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Characterization of Acetogenic and Methanogenic Leachates Generated From a Sanitary Landfill Site

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***Abstract*—**Decomposition processes take place in landfill generate leachates that can be categorized mainly of acetogenic and methanogenic in nature. BOD:COD ratio computed in this study for a landfill site over a 3 years duration revealed as a good indicator to identify acetogenic leachate from methanogenic leachate. Correlation relationships to predict pollutant level taking into consideration of climatic condition are derived.

***Keywords*—**Acetogenic Leachate, Methanogenic Leachate, BOD:COD Ratio.

I. INTRODUCTION

LANDFILLS are major sources of groundwater and land contamination that can cause adverse impacts to the environment. Perforation of pollutants due to waste disposal which passes through as leachate if not properly handled will diffuse through the landfills and contaminate soils and groundwater if left unchecked. The constituents of leachate can be categorized into four types namely organic matter, inorganic matter, heavy metal and xenobiotic organic compounds [1].

The extent of contamination from the leachate depends on the type of control measures used in landfill. Nevertheless, pollutants in the leachate of different composition have different impacts on the environment. Even under controlled conditions such as those of a well planned and well managed landfill, leachate may percolate or penetrate through natural ground and may still contaminate groundwater and ultimate contaminate fresh water supplies over time. The environmental impact is most significant particularly those landfills without integration of engineering controls such as liners and leachate collection system.

The content of leachate generated from most landfill is subject to several factors such as climatic condition, infiltration and waste type. As leachate percolates through waste strata layers that undergo various decomposition high amounts of both organic matter and inorganic matters are found to be higher than those in groundwater [2]-[3].

Both temperature and water content in landfill will affect the rate of waste decomposition which is usually lower in dry weather condition. Dissolved organic matter in leachate consists of various organic and inorganic constituents. Higher organic matter is anticipated in acetogenic phase whereas inorganic matter is lower in methanogenic phase due to lower dissolved organic matter and higher pH [4]-[8].

Leachate content generated from waste landfill can be broadly categorized as organic matters, inorganic matter, xenobiotic organic compounds and other compounds due to various conditions such as weather, infiltration, gravity drainage and groundwater inflow. The strength of leachate is depend on decomposition processes comprising of biological and chemical reactions which vary from pH and high concentration of biodegradable organic pollutant in early methanogenic phase to high pH and lower concentration of biodegradable organic content in later methanogenic phase.

The purpose of this paper is to study the impact of temperature and precipitation on landfill performance that yield various pollutant removal experiencing both acetogenic and methanogenic phases.

II. MATERIALS AND METHOD

Performance data from a landfill site in Toronto, Canada was evaluated over a period of 3 years to assess the range of pollutant in the leachate. The performance data is depicted in Table 1.

The dissolved organic matters were evaluated in terms of BOD (Biochemical Oxygen Demand), COD (Chemical Oxygen Demand) and DOC (Dissolved Organic Carbon). The inorganic matters such as ammonia, calcium, chloride, iron, magnesium, sodium and sulfate of the landfill leachate were evaluated and xenobiotic organic compounds such as phenols were also evaluated. Statistical study using regression analysis to establish correlation relationship to evaluate pollutants from landfill that are leached out from this traditional waste landfill using clay liner taking into consideration of basic properties and factors influencing landfill performance include climatic conditions such as temperature and precipitation and also the organic content of leachate in terms of BOD:COD ratio.

III. RESULTS AND DISCUSSIONS

As water passes through waste strata layer in landfill it triggers and activates decomposition due to present of microorganisms. The decomposition can be defined in two

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phases, firstly soluble organic matter produced due to aerobic decomposition or acetogenic phase and secondly methane and carbon dioxide produce due to anaerobic decomposition or methanogenic phase.

