

NIH Public Access

Author Manuscript

J Am Chem Soc. Author manuscript; available in PMC 2009 June 24.

Published in final edited form as: *J Am Chem Soc*. 2006 March 1; 128(8): 2552–2553. doi:10.1021/ja058337u.

Asymmetric Synthesis of Hydrobenzofuranones via Desymmetrization of Cyclohexadienones using the Intramolecular Stetter Reaction

Qin Liu and Tomislav Rovis

Department of Chemistry, Colorado State University, Fort Collins, Colorado 80523

Abstract

Dearomatization of phenols followed by oxidation affords cyclohexadienyloxyacetaldehydes, which produce hydrobenzofuranones via asymmetric intramolecular Stetter reaction in good to excellent yield. Quaternary as well as up to three contiguous stereocenters may be formed in good to excellent enantioselectivities and high diastereoselectivities.

Dearomatization of aromatic compounds provides a useful alicyclic synthetic building block due to its high economy and simple elegance.¹ When coupled with a stereoselective process, it has the potential for affording enantioenriched material from commonly available precursors in rapid fashion.² Reactivity umpolung³ affords an opportunity to take advantage of unobvious, complementary bond disconnections in the synthesis of complex molecules. One such reaction is the Stetter reaction, the nucleophile-catalyzed addition of an aldehyde to a Michael acceptor. ⁴ If the Michael acceptor involves a prochiral alkene, this reaction generates new stereocenters. ⁵ We have recently developed a family of catalysts capable of inducing highly enantioselective intramolecular Stetter reactions.⁶ As part of an effort to extend the power of this transformation, we considered that cyclohexadienones, readily available from dearomatization of phenols,⁷ could be suitable substrates for a desymmetrizing Stetter reaction. Herein we disclose that treatment of such substrates with chiral triazolium salts affords, in high yields and enantioselectivities, hydrobenzofurans, which are core skeletons found in many natural products.⁸

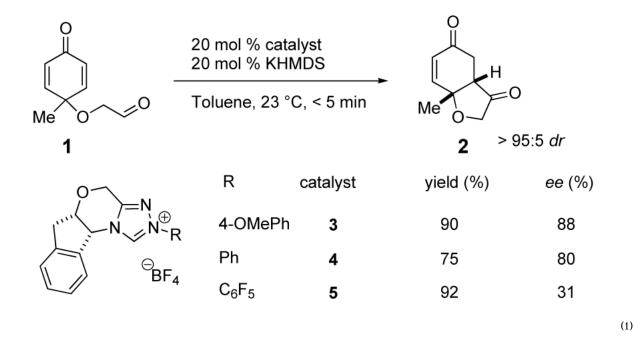
The requisite substrates are readily accessible from the corresponding phenols (Scheme 1). Hypervalent iodine reagents are used in conjunction with glycol to afford the dienone alcohols in good yield, whereupon Dess-Martin oxidation leads to the aldehydes and sets the stage for the asymmetric Stetter reaction. Upon cyclization, the desired product hydrobenzofuranones would contain at least two contiguous stereocenters.

The reactivity of a series of catalysts⁶ was studied using substrate **1** (eq 1). The reactions were conducted in the presence of 20 mol% catalyst and 20 mol% KHMDS in toluene at 25 °C, and proceeded to completion in less than 5 min affording desired product **2**. Aminoindanol-derived triazolium salt **3** bearing an anisyl substituent was found to be the best catalyst precursor for this reaction, again reflecting the subtle impact of electronics arising from aryl substitution.⁶

A range of substrates was synthesized following the general procedure (Scheme 1). Under optimized conditions,⁹ the asymmetric intramolecular Stetter reaction proceeds very well, Table 1. The enantioselectivities are excellent (92-94 % *ee*) when groups at the 4-position of

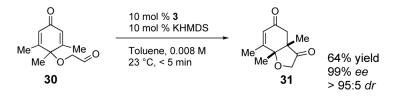
E-mail: rovis@lamar.colostate.edu.

the substrates are methyl, ethyl, isopropyl and tert-butyl (entries 1-4). Aromatic substituents result in slightly lower, but still synthetically useful enantioselectivities (Table 1, entries 5-6). Similar effects are observed with more functionalized sidechains at that position, entries 7-9. With all substrates, the reaction proceeds with very high diastereoselectivities (> 95:5 by ¹H NMR).¹⁰



