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Assessment of heavy metal pollution in water and its effect on Nile tilapia (*Oreochromis niloticus*) in Mediterranean Lakes: a case study at Mariout Lake

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Abstract

Mariout Lake is one of the Northern Nile-Delta Lakes in Egypt that receives agricultural, industrial and domestic effluents through several drains. The present study aims to evaluate the levels of some heavy metals (HMs) in water and edible parts of *Oreochromis niloticus* in Mariout Lake, in addition to studying several pollution indices and potential human health risks. The levels of the studied HMs in water were in the order of Fe > Zn > Mn > Pb > Cu > Ni > Cd. However, results of the pollution index, that concerns the effect of individual metal, concluded that Cd and Pb in water had serious pollution effects for aquatic life, while Cu, Fe, Mn, Ni, and Zn had not any pollution effects at different locations in the lake. The indices of the composite effects of all HMs (Metal Index and Heavy Metal Pollution Index) indicated the high pollution of Mariout Lake water, which may cause adverse effects on fish and different aquatic organisms. On the other side, the bioaccumulation factors of HMs in edible parts of *O. niloticus* were in the order of Zn > Cd > Cu > Ni > Pb > Mn > Fe. Although the target hazard quotient for all metals was less than the non-hazardous limit (THQ < 1), the non-carcinogenic hazard index (HI = 1.24) was classified in the moderate hazard risk level (1 < HI < 10) indicating low potential adverse effects on the exposed population due to consumption of *O. niloticus* caught from Mariout Lake.

Keywords Northern Nile-Delta Lakes \cdot Metal pollution \cdot Pollution indices \cdot Non-carcinogenic health risk \cdot Hazard index \cdot Hazard quotient \cdot Bioaccumulation factor

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Introduction

Potentially toxic element (PTE) pollution in the aquatic environment has become a global challenge in the twenty-first century. This is due to their abundance, persistence, inherent toxicity, non-degradability, and ubiquity in aquatic systems. Heavy metals are present in the environment in different phases: solid phase, liquid phase as free ions, or absorbed as solid colloidal particles. They may enter the aquatic ecosystem through natural and anthropogenic processes, including atmospheric deposition, erosion of the geological matrix, storm runoff, leaching from landfills, mining, shipping and harbor activities, and domestic, industrial, and agricultural runoff (Panico et al. 2018; Abd El-Aal et al. 2020; Bashir et al. 2020; Napoletano et al. 2021; Sánchez et al. 2022), causing worsening of the aquatic environment (Hamed et al. 2020). Intensive research on heavy metals in the aquatic and terrestrial environments has become important due to concerns regarding their overaccumulation and toxic effects on organisms and, eventually, on humans through the food chain (Volpe et al. 2015; Memoli et al. 2017; Kortei et al. 2020).

Heavy metal pollution endangers several aquatic species of fish, and continuous exposure to such toxic metals makes and forces fishes to take them directly from the environment through their blood and thus through contact with their organs and tissues (Goher et al. 2018). This causes physiological, morphological, and behavioral abnormalities alongside reproductive impairment in fish (Bristy et al. 2021). Consequently, there is a concomitant increase in human health risk through the consumption of fish contaminated with heavy metals. The long-term intake of heavy metals through foodstuffs, such as fish, may induce chronic accumulation, thereby damaging the human body (Talab et al. 2016; Adegbola et al. 2021).

The severity of toxic metals in humans has been known for ages, and exposure to these metals is continuously increasing in many areas, such as: Manzala Lake and Abou-Quir Bay. The toxicity of some heavy metals can be acute, while others can be chronic after long-term exposure. Heavy metals negatively affect humans, even at low concentrations, as they can cause abnormal development of a fetus; procreation failure; immune deficiency; carcinoma; organ dysfunction; physical, mental, and neurological disorders; renal tumor; nephritis; osteoporosis; nasopharyngeal congestion; increased blood pressure associated with cardiovascular diseases; reduced life expectancy; and, in some cases, death (Hamadani et al. 2020; Mohanta et al. 2020; Sánchez et al. 2022).

The most studied lagoons in the Mediterranean are located in Albania, Algeria, Egypt, France, Greece, Italy, Montenegro, Morocco, Spain, Tunisia, and Turkey. They provide ecosystem services that are essential for humans in many ecological, cultural, and medical uses. They also contribute to the overall productivity of coastal waters by supporting a variety of habitats and to the reduction in poverty by providing natural resources. Their stressors include overexploitation, pollution, climate changes, and biological invasions, often co-occurring in time and space and having cumulative effects. Such ecosystem changes can have large consequences on species abundance, bio-distributions, as well ecosystem functioning and services. Hence, it is of primary importance to study and protect lagoons (Parisi et al. 2022).

Mariout Lake is one of the five coastal northern lakes in Egypt. It is a brackish water body in the northern part of Egypt, between the Alexandria and El-Beheira governorates. It represents the southern boundary of Alexandria City and is located in the northwestern corner of the Egyptian Nile Delta. It is the smallest of the northern Delta lakes but perhaps the most threatened. Mariout Lake has been in existence for over 6000 years. The current lake is only a small remnant of its precursor, Lake Mareotis (the lake's name in ancient times). During that period, the lake covered a large area south of Alexandria, extending over 40 km southeast and 70 km southwest along the Mediterranean coast. The maximum width of the lake was estimated to be 24.5 km, while its length was about 44.5 km. The lake had eight islands. Four of them disappeared, while the remaining four became peninsulas. There is no evidence in history that the lake had any connection with the sea, but it was always a freshwater lake, getting its water from River Nile through various branches stemming from the Canopy branch, which ends at the Mediterranean Sea at the old Canoup City (now known as Abou-Quir) (Shehata 2014). Fishing is a major activity in the lake that was characterized in the past by large fish harvest rates with reputed quality. Lake fishes are essential to the well-being and livelihood of about 7000 fishers and their families. Until the mid-1970s, Mariout Lake was highly productive, contributing no less than 75% of the national fish catch in the Alexandria area. In 1974, the fish catch attained its peak level of 17,000 t. Since the beginning of the 1980s, fish production has progressively decreased, reaching 5000 t in 2007. The simultaneous reclamation of great areas of Mariout Lake, excessive fishing pressure, and wastewater discharged into the lake have dramatically affected fish production (Omran and Negm 2017). However, there was a fluctuation in fish production of Mariout Lake during the last period, recording 12,301 t, 8058 t, and 15,510 t in 2015, 2017, and 2020, respectively (GAFRD 2020).

This study aims (a) to investigate the heavy metal (Fe, Mn, Zn, Cu, Ni, Pb, and Cd) distributions in water and muscles of the *Tilapia* species (*Oreochromis niloticus*) from Mariout Lake; (b) To track the long-term changes of the HMs levels over previous periods (c) to appraise the extent of direct and indirect pollution effects on the lake's environmental situation due to the disposal of domestic, industrial, and agricultural effluents into the lake through the drains; (e) to determine the bioaccumulation levels of heavy metals in *Oreochromis niloticus* muscles; and (f) to assess the human health risks due to consumption of the edible parts of the *Tilapia* species (*O. niloticus*) caught from Mariout Lake.

Materials and methods

Study region

Mariout Lake lies between Latitude $31^{\circ} 04' 47''$ N and $31^{\circ} 09' 48''$ N, and Longitude $29^{\circ} 51' 40''$ E and $29^{\circ} 55' 14''$ E along the Mediterranean Sea coast of Egypt, as shown in Table 1 and Fig. 1. Its area secured 200km^2 at the beginning of the twentieth century, but at the beginning of the twenty-first, it was only about 50km^2 . Mariout Lake has been decreased by more than 75% and is still shrinking. It has a very shallow bottom reaching 150 cm at its deepest

 Table 1
 Details, latitudes, and longitudes of the sampling locations in Mariout Lake

Basins	Sites	Latitudes (N)	Longitudes (E)
South-West Basin (SWB)	1	31° 04′ 47.31″	29° 53′ 24.54″
	2	31° 06' 34.20"	29° 53′ 26.04″
	3	31° 06' 22.29"	29° 55′ 14.45″
	4	31° 05′ 45.06″	29° 54′ 41.21″
Fisheries Basin (FB)	5	31° 07′ 36.17″	29° 54' 03.35"
	6	31° 08′ 59.41″	29° 55′ 30.87″
Main Basin (MB)	7	31° 09' 18.37"	29° 54′ 57.47″
	8	31° 09' 48.47"	290 54' 14.79"
	9	31° 09' 03.29"	29° 53' 15.17"
	10	31° 07' 48.59"	29° 53' 35.63"
North-West Basin (NWB)	11	31° 07' 35.43"	29° 52′ 46.95″
	12	31° 06' 47.20''	29° 51' 40.52"
	13	31° 07′ 59.98″	29° 51′ 37.05″

location and a depth of 20 cm at shore locations around the lake periphery. It is a closed lake having no direct connection to the Mediterranean Sea. It is separated from the sea (by a 20 m) by the Desert Road, El-Nubaria Canal and El-Umoum Drain. Moreover, during February 2003, the remaining opened disposal sites to the beaches of Alexandria were locked and converted to Mariout Lake (Baraka 2012; Shehata 2014; Donia 2016; El-Kafrawy et al. 2017; Alnagaawy et al. 2018).

