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Bioactive Indole Alkaloids from Croton echioides

Claudio R. Novello,^{*,a} Luís C. Marques,^b Murilo E. Pires,^c Ana P. Kutschenco,^d Celso V. Nakamura,^e Samara Nocchi,^e Maria H. Sarragiotto^c and João C. P. Mello^d

^aDepartamento de Química e Biologia, Universidade Tecnológica Federal do Paraná, 85601-970 Francisco Beltrão-PR, Brazil

^bUniversidade Bandeirante de São Paulo, 04710-000 São Paulo-SP, Brazil

^cDepartamento de Química, ^dDepartamento de Farmácia and ^eDepartamento de Ciências Básicas da Saúde, Universidade Estadual de Maringá, 87020-900 Maringá-PR, Brazil

Bioguided fractionation of a hydroethanolic extract from the stem bark of *Croton echioides* Baill. led to isolation of the new indole alkaloid *N-trans*-feruloyl-3,5-dihydroxyindolin-2-one as a stereoisomeric mixture and the known alkaloids *N-trans-p*-coumaroyl-tryptamine, *N-trans-p*-coumaroyl-5-hydroxytryptamine, *N-trans*-4-methoxy-cinnamoyl-5-hydroxytryptamine, *N-trans*-feruloyl-5-hydroxytryptamine (moschamine), from the ethyl acetate fraction. The flavonoids 3-*o*-methyl kaempferol, 3-*o*-methyl quercetin, 3,7-di-*o*-methyl quercetin and 3,3'-di-*o*-methyl quercetin, together with the benzoic acid derivatives 4-hydroxybenzoic acid, 4-hydroxy-3-methoxybenzoic acid and 4-hydroxy-3,5-dimethoxybenzoic acid were also isolated. Alkaloids and the flavonoids showed strong antioxidant properties *in vitro* radical scavenging assay (2,2-diphenyl-1-picrylhydrazyl, DPPH), with IC₅₀ values ranging from 9.2 to 17.5 µmol L⁻¹, lower than those of the positive control Trolox (IC₅₀ = 17.9 µmol L⁻¹). Alkaloids showed cytotoxic activity against the HCT-116 human cancer cell line, with IC₅₀ values ranging from 86.8 to 210.7 µmol L⁻¹. Compound *N-trans*-4-methoxy-cinnamoyl-5-hydroxytryptamine was the most active, with an IC₅₀ value of 86.8 µmol L⁻¹.

Keywords: *Croton echioides*, Euphorbiaceae, indole alkaloid, cytotoxic activity, antioxidant activity

Introduction

Plants of the genus *Croton* are well known for their many therapeutic uses, including anti-inflammatory, antiulcer, antitumor, antioxidant and cytotoxic activity.¹⁻³ This genus is a rich source of special metabolites, mainly clerodane diterpenes, flavonoids, and different classes of alkaloids.⁴ Alkaloids related to benzylisoquinolines, such as morphinandienones and tetrahydroprotoberberine alkaloids are the most frequent class found in members of the genus *Croton.*⁵ Glutarimide alkaloids and a new class of sesquiterpene guaiane-type alkaloids have recently been reported from *Croton* species.⁶

Croton L. is the second-largest genus in the family Euphorbiaceae, with about 1,200 species, 350 of which occur in Brazil.⁵ *Croton echioides* Baill., popularly

*e-mail: crnovello@utfpr.edu.br

known as "quebra-faca", "caatinga branca", "velame" and "canela-de-velho", is a small native tree found in the North-eastern region of Brazil.^{7,8} The stem bark of this plant has been widely marketed as an aphrodisiac and tonic, as a substitute for the roots of Amazon Marapuama, *Ptychopetalum olacoides* Benth. (Olacaceae).⁹ To our knowledge, no chemical or biological study of this species has been reported.

In this paper, we describe the isolation and structure identification of a new indole alkaloid identified as *N*-*trans*-feruloyl-3,5-dihydroxyindolin-2-one (1) together with the known alkaloids *N*-*trans*-*p*-coumaroyl-tryptamine (2),¹⁰ *N*-*trans*-*p*-coumaroyl-5-hydroxytryptamine (3),¹¹ *N*-*trans*-4-methoxy-cinnamoyl-5-hydroxytryptamine (4)¹² and *N*-*trans*-feruloyl-5-hydroxytryptamine (moschamine) (5),¹³ the flavonoids 3-*o*-methyl kaempferol (6),¹⁴ 3-*o*-methyl quercetin (7),¹⁴ 3,7-di-*o*-methyl quercetin (8)¹⁵ and 3,3'-di-*o*-methyl quercetin (9),¹⁶ and benzoic acid derivatives 4-hydroxybenzoic acid (10), 4-hydroxy-

In memorian of Prof Gentil José Vidotti.

