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Channel Modeling for Diffusive Molecular Communication – A Tutorial Review

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Abstract—Molecular communication (MC) is a new communication engineering paradigm where molecules are employed as information carriers. MC systems are expected to enable new revolutionary applications such as sensing of target substances in biotechnology, smart drug delivery in medicine, and monitoring of oil pipelines or chemical reactors in industrial settings. As for any other kind of communication, simple yet sufficiently accurate channel models are needed for the design, analysis, and efficient operation of MC systems. In this paper, we provide a tutorial review on mathematical channel modeling for diffusive MC systems. The considered end-to-end MC channel models incorporate the effects of the release mechanism, the MC environment, and the reception mechanism on the observed information molecules. Thereby, the various existing models for the different components of an MC system are presented under a common framework and the underlying biological, chemical, and physical phenomena are discussed. Deterministic models characterizing the expected number of molecules observed at the receiver and statistical models characterizing the actual number of observed molecules are developed. In addition, we provide channel models for time-varying MC systems with moving transmitters and receivers, which are relevant for advanced applications such as smart drug delivery with mobile nanomachines. For complex scenarios, where simple MC channel models cannot be obtained from first principles, we investigate simulation-driven and experiment-driven channel models. Finally, we provide a detailed discussion of potential challenges, open research problems, and future directions in channel modeling for diffusive MC systems.

Index Terms—Molecular communications, diffusion, flow, reaction, end-to-end CIR, statistical model, simulation-driven models, and experiment-driven models.

I. INTRODUCTION

Wireless communication networks have permeated throughout modern society, but existing systems are constrained by where conventional radio frequency technologies can be deployed. There are emerging applications where wireless communication could be a vital component, but where conventional implementations would be unsafe or impractical. An alternative approach that has received increasing attention within the communications research community over the last decade is molecular communication (MC), where molecules are

employed as the information carriers¹. MC was first proposed for the design of synthetic communication networks in [1]. The topic has received steady growth since the seminal survey on nanonetworks in [2], which are networks of devices with nanoscale functional components. MC is ubiquitous in natural biological systems, which lends credibility to its potential for biomedical applications such as targeting substances, smart drug delivery, and designing lab-on-a-chip systems [3]. Furthermore, MC could be deployed in industrial settings, including the monitoring of chemical reactors and nanoscale manufacturing, or for larger activities such as monitoring the emission of pollutants or the transport of oil [4]. A network of nanomachines communicating with each other via MC can help realize the Internet-of-BioNanothings and enable nanomachines to perform complex tasks [5].

Motivated by natural MC systems, several different mechanisms have been considered for MC in the literature including free diffusion [6]–[12], gap junctions [13]–[15], molecular motors [16], and bacterial motors [17]; see Fig. 1. In particular, diffusion is referred to as the random movement of small particles suspended in a fluid medium as a result of their collisions with other particles in the fluid. Diffusion is one of the dominant propagation mechanisms in nature including communication inside cells and between cells, e.g., in quorum sensing among bacteria and in the synaptic cleft between neurons. Gap junctions enable another form of communication between cells where the molecules pass through small channels that connect the cytosols of neighboring cells. Calcium signaling is an example of this form of MC that is used by adjacent cells to regulate a large number of cellular processes including fertilization, proliferation, and death of mammalian cells [13], [18]. Molecular motors enable a form of active transportation of large signaling molecules via a special rail-like infrastructure, e.g., actin or microtubule filaments [19]. The motor moves along the rail by using repeated cycles of coordinated binding and unbinding of its legs to the rail. This type of MC is primarily used for intracellular communication among organelles inside a cell [16]. Finally, bacterial motors enable another kind of active transport where the bacteria can pick up large signaling molecules, e.g., deoxyribonucleic acid (DNA), and move in a specific direction, e.g., due to a food concentration gradient, using their tiny propellers (known as flagella) [17].

Diffusion-based MC, sometimes in combination with advection and chemical reaction networks (CRNs), has been the

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¹We note that, in this paper, we use the terms “molecule” and “particle” interchangeably.

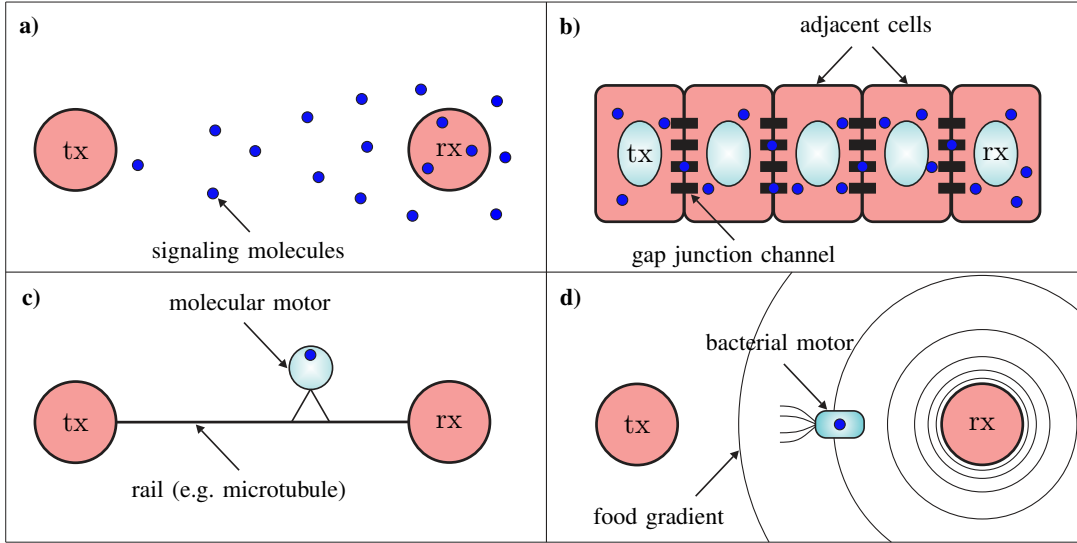


Fig. 1. Bio-inspired mechanisms for MC between a transmitter (tx) and a receiver (rx); a) Free diffusion, b) gap junctions, c) molecular motors, and d) bacterial motors.

prevalent approach considered in the literature thus far; see [3, Table 4]. The main advantages of diffusion-based MC include that, unlike gap junction-based MC, special infrastructure is not needed, and unlike motor-based MC, external energy for propagation of the signaling molecules is not required. Moreover, the simplicity of diffusion makes it an attractive propagation scheme, especially for *ad hoc* networks where mobile nanorobots with limited computational resources form a communication network among themselves and/or with living cells in their close proximity. Hence, in this tutorial, we focus on diffusion-based MC, where we also consider environments with advection and CRNs.

A. Scope

The physics of diffusion and characterizing *expected* diffusion in environments of different shapes have been extensively studied in the physics, biology, and chemistry literature, cf. e.g., [20], [21]. Thereby, the primary goal is to understand how natural phenomena work, e.g., to understand the natural and evolutionary MCs that exist within and among living organisms. In contrast, in the emerging field of engineered MC, the aim is to design, build, and control human-made MC systems for a specific purpose². To this end, the communications research community has expanded the models obtained in other disciplines to account for the behavior of the *end-to-end* system, for the inclusion of non-diffusive phenomena that play important roles in biophysical systems, and for the *statistics* of molecular behavior.

Recent surveys, in particular [3], [23], have provided excellent qualitative summaries of diffusive MC and included some of the most common channel models available thus far. A more complete mathematical treatment of diffusion-based

modeling of MC can be found in [24]. However, there have been significant advances in channel modeling in the years since the publication of [24], and also since the most recent major survey of models in [3]. In particular, non-diffusive effects that can be coupled with diffusion, such as advective flow and chemical reaction kinetics, have been integrated in many channel models to make them more practical and more accurate.

Due to the rapid growth in channel models, it has become difficult for an interested researcher to enter the MC field and become familiar with the state-of-the-art in diffusion-based channel modeling. It has also become more challenging for practitioners in this field to stay up to date. The aim of this tutorial review is to satisfy both audiences. We present a detailed and rigorous mathematical development of diffusive MC channel models. We seek to provide a useful comprehensive reference on channel models that is both approachable for an audience that is new to the field and also convenient for active practitioners to assess and select a model. To do so, we begin with a review of the underlying fundamental laws that govern diffusive MC channels and show how they are used in the literature to derive the channel impulse responses (CIRs) of different MC systems. In addition, we present different deterministic and statistical models developed for the observation signal at the receiver. We also discuss the complementary roles of simulations and physical experiments to both support analytical modeling and provide data-driven models when simple analytical models that capture the underlying complex dynamics of the system cannot be readily obtained.

B. Contributions

In this tutorial review, we make the following contributions:

- 1) By taking a mathematically rigorous approach, we first provide a tutorial on the underlying phenomena from biology, chemistry, and physics, and their effect on the

²Different options to build MC systems exist, e.g., to genetically modify natural cells or to design fully-synthetic MC systems [22]. Therefore, an engineered MC system may also include components that naturally developed via evolution.

components of MC systems. Specifically, we start with Fick's laws of diffusion and build towards the general advection-reaction-diffusion equation. We discuss the common assumptions and special cases that enable the general equation to be solved for the CIR in closed form.

- 2) We review the major end-to-end channel models in the diffusive MC literature including the effects of release mechanisms, the physical channel, and reception mechanisms. In particular, we include the relevant classical models from the physical sciences literature, as well as a comprehensive presentation of the models that have been developed and the equations that have been derived within the communications engineering community over the last few years.
- 3) We present a unified definition for the observed signal at a receiver. The unified definition encompasses both timing and counting receivers and helps to better understand the basic assumptions that have been made to arrive at the well-known signal models used in the MC literature and how they relate to each other. Then, we focus on counting receivers and derive signal models relevant for different time scales. We further generalize these models to account for interfering noise molecules and inter-symbol interference (ISI). Finally, we study the correlation between the received signals observed at different time scales.
- 4) We discuss the integral role of simulations and experiments, in particular to gain insight from a data-driven model when closed-form solutions for the CIR are not readily available. We also describe how to implement simple stochastic simulations as well as how to derive an example data-driven model based on experimental data.

For clarity of presentation, the focus of the channel models presented in this work is on a *single* communication link between one transmitter and one receiver. Many of the envisioned applications of diffusive MC systems will depend on many links within a network of devices. While there have been a number of relevant contributions that consider the propagation of signals over multiple links, such as via relaying and cooperative detection (cf. e.g., [25]–[29]), these models can often be decomposed into a superposition of individual links. In these cases, the analytical models developed in this paper (cf. Sections III and IV) still apply to the individual links. However, it is important to note that single-link analysis cannot always be applied to multi-link systems. For example, when other non-transparent entities (such as reactive receivers) are present in the system and molecules can collide or react with them, each of these entities will impact the signal received at *any* receiver. In general, the impact of other reactive entities on the received signal can be considered by modeling them via additional boundary conditions. Then, the analytical channel modeling methodologies presented in this paper can be used, cf. Sections III and IV. Nevertheless, the resulting systems of partial differential equations (PDEs) are typically too complex to solve and hence, data-driven approaches have to be used in practice, cf. Section V. For example, in [30], the CIR was presented in closed form for the special case of having two

absorbing receivers placed on either side of a transmitter, whereas in [31] a data-driven model was proposed for the more general case of having multiple absorbing receivers at arbitrary positions.

C. Organization

The rest of this tutorial review is organized as follows, and also summarized in Table I to show how the content of Sections II–V is connected. We review the fundamental physical principles that govern diffusion-based MC systems in Section II. In particular, we model diffusion, advection, and chemical reactions, which leads to a general advection-reaction-diffusion PDE to describe the spatio-temporal variation in molecule concentrations.

In Section III, we discuss the components of MC systems and their effect on the end-to-end CIR. Our definition of the end-to-end channel includes the physical and chemical properties of the transmitter and receiver, as well as the fluid medium in which they are located. A table to summarize the reviewed CIRs is also provided.

In Section IV, we present a unified definition for the diffusive signal observed at the receiver. We focus on counting receivers and derive deterministic and statistical signal models that are valid for different time scales.

We discuss simulation- and experiment-driven models in Section V. We describe the different physical scales for simulating diffusion-based systems, summarize existing simulation platforms for each scale, and discuss how to implement simple stochastic simulations. Moreover, we review a selection of experimental platforms and propose to employ either physically-motivated parametric models or neural networks, whose parameters are found using experimental data.

We end this tutorial review with a discussion of future work and open challenges in Section VI before presenting our conclusions in Section VII.

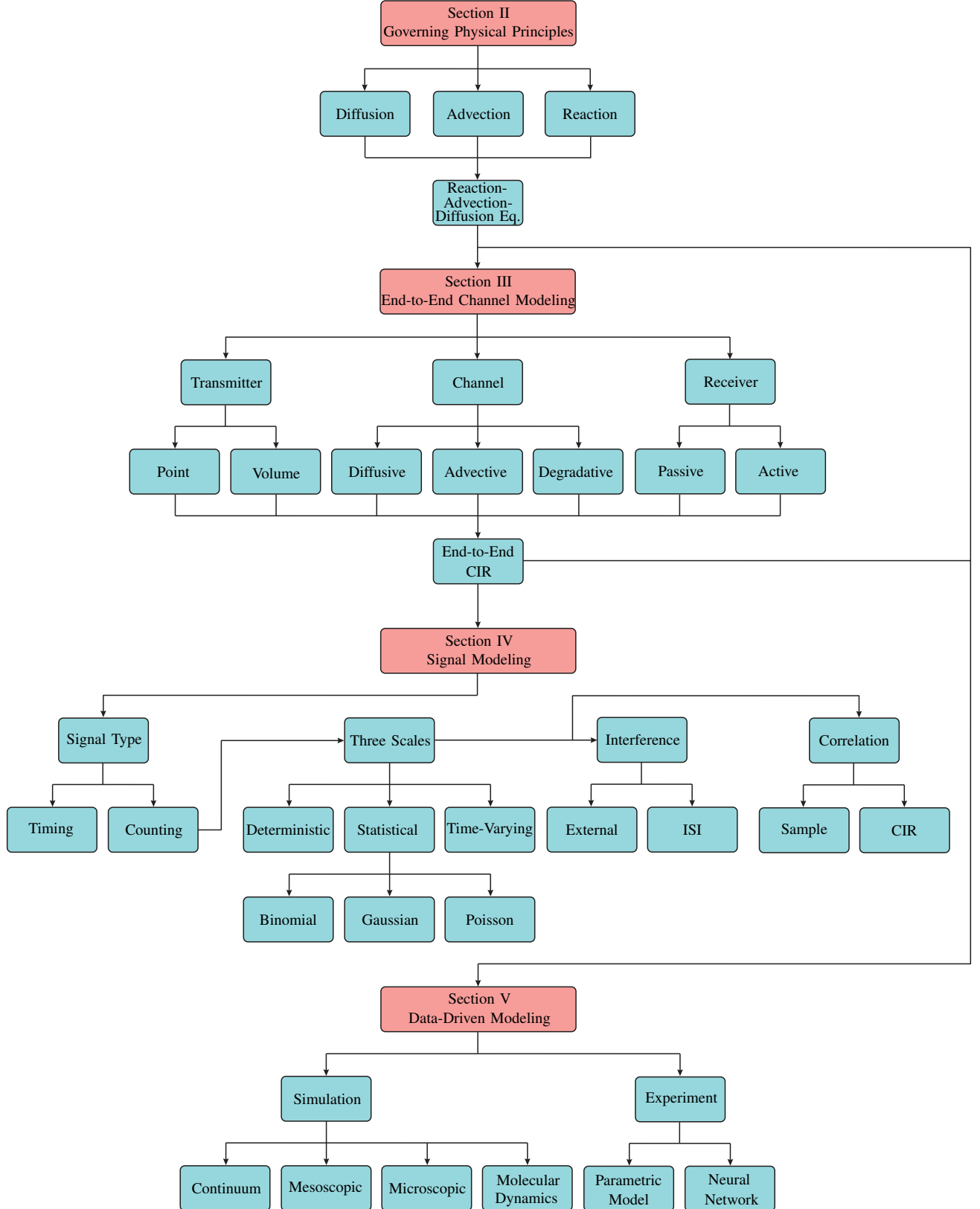
II. FUNDAMENTAL GOVERNING PHYSICAL PRINCIPLES IN MC SYSTEMS

In this section, we review the fundamental laws that govern the propagation of molecules. In particular, we mathematically model the impact of diffusion, advection, and reaction on the spatio-temporal distribution of molecules. This modeling is essential for the development of channel models. A solid understanding of these phenomena is needed to develop intuition for molecule propagation in diffusive MC systems. Furthermore, in Section III, we will use the mathematical tools introduced in this section for the derivation of the CIR for several different diffusive MC systems.

A. Free Diffusion

Molecules in a fluid environment, such as a liquid or a gas, are affected by thermal vibrations and collisions with other molecules. The resulting movement of the molecules is purely random without any preferred direction and is referred to as random walk or Brownian motion. Let $\mathbf{d}_i(t) = [x, y, z]$ denote

TABLE I
ORGANIZATION AND CONTENT OF SECTIONS II-V AND THEIR CONNECTIONS.



a vector specifying the position of the i -th molecule in three-dimensional (3D) Cartesian coordinates at time t . Thereby, the random walk is modeled by [32, Eqs. (1.3) and (1.21)]

$$\mathbf{d}_i(t + \Delta t) = \mathbf{d}_i(t) + \mathcal{N}(\mathbf{0}, 2D\Delta t \mathbf{I}), \quad (1)$$

where Δt is the time step size and D in $[\text{m}^2\text{s}^{-1}]$ is the diffusion coefficient of the i -th molecule. Moreover, $\mathcal{N}(\boldsymbol{\mu}, \boldsymbol{\Sigma})$ denotes a multivariate Gaussian random variable (RV) with mean vector $\boldsymbol{\mu}$ and covariance matrix $\boldsymbol{\Sigma}$, $\mathbf{0}$ represents a vector whose elements are all zeros, and \mathbf{I} is the identity matrix. The diffusion coefficient determines how fast the molecule moves. The larger the diffusion coefficient, the larger the average displacement of the molecule in a given time interval. The value of the diffusion coefficient depends on the environment as well as the shape and the size of the particle. For spherical particles immersed in a fluid continuum, the diffusion coefficient can be determined based on the Einstein relation [33, Chapter 5.2.1]

$$D = \frac{k_B T}{6\pi\eta R}, \quad (2)$$

where $k_B = 1.38 \times 10^{-23} \text{ JK}^{-1}$ is the Boltzmann constant, T is the temperature in kelvin, η is the (dynamic) viscosity of the fluid ($\eta = 10^{-3} \text{ kg m}^{-1}\text{s}^{-1}$ for water at 20°C), and R is the radius of the particle. Note that larger particles have a smaller diffusion coefficient and are hence less affected by diffusion.

Remark 1: From [33, Chapter 5.2.1], the diffusion coefficient can be determined from (2) as long as the surrounding liquid can be modeled as a continuum. By experiment, this is an accurate assumption if the particle size is at least *five* times the size of the molecules of the liquid. For example, in water, (2) is applicable for particles having a diameter larger than 1.5 nm. For small particles not satisfying this condition, the diffusion coefficient tends to be larger than that predicted by (2). Nevertheless, a general formula encompassing all physical regimes does not exist. \square

Remark 2: Besides the ideal free diffusion with constant diffusion coefficient discussed above, there are also other types of diffusion. For instance, in contrast to the typical free diffusion where the mean squared displacement (MSD) is linearly proportional to time, i.e., $\text{MSD} \propto D\Delta t$, in anomalous diffusion, the MSD follows a nonlinear relation, i.e., $\text{MSD} \propto D\Delta t^\gamma$ where $\gamma \neq 1$. Sub-diffusion occurs when $\gamma < 1$ and can be used to model diffusion inside biological cells where the presence of the organelles does not allow ideal free diffusion to take place [34]. Super-diffusion occurs when $\gamma > 1$ and can be used to model *active* cellular transport processes [35]. Moreover, in (1), we assumed the diffusion coefficient to be *constant*. However, the diffusion coefficient may depend on the local concentration of the molecules [33]. For the constant diffusion coefficient assumption to hold, the temperature and viscosity of the environment are assumed to be uniform and constant and all solute molecules (dissolved molecules) are assumed to be locally dilute everywhere, i.e., the number of solute molecules is sufficiently small everywhere. These assumptions allow us to ignore potential collisions between solute molecules such that the diffusion coefficient does not vary with the local concentration [8], [33]. We refer the readers to [36] for the study of diffusion with non-constant diffusion

coefficients. Another example of a complex diffusion process is the diffusion of protons in water. Here, the movement of the protons is a combination of ideal free diffusion and the so-called structural diffusion where protons hop from one water molecule to the next. Nevertheless, it has been shown in [37] that proton transport can be well approximated by free diffusion with an effective diffusion coefficient. \square

We let $c(\mathbf{d}, t)$ denote the concentration of the solute molecules, i.e., the *average number* of solute molecules per unit volume, at coordinate \mathbf{d} and time t . The random movement of molecules due to diffusion, described by (1), leads to variation of $c(\mathbf{d}, t)$ across time and space that obeys Fick's second law of diffusion³

$$\frac{\partial c(\mathbf{d}, t)}{\partial t} = D\nabla^2 c(\mathbf{d}, t), \quad (3)$$

where ∇^2 is the Laplace operator, e.g., $\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$ in Cartesian coordinates. The PDE in (3) can be solved for simple initial conditions (ICs) and simple boundary conditions (BCs). In the following, we consider a simple example, namely diffusion in an unbounded 3D environment with an impulsive point release, which has been the most widely studied case in the MC literature due to its simplicity [3], [38]–[46]. In the remainder of this paper, we denote the solutions of the considered PDEs by $c^*(\mathbf{d}, t)$.

Example 1 (Diffusion in an Unbounded 3D Environment with Impulsive Point Release): Consider a 3D diffusion process with instantaneous release of N solute molecules from \mathbf{d}_0 at time t_0 . To obtain $c^*(\mathbf{d}, t)$, we have to solve (3) with the following initial and boundary conditions

$$\text{IC}_1 : c(\mathbf{d}_0, t \rightarrow t_0) = N\delta(\mathbf{d} - \mathbf{d}_0) \quad (4)$$

$$\text{BC}_1 : c(\|\mathbf{d}\| \rightarrow \infty, t) = 0, \quad (5)$$

where $\delta(\mathbf{d}) = \delta(x)\delta(y)\delta(z)$, and $\delta(\cdot)$ is the Dirac delta function. Solving (3) with IC_1 and BC_1 yields [32, Eq. (2.8)]

$$c^*(\mathbf{d}, t) = \frac{N}{(4\pi D(t - t_0))^{3/2}} \exp\left(-\frac{\|\mathbf{d} - \mathbf{d}_0\|^2}{4D(t - t_0)}\right). \quad (6)$$

\square

In Fig. 2, the molecule concentration $c^*(\mathbf{d}, t)$ [molecules/ m^3] in (6) is plotted versus time $[\mu\text{s}]$ at distance $\mathbf{d} = [d, 0, 0]$ with $d \in \{300, 400, 500\}$ nm for an initial release of $N = 10^4$ molecules with $D = 4.5 \times 10^{-10} \text{ m}^2/\text{s}$ from the origin $\mathbf{d}_0 = [0, 0, 0]$ at time $t_0 = 0$. From Fig. 2, we observe that first $c^*(\mathbf{d}, t)$ increases with time, which is due to the non-zero propagation time that the molecules need to reach \mathbf{d} , before it decreases since the molecules diffuse away. Moreover, as distance increases, the peak of the concentration decreases since the molecules are spread over a larger volume. Furthermore, the time when the concentration peak occurs, denoted by t^p , increases with distance.

The assumption of an unbounded environment is accurate when the actual boundaries of the system are far away from the region of interest (i.e., from transmitter and receiver), such that the impact of the boundaries on the diffusing molecules

³Fick's first law of diffusion relates the diffusive flux, denoted by $\mathbf{J}(\mathbf{d}, t)$, to the concentration as $\mathbf{J}(\mathbf{d}, t) = -D\nabla c(\mathbf{d}, t)$.

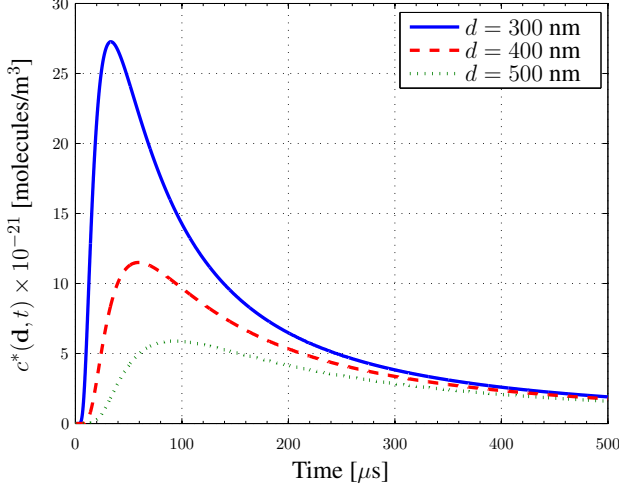


Fig. 2. Molecule concentration $c^*(\mathbf{d}, t)$ [molecules/m³] versus time [μ s] at distance $\mathbf{d} = [d, 0, 0]$ with $d \in \{300, 400, 500\}$ nm for initial release of $N = 10^4$ molecules with $D = 4.5 \times 10^{-10}$ m²/s from the origin $\mathbf{d}_0 = [0, 0, 0]$ at time $t_0 = 0$.

can be neglected. In the following, we present an example where the effect of the boundaries cannot be neglected.

Example 2 (Diffusion in an Unbounded Straight Duct with Impulsive Release from Cross-Section): We assume a straight duct⁴ channel with circular cross-section and for convenience, we employ cylindrical coordinates, i.e., $\mathbf{d} = [\rho, \varphi, z]$ with $0 \leq \rho \leq a_c$, $0 \leq \varphi \leq 2\pi$, $-\infty < z < +\infty$, where a_c denotes the radius of the circular cross-section of the duct. We assume that, at the time of release, t_0 , the molecules are uniformly distributed across the cross-section at $z = z_0$. Therefore, we have the following initial and boundary conditions

$$\text{IC}_2 : c(\mathbf{d}_0 = [\rho, \varphi, z], t \rightarrow t_0) = \frac{N}{\pi a_c^2} \delta(z - z_0) \quad (7)$$

$$\text{BC}_2 : \left. \frac{\partial c(\mathbf{d}, t)}{\partial \rho} \right|_{\rho=a_c} = 0 \quad (8)$$

$$\text{BC}_3 : c(\mathbf{d} = [\rho, \varphi, z \rightarrow \pm\infty], t) = 0, \quad (9)$$

where BC_2 enforces the reflection of the molecules at the wall, i.e., a fully reflective wall is assumed. Solving (3) with IC_2 , BC_2 , and BC_3 yields [47]

$$c^*(\mathbf{d}, t) = \frac{N}{\pi a_c^2 \sqrt{4\pi D(t - t_0)}} \exp\left(-\frac{(z - z_0)^2}{4D(t - t_0)}\right), \quad \rho < a_c. \quad (10)$$

□

As can be seen from (10), $c^*(\mathbf{d}, t)$ does not depend on variables ρ and φ due to the symmetry of the initial condition and the environment with respect to ρ and φ . This model can be used to characterize the propagation of molecules in blood vessels as is necessary for drug delivery applications of MC in the cardiovascular system [48]–[52].

B. Advection

Besides diffusion, advection is another fundamental mechanism for solute particle transport in a fluid environment. In the following, we first specify how mass transport by advection affects a single solute particle. Subsequently, we distinguish between two types of advection, namely drift and fluid flow, and give the particle velocity vector for some example cases. Moreover, we present the advection equation which describes the change in molecule concentration due to advection. Finally, we introduce the advection-diffusion equation, which captures the joint impact of diffusion and advection, and characterize the relative importance of diffusion and advection.