The lake is currently divided into four basins, as shown in Fig. 1, that are dissected by railroads, embankments and highways connecting Alexandria with the desert areas around it and with Cairo. The four basins are interconnected with each other by several breaches in the dykes of El-Umoum Drain and El-Nubaria Canal (EEAA 2011; Khalil et al. 2013; El-Naggar and Rifaat 2017). The four basins are:

- The Main (6000 Feddans) Basin: It is the northernmost water body and the largest basin in Mariout Lake as it covers an area of about 25km² with an average water depth of 1.2 m (El-Shorbagi 2015; Afifi et al. 2016). It receives most of its water from the most heavily polluted drains (El-Qalaa and El-Umoum Drains) (Khalil et al. 2013).
- 2. The Fisheries or Aquacultural (1000 Feddans) Basin: It lies south of the Main Basin. It is bordered by fish farms in the north and agricultural lands in the east. It covers an area of about 9.44km² (849 Feddans) and its average water depth is 1.35 m. It also receives its water from El-Umoum and El-Qalaa Drains (Afifi et al. 2016).
- The North-West (3000 Feddans) Basin: It is a triangular lagoon located in the west of El-Umoum Drain. Its area is about 11.59km² with an average water depth of 0.7 m. It receives its water through El-Nubaria Canal and El-

Max pumping site that is located at the northern side of the basin (El-Naggar and Rifaat 2017).

4. The South-West (5000 Feddans) Basin: It is bordered by agricultural lands in the east, and petroleum companies and human settlements at the west. It is partially divided by El-Nubaria Canal. This basin is very shallow with an average water depth of 0.68 m and a surface area of 33.77km². The main source of water is El-Omoum Drain at the northern region and El-Nubaria Canal at the middle region (EEAA 2011; Donia 2016).

Mariout Lake inflows and outflows

Mariout Lake receives daily a large amount of different pollutants from Alexandria City. The lake receives water from many drains and canals. (a) El-Qalaa Drain, which receives effluents from raw wastewater, irrigation drainage, agricultural runoff, and domestic and industrial waste from the eastern sector of Alexandria. Its water was thrown into the lake on the south-eastern side of the Main Basin. Now, it runs parallel to the southern border of the Aquacultural Basin and connects to El-Umoum Drain. (b) El-Umoum Drain, which represents the main water source for the Main Basin (Shehata 2014; Saad et al. 2017). (c) Petroleum Drain, which receives untreated industrial wastewater from the estimated 775 industries around the Mariout Lake area (Silim 2015; El-Naggar and Rifaat 2017). (d) El-Nubaria Canal, which lines the western side of the Main Basin and may be considered as fresh water source for the lake. (e) East Water Treatment Plant (EWTP), which discharges its treated effluents (90% domestic sewage, and only 10% industrial effluents) into El-Qalaa Drain. (f) West Water Treatment Plant (WWTP), which discharges its treated effluents (66% industrial effluents and 34% domestic sewage) directly into the Main Basin. Mariout Lake is not directly connected to the Mediterranean Sea. The only outflow from Mariout Lake is the El-Max pumping site. It comprises two buildings, each housing six pumps with nominal capacities of 12.5m³/s (Silim 2015). The fact that domestic, municipal, agricultural, and industrial wastes are continuously discharged into the lake makes El-Max pumping site essential to maintain the lake water level about 2.8-3.0 m below the mean sea level (Khalil et al. 2013; El-Khatib et al. 2020).

Sampling protocol

Subsurface water samples were collected seasonally from 13 selected sites in Mariout Lake in addition to El-Nubaria Canal and three drains (El-Qalaa Drain, El-Umoum Drain, and Petroleum Drain) during spring 2017–winter 2018, using a polyvinyl chloride Van-Dorn bottle with a capacity of 2L. Triplicate subsamples were collected from each site. The samples were kept in cleaned stopper plastic bottles

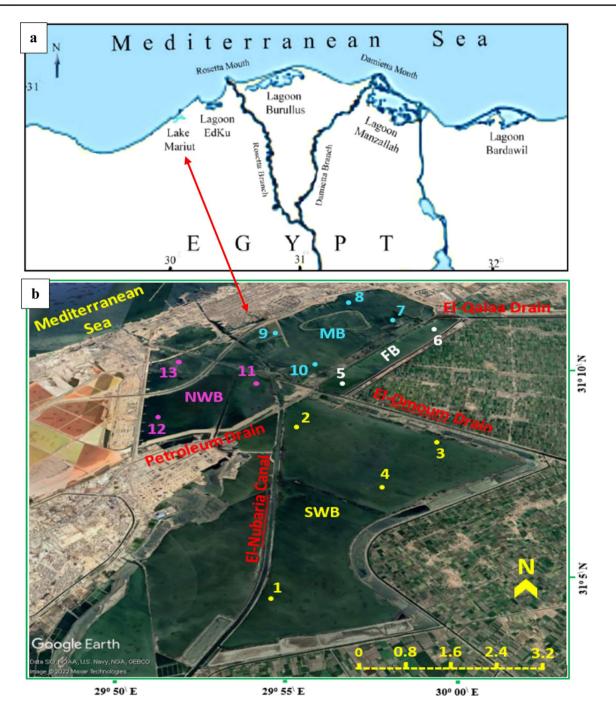


Fig. 1 A map a showing the five Northern Delta Lakes in Egypt, and b illustrating the sampling locations in Mariout Lake

of 1L capacity to analyze later in the laboratory. For heavy metals analyses, water samples were preserved with conc. HNO_3 to reduce pH to be below 2 to stop bacterial growth, block oxidation reactions and prevent precipitation of the metals. The bottles were then stored in a refrigerator at 4°C to prevent change in volume due to evaporation.

Fish samples of *Tilapia* species (*Oreochromis niloticus*), a common and a famous fish species in Egypt; its native name is Bolti, were collected from the sites 2, 3, 6, 7, 10

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and 11 during summer 2017 and winter 2018. The collected fish samples (10 from each site) nearly had the same size, length and weight. The samples were quickly transported into the laboratory for dissection and subsequent analyses. Fish caught from the selected sites were externally dried from water, then lateral white muscles were quickly and carefully removed, and kept frozen at -20° C for determination of heavy metals concentrations. Geographic positions of the samples were appointed using the GPS device (Garmin,

ETrex model). The sampling locations along with their longitudes and latitudes are presented in Table 1.

Methodology

Physicochemical parameters in water

Temperature, pH and dissolved oxygen (DO) content of water samples collected from Mariout Lake were measured *in-situ* using Thermo Orion Star (A 329 multiparameter instrument). Water transparency was also measured *in situ* using a white/black Secchi disk (30 cm in diameter). However, total dissolved solids (TDS) and total suspended solids (TSS) in lake water were determined in the laboratory using the evaporation method where TDS were determined by filtering a volume of a sample with a glass microfiber filter (GF/C), and a known volume of the filtrate was evaporated at 180°C. TSS were calculated as the difference between TS (total solids) and TDS, where TS were measured by evaporating a known volume of a well-mixed sample (APHA 2005).