We have previously documented the ability of the Stetter reaction to form contiguous stereocenters with trisubstituted alkene acceptors.^{6d} The use of 2,4,6-trisubstituted phenols as precursors in this chemistry allow a rapid entry into suitably functionalized substrates to test whether the desymmetrization would proceed. In the event, the dienone derived from 2,4,6-trimethylphenol proved remarkably efficient. This reaction provides hexahydrobenzofuranone **23** possessing three contiguous stereocenters in excellent yield and ee, entry 1 in Table 2. The reaction is tolerant of alkoxymethyl groups at the 2-position, providing **25** in excellent ee and good yield, entry 2. No elimination of the methoxy group is observed under these conditions. Remarkably, this reaction is also tolerant of as many as three *tert*-butyl groups in the vicinity of the reaction center and the substrates provide one (entry 3), or two (entry 4) neopentyl stereocenters in excellent selectivity. In each of the cases examined to date, a single diastereomer is formed.¹⁰

Commercially available 3,4,5-trimethylphenol provides an opportunity to test whether this approach is capable of forming quaternary stereocenters as per our previous report.^{6c} In the event cyclization of substrate **30** affords one quaternary stereocenter adjacent to a tertiary ether in excellent enantioselectivity and good yield (eq 2).



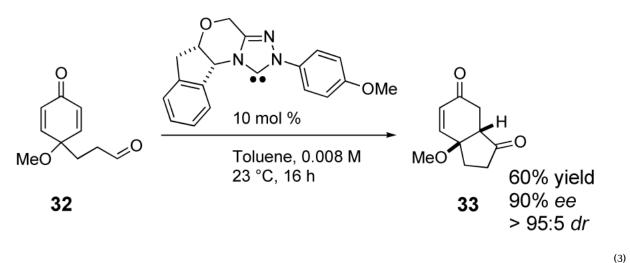
JAm Chem Soc. Author manuscript; available in PMC 2009 June 24.

(2)

Liu and Rovis

Although the bulk of our work focused on oxygen-tethered substrates due to their accessibility, the reaction is also capable of forming carbocycles. To that end, when **32** is subjected to the reaction conditions, hydrindane **33** is isolated in good yield and 90% ee, eq 3.

In conclusion, a family of phenol-derived substrates have been



successfully synthesized and found to undergo asymmetric intramolecular Stetter reactions in very good yields. Quaternary stereocenters and up to three contiguous stereocenters may be formed in excellent enantioselectivities and diastereoselectivities.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgment

Dedicated to Professor David A. Evans on the occasion of his 65th birthday. We thank the National Institute of General Medical Sciences (GM72586) and Johnson and Johnson (Focused Giving) for partial support of this research. T.R. thanks Merck Research Laboratories, GlaxoSmithKline, Eli Lilly, Amgen, and Boehringer Ingelheim for unrestricted support, and the Monfort Family Foundation for a Monfort Professorship. T.R. is a fellow of the Alfred P. Sloan Foundation. We thank Jerry Murry and Michael C. Hillier (Merck Research Laboratories) for a generous gift of aminoindanol. We thank Mark S. Kerr, Oren P. Anderson and Susan M. Miller (CSU) as well as Charles Campana (Bruker AXS) for X-ray analysis.

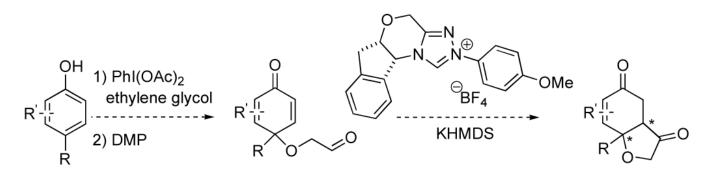
References

- (1)(a). Woodward RB. J. Am. Chem. Soc 1940;62:1208.Waring, AJ. Advances in Alicyclic Chemistry. Hart, H.; Karabatsos, GJ., editors. Academic Press; New York: 1966. p. 129-256. (c) Corey EJ, Girotra NN, Mathew CT. J. Am. Chem. Soc 1969;91:1557–1559. (d) Wiesner K, Tsai TYR, Nambiar KP. Can. J. Chem 1978;56:1451–1454. (e) Semmelhack MF, Harrison JJ, Thebtaranonth Y. J. Org. Chem 1979;44:3275–3277. (f) Corey EJ, Dittami JP. J. Am. Chem. Soc 1985;107:256– 257. (g) Mander LN. Synlett 1991;3:134–144. (h) Kopach ME, Harman WD. J. Am. Chem. Soc 1994;116:6581–6592. (i) Maruoka K, Ito M, Yamamoto H. J. Am. Chem. Soc 1995;117:9091– 9092. (j) Clive DLJ, Fletcher SP, Liu D. J. Org. Chem 2004;69:3282–3293. [PubMed: 15132533] (k) Villar FV, Kolly-Kovac T, Equey O, Renaud P. Chem. Eur. J 2003;9:1566–1577.
- (2)(a). Pearson AJ, Zhu PY, Youngs WJ, Bradshaw JD, McConville DB. J. Am. Chem. Soc 1993;115:10376–10377. (b) Pearson AJ, Milletti MC, Zhu PY. Chem. Commun 1995;8:853–854.
 (c) Schultz AG. Chem. Commun 1999:1263–1271. (d) Imbos R, Minnaard AJ, Feringa BL. J. Am. Chem. Soc 2002;124:184–185. [PubMed: 11782165] (e) Hayashi Y, Gotoh H, Tamura T,

J Am Chem Soc. Author manuscript; available in PMC 2009 June 24.

