Although spectra from PNNL have uniformly lower intensity than NIST, the agreement
23 between all three sources is very good and lies within the combined experimental uncertainties
24 for all spectral features (0.2-4.9%) except the weak feature in the atmospheric window in the

1 860-1050 cm^{-1} spectral range. For this latter band, the disagreement between Vander Auwera
2 (2000) data and both PNNL and NIST is around 19% and can possibly be attributed to the
3 spectral calibration. It should also be noted that the data from Ref. [90] for this molecule does
4 not contain the weak features in the zero-filled areas between the strongest bands in the
5 atmospheric window (which is, however, negligible for the GWP estimation purposes). We have
6 included the PNNL data for all three temperatures in HITRAN*Online*. The 293 K set from
7 Highwood and Shine/Vander Auwera also was included in HITRAN*Online* for the reason of
8 having more resolved spectral structure which can possibly be useful for remote-sensing
9 purposes.



10

11 **Figure 13.** Absorption cross-section comparison for 1,2-dichloroethane.

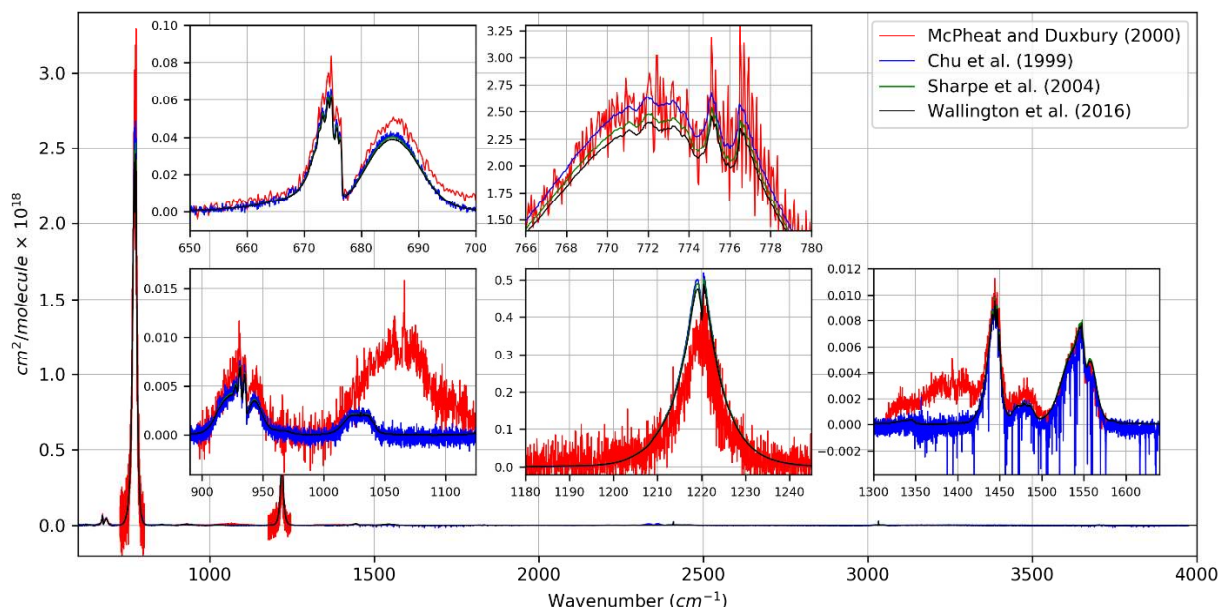
12

13 3.5.2. Chloroform

14 Chloroform (trichloromethane, CHCl_3) is a compound widely used in chemical synthesis, as a
15 reagent, solvent, and anesthetic. In addition to the spectra from Highwood and Shine [90]
16 (McPheat and Duxbury [103]), NIST [17] and PNNL [13], the data from Wallington et al. [113]
17 have been considered in the evaluation for this molecule. In Ref. [113] the scaled PNNL cross-
18 sections at 296K have been supplied for four (hydro) chlorocarbons: chloroform, carbon
19 tetrachloride, methyl chloride, and methylene chloride. Spectra from PNNL are given at three
20 temperatures, whereas the data from MCPheat and Duxbury [103] were supplied only at 295 K,
21 and were split by wavenumber range into five files (for 640-710 cm^{-1} , 725-805 cm^{-1} , 880-1150
22 cm^{-1} , 1175-1250 cm^{-1} , and 1300-1600 cm^{-1} wavenumber ranges). The data from MCPheat and
23 Duxbury has higher spectral resolution than the rest of the data (0.01 cm^{-1}). The comparison
24 shows that for the integrated intensity of the strongest band in the 730-800 cm^{-1} spectral region

1 the agreement between NIST, PNNL, Wallington et al., and McPheat and Duxbury is within 8%.
2 For other regions (**Figure 14**), the agreement between all sources except McPheat and Duxbury
3 is better and lies within 5-6%, while the latter disagrees with the rest by 25%, which can be
4 attributed to calibration problems. Users should keep in mind this discrepancy when using the
5 data for this molecule. Also, the visual comparison of the band shapes shows that the McPheat
6 and Duxbury spectra contain additional absorption, which is missing both in PNNL and NIST
7 spectra (see the insets in **Figure 14**). This additional absorption can be due to presence of
8 isomers or other contaminants. The data from PNNL and McPheat and Duxbury (only the strong
9 feature between 725-805 cm^{-1}) have been included in HITRAN*Online*. Since the data from
10 McPheat and Duxbury is of higher resolution (0.01 cm^{-1} vs 0.1 cm^{-1} in PNNL, NIST, and
11 Wallington et al.), it can be useful in remote sensing and spectral analyses. For this reason, these
12 data (except the 725-805 cm^{-1} feature) have been placed in the supplemental folder.

13



14

15 **Figure 14.** Absorption cross-section comparison for chloroform. The insets show the difference
16 in the band shapes at 1000 cm^{-1} and 1500 cm^{-1} . The contaminant lines at 1400-1600 cm^{-1} are due
17 to water vapor absorption.

18

19 3.5.3. Carbon tetrachloride

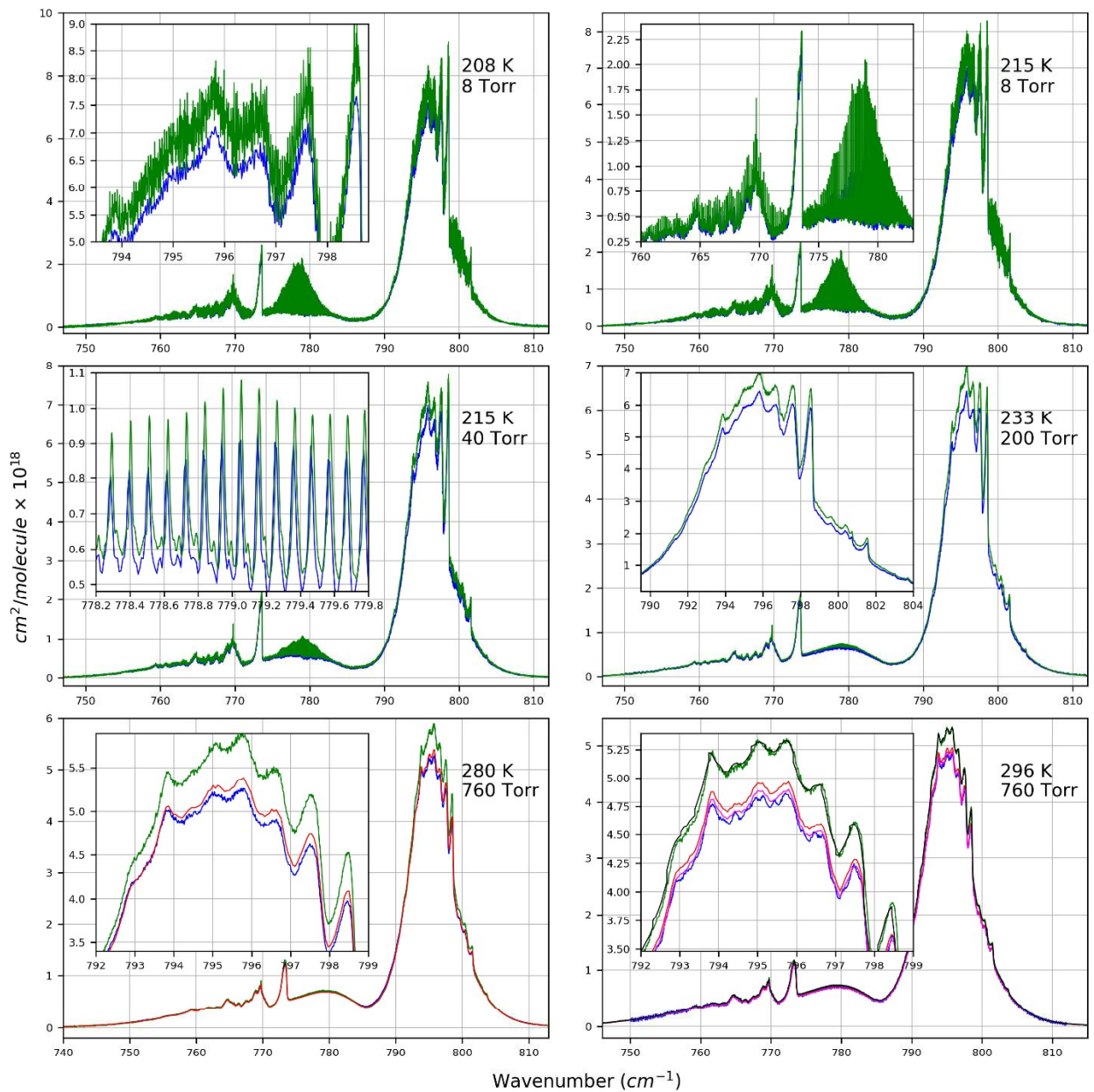
20 As carbon tetrachloride (CCl_4 , CFC-10) does not occur naturally in the Earth system, its
21 presence in the atmosphere, soil and ocean arises solely through its anthropogenic sources. In
22 the early twentieth century, carbon tetrachloride found many uses, e.g. in dry cleaning (excellent
23 solvent properties). However, its largest industrial application was as a feedstock in the
24 production of CFC-12 and CFC-11 (trichlorofluoromethane). The realization that CCl_4 , like
25 CFCs, could destroy stratospheric ozone resulted in its regulation under the Montreal Protocol,

1 leading to a decline in its use. Under the terms of the Protocol, production and consumption of
2 CCl₄ were eliminated for developed countries in 1996 and for developing countries in 2010
3 [114]. There are no limits on the use of CCl₄ as a feedstock, however, so it is still used, for
4 example, in the production of hydrofluorocarbons. Besides being an ozone-depleting substance,
5 with an ozone depletion potential of 0.72 [115], CCl₄ is also a strong greenhouse gas with a
6 lifetime of 26 years and a 100-year GWP of 1730 [3]. Recently there has been renewed interest
7 in carbon tetrachloride, on account of the inconsistency between observations of its abundance
8 and estimated sources and sinks [114,116]. For example, the 2007–2012 emissions estimate
9 based on the UNEP reported production and feedstock usage is near zero. However, the observed
10 inter-hemispheric gradient for CCl₄ (1.5 ± 0.2 ppt for 2000–2012) provides evidence for ongoing
11 emissions. Additionally, the decline of atmospheric concentrations is slower than can be
12 accounted for by our current best estimates of the total CCl₄ lifetime and its uncertainties.

13 Remote-sensing measurements which provide vertical profiles into the stratosphere, e.g. the
14 ACE-FTS [117], are particularly useful in validating stratospheric loss rates in atmospheric
15 models (the atmospheric loss of CCl₄ is essentially all due to photolysis in the stratosphere). The
16 previous absorption cross-section dataset available in HITRAN was provided by Nemtchinov
17 and Varanasi [118]. For HITRAN 2016, this has been replaced by a new dataset from Harrison
18 et al. [28] over the spectral range 700 – 860 cm⁻¹. Spectra were recorded using a high-resolution
19 FTS (Bruker IFS 125HR) for mixtures of CCl₄ with dry synthetic air in a 26-cm-pathlength cell
20 at spectral resolutions between 0.01 and 0.03 cm⁻¹ (calculated as 0.9/MOPD) over a range of
21 temperatures and pressures appropriate for upper troposphere – lower stratosphere conditions
22 (7.5 – 760 Torr and 208 – 296 K).

23 The previous data of Nemtchinov and Varanasi cover 208-297K and 8-760 Torr ranges for
24 temperature and pressure respectively, and have not been changed since HITRAN2000. The
25 detailed comparison of the Varanasi and Harrison et al. spectra for CCl₄ can be found in Ref.
26 [28]. We give only a brief visual comparison for the sets which have the same temperatures and
27 pressures. **Figure 15** displays spectra from Varanasi, Harrison et al. [28], PNNL [13], NIST [17],
28 and Wallington et al. [113]. In **Figure 15**, the panels correspond to the different approximate
29 temperature and pressure conditions, at which the comparisons were made. It is clearly seen (and
30 described furthermore in Ref. [28]), that cross-sections from Harrison et al. have higher (~5.5%)
31 intensities than PNNL, Varanasi, and Wallington et al. spectra. The intensities of the spectra
32 from Harrison et al. were calibrated using the NIST database, which results in a better agreement
33 with ground-based measurements [28]. For this reason, the spectra provided by Harrison et al.,
34 have been included in HITRAN*online*, while the old dataset (Varanasi et al.) have been placed in

- 1 the supplemental folder along with the PNNL set which contains additional 50°C temperature
- 2 spectra.
- 3



4
 5 **Figure 15** . Absorption cross-section comparisons for carbon tetrachloride (CCl_4). Each panel
 6 corresponds to the approximate values of temperature and pressure. Considered spectra: Harrison
 7 et al. [28] (green), Varanasi private communication (blue), PNNL [13] (red), NIST [17] (black),
 8 Wallington et al. [113] (magenta).

9
 10 **3.5.4 Methyl chloride**

11 For methyl chloride (chloromethane, CH_3Cl), the line-by-line part of the HITRAN database
 12 includes the spectral transitions for the two most abundant isotopologues within a broad
 13 wavenumber range ($0.8\text{-}3198\text{ cm}^{-1}$). For the details concerning the line-by-line spectra of CH_3Cl ,

1 the user can refer to the HITRAN2016 [2] and HITRAN2012 [6] papers. The agreement
 2 between the absorption cross-sections generated from HITRAN2016 line-by-line data, and
 3 PNNL spectra, is acceptable both in terms of the line shape and integrated intensities (12% and
 4 6% for the two strongest bands). This is not surprising because for many bands the intensities
 5 were scaled to reproduce the PNNL spectrum. See HITRAN2012 [6] and Nikitin et al. [119] for
 6 details. Since there exist high-quality line-by-line data for this molecule, we omit the absorption
 7 cross-sections from the interactive HITRAN*online* page to encourage the users to take advantage
 8 of the line-by-line data. However, since the PNNL spectra contain additional features in the
 9 higher wavenumber range ($>4000\text{ cm}^{-1}$), these spectra have been placed in the supplemental
 10 folder.

11 **Table 12.** PNNL-DOE data for chlorocarbons and hydrochlorocarbons

Molecule	CAS	Range, cm^{-1}
1,1,1,2-Tetrachloroethane	630-20-6	530-6500
1,1,1-TCE	71-55-6	500-6500
1,1,2,2-Tetrachloroethane	79-34-5	530-6500
1,1,2-Trichloroethane	79-00-5	600-6500
1,1-Dichloroethane	75-34-3	560-6500
1,2-Dichloroethane	107-06-2	600-6500
1,2-Dichloropropane	78-87-5	590-6500
1,2-Dichlorobenzene	95-50-1	600-6500
1,3-Dichlorobenzene	541-73-1	600-6500
1,3-Dichloropropane	142-28-9	570-6500
1,4-Dichlorobenzene †	106-46-7	600-6500
Chlorobenzene	108-90-7	600-6500
Chloroethane	75-00-3	590-6500
Chloroform	67-66-3	580-7200
Hexachloro-1,3-butadiene †	87-68-3	530-6500
Methylene chloride	75-09-2	600-6500
Tetrachloroethylene	127-18-4	600-6500
Trichloroethylene	79-01-6	595-6500
Vinyl chloride	75-01-4	540-6500
cis-1,3-Dichloropropene	10061-01-5	600-6500
trans-1,2-Dichloroethene	156-60-5	540-6500
trans-1,3-Dichloropropene	10061-02-6	600-6500

12 Spectra from Sharpe et al. [13] broadened by N_2 . Spectra were recorded at 278.15 K, 298.15 K,
 13 323.15 K (except for 1,4-dichlorobenzene and hexachloro-1,3-butadiene), 760 Torr total
 14 pressure, and 0.112 cm^{-1} resolution.

