

1 **Assessment of a healthy oil combination structured in ethyl cellulose and beeswax**  
2 **oleogels as animal fat replacers in low-fat, PUFA-enriched pork burgers**

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9 **ABSTRACT**

10 The present work evaluates the suitability of ethyl cellulose and beeswax oleogels  
11 prepared with a healthy lipid mixture (olive, linseed, and fish oils) as fat replacers for  
12 fresh meat product development. The texture, color, thermal properties, and fatty acid  
13 composition of the oleogels indicated their suitability for the intended use, and they  
14 were stable for at least one month of chilled storage ( $3 \pm 1$  °C). However, the oleogels  
15 suffered some lipid oxidation during refrigerated storage, especially in the case of ethyl  
16 cellulose. Low-fat pork burgers formulated with total substitution of pork backfat by the  
17 oleogels developed were softer and without important changes in optical properties, as  
18 compared to the control. Although some lipid oxidation was observed, especially when  
19 ethyl cellulose oleogel was used, the fatty acid profile of the reformulated burgers was  
20 significantly improved, with a 3.6-fold increase of the PUFA/SFA ratio and a 23-fold  
21 decrease of the n-6/n-3 ratio, as compared to the control. A sensory acceptability test  
22 showed high ratings for the burgers made with beeswax oleogel, in contrast to the ones  
23 made with ethyl cellulose, which scored values below the neutral point. Results from  
24 this work indicate the potential of the ingredients developed for the formulation of

25 healthier fresh meat products with an improved fatty acid profile, and the need for  
26 research on strategies to improve oxidative stability and sensory properties.

27 **Keywords:** functional food; meat product; PUFA; n-3; reformulation

28

## 29 **Introduction**

30 In response to recommendations for optimal intake of total and unsaturated fatty acids  
31 proposed by a number of scientific authorities and nutritional organizations (McNeill  
32 2014; Anonymous 2003), various approaches have been considered to optimize the lipid  
33 contents and fatty acid profiles of various foods, including meat products, in order to  
34 achieve a composition more in line with nutrient intake goals. In this regard, various  
35 oils (of plant and marine origin) have been widely used as animal fat replacers in meat  
36 product reformulation strategies and to produce healthier lipid content in meat-based  
37 functional foods (Domínguez et al. 2017; Barbut et al. 2016a; Delgado-Pando et al.  
38 2014; Salcedo-Sandoval et al. 2015; Muguerza et al. 2004; Jiménez-Colmenero 2007).  
39 However, as the functionality and texture of the solid animal fat present in meat  
40 products are quite different from the characteristics of oil (liquid at room temperature),  
41 and these properties have a major effect on several product characteristics (mouthfeel,  
42 juiciness, texture, bite, heat transfer, etc.), the replacement of animal fat with liquid oils  
43 presents a considerable technical challenge, often resulting in a negative effect on the  
44 desired quality attributes of the reformulated products (Grasso et al. 2014; Muguerza et  
45 al. 2004).

46 One of the first developments in this topic was the addition of oils in pre-emulsified  
47 form, as in the work by Martínez et al. (2012), who developed beef burgers with a  
48 mixture of olive, corn and fish oils. Other structuration method that has been employed  
49 for meat products development is encapsulation. Thus, Jiménez-Martín et al. (2016)  
50 developed chicken nuggets containing microencapsulated omega-3 fish oil, achieving  
51 good results for oxidative stability and sensory quality. Enzymatic intersterification and  
52 structuration using mono- and diglycerides have been also employed as strategies to  
53 improve the physical performance of liquid fats (Chen et al. 2012; Lupi et al. 2012).

54 However, newer proposals for oil structuration have recently been considered in order  
55 to create a plastic fat that retains solid-like properties (very similar to those of animal  
56 fat, while possessing a healthier fatty acid profile), and that can be used as an approach  
57 to improving the fat content of meat products (Jimenez-Colmenero et al. 2015).  
58 Organogelation is one of the most promising techniques to give liquid oils solid-fat  
59 functionality. Various organogelators have been used to obtain organogels (oleogels),  
60 including small molecules that crystallize to form colloidal or fibrillar networks, and  
61 hydrophobic polymers that self-assemble under specific processing conditions  
62 (Davidovich-Pinhas et al. 2016). Waxes and ethyl cellulose (EC) are examples of these  
63 two organogelation-lipid structuring systems, respectively. Wax esters previously  
64 solubilized into liquid oil may crystallize upon cooling, trapping the oil molecules in a  
65 structured network. The properties of the resulting oleogels depend on the type of wax  
66 used (candelilla, carnauba, rice bran, beeswax, etc.) and concentration levels (Martins et  
67 al. 2016; Yilmaz and Ögütçü 2014; Ögütçü and Yilmaz 2014; Mert and Demirkesen  
68 2016; Zetzl and Marangoni 2011; Ögütçü et al. 2015). EC oil structuring is based on the  
69 ability of polymers to dissolve in the oil phase while the temperature is increased above  
70 its glass transition, approximately  $\sim 140^{\circ}\text{C}$ , followed by cooling below the gel point  
71 temperature, approximately  $\sim 120^{\circ}\text{C}$ .

72 EC oleogels have been used to replace animal fat in the reformulation of fresh and  
73 cooked meat products such as frankfurters, breakfast sausages, and pâtés (Zetzl et al.  
74 2012; Barbut et al. 2016a, b, c; Gómez-Estaca et al. 2019). The characteristics of  
75 oleogels, and therefore their suitability to be used as animal fat replacers, depends on  
76 factors associated basically with the nature of the compound (polymer molecular  
77 weight), concentration, presence of surfactants, and type of oil (Zetzl et al. 2012; Barbut  
78 et al. 2016a; Gravelle et al. 2014). The use of wax oleogels as fat substitutes in meat

79 products is not common and is very recent. Moghtadaei et al. (2018) formulated beef  
80 burgers in which animal fat was partially replaced by sesame oil-beeswax oleogels,  
81 finding a reduction in cooking loss and fat absorption, although cooking shrinkage and  
82 lipid oxidation increased significantly. Wolfer et al. (2018) substituted pork backfat  
83 with soybean oil oleogels structured with rice bran wax in the formulation of  
84 frankfurter-type sausages, with good results for textural and sensory properties.

85 The characteristics and stability of new lipid materials are fundamental for  
86 understanding and improving their suitability as animal fat replacers in meat products.  
87 The characteristics need to be considered in order to modulate their composition  
88 (healthier lipid presence), understand their role in the meat matrix, and achieve the  
89 appearance and the technological, rheological, and sensory properties required for use  
90 as raw materials to replace animal fats. Further studies are needed because, whether  
91 they are used as non-meat ingredients or as part of meat products, these novel materials  
92 can be affected by various common processing treatments such as chilling storage,  
93 which can alter both their technological suitability and the quality attributes of  
94 reformulated meat products. Therefore, the objective of the present work is to evaluate  
95 the suitability of ethyl cellulose and beeswax oleogels prepared with an olive-linseed-  
96 fish oil mixture as fat replacers in low-fat, PUFA-enriched pork burgers, considering  
97 aspects related to the lipid ingredients (fatty acid profile, texture, thermal properties, and  
98 chilling stability) and the products (composition, technological and sensory properties).

## 99 **Materials and Methods**

### 100 **Materials and reagents**

101 The lipid sources for oleogel development were: olive oil (Carbonell Virgen Extra, SOS  
102 Cuétara SA, Madrid, Spain), linseed oil (Natursoy SL, Alimentos Ecológicos,

103 Castellterçol, Spain), and fish oil (Omevital 18/12 TG Gold) from Cognis GmbH,  
104 Illertissen, Germany. Ethyl cellulose with a viscosity of 10 cP (Aqualon N10 EC) was  
105 kindly donated by Ashland (The Netherlands). Beeswax was acquired from Manuel  
106 Riesgo SA (Madrid, Spain). Sorbitan monostearate was from Sigma-Aldrich (Madrid,  
107 Spain). Raw materials, pork meat, and backfat were acquired and prepared as described  
108 by Salcedo-Sandoval et al. (2015). Finally, the spice mixture and salt were obtained at a  
109 local market.