Heavy metals in water and fish

All glassware used were rinsed and washed with diluted HCl before usage. Merck analytical grades (USA) of all reagents were used to prepare standards for ICP instrument calibration under satisfying clean laboratory environment. A 1000 g/L multi-element certified standard solution was used as a stock solution for the instrument standardization. Standard reference materials, from National Institute of Standards and Technology (NIST), were used to validate analysis. Accuracy and precision of analysis was checked by replicate measurements of these reference materials, with recovery rates for metals between 97.1 and 103.8% which were within the acceptable recovery percentage range (80–110%) according to Huber (2007).

Water samples were digested using the *Nitric Acid Digestion Method* according to APHA (2005) as the following:

- Put a well-mixed 250-ml water sample in a beaker on a hot plate and add 5 ml of conc. HNO₃, then slow boil and evaporate till the lowest volume possible (50 ml) but before precipitation occurs. The digested solutions were cooled to room temperature.
- Transfer to a 50-ml volumetric flask and dilute to the mark with deionized water if needed. Mix thoroughly and filter if necessary. The total concentrations of heavy

metals in water were calculated according to the following equation.

Metal Concentration(
$$\mu g/L$$
) = $\frac{A \times B}{C}$

where *A* is the total concentration of a metal in the digested solution $(\mu g/l)$, *B* is the final volume of the digested solution (ml), and *C* is the initial water sample volume (ml).

Fish muscles were digested according to FAO/SIDA (1983) as in the following:

- The fish flesh of *Tilapia* species (*O. niloticus*) collected from the same site were mixed together. A 10 ml portion of a freshly prepared mixture of nitric and perchloric acids (HNO_3 : $HCIO_3 = 1$: 1) was added to 5 g of muscles from each mixture in tightly closed Teflon vessels.
- Closed vessels holder was placed at 160°C in a microwave oven (model Milestone, MLS-1200 mega, Germany) until the organic matter was broken down which was observed through the reduced volume and clarity of the solution. After the digestion process, the vessels were cooled and carefully opened.
- Solutions were transferred to a 50-ml volumetric flask and diluted to the mark with deionized water, then filtered and transferred to plastic bottles. The concentrations of heavy metals in fish muscles were calculated according to the following equation:

Metal Concentration(
$$\mu g/g$$
 wet wt.) = $\frac{A \times B}{C \times 10^3}$

where A is the metal concentration in the digested solution $(\mu g/l)$, B is the final volume of the digested solution (ml), C is the weight of fish muscle sample (g), and 10^3 is the unit conversion factor.

The total concentrations of the heavy metals (Fe, Mn, Zn, Cu, Ni, Pb, and Cd) in digested water and *O. niloticus* muscles samples collected from Mariout Lake were measured using Inductively Coupled Plasma Emission Spectrometry (ICP-ES) with Ultra Sonic Nebulizer (USN), model Perkin Elmer optima 7000, USA.

Blanks were prepared in each digestion procedure. Water and fish samples were analyzed in three replicates with relative standard deviations (RSD_s) less than 10% of the trace elements. Concentrations below the detection limits were substituted with the reported detection limit for all non-detects, if concentrations in respective replicates of the sample were detected at least once above the detection limits. Wavelengths (nm) and detection limits (µg/L) of ICP-ES are 259.933, 257.604, 213.855, 324.747, 231.602, 220.35 and 226.499 nm, and 0.3, 0.04, 0.2, 0.3,

0.5, 1.5 and 0.1 μ g/l for Fe, Mn, Zn, Cu, Ni, Pb and Cd, respectively.

Statistics

The obtained results were analyzed using the Excel-Stat 2016 software for determination of the spatial and temporal variations by analysis of variance using the one-way ANOVA test ($\alpha = 0.05$) which was also used for determination of the probability values (P-values); significance levels of tests were taken as significant when $P \le 0.05$ and highly significant when $P \leq 0.01$. In addition, the relationships between the studied heavy metals were determined by calculating the Spearman correlation matrix using Minitab 20 software (confidence level = 95). Spearman correlation was specified after application of the normality test (Ryan-Joiner) on our results using the Minitab 20 software. Based on the correlation coefficient values (r) and number of samples (n), highly significant correlations were indicated when $r \ge 0.354$ and significant correlations when $0.273 \le r < 0.354$. Also, Microsoft Excel was used to design some of the figures in this study.

Metal quality indices (MQI) in water

Three different quality indices were used to determine the grades of metals contamination in Mariout Lake water for aquatic life utilizations. These indices are Pollution Index (PI), Metal Index (MI), and Heavy Metal Pollution Index (HPI).

Pollution index (PI)

IT is based on the individual metal concentration which means that each metal has its own PI value. It is categorized into 5 classes, as shown in Table S1. PI was calculated using the equation according to Caerio et al. (2005).

$$\mathrm{PI} = \frac{\sqrt{\left[\frac{C_i}{S_i}\right]_{\max}^2 + \left[\frac{C_i}{S_i}\right]_{\min}^2}}{2}$$

where C_i is the concentration of each heavy metal and S_i is the metal level according to the national water quality criterion.

Metal index (MI)

IT is based on a total trend evaluation of the present status and by that summarizes the status of the study area by one value. The higher the concentration of a metal compared to its respective MAC value, the worse the quality of water. MI value > 1 is a threshold of warning (Bakan et al. 2010). MI was calculated by the equation according to Tamasi and Cini (2004).

$$\mathrm{MI} = \sum_{i=1}^{n} \frac{C_i}{(MAC)_i}$$

where C_i is the concentration of each heavy metal, *n* is the number of measured heavy metals, and *MAC* is the maximum allowable concentration of the *i*th metal.

Heavy metal pollution index (HPI)

IT is a comprehensive tool or a rating model that assesses the overall water quality according to the composite effects of individual heavy metal. It is based on the weighted arithmetic quality mean method (Hassouna et al. 2019) and can be calculated from the following equation:

$$HPI = \frac{\sum_{i=1}^{n} Q_i W_i}{\sum_{i=1}^{n} W_i}$$

where Q_i is the sub-index of the *i*th metal, *n* is the number of measured heavy metals, and W_i is the weight unit of the *i*th metal (between 0 and 1).

 W_i and Q_i were calculated according to the following equations:

$$W_i = \frac{K}{S_i} = \frac{1}{S_i}$$
$$Q_i = \frac{C_i - I_i}{S_i - I_i} \times 100$$

where K is the proportionality constant, C_i is the concentration value of the *i*th metal, S_i is the standard permissible value of the *i*th metal for aquatic life, the sign (–) indicates the numerical differences between the two values ignoring the algebraic sign, and I_i is the ideal value of the *i*th metal; in pure water, $I_i=0$ and the previous equation converts to the following equation:

$$Q_i = \frac{C_i}{S_i} \times 100$$

Finally, the critical heavy metal pollution index (HPI) score for the water suitability to aquatic life is 100 (Milivojević et al. 2016).

Bioaccumulation factor (BAF) estimation

Bioaccumulation is the process that causes an increase in chemical concentrations in aquatic organisms compared to those in water due to uptake by all the exposure routes, including dietary absorption, transport across the respiratory surfaces, and dermal absorption. Thus, it can be viewed as a combination of bio-concentration and food uptake. It is a number that describes the bioaccumulation as a ratio of the concentration of a chemical inside an organism to its concentration in the surrounding environment (Goher et al. 2015). The bioaccumulation factor (BAF) of a heavy metal in fish tissues was determined using the equation according to El-Khatib et al. (2020).

$$BAF = M_{tissue}/M_{water}$$

where M_{tissue} is the metal concentration in fish tissues ($\mu g/g$ wet weight) and M_{water} is the metal concentration in water ($\mu g/l$).

According to Olayinka-Olagunju et al. (2021), BAFs values of heavy metals are categorized as follows: BAF < 1000 means no probability of accumulation; 1000 < BAF < 5000 means bioaccumulative metal and BAF > 5000 means extremely bioaccumulative metal.

Human health risk assessment of fish consumption

The human health risks associated with the intake of *Tilapia* species (*O. niloticus*) from Mariout Lake by approximately one million people living in cities and villages around it were evaluated using Estimated Daily Intake (EDI), Target Hazard Quotient (THQ) and Hazard Index (HI) developed by the United States Environmental Protection Agency (USEPA 2013).

