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# Alternative Gate Dielectrics on Semiconductors for MOSFET Device Applications

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*We have investigated the synthesis and properties of deposited oxides on Si and Ge for use as alternative gate dielectrics in MOSFET applications. The capacitance and leakage current behavior of polycrystalline  $Y_2O_3$  films synthesized by pulsed-laser deposition is reported. In addition, we also discuss the growth of epitaxial oxide structures. In particular, we have investigated the use of silicide termination for oxide growth on (001) Si using laser-molecular beam epitaxy. In addition, we discuss a novel approach involving the use of hydrogen to eliminate native oxide during initial dielectric oxide nucleation on (001) Ge.*

## 1. INTRODUCTION

Metal/oxide/semiconductor (MOS) structures are key elements in microelectronic applications. [1,2] For silicon integrated circuit development, device architecture scaling relationships indicate that the capacitance/area requirement for the 0.1  $\mu\text{m}$  linewidth technology will be on the order of 35  $\text{fF}/\mu\text{m}^2$ . For a  $\text{SiO}_2$ -based capacitor structure, this corresponds to a dielectric oxide thickness of 1 nm. At this thickness, the magnitude of leakage current due to tunneling will be sufficiently high to make the usefulness of such devices doubtful. As such, there is significant interest in replacing  $\text{SiO}_2$  with a material with a higher dielectric constant for use as the gate dielectric in Si-based MOSFET device structures. Figure 1 illustrates how film thickness constraints are relaxed for gate dielectrics with dielectric constants greater than that for  $\text{SiO}_2$ . One significant difficulty in satisfying this metric is in the tendency to form a thin  $\text{SiO}_2$

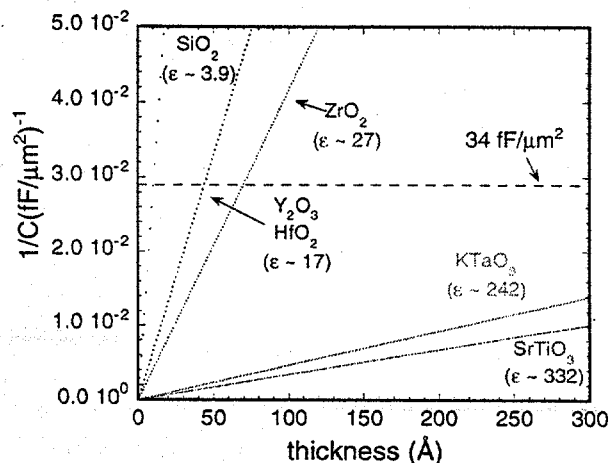


Figure 1 Plot of inverse capacitance versus films thickness for various dielectric materials

layer at the oxide/silicon interface. This directly limits the maximum capacitance/area that can be achieved. If, for example, a SiO<sub>2</sub> layer of only 10 Å forms at the interface due to a moderate reaction between the Si substrate and dielectric film, the metric can not be met. In addition to Si, MOS-type device structures are of interest for other semiconductor systems if a functional dielectric/semiconductor interface can be identified.

## 2. Polycrystalline Y<sub>2</sub>O<sub>3</sub> on Si (001)

We have investigated the properties of polycrystalline Y<sub>2</sub>O<sub>3</sub> films as a candidate for an alternative gate dielectric material. Ytria is attractive based on thermodynamic arguments regarding its stability in contact with silicon.[3] The films were deposited by pulsed laser deposition (PLD) using a KrF excimer laser. Oxide ceramic targets were used as the ablation sources. The Si (001) substrates were dipped in dilute HF/H<sub>2</sub>O solution prior to loading for film deposition. Most of the effort focused on films with thickness of 5-10 nm. Deposition was carried out at a substrate temperature of 100-300°C. Capacitors were fabricated by e-beam evaporating 5 nm Cr/200 nm Au through a contact mask with 1 mm diameter holes. Back contacts were formed by evaporating Cr/Au on the back of the Si wafer following abrasion.