TABLE I
PERFORMANCE DATA OF LANDFILL SITE

Parameter	Year	Month											
		1	2	3	4	5	6	7	8	9	10	11	12
Temperature													
High	1	9	10.5	18.2	25	27.5	31.8	30.5	29.9	29.1	27.2	15	10
	2	18	7.3	12.9	28	24.9	34.6	35.5	34	32	26.1	19.1	4.4
	3	10	6	16	17	33	33	34	36	27	22	16	12
	4	11	4	15	25	29	34	34	33	32	32	19	3
	5	15	6	11.7	23	24.1	30.4	30.0	29.4	29.7	24.4	17.4	6.1
Low	1	-23	-19	-10.3	-7	-2	6.6	11.4	9.3	5.5	0.7	-4.6	-24
	2	-24.2	-15	-14.2	0	-0.4	22.1	12.4	13	6.4	-0.3	-13.3	-15.2
	3	-13	-11	-10	-4	3	8	13	13	5	1	-2	-10
	4	-17	-20	-22	-7	4	7	12	11	5	0	-9	-11
	5	-17	-18.2	-13.4	-3.6	3.1	9.0	12.9	10.9	8.1	-2.1	-9.6	-12.4
Precipitation	1	50.6	22.5	63.2	62.4	98.8	67.5	121.1	69	25.70	35	61	96.6
	2	71.70	69.90	38.30	98.30	14.90	32.50	18.50	139.40	244.30	46.30	104.80	60.10
	3	45.6	45.5	56.9	64.0	66.0	68.9	76.6	84.2	74.2	67.0	70.3	65.5
	4	45.6	45.5	56.9	64.0	66.0	68.9	76.6	84.2	74.2	67.0	70.3	65.5
	5	45.6	45.5	56.9	64.0	66.0	68.9	76.6	84.2	74.2	67.0	70.3	65.5
BOD/COD ratio	1	0.07	0.08	0.10	0.09	0.07	0.06	0.07	0.06	0.07	0.13	0.19	0.15
	2	0.80	0.15	0.59	0.22	0.15	0.08	0.06	0.06	0.10			
	3	0.23	0.11	0.14	0.09	0.13	0.09	0.12	0.08	0.21	0.09	0.09	0.13
	4	0.55	0.12	0.49	0.8	0.28	0.14	0.06	0.09	0.14	0.08	0.1	0.11
	5	0.22	0.12	0.21	0.19	0.49	0.11	0.11	0.09	0.39	0.27	0.28	0.22
Performance Data													
Alkalinity	1	2700	2700	1600	2800	3400	2500	3200	3200	3100	3500	3500	2800
	2	2400	2700	3100	3100	3400	3300	4700	3300	2700	0	0	0
	3	1200	2000	1500	2300	2400	2400	2500	2300	740	1900	2150	1600
	4	1700	1600	1400	2200	3100	3100	3400	3800	2900	3900	3400	2500
	5	2800	3200	2500	2200	2400	2420	1900	1200	1260	1300	980	1670
Ammonia	1	230	270	140	240	330	270	300	330	300	290	320	250
	2	200	260	180	200	340	300	450	370	290	0	0	0
	3	120	212	170	240	270	240	190	220	46.0	190	190	160
	4	140	150	110	200	230	310	320	440	230	380	310	210
	5	360	330	280	210	220	215	299	125	119	92	85	148
BOD	1	31	53	38	53	65	53	55	60	61	160	90	120
	2	120	170	1080	120	140	67	75	59	78	0	0	0
	3	120	78	74	59	110	69	67	58	120	59	50	74
	4	1800	50	430	2500	200	114	60	135	125	100	92	95
	5	545	140	224	118	810	62	60	42	270	100	80	97
COD	1	480	690	410	640	1000	990	800	1000	1000	1300	490	850
	2	150	1200	1840	550	990	830	1230	970	780	0	0	0
	3	540	730	540	680	840	720	580	700	570	660	610	570
	4	3300	450	860	3160	720	850	1000	1500	890	1260	1000	910
	5	2500	1200	1100	628	1680	580	590	470	700	370	290	450
Calcium	1	150	130	170	200	150	178	187	133	112	210	170	243
	2	240	253	641	411	392	275	307	132	155	0	0	0
	3	268	155	117	152	89	117	148	141	261	132	20.8	139
	4	433	144	177	286	225	154	110	130	146	125	120	112
	5	1020	128	135	148	270	162	138	263	213	154	127	165
Chloride	1	960	290	440	640	960	760	853	929	679	890	765	685
	2	627	678	719	660	978	800	1160	630	761	0	0	0
	3	505	573	449	632	681	668	515	640	243	584	569	390
	4	454	360	310	665	759	811	1040	1350	886	1220	990	806
	5	834	990	687	620	868	814	720	438	421	419	366	548
Conductivity	1	6900	7900	4500	6000	7600	7100	6590	7310	7620	8690	7700	6330
	2	5520	5640	33200	33200	49300	8420	12100	10600	8740	0	0	0
	3	4150	5650	4090	5470	5760	5370	5070	5640	3160	5020	5310	4110
	4	4710	3630	3430	6070	6290	7130	8160	11300	7700	10200	8750	6140
	5	7730	8690	6430	5850	7050	6230	5450	3750	3790	3900	3090	4500
DOC	1	1200	190	150	200	380	340	298	320	322	400	280	336
