

Changes in the composition of phenolic compounds and antioxidant properties of grapevine roots and leaves (*Vitis vinifera* L.) under continuous of long-term drought stress

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Abstract Grapevine seedlings *Vitis vinifera* L. were grown in a greenhouse under optimum conditions (soil moisture ca 70 %) and under drought stress (soil moisture ca 30 %). Drought stress caused reduction in total phenolic compounds in grapevine leaves and roots, where were identified tree phenolic acids: caffeic acid, *p*-coumaric acid and ferulic acid. All acids found in leaves and roots occurred in the ester-bound form. Only caffeic acid in leaves appeared in the free and ester-bound form. Caffeic acid was present in the highest concentrations. The content of ferulic acid was the lowest in both tissues. The levels of all phenolic acids in leaves and roots decreased significantly under the drought stress. All the extracts from grapevine leaves and roots had antioxidative properties, but the antiradical activity of the extracts obtained from roots subjected to drought stress was lower to the control. The results of the analysis revealed that long-term drought stress caused a decrease in selected elements of secondary metabolism in such a different plant tissues that are the leaves and roots of the grapevine.

Keywords *Vitis vinifera* L. · Antioxidant system · Drought stress · Phenolic compounds · Phenolic acids

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Introduction

According to estimated data, a 70 % increase in crop yield could be achieved if the environmental conditions were close to optimum for a given plant. That is why identifying and managing the adverse environmental effects is a top priority in many countries in the world (Chaves and Oliveira 2004). The main factors that restrain the plant distribution are environmental stresses, such as drought, low or high temperature and excessive salinity. These abiotic stress factors generate secondary stresses, i.e. osmotic and oxidative stress, which have negative influence on the plant, causing changes in its normal growth, development and metabolism (Bohenert et al. 1995; Kranner et al. 2010). Thus, it is necessary to conduct research focused on obtaining cultivars, which would be more tolerant to adverse environmental factors. Water stress is one of these environmental factors that can considerably limit distribution of crops in the world (Passioura 2007; Cattivelli et al. 2008; Farooq et al. 2009).

Drought stress decreases water absorption in plants. As a result of membrane phase transition from liquid crystalline structure alters to less liquid jelly structure, selective membrane transport is disturbed (Larcher 1995; Bray 2009). Then in turn, destabilization of nucleic acid structure and alteration of enzyme activity may be observed (Lee et al. 2009; Dumont et al. 2011). A plant's ability to function in the environment, that is continuous source of stress, depends in large extend on their genetic programme (Dodd et al. 2006). Stresses cause the disturbance in cellular redox homeostasis, which in turn, may lead to secondary oxidative stress and production of reaction oxygen species (ROS) (Asada 2006). ROS fulfil the roles of signalling molecules in order to initiate synthesis of enzyme (catalase, peroxidase, superoxide dismutase) as well as the

nonenzymatic molecules (glutathione, α -tocopherol, carotenoids) that take part in antioxidant reactions. However, during times of environmental stress, ROS levels can increase dramatically. This may result in significant damage to cell structures including nucleic acid and proteins (Smirnof 1993; Aroca et al. 2003; Murata et al. 2012). Consequently, alterations in cellular genetic programming occur. The synthesis of compounds will help create a new homeostasis (Gilmour et al. 2000). Secondary metabolic products, which are intensively synthesized under drought, are antioxidants (Nascimento and Fett-Neto 2010). This group of compounds protects cells against the negative effects of ROS as well as against lipid peroxidation, protein denaturation and DNA damage (Kranner et al. 2002; Mittler 2002; Allakhverdiev et al. 2008). In turn, environmental stress can cause a decline (Weidner et al. 2007, 2009b) or an increase (Wróbel et al. 2005; Weidner et al. 2009a) in the content of phenolic compounds in a cell. Phenolic compounds can scavenge ROS (Amarowicz et al. 2000, 2004; Negro et al. 2003; Caillet et al. 2006; Amarowicz and Weidner 2009), they form complexes with the metals, which catalyse oxygenation reactions and inhibit activity of oxidizing enzymes (Elavarthi and Martin 2010). Phenolic compounds include many secondary metabolites in plants that exhibit antioxidant properties (Oszmański 1995). Phenols constitute a large group of compounds that may be divided into five subgroups: coumarins, lignins, flavonoids, phenolic acids and tannins (Gumul et al. 2007). Precursors for the synthesis of phenolic compounds are made in the shikimic acid and chorismic acid pathways. The metabolism of chorismic acid leads to L-phenylalanine and L-tyrosine. Tyrosine biosynthesis leads to *p*-coumaric acid and L-phenylalanine synthesis leads to cinnamic acid. As a result of the processes of methylation, hydration and dehydration of cinnamic acid, phenolic acids are produced, as part of the response of the plant to abiotic stresses (Dixon and Paiva 1995). Some of the phenolic compounds, such as phenolic acids or flavonoids, are widely known and present in most of the plant species (Jwa et al. 2006). Phenolic compounds were considered as by-products of metabolic alteration (Solecka 1997). The results investigations unequivocally suggest that phenols not only play a major role in defensive reactions of plants but also influence humans and animals that consume products enriched with phenolic compounds (Franca et al. 2001; Amarowicz and Weidner 2009).