In general, transport by advection can be described by a velocity vector $\mathbf{v}(\mathbf{d}, t)$ which generally may depend on position \mathbf{d} and time t . When considering the movement of the i -th particle at position \mathbf{d}_i due to advection, its position at time $t + \Delta t$ can be modeled by

$$\mathbf{d}_i(t + \Delta t) = \mathbf{d}_i(t) + \mathbf{v}(\mathbf{d}_i(t), t)\Delta t, \quad (11)$$

where Δt should be small enough such that the velocity vector is constant between $\mathbf{d}_i(t)$ and $\mathbf{d}_i(t + \Delta t)$. Next, we discuss what may cause the velocity vector $\mathbf{v}(\mathbf{d}, t)$ and what form it may take.

1) Velocity Vector Field: Transport by advection can be mediated by different physical mechanisms which we categorize as *force-induced drift* and *bulk flow* [53], [54].

Force-Induced Drift: Advection can be caused by external forces acting on the particles but not on the fluid containing the particles. An external force can be modeled by force vector $\mathbf{F}(\mathbf{d}, t)$ which describes the force on a particle at position \mathbf{d} at time t . These external forces can be electrical, e.g., if the particles are ions, or magnetic, e.g., if the particles are magnetic nanoparticles, or gravitational, e.g., if the particles have sufficient mass, or a combination of forces [54], [55]. When the force is not too large, the velocity vector can be determined from the corresponding force by Stokes' law via [56, Eq. (2.65)]

$$\mathbf{v}(\mathbf{d}, t) = \frac{\mathbf{F}(\mathbf{d}, t)}{\zeta}, \quad (12)$$

where ζ is a proportionality constant referred to as the friction coefficient. The friction coefficient can be related to the diffusion coefficient via $\zeta D = k_B T$. In other words, using (2), we obtain $\zeta = 6\pi\eta R$. Force $\mathbf{F}(\mathbf{d}, t)$ may vary with time (e.g., for ions if the electric field changes over time) and space (e.g., for magnetic nanoparticles, the magnetic force generally decreases rapidly with increasing distance from the magnet) [54], [55].

Bulk Flow: If the particle movement is induced by the movement of the fluid, then the resulting transport by advection is referred to as flow. Flow can be encountered in many MC environments such as blood vessels and microfluidic channels [57]. In MC, we typically have dilute particle suspensions, where the flow velocity $\mathbf{v}(\mathbf{d}, t)$ is independent from the particle concentration. Thereby, the velocity vector will depend on space if there are boundaries or obstacles in the environment, e.g., in a duct, the flow velocity is typically largest in the center and smallest at the boundary where the fluid is subject to friction.

⁴A duct is a pipe, tube, or channel which carries a liquid or gas.

The flow may also depend on time, e.g., in a blood vessel the flow is generated by the periodic contractions of the heart.

Remark 3: Although both flow and external force cause the particles to drift, which can be modeled by (11), they may require quite different considerations. For instance, any object in the environment influences the velocity vector caused by bulk flow since the flow may not be able to penetrate the object and has to go around the object. On the other hand, the drift velocity vector caused by an external force is not necessarily influenced by objects in the environment. \square

Flow can be also categorized into two classes, namely *turbulent* and *laminar* flow. In particular, when the variations of the flow velocity, over space and/or time, are stochastic, e.g., due to rough surfaces and high flow velocities [58], we refer to the flow as turbulent. If the flow is not turbulent, it is referred to as laminar. For flow in a bounded environment of effective length d_{eff} and with an effective velocity of v_{eff} , the Reynolds number can be used as a criterion for predicting laminar or turbulent flow and is given by [58, Eq. (1.24)]

$$\text{Re} = \frac{d_{\text{eff}} \cdot v_{\text{eff}}}{\nu}, \quad (13)$$

where ν is the kinematic viscosity [m^2/s] of the fluid⁵. For example, for flow in a straight pipe with circular cross-section of radius a_c , the flow can be assumed to be laminar and turbulent for $\text{Re} \ll 2100$ and $\text{Re} \gg 2100$, respectively, where $d_{\text{eff}} = a_c$ [58]. For microfluidic settings, typically $\text{Re} \ll 10$ and hence laminar flow can be assumed [56]. For most blood vessels, $\text{Re} < 500$ holds and hence the blood flow is typically laminar [59], [60]. Only in large arteries such as the aorta (the largest artery in the human body), the Reynolds number can be in the range [3400, 4500] and thereby blood flow exhibits turbulent behavior [60].

Generally, for a given environment, the flow velocity vector $\mathbf{v}(\mathbf{d}, t)$ as a function of space and time can be determined by solving the so-called Navier-Stokes equation with appropriate boundary conditions, see e.g. [56, Eq. (5.22)]. Let us review two special cases of $\mathbf{v}(\mathbf{d}, t)$, which have been widely studied in the MC literature [3], [44], [53], [61], [62] and are also considered in Section III.

Example 3 (Uniform and/or Constant Advection): For uniform advection, the velocity vector is constant across space but can be time-dependent, i.e., $\mathbf{v}(\mathbf{d}, t) = \mathbf{v}(t)$ [62]. For advection by flow in an unbounded environment, uniform flow solves the Navier-Stokes equation and hence can be physically plausible. Moreover, for advection by drift, uniform drift is applicable when the corresponding force vector does not depend on space, see (12). As a special case, the velocity vector may be constant across both space and time, i.e., $\mathbf{v}(\mathbf{d}, t) = \mathbf{v}$. Due to its simplicity, advection with constant velocity is the most widely-studied advection model in the MC literature [3], [44], [53]. \square

Example 4 (Steady Flow in an Infinite Straight Duct with Circular Cross-Section): For this example, we concentrate on advection by fluid flow because force-induced drift is completely specified by (12). In particular, in this case, the

flow velocity vector in cylindrical coordinates $[\rho, \varphi, z]$ can be obtained as [58, Eq. (4.134)]

$$\mathbf{v}(\rho) = \left[0, 0, v_0 \left(1 - \frac{\rho^2}{a_c^2} \right) \right], \quad 0 \leq \rho \leq a_c, \quad (14)$$

where v_0 is the center velocity. The flow described in (14) is laminar and can be interpreted as follows. For a given ρ , the flow velocity in (14) is constant but increases from the boundary where $\mathbf{v}(a_c) = [0, 0, 0]$ towards the center where $\mathbf{v}(0) = [0, 0, v_0]$, i.e., for each ρ , we can think of a circular layer within the duct that slides along its neighboring layers with a constant velocity. The velocity vector in (14) is known as the Poiseuille flow profile and is a common model for the flow in blood capillaries [61]. \square

While for other environments and boundary conditions the velocity vector can still in principle be obtained from the Navier-Stokes equation, it is often not possible to do so analytically. In these cases, the Navier-Stokes equation can be solved by numerical algorithms that are well-established in computational fluid dynamics [58].

2) *Advection Equation:* Given $\mathbf{v}(\mathbf{d}, t)$, the change in concentration with respect to time due to advective transport is modeled by the following PDE, which is referred to as the advection equation or continuity equation [56, Eq. (4.14)]

$$\frac{\partial c(\mathbf{d}, t)}{\partial t} = -\nabla \cdot (\mathbf{v}(\mathbf{d}, t)c(\mathbf{d}, t)), \quad (15)$$

where $\nabla = [\frac{\partial}{\partial x}, \frac{\partial}{\partial y}, \frac{\partial}{\partial z}]$ denotes the gradient operator and $\mathbf{x} \cdot \mathbf{y}$ denotes the inner product of two vectors \mathbf{x} and \mathbf{y} . In general, (15) cannot be readily solved for a given velocity vector and numerical methods have to be employed [63]. Nevertheless, for the velocity vectors in Examples 3 and 4, (15) can be solved as shown in the following.

Example 5: Assuming initial condition $c(\mathbf{d}, 0)$ at $t = 0$, the advection equation (15) has the following solution for $t > 0$

$$c^*(\mathbf{d}, t) = \begin{cases} c\left(\mathbf{d} - \int_0^t \mathbf{v}(\tau) d\tau, 0\right), & \text{Uniform Flow} \\ c(\mathbf{d} - \mathbf{v}t, 0), & \text{Constant Uniform Flow} \\ c(\mathbf{d} - \mathbf{v}(\rho)t, 0), & \text{Poiseuille Flow.} \end{cases} \quad (16)$$

\square

We note that while the solutions in (16) appear similar, they are actually fundamentally different. In particular, for constant uniform flow and uniform flow (space-independent flow profiles), the initial concentration is simply translated to a different position without changing its shape. However, for Poiseuille flow (space-dependent flow profile), the concentration generally spreads in space over time depending on the initial concentration.

3) *Advection-Diffusion Equation:* In many application scenarios, such as drug delivery via the capillary networks [48]–[52], advection and diffusion are both present in the MC environment. Thereby, the combined effect of both advection and diffusion is characterized by the following PDE known as the advection-diffusion equation

$$\frac{\partial c(\mathbf{d}, t)}{\partial t} = D\nabla^2 c(\mathbf{d}, t) - \nabla \cdot (\mathbf{v}(\mathbf{d}, t)c(\mathbf{d}, t)). \quad (17)$$

⁵Kinematic viscosity ν is related to (dynamic) viscosity η according to $\nu = \eta/\rho_d$ where ρ_d [kg m^{-3}] is the fluid density.

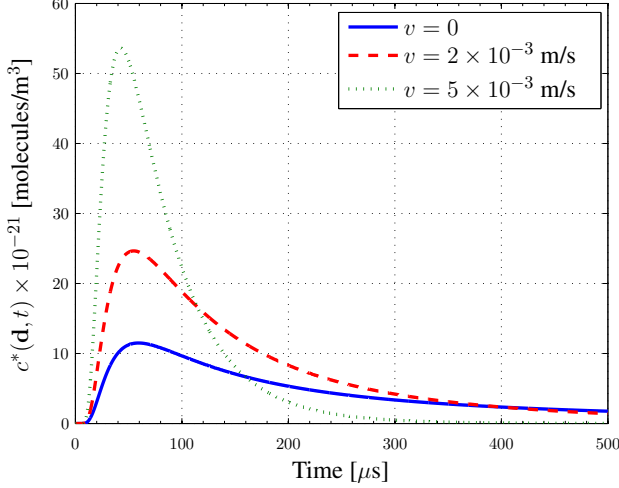


Fig. 3. Molecule concentration $c^*(\mathbf{d}, t)$ [molecules/m³] versus time [μ s] at $\mathbf{d} = [400, 0, 0]$ nm for initial release of $N = 10^4$ molecules with $D = 4.5 \times 10^{-10}$ m²/s from $\mathbf{d}_0 = [0, 0, 0]$ at $t_0 = 0$, and flow velocity $\mathbf{v} = [v, 0, 0]$ with $v \in \{0, 2, 5\} \times 10^{-3}$ m/s.

Similar to diffusion equation (3), (17) cannot be solved analytically for general velocity vectors $\mathbf{v}(\mathbf{d}, t)$ and general boundary and initial conditions. In the following, we first provide the solution of (17) for constant uniform flow in an unbounded environment. Subsequently, we quantify the relative impact of diffusion over advection by introducing the notions of Péclet number and dispersion factor.

Example 6: Consider an unbounded 3D environment with instantaneous release of N solute molecules at \mathbf{d}_0 at time t_0 . Solving (17) with initial condition IC_1 in (4), boundary condition BC_1 in (5), and constant uniform velocity vector \mathbf{v} yields [44, Eq. (18)]

$$c^*(\mathbf{d}, t) = \frac{N}{(4\pi D(t - t_0))^{3/2}} \times \exp\left(-\frac{\|\mathbf{d} - (t - t_0)\mathbf{v} - \mathbf{d}_0\|^2}{4D(t - t_0)}\right), \quad t > t_0. \quad (18)$$

□

In Fig. 3, we show molecule concentration $c^*(\mathbf{d}, t)$ [molecules/m³] in (18) versus time [μ s] at $\mathbf{d} = [400, 0, 0]$ nm for initial release of $N = 10^4$ molecules with $D = 4.5 \times 10^{-10}$ m²/s from $\mathbf{d}_0 = [0, 0, 0]$ at $t_0 = 0$, and flow velocity vector $\mathbf{v} = [v, 0, 0]$ with $v \in \{0, 2, 5\} \times 10^{-3}$ m/s. From Fig. 3, we observe that as the flow velocity increases, the concentration peak increases and t^P decreases. This is mainly due to the fact that the flow is in the same direction as the point where the concentration is measured, i.e., parallel flow is considered. Parallel flow can considerably enhance the coverage of a diffusion-based MC system, e.g., in blood vessels. Moreover, by increasing v , the tail of $c^*(\mathbf{d}, t)$ over time is decreased, which is useful for ISI reduction in MC systems [44], [64].

Relative Importance of Advection over Diffusion for Molecule Transport: Advection and diffusion can both displace and transport molecules, albeit in different ways. An important

question is under what conditions is one more effective than the other. The Péclet number, denoted by Pe , can be used to answer this question. Let us assume a velocity vector with strength v and transport over a distance d_c which is referred to as the characteristic length. The Péclet number quantifies the ratio of time required for particles to be transported by diffusion over distance d_c (which is proportional to d_c^2/D) with the time required for particles to be transported by advection over distance d_c (given by d_c/v). This ratio is given by [56, Eq. (4.44)]

$$Pe = \frac{d_c^2/D}{d_c/v} = \frac{v \cdot d_c}{D}. \quad (19)$$

Note that Pe is a dimensionless number. If $Pe \ll 1$ holds, diffusion dominates advection and the spreading of molecules is almost isotropic despite a weak biased transport in the direction of the flow. In this case, the solution of the diffusion equation (3) provides an accurate estimate of the molecule concentration. On the other hand, if $Pe \gg 1$ holds, advection dominates diffusion and is the main cause for molecule transport. In this case, the advection equation (15) can be solved to obtain an accurate estimate of the molecule concentration. Finally, for $Pe \approx 1$, molecule transport is sensitive to both diffusion and advection and the advection-diffusion equation in (17) should be solved.

Relative Importance of Advection over Diffusion for Dispersion: Let us consider a straight duct with a circular cross-section, see Examples 2 and 4, where advection is the main transport mechanism along the duct. In other words, $Pe_z \triangleq \frac{v_{\text{eff}} d_z}{D} \gg 1$ holds where Pe_z denotes the Péclet number for transport along the z -axis, $v_{\text{eff}} = v_0/2$ is the effective flow velocity in the duct (see (14)), and d_z is the desired transport length along the z -axis. In this case, we are interested in studying the dispersion (spatial spreading) of individual particles across the cross-section over the time when transport along the z -axis occurs. In particular, one may distinguish between the following two extreme regimes, namely the non-dispersive and dispersive regimes:

i) Non-dispersive regime: Here, particles do not considerably diffuse across the cross-section while being transported by advection. Therefore, each particle is simply transported along the z -axis by advection with a velocity strength that depends on the radial position of the particle, ρ , according to (14). We note that although the dispersion of individual particles is negligible in this regime, the shape of the concentration profile varies over time since the flow has a different effect at different radial positions, i.e., particles closer to the center of the duct travel faster.

ii) Dispersive regime: In the dispersive regime, particles fully diffuse across the cross-section while also being transported along the z -axis by advection. In addition to the dispersion across the cross-section, there is also dispersion along the z -axis, due to the combined impact of diffusion and advection with space-dependent flow profile (14).

In the following, we mathematically quantify the dispersive and non-dispersive regimes in terms of system parameters, i.e., v_{eff} , D , d_z , and a_c . We choose the characteristic length d_c as the distance over which the velocity vector changes

(usually a fraction of a_c). Moreover, we define $\bar{d}_z \triangleq d_z/d_c$ as the corresponding dimensionless normalized distance with respect to characteristic distance d_c . Then, we can compare the characteristic time required for particles to be transported by advection over distance d_z (given by d_z/v_{eff}) with the time required for diffusion over distance d_c (which is proportional to d_c^2/D). To compare these two time scales, we can define a dispersion factor α_d as

$$\alpha_d = \frac{d_z}{v_{\text{eff}} d_c} = \frac{D d_z}{v_{\text{eff}} d_c^2} = \frac{\bar{d}_z^2}{\text{Pe}_z}. \quad (20)$$

Here, $\alpha_d \ll 1$ signifies that there is not enough time for particles to diffuse across the cross-section while being transported by advection over distance d_z , i.e., we are in the non-dispersive regime. On the other hand, for $\alpha_d \gg 1$, diffusion causes considerable dispersion across the cross-section, which in turn causes significant dispersion along the z -axis due to space-dependent flow velocity (14), i.e., we are in the dispersive regime. In other words, in terms of the Péclet number Pe_z , we have non-dispersive and dispersive regimes if $\text{Pe}_z \gg \bar{d}_z^2$ and $\text{Pe}_z \ll \bar{d}_z^2$ hold, respectively.

Fig. 4 illustrates different dispersion regimes for a 3D straight duct. For clarity of presentation, we only show those particles for which the x -component of their position lies in interval $[-0.1a_c, 0.1a_c]$. As can be seen from Fig. 4, for $\alpha_d = 0.1$, the positions of the particles simply follow the velocity profile in (14) whereas for $\alpha_d = 10$, the particles are significantly dispersed in the environment.

C. Chemical Reactions

Another important phenomenon affecting the propagation of signaling molecules in diffusive MC systems is chemical reactions. On the one hand, chemical reactions may occur naturally in MC environments and their impact must be taken into account for communication design. On the other hand, chemical reactions have been exploited in the MC literature to achieve certain objectives, such as ISI reduction [43], [44], [65], [66] and ligand-based reception modeling [67], [68]. Therefore, in the following, we first review general chemical reactions, the corresponding reaction equations, and examples of reactions widely considered in the MC literature. Subsequently, we study the joint impact of all three phenomena discussed in this section, namely diffusion, advection, and reaction, on the propagation of the molecules and solve the corresponding advection-reaction-diffusion equation for a simple example.

1) *Reaction Equation:* Consider a general reaction of the form [69, Eq. (13)]

$$\sum_{I \in \mathcal{I}} n_I I \xrightarrow{\kappa} \sum_{J \in \mathcal{J}} n_J J, \quad (21)$$

where $I \in \mathcal{I}$ are reactant molecules, \mathcal{I} is the set of reactant molecules, $J \in \mathcal{J}$ are product molecules, \mathcal{J} is the set of product molecules, n_I and n_J are non-negative integers, and κ is the reaction rate constant. Let $c_I(\mathbf{d}, t)$ and $c_J(\mathbf{d}, t)$ denote the concentration of type- I and type- J molecules at coordinate \mathbf{d} and time t , respectively. Reactions locally change

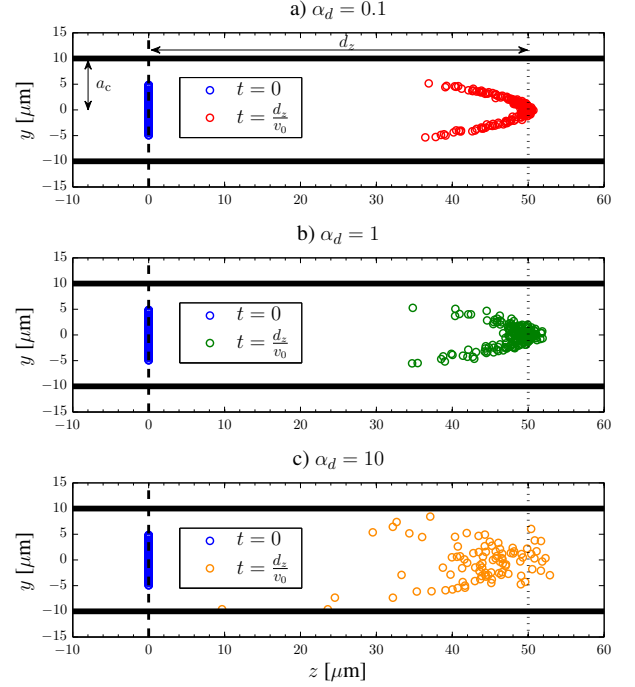


Fig. 4. Illustration of different dispersion regimes in a 3D straight duct with reflective walls, $D = 10^{-11} \text{ m}^2/\text{s}$, $a_c = 10 \mu\text{m}$, $d_c = 0.1a_c$, $d_z = 50 \mu\text{m}$, the flow velocity profile in (14), and $v_0 = 10^{-2}, 10^{-3}, 10^{-4} \text{ m/s}$ which leads to $\alpha_d = 0.1, 1, 10$, respectively. For clarity of presentation, we only show particles whose x -component of the position lies in interval $[-0.1a_c, 0.1a_c]$. The particles are initially placed at $z = 0$ and uniformly distributed in a disk with radius $5 \mu\text{m}$ centered at $(x, y) = (0, 0)$. The solid horizontal lines represent the duct walls, the dashed vertical lines denote the initial positions of the particles on the z -axis, and the dotted vertical lines denote the distance of interest on the z -axis, i.e., d_z .

the concentration of particles over time which is described by the following PDEs, known as reaction equations

$$\frac{\partial c_I(\mathbf{d}, t)}{\partial t} = -n_I f(\kappa, c_I, \forall I \in \mathcal{I}), \quad \forall I \in \mathcal{I} \quad (22a)$$

$$\frac{\partial c_J(\mathbf{d}, t)}{\partial t} = n_J f(\kappa, c_I, \forall I \in \mathcal{I}), \quad \forall J \in \mathcal{J}, \quad (22b)$$

where $f(\kappa, c_I, \forall I \in \mathcal{I})$ denotes the reaction rate function, which depends on the reaction rate constant and the concentrations of the reactant molecules. The reaction rate function has the following general form, known as the rate law [70, Eq. (9.2)]

$$f(\kappa, c_I, \forall I \in \mathcal{I}) = \kappa \prod_{I \in \mathcal{I}} c_I^{\varepsilon_I}(\mathbf{d}, t), \quad (23)$$

where ε_I is the order of the reaction with respect to type- I reactant molecules and typically takes an integer value (but in principle may also assume real values). The overall reaction order is defined as $\sum_{I \in \mathcal{I}} \varepsilon_I$ [47], [70]. Note that the units of reaction rate function $f(\kappa, c_A, c_B)$ and reaction rate constant κ are $\frac{\text{molecule}}{\text{s} \cdot \text{m}^3}$ and $\frac{1}{\text{s}} \left(\frac{\text{molecule}}{\text{m}^3} \right)^{1 - \sum_{I \in \mathcal{I}} \varepsilon_I}$, respectively.

In the following, we present three important classes of reactions, namely unimolecular degradation, bimolecular reactions, and enzymatic reactions, which can all play important roles in MC systems [44], [65], [71]–[73]. In particular, degradation

is a natural characteristic of some types of molecules and its effect has to be accounted for in communication design, see Section III-D and [44], [71]. Bimolecular reactions can be used to analyze ligand-receptor binding [67], [68] and reactive signaling [66], [74]. Enzymatic reactions have been studied in the MC literature for the purpose of ISI reduction [65], [73].

Example 7 (Unimolecular Degradation): This reaction is used to describe the degradation of a desired type of molecule, e.g., type A , into a new type of molecule, denoted by ϕ , which is of no interest for the considered communications. In fact, unimolecular degradation is often used as a first-order approximation of more complex reactions such as bimolecular and enzymatic reactions, see Examples 8 and 9. Unimolecular degradation is modeled by [70, Ch. 9]



where $\kappa \left[\frac{1}{s} \left(\frac{\text{molecule}}{\text{m}^3} \right)^{1-\varepsilon_A} \right]$ is the reaction rate constant, $f(\kappa, c_A) = \kappa c_A^{\varepsilon_A}(\mathbf{d}, t)$ is the reaction rate function, and ε_A is the reaction order. In the MC literature, first-order reactions are used to model degradation, i.e., $\varepsilon_A = 1$ [44], [71]. However, depending on the speed of reaction, higher and lower order reactions may be relevant, e.g., zero-order ($\varepsilon_A = 0$) or second-order (Type-I) ($\varepsilon_A = 2$) reactions [70, Ch. 9]. Assuming an initial condition $c_A(\mathbf{d}, t_0)$ at t_0 , (22) has the following solution for $t > t_0$

$$c_A^*(\mathbf{d}, t) = \begin{cases} [c_A(\mathbf{d}, t_0) - \kappa(t - t_0)]^+, & \text{if } \varepsilon_A = 0 \\ c_A(\mathbf{d}, t_0) \exp(-\kappa(t - t_0)), & \text{if } \varepsilon_A = 1 \\ 1/(\kappa(t - t_0) + 1/c_A(\mathbf{d}, t_0)), & \text{if } \varepsilon_A = 2, \end{cases} \quad (25)$$

where $[x]^+ = \max\{0, x\}$. Note that the speed of molecule concentration decay is hyperbolic for second-order degradations, which is faster than the exponential decay for first-order degradations, which in turn is faster than the linear decay for zero-order degradations. Nevertheless, for sufficiently large t , $c_A^*(\mathbf{d}, t)$ for second-order degradations is larger than that for first-order degradations, whereas $c_A^*(\mathbf{d}, t) = 0$, $t \geq t_0 + \frac{c_A(\mathbf{d}, t_0)}{\kappa}$, holds for zero-order degradations. \square

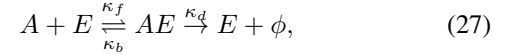
Example 8 (Bimolecular Reactions): Some reactions may involve the interaction of two reactant chemical species, e.g., A and B , to produce product molecule(s), e.g., C . For instance, in [67], the activation of ligand receptors via signaling molecules was modeled by a second-order bimolecular reaction. Moreover, in [66] and [74], acids and bases were used as reactive signaling molecules to reduce ISI. Acids and bases cancel each other out to produce salt and water. This process is modeled by a second-order bimolecular reaction. In particular, the second-order (Type-II) bimolecular reaction is given by [75]



where κ_f is the forward reaction rate constant $\left[\frac{\text{m}^3}{\text{s} \cdot \text{molecule}} \right]$, κ_b $\left[\frac{1}{\text{s}} \right]$ is the backward reaction rate constant, and $f(\kappa, c_A, c_B) = \kappa_f c_A(\mathbf{d}, t) c_B(\mathbf{d}, t)$ is the reaction rate function. The PDEs corresponding to (26) are *nonlinear* and challenging to solve. However, after introducing some approximations, in Section III, we use (26) to derive the CIRs of MC systems affected by bimolecular reactions. Moreover, let us assume $\kappa_b \rightarrow 0$ and that

the concentration of type- B molecules is sufficiently large such that the reaction in (26) does not considerably change $c_B(\mathbf{d}, t)$ over time, i.e., $c_B(\mathbf{d}, t) \approx c_B(\mathbf{d}, t = 0) \triangleq c_B(\mathbf{d})$. In this case, the bimolecular reaction in (26) can be approximated by the first-order unimolecular reaction in (24) with $\kappa = \kappa_f c_B(\mathbf{d})$ [67]. \square

Example 9 (Enzymatic Reactions): For typical scenarios, the speed of natural degradation might be too slow compared to the desired time scale of communication. In this case, enzymes can be used to accelerate the reaction process. Enzymes, denoted by E , are specific proteins that bind to the desired molecule A (also referred to as the substrate), and lower the activation energy needed for a reaction to occur. Enzymatic degradations are modeled by the following reactions [65, Eq. (1)]



where AE is an intermediate chemical species and ϕ is the product molecule. Moreover, $\kappa_f \left[\frac{\text{m}^3}{\text{s} \cdot \text{molecule}} \right]$, $\kappa_b \left[\frac{1}{\text{s}} \right]$, and $\kappa_d \left[\frac{1}{\text{s}} \right]$ denote the reaction rate constants of the forward, backward⁶, and degradation reactions, respectively. As can be seen from (27), the enzyme molecules are *not* consumed in the reaction process. The following set of PDEs, known as Michaelis-Menten kinetics, describe the evolution of the concentrations of the participating molecules

$$\frac{\partial c_A(\mathbf{d}, t)}{\partial t} = -\kappa_f c_A(\mathbf{d}, t) c_E(\mathbf{d}, t) + \kappa_b c_{AE}(\mathbf{d}, t) \quad (28a)$$

$$\frac{\partial c_E(\mathbf{d}, t)}{\partial t} = -\kappa_f c_A(\mathbf{d}, t) c_E(\mathbf{d}, t) + (\kappa_b + \kappa_d) c_{AE}(\mathbf{d}, t) \quad (28b)$$

$$\frac{\partial c_{AE}(\mathbf{d}, t)}{\partial t} = \kappa_f c_A(\mathbf{d}, t) c_E(\mathbf{d}, t) - (\kappa_b + \kappa_d) c_{AE}(\mathbf{d}, t). \quad (28c)$$

Solving the above system of *coupled* and *nonlinear* PDEs is challenging. Let us consider very fast degradation reactions, i.e., $\kappa_d \rightarrow \infty$, slow backward reactions, i.e., $\kappa_b \rightarrow 0$, and that the concentration of enzyme molecules is much larger than the concentration of type- A molecules. In this case, the formation of intermediate AE molecules does not last long and hence, we obtain $c_E(\mathbf{d}, t) \approx c_E(\mathbf{d}, t = 0) \triangleq c_E(\mathbf{d})$. In [65], it was shown that under the aforementioned assumptions, the enzymatic reaction in (27) can be approximated by the first-order unimolecular reaction in (24) with reaction rate constant $\kappa = \frac{\kappa_f \kappa_d}{\kappa_b + \kappa_d} c_E(\mathbf{d}) \approx \kappa_f c_E(\mathbf{d})$. \square

2) *Advection-Reaction-Diffusion Equation:* Next, we consider the joint effects of diffusion, drift, and reactions. For simplicity, we focus on a single molecule type and drop the corresponding subscript. In this case, the general advection-reaction-diffusion equation is given by the following PDE [32], [76]

$$\frac{\partial c(\mathbf{d}, t)}{\partial t} = D \nabla^2 c(\mathbf{d}, t) - \nabla \cdot (\mathbf{v}(\mathbf{d}, t) c(\mathbf{d}, t)) + q f(\kappa, c(\mathbf{d}, t)), \quad (29)$$

where $q = 1$ and $q = -1$ hold if the considered molecule is the product and the reactant of the reaction, respectively. Solving (29) for general initial and boundary conditions is

⁶The forward and backward reaction rate constants are also referred to as binding and unbinding reaction rate constants, respectively.