### 110 **Oleogel development**

111 Two different oleogels were prepared, using ethyl cellulose (EC-OG) and beeswax (W-  
112 OG) as organogelators. In both, the lipid phase was prepared to attain an optimal lipid  
113 profile from a health standpoint (Delgado-Pando et al. 2010), and consisted of a mixture  
114 of olive, linseed, and fish oils in a proportion of 44.39%, 37.87%, and 17.74%,  
115 respectively. Ethyl cellulose oleogel (EC-OG) and beeswax oleogel (W-OG) were  
116 prepared on the basis of the method described by Gómez-Estaca et al. (2019).

### 117 **Burger formulation**

118 Three batches of burgers, each batch weighing 3.5 kg, were prepared as described by  
119 Salcedo-Sandoval et al. (2015), with slight modifications. All of them contained 84%  
120 meat, 8% water/ice mixture, 1% salt, 1% spice mixture (black pepper, white pepper,  
121 dried garlic, dried onion), and 6% fatty ingredient: pork backfat (C-B batch), ethyl  
122 cellulose oleogel (EC-B), or beeswax oleogel (W-B). For the preparation of the burgers,  
123 meat and pork backfat packages were thawed (approx. 18 h in refrigeration, reaching  
124  $\approx -5$  °C), and the oleogels were placed in a freezer ( $-18 \pm 1$  °C) for approximately 2 h  
125 before burger preparation. Meat and fatty ingredients (pork backfat or oleogel) were  
126 minced through a grinder with a 4.5 mm plate (Van Dall Srl, model FTSIII, Treviglio,

127 Italy). For each formulation, the chopped meat and the fatty ingredient were placed in a  
128 mixer (MAINCA, Spain) and homogenized for 1 min. The other ingredients were added  
129 and mixed for 1 min more. The mince temperature was controlled (Daqpro-5300  
130 thermocouple from Omega, Spain) during the process and was always below 3 °C.  
131 Burgers weighting  $\approx 85$  g were prepared using a manual burger former, packaged in  
132 trays under air atmosphere, and stored at  $2 \pm 2$  °C until analysis (Figure 1).

### 133 **Oleogel characterization**

#### 134 *Fatty acid determination*

135 Ten mg of the oil mixture or oleogel was derivatized into fatty acid methyl esters  
136 (FAMES) in triplicate using 0.5 M sodium methoxide in anhydrous methanol and acetyl  
137 chloride in anhydrous methanol. FAMES were extracted with 4 mL hexane and used for  
138 GC analysis (1  $\mu$ L). The fatty acid profile was determined in an Agilent 7820A gas  
139 chromatograph with FID detector. Separation was performed in an Agilent HP-88  
140 column (60 m, 0.32 mm i.d., 0.25  $\mu$ m film thickness, ref. 112-8867) with split injection  
141 (40:1) and helium at a constant flow of 1.2 mL/min. Detector temperature was set at 260  
142 °C and injector temperature at 250 °C. The temperature profile of the oven was 125 °C  
143 for 1 min, then increased by 8 °C/min to 145 °C for 26 min, then increased to 220 °C  
144 for 5 min. Identification was done by comparing retention times with a standard of 37  
145 fatty acids (Supelco 37 FAME Mix 47885-U, USA). The internal standard used for  
146 quantification was C13:0, which was added to the sample in the non-methylated state  
147 before methylation. Results were expressed as mg fatty acid/g oil.

#### 148 *Instrumental color and texture analysis*

149 Color parameters were measured (ten times) using a Konica Minolta CM-3500d  
150 spectrophotometer (Konica Minolta Sensing, Inc., Osaka, Japan) set to D65  
151 illuminant/10° observer. The CIELAB color space was used to obtain the color  
152 coordinates L\* [black (0) to white (100)], a\* [green (–) to red (+)], and b\* [blue (–) to  
153 yellow (+)]. Determinations were performed at different time intervals during 28 days  
154 of chilled storage ( $3 \pm 1$  °C).

155 Texture analysis was performed in a TA-XTplus Texture Analyzer (Stable Micro  
156 Systems Ltd., Godalming, UK) equipped with a 50 N load cell. A penetration test was  
157 carried out at room temperature on each sample placed in a cylindrical-shaped container  
158 (35 mm in diameter and 30 mm high), immediately after refrigeration at 3 °C (six  
159 replications). The analysis was performed with a 12 mm diameter flat probe that  
160 penetrated 10 mm into the sample at a velocity of 0.8 mm/s. The penetration force (PF,  
161 N) was calculated as the force exerted at 10 mm (for the EC-OG sample) or the force at  
162 the point of gel fracture (for the W-OG sample), according to Herrero et al. (2011).  
163 Determinations were performed at different time intervals during 28 days of refrigerated  
164 storage ( $3 \pm 1$  °C).

#### 165 *Lipid oxidation*

166 Thiobarbituric acid reactive substances (TBARS) of the oleogels were determined in  
167 triplicate, based on the method described by Maqsood and Benjakul (2010), with  
168 modifications. One gram of finely comminuted oleogel was dispersed with 3 mL of  
169 0.26% butylhydroxytoluene (BHT) by sonication (Q-Sonica sonicator equipped with a  
170 microtip) at 75 Hz for 5 s; the test tube containing the sample was immersed in an ice  
171 bath to avoid heating. Afterwards, 8 mL of TBARS reagent (15% trichloroacetic acid,  
172 0.375% thiobarbituric acid, 0.25% hydrochloric acid) was added, and the mixture was



173 vortexed and sonicated again (75 Hz, 5 s). The test tubes were placed in a water bath set  
174 at 100 °C for 15 min and then immediately transferred to an ice-water bath to cool  
175 down. Afterwards, 4 mL of 4 M ammonium sulfate and 4 mL of hexane were  
176 sequentially added and the mixture was vigorously vortexed for 30 s. The lower phase  
177 was collected after centrifugation in a Heraeus Multifuge 3L centrifuge (DJB Labcare  
178 Ltd., Buckinghamshire, England) set at 5000 rpm/10 °C/30 min, and the absorbance at  
179 532 nm was measured in a Shimadzu UV-1800 spectrophotometer (Shimadzu Inc.,  
180 Kyoto, Japan). Results were expressed as mg malonaldehyde (MDA)/kg oil, based on a  
181 standard curve prepared from 1,1,3,3-tetraethoxypropane in advance. Determinations  
182 were performed at different time intervals during 28 days of refrigerated storage ( $3 \pm 1$   
183 °C).

#### 184 *Microscopy*

185 Oleogel morphology was studied by optical microscopy with a Leica AF6000 LX  
186 microscope. After organogelator dissolution and prior to setting, samples were poured  
187 directly into the holders and then allowed to gel at room temperature in the same  
188 conditions as described above. Micrographs were taken at 40× magnification after two  
189 days of refrigerated storage ( $3 \pm 1$  °C).

#### 190 *Differential Scanning Calorimetry (DSC)*

191 DSC analysis of the oleogels was performed with a previously calibrated model TA-  
192 Q1000 Differential Scanning Calorimeter (TA Instruments, New Castle, DE, USA).  
193 Pure ethyl cellulose, pure beeswax, and oleogel samples of approximately 5–8 mg ( $\pm$   
194 0.002 mg) were weighed out using a model ME235S electronic balance (Sartorius,  
195 Goettingen, Germany), tightly encapsulated in hermetic aluminum pans, and scanned  
196 under dry nitrogen (50 mL/min) purge. An empty capsule was used as a reference. A

197 heating ramp and a subsequent cooling ramp were performed from 5 to 100 or 200 °C  
198 (W-OG and EC-OG samples, respectively) at 5 °C/min. A second run was also  
199 performed in order to check the reversibility of thermal events. Melting temperatures  
200 ( $T_m$ , °C), crystallization temperatures ( $T_c$ , °C), and transition enthalpies ( $\Delta H$ , J/g, by  
201 linear baseline integration) were calculated. Samples were analyzed in triplicate.  
202 Determinations were performed at different time intervals during 28 days of refrigerated  
203 storage ( $3 \pm 1$  °C).

## 204 **Burger characterization**

### 205 *Composition*

206 Proximate analysis was carried out in triplicate. Moisture and ash content of the burgers  
207 were determined according to AOAC (2005). Protein content was measured with a  
208 LECO FP-2000 Nitrogen Determinator (Leco Corporation, USA), and fat content was  
209 evaluated following the method described by Bligh and Dyer (1959).

210 Fatty acid composition was determined in freeze-dried burgers as described previously  
211 for the oleogels. Results were expressed as g fatty acid/100 g burger.