	2	300	287	510	320	600	480	825	645	185	0	0	0
	3	148	126	133	150	277	193	158	152	200	220	160	160
	4	900	120	300	760	280	240	249	410	260	380	300	215
	5	515	350	270	170	600	132	195	120	98	112	89	140
Hardness	1	828	901	762	882	869	914	1010	630	860	1100	920	1040
	2	1070	1140	2110	1530	1580	1300	1560	840	880	0	0	0
	3	1070	730	610	740	660	650	760	740	1050	780	300	610
	4	1680	770	870	1340	1200	1000	930	1050	960	860	970	740
	5	3720	960	980	890	1230	880	740	1020	860	730	520	750
Iron	1	4.1	25	5.1	3.4	4.5	3.8	4.9	4.86	9.6	2.45	2.57	2.27
	2	3.37	3.12	2.67	2.88	4.12	5.17	6.76	5.92	4.36	0	0	0
	3	4.31	3.37	4.49	7.1	4.8	4.0	4.3	3.9	13.6	5.2	0.67	4.5
	4	14.9	4.6	5.3	17.7	5.7	3.5	2.3	6	14.1	18.6	4.4	2.3
	5	2.41	2.4	4.8	3.3	10	2.2	2.1	36.9	3.6	1.9	2.67	3
Magnesium	1	110	140	82	93	120	114	132	71.2	91.1	140	121	106
	2	113	122	125	122	146	149	194	125	120	0	0	0
	3	97.6	84.2	77.5	88	107	86	101	99	98	57	59.4	64
	4	144	100	104	151	155	149	160	180	144	132	164	113
	5	280	154	137	127	130	116	97	88	79	83	18.4	82
Nitrate	1	<0.5	<0.5	0.51	<0.3	<1	<0.3	<0.5	<0.3	<1	<0.3	<0.5	<0.3
	2	<0.5	<0.3	<1	<0.5	<1	<0.5	<1	<1	<1	0	0	0
	3	<0.5	<1	<1	<1	<0.5	<2	<2	<2	0.1	<2	<1	<2
	4	<0.5	<1	<1	<2	<2	<2	<2	<2	<2	<2	<2	<2
	5	<2	<2	<2	<2	<2	<2	<2	<2	<2	<2	<2	<2
Nitrite	1	<1	<0.5	0.45	<0.3	<1	<1	<1	<0.5	<1	<1	<1	<1
	2	<0.5	<0.5	<1	<0.5	<1	<1	<1	<1	<1	0	0	0
	3	<0.5	<1	<1	<0.5	<2	<2	<2	<2	<0.1	<2	<1	<2
	4	<0.5	<1	<1	<2	<2	<2	<2	<2	<2	<2	<2	<2
	5	<2	<2	<2	<2	<2	<2	<2	<2	<2	<2	<2	<2
pH	1	6.9	6.9	7.0	6.9	7.2	7	6.9	7.1	6.9	7.03	6.96	7.17
	2	7.13	6.84	7.09	7.23	6.97	6.95	7.59	7.08	6.99	0	0	0
	3	7.66	7.12	7.42	7.21	6.93	7.08	7.07	7.24	6.88	7.25	8.35	7.05
	4	6.99	7.13	6.7	7.32	7.27	7.33	7.35	7.24	7.08	7.74	7.53	7.29
	5	7.06	7.47	7.47	7.47	7.59	7.38	7.48	7.3	7.27	6.82	7.78	7.52
Phenols	1	16	25	9	10	<1	25	64	12	83	203	150	203
	2	195	178	945	247	228	135	292	203	208	0	0	0
	3	48	58	221	140	310	264	271	177	92	87	170	316
	4	428	185	431	528	204	9	76	94	212	19	193	160
	5	321	271	271	261	554	141	116	12	220	100	89	208
Phosphorus	1	1.2	2.2	0.15	1.4	1.5	1.8	1.5	4	1.8	1.8	2.2	4
	2	1.4	1.8	2.2	3.1	4.4	4.6	4	3	4.0	0	0	0
	3	1.40	1.90	0.80	4.40	4.00	1.80	1.10	1.40	3.00	1.00	0.80	1.20
	4	4	0.9	0.8	1.8	1.3	1.6	1.4	3.1	1.7	3.1	1.8	1.1
	5	1.3	1.7	1.7	2	2.3	2.3	1.7	1.4	0.8	0.8	0.6	1.4
Sodium	1	960	880	440	540	860	833	700	520	582	923	788	631
	2	632	660	609	773	951	769	1360	899	708	0	0	0
	3	416	483	428	453	696	545	524	569	273	380	1020	310
	4	532	425	311	590	662	811	920	1240	979	1151	974	530
	5	518	515	515	550	800	640	550	380	332	230	305	378
Sulphate	1	31	260	130	60	25	9.1	<10	43	120	23	17	29
	2	11	23	<20	8	<20	<10	<20	<20	<20	0	0	0
	3	486	90	102	41	<40	<40	<40	59	427	114	22	<40
	4	458	20	66	90	<40	<40	<40	71	<40	<40	<40	51
	5	111	<40	<40	46	57	<40	<40	117	277	<40	90	111
Total Kjeldahl Nitrogen	1	240	580	170	290	390	300	300	430	320	300	360	520
	2	260	250	220	300	480	400	610	700	408	0	0	0
	3	120	275	300	350	355	280	265	272	110	350	220	320
	4	175	164	157	309	320	368	372	607	335	520	410	270
	5	430	340	340	290	340	350	240	140	120	98	99	200
Total Suspended Solid	1	43	70	27	8	22	16.3	11	17	35	15	11	<1
	2	4	12	44	35	68	12	29	18	14	0	0	0
	3	35	18	20	83	44	30	19	14	190	31	4	25
	4	90	30	60	68	64	18	13	92	32	55	17	9
	5	140	120	120	22	340	24	24	1000	56	41	95	30