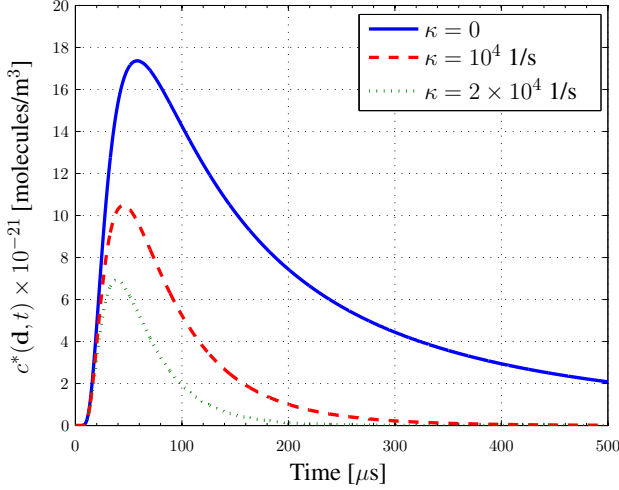


Fig. 5. Molecule concentration $c^*(\mathbf{d}, t)$ [molecules/m³] versus time [μ s] at $\mathbf{d} = [400, 0, 0]$ nm for an initial release of $N = 10^4$ molecules from $\mathbf{d}_0 = [0, 0, 0]$ and at $t_0 = 0$, $D = 4.5 \times 10^{-10}$ m²/s, flow velocity $\mathbf{v} = [10^{-3}, 0, 0]$ m/s, and $\kappa \in \{0, 1, 2\} \times 10^4$ 1/s.

again difficult for most practical MC environments. Hence, in the following, we make some simplifying assumptions that enable us to solve (29) in closed form for one example scenario [65].

Example 10: Let us assume the impulsive release of N molecules at time t_0 by a point source located at \mathbf{d}_0 into an unbounded 3D environment, i.e., initial condition IC₁ in (4) and boundary condition BC₁ in (5) hold. Moreover, we assume uniform flow $\mathbf{v}(\mathbf{d}, t) = \mathbf{v}$ and the first-order degradation reaction in (24), i.e., $q = -1$ and $f(\kappa, c(\mathbf{d}, t)) = \kappa c(\mathbf{d}, t)$. Based on these assumptions, (29) has the following closed-form solution [77], [78]

$$c^*(\mathbf{d}, t) = \frac{N}{(4\pi D(t - t_0))^{3/2}} \times \exp\left(-\kappa(t - t_0) - \frac{\|\mathbf{d} - (t - t_0)\mathbf{v} - \mathbf{d}_0\|^2}{4D(t - t_0)}\right), \quad t > t_0. \quad (30)$$

□

In Fig. 5, the molecule concentration $c^*(\mathbf{d}, t)$ [molecules/m³] is shown versus time [μ s] at $\mathbf{d} = [400, 0, 0]$ nm for an initial release of $N = 10^4$ molecules from $\mathbf{d}_0 = [0, 0, 0]$ and at $t_0 = 0$, $D = 4.5 \times 10^{-10}$ m²/s, flow velocity $\mathbf{v} = [10^{-3}, 0, 0]$ m/s, and $\kappa \in \{0, 1, 2\} \times 10^4$ 1/s. This figure shows that as the degradation rate constant increases, the concentration peak decreases, which is not desirable for an MC system, in general. However, the tail of the concentration for large t fades away much faster for larger degradation rates, which was exploited for ISI reduction in [65].

III. COMPONENT MODELING

In this section, we review the existing component models for the transmitter, receiver, and physical channel of diffusive MC systems. To this end, in Section III-A, we first define the end-to-end CIR of single-link diffusive MC systems, and

discuss the relevant mechanisms of each component and their impact on the end-to-end CIR. We use the CIR to characterize the components of MC systems, since the impulse response fully characterizes the behavior of linear systems, and linearity is commonly assumed in the MC literature⁷. Subsequently, in Sections III-B, III-C, and III-D, we review the existing models that have been developed by taking into account the impact of the receiver, transmitter, and physical channel on the end-to-end CIR, respectively. Finally, in Section III-E, we provide a summary table of all reviewed end-to-end CIR models.

A. Channel Impulse Response

In this subsection, we first briefly discuss the relevant mechanisms that characterize the functionalities of the transmitter and receiver, and the phenomena and impairments that occur in the physical channel of diffusive MC systems. Then, we provide a formal definition of what we refer to as the end-to-end channel of diffusive MC systems and we show how the CIR corresponding to the end-to-end channel can be obtained using the tools introduced in Section II.

Similar to traditional communication systems, the end-to-end chain of diffusive MC systems consists of three components, namely the transmitter, the physical channel, and the receiver; see Fig. 6. Each of these components has unique features and responsibilities, which are outlined below; see also Fig. 7.

- **Transmitter:** The transmitter is responsible for the encoding and modulation of information bits. In MC, the information is typically encoded in the number, type, or time of release of signaling molecules. Furthermore, the transmitter has to generate the signaling molecules (e.g. by CRNs inside the transmitter), store the signaling molecules (e.g. in vesicles), and control their release into the physical channel.
- **Physical Channel:** The physical channel is the environment in which the signaling molecules move and propagate once they leave the transmitter. In diffusive MC systems, the movement of signaling molecules, at its most basic level, is described by the diffusion process. However, during the course of diffusion, the random walk of signaling molecules may be affected by several other factors and noise sources such as advection, CRNs degrading the signaling molecules, environment geometry, and obstacles inside the physical channel, see Section II.
- **Receiver:** Signaling particles that reach the vicinity of the receiver can be observed and processed by the receiver to extract the information that is necessary for performing detection and decoding. The reception mechanism of the receiver *may* include the following functionalities, depending on its structure: *i)* external sensory units for detecting the presence of signaling molecules, membrane receptors of cells in nature, or sensing component(s) of macro-scale receivers such as the alcohol sensor in

⁷Linear models of MC systems provide first-order approximations of the behavior of these systems and enable further investigation and analysis. Capturing the nonlinear dynamics of complex MC systems can be achieved typically only via simulation or direct experimentation, cf. Section V, which may not provide much insight for system design.

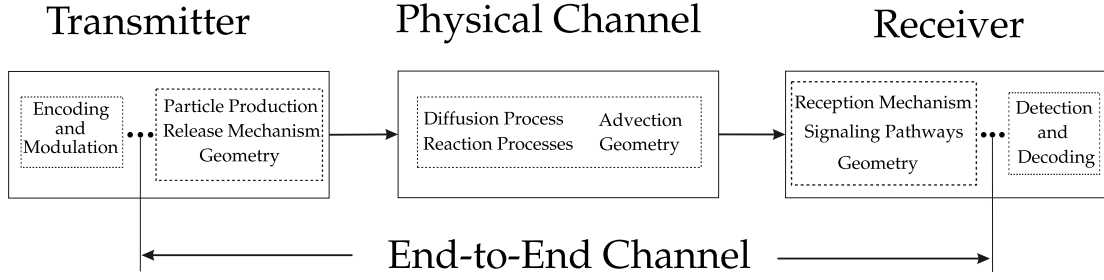


Fig. 6. Schematic presentation of the end-to-end chain of communication in typical diffusive MC systems.

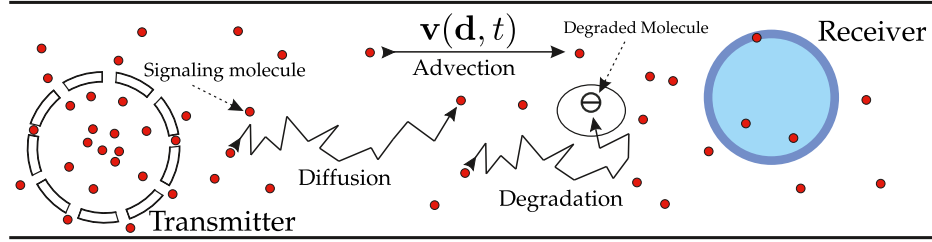


Fig. 7. Example of a physical system model including a transmitter, physical channel, and receiver.

[64] and the magnetic coils of the susceptometer in [79]; *ii*) internal relaying and interface components to convey and convert the measurements of the sensory unit into quantities suitable for detection and decoding of the information bits. For instance, in nature, this task is performed by the CRNs *inside* cells, which are referred to as *downstream signaling pathways* [18]. Downstream signaling pathways may be driven by activated receptors or directly by signaling molecules that passively enter the cells.

In the following, we formally define the end-to-end channel to study the reviewed CIR models in a unified manner.

Definition 1 (End-to-end Channel): We define the *end-to-end channel* as the effective channel that not only includes the physical channel but also the impact of the physical and chemical properties of the transmitter and receiver, including the effects of signaling molecule generation, release mechanisms, sensory units, and internal receiver components. \square

Note that our definition of the end-to-end channel does *not* include the coding, modulation, detection, and decoding operations that the transmitter and receiver may perform; see also Fig. 6. This definition of the end-to-end channel is analogous to that in traditional wireless communication systems, where the antennas, power amplifiers, and filters of the transmitter and receiver are also included in the model for the wireless end-to-end channel. The input to the end-to-end channel is the signal representing the modulated information symbol, which we also refer to as the *stimulation signal*. The stimulation signal can be an electrical (voltage or current), magnetic, mechanical, optical, chemical, or temperature signal. The output of the end-to-end channel is referred to as the observed signal and should be in a form that is suitable for the subsequent detection and decoding operations. Depending on the structure of the receiver, the observed signal can be either a number of *output molecules* or any secondary signal derived

from the output molecules. In particular, output molecules may represent: *i*) signaling molecules that can passively enter the receiver; *ii*) absorbed molecules that hit the receiver surface; or *iii*) activated receptors. Furthermore, the secondary signal derived from output molecules may be an electrical signal, e.g., the output voltage or output current of the alcohol sensor in [64]. In the following, for the definition of the CIR of the end-to-end channel, we emphasize that we consider the number of output molecules as the observed signal, as it is commonly assumed in the MC literature, although our definition can be easily extended to other forms of the observed signal.

Definition 2 (Channel Impulse Response): We define the CIR of the *end-to-end channel*, denoted by $h(t)$, as the probability of *observation* of one output molecule at time t at the receiver when the transmitter is stimulated in an impulsive manner at time $t_0 = 0$. \square

We note that defining the CIR as a probability has several advantages. In particular, it facilitates the definition of the received signal in Section IV. There, we propose a general received signal model that takes into account both the *arrival time* and the *numbers* of observed output molecules. As is shown in Section IV, both of these quantities can be readily obtained from the probability of observation of one output molecule.

In our definition of the CIR, the quantitative meaning of the term *observation* depends on the type of receiver and is defined for each considered receiver model in detail in the next subsection, e.g., for passive receivers the observed signal is defined as the number of signaling molecules inside the receiver, while for reactive receivers it is defined as the number of activated receptor molecules. Furthermore, we assume that the transmitter stimulation is an impulsive input that either controls the opening and closing of the signaling molecule reservoir or drives the CRNs inside the transmitter responsible for the generation of the signaling molecules.

In this section, we assume that the parameters of the considered MC system are constant, i.e., the end-to-end CIR $h(t)$ is time-invariant. In the following, we refer to the signaling molecules as A molecules. The following phenomena may affect the propagation of the A molecules, and as a result, $h(t)$:

- 1) **Particle generation:** Generation of the A molecules is performed, e.g., by the CRNs inside the transmitter.
- 2) **Release mechanism:** The release mechanism can be chemical, electrical, or mechanical and controls the release of the A molecules into the physical channel.
- 3) **Diffusion:** Diffusion refers to the propagation of molecules by Brownian motion.
- 4) **Degradation and production:** CRNs may degrade or produce A molecules in the physical channel.
- 5) **Advection:** Advection may affect the transportation of the A molecules in the physical channel.
- 6) **Geometry:** Potentially, the geometry of the individual components of the end-to-end channel can influence the propagation of signaling molecules.
- 7) **Receptor kinetics:** Receptor kinetics affect the interaction of the A molecules with the receptors of the sensory unit at the receiver.
- 8) **Signaling pathways:** The signaling pathways transducing the observed A molecules into a secondary signal affect the received signal.

In order to obtain $h(t)$ for a specific MC system, one has to solve the advection-reaction-diffusion equation (29) or a simplified version thereof, depending on the MC system under consideration, with the appropriate initial and boundary conditions. The initial conditions of the system capture the initial states of the CRNs, the time of production of the A molecules, and the location of the produced A molecules. The boundary conditions capture the physical and chemical properties of the components of the end-to-end channel. As discussed in the previous section, the solution to this system of PDEs does not exist in closed-form for many environments. However, as we will see in the remainder of this section, in the MC literature, different approximations have been developed to arrive at approximate yet meaningful solutions for $h(t)$ that can still capture the main effects and phenomena of the end-to-end channel. These approximate models focus on one of the components of the MC system and make simplifying assumptions about the other two. Accordingly, we will consider such receiver, transmitter, and channel centric models in the following three subsections.

B. Receiver Models

In this section, we review some of the existing end-to-end CIR models that focus particularly on the properties of the receiver, while simplifying assumptions for the transmitter and MC environment are made. The reception mechanism of the receiver can be categorized into two classes: *i*) passive reception, where the receiver does not impede the movement of signaling molecules; and *ii*) active reception, where the receiver may affect the movement of signaling molecules either by their absorption on its surface, or by chemically reacting with them

via receptors (and thereby forming ligand-receptor complexes) embedded in the receiver surface. For active reception, both mechanisms can be described by a form of chemical reaction. Moreover, the received signaling molecules may be converted via signaling pathways into secondary molecules, which can later be used for detection or decoding of the information. In nature, cells have diverse types of signaling pathways, each of which is responsible for relaying a particular type of measurement taken in the extracellular space to the organelles in the cytosol, which ultimately causes a response by the cell. For more information on the signaling pathways in natural cells, we refer the interested reader to [18].

For the CIR models considered in the following, we adopt rather simple models for the transmitter and the physical channel. Specifically, we assume that the transmitter is a *point* that releases one A molecule *instantaneously* upon stimulation at time $t_0 = 0$ at location \mathbf{d}_{tx} , where \mathbf{d}_{tx} denotes the location of the center of the transmitter; see Section III-C for more details on the point transmitter model. In other words, a point transmitter implicitly implies that upon stimulation, the A signaling molecule is *immediately* produced and enters the physical channel. We denote the location of the center of the receiver by \mathbf{d}_{rx} , and the distance between the center of the transmitter and the center of the receiver by $d_0 = \|\mathbf{d}_{tx} - \mathbf{d}_{rx}\|$. Furthermore, for the physical channel, we consider an unbounded environment affected only by diffusion noise; see Section III-D for more complex MC environments.

Passive receiver: Passive receivers (also referred to as transparent receivers or perfect monitoring receivers) employ passive reception mechanisms and are commonly considered in the MC literature, see e.g. [3], [38]–[46]. In particular, signaling A molecules in the vicinity of the receiver can enter and leave the receiver via free diffusion; see e.g. Fig. 8a). The passive receiver model is a good approximation for the diffusion of small uncharged molecules such as ethanol, urea, and oxygen. These molecules can enter and leave a cell by passive diffusion across the plasma membrane [18]. A passive receiver model is also valid for the experimental system in [79], where the susceptometer that serves as the receiver does not impede the movement of the magnetic nanoparticles passing through it. For passive receivers, the set of all points \mathbf{d} inside the volume of the receiver, \mathcal{V}_{rx} , constitutes the sensing area, and the number of A molecules in \mathcal{V}_{rx} constitutes the observed signal. Let N_{tx} denote the number of molecules that the transmitter releases. Since we are interested in computing CIR $h(t)$, i.e., the probability that a molecule released by the transmitter at $t_0 = 0$ is observed at the receiver at time t , we set $N_{tx} = 1$. Moreover, we define $p(\mathbf{d}, t) = c(\mathbf{d}, t)|_{N_{tx}=1}$ which can be interpreted as the PDF of a molecule released by the transmitter at $t_0 = 0$ with respect to \mathbf{d} at time t . In other words, $p^*(\mathbf{d}, t)dxdydz$ is the probability that the molecule is observed at time t in a rectangular cuboid of length dx , height dy , and depth dz , centered at coordinate \mathbf{d} . Since we focus on linear systems, solving $c^*(\mathbf{d}, t)$ with $N_{tx} \neq 1$ and solving $p^*(\mathbf{d}, t)$ for $N_{tx} = 1$ are related as $p^*(\mathbf{d}, t) = c^*(\mathbf{d}, t)/N_{tx}$. For the considered MC system with a point transmitter and unbounded environment, the CIR of a passive receiver can be obtained by first finding $p(\mathbf{d}, t)$

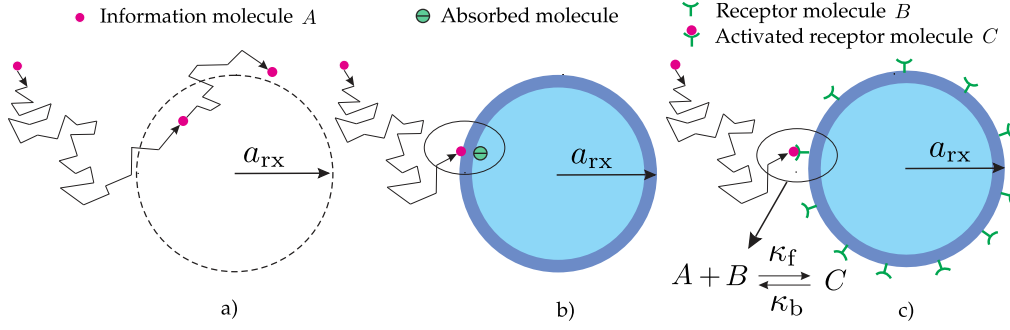


Fig. 8. Schematic depiction of three common receiver models; a) passive receiver, b) fully absorbing receiver, and c) reactive receiver.

from (3) with the following initial and boundary conditions

$$\text{IC}_3 : p(\mathbf{d}, t_0) = \delta(\mathbf{d} - \mathbf{d}_{\text{tx}}) \quad (31)$$

$$\text{BC}_3 : p(\|\mathbf{d}\| \rightarrow \infty, t) = 0. \quad (32)$$

Given the solution of (3), $p^*(\mathbf{d}, t)$, $h(t)$ can be written as

$$h(t) = \int_{\mathbf{d} \in \mathcal{V}_{\text{rx}}} p^*(\mathbf{d}, t) d\mathbf{d}. \quad (33)$$

The solution of the integral in (33) can be readily obtained when the receiver is sufficiently far away from the transmitter, i.e., d_0 is very large relative to the largest dimension of the receiver. In this case, a common approach, which is referred to as the *uniform concentration assumption* (UCA), is to approximate $p^*(\mathbf{d}, t)$ everywhere inside the volume of the receiver by its value at the center of the receiver, i.e., $p^*(\mathbf{d}, t) \simeq p^*(\mathbf{d}_{\text{rx}}, t), \forall \mathbf{d} \in \mathcal{V}_{\text{rx}}$. This leads to the following simple expression for $h(t)$, [3], [38]–[46]

$$h(t) = \frac{V_{\text{rx}}}{(4\pi Dt)^{3/2}} \exp\left(-\frac{d_0^2}{4Dt}\right), \quad (34)$$

where V_{rx} is a constant denoting the volume of the receiver. We note that (34) is valid independent of the geometry of the receiver. Specifically, the UCA is one of the most useful approximation methods in the MC literature, since it directly relates the solution of (3), (17), and (29) to the CIR of the corresponding system. Thus, many results in the rich literature on solving PDEs, see [21], can be used to obtain the CIR in MC systems with passive receivers under the UCA.

The problem of solving (33) may become cumbersome when the receiver is close to the transmitter. In this case, the solution of the integral depends on the geometry of the receiver and the UCA does not hold. It has been shown in [40, Eq. (27)] that for a *spherical* passive receiver with radius a_{rx} , $h(t)$ is given by

$$\begin{aligned} h(t) = & \frac{1}{2} \left(\text{erf}\left(\frac{a_{\text{rx}} - d_0}{\sqrt{4Dt}}\right) + \text{erf}\left(\frac{a_{\text{rx}} + d_0}{\sqrt{4Dt}}\right) \right) \\ & + \frac{\sqrt{Dt}}{a_{\text{rx}}\sqrt{\pi}} \left(\exp\left(-\frac{(a_{\text{rx}} - d_0)^2}{4Dt}\right) \right. \\ & \left. + \exp\left(-\frac{(a_{\text{rx}} + d_0)^2}{4Dt}\right) \right), \end{aligned} \quad (35)$$

where $\text{erf}(\cdot)$ denotes the error function. Eq. (34) provides an accurate approximation for (35) if $a_{\text{rx}} < 0.15 d_0$ [40].

Remark 4: We refer the interested reader to [40] for an analytical expression for $h(t)$ for a passive receiver with rectangular geometry. \square

Fully-absorbing Receiver: For fully-absorbing receivers [31], [71], [80]–[84] (also referred to as perfect sinks), unlike the passive receiver model, the physical and chemical properties of the receiver geometry are taken into account. In particular, the signaling A molecules that reach the receiver via diffusion are absorbed as soon as they hit the receiver surface, see Fig. 8b). The sensing area of a fully-absorbing receiver is defined as all points \mathbf{d} on the surface of the receiver, \mathcal{S}_{rx} , and the observed signal is the number of *absorbed* molecules during an infinitesimally small time dt . Here, a useful quantity that facilitates the derivation of $h(t)$ is the rate of absorption of the A molecule, which we denote by $k(t)$. Given $k(t)$, we have $h(t) = k(t)dt$. In other words, an absorbing receiver that measures the hitting rate of molecules on its surface can be seen as a receiver that counts the number of molecules that it absorbs in each interval of length dt and divides it by dt . Now, to derive $h(t)$, we first have to solve (3) with IC_3 (31), BC_3 (32), and the following boundary condition that models the absorption of the A molecule on the surface of the receiver

$$\text{BC}_4 : p(\mathbf{d} \in \mathcal{S}_{\text{rx}}, t) = 0, \quad (36)$$

where in a spherical coordinate system, $\mathbf{d} = [\rho, \varphi, \theta]$, for a spherical receiver with radius a_{rx} located at the origin of the coordinate system, i.e., $\mathbf{d}_{\text{rx}} = [0, 0, 0]$, we have $\mathcal{S}_{\text{rx}} = \{\mathbf{d} | \rho = a_{\text{rx}}\}$. Given $p^*(\mathbf{d}, t)$, i.e., the solution of (3) with IC_3 , BC_3 , and BC_4 , $k(t)$ is given by [85, Eq. (3.106)]

$$k(t) = 4\pi a_{\text{rx}}^2 D \frac{\partial p^*(\mathbf{d}, t)}{\partial \rho} \bigg|_{\rho=a_{\text{rx}}}. \quad (37)$$

In [80], $p^*(\mathbf{d}, t)$ for a spherical absorbing receiver is provided and $h(t)$ is calculated as [80, Eq. (22)]

$$h(t) = \frac{a_{\text{rx}}(d_0 - a_{\text{rx}})}{td_0\sqrt{4\pi Dt}} \exp\left(-\frac{(d_0 - a_{\text{rx}})^2}{4Dt}\right) dt. \quad (38)$$

Another quantity of interest is the probability that a given A molecule is absorbed by time t , $\tilde{g}(t)$, which can be obtained as

$$\tilde{g}(t) = \int_{t'=0}^t k(t') dt' = \frac{a_{\text{rx}}}{d_0} \text{erfc}\left(\frac{d_0 - a_{\text{rx}}}{\sqrt{4Dt}}\right), \quad (39)$$

where $\text{erfc}(\cdot)$ is the complementary error function.

Remark 5: Alternatively, when the receiver counts the number of absorbed molecules during observation window $[t_u, t_l]$, $h(t)$ can be defined as

$$\begin{aligned} h(t) &= \tilde{g}(t_u) - \tilde{g}(t_l) \\ &= \frac{a_{rx}}{d_0} \left[\operatorname{erfc} \left(\frac{d_0 - a_{rx}}{\sqrt{4Dt_u}} \right) - \operatorname{erfc} \left(\frac{d_0 - a_{rx}}{\sqrt{4Dt_l}} \right) \right]. \end{aligned} \quad (40)$$

□

Remark 6: For a *fully-absorbing receiver*, it is implicitly assumed that the *whole* surface of the receiver is fully-absorbing. The extension of this model to the case where the receiver surface is *partially* covered by *fully* absorbing receptor patches is considered in [81]. Moreover, the extension of the fully-absorbing receiver to take the impact of degradation and production noise into account, is considered in [71]. □

Remark 7: We note that one of earliest CIR models taking the absorption of particles in a *1D* diffusion channel with uniform drift into account is proposed in [86]. There, a closed-form expression is derived for the probability of the time of absorption of the signaling molecules. □

Reactive Receiver: Large or polar signaling molecules cannot passively diffuse through the membrane of cells and are detected by external receptors embedded in the cell membrane. In particular, the diffusive signaling *A* molecules that reach the cell *may* participate in a *reversible* bimolecular second-order reaction with receptor protein *B* molecules on the cell surface and form ligand-receptor complex *C* molecules; see Fig. 8c). The ligand-receptor interaction can be modeled as shown in (26) with binding reaction rate constant κ_f in $[\text{molecule}^{-1} \cdot \text{m}^3 \cdot \text{s}^{-1}]$ and unbinding reaction rate constant κ_b in $[\text{s}^{-1}]$. For such reactive receivers, the sensing area is that part of the receiver surface that is covered by receptors, denoted by \tilde{S}_{rx} , and the number of activated receptors *C* constitute the received signal. We refer the interested reader to [67] for a closed-form CIR expression for reactive receivers.

