

Application of High-Performance Liquid Chromatography with Diode Array Detector for Simultaneous Determination of 11 Synthetic Dyes in Selected Beverages and Foodstuffs

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Abstract A simple, inexpensive and robust high-performance liquid chromatography diode array detector (HPC-DAD) procedures are proposed to analyse food dyes in beverages, hard candy and fish roe samples. An ether-linked phenyl stationary phase provides sufficient selectivity and chromatographic performance for separation of 11 sulfonated azo dyes. Beverage samples were only diluted (and degassed when needed) before analysis. Solidphase extraction (SPE) or matrix solid-phase dispersion (MSPD) procedures are proposed for efficient extraction of the analytes from candies or fish roe samples, respectively. Limits of detection (LODs) were from 0.005 to 0.013 μ g mL⁻¹ and limits of quantification (LOQs) between 0.014 and 0.038 μ g mL⁻¹. HPLC-DAD method was validated in terms of intra- and inter-day accuracy and precision at three concentration levels 2, 1, and 0.1 μ g mL⁻¹. Validation was also performed for SPE and MSPD extraction procedures including intra- and inter-day accuracy (Recovery %) and precision (RSD%), as well as intralaboratory reproducibility. Application to analysis of beverages and food samples available to consumers proved that described methods are suitable for the routine analysis of dyes in food products.

Keywords Synthetic dyes \cdot Food products \cdot Solid-phase extraction (SPE) \cdot Matrix solid-phase dispersion (MSPD) \cdot HPLC-DAD

Introduction

Synthetic dyes are widely used as food additives, which are added to foodstuffs in order to compensate for the loss of natural colours destroyed during processing and/or storage, to enhance natural colour or add colour to foods that would otherwise be colourless or coloured differently (Jia et al. 2014). There are evidences indicating that dyes and their metabolites pose potential health risk to human, including allergy and asthmatic reaction, DNA damage, hyperactivity and carcinogenesis (Zou et al. 2013; Rovina et al. 2016). In order to ensure food safety and control international trade, different legislative efforts were paid to the food colourants regulation (Burrows 2009). Many countries have their own regulations about the food dyes permitted to be used in foods. Both, maximum level of dyes used in different foodstuffs as well as acceptable daily intake (ADI), were established by the respective institutions. According to the Regulation (EC) No 1333/2008 (2008), all food additives authorized for use in the EU before 20 January 2009 should be subjected to a new risk assessment by the European Food Safety Authority (EFSA). In EU, Tartrazine (E 102); Quinoline yellow (E 104); Sunset yellow (E 110); Azorubin (E 122); Ponceau 4R (E 124) and Allura red (E 129) should be labelled with additional information: 'name or E number of the colour(s): may have an adverse effect on activity and attention in children'. Some of these colourants are banned in other countries. Considering wide availability, low cost and chemical stability of the synthetic dyes, there is a risk of their illegal usage by unfair manufacturers. Food producers may exceed the maximum levels of approved dyes or use banned substances in order to increase the attractiveness of food products. There are also possible misstatements by adding additional dyes without their proper labelling. As a consequence, analysis of synthetic dyes in food products

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plays an important role in assessing food quality and evaluating food safety, including possible risks to human. Therefore, reliable analytical methods are needed to monitor dyes in processed products brought to the consumer.

Up to date, several review papers concerning analytical approaches for extraction and reliable identification and quantification of food dyes have been published in the literature. Ahlström et al. (2005) have reviewed the development of analytical procedures for determination of banned azo dyes in consumer goods. Kucharska and Grabka (2010) reviewed chromatographic methods for the determination of synthetic dyes. Rebane et al. (2010) reviewed methods utilizing liquid chromatography (LC) coupled with UV-VIS detection and LC coupled with mass spectrometry (MS) for analysis of Sudan I-IV dyes along with their extraction procedures in various food matrices. Kaur and Gupta (2012) have reviewed the determination of water-soluble and waterinsoluble food dyes by spectrophotometry. Yamjala et al. (2016) comprehensively described methods for the analysis of azo dyes employed in food industry. Liquid chromatography (LC) is preferred analytical technique for artificial colourants analysis. Synthetic food dyes are complex molecules, which may cause some difficulties during chromatographic analysis. These compounds are generally highly polar and elute very fast near the dead volume. Significant influence on chromatographic behaviour have sulfonyl acidic groups in their structure. The presence of these ionisable groups leads to different interactions of neutral and ionic forms with the stationary phase resulting in peak tailing, low system efficiency and poor reproducibility of retention data. Dyes in their anionic forms are poorly retained in RP-LC systems, which may be explained by electrostatic repulsion between anions and the deprotonated residual silanol groups present on the silicabased stationary phase. Separation of different dyes is most often performed in reversed phase (RP) or ion pair (IP) systems, while quantitative analysis is usually based on measurements via UV-VIS detection, especially using diode array detector (DAD) (Fuh and Chia 2002; Kiseleva et al. 2003, Tuzimski and Woźniak 2008; Culzoni et al. 2009; Tuzimski 2011; Yan et al. 2012; Petigara Harp et al. 2013; Zou et al. 2013; De Andrade et al. 2014; Li et al. 2015; Qi et al. 2015; Suleková et al. 2016), or employing MS (Fuh and Chia 2002; Sun et al. 2007; Pardo et al. 2009; Feng et al. 2011; Liu et al. 2011; Chen et al. 2013; Li et al. 2013; Zou et al. 2013; Li et al. 2014; Jia et al. 2014; Li et al. 2015; Tsai et al. 2015; Qi et al. 2015). There are also some literature reports devoted to utilization of gas chromatography coupled with MS (GC-MS) (Otero et al. 2016) or capillary electrophoresis (CE) (Del Giovine and Bocca 2003; Huang et al. 2005; Prado et al. 2006) for food colours analysis.

The presence of chromophore groups in synthetic dyes allowed their determination and quantification at sufficiently low concentration levels by measuring their absorbance in visible wavelengths range. Therefore, matrix effect that can be observed in food analysis is practically limited. Only natural pigments (e.g. anthocyanins) present in food samples may affect proper quantification of analytes. However, these natural pigments often differs significantly in terms of chemical structure and properties from synthetic colourants and can be removed during sample preparation and/or chromatographically separated from peaks of interest. In case of LC-MS/MS, any of co-eluting compounds from a food extract (if present in the sample) can cause significant differences in signals in comparison to solvent-only samples. The co-eluting compounds can either suppress or enhance ionization of the analyte in the electrospray ion source (ESI). These difficulties are partially overcome by plotting of matrixmatched calibration and carful matrix effect assessment. Nevertheless, some differences observed from sample to sample may still occur and affect proper quantification of the analytes. Considering all of the above, HPLC-DAD can be simple, cost-effective, reliable and sufficiently sensitive analytical platform for dyes analysis in beverages and foodstuffs. As a result, it may successfully serve as alternative to expensive and more sophisticated LC-MS/MS instrumentation.

First stage of dyes analysis is extracting them from food sample, which may be, in many cases, more complicated then instrumental analysis itself (Kucharska and Grabka 2010). Selection of proper extraction technique depends on the kind of sample matrix and is also strictly connected with analytical technique that will be used to determinate the food dyes (Kucharska and Grabka 2010). Due to its advantages, like simplicity and rapidity, solid-phase extraction (SPE) is the most commonly used technique to concentrate and purify these target compounds (Tuzimski and Woźniak 2008; Baggiani et al. 2009; Tuzimski 2011; Yan et al. 2012; Chen et al. 2013; Li et al. 2014; Qi et al. 2015; Yamjala et al. 2016). However, in case of solid food samples, dyes ought to be extracted prior to SPE into appropriate solvent-media. For this purpose, solvent extraction is preferably selected (Zou et al. 2013; Tsai et al. 2015; Otero et al. 2016). Solvent extraction could be also assisted by ultrasounds (UAE) (Bonan et al. 2013; Li et al. 2013; Khalikova et al. 2015) or microwaves (MAE) (Sun et al. 2013). There are also literature reports on applying of quick, easy, cheap, effective, rugged and safe (QuEChERS) approach for dyes extraction (Jia et al. 2014; Zhu et al. 2014; Rejczak and Tuzimski 2015).