In the literature, the problem of adverse effects on plant growth under the influence of water deficit is often claimed. However, there is little research on the long-term effects of drought on selected elements of secondary metabolism in grapevine leaves and roots.

The aim of the study has been to investigate the quantitative and qualitative modifications of phenolic compounds

and to study the antioxidative activity and the reducing power of extracts of grapevine roots and leaves, which occur under long-term drought stress. The main aim of the study was to determine the changes in the content of phenolic acids in the tissues treated with drought.

Materials and methods

Plant material

The material for the study consisted of whole root systems and leaves of grapevine *Vitis vinifera* L., cultivar Kiszmisz Łuczystyj. The seedlings of grapevine for testing were purchased from the company “professional grapevine seedlings” in Józefosław (Poland).

Experimental conditions

The seedling were transferred into large pots and grown for the next 8 weeks in a greenhouse under optimal conditions (soil moisture ca 70 %, 25/18 °C day/night cycle and 14-h photoperiod at 180–200 $\mu\text{m m}^{-2} \text{s}^{-1}$ irradiance). During the whole period, the plants were regularly watered to maintain the optimum soil moisture. The soil moisture was monitored using a VP-PRL-Nr 279462 moisture meter (Mera). When the plants adapted to greenhouse conditions had reached proper size, the experiment began. The seedlings were divided into two groups: control (C) and drought stress (S). The experimental conditions were as follows:

1. Control sample (C)—plants were grown for 2 weeks in optimum conditions established in greenhouses (soil moisture content at the level of about 70 %). Optimum soil moisture content was guaranteed by systematic watering.
2. Drought stress sample (S)—plants were grown for 2 weeks under drought stress. In the first week, the moisture level was reduced from 70 to 30 % and was maintained at 30 % for the next week. After 2 weeks, both leaf cuttings (only fully developed leaves) and root seedlings (the whole root system) were cut and cleaned from soil by rinsing with room temperature water. Next, all samples were frozen in liquid nitrogen, lyophilized and fragmented in mixer mill. The samples were stored at -20 °C until the time of analysis.

Extraction of phenolic compounds

Phenolic compounds were extracted from plant material three times into 80 % (v/v) acetone and then three times into 80 % (v/v) methanol for 15 min at 80 °C

(Amarowicz et al. 2000). Extraction was carried out in a shaking water bath. The solutions from all extractions were combined. After evaporating the organic solvent in a rotary evaporator at 45 °C, the remaining aqueous solution was lyophilized. Finally, the prepared crude extracts were stored at –20 °C in the dark until further analysis.

Determination of total phenols

The content of total phenolic compounds in the extracts was determined using the Folin–Ciocalteu's reagent (Naczka and Shahidi 1989), and (+)-catechin was used as a standard.

Determination of antioxidant activity of extracts by two distinct methods

1. DPPH radical scavenging activity of the extracts.

A fresh solution of DPPH• free radical is purple in colour and reaches maximum absorbance at wavelength 517 nm. The colour of the solution resulting from the presence of free radical disappears under the effect of antioxidative compounds, which extinguish DPPH•. When a free radical is bound to an antioxidant, a colourless product appears. In consequence, the solution becomes colourless and its absorbance is lower.

