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Atomic-layer soft plasma etching of MoS₂

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Transition from multi-layer to monolayer and sub-monolayer thickness leads to the many exotic properties and distinctive applications of two-dimensional (2D) MoS₂. This transition requires atomic-layer-precision thinning of bulk MoS₂ without damaging the remaining layers, which presently remains elusive. Here we report a soft, selective and high-throughput atomic-layer-precision etching of MoS₂ in SF₆ + N₂ plasmas with low-energy (<0.4 eV) electrons and minimized ion-bombardment-related damage. Equal numbers of MoS₂ layers are removed uniformly across domains with vastly different initial thickness, without affecting the underlying SiO₂ substrate and the remaining MoS₂ layers. The etching rates can be tuned to achieve complete MoS₂ removal and any desired number of MoS₂ layers including monolayer. Layer-dependent vibrational and photoluminescence spectra of the etched MoS₂ are also demonstrated. This soft plasma etching technique is versatile, scalable, compatible with the semiconductor manufacturing processes, and may be applicable for a broader range of 2D materials and intended device applications.

The electronic, mechanical, optoelectronic and catalytic properties of two-dimensional (2D) transition metal dichalcogenides (TMDs) such as MoS₂ critically depend on the number of atomic stacking layers^{1–10}. MoS₂ has attracted rapidly growing attention from both academia and industry owing to its atomic-layer-dependent properties. Indeed, the electronic bandgap of MoS₂ increases and transits from indirect to direct as the number of stacking layers decreases from multiple layers to monolayer^{11–13}. Phase transition from semiconducting 2H to metallic 1T phases may also occur when the thickness is reduced to monolayer^{14,15}. As such, atomically-thin MoS₂ is a promising material for applications in valleytronic and optoelectronic devices, such as photodetectors, photovoltaics, and light emitters^{11,16,17}; while multilayer MoS₂ is suited for electronic and energy storage devices owing to its tunable bandgaps and loosely bound 2D atomic layers^{18–20}.

However, exfoliated or synthesized MoS₂ films usually contain non-uniform domains with different number of layers²¹. Post-growth thinning by methods of thermal annealing, laser and plasma etching has been actively pursued, but with only limited success^{20–23}. For instance, thermal annealing has been applied to thin few-layer MoS₂ flakes; yet the process was very time- and energy-consuming^{22,23}. The thinned MoS₂ films also possibly formed MoO₃ due to high-temperature oxidation²³ and showed non-uniform thickness across the surface and shrank laterally due to the uneven thermal sublimation effect²². Laser heating was also demonstrated to reduce the MoS₂ thickness through thermal sublimation induced by light absorption²¹. However, monolayer domains were only achieved with careful control of the uniformity and dose of the laser exposure, which is challenging for scale-up^{8,21}. In addition, argon-based plasmas facilitated layer-by-layer MoS₂ thinning²⁰, but may leave unwanted residues on the etched surface due to energetic Ar⁺ ion bombardment. It is therefore highly desired to develop a MoS₂ thinning process with time- and energy-efficiency, ease to control and scale-up, and non-destruction to the remaining layers and substrate.

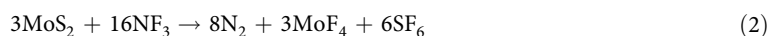
Here we report a soft, selective, high-throughput and uniform large-area plasma etching of MoS₂. Compared to conventional plasma processes where energetic ions cause unwanted damage during etching, the present

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approach is based on reactive plasmas where the ion impact is intentionally reduced by generating the majority of the electrons with the energies insufficient for the effective ion generation or causing damage to the remaining MoS₂ layers. The reactive radicals generated in such plasmas uniformly remove equal numbers of MoS₂ layers irrespective of the initial thickness, without affecting the underlying SiO₂ substrate and the exposed MoS₂ layers. The etching rates can be tuned to achieve complete MoS₂ removal and any desired number of atomic MoS₂ layers including monolayer, starting from the pristine MoS₂ domains of virtually any thickness. Layer-dependent vibrational and photoluminescence spectra of the etched MoS₂ are also demonstrated. The process is fast, reproducible, and compatible with the established semiconductor microfabrication technologies, thus holding great promise to enable next-generation TMD-based devices.

Results

In contrast to the traditional plasma etching processes, the soft plasma etching relies on SF₆+N₂ precursors dissociated in low-power, low-density radio-frequency plasma source (Supplementary Figure S1). Energetic ions can be minimized in such plasmas to significantly reduce structural damage to the 2D films. Specifically, the etching reactions used in this process are:



The key features of this plasma-based process include: (i) strong oxidant NF₃ from reaction (1) can simultaneously react with both Mo and S; (ii) by-products MoF₄, MoF₃, F₂ and SF₄ of reactions (1–3) are volatile and are removed without residues on the etched surface, although some of these by-products are not so environment-friendly; (iii) the precursors and reaction by-products have negligible etching effect on the SiO₂/Si substrate so that a highly-selective etching between MoS₂ and SiO₂ can be achieved; (iv) the rates of the soft etching can be effectively controlled by the plasma power density to enable *fine* (~0.8 mW/cm³) and *fast* (~1.2 mW/cm³) etching modes where any pre-determined number of MoS₂ layers can be removed; (v) same number of MoS₂ layers is removed uniformly across the domains with very different original number of layers. The soft etching mode is effective below the critical value (~1.5 mW/cm³) of the input power density. Beyond this critical value, structural damages to MoS₂ emerge, *i.e.*, harsh etching mode. More details are provided below.