In this study, authors developed an easy and cost-effective HPLC-DAD method for analysis of 11 synthetic dyes. Extraction of the analytes was carried out by means of anion exchange SPE or matrix solid-phase dispersion (MSPD). The procedures were preliminary validated and applied to analysis of different beverages, as well as solid food samples, such as hard candies and fish roe.

Experimental

Chemicals and Reagents

Solvents and Mobile-Phase Solutions

Acetonitrile (MeCN) and methanol (MeOH) were pro chromatography grade and were obtained from E. Merck. Deionized water (0.07–0.09 μ S cm⁻¹) was obtained by means of Hydrolab System (Gdansk, Poland) in our laboratory. Ammonium acetate and aqueous ammonia solution were obtained from POCH (Gliwice, Poland).

Dyes Standards

Standards for the 11 synthetic colourants under investigation, such as Allura Red AC (E129), Amaranth (E123), Azorubin (E122), Brilliant Black PN (E151), Brilliant Blue FCF (E133), Brilliant Green BS (E142), Patent Blue V (E131), Ponceau 4R (E124), Red 2G (E128), Sunset Yellow FCF (E110) and Tartrazine (E102), were obtained from Institute for Engineering of Polymer Materials and Dyes (Zgierz, Poland).

Individual stock standard solutions (200 mg L⁻¹) were prepared in methanol and were stored at 6 ± 2 °C. The working standard solutions and their mixtures were prepared by combining suitable aliquots of each individual standard stock solution and diluting them with 50 mM ammonium acetate.

Solid-Phase Extraction Materials

Strata X-AW cartridges, containing weak anion exchange functionalized polymeric sorbent, were obtained from Phenomenex Inc. (Torrance, CA, USA).

Sample Preparation

Beverages Procedure

Several different types of beverages were selected for the analysis including isotonic drinks (OSHEE Multifruit, OSHEE Red, OSHEE Pink, OSHEE Orange for Runners and 4Move Lime&Mint Flavour), carbonated alcoholic beverages (Sobieski Impress Cranbery and Sobieski Impress Kamikaze), flavoured vodka (Lubelska Grapefruit) and syrups (Victoria's Blue Curacao and Bols Grenadine). Simple dilute and shoot method was applied for beverage analysis. Before injection, samples were prepared by combining suitable aliquots of each beverage and diluting them with 50 mM ammonium acetate up to 5 mL in volumetric flasks. Isotonic drinks and flavoured vodka were diluted fiftyfold, when syrup samples were diluted hundredfold before analysis. In the case of carbonated alcoholic drinks, gas in beverages was removed by placing them into an ultrasonic bath at room temperature for 20 min. Afterwards, fiftyfold dilution was used to prepare final samples. Each beverage sample was prepared in triplicates (n = 3).

Hard Candy Procedure

Hard candies were thoroughly grinded down in a ceramic mortar with a pestle. Then, 100 mg of the crumbled candy was weighted into a 12 mL polypropylene (PP) tube and 10 mL of deionized water was added. Closed tube was shaken by hand for 5 min in order to dissolve the sample. Afterwards, SPE procedure using Strata X-AW cartridges was conducted.

Initially, a SPE cartridge was conditioned with 5 mL of methanol and 5 mL of deionized water. Next, sample was loaded onto the cartridge (negative pressure 300–400 mbar on the pump connected to the SPE vacuum manifold). Afterwards, cartridge was washed with 2 mL of 25 mM ammonium acetate and 3 mL of methanol and dried for 1 min (negative pressure about 750 mbar). Finally, dyes were eluted with 5% (ν/ν) solution of aqueous ammonia in methanol up to 5 mL in volumetric flask. The eluate was transferred into an evaporating dish and evaporated to dryness under a fume hood with air intake switched on. Dry residues were reconstituted in 1 mL of 50 mM ammonium acetate producing final sample for HPLC-DAD analysis.

Caviar/Fish Roe Procedure

Matrix solid-phase dispersion (MSPD) procedure was developed for caviar/fish roe samples. First, 250 mg of the sample was weighted into ceramic mortar and 200 mg of Strata X-AW sorbent was added. The whole content of the mortar was blended using a pestle for approximately 2 min until a visually homogeneous mixture was obtained. Then, 1 mL of deionized water was added and the mixture was blended for additional 1 min in order to enable/improve retention of dyes by anion exchange mechanism. The mixture was transferred into a 6 mL empty SPE syringe with a frit disk packed at the bottom. Water was pulled away from the sample in the SPE vacuum manifold and discarded. Next, 1 mL 5% (v/v) solution of aqueous ammonia in acetonitrile was used to wash the mortar carefully and transferred to the syringe with packed sample. Finally, dyes were eluted with another portion of 5% (v/v)aqueous ammonia solution in acetonitrile, together up to 5 mL in volumetric flask. The eluate was transferred into 12 mL PP centrifuge tube and stored in a freezer compartment at temperature about -20 °C for 15 min in order to improve protein/peptide precipitation. Afterwards, the sample was centrifuged for 5 min at 6000 rmp (3480 rcf) using a laboratory centrifuge (MPW-223e, Warsaw, Poland). After centrifugation, 200 µL of extract was transferred into an evaporating dish and evaporated to dryness under a fume hood with air intake switched on. Dry residues were reconstituted in 1 mL of 50 mM ammonium acetate producing final sample for HPLC-DAD analysis.

HPLC-DAD Procedure

Agilent Technologies 1200 HPLC system with a quaternary pump was used for the LC analysis. Analytes were separated using a Synergi Polar RP 150 mm × 4.6 mm column, with 4-µm particle size (Phenomenex, Torrance, CA, USA). The column was thermostated at 22 °C. Mobile phase consisted of 50 mM CH₃COONH₄ in water (component A) and acetonitrile (component B). Linear gradient elution programme was used (5% B at start; linear to 40% B in 20 min). Mobile phase flow was 1 mL min⁻¹. Final samples were injected onto the column using a Rheodyne manual injector with 20 μ L loop. The column was re-equilibrated for 15 min using initial mobile phase composition between subsequent analysis.

UV–VIS spectra library for dyes under investigation was created by collecting analyte spectra (200–900 nm) from the analysis of individual standards at 5 μ g mL⁻¹ concentration (Fig. 1). Four different wavelengths were selected for simultaneous detection and quantification of the analytes (420 nm for yellow dyes; 504 and 515 nm for red dyes and 630 nm for blue dyes). Identification of colourants was accomplished on the basis of their retention times and by comparison between the UV–VIS spectra of the reference compounds in the chromatograph library and the UV spectra of the detected peaks in the samples.