For deposited film thicknesses on the order of 5 to 10 nm, relatively high leakage currents make capacitance measurements rather difficult. In particular, the conventional equivalent circuits consisting of a capacitor with either a series or parallel resistance are not sufficient. In order to extract capacitance in the presence of a high leakage current, the complex impedance of the structures was modeled as a capacitor with both a series and parallel resistance. With three unknowns, this required measuring the complex impedance at two frequencies. Measurements were made at 50 and 100 kHz in order to derive the high frequency (100 kHz) capacitance, assuming that the elements within the model are frequency-independent for the frequency range considered.

Figure 2 shows the capacitance and leakage current behavior for a 12 nm thick Y<sub>2</sub>O<sub>3</sub> film deposited on (001) Si. The substrate was p-type with a resistivity of 10-20 Ω-cm. There is a noticeable shift in the flat-band voltage that may result either from fixed charges in the oxide film or charges due to the e-beam evaporation of the metal gates. For the structure shown, the capacitance/area in accumulation is 14 fF/μm<sup>2</sup>. Leakage current is 5x10<sup>-6</sup> A/cm<sup>2</sup> at a gate bias of -1 volt, and 4x10<sup>-4</sup> A/cm<sup>2</sup> at -2 Volts (accumulation). In the as-deposited films, there is significant dispersion in the C-V behavior. This may indicate ionic motion under bias due to oxygen vacancies within the Y<sub>2</sub>O<sub>3</sub> matrix. This frequency dispersion can be minimized by

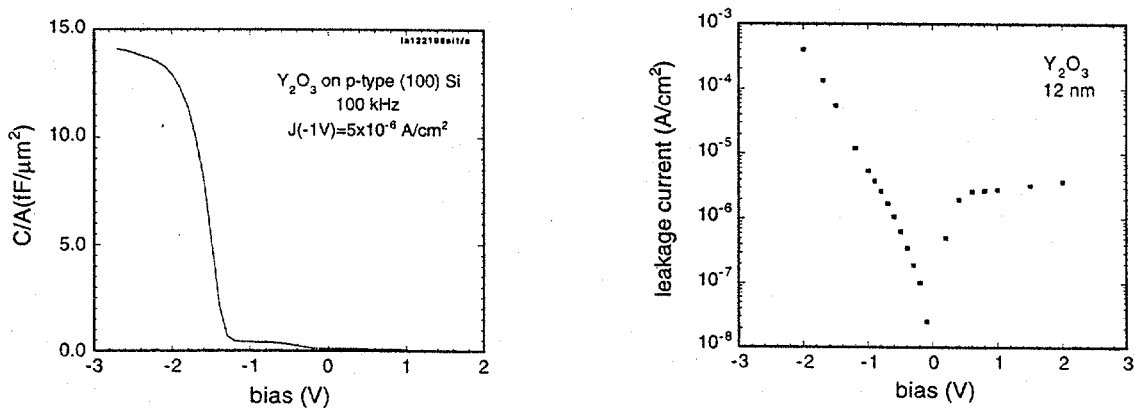
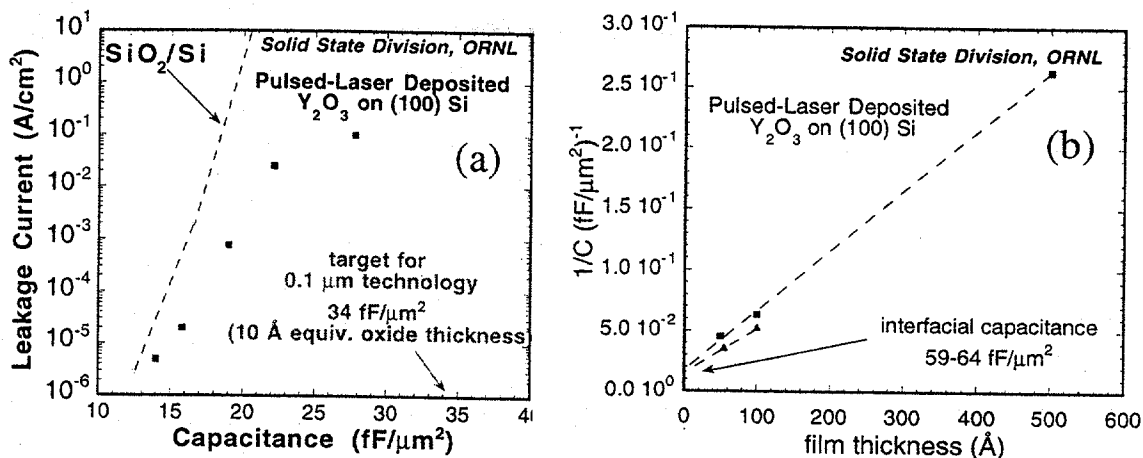