Figure 1a–c show the etching result of a 4-layer MoS₂ flake at an input power of 0.8 mW cm⁻³. The number of layers of pristine MoS₂ flake was determined by optical contrast and atomic force microscopy (AFM), by assuming 0.7 nm as the thickness of a single S-Mo-S layer. From these images, it is found that 2 layers (~1.4 nm) and 1 layer (~0.7 nm) of MoS₂ were successfully removed in two consecutive steps of 4 and 3 min, respectively. For comparison, thermal annealing in vacuum required 1 hr to remove 1 layer of MoS₂²². Figure 1d–f show the etching result of a thick MoS₂ flake starting with ~40 layers at an input power density of 1.2 mW cm⁻³. ~20 layers (~15 nm) of the MoS₂ flake were removed in each step with the duration of 8 min, demonstrating the fast rate and high efficiency of this technique.

Remarkably, the present plasma thinning method had negligible etching effect to the SiO₂/Si substrate. In a control experiment, we have covered half area of the SiO₂/Si substrate by a shadow mask and treated it under the same SF₆+N₂ plasma conditions at either 0.8 or 1.2 mW cm⁻³ for 2 hr. No obvious boundaries or steps between the treated and the covered areas could be found from both AFM and optical images (Supplementary Figure S2), suggesting that the SF₆+N₂ plasma chemistry is highly selective to MoS₂. As the SiO₂/Si substrate is generally used as a reference in the AFM measurements to calculate the number of MoS₂ layers, this allowed us to precisely determine the etching rate at different input power densities.

Figure 1g plots the etched thickness as a function of etching time at the power densities of 0.8 and 1.2 mW/cm³. Interestingly, the etching rates were low in the first 2–3 min in both cases. This can be explained by considering that reaction (1) requires certain time for the plasma to dissociate the SF₆ molecules and produce enough NF₃ radicals to etch MoS₂, particularly at the beginning of the plasma discharge. The etching rates became stable after 3 min, as demonstrated by the linear fitting curves in Fig. 1g. The etching rates were about 2.8 and 3.6 nm/min for the power densities of 0.8 and 1.2 mW/cm³, respectively. Therefore, the former is referred as the fine etching mode because the low etching rate is best suited for high-precision removal of MoS₂ atomic layers. On the other hand, the latter case is referred to as the fast etching mode which has higher etching rates and is more efficient for etching thicker MoS₂ films.

By combining the fine and fast etching modes, domains of the arbitrary number of layers of MoS₂ can be uniformly thinned across the sample. As shown in Fig. 1h,i and S3, a pristine MoS₂ flake with 90 layers (~63 nm) and a size of ~60 μm was firstly subjected to the fast etching mode at 1.2 mW/cm³ for 20 min. This led to a thinned MoS₂ flake with only 3 layers (~2.4 nm). Then, the MoS₂ flake was further etched under the fine mode for another 4 min, resulting in the formation of monolayer MoS₂. The height of the MoS₂ monolayer (~1.0 nm) was slightly larger compared to the monolayer MoS₂ obtained by mechanical exfoliation²⁴, which can be attributed to surface corrugation or the presence of adsorbed or trapped molecules¹³.

Importantly, the surface roughness of etched samples remained unchanged as compared to the pristine samples (Supplementary Figure S3), suggesting that the plasma etching process is indeed soft with the minimum damage induced. The layers that were left underneath after the etching were homogeneous. This is in stark contrast to the laser thinning method where ~3 times larger surface roughness was observed, possibly due to the different reaction rates of S and Mo atoms and/or the unremoved MoS₂ traces on the surface²¹. The smooth