During acenogenic phase microorganisms convert insoluble organic compounds to acetic acid, carbon dioxide and hydrogen by acetogenic bacteria. Leachate generated in the phase typically are characterized by high BOD value, high BOD:COD ratio illustrating a high concentration of soluble organic matters that are biodegradable and pH value that is acidic and high ammonia concentration. Such aggressive nature of leachate prompts dissolution of other components in wastes resulting in high concentration of iron, magnesium, zinc and calcium.

On the other hand, microorganisms in methanogenic phase that can remove soluble organic matter are gradually established due to absence of oxygen. The decomposition in this phase thrives to convert soluble organic matter to methane and carbon dioxide thus release as landfill gas. Methanogenic leachate generated is typically characterized by low BOD value and low ratio of BOD:COD with high concentration of ammonical nitrogen and inorganic matters such as iron, sodium, potassium, sulphate and chloride due to active dissolution.

BOD:COD ratio is thus defined as an indicator for degradation of organic matter that differentiate the acetogenic phrase from methanogenic phase in landfill which trickling leachate. Figure 1 depicts the ratio of BOD:COD spread over the 3 years in the landfill studied.

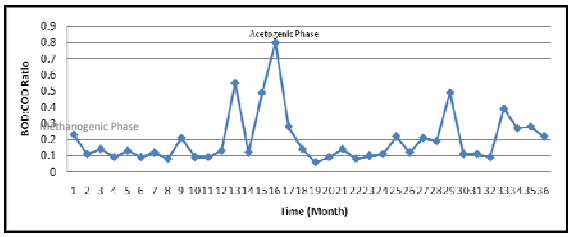


Fig. 1 Ratio of BOD:COD Over 3 Years Duration

The ratio of BOD:COD computed for the landfill exhibited that more than 90% of time leachate in the landfill experienced methanogenic phase.

The results of leachate generated containing low ratio of BOD:COD indicate that the decomposition is stabilized and is biological active and is dynamically in equilibrium.

It is reported that the intensity of decomposition is significantly affected by various external forces such as climatic conditions in terms of environmental temperature and precipitation [9].

Correlation relationship of leachate concentration to temperature is evaluated in terms of physical properties of pH, alkalinity, hardness, conductivity and total suspended solid; dissolved organic matters of BOD, COD and DOC; inorganic matters of sulphate, chloride, ammonia, calcium, magnesium, sodium, iron, nitrate, nitrite and Total Kjeldahl Nitrogen and xenobiotic organic compounds of phenols.

From the data evaluation as depicted in Figures 2 to 5, all physical properties ($r>0.118$) except hardness ($r=0.004$), all inorganic matters ($r>0.100$) except iron ($r=0.071$) and xenobiotic organic compounds of phenols ($r=0.200$) depict there is some degree of significance of correlation relationship of leachate to temperature in the environment. This can be explained by the active decomposition rate in the

waste due to high temperature that facilitates both biological and chemical reaction inside the mass.