Figure 2 C-V and leakage current behavior for 12 nm thick Y<sub>2</sub>O<sub>3</sub> on (001) Si



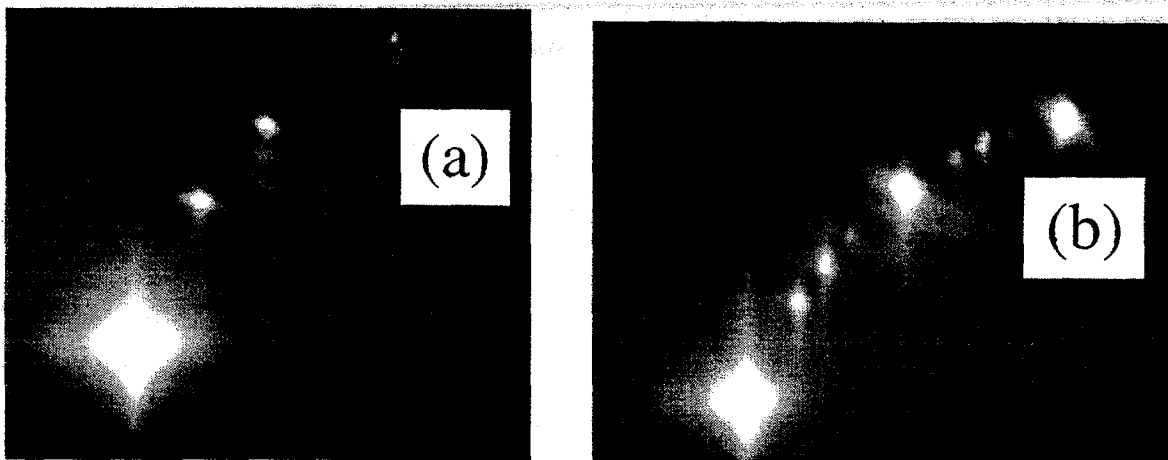
**Figure 3** Plot of leakage current versus areal capacitance (a) and inverse capacitance versus film thickness for polycrystalline  $Y_2O_3$  films on *p*-type (001) Si.

annealing in oxygen. Note that some hysteresis is also observed in the C-V with bias sweep. This has been attributed to the slow movement of mobile charges in the oxide at or near the oxide/semiconductor interface. Also, an anomaly in the 1 kHz C-V behavior appears upon annealing. This indicates the likely presence of interface charges at the oxide/semiconductor interface. The capacitance/area decreases with oxygen annealing presumably due to the growth of an interfacial SiO<sub>2</sub> layer. It is interesting to compare the performance of these deposited polycrystalline oxides with thermally grown SiO<sub>2</sub> on Si (001). Figure 3a) shows a plot of leakage current versus capacitance/area for a number of Y<sub>2</sub>O<sub>3</sub> structures. For convenience, the behavior of SiO<sub>2</sub>/Si is shown. Note that the data for the Y<sub>2</sub>O<sub>3</sub> films lies to the high-capacitance side of the SiO<sub>2</sub>/Si curve indicating promising performance. In addition, we have also plotted the inverse capacitance versus film thickness in Fig. 3b). The intercept at zero film thickness indicates the interfacial capacitance. For the Y<sub>2</sub>O<sub>3</sub> films, this interfacial capacitance is ~ 60 fF/μm<sup>2</sup> which is encouraging in terms of ultimately achieving Y<sub>2</sub>O<sub>3</sub> capacitors with high capacitance/area.