### 212 *Technological properties*

213 Color parameters were measured (ten times) after one day of preparation and in  
214 refrigerated conditions in raw burgers as described for the oleogels. For the burgers,  
215 simple transformations were used to convert  $a^*$  and  $b^*$  coordinates to  $C^*$  and  $h^\circ$   
216 chromatic parameters (Gómez-Estaca et al. 2015).

217 For texture analysis, Kramer shear force (KSF) was determined (six replications) in a  
218 TA-XTplus Texture Analyzer (Stable Micro Systems Ltd., UK) equipped with a 50 N

219 load cell using a miniature Kramer/Ottawa cell (HDP/MK05). Samples measuring 2×2  
220 cm were cut from two burgers, accurately weighed, and placed into the cell, at room  
221 temperature. Samples were penetrated 20 mm, at a speed of 0.8 mm/s. KSF values were  
222 calculated as the maximum force per g of sample (N/g).

223 For lipid oxidation analysis, TBARs were determined in triplicate using the method  
224 described by Delgado-Pando et al. (2012). Briefly, 4 g of each sample was placed in  
225 centrifuge tubes, 1 mL of distilled water and 10 mL of 10% trichloroacetic acid were  
226 added, and the mixture was homogenized for 30 s with a vortex stirrer. Then 5 mL of 20  
227 mM 2-thiobarbituric acid was added and the mixture was stirred again for 30 s. Each  
228 tube was centrifuged for 5 min at 2600 g (Heraeus Multifuge 3 L-R, DJB Labcare Ltd.,  
229 Buckinghamshire, UK), and the supernatant was collected and kept in darkness for 20 h  
230 at 20 ± 2 °C. The absorbance at 532 nm was measured in a Shimadzu UV/VIS 1203  
231 spectrophotometer, and the results were expressed as mg malonaldehyde (MDA)/kg  
232 burger, based on a standard curve prepared from 1,1,3,3-tetraethoxypropane in advance.

### 233 *Sensory analysis*

234 A hedonic sensory analysis was performed by an untrained panel of 40 people selected  
235 from staff of ICTAN-CSIC, in two independent sessions. Immediately after being  
236 cooked on a grill plate for 4 min (2 min per side), each burger was cut into 4 pieces and  
237 the three samples (C-B, EC-B, W-B) were presented to the panelists, who were  
238 instructed to rinse their mouth with bread and water between samples. Odor, color,  
239 texture, and overall acceptability were evaluated on a 10-point scale, 0 being considered  
240 as “dislike strongly” and 10 as “like strongly.” The panelists were also asked to make  
241 any comments they considered appropriate about their sensory perception of the  
242 samples.

## 243 **Statistical analysis**

244 Statistical tests were performed using the SPSS computer program (SPSS Statistical  
245 Software, Inc., Chicago, IL, USA). One-way and/or two-way analyses of variance  
246 (ANOVA) were conducted. Differences between pairs of means were assessed on the  
247 basis of confidence intervals using the Tukey-b test. The level of significance was  $p \leq$

## 248 **0.05. Results and Discussion**

### 249 **Fatty acid composition, characterization, and chilling stability of oleogels**

250 A study of the fatty acid composition is not generally conducted in the literature,  
251 because it seems evident that it corresponds to the composition of the lipid material of  
252 which it consists, but in this case it is useful to make this analysis because the  
253 preparation conditions required (high processing temperatures) may induce changes.  
254 The fatty acid profiles of the oleogels developed and the oil mixture are shown in Table  
255 1. The fatty acid composition of the oil mixture was similar to that previously reported  
256 by Delgado-Pando et al. (2010), showing 15.5% saturated fatty acid (SFA), 48.5%  
257 monounsaturated fatty acid (MUFA), and 36% polyunsaturated fatty acid (PUFA)  
258 according to the composition of the oils in the mixture (olive, linseed, and fish). The  
259 fatty acid profile was scarcely affected by organogelation, the main change being an  
260 increase in  $\Sigma$ SFA ( $p \leq 0.05$ ). A possible explanation is that these fatty acids were  
261 formed from the degradation of other fatty acids, not only by oxidative reactions, but  
262 also by decarboxylation or degradation by carbon-carbon cleavage (Nawar 1969), as the  
263 content of  $\Sigma$ MUFA and  $\Sigma$ PUFA decreased concomitantly, although the decrease was  
264 not significant ( $p > 0.05$ ). In consonance with these findings, Tenyang et al. (2017)  
265 found that roasting two varieties of sesame (120 °C/30 min) caused a decrease in  
266  $\Sigma$ PUFA and/or  $\Sigma$ MUFA and a related increase in  $\Sigma$ SFA. Along the same lines, Blasi et

267 al. (2018) reported an increase in  $\Sigma$ SFA of olive oil during frying. As a result of these  
268 fatty acid modifications, the  $\Sigma$ PUFA/ $\Sigma$ SFA ratio decreased ( $p \leq 0.05$ ) for both oleogels,  
269 especially for EC-OG, but the  $\Sigma$ PUFAn-6/ $\Sigma$ PUFAn-3 ratio remained unchanged. As the  
270 changes in the fatty acid profile were small, the lipid composition of the oleogels  
271 developed was optimal from a nutritional standpoint, with a view to their intended use  
272 as ingredients for the development of healthier meat products.

273 A fuller knowledge of oleogel characteristics (appearance, morphology, mechanical  
274 properties, thermal behavior, oxidation, etc.) will therefore facilitate their use, help to  
275 elucidate their role in the protein matrix structure, and help to improve the quality of  
276 healthy meat-based food systems in which they are used. Both oil-structuring methods  
277 produced solid-like structures that showed a yellowish color to the naked eye (Figure 2).  
278 The objective measure of color showed that W-OG was lighter (Table 2) and more  
279 yellowish ( $p \leq 0.05$ ), whereas EC-OG showed higher redness ( $p \leq 0.05$ ). With regard to  
280 the effect of storage time, no changes ( $p > 0.05$ ) were observed, irrespective of the type  
281 of organogelator used, indicating the stability of the oleogels developed in terms of the  
282 optical properties considered. For this reason and for further simplification, Table 2  
283 shows the mean value of all the determinations performed during storage.

284 The microscopic analysis of the oleogels is shown in Figure 3. EC-OG had a compact,  
285 granular structure in which no crystalline structures were observed (2A). This was  
286 especially evident in the photograph taken under polarized light (2C). Zetzl et al. (2012)  
287 reported that the ethyl cellulose gel consists of an extensive polymer network with small  
288 pockets or holes where oil would be entrapped; the image observed here may be  
289 consistent with the microstructure reported there. The beeswax oleogel had needle-like  
290 structures (2B and 2D), which are characteristic of oleogels prepared with this type of  
291 wax (Yilmaz and Öğütçü 2014; Martins et al. 2016). The gelation process of waxes is

292 the result of the association of microcrystalline structures that build up a three-  
293 dimensional network that, if strong enough, is capable of restraining the oil phase,  
294 resulting in an oleogel (Toro-Vazquez et al. 2007). Needle-like microcrystals enable  
295 large volumes of oil to become entrapped between the crystalline strands, unlike other  
296 waxes such as candelilla or carnauba, which form smaller spherulitic crystals (Zetzl and  
297 Marangoni 2011).

298 Initially, EC-OG showed higher ( $p < 0.05$ ) oxidation levels than W-OG (Figure 4),  
299 probably owing to the higher temperatures (up to 170 °C) needed to unfold ethyl  
300 cellulose for gel formation (Gravelle et al. 2012), in contrast to W-OG, which is  
301 produced at 65 °C, a temperature that is sufficient to ensure effective wax melting and  
302 mixing with the oil. The two samples had similar oxidation rates during storage, with a  
303 progressive accumulation of TBARS that was significantly different from the initial  
304 point after 28 days of storage and with higher values for EC-OG, although the oxidation  
305 values were below the reported minimum needed to detect objectionable flavors  
306 (Ockerman 1985). Lipid oxidation is an important aspect to be taken into consideration  
307 when producing plastic fats from unsaturated oils owing to the susceptibility of these  
308 kinds of lipids to oxidation and to the high processing temperatures used, especially in  
309 the case of ethyl cellulose oleogels. Surprisingly, there is very little literature about the  
310 chilling stability of these materials, which can be processed in a similar way to other  
311 raw materials used in food (meat) processing. Gravelle et al. (2012) reported a  
312 progressive accumulation of lipid oxidation products (hydroperoxides and TBARS)  
313 when canola oil was heated to 140 °C during a 120 min experiment. They found that  
314 holding heating times beyond 20 min resulted in an accumulation of  $\approx 10$  meq  
315 hydroperoxides/kg oil, which, according to the authors, is the upper limit for oil to be  
316 considered “fresh.” On the basis of these results, the authors stated that a processing

317 time of 15–20 min would be optimum for ethyl cellulose unfolding and mixing with the  
318 oil while limiting lipid oxidation. However, to date no studies have been done on the  
319 stability of ethyl cellulose oleogels during storage. Yilmaz and Öğütçü (2014) found  
320 that beeswax-structured hazelnut oil stored at 4 °C did not oxidize after 3 months  
321 (according to the peroxide value), whereas lipid oxidation was observed when it was  
322 stored at 20 °C.