The antiradical activity was analysed with the method described by Yen and Chen (1995). Briefly, 0.1–0.5 mg of an extract dissolved in 0.1 cm³ of methanol was added to 2 cm³ of methanol; then, 0.25 cm³ of a DPPH (2,2-diphenyl-1-picrylhydrazyl) solution at a concentration of 1 mg cm⁻³ methanol was added. The mixture was left in the dark at room temperature for 20 min. After that time, absorbance was read at a wavelength of 517 nm. The analysis was performed in three replications.

2. Trolox equivalent antioxidant capacity (TEAC).

The antiradical activity also was analysed with the method described by Huang et al. (2005). 0.1–0.5 mg of raw extract was dissolved in 0.1 cm³ methanol, and then 2 cm³ ABTS⁺ radical (2,2'-azinobis(3-ethylbenzthiazoline-6-sulphonic acid) diammonium salt) was added. The mixture was left in the dark at 32 °C for 20 min. After incubation, the absorption was quantified at 734 nm. Determination of the calibration curve was obtained with Trolox as a standard. A colour solution of ABTS⁺ becomes colourless when sample has been added to it. The rate at which the solution loses its colour depends on the amount of added extract and its free radical scavenging activity.

Determination of the reducing power of extract

Reducing power of extracts was measured using the FRAP method (ferric ion reducing antioxidant power) described by Benzie and Strain (1999). 0.1–0.5 mg of raw extract was dissolved in 0.1 cm³ water, and then 3 cm³ FRAP reagent was added (acetic acid buffer of pH 3.6 TPTZ (2,4,6-tris(2-pyridyl)-s-triazine). After incubation, the absorption was quantified at 593 nm. Determination of the calibration curve was obtained with Trolox as a standard.

Separation and identification of free, ester-bound and glycoside-bound phenolic acids

Phenolic acids (free and those liberated from soluble esters and from soluble glycosides) were isolated from the extracts according to the method previously described by Zadernowski and Kozłowska (1983) and modified by Weidner et al. (2000). An aqueous suspension of the methanolic extract (400 mg in 10 cm³) was adjusted to pH 2 with 6 M HCl, and free phenolic acids were extracted five times into 10 cm³ diethyl ether using a separatory funnel. The ether extract was evaporated to dryness under vacuum at room temperature. The water solution was neutralized and then lyophilized. The residue was dissolved in 10 cm³ of 2 M NaOH and hydrolysed for 3.5 h under a nitrogen atmosphere at room temperature. After acidification to pH 2 using 6 M HCl, phenolic acids released from soluble esters were extracted from the hydrolysate five times into 15 cm³ diethyl ether. 9 cm³ of 6 M HCl was added to the water solution, which was then placed in nitrogen atmosphere and hydrolysed for 1 h at 100 °C. Phenolic acids released from soluble glycosides were separated from the hydrolysate five times into 20 cm³ diethyl ether. After ether evaporation, the dry residue was dissolved in 2 ml methanol and filtered through a 0.45-µm nylon filter. The sample was injected onto an HPLC column. A Shimadzu HPLC system was employed: LC-10ADVP pump, photodiode array detector UV–VIS SPD-M10AVP, oven CTO-10ASVP, controller SCL-10AVP. The conditions of the separations were as follows: pre-packed LUNA C₁₈ column (5 µm, 4.6 × 250 mm; Phenomenex); mobile phase water–acetonitrile–acetic acid (88:10:2, v/v/v) (Amarowicz and Weidner 2001); flow rate of 1 cm³ min⁻¹; injection volume of 20 µl; the detector was set at 280 and 320 nm; oven temperature was 20 °C.

Statistical analysis

All experiments were repeated four times. The results are reported as the mean ± SD. Statistically significant differences between the mean values were tested by Student's *t* test.

