# Clausenlanins A and B, Two Leucine-Rich Cyclic Nonapeptides from Clausena lansium 

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#### Abstract

Two new cyclic nonapeptides, named clausenlanins A (1) and B(2), were isolated from the roots and rhizomes of Clausena lansium. Their structures were elucidated as cyclo-(Gly ${ }^{1}-\mathrm{L}-\mathrm{Leu}^{2}-\mathrm{L}-\mathrm{Ile}^{3}-\mathrm{L}-\mathrm{Leu}^{4}-\mathrm{L}-\mathrm{Leu}^{5}-\mathrm{L}-\mathrm{Leu}^{6}-\mathrm{L}-\mathrm{Leu}^{7}-\mathrm{L}-\mathrm{Leu}^{8}$ -L-Leu ${ }^{9}$ ) (1) and cyclo-(Gly $\left.{ }^{1}-\mathrm{L}-\mathrm{Leu}^{2}-\mathrm{L}-\mathrm{Val}^{3}-\mathrm{L}-\mathrm{Leu}^{4}-\mathrm{L}-\mathrm{Leu}^{5}-\mathrm{L}-\mathrm{Leu}^{6}-\mathrm{L}-\mathrm{Leu}^{7}-\mathrm{L}-\mathrm{Leu}^{8}-\mathrm{L}-\mathrm{Leu}^{9}\right)(\mathbf{2})$ respectively on the basis of extensive spectroscopic analysis, particularly 2D NMR spectra taken at the temperature of 338 or 303 K and MS.


## Graphical Abstract



Keywords Clausena lansium • Cyclopeptides • Clausenlanin A • Clausenlanin B

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1


2

Fig. 1 Structures of compounds $\mathbf{1}$ and $\mathbf{2}$ from C. lansium

## 1 Introduction

About 30 species of Clausena (Rutaceae) are widely distributed in the world, and 10 of them exist in China. Clausena lansium (Lour.) Skeels is a fruit tree and distributes widely in south of China [1]. Its leaves and roots have been used as a folk herb for the treatment of cough, asthma, dermatological disease, viral hepatitis, and gastrointestinal disease; and its seeds for treating acute and chronic gastrointestinal inflammation, and ulcer [2]. Caryophyllaceae-type cyclopeptides (CPs), carbazole alkaloids, coumarins, amides, and terpenoids have been isolated from C. lansium [3-8]. Among them, CPs are formed with the peptide bonds of protein or non-proten $\alpha$ amino acid residues, which are homomonocyclopeptides with mainly five to twelve $\alpha$-amino acid residues [9]. During this work, two new cyclic nonapeptides, named clausenlanins A (1) and B (2) (Fig. 1), were isolated from the roots and rhizomes of $C$. lansium. Because the ${ }^{1} \mathrm{H}$ NMR signals are weak and severely overlapped taken at room temperature, variable temperature NMR experiments were performed [10]. In this paper, their separation and structure elucidation are described.

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## 2 Results and Discussion

Clausenlanin $\mathrm{A}(\mathbf{1})$ was obtained as an amorphous solid. Its molecular formula was shown as $\mathrm{C}_{50} \mathrm{H}_{91} \mathrm{~N}_{9} \mathrm{O}_{9}$ by its negative HRESIMS ([M-H] ${ }^{-}$, 960.6876, calcd 960.6867), indicating the $10^{\circ}$ of unsaturation. The IR spectrum exhibited the absorption bands at 3429 and $1661 \mathrm{~cm}^{-1}$ ascribable to NH and CO groups. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1}$ in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}$ (Table 1) displayed the characteristic signals of typical CPs.