### 3. Epitaxial Oxides on (001) Si

The formation of epitaxial oxide structures on semiconductor surfaces represents an enabling development in the integration of electronic oxides on a semiconductor platform. This is particularly true for metal-oxide-semiconductor field effect transistor structures in which the oxide/semiconductor interface largely determines the properties of the device. Recently, there has been significant interest in the formation of epitaxial oxides on silicon for this application. One interesting approach is to terminate the Si(001) surface with an ordered silicide as an initial step in nucleating epitaxial oxides.[4] Silicide termination is achieved via high temperature epitaxy on an atomically-clean silicon surface. We have investigated the formation of SrSi<sub>2</sub> on Si (001) by laser molecular beam epitaxy (laser-MBE) using reflection high energy electron diffraction.

Laser molecular beam epitaxy simply refers to the growth of films by pulsed-laser deposition in a ultra-high vacuum environment. Laser-MBE has several attractive advantages over conventional MBE in that it is a truly digital film growth process with flux delivered to the substrate in a pulse-by-pulse mode. This not only allows for significant control of the film growth process, but also enables one to study the relaxation of the surface between flux delivery.



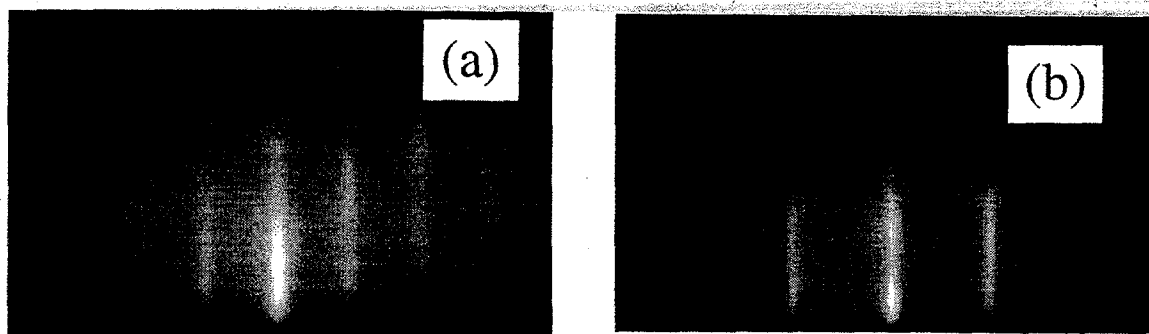
**Figure 4** RHEED images slightly off the Si (110) direction for clean (a) and sub-monolayer SrSi<sub>2</sub> covered surface

L-MBE can utilize simple metal targets that are not intentionally heated, as is necessary for an evaporation cell. This greatly simplifies the hardware requirements for film growth. It also eliminates a heated source of potential impurities within the vacuum growth chamber. L-MBE also provides an energetic flux to the film growth surface which can impact film kinetics. The laser molecular beam epitaxy system used in this work consists of a UHV system with a base pressure of  $4 \times 10^{-10}$  Torr equipped with a liquid nitrogen-cooled shroud. 1" diameter Sr and Ba metal targets were mounted on a target carousel for in situ target selection. A KrF excimer laser was used as the ablation source. A laser energy of less than  $0.4 \text{ J/cm}^2$  was used in order to minimize the density of ablation particulates. Metal flux was provided as a series of 5-10 shot bursts with a laser repetition rate of 100 Hz.

The Si(001) substrates were cleaned in trichloroethylene, acetone, and methanol in an ultrasonic bath. A UV-ozone treatment was then used to remove hydrocarbon contamination prior to film growth. The substrates were then loaded via a load-lock into the film growth chamber. The substrates were preannealed for  $\sim 30$  minutes at  $350^\circ\text{C}$  before heating to  $825\text{-}900^\circ\text{C}$  for SiO<sub>2</sub> desorption as confirmed by a  $2 \times 1$  reconstructed RHEED pattern. The substrates were then cooled to  $650\text{-}725^\circ\text{C}$  for deposition of Sr metal. Upon completion of silicide nucleation, the substrates were cooled to  $100\text{-}200^\circ\text{C}$  for additional metal and oxide growth.