Yamaguchi H, Masui R, Shoji M. J. Am. Chem. Soc 2005;127:16028–16029. [PubMed: 16287281] (f) Studer A, Schleth F. Synlett 2005;20:3033–3041.

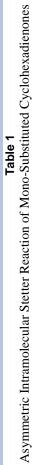
- (3). Seebach D. Angew. Chem., Int. Ed. Engl 1979;18:239-258.
- (4)(a). Stetter H. Angew. Chem., Int. Ed. Engl 1976;15:639–648. (b) Stetter H, Kuhlmann H. Org. React 1991;40:407–496.
- (5)(a). Enders D, Breuer K, Runsink J, Teles JH. Helv. Chim. Acta 1996;79:1899–1902. (b) Enders D, Balensiefer T. Acc. Chem. Res 2004;37:534–541. [PubMed: 15311952] (c) Christmann M. Angew. Chem. Int. Ed 2005;44:2632–2634. (d) Nair V, Bindu S, Sreekumar V. Angew. Chem. Int. Ed 2004;43:5130–5135. (e) Zeitler K. Angew. Chem. Int. Ed 2005;44:7506–7510. (f) Pesch J, Harms K, Bach T. Eur. J. Org. Chem 2004;9:2025–2035. (g) Mennen SM, Blank JT, Tran-Dube MB, Imbriglio JE, Miller S. J. Chem. Commun 2005:195–197. (h) Nahm MR, Linghu X, Potnick JR, Yates CM, White PS, Johnson JS. Angew. Chem. Int. Ed 2005;44:2377–2379.
- (6)(a). Kerr MS, Read de Alaniz J, Rovis T. J. Am. Chem. Soc 2002;124:10298–10299. [PubMed: 12197730] (b) Kerr MS, Rovis T. Synlett 2003;12:1934–1936. (c) Kerr MS, Rovis T. J. Am. Chem. Soc 2004;126:8876–8877. [PubMed: 15264801] (d) Read de Alaniz J, Rovis T. J. Am. Chem. Soc 2005;127:6284–6289. [PubMed: 15853335]
- (7)(a). Pelter A, Elgendy SMA. J. Chem. Soc. Perkin Trans. 1 1993:1891–1896. (b) Moriarty RM, Prakash O. Org. React 2001;57:327–415. (c) Trân-Huu-Dâu M, Wartchow R, Winterfeldt E, Wong Y. Chem. Eur. J 2001;7:2349–2369. (d) Canesi S, Bouchu D, Ciufolini MA. Org. Lett 2005;7:175–177. [PubMed: 15646951]
- (8)(a). Chu-Moyer MY, Danishefsky SJ, Schulte GK. J. Am. Chem. Soc 1994;116:11213–11228. (b) Guth H. Helv. Chim. Acta 1996;79:1559–1571. (c) Yamashita M, Ohta N, Shimizu T, Matsumoto K, Matsuura Y, Kawasaki I, Tanaka T, Maezaki N, Ohta S. J. Org. Chem 2003;68:1216–1224. [PubMed: 12585858] (d) Clive DLJ, Fletcher SP. Chem. Commun 2003:2464–2465. (e) Booker-Milburn KI, Hirst P, Charmant JPH, Taylor LHJ. Angew. Chem. Int. Ed 2003;42:1642–1644. (f) Taber DF, Neubert TD, Rheingold AL. J. Am. Chem. Soc 2002;124:12416–12417. [PubMed: 12381175] (g) Takao K, Tsujita T, Hara M, Tadano K. J. Org. Chem 2002;67:6690–6698. [PubMed: 12227798] (h) Burke SD, Cobb JE, Takeuchi K. J. Org. Chem 1985;50:3420–3421. (i) Yao S, Johannsen M, Hazell RG, Jørgensen KA. J. Org. Chem 1998;63:118–121. [PubMed: 11674051] (j) Jonasson C, Rönn M, Bäckvall J. J. Org. Chem 2000;65:2122–2126. [PubMed: 10774035] (k) Germain J, Deslongchamps P. J. Org. Chem 2002;67:5269–5278. [PubMed: 12126415]
- (9). Lower catalyst loading or higher concentration result in comparable yields of product but lower enantioselectivity (2 mol% catalyst: 84% yield, 83% ee; 0.04 M: 90% yield, 88% ee).
- (10). Relative and absolute configurations determined by nOe and X-ray for several compounds and the rest assigned by analogy (see SI for details). We have been unable to form the complementary diastereomers even by base-induced epimerization. A semi-empirical calculation of 23, 25, 27 and 29 suggests the minor diastereomers are as much as 9 kcal/mol less stable. Examination of GC traces suggests that this reaction is likely far more selective than 95% of a single diastereomer.