Estimated daily intake (EDI)

IT is a commonly used method that helps identify the number of pollutants consumed daily. The EDI of potentially toxic elements (PTEs) is directly proportional to the concentrations of PTEs in food and the amount of daily food consumption. Furthermore, human body weight significantly affects the tolerance of contaminants. EDI of heavy metals (mg/day/70 kg Egyptian person) was calculated by multiplying the overall average level of the heavy metals concentrations detected in the *Tilapia* species (*O. niloticus*) by the average daily consumable rate of the fish muscles for the general adult consumers (Ullah et al. 2017). It was determined using the following formula (Bristy et al. 2021):

$$EDI = \frac{MC \times IR}{BW}$$

Target hazard quotient (THQ)

IT is a ratio of the estimated exposure dose of a pollutant to the reference dose. It represents the risk of non-carcinogenic effects (Kortei et al. 2020). It must be noted that THQ is not a measure of risk, but indicates a level of concern. It was calculated for the individual heavy metal using the following equation:

$$THQ = \frac{MC \times IR \times EF \times ED}{RfD \times BW \times ATn} \times 10^{-3}$$

where *MC* is the heavy metal concentration (mg/kg) in the fish flesh in wet weight basis, *IR* is the fish ingestion rate by a person per day, the per capita consumption of fish flesh in Egypt for human food is averaged to be 57 g/day for general population (FAO 2014), *EF* is the exposure frequency (365 days/year), *ED* is the life time exposure duration (70 years), *RfD* is the oral reference dose of individual heavy metal (mg/kg/day), as reported by USEPA (2013), *BW* is the average adult body weight of a consumer in Egypt (70 kg), as a default value suggested by USEPA (2013), *ATn* is the average exposure time for non-carcinogens (365 days/year × number of exposure years), and 10^{-3} is the unit conversion factor. It was considered that the fillet yield for *Tilapia* species (*O. niloticus*) is within 39.1% (Neira et al. 2016).

THQ < 1 means that adverse health effects are unlikely. THQ > 1 reveals probable adverse health effects. THQ > 10 indicates high chronic risks (Custodio et al. 2021). Although the THQ values are additive, they are not multiplicative; e.g., the level of concern at THQ of 20 is larger but not tenfold of that at THQ=2 (Osakwe et al. 2014).

Hazard index (HI)

The overall potential for non-carcinogenic effects from all studied metals was assessed using the hazard index (HI). It was denoted as the sum of the target hazard quotient (THQ) values of all the tested heavy metals and was obtained using the equation according to USEPA (2011).

$$HI = \sum_{i=1}^{n} HQi$$

where *i* is the individual heavy metal.

If HI < 1, it is assumed that the non-carcinogenic adverse effects due to the fish consumption are negligible, while the potential for chronic effects may be a concern when HI > 1 (Custodio et al. 2021). Hence, HI < 1 means no hazard; 1 <HI < 10 means moderate hazard, while HI > 10 means high chronic hazard (Ukoha et al. 2014). Like THQ, HI does not measure the risk directly as it cannot define the dose–response relationship (Bristy et al. 2021).

Results and discussion

General physicochemical features of Mariout Lake water

The ranges and means of temperature, transparency, TDS, TSS, pH, and DO of Mariout Lake water during 2017–2018 are presented in Table 2. Temperature ranged between 15.6°C at site (1) during winter and 31.4°C at site (13) during summer season with highly temporal significant differences ($p \le 0.01$). Mariout Lake is characterized by water turbidity due to the deterioration effects of El-Qalaa and El-Umoum Drains. Transparency fluctuated between 15 cm at site (8) in spring and 65 cm in summer at site (11) with highly temporal significant differences $(p \le 0.01)$ and significant differences $(p \le 0.05)$ among sites. However, the maximum TDS value (9.68 g/L) was observed in winter at site (6), but the minimum value (2.78 g/L) was at site (8) during summer season with highly spatial significant differences ($p \le 0.01$). Conversely, the highest TSS value (148.62 mg/L) was recorded at site (8) during spring, but the lowest value (35.18 mg/L) was at site (6) in summer season with highly temporal significant differences ($p \le 0.01$) and spatial significant differences ($p \le 0.05$). TSS levels were higher than the permissible limits for aquatic organisms during all seasons and sites. On the other hand, Mariout Lake water is alkaline, whereas pH values fluctuated between 7.22 in summer at site (2) and 8.68 at site (12) in spring with highly significant differences ($p \le 0.01$) among seasons; the values were within the permissible levels for aquatic life (Elsayed et al. 2019). pH was positively correlated with DO (r=0.3, $p \le 0.05$) which may be related to the photosynthetic action (Imam et al. 2020). Our results showed that DO ranged from 0.5 mg/L during spring to 9.14 mg/L during winter at sites (8) and (4), respectively, with significant differences $(p \le 0.05)$ between seasons and sites.

 Table 2
 General physicochemical parameters in Mariout Lake water during 2017–2018

Parameter	Range	Mean±SD	Aquatic life*
Temperature (°C)	15.6–31.4	23.69±5.73	8–28
Transparency (cm)	15-65	38.08 ± 11.17	
TDS (g/L)	2.78-9.68	5.7 ± 1.71	
TSS (mg/L)	35.18-148.62	59.19 ± 18.86	25
pH	7.22-8.68	7.83 ± 0.33	6.5–9
DO (mg/L)	0.5–9.14	5.26 ± 2.16	>5.5

*CCME-Canadian Council of Ministers of the Environment (2021)

Heavy metals in Mariout Lake water

The occurrence of metal contaminants in Mariout Lake water has become an increasing concern. This situation has arisen due to the rapid growth of the population and, thus, the increase in human activity (Silim 2015). The concentrations of the studied heavy metals in Mariout Lake water were found in the ranges of 266.50-1145.40, 18.60-103.20, 18.00-130.40, 5.00-23.23, 5.60-19.62, 10.55-59.51, and 3.30-9.90 µg/l for Fe, Mn, Zn, Cu, Ni, Pb, and Cd, respectively, over the study year. Generally, the annual average contents were in the order of Fe > Zn > Mn > Pb > Cu > Ni >Cd with respective values of $682.70 \pm 248.87, 60.42 \pm 31.55,$ 58.91 ± 22.42 , 33.55 ± 11.47 , 12.40 ± 4.37 , 12.09 ± 3.6 , and $6.35 \pm 1.39 \,\mu\text{g/l}$, respectively, as shown in Table 3. ANOVA results indicated that there were highly significant spatial differences ($p \le 0.01$) for Fe and Mn concentrations in Mariout Lake water. However, Pb and Cd concentrations showed highly temporal significant differences (p < 0.01), while Cu concentrations showed significant differences between seasons ($p \le 0.05$). Generally, the seasonal variations in heavy metal content may be attributed to the fluctuation of the amounts of agricultural drainage water, industrial wastes, and illegal untreated domestic sewage discharged into the lake. Domestic effluents can contain fairly high concentrations of metals, such as Fe, Zn, Cu, Ni, and Pb, which are derived from various household products such as cleaning materials, toothpaste, and cosmetics (Goher et al. 2019a).

The maximum concentrations of all heavy metals in Mariout Lake water were observed in the Main Basin (sites 8 and 9) due to the effects of domestic, agricultural, and industrial effluents from El-Qalaa Drain, El-Umoum Drain, and the WWTP located in front of the site (8). These data agreed with those obtained by El-Rayis and Saad (1990) and Ali et al. (2020); they concluded that heavy metals increased in the Main Basin more than in the other three basins. However, the minimum concentration values of Fe, Mn, Cu, and Pb were recorded in the North-West Basin (sites 12 and 13). The lowest value of Ni concentrations (5.60 µg/l) was found at site (1) in the South-West Basin, and that of Zn concentrations (18.00 µg/l) was determined at site (5) in the Aquacultural Basin.

