

Basis set effects on the hyperpolarizability of CHCl_3 : Gaussian-type orbitals, numerical basis sets and real-space grids

Fernando D. Vila,¹ David A. Strubbe,² Yoshinari Takimoto,^{1,3} Xavier Andrade,⁴ Angel Rubio,^{4,5} Steven G. Louie,² and John J. Rehr^{1,a)}

¹Department of Physics, University of Washington, Seattle, Washington 98195-1560, USA

²Department of Physics, University of California, Berkeley, California, USA and Materials Sciences Division, Lawrence Berkeley National Laboratory, California 94720, USA

³Institute for Solid State Physics, University of Tokyo, Kashiwa, Chiba 277-8581, Japan

⁴Nano-Bio Spectroscopy group and ETSF Scientific Development Centre, Dpto. Física de Materiales, Universidad del País Vasco, Centro de Física de Materiales CSIC-UPV/EHU-MPC, Spain and DIPC, Av. Tolosa 72, E-20018 San Sebastián, Spain

⁵Fritz-Haber-Institut der Max-Planck-Gesellschaft, Berlin D-14195, Germany

(Received 27 October 2009; accepted 1 June 2010; published online 19 July 2010)

Calculations of the hyperpolarizability are typically much more difficult to converge with basis set size than the linear polarizability. In order to understand these convergence issues and hence obtain accurate *ab initio* values, we compare calculations of the static hyperpolarizability of the gas-phase chloroform molecule (CHCl_3) using three different kinds of basis sets: Gaussian-type orbitals, numerical basis sets, and real-space grids. Although all of these methods can yield similar results, surprisingly large, diffuse basis sets are needed to achieve convergence to comparable values. These results are interpreted in terms of local polarizability and hyperpolarizability densities. We find that the hyperpolarizability is very sensitive to the molecular structure, and we also assess the significance of vibrational contributions and frequency dispersion. © 2010 American Institute of Physics. [doi:10.1063/1.3457362]

I. INTRODUCTION

Chloroform (CHCl_3) is a widely used solvent in measurements of nonlinear optical properties of organic chromophores, using techniques such as electric-field-induced second-harmonic generation (EFISH) and hyper-Rayleigh scattering (HRS).^{1,2} It is sometimes also used as an internal reference.³ However, assumptions have to be made to extract molecular hyperpolarizabilities from these measurements, in particular from EFISH which actually measures a third-order response function. For calibration purposes, either absolute measurements or *ab initio* calculations are needed to convert between the different combinations of tensor components of the hyperpolarizability measured in the EFISH and HRS experiments. However, very early calculations, which have been recognized as unsatisfactory by their authors,⁴ have heretofore been used for such conversions.¹ Consequently there is a need for better understanding of the convergence issues and for more accurate *ab initio* calculations. Toward this end we have carried out a systematic study of the second-order hyperpolarizability β of chloroform using several theoretical methods. In an effort to obtain high-quality results which improve on earlier calculations, we have based our calculations on an accurate experimental structure and also considered both frequency-dependence and vibrational contributions, effects which are typically neglected in other calculations.

Although our presentation is restricted to a single system, the methodology is more general and hence many of our

results will likely be applicable to many other cases. Likewise, the quantitative comparison of several theoretical methods is both novel and of general interest. Finally we believe our interpretation of the linear and nonlinear response in terms of local polarizability densities provides a useful way of understanding the local contributions to the polarizability from various parts of a molecule.

Chloroform is of particular theoretical interest because its hyperpolarizability is challenging to measure experimentally due to its small magnitude compared to typical experimental errors, and hence the available measurements have both positive and negative values, with large relative error bars.^{1,5} Similarly, this nonlinear property has proved to be quite difficult to calculate theoretically, as the results exhibit a large dependence on the quality of the basis set used, both for density functional theory (DFT) and coupled-cluster methods,^{6,7} as well as the molecular geometry. One of the main purposes of this paper is to investigate the reasons for these difficulties using three different basis set approaches: (i) Gaussian-type orbitals (GTOs), (ii) numerical basis sets (NBSs), and (iii) real-space grids, with a common treatment of exchange and correlation given by the Perdew–Burke–Ernzerhof⁸ (PBE) functional. The importance of different aspects of the basis sets (diffuseness, polarization, etc.) was studied systematically by changing the number of GTOs, the cutoff radii of the NBSs, and the extent and density of the real-space grids. In order to interpret these results, we also studied the spatial distribution of the dielectric properties using the concepts of polarizability- and hyperpolarizability-densities, as well as first- and second-

^{a)}Electronic addresses: jjr@uw.edu and jjr@phys.washington.edu.

order electric-field-perturbed densities.^{9,10} In addition, we also investigated the dependence of polarization properties on the choice of exchange-correlation (XC) functionals and the correlation level by means of DFT, Hartree Fock (HF), second-order Møller-Plesset (MP2), and coupled clusters with single and double excitations (CCSD) methods. We also briefly discuss the dependence of the results on the molecular geometry, which was found to have a significant influence on the calculated hyperpolarizability.

Unless noted explicitly, the experimental molecular geometry of Colmont *et al.*¹¹ was used throughout this work: $r_{\text{CH}}=1.080$ Å, $r_{\text{CCl}}=1.760$ Å, and $\angle\text{HCCl}=108.23^\circ$. The molecule was located with its center of mass at the origin, and oriented with the CH bond along the positive z -direction and one HCCl angle in the yz plane. Since chloroform has C_{3v} point-group symmetry, the following symmetry relations hold for the linear polarizability α and hyperpolarizability β : $\alpha_{xx}=\alpha_{yy}$, $\beta_{xxy}=-\beta_{yyy}$, and $\beta_{xxz}=\beta_{yyz}$. In the static case Kleinman symmetry¹² also applies. Thus the α_{yy} , α_{zz} , β_{yyy} , β_{yyz} , and β_{zzz} components fully describe the polarizability and hyperpolarizability tensors; all other permutations of the indices are equivalent. In the dynamic case at nonzero frequency, however, the components of $\beta_{ijk}(-2\omega; \omega, \omega)$ are not all equivalent: $\beta_{yyz}=\beta_{zyy} \neq \beta_{zyy}$. Here we use the Taylor convention for hyperpolarizabilities.¹³

From our calculated tensor components we can calculate the isotropically averaged polarizability $\bar{\alpha}=\frac{1}{3}\sum_i\alpha_{ii}$ and the EFISH hyperpolarizability in the direction of the dipole moment $\beta_{\parallel i}=\frac{1}{5}\sum_j(\beta_{ijj}+\beta_{jij}+\beta_{jji})$. In the C_{3v} point group, these relations reduce to $\bar{\alpha}=\frac{1}{3}(2\alpha_{yy}+\alpha_{zz})$, $\beta_{\parallel}=\beta_{\parallel z}=\frac{3}{5}(2\beta_{yyz}+\beta_{zzz})$. We also calculate the hyperpolarizability for HRS in the vertical/vertical (VV) polarization, as given by Cyvin *et al.*¹⁴ for the static case (where Kleinman symmetry holds) and the generalization of Bersohn *et al.*¹⁵ for the dynamic case. In the static case for C_{3v} symmetry

$$[\beta_{\text{HRS}}^{\text{VV}}]^2 = \frac{8}{35}\beta_{yyz}^2 + \frac{1}{7}\beta_{zzz}^2 + \frac{24}{35}\beta_{yyz}^2 + \frac{12}{35}\beta_{yyz}\beta_{zzz}. \quad (1)$$

This quantity has only been measured for liquid chloroform;¹ measurements are not available for the gas phase.

