

# Cerium oxide and iron oxide nanoparticles abolish the antibacterial activity of ciprofloxacin against gram positive and gram negative biofilm bacteria

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**Abstract** Metal oxide nanoparticles have been suggested as good candidates for the development of antibacterial agents. Cerium oxide ( $\text{CeO}_2$ ) and iron oxide ( $\text{Fe}_2\text{O}_3$ ) nanoparticles have been utilized in a number of biomedical applications. Here, the antibacterial activity of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles were evaluated on a panel of gram positive and gram negative bacteria in both the planktonic and biofilm cultures. Additionally, the effect of combining  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles with the broad spectrum antibiotic ciprofloxacin on tested bacteria was investigated. Thus, minimum inhibitory concentrations (MICs) of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles that are required to inhibit bacterial planktonic growth and bacterial biofilm, were evaluated, and were compared to the MICs of the broad spectrum antibiotic ciprofloxacin alone or in the presence of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$

nanoparticles. Results of this study show that both  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles fail to inhibit bacterial growth and biofilm biomass for all the bacterial strains tested. Moreover, adding  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles to the broad spectrum antibiotic ciprofloxacin almost abolished its antibacterial activity. Results of this study suggest that  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles are not good candidates as antibacterial agents, and they could interfere with the activity of important antibiotics.

**Keywords** Cerium oxide · Iron oxide · Nanoparticles · Antibacterial activity · Ciprofloxacin · Biofilm bacteria

## Introduction

The introduction of antimicrobial agents had a vital role in decreasing the total deaths from infectious diseases during the mid-twentieth century (Cohen 2000). However, the emergence of bacterial resistance to antibacterial drugs has become a serious problem for public health (Kurek et al. 2011). Many traditionally used antibiotics are not effective anymore in managing drug-resistant bacteria. This might lead to the re-emergence of once controlled microbial diseases (Cohen 2000). Additionally, many bacteria escape most antibiotic treatments and host defense systems by forming a protective matrix of

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exopolymeric substances called biofilm (Subbiahdoss et al. 2012; Weir et al. 2008). Moreover, the horizontal gene transfer between bacteria within biofilms can increase the spread of antibiotic resistance (Fux et al. 2005; Weir et al. 2008). Adding to the antibiotic resistance problem is the decline in the development of new antibacterial agents, with only few newly approved agents introduced to the pharmaceutical market (Donadio et al. 2010). Therefore, there is a great need to develop new antibacterial agents.

Over the last decade, many researchers have been evaluating potential antibacterial effect of metals in their nanoparticle form. Metals including zinc, silver, and copper have been used as antibacterial agents for long time (Subbiahdoss et al. 2012). The advantage of using metals in their nanoparticle form is that these particles can be prepared to have very small diameter, and therefore to have a high surface area to volume ratio. It is thought that the high surface area to volume ratios and the resultant unique chemico-physical properties of the nanoparticles could contribute to their antimicrobial activities (Huh and Kwon 2011; Pal et al. 2007). Moreover, antibacterial nanoparticles influence several structures and biological pathways found in a wide range of pathogenic bacteria. This makes it harder for bacteria to develop resistance against nanoparticles (Nel et al. 2009; Huang et al. 2008; Pal et al. 2007).

Cerium oxide ( $\text{CeO}_2$ ) nanoparticles are metal oxide nanoparticles that have been exploited in a number of biomedical applications. For example, they have been used as a UV light absorber in sunscreens (Wu et al. 2010).  $\text{CeO}_2$  nanoparticles exhibit an antioxidant activity at physiological pH, and were shown to protect cells against oxidative stress, inflammation, or damage caused by radiation (Tarnuzzer et al. 2005; Niu et al. 2007; Perez et al. 2008). Studies on the antibacterial activity of  $\text{CeO}_2$  nanoparticles have shown mixed results as well. While some studies have suggested antibacterial activity for  $\text{CeO}_2$  nanoparticles (Shah et al. 2012), others have indicated no toxic effect of  $\text{CeO}_2$  nanoparticles on bacteria (Negahdary et al. 2012; Pelletier et al. 2010; Thill et al. 2006).