Figure 4 shows the RHEED patterns from the substrate surface taken at various stages of the silicide formation. The images were acquired slightly off the Si(110) direction with an electron acceleration voltage of 10 kV. Fig. 4a) shows a clean  $2 \times 1$  reconstructed Si(001) surface following the thermal desorption of a sacrificial SiO<sub>2</sub> layer at  $850\text{-}900^\circ\text{C}$ . Following desorption of the native oxide, the substrate is cooled to  $\sim 650\text{-}725^\circ\text{C}$  for Sr deposition. Upon deposition of Sr, additional higher order reflections appear in the RHEED pattern as seen in Fig. 4b). This has been associated with formation of an ordered SrSi<sub>2</sub> phase on the Si surface. Upon deposition of additional Sr, the additional reflections disappear, resulting in the image identical to that shown in Fig. 4a). The appearance and disappearance of higher order spots indicates a  $1/4$  monolayer coverage of the Si(100) surface.[4] Additional Sr deposition does not result in a reappearance of the higher order reflections. It should be noted to the realization of this pattern was highly dependent on background contaminants. The use of a liquid nitrogen cooled shroud to reduce hydrocarbon impurities was essential to achieving reproducible results.

With a surface coverage of  $1/4$  monolayer, the substrate is cooled to  $100\text{-}200^\circ\text{C}$  for additional metal and subsequent oxide growth. Fig. 5 shows the RHEED patterns along the Si (110) and (100) directions achieved for an epitaxial (001) BaO layer growth on a SrSi<sub>2</sub> terminated Si surface at  $150^\circ\text{C}$ . The oxygen pressure during the initial oxide nucleation was



**Figure 5** RHEED images for BaO on Si(001) taken along the Si(110) (a) and (100) (b) directions.

$5 \times 10^{-8}$  Torr. It should be noted that the RHEED patterns were obtained over a limited region of the 1" diameter substrate presumably where near-ideal SrSi<sub>2</sub> sub-monolayer coverage occurred.

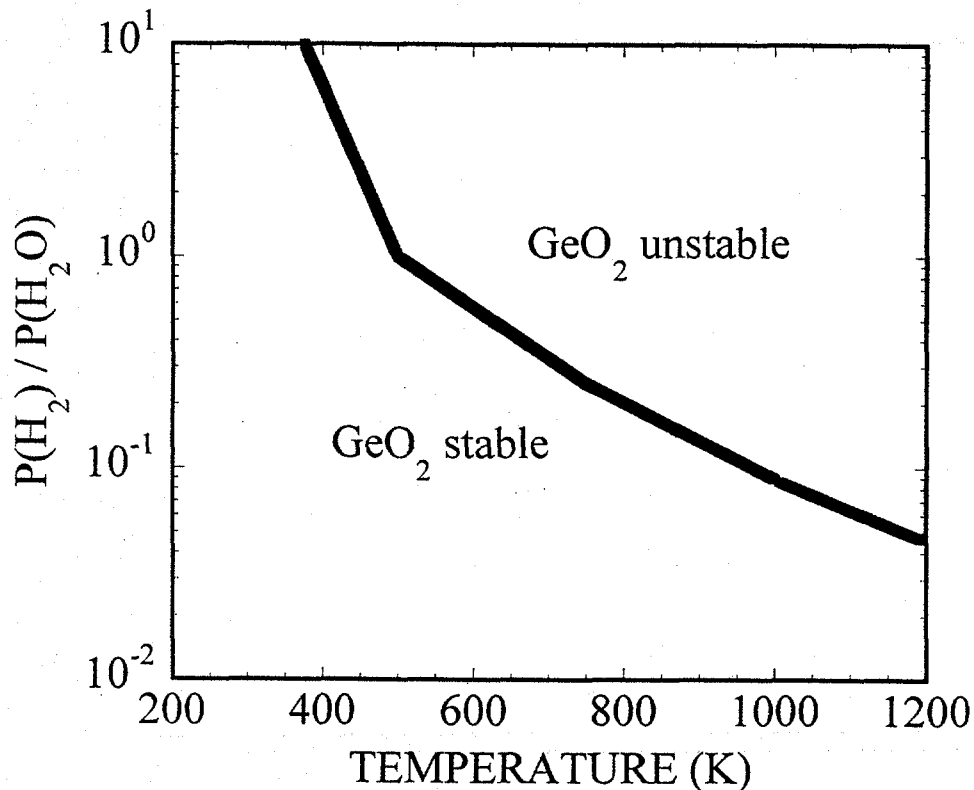
#### 4. CeO<sub>2</sub> on (001) Ge by Hydrogen-Assisted Pulsed-Laser Deposition