Our best results for each method generally exhibit a consistent agreement among themselves and lend confidence to the overall quality of our calculations compared to earlier work. Achieving this consistency points to the need for a comprehensive and well balanced description of all regions of the system: namely, the outlying regions of the molecule, the short C–H bond, and the Cl atoms. A key finding is that the local contributions to the β_{zzz} response of the Cl atoms and the C–H bond are of opposite sign and tend to cancel, thus explaining the relatively slow convergence of this component with respect to the basis set size. This behavior, together with the near cancellation of the β_{yyz} and β_{zzz} components, leads to the relatively small value of β_{\parallel} of chloroform. By contrast, the HRS hyperpolarizability converges much more quickly since it is an incoherent process which is mostly given by a sum of squares of tensor components that do not cancel.

II. METHODS

A. GTOs

The GTO polarization properties were calculated using finite-field perturbation theory (FFPT). The electric-field strengths \mathcal{E} used were 0.00, ± 0.01 , and ± 0.02 a.u. The different components of the induced dipole moment were fit to a fourth-order polynomial to obtain the polarizability and hyperpolarizability tensors. Using analytic derivatives available at the Hartree–Fock level, we find that the properties obtained with the FFPT method agree to 0.1% or better. The effects of the basis set size and diffuse quality were studied using Dunning’s double-through quintuple- ζ correlation-consistent sets,^{16,17} with and without augmentation exponents, and with additional diffuse exponents. We also used Sadlej’s HyPol basis set,¹⁸ which is specifically designed for the calculation of nonlinear response properties. In this paper, for simplicity, these basis sets will be referred to as aVDZ and VDZ (for aug-cc-pVDZ and cc-pVDZ, respectively), aVTZ and VTZ (for aug-cc-pVTZ and cc-pVTZ), etc. The HyPol set will be abbreviated as HP. The basis set labeled aV5Zs corresponds to the aV5Z set, where the g and h functions were removed from the C and Cl atoms and the f and g were removed from the H atom. The d-aV5Zs basis set corresponds to the aV5Zs set augmented with (0.014184, 0.009792, and 0.025236) and (0.017244, 0.012528, and 0.036108) (s, p, d) exponents on the C and Cl atoms, respectively, and (0.004968 and 0.026784) (s, p) exponents on the H atom.

Throughout this paper, unless otherwise specified, we use Hartree atomic units $e=\hbar=m=1$ with distances in bohr ($a_0 \approx 0.529$ Å) and energies in hartrees ≈ 27.2 eV. The effect of electron correlation on these all-electron calculations was studied with the local-density approximation (LDA),¹⁹ PBE,⁸ B3LYP,^{20,21} BMK,²² and CAM-B3LYP (Ref. 23) XC density functionals, as well as with the HF, MP2, and CCSD methods. All GTO calculations were performed with GAUSSIAN 03 (Ref. 24) except the CAM-B3LYP ones which used GAUSSIAN 09.²⁵

B. Real-space grids

For the real-space grid calculations we used *ab initio* DFT with a real-space basis, as implemented in OCTOPUS.^{26,27} The polarizability and hyperpolarizability were calculated by linear response via the Sternheimer equation and the $2n+1$ theorem.²⁸ This approach, also known as density-functional perturbation theory, avoids the need for sums over unoccupied states. The PBE XC functional was used for the ground state, and the adiabatic LDA kernel was used for the linear response. All calculations used Troullier–Martins norm-conserving pseudopotentials.²⁹

The molecule was studied as a finite system, with zero boundary conditions for the wave function on a large sphere surrounding the molecule, as described below. Convergence was tested with respect to the real-space grid spacing and the radius of the spherical domain. The grid spacing required is determined largely by the pseudopotential, as it governs the fineness with which spatial variations of the wave functions can be described as well as the accuracy of the finite-

TABLE I. Parameters used in the definition of the NBSs: energy shifts $\delta\epsilon$ and cutoff radii $r_c^l(X)$ of the first- ζ for angular momentum l and atom X . All values are in atomic units.

Parameter	DZP	QZTPe4	5Z4Pe5	5Z4Pe6	5Z4Pe7	5Z4Pe8
$\delta\epsilon$	10^{-2}	10^{-4}	10^{-5}	10^{-6}	10^{-7}	10^{-8}
$r_c^s(\text{C})$	4.088	6.574	7.832	8.875	10.056	11.114
$r_c^p(\text{C})$	4.870	8.655	10.572	12.283	14.271	15.772
$r_c^s(\text{H})$	4.709	8.164	9.972	11.586	13.461	14.877
$r_c^s(\text{Cl})$	3.826	5.852	6.799	7.704	8.730	9.410
$r_c^p(\text{Cl})$	4.673	7.514	8.951	10.400	11.784	13.024

difference evaluation of the kinetic-energy operator. The spacing λ can be converted to an equivalent plane-wave cutoff via $E_c = (\hbar^2/2m)(2\pi/\lambda)^2$, where E_c is the cutoff energy for both the charge density and wave functions. The sphere radius determines the maximum spatial extent of the wave functions.

With tight numerical tolerances in solving the Kohn-Sham and Sternheimer equations, we can achieve a precision of 0.01 a.u. or better in the converged values of the tensor components of β . We also did two additional kinds of calculations. For comparison to the nonlinear experiments, which used incoming photons of wavelength 1064 nm (energy of 1.165 eV), we also performed dynamical calculations at this frequency via time-dependent DFT (TDDFT). To compare directly to the results from finite-field perturbation theory with the other basis sets, we also calculated the dielectric properties via finite differences using electric-field strengths of ± 0.01 and ± 0.015 a.u.

C. NBSs

The NBS calculations were performed with the SIESTA (Refs. 30 and 31) code and used Troullier–Martins norm-conserving pseudopotentials.²⁹ As in the GTO calculations described in Sec. I, the polarization properties were obtained using FFPT with electric-field strengths of 0.00, ± 0.01 , and ± 0.02 a.u. The NBSs use a generic linear combination of numerical atomic orbitals that are forced to be zero beyond a cutoff radius r_c .³² Rather than using a fixed r_c for all atoms, a common confinement-energy shift is usually enforced, resulting in well-balanced basis sets.³¹ In general, multiple radial functions with the same angular dependence are introduced to enhance the flexibility of the basis set. This results in a so-called multiple- ζ scheme similar to the standard split-valence approach used for GTOs.^{33,34} Polarization functions can also be added using the approach described in Ref. 31. Typically, double- ζ sets with a single polarization function (DZP) are sufficient in linear-response calculations. However, we found that a DZP set is insufficient for nonlinear properties. Instead of performing an optimization of the NBSs, we improved their flexibility by adding ζ -functions, and controlling their splitting.³¹ By varying the “norm-splitting” parameter in SIESTA, we can control the flexibility in different regions of the radial functions, resulting in the cutoff radii and energy shifts shown in Tables I and II. Finally, the NBS calculations used a common $(37.79 \text{ Bohr})^3$ cell and real-space grid with a plane-wave-equivalent cutoff

of 250 Ry for the calculation of the Hartree and XC potentials. This corresponds to a real-space mesh spacing of about 0.189 Bohr.