15 † Spectra for 1,4-dichlorobenzene and hexachloro-1,3-butadiene were recorded at 25 and 50°C
 16

17 **Table 13.** Measurements from Harrison et al [28] and Hodnebrog et al [1]

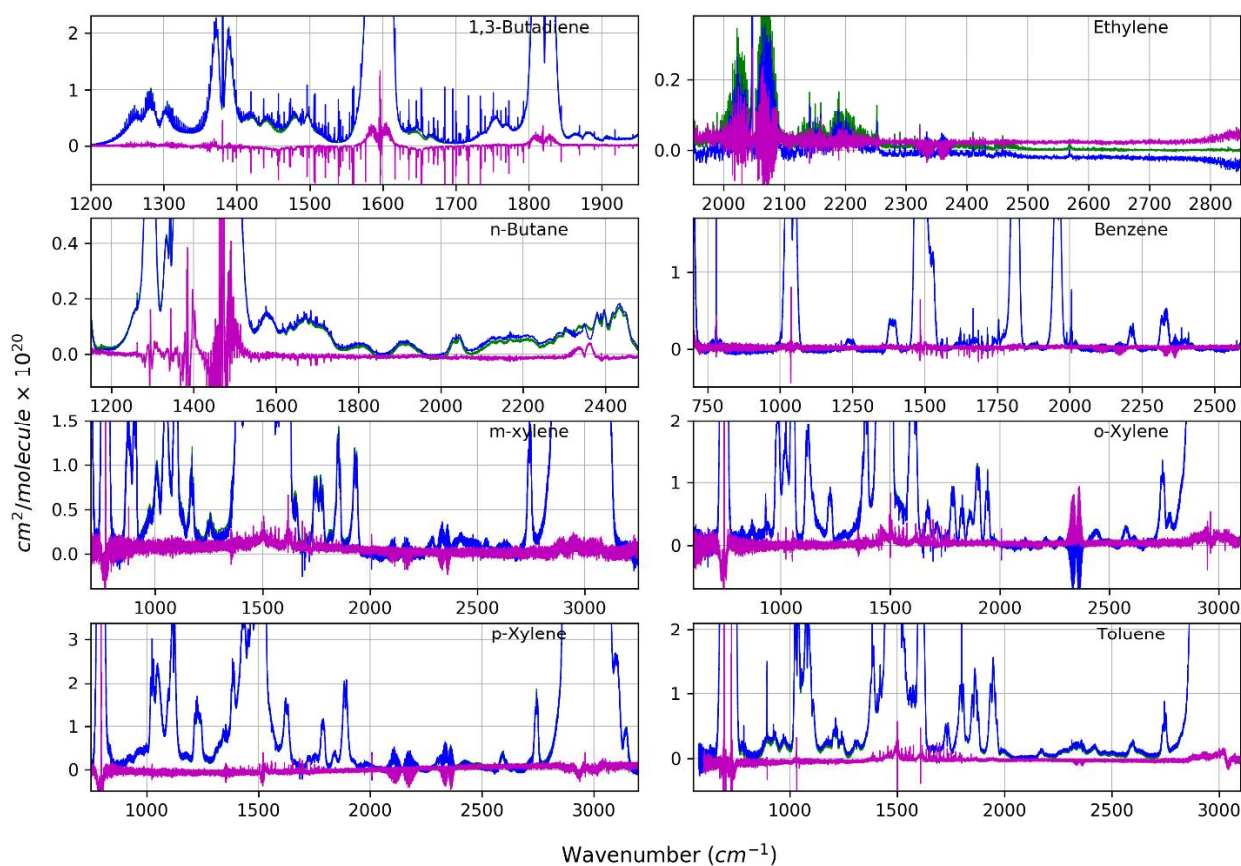
Molecule	CAS	Range (cm^{-1})	T (K)	P (Torr)	Resolution (cm^{-1})	Source
1,2-Dichloroethane	107-06-2	620-1570	293			Vander Auwera (2000) [95]
Chloroform	67-66-3	725-805	295		0.01	McPheat and Duxbury (2000) [103]

1 Spectra from Ref. [95] are self-broadened. Cross-sections from the work by Harrison et al. were
2 recorded with air as a bath gas with 0.01-0.03 cm⁻¹ resolution range. The full list of these cross-
3 sections and a more detailed description of the experimental conditions can be found in Ref.
4 [28].

5 3.6. Hydrocarbons

6 Hydrocarbons have short atmospheric lifetimes and are not well mixed in the atmosphere.
7 Hydrocarbons are important molecules in many fields. In previous HITRAN editions, the
8 hydrocarbons have been predominantly in the line-by-line section, which included the smallest
9 alkanes methane and ethane and selected unsaturated compounds (acetylene, ethylene, and
10 diacetylene). The cross-section part included data for propane and ethane in the 2500-3300 cm⁻¹
11 spectral window. The data for three isomers of xylene (meta-, ortho-, and para-) were present
12 only in the UV range. In the recent HITRAN2016 [2] edition, high-resolution infrared spectra
13 have been added for around thirty alkanes, alkenes, and aromatic hydrocarbons. **Table 14** and
14 **Table 15** present the summary of all these compounds available via HITRAN*online* web
15 interface.

16 For most of the molecules from this category, the data from PNNL [13,22] have been added
17 to the database. For ethane, high-quality frequency-stabilized cavity ring-down (FS-CRDS)
18 spectra have also been included. In the case of acetylene, when the line-by-line data superseded
19 the cross-sections, the data on the cross-section have been omitted. Below we give a brief
20 characterization of the comparisons for a number of molecules. For the spectra for eight
21 compounds originated from PNNL, the comparisons with the NIST [17] infrared database have
22 been made to validate the spectra (**Figure 16**). In comparisons in this figure, all spectra are given
23 at 296K and 760 Torr. NIST spectra used in this comparison are taken with the Boxcar
24 apodization and 0.125 cm⁻¹ resolution. The residual between PNNL and NIST is in magenta. The
25 agreement between the integrated intensities is good and lies within the experimental
26 uncertainties stated in both sources.



1

2 **Figure 16.** Comparison of PNNL [13] and NIST [17] spectra for eight hydrocarbons (see labels
 3 in each panel). NIST spectra are in blue, PNNL are in green, residual (PNNL-NIST) are in
 4 magenta. All spectra are recorded at 296K and 760 Torr. The residual shows traces of minor
 5 contamination in the NIST spectra (possibly due to H₂O, CO₂, and CO).

6 3.6.1. 1,3-butadiene

7 PNNL and NIST spectra for 1,3-butadiene (C₄H₆) agree very well in terms of integrated
 8 intensities. For the region with strongest absorption, 700-1200 cm⁻¹, the agreement is within
 9 2.2%, while in the rest of the range it is 0.11%. However, the NIST spectrum has additional
 10 sharp spectral lines which are not present in the PNNL spectrum. Those lines potentially signify
 11 the presence of contaminants in the gas mixtures in the NIST experiment. In particular, NIST
 12 spectra have a well-resolved additional band in the 2300-2380 cm⁻¹ region, which belongs to
 13 carbon dioxide.

14 3.6.2. Ethylene

15 Ethylene (ethene, C₂H₄) spectra are present in the line-by-line part of the HITRAN database
 16 and were updated in the recent HITRAN2016 edition. The ethylene transitions cover all strong
 17 spectral bands in the region below 3200 cm⁻¹. Since the PNNL spectra have additional low-
 18 intensity spectral features, we have included the cross-sections for three temperatures from
 19 PNNL in the HITRAN2016 edition. To encourage the user to take advantage of the line-by-line
 20 data, the PNNL cross-sections have been placed in the supplemental folder. With the update of

1 the line-by-line data in the HITRAN2016 edition, the agreement with the PNNL cross-sections
2 became significantly better (see the comparison of the update with the HITRAN2012 edition
3 given in Figure 25 in Ref. [2]). For the PNNL and NIST comparison, the agreement of the
4 integrated intensities is also good for the two strongest bands (less than 0.6% difference in the
5 600-1300 cm^{-1} and 2800-3400 cm^{-1} regions). For the weaker bands, integrated intensities agree
6 within 2.2% in the interval 1350-1550 cm^{-1} , and worse in the 1840-1940 cm^{-1} range (4.4%). The
7 latter can be due to the influence of the minor negative baseline drift for NIST spectrum for this
8 molecule in the full spectral range.

9 **3.6.3. *n*-Butane**

10 Similar to 1,3-butadiene, the spectrum for *n*-butane (C_4H_{10}) from NIST contains additional
11 spectral lines which are likely due to contaminants. This molecule absorbs intensively in the full
12 spectral range. We have compared the integrated intensities between NIST and PNNL in two
13 regions: 500-2000 cm^{-1} and 2000-4000 cm^{-1} . The discrepancies in intensities are 0.35% and
14 1.53% respectively. The PNNL spectra for *n*-butane have been included in HITRAN*Online*.

15 **3.6.4. Benzene**

16 For benzene (C_6H_6), the spectral data are available in both PNNL and NIST databases.
17 Spectra from PNNL have already been included in previous HITRAN editions. The
18 HITRAN2016 edition has the data for benzene unchanged. The overall wide-range agreement of
19 the integrated intensities between PNNL and NIST is excellent (0.2%). However, for most of the
20 features, the NIST line intensity is systematically stronger than that of PNNL which is
21 compensated by the minor negative baseline drift in the NIST spectrum. Like some of the
22 previous molecules mentioned in this section, the NIST data contains additional sharp spectral
23 lines possibly due to contaminants.

24 **3.6.5. *m*-Xylene, *o*-Xylene, and *p*-Xylene**

25 For *m/o/p*-isomers of xylene (dimethylbenzene, $\text{C}_6\text{H}_5(\text{CH}_3)_2$), previous HITRAN editions
26 included only spectra in the UV range. In HITRAN2016 we have included IR spectra from
27 PNNL for these molecules in the HITRAN*Online* interface. The spectra for these molecules have
28 also been measured by NIST. From the comparisons of PNNL and NIST spectra, it is clear that
29 the integrated intensities over a wide range (500-3500 cm^{-1}) are in very good agreement (2.5%
30 1.8%, and 0.2% for *m*-, *o*-, and *p*-xylene respectively). Again, NIST spectra have presumably
31 minor artifacts such as additional contaminant bands in the 2000-2500 cm^{-1} range.

1 3.6.6 Toluene

2 For toluene ($C_6H_5CH_3$), the previous HITRAN editions included only the UV spectra. The IR
3 data from PNNL have been included in HITRAN*online*. For toluene, the agreement between
4 PNNL and NIST integrated cross-sections is 1.6%. The NIST spectrum has slight line-shape
5 artifacts at the top of the Q-branches, and slightly higher absorption intensity in the area of the
6 band wings, as well as a minor contaminant band near 2350 cm^{-1} .

7 3.6.7. Ethane

8 The ethane molecule, C_2H_6 , is the second most abundant hydrocarbon in the terrestrial
9 atmosphere after methane. Ethane is concentrated mostly in the lower troposphere due to its
10 short lifetime [49]. It also has been observed in the atmospheres of gas giants and comets (see
11 Ref. [49] and links within).

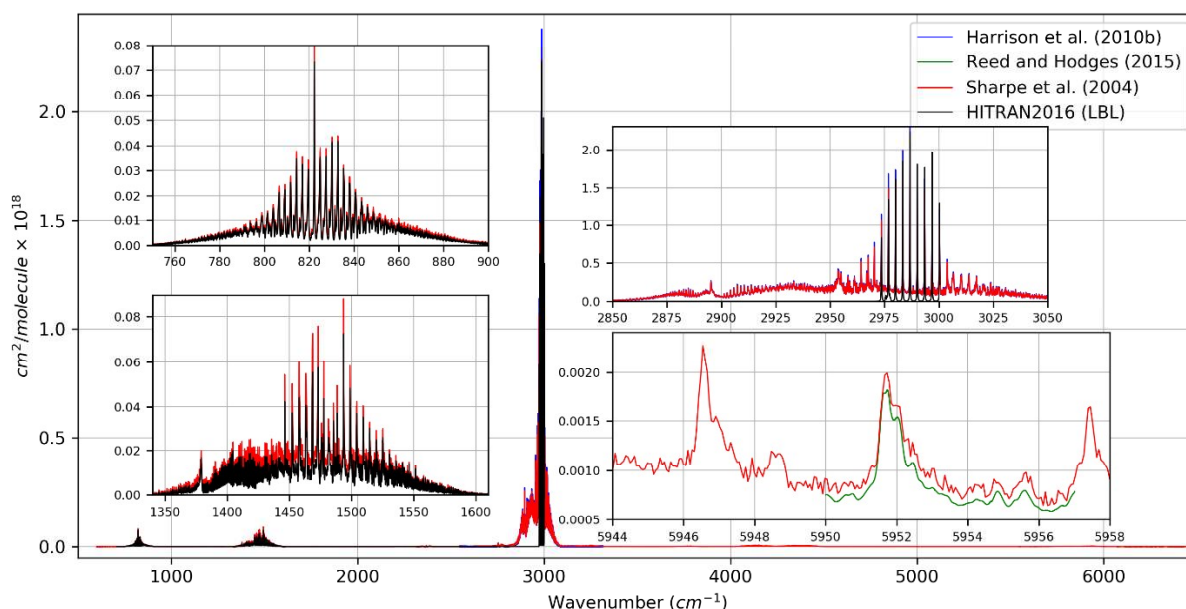
12 In HITRAN, the ethane spectra are contained in both line-by-line and cross-section formats.
13 The line-by-line spectral transition data and its latest update were described in the HITRAN2016
14 paper [2]. The line-by-line data cover the three strongest spectral features of ethane in the IR at
15 3, 6.5, and $11.7\text{ }\mu\text{m}$. Along with line-by-line data, 14 absorption cross-sections provided by
16 Harrison et al. [42] at $3\text{ }\mu\text{m}$ have been adopted in HITRAN2016 from the previous database
17 release without any changes. These data cover the full range of the spectral feature at $3\text{ }\mu\text{m}$
18 which compensates for the incompleteness of the line-by-line data in this region. Intensities were
19 calibrated using the PNNL spectra at three temperatures [42].

20 For the HITRAN2016 update, the data from two sources were considered. First, the PNNL
21 data for 25°C and atmospheric pressure have been compared to the spectra from the line-by-line
22 part [2] and from Ref. [42], as shown in **Figure 17**. Both of the latter sources cover all the
23 strongest bands of ethane in the IR. The PNNL dataset for three temperatures (278.15 , 298.15 ,
24 323.15 K) has been placed in the supplemental folder for the reason of having wider temperature
25 range (up to 50°C) and minor absorption features above 3500 cm^{-1} .

26 More recent data on ethane in the near-infrared range $5950\text{--}5967\text{ cm}^{-1}$ were provided by Reed
27 and Hodges [49]. The spectra were recorded using frequency-stabilized cavity ring-down
28 technique (FS-CRDS) with 200 kHz resolution. The spectra were measured both air- and self-
29 broadened, and the summary on the temperature and pressure sets is given in **Table 15**. The self-
30 broadened and low-pressure air-broadened spectra (at 5, 10, and 50 Torr) had minor artifacts,
31 namely the similar spectral points with different absorptions, which occurred possibly due to the
32 interruptions in data acquisition, which led to changes in the cavity base losses. These artifacts

1 were smoothed using the Akima spline provided in the Scipy package (<https://www.scipy.org/>),
 2 and the spectra were interpolated on an equidistant grid.

3 The **Figure 17** inset shows the comparison of the spectra at room temperature and
 4 atmospheric pressure from PNNL [13] and Reed and Hodges [49]. The insets show three strong
 5 absorption bands in the atmospheric window, and the narrow near-infrared region with the new
 6 cavity ring-down data from Reed and Hodges [49]. The line shapes of FTS and CRDS spectra
 7 are similar with a slight shift in intensity, which can be due to micro-fluctuations of the FTS
 8 baseline. The near-infrared spectrum can be useful in remote sensing of hydrocarbons since
 9 interference by other molecules is minor in this region [49]. The data from Reed and Hodges
 10 have been provided to the users via the HITRAN*Online* web interface.



11
 12 **Figure 17.** Ethane spectra at 296 K and 760 Torr. The absorption cross-sections generated from
 13 the HITRAN2016 line-by-line transitions are given in black.

14 **Table 14.** PNNL data for hydrocarbons

Molecule	CAS	Range, cm ⁻¹	T, K
<i>PNNL – SERDP / Johnson et al. (2010) [22]</i>			
2-Carene	554-61-0	600-7300	298.15, 323.15
3-Carene	13466-78-9	600-7300	298.15, 323.15
D-Limonene	5989-27-5	520-6500	298.15, 323.15
DL-Limonene	138-86-3	580-6500	298.15, 323.15
n-Hexadecane	544-76-3	550-6500	323.15
n-Pentadecane	629-62-9	570-6500	298.15, 323.15
Myrcene	123-35-3	520-6500	298.15, 323.15
Vinyl toluene	100-80-1	550-6500	298.15, 323.15
<i>PNNL – DOE / Sharpe et al. (2004) [13]</i>			
1,3-Butadiene	106-99-0	530-6500	278.15, 298.15, 323.15

1-Butene	25167-67-3	525-6500	278.15, 298.15, 323.15
1-Nonene	124-11-8	550-6500	298.15, 323.15
2-Ethyltoluene	611-14-3	520-6500	298.15, 323.15
4-Ethyltoluene	622-96-8	600-6500	298.15, 323.15
alpha-Pinene (1S)(-)	7785-26-4	600-6500	298.15, 323.15
beta-Pinene (1S) (-)	18172-67-3	600-6500	298.15, 323.15
n-Heptane	142-82-5	600-6500	278.15, 298.15, 323.15
n-Hexane	110-54-3	600-6500	278.15, 298.15, 323.15
Isobutane	75-28-5	600-6500	278.15, 298.15, 323.15
Isobutene	115-11-7	600-6500	278.15, 298.15, 323.15
Isopentane	78-78-4	600-6500	278.15, 298.15, 323.15
Isoprene	78-79-5	600-6500	278.15, 298.15, 323.15
Naphthalene	91-20-3	600-6500	323.15
n-Butane	106-97-8	600-6500	278.15, 298.15, 323.15
Octane	111-65-9	600-6500	278.15, 298.15, 323.15
Pentane	109-66-0	570-6500	278.15, 298.15, 323.15
Propane	74-98-6	600-6500	278.15, 298.15, 323.15
Styrene (monomer)	100-42-5	600-6500	298.15, 323.15
Toluene	108-88-3	600-6500	278.15, 298.15, 323.15
m-Xylene	108-38-3	580-6500	298.15, 323.15
o-Xylene	95-47-6	600-6500	298.15, 323.15
p-Xylene	106-42-3	580-6500	298.15, 323.15

1 Spectra from the PNNL database. All spectra are broadened by nitrogen and recorded at 0.112
2 cm^{-1} resolution.

3 **Table 15.** Ethane absorption cross-sections in the near-infrared range (Reed and Hodges [49]).