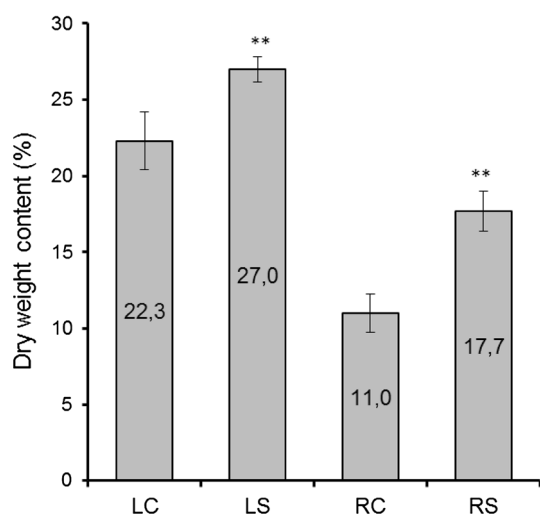


Fig. 1 Percentage content of dry matter in control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in “Materials and methods”. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.01$ (double asterisks)

Results

Dry matter content

The changes of percentage dry matter in grapevine leaves and roots are illustrated in Fig. 1. Grapevine leaves (both LC and LS) have significantly higher dry matter content than grapevine roots (RC and RS). The results show that dry matter in leaves of grapevine seedlings subjected to drought stress (LS) was significantly higher (26.24 %) compared to the control samples of leaves (LC), in which the dry matter content was at the level of 22.31 % (Fig. 1). The same trend was noted in roots, where the dry matter of the stress samples of roots (RS) content was at the level of 17.74 % and was significantly higher than in control samples (RC) (10.98 %).

Content of phenolic compounds

In Table 1 are reported content of phenolic compounds of all samples recalculated per gram fresh matter, dry matter of tissues and per mg of extract. The results show also that leaves have a lower content of phenolic compounds than roots. Moreover, the content of total phenolic compounds in the samples of leaves and roots subjected to drought stress was lower than in the control samples. The total phenolic compounds in grapevine leaves after drought stress (LS) were significantly lower according to several calculations (272 mg g⁻¹ of the extract, 62 mg g⁻¹ of the dry matter,

Table 1 Content of total phenolic compounds in control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*

Sample	mg g ⁻¹ of extract	mg g ⁻¹ DW	mg g ⁻¹ FW
LC	389 ± 17	80.4 ± 8	19.37 ± 1.47
LS	272 ± 13**	62.7 ± 5.59***	15.94 ± 1.38**
RC	944 ± 12	286 ± 18	39.22 ± 2.73
RS	861 ± 16**	222 ± 10**	27.63 ± 1.09***

The experimental conditions are explained in “Materials and methods”

The vertical lines mark standard deviation ($n = 4$)

The asterisks designate significant differences for dependent samples at $p < 0.01$ (**) and $p < 0.001$ (***)

16 mg g⁻¹ of the fresh matter) versus the control sample (LC) (respectively—389 mg g⁻¹ extract, 80 mg g⁻¹ dry matter, 19 mg g⁻¹ fresh matter). The total phenolic compounds in grapevine roots after drought stress (RS) were significantly lower according to several calculations (891 mg g⁻¹ of the extract, 222 mg g⁻¹ of the dry matter, 27 mg g⁻¹ of the fresh matter) versus the control sample (RC) (respectively: 944 mg g⁻¹ extract, 268 mg g⁻¹ dry matter, 39 mg g⁻¹ fresh matter).

The DPPH• scavenging activity of extracts

Extracts of all the samples demonstrated DPPH• scavenging activity, which suggests that they all contained antioxidants. The samples possessing the highest antioxidant activity were characterized by the lowest absorbance (Fig. 2).

At the highest extract concentration (0.5 mg/sample) in control samples of leave (LC), the absorbance was at the level of 0.34 and it was significantly lower that stress sample of leaves (LS) (0.42). However, at the highest extract concentration in control sample of roots (RC) (0.05 mg/sample), the absorbance was 0.14. In the same extract concentration from stress sample of roots (RS), the absorbance was at the level of 0.19. Between these samples are significant differences. Results of revealed that control samples of leaves and roots were a higher antioxidant activity than stress samples. The highest capacity to scavenge DPPH• was seen in the extract from roots.

Trolox equivalent antioxidant capacity (TEAC)

The ABTS⁺ scavenging activity is illustrated in Fig. 3. The results showing the antioxidative activity of the extracts have been presented as Trolox equivalents.