323 For a better understanding of the mechanical properties of ethyl cellulose and beeswax  
324 oleogels, examples of the typical penetration curves obtained are plotted in Figure 5.  
325 EC-OG showed a typical viscoelastic pattern, with an elastic region followed by a  
326 plateau, whereas W-OG behaved quite differently, with plunger penetration producing a  
327 breaking point characteristic of gel fracture. W-OG attained a significantly ( $p \leq 0.05$ )  
328 higher penetration force (measured at the breaking point,  $11.7 \pm 2.3$  N) than EC-OG  
329 (measured at 10 mm of penetration,  $6.5 \pm 0.8$  N). These results are in accordance with a  
330 previous work in which a rheological characterization of these oleogels was performed,  
331 finding higher values of complex modulus ( $G^*$ ) and lower values of  $\gamma_{\max}$  for beeswax  
332 oleogels, which indicated that beeswax produced denser but less deformable gels than  
333 those produced with ethyl cellulose (Gómez-Estaca et al. 2019).

334 Other researchers working with ethyl cellulose oleogels observed similar mechanical  
335 behavior to that observed in this work: using back extrusion, they found a plateau zone  
336 in which the force was constant from a certain penetration depth onwards; however,  
337 texture profile analysis revealed elastic behavior (Zetzl et al. 2012). In other works, the  
338 plateau zone was not observed when the mechanical properties of ethyl cellulose  
339 oleogels were determined, the gel being clearly fractured as the probe penetrated  
340 (Gravelle et al. 2014; Gravelle et al. 2013). These different results may be related to the  
341 fact that the mechanical properties of ethyl cellulose oleogels are strongly dependent on

342 processing and compositional parameters such as processing time and temperature,  
343 heating/cooling rates, ethyl cellulose molecular weight and concentration, the presence,  
344 type, and concentration of plasticizer, and the degree of oil unsaturation (Gravelle et al.  
345 2014; Davidovich-Pinhas et al. 2015; Gravelle et al. 2012). As in the case of lipid  
346 oxidation, studies of the effect of chilling storage on EC oleogel mechanical properties  
347 have not been reported. After 28 days of chilled storage, neither the mechanical profile  
348 nor the maximum penetration force or gel strength had changed, irrespective of the type  
349 of organogelator used (results not shown), bearing witness to the high stability of the  
350 oleogels developed. In consonance with this experiment, Yilmaz and Ögütçü (2014),  
351 who studied the effect of storage time (up to 3 months) on the mechanical properties of  
352 hazelnut oil-beeswax oleogels, did not find significant changes, although some  
353 fluctuations were observed during the storage period.

354 The thermal transitions of the oleogels were studied by DSC, owing to the importance  
355 of their physical state on the behavior during production, storage, and cooking of the  
356 meat product (Figure 6). The oil mixture, which was analyzed as a control, exhibited an  
357 exothermic event at  $151.5 \pm 10.4$  °C (Figure 6A) in the first heating scan, which did not  
358 appear either in the cooling or the second heating scans (data not shown) and is  
359 attributable to partial thermal degradation. EC-OG showed a similar trend, with an oil  
360 thermal degradation peak at  $147.5 \pm 9.8$  °C (Figure 6A). The same behavior was  
361 observed by Dey et al. (2011) working with flaxseed oil structured in 22 cP ethyl  
362 cellulose oleogel. In that case, the thermal degradation peak was found at 130–135 °C  
363 for unstructured oil, whereas it appeared at lower temperatures in the oleogel (125–130  
364 °C). The absence of any other thermal transition event in EC-OG during heating or  
365 melting indicates that the gelation mechanism and structure do not involve a highly  
366 ordered secondary structure formation (Davidovich-Pinhas et al. 2015). Radically



367 different thermal behavior was observed in the beeswax oleogel, showing melting and  
368 crystallization peaks at  $54.2 \pm 0.1$  °C and  $49.0 \pm 0.2$  °C, respectively. The thermal  
369 behavior of the pure beeswax revealed a similar trend, with melting and crystallization  
370 peaks  $\approx 10$  °C higher ( $63.9 \pm 0.2$  °C and  $58.5 \pm 0.7$  °C, respectively) (traces not shown).  
371 Melting and crystallization enthalpies were also higher for pure polymer than for W-  
372 OG:  $155.2 \pm 3.2$  J/g and  $136.9 \pm 0.4$  J/g, respectively, for pure beeswax versus  $8.9 \pm 1.0$   
373 J/g and  $8.4 \pm 0.3$  J/g, respectively, for W-OG owing to a dilution effect of the polymer.  
374 It was observed that melting enthalpies are dependent on organogelator concentration,  
375 suggesting the possibility of adjusting the organogelator concentration depending on the  
376 intended application of the oleogel (Yilmaz and Öğütçü 2014; Yi et al. 2017). These  
377 results are in agreement with a previous report by Yilmaz and Öğütçü (2014), who  
378 found  $T_m$  and  $T_c$  peak values and enthalpies for beeswax and 10% hazelnut oil-  
379 beeswax oleogels quite similar to those found in this paper. As for the effect of storage  
380 time (28 days) on the thermal properties of the oleogels, no differences were observed  
381 for thermal profile or melting/crystallization temperatures or enthalpies (data not  
382 shown). Therefore, the ingredients developed could be stored for at least one month  
383 without significant changes in their thermal behavior. From these results it can be  
384 deduced that both organogels retain their solid-like properties at room temperature, but  
385 the beeswax oleogel will melt upon cooking or thermal processing. However, upon  
386 cooling, the oil will be structured again to form the oleogel.

### 387 **Suitability of oleogels as fat substitutes for healthier burgers**

388 According to the target formulation, generally the three burgers had a similar proximate  
389 composition (Table 3). Although some significant variations were observed in fat and  
390 ash contents, they can be considered of minor technological or nutritional importance  
391 owing to their small magnitudes.

392 The fatty acid profile of the burgers is shown in Figure 7. Significant differences were  
393 observed among the samples for all fatty acid groups. Thus,  $\Sigma$ SFA decreased and  
394  $\Sigma$ MUFA and  $\Sigma$ PUFA increased in the EC-B and W-B batches, as compared to the  
395 control batch formulated with animal fat (C-B). MUFAs were the most abundant fatty  
396 acid group in all samples, which is consistent with the lipid composition of the pork  
397 meat used. There was a notable increase in  $\Sigma$ PUFA in the EC-B and W-B samples,  
398 mainly attributed to linolenic acid, and this increase, together with a concomitant  
399 decrease in  $\Sigma$ SFA, resulted in an increase in the  $\Sigma$ PUFA/ $\Sigma$ SFA ratio by  $\approx$ 3.6-fold.  
400 Furthermore, the increase in the content of n-3 fatty acids in oleogel-added batches, as  
401 compared to the control, gave rise to a considerable decrease in the  $\Sigma$ PUFAn-  
402 6/ $\Sigma$ PUFAn-3 ratio ( $\approx$ 23-fold). These changes are consistent with the fat substitution  
403 level and the fatty acid composition of the oleogels developed (Table 1). According to  
404 European regulations (1924/2006 and 432/2012), nutritional claims may be made  
405 concerning burgers made with oleogels, such as *high  $\alpha$ -linolenic acid content* and *high*  
406 *omega-3 fatty acids content*, and the corresponding health claims, among others. There  
407 is abundant evidence associating a higher n-6/n-3 ratio with the promotion of  
408 pathogenesis of many diseases, including cardiovascular diseases, cancer, etc., and  
409 lower ratios with a suppressive effect (Simopoulos 2002). The results obtained are in  
410 line with a previous work in which functional pork patties were developed by  
411 substitution of animal fat by the same oil mixture as in the present work, structured in a  
412 konjac-based bulking system (Salcedo-Sandoval et al. 2014). However, the ratios  
413 obtained here were even better, owing to the higher animal fat substitution level. Similar  
414 results were obtained for cooked meat products (frankfurters and pâtés) in which animal  
415 fat was totally or partially replaced by an oil-in-water emulsion or ethyl cellulose or  
416 beeswax oleogels containing the same vegetable/fish oil combination as in the present

417 work (Delgado-Pando et al. 2010; Delgado-Pando et al. 2011; Gómez-Estaca et al.  
418 2019).