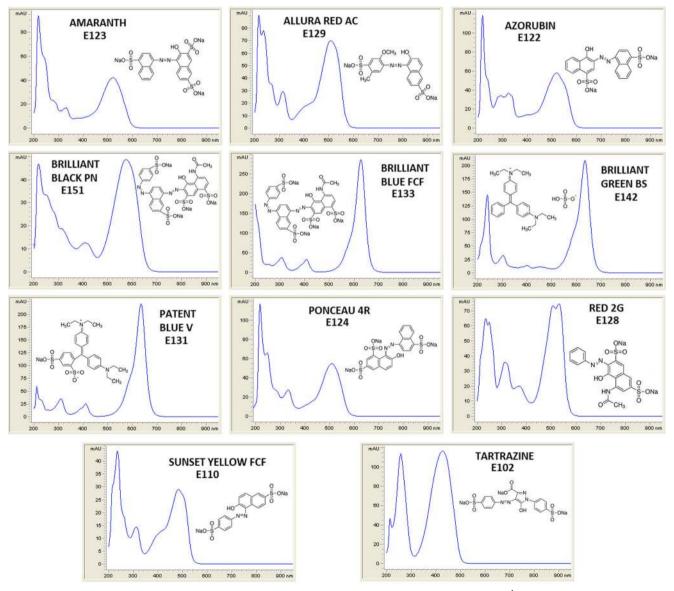


Fig. 1 UV–VIS spectra (200–900 nm) of 11 synthetic food dyes collected from individual solutions (5 μ g mL⁻¹) prepared and injected in 50 mM ammonium acetate

HPLC Method Validation

The standard calibration curves of the analytes were constructed by plotting analyte concentration against peak area. Dye standards were prepared as solutions in mobile phase component A at ten concentrations in the range of 0.01–2 µg mL⁻¹ and injected in triplicate under the same chromatographic conditions. The individual calibration concentrations were 0.01; 0.02; 0.04; 0.06; 0.08; 0.1; 0.2; 0.5; 1 and 2 µg mL⁻¹. The calibration curves of dyes under investigation showed satisfactory linearity and correlation between concentration and peak area over the studied range with the correlation coefficients, r^2 , ≥0.9998. The instrumental limits of detection (LOD) and quantification (LOQ) for all analytes were calculated using following formulas (1 and 2) (ICH guidelines 2005):

$$LOD = 3.3 \frac{SD}{S}$$
(1)

$$LOQ = 10 \frac{SD}{S}$$
(2)

where SD is the standard deviation of *y*-intercept of regression lines (calculated using LINES function in MS Excel 2010), and *S* is the slope of the calibration plot. Retention times and full calibration data including LODs and LOQs are presented in Table 1.

The HPLC-DAD procedure was validated in terms of accuracy as well as intra- and inter-day precision. Validation study was conducted for three concentration levels of 2, 1 and 0.1 µg mL⁻¹. During three following days, six different mixtures of dye standards for each concentration level were prepared in 50 mM CH₃COONH₄ and analysed per day. Average concentration measured, its standard deviation (SD) as well as accuracy (%) and precision (RSD%) were calculated for each dye standard. Inter-day accuracy (%) and precision (RSD%) was calculated for each following day (n = 6) and is given in Table 2, as well as intra-day data calculated as a mean results for all replicates (n = 18).

Validation Studies for Sample Preparation Procedures

Validation study was performed for SPE using Strata X-AW columns (Table 3) as well as for MSDP extraction for caviar/ fish roe samples (Table 4). Intra-day accuracy and precision was studied by analysing samples in six replicates (n = 6). The experiments were repeated in three following days, producing results (n = 18) for inter-day accuracy and precision evaluation. Accuracy in all cases was expressed as percentage recovery of the analyte using equitation (3):

$$Recovery\% = \frac{Average analyte concentration found in the sample}{Analyte concentration added to the sample} \times 100\%$$

Precision was expressed as RSD% calculated as follows (4):

$$RSD\% = \frac{SD \text{ of the recovery}\%}{Mean \text{ recovery}\%} \times 100\%$$
(4)

Intra-laboratory reproducibility was studied for both extraction procedures (SPE and MSPD). Experiments in six replicates (n = 6) were conducted by two different analysts using the same instrumentation and calibration data, but using self-prepared dyes solution for spiking procedures. In total, 30 replicates (n = 30) were analysed for each procedure (and each spiking level) and on their basis overall accuracy (Recovery %) and precision (RSD%) were determined. MSPD procedure for extraction of dyes form fish roe/ caviar samples was validated at four spiking levels; at 10 μ g g⁻¹ (for the mixture of all eleven standards under investigation); and additionally at 20, 50 and 100 $\mu g g^{-1}$ for the Brilliant Black PN and Brilliant Blue FCF, which were expected to be found in fish roe samples bought from the market to the analysis. For beverages analysis, only simple dilution was proposed as sample preparation. Therefore, validation studies for this procedure are not necessary, since HPLC-DAD method is validated (Table 2).

Results and Discussion

The elaborated HPLC-DAD method and three different sample preparation solutions was successfully applied to the analysis of ten different beverages (including isotonic drinks, flavoured vodka, carbonated alcoholic drinks and syrups), two different hard candy products and three fish roe/caviar samples. Eight out of 11 colourants under investigation including Tartrazine (E 102), Sunset Yellow FCF (E110), Azorubin (E 122), Ponceau 4R (E 124), Allura Red AC (E 129), Patent Blue V (E 131), Brilliant Blue FCF (E 133) and Brilliant Black PN (E 151) were found in food products at concentration from 0.32 to 237.8 μ g mL⁻¹ or gram of the sample, respectively. The detailed results and most important information on method development and validation are provided in subsequent subsections devoted to HPLC-DAD method development and validation, and analysis of beverage; hard candy and fish roe/caviar samples.

HPLC-DAD Method Development and Validation

In this study, ether-linked phenyl phase (Synergi Polar-RP) was applied for chromatographic method development. Retention of dyes on this stationary phase occurs due to $\pi - \pi$ interactions. Generally, dyes with naphthalene ring show grater retention in comparison to dyes containing benzene ring in their molecules. Addition of ammonium acetate into mobile phase caused increase in the retention of investigated food

E number	Name	$t_{ m R}$ (min)	Calibration data	on data						
			λ (nm)	Range ($\mu g \ m L^{-1}$)	Linear regression	r	SD of slope ^a	SD of intercept ^a	$LOD~(\mu g~mL^{-1})$	LOQ ($\mu g m L^{-1}$)
E102	Tartrazine	4.140-4.220	420	0.01–2	y = 59.458x + 0.1226	0.9999	0.26519	0.19329	0.011	0.033
E123	Amaranth	5.875-5.970	504	0.01 - 2	$y = 34.91 \mathrm{x} - 0.0039$	0.9999	0.14773	0.10767	0.010	0.031
			515	0.01 - 2	y = 37.754x + 0.0236	0.9999	0.17302	0.12610	0.011	0.033
E124	Ponceau 4R	7.995-8.130	504	0.01 - 2	y = 41.222x + 0.0984	0.9999	0.13340	0.09723	0.008	0.024
			515	0.01 - 2	y = 41.098x + 0.1045	0.9999	0.15705	0.11447	0.009	0.028
E151	Brilliant Black PN	8.579-8.815	515	0.01 - 2	y = 19.346x + 0.0569	0.9999	0.03830	0.02791	0.005	0.014
			630	0.01 - 2	y = 19.705 x + 0.0505	6666.0	0.04638	0.03563	0.006	0.018
E110	Sunset Yellow FCF	9.185-9.312	420	0.04-2	y = 19.93 x - 0.0157	0.9999	0.07172	0.05844	0.010	0.029
			504	0.01-2	y = 39.946x + 0.1245	0.9999	0.13185	0.09610	0.008	0.024
			515	0.01-2	y = 33.141x + 0.0845	6666.0	0.08327	0.06069	0.006	0.018
E129	Allura Red AC	10.792 - 10.895	504	0.01-2	y = 51.807x + 0.1114	0.9999	0.25318	0.18452	0.012	0.036
			515	0.01 - 2	y = 51.732x + 0.1352	6666.0	0.26105	0.19026	0.012	0.037
E128	Red 2G	11.089-11.186	504	0.01-2	y = 64.266x + 0.2334	0.9999	0.21101	0.15378	0.008	0.024
			515	0.01-2	y = 64.573x + 0.2218	0.9999	0.19891	0.14497	0.007	0.022
E122	Azorubin	13.258-13.384	504	0.01-2	y = 42.407x + 0.1271	6666.0	0.17138	0.12491	0.010	0.029
			515	0.01-2	y = 45.747x + 0.1428	6666.0	0.24049	0.17528	0.013	0.038
E133	Brilliant Blue FCF	14.964–15.157	630	0.01 - 2	y = 150.98x + 0.8019	0.9999	0.40999	0.29881	0.006	0.020
E142	Brilliant Green BS	16.236–16.329	630	0.01 - 2	y = 172.72x + 1.0567	0.9999	0.68537	0.49953	0.009	0.029
E131	Patent Blue V	18.822-18.950	630	0.01 - 2	y = 203x + 1.1198	0.9999	0.68283	0.49767	0.008	0.025