3-methoxybenzoic acid (11) and 4-hydroxy-3,5dimethoxybenzoic acid (12)¹⁷ from the stem bark of *Croton echioides*. The *in vitro* cytotoxicity against the HCT-116 human cancer cell line was determined for the isolated alkaloids 1-5 and the free radical (2,2-diphenyl-1-picrylhydrazyl, DPPH) scavenging activities were also determined for the compounds 1-12.

Experimental

General experimental procedures

Infrared (IR) spectra were recorded on a Bomem MB-Series spectrometer (Bomem, Quebec, CA, USA) in a 400 to 4000 cm⁻¹ spectral region using KBr matrix. Shimadzu UV-1650-PC spectrometer (Shimadzu, Kyoto, Japan) was used for UV-Vis determinations. Optical rotation was measured on a JASCO P-2000 polarimeter (Jasco, Tokyo, Japan) in MeOH. High-resolution electrospray ionization mass spectrometry (HR-ESIMS) were recorded on a Bruker-Daltonics MicroTof (Bruker-Daltonics, Bremen, Germany), and the ESIMS were recorded on a Thermo Scientific Orbitrap LTQ XL spectrometer (Thermo Fisher System, San Jose, CA, USA) performed at University of Münster, Germany. Nuclear magnetic resonance (NMR) spectra were recorded on a Varian Mercury Plus 300 MHz spectrometer (7.02 T) (Varian, Palo Alto, CA, USA) operated at 75 MHz for ¹³C and 300 MHz for ¹H and 2D NMR, using deuterated MeOH as solvent and tetramethylsilane (TMS) as internal standard. Medium pressure liquid chromatography (MPLC) separation was performed using a Waters 6000A (Waters, Milford, Massachusetts, USA) pump with a manual sample injection valve 10 mL loop coupled with different columns $(450 \times 10 \text{ mm internal diameter (i.d.)};$ 340×30 mm i.d.; 300×50 mm i.d.) loaded with Sepra C18-E 50 µm (Phenomenex, Torrance, CA, USA) as the stationary phase. The MPLC system is equipped with a Pharmacia-Biotech FRAC-200 (Amersham Biosciences KK, Tokyo, Japan) fraction collector. Preparative counter-current chromatography was performed using a high-speed counter-current chromatography (HSCCC) (P. C. Inc., Potomac, USA) chromatograph equipped with a polytetrafluoroethylene (PTFE) column (2.5 mm i.d.; 200 mL total capacity) coupled to a Waters Mod. 6000A (Waters, Milford, Massachusetts, USA) pump and a manual sample injection valve with a 10 mL loop. Column chromatography was carried out on Sephadex LH 20 (Pharmacia Fine Chemicals Inc., New Market, NJ, USA). Pharmacia-Biotech FRAC-200 fraction collector (Amersham Biosciences KK, Tokyo, Japan) was used for isolations procedures. Thin-layer chromatography (TLC) was performed on precoated silica gel aluminum sheets (Kieselgel 60 F254, 0.20 mm, Merck, Darmstadt, Germany). Fractions and pure compounds were monitored by TLC, and spots were revealed by heating (110 $^{\circ}$ C) the plates sprayed with a 2% vanillin in chloride acid/EtOH (1:2) solution.