D. Linear and nonlinear response densities

The origin of the slow convergence of the hyperpolarizability with respect to the quality of the basis set is difficult to understand by studying only the total quantities. A more informative analysis can be obtained from the spatial distribution of the dielectric properties. Thus, we have calculated the linear and nonlinear response densities, as well as their associated properties. Here we will focus on the response densities induced by an electric field in the z -direction. The first-order density is defined as

$$\rho_z^{(1)}(r) = \frac{\partial \rho}{\partial \mathcal{E}_z} \quad (2)$$

and the linear polarizability $\alpha_{zz}(r)$ as

$$\alpha_{zz}(r) = \rho_z^{(1)}(r)z. \quad (3)$$

The second-order response density and associated hyperpolarizability are defined similarly

$$\rho_{zz}^{(2)}(r) = \frac{\partial^2 \rho}{\partial \mathcal{E}_z^2}, \quad (4)$$

$$\beta_{zzz}(r) = \rho_{zz}^{(2)}(r)z. \quad (5)$$

These response densities are all calculated using finite differences. For the real-space grids, our Sternheimer approach provides only the linear response density and polarizability density, but not the nonlinear response and hyperpolarizability densities.

Unlike the total properties, the spatial distributions of polarizabilities and hyperpolarizabilities as defined above depend on the origin of coordinates. Throughout this work we

TABLE II. Parameters used in the definition of the NBSs: splitting radii $r_c^l(X)$ associated with a given norm splitting for angular momentum l and atom X . All values are in atomic units.

Norm splitting	0.0015	0.0150	0.1500	0.6000
$r_c^s(\text{C})$	6.574	5.120	3.519	2.272
$r_c^p(\text{C})$	8.548	6.332	3.841	2.005
$r_c^s(\text{H})$	8.690	6.600	4.155	2.223
$r_c^s(\text{Cl})$	5.707	4.557	3.252	2.292
$r_c^p(\text{Cl})$	7.328	5.566	3.639	2.235

TABLE III. Structural parameters used in the study of the variation of the dielectric properties of CHCl_3 with structure. PBE/aVQZ and PBE/RS refer to PBE-optimized structures using the aVQZ GTO in Gaussian and a real-space grid in OCTOPUS, respectively. Bond lengths are in Angstroms and angles in degrees. The experimental structure from Ref. 11 was used for all our subsequent calculations.

Source	$r(\text{C-H})$	$r(\text{C-Cl})$	$\angle\text{HCCl}$
Expt. ^a	1.100 ± 0.004	1.758 ± 0.001	107.6 ± 0.2
Expt. ^b	1.080 ± 0.002	1.760 ± 0.002	108.23 ± 0.02
PBE/aVQZ ^c	1.090	1.779	107.7
PBE/RS	1.084	1.769	107.6

^aReference 35.

^bReference 11.

^cReference 6.

chose a center-of-mass reference for the spatial distributions. To understand the role that different regions of the molecule play in the total properties, we have devised a partitioning scheme for the spatial distribution corresponding to the spaces occupied roughly by the Cl atoms and C–H bond. That is, we divide space into two regions by constructing three planes, each orthogonal to one of the C–Cl bonds, and passing through a point located 40% along the C–Cl bond from the C atom, which corresponds approximately to the density minimum along the C–Cl bond. The first region (“CH”) consists of all the space above the three planes, and contains the C–H bond, while the second (“Cl”) covers the remainder of the space, including the three Cl atoms. We have integrated the various densities in each region numerically to find its contribution to the total.

III. RESULTS AND DISCUSSION

A. Structure

Although all the results presented in later sections were obtained for the experimental geometry determined by Colmont *et al.*,¹¹ here we briefly discuss the effect of the structural parameters on the dielectric properties. Experimental structures have been obtained from spectroscopy of rotational transitions in isotopically substituted chloroform molecules, in 1962 by Jen and Lide³⁵ and in 1998 by Colmont *et al.* The newer experimental work used more sophisticated

methods than the earlier experiment, and has more precise results, so we consider it the more reliable experimental result. (Best-fit structure of Colmont *et al.*¹¹ was slightly modified from the purely experimental result by inclusion of data from *ab initio* MP2 aug-cc-PVTZ calculations.) We compared these experimental structures with theoretical structures optimized using the PBE functional, one obtained with the aVQZ basis in GAUSSIAN 03,⁶ and the other with a real-space grid in OCTOPUS. The parameters for each structure are listed in Table III. The linear and nonlinear properties for each structure were calculated with the Sternheimer method in OCTOPUS, using a radius of $17 a_0$ and a spacing of $0.25 a_0$ and the results are summarized in Table IV. Our calculations show that the dipole moment and polarizability are not affected much, but the hyperpolarizability varies significantly with structure. Individual tensor components of the hyperpolarizability do not change by more than $\sim 30\%$, but since $\beta_{||}$ is a sum of large positive and negative components, it can change sign, and change by orders of magnitude depending on the structure. Therefore use of an accurate experimental structure is crucial for reliable results, as the effect of structure can be as large as those from basis set or theoretical method. Unfortunately previous work has often not mentioned the structure used, complicating comparison of results from different methods.⁷ The experimental error bar on the gas-phase experimental value of $\beta_{||} = 1 \pm 4$ a.u. (Ref. 1) is so large relative to the value that the results for all the structures are consistent with the experiment, despite their significant variation. We used the experimental structure from Ref. 11 for all subsequent calculations in this work.

B. GTOs

Table V shows the effect of the basis set size on the dielectric properties calculated with the PBE functional and GTOs. The results from this work agree well with those reported by Davidson *et al.*⁶ In the case of the dipole moment, we also find agreement to within 1% of the experimental value of 0.409 ± 0.008 a.u. (Ref. 36) for every basis set except VDZ. (The quality of the correlation-consistent basis sets roughly increases down the table, with the HP basis set giving results that are roughly equivalent to those obtained with the best correlation-consistent set, i.e., d-aV5Zs.) The

TABLE IV. Dielectric properties of various structures for CHCl_3 described in Table III, as calculated by DFT on a real-space grid with radius $17 a_0$ and spacing $0.25 a_0$, compared with the experimental values of the dipole moment and the electronic contribution to the polarizability. PBE/aVQZ and PBE/RS refer to the structures described in Table III. All values are in atomic units (a.u.).

Structure	μ_z	α_{yy}	α_{zz}	β_{yyy}	β_{yyz}	β_{zzz}	$\bar{\alpha}$	$\beta_{ }$	$\beta_{\text{HRS}}^{\text{VV}}$
1962 Expt. Struct. ^a	0.395	66.14	47.22	27.09	−14.41	28.47	59.83	−0.21	16.89
1998 Expt. Struct. ^b	0.399	66.02	47.00	27.12	−16.36	26.92	59.68	−3.49	17.44
PBE/aVQZ ^c	0.401	67.17	47.35	27.23	−14.11	27.76	60.57	−0.27	16.79
PBE/RS	0.397	66.66	47.12	27.29	−14.26	28.92	60.15	0.24	16.96
Expt.	0.409 ± 0.008^d	61 ± 5^e	45 ± 3^e				56 ± 4^e	1 ± 4^f	

^aReference 35.

^bReference 11.