1, and 0.1 $\mu g m L^{-1}$
HPLC-DAD method at 2,
or the proposed HPLC
and precision studied fe
Intra- and inter-day accuracy a
Table 2 h

Dye	Conc. (µg	Intra-day							
	mL)	Day 1 $(n=6)^a$				Day 2 $(n=6)^a$			
		Average concentration measured ($\mu g m L^{-1}$) ^c	SD	Accuracy (%) ^d	Precision (RSD%) ^e	Average concentration measured ($\mu g m L^{-1}$) ^c	SD	Accuracy (%) ^d	Precision (RSD%) ^e
Tartrazine (E102)	1 2	1.9928 1.0184 0.1045	0.0085 0.0091 0.0034	99.64 101.84 104 53	0.43 0.89 3.25	1.9915 1.0232 0.1029	0.0017 0.0068 0.0074	99.58 102.32 102 87	0.08 0.00 0.02
Amaranth (E123)		2.0425	0.0019	102.13	2:05 0.89	2.0087 1.0226	0.0052	102.26	1.46 0.50
Ponceau 4R (E124)	0.1 1	$\begin{array}{c} 0.1048 \\ 1.9793 \\ 1.0074 \end{array}$	$0.0043 \\ 0.0200 \\ 0.0066$	104.78 98.96 100.74	4.13 1.01 0.66	0.1004 1.9895 1.0126	0.0040 0.0090 0.0047	100.39 99.48 101.26	3.94 0.45 0.47
Brilliant Black PN (E151)	0.1 2 1	0.1005 2.0426 1.0070	$\begin{array}{c} 0.0012 \\ 0.0139 \\ 0.0046 \end{array}$	100.54 102.13 100.70	1.24 0.68 0.46	0.1014 2.0211 1.0045	0.0008 0.0303 0.0023	101.38 101.06 100.45	$0.74 \\ 1.50 \\ 0.23$
Sunset Yellow FCF (E110)	0.1 2 1	0.1030 1.9972 1.0027	0.0018 0.0207 0.0077	102.97 99.86 100.27	1.70 1.03 0.76	0.1020 1.9952 1.0088	0.0014 0.0023 0.0059	102.02 99.76 100.88	$ \begin{array}{c} 1.38 \\ 0.11 \\ 0.58 \\ 0.58 \\ \end{array} $
Allura Red AC (E129)	0.1 1	0.1005 1.9795 1.0069	0.0032 0.0196 0.0092	100.52 98.98 100.69	3.18 0.99 0.92	0.1002 1.9855 1.0171	0.0009 0.0052 0.0088	100.17 99.28 101.71	0.85 0.26 0.87
Red 2G (E128)	0.1 2 1	0.1004 2.0353 1.0212	$\begin{array}{c} 0.0033 \\ 0.0404 \\ 0.0075 \end{array}$	100.39 101.77 102.12	3.24 1.98 0.74	0.0983 2.0072 1.0163	0.0018 0.0243 0.0043	98.30 100.36 101.63	1.88 1.21 0.42
Azorubin (E122)	0.1 2 1	0.1033 2.0115 1.0138	0.0018 0.0100 0.0084	103.30 100.57 101.38	1.79 0.50 0.83	0.1020 1.9971 1.0184	0.0011 0.0125 0.0047	102.03 99.86 101.84	1.08 0.62 0.46
Brilliant Blue FCF (E133)	0.1 2 1	0.1012 1.9803 1.0041	$\begin{array}{c} 0.0016 \\ 0.0087 \\ 0.0071 \end{array}$	101.15 99.01 100.41	1.59 0.44 0.71	0.1015 1.9874 1.0065	0.0007 0.0101 0.0034	101.45 99.37 100.65	$0.64 \\ 0.51 \\ 0.34$
Brilliant Green BS (E142)	0.1 2 1	0.1014 1.9955 1.0099	0.0014 0.0081 0.0065	101.40 99.78 100.99	1.33 0.40 0.64	0.1019 1.9933 1.0130	0.009 0.0031 0.0044	101.87 99.67 101.30	0.88 0.16 0.44
Patent Blue V (E131)	0.1 2 1 0.1	0.1011 1.9752 1.0069 0.1002	$\begin{array}{c} 0.0016\\ 0.0190\\ 0.0049\\ 0.0014\end{array}$	101.07 98.76 100.69 100.17	$1.60 \\ 0.96 \\ 0.48 \\ 1.44 $	0.1019 1.9838 1.0096 0.1006	0.0012 0.0122 0.0038 0.0006	101.93 99.19 100.96 100.59	$ \begin{array}{c} 1.20 \\ 0.61 \\ 0.38 \\ 0.59 \\ \end{array} $
Dye	Intra-day				Inter	Inter-day $(n = 18)^{\rm b}$			
	Day 3 $(n=6)^a$								
	Average concentration measured ($\mu g m L^{-1}$) ^c	SD ¹ ⁵	Accuracy (%) ^d	Precision (RSD%) ^e		Average concentration measured ($\mu g m L^{-1}$) ^c	SD	Accuracy (%) ^d	Precision (RSD%) ^e
Tartrazine (E102)	1.9906 1.0220 0.1033	0.0083 0.0055 0.0023	99.53 102.20 103.30	0.42 0.54 2.25	1.9916 1.0210 0.1037	16 10 37	0.0066 0.0066 0.0025	99.58 102.10 103.65	0.33 0.65 2.40
Amaranth (E123)	2.0457	0.0389	102.29	1.90	2.03	44	0.0378	101.72	1.86