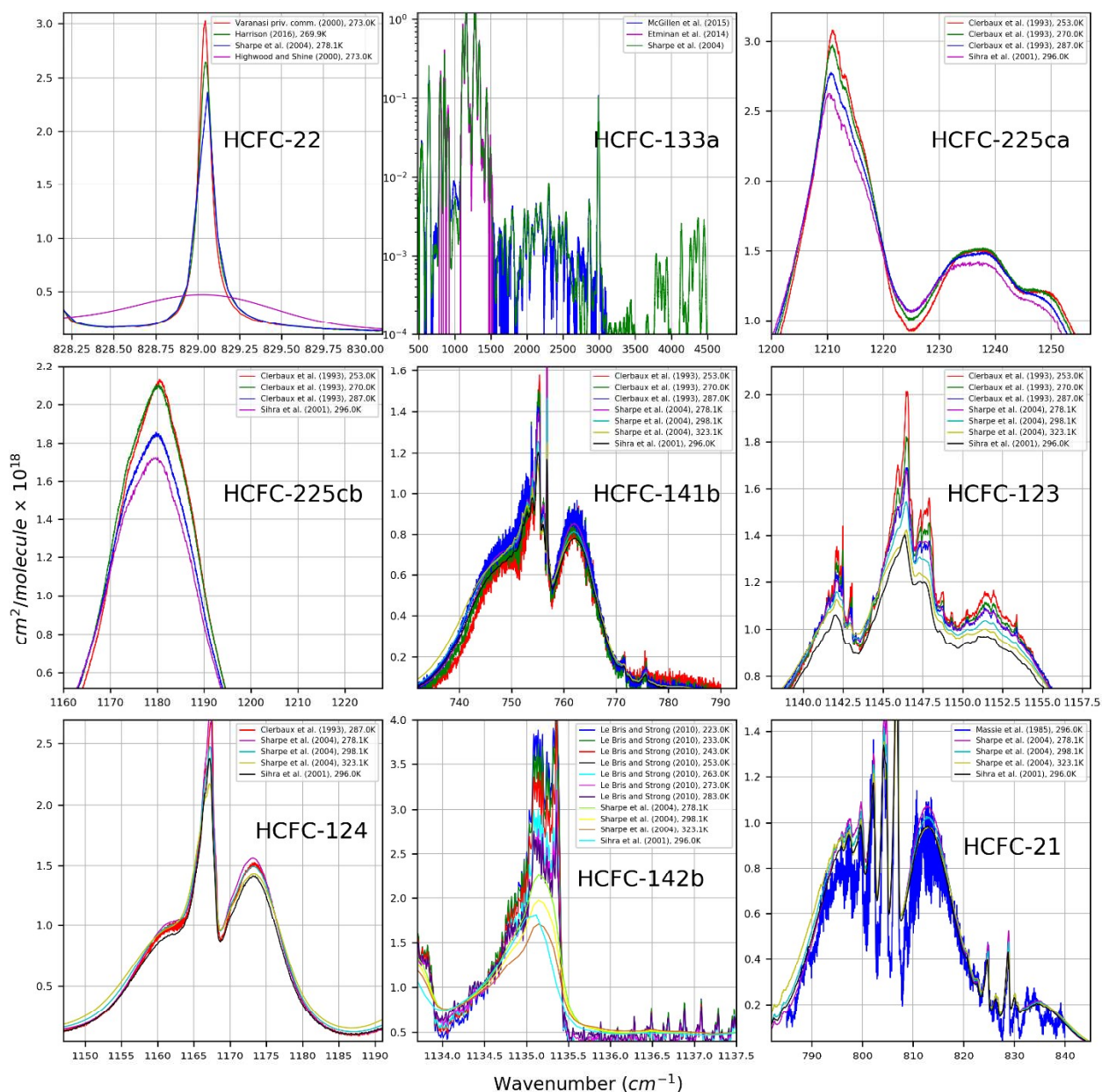
Range, cm^{-1}	T, K	p, Torr	Broadener
5950-5968	296	5	air
5950-5957	295.6	10	air
5950-5957	295.8	50	air
5951-5957	295.8	300	air
5950-5957	295.9	759.9	air
5948-5967	296.1	0.022	self

4 Cavity Ring Down Spectroscopy (CRDS) measurements of ethane air and self-broadened spectra
5 published by Reed and Hodges [49]. Measurements were made at 200 kHz frequency resolution.

1 3.7. Hydrochlorofluorocarbons

2 HCFCs were primarily viewed as transitional replacements for the CFCs in industrial and
3 domestic applications. While still containing chlorine, and hence being ODSs, controlled by the
4 Montreal Protocol, the presence of at least one hydrogen atom makes them susceptible to
5 reaction with OH in the troposphere, and thus they have shorter lifetimes than CFCs, ranging
6 from about 1 month to 2 decades [1] and, for a given emission, are less able to accumulate in the
7 stratosphere. With the worldwide CFC phase-out achieved, the focus has shifted to ending the
8 use of HCFCs themselves, with a final phase-out currently scheduled for 2040 in the developing
9 world and 2030 for developed countries. By far the most abundant HCFC in the atmosphere is
10 HCFC-22, with current concentrations of about 240 pptv and still rising, followed by HFC-141b
11 and HFC142b (concentrations of 20-25 pptv, which have leveled off in recent years) [94]. The
12 2011 radiative forcing due to the HCFCs is about 0.05 W m^{-2} , about 3% of that due to CO_2 [3].
13 Their contribution to ozone depletion, measured by the EESC, is about 3% of the total mid-
14 latitude EESC [107]. In this section, we give a summary for newly added compounds (**Table 16**
15 and **Table 17**), as well as compare the cross-section datasets for some of the molecules. For each
16 sub-section, the cross-section comparison is displayed in the corresponding panel of **Figure 18**.

17



1

2 **Figure 18.** Cross-section comparison for selected bands of hydrochlorofluorocarbons (HCFCs).
 3 In each group, the comparison was taken between the datasets with similar conditions.

4 **3.7.1. HCFC-22**

5 Chlorodifluoromethane (HCFC-22) is the most widely-used HCFC and has been
 6 commercially available since the 1930s. HCFC-22 is a strong greenhouse gas with a 100-year
 7 GWP of 1760 [3], an atmospheric lifetime of 11.9 years [3], and an ODP of 0.034 [115].
 8 Remote-sensing measurements of HCFC-22 in the atmosphere have been made by a number of
 9 instruments, for example ATMOS [120], MIPAS [121], and the ACE-FTS [122].

10 The absorption cross-sections for HCFC-22 have been represented in previous HITRAN
 11 compilations by the air-broadened dataset from Varanasi et al. (supplied to the database in 2000
 12 as a private communication), as well as self-broadened data from Clerbaux et al. [18]. For
 13 HITRAN 2016, the air-broadened dataset from Varanasi et al. has been replaced by a new

1 dataset from Harrison [27]. New high-resolution cross-sections for air-broadened HCFC-22 have
2 been determined over the spectral range 730 – 1380 cm^{-1} . Spectra were recorded using a high-
3 resolution Fourier transform spectrometer (Bruker IFS 125HR) for mixtures of HCFC-22 with
4 dry synthetic air in a 26-cm-pathlength cell at spectral resolutions between 0.01 and 0.03 cm^{-1}
5 (calculated as 0.9/MOPD) over a range of temperatures and pressures appropriate for upper
6 troposphere – lower stratosphere conditions (7.5 – 762 Torr and 191 – 295 K). The intensity was
7 calibrated using the spectra from PNNL [13].

8 The detailed comparison of the new data from Harrison with Varanasi et al. is provided in
9 Ref. [27]. **Figure 18** contains the comparison for the datasets at the 270K temperatures with
10 additional data from Highwood and Shine [90], which were an average over measurements from
11 5 laboratories in the intercomparison reported by Ballard et al. [123]. All cross-sections were
12 compared in the region, and good agreement has been demonstrated (around 3.2% in the worst
13 case between the data from Ref. [27] and Ref. [90]). The spectrum from Ref. [90] was supplied
14 with lower resolution and contains no additional features; thus it was omitted. The PNNL data
15 were also included in HITRAN*Online* as having a higher range of temperature. The dataset from
16 Varanasi et al. is available in the supplemental folder.

17 **3.7.2. HCFC-133a**

18 HCFC-133a (2-chloro-1,1,1-trifluoroethane, $\text{C}_2\text{H}_2\text{ClF}_3$) is used in pharmaceuticals and
19 chemical synthesis [124] and has an atmospheric lifetime of 4 years. It is removed from the
20 atmosphere primarily by reactions with OH radicals and by UV photolysis. Recently it was
21 detected in the atmosphere for the first time (see the Nature review by Laube et al. [125]). We
22 have compared data from three sources for this molecule: PNNL [13], and more recently
23 published data from McGillen et al. [124] and Etminan et al. [43]. The stated resolution for
24 spectra from McGillen et al. is 1 cm^{-1} , which is lower than for both the other sources, and has
25 somewhat less agreement with Refs. [13,43] for intensities, integrated in two ranges: ~6% in
26 780-920 cm^{-1} range, and ~2% in 1080-1520 cm^{-1} range. Both recently-measured data from Refs.
27 [43,124] are recorded at low pressures (20 Torr He-broadened and 6.96 Torr self-broadened
28 respectively). The data from Refs. [13] and [43] have been included in HITRAN*Online*. The He-
29 broadened data from McGillen et al. have a similar line shape as the PNNL spectra, and do not
30 contain additional features. Since the overall agreement is good, it has been decided to include
31 these data in the supplemental folder, to demonstrate the independence of these spectra on
32 broadener and pressure.

3.7.3. HCFC-225ca and HCFC-225cb

The spectra for HCFC-225ca ($\text{CHCl}_2\text{CF}_2\text{CF}_3$) and HCFC-225cb ($\text{CClF}_2\text{CF}_2\text{CHClF}$) have been included in the previous versions of HITRAN. These data have been published by Clerbaux et al. [18] and span the 253-287 K temperature range. No dependence on pressure of these cross-sections was found [18], so in principle these cross-sections can be used at higher pressures. The compared data from Sihra et al. [61] shows good agreement for both molecules, and the integrated intensities agree well within experimental uncertainties of Ref. [18]. The data from the latter source can be used in conjunction with the set from Clerbaux et al. since it agrees well with the temperature dependence of the spectra from Clerbaux et al. The spectra from Sihra et al. were included in HITRAN*online* for this molecule.

3.7.4. HCFC-141b

For the HCFC-141b (1,1-dichloro-1-fluoroethane, $\text{CH}_3\text{CCl}_2\text{F}$) molecule, the HITRAN database includes high-resolution self-broadened data from Clerbaux et al. [18] for three temperatures (253, 270, and 287 K), having each temperature dataset split into three regions (710-790, 895-1210, and 1325-1470 cm^{-1}), corresponding to the strong, medium, and weak absorption regions respectively. The new data for this comparison come from PNNL [13] and Sihra et al. [61] which provide spectra at atmospheric pressure broadened by nitrogen and synthetic air, and Le Bris et al. [34] which provide pure vapor spectra.

The comparison of the old spectra from Ref. [18] shows several things to note. First, the 270K spectra from Clerbaux et al. contain a unique spectral feature in the region between 990-1070 cm^{-1} , which is not contained in the datasets for different temperatures from Clerbaux et al. as well as from Refs. [13,61]. This artifact has been identified as a contaminant [34]. Users should be aware of that when processing these spectra. The PNNL data (given at three standard temperatures), and data from Clerbaux agree well for the temperatures that can be compared (i.e. one set around 270K, and another around 290K) with integrated intensities within 3% at 296K and within 5.5% at 270K for all spectral features except the weakest one (at 1400 cm^{-1}). The data from Le Bris et al. also agree very well with those from PNNL (within 1.5%) and Clerbaux et al. (within 2%) around 283 K. At lower temperatures, however, the integrated intensities of Clerbaux et al. decrease by 12% at 253 K, while those from Le Bris et al. do not change appreciably. The Le Bris et al. data cover a larger spectral range (570-3100 cm^{-1}) and a wider temperature range (223K to 283K) and this dataset is preferred.

The spectra from Refs [13,34] have been included in HITRAN*online* (Ref. [34] as a post-release update) because of the good agreement with the spectra from Clerbaux et al. The data

1 from Sihra et al. [61] disagree somewhat with the two other sources (8-10% for integral
2 intensities) and have been omitted. Self-broadened data from Ref. [18] was relegated to the
3 supplemental folder.

4 **3.7.5. HCFC-123**

5 For HCFC-123 (2,2-dichloro-1,1,1-trifluoroethane, CHCl_2CF_3), the previous HITRAN issues
6 had a set of spectra from Clerbaux et al. [18] for three temperatures (253, 270, and 287K), each
7 set split into two regions (740-886 cm^{-1} and 1080-1450 cm^{-1}). New data come from PNNL [13]
8 and Sihra et al. [61]. The agreement between the PNNL and Clerbaux spectra is excellent (1.4%
9 for the region of the strong absorption, and less than a percent difference for weak absorption
10 region at lower wavenumbers). The spectra from Sihra et al. [61] agree less well with other
11 sources, with a discrepancy of approximately 9%. It should be noted that in the region 960-1150
12 cm^{-1} , data from Sihra et al. contain additional spectral features at the strong band near 1200 cm^{-1} .
13 These artifacts are absent in other sources and possibly are due to the presence of contaminants.
14 In addition, the PNNL spectra have an additional strong feature at 670 cm^{-1} , while the spectrum
15 from Sihra et al. is zeroed in this region. The spectra from PNNL have been included in
16 HITRAN*online*, while the Sihra et al. data were omitted.

17 **3.7.6. HCFC-124**

18 For HCFC-124 (2-chloro-1,1,1,2-tetrafluoroethane, CHClCF_3), data from previous HITRAN
19 issues include only Clerbaux et al. [18] for 287 K. New data from PNNL [13] are given for three
20 standard temperatures, and data from Sihra et al. [61] for 296K. PNNL have been included in
21 HITRAN*online* owing to the good agreement with Ref. [18] (~2.8%) for the 790-430 cm^{-1} range.
22 Data from Sihra et al. have been omitted.

23 **3.7.7. HCFC-142b**

24 HCFC-142b (1-chloro-1,1-difluoroethane, CH_3CClF_2) is a compound that has been detected
25 in the Earth's atmosphere (see [126] and references within). The previous HITRAN editions
26 contained cross-sections for this molecule provided by Le Bris and Strong [33]. These data are
27 self-broadened and recorded at multiple temperatures in the 223-283K range. The new data in
28 this comparison includes PNNL [13] and Sihra et al. [61]. The comparison of the integrated
29 intensities in two ranges (**I**: 860-1000 cm^{-1} , **II**: 1050-1300 cm^{-1}) shows good agreement between
30 Le Bris and Strong, and PNNL (~3.2% for region **I**, and 1.9% for region **II**). The spectrum from
31 Sihra et al. agrees within 6% with both of these sources, but lacks the feature at 680 cm^{-1} . The
32 spectrum from Sihra et al. shows a slight wavenumber shift (~0.25 cm^{-1}) in comparison to other

1 sources. The data from PNNL were included in HITRAN*Online*, while the spectra from Sihra et
 2 al. were omitted.

3 3.7.8. HCFC-21

4 The HCFC-21 (dichlorofluoromethane, CHCl₂F) molecule was included in previous HITRAN
 5 issues as a single high-resolution set for 1 Torr pressure and 296K temperature provided by
 6 Massie et al. [12]. The stated intensity accuracy was 10-25% [12]. The spectra from this source
 7 cover only a small region, 780-840 cm⁻¹. As for most of the molecules from this sub-section, new
 8 data come from PNNL [13] and Sihra et al. [61]. The agreement is generally good for the data
 9 from Refs. [13,61] (~5.5%), and for Ref. [12] it is within 17%, which is consistent with the
 10 stated accuracy. The Sihra et al. spectrum was omitted since it does not contain information
 11 additional to that from PNNL. The PNNL data have been included in HITRAN*Online*.
 12 Considering the high resolution of the data from Massie et al. compared to PNNL, these data
 13 have been retained available through HITRAN*Online* interface and will be considered to be
 14 moved to the supplemental folder once more recent high-resolution data can be provided to the
 15 database.

16 **Table 16.** PNNL-DOE data for hydrochlorofluorocarbons

Molecule	CAS	Range, cm ⁻¹
1,2-Dichloroethane	107-06-2	600-6500
HCFC-132	431-06-1	550-6500
HCFC-121	354-14-3	560-6500
HCFC-123	306-83-2	500-6500
HCFC-123a	354-23-4	520-6500
HCFC-124	2837-89-0	500-6500
HCFC-141b	1717-00-6	550-6500
HCFC-142b	75-68-3	600-6500
HCFC-21	75-43-4	600-7300
HCFC-124a	354-25-6	525-6500
HCFC-132a	471-43-2	510-6500
HCFC-133a	75-88-7	500-7100
HCFC-141	430-57-9	550-6500

17 Spectra from Sharpe et al. (2004) [13] broadened by N₂. Spectra were recorded at 278.15 K,
 18 298.15 K, and 323.15 K, 760 Torr total pressure, and 0.112 cm⁻¹ resolution.

19 **Table 17.** Spectra from the halocarbon review [1] and more recently-measured data for
 20 hydrochlorofluorocarbons

Molecule	CAS	Range (cm ⁻¹)	T (K)	P (Torr)	Broader	Resolution (cm ⁻¹)	Source
HCFC-141b	1717-00-6	560-3100	223-283			0.02	Le Bris et al. (2012) [34]
1,2-Dichloroethane	107-06-2	620-1570	293				Vander Auwera (2000) [95]
HCFC-225ca	422-56-0	650-1400	296	700	air	0.5	Sihra et al. (2001) [61]
HCFC-225cb	507-55-1	500-1500	296	700	air	0.5	Sihra et al. (2001) [61]

(E)-1-Chloro-3,3,3-trifluoroprop-1-ene	102687-65-0	600-2000	295	700	air	0.25	Andersen et al. (2008) [77]
HCFC-133a	75-88-7	690-1730	295	6.96		0.01	Etminan et al. (2014) [43]
HCFC-22	75-45-6	730-1380	190.8-294.8	7.496-761.6	air	0.01-0.03	Harrison (2016) [27]†

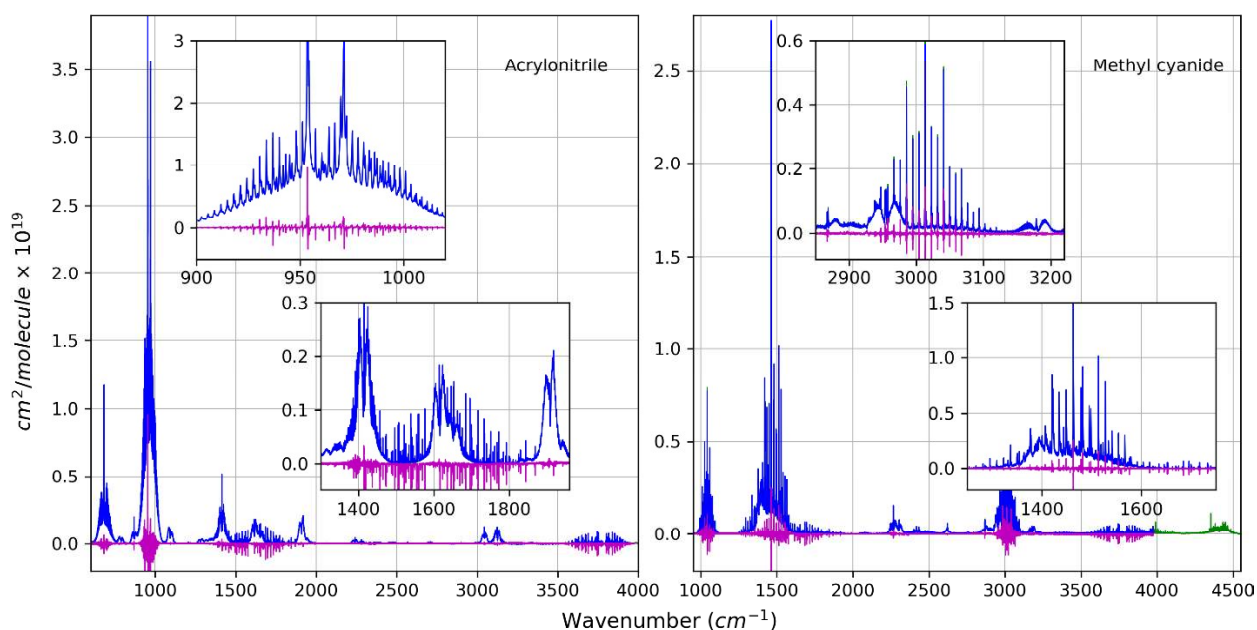
1 † The full list of cross-sections and a more detailed description of the experimental conditions
2 can be found in Ref. [27].