Remark 8: In [68], a reactive receiver with an infinite number of receptor *B* molecules covering the whole surface of the receiver, \tilde{S}_{rx} (i.e., a homogeneous receiver surface, which is a special case of [67]), was considered and the corresponding CIR was numerically evaluated. Furthermore, in the MC literature, first steps to take the impact of ligand-receptor interaction on the CIR into account are made in [6] and [87]. There, for the evaluation of $h(t)$, the diffusion equation and the reaction equation are solved separately, unlike [67], [68] where a *coupled* diffusion-reaction equation is considered. □

Remark 9: The fully-absorbing receiver is a special case of the reactive receiver when the whole surface of the receiver is covered with infinitely many *B* molecules, $\kappa_b = 0$, and $\kappa_f \rightarrow \infty$. In this case, reaction equation (26) becomes a pseudo first-order reaction of the form $A \rightarrow C$, with binding reaction rate constant $\kappa_f \rightarrow \infty$, where now *C* corresponds to the number of absorbed molecules. However, $\kappa_f \rightarrow \infty$ implies that any collision of a signaling *A* molecule with the receiver surface leads to the formation of a *C* molecule, i.e., the reaction is deterministic. We refer the interested reader to [67] where it is shown how the CIR of the reactive receiver, under the above assumptions, simplifies to the CIR of the fully-absorbing receiver. □

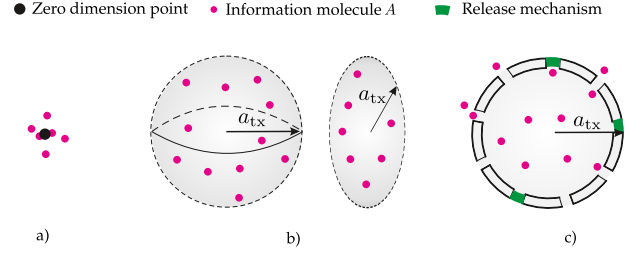


Fig. 9. Schematic depiction of transmitter models; a) point transmitter, b) volume transmitter, and c) ion-channel based transmitter.

Remark 10: A receiver model that, unlike the CIR models reviewed in this section so far, also accounts for the impact of the *signaling pathways*, is proposed in [88]. In that model, two simple approximate signaling pathways, modeled via first-order and second-order CRNs, are considered. The CIR model in [88] is derived based on a mesoscopic modeling approach; see Section V for more details on mesoscopic modeling. □

C. Transmitter Models

In this section, we review some of the existing end-to-end CIR models developed in the MC literature that mainly focus on the properties of the transmitter. The main features of the transmitter that can potentially affect the end-to-end CIR include: *i*) the geometry of the transmitter, i.e., the volume, boundaries, and shape of the transmitter [2], [23]; *ii*) the particle generation via chemical reactions, which can take different forms ranging from a simple zero-order production reaction to more complex CRNs that take several aspects of *A* molecule generation into account including, e.g., energy consumption via hydrolyzation of adenosine triphosphate (ATP) molecules [18]; and *iii*) the release mechanism controlling the release of the *A* molecules into the physical channel. In particular, after production, the *A* molecules can leave the transmitter either *passively*, for instance by passive diffusion through channels or gates embedded in the hull of the transmitter, or *actively*, for example via pumps integrated in the hull of the transmitter. In nature, passive and active transportation occur in cells via ion channels and transporters, respectively, see [18]. In the following, we study transmitter models that partially take the effects of the geometry, release mechanisms, and particle generation into account.

Point Transmitter: The point transmitter is the most widely used transmitter model in the MC literature mainly due to its simplicity, see [3]. However, this model takes none of the above mentioned features into account. In particular, the point transmitter, as the name suggests, is modelled as a zero-dimensional point, i.e., the impact of the geometry of a physical transmitter is not included in the model; see Fig. 9a). Furthermore, it is commonly assumed that the *A* molecules are produced *instantaneously* and enter the physical channel *immediately*. These assumptions imply that the effects of the particle generation and the release mechanism on $h(t)$ are neglected.

Volume Transmitter: Unlike point transmitters, where all *A* molecules are generated at the same location, volume

transmitter models take the transmitter geometry into account by assuming that the A molecules are initially distributed over the transmitter volume⁸ [89]; see Fig. 9b). This leads to more realistic models since, in reality, signaling molecules are physical quantities that occupy space. However, volume transmitter models assume that the A molecules are generated *instantaneously*, and that the surface of the transmitter is transparent and does not impede the diffusion of the A molecules. With these two assumptions, volume transmitters neglect the effect of the particle generation and the impact of the release mechanisms. Let us, for the moment, denote the CIR models obtained for a point transmitter model, e.g., (35), (34), (38), by $h^\bullet(t, d_0)$. Then, employing the principle of superposition and assuming a uniform particle distribution over the volume of the transmitter, V_{tx} , the CIR of the corresponding volume transmitter can be written as [89, Eq. (12)]

$$h(t) = \frac{1}{V_{tx}} \int_{\mathbf{d} \in V_{tx}} h^\bullet(t, \|\mathbf{d} - \mathbf{d}_{rx}\|) d\mathbf{d}, \quad (41)$$

where V_{tx} denotes the volume of the transmitter.

Remark 11: In [89], (41) is solved numerically for a 3D spherical transmitter and both passive and fully-absorbing receivers. Furthermore, in [89], closed-form expressions are given for corresponding *one-dimensional* scenarios. \square

One useful approximation of (41) can be obtained when the transmitter is sufficiently far away from the receiver. Then, the distance of any point inside the transmitter to the receiver can be approximated by d_0 and (41) simplifies to

$$h(t) \approx \frac{h^\bullet(t, d_0)}{V_{tx}} \int_{\mathbf{d} \in V_{tx}} d\mathbf{d} = \frac{h^\bullet(t, d_0)}{V_{tx}} \times V_{tx} = h^\bullet(t). \quad (42)$$

We refer the interested reader to [89], where the accuracy of the above approximation has been investigated for several environments.

Remark 12: The analytical transmitter models in [89] assume that the A molecules are generated throughout V_{tx} . The authors of [89] and [90] simulated a volume transmitter model where the A molecules are generated on the surface of a reflective spherical transmitter. In [90], a parametric model is proposed for the CIR of an MC system employing the considered transmitter and a fully-absorbing receiver. A machine learning approach is used to obtain the parameters of the parametric model. \square

Ion-Channel Based Transmitter: Ion-channel based (IC) transmitters are considered in [91] to model the effect of the release of the signaling molecules into the physical channel. IC transmitters take both the transmitter geometry and the release mechanism into account. In particular, IC transmitters are modelled as spherical objects with *ion-channels* embedded in their membrane; see Fig. 9c). The opening and closing of the ion-channels is controlled via a so-called gating parameter such as a voltage or a ligand. When the gating parameter is applied, e.g., in the form of a voltage across the transmitter membrane, the ion-channels open and the A molecules can leave the transmitter via passive diffusion. The impact of the particle

generation is neglected in [91]. In particular, it is assumed that the A molecules diffuse with different diffusion coefficients inside and outside the transmitter. In [91] an expression is derived for the average rate of signaling molecules entering the physical channel upon transmitter stimulation. Furthermore, an approximate solution for the CIR of the corresponding end-to-end channel is provided under the conditions that the entire surface of the transmitter is covered by a large number of open ion-channels and that the signaling molecules diffuse with the same diffusion coefficient inside and outside the transmitter. There, a passive receiver under the UCA and an unbounded environment are assumed. Then, the CIR is approximated as [91, Eq. (42)]

$$h(t) = \frac{a_{tx}}{d_0 \sqrt{2Dt}} \exp\left(\frac{-(d_0^2 + a_{tx}^2)}{4Dt}\right) \sinh\left(\frac{d_0 a_{tx}}{2Dt}\right), \quad (43)$$

where $\sinh(\cdot)$ denotes the hyperbolic sine function. In fact, (43) is actually the CIR of a volume transmitter, since the assumption of having many open ion-channels is equivalent to assuming that the entire surface of the transmitter is a transparent membrane.

Remark 13: The transmitter models reviewed so far do not consider the impact of particle generation via CRNs inside the transmitter. This is mainly due to the fact that to take particle generation into account, a coupled reaction-diffusion equation has to be solved, which is a challenging task. Nevertheless, the effect of particle generation has been studied in [9], [92], [93]. A common methodology for solving the corresponding reaction-diffusion equation is to adopt mesoscopic models and to *numerically* solve the problem; see Section V-A for a review of numerical methods. \square

D. Physical Channel Models

In this section, we review some of the existing end-to-end CIR models that emphasize the phenomena or impairments of the physical channel. In diffusive MC systems, the signaling molecules that enter the physical channel may be affected by several factors and noise sources besides diffusion, including: *i*) advection that can be constructive or destructive depending on the direction and strength of the velocity vector; *ii*) the geometry of the physical channel, e.g., bounded or unbounded environments, constraining the dispersion of the particles; and *iii*) degradation and production of A molecules. For the CIR models reviewed in this section, in order to be able to focus on how $h(t)$ is affected by the phenomena in the physical channel, we adopt the point or volume transmitter model and the passive receiver model.

Bounded Diffusion Channels: The CIR models reviewed previously were obtained under the assumption of an unbounded physical channel. Now, we focus on CIR models that assume a more elaborate physical channel geometry. To determine $h(t)$ for *bounded* physical channels, generally, one has to solve a diffusion equation (3) with appropriate boundary conditions reflecting the physical and chemical properties of the geometry of the channel. Unfortunately, for many practical geometries, simple and insightful solutions of (3) do not exist. Thus, approximations are needed to model practical geometries.

⁸We note that, here, the term “volume” is generic and may refer to a volume or a surface in a 3D space, a surface or a line in a 2D space, and a line in a 1D space.

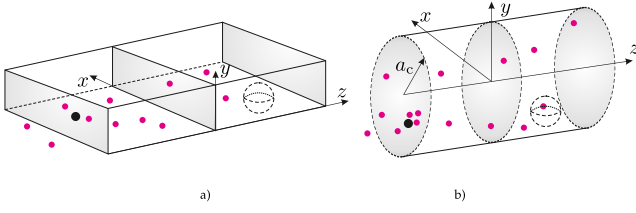


Fig. 10. Schematic presentation of two duct channels with a) rectangular and b) circular cross sections.

In the following, we focus on a class of bounded physical channels that are referred to as *duct channels*. In particular, we consider duct channels with *rectangular* and *circular* cross sections; see Fig. 10. These two duct channels are of particular importance since they approximate the geometry of microfluidic channels and blood vessels, respectively.

- **Rectangular Duct Channel:** For a rectangular duct channel with dimensions $-\infty < z < +\infty$, $0 < x < l_x$, $0 < y < l_y$, fully reflective walls, a point transmitter at $\mathbf{d}_{tx} = [x_{tx}, y_{tx}, z_{tx}]$, and a receiver at $\mathbf{d}_{rx} = [x_{rx}, y_{rx}, z_{rx}]$, the CIR can be obtained by solving (3) for $p(\mathbf{d}, t)$ with IC_3 and the following boundary conditions

$$BC_5 : \left. \frac{\partial p(\mathbf{d}, t)}{\partial x} \right|_{x=\{0, l_x\}} = 0, \quad (44)$$

$$BC_6 : \left. \frac{\partial p(\mathbf{d}, t)}{\partial y} \right|_{y=\{0, l_y\}} = 0, \quad (45)$$

$$BC_7 : p(\mathbf{d} = [x, y, z \rightarrow \pm\infty], t) = 0, \quad (46)$$

where BC_5 and BC_6 capture the reflection of the A molecule on the duct walls. Since, for the considered geometry, the diffusion of the A molecule in one Cartesian coordinate does not influence its diffusion in the other coordinates, we can write $p(\mathbf{d}, t) = p(x, t) \times p(y, t) \times p(z, t)$. Now, solving (3) for $p(x, t)$, $p(y, t)$, and $p(z, t)$ with BC_5 , BC_6 , and BC_7 , respectively, and considering a passive receiver under the UCA, $h(t)$ can be obtained as follows [21, Eq. (14.4.4)]

$$h(t) = \frac{V_{rx}}{l_x l_y} \left[1 + 2 \sum_{n=1}^{\infty} e^{-Dn^2\pi^2 t/l_x^2} \cos\left(\frac{n\pi x_{rx}}{l_x}\right) \times \cos\left(\frac{n\pi x_{tx}}{l_x}\right) \right] \times \left[1 + 2 \sum_{n=1}^{\infty} e^{-Dn^2\pi^2 t/l_y^2} \times \cos\left(\frac{n\pi y_{rx}}{l_y}\right) \cos\left(\frac{n\pi y_{tx}}{l_y}\right) \right] \times \left[\frac{1}{\sqrt{4D\pi t}} \exp\left(\frac{-(z_{rx} - z_{tx})^2}{4Dt}\right) \right]. \quad (47)$$

- **Circular Duct Channel:** For a circular duct channel with dimensions $0 < \rho < a_c$, $0 < \theta < 2\pi$, $-\infty < z < +\infty$ in cylindrical coordinates, fully reflective walls, a point transmitter at $\mathbf{d}_{tx} = [\rho_{tx}, \varphi_{tx}, z_{tx}]$, and a receiver at $\mathbf{d}_{rx} = [\rho_{rx}, \varphi_{rx}, z_{rx}]$, the CIR can be derived by solving (3) with IC_3 (31) and the following boundary conditions

$$BC_8 : \left. \frac{\partial p(\mathbf{d}, t)}{\partial \rho} \right|_{\rho=a_c} = 0, \quad (48)$$

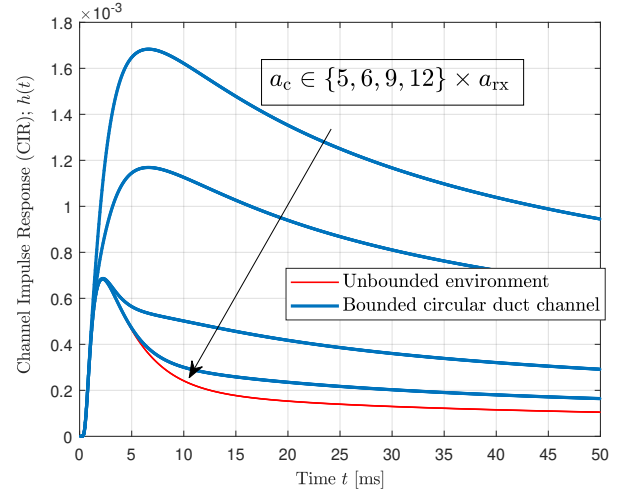


Fig. 11. Channel impulse response, $h(t)$, as a function of time t , for an unbounded environment and a bounded circular duct channel. The duct radius increases in the direction of the arrow.

$$BC_9 : p(\mathbf{d} = [\rho, \varphi, z \rightarrow \pm\infty], t) = 0. \quad (49)$$

Here, BC_8 models the reflection of the A molecule at the wall of the duct with radius a_c . Employing the same technique as for rectangular duct channels, using [21, Eq. (14.13.7)] and considering a passive receiver under the UCA, $h(t)$ can be obtained as follows

$$h(t) = \frac{V_{rx} \exp(-(z_{rx} - z_{tx})^2/4Dt)}{2\pi a_c^2 \sqrt{\pi Dt}} \times \left[1 + \sum_{n=-\infty}^{+\infty} \cos(n(\varphi_{rx} - \varphi_{tx})) \right] \times \sum_{\alpha} \exp(-D\alpha^2 t) \frac{\alpha^2 J_n(\alpha \rho_{rx}) J_n(\alpha \rho_{tx})}{(\alpha^2 - n^2/a_c^2) J_n^2(a_c \alpha)}, \quad (50)$$

where the summation in α is over the positive roots of $J'_n(\alpha a_c) = 0$. Here, $J_n(\cdot)$ denotes the n -th order Bessel function of the first kind and $J'_n(\cdot)$ denotes its derivative.

The CIR expressions in (47) and (50) indicate that even for simple bounded environments, the solution to $h(t)$ may be quite complicated and difficult to interpret. To gain more insight, in Fig. 11, we compare the CIR of an unbounded physical channel to that of a circular duct channel for system parameters $\mathbf{d}_{tx} = [0, 0, -1.15] \mu\text{m}$, $\mathbf{d}_{rx} = [0, 0, 0] \mu\text{m}$, receiver radius $a_{rx} = 0.15 \mu\text{m}$, and $a_c \in \{5, 6, 9, 12\} \times a_{rx}$. Fig. 11 shows that when duct radius a_c is small, the CIR of the duct channel is much larger than the CIR of the unbounded channel, i.e., for a given time t it is more likely to observe the signaling molecule. This is because when a_c is small, the signaling molecule is reflected more frequently on the duct walls which increases its chance of being observed at the receiver compared to the unbounded case where the A molecules can diffuse away. However, for large a_c , the CIR of the duct channel approaches the CIR of the unbounded environment, i.e., the CIR of the unbounded channel provides a good approximation for the CIR of a large bounded circular duct channel.

Remark 14: We note that the necessary condition for the validity of the UCA developed for passive receivers in unbounded channels, i.e., $a_{\text{rx}} < 0.15d_0$, is not applicable for bounded physical channels. However, we expect that as $a_{\text{rx}}/d_0 \rightarrow 0$, the accuracy of (47) and (50) improves. \square

Advection Channels: Next, we consider physical channels in which the signaling molecules experience advection in addition to their random walk. In particular, for the CIR models reviewed in this section, for mathematical tractability, we consider advection processes with a time-invariant velocity field, i.e., we assume $\mathbf{v}(\mathbf{d}, t) = \mathbf{v}(\mathbf{d})$, $\forall t > t_0$.

- **Uniform constant advection:** In this case, the magnitude and the direction of the velocity field are identical at any point \mathbf{d} in space, i.e., $\mathbf{v}(\mathbf{d}) = \mathbf{v} = [v_x, v_y, v_z]$, $\forall \mathbf{d} \in \mathcal{R}^3$, where \mathcal{R}^3 is the set of all points in the 3D Cartesian coordinate system, see Example 3 in Section II. Vector \mathbf{v} can be effectively decomposed into two components, a parallel component v_{\parallel} and an orthogonal component v_{\perp} with respect to $\mathbf{d}_0 = \mathbf{d}_{\text{tx}} - \mathbf{d}_{\text{rx}}$. Let us assume, without loss of generality, a point transmitter at $\mathbf{d}_{\text{tx}} = [0, 0, -z_{\text{tx}}]$ and a passive receiver located at $\mathbf{d}_{\text{rx}} = [0, 0, 0]$, such that $\mathbf{d}_0 = [0, 0, -z_{\text{tx}}]$. Then, $v_{\parallel} = v_z$, and we can write $v_{\perp} = \sqrt{v_x^2 + v_y^2}$.

- **Unbounded Channel with UCA:** For an unbounded channel and a passive receiver under the UCA, $h(t)$ can be obtained by solving advection-diffusion equation (17). Using the method of *moving reference frame*, i.e., assuming that the reference of the coordinate system is moving with \mathbf{v} , it can be readily verified that $h(t)$ can be obtained from (34) as [44, Eq. (18)]

$$h(t) = \frac{V_{\text{rx}}}{(4\pi Dt)^{3/2}} \times \exp\left(-\frac{(v_{\perp}t)^2 + (z + z_{\text{tx}} - v_{\parallel}t)^2}{4Dt}\right). \quad (51)$$

Eq. (51) can be also directly obtained from (18) after setting $N = 1$, multiplying $c^*(\mathbf{d}, t)$ with V_{rx} , and using the \mathbf{v} , \mathbf{d}_{tx} , and \mathbf{d}_{rx} mentioned above.

- **Unbounded channel without UCA:** For the case when the UCA does not hold, $v_{\parallel} \neq 0$, and $v_{\perp} \neq 0$, $h(t)$ can be solved numerically. However, it is shown in [44] that for the special case of $v_{\perp} = 0$, $h(t)$ can be obtained from (35) after substituting d_0 with $-(z_{\text{tx}} - v_{\parallel}t)$.
- **Bounded channel with UCA:** In this case, i.e., when we have bounded channels such as duct channels, and for the general case where $v_{\parallel} \neq 0$, $v_{\perp} \neq 0$, we cannot apply the technique of *moving reference frame* in the dimensions of the coordinate system where the physical channel is bounded. Thus, $h(t)$ has to be directly evaluated via the corresponding advection-diffusion equation. However, for the special case of $v_{\perp} = 0$, after substituting z_{tx} with $z_{\text{tx}} - v_{\parallel}t$, the corresponding CIRs of the rectangular and circular duct channels can be obtained from (47) and (50), respectively.
- **Bounded channel without UCA:** In this case, the general form of $h(t)$ depends on the geometries of the bounded physical channel and the passive receiver. However,

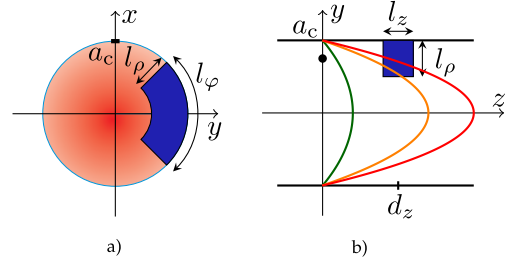


Fig. 12. Schematic presentation of a circular duct channel with radius a_c and laminar flow; a) cross-section and b) along the z axis. The receiver is depicted in blue color.

for a rectangular duct channel, a rectangular passive receiver, and $v_{\parallel} \neq 0$, $v_{\perp} \neq 0$, an analytical expression for $h(t)$ is derived in [54]. We note that in [54], it is assumed that v_{\parallel} and v_{\perp} are a fluid velocity field and a drift velocity caused by a magnetic field, respectively. However, the derived expression for $h(t)$ is valid independent of the origin of v_{\parallel} and v_{\perp} . Furthermore, in [54], the case of partially absorbing duct channel walls is also considered.

- **Laminar flow:** In this case, we only focus on bounded channels, and in particular on circular duct channels, since laminar flow arises in bounded environments. Thus, we consider $\mathbf{v}(\mathbf{d})$ given in (14). For the CIR models reviewed here, we distinguish between point and volume transmitter models with axial position $z_{\text{tx}} = 0$, and consider the passive receiver model with the following dimensions in cylindrical coordinates $a_c - l_\rho \leq \rho_{\text{rx}} \leq a_c$, $|\varphi_{\text{rx}}| \leq l_\varphi/2$, $|z_{\text{rx}} - d_z| \leq l_z/2$; see Fig. 12. In particular, we distinguish between two cases, namely the dispersion regime ($\alpha_d \gg 1$) and the flow dominant regime ($\alpha_d \ll 1$), see (20).
- **Dispersion regime with UCA:** In the dispersion regime, $\alpha_d \gg 1$ holds in (20). As a result, the time required for transportation of the A molecule in the z direction via flow, d_z/v_{eff} , is much larger than a_c^2/D , which is the characteristic time for diffusion of the A molecule over distance a_c . This fact has two immediate consequences: *i*) by the time that the released A molecule reaches the receiver, it experiences the average flow velocity, i.e., v_{eff} , due to its fast diffusion across the cross section; *ii*) there is no difference between point and uniform release and $h(t)$ only depends on z . Thus, the corresponding advection-diffusion equation in three dimensional space can be effectively approximated by its one dimensional equation with effective velocity v_{eff} and effective diffusion coefficient D_{eff} as follows

$$\partial_t p(z, t) = D_{\text{eff}} \partial_z^2 p(z, t) - v_{\text{eff}} \partial_z p(z, t), \quad (52)$$

where D_{eff} is the Aris-Taylor effective diffusion coefficient and can be obtained as [57, Eq. (4.6.35)]

$$D_{\text{eff}} = \left(1 + \frac{1}{48} \left(\frac{(v_{\text{eff}} a_c)^2}{D}\right)\right). \quad (53)$$

Solving (52) with the UCA approximation, BC₉, and the following initial condition for uniform release across the cross section

$$\text{IC}_4 : p(z, t_0) = \frac{1}{\pi a_c^2} \delta(z) \quad (54)$$

leads to [61, Eq. (11)]

$$h(t) = \frac{V_{\text{rx}}}{\pi a_c^2} \times \frac{1}{\sqrt{4\pi D_{\text{eff}} t}} \exp\left(-\frac{(d_z - v_{\text{eff}} t)^2}{4D_{\text{eff}} t}\right). \quad (55)$$

- *Dispersion regime without UCA*: In this case, $h(t)$ can be obtained by taking the integral of the solution of (52) over the volume of the receiver, which leads to [61, Eq. (13)]

$$h(t) = \frac{l_\varphi(2a_c l_\rho - l_\rho^2)}{2\pi a_c^2} \times \left[Q\left(\frac{d_z - l_z/2 - v_{\text{eff}} t}{2D_{\text{eff}} t}\right) + Q\left(\frac{d_z + l_z/2 - v_{\text{eff}} t}{2D_{\text{eff}} t}\right) \right], \quad (56)$$

where $Q(\cdot)$ denotes the Gaussian Q-function.

Remark 15: We note that the accuracy of both (55) and (56) depends on the value of α_d in (20). For example, by increasing D and d_z , α_d increases and the accuracy of the dispersion regime improves; see [61]. \square

- *Flow dominant regime with volume transmitter*: In this case, i.e., $\alpha_d \ll 1$, the impact of diffusion is negligible. Thus, the signaling molecules do not have sufficient time to disperse across the cross section of the duct channel before they arrive at the receiver. As a result, a particle released at radial position ρ is observed approximately at the same radial position at the receiver. Thus, we have to distinguish between the volume and point transmitter models. For the case of uniform release, $h(t)$ can be approximated as [61, Eq. (16)]

$$h(t) = \begin{cases} 0, & t \leq t_1 \\ \frac{c_\varphi(2a_c l_\rho - l_\rho^2)}{2\pi a_c^2} - \frac{l_\varphi}{2\pi} \frac{d_z - l_z/2}{2v_{\text{eff}} t}, & t_1 < t < t_2 \\ \frac{l_\varphi}{2\pi} \frac{l_z}{2v_{\text{eff}} t}, & t \geq t_2. \end{cases} \quad (57)$$

where

$$t_{1,2} = \frac{d_z \mp l_z/2}{2v_{\text{eff}}(1 - (1 - l_\rho/a_c)^2)}. \quad (58)$$

In (58), t_1 and t_2 are the times when the parabolic velocity field first hits and leaves the receiver volume, respectively.

- *Flow dominant regime with point transmitter*: For the case of a point transmitter, when the A molecule is released d_z away from the receiver but within the ρ and φ coordinates defining the dimensions of the receiver, i.e., at $\rho_{\text{tx}} \in [a_c - l_\rho, a_c]$ and $\varphi_{\text{tx}} \in [-l_\varphi/2, l_\varphi/2]$, we observe the A molecule with certainty if $d_z - l_z/2 \leq v(\rho_{\text{tx}})t \leq d + l_z/2$

$$h(t) = \text{rect}\left(\frac{v(\rho_{\text{tx}})t - d_z}{l_z}\right), \quad (59)$$

where $\text{rect}(x) = 1$ if $-1/2 \leq x \leq 1/2$.

Remark 16: In the MC literature, first steps towards the extension of some of the CIR models reviewed for the advection channel to more complex networks of interconnected bounded duct channels are provided in [49], [52], [94], [95]. For example, in [94], the Aris-Taylor effective diffusion coefficient approximation is employed to calculate the end-to-end CIR of multiple interconnected blood vessels for drug delivery applications. Furthermore, in [52], the uniform advection model is adopted to model blood vessel networks for abnormality detection applications in biological systems. \square

Degradation Channels: In degradation channels, the arrival of the signaling molecules is affected by their possible degradation and production. In this case, $h(t)$ can be obtained by solving the diffusion-reaction equation (29) (with $\mathbf{v}(\mathbf{d}, t) = 0$) given appropriate initial and boundary conditions. However, the solution to (29) depends very much on the structure of the corresponding CRN described by reaction rate function $f(\cdot)$. Often, the reaction terms in $f(\cdot)$ are highly nonlinear and coupled, which makes the problem of solving (29) very challenging. Here, in order to arrive at mathematically tractable and insightful results describing the *general* behaviour of degradation channels, we focus on two particular forms of degradation and production noise, namely *first-order degradation* and *enzymatic degradation*; see Examples 7 and 9, and [27], [65], [67], [71].

- *First-order degradation*: Let us for the moment denote the CIR expressions developed in the previous sections by $\tilde{h}(t)$. It can be shown that $h(t)$, for a physical channel with first-order degradation reaction of the form (24) and reaction rate constant κ can be readily obtained from $\tilde{h}(t)$ when the following assumptions hold. A1) The signaling A molecule is affected by the degradation reaction uniformly and equally throughout the entire *end-to-end channel*, and A2) it is not involved in any other CRN from stimulation time t_0 until observation time t . In this case, we can write

$$h(t) = \tilde{h}(t) \times \exp(-\kappa t). \quad (60)$$

In (60), the term $\exp(-\kappa t)$ captures the *surviving probability* of the signaling molecule, which is a monotonically decreasing function of time, i.e., as t increases, it becomes more likely that the signaling molecule A is degraded in the channel. As a result, for degradation channels, at any instant, $h(t)$ is smaller than the corresponding CIR without degradation.