The ${ }^{1} \mathrm{H}$ NMR signals of the amino acid residues of $\mathbf{1}$, especially the signals of NH and $\alpha-H$, were severely overlapped taken at room temperature. The significant improvement of the ${ }^{1} \mathrm{H}$ NMR signals was observed by increasing the temperatures from 243 to 338 K. Finally a well-resolved ${ }^{1} \mathrm{H}$ NMR spectrum with sharp proton signals (Fig. 2a) was obtained at 338 K in pyridine $-\mathrm{d}_{5}$. Then the assignment of the ${ }^{1} \mathrm{H}$ NMR signals of the amino acid residues was obtained by analyzing the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectrum, particularly amide proton NH and $\alpha-\mathrm{H}$ signals. The corresponding ${ }^{13} \mathrm{C}$ NMR assignments were determined on the basis of the HSQC and HMBC experiments, particularly $\alpha-C$ signals (Table 1). The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{1}$ showed the presence of nine $\mathrm{NH}\left(\delta_{\mathrm{H}} 8.91,8.75,8.73,8.72,8.52,8.46\right.$, $8.36,8.23,8.10)$ and ten $\alpha-\mathrm{H}\left(\delta_{\mathrm{H}} 4.81,4.77,4.68,4.59,4.58\right.$, $4.57,4.51,4.46,4.46,3.85$ ), respectively. The ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum of $\mathbf{1}$ displayed nine carbonyl CO signals at $\delta_{\mathrm{C}}$ 175.6, 174.7, 174.5, 174.4, 174.1, 173.9, 173.9, 173.5, 171.1, eight $\alpha-\mathrm{CH}$ signals at $\delta_{\mathrm{C}} 61.2,54.9,54.8,54.7,54.5$, $54.0,53.7,53.6$, one $\alpha-\mathrm{CH}_{2}$ signal at $\delta_{\mathrm{C}}$ 44.8. These data indicated that $\mathbf{1}$ might be a cyclic nonapeptide. Analysis of the HSQC, HMBC and COSY spectra revealed that $\mathbf{1}$ consisted of one glycine ( $\delta_{\mathrm{H}} 4.51$ and $3.85\left(\alpha-\mathrm{H}_{2}\right), 8.75(\mathrm{NH}) ; \delta_{\mathrm{C}}$ $\left.171.1(\mathrm{CO}), 44.8\left(\alpha-\mathrm{CH}_{2}\right)\right)$, and one isoleucine $\left(\delta_{\mathrm{H}} 4.46(\alpha-\right.$