419 With regard to the effect of lipid substitution on the optical properties of the burgers, no  
420 significant differences in lightness were found between the samples ( $p > 0.05$ ), but  
421 higher ( $p \leq 0.05$ ) redness and yellowness were observed in the EC-B and W-B samples  
422 compared with the control sample (Table 4). In order to make a deeper analysis and  
423 interpretation of the optical properties of the burgers developed, hue angle and  
424 chromaticity were calculated, showing that the W-B sample did not differ significantly  
425 from the control formulated with animal fat in hue ( $p > 0.05$ ), but the color was  
426 significantly more intense. In the case of the EC-B sample, the hue shifted to lower  
427 values (nearer to the yellow region) ( $p \leq 0.05$ ) and the color was more intense than that  
428 of the control burger ( $p \leq 0.05$ ), indicating a higher impact of the ethyl cellulose oleogel  
429 on the optical properties of the burgers than the beeswax one. In any case, the changes  
430 in the optical properties were of low magnitude and of little importance from a  
431 technological point of view.

432 The Kramer shear force of the burgers developed is also shown in Table 4. The control  
433 burger was firmer than the one formulated in previous works (Freire et al. 2017;  
434 Salcedo-Sandoval et al. 2015), which is consistent with the lower fat content of the  
435 present formulation ( $\approx 7\%$  to  $\approx 15\%$ ). The incorporation of oleogels produced a decrease  
436 in shear force ( $p \leq 0.05$ ), without differences as a function of organogelator system ( $p >$   
437  $0.05$ ). The texture of restructured meat products is affected both by matrix  
438 characteristics (protein/water/fat composition) and by the physicochemical properties of  
439 the fat (Jiménez-Colmenero et al. 1995). The substitution of animal fat by fish or  
440 vegetable oils in burgers also resulted in a decrease in hardness (Keenan et al. 2015;

441 Lurueña-Martínez et al. 2004), whereas when oils were incorporated into a konjac oil  
442 bulking system no significant differences were found (Salcedo-Sandoval et al. 2015).  
443 There is little literature on the substitution of animal fat by oleogels in fresh  
444 comminuted meat products. Recently, Moghtadaei et al. (2018) developed beef burgers  
445 in which animal fat was partially substituted by sesame oil-beeswax oleogel, finding a  
446 decrease in hardness that was directly related to the substitution level in the raw  
447 samples. The authors attributed this to the lower fat globule size in oleogels as  
448 compared to animal fat. With regard to the use of ethyl cellulose oleogels as fat  
449 substitutes in meat products, the literature is limited to finely comminuted cooked  
450 products. Barbut et al. (2016a) formulated frankfurters in which beef fat was substituted  
451 by ethyl cellulose oleogels made from canola oil, obtaining similar hardness values in  
452 both samples. Similarly, Gómez-Estaca et al. (2019) did not find differences in pork  
453 liver pâtés formulated by partial or total substitution of pork backfat by ethyl cellulose  
454 or beeswax oleogels. From the few results found in the literature and those of the  
455 present work, it seems that the effect of the ethyl cellulose oleogel on texture is  
456 dependent on the type of product developed.

457 The oxidative status of the burgers developed was determined. The control sample  
458 showed the lowest value among the three samples, followed by the sample with  
459 beeswax oleogel and the one with ethyl cellulose (Table 4). This is consistent with the  
460 results of lipid oxidation of the oleogels shown in Figure 4, as the ethyl cellulose  
461 oleogel was the one that showed the highest lipid oxidation, probably owing to the  
462 higher processing temperature and time during the manufacturing process, as compared  
463 to the beeswax oleogel. These results indicate the need to make a deeper study of the  
464 oxidative stability of meat products containing oleogels with this oil mixture. Despite  
465 this, the oxidation values were relatively low and below the reported minimum needed

466 to detect objectionable flavors in processed meat products (Delgado-Pando et al. 2012),  
467 although it is expected that they would increase with storage time. A possible way to  
468 ameliorate this effect would be the addition of antioxidant compounds to the oleogels,  
469 as shown for other animal fat analogues containing unsaturated lipids (Flaiz et al. 2016).

470 The sensory evaluation of the cooked burgers is shown in Figure 8. The control sample  
471 showed a relatively good overall acceptability, in spite of its low fat content.  
472 Substituting animal fat by the oleogels developed had no effects on color and texture  
473 acceptability ( $p > 0.05$ ), and these two samples also obtained good ratings. With regard  
474 to flavor, animal fat substitution by the two oleogels developed had a negative effect ( $p$   
475  $\leq 0.05$ ); this was especially evident in the EC-B batch, which attained values below 5  
476 (considered as “neutral”). The overall acceptability of the burgers showed a similar  
477 trend, i.e., a lower acceptability in the reformulated samples than in the control ( $p \leq$   
478  $0.05$ ), but no differences were observed between the two oleogelation systems ( $p >$   
479  $0.05$ ). From these results it can be deduced that the oleogelation systems employed in  
480 the present work were successful for the development of reformulated burgers whose  
481 color and texture resemble those of a control product, but the lipid source has a major  
482 impact on flavor that ultimately impairs overall acceptability. Salcedo-Sandoval et al.  
483 (2015), who developed pork burgers by substitution of pork backfat by an oil bulking  
484 system based on konjac and including the same oil mixture as in the present work,  
485 found similar results and attributed them to the intrinsic sensory properties of the lipid  
486 material, especially fish oil. The negative effect on sensory properties could be  
487 mitigated by reformulating the seasoning. In a previous work in which pork liver pâtés  
488 were formulated by substituting pork backfat by the same oleogels as those used in the  
489 present work good acceptability was achieved, as compared to a sample without

490 substitution, probably owing to the addition of seasoning in the formulation (Gómez-  
491 Estaca et al. 2019).

## 492 **CONCLUSION**

493 These results demonstrate the suitability of ethyl cellulose and beeswax organogelation  
494 systems to structure an oil combination with an optimal fatty acid profile from a health  
495 standpoint into solid-like structures that behave in a similar way to animal fat when  
496 incorporated in a fresh comminuted product such as pork burgers. From this study and  
497 under the experimental conditions used, beeswax seems to be a better alternative than  
498 ethyl cellulose for the development of substitutes for animal fat to be employed in the  
499 reformulation of fresh comminuted meat products with an improved fatty acid profile.  
500 Despite this, further research on this topic is being conducted, focusing on improving  
501 the oxidative stability and sensory properties of beeswax and ethyl cellulose oleogels.

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686

687 **Figure captions**

688 Figure 1. Schematic drawing of burger preparation.

689 Figure 2. Pictures of ethyl cellulose (A) and beeswax (B) oleogels.

690 Figure 3. Microscopic images of ethylcellulose (A, C) and beeswax (B, D) oleogels.

691 Pictures C and D were taken under polarized light.

692 Figure 4. TBARS of ethyl cellulose (EC-OG) and beeswax (W-OG) oleogels as a  
693 function of storage time at  $3 \pm 1$  °C. Different letters indicate significant differences for  
694 each oleogel as a function of storage time.

695 Figure 5. Typical penetration force analysis curves of ethyl cellulose (EC-OG, solid  
696 line) and beeswax (W-OG, dotted line) oleogels.

697 Figure 6. DSC traces of ethyl cellulose (A) and beeswax (B) oleogels during heating  
698 (solid lines) and cooling (dotted lines) ramps (1 day after preparation). The DSC trace  
699 of the oil mixture during the heating ramp is also plotted (A).

700 Figure 7.  $\Sigma$ SFA,  $\Sigma$ MUFA,  $\Sigma$ PUFA, and ALA (g /100 g product) and nutritional ratios of  
701 the burgers developed.

702 Figure 8. Sensory evaluation of the burgers developed. C-B, control with pork backfat;  
703 EC-B, olive oil/linseed oil/fish oil structured with ethyl cellulose; W-B, olive oil/linseed  
704 oil/fish oil structured with beeswax.