The distribution patterns of iron and manganese concentrations in Mariout Lake water increased during the cold rainy seasons (autumn and winter). Generally, Fe concentrations in Mariout Lake water increase in the rainy seasons due to water flowing from nearby shores carrying soil along with it, as the earth's crust contains abundant amounts of metals (Vasistha and Ganguly 2020). However, increasing Mn values in lake water are mainly attributed to the decomposition of organic debris by microbial activity (Ali et al. 2020). Conversely, the highest concentrations of Zn, Ni, Pb, and Cd were observed in the dry hot seasons, which may be

Table 3 Ranges, annual means, and standard deviations of heavy metals concentrations ($\mu g/l$) in Mariout Lake water compared to the standard permissible guidelines for aquatic life ($\mu g/l$)

Sites	Fe	Mn	Zn	Cu	Ni	Pb	Cd
1	397.6-818.2	49.0–73.62	71.2–113.29	8.80-18.60	5.60-14.61	20.55-28.88	6.62-8.30
	648.0 ± 193.0	60.33 ± 10.46	94.07 ± 18.01	11.74 ± 4.62	10.10 ± 4.21	24.24 ± 3.44	7.71 ± 0.80
2	272.0-1039	24.80-90.8	28.20-128.0	7.2-10.4	7.0–15.6	12.46-40.31	4.70-8.20
	613.98 ± 348.73	51.59 ± 28.07	73.65 ± 43.35	9.15 ± 1.37	10.80 ± 3.67	31.64 ± 13.05	6.60 <u>±</u> 1.44
3	812.22-963.17	27.00-71.80	30.60-87.40	7.60-17.20	9.00-17.80	16.86-46.47	3.70-5.90
	896.50 ± 70.20	60.15 ± 22.11	54.05 ± 25.27	11.85 ± 4.46	12.59 ± 4.12	32.03 ± 12.10	5.03 ± 0.94
4	581.60-796.08	40.20-59.80	25.80-97.80	11.2-6.60	10.2-15.2	21.16-40.60	5.20-7.33
	671.13 ± 90.05	51.20 ± 8.14	50.80 ± 32.04	13.53 ± 2.26	12.95 ± 2.11	29.90 ± 8.20	6.11±0.93
5	523.80-846.78	26.30-77.17	18.00-78.80	6.40-18.20	7.40-14.00	18.03-40.16	3.60-8.50
	708.33 ± 143.21	54.57 ± 22.06	49.30 ± 30.00	11.05 ± 5.67	11.50 ± 2.86	31.58 ± 9.48	6.10 ± 2.12
6	751.0-1044.48	27.20-61.98	34.12-63.20	8.20-12.84	6.20-15.80	28.58-42.27	4.70-7.20
	854.06 ± 130.42	44.65 ± 18.38	49.37 ± 14.94	10.81 ± 2.01	11.26 ± 4.23	35.19 ± 6.49	6.25 ± 1.10
7	271.60-875.23	67.40-100.4	18.40-68.60	6.20-19.00	9.14-19.40	21.99-53.50	3.30-8.10
	550.46 ± 249.12	82.45 ± 15.93	46.05 ± 20.94	11.60 ± 5.51	12.93 ± 4.47	37.46 ± 17.31	6.10 ± 2.24
8	959.20-1145.4	64.4-103.20	37.62-122.8	11.0-23.23	7.40-15.40	33.18-59.51	5.90-9.90
	1027.10 ± 85.17	82.75 ± 16.34	75.36 ± 35.90	18.11 ± 5.40	13.15 ± 3.85	43.14 ± 11.96	7.88 ± 1.63
9	580.0-1049.01	68.20-96.11	27.0-130.40	9.80-17.60	7.43-19.62	24.26-57.17	5.10-8.40
	866.76 ± 207.49	85.67±12.34	56.79 ± 49.49	13.78 ± 3.86	15.22 ± 5.78	39.30 ± 13.56	6.80 ± 1.48
10	544.40-875.71	27.00-84.20	19.2-100.60	7.80-18.40	13.64-15.16	20.70-47.20	5.10-6.60
	756.05 ± 154.83	56.71 ± 28.71	66.68 ± 39.59	12.57 ± 4.52	14.35 ± 0.79	34.42 ± 12.32	5.85 ± 0.81
11	303.4-774.80	34.20-71.00	19.20-51.80	5.80-15.40	6.16-14.60	28.00-35.77	4.40-6.80
	496.35 ± 201.4	51.95 ± 17.91	40.35 ± 15.30	12.55 ± 4.53	10.14 ± 3.67	32.62 ± 3.50	5.80 ± 1.04
12	266.5-298.40	23.60-53.22	28.80-83.60	5.20-15.29	11.80-14.61	13.63-52.04	5.10-7.80
	279.96 ± 14.62	44.05 ± 14.03	61.99 ± 23.76	11.18 ± 4.43	13.06 ± 1.19	34.99 ± 15.89	6.34 ± 1.15
13	304.6-604.6	18.6-64.0	26.2-122.4	5.0-17.6	6.20-12.06	10.55-48.67	5.50-6.72
	506.35 ± 139.64	39.75 ± 22.20	66.95 ± 43.56	13.25 ± 5.73	9.16 ± 2.90	29.62 ± 17.44	6.01 ± 0.52
Min	266.50	18.60	18.00	5.00	5.60	10.55	3.30
Max	1145.40	103.20	130.40	23.23	19.62	59.51	9.90
Mean \pm SD	682.70 ± 248.87	58.91 ± 22.42	60.42 ± 31.55	12.40 ± 4.37	12.09 ± 3.60	33.55 ± 11.47	6.35 ± 1.39
CCME (2021)	300	430	7	2–4	25-150	1–7	0.09–0.37
USEPA (2021)	1000	100	120	4	52	2.50	0.72

Min. minimum, Max. maximum, SD standard deviation

attributed to metal liberation from bottom sediment into the overlying water column under the effects of high temperatures and the fermentation process (Tafa and Assefa 2014; Abd El-Aal et al. 2020).

Table 3 illustrates that the concentrations of Cu, Pb, and Cd in Mariout Lake water exceeded the maximum permissible limits of the USEPA (2021) and CCME (2021) guidelines for aquatic life, but Mn and Ni concentrations were below the standard permissible limits of the USEPA (2021) and CCME (2021) guidelines. However, following the USEPA (2021) guidelines, the concentrations of Fe and Zn in lake water were below the permissible levels (1000 and 120) μ g/l, respectively, but exceeded the permissible standards for aquatic life of CCME (2021) guidelines, which are (300 and 7) μ g/l, respectively. This indicates the poor water quality of Mariout Lake, especially in the sites facing the outlets of the drains, and this may cause negative human health effects (Ghannam 2021).

The statistical analysis of the Spearman correlation coefficient matrix in Table 4 showed that the positive significant correlations among the metals Fe/Mn (r=0.396, $p \le 0.01$) and Fe/Cu (r=0.333, $p \le 0.05$) suggest that there are associations between these metals as they have similar distribution dynamics and mutual dependence as well as they originate from a common source in the aquatic ecosystem during transportation and/or the deposition reactions (Masoud et al. 2005; Ali et al. 2020). The chemical behavior of nickel as a strong chelating transition element is clear from its strong correlation with other metals, such

Table 4Spearman correlationmatrix of the studied heavymetals in Mariout Lake waterduring 2017–2018

Metal	Fe	Mn	Zn	Cu	Ni	Pb
Mn	0.396					
Zn	-0.061	-0.091				
Cu	0.333	0.163	-0.065			
Ni	0.329	0.193	0.074	0.234		
Pb	0.096	0.068	-0.066	0.160	0.141	
Cd	0.154	0.145	-0.061	0.259	-0.031	0.2

as iron [Ni/Fe (r = 0.329, $p \le 0.05$)], due to its tendency to form more stable organic complexes (Wangersky 1986).

Generally, heavy metals have high ecological significance because they persist in the environment and can remain for a very long time; they can potentially accumulate in water reservoirs and thus enter the food chain. Under certain environmental conditions, heavy metals may accumulate and concentrate in living organisms in highly toxic concentrations and cause serious harm and ecological damage (Bashir et al. 2020; Sánchez et al. 2022).

Some PTEs, such as Cu, Zn, Fe, Ni, and Mn, benefit humans at low concentrations and play a biochemical role in the life processes of all aquatic organisms but become toxic in unregulated quantities. However, the intake of some others, such as Cd and Pb, is highly poisonous to humans, even at quite low concentrations (Ghannam 2021; Sánchez et al. 2022).