Ge and SiGe alloys are attractive semiconductor materials for electronic applications, possessing higher carrier mobilities and thermal conductivities than that of silicon. Germanium possesses a simple cubic crystal structure with  $\mu_n = 3900 \text{ cm}^2/\text{V}\cdot\text{sec}$ ,  $\mu_p = 1800 \text{ cm}^2/\text{V}\cdot\text{sec}$ , and a thermal conductivity of  $0.6 \text{ W/cm}\cdot\text{K}$ . Unfortunately, the native germanium oxides are not suitable for MOS-type device structures.[5,6] The formation of stable metal oxides on Ge could prove instrumental in the development of Ge and/or SiGe alloy integrated circuits. In addition, a method to form oxides that are epitaxial on these semiconductors would enable the integration of various epitaxial oxide materials and device structures with semiconductor electronics by providing a crystalline oxide template for additional epitaxial oxide film growth.

We report on the epitaxial growth and properties of CeO<sub>2</sub> on (001) Ge using pulsed-laser deposition. Hydrogen is introduced as a background gas during the film nucleation in order to eliminate the native GeO<sub>2</sub> on the Ge surface and achieve epitaxy.[7,8] The use of hydrogen greatly relaxes vacuum requirements needed for the formation of a GeO<sub>2</sub>-free surface. Using this approach, (001)-oriented CeO<sub>2</sub> thin films were obtained on the (001) Ge surface. The resulting metal oxide/semiconductor interface is atomically abrupt, with no apparent native oxide present at the semiconductor/metal oxide interface. These results differ from that observed for CeO<sub>2</sub> on (001) Si, where the film is (110)-oriented with significant SiO<sub>2</sub> formation at the film/substrate interface.[9-11]

The deposition of (001) epitaxial cerium oxide on single crystal Ge (001) was performed by pulsed-laser deposition using a KrF excimer laser. Single crystal (001) Ge substrates were cleaned by successive rinsing in trichloroethylene, acetone, and methanol, followed by rinsing in deionized water. The native oxide was then removed by a 30 sec dipping in a 1:10 HF:H<sub>2</sub>O solution. The substrate was blown dry with dry nitrogen and mounted on the heater platen using silver paint. The sample was loaded into the vacuum chamber for pulsed-laser deposition of the oxide film. The chamber was evacuated to an initial base pressure that ranged from  $5 \times 10^{-6}$  Torr to  $2 \times 10^{-9}$  Torr. The mounted Ge substrate was annealed in vacuum at 350°C for 2-12 hrs in order to decompose the organic binder in the silver paint. Prior to heating, the ablation target was in situ cleaned by laser ablation with a shutter between the substrates and ablation target.

In order to minimize or eliminate any native germanium oxide on the substrate surface prior to growth, hydrogen gas was introduced into the chamber to a sufficient pressure such that the



**Figure 6** Plot showing  $\text{GeO}_2$  stability line with respect to the ratio of hydrogen gas to water vapor. The conditions for which  $\text{GeO}_2$  is unstable are explicitly denoted.