Scheme 1. Asymmetric Synthesis of Hydrobenzofuranones

NIH-PA Author Manuscript

NIH-PA Author Manuscript



	,				
	∘={	10 mol % 3 10 mol % KHMDS	0		
		Toluene, 0.008 M 23 °C, < 5 min	\ 	R 05:5 dr	
entry ^a	м	substrate	product	Yield (%)	ee (%)p
_	Me	1	2	06	92
2	Et	6	7	86	94
3	¹ Pr	8	6	87	94
4	'Bu	10	11	86	94
5	Ph	12	13	87	88
9	4-BrPh	14	15	78	85
7	CH ₂ OAc	16	17	86	83
8	CH_2CH_2OMe	18	19	86	82
6	CH ₂ CH ₂ CO ₂ Me	20	21	94	87

J Am Chem Soc. Author manuscript; available in PMC 2009 June 24.

 a All reactions conducted in the presence of 10 mol% catalyst and 10 mol% KHMDS in toluene at 23 $^{\circ}$ C.

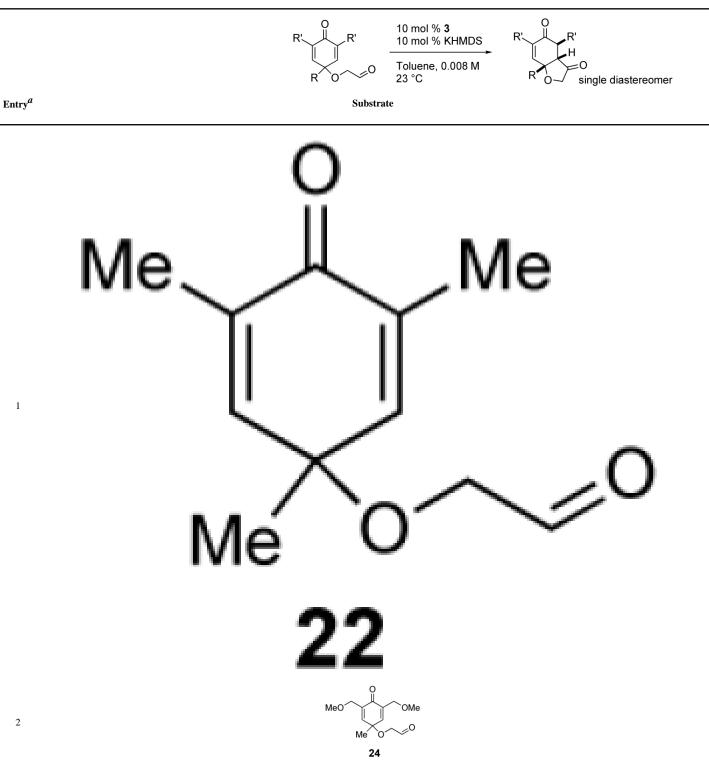
 $b_{\mbox{Determined}}$ betermined by HPLC or GC using a chiral stationary phase.

Liu and Rovis

Liu and Rovis

Reaction of Tri-Substituted Substrates

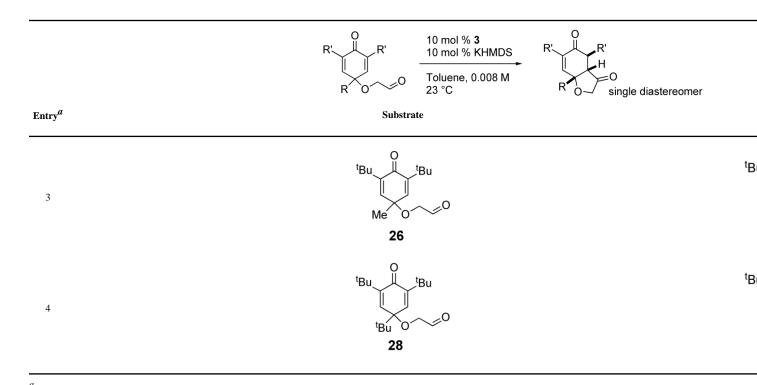




Μ

Me

Liu and Rovis



^aSee Table 1.

^bSee Table 1.

^cReaction time: 5 min.

^dReaction time: 2h.