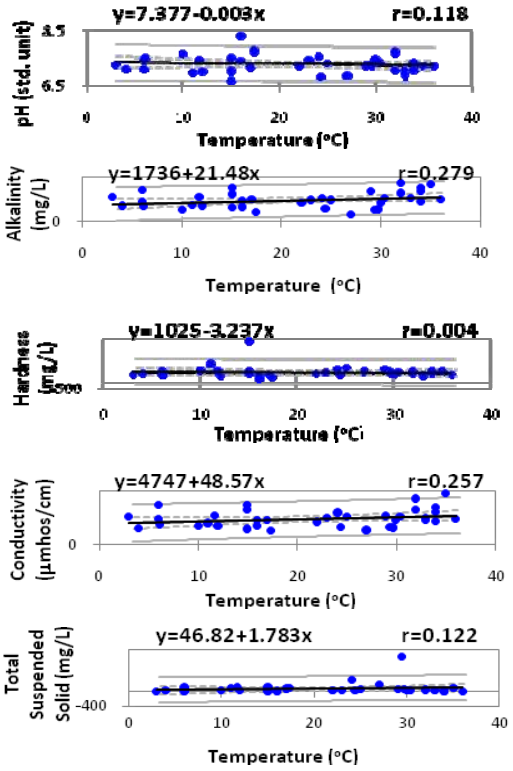


Fig. 2 Physical Properties of Leachate Versus Temperature

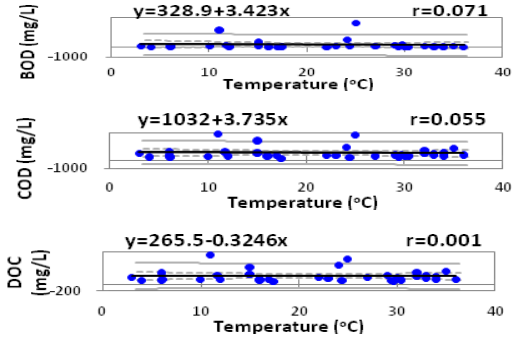
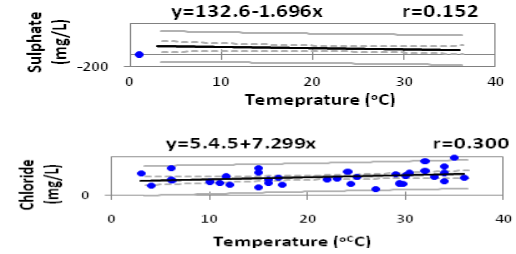


Fig. 3 Dissolved Organic Matters of Leachate Versus Temperature



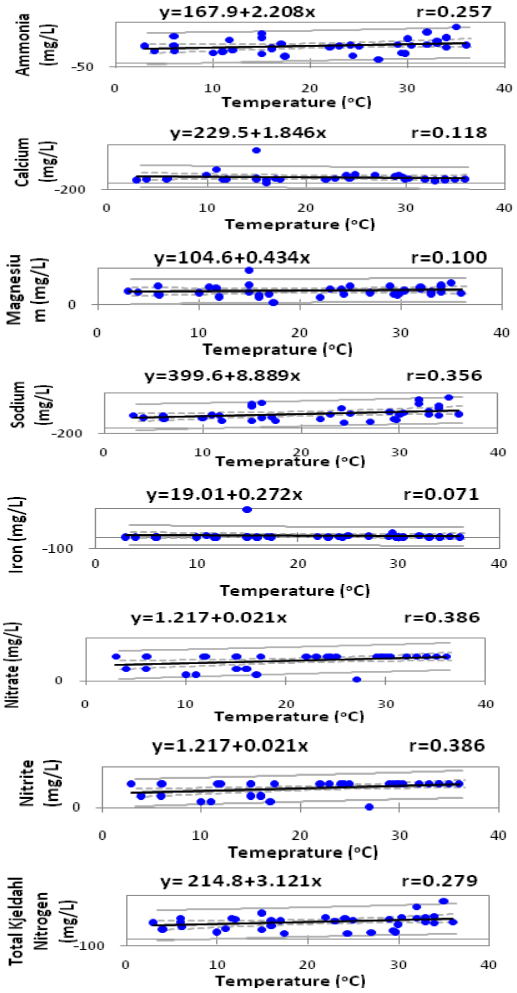


Fig. 4 Inorganic Matters of Leachate Versus Temperature

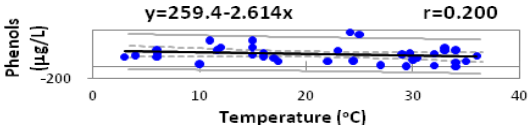


Fig. 5 Xenobiotic Organic Compounds of Leachate Versus Temperature

Figures 6 to 9 depict the correlation relationship of leachate concentration to precipitation. Results illustrate that all physical properties (r>0.114) except pH (r=0.077); all dissolved organic matters (r>0.257); all inorganic matters (r>0.118) except ammonia (r=0.077) and xenobiotic organic compound of phenol (r=0.339) are relatively quite correlated to precipitation as excessive precipitation is likely to slow down decomposition rate in the waste environment which leachate is percolated through.