The results confirm that all the analysed samples were able to scavenge ABTS free cation radical. Moreover, control sample and stressed samples of roots had more the ABTS⁺ free radical scavenging activity compared to

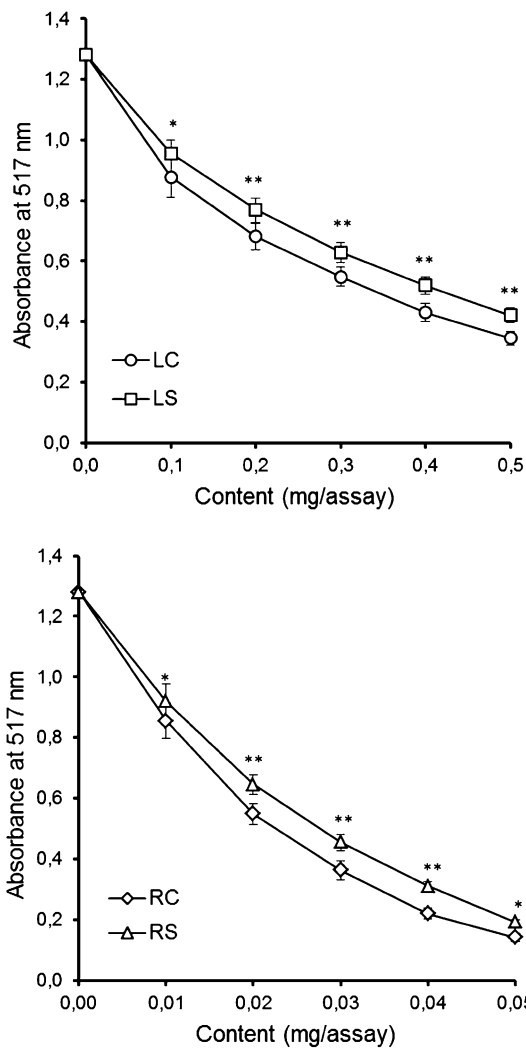


Fig. 2 DPPH[•] free radical scavenging activity of extracts from control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in “Materials and methods”. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.05$ (single asterisks), at $p < 0.01$ (double asterisks)

control and stressed samples of leaves. Analogously to the tests on DPPH[•] scavenging activity, described above, the samples of leaves (LS) subjected drought stress (LS) showed a significantly lower free radical scavenging activity than the control sample of leaves (LC): respectively 0.33 mmol Trolox g⁻¹ extract (LS) and 0.42 mmol Trolox g⁻¹ extract (LC). The same relationship was observed in the case of the roots where the able to scavenge ABTS^{•+} free cation radical of stress sample (RS) was significantly lower than control sample (RC): respectively 1.07 mmol Trolox g⁻¹ extract from RS and 1.25 mmol Trolox g⁻¹ extract from RC. These results indicate that control sample of leaves and roots had more the ABTS free radical scavenging activity, compared to the stress samples.

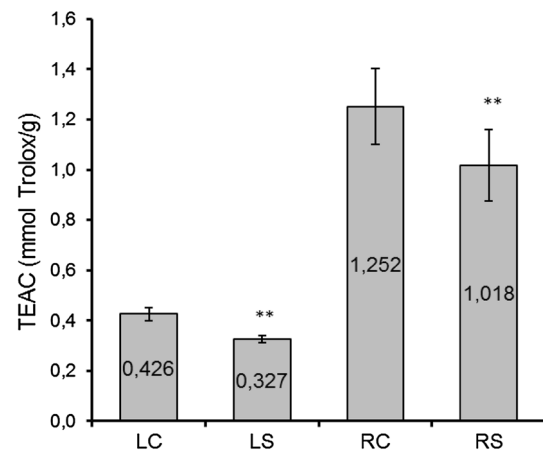


Fig. 3 ABTS^{•+} free cation radical scavenging activity of extracts from control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in “Materials and methods”. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.01$ (double asterisks)

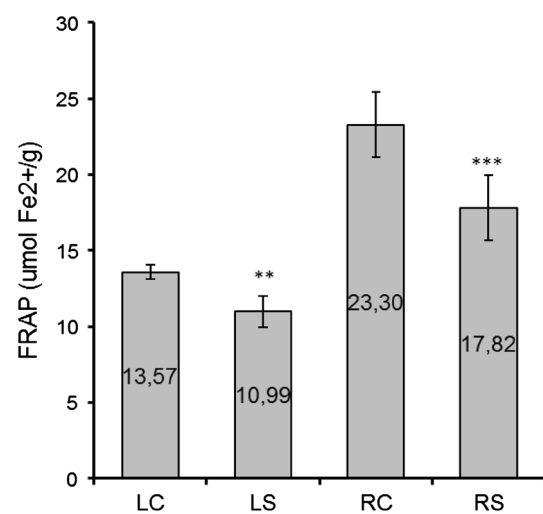


Fig. 4 Reducing power of extract from control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in Table 1. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.01$ (double asterisks), at $p < 0.001$ (triple asterisks)