Assumption A2) holds for all CIR models presented so far except the CIR model for the reactive receiver in Section III-B. This is because for reactive receivers, the signaling molecule is involved in a ligand-receptor kinetic reaction (26), in addition to the degradation reaction, and may experience several binding/unbinding events before reception time t . We refer the interested reader to [67], where a closed-form expression is derived for the CIR of an MC system with a reactive receiver in a first-order degradation channel.

- *Enzymatic degradation*: The impact of enzymatic degradation reactions in the channel is studied for passive receiver and point transmitter models in [65]. Enzymatic reactions include a second-order reaction, and as a result, in order

to obtain $h(t)$, we have to solve (29) with IC_3 , BC_3 , and $f(\cdot)$ driven by (28). However, this system of nonlinear equations does not facilitate a closed-form solution for $h(t)$. As a result, in [65], several approximate solutions were proposed for the CIR of an MC system with point transmitter, passive receiver, and unbounded environment as follows:

- If the concentration of signaling molecules, $c_A(\mathbf{d}, t)$, and the concentration of the intermediate species AE , $c_{AE}(\mathbf{d}, t)$, remain constant over time and space, then $f(\cdot)$ is described only by (28a) and $h(t)$ has the following solution

$$h(t) \approx \frac{V_{rx}}{(4\pi Dt)^{3/2}} \exp\left(-\kappa_f c_{E_t} t - \frac{d_0}{4Dt}\right) + \kappa_b c_{AE} t. \quad (61)$$

- If $\kappa_d \rightarrow \infty$ and $\kappa_b \rightarrow 0$, then the total concentration of the enzyme E molecules remains constant and $c_{AE} = 0$. Then, a lower bound for $h(t)$ is obtained via (61) after setting $c_{AE} = 0$.
- Another useful approximation is obtained by assuming that c_{AE} is constant [65]. This is a valid assumption when $\kappa_f \rightarrow \infty$ and $\kappa_b \rightarrow 0$. Then, as explained in Section II, the enzymatic reaction in (27) can be approximated by the first-order unimolecular reaction in (24) and the corresponding $h(t)$ can be written as

$$h(t) \approx \frac{V_{rx}}{(4\pi Dt)^{3/2}} \exp\left(-\frac{\kappa_f \kappa_d}{\kappa_b + \kappa_d} c_{E_t} t - \frac{d_0}{4Dt}\right), \quad (62)$$

where c_{E_t} denotes the total concentration of the enzyme E molecules, including both bound and unbound enzyme. We refer the interested reader to [65] for verification of the accuracy of the proposed approximate expressions for $h(t)$.

Remark 17: Signaling molecules of different types may also degrade each other. For instance, the MC testbed in [66] uses acids and bases as signaling molecules that can participate in a bimolecular reaction and cancel each other out, cf. (26). Unfortunately, the underlying PDEs that describe the bimolecular reaction are coupled and nonlinear and closed-form expressions for the CIR are not available. In [74], a numerical method was developed which decouples reaction and diffusion in each time slot and computes the channel response in an iterative manner. \square

E. Summary of End-to-End CIR Models

To conclude this section, we provide a summary of the reviewed CIR models in Table II. Although the keywords, notations, and variables used in this table are mostly self-explanatory, for clarity and completeness, we briefly explain them in the following. For the transmitter, “ Ω_{tx} ” indicates whether a point transmitter is assumed or a volume transmitter releasing the molecules from its volume (\mathcal{V}_{tx}) or surface (\mathcal{S}_{tx}). In the latter case, we also specify whether the surface is

“transparent” or “reflective”. Furthermore, we specify whether “Particle generation” and the “Release mechanism” are taken into account in the CIR model, respectively. For the physical channel, we indicate whether “Diffusion” and “Advection” processes are taken into account. In the case of advection, we distinguish between “uniform” and “laminar” advection. The category “Geometry” specifies whether a “bounded” or an “unbounded” environment is considered. Reactions inside the physical channel involving the signaling molecules are indicated in the column “Degradation & production”. For the receiver, “ Ω_{rx} ” indicates whether the volume of the receiver, \mathcal{V}_{rx} , a surface, \mathcal{S}_{rx} , or a partial surface, $\tilde{\mathcal{S}}_{rx}$, constitute the sensing area of the receiver. Furthermore, “Passive” and “Active” refer to the reception mechanism of the receiver. In the latter case, “Deterministic” and “Stochastic” specify whether the corresponding reaction for active reception is modeled deterministically or stochastically, respectively. We also indicate whether “Signaling pathways” in the receiver are considered. Moreover, we provide the “Dimension” of the considered end-to-end channel. “Numerical” indicates that the CIR $h(t)$ is obtained numerically. Whenever possible, we also provide the equation number of the corresponding CIR $h(t)$. Finally, whenever the reaction-diffusion equation (29) was employed to obtain $h(t)$, we highlight whether the reaction and diffusion processes were considered “Separately” or “Jointly”.

IV. RECEIVED SIGNAL MODELING

In this section, we provide mathematical models for the signals used for estimation of the system parameters and detection of the transmitted data by MC receivers. To this end, we first present a unified signal representation for MC systems. Next, we introduce three time scales for the signal observed at the receiver, and subsequently, we provide signal models for each of these time scales. In addition, we generalize these models to account for the interfering noise molecules in the environment. Subsequently, time-slotted communication is considered and a corresponding signal model is developed which accounts for the impact of ISI. Finally, the correlation of the signals received at different time instants is discussed for the considered time scales.

The models that we present in this section are general in the sense that they apply to all MC systems discussed in Section III. More specifically, these models only depend on the CIR $h(t)$ within the considered observation window or at the considered sampling times. We note that for most MC environments, derivation of the CIR in closed form, as was done for specific cases in Section III, is challenging. In Section V, we present numerical and simulation methods to obtain the CIR of more complex MC systems. In addition, in practical MC systems, the transmitter may send known pilot symbols that enable the receiver to estimate the CIR from its observations (see e.g. [45] and [96] for channel parameter estimators for MC systems). The models developed in this section are applicable for analytically derived, simulated, and estimated CIRs.

A. Unified Signal Definition

In the MC literature, different physical quantities have been modeled as the received signal. Important examples

TABLE II
SUMMARY OF CIR MODELS REVIEWED IN SECTION III.

$h_t(t)$	Reference	Transmitter				Physical Channel				Receiver						(29)		
		Surface	Ω_{tx}	Particle generation	Release mechanism	Diffusion	Advection	Geometry	Degradation & production	Ω_{rx}	Passive		Active		Signaling pathways	Dimension	Separately	Jointly
(34), (35)	[40]		point			✓		unbounded		\mathcal{V}_{rx}	✓					3D		
(38)	[80]		point			✓		unbounded		\mathcal{S}_{rx}				1st-order		3D	✓	
[81, Eq. (13)]	[81]		point			✓		unbounded		$\tilde{\mathcal{S}}_{rx}$				1st-order		3D	✓	
[71, Eq. (12)]	[71]		point			✓		unbounded	(24)	\mathcal{S}_{rx}				1st-order		3D	✓	
Numerical	[68]		point			✓		unbounded		\mathcal{S}_{rx}			2nd-order			3D	✓	
[67, Eq. (29)]	[67]		point			✓		unbounded	(24)	$\tilde{\mathcal{S}}_{rx}$			2nd-order			3D	✓	
Numerical	[88]	transparent	\mathcal{V}_{tx}			✓		unbounded		\mathcal{V}_{rx}		✓			✓	3D	✓	
(41)	[89]	transparent/reflective	$\mathcal{V}_{tx}/\mathcal{S}_{tx}$			✓		unbounded		$\mathcal{V}_{rx}/\mathcal{S}_{rx}$		✓		1st-order		1D / 3D	✓	
Numerical	[90]	reflective	\mathcal{S}_{tx}			✓		unbounded		\mathcal{S}_{rx}				1st-order		3D	✓	
Numerical	[91]	reflective	\mathcal{V}_{tx}		✓	✓		unbounded		\mathcal{V}_{rx}		✓				3D	✓	
Numerical	[91, [92], [93]	transparent	\mathcal{V}_{tx}	✓		✓		unbounded		\mathcal{V}_{rx}		✓			✓	3D	✓	
(47), (50)	This article		point			✓	uniform	bounded		\mathcal{V}_{rx}		✓				3D		
(51)	[44]		point			✓	uniform	unbounded		\mathcal{V}_{rx}	✓	✓				3D		
[54, Eq. (15)]	[54]		point			✓	uniform	bounded	(24)	\mathcal{V}_{rx}	✓					3D	✓	
(55), (56), (57), (59)	[61]	transparent	point / \mathcal{S}_{tx}			✓	laminar	bounded		\mathcal{V}_{rx}	✓					3D		
	[6]	transparent	\mathcal{V}_{tx}			✓		unbounded		\mathcal{V}_{rx}		✓	2nd-order			3D	✓	
[87, Eq. (8)]	[87]		point			✓		unbounded		\mathcal{V}_{rx}		✓		2nd-order		3D	✓	
(61), (62)	[65]		point			✓		unbounded	(28)	\mathcal{V}_{rx}		✓				3D	✓	
[86, Eq. (12)]	[86]		point			✓	uniform	unbounded		\mathcal{V}_{rx}				1st-order		1D	✓	

include *i*) the number of molecules observed at a given time within the volume of a transparent receiver [38]–[46], *ii*) the number of molecules bound at a given time to the receptors of a reactive receiver [67], [68], [81], *iii*) the accumulated number of molecules observed by a fully-absorbing receiver within a given observation time window [11], [80], [97]–[99], and *iv*) the arrival times of the molecules at a fully-absorbing receiver [10], [100]–[104]. In the following, we first provide a unified definition of the received signal of general MC receivers including the aforementioned special cases. Since the presented general signal model is difficult to analyze, subsequently, we introduce the concept of counting receivers, which are widely considered in the literature and allow for simple mathematical modeling. The main purpose for introducing a general representation of the received signal is to highlight the basic assumptions that have been made to arrive at specific signal models used in the literature and to unveil the connections between different signal models.

1) Generalized Receivers: Since different molecules of the same type are indistinguishable for the receiver, the most detailed information that the receiver could access at a given time t is the arrival (and departure, if relevant) times of the molecules at (or from) the receiver up to that time. We use this fact to introduce a unified representation of the received signal of general MC receivers. For mathematical rigor, let us first formally distinguish between two types of receivers, namely recurrent and non-recurrent receivers.

Definition 3: If a given molecule can be observed by the receiver at most once, then the receiver is referred to as *non-recurrent*; otherwise, it is referred to as *recurrent*.

Transparent and reactive receivers with unbinding rate $\kappa_b \neq 0$ are recurrent since a given molecule can be observed multiple times by the receiver. On the other hand, fully-absorbing receivers and reactive receivers with $\kappa_b = 0$ are non-recurrent since after a molecule has been observed at the receiver, it cannot be observed again. For non-recurrent receivers, the time instants at which the molecules are observed constitute the most general signal representation. Let us define

$$\vec{\mathbf{T}}^{\text{arv}}(t) = [t_1, t_2, \dots, t_{n^{\text{arv}}(t)}], \quad (63)$$

as the vector containing the arrival times t_n , $n = 1, 2, \dots, n^{\text{arv}}(t)$, of all $n^{\text{arv}}(t)$ molecules *observed* by time t in an ascending order. We note that both the number of molecules observed by time t , i.e., $n^{\text{arv}}(t)$, and their arrival times t_n , $n = 1, \dots, n^{\text{arv}}(t)$, are RVs. On the other hand, for recurrent receivers, in addition to $\vec{\mathbf{T}}^{\text{arv}}(t)$, we need to keep track of the molecules that have been *un-observed*, i.e., have left the receiver. To this end, let us define

$$\vec{\mathbf{T}}^{\text{dpr}}(t) = [t'_1, t'_2, \dots, t'_{n^{\text{dpr}}(t)}], \quad (64)$$

as the vector containing the departure times t'_n , $n = 1, 2, \dots, n^{\text{dpr}}(t)$, of all $n^{\text{dpr}}(t)$ molecules that have left the receiver by time t in an ascending order. We note that by the above formulation, non-recurrent receivers can be seen as a special case of recurrent receivers where $n^{\text{dpr}}(t) = 0$, $\forall t$. In summary, $\vec{\mathbf{T}}^{\text{arv}}(t)$ and $\vec{\mathbf{T}}^{\text{dpr}}(t)$ constitute a complete and unified representation of the received signal of MC receivers.

As will be shown in the following, different notions of received signal used in the MC literature can be seen as special cases of $\vec{\mathbf{T}}^{\text{arv}}(t)$ and $\vec{\mathbf{T}}^{\text{dpr}}(t)$.

2) Timing-based Receivers: In the MC literature, timing channels have been used as a model for non-recurrent receivers [10], [100]–[103]. Let \mathbf{T}^{rls} denote the vector containing the release times of the molecules by the transmitter and let \mathbf{T}^{arv} be the vector containing the respective arrival times of the molecules at the receiver. Thus, \mathbf{T}^{arv} is related to \mathbf{T}^{rls} according to [10], [100]–[104]

$$\mathbf{T}^{\text{arv}} = \mathbf{T}^{\text{rls}} + \mathbf{T}^{\text{dly}}, \quad (65)$$

where \mathbf{T}^{dly} is a vector containing the random delays between the release of the molecules by the transmitter and their observation at the receiver. Moreover, it is typically assumed that the release, propagation, and reception of molecules are independent from each other, which we refer to as the *independent molecule behavior* assumption [102], [103]. Based on this assumption, the elements in \mathbf{T}^{dly} are independent and identically distributed and assume only non-negative real values. For an unbounded 1D environment, the random observation delay follows a Levy distribution if no flow is present [10] and the inverse Gaussian distribution if flow in the direction of the receiver is present [100].

We note that, in practice, \mathbf{T}^{arv} is not available at the receiver since *i*) different molecules of the same type are indistinguishable by the receiver and *ii*) out of the total number of released molecules, only $n^{\text{arv}}(t)$ molecules are observed by time t . In fact, $\vec{\mathbf{T}}^{\text{arv}}(t)$ is the actual observation signal available to the receiver. To arrive at a model for $\vec{\mathbf{T}}^{\text{arv}}(t)$, we introduce the following definitions and assumptions. Let us assume that N_{tx} molecules are released by the transmitter within interval $[0, t]$ and their release times are collected in \mathbf{T}^{rls} . Since the $n^{\text{arv}}(t)$ molecules observed at the receiver are indistinguishable, we do not know which $n^{\text{arv}}(t)$ molecules out of the total N_{tx} released molecules have been observed. In general, there are at most $\frac{N_{\text{tx}}!}{(n^{\text{arv}}(t)-1)!}$ possibilities for selecting $n^{\text{arv}}(t)$ (observed) molecules from the N_{tx} (released) molecules. Therefore, we define \mathbf{p}_p , $p = 1, \dots, \frac{N_{\text{tx}}!}{(n^{\text{arv}}(t)-1)!}$, as a vector which contains the p -th possible order index of the observed molecules. Moreover, let $f_X(x)$ and $F_X(x)$ denote the probability density function (PDF) and cumulative density function (CDF) of RV X at $X = x$, respectively. We note that due to causality, $f_{T^{\text{dly}}}(t) = 0$, $t < 0$, has to hold where RV T^{dly} denotes the random delay of a given molecule. Following a similar framework as developed in [102], [103], the PDF of the observation vector $\vec{\mathbf{T}}^{\text{arv}}(t) = \vec{\mathbf{t}}^{\text{arv}}$ conditioned on the molecule release time vector \mathbf{T}^{rls} , denoted by $f_{\vec{\mathbf{T}}^{\text{arv}}(t)|\mathbf{T}^{\text{rls}}}(\vec{\mathbf{t}}^{\text{arv}}|\mathbf{T}^{\text{rls}})$, is obtained as

$$\begin{aligned} & f_{\vec{\mathbf{T}}^{\text{arv}}(t)|\mathbf{T}^{\text{rls}}}(\vec{\mathbf{t}}^{\text{arv}}|\mathbf{T}^{\text{rls}}) \\ &= \sum_{p=1}^{\frac{N_{\text{tx}}!}{(n^{\text{arv}}(t)-1)!}} f_{\vec{\mathbf{T}}^{\text{arv}}(t)|\mathbf{T}^{\text{rls}}}(\vec{\mathbf{t}}^{\text{arv}}|\mathbf{T}^{\text{rls}}, \mathbf{p}_p) \\ &= \sum_{p=1}^{\frac{N_{\text{tx}}!}{(n^{\text{arv}}(t)-1)!}} \left[\prod_{m=1}^{n^{\text{arv}}(t)} f_{T^{\text{dly}}}(t_m - \mathbf{T}^{\text{rls}}[\mathbf{p}_p[m]]) \right] \end{aligned}$$

$$\begin{aligned}
& \times \prod_{m=n^{\text{arv}}(t)+1}^{N_{\text{tx}}} \left[1 - F_{T^{\text{dly}}} \left(t - \mathbf{T}^{\text{rls}}[\mathbf{p}_p[m]] \right) \right] \\
& \stackrel{(a)}{=} \frac{N_{\text{tx}}!}{(n^{\text{arv}}(t) - 1)!} (f_{T^{\text{dly}}}(t_m))^{n^{\text{arv}}(t)} \\
& \quad \times (1 - F_{T^{\text{dly}}}(t_m))^{N_{\text{tx}} - n^{\text{arv}}(t)}, \quad (66)
\end{aligned}$$

where equality (a) holds when all N_{tx} molecules are released at time zero. The above formulation provides a general framework for modeling the arrival times of non-recurrent receivers. Unfortunately, (66) cannot be easily simplified and its generalization to recurrent receivers or the cases when interfering noise molecules or ISI are present is cumbersome. In fact, the results reported in [10], [100]–[104] are valid for non-recurrent receivers when ISI and interfering noise molecules do not exist. In addition, perfect synchronization is a key underlying assumption for most timing channels considered in the literature [10], [100]–[104] and hence the performance of timing receivers is very sensitive to synchronization errors. Therefore, in the remainder of this section, we consider special receivers, namely *molecule counting* receivers, whose signal is a function of $n^{\text{arv}}(t)$ and $n^{\text{dpr}}(t)$ only. Molecule counting receivers are widely adopted in the literature and the corresponding received signal lends itself to more tractable models and analysis.

3) *Counting Receivers*: These receivers consider the number of observed molecules as the received signal. In general, the receiver may count the number of observed molecules multiple times, which is referred to as a multi-sample detector [38], [39], [42], [53], [90], [105]. Let $r(t_m)$ denote the received signal at sample time $t_m = m\Delta t$, $m = 1, 2, \dots$, where Δt is the sample interval. To formally characterize $r(t_m)$, we distinguish two types of counting receivers, namely arrival-counting and observation-counting receivers.

Definition 4: If a receiver counts the number of molecules that have *arrived within the observation window* $(t_m - \Delta t, t_m]$ at its reception site, i.e., $r(t_m) = n^{\text{arv}}(t_m) - n^{\text{arv}}(t_m - \Delta t)$, then it is referred to as an *accumulative-molecule-counting* (AMC) receiver, whereas if the receiver counts the number of molecules that are *observed at a given time* t at its reception site, i.e., $r(t_m) = n^{\text{arv}}(t_m) - n^{\text{dpr}}(t_m)$, then it is referred to as an *instantaneous-molecule-counting* (IMC) receiver.

In general, there are four types of receivers based on the recurrent/non-recurrent and AMC/IMC classifications. In the following, we present the different counting receivers used in the MC literature as special cases of these four categories:

Non-Recurrent Accumulative-Molecule-Counting (nR-AMC) Receivers: The signal in this case is $r(t_m) = n^{\text{arv}}(t_m) - n^{\text{arv}}(t_m - \Delta t)$ where $n^{\text{arv}}(t_m) \geq n^{\text{arv}}(t_m - \Delta t) \geq 0$. For instance, for fully-absorbing receivers, $r(t_m)$ denotes the number of molecules that have arrived within interval $(t_m - \Delta t, t_m]$ [11], [80], [97]–[99].

Recurrent Accumulative-Molecule-Counting (R-AMC) Receivers: The signal in this case is $r(t_m) = n^{\text{arv}}(t_m) - n^{\text{arv}}(t_m - \Delta t)$ where $n^{\text{arv}}(t_m) \geq n^{\text{arv}}(t_m - \Delta t) \geq 0$. Although the mathematical form looks identical to that for nR-AMC receivers, the modeling for R-AMC receivers is much more cumbersome since one molecule might be counted

multiple times within the observation window $(t_m - \Delta t, t_m]$. Furthermore, we note that the expected number of observed molecules for R-AMC receivers is larger than that for nR-AMC receivers because some molecules may be counted multiple times.

Recurrent Instantaneous-Molecule-Counting (R-IMC) Receivers: The signal in this case is $r(t_m) = n^{\text{arv}}(t_m) - n^{\text{dpr}}(t_m)$ where $n^{\text{arv}}(t_m) \geq n^{\text{dpr}}(t_m) \geq 0$. For instance, for transparent receivers, $r(t_m)$ denotes the number of molecules within the receiver volume at time t_m [38]–[46], and for reactive receivers, $r(t_m)$ is the number of molecules bound to the receiver's receptors at time t_m [67], [68], [81].

Non-Recurrent Instantaneous-Molecule-Counting (nR-IMC) Receivers: The signal in this case is $r(t_m) = n^{\text{arv}}(t_m) - n^{\text{dpr}}(t_m) = n^{\text{arv}}(t_m)$ where $n^{\text{arv}}(t_m) \geq 0$ and $n^{\text{dpr}}(t_m) = 0$. We note that since the received molecules do not leave the receiver, $r(t_m)$ is a non-decreasing function of time.

In the remainder of this section, we focus on the modeling of $r(t_m)$ for R-IMC receivers as a function of CIR $h(t)$, i.e., the probability of a molecule being observed at time t seconds after its release by the transmitter; see Section III. This model is also valid for nR-AMC and nR-IMC receivers if $h(t_m)$ is substituted by the probability of observing a molecule within intervals $(t_m - \Delta t, t_m]$ and $(0, t_m]$, respectively, after its release by the transmitter at time $t = 0$, cf. (40). Modeling of $r(t_m)$ for R-AMC receivers is cumbersome due to the possibility of counting a given molecule multiple times within the observation window. This type of signal is relevant, e.g., for ligand-based receivers when a ligand molecule can activate the receptors on the receiver surface multiple times. However, this problem has not yet been studied in the MC literature and is a potential topic for future research. Finally, in the following, we assume that the sampling interval Δt is sufficiently large such that consecutive samples are statistically independent. Therefore, we drop index m in Sections IV-B, IV-C, and IV-D for simplicity. How large Δt should be chosen to guarantee sample independence will be discussed in Section IV-F.

B. Three Time-Scale Signal Representation

Let us define $r(t, \tau)$ as the number of molecules observed at the receiver t seconds after their release is stimulated by the transmitter at time τ . Then, $r(t, \tau)$ can be modeled as

$$r(t, \tau) = \bar{r}(t, \tau) + w(t, \tau), \quad (67)$$

where $\bar{r}(t, \tau) = \mathbb{E}\{r(t, \tau)\}$ denotes the mean of the signal for a fixed set of channel parameters, $w(t, \tau)$ denotes the random fluctuations around the mean (e.g., caused by diffusion), and $\mathbb{E}\{\cdot\}$ denotes expectation. We note that the channel parameters may also change over time; however, this is over a scale that is much slower than the signal variations. In other words, the mean of the signal, $\bar{r}(t, \tau)$, varies over time t due to diffusion, advection, and reactions in the channel, but it also varies over the larger time scale τ due to variations of system parameters such as temperature, viscosity, and the distance between a mobile transmitter and receiver [72], [106], [107]. In summary, we have variations on three time scales in $r(t, \tau)$:

- *Time Scale 1:* Variations of $r(t, \tau)$ around its mean $\bar{r}(t, \tau)$, i.e., noisy fluctuations $w(t, \tau)$.

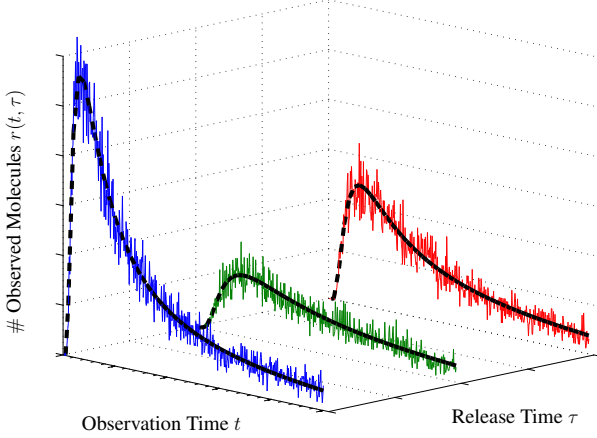


Fig. 13. Schematic illustration of the number of molecules observed at the receiver t seconds after their release by the transmitter at time τ . The three time scales are illustrated as follows: 1) the actual received signals, $r(t, \tau)$, are denoted by colored solid lines; 2) the black dashed lines denote the signal mean $\bar{r}(t, \tau)$; and 3) the variations of the signal due to changes in the system parameters over time scale τ are represented by different colors.

- *Time Scale 2:* Variations of the signal mean $\bar{r}(t, \tau)$ over observation time t , which are slower than the variations of $w(t, \tau)$.
- *Time Scale 3:* Variations of $\bar{r}(t, \tau)$ over the release time τ , which are slower than the variations of $\bar{r}(t, \tau)$ with respect to t .

For instance, for typical MC systems at microscale, e.g., cell-to-cell communication, the noisy fluctuations vary on the order of a few μs , the variations of the signal mean over time t are on the order of tens or hundreds of μs , and the change in the parameters, e.g., due to the mobility of the nodes, can be on time scales much larger than ms [67]. Fig. 13 illustrates $r(t, \tau)$ versus t for three values of τ . The aforementioned three time scales are illustrated in this figure: 1) the actual received signals, $r(t, \tau)$, are denoted by colored solid lines; 2) the black dashed lines denote the signal mean $\bar{r}(t, \tau)$; and 3) the variations of the signal due to changes in the system parameters over time scale τ are represented by different colors.

Remark 18: The three time-scale signal representation for MC is analogous to a similar signal representation in wireless communications. In particular, in a wideband wireless communication system impaired by additive white Gaussian noise (AWGN) and fading, the AWGN is analogous to the random fluctuations of the signal in MC, the CIR of the wireless communication channel is analogous to the signal mean in MC, and the variations of the CIR over time (due to the movement of the transmitter and/or receiver) are analogous to the time-variant signal mean in MC [108, Chapter 4]. \square

C. Signal Models

In the following, we first derive the expected number of molecules observed at the receiver, which we refer to as the deterministic model of the received signal. Subsequently, we

derive statistical models of the received signal that capture the random fluctuations of the observed molecules. Finally, we study time-variant channels and derive stochastic models to capture the effect of the time variance.

1) *Deterministic Models:* In Section III, we derived the CIR $h(t)$ which can be interpreted as the probability of a molecule released at time $t = 0$ being observed at the receiver at time t . Let us define $h(t, \tau)$ as the probability of a molecule released by the transmitter at time τ being observed at the receiver at time t . In the following, we first assume a time-invariant MC channel which leads to $h(t, \tau) = h(t - \tau)$, $t \geq \tau$. Then, in Section IV-C3, we analyze the impact of time variance of the channel. Following the independent molecule behavior assumption [8], [11], the *expected* number of molecules observed at the receiver at time t due to the release of N_{tx} molecules by the transmitter at time $\tau = 0$ is readily obtained as

$$\bar{r}(t, \tau) = N_{\text{tx}} h(t, \tau). \quad (68)$$

For a given set of system parameters, the expected behavior is non-random and we have a deterministic signal model. Thus, each of the CIR expressions derived in Section III constitutes a deterministic representation of the respective MC system.