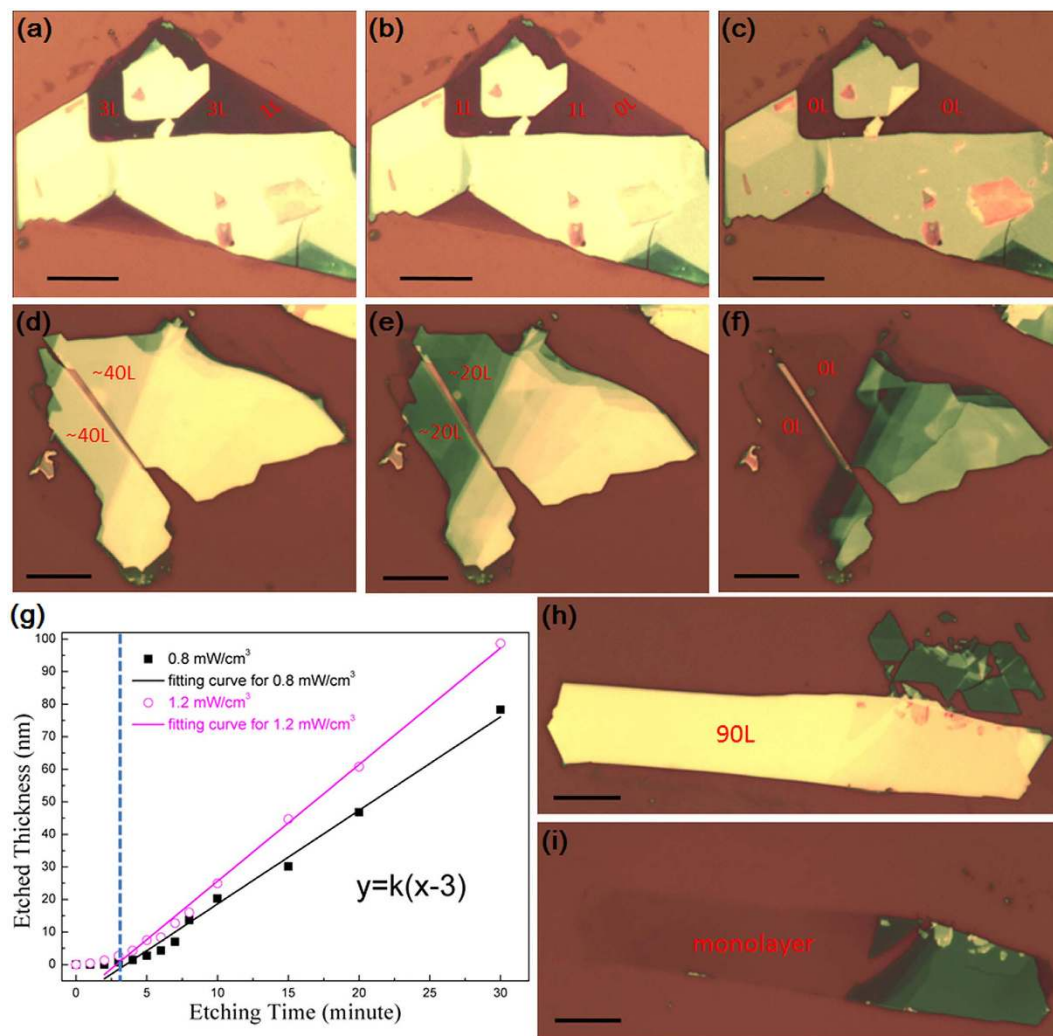


Figure 1. Plasma-etched MoS₂ flakes at different input power densities. (a) Pristine MoS₂ flake showing 1 and 3 layers; (b) MoS₂-free (0L) and monolayer (1L) after 4 min *fine* plasma etching at 0.8 mW/cm²; and (c) MoS₂-free surface after another 3 min *fine* etching at 0.8 mW/cm². (d) Pristine MoS₂ flake with ~40 layers; (e) ~20 layers after 8 min *fast* plasma etching at 1.2 mW/cm²; (f) MoS₂-free surface after another 8 min *fast* etching at 1.2 mW/cm². (g) The etched thickness as a function of time at 0.8 and 1.2 mW/cm². (h) Large-area pristine MoS₂ flake with ~90 layers; (i) MoS₂ monolayer etched from the sample (h) using a combination of *fast* (first) and *fine* etching processes. All the scale bars in (a–i) are 10 μm.

surface obtained in the present soft plasma etching processes can be largely attributed to the NF₃ radicals which etch both Mo and S uniformly to form volatile by-products. In addition, the domain sizes of the MoS₂ flakes were not noticeably affected throughout the whole process, as compared to the conventional thermal or laser-based methods which significantly reduce the surface area of MoS₂ domains^{21,22}.

Our plasma etching technique can also realize layer-by-layer thinning of MoS₂ films with the domains of different number of layers. This aspect is particularly important because of the uneven thickness and patchy domains of the commonly produced MoS₂ films. Figure 2a shows a representative etched sample consisting of sub-monolayer (SM), single, bi-, tri-, and multilayers, where the number of layers at distinct regions can be easily identified by both AFM and optical measurements (Fig. 2a–c). Notably, *SiO₂ in the figure represents the exposed SiO₂ surface after the complete removal of top MoS₂ layers. It was also found that no residues were left for any layers with above 3 layers. However, for the incomplete removal of monolayer MoS₂ that resulted in the SM area, a large number of MoS₂ residues were identified (Fig. 2b). Scattered MoS₂ residues were also found on 1-layer (1L) and 2-layer (2L) areas from Fig. 2b, although these residues were far less than that on the SM area. These observations suggested that the substrate had an impact in the etching process, particularly when the number of MoS₂ layers is less than 3 layers. This substrate effect may be caused by the interactions between long-range van der Waals (vdW) and short-range polar and molecular forces^{25,26}, which becomes negligible when the number of layers is large. Figure 2d shows the sample further subjected to a *fine* etching mode for another 7 min. Seven atomic layers were removed uniformly from the domains with more than 7 layers prior to etching, while only smooth