Iron Oxide nanoparticles represent another nanoparticle that has been utilized in biomedical research due to its biocompatibility, ease to functionalize for many applications, and magnetic characteristics (Gupta and Gupta 2005). Clinical and experimental applications of iron oxide nanoparticles include its

usage in magnetic resonance imaging (MRI) as a contrast agent (Babes et al. 1999), magnetic fluid hyperthermia (Khandhar et al. 2012; Gonzales-Weimuller et al. 2009), targeted drug therapy as a drug carrier (Chertok et al. 2008), immunoassays, detoxification of biological fluids, tissue repair, and cell separation (Gupta and Gupta 2005; Pareta et al. 2008). Previous studies have suggested the antibacterial activity of iron oxide nanoparticles in the form of  $\text{Fe}_3\text{O}_4$  against some bacteria including *Staphylococcus aureus* and *Staphylococcus epidermidis* (Taylor and Webster 2009; Tran et al. 2010). A study by Ravikumar et al., reported antibacterial effect for  $\text{Fe}_3\text{O}_4$  nanoparticles against *Pseudomonas aeruginosa*, *Klebsiella pneumoniae* and *Streptococcus pyogenes*, but had no effect on *Escherichia coli*, *Streptococcus viridans*, *Acinetobacter* sp. and other *Klebsiella* sp. (Ravikumar et al. 2011). A recent study by Gokulakrishnan et al. investigated the antibacterial activity of  $\text{Fe}_2\text{O}_3$  nanoparticles on a number of bacteria in their planktonic forms (Gokulakrishnan et al. 2012). In the present study, antibacterial activity of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles on a larger panel of gram-positive and gram-negative bacteria in both the planktonic and biofilm cultures were evaluated. Moreover, the effect of combining  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles with the broad spectrum antibiotic ciprofloxacin on tested bacteria was investigated.

## Materials and methods

### Bacterial strains and maintenance

Bacterial strains were obtained from the American Type Culture Collection (ATCC, Manassas, VA, USA). The following microorganisms were included in this study: *E. coli* (ATCC 25922), *P. aeruginosa* (ATCC 27853), methicillin-sensitive *S. aureus* (MSSA) (ATCC 29213), methicillin-resistant *S. aureus* (MRSA) (ATCC 43300), *Streptococcus pneumoniae* (ATCC 25923), vancomycin-sensitive *Enterococcus faecalis* (VSE) (ATCC 19433), vancomycin-resistant *E. faecalis* (VRE) (ATCC 51299), *Acinetobacter baumannii* (ATCC 17978), *Proteus mirabilis* (ATCC 12459), *K. pneumoniae* (ATCC 13833), *S. pyogenes* (ATCC 19615), *Haemophilus influenzae* (ATCC 29247), *S. epidermidis* (ATCC 12228), *Enterobacter aerogenes* (ATCC 29751), *Citrobacter freundii* (ATCC 8090), and

*Enterobacter cloacae* (ATCC 13047). The organisms were stored at  $-70\text{ }^{\circ}\text{C}$  in trypticase-soy broth with 20 % glycerol (BBL Microbiology Systems, Cockeysville, Md., USA) until ready for batch susceptibility testing. They were thawed and passed 3 times to assure purity and viability.

#### Preparation of the $\text{CeO}_2$ and $\gamma\text{-Fe}_2\text{O}_3$ nanoparticles

$\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles prepared as previously described by Aljarrah et al. (2012). Briefly, equimolar amounts (0.1 M) of  $\text{Ce}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$  and  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  (Sigma Aldrich, St. Louis, MO, USA, >99 %) were added into a two separate 100 ml glass flasks containing 50 ml of 0.2 M glycine (Sigma Aldrich, 99 %). Each solution was rigorously mixed to generate a 0.1 M  $\text{Fe}^{3+}$  and 0.1 M  $\text{Ce}^{4+}$  solutions and were, then, transferred into two separate 100 ml Teflon-lined stainless steel vessels. The vessels were tightly sealed and heated to  $150\text{ }^{\circ}\text{C}$  for 10 h. They were, then, slowly cooled to room temperature. Precipitated powders were washed several times using deionized water and absolute ethanol. The precipitates were sonicated for 5 min prior to filtering, annealed at  $250\text{ }^{\circ}\text{C}$  in oxygen for 2 h, cooled to room temperature, and dried in air for 10 h.

The morphology and the microstructure of samples was observed using field emission scanning electron microscope (FE-SEM, JEOL, Peabody, MA, USA). The crystal structure of the samples was measured using an X-ray diffractometer (XRD, Shimadzu 6000, Kyoto, Japan) with  $\text{CuK}\alpha$  ( $\lambda = 1.5418\text{ \AA}$ ) radiation in the  $2\theta$  range of  $20\text{--}70^{\circ}$ . The scan rate was  $5^{\circ}/\text{min}$ .