Remark 19: The independent molecule behavior assumption has to hold for (68) to be valid. However, for some practical MC systems, this assumption does not hold. For instance, if a high fraction of receptors on the surface of a reactive receiver is occupied, $\bar{r}(t, \tau)$ becomes a nonlinear function of the released molecules N_{tx} and cannot be described by the simple linear expression in (68). This effect is known as *receptor occupancy* [67]. In these cases, $\bar{r}(t, \tau)$ has to be found for a given N_{tx} either numerically or via simulation, cf. Section V-A for a detailed discussion on simulation methods. \square

Remark 20: The deterministic model in (68) assumes an impulsive release of N_{tx} molecules at time $\tau = 0$ by the transmitter. In general, the transmitter may release the molecules continuously over a finite time interval $[0, T^{\text{rls}}]$ of length T^{rls} . Let $g(t)$ denote the release function satisfying $\int_{t=0}^{T^{\text{rls}}} g(t) dt = N_{\text{tx}}$ and $g(t) = 0$, $t \notin [0, T^{\text{rls}}]$. Then, the expected number of molecules observed at the receiver at time t due to the release of molecules by the transmitter with release function $g(t)$ is given by

$$\bar{r}(t, \tau) = \int_{t'=0}^t g(t') h(t - t', \tau) dt'. \quad (69)$$

We note that (69) reduces to (68) for $g(t) = N_{\text{tx}} \delta(t)$. In the remainder of this section, we focus on impulsive release, as this is typically assumed in the MC literature. \square

2) *Statistical Models:* In the following, we develop statistical models for the number of molecules observed at the receiver as a function of $h(t, \tau)$ for time-invariant MC channels.

Binomial Model: Based on the independent molecule behavior assumption and since any given molecule released by the transmitter is either observed by the receiver or not, a binary state model applies and the number of observed molecules follows the Binomial distribution with N_{tx} trials and success probability $h(t, \tau)$, i.e.,

$$r(t, \tau) \sim \mathcal{B}(N_{\text{tx}}, h(t, \tau)), \quad (70)$$

where $\mathcal{B}(N, p)$ represents a Binomial distribution with parameters N and p denoting the number of trials and the success probability, respectively. Under the Binomial model, the probability mass function (PMF) of $r(t, \tau)$, denoted by $f_r^{\mathcal{B}}(n)$, is given by

$$f_r^{\mathcal{B}}(n) = \binom{N_{\text{tx}}}{n} (h(t, \tau))^n (1 - h(t, \tau))^{N_{\text{tx}} - n}, \quad (71)$$

for $n \in \{0, 1, \dots, N_{\text{tx}}\}$. Unfortunately, the Binomial distribution considerably complicates the analysis of MC systems. Therefore, in the following, we present two approximations of the Binomial model with better mathematical tractability.

Gaussian Model: If the expected number of molecules observed at the receiver, i.e., $\bar{r}(t, \tau)$, is sufficiently large, then we can apply the central limit theorem (CLT) and approximate $r(t, \tau)$ by a Gaussian RV with mean and variance identical to that of the Binomial RV. This leads to

$$r(t, \tau) \sim \mathcal{N}(N_{\text{tx}}h(t, \tau), N_{\text{tx}}h(t, \tau)(1 - h(t, \tau))). \quad (72)$$

Under the Gaussian model, the PDF of $r(t, \tau)$, denoted by $f_r^{\mathcal{N}}(n)$, is given by

$$f_r^{\mathcal{N}}(n) = \frac{1}{\sqrt{2\pi N_{\text{tx}}h(t, \tau)(1 - h(t, \tau))}} \times \exp\left(-\frac{(n - N_{\text{tx}}h(t, \tau))^2}{2N_{\text{tx}}h(t, \tau)(1 - h(t, \tau))}\right), \quad n \in \mathbb{R}. \quad (73)$$

The Gaussian distribution is much more amenable to analysis than the Binomial distribution. However, the basic assumption behind the applicability of the Gaussian distribution, namely large $\bar{r}(t, \tau)$, may not hold in MC systems. In fact, although the number of released molecules N_{tx} can be quite large, the expected number of observed molecules $\bar{r}(t, \tau)$ can be very small. Moreover, Gaussian RVs are continuous and can assume non-integer and negative values, which contradicts the true nature of RV $r(t, \tau)$ as discrete and non-negative.

Poisson Model: For the case when the number of trials is large and the mean of the Binomial RV is small, the Binomial distribution can be well approximated by a Poisson distribution with the same mean $\bar{r}(t, \tau) = N_{\text{tx}}h(t, \tau)$, i.e.,

$$r(t, \tau) \sim \mathcal{P}(N_{\text{tx}}h(t, \tau)), \quad (74)$$

where $\mathcal{P}(\lambda)$ represents the Poisson distribution with parameter λ denoting the mean of the RV. Under the Poisson model, the PMF of $r(t, \tau)$, denoted by $f_r^{\mathcal{P}}(n)$, is given by

$$f_r^{\mathcal{P}}(n) = \frac{(N_{\text{tx}}h(t, \tau))^n}{n!} \exp(-N_{\text{tx}}h(t, \tau)), \quad n \in \mathbb{N}. \quad (75)$$

In fact, assuming $\bar{r}(t, \tau)$ is fixed, the proof simply follows from [47]

$$\begin{aligned} & \lim_{N_{\text{tx}} \rightarrow \infty} f_r^{\mathcal{B}}(n) \\ &= \lim_{N_{\text{tx}} \rightarrow \infty} \binom{N_{\text{tx}}}{n} \left(\frac{\bar{r}(t, \tau)}{N_{\text{tx}}}\right)^n \left(1 - \frac{\bar{r}(t, \tau)}{N_{\text{tx}}}\right)^{N_{\text{tx}} - n} \\ &\stackrel{(a)}{=} \frac{(\bar{r}(t, \tau))^n}{n!} \exp(-\bar{r}(t, \tau)) = f_r^{\mathcal{P}}(n), \end{aligned} \quad (76)$$

where for equality (a) we used $\lim_{x \rightarrow \infty} \binom{x}{y} = \frac{x^y}{y!}$ and $\lim_{x \rightarrow \infty} \left(1 - \frac{y}{x}\right)^x = \exp(-y)$ [109].

Comparison: In order to quantify the accuracy of the Gaussian and Poisson approximations, we define the root mean square error (RMSE) between the approximated Gaussian and Poisson CDFs, denoted by $F_r^x(n)$, $x \in \{\mathcal{N}, \mathcal{P}\}$, and the Binomial CDF, denoted by $F_r^{\mathcal{B}}(n)$, as [11], [98], [110]

$$\text{RMSE}^x = \sqrt{\frac{1}{N_{\text{tx}} + 1} \sum_{n=0}^{N_{\text{tx}}} |F_r^x(n) - F_r^{\mathcal{B}}(n)|^2}. \quad (77)$$

In Fig. 14, the RMSE between the approximate Gaussian and Poisson CDFs and the Binomial CDF versus $h(t, \tau)$ is shown for $N_{\text{tx}} \in \{10^2, 10^3, 10^4, 10^5\}$. We observe from this figure that by increasing $h(t, \tau)$, the accuracy of the Poisson model deteriorates whereas the accuracy of the Gaussian model improves, which is consistent with the respective assumptions that led to their derivation. Moreover, as N_{tx} increases, the Gaussian model becomes more accurate whereas this is not true for the Poisson model if $h(t, \tau)$ is very small. In fact, for small $h(t, \tau)$ and small N_{tx} , both the Binomial and Poisson distributions approach the binary distribution, i.e., either zero or one molecule is observed and the probability of observing more than one molecule becomes negligible, i.e., $\bar{r}(t, \tau) \ll 1$. Thus, for a fixed value of $h(t, \tau)$, the accuracy of assumption $\bar{r}(t, \tau) \ll 1$ improves as N_{tx} decreases. Since for typical MC systems, the value of $h(t, \tau)$ is expected to be much smaller than 0.1, the Poisson model is generally a more accurate model. Nevertheless, the fact that the accuracy of the Gaussian model increases with increasing N_{tx} makes it a suitable model for macroscale applications when N_{tx} is potentially very large. Moreover, the Gaussian model is attractive for asymptotically high signal-to-noise ratio (SNR) analysis. These observations are consistent with the results reported in [11].

3) *Time-Variant Models:* Until now, we have assumed time-invariant MC channels where the channel parameters are fixed. Hence, $h(t, \tau)$ and consequently $\bar{r}(t, \tau)$ were only functions of $t - \tau$. Now, we consider time-variant MC channels where $h(t, \tau)$ and $\bar{r}(t, \tau)$ are in general functions of both t and τ . More specifically, we study the impact of system parameter variations on the mean received signal $\bar{r}(t, \tau)$. In principle, each of the system parameters such as D , $\mathbf{v}(\mathbf{d}, t)$, the physical and chemical properties of the boundaries of the end-to-end channel, the reaction rates of the involved CRNs, and \mathbf{d}_{tx} and \mathbf{d}_{rx} can potentially vary over time, which in turn leads to a variation of $\bar{r}(t, \tau)$. For instance, the diffusion coefficient D appears in the expressions for $h(t, \tau)$ for all diffusive MC systems, and consequently in $\bar{r}(t, \tau)$. As we can see from (2), changes in the parameters of the fluid environment, e.g., the viscosity or temperature, will result in a change in D . In fact, the impact of variations in D on $\bar{r}(t, \tau)$ for a point transmitter and passive receiver in 1D is investigated in [107]. The authors in [72] consider a point transmitter with impulsive release, a passive receiver with the UCA, and an unbounded 3D environment with uniform flow and uniformly distributed enzymes, cf. (33) and (30). There, the impact of Gaussian variations in the diffusion coefficient, flow velocity, and enzyme

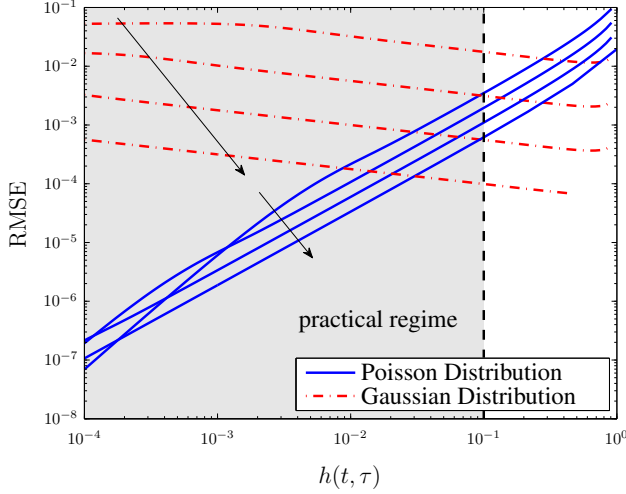


Fig. 14. RMSE between the approximate Gaussian/Poisson CDF and the Binomial CDF versus $h(t, \tau)$ for $N_{tx} \in \{10^2, 10^3, 10^4, 10^5\}$. The number of released molecules increases in the direction of the arrows. For typical MC systems, $h(t, \tau) < 0.1$ holds, which is indicated by the shaded area and the dashed vertical line and denoted by *practical regime*.

concentration is modeled by a parametric model where the parameters of the model are obtained via curve fitting. The impact of the mobility of a point transmitter and a passive receiver on the CIR $\bar{r}(t, \tau)$ is studied in [106] and a stochastic model for $\bar{r}(t, \tau)$ is derived. Similarly, a stochastic model for mobile MC systems with a point transmitter and fully-absorbing receiver is derived in [99]. We note that mobile transceivers are relevant for many envisioned applications of synthetic MC systems such as targeted drug delivery and health monitoring [49]–[52]. Therefore, in the following, we focus on diffusive mobile MC systems and review some of the results reported in [106].

We assume a point transmitter, a passive receiver with the UCA approximation, and an unbounded diffusive channel without advection. Furthermore, we model the mobility of transmitter and receiver via 3D diffusion, since diffusion is a common cause of mobility and can also be used to model more elaborate movements such as cell migrations and bacteria chemotaxis [111], [112]. In particular, we denote the diffusion coefficients of transmitter and receiver by D_{tx} and D_{rx} , respectively, and their corresponding locations at time τ by $\mathbf{d}_{tx}(\tau)$ and $\mathbf{d}_{rx}(\tau)$, respectively. Then, it can be shown that $\mathbf{d}(\tau) = \mathbf{d}_{tx}(\tau) - \mathbf{d}_{rx}(\tau)$ follows a Gaussian distribution [106, Eq. (2)]

$$f_{\mathbf{d}(\tau)}(\mathbf{d}) = \frac{1}{(4\pi D_2 \tau)^{3/2}} \exp\left(-\frac{\|\mathbf{d} - \mathbf{d}(0)\|^2}{4D_2 \tau}\right), \quad (78)$$

where $D_2 = D_{tx} + D_{rx}$ is an effective diffusion coefficient capturing the relative motion of transmitter and receiver, see [113, Eq. (10)]. Then, given (34), the CIR of the end-to-end channel can be rewritten as

$$h(t, \tau) = \frac{V_{rx}}{(4\pi D_1 t)^{3/2}} \exp\left(-\frac{d^2(\tau)}{4D_1 t}\right), \quad (79)$$

where $d(\tau) = \|\mathbf{d}(\tau)\|$ and $D_1 = D + D_{rx}$ is the effective diffusion coefficient capturing the relative motion of the signaling molecules and the receiver, see [113, Eq. (8)]. The movement of the receiver affects both (78) and (79) via D_2 and D_1 , respectively, as long as its movement with respect to the transmitter and the signaling molecules is accounted for. For any given t , $h(t, \tau)$ is a stochastic process with RVs $h(t, \tau_i)$, $i \in \{1, 2, 3, \dots\}$. In the following, we analyze the mean, the variance, and an approximate expression for the PDF of $\bar{r}(t, \tau)$.

Mean: Let d_0 denote the distance between the transmitter and receiver at $\tau = 0$, i.e., $\|\mathbf{d}(0)\| = d_0$. Given (78) and (79), the mean of the time-variant channel, denoted by $m(t, \tau)$, can be evaluated as [106, Eq. (14)]

$$\begin{aligned} m_{\bar{r}}(t, \tau) &= \mathbb{E}\{\bar{r}(t, \tau)\} \\ &= \int_{\mathbf{d} \in \mathbb{R}^3} \bar{r}(t, \tau | d_0) f_{\mathbf{d}(\tau)}(\mathbf{d}) d\mathbf{d}, \\ &= \frac{N_{tx} V_{rx}}{(4\pi (D_1 t + D_2 \tau))^{3/2}} \exp\left(\frac{-d_0^2}{4(D_1 t + D_2 \tau)}\right), \end{aligned} \quad (80)$$

where in $\mathbb{E}\{\bar{r}(t, \tau)\}$, the expectation is taken with respect to the RV $\mathbf{d}(\tau)$. As we expected, $\bar{r}(t, \tau)$ is a function of τ , because of the mobility of transmitter and receiver. As a result, $\bar{r}(t, \tau)$ is a non-stationary stochastic process. Moreover, due to the assumption of an unbounded environment, on average the transmitter and receiver diffuse away from each other. Therefore, when at least one of the transceivers is mobile, i.e., $D_2 \neq 0$, we obtain $\bar{r}(t, \tau) \rightarrow 0$ as $\tau \rightarrow \infty$.

Variance: The variance of $\bar{r}(t, \tau)$, denoted by $\sigma_{\bar{r}}^2(t, \tau)$, is given by

$$\sigma_{\bar{r}}^2(t, \tau) = \mathbb{E}\{\bar{r}^2(t, \tau)\} - m_{\bar{r}}^2(t, \tau), \quad (81)$$

where the second order moment $\phi_{\bar{r}}(t, \tau) \triangleq \mathbb{E}\{\bar{r}^2(t, \tau)\}$ is obtained as [106, Eq. (21)]

$$\phi_{\bar{r}}(t, \tau) = \frac{(N_{tx})^2 V_{rx}^2 \exp\left(\frac{-d_0^2}{2(D_1 t + 2D_2 \tau)}\right)}{(4\pi D_1 t)^{3/2} (4\pi (D_1 t + 2D_2 \tau))^{3/2}}. \quad (82)$$

We note that $\sigma_{\bar{r}}^2(t, \tau) \rightarrow 0$ as $\tau \rightarrow \infty$, which is due to the fact that $\bar{r}(t, \tau) \rightarrow 0$ as $\tau \rightarrow \infty$. On the other hand, it can be shown that the normalized variance $\frac{\sigma_{\bar{r}}^2(t, \tau)}{m_{\bar{r}}^2(t, \tau)} \rightarrow \infty$ as $\tau \rightarrow \infty$. In other words, the normalized variance increases as τ increases. This in turn implies that due to the random walk, the uncertainty that we have about $\bar{r}(t, \tau)$ increases as τ increases.

Approximate PDF: In the following, we present the approximated PDF of the considered time-variant channel with mobile transceivers and refer the interested reader to [106] for the exact expressions of the CDF and PDF. In particular, it is shown in [106] that when $D_2 \tau \leq d_0^2/200$ holds for any $\tau > 0$, then the PDF of the CIR can be accurately approximated via a log-normal distribution [106, Eq. (29)]

$$f_{h(t, \tau)}(h) \sim \text{Lognormal}(\tilde{\mu}, \tilde{\sigma}^2) \quad (83)$$

with

$$\begin{cases} \tilde{\mu} = \ln\left(\frac{V_{rx}}{(4\pi D_1 t)^{3/2}}\right) - \frac{D_2 \tau}{4D_1 t} \left(6 + \frac{d_0^2}{D_2 \tau}\right) \\ \tilde{\sigma}^2 = \left(\frac{D_2 \tau}{2D_1 t}\right)^2 \left(6 + \frac{2d_0^2}{D_2 \tau}\right) \end{cases}$$

where $\tilde{\mu}$ and $\tilde{\sigma}^2$ denote the mean and the variance of the log-normal distribution. Given (83), the PDF of $\bar{r}(t, \tau)$, denoted by $f_{\bar{r}(t, \tau)}(\bar{r})$, can be written as

$$f_{\bar{r}(t, \tau)}(\bar{r}) = \frac{1}{N_{\text{tx}}} \times f_{h(t, \tau)}\left(\frac{\bar{r}}{N_{\text{tx}}}\right). \quad (84)$$

The above stochastic model can be used for the design and performance analysis of time-variant MC systems. For instance, (84) was used in [113] to compute the expected error probability of a mobile MC system when the knowledge of CIR $h(t, \tau)$ used for detection becomes gradually outdated due to the mobility of the transceivers. Moreover, in [72], a stochastic channel model was used to develop non-coherent detectors. In contrast to the stochastic model in (84) for mobile MC systems, it was shown in [72] that the Gamma distribution is a good fit for (Gaussian) variations in the diffusion coefficient, flow velocity, and enzyme concentration in a non-mobile MC system.

D. Interfering Noise Molecules

In the previous section, we have considered statistical models for the number of molecules observed at the receiver due to the release of signaling molecules by the transmitter. However, MC systems may be impaired by noise molecules that are not released by the transmitter but originate from interfering natural or synthetic sources. In the following, we introduce statistical models to account for the number of noise molecules that are observed at the receiver. Since information and noise molecules are indistinguishable, the receiver treats the total numbers of observed signaling molecules, denoted by $r_{\text{sig}}(t, \tau)$, and interfering noise molecules, denoted by $r_{\text{int}}(t, \tau)$, as the received signal $r(t, \tau)$, i.e.,

$$r(t, \tau) = r_{\text{sig}}(t, \tau) + r_{\text{int}}(t, \tau). \quad (85)$$

To derive a statistical model for $r_{\text{int}}(t, \tau)$, we focus on a passive receiver. Similar arguments apply for other receiver types. We make the following assumptions. A1) Let $\bar{r}_{\text{int}}(\tau)$ denote the expected number of noise molecules observed within the receiver volume V_{rx} at a given sample time t . We assume that the value of $\bar{r}_{\text{int}}(\tau)$ is constant over observation time t . Nevertheless, $\bar{r}_{\text{int}}(\tau)$ may change over larger time scale τ due to variations in the system parameters such as the temperature, cf. Section IV-B and Section IV-C3. A2) It is further assumed that the observation of one noise molecule at the receiver is independent from the observations of other noise molecules. A3) Finally, we assume that the expected number of noise molecules observed within a given volume in space is proportional to the size of that volume.

Based on assumptions A1-A3, the statistics of the observed noise molecules is Poisson following the law of rare events (LRE) [114]. In particular, suppose the receiver volume is divided into J subvolumes where $J \gg \bar{r}_{\text{int}}(\tau)$. Thus, $\frac{\bar{r}_{\text{int}}(\tau)}{J}$ can be interpreted as the probability that one noise molecule is observed in one of these subvolumes at the receiver. The probability that two noise molecules are simultaneously observed in one subvolume becomes negligible for large J . Therefore, the number of noise molecules observed over the

entire volume of the receiver follows a Binomial distribution $\mathcal{B}\left(J, \frac{\bar{r}_{\text{int}}(\tau)}{J}\right)$ with J trials and success probability $\frac{\bar{r}_{\text{int}}(\tau)}{J}$. Consequently, since J is a free variable, one can assume $J \rightarrow \infty$ such that the Binomial distribution approaches the Poisson distribution $\mathcal{P}(\bar{r}_{\text{int}}(\tau))$, cf. (75). In summary, under assumptions A1-A3, we obtain $r_{\text{int}}(t, \tau) \sim \mathcal{P}(\bar{r}_{\text{int}}(\tau))$.

Remark 21: The choice of the Poisson distribution for the number of environmental noise molecules observed at the receiver, $r_{\text{int}}(t, \tau)$, can be further justified from an information-theoretic perspective [115]. Let us define RV $\mathbf{D} = [\mathbf{d}_1, \mathbf{d}_2, \dots, \mathbf{d}_{r_{\text{int}}}]$ where \mathbf{d}_i denotes the coordinates of the i -th noise molecule observed at the receiver and we drop argument (t, τ) of $r_{\text{int}}(t, \tau)$ in \mathbf{D} for notational simplicity. In particular, the maximum entropy distribution for \mathbf{D} corresponds to a Poisson distribution for the number of observed noise molecules $r_{\text{int}}(t, \tau)$. Therefore, the most random noise under assumptions A1-A3 is Poisson noise, i.e., a worst-case scenario. To see this, let $f_{\mathbf{D}}(\mathbf{D})$ denote the distribution of RV \mathbf{D} . Using the chain rule, we have $f_{\mathbf{D}}(\mathbf{D}) = f_{\mathbf{D}|r_{\text{int}}}(\mathbf{D}|r_{\text{int}})f_{r_{\text{int}}}(r_{\text{int}})$ where $f_{\mathbf{D}|r_{\text{int}}}(\mathbf{D}|r_{\text{int}})$ is the conditional distribution of \mathbf{D} given $r_{\text{int}}(t, \tau)$, and $f_{r_{\text{int}}}(r_{\text{int}})$ denotes the distribution of $r_{\text{int}}(t, \tau)$. For maximum entropy, $f_{\mathbf{D}|r_{\text{int}}}(\mathbf{D}|r_{\text{int}})$ should be a uniform distribution across the receiver volume. Substituting this result in $f_{\mathbf{D}}(\mathbf{D})$, we obtain that $f_{r_{\text{int}}}(r_{\text{int}})$ has to be the Poisson distribution to maximize the entropy of \mathbf{D} [115, Appendix 8]. \square

We note that the Poisson distribution $\mathcal{P}(\lambda)$ approaches a Gaussian distribution $\mathcal{N}(\lambda, \lambda)$ for $\lambda \rightarrow \infty$. Therefore, for very noisy environments, the approximation $r_{\text{int}}(t, \tau) \sim \mathcal{N}(\bar{r}_{\text{int}}(\tau), \bar{r}_{\text{int}}(\tau))$ becomes valid.

Remark 22: Assumption A1 states that the mean of the observed interfering molecules is constant, i.e., $\bar{r}_{\text{int}}(t, \tau) = \bar{r}_{\text{int}}(\tau)$, $\forall t$. This assumption is accurate for natural sources that continuously secrete molecules. However, for multi-user interfering sources, the mean number of observed molecules is in general time-dependent. Nevertheless, when no information about the activity of the interfering users is available, it is reasonable to assume that the mean number of observed interference molecules is constant. We note that examples of time-dependent interference were studied in [78], [116], [117]. In [78], the authors assumed that an external noise source starts to release molecules into the environment with a constant rate at $t = 0$. They derived the expected number of noise molecules observed at the receiver at time $t > 0$, denoted by $\bar{r}_{\text{int}}(t, \tau)$, as a function of the system parameters such as the distance between the noise source and the receiver. It was shown that asymptotically as $t \rightarrow \infty$, $\bar{r}_{\text{int}}(t, \tau)$ converges to a constant value, i.e., $\bar{r}_{\text{int}}(\tau)$, which is consistent with assumption A1 made earlier in this section. In [116], [117], statistical models for the number of received noise molecules originating from multiple interfering sources were derived for various scenarios regarding the distribution of interfering sources in the environment and their molecule release patterns. \square

E. Continuous Transmission

The statistical models developed so far are appropriate for one-shot transmission. Nevertheless, in most communication

systems, the transmitter may send multiple symbols consecutively to the receiver. To develop a model valid for continuous transmission, we consider a time-slotted communication system where one symbol is transmitted in each time slot, also referred to as a symbol interval, of length T^{symp} . We focus on concentration shift keying (CSK) modulation where the transmitter releases $s[k]N_{\text{tx}}$ molecules at the beginning of the k -th symbol interval to convey information symbol $s[k] \in [0, 1]$ [3]. We assume synchronous transmission and that the receiver counts the number of observed molecules multiple times in each symbol interval with sampling interval Δt [105]. Because of the memory of the MC channel, ISI occurs. To take this into account, we assume that the MC channel has a memory of L symbol intervals, i.e., the ISI in symbol interval k originates from the symbols transmitted in the $L - 1$ previous symbol intervals. We further take into account that communication may be impaired by noise molecules that originate from interfering natural or synthetic sources. Finally, we assume that the MC channel parameters remain unchanged for the considered observation window, and hence, we drop argument τ in $r(t, \tau)$, $\bar{r}_{\text{int}}(\tau)$, and $h(t, \tau)$ for notational simplicity. In the following, we first provide the signal model for a general case and subsequently simplify it for extreme SNR regimes to obtain further insight.

1) *General Case:* Let $r[k, m]$ denote the total number of molecules observed at the receiver for sample m in symbol interval k , i.e., $r[k, m] = r(t_{k,m})$ where $t_{k,m} = (k - 1)T^{\text{symp}} + m\Delta t$ and $r(t_{k,m})$ is given in (67). Then, following the discussion in Section IV-C2, $r[k, m]$ can be accurately modeled as a Poisson RV, i.e.,

$$r[k, m] \sim \mathcal{P} \left(\sum_{l=1}^L \bar{r}_{\text{sig}}[l, m] s[k - l + 1] + \bar{r}_{\text{int}} \right), \quad (86)$$

where $\bar{r}_{\text{sig}}[l, m] = N_{\text{tx}} h(t_{l,m})$. Moreover, we used the superposition property of Poisson RVs, i.e., if X and Y are two independent Poisson RVs with means λ_x and λ_y , respectively, then $X + Y$ is also a Poisson RV with mean $\lambda_x + \lambda_y$ [118]. Alternatively, defining $\tilde{r}[k, m] = r[k, m] - \bar{r}_{\text{int}}$, one can obtain the following more familiar additive signal model

$$\begin{aligned} \tilde{r}[k, m] = & \underbrace{\sum_{l=1}^L \bar{r}_{\text{sig}}[l, m] s[k - l + 1]}_{\text{signal component}} \\ & + \underbrace{r_{\text{dfn}}[k, m]}_{\text{diffusion noise}} + \underbrace{r_{\text{int}}[k, m]}_{\text{interference noise}}, \end{aligned} \quad (87)$$

where $r_{\text{dfn}}[k, m] \sim \mathcal{P}_0 \left(\sum_{l=1}^L \bar{r}_{\text{sig}}[l, m] s[k - l + 1] \right)$ denotes the diffusion noise and $r_{\text{int}}[k, m] \sim \mathcal{P}_0(\bar{r}_{\text{int}})$ denotes the interfering noise molecules. Here, we use the notation $X \sim \mathcal{P}_0(\lambda)$ when $X = Y - \lambda$ where $Y \sim \mathcal{P}(\lambda)$, i.e., X is a Poisson RV whose mean has been subtracted.