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Dye	Intra-day				Inter-day $(n = 18)^{b}$			
	Day 3 $(n = 6)^{a}$							
	Average concentration measured ($\mu g m L^{-1})^c$	SD	Accuracy (%) ^d	Precision $(RSD\%)^{e}$	Average concentration measured ($\mu g m L^{-1}$) ^c	SD	Accuracy (%) ^d	Precision $(RSD\%)^{e}$
Ponceau 4R (E124)	1.0117 0.1042 1.9863 1.1004	0.0091 0.0035 0.0076 0.0091	101.17 104.20 99.32 100.04	0.90 3.33 0.38 0.91	1.0207 0.1035 1.9846 1.0059	0.0110 0.0040 0.0131 0.0084	102.07 103.54 99.23 100.59	1.07 3.88 0.66 0.84
Brilliant Black PN (E151)	0.1006 2.0150 1.0056	$\begin{array}{c} 0.0010 \\ 0.0239 \\ 0.0072 \end{array}$	100.56 100.75 100.56	1.03 1.19 0.71	0.1007 2.0246 1.0059	0.0011 0.0231 0.0051	100.74 101.23 100.59	$1.04 \\ 1.14 \\ 0.51$
Sunset Yellow FCF (E110)	0.1026 1.9222 1.0009	0.0008 0.0137 0.0011	102.57 99.61 100.09	$0.73 \\ 0.69 \\ 1.10$	0.1026 1.9948 1.0035	0.0013 0.0138 0.0088	102.60 99.74 100.35	$\begin{array}{c} 1.26 \\ 0.69 \\ 0.88 \end{array}$
Allura Red AC (E129)	0.1010 1.9810 0.9978	0.0039 0.0084 0.0137	101.00 99.05 99.78	3.83 0.42 1.38	0.1006 1.9817 1.0058	0.0029 0.0122 0.0128	100.62 99.09 100.58	2.92 0.61 1.27
Red 2G (E128)	0.0990 2.0378 1.0126	0.0013 0.0327 0.0056	99.01 101.89 101.26	1.31 1.60 0.55	0.0994 2.0286 1.0167	0.0023 0.0334 0.0069	99.38 101.43 101.67	2.34 1.65 0.68
Azorubin (E122)	0.1041 2.0126 1.0064	0.0016 0.0156 0.0090	104.10 100.63 100.64	1.54 0.78 0.89	0.1033 2.0079 1.0120	0.0017 0.0135 0.0089	103.32 100.40 101.20	$1.64 \\ 0.67 \\ 0.88 \\ $
Brilliant Blue FCF (E133)	0.1019 1.9861 0.9989	0.0013 0.0075 0.0089	101.92 99.30 99.89	1.30 0.38 0.89	0.1015 1.9848 1.0028	0.0012 0.0076 0.0071	101.55 99.24 100.28	1.22 0.38 0.70
Brilliant Green BS (E142)	0.1008 1.9937 1.0056	0.0023 0.0023 0.0094	99.68 100.56	0.47 0.11 0.94	0.1015 1.9941 1.0090	0.0010 0.0039 0.0071	99.70 99.70 100.90	0.20 0.70 0.70
Patent Blue V (E131)	0.1015 1.9861 1.0036 0.1009	0.0013 0.0095 0.0075 0.0018	101.53 99.30 100.36 100.87	1.28 0.48 0.75 1.79	0.1015 1.9823 1.0063 0.1005	0.0013 0.0118 0.0056 0.0013	$101.46 \\ 99.12 \\ 100.63 \\ 100.53$	1.24 0.59 1.29
^a Six different samples analysed per day for ^b $n = 18$ (3 days with six replicates per day)	^a Six different samples analysed per day for intra-day accuracy and precision $^{b}n = 18$ (3 days with six replicates per day)	and precision						

to (2 days will six replicates per day) 2

^c Concentrations calculated on the basis of calibration curves presented in Table 1. For Tartrazine $\lambda = 420$ nm; for Sunset Yellow FCF mean concentrations from 420, 504, and 515 nm; for Amaranth, Ponceau 4R, Allura Red AC, and Azorubin mean concentrations from 504 and 515 nm; for Brilliant Black PN mean concentrations from 515 and 630 nm; and for Brilliant Blue FCF, Brilliant Green BS, and Patent Blue V $\lambda = 630 \text{ nm}$

 $^{\rm d}$ Calculated as (mean determined concentration/nominal concentration) \times 100%

^e Expressed as RSD% = (SD/mean) × 100%

Dye		Intra-day ^a						Inter-day ^b $(n = 18)$	n = 18	Intra-labor:	ttory repi	Intra-laboratory reproducibility ^c				Overall ^d $(n = 30)$	(0)
Ц	Name	Day 1 $(n = 6)$	= 6)	Day 2 $(n = 6)$	(9	Day 3 $(n = 6)$	(9			Analyst 1 $(n = 6)$	(9 = 0)	Analyst 2 $(n = 6)$	<i>n</i> = 6)	Mean $(n = 12)$	12)		
number		Recovery %	RSD%	Recovery %	RSD %	Recovery %	RSD %	Recovery %	RSD %	Recovery %	RSD %	Recovery %	RSD %	Recovery %	RSD %	Recovery %	RSD %
E102	Tartrazine	104.3	3.4	102.4	5.5	103.4	3.9	103.3	4.2	103.9	3.2	97.8	4.5	101.9	4.5	102.8	4.3
E123	Amaranth	102.5	4.2	100.6	5.9	101.5	4.4	101.5	4.7	101.9	4.0	95.9	4.1	9.99	4.9	100.9	4.7
E124	Ponceau 4R	100.9	4.0	99.4	5.2	100.3	3.9	100.2	4.2	100.5	3.7	95.8	3.6	98.9	4.2	99.7	4.1
E151	Brilliant Black PN	100.6	3.8	99.3	4.8	100.4	3.5	100.0	3.9	100.4	3.5	97.0	4.0	99.2	3.8	96.8	3.8
E110	Sunset Yellow FCF	98.9	2.4	97.6	3.8	98.6	2.3	98.3	2.8	98.6	2.3	95.2	3.5	97.5	3.0	98.1	2.9
E129	Allura Red AC	101.3	3.6	99.7	5.1	100.7	3.5	100.6	3.9	100.9	3.4	96.1	3.9	99.3	4.1	100.1	4.0
E128	Red 2G	102.8	2.8	100.9	5.2	101.8	3.4	101.8	3.8	102.6	2.6	96.8	5.2	100.7	4.3	101.4	4.0
E122	Azorubin	101.6	2.3	99.3	6.2	100.3	4.0	100.3	4.4	101.7	2.1	94.2	7.7	99.2	5.5	99.9	4.7
E133	Brilliant Blue FCF	101.1	7.1	99.1	8.1	100.3	6.7	100.1	6.9	101.0	6.4	95.3	6.0	99.1	6.5	99.8	6.7
E142	Brilliant Green BS	98.5	5.6	95.4	7.4	9.96	4.6	96.7	5.1	98.6	5.5	90.4	8.9	95.5	6.4	96.3	5.5
E131	Patent Blue V	100.9	4.2	98.7	6.8	6.66	4.6	99.7	5.1	100.6	3.9	93.6	6.1	98.2	5.6	99.2	5.2

^c Mean recovery % and RSD% from experiments conducted by two different analysts (n = 6 for each operator) and mean results (n = 12)

^d Average recovery % and RSD% from all experiments (n = 30)

ıl. Met	linod	15 (2	20.	.,,	10									each																								
18)		KSD%	4.9	3.6	5.2	3.8	4.0	5.5	5.1	10.9	5.4	14.4	15.3	Inter-dav ^b $(n = 18$ for each	spiking level)		Recovery %	85.7	85.8	89.8	90.5	1.08	84.2															
Inter-day ^b $(n = 18)$	Ę	Kecovery %	1.66	93.3	95.4	92.8	98.5	94.6	94.4	85.5	98.6	76.3	81.5		each		RSD%	6.2	6.3	6.2	8.8	2.8	4.0		30)	(nc	RSD%	5.1	3.7	5.5	4.1	4.3	5.8	5.5	11.9	5.1	15.6	
		KSD%	C.0	4.9	7.0	4.9	6.0	7.8	6.9	13.3	7.0	16.9	18.7		Day 3 $(n = 6 \text{ for each})$	spiking level)	Recovery %	85.3	84.0	88.7	89.4	83.0 07 1	83.1		Originally $(n - 30)$		Recovery %	101.0	94.1	96.6	93.6	99.3	95.8	95.6	88.0	99.3	79.3	
	Day 3 $(n = 6)$	Kecovery %	100.4	93.9	96.2	93.2	99.4	95.5	95.1	86.2	9.66	76.8	82.4		or each			9.4	9.6	8.2	12.1).C	<i>e.</i> c				RSD%	5.2	3.9	5.8	4.3	4.6	6.0	5.8	12.6	4.7	16.2	
		KSD%	0.0	3.4	5.5	3.8	1.7	5.4	5.4	11.8	5.1	15.7	16.7		Day 2 ($n = 6$ for each	spiking level)	Recovery %	86.3	87.4	91.6	91.6	80.0 05.6	0.08			Mean $(n = 12)$	Recovery %	102.8	95.2	98.4	94.8	100.5	97.7	97.4	91.8	100.4	83.9	
	Day 2 $(n = 6)$	Kecovery %	1.99	93.2	95.5	92.7	97.5	94.4	94.5	85.6	98.1	76.4	81.6		or each			9.6	9.3	3.8	11.5	0.0	2.4				RSD%	4.3	3.7	5.4	4.4	5.3	5.4	5.7	12.0	3.6	14.2	
		KSD%	1.0	2.8	3.4	3.4	3.7	3.9	3.3	9.6	5.1	13.6	13.7	Intra-dav ^a	Day 1 ($n = 6$ for each	spiking level)	Recovery %	85.5	85.9	89.2	90.4	7.08	83.9			Analyst 2 $(n = 6)$	Recovery % R											
Intra-day ^a	Day 1 $(n = 6)$	Kecovery %	99.1	92.9	94.5	92.5	98.4	93.7	93.5	84.7	98.1	75.6	.4				Spiking level ($\mu g g^{-1}$)			00		00	0ſ		, C		RSD% Recc	9 106.1	4 97.4	2 101.8	96.8	4 103.2	2 101.5	1 100.6	12.1 97.6	7 103.1	16.3 90.9	
Ini							Sunset Yellow FCF 98		Red 2G 93			1 BS	Patent Blue V 80					Brilliant Black PN 20	50		Brilliant Blue FCF 20	00	10		Intra Jahomatomy nannodaloihility ^c	Analyst 1 $(n = 6)$	Recovery % R5	100.5 4.9	93.8 3.4	96.1 5.2	93.5 3.9	98.7 3.4	95.2 5.2	95.2 5.1	87.9 12	98.6 4.7	79.3 16	
(a) Dye	-	E number	E102	E123	E124	E151	E110	E129	E128	E122	E133	E142	E131	Dve			E number	E151			E133			(a)			E number Ro	E102 10	E123 93	E124 96	E151 93	E110 98	E129 95	E128 95	E122 87	E133 98	E142 79	