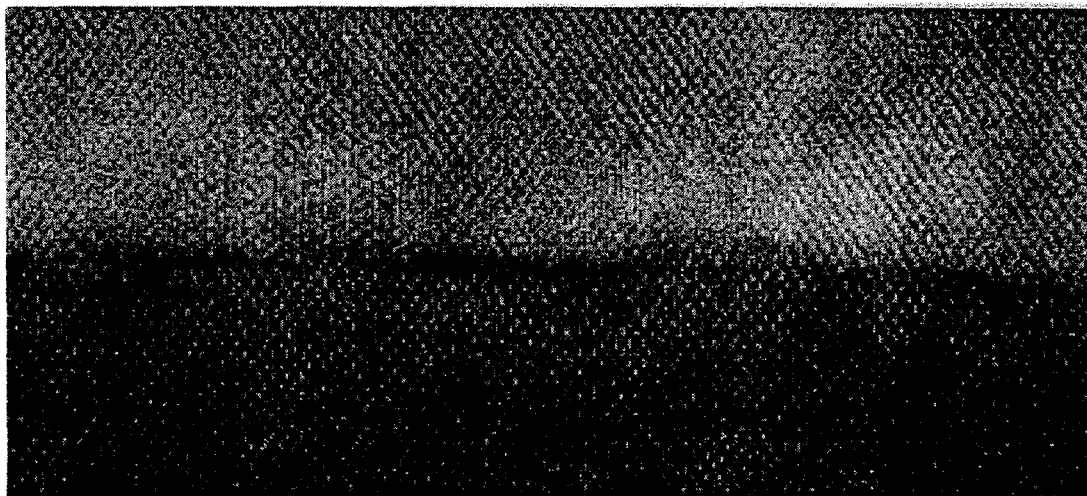
ratio of hydrogen to water vapor partial pressure  $P(\text{H}_2) / P(\text{H}_2\text{O})$  was approximately at or above the  $\text{GeO}_2$  stability curve at the anticipated oxide film growth temperature as shown in Fig. 6. This curve was derived from Ellingham diagrams for oxide materials.[12,13] The oxide stability line can be estimated from the temperature-dependent Gibbs free energy of the chosen native oxide when compared to the  $\text{H}_2\text{O}/\text{H}_2$  equilibrium behavior. Depending on the anticipated metal oxide deposition temperature, the  $\text{GeO}_2$  instability criterion translates into a value of  $P(\text{H}_2)/P(\text{H}_2\text{O}) > 0.04$ , preferably greater than 1.0. As a practical matter, the base pressure of the vacuum systems consists mostly of  $\text{H}_2\text{O}$ .

For the experiments reported here, a flow of 4%  $\text{H}_2$  / 96% Ar was introduced into the chamber with the  $\text{H}_2/\text{Ar}$  pressure ranging from  $10^{-5}$  to  $10^{-1}$  Torr, depending on the base pressure. For example, an  $\text{H}_2/\text{Ar}$  pressure of 0.1 Torr yields a hydrogen partial pressure of  $4 \times 10^{-3}$  Torr, and a value of  $P(\text{H}_2)/P(\text{H}_2\text{O})$  of  $8 \times 10^2$  for a base pressure of  $5 \times 10^{-6}$  Torr. The substrate was then heated to the growth temperature in the 4%  $\text{H}_2$  / 96% Ar background. As the substrate is heated, the hydrogen reduces any  $\text{GeO}_2$  that resides or forms on the substrate surface, resulting in the gas-phase etching of the native oxide. The final growth temperature must be consistent with the requirement that the conditions (temperature, water vapor partial pressure, hydrogen partial pressure) be above the  $\text{GeO}_2$  stability line where the formation of  $\text{GeO}_2$  is thermodynamically unfavored. Under these conditions, a metal oxide material that is stable for the chosen temperature/water vapor/hydrogen conditions can be deposited onto the heated substrate by means of pulsed-laser deposition. This oxide material should be thermodynamically stable in contact with Ge.[3] Cerium oxide satisfies this criterion.