When the expected numbers of information and interfering noise molecules are large, one may use the Gaussian model for the number of observed molecules, i.e., $r[k, m] \sim \mathcal{N}(\bar{r}[k, m], \bar{r}[k, m])$ where $\bar{r}[k, m] = \sum_{l=1}^L \bar{r}_{\text{sig}}[l, m] s[k - l + 1] + \bar{r}_{\text{int}}$. One can also write $\tilde{r}[k, m]$ in the form of (87) where for the Gaussian model, we

have $r_{\text{dfn}}[k, m] \sim \mathcal{N} \left(0, \sum_{l=1}^L \bar{r}_{\text{sig}}[l, m] s[k - l + 1] \right)$ and $r_{\text{int}}[k, m] \sim \mathcal{N}(0, \bar{r}_{\text{int}})$. We note that unlike for the AWGN channel in conventional wireless communication, the Gaussian diffusion noise in MC is signal dependent.

Remark 23: In (87), we distinguish between two types of additive noise, namely $r_{\text{dfn}}[k, m]$, which originates from signaling molecules, and $r_{\text{int}}[k, m]$, which originates from external interfering noise molecules. We note that the randomness of $r_{\text{dfn}}[k, m]$ and $r_{\text{int}}[k, m]$ can be attributed to the random Brownian motion of the signaling and noise molecules, respectively. In addition to the aforementioned noises, other types of noises may be present. For instance, in a reactive receiver, the noisy measurements of the activated receptors, caused by the randomness of diffusion and ligand-receptor interactions, may be relayed by signaling pathways to the interior of the receiver (e.g. a cell), which may add extra noise [6], [110]. We refer to this noise as *counting noise* to contrast it with the diffusion noise. \square

2) *Simplifications for Extreme SNR Regimes:* In the following, we further simplify the model in (87) for two asymptotic SNR regimes, namely the diffusion-noise-limited and interference-limited regimes. To do so, we first formally define SNR as [119]

$$\begin{aligned} \text{SNR} &= \frac{\text{Power of Signal}}{\text{Variance of Diffusion Noise} + \text{Variance of Interfering Noise}} \\ &= \frac{\bar{r}_{\text{sig}}^2}{\bar{r}_{\text{sig}} + \bar{r}_{\text{int}}}, \end{aligned} \quad (88)$$

where \bar{r}_{sig} denotes the expected number of signaling molecules received at the sampling time. In the following, we focus on the ISI-free channel, i.e., $L = 1$, and a single-sample detector. Therefore, we drop indices l and m for notational simplicity.

Remark 24: One approach to obtain an approximately ISI-free channel is to choose a sufficiently large symbol interval such that the CIR practically fully decays to zero within one symbol interval. In such a case, the transmission rate may be severely reduced which may lead to an inefficient system design. Fortunately, it has been shown in the literature that reactions can be beneficial for ISI mitigation [65], [66], [74]. In particular, enzymes [65] and reactive signaling molecules, such as acid and base molecules [66], [74], may be used to speed up the decay of the CIR as a function of time, which would increase the accuracy of the assumption of an ISI-free channel, see Fig. 5. For instance, in [74], a reactive signaling MC system was assumed where the transmitter employs different molecules that react with each other, e.g., acids and bases. Then, after the release of the signaling molecules (e.g., an acid), the transmitter may release so-called cleaning molecules (e.g., a base). It is shown that the resulting CIR is considerably shortened, which makes the ISI-free channel an accurate model, see Fig. 15. Moreover, the peak of the received signal remains unchanged since the cleaning molecules are released after the peak is observed at the receiver. \square

Diffusion-Noise-Limited SNR Regime: In this case, we assume $\bar{r}_{\text{sig}} \gg \bar{r}_{\text{int}}$ holds. Thus, the model in (87) simplifies

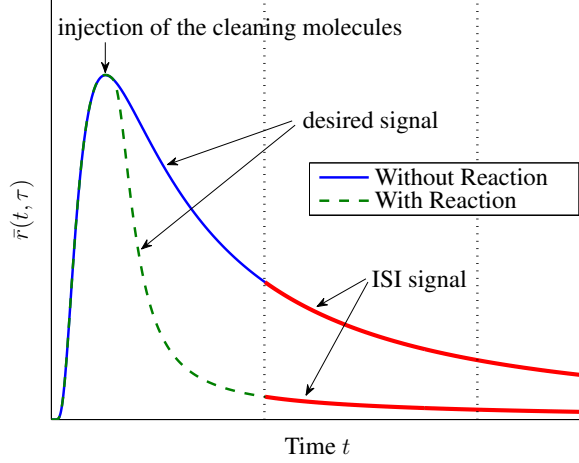


Fig. 15. Expected number of molecules observed at the receiver versus time. The dotted vertical lines indicate the beginning of symbol intervals. The injection of the reactive cleaning signal helps to shorten the CIR.

to

$$\tilde{r}[k] = \tilde{r}_{\text{sig}}s[k] + r_{\text{dfn}}[k], \quad (89)$$

where $r_{\text{dfn}}[k] \sim \mathcal{P}_0(\tilde{r}_{\text{sig}}s[k])$ and $r_{\text{dfn}}[k] \sim \mathcal{N}(0, \tilde{r}_{\text{sig}}s[k])$ hold for the Poisson and Gaussian models, respectively. The SNR in this case is obtained as $\text{SNR} = \tilde{r}_{\text{sig}}$.

Interference-Limited SNR Regime: In this case, we assume $\tilde{r}_{\text{sig}} \ll \tilde{r}_{\text{int}}$ holds. Thus, the model in (87) simplifies to

$$\tilde{r}[k] = \tilde{r}_{\text{sig}}s[k] + r_{\text{int}}[k], \quad (90)$$

where $r_{\text{int}}[k] \sim \mathcal{P}_0(\tilde{r}_{\text{int}})$ and $r_{\text{int}}[k] \sim \mathcal{N}(0, \tilde{r}_{\text{int}})$ hold for the Poisson and Gaussian models, respectively. The SNR in this case is obtained as $\text{SNR} = \tilde{r}_{\text{sig}}^2/\tilde{r}_{\text{int}}$. We note that this special case yields a signal-independent (Gaussian) model as it is widely adopted in conventional wireless communications.

Finally, we note that it may be necessary to use a combination of both of the above special cases for the analysis of MC systems. For instance, for a simple on-off keying (OOK) modulation, i.e., $s[k] \in \{0, 1\}$, the interference noise molecules are dominant for bit $s[k] = 0$ whereas the diffusion noise is dominant for bit $s[k] = 1$. This is schematically illustrated in Fig. 16 where it can be observed that the noise power for symbol $s[k] = 1$ (diffusion-noise-limited regime) is larger than that for symbol $s[k] = 0$ (interference-limited regime). A similar observation has also been reported for photon-counting receivers in optical wireless communications where the shot noise at the receiver has two components, one generated by the laser transmitter (analogous to diffusion noise) and one generated by the ambient background light (analogous to interfering noise molecules) [120].

F. Time Correlation

In the following, we discuss the signal correlation with respect to observation time scale t and release time scale τ .

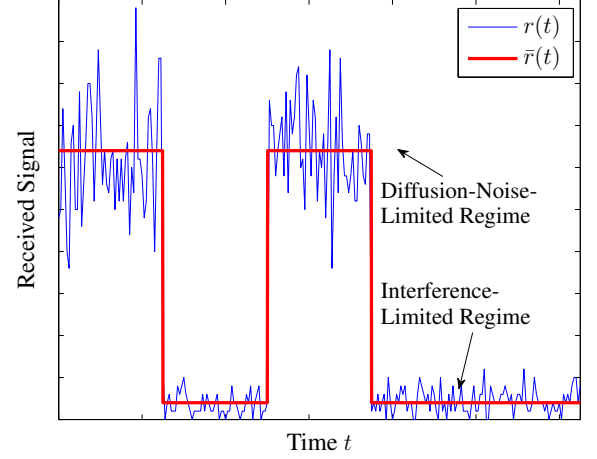


Fig. 16. Received signal versus time. Illustration of diffusion-noise-limited and interference-limited regimes.

1) Sample Correlation: In (86) and (87), we assume that the number of molecules counted at different time instants t within one symbol interval or in different symbol intervals are independent from each other. However, this assumption holds only if the sampling interval is chosen large enough such that the independence of consecutive samples is guaranteed. In [53], the mutual information between two samples $r(t_1, \tau)$ and $r(t_2, \tau)$ was numerically computed and the minimum spacing needed to ensure independence between consecutive samples was found such that the corresponding mutual information is below some threshold. Since the mutual information between two samples is difficult to derive in closed form, one may consider the Pearson correlation coefficient [121] among two consecutive samples instead, i.e.,

$$\begin{aligned} \rho_t(t_1, t_2) &= \frac{\mathbb{E}\{(r(t_1, \tau) - \bar{r}(t_1, \tau))(r(t_2, \tau) - \bar{r}(t_2, \tau))\}}{\sqrt{\mathbb{E}\{(r(t_1, \tau) - \bar{r}(t_1, \tau))^2\} \mathbb{E}\{(r(t_2, \tau) - \bar{r}(t_2, \tau))^2\}}} \\ &\stackrel{(a)}{=} \frac{\mathbb{E}\{r(t_1, \tau)r(t_2, \tau)\} - \bar{r}(t_1, \tau)\bar{r}(t_2, \tau)}{\sqrt{\bar{r}(t_1, \tau)\bar{r}(t_2, \tau)}}, \end{aligned} \quad (91)$$

where equality (a) follows from the fact that under both Poisson and Gaussian statistics, the variance of $r(t, \tau)$ is $\bar{r}(t, \tau)$. The cross-correlation term $\mathbb{E}\{r(t_1, \tau)r(t_2, \tau)\}$ depends on the specific adopted receiver type. Note that by definition, $-1 \leq \rho_t(t_1, t_2) \leq 1$ holds. Typically, the sample times t_1 and t_2 should be separated such that $\rho_t(t_1, t_2)$ falls below a certain threshold, denoted by ζ_t , i.e., $|t_2 - t_1|$ should be large enough such that $\rho_t(t_1, t_2) < \zeta_t$ holds. In Fig. 17, we show the absolute correlation $|\rho_t(t^p, t^p + \Delta t)|$ versus Δt where t^p denotes the peak of the expected received signal. As can be seen from this figure, the correlation decreases as Δt increases. Moreover, as an example, we choose the value of the threshold as $\zeta_t = 0.2$. One can observe from Fig. 17 that as the diffusion coefficient of the molecules increases, the minimum sample spacing Δt needed to ensure $|\rho_t(t^p, t^p + \Delta t)| < \zeta_t$ decreases.

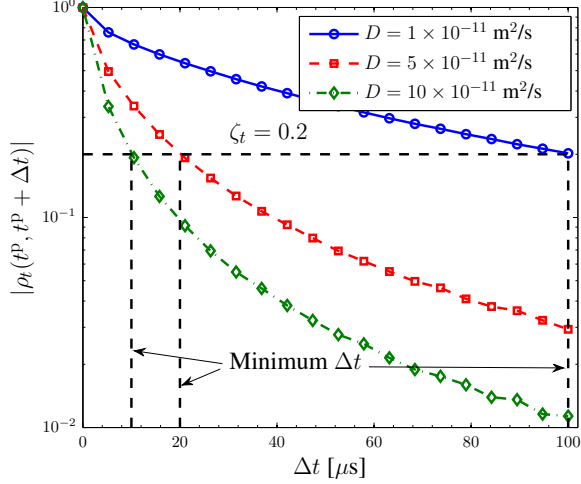


Fig. 17. Absolute correlation $|\rho_t(t^P, t^P + \Delta t)|$ versus Δt [μs] where $t^P = \max_t \bar{r}(t, \tau)$ for a point transmitter, an unbounded environment, a passive receiver of radius $a_{\text{rx}} = 50$ nm, $d = 200$ nm, $N_{\text{tx}} = 2000$, and $D = \{1, 5, 10\} \times 10^{-11} \text{ m}^2/\text{s}$.

2) *Mean Correlation*: Recall that if the system parameters change, the mean signal $\bar{r}(t, \tau)$ varies over time scale τ . In a similar manner as for sample correlation, one can define a correlation factor $\rho_\tau(\tau_1, \tau_2)$ between the mean signals at time τ_1 and τ_2 as follows

$$\rho_\tau(\tau_1, \tau_2) = \frac{\mathbb{E}\{\bar{r}(t, \tau_1)\bar{r}(t, \tau_2)\} - m_{\bar{r}}(t, \tau_1)m_{\bar{r}}(t, \tau_2)}{\sigma_{\bar{r}}(t, \tau_1)\sigma_{\bar{r}}(t, \tau_2)}. \quad (92)$$

For the case when transmitter and receiver mobility are the cause of the variations in $\bar{r}(t, \tau)$, cf. Section IV-C3, $m_{\bar{r}}(t, \tau)$ and $\sigma_{\bar{r}}(t, \tau)$ are given by (80) and (81), respectively. Moreover, the cross-correlation, denoted by $\phi_{\bar{r}}(t, \tau_1, \tau_2) \triangleq \mathbb{E}\{\bar{r}(t, \tau_1)\bar{r}(t, \tau_2)\}$, for two arbitrary times τ_1 and $\tau_2 > \tau_1$ is derived in [106, Eq. (19)]

$$\begin{aligned} & \phi_{\bar{r}}(t, \tau_1, \tau_2) \\ &= \iint_{\mathbf{d}_1, \mathbf{d}_2 \in \mathbb{R}^3} \bar{r}(t, \tau_1 | \mathbf{d}(\tau_1) = \mathbf{d}_1) \bar{r}(t, \tau_2 | \mathbf{d}(\tau_2) = \mathbf{d}_2) \\ & \quad \times f_{\mathbf{d}(\tau_1), \mathbf{d}(\tau_2)}(\mathbf{d}_1, \mathbf{d}_2) d\mathbf{d}_1 d\mathbf{d}_2, \\ &= \frac{(N_{\text{tx}})^2 (2\pi)^3 \phi^2 \lambda(\tau_1) \lambda(\tau_2 - \tau_1)}{(4\theta(\tau_1, \tau_2))^{3/2}} \\ & \quad \times \exp\left(-\beta(\tau_1) d_0^2 \left[1 - \frac{(\alpha + \beta(\tau_2 - \tau_1))\beta(\tau_1)}{\theta(\tau_1, \tau_2)}\right]\right), \quad (93) \end{aligned}$$

where for compactness ϕ , $\lambda(\tau)$, α , $\beta(\tau)$, and $\theta(\tau_1, \tau_2)$ are respectively defined as

$$\begin{aligned} \phi &= \frac{V_{\text{rx}}}{(4\pi D_1 t)^{3/2}}, \quad \lambda(\tau) = \frac{1}{(4\pi D_2 \tau)^{3/2}}, \\ \alpha &= \frac{1}{4D_1 t}, \quad \beta(\tau) = \frac{1}{4D_2 \tau}, \quad \text{and} \\ \theta(\tau_1, \tau_2) &= (\alpha + \beta(\tau_1))(\alpha + \beta(\tau_2 - \tau_1)) + \alpha\beta(\tau_2 - \tau_1). \end{aligned}$$

In order to quantify the time variations of the end-to-end MC channel, we define the coherence time, T_c , as the minimum time

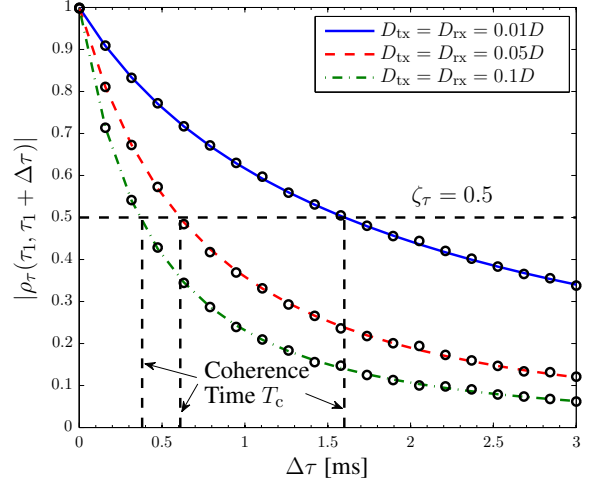


Fig. 18. Absolute correlation $|\rho_\tau(\tau_1, \tau_1 + \Delta\tau)|$ versus $\Delta\tau$ [ms] for a point transmitter, an unbounded environment, a passive spherical receiver of radius $a_{\text{rx}} = 50$ nm, $d_0 = 200$ nm, $\tau_1 = 1$ ms, observation time at $t^P = \max_t \bar{r}(t, \tau = 0)$, $N_{\text{tx}} = 2000$, $D = 10^{-11} \text{ m}^2/\text{s}$, and $D_{\text{tx}} = D_{\text{rx}} = \{0.01, 0.05, 0.1\} \times D$. Markers denote simulation results and lines denote the analytical results based on (92).

$\Delta\tau$ for which $\rho_\tau(\tau_1, \tau_1 + \Delta\tau)$ falls below a certain threshold value $0 < \zeta_\tau < 1$, i.e., [106]

$$T_c = \arg \min_{\forall \Delta\tau > 0} (\rho_\tau(\tau_1, \tau_1 + \Delta\tau) < \zeta_\tau). \quad (94)$$

The coherence time of the channel is a metric which determines the time over which the channel does not change substantially. As such, the particular choice of ζ_τ depends on the application of interest. Future applications of synthetic MC systems that are more robust to CIR variations can assume smaller values of ζ_τ , whereas applications that are more sensitive to CIR variations may require larger values of ζ_τ . For example, typical values of ζ_τ reported in the conventional wireless communications literature span the range from 0.5 to 1 [122]–[124]. Smaller values of ζ_τ are often employed for resource allocation problems, while larger values of ζ_τ are used for channel estimation. In Fig. 18, we show the absolute correlation $|\rho_\tau(\tau_1, \tau_1 + \Delta\tau)|$ versus $\Delta\tau$ for different scenarios of transmitter and receiver mobility, i.e., $D_{\text{tx}} = D_{\text{rx}} = \{0.01, 0.05, 0.1\} \times D$. As can be seen from Fig. 18, the channel mean decorrelates as $\Delta\tau$ increases. Moreover, assuming a fixed threshold $\zeta_\tau = 0.5$, the coherence time decreases as the diffusion coefficients of transmitter and receiver increase.

V. SIMULATION- AND EXPERIMENT-DRIVEN MODELS

The analytical results presented thus far in this tutorial have focused on tractable solutions based on the underlying physical principles of advection, reaction, and diffusion. In order to arrive at these results, we often had to make assumptions that simplify the physical transmitter, receiver, and channel. However, this approach has limitations. Assumptions are generally constrained by specific channel parameters or the conditions for which they accurately apply. For example, we can assume that an environment's outer boundary is

unbounded if it is sufficiently larger than the signaling range; see Fig. 11 and [125]. Similarly, we can model the locally-varying concentration due to a molecule source as uniform if we are observing from a distance that is sufficiently far from the source; see the UCA in Section III and [40].

Sometimes we are able to relax assumptions and still maintain analytical tractability, cf. Section III. When this occurs, we can define a reliable rule of thumb that dictates explicit conditions under which the assumption can be satisfied to some degree of accuracy. For instance in [40], it was shown that the simplified CIR with UCA was within 2 % of the ideal CIR for most of the time of interest if the radius of a spherical receiver was no more than 15 % of the distance from the molecule source to the center of the receiver. However, we generally do not have the option to relax assumptions for analytical tractability while maintaining sufficient accuracy. Furthermore, we might encounter a channel with complex or novel phenomena where we do not yet know what suitable assumptions might be.

In the absence of reliable analytical results, we must rely on data-driven approaches to model a channel. Such approaches can also be used to help verify analytical results. This section reviews simulation and experimental approaches for generating data. Simulations can provide an efficient means for channel modeling, even in the presence of complex and coupled physical phenomena. Reliable experimental data may be preferred, but can be time-consuming and expensive to obtain.

A. Simulation-Driven Models

Simulations of reaction-diffusion systems can be performed over a range of physical scales. As such, there are a range of simulation classes available, which we summarize in Fig. 19 and also discussed in [125]. We refer to these classes as *continuum simulations*, *mesoscopic simulations*, *microscopic simulations*, and *molecular dynamics simulations*. Generally, each class is suitable for a particular scale. Not surprisingly, there is an inherent trade-off between the physical resolution of a simulation and the computational resources (whether measured in time or memory) that are required to simulate it. The continuum approach is most suitable for macroscale systems. Both the microscopic and mesoscopic approaches can be appropriate for microscale systems. The molecular dynamics approach is most suitable for systems at the nanoscale and smaller. While the microscopic approach has been the most common simulation method within the MC research community, here we discuss all four approaches, their relevance, and also the potential to combine them in a single simulation. For the microscopic and mesoscopic approaches, we also describe how to implement a simple simulation.

1) *Continuum Simulations*: When the physical scale of a simulation, including the number of molecules, is sufficiently large, then the evolution of the system can be directly described using the corresponding spatio-temporal PDEs, see e.g. (29). We refer to these as continuum simulations. Specifically, finite element analysis is used to spatially partition the system into a grid (see Fig. 19a)), and the system is simulated over a sequence of time steps. The molecule concentrations at each node in the grid are updated in every time step according to the

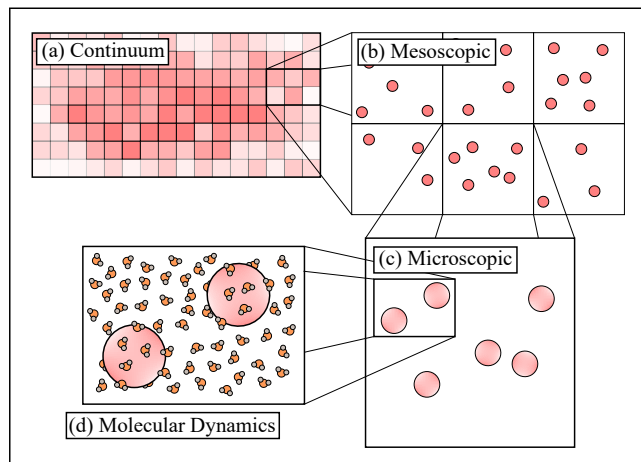


Fig. 19. Physical scales of molecular simulation. a) Continuum simulations solve the PDEs that describe the system. Molecular concentrations are non-negative and real-valued. b) Mesoscopic simulations proceed as a sequence of events, where each event is an occurrence of a chemical reaction or a molecule moving between adjacent subvolumes. c) Microscopic simulations individually track each molecule of interest. The solute molecules diffuse within a continuum of solvent molecules. d) Molecular dynamics simulations model *all* individual atoms and molecules, including intermolecular forces and collisions.

differential equations that describe the phenomena. The updated concentrations are always non-negative real values. Popular commercial solvers that follow this approach include COMSOL Multiphysics [126] and ANSYS [127]. This approach was used in an MC context in [128] for the characterization of the diffusion of autoinducer molecules in a bacterial environment.

Unless the differential equations are stochastic or explicit noise sources are introduced, the continuum simulation of a system is deterministic. Generally, the accuracy depends on the resolution of the grid and the size of the time step; more accurate simulations can be performed by increasing the grid resolution and decreasing the size of the time step. However, as the nodes in the grid become increasingly close, the number of molecules associated with each node decreases. When the molecule concentrations get sufficiently small, it becomes more appropriate to consider *integer* numbers of molecules instead of continuous-valued concentrations. Thus, we next discuss mesoscopic simulations.

2) *Mesoscopic Simulations*: Like continuum simulations, mesoscopic simulations also partition the system into a grid. The resulting containers are commonly referred to as subvolumes or voxels, and have also been referred to as lattices; see [129]. However, instead of tracking continuous molecule concentrations, mesoscopic modeling counts discrete numbers of molecules in each subvolume; see Fig. 19b) and implementations for MC systems in [125], [129]. Instead of deterministically solving the system's set of PDEs, mesoscopic simulations proceed by stochastically generating *event* times, where each event is the occurrence of a chemical reaction or a molecule's transition between two subvolumes. The key physical assumptions to justify using a mesoscopic approach are that *i*) molecules within a given subvolume are uniformly distributed, and *ii*) the solvent molecules in a subvolume can be treated as a homogeneous continuum that is in

thermal equilibrium. When these assumptions are satisfied, the mesoscopic approach can simulate chemical reactions *exactly* (in a statistical sense, as proven in [130]). Furthermore, if the subvolume sizes are appropriately chosen, advection-reaction-diffusion systems can also be simulated exactly; see [131], [132]. An important constraint is that the subvolume size (e.g., cube length) ℓ must be much smaller than both $\sqrt{2nDt_r}$ and $2D/|v|$, where n is the dimension of the subvolume, t_r is the characteristic time of the fastest reaction in the system, D is the diffusion coefficient of the largest corresponding reactant, and $|v|$ is the magnitude of the flow velocity. If this is not satisfied, then we cannot safely assume that the subvolumes are well-stirred (i.e., that the molecules are uniformly distributed).

Simple Implementation: A basic implementation of a mesoscopic simulation with equal-sized subvolumes (of length ℓ) is as follows. Let $U_{s,m}$ be the number of molecules of the m -th type that are in the s -th subvolume. Events are associated with propensities. The propensity $\alpha_{s,q,m}$ of a transition of a molecule of the m -th type to diffuse from the s -th subvolume to the q -th subvolume, where these two subvolumes are adjacent and share a face, is [133, Eq. (1.6)]

$$\alpha_{s,q,m} = \frac{D_m}{\ell^2} U_{s,m}, \quad (95)$$

where D_m is the diffusion coefficient of the m -th molecule type. The propensity $\beta_{s,p}$ of the p -th chemical reaction in the s -th subvolume is [134, Eq. (6)]

$$\beta_{s,p} = \kappa_p V, \quad (96)$$

$$\beta_{s,p} = \kappa_p U_{s,m}, \quad (97)$$

$$\beta_{s,p} = \frac{\kappa_p U_{s,m} U_{s,n}}{V}, \quad (98)$$

for zeroth-, first-, and second-order reactions, respectively, where the order corresponds to the number of reactants. κ_p is the corresponding reaction rate constant, V is the subvolume volume, and $U_{s,m}$ and $U_{s,n}$ are the corresponding numbers of reactant molecules. For the entire system, the *total* propensity γ_{tot} is then

$$\gamma_{\text{tot}} = \sum_{s,q,m} \alpha_{s,q,m} + \sum_{s,p} \beta_{s,p}, \quad (99)$$

and we can simulate the time t_{next} of the next event in the system by generating exponential random variable

$$t_{\text{next}} = -\frac{\log u}{\gamma_{\text{tot}}}, \quad (100)$$

where u is a random number uniformly distributed between 0 and 1. We can determine which of the possible events occurred by tossing a weighted die, where the likelihood of each event is proportional to its associated propensity. Once the event is determined, we update the molecule counts, update the corresponding propensities, and repeat the process to find the next event.

Remark 25: We note that there are mathematically equivalent but more computationally efficient implementations, particularly when updating propensities. These include Gibson and Bruck’s Next Reaction Method; see [135]. Furthermore, different accuracy-efficiency trade-offs can be introduced to

provide more flexible scalability. For example, *tau-leaping* can be used to execute multiple events in a constant time step, where “tau” refers to the time step size; see [136]. Tau-leaping enables a transition between continuum and mesoscopic simulations; if the number of events during one “leap” is sufficiently large, then it can be treated as a deterministic value. As long as the propensities do not significantly change between time steps, then tau-leaping’s computational efficiency gains can be made with minimal losses in accuracy. \square

3) Microscopic Simulations: In some sense, microscopic simulations are the dual of the mesoscopic approach. Whereas the (non-leaping) mesoscopic approach is continuous over time and discrete over space, the common microscopic approach implementation is discrete over time and continuous over space; see [76]. Instead of relying on well-stirred subvolumes, microscopic simulations track every molecule individually (i.e., particle-based simulation); see Fig. 19c). Nevertheless, they still assume that the solvent is a continuum of molecules, which means that the diffusion of the molecules of interest is still governed by a diffusion coefficient.