Table 1 NMR data of compounds 1 and 2

| 1* |  |  |  | $2^{\text {\# }}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Residue | Position | $\delta_{\text {H }}$ | $\delta_{\text {C }}$ | Residue | Position | $\delta_{\text {H }}$ | $\delta_{\text {C }}$ |
| Gly-1 | $\alpha$ | 4.51 (d, 5.1) | 44.8 (t) | Gly-1 | $\alpha$ | 4.57 (overlap) | 44.6 (t) |
|  |  | 3.85 (dd, 5.1, 16.3) |  |  |  | $3.91 \text { (dd, 5.2, 16.4) }$ |  |
|  | NH | 8.75 (overlap) |  |  | NH | 9.03 (br.s) |  |
|  | $\mathrm{C}=\mathrm{O}$ |  | 171.1 (s) |  | $\mathrm{C}=\mathrm{O}$ |  | 171.2 (s) |
| Leu-2 | $\alpha$ | 4.77 (d, 7.0) | 53.7 (d) | Leu-2 | $\alpha$ | 4.87 (br.s) | 53.3 (d) |
|  | $\beta$ | 1.93-2.21 (overlap) | 40.6 (t) |  | $\beta$ | 1.93 (overlap) | 40.4 (t) ${ }^{\text {a }}$ |
|  |  |  |  |  |  | 2.07 (overlap) |  |
|  | $\gamma$ | 1.87-1.99 (overlap) | 25.8 (d) ${ }^{\text {a }}$ |  | $\gamma$ | 1.89-2.00 (overlap) | 25.8 (d) ${ }^{\text {b }}$ |
|  | $\delta$ | 0.93-1.06 (overlap) | 23.7 (q) ${ }^{\text {b }}$ |  | $\delta$ | 0.86-1.03 (overlap) | 23.4 (q) ${ }^{\text {c }}$ |
|  |  | 0.91-1.02 (overlap) | $22.3(\mathrm{q})^{\mathrm{c}}$ |  |  | 0.86-1.03 (overlap) | 22.0 (q) ${ }^{\text {d }}$ |
|  | NH | 8.23 (d, 7.0) |  |  | NH | 8.42 (br.s) |  |
|  | $\mathrm{C}=\mathrm{O}$ |  | 175.6 (s) |  | $\mathrm{C}=\mathrm{O}$ |  | 174.7 (s) ${ }^{\text {e }}$ |
| Ile-3 | $\alpha$ | 4.46 (overlap) | 61.2 (d) | Val-3 | $\alpha$ | 4.42 (br.s) | 62.6 (d) |
|  | $\beta$ | 2.30 (overlap) | 36.7 (d) |  | $\beta$ | 2.57 (m) | 30.6 (d) |
|  | $\gamma$ | 1.39 (m) | 26.7 (t) |  | $\gamma$ | 1.18 (d, 4.4) | 20.3 (q) |
|  |  | 1.87 (overlap) |  |  |  | $1.19 \text { (d, 4.4) }$ | $20.2 \text { (q) }$ |
|  | Me $\gamma$ | 1.18 (d, 6.7) | 16.6 (q) |  |  |  |  |
|  | Me $\delta$ | 0.92 (overlap) | 11.5 (q) |  |  |  |  |
|  | NH | 8.72 (overlap) |  |  | NH | 9.22 (overlap) |  |
|  | $\mathrm{C}=\mathrm{O}$ |  | 173.9 (s) |  | $\mathrm{C}=\mathrm{O}$ |  | 173.9 (s) |
| Leu-4 | $\alpha$ | 4.58 (overlap) | 54.5 (d) | Leu-4 | $\alpha$ | 4.65 (overlap) | 54.6 (d) |
|  | $\beta$ | 1.93-2.21 (overlap) | 40.0 (t) ${ }^{\text {d }}$ |  | $\beta$ | 2.07 (overlap) | 39.9 (t) ${ }^{\text {a }}$ |
|  | $\gamma$ | 1.87-1.99 (overlap) | 25.9 (d) ${ }^{\text {a }}$ |  | $\gamma$ | 1.89-2.00 (overlap) | 25.7 (d) ${ }^{\text {b }}$ |