#### Biofilm formation and screening

Bacterial biofilms were prepared as described previously (Masadeh et al. 2013; Cernohorska and Votava 2008). Briefly, 100  $\mu\text{l}$  of bacterial suspension from each of the bacterial strains tested were cultivated in polypropylene tubes containing 2 ml of Trypticase Soy Broth (TSB) supplemented with 1 % glucose for 48 h at  $37\text{ }^{\circ}\text{C}$ . Culture media was refreshed after 24 h of incubation. In order to screen for biofilm formation, some of the cultivated tubes were stained as described previously (Christensen et al. 1985). Briefly, after being emptied from their content, culture tubes were stained with trypan blue or safranin. Biofilms were judged by the appearance of a visible film that lined the walls of the tube. Observations were carried out by

three independent observers. Biofilms were scored as absent (score 0), weak (score 1), moderate (score 2), or strong (score 3). The average scores were used.

#### Minimum inhibitory concentration (MIC)

The MICs of the broad spectrum antibiotic ciprofloxacin,  $\text{CeO}_2$  nanoparticles,  $\text{Fe}_2\text{O}_3$  nanoparticles, ciprofloxacin mixed with  $\text{CeO}_2$  nanoparticles, and ciprofloxacin mixed with  $\text{Fe}_2\text{O}_3$  nanoparticles were evaluated. MICs were determined using broth macrodilution method according to the Clinical and Laboratory Standards Institute guidelines (Clinical and Laboratory Standards Institute (CLSI) 2012). Aliquots of 10  $\mu\text{l}$  from each bacterial strain tested were inoculated in 10 ml of Muller-Hinton Broth (MHB), and incubated for 24 h at  $37\text{ }^{\circ}\text{C}$ . After 24 h, bacterial suspensions from each bacterial strain tested were adjusted to 0.5 McFarland ( $1.5 \times 10^8$  colony forming units (CFU)/ml).

Five different samples were performed in aqueous solution: bacterial suspension, ciprofloxacin suspension,  $\text{CeO}_2$  nanoparticles suspension,  $\text{Fe}_2\text{O}_3$  nanoparticles suspension, ciprofloxacin mixed with  $\text{CeO}_2$  nanoparticles, and ciprofloxacin mixed with  $\text{Fe}_2\text{O}_3$  nanoparticles. Nanoparticles samples were sonicated for 1 h to get a well-dispersed and clear suspension. All samples were incubated for 1 h at  $37\text{ }^{\circ}\text{C}$  with gentle shaking every 15 min prior to biofilm cultures. Adjusted bacterial suspensions (100  $\mu\text{l}$ ) were added to ciprofloxacin,  $\text{CeO}_2$  nanoparticles,  $\text{Fe}_2\text{O}_3$  nanoparticles, ciprofloxacin mixed with  $\text{CeO}_2$  nanoparticles, and ciprofloxacin mixed with  $\text{Fe}_2\text{O}_3$  nanoparticles. Bacterial suspensions were incubated at  $37\text{ }^{\circ}\text{C}$  for 24 h for planktonic cultures, and for 48 h for biofilm cultures as described above. To determine the MIC of ciprofloxacin treatment, the same amount of adjusted bacterial suspension (0.5 McFarland) from each examined bacterial strain were added to serial dilutions of ciprofloxacin. Ciprofloxacin concentrations from 0.015 to 0.96  $\mu\text{g}/\text{ml}$  were tested. To determine the MIC of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles treatment, adjusted bacterial suspension (0.5 McFarland) from each examined bacterial strain, were added to twofold serial dilutions of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles.  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles concentrations from 8.25 to 528  $\mu\text{g}/\text{ml}$  were tested. Similarly, to determine the MIC of ciprofloxacin in the presence of  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles adjusted bacterial suspension (0.5 McFarland) from each examined bacterial strain were

added to serial dilutions of ciprofloxacin that contains either  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles. MICs were determined as the lowest concentration of ciprofloxacin or nanoparticles at which there was no growth, a faint haze or fewer than 3 discrete colonies. Tubes of bacterial suspensions without nanoparticles served as control. As a particle control, tubes of  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles solutions were added to tubes containing only MHB at the same concentrations that were used above. Plates were read in triplicate and the higher MIC value was recorded (Clinical and Laboratory Standards Institute (CLSI) 2012).

### Statistical analysis

Analysis was performed using GraphPad Prism software (version 4.0, GraphPad software, La Jolla, CA, USA). One-way ANOVA followed by Dunnett's post-test were used to determine significant difference.  $P$  values  $<0.05$  was considered significant.