Heavy metals in Mariout Lake inflows

The seasonal variations of heavy metal concentrations in Mariout Lake inflows are graphically represented in Fig. 2. The concentrations of the studied metals in the lake drains were found in the ranges of 271.00-1970.60, 23.00-97.40, 46.12-97.20, 5.40-32.00, 4.80-25.60, ND-56.93, and 9.40-17.60 µg/l for Fe, Mn, Zn, Cu, Ni, Pb, and Cd, respectively, over the study year. The annual average contents were in the order of Fe > Zn > Mn > Pb > Cu > Ni > Cd. The highest concentration values of Zn, Ni, and Cd were observed in the petroleum drain, but the maximum concentrations of Mn, Cu, and Pb were observed in the El-Umoum Drain. However, the highest iron concentration value was recorded in the El-Qalaa Drain. The discharged heavy elements into the lake either precipitate to the bottom sediment, are uptaken by living organisms, or increase the levels of these metals in the lake water (Goher et al. 2019b).

Long-term changes of heavy metals in Mariout Lake water

The long-term analyses of the measured heavy metals (Fe, Mn, Zn, Cu, Ni, Pb, and Cd) in Mariout Lake water showed significantly irregular vibrations from 1981 to 2020, as

indicated in the supplementary file (Table S2). The high fluctuation in the concentrations of all the measured metals may be attributed to the quality and quantity of the waste discharged into the lake via drains, especially El-Qalaa and El-Umoum Drain. In addition, some studies focused only on the Lake Main Basin, like El-Bestauy (1993), Amr et al. (2005), Abaza et al. (2009), and Afifi et al. (2016), but others studied the fishery basin, like Khalil (1998) and El-Khatib et al. (2020). However, some researchers collected water samples monthly from the lake, for example, Saad et al. (1981) and El-Bestauy (1993), while others studied seasonal variations, like Khalil (1998), El-Shorbagi (2015), Ali et al. (2020), and El-Khatib et al. (2020), but Afifi et al. (2016) explored heavy metal concentrations of water samples collected only in the winter season. In addition, some studies determined only the concentrations of dissolved heavy metals, as in Saad et al. (1981), El-Bestauy (1993), and El-Shorbagi (2015), while others measured the total heavy metal concentrations in Mariout Lake water, like Ashmawy et al. (2018), Abd Elalkhoris et al. (2020), and Morsy et al. (2020). On the other hand, Table S2 (in the supplementary file) shows the levels of the studied metals in Mariout Lake water in comparison to the other Northern Nile-Delta Lakes.

Metal quality indices

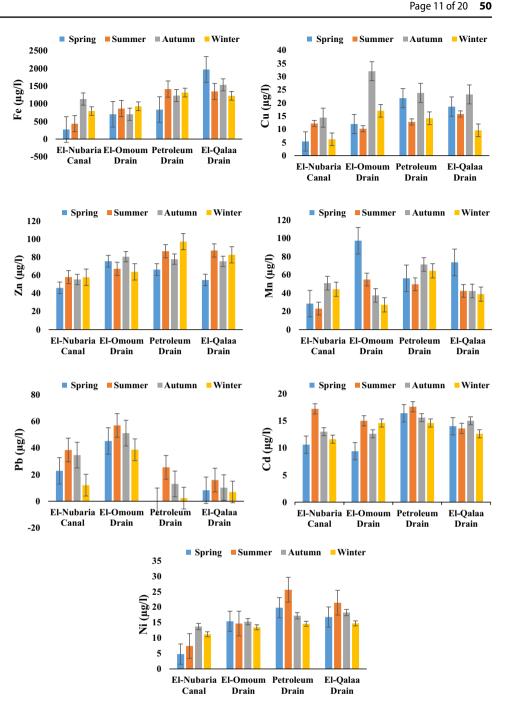
Pollution index (PI)

According to the PI values, Mariout Lake water suffers from obviously different contamination grades with the measured metals for aquatic life utilization. Mn, Fe, Zn, Cu, and Ni recorded no pollution effects at any sites along the lake, but Cd and Pb exhibited serious pollution effects at all the studied locations according to the aquatic life criteria, as indicated in Table 5.

Metal index (MI)

According to the MI values, all the selected sites along Mariout Lake were seriously threatened by metal pollution for aquatic life usage (MI>1). The MI values ranged from 24.48 to 35.34 along the studied sites in Mariout Lake. As

Fig. 2 Seasonal variations of the studied heavy metals concentrations (µg/l) in Mariout Lake inflows during 2017-2018



indicated in Table 6, the most polluted site was site (8) in the Main Basin.

Lake may be exposed to greater risks. Nadmitov et al. (2015) reported that at an HPI > 100, the overall pollution levels must be assessed as undesirable for an aquatic ecosystem.

Heavy metal pollution index (HPI)

Table 7 shows that the HPI values in Mariout Lake water ranged from 772.24 to 1151.33. The most polluted site was site (8) in the Main Basin. These results demonstrate that all the studied metals have significant polluting effects on aquatic life usage. Following a critical HPI value of 100, the data indicate that aquatic organisms living in Mariout

Heavy metals in fish

In this study, Tilapia species (O. niloticus) caught from Mariout Lake was selected to assess the metals' accumulation pattern because Egypt is the second-largest producer of Tilapia after China (FAO 2014). Besides its ecological value, Tilapia is economically important as a food source

 Table 5
 Pollution index (PI) of the measured heavy metals in Mariout Lake water according to the guideline levels for aquatic life

Site	Fe		Mn		Zn		Cu	
	PI value	Effect	PI value	Effect	PI value	Effect	PI value	Effect
1	0.45	No effect	0.44	No effect	0.56	No effect	1.14	No effect
2	0.54	No effect	0.47	No effect	0.55	No effect	0.70	No effect
3	0.63	No effect	0.38	No effect	0.39	No effect	1.04	No effect
4	0.49	No effect	0.36	No effect	0.42	No effect	1.11	No effect
5	0.50	No effect	0.41	No effect	0.34	No effect	1.07	No effect
6	0.46	No effect	0.34	No effect	0.30	No effect	0.85	No effect
7	0.64	No effect	0.60	No effect	0.30	No effect	1.11	No effect
8	0.75	No effect	0.59	No effect	0.54	No effect	1.43	No effect
9	0.60	No effect	0.61	No effect	0.55	No effect	1.12	No effect
10	0.52	No effect	0.44	No effect	0.43	No effect	1.11	No effect
11	0.42	No effect	0.39	No effect	0.23	No effect	0.91	No effect
12	0.20	No effect	0.29	No effect	0.37	No effect	0.90	No effect
13	0.34	No effect	0.33	No effect	0.52	No effect	1.02	No effect
$Mean \pm SD$	0.50 ± 0.14		0.44 ± 0.11		0.42 ± 0.11		1.04 ± 0.18	
Site	Ni			Cd		Pb		
	PI value	e E	ffect	PI value	Effect	PI	value	Effect
1	0.15	N	o Effect	7.37	Seriously	7.	09	Seriously
2	0.16	Ν	o Effect	6.56	Seriously	8.	44	Seriously
3	0.19	Ν	o Effect	4.84	Seriously	9.	89	Seriously
4	0.18	Ν	o Effect	6.24	Seriously	9.	16	Seriously
5	0.15	Ν	o Effect	6.41	Seriously	8.	80	Seriously
6	0.16	Ν	o Effect	5.97	Seriously	10.	21	Seriously
7	0.21	Ν	o Effect	6.07	Seriously	11.	57	Seriously
8	0.16	Ν	o Effect	8.00	Seriously	13.	63	Seriously
9	0.20	Ν	o Effect	6.82	Seriously	12.	42	Seriously
10	0.20	Ν	o Effect	5.79	Seriously	10.	31	Seriously
11	0.15	Ν	o Effect	5.62	Seriously	9.	08	Seriously
12	0.18	Ν	o Effect	6.47	Seriously	10.	76	Seriously
13	0.13	Ν	o Effect	6.03	Seriously	9.	96	Seriously
Mean \pm SD	0.17 ± 0	0.02		6.32 ± 0.79		10.	10 ± 1.73	