705 **Table captions**

706 Table 1. Most abundant fatty acids (mg/g oil) and nutritional ratios of ethyl cellulose  
707 (EC-OG) and beeswax (W-OG) oleogel samples compared to those of the oil mixture.

708 Table 2. Optical properties (lightness,  $L^*$ ; redness,  $a^*$ ; yellowness,  $b^*$ ).

709 Table 3. Proximate composition (%) of burgers formulated with different lipid sources  
710 or oil structuring methods: C-B, control with pork backfat; EC-B, olive oil/linseed  
711 oil/fish oil structured with ethyl cellulose; W-B, olive oil/linseed oil/fish oil structured  
712 with beeswax.

713 Table 4. Some physicochemical properties of the burgers developed: luminosity (L\*),  
714 redness (a\*), yellowness (b\*), hue angle, chromaticity, Kramer shear force, and  
715 TBARS. Batch denominations as per Table 3.

716

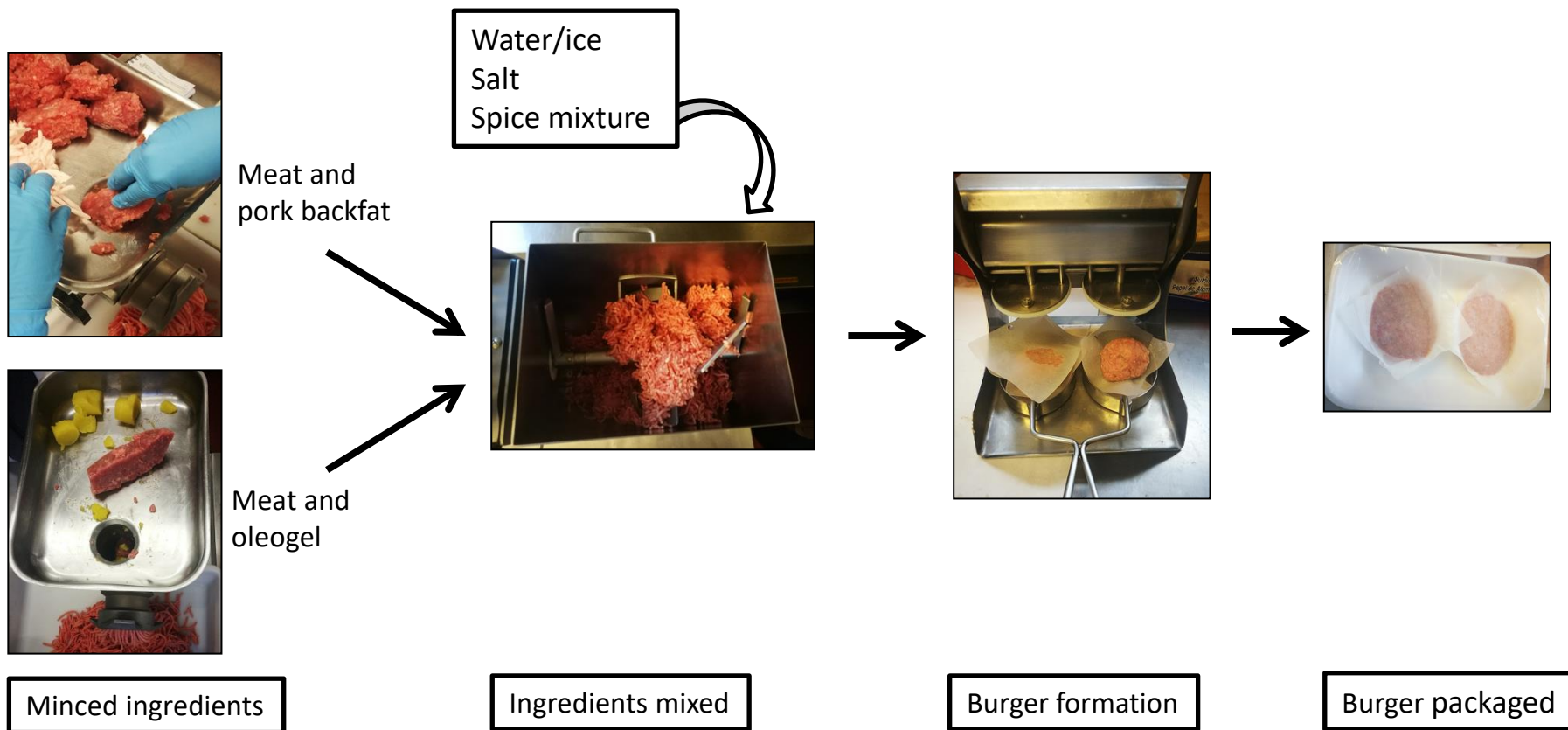


Figure 1. Schematic drawing of burger preparation

Table 1. Most abundant fatty acids (mg/g oil) and nutritional ratios of ethylcellulose (EC-OG) and beeswax (W-OG) oleogel samples compared to those of the oil mixture.

	<b>Oil mixture</b>	<b>EC-OG</b>	<b>W-OG</b>
<b>Palmitic C16:0</b>	87.8 ± 1.8a	102.9 ± 0.2c	98.1 ± 1.7b
<b>Stearic C18:0</b>	38.3 ± 0.8a	53.0 ± 0.04b	38.4 ± 0.6a
<b>ΣSFA</b>	147.2 ± 2.8a	177.1 ± 0.1c	163.1 ± 2.5b
<b>Oleic C18:1n9c</b>	421.6 ± 8.7a	412.4 ± 0.6a	412.6 ± 7.1a
<b>ΣMUFA</b>	459.3 ± 9.8a	449.8 ± 1.4a	452.2 ± 8.1a
<b>Linoleic C18:2n6c</b>	82.9 ± 1.7a	81.0 ± 0.2a	80.9 ± 1.3a
<b>Linolenic C18:3n3</b>	209.2 ± 4.6a	203.4 ± 0.5a	204.4 ± 3.5a
<b>EPA C20:5n3</b>	28.9 ± 0.6a	28.1 ± 0.1a	28.4 ± 0.5a
<b>DHA C22:6n3</b>	19.7 ± 0.5a	19.5 ± 0.4a	19.7 ± 0.5a
<b>ΣPUFA</b>	341.1 ± 7.6a	332.5 ± 0.8a	334.9 ± 5.8a
<b>ΣPUFA/ΣSFA</b>	2.32 ± 0.01c	1.88 ± 0.01a	2.05 ± 0.01b
<b>ΣPUFAn-3</b>	267.8 ± 6.2a	260.9 ± 0.2a	263.2 ± 5.0a
<b>ΣPUFAn-6</b>	85.6 ± 1.7a	83.7 ± 0.3a	84.2 ± 1.5a
<b>Σn-6/Σn-3</b>	0.32 ± 0.0a	0.32 ± 0.0a	0.32 ± 0.0a

ΣPUFA: total amount of polyunsaturated fatty acids. ΣPUFAn-3: total amount of n-3 polyunsaturated fatty acids. ΣPUFAn-6: total amount of n-6 polyunsaturated fatty acids. ΣPUFA/ΣSFA: polyunsaturated/saturated fatty acid ratio. Σn-6/Σn-3: n-6/n-3 polyunsaturated fatty acid ratio. Means ± standard deviation. Different letters (a, b, c) in the same row indicate significant differences (p≤0.05).

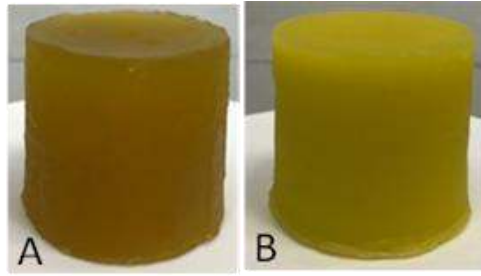


Figure 2. Pictures of ethyl cellulose (A) and beeswax (B) oleogels.

Table 2. Optical (lightness, L\*; redness, a\*; yellowness, b\*).

		EC-OG	W-OG
Optical properties	L*	25.9 ± 0.1a	36.7 ± 0.1b
	a*	-0.1 ± 0.1b	-1.40 ± 0.01a
	b*	2.7 ± 0.1a	20.9 ± 0.1b

As no significant differences were observed as function of storage time (28 days at 3 °C ± 1), results are expressed means of all sampling dates ± standard deviation. Different letters (a, b) indicate significant differences ( $p \leq 0.05$ ) as a function of the oleogelator system.