## Results

### Nanoparticles synthesis and characterization

Figure 1a shows X-ray powder diffraction patterns of the prepared  $\gamma\text{-Fe}_2\text{O}_3$  nanoparticles. The X-ray powder diffraction patterns of the material proved its crystalline nature and all the peaks matched well with standard  $\gamma\text{-Fe}_2\text{O}_3$  reflections. The sharpness of XRD peaks reveals high crystallinity of the nanoparticles. No traces of other phases have been detected in the pattern. SEM image of prepared  $\gamma\text{-Fe}_2\text{O}_3$  (Fig. 1b) indicates the presence of spherical-shaped nanoparticles. The grain boundaries are clean and round with no presence of other phases or salts in microstructure. The mean size of the particles varies from 40 to 50 nm. Figure 2a, b shows the XRD and SEM data for the  $\text{CeO}_2$  nanoparticles, respectively. Similarly, XRD peaks show quite high degree of crystallinity of the nanoparticles. No traces of other phases have been detected in the pattern. The SEM shows the presence of spherical-shaped  $\text{CeO}_2$  nanoparticles of homogeneous morphology with a grain size from 25 to 50 nm. However, traces of salt washing residues with smaller nanoparticle size are present between the grains and on the grain boundaries of  $\text{CeO}_2$  nanoparticles.

It has been noticed that the pH values of the colloidal solutions plays an important role in the precipitation process and was controlled before and after the hydrothermal process. However, in many cases (especially for the Fe colloidal solution), it is not easy to precipitate specific iron oxide particles directly in the desired size and shape. The size and shape of the nanoparticles can be tailored with relative success by adjusting pH, ionic strength, temperature, nature of the salts (chlorides, sulfates, and nitrates), or the  $\text{Fe}^{2+}/\text{Fe}^{3+}$  concentration ratio (Issa et al. 2013). Moreover, based on the careful handling of the hydrothermal process carried out in the current study, we believe that final annealing in oxygen for resultant iron oxide powder facilitates oxidation of  $\text{Fe}_3\text{O}_4$  to  $\gamma\text{-Fe}_2\text{O}_3$  and produce a monodisperse, porous and magnetic  $\gamma\text{-Fe}_2\text{O}_3$  nanoparticles. This was supported by the X-ray and the SEM data shown in Figs. 1 and 2.

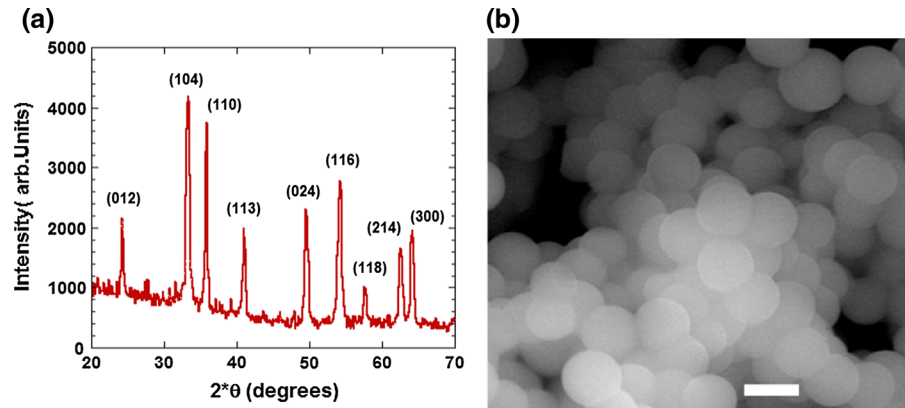
$\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles have no inhibitory effect on a panel of planktonic gram positive and gram negative bacteria

In order to assess the antibacterial effect of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles against planktonic bacteria, the MICs of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles were evaluated on a panel of planktonic gram positive and gram negative bacteria. Both  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles showed no antibacterial effect against all planktonic bacterial strains that were tested (Table 1). MICs of ciprofloxacin against all the bacterial strains tested were used as control. Ciprofloxacin treatment inhibited the planktonic bacterial growth of all selected strains (Table 1). Further controls included bacterial suspensions without nanoparticles, as well as,  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles suspensions without bacteria.