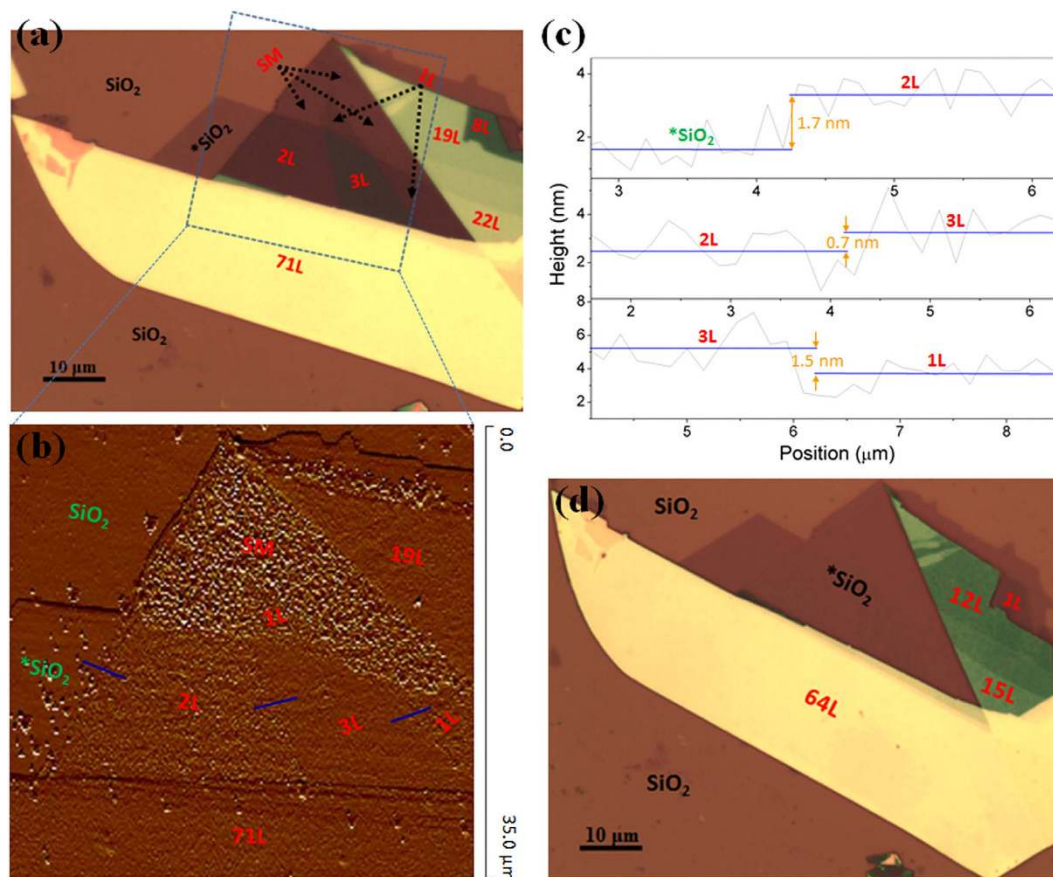


Figure 2. (a) The optical image of a representative etched sample consisting of flakes with the different number of layers including 1, 2, 3, 8, 19, 22, and 71 layers. SM denotes the MoS₂ residues due to incomplete removal of monolayer. (b) AFM image of the area squared in (a). (c) AFM depth profiles showing 1, 2 and 3 layers of MoS₂ as marked by blue lines in (b). (d) The optical image of the same sample after another 7 min *fine* etching showing that 7 MoS₂ layers have been removed uniformly from all domains starting with more than 7 layers.

SiO₂ on the surface were found for domains with less than 7 layers. These results corroborate the large-area, uniform etching of MoS₂ films using the soft plasma technique.

It should be noted that a height difference was observed between the bottom *SiO₂ (0 layer areas) and the surrounding SiO₂ (labeled as SiO₂ in both optical and AFM images) in Figs 1b,c and 2a,d. This is because the pristine samples used for soft etching had been subjected to SF₆+N₂+H₂ plasma treatment in the beginning at an input power density of 4 mW/cm³ for 20 min. Adding H₂ into the plasma can effectively promote the etching of SiO₂ and cleaning of substrate. About 20 nm SiO₂ was etched after this step while no etching of MoS₂ was found. The optical images of MoS₂ sample before and after SF₆+N₂+H₂ plasma treatment are shown in Figure S4. This observation demonstrated the easy control over selective etching between SiO₂ and MoS₂ by simply changing the plasma etchant gases. Details are out of scope of the present work but will be a subject of future studies.

The etched MoS₂ is further characterized by Raman spectroscopy, which is a sensitive technique to determine the number of layers in ultrathin MoS₂ flakes^{24,27–29}. According to previous studies^{24,28}, the peak positions of in-plane vibrational E_{2g}¹ mode and out-of-plane vibrational A_{1g} mode vary monotonously with the number of MoS₂ layers. Specifically, when the number of layers is decreased, A_{1g} shifts to lower frequencies due to weaker van der Waals forces which mostly affect the out-of-plane vibrations, while E_{2g}¹ shifts to higher frequencies owing to the decreased dielectric screening of the long-range Coulombic interactions^{22,30,31}. Figure 3a compares the Raman spectra obtained from different regions of an etched MoS₂ sample (displayed in Fig. 2a), including SM area and numbers of 1, 2, 3, 8, 19, 22 and 71 layers. It is found that the A_{1g} peak shifted slightly to lower frequencies as the number of layers decreased, especially in the 1–8 layers range, while the E_{2g}¹ peak shifted up to higher frequencies. These Raman shifts are consistent with the predicted characteristics of thinned MoS₂ layers.

Quantitatively, we also calculated the frequency difference between E_{2g}¹ and A_{1g} in the plasma etched samples, as shown in Fig. 3b. The frequency difference decreased gradually from about 25 cm⁻¹ for bulk film to 21.7 cm⁻¹ for monolayer MoS₂. This frequency shift is similar to the results obtained by other post-growth thinning methods, but the frequency difference value of the plasma-thinned monolayer or bilayer is larger than that of the pristine MoS₂ produced by mechanical exfoliation^{20,22,24,28}. Like other post-growth thinning methods such as laser thinning or chemical methods^{13,21}, this discrepancy can be attributed to the presence of a small number of traces of MoS₂ residues on the surface of the plasma-thinned monolayer or bilayer, as demonstrated in Fig. 2b. While