3 3.8. Nitriles, amines and other nitrogenated hydrocarbons

4 Nitrogenated hydrocarbons are an atmospherically-important class of molecules; they have
5 both anthropogenic and natural sources. These compounds, in particular, are produced by
6 biomass burning.

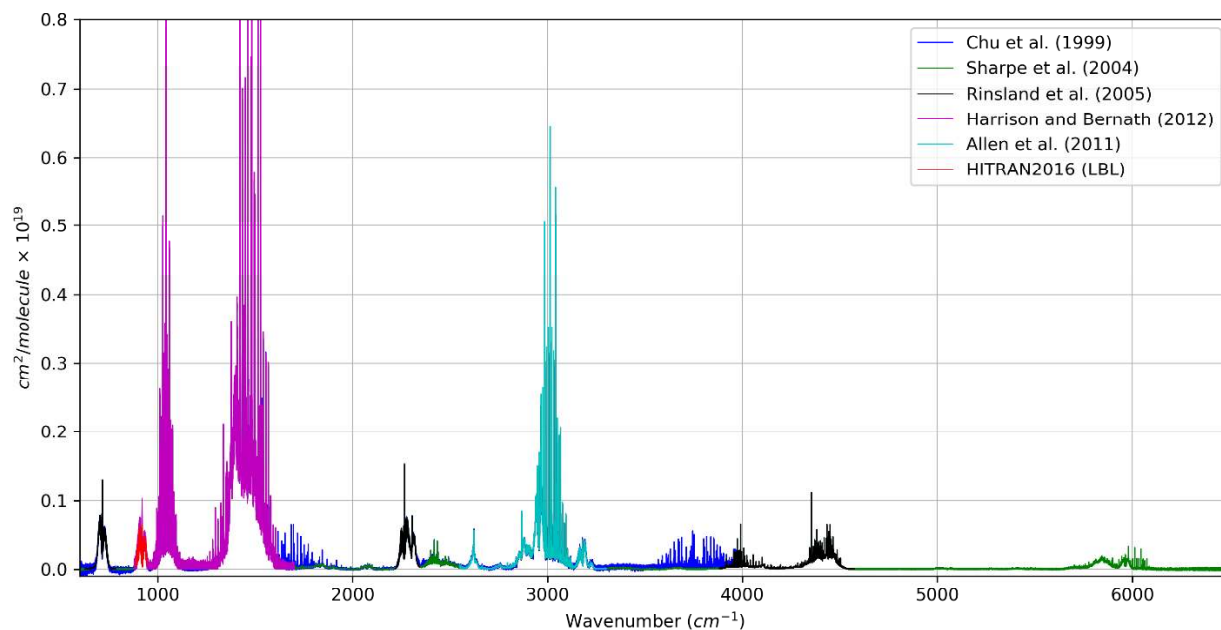
7 In previous HITRAN editions, this category was limited to methyl cyanide (acetonitrile,
8 CH_3CN), which was presented both in the line-by-line part in the far infrared region, and cross-
9 sections in the range up to 4580 cm^{-1} [127–129]. The HITRAN2016 edition expands this
10 category to more than 25 molecules including nitrogen-containing acids, -alkyl nitrites, amines
11 and aromatic amines, and nitriles.

12 The summary of the molecules in the update is given in the **Table 18**. It should be noted that
13 the data for most of the molecules come from PNNL measurements [13,22]. The spectra for
14 some of the molecules from the PNNL database, such as 2-vinylpyridine and N,N-diethyl
15 formamide, have been provided in the newer set of data, the PNNL-SERDP database by Johnson
16 et al. [22].



17
18 **Figure 19.** NIST (blue) and PNNL (green) spectra for acrylonitrile (left panel) and methyl
19 cyanide (right panel). PNNL-NIST differences in magenta. The insets show the most noticeable
20 discrepancies which include minor water contamination in the spectra from Ref. [17].

1



2

3 **Figure 20.** Spectra for methyl cyanide. The spectrum generated from the line-by-line
4 HITRAN2016 data is given in red.

5 PNNL [13] and NIST [17] spectra for acrylonitrile and methyl cyanide were compared. For
6 acrylonitrile, very good agreement in the integrated intensities were observed ($\sim 0.5\%$ for the
7 strongest features near 800 cm^{-1}). For methyl cyanide, the agreement is also very good ($\sim 1.5\%$).
8 From the visual comparison given in **Figure 19**, it can be seen that for line shapes, the agreement
9 is worse and in some ranges (see insets in the figure) there are substantial differences, mostly due
10 to presence of the sharp lines in NIST spectra that are absent in PNNL (likely trace H_2O
11 contaminants). For acrylonitrile, the PNNL data have been included in *HITRANonline*.

12 For methyl cyanide, the previous HITRAN edition included the cross-sections from Rinsland
13 et al. [127], Allen et al. [129], and Harrison and Bernath [128], which cover all strong features
14 in the spectral range from 600 to 5000 cm^{-1} , and a wide range of pressures and temperatures. The
15 full range can be seen in the corresponding papers, or at the *HITRANonline* web-site. Besides
16 the wide-range cross-section data, the HITRAN database contains the set of transitions in the
17 line-by-line part in the narrow spectral region (890 - 946 cm^{-1}) which are plotted in **Figure 20** in
18 red. The cross-section datasets are of high quality and are retained in the HITRAN2016 database,
19 while the PNNL dataset has been omitted since it does not contain additional strong features to
20 those sources already present in the database.

21

1 **Table 18.** PNNL data for nitriles, amines and other nitrogenated hydrocarbons

Molecule	CAS	Range, cm ⁻¹	T, K
<i>PNNL-SERDP / Johnson et al. (2010) [22]</i>			
2-Vinylpyridine	100-69-6	540-7300	298.15, 323.15
N,N-diethyl formamide	617-84-5	520-6500	298.15, 323.15
<i>PNNL-DOE / Sharpe et al. (2004) [13]</i>			
Acrylonitrile	107-13-1	550-6500	278.15, 298.15, 323.15
Diethylamine	109-89-7	590-7000	278.15, 298.15, 323.15
Diisopropylamine	108-18-9	550-6500	278.15, 298.15, 323.15
Dimethylamine	124-40-3	550-6700	278.15, 298.15, 323.15
Ethylamine	75-04-7	600-7000	278.15, 298.15, 323.15
Ethylenediamine	107-15-3	550-6500	278.15, 298.15, 323.15
Hydrazine	302-01-2	600-7000	298.15, 323.15
Isocyanic acid	75-13-8	500-7000	298.15
Isopropylamine	75-31-0	600-6800	278.15, 323.15
Methyl isothiocyanate	556-61-6	600-6500	278.15, 298.15, 323.15
Methyl nitrite	624-91-9	520-6500	278.15, 298.15, 323.15
N,N-Diethylaniline	91-66-7	550-6500	298.15, 323.15
n-Butylamine	109-73-9	600-7000	278.15, 298.15, 323.15
Nitrobenzene	98-95-3	550-6500	298.15, 323.15
Nitrogen trifluoride	7783-54-2	600-6500	278.15, 298.15, 323.15
Nitrous acid	7782-77-6	550-6500	298.15
o-Toluidine	95-53-4	550-7300	298.15, 323.15
Piperidine	110-89-4	510-6700	278.15, 298.15, 323.15
Propylenimine	75-55-8	600-6700	278.15, 298.15, 323.15
sec-Amylamine	625-30-9	540-6900	278.15, 298.15, 323.15
Triethylamine	121-44-8	575-7100	278.15, 298.15, 323.15
Trimethylamine	75-50-3	600-6500	278.15, 298.15, 323.15

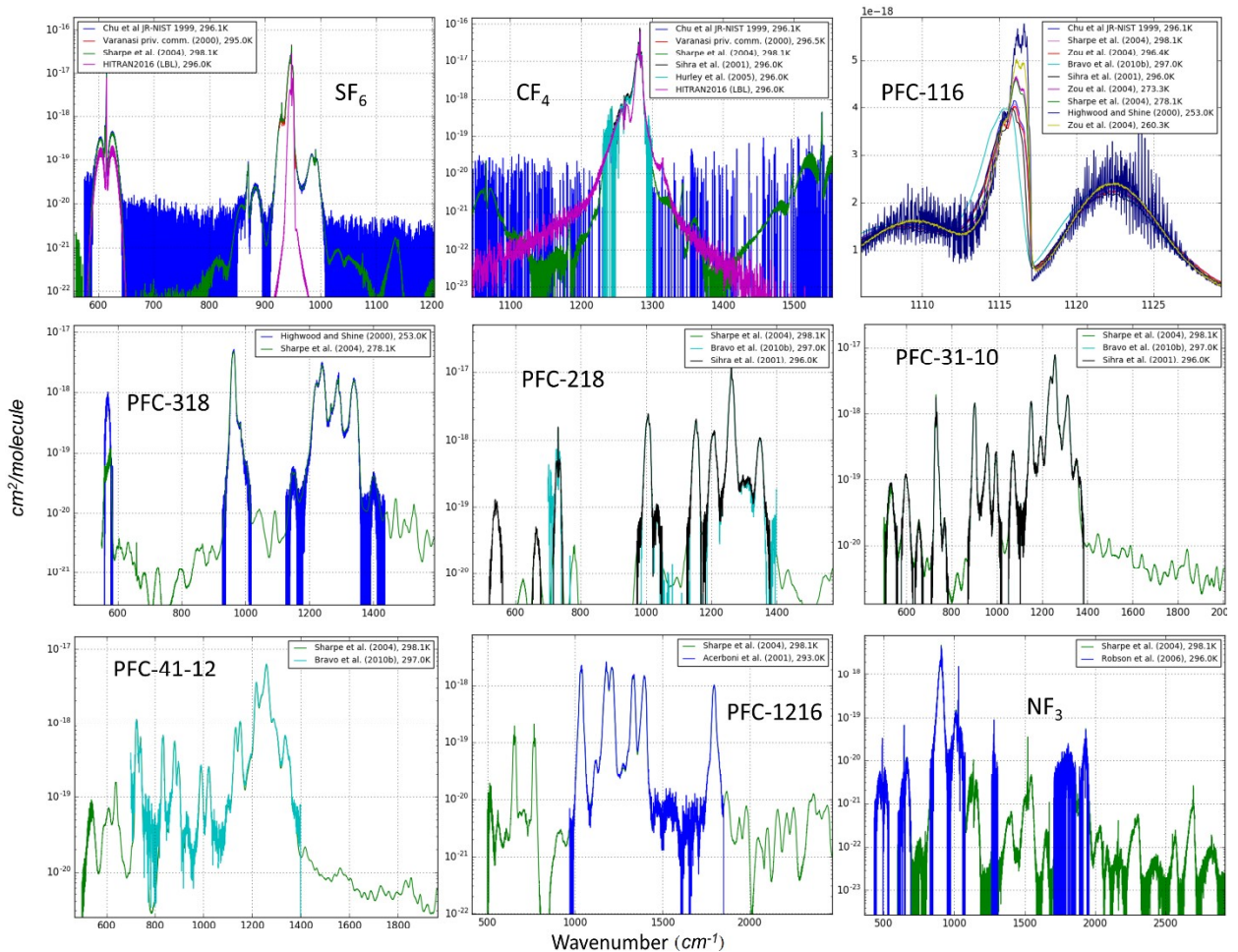
2 Spectra were recorded in 760 Torr total pressure of N₂ with 0.112 cm⁻¹ resolution.

3 **3.9. Fully-Fluorinated Species**

4 The fully-fluorinated species in HITRAN2016 consist predominantly of perfluorocarbons
5 (PFCs), but include four other species: nitrogen trifluoride (NF₃), sulfur hexafluoride (SF₆),
6 sulfuryl fluoride (SO₂F₂), and trifluoromethyl sulfur pentafluoride (SF₅CF₃). The listed
7 molecules also belong to another category (nitriles, amines, and other nitrogenated compounds
8 for NF₃, and sulfur-containing species for the remaining three). The saturated PFCs are very
9 long-lived (more than a thousand years, and up to 50,000 years in the case of CF₄), while
10 unsaturated PFCs can have lifetimes of approximately one day [1] but they are generally
11 characterised by particularly strong absorption in the 1200-1300 cm⁻¹ region, making them
12 potent greenhouse gases. They are not ODSs. The most abundant PFC is CF₄ with concentrations
13 of approximately 80 pptv and rising, and is released during aluminum production. CF₄ has a
14 significant natural background concentration estimated to be 35 pptv [3]. SF₆, used extensively
15 in electrical insulation, and NF₃, used in the electronics industry, have abundances of around 7
16 and 0.9 pptv [3], respectively and rising. Emissions of many PFCs and SF₆ are part of the

1 collection of gases whose emissions fall under the Kyoto Protocol, with NF_3 added in the second
 2 commitment period of the protocol. The 2011 radiative forcing due to these fully fluorinated
 3 species was about 0.01 W m^{-2} , about 0.5% of the forcing due to CO_2 [3].

4 As in other sections, we present comparisons of the spectra for the molecules for which a
 5 number of different sources were given. In particular, it includes the recent measurements of
 6 perfluorotributylamine [21], made self-broadened and provided at zero pressure limit.
 7 Measurements of isomers of perfluorodecalin, a compound of medical use [38,130], are also
 8 supplied. The spectra for several perfluorinated alkanes [53]) and alkenes [5] are also available
 9 in HITRAN for the first time. The full list of these fluorinated compounds available via
 10 HITRANonline is given in **Table 19** and **Table 20**. Below we give the description of some of the
 11 compounds of the most atmospheric importance, comparisons for which are also provided in
 12 **Figure 21**.



13
 14 **Figure 21.** Cross-section comparison for fully fluorinated species.

15 **3.9.1. Sulfur hexafluoride**

16 Sulfur hexafluoride (SF_6) is a gas included in the Kyoto Protocol, with a lifetime of 3200
 17 years and a 100-year GWP of 23,500 [3]. The HITRAN2016 database contains an extensive list

1 of transitions for two regions (580-650 cm^{-1} and 900-1000 cm^{-1}), and for three bands (ν_3 , ν_4 , and
2 $\nu_4+\nu_6-\nu_6$). This effective Hamiltonian dataset was described in the HITRAN2008 paper [131].

3 Since SF_6 is a heavy polyatomic molecule, for which the spectral analysis based on the
4 molecular quantum structure is quite complicated, the cross-section data have been included in
5 previous HITRAN issues. Data have been provided by Varanasi and co-workers [132], and via
6 private communication [133]). The data from Ref. [132] are self-broadened low-pressure sets for
7 216K, given for three pressures from 25-90 Torr. The air-broadened data, provided by Varanasi
8 to HITRAN2000, span a wide range of temperatures and pressures (180-295K, 20-60 Torr) and
9 are measured with a 0.03 cm^{-1} resolution in the 925-925 region. All these data reside in the
10 recent version of HITRAN. Other data for the comparison, besides the line-by-line and Varanasi
11 datasets, are from the PNNL [13] and NIST [17] databases.

12 Comparison of the line-by-line data to experimental cross-sections demonstrates the lack of
13 bands in the line-by-line data, and possible line-mixing effects. We have compared the cross-
14 section data at room temperature (i.e. not including Ref. [132]) in the region of the strongest
15 absorption 925-955 cm^{-1} . The best agreement for the intensities, integrated in this region, is
16 between PNNL and Varanasi ($\sim 1.6\%$), while the NIST data, although agreeing well, differs by 5-
17 7% and has systematically higher absorption. The PNNL data have been included in
18 HITRAN*online* to enable the higher-temperature interpolation.

19 **3.9.2. Carbon tetrafluoride**

20 Carbon tetrafluoride (CF_4) is a long-lived substance with an estimated lifetime of 50000 years
21 and 100-yr GWP of 6630 [3] . Its strongest absorption is situated in the 1000-1500 cm^{-1}
22 atmospheric window and it is included in the Kyoto protocol basket of gases [134].

23 The line-by-line part of the HITRAN2016 edition had been updated with a new line list from
24 the global Effective Hamiltonian fit [2,135]. However, due to the overall complexity of the
25 spectra of this molecule, the HITRAN database also includes the IR cross-section datasets.

26 In previous HITRAN editions, the air-broadened cross-sections for CF_4 have been included
27 from Varanasi and co-workers (HITRAN2000 [133], private communication). These data have
28 been recorded at a set of pressures and temperatures (180-296 K and 22-761 Torr) with a
29 resolution of 0.03 cm^{-1} . In addition, in this comparison we have the data from Hurley et al. [134],
30 which provide revised room-temperature cross-sections, measured at Ford in Sihra et al. [61].

31 The integrated intensities for the room-temperature cross-section data have been compared in
32 the 1250-1300 cm^{-1} interval. There is good agreement between the data from Varanasi, PNNL,

1 and NIST (~1.6%), while the spectrum from Hurley et al. lies outside of this group by 5.3%.
2 From the visual comparison given in the [Figure 21](#), the good agreement of the band shapes can
3 be seen for the cross-section data. The data from Sihra et al. [61] is shifted in wavenumber scale,
4 and the integrated cross-section disagrees with other data by more than 10%. More detailed
5 discussion and comparisons of the datasets from Hurley et al. and from Sihra et al. is provided in
6 Ref. [134].

7 The spectra generated from the line list [2,135] lacks absorption in the region where it is
8 strong, which can be both due to missing bands or line-shape effects, e.g. line mixing. For the
9 weak absorption, the line-by-line data give a stronger absorption than that which is measured
10 directly, see [Figure 21](#). The data from PNNL have been included in HITRAN*Online*, while data
11 from other sources (Refs. [61,134]) have been omitted.

12 **3.9.3. Perfluoroethane**

13 Perfluoroethane (PFC-116, C_2F_6) has a lifetime of 10000 years and 100-year GWP of 11100
14 [3]. Previous HITRAN editions included the data for this molecule supplied by Zou et al. [46].
15 This dataset is air-broadened and measured with 0.01-0.03 cm^{-1} resolution, spanning a wide
16 range of temperatures and pressures (180-296 K and 5-760 Torr). This dataset has not undergone
17 changes in the HITRAN2016 edition.