(q)									
Dye	Inter-day ^b ($n = 18$ for each Intra-laboratory reproducibility ^c spiking level) Analyst 1 ($n = 6$ for each spikin,	Intra-laboratory 1 Analyst 1 $(n = 6)$	Intra-laboratory reproducibility ^c Analyst 1 ($n = 6$ for each spiking level) Analyst 2 ($n = 6$ for each spiking level) Mean ($n = 12$ for each spiking level)	Analyst 2 ($n = 6$	for each spiking level)	Mean $(n = 12$ fo	r each spiking level)	Overall ^d $(n = 30 \text{ f})$	Overall ^d $(n = 30$ for each spiking level)
E number RSD%	- RSD%	Recovery %	RSD%	Recovery %	RSD%	Recovery %	RSD%	Recovery %	RSD%
E151	8.0	85.5	9.6	86.0	5.0	85.7	7.1	85.7	7.4
	7.6	85.1	6.1	86.5	9.3	85.7	6.9	85.8	7.1
	5.7	92.6	5.9	89.8	5.6	91.3	5.6	90.6	5.6
E133	10.2	90.4	11.5	93.2	6.4	91.9	8.7	91.1	9.4
	4.5	84.9	3.7	85.2	5.5	85.0	4.1	85.1	4.2
	4.0	86.0	4.2	83.9	3.7	85.0	4.0	84.6	3.9

conducted by two different analysts (n = 6 for each operator) and mean results (n = 12)

from 18 samples analysed in three different days (n = 6 for each day)

% and RSD% from all experiments (n = 30)

experiments

from

RSD%

and

^c Mean recovery % ^d Average recovery

² Mean recovery % and RSD%

colouring agents. This was important, e.g. for Tartrazine, because in only water-organic modifier system, it elutes close to the dead volume. Another advantage of ammonium acetate additive to the mobile phase was observed in terms of peak shape and chromatographic system efficiency. Finally, separation of 11 synthetic food dyes is possible applying gradient elution mode in less than 20 min obtaining satisfactory selectivity, theoretical plate number (N/m) and peak symmetry for all of the analytes (Fig. 2). All obtained peaks were narrow with widths at their half heights from 0.075 to 0.088, which results in high efficiency of the chromatographic system with N/m from 108,800 to 1,713,800 being observed. Symmetry factors (A_S) and tailing factors (t_F) were from 0.799 to 0.932 and 1.050 to 1.312, respectively (Fig. 2). All these parameters proved that chromatographic system applied in this study is suitable and demonstrate satisfactory performance for analysis of synthetic dyes.

HPLC-DAD provides sufficient sensitivity for dyes under investigation with LODs from 0.005 to 0.013 μ g mL⁻¹ and LOQs between 0.014 and 0.038 μ g mL⁻¹ being observed (Table 1). These LOD/LOQ values are lower than those obtained by Zou et al. using HPLC-DAD and similar to those observed by the authors for HPLC-MS/MS (Zou et al. 2013). Feng et al. developed method for screening of 40 dyes in soft drinks by liquid chromatography-tandem mass spectrometry (LC-MS/MS) (Feng et al. 2011). For Sunset Yellow FCF, Allura Red AC, Azorubin, Brilliant Blue FCF and Patent Blue V, the authors obtained LOD ranging from 0.3 to 10 ng mL^{-1} , which are similar or lower than those in our study (Feng et al. 2011). However, for Tartrazine, Amaranth and Ponceau 4R reported LODs were relatively high with values of 500, 250 and 125 ng mL⁻¹, respectively, which means detection limits from 15.6 to 45.45-fold higher than those obtained via HPLC-DAD in our study (Feng et al. 2011).

The proposed HPLC-DAD method for synthetic dyes analysis undergone validation study including inter- and intra-day accuracy (%) and precision (RSD%) measurements (Table 2). Inter-day accuracy (%) and precision (RSD%) were investigated over three following days on the basis of analysis of six different solution of dye standards (n = 6) for three concentration levels of 2, 1 and 0.1 μ g mL⁻¹. Inter-day accuracy (%) and precision (RSD%) at the highest concentration level (2 μ g mL⁻¹) were from 98.76 to 102.13% and 0.40 to 2.05% (day 1; n = 6); from 99.19 to 101.06% and 0.08 to 1.50% (day 2; n = 6); and from 99.09 to 101.72% and 0.20 to 1.86% (day 3; n = 6), respectively. At the middle concentration level (1 μ g mL⁻¹), inter-day accuracy (%) and precision (RSD%) were from 100.27 to 102.85% and 0.46 to 0.92% (day 1; n = 6); from 100.45 to 102.32\% and 0.23 to 0.87% (day 2; *n* = 6); and from 99.78 to 102.20% and 0.54 to 1.38% (day 3; n = 6), respectively. At the low concentration level (0.1 μ g mL⁻¹), inter-day accuracy (%) and precision (RSD%) were from 101.17 to 104.78% and 1.24 to 4.13%

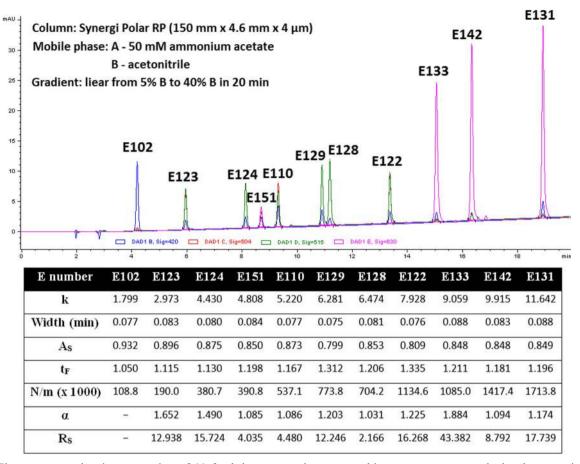


Fig. 2 Chromatogram showing separation of 11 food dyes at concentration of 1 μ g mL⁻¹ (overlaid signals from 420, 504, 515 and 630 nm wavelengths) and table showing chromatographic parameters for individual peaks (*k* capacity factor, *width* (min) peak width at half height, *A_S* symmetry factor, *t_F* tailing factor, *N/m* theoretical plate numbers calculated by halfwidth method, α selectivity to proceeding peak, *R_S* resolution to proceeding peak calculated by halfwidth method; Note: all

chromatographic parameters were calculated automatically by ChemStation® software operating Agilent 1200 chromatograph; for more details about respective formulas used for calculation, please see the manual linked (https://www.agilent.com/cs/library/usermanuals/Public/ G2070-91126_Understanding.pdf); *E* numbers for food dyes are explained in the text)

(day 1; n = 6); from 98.30 to 102.87% and 0.59 to 3.94% (day 2; n = 6); and from 99.01 to 104.20% and 0.47 to 3.83% (day 3; n = 6), respectively. Intra-day accuracy (%) and precision (RSD%) (three following days with six replicates; n = 18) were from 99.09 to 101.72% and 0.20 to 1.86% at 2 µg mL⁻¹, from 100.28 to 102.10% and 0.51 to 1.27% at 1 µg mL⁻¹, and 99.38 to 103.65% and 0.96 to 3.88% at 0.1 µg mL⁻¹, respectively. The HPLC-DAD method showed satisfactory accuracy and precision for the analysis of synthetic dyes in beverage and food samples.