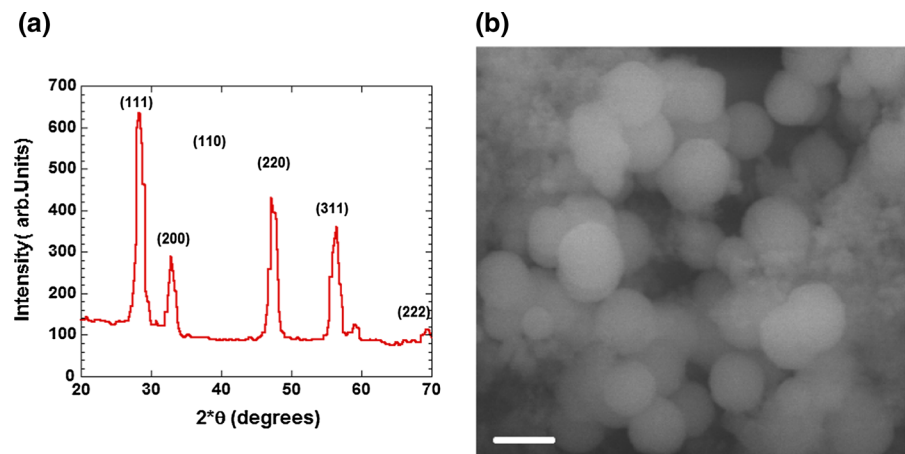
$\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles have no inhibitory effect on the formation of a panel of gram positive and gram negative bacterial biofilms

One way of bacteria to resist aggressive antibiotic treatment and protect themselves against the host immune system is by forming biofilm. Biofilm is a matrix of exopolymeric substances that is impenetrable by most antibiotics and immune cells (Subbiahdoss et al. 2012). Metal nanoparticles, including zinc oxide and selenium nanoparticles, have been

**Fig. 1** **a** XRD of  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> nanoparticles, all major diffraction peaks are indexed, **b** SEM of  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> nanoparticles with a scale bar of 50 nm



**Fig. 2** **a** XRD of CeO<sub>2</sub> nanoparticles, all major diffraction peaks are indexed, **b** SEM of CeO<sub>2</sub> nanoparticles with a scale bar of 50 nm



suggested to possess characteristics that enable them to inhibit bacterial biofilm formation (Applerot et al. 2012; Wang and Webster 2013). In order to examine the effect of CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles on bacterial biofilms biomass, MICs of CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles were evaluated on a panel of gram positive and gram negative bacterial biofilms, including antibiotic resistant strains. Bacterial suspensions were added to two-fold serial dilutions of CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles, and suspensions were incubated for 48 h at 37 °C. Bacterial biofilm formation of nanoparticles treated bacteria was compared with biofilm formation in untreated bacterial suspensions. CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles showed no antibacterial effect on biofilm biomass of all tested bacterial strains (Table 2). The MICs of ciprofloxacin were used as control and showed significant inhibitory effect on biofilm biomass of all tested bacterial strains (Table 2). Further controls included CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles suspensions without bacteria.

CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles dramatically reduce antibacterial effect of ciprofloxacin against a panel of gram positive and gram negative planktonic and biofilm bacterial cultures

In order to evaluate whether combining either CeO<sub>2</sub> or Fe<sub>2</sub>O<sub>3</sub> nanoparticles with ciprofloxacin would influence the antibacterial effect of ciprofloxacin against the tested bacterial strains, the effect of CeO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub> nanoparticles combined with ciprofloxacin was examined on bacterial growth. The MICs of ciprofloxacin in the presence of either CeO<sub>2</sub> or Fe<sub>2</sub>O<sub>3</sub> nanoparticles against planktonic and biofilm cultures of tested gram positive and gram negative bacteria were examined. Bacterial suspensions were added to two-fold serial dilutions of ciprofloxacin that contain either CeO<sub>2</sub> or Fe<sub>2</sub>O<sub>3</sub> nanoparticles. Bacterial cultures were allowed to grow to planktonic cultures, or were left to form biofilm cultures and the MICs of ciprofloxacin were assessed. Interestingly, the



**Table 1** A comparison between the minimum inhibitory concentrations (MIC;  $\mu\text{g/mL}$ ) of ciprofloxacin,  $\text{CeO}_2$  nanoparticles,  $\text{Fe}_2\text{O}_3$  nanoparticles, and ciprofloxacin in the presence of  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles against planktonic gram positive and gram negative bacterial cultures