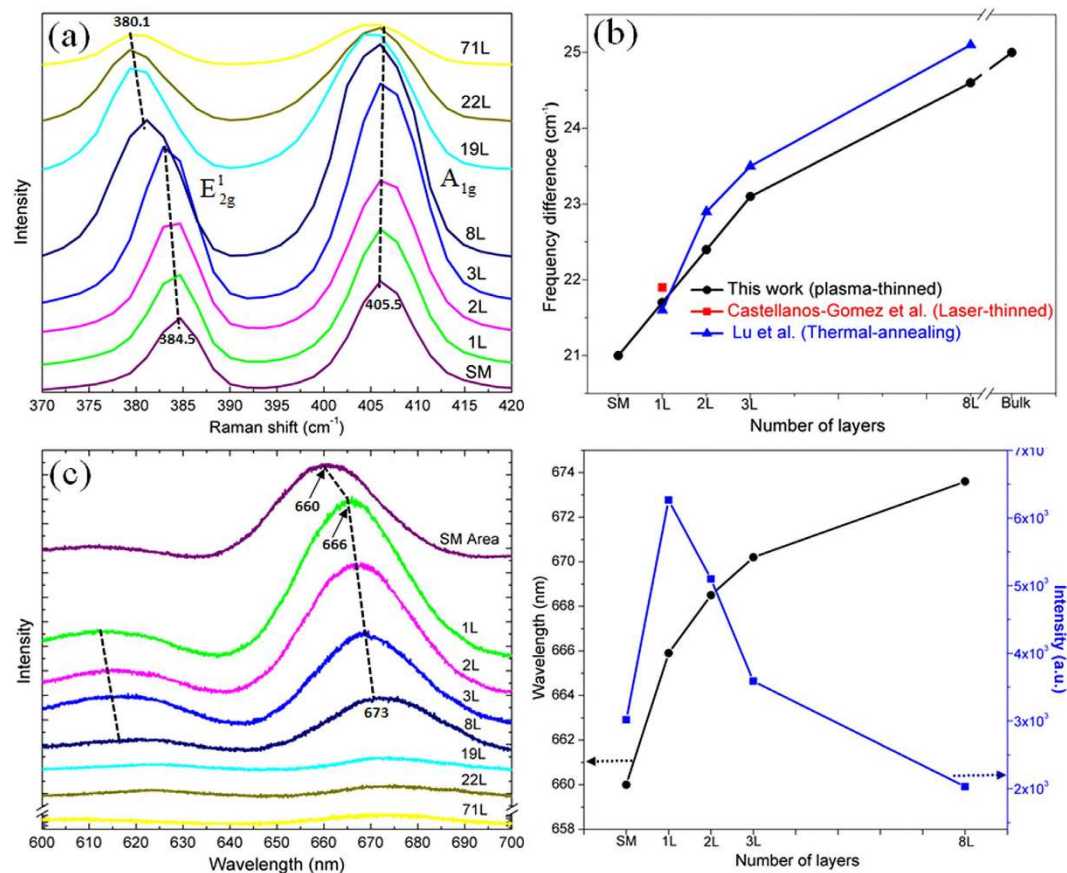


Figure 3. (a) Raman spectra of the etched MoS₂ flakes with different thickness and MoS₂ sub-monolayer. (b) The difference in Raman shifts between the E_{2g}¹ and A_{1g} peaks plotted as a function of the number of MoS₂ layers. MoS₂ layers thinned by the laser and thermal methods^{21,22} are also included for comparison. (c) PL spectra of the etched MoS₂ flakes with different thickness and MoS₂ sub-monolayer. (d) The wavelength and intensity of the prominent PL peak plotted as a function of the number of MoS₂ layers.

this made the present etching method slightly inferior to Ar⁺ plasma method²⁰ in thinning monolayer or bilayer, it is more efficient and credible in thinning thick layers. Interestingly, the frequency difference of SM area showed a further decrease as compared to that of monolayer MoS₂, possibly due to the quantum confinement effect arising from the reduced lateral size of MoS₂ residues.

In addition to the Raman vibrational fingerprints, ultrathin MoS₂ layers also exhibit unique signatures in their PL spectra. Figure 3c shows the PL spectra of MoS₂ with 1, 2, 3, 8, 19, 22 and 71 layers, as well as SM area. Two peaks located at ~670 and ~615 nm were identified in ultrathin layers (<8 layers), corresponding to the excitonic transitions between the minimum of the conduction band and the maxima of the two splitted valence bands, *i.e.*, the A₁ and B₁ excitons in MoS₂^{11,16}. In particular, the PL intensity increased gradually along with the decrease of the number of MoS₂ layers, in a good agreement with previous reports^{11,16}. Figure 3d shows the variations of peak position and intensity of the dominant PL peak as a function of the atomic layer numbers. As seen, the peak showed a blueshift from 673 to 666 nm as MoS₂ was thinned to monolayer, suggesting the increased bandgap energy. The monolayer produced the highest PL intensity, which can be ascribed to the indirect to direct bandgap transition^{8,16}. Interestingly, the SM area also exhibited a broad PL peak at ~660 nm with slightly reduced intensity as compared to that of monolayer MoS₂²⁰. Such observation may be attributed to the additional quantum confinement effect arising from the reduced lateral size and/or the distorted S-Mo-S sandwich structure in MoS₂ residues.

It was noted that our soft plasma etching regime of MoS₂ films was controlled by the input power density of the radio-frequency ICP source. The etching becomes harsh at a critical value of about 1.5 mW/cm². Below this critical input power density, the MoS₂ thinning process is dominated by radical reactions based on the SF₆+N₂ plasma chemistry. While at higher power densities, more high-energy electrons are generated, which enhances ion production and leads to uncontrollable damage to the samples through ion impact. In particular, at the input power density of 2 mW/cm² the roughness of etched MoS₂ samples increased significantly, with craters and pinholes clearly seen on the surface (Supplementary Figure S5).

To further explore the effect of the plasma electrons on the specific etching mode, we measured the electron energy distribution function (EEDF) using a Langmuir probe. The measurements were performed in a wide range of bias voltage, providing the I-V characteristic. The EEDF is generally defined as the second-order derivative of the measured I-V curve and can be expressed by the following equation^{32,33}:

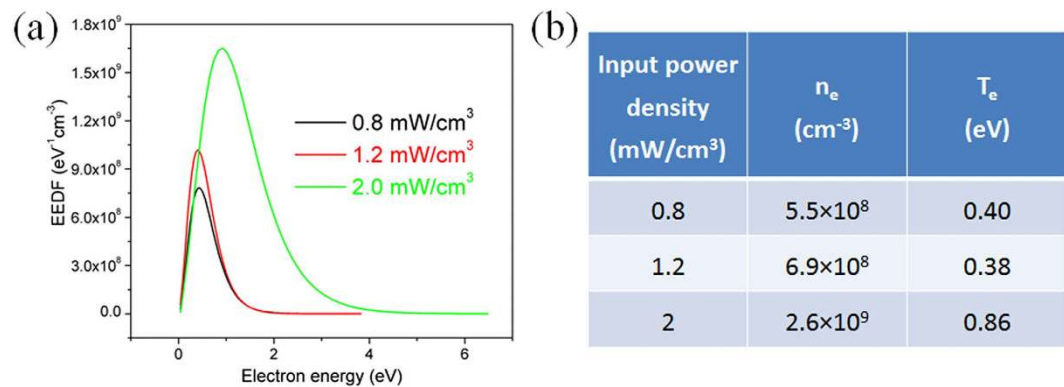


Figure 4. (a) The electron energy distribution function (EEDF) of the E-mode SF₆+N₂ discharges at three typical input power densities; (b) the corresponding calculated values of electron density (n_e) and electron temperature (T_e).

$$g_e(V) = \frac{2m_e}{e^2A} \left(\frac{2eV}{m_e} \right)^{1/2} \frac{d^2I_e}{dV^2} \quad (4)$$

where A is the surface collection area of the Langmuir probe, e is the electron charge, and m_e is electron mass. The electron density n_e and electron temperature T_e can be derived from the measured EEDF by assuming a Maxwellian distribution^{32,33}:

$$n_e = \int g_e(\varepsilon) d\varepsilon \quad (5)$$

$$T_e = \frac{2}{3n_e} \int \varepsilon g_e(\varepsilon) d\varepsilon \quad (6)$$

Figure 4a shows the experimental EEDF curves at the three input power densities of 0.8, 1.2 and 2 mW/cm³, corresponding to the fine and fast regimes of the soft etching mode, and the harsh etching mode, respectively. Figure 4b shows the values of n_e and T_e calculated based on Equations (5) and (6) respectively. It can be clearly seen that as long as the input power density was lower than the critical value of 1.5 mW/cm³, the electron temperature and density were below 0.4 eV and $\sim 10^8$ cm⁻³, respectively. These values were much lower compared to the conventional plasmas generated in the capacitively or inductively coupled configurations. As such, the fraction of electrons causing precursor ionization and dissociation was lower, so was the probability of ion-induced damage of underlying MoS₂ layers. Furthermore, negligible high-energy electron tail (> 2 eV) was observed in the EEDF, suggesting that electron-impact ionization and dissociation of the reactive radicals was not effective. Consequently, the effects of chemical reactions dominated over the ion bombardment in the MoS₂ thinning process.

When the power density exceeded the critical value, the energy and density of electrons and ions increased significantly which may start to cause structural damage to the sample. The presence of higher-energy electrons (> 2.0 eV) may also disrupt the S-Mo-S structure as the energy required to create vacancies in MoS₂ is as low as 2.12 eV³⁴. Another adverse effect is heat-induced evaporation, which may lead to surface defects at random positions. Nevertheless, as long as the input power density is lower than the critical value, the surface of multi-layer MoS₂ obtained by the soft plasma etching remains very smooth and the number of removed atomic layers is uniform across the whole sample surface. Furthermore, some encouraging results have been obtained on the atomic-layer soft etching of MoSe₂ thin films (Figure S6), suggesting that this technique is also applicable to other TMDs. The comparative advantages of the present soft plasma etching techniques over other existing techniques are also summarized in Supplementary Table S1.

In summary, we have demonstrated a versatile and effective plasma technique for the soft, selective, and uniform layer-by-layer etching of MoS₂ using SF₆ + N₂ precursors. Using the plasma power densities below the critical value of ~ 1.5 mW/cm³, equal number of MoS₂ atomic layers can be removed from all the MoS₂ areas irrespective of their original thickness. Large-area MoS₂ flakes with arbitrary number of layers can be effectively thinned by combining the fine and fast etching modes switched-over by the power density of the plasma discharge. The plasma chemistry is highly-selective to MoS₂ with negligible etching of the SiO₂/Si substrate. The surface of etched MoS₂ samples remained homogeneous and smooth with no shrinkage in the original domain sizes. The present approach is generic and may be used in the development of plasma-based etching processes of other TMDs pursued for applications in next-generation electronic, optoelectronic and other integrated devices and systems. It could also be promising for the catalytic applications of MoS₂ since plasma treatment is one of the important techniques for generating active sites in MoS₂ layers³⁵. Furthermore, our results may contribute to the development of soft plasma etching processes for large-scale semiconductor microfabrication technologies.

Methods

MoS₂ preparation and plasma thinning. Thick MoS₂ layers were fabricated by mechanical exfoliation from bulk single-crystal MoS₂ and deposited onto a Si/SiO₂ (300 nm) substrate. These multilayers were characterized by a combination of optical microscopy, atomic force microscopy (AFM) and Raman spectroscopy. A planar low-frequency (0.5 MHz) inductively-coupled plasma (ICP) source was applied to etch the MoS₂ multilayers without any external heating. The plasma was excited in the E-mode of ICP with the precursor gases of N₂ and SF₆ fed at the flow rates of 1.0 and 4.5 sccm, respectively. The E-mode discharge can be stably maintained at very low input power densities so that the ion density was too low to induce destructive ion bombardment onto the processed samples³⁶. Whether the plasma etching is *soft* or *harsh* depended strongly on the input power density, with a critical value of ~1.5 mW/cm³. Therefore, we adopted two low input power densities at 0.8 and 1.2 mW/cm³ for the *fine* and *fast* etching processes, respectively. The schematics of experimental setup of the E-mode ICP source is presented in Figure S1.