^cReference 6.

^dReference 36.

^eReference 37.

^fReference 1.

TABLE V. Effect of the GTO basis set quality on the components of the dielectric properties of CHCl_3 calculated with the PBE functional. All values are in atomic units.

Basis Set	μ_z	α_{yy}	α_{zz}	β_{yyy}	β_{yyz}	β_{zzz}	$\bar{\alpha}$	β_{\parallel}	$\beta_{\text{HRS}}^{\text{VV}}$
VDZ	0.426	46.36	25.82	19.64	-8.43	-44.25	39.51	-36.66	23.33
VTZ	0.417	54.57	34.08	0.52	-3.35	-37.20	47.74	-26.33	15.75
VQZ	0.407	59.92	39.76	-5.83	-2.88	-25.57	53.20	-18.80	10.90
V5Z	0.405	61.65	41.93	-6.12	-5.70	-25.85	55.08	-22.35	12.64
aVDZ	0.412	63.29	44.35	8.53	-11.35	2.62	56.98	-12.05	9.78
aVTZ	0.408	64.91	46.03	14.67	-11.81	11.72	58.62	-7.14	10.82
aVQZ	0.406	65.50	46.61	21.15	-13.81	19.52	59.21	-4.86	13.96
aV5Z	0.404	65.69	46.71	22.02	-14.26	18.65	59.36	-5.92	14.45
aV5Zs	0.404	65.67	46.70	21.78	-14.01	17.92	59.34	-6.07	14.24
d-aV5Zs	0.404	65.70	46.79	27.35	-15.31	22.27	59.40	-5.01	16.90
HP	0.405	65.51	46.60	27.64	-15.54	22.93	59.21	-4.89	17.12
Expt.	$0.409 \pm 0.008^{\text{a}}$	$61 \pm 5^{\text{b}}$	$45 \pm 3^{\text{b}}$				$56 \pm 4^{\text{b}}$	$1 \pm 4^{\text{c}}$	

^aReference 36.^bReference 37.^cReference 1.

polarizabilities agree well with the experimentally measured values at 546 nm (2.27 eV);³⁷ these quantities are optical polarizabilities which contain only an electronic contribution and have minimal dispersion. Indeed, our real-space TDDFT calculations (Sec. III C) at 532 nm (2.24 eV) give $\alpha_{yy} = 68.827$ a.u. and $\alpha_{zz} = 48.405$ a.u., a small increase from the static and 1064 nm results, and basically consistent with the experimental values. We can also compare to a measurement of the static isotropically averaged polarizability of 64 ± 3 a.u. (Ref. 38). To compare with our result for the average electronic polarizability, we subtract the estimated vibrational contribution of 4.5 a.u. calculated from experimental data (no error bar provided),³⁹ yielding 60 a.u., which agrees with the predicted values within 0.4%. To verify this comparison, we have also computed the vibrational component of the polarizability from first principles, obtaining a value of 6.3 a.u., in reasonable agreement with the above estimate. This value was obtained by first doing a standard GAUSSIAN 03 (Ref. 24) calculation of vibrational frequencies at the PBE/aug-cc-pVTZ level, using a molecular structure optimized at the same level. The polarizability is then calculated as described in Ref. 40.

For the hyperpolarizability, Davidson *et al.*⁶ also obtain good agreement between theory and experiment provided the values of β_{\parallel} calculated with the aVDZ, aVTZ, and aVQZ correlation-consistent basis sets are smoothly extrapolated to the complete basis set limit.⁴¹ If the same extrapolation scheme is applied to our results for β_{\parallel} given in Table V, however, we obtain a value of -2.89 a.u., which is barely within the error of the experimental value of 1 ± 4 a.u. This extrapolated value also differs from the value of 0.35 a.u. obtained by Davidson *et al.*⁶ The difference can be attributed to the different geometries used: in this work we used the experimentally determined structure, while Davidson *et al.*⁶ used the theoretical structures obtained with the aVDZ, aVTZ, and aVQZ basis sets. When we use the aVTZ-optimized geometry in our calculations instead of the experimental one, we obtain an extrapolated value of 0.6 a.u., which is consistent with the Davidson *et al.*⁶ value. Our calculations also included the aV5Z basis set, the next basis

set in the correlation-consistent series. When this set is included in the extrapolation, our predicted value is lowered to -5.29 a.u. The sudden reduction of the estimated complete basis set value upon inclusion of the aV5Z set indicates that the convergence of β_{\parallel} is not smoothly monotonic. Therefore the basis set sequence aVDZ-aV5Z is not well adapted to the extrapolation of this property, a fact that has been noticed with other methods.^{42,43} This stems from the fact that although the individual components β rise in a reasonably monotonic way, small deviations in their progression and their near cancellation lead to sudden changes in β_{\parallel} .

To better understand the convergence of the nonlinear properties with respect to the degree of polarization and diffuseness of the basis sets, we also performed calculations using simplified and enhanced versions of the correlation-consistent basis sets. The results for the nonaugmented sets (labeled VDZ-V5Z in Table V) indicate, as is widely known, that the diffuse exponents are important for the polarizability and crucial for the hyperpolarizability. Even the very large V5Z basis set yields β_{\parallel} values that are too low. These values are also less well converged than the much smaller aVDZ augmented set. The polarization functions with high angular momentum (i.e., larger than d and f for the hydrogen and heavy atoms, respectively) play a very small role, as seen from the similarity between the results obtained with the aV5Z set, and the aV5Zs set, where such functions were removed. The results in Table V show that this simplification has a negligible effect on the dipole moment and linear polarizability. It also has a fairly small effect ($\sim 3\%$) on the hyperpolarizability.

The hyperpolarizability values can be further improved by enhancing the diffuseness of the basis set beyond that of the standard Dunning correlation-consistent sets. When the aV5Zs set is extended with a single d -function with exponent 0.036 108 localized on the Cl atoms, the value of β_{zzz} was increased by almost 20%. Further extending the basis set with diffuse functions in the C and H atoms, resulting in the d-aV5Zs set, increases β_{zzz} slightly. The d-aV5Zs basis set provides the most saturated GTO results obtained in this

TABLE VI. Effect of the XC treatment on the components of the dielectric properties of CHCl_3 calculated with the HP GTO basis set. All values are in atomic units.

XC	μ_z	α_{yy}	α_{zz}	β_{yyy}	β_{yyz}	β_{zzz}	$\bar{\alpha}$	$\beta_{ }$	$\beta_{\text{HRS}}^{\text{VV}}$
LDA	0.414	65.95	47.03	28.89	-16.21	22.56	59.64	-5.91	17.83
PBE	0.405	65.51	46.60	27.64	-15.54	22.93	59.21	-4.89	17.12
BMK	0.453	62.13	44.77	22.56	-11.58	18.63	56.35	-2.72	13.56
B3LYP	0.423	63.60	45.60	21.90	-12.99	18.98	57.60	-4.21	13.87
CAM-B3LYP	0.435	62.20	45.01	19.91	-11.37	17.66	56.47	-3.05	12.45
HF	0.476	58.98	42.76	13.57	-7.58	13.02	53.57	-1.28	8.48
MP2	0.424	62.13	45.18	15.03	-9.75	16.47	56.48	-1.82	10.03
CCSD	0.425	61.54	44.88	16.51	-10.31	16.81	55.99	-2.29	10.78
Expt.	0.409 ± 0.008^a	61 ± 5^b	45 ± 3^b				56 ± 4^b	1 ± 4^c	

^aReference 36.^bReference 37.^cReference 1.

work. We note that, from the point of view of diffuse function augmentation, the basis set d-aV5Zs is equivalent to a d-aug-cc-pV5Z set. That is, the diffuse exponents used are the same as those in the Dunning set. The main difference between the two sets is in the high-angular-momentum exponents, which play a very small role as discussed above. We also attempted to perform the calculations with the t-aug-cc-pV5Z set; however, GAUSSIAN 03 calculations have convergence problems due to the extremely diffuse exponents.