HSCCC separations

The elution systems used in HSCCC separations were optimized by determination of the partition coefficient (Kp) of target compounds in a series of biphasic liquid systems by UV-Vis determinations. Sets of biphasic solvent systems composed of hexane, petroleum ether, ethyl acetate, methanol, ethanol and water was examined at different arrangements and volume ratios. The biphasic liquid system was prepared by adding portions of the solvents in a separatory funnel and, after the equilibrium was reached, 1 mL of each phase (upper and lower) were added in a test tube. After that, 1 mg of the solute was introduced. The test tube was shaken manually until equilibrium was reached. The system was equilibrated for at least 1 h and the concentration of the solute in the upper and the lower phase was analyzed by UV-Vis. The Kp was calculated as the ratio of the solute concentration in the upper and the lower phase. For the samples purification process, the most suitable solvent system (H₂O/EtOH/EtOAc/hexane; 2.5:1:1.5:1.5 v/v) was prepared and thoroughly equilibrated in a separatory funnel at room temperature and the two phases were separated shortly before use. The sample solution was prepared by dissolving the crude sample in 10 mL of the stationary lower phase of the solvent system. The multilayer coiled column was first entirely filled with the lower phase. The upper mobile phase was then pumped into the inlet of the column on tail to head option at the flow rate of 1.0 mL min⁻¹, while the apparatus was run at 800 rpm. After hydrodynamic equilibrium was reached, a sample was injected. The eluent was continuously monitored by TLC and the fractions were collected according to the chromatographic profile. After target compounds were eluted, the HSCCC apparatus was stopped and the column contents were fractionated.

Plant material

The aerial parts of flowering plants of *Croton echioides* Baill. were collected in Lagoa Real County, Bahia State, Brazil, in November 2008, and identified by Prof Daniela Santos Carneiro-Torres. A voucher specimen (No. HUEFS 139049) was deposited in the herbarium of the State University of Feira de Santana, Bahia State, Brazil.

Extraction and isolation

Air-dried and ground stem bark (7.6 kg) of C. echioides were extracted with EtOH/H₂O (70%, 76 L) through an Ultra-Turrax Ika Works UTC 115/KT (Ika Works, Wilmington, NC, USA) at room temperature for 48 h, subdivided into 5 min of Ultra-Turrax turbo extraction followed by 2 h of rest, successively. The crude extract was filtered, the solvent was evaporated under reduced pressure, and the residue was lyophilized to give the crude extract (383 g, yield 5.04%). Part of the crude extract (300 g) was suspended in water (2.5 L) and partitioned with hexane, CH₂Cl₂, EtOAc and BuOH (2.5 L) to give the fractions respectively termed HF (62.6 g, yield 20.9%), DF (44.2 g, yield 14.7%), ethyl acetate fraction (EAF) (23.5 g, yield 7.8%) and BF (53.6 g, yield 17.9%). The EAF was subjected to chromatography column over Sephadex LH-20 ($460 \times 55 \text{ mm i.d.}$) eluting with a gradient of EtOH/H₂O, MeOH, Me₂CO/H₂O and AcOH/H₂O to give 26 fractions (EAF-1 to EAF-26). Fraction EAF-9 (2.43 g, yield 10.3%) was separated by preparative MPLC (column 300 \times 50 mm i.d., MeOH/H₂O 1:1 v/v, 3 mL min⁻¹) to give four subfractions (EAF-9.1 to EAF-9.4). Subfraction AF-9.2 (190 mg, yield 7.8%) was purified by HSCCC using H₂O/EtOH/EtOAc/hexane (2.5:1:1.5:1.5 v/v, 1 mL min⁻¹), to yield 1 (47.2 mg, yield 24.8%), 4-hydroxy-3-methoxybenzoic acid (35.0 mg, yield 18.4%) and 4-hydroxy-3,5-dimethoxybenzoic acid (15.0 mg, yield 7.9%). Subfraction EAF-9.3 (241 mg, yield 9.9%) was purified by HSCCC in the same conditions as AF-9.2 to provide 4-hydroxybenzoic acid (5.0 mg, yield 2.1%). The EAF-17 (1.04 g, yield 4.4%) fraction was subjected to MPLC in a C-18 column (340 \times 30 mm i.d., MeOH/H₂O, 3:7 v/v, 2 mL min⁻¹), to give two subfractions (AF-17.1 and AF-17.2). Subfractions EAF-17.1 (106 mg, yield 10.2%) and AF-17.2 (60 mg, yield 5.8%) were purified by MPLC in a C-18 column ($450 \times 10 \text{ mm i.d.}$, MeOH/H₂O, 1:1 v/v, 1.5 mL min⁻¹) to yield **4** (5.0 mg, yield 8.3%) and 5 (13.4 mg, yield 22.3%). Purification of the EAF-18 fraction (398.8 mg, yield 1.7%) by MPLC in a C-18 column $(340 \times 30 \text{ mm i.d.}, \text{MeOH/H}_2\text{O}, 7:3 \text{ v/v}, 2 \text{ mL min}^{-1}),$ afforded 8 (28.4 mg, yield 7.1%) and 9 (25.2 mg, yield 6.3%). Purification of EAF-19 (283 mg, yield 1.20%) by MPLC (340 \times 30 mm i.d., EtOH/H₂O, 3:7 to 9:1 v/v, 2 mL min^{-1}) provided 6 (10.5 mg, yield 3.7%). Subfraction EAF-19.1 (65.0 mg, yield 23.0%) was purified by MPLC $(450 \times 10 \text{ mm i.d.}, \text{MeOH/H}_2\text{O}, 5:5 \text{ v/v}, 1.5 \text{ mL min}^{-1})$ to provide 3 (51.6 mg, yield 79.4%). Purification of EAF-20 (220.6 mg, yield 0.9%) by MPLC (340 × 30 mm i.d., MeOH/H₂O, 7:3 v/v, 2 mL min⁻¹) afforded compounds 7 (78.2 mg, yield 35.4%) and 2 (16.2 mg, yield 7.3%).