Simple Implementation: A basic implementation of a microscopic simulation with flow and first-order reactions in the propagation environment is as follows. In each time step Δt , every molecule is tested for every possible first-order reaction. If there is only one potential reaction, and the associated reaction rate is κ , then the corresponding reaction probability P_{rxn} is [76, Eq. (14)]

$$P_{\text{rxn}} = 1 - \exp(-\kappa \Delta t). \quad (101)$$

If a coin flip with this probability is successful, then the molecule is converted to the corresponding reaction product. After all of the possible reactions have been tested, the remaining molecules are diffused along every available dimension by adding a displacement of $\sqrt{2D\Delta t} \times \mathcal{N}(0, 1)$ towards each dimension of the Cartesian coordinate system, cf. (1). The realizations are independent for every molecule and along every dimension. Furthermore, if the environment has a bulk flow with a component v along a particular dimension, then every molecule should have an additional displacement of $v\Delta t$ along that dimension, cf. (11). Diffusion should be unimpeded, unless there are boundaries in the environment. For example, if a molecule crosses a solid reflective surface, then the coordinate that is normal to the surface is reverted to its value before diffusion. If a molecule crosses an absorbing surface, then it should be consumed by the absorbing reaction.

Due to their simplicity and their suitability for simulations over a range of nanometers to micrometers, microscopic simulations have been common for cellular systems and also specifically for MC systems. Mature tools from the physical chemistry community include Smoldyn (see [76], [145]). Microscopic tools that have been developed specifically for the MC community include BiNS2 [142], N3Sim [143], MUCIN [97], and AcCoRD [125].

4) Molecular Dynamics Simulations: At a more precise scale, solvent molecules and their interactions with solute molecules and with each other can be modeled in detail; see Fig. 19d). These are molecular dynamics simulations, and they might account for intermolecular forces (including those

TABLE III
SUMMARY OF THE SIMULATION METHODS REVIEWED IN SECTION V-A.

Class	Typical System Scale	Physical Chemistry Examples and Algorithms	MC Community Examples	Notes
Continuum	Greater than micron	COMSOL Multiphysics [126], ANSYS [127], Virtual Cell [137]	nanoNS3 [138]	Solving PDEs. Requires large concentrations.
Mesoscopic	Micron and larger	Gillespie's Method [130], [136], URDME [139]	BNSim [140]	Counting molecules inside subvolumes. Requires homogeneous concentration within a subvolume.
Microscopic	Micron and smaller	Smoldyn [137], [141]	BiNS2 [142], N3Sim [143], MUCIN [97], AcCoRD [125]	Tracking molecules in solvent. Most common approach for MC simulation.
Molecular Dynamics	Nanometer	LAMMPS [144]	n/a	Tracking all molecules and intermolecular forces.
Hybrid	Multiple	Smoldyn (+mesoscopic), URDME (+microscopic), LAMMPS (+Continuum), Virtual Cell (+microscopic)	AcCoRD (+mesoscopic)	Combining multiple classes. Needs special treatment at interface between classes. Can improve scalability.

imposed by charge potentials) and collision dynamics. One such example is the Large-scale Atomic/Molecular Massively Parallel Simulator (LAMMPS); see [144]. Due to the very large density of molecules to be considered, molecular dynamics simulations are best suited for very small systems, e.g., on a nanoscale. For instance, molecular dynamic simulations can be used to study how the conformation of receptor proteins change after binding to a specific molecule. Thus, they have generally not been applied to study MC systems.

Remark 26 (Hybrid Simulations): The aforementioned discussion of simulation classes has emphasized their suitability for simulations over different physical scales. However, a particular system might have multiple scales of interest. In order to avoid constraining the entire simulation by the most granular approach needed, hybrid simulation tools have sought to integrate different classes within a single simulation. One approach has been to combine microscopic and mesoscopic models, using hybrid interfaces such as that proposed in [133] and later implemented in Smoldyn (see [141]) and AcCoRD (see [125]). Other examples include the integration of the molecular dynamics solver LAMMPS with a continuum model (see [146]), and the integration of the continuum solver Virtual Cell with the microscopic approach in Smoldyn (see [137]). \square

The aforementioned different simulation methods, their characteristics, and example implementations are concisely summarized in Table III. In the following, we present an example for CIR characterization of an MC system using the AcCoRD simulator.

Example 11 (Example Simulation): We complete our discussion of simulations with a brief demonstration. We consider an extension of the bounded rectangular-duct channel discussed in Section III-D that has no readily available analytical channel response. Nevertheless, we can simulate the system. The environment is a microfluidic system where two chambers are connected via a long pipe, as shown and described in Fig. 20. We place $N_{tx} = 500$ molecules uniformly within one of the chambers (i.e., a cube). These molecules can diffuse out through the rectangular pipe and into the other chamber and no flow is considered. The second chamber has a perfectly-absorbing surface and we count the number of molecules that are absorbed. We assume that the receiver counts the number of molecules absorbed by time t . Therefore, the receiver can be classified as nR-AMC, i.e., non-recurrent and accumulative-molecule-counting, with received signal $r(t) = n^{arv}(t)$, cf. Section IV-A. A realization of this system is simulated using a microscopic approach in the AcCoRD simulator with a simulation time step of $\Delta t = 1$ ms. The number of absorbed molecules $r(t)$

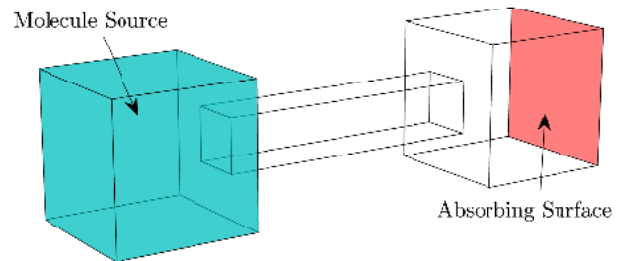


Fig. 20. Example environment to simulate. This “dumbbell”-shaped environment represents two connected microfluidic chambers with a rectangular duct between them. It is composed of two cubes of length $32 \mu\text{m}$ that are connected by a rectangular pipe of size $60 \mu\text{m} \times 12 \mu\text{m} \times 12 \mu\text{m}$. We can also vary the length of the pipe. The left cube has molecules initialized throughout it. The right cube has an absorbing surface on the far side. An analytical channel response for this environment is not readily available.

is plotted in Fig. 21 for different pipe lengths. We see that all released molecules are absorbed within about 850 s for the shortest pipe length (i.e., $60 \mu\text{m}$). As the distance between the two chambers increases, fewer molecules get absorbed within the same time. We note that one can obtain the CIR of this system, i.e., $h(t)$ defined as the probability of a molecule being absorbed at the receiver in interval $(0, t]$ after its release by the transmitter at $t = 0$, by simulating the system for many realizations and averaging the result, i.e., $h(t) = \mathbb{E} \{r(t)/N_{tx}\}$. \square

B. Experiment-driven Models

In the previous subsection, we have seen how elaborate simulations can be used for scenarios where it is difficult or even impossible to derive an analytical model based on physical principles. However, for practical systems, we may also face situations where even simulation of certain phenomena is challenging. In fact, even complex simulation methods typically cannot account for all characteristics of a real experimental environment. In the following, we first highlight some of the unique characteristics of two existing experimental platforms for MC [64], [147] that cannot be easily modeled or simulated. Subsequently, to cope with the aforementioned challenges, we present a general data-driven modeling approach which is then applied to an example experimental system.

1) Challenges in Modeling Existing Experimental Systems: Several experimental systems exist for demonstrating MC. These testbeds include both non-biological systems [64], [66], [79], [148]–[152] and biological systems [147], [153]–[156]. To show the need for experiment-driven models, we review the challenges of channel modeling for two of these testbeds.

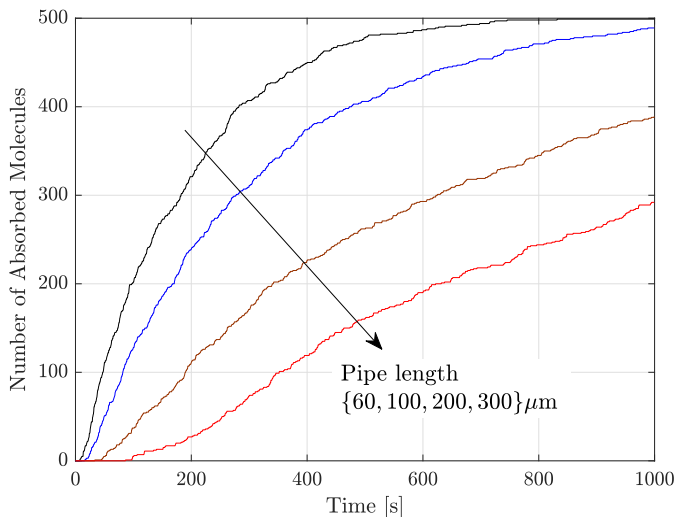


Fig. 21. Number of absorbed molecules in the simulation of the system described in Fig. 20. 500 molecules are instantaneously released throughout the molecule source, which freely diffuse with a diffusion coefficient of $10^{-10} \frac{\text{m}^2}{\text{s}}$. A single realization is shown for different pipe lengths using a simulation time step of 1 ms.

Example 12 (Non-biological Testbed [64]): An early experimental MC system was presented in [64] and is based on spraying and detecting alcohol in open space. In [157], it was shown that a simple model based on diffusion and the flow generated by a fan cannot accurately explain the measurements obtained from the testbed in [64] due to system nonlinearities whose exact cause is not known. For example, the spray that is used for releasing the chemicals may not produce consistently-sized droplets in the spray stream across different trials, the flow may show turbulent behavior that is difficult to model, and the receiver sensor is prone to long response and recovery times. \square

Example 13 (Biological Testbed [147]): The biological MC testbed reported in [147] converts an electrically controlled optical signal into a chemical signal. In particular, for this testbed, *E. coli* bacteria were genetically modified to incorporate light-driven proton pumps in their cell membranes. Upon a light stimulus, the modified bacteria then pump protons out into the environment which increases the proton concentration outside the bacteria. The resulting proton concentration was measured by a pH sensor playing the role of the receiver. Although complex models were developed in the biology literature for describing the proton release rate of proton pumps as a function of a given induced optical intensity [158]–[160], they typically do not account for all of the dynamics inherent to living cells. In fact, the growth, dying, and varying living conditions of the bacteria due to constant exposure to light may impact the channel model of the MC system in [147] and cannot be easily captured analytically or via simulation. \square

We note that similar inherent randomness and nonlinearities as discussed for the two examples above also exist for other experimental testbeds [66], [79], [148]–[156], [161] and are challenging to model analytically or even simulate since their exact cause is unknown.

2) Data-Driven Model: To address the aforementioned shortcomings of analytical and simulation models, we propose to employ data-driven models to account for the unpredictable randomness and nonlinearities of real MC systems. The basic idea behind these models is to select an appropriate parametric model and choose its corresponding parameters to fit the measurement data. In the following, we describe two different approaches for selecting a suitable parametric model.

Physically-Motivated Parametric Models: Here, the model is chosen based on physics' first principles. For instance, in [157], a mathematical model is developed for the testbed in [64] which is based on the solution to the advection-diffusion equation with uniform flow, cf. (18). Nevertheless, the parameters of the original analytical model were modified to fit the model to the experimental data. As another example, we consider the system in [79]. The model uses magnetic nanoparticles in duct flow that are detected upon moving through a coil enclosing the duct. Here, the parametric model is based on laminar flow, cf. (14), and depends on the initial distribution of the particles released across the cross-section of the duct. The adopted parametric model was then shown to accurately model the complex advection-diffusion process in the duct after fitting its parameters to the measurement data. In general, after choosing the parametric model, standard curve fitting toolboxes can be employed to find the model parameters. One common approach is to use the parameter set that minimizes the mean square error between the model and the measurement data [79], [147], [157].

Blind Models based on Neural Networks: In the absence of an appropriate physically-motivated model, an alternative option is to employ blind models based on neural networks to jointly learn the model and its parameters [169], [170]. One suitable network architecture for this purpose is the generative adversarial network (GAN) which is able to generate a model that creates artificial data very similar to the measurement data [171]. The advantage of such blind parametric models is that they can be universally applied to general MC systems, whereas physically-motivated models have to be carefully chosen according to the MC system under consideration. On the other hand, the parameters of a physically-motivated model have physical meaning, which is not the case for the parameters of a trained neural network. The other challenge of channel modeling based on neural networks is that they typically require much more experimental data than parametric models to construct the model. This is not surprising since without domain knowledge, the number of parameters to be learned for a neural network is much larger than that for a parametric model.

Table IV summarizes the components (i.e., the transmitter, the receiver, the channel, and the signaling messenger), the characteristics (synthetic versus biological), and if available, the corresponding data-driven channel models of several MC testbeds that have been reported in recent years. In the following, we explain a data-driven modeling methodology for one example in detail.

3) Example of an Experiment-driven Model: In order to further familiarize the reader with the main steps of developing an experiment-driven channel model and to highlight some

TABLE IV
SUMMARY OF COMPONENTS, CHARACTERISTICS, AND CORRESPONDING DATA-DRIVEN MODELS OF SEVERAL MC TESTBEDS.

Testbed	Messengers	Channel	Transmitter	Receiver	Model
Synthetic Systems					
Farsad et al. [64]	Ethanol	Open air; turbulent flow via fan	Spray pump	Semiconductor gas sensor	Parametric analytical model; unbounded diffusion; constant uniform flow [157]
Farsad et al. [66]	Acid and base	Water in vessel; laminar flow via water pump	Injection pump	pH electrode	Continuum simulation; unbounded 1D; finite difference method [162]
					Continuum and microscopic simulations; unbounded 1D-3D; hybrid analytical-numerical method [74]
Unterweger et al. [79]	Magnetic nanoparticles	Water in vessel; laminar flow via water pump	Injection pump	Magnetic susceptometer	Parametric analytical model; flow-dominated
Kennedy et al. [150], [161]	Isopropyl alcohol	Air in pipe; flow via fan	Spraying pump	Photoionization gas sensor	Parametric data-driven model [163]
Giannoukos et al. [151]	Chemical odorants (e.g. Acetone, n-Hexane)	Nitrogen gas in pipe; flow via mass flow controller	Spraying pump	Mass spectrometer	Parametric analytical model; 1D diffusion; constant uniform flow; transparent receiver [164]
					Approximate analytical model; 3D diffusion; constant uniform flow; absorbing receiver [165]
Atthanayake et al. [148]	Fluorescent dye	Water in chamber; turbulent flow	Injection pump	Fluorescence detector	-
Abbaszadeh et al. [166]	Vortex ring (specific structure of gas molecules)	Open air; turbulent flow	Injection piston	Motion camera	-
Leo et al. [152]	Microfluidic droplet	Fluorinated oil FC-3283 in microfluidic channel; turbulent flow	Injection pump	Camera	Continuum simulation using OpenFOAM [167]
Tuccitto et al. [168]	Fluorescent chemicals	Liquid pH buffer in tube; laminar flow	Injection pump	Fluorescence detector	Continuum simulation
Biological Systems					
Krishnaswamy et al. [153]	Molecule C6-HSL	2×YT broth in microfluidic channel; diffusion and laminar flow	Injection pump	<i>E. coli</i> bacteria	Modeling GFP production via bacterial CRNs [156]
Felicetti et al. [154]	Protein CD40L	Liquid in confinement; diffusion	Platelet cells via manual stimulation	Endothelial cells visualized by fluorescence detector	Approximate model; diffusion in infinite cylinder
Grebenstein et al. [147]	Proton H ⁺	Liquid pH buffer in confinement; diffusion	<i>E. coli</i> bacteria via light stimulus	pH electrode	Parametric analytical model; state-dependent diffusion
Nakano et al. [155]	Lucifer yellow	Diffusion through gap junctions expressed by HeLa cells	Manual injection	Fluorescence detector	-
	Calcium ion C ²⁺		Photolysis of caged-ATP	Calcium imaging	

peculiarities that may arise, we present the modeling methodology for the biological testbed in [147] in some detail, cf. Example 13 and Fig. 22.

Simple Physically-Motivated Parametric Model: In order to arrive at an analytical model, the following assumptions are made in [147]. It is assumed that the bacteria (i.e., the transmitter) are uniformly distributed in their container and that all bacteria are subject to the same light stimulus at the same intensity because the bacteria suspension is continuously stirred. It is further assumed that the bacteria begin and stop pumping protons (i.e., signaling molecules) instantly when the light is activated and deactivated, respectively. Furthermore, it is assumed that the bacteria take up protons in a passive manner, i.e., protons are consumed by the bacteria which lowers the measured proton concentration. Finally, it is assumed that the pH measuring device (i.e., the receiver) is passive and that its presence does not change the proton concentration or the behavior of the bacteria. These assumptions do not strictly hold but are reasonable in consideration of the size of the setup, the pumping speed of a proton pump, and the characteristics of the bacteria [147]. We note that counting the individual molecules observed at the receiver, $r(t)$, might be a reasonable assumption for nanomachines; however, for experimental testbeds such as [147], computing $r(t)$ is not feasible. Hence, in [147], the proton concentration obtained from the measured pH was considered as the received signal and was modeled as

$$r_c(t) = \frac{r(t)}{V_{rx}} = c_b(t) + w(t), \quad (102)$$

where $c_b(t)$ is the expected proton concentration and $w(t)$ is a random additive noise. It was shown that $w(t)$ follows a Gaussian distribution and this was justified using the CLT since $w(t)$ consists of different types of noises including diffusion (counting) noise, pH sensor circuitry noise, and the noise

inherent to the biological machinery of the bacteria. We assume that the light stimulates the bacteria over a time interval (i.e., a rectangular input pulse) since an impulsive stimulus (i.e., a delta input) does not effectively stimulate the bacteria to release a sufficient number of protons into the MC channel, see Remark 20. Under the aforementioned assumptions, the expected proton concentration $c_b(t)$, depending on whether the light is on ($i = 1$) or off ($i = 0$), can be obtained as

$$c_b(t) = c_b(t_0) + (c_i^\infty - c_b(t_0)) \left(1 - \exp\left(-\frac{t - t_0}{\tau_i}\right) \right), \quad (103)$$

where $c_b(t_0)$ is the initial concentration at starting time t_0 , c_i^∞ is the saturation concentration, and τ_i is a time constant. The parameters of this model are $c_b(t_0)$, c_i^∞ , t_0 , and τ_i , which are found using nonlinear least square error minimization to fit the measurement data. For example, in Fig. 23a), we apply constant illumination for 54 minutes followed by darkness. The corresponding measurement signal and the fitted model using (103) are shown in Fig. 23b) using blue and black lines, respectively. As expected, the concentration increases upon illumination and decreases quickly in darkness. Nevertheless, the model in (103) fails to accurately follow the measurement data. In fact, there exists an additional persistent decreasing bias in the measurement signal which is not anticipated by the saturation model in (103).

Enhanced Parametric Model: The assumptions made to arrive at (103) do not account for the dynamics inherent to living cells. As such, cells can be growing in number or dying, or their fidelity can change. Motivated by the observations from the measurement data in Fig. 23b), it was suggested in [147] to enhance the model (103) with a simple additive linear offset as follows

$$c_d(t) = m_d \cdot (t - t_0), \quad (104)$$

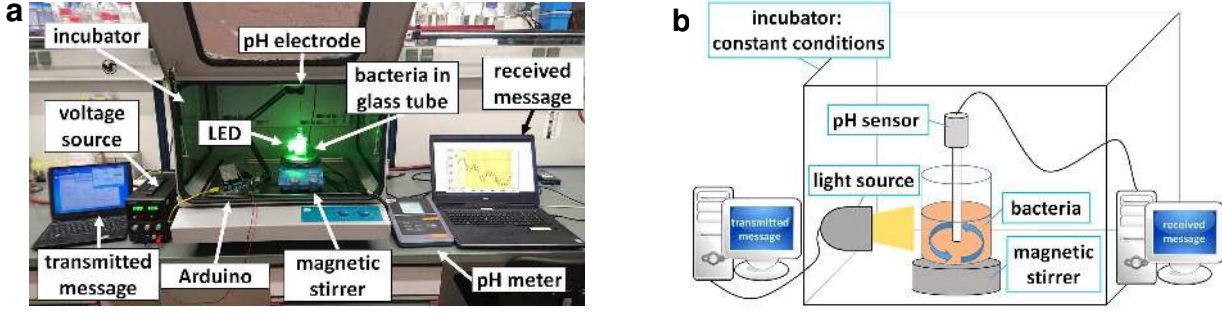


Fig. 22. Biological testbed. (a) Benchtop experimental setup; (b) Schematic illustration. Taken from [147].

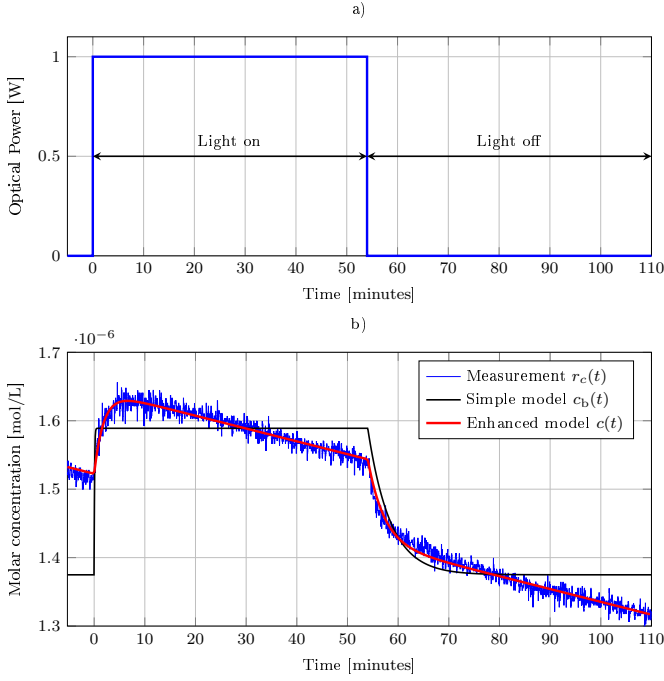


Fig. 23. Experimental data. a) Optical signal versus time. b) The measured proton concentration as well as the simple and enhanced models versus time. Data taken from [147].

where m_d is a parameter controlling the slope of the bias. The extended model is then given by $r_c(t) = c(t) + w(t)$ where $c(t) = c_b(t) + c_d(t)$. From Fig. 23b), we can observe that the enhanced model, shown in red, fits the measurement data well. This example shows that further modification of a model that was obtained solely based on physical principles may be needed to arrive at an appropriate parametric model for an experimental system.

VI. CHALLENGES AND DIRECTIONS FOR FUTURE WORK

MC is still in its early stages of development and our understanding of MC channels is still quite limited. In the following, we review some potential challenges and open research problems which have to be addressed for successful deployment of MC systems.

Particle Generation and Signaling Pathways: Although the impact of CRNs in the *physical channel* (e.g. degradation reactions) on the CIR of MC systems has been studied, see

Section III-D and [43], [44], [65]–[68], [74], less attention has been dedicated to the analysis of the influence of the CRNs at the transmitter and receiver. Such CRNs include particle generation reaction networks at the transmitter and signaling pathways at the receiver. In the MC literature, there are some preliminary works that have studied the impact of particle generation reaction networks and also simplified signaling pathways; see e.g. [9], [88]. However, the corresponding models are derived with a mesoscopic modeling approach. Analytical CIR models that take the impact of these CRNs into account are crucial for system design and hence constitute an interesting research challenge.

Turbulent Flow: The majority of the CIR models in the MC literature for advection channels have been developed based on the assumption of a uniform or laminar flow velocity field, cf. Section III-D. However, for several MC environments, such as large arteries (e.g. the aorta) and macroscale environments (e.g. oil pipe lines), flow may exhibit turbulent behavior [60]. In particular, turbulence can occur when the MC channel is non-homogeneous, e.g., due to the presence of obstacles in the physical channel. Therefore, studying and analyzing advection channels with a *turbulent* velocity field is an important open research problem. First results towards analyzing turbulent flow for MC systems were reported in [148].

Sample Correlation: Multiple-sample detectors are used in the MC literature to improve the detection performance [38], [39], [42], [44], [53], [65], [90], [96], [105]. It is typically assumed that different samples are statistically independent from each other. However, this assumption holds only when the sampling interval is chosen large enough such that the independence of consecutive samples is ensured. In Section IV-F1, we have numerically evaluated the correlation among consecutive samples, and in [53], the mutual information between consecutive samples is numerically evaluated. We note that sample correlation significantly depends on the type of receiver, e.g., a recurrent, non-recurrent, AMC, or IMC receiver, and its physical and chemical properties, e.g., the size, the number of receptors, and the reaction rate constants of the binding and unbinding reactions for a reactive receiver. Therefore, a careful study of sample correlation and the minimum sampling interval needed to ensure sample independence for different receiver types is essential for the applicability and performance analysis of the multiple-sample detectors proposed in the literature.

Complex Networks: In this tutorial, we focused on a single

one-way communication link from a transmitter to a receiver. This is the simplest communication architecture and hence the basis for more complex network typologies. We note that although multi-node networks can often be decomposed into a superposition of individual links, there are certain scenarios where such a decomposition is invalid. For instance, if multiple reactive receivers are in the environment, the presence of each receiver will impact the signal received at *any* other receiver, see e.g. [30], [31], and the discussion in Section I-B. Moreover, in Section III, we considered MC environments with simple boundary and initial conditions in order to derive analytical channel models. However, some important MC environments, such as the cardiovascular system, are quite complex and cannot be fully modeled based on physics' first principals. One approach is to develop simulation environments for such complex networks, see [142] and Section V-A. Nevertheless, for system design, it is desirable to have simple yet sufficiently accurate analytical models for complex multi-node networks. Developing such analytical models constitutes an important future research topic, see [49], [82], [154], [172] for some related works.

Microscale and Macroscale Models: MC systems have numerous potential applications which range from targeted drug delivery and health monitoring for microscale systems to communication in oil pipelines or chemical reactors and environmental monitoring for macroscale systems. Nevertheless, most of the current literature has targeted microscale applications and the available models are typically developed for microscale MC environments. However, macroscale and microscale MC systems may require quite different considerations. For instance, the number of molecules needed for communication at macroscale is typically much larger than that needed for communication at microscale. Moreover, while at microscale, molecules can be counted at the receiver (e.g. via ligand-receptors), at macroscale, receivers usually measure a quantity that is a function of the molecule concentration (e.g. a pH sensor was used in [66], [79] and mass spectroscopy was used in [151], [165]), see also Table IV. In summary, the development of channel models for macroscale MC systems is an important and interesting topic for future research. We refer the interested reader to [4], [79], [151], [157], [165], [173] for preliminary results on channel modeling for macroscale MC systems.

Generally-Accepted and Experimentally-Verified Models: Over the past years, several non-biological experimental testbeds [64], [66], [79], [148]–[152] and biological experimental testbeds [147], [153]–[156] have been developed to demonstrate MC, see Table IV. Most of these experimental testbeds were developed as proofs-of-concept for human-designed MC and mathematical models that explain the corresponding measurement data are usually too simplistic if available at all, see Table IV. However, for the advancement of MC research, it will be crucial to specify generally-accepted test channels with corresponding experimentally-verified mathematical channel models. Then, researchers in the MC community can use these established models for the design and performance analysis of newly-developed communication schemes.

VII. CONCLUSIONS

This paper provided a comprehensive tutorial review of the diffusive MC channel models available in the literature. To this end, we first presented the underlying fundamental laws that govern diffusion, advection, and chemical reactions in MC channels and constitute the essential mathematical tools from biology, chemistry, and physics required for the development of MC channel models. Subsequently, we reviewed the main end-to-end channel models reported in the diffusive MC literature and showed how they were developed from basic physical principles. The reviewed end-to-end channel models included the joint effects of release mechanisms, the physical channel, and reception mechanisms. Moreover, we provided a unified definition for the received signal that included the representation obtained by both timing and counting receivers as special cases. Furthermore, for counting receivers, we derived signal models relevant for different time scales. We generalized these models to account for interfering noise molecules and ISI and studied the correlation among the received signals observed at different time scales. In addition, simulation-driven and experiment-driven channel models were investigated for complex scenarios where simple MC channel models cannot be obtained from basic physical principles. Finally, we provided a discussion of challenges, open research problems, and future directions for channel modeling of diffusive MC systems.

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