 Table 6
 Metal index (MI) of the measured heavy metals in Mariout

 Lake water according to the guideline levels for aquatic life

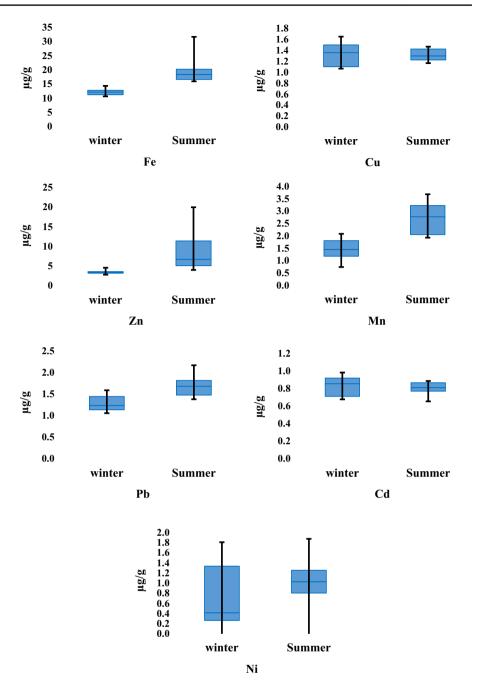
 Table 7
 Heavy metal pollution index (HPI) of the measured metals in

 Mariout Lake water according to the guideline levels for aquatic life

Site	MI value	Rank	Site	MI value	Rank
1	25.44	Polluted	7	29.05	Polluted
2	26.23	Polluted	8	35.34	Polluted
3	25.39	Polluted	9	31.21	Polluted
4	25.37	Polluted	10	27.21	Polluted
5	25.87	Polluted	11	25.20	Polluted
6	26.87	Polluted	12	26.18	Polluted
7	29.05	Polluted	13	24.48	Polluted
Mean \pm SD	27.22 ± 3.0	4			

Site	HPI value	Category	Site	HPI value	Category
1	974.25	Polluted	8	1151.33	Polluted
2	923.97	Polluted	9	1010.02	Polluted
3	772.24	Polluted	10	874.37	Polluted
4	863.14	Polluted	11	854.23	Polluted
5	874.69	Polluted	12	926.14	Polluted
6	919.11	Polluted	13	850.32	Polluted
7	923.81	Polluted			
Mean \pm SD	916.74±92				

Fig. 3 Multiple box and whisker plots of the measured heavy metals concentrations in muscles of Oreochromis niloticus collected from Mariout Lake



for low-income families. It is one of the most common fish species used in toxicological studies (Osman et al. 2010) because it presents some characteristics that may make it a good biomonitor for describing the quality of aquatic systems and testing levels of metal pollution (Ghannam 2021).

The seasonal variations of heavy metal concentrations in edible tissues of the *Tilapia* species (*O. niloticus*) caught from Mariout Lake are graphically represented in Fig. 3. The concentrations of the studied metals were found in the ranges of 14.79–30.54 and 10.25–13.96, 1.33–3.08 and 0.9–2.24, 3.44–19.4 and 2.93–4.7, 1.15–1.45 and 0.84–1.43, 0.58–2.49 and 0.11–2.47, 1.27–2.06 and 1.04–1.57, and 0.73–0.96 and

0.56–0.87 μ g/g for Fe, Mn, Zn, Cu, Ni, Pb, and Cd during the summer and winter seasons, respectively. In general, the annual average contents of *O. niloticus* muscles in the lake were in the order of Fe > Zn > Mn > Pb > Ni > Cu > Cd.

The maximum concentration values of all the measured heavy metals in *O. niloticus* muscles from Mariout Lake were observed in the summer season at site (2) in the South-West Basin for Fe, Mn, and Pb; at site (11) in the North–West Basin for Zn and Cd; and in the Main Basin at sites (7 and 10) for Cu and Ni, respectively. In general, higher concentrations of heavy metals in fish are mainly attributed to industrial effluents, municipal sewage runoff,

Organization/Country	References	Fe	Mn	Zn	Cu	Ni	Pb	Cd
FAO	FAO (1983)			30	30	0.5–0.6	0.5	
FAO/WHO	FAO/WHO (1989)			40	30		0.5	0.5
WHO	WHO (1989)	100	1	100	30			
USFDA					70-80			
Australia and New Zealand*	ANZECC/ARMCANZ (2000)						2	0.2
England	MAFF (2000)			50	20		2	0.2
EU	EU (2001)				10		0.1	0.1
Europe	EC (2005)						0.2	0.05
Turkey	Dural et al. (2007)		20					
Present Study	Min	10.25	0.9	2.93	0.84	0.11	1.04	0.56
	Max	30.54	3.08	19.4	1.45	2.49	2.06	0.96
	Annual Mean	15.60	1.92	6.14	1.23	1.32	1.44	0.80

Table 8 Minimum, maximum and annual mean of heavy metals concentrations ($\mu g/g$ wet weight) in muscles of Oreochromis niloticus collected from Mariout Lake in comparison with the international standard permissible limits

*µg/g dry weight

untreated waste dumping, and urbanization (Ali and Chidambaram 2021). On the other hand, the minimum concentrations of all metals were recorded in the winter season at site (10) in the Main Basin for Fe, Mn, and Cu; at site (6) in the Aquacultural Basin for Zn and Cd; at site (2) in the South–West Basin for Ni; and at site (7) in the Main Basin for Pb. ANOVA results indicated that there were temporal significant differences ($p \le 0.05$) for Fe, Mn, and Cu concentrations in muscles of *O. niloticus* from Mariout Lake. However, Pb concentrations showed high temporal significant differences ($p \le 0.01$).

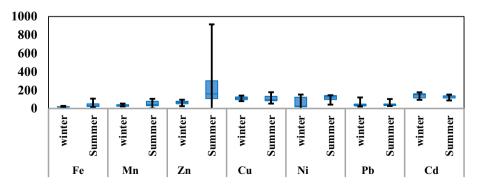
Table 8 illustrates the minimum, maximum, and annual means of the measured heavy metal concentrations in edible tissues of the *Tilapia* species (*O. niloticus*) caught from Mariout Lake compared to the international standard levels. It is clear that the concentrations of Fe, Zn, and Cu were below the international permissible limits, but that of Cd exceeded the corresponding standard guidelines. However, Pb concentrations were below the guideline value of 2.0 μ g/g (ANZECC 2000; MAFF 2000) but exceeded the other international permissible levels. Furthermore, Ni concentrations were found to be higher than the estimated guideline of

 $(0.5-0.6) \mu g/g$ (FAO 1983) but were below the standard level of (70–80) $\mu g/g$ (USFDA 1993). The concentrations of Mn also exceeded the guideline of 1.0 $\mu g/g$ (WHO 1989) but were below the permissible limit of 20 $\mu g/g$ stipulated by Dural et al. (2007). On the other hand, Table S3 (in the supplementary file) shows the levels of the studied metals in *O. niloticus* muscles from Mariout Lake and in other Northern Nile-Delta Lakes.

Bioaccumulation factor (BAF)

Kucuksegin et al. (2006) have identified the bioaccumulation of heavy metals in the tissues of marine organisms as an indirect measure of the abundance and availability of metals in the marine environment. It is always important to determine the bioaccumulation capacity of heavy metals by fish to assess the potential risks to human health and to take appropriate actions to protect public health and the environment (Ambedkar and Muniyan 2011). Generally, many factors may affect metals' uptake and accumulation in aquatic organisms, including the organism characteristics, metal concentrations, exposure time, mode of metals'

Fig. 4 Bioaccumulation factors of the measured heavy metals in muscles of Oreochromis niloticus collected from Mariout Lake



uptake, and physiochemical conditions of water (Goher et al. 2019b; Ghannam 2021).

The BAFs of the measured heavy metals in the edible tissues of the Tilapia species (O. niloticus) caught from Mariout Lake are represented in Fig. 4. Results showed that the BAF values were 12.48–112.26, 10.3–124.22, 35.46-1010.66, 76.4-200, 11.96-210.45, 20.18-118.79, and 78.4-160.78 for Fe, Mn, Zn, Cu, Ni, Pb, and Cd, respectively. The highest BAF (1010.66) was observed for Zn in the summer season at site (11) in the North-West Basin, but the lowest BAF (10.3) was recorded for Mn in the winter season at site (7) in the Main Basin. Generally, the accumulation order of metals in O. niloticus muscles from the lake was Zn > Cd > Cu > Ni > Pb > Mn > Fe. Cadmium bioaccumulation is generally hazardous, as it is a nonessential trace element. According to the categories of BAFs obtained by Olayinka–Olagunju et al. (2021), all the studied heavy metal BAF values were less than 1000 (BAFs < 1000), indicating no probability of accumulation, except the highest BAF for Zn in the summer season (1010.66) at site (11), which indicates that Zn may be a bioaccumulative metal because it was identified that phytoplanktons bloom during summer, and subsequently, its consumption results in bioaccumulation (Ali and Chidambaram 2021).