We have also investigated the effect of the tight d -functions on Cl by carrying out calculations with the cc-pV(5+d)Z basis set for Cl augmented with diffuse functions and paired with the d-aV5Zs set from our original calculations for the C and H atoms. However, we find that the tight d functions change the individual components by only about 0.3% and $\beta_{||}$ by only about 2%. Finally, we find that the HyPol basis set, which is approximately equivalent in size to the aVTZ set, yields results that are equivalent to the much larger d-aV5Zs set. This basis set was designed with the explicit purpose of efficiently calculating nonlinear properties. Thus it is not surprising that it provides converged results for a smaller size.

The large variations in $\beta_{||}$ are due largely to changes in β_{zzz} , for which completing the basis set produces a change of sign and an absolute change of nearly two orders of magni-

tude. This effect translates into a change of $\beta_{||}$ of nearly an order of magnitude! The same behavior is observed as we complete the NBSs (see Table VIII) but not in the case of the real-space calculations (see Table VII). Then the two basis set approaches (GTO and NBS) show similar convergence features as we increase the quality of the basis set.

The variation in the results with the quality of the XC treatment is shown in Table VI. Note first that the *linear* properties are rather insensitive to the treatment of exchange and correlation. Also the higher-quality *ab initio* correlation treatments (MP2 and CCSD) are more or less consistent, while the DFT functionals LDA, PBE, BMK, B3LYP, and CAM-B3LYP vary considerably for the nonlinear properties. In the case of the HF, MP2, and CCSD methods, the properties follow the usual pattern of decreasing the dipole moment and enhancing the polarizability and hyperpolarizability tensor components as the treatment of correlation is improved. Surprisingly, the HF method provides the value of $\beta_{||}$ closest to experiment. The inclusion of correlation with the MP2 and CCSD methods results in poorer agreement. Compared to the CCSD values, the components of the hyperpolarizability calculated with the DFT methods vary by as much as 75%. Judging by the values of $\beta_{||}$, the predicted value increases with improvements in the treatment of correlation. The best overall DFT agreement with the CCSD re-

TABLE VII. Effect of the real-space-grid quality (radius R and spacing λ) on the components of the dielectric properties of CHCl_3 calculated with the PBE functional and LDA kernel. All values are in atomic units.

R	λ	μ_z	α_{yy}	α_{zz}	β_{yyy}	β_{yyz}	β_{zzz}	$\bar{\alpha}$	$\beta_{ }$	$\beta_{\text{HRS}}^{\text{VV}}$
12	0.25	0.398	65.921	46.924	27.975	-17.232	22.975	59.589	-6.921	17.106
15	0.25	0.399	66.019	46.993	27.159	-16.401	26.758	59.677	-3.629	17.461
17	0.25	0.399	66.022	46.995	27.123	-16.363	26.921	59.680	-3.485	17.443
20	0.25	0.399	66.023	46.995	27.119	-16.358	26.940	59.680	-3.469	17.441
22	0.25	0.399	66.023	46.995	27.119	-16.358	26.940	59.680	-3.468	17.441
17	0.35	0.397	66.032	47.002	27.181	-16.233	26.921	59.689	-3.351	17.415
17	0.30	0.399	66.029	46.989	27.168	-16.357	26.893	59.683	-3.492	17.455
17	0.25	0.399	66.022	46.995	27.123	-16.363	26.921	59.680	-3.485	17.443
17	0.20	0.398	66.021	46.993	27.091	-16.355	26.903	59.678	-3.488	17.427
Expt.		0.409 ± 0.008^a	61 ± 5^b	45 ± 3^b				56 ± 4^b	1 ± 4^c	

^aReference 36.^bReference 37.^cReference 1.

TABLE VIII. Effect of the NBS quality on the components of the dielectric properties of CHCl₃ calculated with the PBE functional. All values are in atomic units. See Table I for a detailed description of each basis set. Note that DZP and DZPe4 are both double- ζ basis with the same split radius except the first- ζ 's cutoff radius is the same as QZTPe4 for DZPe4.

Basis set	μ_z	α_{yy}	α_{zz}	β_{yyy}	β_{yyz}	β_{zzz}	$\bar{\alpha}$	$\beta_{ }$	$\beta_{\text{HRS}}^{\text{VV}}$
DZP	0.240	47.30	29.99	11.68	-5.35	-29.15	41.53	-23.91	15.02
DZPe4	0.361	56.45	37.62	6.05	-9.00	-39.98	50.17	-34.79	20.39
QZTPe4	0.392	63.71	44.69	12.18	-10.52	-9.90	57.37	-18.57	12.63
5Z4Pe5	0.397	65.14	46.02	21.93	-13.70	14.16	58.77	-7.96	14.17
5Z4Pe6	0.398	65.41	46.23	23.71	-14.77	19.51	59.02	-6.02	15.29
5Z4Pe7	0.398	65.45	46.28	24.48	-14.93	20.18	59.06	-5.22	15.64
5Z4Pe8	0.398	65.45	46.28	24.54	-14.90	21.37	59.06	-5.07	15.68
Expt.	0.409 ± 0.008^a	61 ± 5^b	45 ± 3^b				56 ± 4^b	1 ± 4^c	

^aReference 36.

^bReference 37.

^cReference 1.

sults for CHCl₃ is provided by the CAM-B3LYP functional. However, other studies show substantial variation of the accuracy of CAM-B3LYP results for different molecules and basis sets.⁷

The variation among the values can be regarded as an estimate of the error in the results due to approximations in the treatment of exchange and correlation.

C. Real-space grids

Convergence of the dipole moment, polarizability, and hyperpolarizability is illustrated in Table VII. The total energy was well converged for a spacing of 0.35 a_0 (equivalent plane-wave cutoff=20 Ry) and a sphere radius of 12 a_0 . The dipole moment was also well converged with that basis. However, to converge $\beta_{||}$ to better than 0.01 a.u., a spacing of 0.25 a_0 (equivalent plane-wave cutoff=40 Ry) and a sphere radius of 22 a_0 was required. The convergence of the tensor components of β is similar to that of $\beta_{||}$ in absolute terms, i.e., they are also converged to 0.01 a.u. or better with these parameters. Generally, the magnitude of $\beta_{||}$ declines with smaller spacing and larger radius, as the cancellation between the tensor components becomes closer.