BOD:COD ratio is also evaluated to established the correlation relationship to leachate concentration obtained from the landfill. Figures 10 to 13 depict the correlation relationship of leachate concentration to BOD:COD ratio.

Results reveals that all physical properties (r>0.146) except total suspended solid (r=0.045); dissolved organic matters (r>0.633); inorganic matter (r>0.118) except iron (r=0.063) and xenobiotic organic compound (r=0.688) are correlated significantly to BOD:COD ratio computed for the leachate concentration obtained for the landfill in this study.

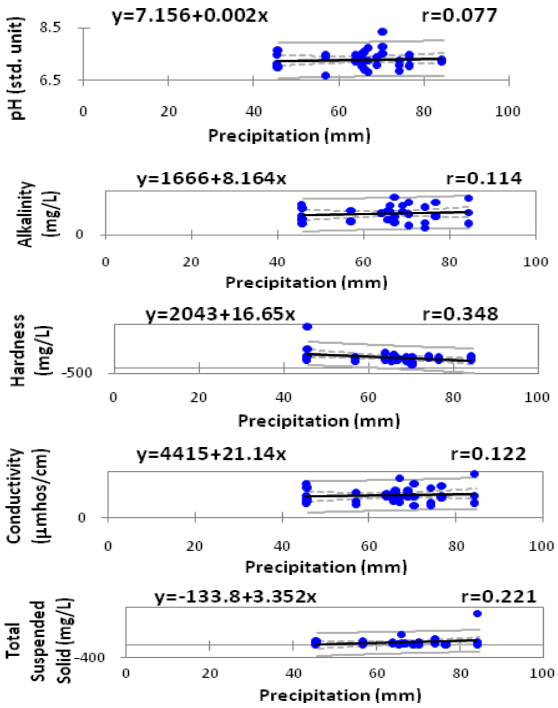


Fig. 6 Physical Properties of Leachate Versus Precipitation

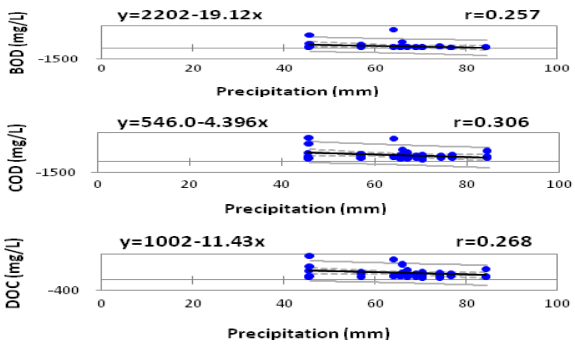
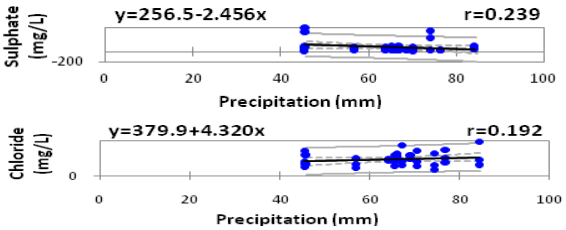