N-trans-Feruloyl-3,5-dihydroxyindolin-2-one (1)

Amorphous gray solid; mp: 156-158 °C; $[\alpha]_D^{25} = 0^\circ$ (*c* 0.25, MeOH); IR (KBr) v_{max} / cm⁻¹ 3390, 1711, 1650, 1600, 1208; NMR data see Table 1; HR-ESIMS, positive ion mode, *m/z* 407.1220 [M + Na]⁺; calcd. for C₂₀H₂₀N₂O₆: 407.1219.

N-trans-p-CoumaroyI-tryptamine (2)

Amorphous yellow solid; NMR data were in agreement with literature values;¹⁰ HR-ESIMS, positive ion mode, m/z 329.1257 [M + Na]⁺; calcd. for C₁₉H₁₈N₂O₂: 329.1266.

N-trans-p-CoumaroyI-5-hydroxytryptamine (3)

Amorphous gray solid; NMR data were in agreement with literature values;¹¹ HR-ESIMS, positive ion mode, m/z 345.1212 [M + Na]⁺; calcd. for C₁₉H₁₈N₂O₃: 345.1215.

N-trans-4-Methoxy-cinnamoyl-5-hydroxytryptamine (4)

Amorphous gray solid; mp 160-162 °C; ¹H NMR (300 MHz, CD₃OD) δ 7.47 (d, 1H, *J* 16.2 Hz, H-15), 7.45 (d, 2H, *J* 8.4 Hz, H-17, H-21), 7.15 (d, 1H, *J* 8.7Hz, H-7), 7.02 (s, 1H, H-2), 6.95 (d, 1 H, *J* 2.4 Hz, H-4), 6.93 (d, 2H, *J* 9.0 Hz, H-18, H-20), 6.65 (dd, 1H, *J* 8.7, 2.4 Hz, H-6), 6.44 (d, 1H, *J* 15.6 Hz, H-14), 3.81 (s, 3H, OCH₃), 3.56 (t, 2H, *J* 7.2 Hz, H-11), 2.92 (t, 2H, *J* 7.2 Hz, H-10); ¹³C NMR (75 MHz, CD₃OD) 168.0 (C, C-13), 161.3 (C, C-19), 149.8 (C, C-5), 140.3 (CH, C-15), 131.9 (C, C-8), 129.3 (CH, C-17, C-21), 128.2 (C, C-9), 127.7 (C, C-16), 123.2 (C, C-2), 118.2 (CH, C-14), 114.1 (CH, C-18, C-20), 111.6 (CH, C-7), 111.3 (C, C-3), 111.2 (CH, C-6), 102.4 (CH, C-4), 54.7 (CH₃, OCH₃), 40.3 (CH₂, C-11), 25.2 (CH₂, C-10); ESIMS, positive ion mode, *m*/z 359.5 [M + H]⁺; calcd. for C₂₀H₁₉N₂O₃: 337.5.

N-trans-FeruloyI-5-hydroxytryptamine (5)

Amorphous gray solid; NMR data were in agreement with literature values;¹¹ HR-ESIMS, positive ion mode, m/z 375.1310 [M + Na]⁺; calcd. for C₂₀H₂₀N₂O₄: 375.1321.