	Ciprofloxacin MIC ( $\mu\text{g/ml}$ )	$\text{CeO}_2$ NP MIC ( $\mu\text{g/ml}$ )	$\text{Fe}_2\text{O}_3$ NP MIC ( $\mu\text{g/ml}$ )	Ciprofloxacin + $\text{CeO}_2$ NP MIC ( $\mu\text{g/ml}$ )	Ciprofloxacin + $\text{Fe}_2\text{O}_3$ NP MIC ( $\mu\text{g/ml}$ )
Gram +ve planktonic bacterial cultures					
<i>MSSA</i>	0.10 $\pm$ 0.04	50 $\pm$ 20	20 $\pm$ 0.0	50 $\pm$ 20	30 $\pm$ 10
<i>MRSA</i>	0.40 $\pm$ 0.20	70 $\pm$ 0.0	60 $\pm$ 20	30 $\pm$ 10	50 $\pm$ 20
<i>S. pneumonia</i>	0.40 $\pm$ 0.20	110 $\pm$ 40	60 $\pm$ 20	70 $\pm$ 0.0	50 $\pm$ 20
<i>VSE</i>	0.40 $\pm$ 0.20	50 $\pm$ 20	30 $\pm$ 10	20 $\pm$ 0.0	30 $\pm$ 10
<i>VRE</i>	0.40 $\pm$ 0.20	110 $\pm$ 40	70 $\pm$ 0.0	60 $\pm$ 20	60 $\pm$ 20
<i>S. pyogenes</i>	0.20 $\pm$ 0.07	30 $\pm$ 10	50 $\pm$ 20	30 $\pm$ 0.0	30 $\pm$ 10
<i>S. epidermidis</i>	0.20 $\pm$ 0.07	20 $\pm$ 5	40 $\pm$ 0.0	10 $\pm$ 0.0	20 $\pm$ 5
Gram -ve planktonic bacterial cultures					
<i>E. coli</i>	0.03 $\pm$ 0.01	50 $\pm$ 10	20 $\pm$ 5	30 $\pm$ 10	20 $\pm$ 5
<i>P. aeruginosa</i>	0.08 $\pm$ 0.04	20 $\pm$ 5	50 $\pm$ 20	30 $\pm$ 10	20 $\pm$ 5
<i>A. baumannii</i>	0.20 $\pm$ 0.07	70 $\pm$ 0.0	110 $\pm$ 40	40 $\pm$ 0.0	60 $\pm$ 20
<i>P. mirabilis</i>	0.20 $\pm$ 0.07	30 $\pm$ 10	220 $\pm$ 80	8 $\pm$ 0.0	140 $\pm$ 0.0
<i>K. pneumonia</i>	0.70 $\pm$ 0.30	140 $\pm$ 0.0	110 $\pm$ 40	70 $\pm$ 0.0	50 $\pm$ 20
<i>H. influenzae</i>	0.40 $\pm$ 0.20	360 $\pm$ 160	360 $\pm$ 160	110 $\pm$ 40	60 $\pm$ 20
<i>E. aerogenes</i>	0.70 $\pm$ 0.30	70 $\pm$ 0.0	60 $\pm$ 20	40 $\pm$ 0.0	30 $\pm$ 10
<i>C. freundii</i>	0.40 $\pm$ 0.20	70 $\pm$ 0.0	40 $\pm$ 0.0	50 $\pm$ 20	20 $\pm$ 0.0
<i>E. cloacae</i>	0.40 $\pm$ 0.20	70 $\pm$ 0.0	50 $\pm$ 20	50 $\pm$ 20	30 $\pm$ 10

The MIC values for ciprofloxacin alone were significantly ( $P < 0.05$ ) lower than those of  $\text{CeO}_2$  NP or  $\text{Fe}_2\text{O}_3$  NP alone, Ciprofloxacin +  $\text{CeO}_2$  NP, and Ciprofloxacin +  $\text{Fe}_2\text{O}_3$  NP for all tested bacterial strains. NP stands for nanoparticles. Results are presented as mean  $\pm$  SD of 3 independent experiments

addition of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles dramatically reduced the antibacterial activity of ciprofloxacin against all bacterial strains tested whether these bacteria were grown as planktonic (Table 1) or biofilm cultures (Table 2). These results combined with those discussed above suggest that  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles treatment not only fails to provide antibacterial effect, but also inhibits the antibacterial activity of ciprofloxacin. The MICs of ciprofloxacin only treated bacterial suspensions were used as control and showed significant inhibitory effect against all planktonic and biofilm cultures of tested bacterial strains. Further controls include bacterial suspensions without  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles, as well as,  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles suspensions without bacteria.

## Discussion

Metal oxide nanoparticles have been suggested as an important candidate for tackling the healthcare

problem of increasing number of antibiotic resistant and biofilm forming bacteria. In this study, two metal oxide nanoparticles,  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles, were tested for efficacy as antibacterial agents against a list of gram positive and gram negative bacteria. This list of bacteria included strains that are known to be antibiotic resistant such as MRSA and VRE. Our study has shown that  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles failed to inhibit bacterial planktonic growth and biofilm formation as compared to bacterial growth inhibition resulting from ciprofloxacin treatment. Ciprofloxacin, however, inhibited the growth of all gram positive and gram negative bacteria tested.