Fig. 7 Dissolved Organic Matters of Leachate Versus Precipitation



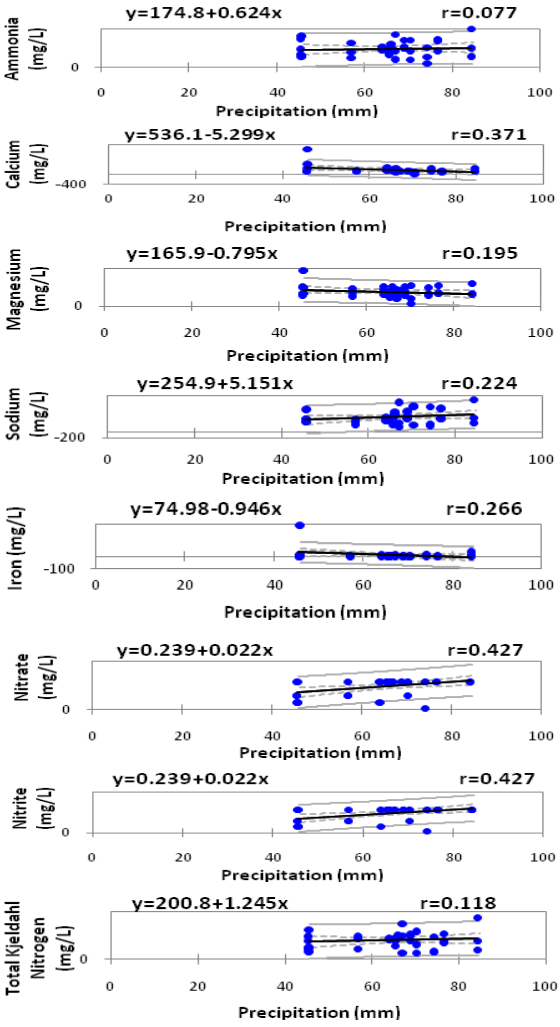


Fig. 8 Inorganic Matters of Leachate Versus Precipitation

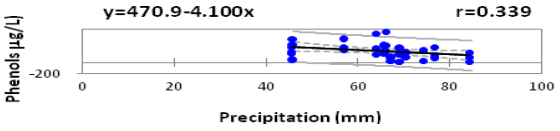
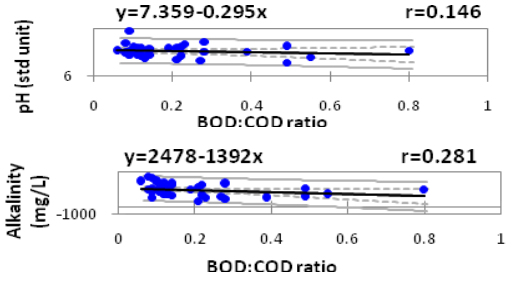


Fig. 9 Xenobiotic Organic Compounds of Leachate Versus Precipitation



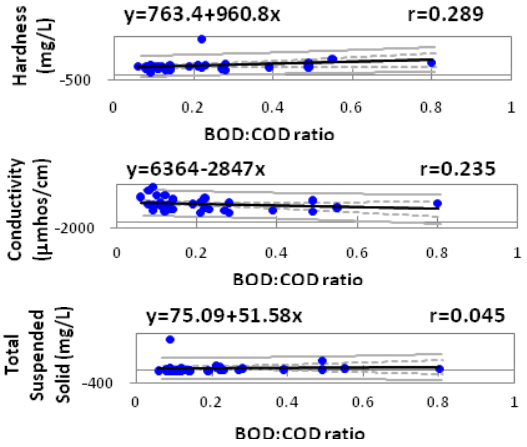


Fig. 10 Physical Properties of Leachate Versus BOD:COD Ratio

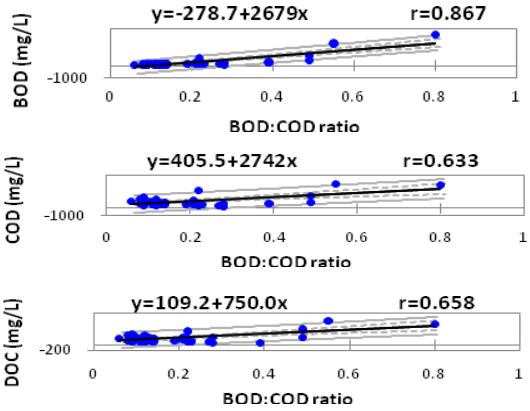
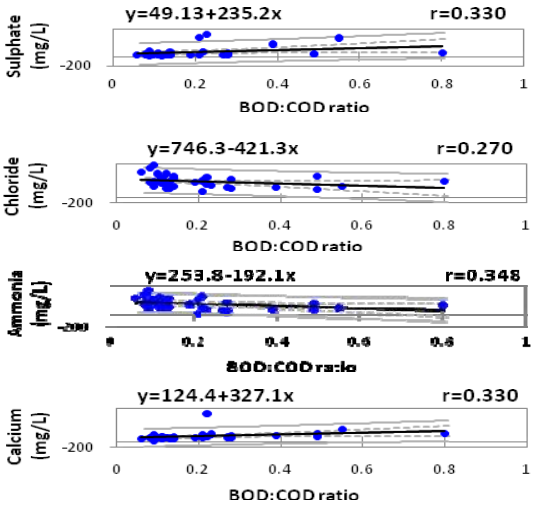