18 The PNNL [13] and NIST [17] databases provide cross-section for this comparison: PNNL at
19 three temperatures (273K, 296K, and 323K), and NIST at 296K.

20 Two data sources from Ref. [1] provided the room temperature cross-sections. In Sihra et al.
21 [61], the atmospheric pressure air-broadened spectra had been measured at Ford. Measurement
22 resolution for this source is 0.5 cm^{-1} . The second source, Bravo et al. [53], provided the self-
23 broadened spectrum measured in the Rutherford Appleton Laboratories (RAL).

24 The lower-temperature spectra at 253K have been provided in the compilation by Highwood
25 and Shine [90]. These data originate from the work by Ballard et al. [136]. The spectra were
26 measured self-broadened with 0.03 cm^{-1} resolution. The cross-sections have been split into three
27 regions containing the strong absorption bands: 680-750 cm^{-1} , 1075-1165 cm^{-1} , and 1170-1380
28 cm^{-1} .

29 Comparisons of the integrated intensities were made at three different temperatures. At
30 $T \sim 253K$, only data from Ref. [90] (Highwood and Shine, see Ballard et al. for measurement
31 details) and Ref. [46] (Zou et al.) are available. The agreement is around 3%, which is inside the
32 combined experimental uncertainties of these sources. For $T \sim 275K$, the data were available from

1 Zou et al. and PNNL, which also show a good agreement (1.5% for the strongest feature at 1250
2 cm^{-1}). At the 296K temperature, all the above listed sources (except Highwood and Shine which
3 do not contain room temperature data) have been compared. The situation here is less
4 straightforward, although it can be seen that two groups of sources are close together: 1) NIST
5 and Bravo et al., 2) Zou et al. and Sihra et al. cross-sections from these two groups agree within
6 0.6% on the intensity, integrated in 1170-1400 cm^{-1} wavenumber range. The PNNL data have a
7 reasonable residual compared to all the sources, with the worst case of $\sim 6\%$ in comparison with
8 Sihra et al. The agreement with NIST is 1.1%, and with Zou et al. is 1.8-5%. It should be
9 pointed out that the spectrum from Bravo et al. has a wavenumber shift which is displayed in
10 **Figure 21**. The spectra from PNNL and Ref. [90] were included in HITRAN*Online*.

11 **3.9.4. PFC-318**

12 Data for PFC-318 (perfluorocyclobutane, C_4F_8) are available from two sources: PNNL [13],
13 and Highwood and Shine [90]. The data from Highwood and Shine are for pure gas, and
14 measured at 253K temperature with 0.03 cm^{-1} resolution. These data were initially supplied by
15 Ballard et al. [136]. The agreement of the integrated intensities is good between these two
16 sources (within 1.7%). The PNNL data has an artifact at the lower wavenumber range (see
17 **Figure 21**) which is due to the signal saturation. In this region, using Highwood and Shine data
18 is preferable. Both sources were included in HITRAN*Online* as having different broadening
19 conditions.

20 **3.9.5. PFC-218**

21 Cross-sections for PFC-218 (perfluoropropane, C_3F_8) were published in PNNL [13], Bravo et
22 al. [53], and Sihra et al. [61]. Spectra published in Bravo et al. have been measured in RAL as a
23 pure vapor. The agreement of integrated intensities is 1.2% between PNNL and Sihra et al., and
24 $\sim 3\%$ between Bravo et al. and Sihra et al. The spectrum from Sihra et al. is better resolved than
25 both of the other sources. PNNL spectra have been included in HITRAN*Online*, while the data of
26 Sihra et al. were placed into the supplemental folder. The spectrum from Bravo et al. has been
27 omitted due to the presence of digitization artifacts.

28 **3.9.6. PFC-31-10**

29 Cross-sections for PFC-31-10 (perfluorobutane, C_4F_{10}) have been published in PNNL [13],
30 Bravo et al. [53], and Sihra et al. [61]. The spectrum from Bravo et al. contains data from Sihra
31 et al. with a wavenumber scale correction. These data have more resolved spectral features but
32 lack temperature dependence; they have thus been placed into the supplemental folder. The

1 PNNL data have been included in HITRAN*Online*, while the spectrum from Sihra et al. has been
2 omitted.

3 **3.9.7. PFC-41-12**

4 For PFC-41-12 (perfluoropentane, $n\text{-C}_5\text{F}_{12}$), spectra were supplied by PNNL [13] (for three
5 temperatures) and Bravo et al. [53]. The latter data were given only at room temperature and
6 measured self-broadened at RAL. These data are more resolved, but the wavenumber scale
7 differs from PNNL by a shift of approximately 1 cm^{-1} . Since they were measured at higher
8 resolution, these data can possibly be useful in retrievals. For this reason, they have been placed
9 into the supplemental folder, while the PNNL dataset was included in HITRAN*Online*.

10 **3.9.8. PFC-1216**

11 PFC-1216 (perfluoropropene, C_3F_6) spectra were supplied by PNNL [13] (at three
12 temperatures) and Acerboni et al. [5]. Data from Acerboni et al. are air-broadened and measured
13 at 293 K and 740 Torr. The agreement between the cross-section from Acerboni et al. and PNNL
14 is 3.83%, which is well within the combined experimental uncertainties [5]. The data from
15 Acerboni et al. are more resolved. They have been placed in the supplemental folder, while
16 PNNL spectra were included in HITRAN*Online*.

17 **3.9.9. Nitrogen trifluoride**

18 Spectral data for nitrogen trifluoride (NF_3) are available from PNNL [13] and Robson et al.
19 [137]. The agreement of the integrated intensities within the $800\text{-}1100\text{ cm}^{-1}$ interval is 2.5%. The
20 spectrum from Robson et al. was measured only at room temperature and 700 Torr pressure. This
21 spectrum, however, provides a band missing in PNNL data at 490 cm^{-1} . For this reason, it has
22 been put into the supplemental folder. The PNNL spectra have been included in HITRAN*Online*.

23 **3.9.10. Perfluorodecalin**

24 Perfluorodecalin (PFD, $\text{C}_{10}\text{F}_{18}$) is a cyclic fluorocarbon used in medical and cosmetic
25 applications. It is usually sold as a mixture of *cis*- and *trans*-isomers. Spectra have been
26 published by Shine et al. [130] and Le Bris et al. [38]. In both studies the spectra of the *trans*-
27 isomer have been extracted by subtracting the spectra of a sample of *cis*-PFD solution from the
28 spectra of a PFD mixture. This leads to higher uncertainties for the *trans*-PFD cross-sections.
29 The comparison of the integrated cross-sections around 300K shows a difference between the
30 two studies of 6.5% for *cis*-PDF and 11% for *trans*-PDF. The data from Le Bris et al. have been
31 added as a post-release update, while the data from Ref. [130] were placed in a supplemental
32 folder.

1 **Table 19.** PNNL-DOE data for fully fluorinated species

Molecule	CAS	Range, cm ⁻¹
Hexafluorobenzene †	392-56-3	580-6500
Nitrogen trifluoride	7783-54-2	600-6500
PFC-1216	116-15-4	500-6500
PFC-14	75-73-0	570-6500
PFC-218	76-19-7	600-6500
PFC-31-10	355-25-9	500-6500
PFC-318	115-25-3	550-6500
PFC-41-12 ‡	678-26-2	500-6500
Perfluoroethane	76-16-4	500-6500
Sulfonyl fluoride	2699-79-8	500-6500
Sulphur hexafluoride	2551-62-4	560-6500

2 Spectra from Sharpe et al. [13] broadened by N₂. Spectra were recorded at 278.15 K, 298.15 K,
 3 and 323.15 K (except hexafluorobenzene and PFC-41-12), 760 Torr total pressure, and 0.112 cm⁻¹
 4 resolution.

5 † Spectra for hexafluorobenzene were recorded at 298.15 K and 323.15 K

6 ‡ Spectra for PFC-41-12 were recorded at 278.15 K and 323.15 K

8 **Table 20.** Spectra from the halocarbon review [1] and recently published data for fully
 9 fluorinated species

Molecule	CAS	Range (cm ⁻¹)	T (K)	P (Torr)	Brd	Res	Source
Perfluorodecalin (E,Z)	306-94-5	560-3000	300-340			0.1	Le Bris et al. (2017) [38]
PFTBA †	311-89-7	550-2000	298-344			0.1	Godin et al. (2016) [21]
Perfluorobuta-1,3-diene	685-63-2	100-2600	293	740	air	1	Acerboni et al. (2001) [5]
PFC-1114	116-14-3	100-2600	293	740	air	1	
PFC-41-12	678-26-2	700-1400	297			0.01	Bravo et al. (2010) [53]
PFC-51-14	355-42-0	700-1400	297			0.01	
PFC-71-18	307-34-6	700-1400	297			0.01	
Perfluoroethane	76-16-4	680-1380	253			0.03	Ballard et al. (2000) [136]
PFC-318	115-25-3	555-1460	253			0.03	
E-C10F18♦	60433-12-7	650-1500	296	700	air	0.5	Shine et al. (2005) [130]
PFC-91-18♦	306-94-5	0-2000	296	700	air	0.5	
Z-C10F18♦	60433-11-6	650-1500	296	700	air	0.5	
Perfluorobut-1-ene	357-26-6	600-2000	296	700	N ₂	0.25	Young et al. (2009) [69]

10 † A summary of temperature and pressure sets and experimental conditions is given in Ref. [55].

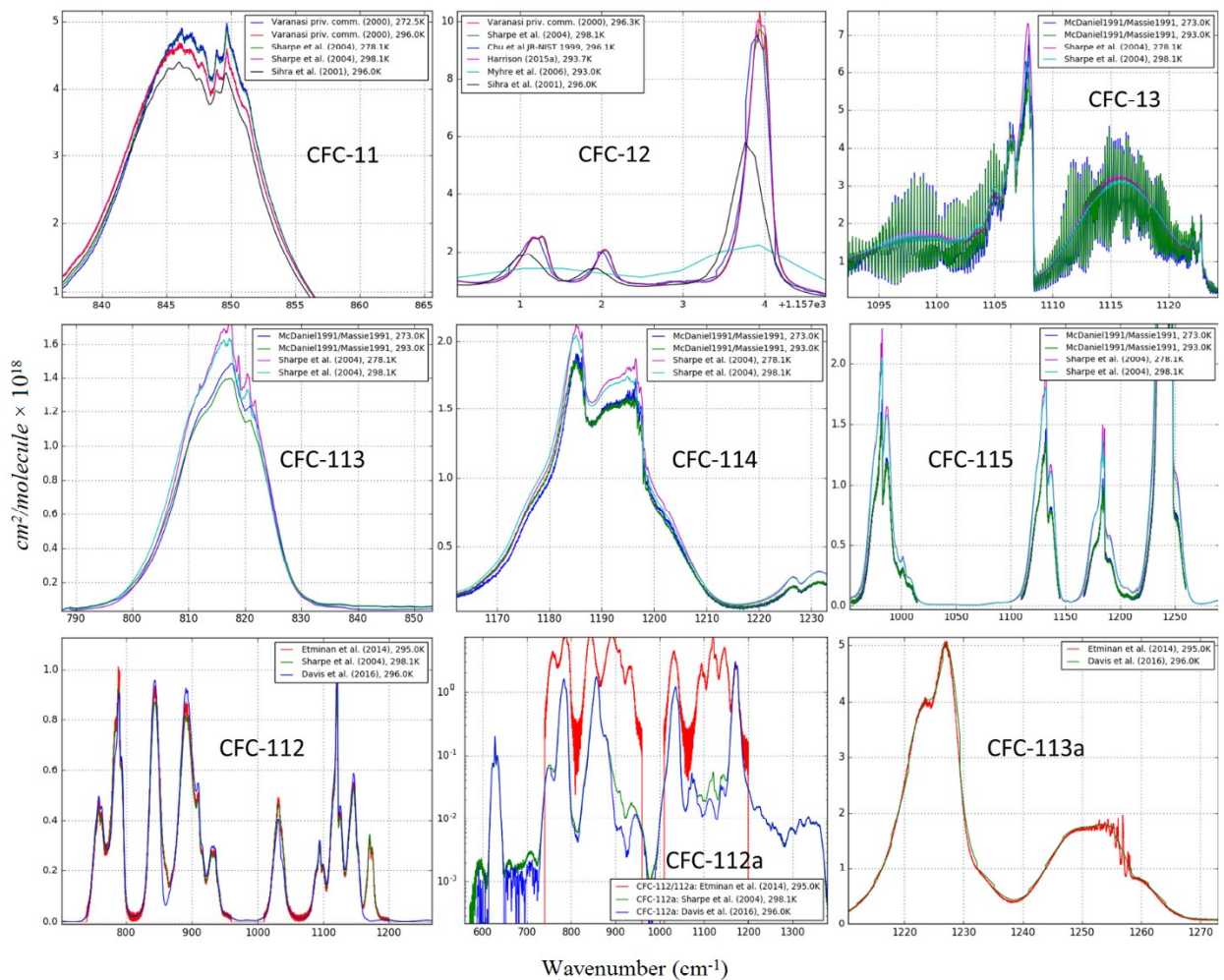
11 ♦ Supplemental folder

12 3.10. Chlorofluorocarbons

13 CFCs were widely-used in industrial and domestic applications before concerns about their
 14 role in ozone depletion led to a phase-out of their production and consumption via the Montreal
 15 Protocol. They are generally long-lived compounds (atmospheric lifetimes from decades to
 16 centuries [1]) which are mostly removed by either UV photolysis or reaction with O(¹D) in the
 17 stratosphere. Despite the production phase-out, CFCs are still the most abundant in the
 18 atmosphere of all the molecules considered in this paper due to their long atmospheric lifetimes
 19 and large historic emissions. Atmospheric concentrations of CFC-11 and CFC-12 are in excess
 20 of 200 ppt and 500 ppt, and are falling slowly from their peaks in about 1994 and 2003

1 respectively [94]. The rate of decline of CFC-11 atmospheric concentration has slowed down,
 2 which presumably reflects an increase in emissions [138]. Some CFCs (CFC-112, CFC-112a
 3 and CFC-113a) have recently been detected in the atmosphere [125], with CFC-113a
 4 concentrations increasing [139], the only known CFC to be doing so, now with typical values of
 5 0.7 ppt. The 2011 radiative forcing due to CFCs is about 0.26 W m^{-2} , about 14% of that due to
 6 CO_2 [3]. Their contribution to ozone depletion is measured via the Equivalent Effective
 7 Stratospheric Chlorine (EESC) metric [115]. Recent data (2017) [107] for the mid-latitude
 8 EESC indicate that the CFCs contribute about 40% of the total EESC.

9 For many of the CFCs considered in this paper, previous HITRAN issues contain high-quality
 10 cross-sections. In HITRAN2016 some datasets, such as CFC-12, have been improved by
 11 integrating recent measurements [26], and for some of the molecules, such as CFC-114a, CFC-
 12 112, CFC-112a, CFC-113a, new room temperature data have been included. The full summary
 13 of CFCs included in HITRAN*Online* is given in [Table 21](#) and [Table 22](#). The next subsections
 14 provide comparisons for several CFCs where cross-evaluation of different sources has been
 15 made (see [Figure 22](#)).



1 **Figure 22.** Cross-section comparison for selected CFC bands. In each group, the comparison
2 was taken between the datasets with the closest temperature and pressure conditions. The CFC-
3 112/112a spectra from Etminan et al. [43] is scaled by 9 to visually match the isolated CFC-112a
4 feature at 1170 cm^{-1} with other sources on the plot.

5 **3.10.1. CFC-11**

6 CFC-11 (trichlorofluoromethane, CCl_3F) has been included in the cross-section part of
7 HITRAN in previous issues. The air-broadened dataset measured by Varanasi et al. was included
8 in HITRAN2000 [133] and covers a wide range of temperatures and pressures (190-296 K and
9 7.5-760 Torr). Datasets from two new sources, PNNL [13] and Sihra et al. [61], have been
10 considered in the current evaluation. Data from PNNL were provided at standard temperatures
11 (278.15, 298.15, and 323.15 K) and at atmospheric pressure (760 Torr) broadened by nitrogen.
12 The spectrum from Sihra et al. was measured only at room temperature [61]. Comparing the
13 intensities, integrated in two regions, $810\text{-}880\text{ cm}^{-1}$ and $1050\text{-}1120\text{ cm}^{-1}$, we found an excellent
14 agreement between Varanasi and PNNL data for 273K and 296K ($\sim 0.7\%$ for both temperatures
15 and ranges). The data from Sihra et al. agree with the rest of the sources at the level of $\sim 8\%$
16 which is within the combined experimental accuracies. Data from PNNL include cross-sections
17 at higher temperature (323.15 K) which is not available in the set of Varanasi et al. To facilitate
18 the extrapolation, and taking into account very good agreement between these two sources, the
19 PNNL data have been included into the HITRAN database and included in *HITRANonline*. The
20 data from Sihra et al. were omitted, while the data provided by Varanasi et al. remain available
21 through the web interface.

22 **3.10.2. CFC-12**

23 CFC-12 (dichlorodifluoromethane, CCl_2F_2) is the most abundant CFC in the Earth's
24 atmosphere and the single largest contributor to the atmospheric loading of chlorine. A very
25 strong greenhouse gas with a 100-yr GWP of 10200 [3], a strong ODS with an ozone depletion
26 potential (ODP) of 0.73 [115], and a long atmospheric lifetime of 100 years [3] make this,
27 unsurprisingly, a molecule of much interest to the atmospheric monitoring community,
28 particularly as its decline, due to the success of the Montreal Protocol, is linked to stratospheric
29 ozone recovery.

30 A number of IR remote-sensing instruments have taken atmospheric measurements of CFC-
31 12 over the years, including ATMOS (Atmospheric Trace MOlecule Spectroscopy) [120],
32 MIPAS (Michelson Interferometer for Passive Atmospheric Sounding) [140], and the ACE-FTS
33 [122]. The CFC-12 cross-section dataset in previous HITRAN compilations was provided by
34 Varanasi and Nemtchinov [133]. For this HITRAN 2016 compilation, we provide a new and

1 improved cross-section dataset from Harrison [26] over the spectral range 800 – 1270 cm⁻¹,
2 determined from spectra recorded using a high-resolution Fourier transform spectrometer
3 (Bruker IFS 125HR) and a 26-cm-pathlength cell. Spectra of CFC-12 / dry synthetic air
4 mixtures were recorded at resolutions between 0.01 and 0.03 cm⁻¹ over a range of temperatures
5 and pressures (7.5 – 761 Torr and 190 – 294 K) appropriate for atmospheric conditions.