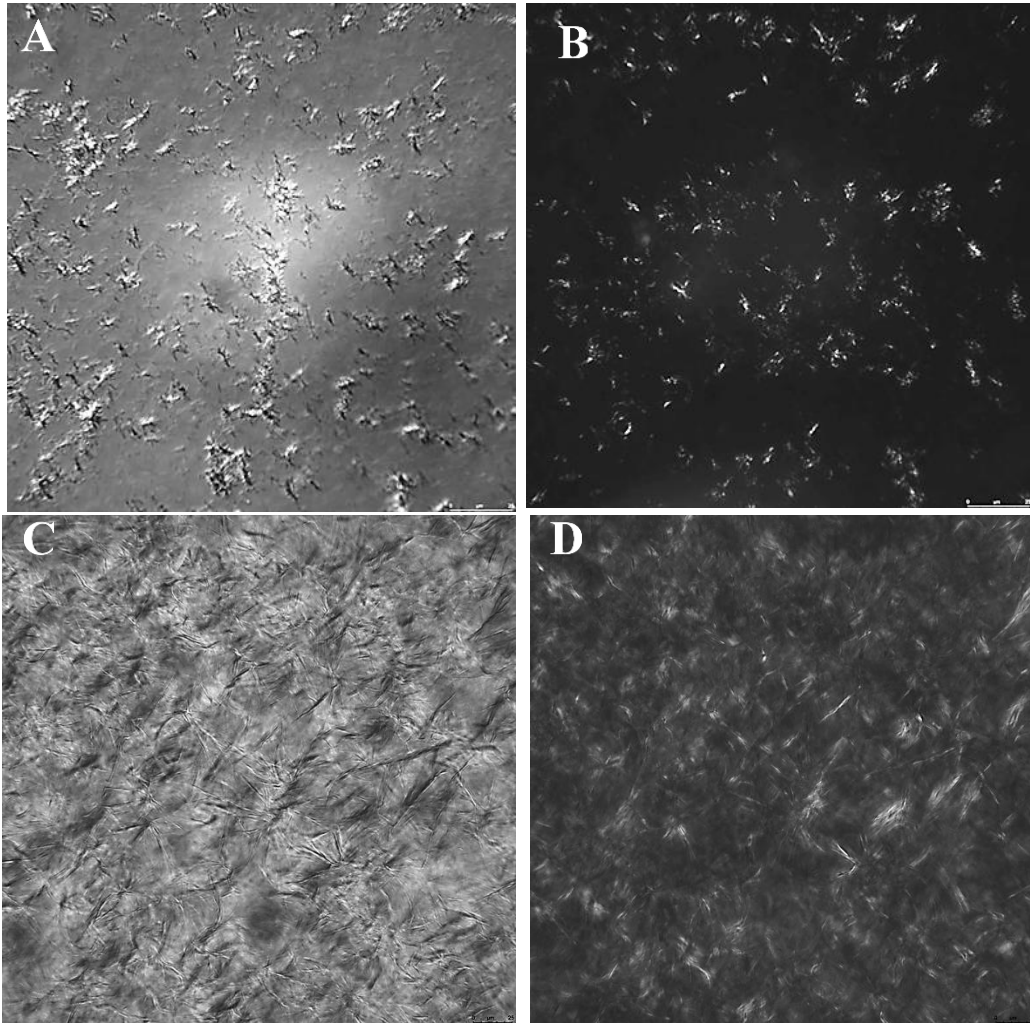


Figure 3. Microscopic images of ethylcellulose (A, C) and beeswax (B, D) oleogels.

Pictures C and D were taken under polarized light.



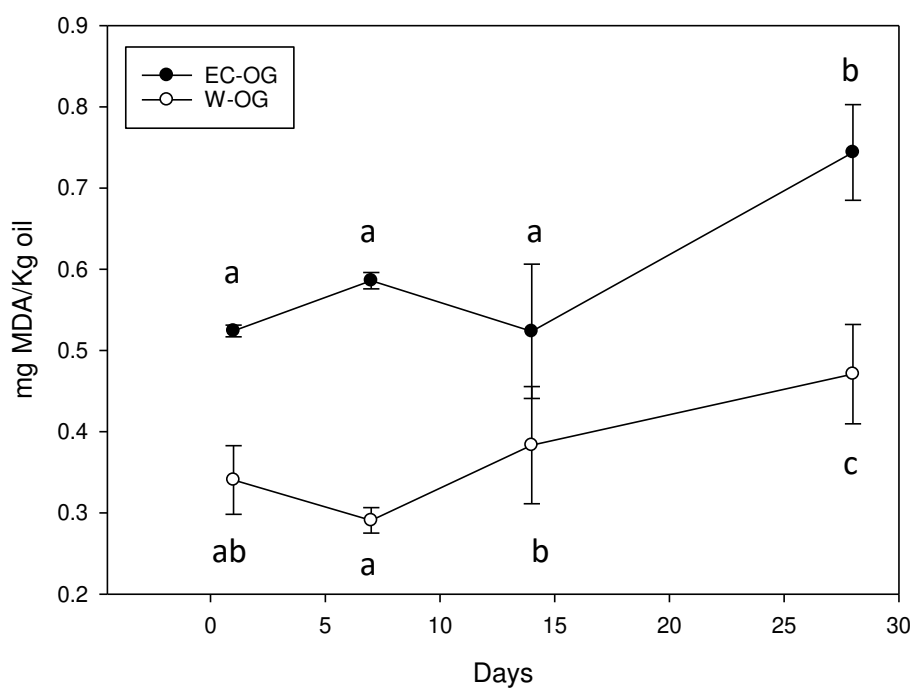


Figure 4. TBARS of ethyl cellulose (EC-OG) and beeswax (W-OG) oleogels as a function of storage time at  $3 \pm 1$  °C. Different letters indicate significant differences for each oleogel as function of storage time.

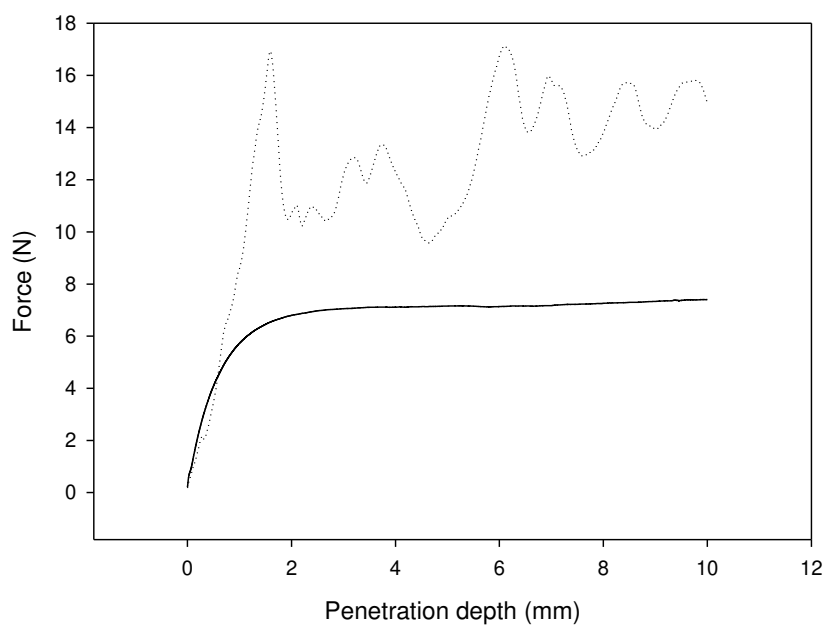


Figure 5. Typical penetration force analysis curves of ethyl cellulose (EC-OG, solid line) and beeswax (W-OG, dotted line) oleogels.