Characterization. The optical contrast images were obtained using a Leica 4200 Optical Microscopy. The Raman and photoluminescence (PL) spectra were recorded using a LabRAM HR Evolution Raman system with 532 nm laser excitation. The laser power at the sample was lower than 0.5 mW to avoid any laser-induced heating. To obtain the Raman images, an X-Y stage was used to move the sample with a 200 nm step, and the corresponding Raman spectrum was recorded at every point. AFM is carried out using a Bruker Dimension ICON system in the tapping mode.

References

- Berkdemir, A., Cutierrez, H. R., Betello-Mendez, A. R., Perea-Lopez, N., Elias, A. L., Chia, C. I., Wang, B., Crespi, V. H., Lopez-Urias, F. & Charlier, J. C. Identification of Individual and Few Layers of WS₂ Using Raman Spectroscopy. *Scientific Reports* **3**, 1755 (2013).
- McCreary, K. M., Hanbicki, A. T., Robinson, J. T., Cobas, E., Culbertson, J. C., Friedman, A. L., Jernigan, G. G. & Jonker, B. T. Large-Area Synthesis of Continuous and Uniform MoS₂ Monolayer Films on Graphene. *Advanced Functional Materials* **24**, 6449–6454 (2014).
- Huang, J. K., Pu, J., Hsu, C. L., Chiu, M. H., Juang, Z. Y., Chang, Y. H., Chang, W. H., Iwasa, Y., Takenobu, T. & Li, L. J. Large-Area Synthesis of Highly Crystalline WSe₂ Mono layers and Device Applications. *ACS Nano* **8**, 923–930 (2014).
- Xu, M. S., Liang, T., Shi, M. M. & Chen, H. Z. Graphene-Like Two-Dimensional Materials. *Chemical Reviews* **113**, 3766–3798 (2013).
- Lee, Y. H., Yu, L. L., Wang, H., Fang, W. J., Ling, X., Shi, Y. M., Lin, C. T., Huang, J. K., Chang, M. T., Chang, C. S., Dresselhaus, M., Palacios, T., Li, L. J. & Kong, J. Synthesis and Transfer of Single-Layer Transition Metal Disulfides on Diverse Surfaces. *Nano Letters* **13**, 1852–1857 (2013).
- Huang, X., Zeng, Z. Y. & Zhang, H. Metal Dichalcogenide Nanosheets: Preparation, Properties and Applications. *Chemical Society Reviews* **42**, 1934–1946 (2013).
- Chhowalla, M., Shin, H. S., Eda, G., Li, L. J., Loh, K. P. & Zhang, H. The Chemistry of Two-dimensional Layered Transition Metal Dichalcogenide Nanosheets. *Nature Chemistry* **5**, 263–275 (2013).
- Wang, Q. H., Kalantar-Zadeh, K., Kis, A., Coleman, J. N. & Strano, M. S. Electronics and Optoelectronics of Two-dimensional Transition Metal Dichalcogenides. *Nature Nanotechnology* **7**, 699–712 (2012).
- Lee, Y. H., Zhang, X. Q., Zhang, W. J., Chang, M. T., Lin, C. T., Chang, K. D., Yu, Y. C., Wang, J. T. W., Chang, C. S., Li, L. J. & Lin, T. W. Synthesis of Large-Area MoS₂ Atomic Layers with Chemical Vapor Deposition. *Advanced Materials* **24**, 2320–2325 (2012).
- Shi, Y. M., Huang, J. K., Jin, L. M., Hsu, Y. T., Yu, S. F., Li, L. J. & Yang, H. Y. Selective Decoration of Au Nanoparticles on Monolayer MoS₂ Single Crystals. *Scientific Reports* **3**, 1839 (2013).
- Mak, K. F., Lee, C., Hone, J., Shan, J. & Heinz, T. F. Atomically Thin MoS₂: A New Direct-Gap Semiconductor. *Physical Review Letters* **105**, 136805 (2010).
- Kuc, A., Zibouche, N. & Heine, T. Influence of Quantum Confinement on the Electronic Structure of the Transition Metal Sulfide TS₂. *Physical Review B* **83**, 245213 (2011).
- Eda, G., Yamaguchi, H., Voiry, D., Fujita, T., Chen, M. W. & Chhowalla, M. Photoluminescence from Chemically Exfoliated MoS₂. *Nano Letters* **11**, 5111–5116 (2011).
- Lin, Y. C., Dumcencou, D. O., Huang, Y. S. & Suenaga, K. Atomic Mechanism of the Semiconducting-to-Metallic Phase Transition in Single-layered MoS₂. *Nature Nanotechnology* **9**, 391–396 (2014).
- Yamaguchi, H., Blancon, J. C., Kappera, R., Lei, S. D., Najmaei, S., Mangum, B. D., Gupta, G., Ajayan, P. M., Lou, J., Chhowalla, M., Crochet, J. J. & Mohite, A. D. Spatially Resolved Photoexcited Charge-Carrier Dynamics in Phase-Engineered Mono Layer MoS₂. *ACS Nano* **9**, 840–849 (2015).
- Splendiani, A., Sun, L., Zhang, Y. B., Li, T. S., Kim, J., Chim, C. Y., Galli, G. & Wang, F. Emerging Photoluminescence in Monolayer MoS₂. *Nano Letters* **10**, 1271–1275 (2010).
- Tonndorf, P., Schmidt, R., Bottger, P., Zhang, X., Borner, J., Liebig, A., Albrecht, M., Kloc, C., Gordan, O., Zahn, D. R. T., de Vasconcellos, S. M. & Bratschitsch, R. Photoluminescence Emission and Raman Response of Monolayer MoS₂, MoSe₂, and WSe₂. *Optics Express* **21**, 4908–4916 (2013).
- Kibsgaard, J., Chen, Z. B., Reinecke, B. N. & Jaramillo, T. F. Engineering the Surface Structure of MoS₂ to Preferentially Expose Active Edge Sites for Electrocatalysis. *Nature Materials* **11**, 963–969 (2012).
- Xu, X., Liu, W., Kim, Y. & Cho, J. Nanostructured Transition Metal Sulfides for Lithium Ion Batteries: Progress and Challenges. *Nano Today* **9**, 604–630 (2014).
- Liu, Y. L., Nan, H. Y., Wu, X., Pan, W., Wang, W. H., Bai, J., Zhao, W. W., Sun, L. T., Wang, X. R. & Ni, Z. H. Layer-by-Layer Thinning of MoS₂ by Plasma. *ACS Nano* **7**, 4202–4209 (2013).
- Castellanos-Gomez, A., Barkelid, M., Goossens, A. M., Calado, V. E., van der Zant, H. S. J. & Steele, G. A. Laser-Thinning of MoS₂: On Demand Generation of a Single-Layer Semiconductor. *Nano Letters* **12**, 3187–3192 (2012).
- Lu, X., Utama, M. I. B., Zhang, J., Zhao, Y. Y. & Xiong, Q. H. Layer-by-layer Thinning of MoS₂ by Thermal Annealing. *Nanoscale* **5**, 8904–8908 (2013).
- Wu, J., Li, H., Yin, Z. Y., Li, H., Liu, J. Q., Cao, X. H., Zhang, Q. & Zhang, H. Layer Thinning and Etching of Mechanically Exfoliated MoS₂ Nanosheets by Thermal Annealing in Air. *Small* **9**, 3314–3319 (2013).
- Zhang, L. L., Liang, G., Peng, G., Zou, F., Huang, Y. H., Croft, M. C. & Ignatov, A. Significantly Improved Electrochemical Performance in Li₃V₂(PO₄)₃/C Promoted by SiO₂ Coating for Lithium-Ion Batteries. *Journal of Physical Chemistry C* **116**, 12401–12408 (2012).
- Seemann, R., Herminghaus, S. & Jacobs, K. Dewetting Patterns and Molecular Forces: A Reconciliation. *Physical Review Letters* **86**, 5534–5537 (2001).
- Han, Z. J. & Tay, B. K. Dewetting of polymer films by ion implantation. *European Physical Journal E* **28**, 273–278 (2009).