|  | $\delta$ | 0.93-1.06 (overlap) | $\begin{aligned} & 23.6(\mathrm{q})^{\mathrm{b}} \\ & 22.3(\mathrm{q})^{\mathrm{c}} \end{aligned}$ |  | $\delta$ | 0.86-1.03 (overlap) | 23.6 (q) ${ }^{\text {c }}$ |
|  |  | 0.91-1.02 (overlap) |  |  |  | 0.86-1.03 (overlap) | 22.0 (q) ${ }^{\text {d }}$ |
|  | NH | 8.73 (overlap) |  |  | NH | 9.23 (overlap) |  |
|  | $\mathrm{C}=\mathrm{O}$ |  | 173.5 (s) |  | $\mathrm{C}=\mathrm{O}$ |  | 173.8 (s) ${ }^{\text {e }}$ |
| Leu-5 | $\alpha$ | 4.81 (d, 7.2) | 54.0 (d) | Leu-5 | $\alpha$ | 4.93 (br.s) | 53.7 (d) |
|  | $\beta$ | 1.93-2.21 (overlap) | 40.6 (t) |  | $\beta$ | 2.17 (overlap) | 40.4 (t) ${ }^{\text {a }}$ |
|  |  |  |  |  |  | 2.31 (overlap) |  |
|  | $\gamma$ | 1.87-1.99 (overlap) | 25.5 (d) ${ }^{\text {a }}$ |  | $\gamma$ | 1.89-2.00 (overlap) | 25.6 (d) ${ }^{\text {b }}$ |
|  | $\delta$ | 0.93-1.06 (overlap) | 23.8 (q) ${ }^{\text {b }}$ |  | $\delta$ | 0.86-1.03 (overlap) | 23.8 (q) ${ }^{\text {c }}$ |
|  |  | 0.91-1.02 (overlap) | 22.6 (q) ${ }^{\text {c }}$ |  |  | 0.86-1.03 (overlap) |  |
|  | NH | 8.10 (d, 7.2) |  |  | NH | 8.35 (br.s) | 22.4 (q) ${ }^{\text {d }}$ |
|  | $\mathrm{C}=\mathrm{O}$ |  | 174.4 (s) |  | $\mathrm{C}=\mathrm{O}$ |  | 174.7 (s) ${ }^{\text {e }}$ |
| Leu-6 | $\alpha$ | 4.59 (overlap) | 54.8 (d) | Leu-6 | $\alpha$ | 4.66 (overlap) | 54.5 (d) |
|  | $\beta$ | 1.93-2.21 (overlap) | $40.3(\mathrm{t})^{\mathrm{d}}$ |  | $\beta$ | 2.07 (overlap) | $39.8(t)^{\mathrm{a}}$ |
|  |  |  |  |  |  | 2.28 (overlap) |  |
|  | $\gamma$ | 1.87-1.99 (overlap) | 25.8 (d) ${ }^{\text {a }}$ |  | $\gamma$ | 1.89-2.00 (overlap) | 25.2 (d) ${ }^{\text {b }}$ |
|  | $\delta$ | 0.93-1.06 (overlap) | 23.7 (q) ${ }^{\text {b }}$ |  | $\delta$ | 0.86-1.03 (overlap) | 23.4 (q) ${ }^{\text {c }}$ |
|  |  | 0.91-1.02 (overlap) | $22.3(\mathrm{q})^{\mathrm{c}}$ |  |  | 0.86-1.03 (overlap) | $22.0(\mathrm{q})^{\mathrm{d}}$ |
|  | NH | 8.52 (d, 6.3) |  |  | NH | 8.85 (br.s) |  |
|  | $\mathrm{C}=\mathrm{O}$ |  | 174.1 (s) |  | $\mathrm{C}=\mathrm{O}$ |  | $174.1(s)^{\text {e }}$ |
| Leu-7 | $\alpha$$\beta$ | 4.46 (overlap) | 54.9 (d) | Leu-7 | $\alpha$ | 4.50 (br.s) | 54.7 (d) |
|  |  | 1.93-2.21 (overlap) | $40.1(\mathrm{t})$ |  | $\beta$ | 2.28 (overlap) | 39.7 (t) ${ }^{\text {a }}$ |
|  |  |  |  |  |  | 2.36 (br.s) |  |
|  | $\gamma$ | 1.87-1.99 (overlap) | 26.0 (d) ${ }^{\text {a }}$ |  | $\gamma$ | 1.89-2.00 (overlap) | 25.7 (d) ${ }^{\text {b }}$ |