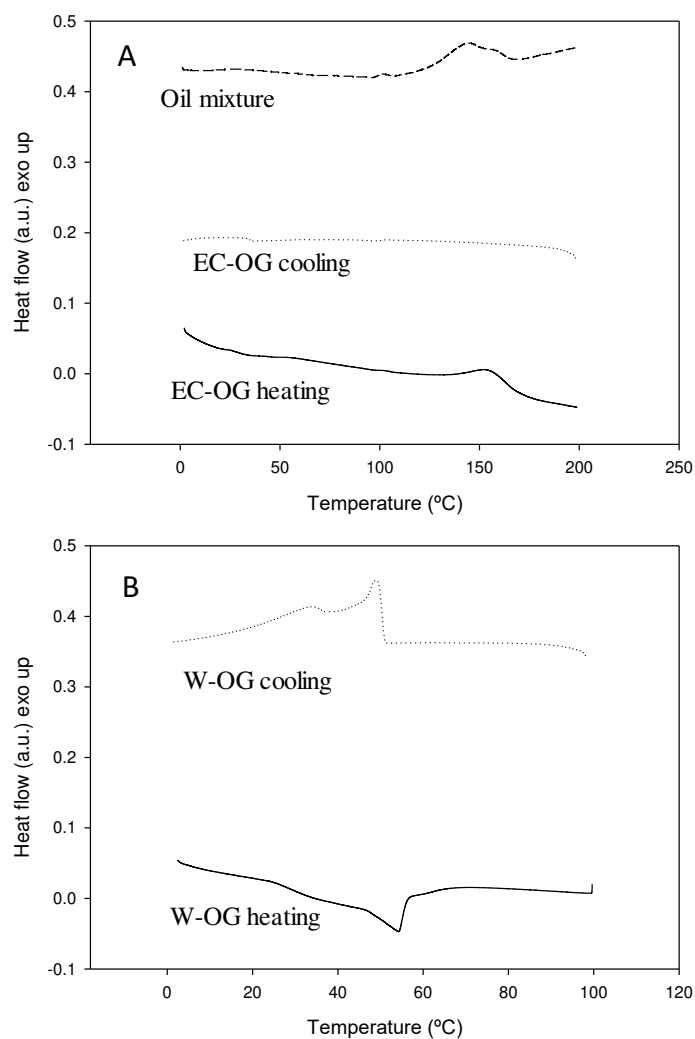


Figure 6. DSC traces of ethyl cellulose (A) and beeswax (B) oleogels during heating (solid lines) and cooling (dotted lines) ramps (1 day after preparation). DSC trace of the oil mixture during heating ramp is also plotted (A).

Table 3. Proximate composition (%) of burgers formulated with different lipid source or oil structuring method: C-B, control with pork backfat; EC-B, olive oil/linseed oil/fish oil structured with ethylcellulose; W-B, olive oil/linseed oil/fish oil structured with beeswax.

	Moisture	Fat	Ash	Protein
<b>C-B</b>	72.30±0.37a	7.21±0.20a	1.87±0.08a	18.31±0.17a
<b>EC-B</b>	71.84±0.21a	7.87±0.04b	2.03±0.03b	18.04±0.40a
<b>W-B</b>	71.92±0.19a	7.72±0.15b	1.88±0.06a	17.87±0.08a

Means ± standard deviation. Different letters in the same column (a, b) indicate significant differences ( $p \leq 0.05$ ).

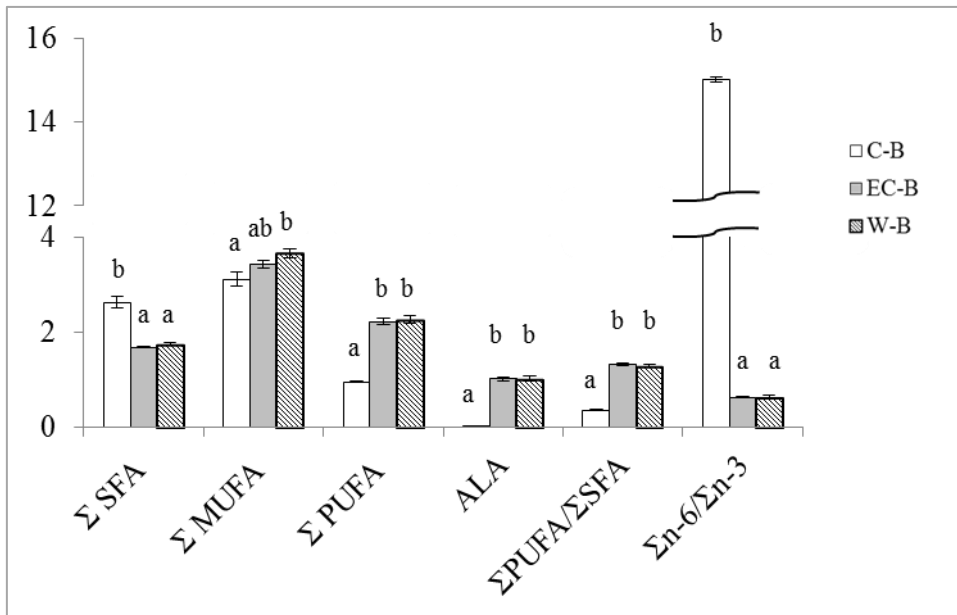


Figure 7. ΣSFA, ΣMUFA, ΣPUFA and ALA (g /100 g product) and nutritional ratios of the burgers developed

Table 4. Some physico-chemical properties of the burgers developed: luminosity (L\*), redness (a\*), yellowness (b\*), hue angle, chromaticity, Kramer shear force, and TBARS. Batches denomination as per Table 3.

	<b>C-B</b>	<b>EC-B</b>	<b>W-B</b>
<b>L*</b>	51.31±1.77a	52.32±1.67a	52.12±2.07a
<b>a*</b>	2.13±0.28a	3.35±0.29c	2.55±0.40b
<b>b*</b>	9.22±0.82a	11.49±0.89b	11.00±0.95b
<b>Hue angle (°)</b>	76.9 ± 1.9b	73.7 ± 1.7a	76.9 ± 1.9b
<b>Chromaticity</b>	9.5 ± 0.8a	12.0 ± 0.9b	11.3 ± 1.0b
<b>Kramer shear force (N/g)</b>	3.48±0.46b	2.61±0.21a	2.23±0.23a
<b>TBARS (mg MDA/Kg sample)</b>	0.11±0.02a	0.30±0.13c	0.14±0.01b

Means ± standard deviation. Different letters (a, b, c) in the same row indicate significant (p≤0.05) differences among samples.

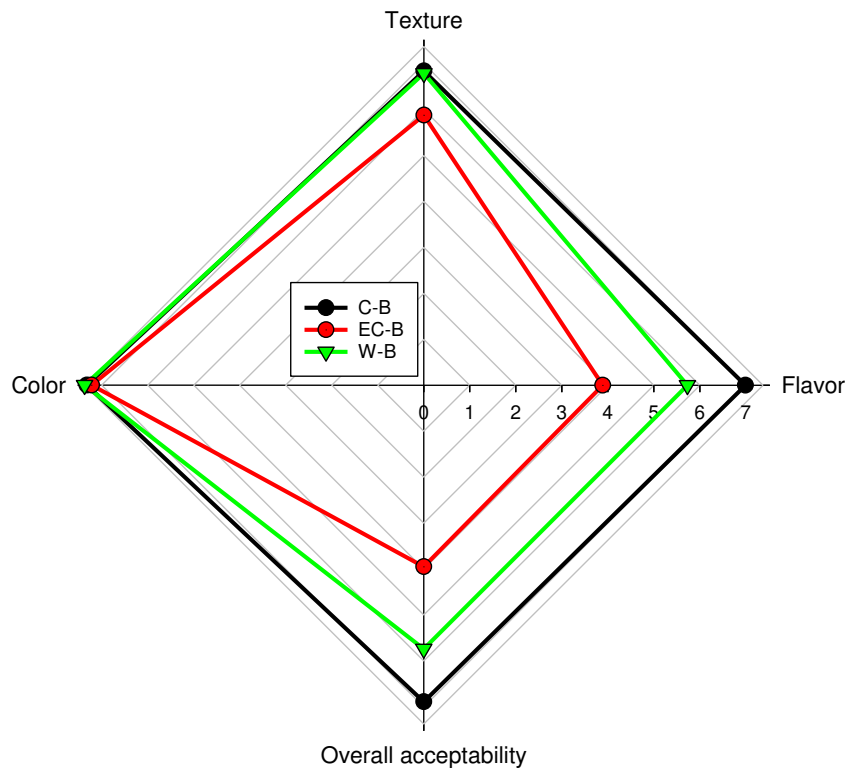


Figure 8. Sensory evaluation of the burgers developed. C-B, control with pork backfat; EC-B, olive oil/linseed oil/fish oil structured with ethylcellulose; W-B, olive oil/linseed oil/fish oil structured with beeswax