Cytotoxic assays

The cytotoxicity activity against HCT-116 cells was determined by *in vitro* sulforhodamine B test (SRB-test).¹⁸ HCT-116 cell lines, originated from a human colorectal carcinoma, were cultured and maintained in Dulbecco's modified Eagle's medium (DMEM, Gibco, Grand Island, NY, USA) supplemented with 10% heat-inactivated fetal bovine serum (FBS, Gibco) and 50 µg mL⁻¹ gentamycin, in incubator set at 37 °C, 5% CO₂ and 95% relative humidity. The cells were expanded when monolayer reached confluence after 3. After reaching 80% confluence, cells

were digested by using 0.25% trypsin (Gibco) containing 1 mmol EDTA (ethylenediamine tetraacetic acid). The cells were seeded in 96-well tissue plates (TPP-Techno Plastic Products, Trasadingen, Switzerland) at a density of 8×10^5 cell mL⁻¹ in 100 µL of medium for 24 h in the CO₂ incubator. The sample solutions with different concentrations (3.12, 6.25, 12.5, 25, 50 and 100 µg mL⁻¹) were prepared and added to the culture media. After incubation for 48 h, the cell monolayers were washed with 100 µL phosphate buffered saline (PBS) sterile, fixed with trichloroacetic acid and stained for 30 min with SRB (7.16 mmol L⁻¹) (Sigma Chemical Co., St. Louis, MO, USA) in acetic acid solution. The dye was removed by four washes with acetic acid solution (175 mmol L⁻¹), and protein-bound dye was extracted with 10 mmol L⁻¹ unbuffered tris base [tris(hydroxymethyl)aminomethane] for determination of optical density in a computer-interfaced, 96-well microtiter plate reader (Power Wave XS, BIO-TEK, Winooski, VT, USA). The IC₅₀ values were determined from a percentage of cells relative to untreated control and the half-life of the HTC cells in the presence and absence of each tested sample was estimated as the time point at which the cell viability decreased to 50% in relation to the beginning of the test. All the tests were performed in triplicate.

DPPH assay

The free radical scavenging activity was determined by in vitro DPPH (2,2-diphenyl-1-picrylhydrazyl) assay.¹⁹ Sample stock solutions (1.0 mg mL⁻¹) of analyte and Trolox (positive control) were diluted to final concentrations of 250, 125, 50, 25, 10 and 5 µmol L⁻¹, in methanol. 1 mL

of a 0.3 mmol L⁻¹ DPPH solution in methanol was added to 2.5 mL of sample solutions of different concentrations, and allowed to react at room temperature. After 30 min the absorbance values were measured at 518 nm and converted into the percentage antioxidant activity. 1 mL of methanol DPPH solution 0.3 mmol L⁻¹ plus methanol (2.5 mL) was used as a negative control. The IC50 values were calculated by linear regression. All the tests were performed in triplicate.

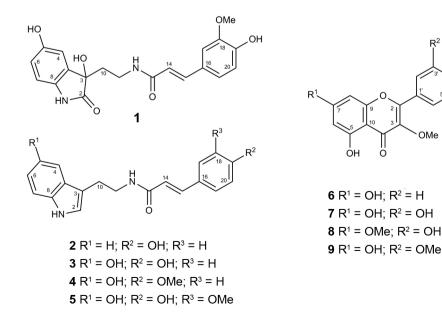
Results and Discussion

The ethanol-aqueous extract of stem bark of C. echioides was successively partitioned into hexane, dichloromethane, ethyl acetate and butanol. The resulting fractions were subjected to free radical DPPH scavenging assay, which demonstrated that the ethyl acetate fraction was the most active, with IC₅₀ = 48.9 \pm 0.2 µg mL⁻¹. In order to isolate bioactive compounds, this fraction was submitted to fractionation on a chromatographic column and the subfractions were also assaved in the DPPH. From that results, the most active subfractions were purified by chromatographic methods affording indolinone and N-substituted tryptamine alkaloid type compounds (1-5) which chemical structures are shown in Figure 1. In addition, were isolated flavonoids derived from kaempferol and quercetin (6-9), as well was benzoic acid derivatives (10-12). The structures of the isolated compounds were elucidated by analysis of their NMR spectroscopic data (including NMR 2D) and HR-ESIMS, and by comparison with those reported.