Analysis of Dyes in Beverage Samples

In this study, sample preparation for beverages was practically limited to their dilution prior to chromatographic analysis. Carbonated drinks were additionally degassed in the ultrasonic bath before preparing their final solutions in 50 mM ammonium acetate. On the one hand, dilution factor (from twentyfold to hundredfold) was selected in terms of dyes concentration, and on the other hand, according to sugar content in particular beverage. In total, ten different drinks were analysed. Concertation of dyes found in the samples (recalculated to 1 mL of the beverage) were from 0.32 to 192.36 μ g mL⁻¹ of the beverage with RSD% less than 1.5% being observed for particular beverages on the basis of samples prepared in triplicates by appropriate dilution of the product. These data are shown in details in Table 5. These detected and quantified amounts of colours are in agreement with maximum level according to the Regulation (EC) No 94/36 (1994).

It should be noted that drinks containing solid components from fruits (e.g. juices) should be at least filtered prior to analysis, or, especially in case of complex matrix (e.g. milkbased drinks), solid-phase extraction clean-up ought to be introduced. SPE procedure using Strata X-AW cartridges is described further in the text.

Beverage			Dye declared	Dye detected	Average concentration $(\mu g m L^{-1} of the$	RSD%
Туре	Name	Origin			(μ g mL of the beverage; $n = 3$)	(<i>n</i> = 3)
Isotonic drinks	OSHEE Multifruit	EU ^a	E133	E133	5.48	1.2
	OSHEE Red	EU	E129	E129	87.33	0.5
	OSHEE Pink	EU	E129	E129	4.63	0.8
	OSHEE Orange for Runners	EU	E110	E110	14.90	1.2
	4Move Lime&Mint Flavour ^b	Poland	E102	E102	3.25	0.3
			E133	E133	0.32	1.1
Carbonated alcoholic beverages	Sobieski Impress Cranberry	Poland	_	E129	94.44	0.7
	Sobieski Impress Kamikaze	Poland	_	E133	5.04	0.4
Flavoured Vodka	Lubelska Grapefruit	Poland	_	E124	10.00	0.6
Syrup	Victoria's Blue Curacao	Poland	E133	E133	99.84	0.4
	Bols Grenadine	The Netherlands	E102	E102	41.44	0.8
			E122	E122	192.36	0.7

Table 5 Average concentration ($\mu g \ mL^{-1}$) with RSD% of dyes detected in samples of investigated beverages (data based on triplicates (n = 3))

^a EU European Union

^b Dilution factor for this isotonic drink was equal 20 in order to achieve concentration of E 133 dye at level higher than its LOQ. All others isotonic drinks were diluted fiftyfold prior to HPLC-DAD analysis (as stated in the text)

Analysis of Dyes in Hard Candy Samples

Sample preparation for hard candies was based on SPE employing weak anion exchange retention of dyes on Strata X-AW sorbent. First, the SPE protocol was developed and evaluated for water samples spiked with known amount of standards (2 µg of each dye). Due to the presence of ionized sulfonyl groups in molecules, dyes are strongly retained in polymeric sorbent functionalized with primary-secondary amine ligands. Therefore, two possible elution approach were tested. Application for this purpose of 5% (ν/ν) formic acid in methanol should suppress the ionization of SO₃⁻ groups of the dyes and cause their elution. However, this approach was insufficient and elution was carried out using 5% (v/v) aqueous ammonia solution in methanol. In this case, elution is forced by deactivating of anion exchanger ligands in SPE sorbent. Since dyes are retained strongly on Strata X-AW sorbent, there is possibility to introduce into the procedure an additional wash step using pure organic solvent (methanol). This is beneficial in terms of extraction selectivity, because majority of matrix component can be easy removed. However, in case of hard candy samples, this was not mandatory.

Satisfactory recovery, intra- and inter-day repeatability and intra-laboratory reproducibility were obtained for spiked water samples (2 µg of each dye standard) (Table 3). Intra-day accuracy (Recovery %) and precision (RSD%) were from 98.5 to 104.3% and 2.3 to 7.1% (day 1; n = 6), from 95.4 to 102.4% and 3.8 to 8.1% (day 2; n = 6) and from 96.6 to 103.4% and 2.3 to 6.7%, respectively. On the basis of all 18 replicates analysed in 3 days (n = 18), inter-day repeatability

was calculated obtaining recovery in the range of 96.7-103.3% with RSD% values ranging from 2.8 to 6.9%. As for intra-laboratory reproducibility (experiments conducted by two different analysts, n = 12 in total), recoveries were in the range of 95.5-101.9% with RSD% values ranging from 3.0 to 6.5%. In total, 30 different samples were analysed with overall accuracy and precision ranging from 96.3 to 102.8% and 2.9 to 6.7%, respectively. Detailed validation data of SPE procedure are provided in Table 3. Since sample preparation for hard candies involves dissolving 100 mg of the grinded down sweet in deionized water and then performing SPE, there was no need for additional recovery evaluation for this particular protocol. Accordingly, recoveries summarized in Table 3 are also applicable to the hard candy samples. Two different hard candy products of various manufacturers were analysed. Two dyes (E131 and E133) were detected at concentrations ranging from 15.7 to 29.6 μ g g⁻¹ of the product. One of the hard candy products contained both dyes (E131 and E133) in spite of the fact that only one colourant (E131) was declared on the label. Nevertheless, detected amounts of dyes in hard candies are lower than maximum level $(300 \ \mu g \ g^{-1})$ established and approved for confectionery by EU Regulation (EC) No 94/36 (1994). Detailed data on dyes analysis in hard candy samples are provided in Table 6.

Analysis of Dyes in Fish Roe/Caviar Samples

Sample preparation method development for caviar/fish roe food products is more challenging. Due to the complex matrix of fish roe, extraction of dyes using selective solid adsorbents or ion exchangers prior to HPLC analysis is necessary.

3585

Table 6 Average concentrations (μg^{-1}) with RSD% of dyes detected in investigated solid food samples. Data obtained after SPE-HPLC-DAD for hard candies and MSPD-HPLC-DAD for caviar/fish roe samples on the basis of triplicates (n = 3)

Food sample			Dye declared	Dye detected	Average concentration $(x_1, x_2) = \int dt dt dt dt$	RSD%
Туре	Name	Origin			$(\mu g g^{-1} \text{ of the product; } n = 3)$	(<i>n</i> = 3)
Hard candy	Goplana Minties	Poland	E133	E133	15.7	4.8
	Secretto Ice	Poland	E131	E131	29.6	5.0
			_	E133	21.1	5.4
Caviar/fish roe	Caviar from lumpfish roe	Sweden	E133	E133	67.3	1.1
	Caviar from capelin roe and herring roe	Sweden	E151	E151	211.3	4.0
	Caviar from lumpfish roe	Germany	E151	E151	237.8	8.5

Ultrasound-assisted extraction (UAE) applying different solvents, such as water, methanol, acetonitrile, water-methanol, water-acetonitrile, acidified water-acetonitrile or alkalized water-acetonitrile, was tested for initial dyes extraction before SPE on Strata X-AW. Nevertheless, all of preliminarily tested condition (including also different variants in terms of extraction duration and temperature) failed to extract colourants with high recovery rates from caviar/fish roe samples. This problem could be explained by the fact that the dyes bind strongly to the food matrix (e.g. to proteins). To overcome these limitations, MSPD with Strata X-AW sorbent is proposed.