Determination of the reducing power of extract

Determination of the reducing power of extract in leaves and roots grapevine is illustrated in Fig. 4. Under the influence of the reducing power of the extracts, Fe³⁺ ions were reduced to Fe²⁺ ions, and the colour of the mixture changed from light orange blue to dark blue. The results confirm that all the analysed samples were able to reduce Fe³⁺ ions. It was shown that control sample of leaves and roots were more

capable of reducing Fe^{3+} ions, compared to the stress sample. The samples of leaves (LS) subjected drought stress showed a significantly lower reducing power than the control samples of leaves (LC), respectively, 10.99 mmol Trolox g^{-1} extract (LS) and 13.56 mmol Trolox g^{-1} extract (LC). The same relationship was observed in the case of the roots where the reducing power ability of the stressed samples (RS) was significantly lower than control samples (RC), respectively, 17.81 mmol Trolox g^{-1} extract from RS and 23.3 mmol Trolox g^{-1} extract from RC.

Content of phenolic acids

The assays performed on extracts from grapevine leaves and roots demonstrated that they contained *p*-coumaric, ferulic and caffeic acids. All these phenolic acids appeared in the leaves and roots were in the ester-bound form. Free and glycoside-bound forms of phenolic acids were not also detected in roots and leaves samples. Control and stressed samples of leaves were additionally identified free form of caffeic acid. Among the analysed samples (Figs. 5, 6, 7), ester-bound form of caffeic acid had the highest concentration- in sample of leaves from 5.225 (LS) to 8.03 $\mu\text{g g}^{-1}$ fresh matter (LC) and in sample of roots from 0.412 (RS) to 0.534 $\mu\text{g g}^{-1}$ fresh matter (RC). The free form of caffeic acid had significant lower concentration in control sample of leaves (0.195 $\mu\text{g g}^{-1}$ fresh matter) and stress samples of leaves (0.08 $\mu\text{g g}^{-1}$ fresh matter). The next most abundant compound was *p*-coumaric acid, whose content in leaves was from 0.285 (LS) to 0.309 $\mu\text{g g}^{-1}$ fresh matter (LC) and in roots was from 0.1045 (RS) to 0.1375 $\mu\text{g g}^{-1}$ fresh matter (RC). The content of ferulic acid in all samples was

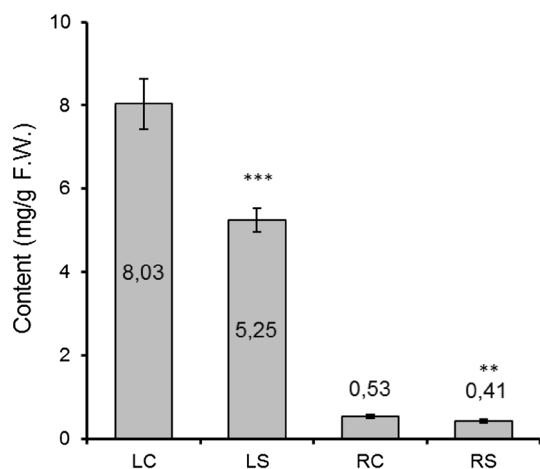


Fig. 5 Content of ester-bound form of caffeic acid in control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in Table 1. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.01$ (double asterisks), at $p < 0.001$ (triple asterisks)

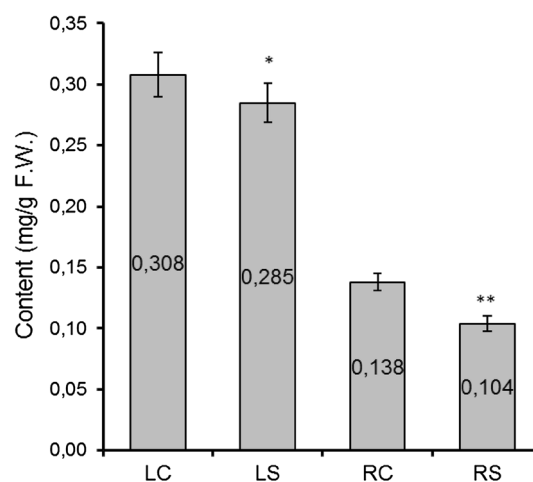


Fig. 6 Content of ester-bound form of *p*-coumaric in control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in Table 1. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.05$ (single asterisks), at $p < 0.01$ (double asterisks)