6 CFC-12 spectra from Harrison et al., Chu et al. (NIST IR database) [17], Sharpe et al. (PNNL
7 database) [13], Sihra et al. [61], and Myhre et al. [141] are shown in **Figure 22**. The PNNL
8 spectrum has the highest signal-to-noise ratio. The agreement between all sources except Sihra et
9 al. is very good and discrepancies do not exceed 2.25% for the integrated intensities for room
10 temperature (two regions were used in the comparison: 850-950 cm⁻¹ and 1050-1200 cm⁻¹). It
11 should be noted that, for spectra from Harrison et al., intensity was calibrated using the PNNL
12 spectra, thus the best agreement is obviously between those two sources. The 296K spectrum
13 from Sihra et al. disagrees with the rest of the data by ~12% and has a slight shift in
14 wavenumber. Besides the new data from Harrison [26], which were included in *HITRANonline*,
15 the spectra from PNNL have been placed in the supplemental folder. In the HITRAN2016 issue,
16 the data from Varanasi et al. for this molecule have been placed into the supplemental folder.
17 Data from the other sources were omitted.

18 **3.10.3. CFC-13**

19 Temperature-dependent infrared absorption cross-sections for CFC-13
20 (chlorotrifluoromethane, CClF₃) for previous HITRAN editions were taken from McDaniel et al.
21 [44] and Massie et al. [45]. These cross-sections were recorded self-broadened for 203, 213, 233,
22 253, 273, and 293K, and split into three regions (I: 765-805, II: 1065-1140, and III: 1170-1235
23 cm⁻¹). The data have been measured with a total estimated error of 15% [44]. In Ref. [44],
24 investigations were made to establish the effects of a foreign gas on the infrared spectra of CFC-
25 13 and related compounds (CFC-113, CFC-114, and CFC-115). “The resultant spectra did show
26 line broadening with increasing foreign gas pressure” [44]. For CFC-13, the temperature
27 dependence of the integrated intensities was pointed out in Ref. [44]. The authors supposed that
28 this could be due to the “anharmonicity in gas phase polyatomic molecules” [44].

29 More recent data for CFC-13 were provided by PNNL [13] at 278.15, 298.15, and 323.15 K
30 and atmospheric pressure. Here we have performed comparisons of the integrated intensities for
31 regions I, II, and III between data from Refs. [44,45] and PNNL. The integrated intensities agree
32 on the order of 5% for region I for the weakest absorption, and of 15% for regions II and III.
33 Also, it can be seen from the visual comparison (**Figure 22**), that the PNNL data are
34 insufficiently resolved.

1 The PNNL spectra for CFC-13 contain additional spectral features. They have been placed in
2 the supplemental folder to encourage using the data with higher resolution, although the user
3 should account for the most appropriate temperature, pressure, and broadening conditions to pick
4 the right dataset.

5 **3.10.4. CFC-113**

6 For CFC-113 (1,1,2-Trichlorotrifluoroethane, $\text{CCl}_2\text{FCClF}_2$) a comparison of the data from
7 Refs. [44,45] with PNNL [13] was performed in two spectral regions (I: $780.5\text{-}1000\text{ cm}^{-1}$, and II:
8 $1005.5\text{-}1232\text{ cm}^{-1}$), which correspond to the splitting of the data from Refs. [44,45]. For the first
9 region, the discrepancy for integrated intensities is 5-7%, while for the second region it goes up
10 to 12.5% which is within the uncertainty range for Refs. [44,45]. Visual comparison (see [Figure](#)
11 [22](#)) shows some minor discrepancies in the band heads. The spectrum at 203 K from Ref. [44]
12 features an important increase of its integrated intensities (15-19% relative to its values at 213 K)
13 and should be used with caution. The resolution value for CFC-113 data from Refs. [44,45] have
14 been corrected to 1.0 cm^{-1} in the HITRAN database according to Ref. [45]. A more recent dataset
15 has been introduced by Le Bris et al. [32]. For the first region, the discrepancy for integrated
16 intensities is 3% with Ref. [44] and 5% with PNNL. For the second region, the integrated
17 intensities differ by 6% relative to Ref. [44] and 4% relatively to PNNL. At lower temperatures,
18 Refs. [32] and [44] are in agreement for comparable temperatures. However, the integrated
19 cross-sections of Le Bris et al. do not show the upward trend with decreasing temperature
20 observed in the data from McDaniels et al.

21 It should be mentioned that the data from PNNL includes additional spectral features in the
22 range below 780 cm^{-1} and above 1230 cm^{-1} . The cross-sections from Refs. [13,32] have been
23 included in HITRAN*Online* (latter source as a post-release update), while the data from Refs.
24 [44,45] were relegated to the supplemental folder

25 **3.10.5. CFC-114**

26 The data for CFC-114 (1,2-dichlorotetrafluoroethane, $\text{CClF}_2\text{CClF}_2$) from Refs. [44,45] were
27 split into four regions (I: $815\text{-}860\text{ cm}^{-1}$, II: $870\text{-}960\text{ cm}^{-1}$, III: $1030\text{-}1067\text{ cm}^{-1}$, and IV: 1095-
28 1285 cm^{-1}) for each temperature set. The PNNL data agree well with the old data from HITRAN
29 for all regions (2-5% for I and II, and 9-12% in III and IV), which is within the experimental
30 uncertainty of the old data (15%). The PNNL dataset has additional spectral features below 815
31 cm^{-1} and above 1285 cm^{-1} , and has been included in HITRAN*Online*.

1 3.10.6. CFC-115

2 For CFC-115 (chloropentafluoroethane, CClF_2CF_3), the HITRAN data from Refs. [44,45] are
3 split into three regions (I: 955-1015 cm^{-1} , II: 1110-1145 cm^{-1} , and III: 1167-1260 cm^{-1}). For all
4 regions, the integrated intensity differs by more than 30 percent. The discrepancy can possibly be
5 due to different concentrations (zero pressure limit in McDaniel et al [115], 1 ppm-meter for
6 PNNL). The PNNL spectrum is less resolved than that from McDaniel et al. which has 0.03 cm^{-1}
7 resolution. Since PNNL provides the data with better accuracy, it was decided to include it into
8 the supplemental folder. For future HITRAN editions, it should be considered whether to scale
9 the data from Refs. [44,45] to match the PNNL integrated intensities.

10 3.10.7. CFC-112

11 The CFC-112 (1,2-difluorotetrachloroethane, $\text{C}_2\text{Cl}_4\text{F}_2$) spectra were included into HITRAN
12 for the first time. Cross-sections for this molecule were published by Sharpe et al. [13], and more
13 recently by Davis et al. [80] and Etminan et al. [43]. It should be noted that Etminan et al.
14 provide the spectra of a mixture of CFC-112/CFC-112a. All sources have a good total accuracy
15 (2.1% for PNNL, 2.7% for Etminan et al., and 4% for Davis et al.). The intensities, integrated in
16 the 700-1158 cm^{-1} range, also give good agreement (within 4%). However, the comparison
17 shows that the PNNL data also represent a mixture of CFC-112 and CFC-112a, but not a pure
18 CFC-112. This can be demonstrated in [Figure 22](#), where significant spectral features near 1200
19 cm^{-1} and 600 cm^{-1} can be seen both in PNNL and Etminan et al., and are completely absent in
20 Davis et al. In addition, the PNNL data are, as stated in the original description, “contaminated
21 with R-141B [0.481%] and corrected by rescaling and subtraction on the composite spectrum”.
22 However, since the PNNL data have more complete temperature coverage than both Davis et al.
23 and Etminan et al., they have been placed into the supplemental folder. The data from Davis et
24 al., as containing the pure CFC-112 spectrum, have been included in HITRAN*Online*. It should
25 also be noted that for CFC-112 and related molecules (CFC-112a, CFC-113a, CFC-114a) from
26 Ref. [80], the measured infrared spectra were independent of total pressure over the pressure
27 range 20–250 Torr with He as a bath gas.

28 3.10.8. CFC-112a

29 For CFC-112a (2,2-difluorotetrachloroethane, $\text{C}_2\text{Cl}_4\text{F}_2$), the data from PNNL [13] and Davis
30 et al. [80] have been compared. In [Figure 22](#), we plot the spectra from these sources (the PNNL
31 one is for 278.15 K) and overlay them on the mixture spectra from Etminan et al. [43]. Excellent
32 agreement (<1%) between PNNL and Davis et al. can be found for the intensities, integrated in
33 the 700-1300 cm^{-1} spectral region. The PNNL data were supplied at three standard temperatures

(278.15, 298.15, and 323.15 K), and were included in HITRAN*Nonline*. Data from Davis are given only at 296K and lower resolution (however sufficient for the purposes of Ref. [80]). As seen on the log scale comparison (**Figure 22**), the PNNL data have very minor contaminants, which are absent in the Davis et al. spectra. The data from the latter source have been placed into the supplemental folder as the one having lower resolution and only 296K temperature.

3.10.9. CFC-113a

The data for CFC-113a (1,1,1-trichlorotrifluoroethane, C₂Cl₃F₃) have been added to the HITRAN database for the first time. The comparison consisted of two sources of data: Davis et al. [80] and Etminan et al. [43]. Both sources provide room-temperature spectra at low pressures (zero-pressure limit for Davis et al., and 2.7 Torr for Etminan et al.). The description of the spectroscopic assignment for the strongest features of this molecule is given in Ref. [43]. The integrated intensity agreement between these two sources is good and goes from 2.5% for the strongest feature at 1250 cm⁻¹ to 8.2% for the weakest one at 910 cm⁻¹. The data from Etminan et al. have been included in HITRAN*Nonline*, while data from Davis et al. have been placed in supplemental folder. However, the latter is smoother (although with less stated total accuracy and resolution), and displays more of the weaker features.

Table 21. PNNL-DOE data for chlorofluorocarbons

Molecule	CAS	Range, cm ⁻¹
CFC-11	75-69-4	570-6500
CFC-113	76-13-1	620-5000
CFC-114	76-14-2	600-5000
CFC-112a	76-11-9	570-6500
CFC-112	76-12-0	550-6500
C-56 †	77-47-4	575-6500

Spectra from Sharpe et al. [13] broadened by N₂. Spectra were recorded at 278.15 K, 298.15 K, and 323.15 K temperatures (except for hexachlorocyclopentadiene), 760 Torr total pressure, and 0.112 cm⁻¹ resolution.

† Spectra for C-56 (hexachlorocyclopentadiene) were recorded at 323.15 K

Table 22. More recent data on chlorofluorocarbons.

Molecule	CAS	Range (cm ⁻¹)	T (K)	P (Torr)	Brd	Resolution (cm ⁻¹)	Source
CFC-12	75-71-8	800-1270	189.5-294.2	7.492-761	air	0.01-0.03	Harrison (2015) [26]†
CFC-113	76-13-1	600-1250	223-283			0.05	Le Bris et al (2011) [32]†
CFC-113a	354-58-5	690-1730	295	2.7		0.01	Etminan et al. (2014) [43]
CFC-114a	374-07-2	550-1397	296			1	Davis et al. (2016) [80]

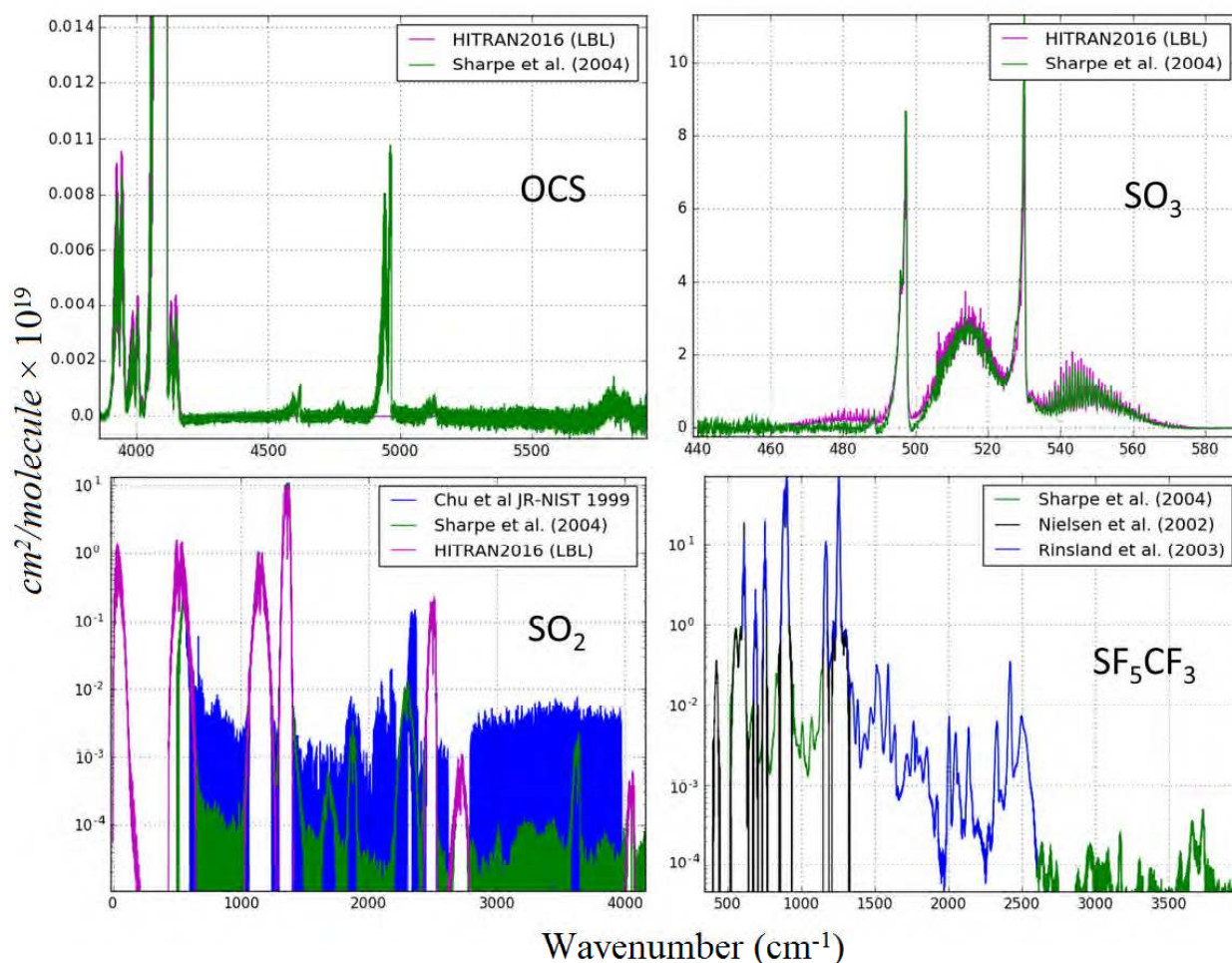
† A detailed summary on temperature and pressure sets and experimental conditions is available in the corresponding references.

1 3.11 Sulfur-containing species

2 The sulfur-containing species in this edition include compounds that are released into the
3 atmosphere due to industrial activities. Many species from this category strongly absorb in the
4 atmospheric window (800-1800 cm^{-1}). A number of species from this category have large
5 lifetimes, such as trifluoromethyl sulfur pentafluoride (SF_5CF_3) and sulfur hexafluoride.
6 Although these molecules so far have minor abundances in the terrestrial atmosphere, they can
7 be considered as species of potential interest in climate applications.

8 A number of sulfur-containing species existed before in previous HITRAN issues. For
9 instance, the line-by-line part of HITRAN contains transitions for carbonyl sulfide (OCS), sulfur
10 dioxide (SO_2), sulfur trioxide (SO_3), and sulfur hexafluoride (SF_6). The discussion of the cross-
11 sections of SF_6 is given in the Section 3.9.1 as it is also categorized as a fully fluorinated species.

12 In the HITRAN2016 update, several new molecules were added belonging to such sub-classes
13 as sulfates, sulfites, and mercaptans. The set of cross-section available through HITRAN*online*
14 incorporates the PNNL spectral data. In addition, the spectra provided in Hodnebrog et al. [1]
15 have been included in the supplemental folder in the case where they had additional features in
16 the lower wavenumber range. The summary of the sulfur-containing molecules available via
17 HITRAN*online* interface is given in **Table 23**. Below we give comparisons for each molecule
18 for which the data from different sources were available (including the line-by-line part of
19 HITRAN).



1

2 **Figure 23.** Cross-section comparison for selected bands of carbonyl sulfide (OCS), sulfur
 3 dioxide (SO₂), sulfur trioxide (SO₃), and trifluoromethyl sulfur pentafluoride (CF₅CF₃). The
 4 cross-sections generated from the line-by-line data from HITRAN2016 [2] are highlighted in
 5 magenta. Note that the spectral ranges of the line-by-line data for some of the molecules in
 6 HITRAN are larger than ones covered in this figure.

7 3.11.1. Carbonyl sulfide

8 Carbonyl sulfide (OCS) is an important molecule of astrophysical and atmospheric interest.

9 The HITRAN database has high-quality line-by-line data for this molecule spanning the wide
 10 wavenumber range from microwave to near-infrared [2]. The infrared spectra for this molecule
 11 have a strong feature centered near 2000 cm⁻¹; the rest of the bands are relatively weak. For this
 12 molecule, only the data from PNNL are presented in the cross-section format for 278.15, 298.15,
 13 and 323.15 K temperature and 760 Torr pressure. The comparison of these data with cross-
 14 sections, generated from the OCS line list from HITRAN2016, is displayed in **Figure 23** and
 15 shows generally good agreement of the spectral band shapes and integrated intensities. For bands
 16 in the region above 1900 cm⁻¹ containing the strong spectral feature, the integrated intensities
 17 agree within 2.4-3.7%, while the agreement is worse (~9%) in the lower wavenumber region
 18 (800-1100 cm⁻¹). It should be noted that the group of bands near 520 cm⁻¹ in PNNL suffers from
 19 distortion caused by the fall-off of the detector response. Second, the PNNL data contains weak

1 but noticeable spectral features which are lacking in the line-by-line spectra (for instance, the
2 ones centered near 3700, 5000, and 6000 cm^{-1}). We consider the line-by-line data preferable over
3 the full spectral range, thus relegating the PNNL data to the supplemental folder.