Human health risk assessment

Risk assessment is a methodology that identifies, characterizes, and analyzes toxic elements to qualify their adverse effects within a specified timeframe and to estimate the potential risk levels for humans. This approach was applied to identify the exposure and tendency of the contaminated fish in Mariout Lake to the human body (Sánchez et al. 2022).

Estimated daily intake (EDI)

As fish consumption is a possible source of heavy metal accumulation in humans, it is important to consider the daily intake of metals through fish consumption. The following assumptions were made in this study to estimate the risks of heavy metals from Tilapia species (O. niloticus) consumption at the extreme: the ingested dose was equal to the absorbed pollutant dose, and cooking did not affect the pollutants, as assumed by Kortei et al. (2020) and Yacoub et al. (2021). Table 9 shows that the EDI values of all heavy metals through consumption of 0.057 kg of fish flesh by a 70 kg Egyptian person per day are lower than their provisional tolerable daily intake (PTDI₇₀) values for each metal. The results revealed that Mn, Ni, Cu, Pb, and Cd constituted the lowest daily intake, while Fe and Zn constituted the highest daily intake, as shown in Fig. 5. These results are supported by Yacoub et al. (2021) for El-Manzala Lake.

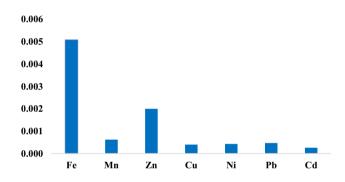


Fig. 5 Estimated daily intake (EDI) of heavy metals via the consumption of the muscles of Oreochromis niloticus collected from Mariout Lake

 Table 9
 Estimated daily intake (EDI) and target hazard quotient (THQ) values of the studied heavy metals due to consumption of muscles of Oreochromis niloticus collected from Mariout Lake

Metal	MC (mg/kg)	RfD (a) (mg/ kg/day)	EDI (mg/kg/day)	PTDI (b) (mg/day)	PTDI70 (c)	THQ	IRFb (mg/day)
Fe	15.60	0.7	0.00508	0.8	56	0.018	3420
Mn	1.92	0.14	0.00062	0.14	9.8	0.112	580
Zn	6.14	0.3	0.00200	1	70	0.017	4680
Cu	1.23	0.04	0.00040	0.5	35	0.025	2325
Ni	1.32	0.02	0.00043	0.005	0.35	0.054	2733
Pb	1.45	0.0035	0.00047	0.0036	0.25	0.338	175
Cd	0.80	0.001	0.00026	0.001	0.07	0.651	89
HI						1.214	

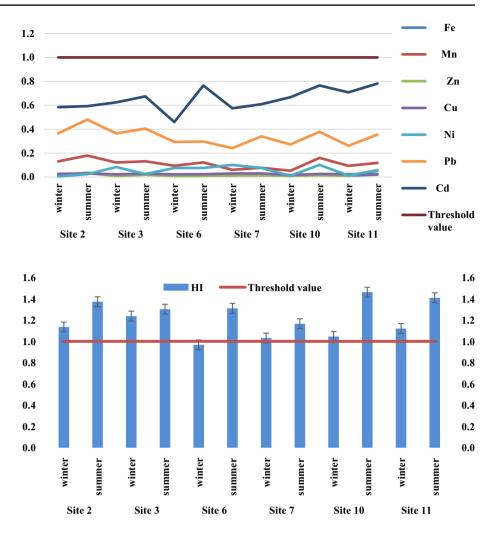
MC average of metal concentration

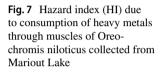
(a) RfD = oral reference dose of individual heavy metal according to USEPA (2013)

(b) PTDI = provisional tolerable daily intake according to Türkmen et al. (2009) and FAO/WHO (2010)

(c) $PTDI70 = PTDI \times 70 \text{ kg}$

Fig. 6 Target hazard quotient (THQ) in relation to consumption of heavy metals through muscles of Oreochromis niloticus collected from Mariout Lake





Target hazard quotient (THQ) and hazard index (HI)

Table 9 and Figs. 6, 7 show the non-carcinogenic human hazard risks of heavy metals through edible tissue exposure routes using the seven measured metals to calculate the THQ and HI values. The present results indicate that the obtained THQ values due to the exposure route for all investigated heavy metals do not demonstrate the risks (THQ < 1) associated with the consumption of 57 g/day of *O. niloticus* from Mariout Lake. The THQ values were in the order of Cd (0.651) > Pb (0.338) > Mn (0.112) > Ni (0.054) > Cu (0.025) > Fe (0.018) > Zn (0.017). Results showed that the THQ values of the most studied metals were ten times less than the non-hazardous limit (THQ < 1). Therefore, there are no non-carcinogenic health risks from ingestion of these metals individually through *Tilapia* species (*O. niloticus*) consumption, according to Custodio et al. (2021).

The IRF_b value is the amount of edible fish tissues that must be consumed daily for the health risks of each element to become evident (THQ \geq 1). IRF_b-calculated values in Table 9 revealed that only a small amount of edible tissues of the *Tilapia* species (*O. niloticus*) caught from Mariout Lake is required to produce adverse health effects for Cd and Pb, whereas larger amounts of edible fish tissues are required to produce health risks for Fe, Mn, Zn, Cu, and Ni.

The health risk assessment of metal exposure from the consumption of *Tilapia* fish species from Mariout Lake should allow for the combined effects of the various heavy metals studied. Therefore, the HI value is necessary to assess the health risk associated with fish consumption (Zhu et al. 2016). The total non-carcinogenic HI value for the analyzed heavy metals was 1.241, as shown in Table 9 and Fig. 7.

This result revealed that the risks of consuming edible tissues of *Tilapia* species (*O. niloticus*) caught from Mariout Lake slightly exceed the threshold value of 1. The human health risk assessment for heavy metal contamination deline-ated low-hazard risks in edible tissues. According to Ukoha et al. (2014), the HI value (HI = 1.241) is classified in the moderate hazard risk level (1 < HI < 10). This moderate value is mainly based on the THQ values of Cd (0.651) and Pb (0.338). In this study, the major health risk-contributing metal was Cd (53.6%), followed by Pb (27.82%), Mn

(9.19%), Ni (4.43%), Cu (2.06%), and Fe (1.5%), and Zn (1.37%).

Conclusion

Mariout Lake is one of the five Egyptian Mediterranean lagoons. Although it is the smallest one of them, it is the most polluted. It is a closed lake that has no direct connection with the Mediterranean Sea. It receives a huge amount of wastes via different drains. The results indicated that the highest concentrations of the measured heavy metals are observed in the Main Basin due to the effects of domestic, agricultural, and industrial effluents from El-Qalaa and El-Umoum Drains. The long-term changes of the studied metals in Mariout Lake water showed significantly irregular vibrations from 1981. The pollution index detected that Cd and Pb have serious pollution effects on aquatic life, while the metal index and heavy metal pollution index indicated the high-water pollution of Mariout Lake and it is inappropriate for fish and different aquatic organisms. In spite of that, the levels of most metals in O. niloticus muscles were within the international permissible limits. Generally, the bioaccumulation order of heavy metals in O. niloticus muscles was Zn > Cd > Cu > Ni > Pb > Mn > Fe. The obtained results showed that the target hazard quotient due to the ingestion of HMs through O. niloticus was less than the hazard limit (THQ < 1), while the hazard index recorded a slight increase (HI = 1.241), indicating moderate potential non-carcinogenic health risks from the collective ingestion of these metals through consumption of muscles of O. niloticus from Mariout Lake. Our study recommends concerted efforts of all the concerned and responsible authorities to enforce law 48/1982 that prevents the dumping of pollutants into the water bodies, in addition to the application of biological and chemical treatment for wastewater discharged into Mariout Lake to rapidly rehabilitate it and improve its water quality.

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Code availability Non applicable.

Declarations

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Consent to participate All authors voluntarily agree to participate in this research study.

Consent to publish All authors voluntarily approved the publication of this research study.

Ethical approval Not applicable.

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