27. Li, S. L., Miyazaki, H., Song, H., Kuramochi, H., Nakaharai, S. & Tsukagoshi, K. Quantitative Raman Spectrum and Reliable Thickness Identification for Atomic Layers on Insulating Substrates. *Acs Nano* **6**, 7381–7388 (2012).
28. Lee, C., Yan, H., Brus, L. E., Heinz, T. F., Hone, J. & Ryu, S. Anomalous Lattice Vibrations of Single- and Few-Layer MoS₂. *Acs Nano* **4**, 2695–2700 (2010).
29. Ferrari, A. C., Meyer, J. C., Scardaci, V., Casiraghi, C., Lazzeri, M., Mauri, F., Piscanec, S., Jiang, D., Novoselov, K. S., Roth, S. & Geim, A. K. Raman Spectrum of Graphene and Graphene Layers. *Physical Review Letters* **97**, 187401 (2006).
30. Molina-Sanchez, A. & Wirtz, L. Phonons in Single-layer and Few-layer MoS₂ and WS₂. *Physical Review B* **84**, 155413 (2011).
31. Li, H., Zhang, Q., Yap, C. C. R., Tay, B. K., Edwin, T. H. T. & Olivier, A. & Baillargeat, D. From Bulk to Monolayer MoS₂: Evolution of Raman Scattering. *Advanced Functional Materials* **22**, 1385–1390 (2012).
32. Godyak, V. A., Piejak, R. B., Alexandrovich, B. M. & Kolobov, V. I. Hot Plasma and Nonlinear Effects in Inductive Discharges. *Physics of Plasmas* **6**, 1804–1812 (1999).
33. Lieberman, M. A. & Lichtenberg, A. J. Principles of Plasma Discharges and Materials Processing. 2nd ed., Wiley-Interscience: Hoboken, N.J., 2005.
34. Hong, J. H., Hu, Z. X., Probert, M., Li, K., Lv, D. H., Yang, X. N., Gu, L., Mao, N. N., Feng, Q. L., Xie, L. M., Zhang, J., Wu, D. Z., Zhang, Z. Y., Jin, C. H., Ji, W., Zhang, X. X., Yuan, J. & Zhang, Z. Exploring Atomic Defects in Molybdenum Disulphide Monolayers. *Nature Communications* **6**, 6293 (2015).
35. Li, H., Tsai, C., Koh, A. L., Cai, L., Contryman, A. W., Fragapane, A. H., Zhao, J., Han, H. S., Manoharan, H. C., Abild-Pedersen, F., Norskov, J. K. & Zheng, X. Activating and Optimizing MoS₂ Basal Planes for Hydrogen Evolution Through the Formation of Strained Sulphur Vacancies. *Nature Materials* doi: 10.1038/nmat4465 (2015).
36. Xiao, S. Q., Xu, S. & Ostrikov, K. Low-temperature Plasma Processing for Si Photovoltaics. *Materials Science & Engineering R-Reports* **78**, 1–29 (2014).

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Author Contributions

P.X. and S.X. initiated the research and worked on plasma treatment, Raman and photoluminescence measurements and analysis of MoS₂ properties. X.Z. and D.Y. performed optical microscopy measurements. F.Q. performed AFM measurements. X.G., Z.N. and K.O. advised on planning and executing the research. All authors discussed the results. S.X., Z.J.H. and K.O. wrote the manuscript.

Additional Information

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