After heating to the selected growth temperature, a  $\text{CeO}_2$  film was deposited on the Ge surface. The KrF excimer laser energy density was  $\sim 1.5 \text{ J/cm}^2$  with a laser repetition rate of 1 Hz. These conditions yielded a deposition rate of  $\sim 0.1 \text{ nm/laser pulse}$  when using a pressed and sintered  $\text{CeO}_2$  ablation target. Both the metal cation and oxygen atoms are provided by laser ablation as short, discrete pulses. Between the laser pulses, the metal oxide / semiconductor system can relax to the conditions that thermodynamically favor instability of the native oxides. After the initial film nucleation, additional  $\text{CeO}_2$  could be deposited on the initial oxide film template using deposition conditions that do not necessarily coincide with the requirements of  $\text{GeO}_2$  thermodynamic instability as outlined for the template oxide layer, thus allowing the  $\text{CeO}_2$  stoichiometry to be controlled. After film growth, the germanium substrate was typically cooled in vacuum.

X-ray diffraction data were taken for a series of 50 nm thick  $\text{CeO}_2$  films on (001) Ge grown at temperatures ranging from 450 to 775°C. The  $\text{CeO}_2$  films were deposited with  $P(\text{H}_2) = 4 \times 10^{-7}$  Torr and a base pressure =  $2 \times 10^{-9}$  Torr. Approximately 5 nm was deposited in the presence of hydrogen. The remaining 45 nm of  $\text{CeO}_2$  was deposited with no hydrogen flow. For temperatures greater than 550°C, only the (001) orientation is observed. In-plane XRD scans confirm that the  $\text{CeO}_2$  films are in-plane aligned. Note that some broadening in  $2\theta$  is observed at the lower growth temperature, indicating that strain is present in the film.

In addition to x-ray diffraction, the properties of the  $\text{CeO}_2$  / Ge interface were investigated using high-resolution scanning transmission electron microscopy (STEM). Cross-section images reveal that  $\text{CeO}_2$  films deposited at 750°C possess a 3D island-like morphology that is faceted, with extended defects (pinholes) extending to the substrate at some of the faceted boundaries. Disruption of the  $\text{CeO}_2/\text{Ge}$  interface in the form of etched holes or amorphous material was evident in the proximity of these defects. Between the defects, the  $\text{CeO}_2/\text{Ge}$  interface is atomically abrupt and free of  $\text{GeO}_2$ . At slightly lower temperatures, faceting is significantly diminished. Figure 7 shows a cross-section Z-contrast STEM image of a  $\text{CeO}_2$  film that was grown at 650°C. The absence of an amorphous native oxide layer at the interface differs from that observed for  $\text{CeO}_2$  films on Si, where significant  $\text{SiO}_2$  is observed at the film/substrate interface. The fact that  $\text{GeO}_2$  is thermodynamically less stable than  $\text{SiO}_2$  suggests that the formation of  $\text{GeO}_2$  at the interface should be less likely.



**Figure 7**  
interface.

*High resolution STEM image of the atomically abrupt  $\text{CeO}_2/\text{Ge}$  crystalline*



## 5. CONCLUSIONS

In conclusion, we have investigated the synthesis and properties of moderate to high- $k$  dielectric materials on Si and Ge for MOSFET applications. Polycrystalline yttria/Si interfaces suffer from charge trapping and relatively high interface trap states, although the capacitance/leakage current behavior is encouraging. The termination of Si(001) with strontium silicide for subsequent growth of epitaxial oxide thin films have also been examined. Laser-MBE has been successfully used to form both the ordered silicide layer and epitaxial alkaline earth oxides. In addition, the growth of (001) epitaxial  $\text{CeO}_2$  on a (001) Ge surface using pulsed-laser deposition in a hydrogen ambient has been realized. By using hydrogen to eliminate  $\text{GeO}_2$  from the surface during film nucleation, a  $\text{CeO}_2$  / Ge interface that is essentially free of  $\text{GeO}_2$  can be formed. The use of hydrogen to promote the epitaxial growth of oxides on semiconductor surfaces should be applicable not only to  $\text{CeO}_2$  on Ge by PLD, but to other material systems and physical vapor deposition techniques. The resulting epitaxial metal oxide film should be useful as a crystalline oxide template for subsequent growth of additional epitaxial oxide layers on the semiconductor surface. These structure may prove useful for numerous electronic and optoelectronic devices, including metal-oxide-semiconductor field-effect transistors, random-access memory devices, and optical waveguide structures.

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