A related interesting finding of the current study is that combining  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles with ciprofloxacin reduced significantly the antibacterial effect of ciprofloxacin. Ciprofloxacin is a second-generation fluoroquinolone antibacterial agent (Drlica and Zhao 1997). It kills bacteria by inhibiting DNA gyrase, and topoisomerase IV enzymes that are necessary to separate bacterial DNA, therefore,

**Table 2** A comparison between the minimum inhibitory concentrations (MIC;  $\mu\text{g}/\text{mL}$ ) of ciprofloxacin,  $\text{CeO}_2$  nanoparticles,  $\text{Fe}_2\text{O}_3$  nanoparticles and ciprofloxacin in the presence of  $\text{CeO}_2$  or  $\text{Fe}_2\text{O}_3$  nanoparticles against gram positive and gram negative bacterial biofilm cultures

	Ciprofloxacin MIC ( $\mu\text{g}/\text{ml}$ )	$\text{CeO}_2$ NP MIC ( $\mu\text{g}/\text{ml}$ )	$\text{Fe}_2\text{O}_3$ NP MIC ( $\mu\text{g}/\text{ml}$ )	Ciprofloxacin + $\text{CeO}_2$ NP MIC ( $\mu\text{g}/\text{ml}$ )	Ciprofloxacin + $\text{Fe}_2\text{O}_3$ NP MIC ( $\mu\text{g}/\text{ml}$ )
Gram +ve bacteria biofilm cultures					
<i>MSSA</i>	0.70 $\pm$ 0.30	180 $\pm$ 80	30 $\pm$ 10	50 $\pm$ 20	30 $\pm$ 10
<i>MRSA</i>	0.80 $\pm$ 0.30	180 $\pm$ 80	360 $\pm$ 160	90 $\pm$ 40	180 $\pm$ 80
<i>S. pneumonia</i>	1.0 $\pm$ 0.00	180 $\pm$ 80	180 $\pm$ 80	70 $\pm$ 0.0	90 $\pm$ 40
<i>VSE</i>	0.70 $\pm$ 0.30	270 $\pm$ 0.0	360 $\pm$ 160	140 $\pm$ 0.0	90 $\pm$ 40
<i>VRE</i>	1.0 $\pm$ 0.0	530 $\pm$ 0.0	530 $\pm$ 0.0	110 $\pm$ 40	360 $\pm$ 160
<i>S. pyogenes</i>	0.40 $\pm$ 0.20	70 $\pm$ 0.0	110 $\pm$ 40	20 $\pm$ 0.0	40 $\pm$ 0.0
<i>S. epidermidis</i>	0.40 $\pm$ 0.20	90 $\pm$ 40	270 $\pm$ 0.0	50 $\pm$ 20	180 $\pm$ 80
Gram -ve bacteria biofilm cultures					
<i>E. coli</i>	0.40 $\pm$ 0.20	90 $\pm$ 40	50 $\pm$ 20	40 $\pm$ 0.0	20 $\pm$ 5
<i>P. aeruginosa</i>	0.80 $\pm$ 0.30	70 $\pm$ 0.0	100 $\pm$ 50	30 $\pm$ 10	60 $\pm$ 20
<i>A. baumannii</i>	1.0 $\pm$ 0.0	360 $\pm$ 160	360 $\pm$ 160	360 $\pm$ 160	90 $\pm$ 40
<i>P. mirabilis</i>	0.70 $\pm$ 0.30	360 $\pm$ 160	530 $\pm$ 0.0	140 $\pm$ 0.0	110 $\pm$ 40
<i>K. pneumonia</i>	0.70 $\pm$ 0.30	360 $\pm$ 160	360 $\pm$ 160	50 $\pm$ 20	90 $\pm$ 40
<i>H. influenzae</i>	1.0 $\pm$ 0.0	530 $\pm$ 0.0	530 $\pm$ 0.0	140 $\pm$ 0.0	180 $\pm$ 80
<i>E. aerogenes</i>	1.0 $\pm$ 0.0	140 $\pm$ 0.0	110 $\pm$ 40	20 $\pm$ 0.0	50 $\pm$ 20
<i>C. freundii</i>	0.50 $\pm$ 0.00	220 $\pm$ 80	360 $\pm$ 160	180 $\pm$ 80	90 $\pm$ 40
<i>E. cloacae</i>	0.80 $\pm$ 0.30	220 $\pm$ 80	110 $\pm$ 40	110 $\pm$ 40	60 $\pm$ 20