Compound 1 was isolated as a stereoisomeric mixture at C-3 position due to the absence of optical rotation

OMe

OH



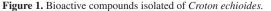


Table 1. ¹H, ¹³C NMR and HMBC data of the compound 1 in CD₃OD

deviation. The IK spectrum showed manny two strong
bands at 1711 and 1650 cm ⁻¹ corresponding to the carbonyl
stretching vibrations of (γ) lactam and NHCO, respectively.
A broad band was found at 3390 cm ⁻¹ due to N–H and OH
stretching. The positive HR-ESIMS spectrum gave the ion
peak at m/z 407.1220 [M + Na] ⁺ (calculated 407.1219),
corresponding to the molecular formula $C_{20}H_{20}N_2O_6$.
The NMR spectrum showed signals for oxindole and
<i>trans</i> -feruloyl moieties which were confirmed by ¹ H and
¹³ C NMR data assignments (Table 1). Specifically, from
the ¹ H NMR spectra three aromatic hydrogens at δ 6.87
(d, J 2.1 Hz, H-4), 6.68 (dd, J 8.4 and 2.1 Hz, H-6) and
6.72 (d, J 8.4 Hz, H-7), which evidenced the presence of
the 1,3-dihydro-3,5-dihydroxy-2 <i>H</i> -indol-2-one moiety
substituted at C-3 position by one aminoethyl unit at
3.26 (m, H-11) and δ 2.13 (t, J 7.8 Hz, H-10). This was
confirmed by the signals for a carbonyl group at δ 181.7
(C-2), a quaternary carbon at δ 77.0 (C-3), together with
the signals for the aromatic ring at 154.8 (C-5), 134.6 (C-9),
133.7 (C-8), 116.7 (C-6), 113.0 (C-4) and 112.0 (C-7), and
for an aminoethyl group at δ 38.2 (C-10) and 35.7 (C-11),
in the ¹³ C NMR spectra. The presence of a <i>trans</i> -feruloyl
group was evidenced by the signals for the aromatic system
at $\delta_{\rm H}/\delta_{\rm C}$ 7.09 (d, J 1.8 Hz, H-17)/ 111.6 (C-17), 6.78 (d,
$J = 8.4 \text{ Hz}, H=20/116.5 \text{ (C-20) and } \delta_{\text{H}}/\delta_{\text{C}} = 6.99 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/116.5 \text{ (C-20) and } \delta_{\text{H}}/\delta_{\text{C}} = 6.99 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/116.5 \text{ (C-20) and } \delta_{\text{H}}/\delta_{\text{C}} = 6.99 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/116.5 \text{ (C-20) and } \delta_{\text{H}}/\delta_{\text{C}} = 6.99 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/116.5 \text{ (C-20) and } \delta_{\text{H}}/\delta_{\text{C}} = 6.99 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/116.5 \text{ (C-20) and } \delta_{\text{H}}/\delta_{\text{C}} = 6.99 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/116.5 \text{ (dd, } J = 8.4 \text{ Hz}, H=20/1$
and 1.8 Hz, H-21)/123.2 (C-21), a <i>trans</i> -olefin system at
$\delta_{\rm H}/\delta_{\rm C}$ 7.36 (d, <i>J</i> 15.6 Hz, H-15)/142.0 (C-15) and 6.32 (d,
J 15.6 Hz, H-14)/118.7 (C-14), and for a methoxy group at $\frac{1}{2}$ ($\frac{1}{2}$)/56.4. The signals of the carbonul
at $\delta_{\rm H}/\delta_{\rm C}$ 3.87 (3H, s)/56.4. The signals of the carbonyl
group and aromatic carbons of the <i>trans</i> -feruloyl moiety
were observed at δ_c 169.2 (C-13), and at δ_c 149.8 (C-19),
149.3 (C-18) and 128.3 (C-16), respectively. The linkage of
the moieties <i>trans</i> -feruloyl and 1,3-dihydro-3,5-dihydroxy-
2 <i>H</i> -indol-2-one was confirmed by the ${}^{3}J$ correlation of the
H-15 (δ 7.36) and H-11 (δ 3.26) signals with the carbon
carbonyl group at δ 169.2 (C-13) in the HMBC spectra. In
addition, it was observed an HMBC correlation between
the methylene hydrogens at δ 2.13 (H-10) and the carbonyl
and quaternary carbons at C-2 and C-3, respectively. These
data supported the structure determination of 1 as N-trans-
feruloyl-3,5-dihydroxyindolin-2-one.
Compound 4 has previously been synthesized but no

deviation. The IR spectrum showed mainly two strong

Compound **4** has previously been synthesized, but no complete NMR assignments were reported.¹² Here, we report its isolation and characterization from a natural source.