Finite-difference calculations were done with the converged grid spacing of 0.25 a_0 , and a sphere radius of 22 a_0 , for comparison to the Sternheimer calculation with the same grid parameters (Table X). The differences between the linear-response and finite-difference calculations are small, allowing a direct comparison between the results with different basis sets. The use of the LDA kernel in the linear-response results gives a small discrepancy compared to the purely PBE finite-difference results. Fields of 0.015 a.u. rather than 0.02 a.u. as for the other basis sets were used because 0.02 a.u. was out of the linear regime in the real-space calculation. The linear response density $\rho_z^{(1)}(r)$ and polarizability density $\alpha_{zz}(r)$ are virtually identical between the finite-difference and linear-response calculations.

Calculations at 1064 nm with the same grid parameters show increases of about 1% in the polarizability, and 10%–20% in the hyperpolarizability, as compared to the static case. We find a small violation of Kleinman symmetry here, in that $\beta_{yyz} = -18.945$ a.u. whereas $\beta_{zyy} = -19.448$ a.u.

D. NBSs

Table VIII shows the variation of the dielectric properties as a function of the quality of the NBS; this is determined by the “energy shift” parameter $\delta\epsilon$ (which is a measure of the spatial extent of the basis) and the number of atomic orbitals.³⁰ The default DZP basis set, using the default energy shift of 0.01 a.u., produces results that are clearly inadequate, being unable to reproduce even the dipole moment. We have found for CHCl₃, that $\delta\epsilon$ must be at least 0.5×10^{-6} a.u. to obtain a value of $\beta_{||}$ that approaches those obtained with GTOs and real-space grids. NBS convergence is apparently achieved with $\delta\epsilon = 1 \times 10^{-8}$ a.u. The corresponding cutoff radii are two to three times larger than those produced by the default value of $\delta\epsilon$, clearly matching the need for very diffuse GTOs and large confining spheres for the real-space grids. Moreover, both the valence and polarization parts of the NBS have to be sufficiently flexible to represent the response properties accurately. For instance, at least five valence and four polarization atomic orbitals are needed to obtain accurate results, as is also the case for the GTOs (e.g., the d-aV5Zs set has five *sp* valence functions plus four *d*- and three *f*-polarization functions). The best value of $\beta_{||}$ is -5.07 a.u., obtained with the 5Z4Pe8 NBS. This is in good agreement with the value of -5.01 a.u. obtained with the d-aV5Zs GTO set. The close agreement with the GTO results seems to be fortuitous since the components of β differ by about 6%.

E. Linear and nonlinear response densities

The origin of the slow convergence of the response properties is difficult to determine just by analyzing their total values for different basis sets. We have found that the differences and similarities between those values can be visualized by computing the real-space distribution of the linear and nonlinear response densities, as well as the associated polarizability and hyperpolarizability densities. For example, Fig. 1 shows the linear response density $\rho_z^{(1)}(r)$ calculated with the PBE functional for both the HP GTO basis set and a real-space grid. Clearly, the nearly identical plots confirm that the linear response is equally well represented by both basis sets. Also shown in Fig. 1, the polarizability density

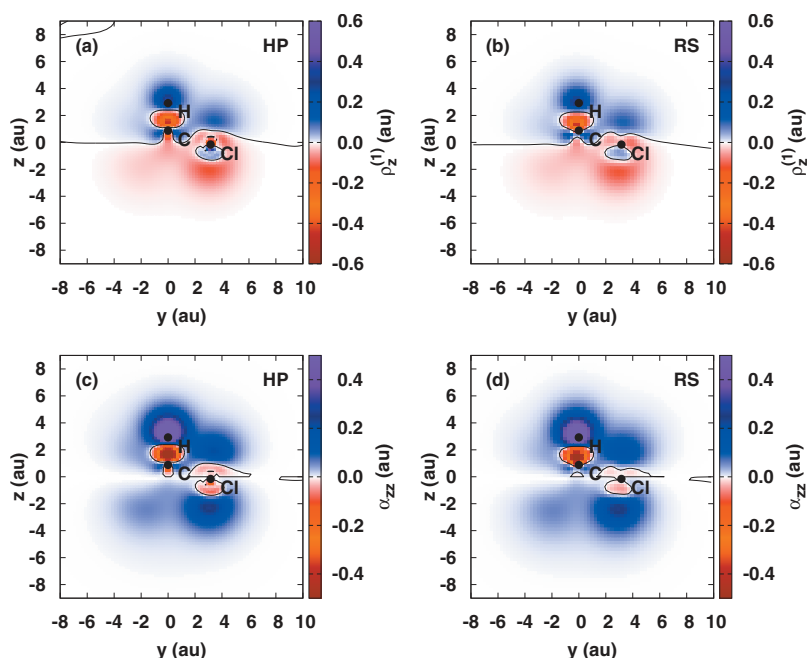


FIG. 1. (a) and (b) Linear response density $\rho_z^{(1)}(r)$ and (c) and (d) polarizability density $\alpha_{zz}(r)$ on one of the HCl planes of the molecule calculated with a GTO basis set (HP) and a real-space grid (RS) using the PBE functional. The positions of the nuclei are indicated with black dots, and the black lines are zero isolines. All quantities are in atomic units. Note that the linear response density is quite similar for both methods.

$\alpha_{zz}(r)$ reveals the spatial contributions to the total polarizability. For the most part, this property is localized to within ~ 6 a.u. of the center of the molecule, explaining its rapid convergence with respect to the diffuseness of the basis set. Our partitioning scheme for $\alpha_{zz}(r)$ (Table IX) shows that most of the positive contribution to α_{zz} arises from the Cl atoms, in accord with the larger polarizability of the Cl atom. The contribution from the CH bond is significantly smaller and, as can be clearly seen in Fig. 1, is the result of counteracting contributions: positive from the H atom and negative from the C–H bond region. Our partitioning also shows that the deficiencies of the GTOs are due mostly to a poorer representation of the Cl atoms.

The spatial distributions of the nonlinear response density $\rho_{zz}^{(2)}(r)$ and hyperpolarizability density $\beta_{zzz}(r)$, shown in Fig. 2, are also very similar for both the HP GTO set and the real-space grid. The hyperpolarizability density, however, is more delocalized than the polarizability, extending up to ~ 8 a.u. from the center, thus stressing the importance of the diffuse functions in calculations of nonlinear properties. The

spatial distribution is also much more complex, with several regions of counteracting contributions. The decomposition shown in Table IX significantly simplifies the analysis of the densities. It shows that the overall contribution from the C–H bond is negative. This contribution also varies very little with respect to the quality of the basis set. The contribution from the Cl atoms, on the other hand, is positive and varies significantly with the basis set used. For instance, the value obtained for the aVDZ set is almost 30% lower than the aV5Z set. Even the aV5Z basis set does not provide converged results, since the inclusion of extra diffuse exponents (d-aV5Zs set) results in a further increase of 5%. This change is small for the individual Cl contribution, but changes the total β_{zzz} component by 16%. Finally, it can again be seen that, although significantly smaller, the HP set provides results that are almost identical to those obtained with the d-aV5Zs one. The plots shown in Fig. 2 indicate that the CH bond is well saturated with respect to the extent of the diffuse functions. This observation was confirmed by us-

TABLE IX. Partitioning of the linear and nonlinear response properties calculated numerically with GTOs and real-space grids (RS) using the PBE functional together with the numerical sum over the CH and Cl₃ partitions. For comparison the “Analytic” results are given for the integral over all space (without partitioning) from Table V.