The MIC values for ciprofloxacin alone were significantly ( $P < 0.05$ ) lower than those of  $\text{CeO}_2$  NP or  $\text{Fe}_2\text{O}_3$  NP alone, Ciprofloxacin +  $\text{CeO}_2$  NP, and Ciprofloxacin +  $\text{Fe}_2\text{O}_3$  NP for all tested bacterial strains. NP stands for nanoparticles. Results are presented as mean  $\pm$  SD of 3 independent experiments

inhibiting cell division (Drlica and Zhao 1997). It is possible that these nanoparticles interact with ciprofloxacin in a way that prevents its absorption by the bacterial cell. Another possibility is that  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles interact directly or indirectly with ciprofloxacin in a way that interferes with ciprofloxacin activity on bacterial DNA inside the bacterial cell. Interestingly, studies have shown that the bioavailability of ciprofloxacin is reduced by 50 % when co-administered with iron compounds (Lode 2001). Characterization of suggested mechanisms for the interaction of  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles with ciprofloxacin is a warranted future study.

Studies that evaluated the antibacterial effect of  $\text{CeO}_2$  nanoparticles are limited. Some of these studies suggest antibacterial effect for  $\text{CeO}_2$  nanoparticles, while other studies show no toxic or inhibitory effect for  $\text{CeO}_2$  nanoparticles against bacteria. Thill et al., Pelletier et al., and Kuang et al., have suggested antimicrobial activity of  $\text{CeO}_2$  nanoparticles against *E. coli*, whereas Shah et al., observed that dextran

coated  $\text{CeO}_2$  nanoparticles are non-lethal to *E. coli* under various experimental conditions examined (Kuang et al. 2011; Pelletier et al. 2010; Shah et al. 2012; Thill et al. 2006). Moreover Shah et al., reported that  $\text{CeO}_2$  nanoparticles can reduce magnesium and potassium salts antibacterial activity, which is similar to the observation in the current study where  $\text{CeO}_2$  nanoparticles almost abolished the antibacterial activity of ciprofloxacin (Shah et al. 2012). In addition,  $\text{CeO}_2$  nanoparticles have been shown to inhibit *Bacillus subtilis*, but have no inhibitory effect on *Shewanella oneidensis* (Pelletier et al. 2010). These studies utilized different synthesis methods, used  $\text{CeO}_2$  nanoparticles of various sizes, and exploited different methods to evaluate  $\text{CeO}_2$  nanoparticles antibacterial effect. It has been suggested that a change in the physical and chemical environment can significantly influence nanoparticles bacterial toxicity (Deshpande et al. 2005; Rispoli et al. 2010).

Although several studies have focused on evaluating the antibacterial effect of nanosized magnetic iron

oxide particle magnetite ( $\text{Fe}_3\text{O}_4$ ), much less is known about the antibacterial effect of the other type of magnetic iron oxide nanoparticles maghemite ( $\text{Fe}_2\text{O}_3$ ) (Ravikumar et al. 2011; Taylor and Webster 2009; Tran et al. 2010). One study has reported antibacterial activity of  $\text{Fe}_2\text{O}_3$  nanoparticles on a number of bacteria in their planktonic forms (Gokulakrishnan et al. 2012). Another study by He et al., found no inhibitory effect of  $\text{Fe}_2\text{O}_3$  nanoparticles on the growth of *E. coli*. In contrast, the results of the mentioned study suggest an increase in bacterial growth upon  $\text{Fe}_2\text{O}_3$  nanoparticles treatment (He et al. 2011). In the current study,  $\text{Fe}_2\text{O}_3$  nanoparticles showed no inhibitory effect on bacterial growth and biofilm forms of all the bacterial strains tested.

In this study, both  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles were tested at a wide range of serial twofold concentrations, which is a standard procedure to estimate MIC value for compounds with previously unknown antibacterial activity (Clinical and Laboratory Standards Institute (CLSI) 2012). Future work about the possible antibacterial activity of these nanoparticles should be targeted toward studying concentrations that are above their MIC values.

In conclusion, the current study provides evidence that  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles fail to inhibit bacterial planktonic growth and biofilm biomass for all examined gram positive and gram negative bacterial strains. Moreover,  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles when combined with the broad spectrum antibiotic ciprofloxacin almost abolished its inhibitory effect on bacterial growth and biofilm formation. Therefore, this study suggests that  $\text{CeO}_2$  and  $\text{Fe}_2\text{O}_3$  nanoparticles are not good candidates as antibacterial agents.

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