The alkaloids (1-5), flavonoids (6-9) and benzoic acid derivatives (10-12) were evaluated for their antioxidant activity by the free scavenging DPPH method. The results for compounds 1-5, expressed by IC_{50} , are shown in Table 2. Compounds 3-5 and the flavonoids 7 and 8 showed prominent antioxidant properties, with IC_{50} values ranging from 9.2 to 17.5 µmol L⁻¹, lower than those of the

Position	$\delta_{ m H}$ mult. (J in Hz) / ppm	$\delta_{ m c}$ / ppm	HMBC
2		181.7	H-10
3		77.0	H-10, H-11
4	6.87, d (2.1)	113.0	
5		154.8	H-4, H-6, H-7
6	6.68, dd (8.4, 2.1)	116.7	H-4
7	6.72, d (8.4)	112.0	
8		133.7	H-7
9		134.6	H-4, H-10
10	2.13, t (7.8)	38.2	H-11
11	3.26, m	35.7	H-10
13		169.2	H-11, H-14, H-15
14	6.32, d (15.6)	118.7	H-15
15	7.36, d (15.6)	142.0	H-17, H-21
16		128.3	H-14, H-15, H-20
17	7.09, d (1.8)	111.6	H-15, H-21
18		149.3	-OCH ₃
19		149.8	H-17, H-20, H-21
20	6.78, d (8.4)	116.5	
21	6.99, dd (8.4, 1.8)	123.2	H-15, H-17
-OCH ₃	3.87 (s)	56.4	

HMBC: heteronuclear multiple bond correlation.

positive control Trolox (IC₅₀ = 17.9 μ mol L⁻¹). Comparison of the IC₅₀ value of compounds **2** (203.7 μ mol L⁻¹) and **3** (10.7 μ mol L⁻¹) showed that the hydroxy group at the C-5 of the tryptamine moiety is fundamental for the antioxidant activity of these indole alkaloids.

The cytotoxic activity of isolated indole alkaloids (1-5) was tested *in vitro* against the HCT-116 human cancer cell line, and the results are expressed in Table 2, with IC_{50} values ranging from 86.8 to 210.7 µmol L⁻¹. Compound **4** was the most bioactive, with an IC_{50} value of 86.8 µmol L⁻¹. According to the texted protocol, the alkaloid **1** was not bioactive ($IC_{50} > 250 \mu mol L^{-1}$), which suggests that the indole unit present in alkaloids (**2-5**) is important for the cytotoxic activity.

 Table 2. DPPH antioxidant activity and in vitro cytotoxicity against HCT-116 cells of compounds 1-5

Compound	Antioxidant activity IC_{50} / (µmol L ⁻¹)	$\begin{array}{c} Cytotoxicity\\ IC_{50}/(\mu mol\;L^{\text{-1}}) \end{array}$
1	30.0 ± 0.7	n.a.
2	203.7 ± 25.3	144.3 ± 28.6
3	10.7 ± 0.4	210.7 ± 1.4
4	17.5 ± 0.4	86.8 ± 9.7
5	14.5 ± 0.4	142.1 ± 60.5
Trolox	17.9 ± 1.1	_

n.a. = not active (IC₅₀ > 250 μ mol L⁻¹).

To our knowledge, this is the first report of indolinone alkaloid from a species of *Croton*, which is an important contribution to chemotaxonomic knowledge of this genus. Concerning the reported activities for the known compounds, the alkaloids have mainly antioxidant and tyronase inhibitory activities.²⁰⁻²² Also, anti-inflammatory activity (compounds **2** and **3**),²³ growth-promoting activity for fibroblasts (compound **3**)^{24,25} and *in vitro* cytotoxic activity against the CaCo2 colon cancer cells (compound **5**)²¹ were previously reported. This is the first report on the cytotoxicity of compounds **1-5** against the HCT-116 human cancer cell line.

Conclusions

The present report is the first chemical study of *C. echioides* Baill., despite this plant species being widely marketed as a substitution to the traditional Amazon Marapuama, *Ptychopetalum olacoides*. Five indole alkaloids were isolated and characterized by NMR data. One of them (*N-trans*-feruloyl-3,5-dihydroxyindolin-2-one) is a new natural product from *Croton* species. Their antioxidant and cytotoxicity activity were reported together with other phenolic compound types. These results represent a relevant contribution to the chemical and biological knowledge of this vegetal species and can support its correct and safe medicinal uses.

Supplementary Information

The NMR spectra of compounds 1 to 5 are available free of charge at http://jbcs.sbq.org.br as a PDF file.

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