Fig. 11 Dissolved Organic Matters of Leachate Versus BOD:COD Ratio



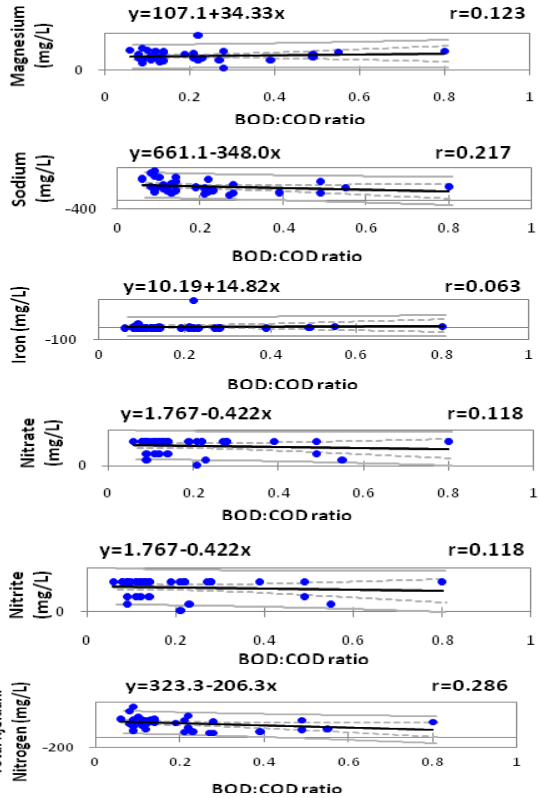


Fig. 12 Inorganic Matters of Leachate Versus BOD:COD Ratio

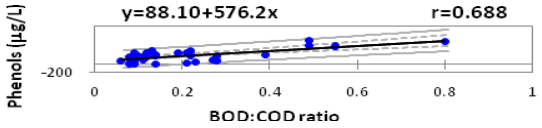


Fig. 13 Xenobiotic Organic Compounds of Leachate Versus BOD:COD Ratio

IV. Conclusion

The study concluded that in an actively decomposing waste landfill, leachate generated can be characterized as acetogenic and methanogenic and BOD:COD ratio of leachate is a good indicator to illustrate the degree of stabilization in landfill. The BOD:COD ratio of leachate computed indicate if sufficient biological and chemical decomposition as well as biodegradion are carried out under changing ambience conditions in the landfill body. It is referred that decreasing BOD:COD ratio is taken as an indicator of degradion of organic substrate due to decomposition.

Acetogenic leachates are typically characterized by its high BOD value and high BOD:COD ratio due to rapid hydrolysis of insoluble organic matters that make it readily degradable. On the other hands, methanogenic leachates are characterized by its relatively low BOD values and low ratios of BOD:COD due to the active dissolution of soluble organic matters present as well as inorganic matter, sulphate, chloride and calcium.

The study also reveals that the waste decomposition in landfill is influenced by climatic condition such as temperature and precipitation based on correlation relationship established. The intensity of decomposition is observed to be significantly affected by amount of precipitation and the temperature inside the landfill mass. Rises in temperature accelerate decomposition while precipitations slow down decomposition to anaerobic condition.

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REFERENCES

[1] Barlaz, M.A, Schaefer, D.M., and Ham, R.K., Bacterial Population Development and chemical Characteristics of Refuse Decomposition in a Simulated Sanitary landfill, Appl. Environ. Microbiol., 55, 55, 1989a.

[2] Assumuth, T.W. and Strandberg, T., Ground-water contamination at Finnish landfills. Water, Air Soil Pollut., 69, 179, 1993.

[3] Kjeldsen, P. and Christophersen, M., Composition of leachate from old landfills in Denmark, Waste Manag. Res., 19, 24-256, 2001.

[4] Barlaz, M.A., Ham, R.K. and Shaefer, D.M., Methane Production from Municipal Refuse : A Review of Enhancement Techniques and Microbial Dynamics, CRC Crit. Rev. Environ. Contr., 19, 6, 557, 1990

[5] Kjeldsen, P., Barlaz, M., Rooker, A., Baun, A., Ledin, A. and Christensen, T., Present and Long-Term Composition of MSW Landfill Leachate : A Review. Critical Reviews in Environmental Science and Technology, 32, No. 4, 2002.

[6] Chian, E.S.K., and DeWalle, F.B., Characterization of Soluble organic matter in leachate. Environ Sci. Techol., 11, 158, 1977.

[7] Ehrig, H.-J., quality and quantity of sanitary landfill leachate, Waste Manag. Res., 1, 53, 1983.

[8] Martensson, A.M., aulin, C, Wahlberg, O., and Argen, S., Effect of humic substances on the mobility of toxic metals in a mature landfill, Waste Manag. Res., 17, 296, 1999.

[9] Ritzkowski M., Heyer, K.-U, Stegmann, R. (2003) : Insitu aeration of Old landfills : Carbon Balances, temperatures and settlements. Proceedings of Sardia, 2003-Ninety International Waste Management and landfill Symposium, Cagliari, Italy, 06-10.10.2003.