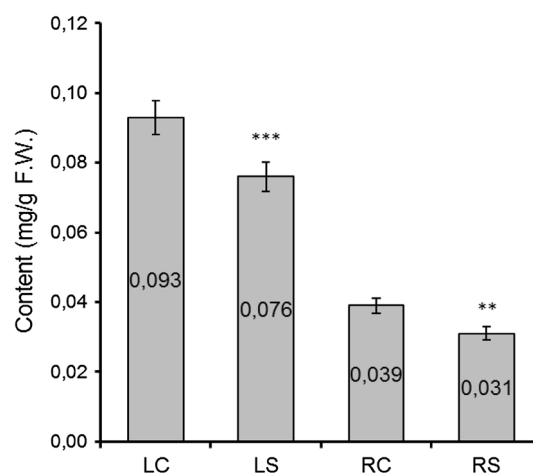


Fig. 7 Content of ester-bound form of ferulic acid in control and drought stressed samples leaves (LC, LS) and roots (RC, RS) of *V. vinifera*. The experimental conditions are explained in Table 1. The vertical lines mark standard deviation ($n = 4$). The asterisks designate significant differences for dependent samples at $p < 0.01$ (double asterisks), at $p < 0.01$ (triple asterisks)

the lowest of all identified acids. In grapevine leaves, it varied from 0.076 (LS) to 0.093 $\mu\text{g g}^{-1}$ fresh matter (LC). In roots it varied from 0.0305 (RS) to 0.0385 $\mu\text{g g}^{-1}$ fresh matter (RC). Indeed, the content of all acids in the sample subjected to drought stress (LS, RS) dropped significantly compared to the control sample (LC, RC).

Discussion

Long-term and continuous drought stress inhibits the total synthesis of phenolic compounds in leaves and roots.

Similar results were obtained in earlier research on seedlings of *V. vinifera* exposed to chilling stress. The total content of phenolics, tannins and phenolic acids in stress sample was lower than in control sample (Weidner et al. 2009b). Furthermore, Amarowicz et al. (2010) observed the same trend in leaves of *V. vinifera* subjected to low-temperature stress. On the other hand, some researchers have observed quite the opposite effect. Chung et al. (2006) found out that chilling stress and water deficit caused an increase in the total content of phenolic compounds in *Rehmannia glutinosa*. These results were confirmed in seedling of soybean subjected to chilling stress (Posmyk et al. 2005). Other experiments have demonstrated that chilling stress did not have any major effect on the total content of phenolic compounds in pea roots, but it significantly decreased the content of flavone (Rudikowskaya et al. 2008). Many authors demonstrated that the production of phenols in plant tissues rises under abiotic stress conditions (Dixon and Paiva 1995; Janas et al. 2002; Wróbel et al. 2005; Weidner et al. 2009a). Such large discrepancies in experimental results can be attributed to differences in abiotic stresses, e.g. type of stress, its intensity, its duration, the stages of plant development (in general, the early stages of germination and plant development are the least tolerant to stress) and the biological material, e.g. whole seedlings or different parts of plants, such as roots or leaves, which are characterized by a great diversity of secondary metabolites (Weidner et al. 2009a).

In response to changing environmental conditions, plants evolved the capacity to biosynthesize different phenolic acids (Cooper-Driver and Bhattacharya 1998; Bieza and Lois 2001; Caldwell et al. 2007). In the present study, three phenolic acids were identified in *V. vinifera* leaves and roots: caffeic, *p*-coumaric and ferulic. The most abundant was caffeic acid, while ferulic acid was the least abundant. All phenolic acids were present as ester-bounds form. Only caffeic acid in leaves appeared in the free form. Under long-term drought stress condition, the level all identified phenolic acids in leaves and roots were reduced. Similar results were shown by Weidner et al. (2011) in germinating seeds of *Vitis californica* subjected to osmotic stress, but the ester-bound form of caffeic acid increased under stress conditions. In leaves of *V. vinifera* exposed to low temperature, the content of ester- and glycoside-bound caffeic and ferulic acid rose, while that of *p*-coumaric acid declined (Amarowicz et al. 2010). In addition, depressed levels of the ester-bound form of *p*-coumaric, ferulic and caffeic acids have also been observed in roots of *V. vinifera* exposed to chilling stress. *p*-Coumaric acid was the most abundant and caffeic acid the least abundant phenolic in tissues of such roots (Weidner et al. 2009b). Solecka (1997) claims that the presence of larger concentrations of phenolic acids in ester- or glycoside-bound forms in tissues