4 **3.11.2. Sulfur dioxide**

5 The line-by-line data for sulfur dioxide (SO_2) exist in HITRAN2016 in a wide spectral range,
6 from the microwave up to 4093 cm^{-1} . HITRAN has included SO_2 UV cross-sections since 2012.
7 The cross-sectional data considered in this comparison were provided by PNNL and NIST, with
8 the spectral coverage of all datasets presented in **Figure 23**. It should be noted that both PNNL
9 and NIST data near 550 cm^{-1} have an intensity decline due to the nonlinear detector response
10 (MCT is not sensitive to light at longer wavelengths). Again, despite the good agreement
11 between the integrated intensities, all three sources (PNNL, NIST and HITRAN2016 line-by-
12 line) lie within 4% for common features near 1300 cm^{-1} and 2500 cm^{-1} . The NIST spectra
13 contain additional spectral bands probably due to other molecules, as well as the baseline slightly
14 shifted to the negative values. When comparing the line-by-line data and absorption cross-
15 sections in a given spectral interval, the former is usually favored for the reasons explained in
16 Section 2 (unless the line-by-line parameters have low accuracy). For SO_2 , the line-by-line data
17 have most of the bands in the IR region, while PNNL spectra have additional weak features (e.g.
18 at 1750 cm^{-1} , 1850 cm^{-1} , 2300 cm^{-1} , and 3600 cm^{-1}) which can be useful in some applications.
19 Thus, we have added the PNNL data to the supplementary folder.

20 **3.11.3. Sulfur trioxide**

21 Sulfur trioxide (SO_3) is an atmospherically-relevant molecule produced during the oxidation
22 of SO_2 . The HITRAN2016 edition provides an updated line list for the principal isotopologue of
23 SO_3 [2]. The absorption cross-sections come only from PNNL which has only one cross-section
24 set at room temperature. The agreement between the cross-sectional and line-by-line data for
25 SO_3 is worse than for OCS. Similar to the case of carbonyl sulfide, the PNNL spectra for SO_3
26 distort at the beginning of the wavenumber range ($\sim 450 \text{ cm}^{-1}$). The corresponding panel in
27 **Figure 23** shows the comparison of the band feature around 500 cm^{-1} which was partly distorted
28 in the PNNL spectra due to the spectrometer response fall-off. Surprisingly, the best agreement
29 of the integrated intensities with the line-by-line data can be observed for this low-wavenumber
30 feature ($\sim 9\%$), while the disagreement is around 17% for the strongest band near 1400 cm^{-1} ,
31 which corresponds well to the HITRAN uncertainty codes for this region. The PNNL spectra
32 contain an additional feature near 2450 cm^{-1} which is absent in the line-by-line. Again, as for
33 OCS, to encourage the user to take advantage of the line-by-line data (which offers more

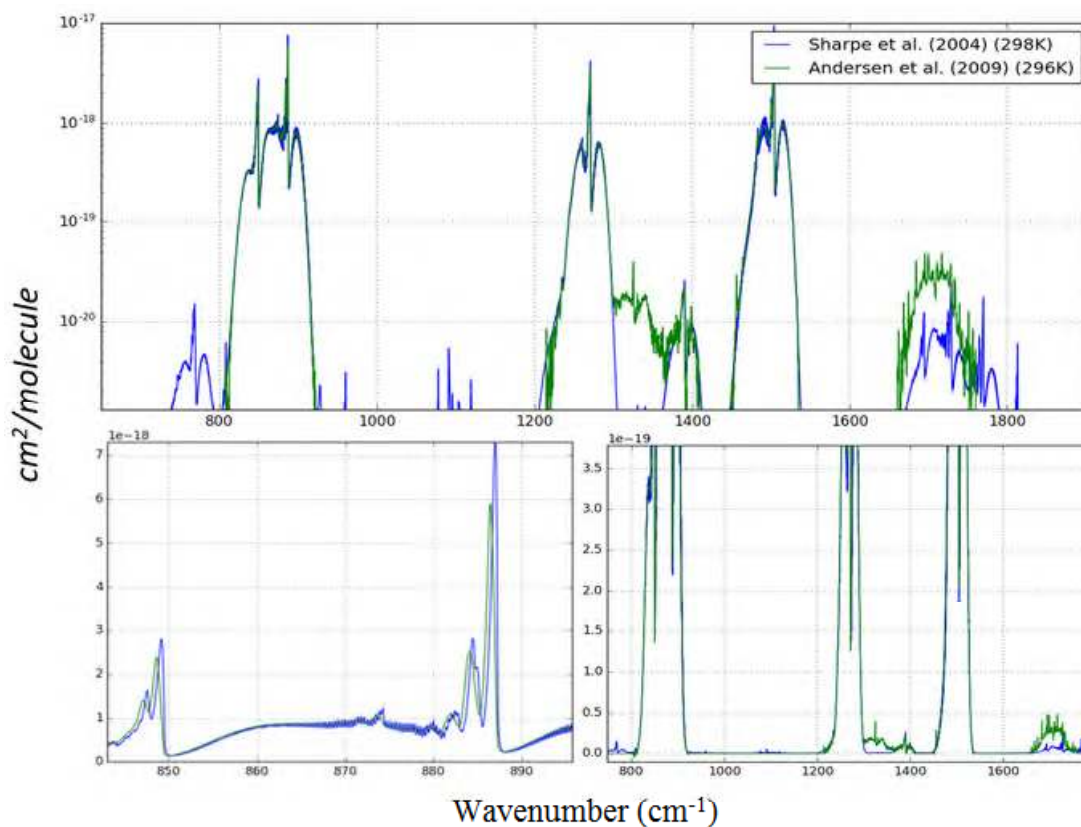
1 flexibility), the PNNL spectrum for 296 K and 760 Torr has been placed in the supplemental
2 folder.

3 **3.11.4. Trifluoromethyl sulfur pentafluoride**

4 Trifluoromethyl sulfur pentafluoride (SF_5CF_3) is a long-lived greenhouse gas with an 800
5 year lifetime and 17400 100-yr GWP [3]. For this molecule, the HITRAN database included the
6 air-broadened cross-sections for 30 temperatures (213-323K) and atmospheric pressure provided
7 by Rinsland et al. [142]. These data coincide with the spectra from Sharpe et al. [13] for 278K,
8 298K, and 323K. Thus the PNNL spectra were omitted from inclusion into the HITRAN2016
9 issue for this molecule. The third dataset in this comparison is Nielsen et al. [143]. The data from
10 this source were measured at Ford at 296K. The authors of this paper remark that “because the
11 spectral line shape was independent of pressure and resolution, the reported spectral data are
12 recorded without air as diluent” [143]. For this reason, we did not fill the broadener and pressure
13 fields, but mention that these spectra can be used at atmospheric pressure as well. Data from
14 Nielsen et al. are in the good agreement with PNNL and have additional features at lower
15 wavenumber range. These data have been added to the supplemental folder. The log-scale
16 comparison of the datasets is presented in the **Figure 23**.

17 **3.11.5. Sulfuryl fluoride**

18 Sulfuryl fluoride (SO_2F_2) is a greenhouse gas with a lifetime of 36 years and a 4090 100-year
19 GWP [1]. It has three strong spectral features in the 800-1800 cm^{-1} spectral window. Data for
20 this molecule are new to HITRAN, and two sources are considered in the comparison given in
21 the **Figure 24**. The data measured at atmospheric pressure have been provided by PNNL at three
22 standard temperatures (278.15, 298.15, and 323.15 K), and by Andersen et al. [1,144]. Despite
23 good agreement in the integrated intensities (2.42%), there is a significant discrepancy ($\sim 1.1 \text{ cm}^{-1}$)
24 on the wavenumber scale which can point to differences in the wavenumber calibration
25 between these two sources. The spectrum from Andersen et al. contains additional absorption in
26 the 1300-1700 cm^{-1} range (see the **Figure 24**) possibly due to contaminants. We have placed the
27 PNNL spectra in HITRAN*online* due to higher resolution and wider spectral coverage. The data
28 from Andersen et al. have been omitted.



1

2 **Figure 24.** Comparison of sulfuranyl fluoride (SO_2F_2) cross-sections at room temperature and
 3 atmospheric pressure.

4 **Table 23.** PNNL-DOE data for sulfur-containing species

Molecule	CAS	Range, cm^{-1}
Carbon disulfide	75-15-0	600-6500
Dimethyl disulfide	624-92-0	600-6500
Dimethyl sulfate †	77-78-1	525-6500
Dimethyl sulfide	75-18-3	600-6500
Dimethyl sulfoxide †	67-68-5	575-6500
Methyl isothiocyanate	556-61-6	600-6500
Methyl mercaptan	74-93-1	600-6500
Sulfuryl fluoride	2699-79-8	500-6500
Sulphur hexafluoride	2551-62-4	560-6500

5 Spectra from Sharpe et al. [13] broadened by N_2 . Spectra were recorded at 278.15 K, 298.15 K,
 6 and 323.15 K temperatures (except for dimethyl sulfate and dimethyl sulfoxide), 760 Torr total
 7 pressure, and 0.112 cm^{-1} resolution.

8 † Spectra for dimethyl sulfate and dimethyl sulfoxide were recorded at 298.15 K and 323.15 K

9 3.12. Iodocarbons and hydroiodocarbons

10 For iodine containing species, previous editions of the HITRAN database included only HI
 11 (hydrogen iodide) in the line-by-line part. The HITRAN2016 update brings the iodocarbons and
 12 hydroiodocarbons into the database for the first time. We have included the three most
 13 atmospherically-important species from this category found in the PNNL [13] database: methyl
 14 iodide (CH_3I), diiodomethane (CH_2I_2), and ethyl iodide ($\text{C}_2\text{H}_5\text{I}$). A brief summary of the
 15 included molecules is given in **Table 24**.

1 **Table 24.** PNNL-DOE data for iodocarbons and hydroiodocarbons

Molecule	CAS	Range, cm ⁻¹
Diiodomethane†	75-11-6	530-7100
Ethyl iodide	75-03-6	600-6500
Methyl iodide	74-88-4	500-6900

2 Spectra from Sharpe et al. [13] broadened by N₂. Spectra were recorded at 278.15 K, 298.15 K,
3 and 323.15 K temperatures (except for Diiodomethane), 760 Torr total pressure, and 0.112 cm⁻¹
4 resolution.

5 † Spectra for Diiodomethane were recorded at 298.15 K and 323.15 K

6 **4. Searching the cross-sections with HITRANonline**

7 Access to the cross-section data from the HITRAN2016 edition is provided by the open
8 information system HITRANonline (<http://hitran.org>) [20], or by the HITRAN Application
9 Programming Interface [9].

10 HITRANonline is a new information system which provides access to the data from the
11 HITRAN [2] database. This system enables the user to select, download, and visualize the
12 spectral line-by-line and cross-section data. This system also allows access to collision induced
13 absorption, aerosol refractive indices, and high-temperature line lists from the HITEMP database
14 [19]. For line-by-line transitions, which are the most frequently used part of the HITRAN
15 database, a flexible approach to the data selection and representation is implemented: the user
16 can create custom ASCII and binary output formats, specify the parameters for output, browse
17 interactively the resulting line lists, impose search conditions (for example, for isotopologues,
18 bands, and wavenumber ranges), and generate the bibliography in different formats. Hill et al.
19 [20] provide a detailed description of the line-by-line part of HITRANonline. This
20 HITRANonline information system is recommended for users to access the HITRAN database
21 since it always provides the most up-to-date database version. In the following sections we
22 describe how the cross-section data can be found and downloaded using HITRANonline and
23 HAPI.

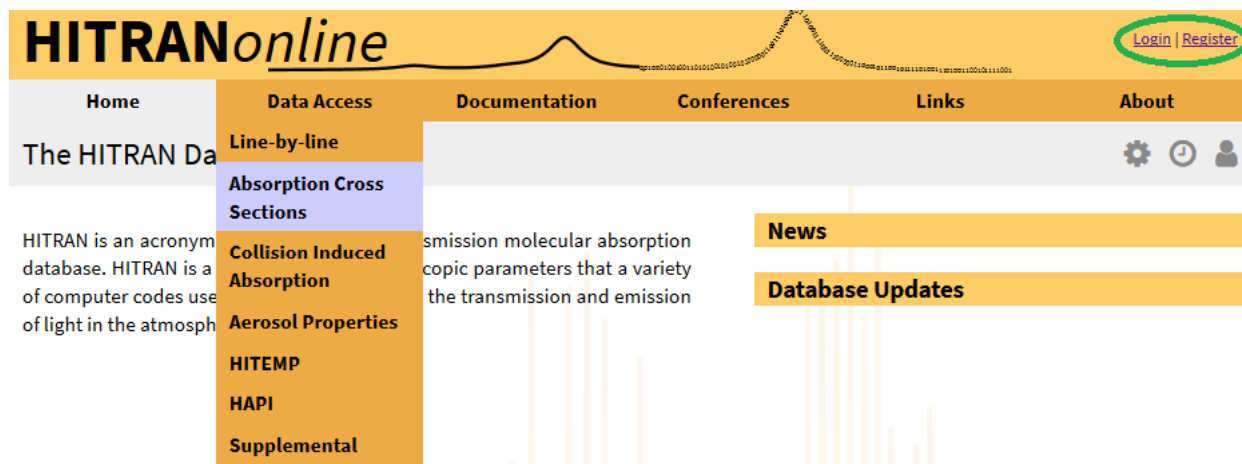
24 **4.1. Interactive data search**

25 To view and download the cross-section data, a user must be logged in on the internet
26 browser. For anonymous users, an automatic invitation for logging in will appear after requesting
27 certain cross-section temperature and pressure sets. To log in, the user should be registered in the
28 system. This can be done either by clicking on the “register” link on the main page (see **Figure**
29 **25**), or on the invitation page as mentioned above.

30 After registering, a user account should be activated. After successful activation of the
31 account, a user will get full access to all the data in the interactive part of the HITRANonline

1 system. The link to the cross-sections main search page is located in a menu, which can be
2 opened either by hovering a mouse over the “Data Access” menu item, or by clicking on it
3 (Figure 25).

4



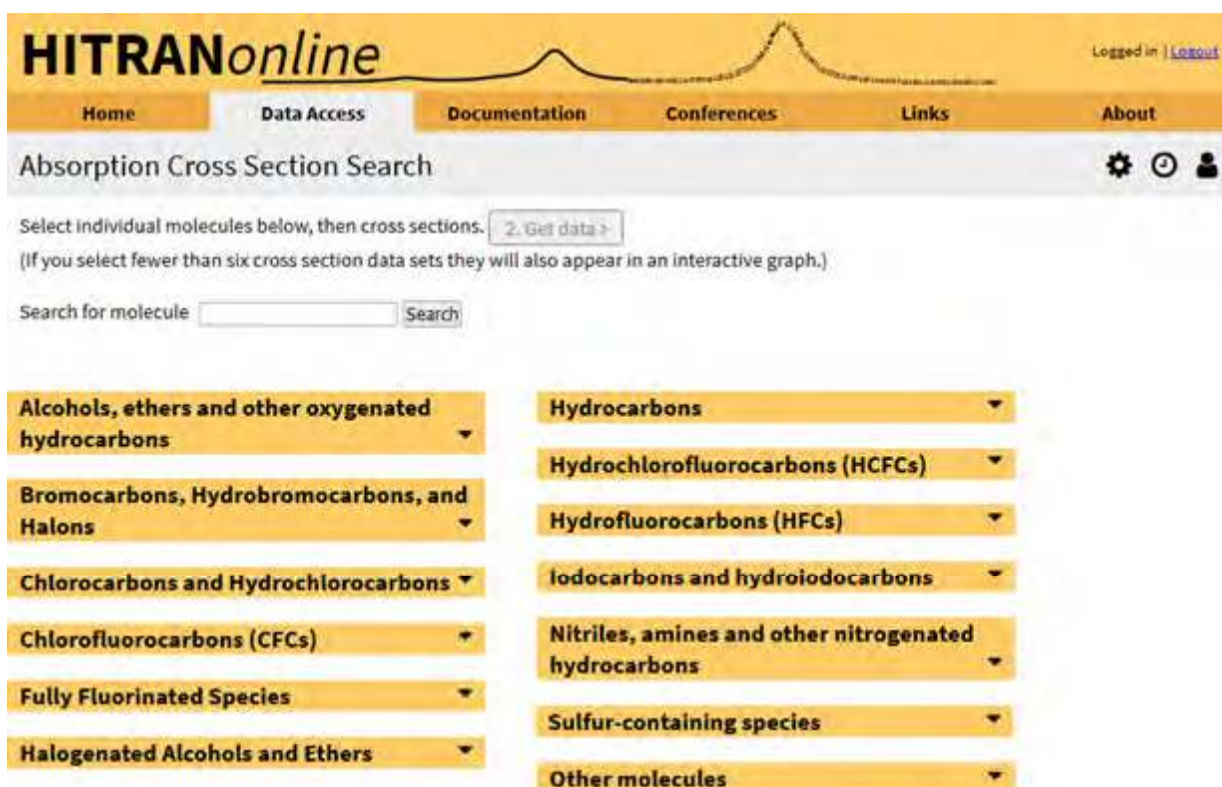
5

6 **Figure 25.** HITRANonline main page.

7 The cross-section search main page of HITRANonline is shown in Figure 25. As with the
8 initial version of this page, described in Hill et al. (2016) [20], the new version contains the
9 molecule search form.

10 Molecules can be searched by a number of *synonyms* or *aliases*, such as those given in Table
11 25. Apart from these names, other types of molecular identifiers will be added to
12 HITRANonline. Although the search is robust, the user should be aware of using multiple
13 synonyms to ensure that the molecule of interest is present in the database. Also, since in some
14 cases, for instance for isomers, the common molecular formulae are the same for different
15 species (e.g. for propene and cyclopropane the molecular formula is C₃H₆), it is advisable to use
16 more reliable aliases in the search, such as molecule identifiers (CAS, InChi, or InChiKey).

17



1
2 **Figure 26.** Molecular search and category browser.

3 **Table 25.** Molecule aliases available in HITRANonline.

TYPE	EXPLANATION	EXAMPLE
Name	Common molecule name	Dichlorodifluoromethane
Formula	Molecular formula not containing atomic weights.*	CCl2F2
Tradename or acronym	Tradename or conventional molecular acronym (such as defined by IUPAC†)	CFC-12
CAS	Chemical Abstract Service registry number	75-71-8
InChI	International chemical identifier (IUPAC)	InChI=1S/CCl2F2/c2-1(3,4)5
InChiKey	Machine-friendly hash string produced for the InChI	PXBRQCKWGAHEHS-UHFFFAOYSA-N

4 * Depends on the order of atoms in the molecule formula. † International Union of Physics and Chemistry

5 For the user's convenience, the molecules are grouped into a number of chemical categories.
6 The list of the categories is growing and an up-to-date list of the categories is available on the
7 HITRANonline site. Most categories were adopted from the Hodnebrog et al. halocarbon review
8 [1], but several new categories were added prior to the HITRAN2016 database release, such as:
9 hydrocarbons, alcohols, ethers and other oxygenated hydrocarbons, nitriles, amines and other

1 nitrogenated hydrocarbons, sulfur-containing species, and iodocarbons and hydroiodocarbons. In
2 the section devoted to the spectral data, we discussed a number of molecules from each category
3 in detail. Note that some molecules belong to several categories.