Property	Basis set	CH	Cl ₃	CH+Cl ₃	Analytic
α_{zz}	aVDZ	8.91	35.42	44.33	44.35
	aV5Z	8.96	37.71	46.67	46.71
	d-aV5Zs	8.96	37.79	46.75	46.79
	HP	8.92	37.67	46.59	46.60
	RS	8.85	38.02	46.87	46.87
β_{zzz}	aVDZ	-43.82	46.46	2.63	2.62
	aV5Z	-44.48	63.08	18.61	18.65
	d-aV5Zs	-44.31	66.54	22.24	22.27
	HP	-44.55	67.49	22.94	22.93
	RS	-43.41	67.12	23.71	23.89

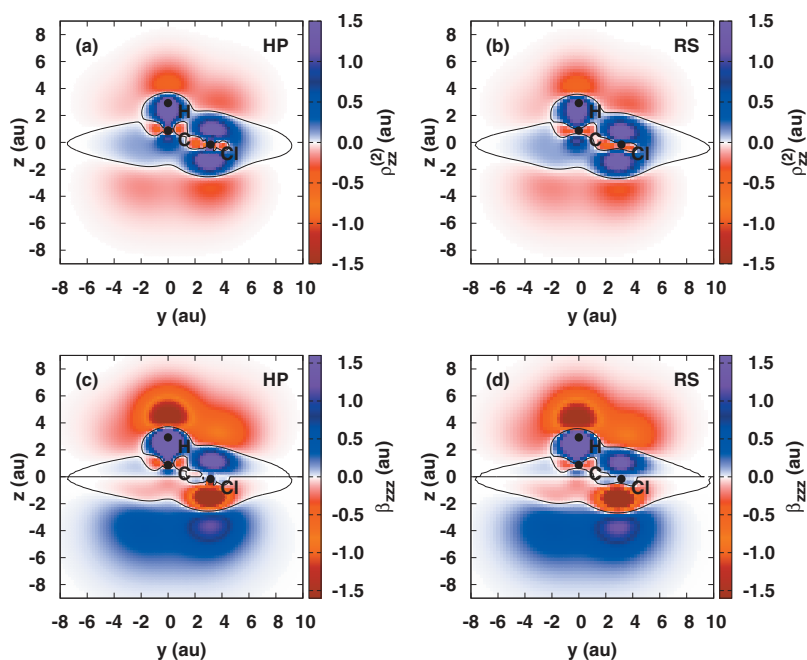


FIG. 2. (a) and (b) Nonlinear response density $\rho_{zz}^{(2)}(r)$ and (c) and (d) hyperpolarizability density $\beta_{zzz}(r)$ on one of the HCCI planes of the molecule calculated with a GTO basis set (HP) and a real-space grid (RS) using the PBE functional. The positions of the nuclei are indicated with black dots, and the black lines are zero isolines. All quantities are in atomic units. The nonlinear densities extend much further into space than the linear densities. The agreement between the real-space and GTO methods is nevertheless quite good. The contributions to the hyperpolarizability from the Cl atoms and the CH bond are of opposite sign and, as indicated by the nonlinear response density, have contributions that extend even further into space.

ing the d-aV5Zs basis set, which also includes diffuse functions on both the C and H atoms and only slightly improves the results.

IV. CONCLUSIONS

We have carried out calculations of the static hyperpolarizability of the gas-phase CHCl_3 molecule, using three different kinds of basis sets: Gaussian-type orbitals, NBSs, and real-space grids. We find that all of these methods can yield quantitatively similar results provided sufficiently large, diffuse basis sets are included in the calculations. In particular diffuse functions are important to obtain accurate results for the polarizability and are crucial for the hyperpolarizability. For GTOs, the standard versions of the augmented Dunning basis set are not adequate to converge the β_{zzz} component. However, convergence can be achieved by increasing the diffuse d space on the Cl atoms. Other diffuse functions play a smaller role. The overall consistency among the results gives confidence to their reliability and overall accuracy. Based on the size of the basis sets and degree of convergence, the linear-response (LR) real-space values in Table X are likely the most reliable PBE results obtained in

this study. However, the variation among our results also provides a gauge of their overall theoretical accuracy.

Although the treatment here has been restricted to chloroform, many of the results are more generally applicable. For example, the spatial distributions provided by the linear and nonlinear response densities provides a good visualization tool to understand the basis set requirements for the simulation of linear and nonlinear response. A key finding for chloroform is that the local contributions near the Cl atoms and the CH bond are of opposite signs and tend to cancel, thus explaining the overall weakness of the hyperpolarizability. The molecule's response is quite extended in space and so real-space grids on a large domain, as well as very extended local orbitals, are required to describe it properly. The frequency-dependence of the polarizability and hyperpolarizability is small, as verified by our time-dependent calculations, and so dispersion is not very important in comparing static theoretical results to experimental measurements.

The discrepancy between the experimentally determined linear polarizability and our theoretical results is essentially eliminated when the vibrational component is taken into account. Our results for the hyperpolarizability for all three

TABLE X. Summary of the best PBE results obtained with the GTOs, NBSs and real-space grids. Real-space grids (LR denotes linear response, and FD finite difference) have radius of $22 a_0$ and spacing of $0.25 a_0$. All values are in atomic units.

Basis set	μ_z	α_{yy}	α_{zz}	β_{yyy}	β_{yyz}	β_{zzz}	$\bar{\alpha}$	$\beta_{ }$	$\beta_{\text{HRS}}^{\text{VV}}$
GTO d-aV5Zs	0.404	65.70	46.79	27.35	-15.31	22.27	59.40	-5.01	16.90
NBS 5Z4Pe8	0.398	65.45	46.28	24.54	-14.90	21.37	59.06	-5.07	15.68
RS LR	0.399	66.02	47.00	27.12	-16.36	26.94	59.68	-3.47	17.44
RS FD	"	66.05	46.87	24.74	-15.17	23.89	59.66	-3.87	15.97
RS 1064 nm	"	66.69	47.34	30.35	-18.95	31.56	60.24	-4.01	19.91
Expt.	0.409 ± 0.008^a	61 ± 5^b	45 ± 3^b				56 ± 4^b	1 ± 4^c	

^aReference 36.

^bReference 37.

^cReference 1.

basis sets are all consistent with each other. Given the error bars in the experimental result our PBE hyperpolarizability results are smaller, though essentially consistent with the experimental measurements for $\beta_{||}$, even without taking into account vibrational contributions. With better treatments of exchange and correlation (Table VI) the agreement is expected to be further improved. Experimental results indicate that the vibrational contribution is small for the hyperpolarizability: differences in the hyperpolarizability of isotopically substituted molecules show the vibrational contributions. Although measurements at the same frequency are not available for CHCl_3 and CDCl_3 , Kaatz *et al.*¹ found that at 694.3 nm, CHCl_3 has $\beta_{||}=1.2 \pm 2.6$ a.u.; at 1064 nm CDCl_3 has $\beta_{||}=1.0 \pm 4.2$ a.u. Given that the frequency-dispersion of $\beta_{||}$ between zero frequency and 1064 nm is only about +15% in our calculations (much smaller than the error bars), this isotopic comparison shows that the vibrational contributions are less than the error bars. Therefore vibrational contributions are not significant in comparing the *ab initio* results to the available experimental measurements. We find additionally that the molecular structure has a significant influence on the calculated value of $\beta_{||}$, and so it is crucial to use an accurate structure for theoretical calculations.