Intra- and inter-day accuracy and precision as well as intralaboratory reproducibility was evaluated for the proposed MSPD procedure (Table 4). Samples were spiked with mixture of 11 dyes at 10 µg g⁻¹ level and left at room temperature for 30 min in order to allow dyes to bind to the matrix and to evaporate the solvent (methanol). Chromatogram obtained after MSPD extraction of spiked fish roe sample is presented (Fig. 3). Intra-day accuracy (Recovery %) and precision (RSD%) were from 75.6 to 99.7% and 2.8 to 13.7% (day 1; n = 6), from 76.4 to 99.1% and 1.7 to 16.7% (day 2; n = 6) and from 76.8 to 100.4% and 4.9 to 18.7%, respectively. On the basis of all 18 replicates analysed in 3 days (n = 18), inter-day repeatability was calculated obtaining recovery in the range of 76.3-99.7% with RSD% values ranging from 3.6 to 15.3%. As for intra-laboratory reproducibility (experiments conducted by two different analysts, n = 12 in total), recoveries were in the range of 83.9-102.8% with RSD% values ranging from 3.9 to 17.2%. In total, 30 different samples were analysed with overall accuracy and precision ranging from 79.3 to 101.0% and 3.7 to 16.6%, respectively (Table 4(a)). The validation study was extended for Brilliant Black PN (E151) and Brilliant Blue FCF (E133) (Table 4(b)), which were expected to be found in fish roe samples bought from the market. Overall recovery (n = 30) of Brilliant Black PN from samples spiked at 20, 50 and 100 μ g g⁻¹ level were 85.7% with RSD% 7.4%; 85.8% with RSD% 7.1; and 90.6% with RSD% 5.6%, respectively. Similarly, overall recovery (n = 30) of Brilliant Blue FCF from samples spiked at 20, 50 and 100 μ g g⁻¹ level were 91.1% with RSD% 9.4%; 85.1% with RSD% 4.2; and 84.6% with RSD% 3.9%, respectively (Table 4(b)). Finally, the procedure was employed for analysis of coloured fish roe products available to consumers. Chromatograms and correlations between spectra of library standards and detected

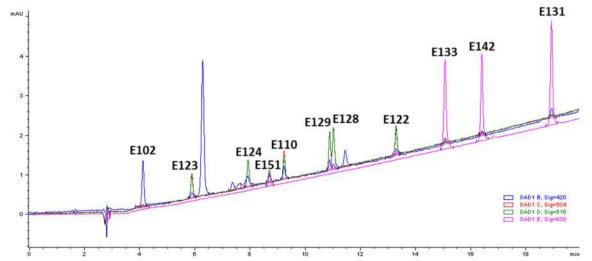


Fig. 3 Chromatogram of fish roe sample spiked at 10 μ g g⁻¹ level (overlaid signals from 420, 504, 515 and 630 nm wavelengths) after proposed MSPD extraction procedure

400 500 500 700 800

9.3

300

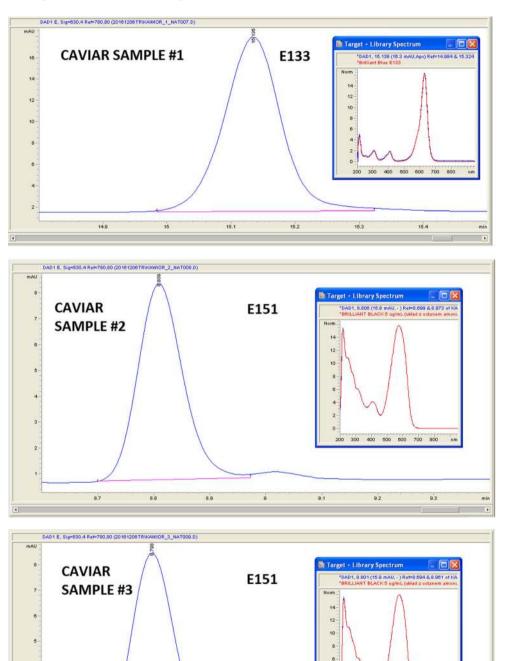
9.2

9.1

peaks is shown (Fig. 4). Each product samples were extracted and analysed via HPLC-DAD in triplicates and on this basis concentration of E133 and E151 dyes per 1 g of foodstuff were calculated and are presented in Table 6. Detected amounts of dyes in fish roe samples (from 67.3 to 237.8 μ g g⁻¹) are lower than maximum level (300 μ g g⁻¹) established and approved by EU Regulation (EC) No 94/36 (1994) for this kind of food product.

To the authors best knowledge, this is the second analytical procedure devoted to artificial colourants analysis in caviar/

Fig. 4 MSPD-HPLC-DAD chromatogram regions (630 nm) of selected from the market fish roe samples with identified peaks of Brilliant Blue FCF (E 133) and Brilliant Black PN (E 151). Identification of dyes was accomplished on the basis of retention times and spectra correlation between library standards and spectra within detected peaks (shown on figures). Caviar sample #1caviar from lumpfish roe (Sweden); Caviar sample #2caviar from capelin roe and herring roe (Sweden); and Caviar sample #3-caviar from lumpfish roe (Germany). Average concentration ($\mu g g^{-1}$) and RSD% on the basis of triplicates are given in Table 6



8.7

8,8

8.9

9

fish roe samples. According to Web of Science search (queries in topic area: 'caviar and dye' or 'fish roe and dye' or 'caviar and colourants'), only one article by Kirschbaum et al. is found (Kirschbaum et al. 2006). The method elaborated by the authors relays on ultrasound-assisted extractions with 1 M aqueous ammonia followed by extraction of dyes using polyamide A sorbent. The extraction steps were repeated until the resulting aqueous layer (1st step) or polyamide (2nd step) were colourless (for more details, please see referenced article) (Kirschbaum et al. 2006). This all makes the procedure reported by Kirschbaum et al. more labour-intense than the one developed in this study's MSPD approach.

Conclusions

In this study, we propose cost-effective and reliable analytical method for food dyes analysis. Ether-linked phenyl stationary phase provide satisfactory selectivity and chromatographic performance for 11 dyes separation using mobile phase of simple composition. Considering low values of LODs and LOQs obtained via HPLC-DAD, this relatively inexpensive technique may be useful and alternative platform to LC-MS/MS. Weak anion exchange SPE polymeric sorbent (STRATA X-AW) can be easily and successfully utilized in MSPD procedure for extraction of colouring factors in fish roe samples. Developed procedures were successfully applied for analysis of beverages, hard candies and fish roe samples available to consumers. In all cases detected, the amounts of dyes were in agreement with the maximum levels established by the European Union.

It is also worth to highlight that the scope of analytical procedures developed in this study may be easily expanded. SPE or MSPD extraction/clean-up procedures can be applied in order to cover broader spectrum of food types. In some cases, only initial extraction of dyes should be taken into optimization.

Compliance with Ethical Standards

Conflict of Interest Tomasz Tuzimski declares that he has no conflict of interest. Tomasz Rejczak declares that he has no conflict of interest

Ethical Approval This article does not contain any studies with human participants or animals performed by any of the authors.

Informed Consent Not applicable.

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