is less toxic to plants, than the free forms. Worth noticing is the fact that the post-stress recovery process caused an increase in the content of the free and ester-bound forms of caffeic and *p*-coumaric acids in germinating seeds of *V. californica*. Also, an increase in the content of ester-bound gallic acid, accompanied by a decrease in its free form, was observed. The post-stress recovery process decreased the ester-bound form of ferulic acid (Weidner et al. 2011). Increased concentrations of ester-bound forms of *p*-coumaric, caffeic and ferulic acids during recovery from chilling stress had been previously detected in roots of *V. vinifera* (Weidner et al. 2009b). Ferulic acid and *p*-coumaric acid are characterized by the highest reducing power of free radicals (Szwajgier et al. 2005).

Antioxidant and antiradical activity of leaves and roots extracts of *V. vinifera* was investigated using several methods such as ABTS⁺•, scavenging DPPH• radical and the FRAP assay. The above results suggest that all the extracts from grapevine leaves and roots had antioxidative properties. All extracts from leaves and roots growing under optimum conditions proved to be more efficient in scavenging DPPH• free radicals and had a higher reducing power than extracts from tissues exposed to drought stress. This finding provides evidence that tissues of grapevine seedlings subjected to stress contain less antioxidants and reducing compounds. Zainol et al. (2003) showed that the phenolic compounds are the major contributors to the antioxidative activities. The extent of antioxidative effects produced by phenols largely depends on the structure of given compound and especially on the number and distribution of hydroxyl groups (–OH). Antioxidative activity is much higher when a compound has two –OH groups in the ortho position (e.g. caffeic acid) (Rosicka-Kaczmarek 2004). Our study also showed that the roots of the grapevine contain more antioxidants than the leaves. There appears to be a significant correlation between the content of total phenolics and antioxidant activity. Xu et al. (2010) claimed the same effect in experiments on different grape cultivars. Similar results were presented in our earlier work (Amarowicz and Weidner 2009) and by Alonso et al. (2002).

When stresses act upon plants, changes in gene expression take place. Within the first 12 h, the expression of many genes is increased (Kilian et al. 2007). It should be noted that this process requires a lot of energy input. But when the stress lasts a long time, the energy-intensive processes are limited. Kilian et al. (2007) claimed that after 24 h, the expression of a large number of genes was inhibited or significantly reduced. Xiong et al. (2006) demonstrated that during long periods of drought stress is a significant decrease in activity of phenylalanine ammonia-lyase (PAL) in *Arabidopsis thaliana* roots. PAL (E.C.4.3.1.5) is the main enzyme in the synthesis pathway of phenolic compounds in plants (Szafrńska et al. 2002).

Numerous studies have revealed that environmental stress often raise the accumulation of phenolic compounds and phenolic acids. It should be noted that most of the studies are concentrated on short-term stress. However, the mechanism of plant response to long and continuous stressor is different. Also the research presented in this paper has demonstrated that the total content of phenolics, phenolic acids and antioxidant activity and reducing power decrease in grapevine leaves and roots under long-term drought stress. It indicates a slowdown in selected elements of secondary metabolism in plants during long-term drought stress. So, the plant reduces energy expenditure until the end of the activity stressor. This is one of strategy for survival in disadvantaged environment conditions.

Author contribution The maintainers of the creation of the manuscript: Professor Stanisław Weidner was: author of research concept and design, critical revision of article, final approval of article. Professor Ryszard Amarowicz was in charge of: critical revision and final approval of article, made measurement of phenolic acids content using HPLC system. Angelika Król was in charge of: collection analysis, data analysis and interpretation, writing the article.

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