4 Clicking on a category name (**Figure 26**) gives a drop-down menu containing all molecules
5 from this category (**Figure 27**). In the dropdown list, general information on each molecule is
6 presented, such as spectral coverage (IR or UV), molecule common name, and ordinary formula.
7 Clicking on a molecule in the menu opens a panel in the right part of the screen containing a full
8 list of cross-sections temperature and pressure sets (T,p -sets). The supplemental folder contains
9 the alternate data in static text files.

10 Once the list of the T,p -sets for the chosen molecule is opened, a summary *metadata* on each
11 set is displayed. Such metadata includes a wavenumber range, temperature, pressure, resolution,
12 broadener, and the number of data points in each cross-section set. By checking the T,p -sets in
13 this list, the user defines which data should be downloaded and visualized. The source of the data
14 can be viewed by hovering a mouse over the corresponding T,p dataset.

15

HITRANonline Logged in | Logout

Home Data Access Documentation Conferences Links About

Absorption Cross Section Search

Select individual molecules below, then cross sections. [2. Get data >](#)
 (If you select fewer than six cross section data sets they will also appear in an interactive graph.)

Search for molecule

[Supplemental folder](#)

Alcohols, ethers and other oxygenated hydrocarbons

Bromocarbons, Hydrobromocarbons, and Halons

Chlorocarbons and Hydrochlorocarbons

Chlorofluorocarbons (CFCs)

IR	UV/Vis	Common Name	Formula
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-11	CCl ₃ F
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-13	CClF ₃
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-113	C ₂ Cl ₃ F ₃
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-114	C ₂ Cl ₂ F ₄
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-115	C ₂ ClF ₅
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-12	CCl₂F₂
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-113a	C ₂ Cl ₃ F ₃
<input checked="" type="checkbox"/>	<input type="checkbox"/>	CFC-114a	C ₂ Cl ₂ F ₄
<input checked="" type="checkbox"/>	<input type="checkbox"/>	Hexachlorocyclopentadiene	C ₅ Cl ₆

CFC-12: CCl₂F₂

IR cross sections						
<input type="checkbox"/>	ν range /cm ⁻¹	T/K	p/Torr	Resolution	npts	Broadener
<input checked="" type="checkbox"/>	800.0 - 1270.0	216.2	129.4	0.02 cm ⁻¹	187181	air
<input checked="" type="checkbox"/>	800.0 - 1270.0	189.5	170.4	0.03 cm ⁻¹	187181	air
<input checked="" type="checkbox"/>	800.0 - 1270.0	200.4	281.5	0.03 cm ⁻¹	187181	air
<input checked="" type="checkbox"/>	800.0 - 1270.0	200.6	170.4	0.03 cm ⁻¹	187181	air
<input checked="" type="checkbox"/>	800.0 - 1270.0	216.2	280.9	0.03 cm ⁻¹	187181	air
<input type="checkbox"/>	800.0 - 1270.0	216.2	379.5	0.03 cm ⁻¹	187181	air
<input type="checkbox"/>	800.0 - 1270.0	233.1	197.2	0.03 cm ⁻¹	187181	air
<input type="checkbox"/>	800.0 - 1270.0	233.2	380.8	0.03 cm ⁻¹	187181	air
<input type="checkbox"/>	800.0 - 1270.0	244.8	171.0	0.03 cm ⁻¹	187180	air
<input type="checkbox"/>	800.0 - 1270.0	244.8	444.8	0.03 cm ⁻¹	187180	air
<input type="checkbox"/>	800.0 - 1270.0	244.8	550.0	0.03 cm ⁻¹	187180	air
<input type="checkbox"/>	800.0 - 1270.0	289.0	189.8	0.03 cm ⁻¹	187180	air
<input type="checkbox"/>	800.0 - 1270.0	289.0	350.9	0.03 cm ⁻¹	187180	air

1
2 **Figure 27.** Cross-sections menu. Example of accessing CFC-12.

3 Clicking on the “Get data” link opens the search result page as shown in **Figure 28**. The page
 4 provides separate links to the HITRAN-formatted files for each T,p -set along with the single
 5 bibliography file. To simplify the download of a large number of cross-section sets they are
 6 compressed into a zip file. **Figure 28** displays the spectra from the requested T,p -sets. It is worth
 7 noting that the cross-sections displayed in the interactive plot are a downsampled version of the
 8 actual spectra stored in the database, since plotting the spectra in their original resolution can
 9 slow down or freeze the internet browser. The user should be aware of that when visualizing the
 10 subtle spectral features, because the “light” versions of the spectra were obtained by down-
 11 sampling of the original data.

CCl₂F₂

5 cross sections were found.

[CCl2F2_189.5K-170.4Torr_800.0-1270.0_00.xsc](#) [1.8 MB] Cross section file (HITRAN .xsc format)

[CCl2F2_200.4K-281.3Torr_800.0-1270.0_00.xsc](#) [1.8 MB] Cross section file (HITRAN .xsc format)

[CCl2F2_200.6K-170.4Torr_800.0-1270.0_00.xsc](#) [1.8 MB] Cross section file (HITRAN .xsc format)

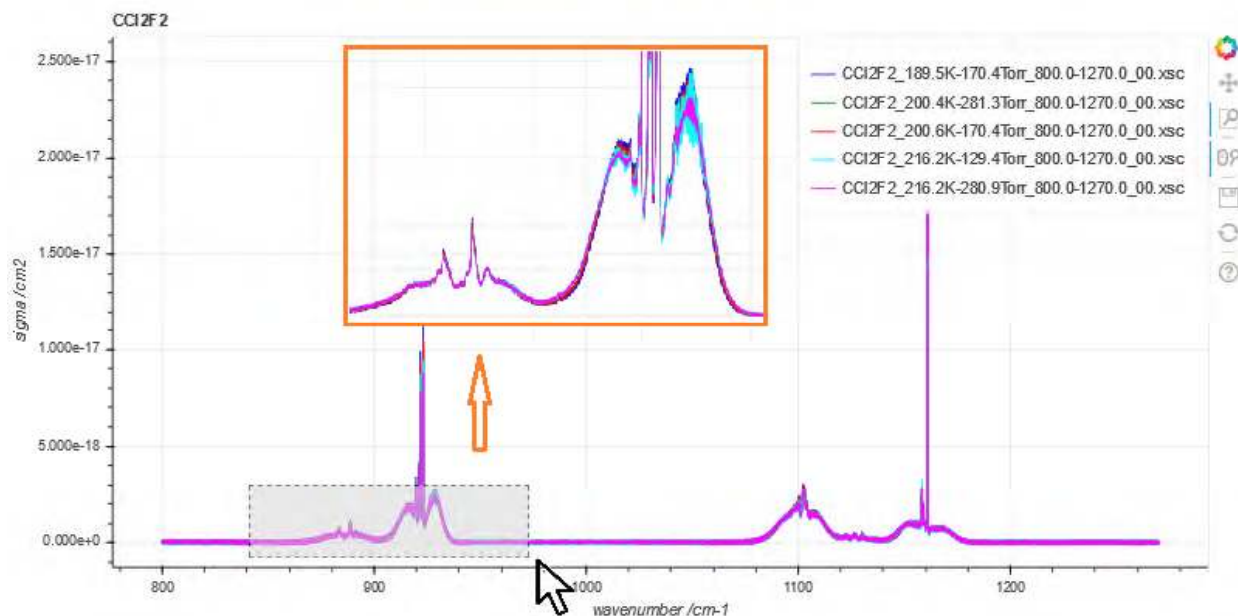
[CCl2F2_216.2K-129.4Torr_800.0-1270.0_00.xsc](#) [1.8 MB] Cross section file (HITRAN .xsc format)

[CCl2F2_216.2K-280.9Torr_800.0-1270.0_00.xsc](#) [1.8 MB] Cross section file (HITRAN .xsc format)

[5a1f781f.bib.html](#) [1.9 KB] List of sources (HTML format)

[5a1f781f.bib](#) [1.2 KB] List of sources (BibTeX format)

ZIP archive: [5a1f781f.zip](#)



1
2 **Figure 28.** Interactive spectra view.

3 A sample of the HITRAN-formatted file is shown in **Figure 29**. The plain text file in this
4 format (an extension .xsc has been used) contains header and data. The header incorporates the
5 cross-section T,p -set parameters which were explained in **Table 3**.

```
CCl2F2 799.9948 1270.0092 187181 216.2 129.4 2.364E-170.020 CFC-12 air 54
0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 2.421E-21 4.848E-21
6.741E-21 8.211E-21 9.483E-21 1.068E-20 1.167E-20 1.218E-20 1.193E-20 1.094E-20 9.616E-21 8.628E-21
8.625E-21 9.856E-21 1.194E-20 1.394E-20 1.465E-20 1.318E-20 9.315E-21 3.712E-21 0.000E+00 0.000E+00
0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 2.238E-21 3.642E-21 3.958E-21 3.620E-21 3.046E-21
2.431E-21 1.706E-21 6.623E-22 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00
0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00
0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 2.676E-21 5.600E-21 7.087E-21 6.969E-21
5.534E-21 3.368E-21 1.114E-21 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00
0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 1.163E-21 2.621E-21 3.499E-21
4.078E-21 4.613E-21 5.186E-21 5.670E-21 5.817E-21 5.421E-21 4.484E-21 3.275E-21 2.253E-21 1.877E-21
2.375E-21 3.613E-21 5.119E-21 6.285E-21 6.647E-21 6.126E-21 5.078E-21 4.141E-21 3.907E-21 4.586E-21
5.831E-21 6.825E-21 6.617E-21 4.576E-21 7.456E-22 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00
0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00 0.000E+00
```

6
7 **Figure 29.** Example of cross-section in HITRAN format (.xsc)

8 4.2. Sources and bibliography

9 The references for the source papers for each cross-section T,p -set are given in multiple
10 formats. The user is able to download the bibliography for the selected cross-sections in HTML
11 and BibTeX formats, as shown in **Figure 30** and **Figure 31**. In the generated bibliography file, all
12 references are listed along with the initial reference for the current HITRAN issue. For each

1 reference, its local database ID number is given in that file. These ID numbers set up the
2 correspondence between cross-section T,p -sets and the references given in the file headers
3 (Figure 29).

The latest full release of the HITRAN database is described in:
I.E. Gordon, L.S. Rothman, C. Hill, R.V. Kochanov, Y. Tan, P.F. Bernath, M. Birk, V. Boudon, A. Campargue, K.V. Chance, B.J. Drouin, J.-M. Flaud, R.R. Gamache, J.T. Hodges, D. Jacquemart, V.I. Perevalov, A. Perrin, K.P. Shine, M.-A.H. Smith, J. Tennyson, G.C. Toon, H. Tran, V.G. Tyuterev, A. Barbe, A.G. Csaszar, V.M. Devi, T. Furtenbacher, J.J. Harrison, J.-M. Hartmann, A. Jolly, T.J. Johnson, T. Karman, I. Kleiner, A.A. Kyuberis, J. Loos, O.M. Lyulin, S.T. Massie, S.N. Mikhailenko, N. Moazzen-Ahmadi, H.S.P. Muller, O.V. Naumenko, A.V. Nikitin, O.L. Polyansky, M. Rey, M. Rotger, S.W. Sharpe, K. Sung, E. Starikova, S.A. Tashkun, J. Vander Auwera, G. Wagner, J. Wilzewski, P. Wcislo, S. Yu, E.J. Zak, "The HITRAN2016 Molecular Spectroscopic Database", *Journal of Quantitative Spectroscopy and Radiative Transfer*, Available online 5 July 2017, ISSN 0022-4073, DOI 10.1016/j.jqsrt.2017.06.038. [[link to article](#)]

References by integer ID in data file
1046. J.J. Harrison, "New and improved infrared absorption cross sections for dichlorodifluoromethane (CFC-12)", *Atmospheric Measurement Techniques* 8, 3197-3207 (2015). [[link to article](#)] [[ADS](#)]

4
5 **Figure 30.** References view (HTML)

6

```
|@This is the latest full HITRAN paper
|@article{HITRAN2016,
|author = {I.E. Gordon, L.S. Rothman, C. Hill, R.V. Kochanov, Y. Tan, P.F. Bernath, M. Birk, V. Boudon, A.
|Campargue, K.V. Chance, B.J. Drouin, J.-M. Flaud, R.R. Gamache, J.T. Hodges, D. Jacquemart, V.I.
|Perevalov, A. Perrin, K.P. Shine, M.-A.H. Smith, J. Tennyson, G.C. Toon, H. Tran, V.G. Tyuterev, A. Barbe,
|A.G. Csaszar, V.M. Devi, T. Furtenbacher, J.J. Harrison, J.-M. Hartmann, A. Jolly, T.J. Johnson, T. Karman,
|I. Kleiner, A.A. Kyuberis, J. Loos, O.M. Lyulin, S.T. Massie, S.N. Mikhailenko, N. Moazzen-Ahmadi, H.S.P.
|Muller, O.V. Naumenko, A.V. Nikitin, O.L. Polyansky, M. Rey, M. Rotger, S.W. Sharpe, K. Sung, E.
|Starikova, S.A. Tashkun, J. Vander Auwera, G. Wagner, J. Wilzewski, P. Wcislo, S. Yu, E.J. Zak},
|title = {The HITRAN2016 Molecular Spectroscopic Database},
|journal = {Journal of Quantitative Spectroscopy and Radiative Transfer},
|year = {2017},
|doi = {10.1016/j.jqsrt.2017.06.038},
|}

|@article{HITRAN-1046,
|author = {J.J. Harrison},
|title = {New and improved infrared absorption cross sections for dichlorodifluoromethane (CFC-12)},
|journal = {Atmospheric Measurement Techniques},
|year = {2015},
|volume = {8},
|pages = {3197-3207},
|doi = {10.5194/amt-8-3197-2015},
|}
```

7
8 **Figure 31.** References view (Bibtex)

9 The referencing system used in the present paper corresponds to the HITRAN nomenclature.
10 According to it, each source has its own unique *local ID number*. The numbering is organized so
11 different parts of HITRAN such as line-by-line parameters and cross-sections do not depend on
12 each other. The local ID numbers for reference are given in the last field of the header of each
13 .xsc file containing cross-sections (see Figure 29 and Table 3). We provide a list of references
14 for the cross-section part of the HITRAN database in the supplementary Table S1, containing
15 local IDs and *aliases* for each source. Aliases are made in correspondence with the Hodnebrog et
16 al. (2013) [1] review, and represent short unique names, by which the sources are referenced
17 throughout the paper (e.g. "Hashikawa et al. (2004)"). Users can obtain more information on
18 sources such as direct links, DOIs, and BIBCOCODEs through the interactive search interface.
19

5. Conclusions and Future Work

This paper presents a substantial update of the infrared absorption cross-sections in the HITRAN database, which at the present time incorporates more than 320 compounds of primarily atmospheric and climate interest. More than 260 of these compounds were included in the database for the first time. An extensive effort was made to collect and systematize self-consistent datasets of cross-sections for those molecules which provide the most complete spectral coverage. Whenever possible, extensive evaluations of the datasets were performed.

A significant effort has been made to facilitate user access to those data using the open infrastructure provided by HITRAN*Online* [20] (at www.hitran.org) and HAPI (HITRAN Application Programming Interface) [9]. In the first place, the purpose of this data collection is to help researchers in climate and atmospheric science, industry, remote sensing, meteorology, and a variety of other applications, to assess the impact on the climate and environment, track down the possible associated environmental processes (such as, for instance, forest fires), and estimate the atmospheric abundance of the compounds in question. Among the sources of cross-section data one can mention around 220 molecules adopted from both the PNNL – DOE database [13], and the PNNL – SERDP dataset of 55 molecules from Johnson et al [22] that extended the set to include biomass burning molecules, as well as cross-sections from the review of halocarbons (Hodnebrog et al. [1] and references therein), and finally recently published data on CFCs, HCFCs, HFCs, hydrocarbons, and other related species [21,24,36,37,43,49,80,91,92,113,124,25–31,35].

All IR data, provided as cross-sections in HITRAN, were measured using the FTS technique. The one exception was the air- and self-broadened ethane spectra included in HITRAN2016 that were obtained from more accurate laser CRDS measurements [49]. Due to the heterogeneous nature of the sources, data for some molecules are more complete in terms of temperature and pressure ranges than others. However, in most cases the spectral coverage is full, and includes the near-, mid-, and far-IR range. Considering the initial assignment of the data for climate and atmospheric applications, the emphasis was placed on the mid-infrared “atmospheric window” wavenumber range [1] (approximately 700-1500 cm^{-1}), where most of the species given in the paper have significant absorption. However, many cross-sections cover a much wider spectral region (such as, for instance, the data from PNNL). Thus, the extended coverage can serve a broader range of applications. As the effects of climate change continue to grow in importance, it would be beneficial to have many more measurements in the far-infrared ranged between 0 and 600 cm^{-1} to better estimate the greenhouse effect of these molecules.

1 While the update of the cross-sectional part of the database in the near-, mid-, and far-IR
2 range is quite extensive, this work is by no means complete. In the future we plan further
3 revisions of high-quality data that are or will become available in the literature or from new
4 experiments. This task is often hampered by the fact that some reports and publications do not
5 provide supplements, and spectral data can be hard to track down or are permanently lost. We
6 would like to remind authors about the importance of making the data available with their
7 publications (see for instance discussion in Ref. [145]). In addition, the utility of theoretical
8 calculations of absorption cross-sections is increasing as such calculations become more refined,
9 and their use in future HITRAN editions should be assessed [88,146–148].

10 In the near future we also plan to extend the database to include planetary-relevant cross-
11 sections that became available recently, including those given in Refs. [149–151]. The update
12 described here did not include any data in the UV part of the spectrum. For the next edition of
13 HITRAN we plan an extensive effort to update existing UV cross-sections and add many new
14 datasets using approaches similar to those described here.

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27 Security Administration, Office of Defense Nuclear Nonproliferation R&D (NA-22), as well as
28 U.S. Department of Defense's Strategic Environmental Research and Development Program
29 (SERDP), resources conservation project RC-2640. PNNL gratefully thanks both sponsors for
30 their support. PNNL is operated for the U.S. Department of Energy by the Battelle Memorial
31 Institute under contract DE-AC06-76RLO 1830.

32 We thank reviewers for many helpful comments.

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