ACKNOWLEDGMENTS

This work was supported in part by U.S. Department of Energy Grant No. DE-FC36-08GO18024 (F.V.) and DE-FG03-97ER45623 (J.J.R.), and the National Science Foundation Grant No. 0120967 through the STC MDITR (Y.T. and F.V.). D.S. was supported by a NSF IGERT fellowship. This work was also supported by National Science Foundation Grant No. DMR07-05941 (D.S.) and by the Director, Office of Science, Office of Basic Energy Sciences, Materials Sciences and Engineering Division, U.S. Department of Energy under Contract No. DE-AC02-05CH11231 (S.G.L.). Computational resources have been provided by DOE at Lawrence Berkeley National Laboratory's NERSC facility and Lawrence Livermore National Laboratory's NERSC facility and Lawrence Livermore National Laboratory's NERSC facility and Lawrence Livermore National Laboratory's NERSC facility. We acknowledge funding by the Spanish MEC (Grant No. FIS2007-65702-C02-01), "Grupos Consolidados UPV/EHU del Gobierno Vasco" (Grant No. IT-319-07) and the European Community through e-I3 ETSF project (Contract No. 211956). We acknowledge support by the Barcelona Supercomputing Center, "Red Española de Supercomputación," SGIker ARINA (UPV/EHU), and ACI-Promociona (Grant No. ACI2009-1036).

¹P. Kaatz, E. A. Donley, and D. P. Shelton, *J. Chem. Phys.* **108**, 849 (1998).

²P. Das, *J. Phys. Chem. B* **110**, 7621 (2006).

³K. Clays and A. Persoons, *Phys. Rev. Lett.* **66**, 2980 (1991).

- ⁴S. P. Karna and M. Dupuis, *Chem. Phys. Lett.* **171**, 201 (1990).
- ⁵C. K. Miller and J. F. Ward, *Phys. Rev. A* **16**, 1179 (1977).
- ⁶E. R. Davidson, B. E. Eichinger, and B. H. Robinson, *Opt. Mater.* **29**, 360 (2006).
- ⁷J. Hammond and K. Kowalski, *J. Chem. Phys.* **130**, 194108 (2009).
- ⁸J. P. Perdew, K. Burke, and M. Ernzerhof, *Phys. Rev. Lett.* **77**, 3865 (1996).
- ⁹P. Chopra, L. Carlucci, H. F. King, and P. N. Prasad, *J. Phys. Chem.* **93**, 7120 (1989).
- ¹⁰M. Nakano, R. Kishi, T. Nitta, B. Champagne, E. Botek, and K. Yamaguchi, *Int. J. Quantum Chem.* **102**, 702 (2005).
- ¹¹J.-M. Colmont, D. Priem, P. Dréan, J. Demaison, and J. E. Boggs, *J. Mol. Spectrosc.* **191**, 158 (1998).
- ¹²D. A. Kleinman, *Phys. Rev.* **126**, 1977 (1962).
- ¹³A. Willetts, J. E. Rice, D. M. Burland, and D. P. Shelton, *J. Chem. Phys.* **97**, 7590 (1992).
- ¹⁴S. J. Cyvin, J. E. Rauch, and J. C. J. Decius, *J. Chem. Phys.* **43**, 4083 (1965).
- ¹⁵R. Bersohn, Y.-H. Pao, and H. L. Frisch, *J. Chem. Phys.* **45**, 3184 (1966).
- ¹⁶T. H. Dunning, Jr., *J. Chem. Phys.* **90**, 1007 (1989).
- ¹⁷D. E. Woon and T. H. Dunning, Jr., *J. Chem. Phys.* **98**, 1358 (1993).
- ¹⁸T. Pluta and A. J. Sadlej, *Chem. Phys. Lett.* **297**, 391 (1998).
- ¹⁹J. P. Perdew and A. Zunger, *Phys. Rev. B* **23**, 5048 (1981).
- ²⁰A. D. Becke, *J. Chem. Phys.* **98**, 5648 (1993).
- ²¹P. Stephens, F. Devlin, C. Chabalowski, and M. Frisch, *J. Phys. Chem.* **98**, 11623 (1994).
- ²²A. D. Boese and J. M. L. Martin, *J. Chem. Phys.* **121**, 3405 (2004).
- ²³T. Yanai, D. P. Tew, and N. C. Handy, *Chem. Phys. Lett.* **393**, 51 (2004).
- ²⁴M. J. Frisch, G. W. Trucks, H. B. Schlegel *et al.*, GAUSSIAN 03, Revision B.04, Gaussian Inc., Pittsburgh PA, 2003.
- ²⁵M. J. Frisch, G. W. Trucks, H. B. Schlegel *et al.*, GAUSSIAN 09, Revision A.1, Gaussian Inc., Wallingford, CT, 2009.
- ²⁶A. Castro, H. Appel, M. Oliveira, C. A. Rozzi, X. Andrade, F. Lorenzen, M. A. L. Marques, E. K. U. Gross, and A. Rubio, *Phys. Status Solidi B* **243**, 2465 (2006).
- ²⁷M. A. L. Marques, A. Castro, G. F. Bertsch, and A. Rubio, *Comput. Phys. Commun.* **151**, 60 (2003).
- ²⁸X. Andrade, S. Botti, M. A. L. Marques, and A. Rubio, *J. Chem. Phys.* **126**, 184106 (2007).
- ²⁹N. Troullier and J. L. Martins, *Phys. Rev. B* **43**, 1993 (1991).
- ³⁰E. Artacho, D. Sánchez-Portal, P. Ordejón, A. García, and J. M. Soler, *Phys. Status Solidi B* **215**, 809 (1999).
- ³¹J. M. Soler, E. Artacho, J. D. Gale, A. García, J. Junquera, P. Ordejón, and D. Sánchez-Portal, *J. Phys.: Condens. Matter* **14**, 2745 (2002).
- ³²O. F. Sankey and D. J. Niklewski, *Phys. Rev. B* **40**, 3979 (1989).
- ³³S. Huzinaga, *Comput. Phys. Rep.* **2**, 281 (1985).
- ³⁴A. Szabo and N. Ostlund, *Modern Quantum Chemistry* (Mac-Millan, New York, 1982).
- ³⁵M. Jen and D. R. Lide, Jr., *J. Chem. Phys.* **36**, 2525 (1962).
- ³⁶P. Reinhardt, Q. Williams, and T. Weatherly, *J. Chem. Phys.* **53**, 1418 (1970).
- ³⁷H. A. Stuart and H. Volkman, *Ann. Phys.* **410**, 121 (1933).
- ³⁸C. K. Miller, B. J. Orr, and J. F. Ward, *J. Chem. Phys.* **74**, 4858 (1981).
- ³⁹D. M. Bishop and L. M. Cheung, *J. Phys. Chem. Ref. Data* **11**, 119 (1982).
- ⁴⁰B. Kirtman, B. Champagne, and J.-M. Andre, *J. Chem. Phys.* **104**, 4125 (1996).
- ⁴¹D. Feller, *J. Chem. Phys.* **96**, 6104 (1992).
- ⁴²D. E. Woon and T. H. Dunning, Jr., *J. Chem. Phys.* **100**, 2975 (1994).
- ⁴³D. Feller and K. A. Peterson, *J. Mol. Struct.: THEOCHEM* **400**, 69 (1997).