Table 1 continued


J values given in Hz in parentheses
a, b, c, d, e Chemical shifts can be exchanged with each other in the column

* In $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}, 338 \mathrm{~K}, 400 \mathrm{MHz}$ for ${ }^{1} \mathrm{H}$ and 100 MHz for ${ }^{13} \mathrm{C}$
\# In $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}, 303 \mathrm{~K}, 800 \mathrm{MHz}$ for ${ }^{1} \mathrm{H}$ and 200 MHz for ${ }^{13} \mathrm{C}$


Fig. $2{ }^{1} \mathrm{H}$ NMR of compounds 1 (a) and 2 (b) at different temperatures
$\mathrm{CH}), 2.30(\beta-\mathrm{CH}), 1.39$ and $1.87\left(\gamma-\mathrm{CH}_{2}\right), 1.18\left(\gamma-\mathrm{CH}_{3}\right)$, $0.92\left(\delta-\mathrm{CH}_{3}\right), 8.72(\mathrm{NH}) ; \delta_{\mathrm{C}} 173.9(\mathrm{CO}), 61.2(\alpha-\mathrm{CH}), 36.7$ $\left.(\beta-\mathrm{CH}), 26.7\left(\gamma-\mathrm{CH}_{2}\right), 16.6\left(\gamma-\mathrm{CH}_{3}\right), 11.5\left(\delta-\mathrm{CH}_{3}\right)\right)$. The remaining signals mentioned-above of seven NH and seven $\alpha-\mathrm{H}$ signals, seven CO and seven $\alpha-\mathrm{C}$ signals, and other
signals including seven methylenes at $\delta_{\mathrm{C}} 40.0-40.6$, seven methines at $\delta_{\mathrm{C}} 25.5-26.0$, two kinds of fourteen methyls at $\delta_{\mathrm{C}} 23.5-23.8$ and at $\delta_{\mathrm{C}} 22.1-22.6$, indicated that $\mathbf{1}$ contained other seven leucines. Therefore $\mathbf{1}$ consisted of seven leucines, one glycine, and one isoleucine (Table 1; Fig. 3).


1


2

$$
\rightarrow \mathrm{HMBC}(\mathrm{H} \rightarrow \mathrm{C}) \quad-{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H} \text { COSY } \leftrightarrow \text { ROESY/NOESY }
$$

Fig. 3 Key $\mathrm{HMBC},{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY, and ROESY/NOESY correlations of $\mathbf{1}$ and $\mathbf{2}$

The sequence of the nine amino acid residues in $\mathbf{1}$ was determined by analyzing the ROESY correlations between the $\alpha$-H of one amino acid residue and the amide proton NH of the next amino acid residue (Fig. 3). The ROESY correlations of Gly ${ }^{1}-\alpha \mathrm{H} /$ Leu $^{2}-\mathrm{NH}$, $\mathrm{Leu}^{2}-\alpha \mathrm{H} / \mathrm{Ile}^{3}-\mathrm{NH}$, $\mathrm{Ile}^{3}-\alpha \mathrm{H} /$ Leu $^{4}-\mathrm{NH}$, $\mathrm{Leu}^{4}-\alpha \mathrm{H} / \mathrm{Leu}^{5}-\mathrm{NH}, \mathrm{Leu}^{5}-\alpha \mathrm{H} / \mathrm{Leu}^{6}-\mathrm{NH}, \mathrm{Leu}^{6}-\alpha \mathrm{H} / \mathrm{Leu}^{7}-\mathrm{NH}$, Leu $^{7}-\alpha \mathrm{H} /$ Lue $^{8}-\mathrm{NH}, \quad \mathrm{Leu}^{8}-\alpha \mathrm{H} / \mathrm{Leu}^{9}-\mathrm{NH}, \quad \mathrm{Leu}^{9}-\alpha \mathrm{H} / \mathrm{Gly}^{1}-\mathrm{NH}$ indicated that the structure of $\mathbf{1}$ is cyclo-(Gly ${ }^{1}$-Leu ${ }^{2}-\mathrm{Ile}^{3}-$ $\left.\mathrm{Leu}^{4}-\mathrm{Leu}^{5}-\mathrm{Leu}^{6}-\mathrm{Leu}^{7}-\mathrm{Leu}^{8}-\mathrm{Leu}^{9}\right)$. This sequence of $\mathbf{1}$ was confirmed by the fragment ion peaks at $962.96[\mathrm{M}+\mathrm{H}]^{+}$, $849.83[\mathrm{M}+\mathrm{H}-113]^{+}, 736.72[\mathrm{M}+\mathrm{H}-2 * 113]^{+}, 623.64$ $\left[\mathrm{M}+\mathrm{H}-3^{*} 113\right]^{+}, \quad 510.52 \quad[\mathrm{M}+\mathrm{H}-4 * 113]^{+}, \quad 453.49$ $\left[\mathrm{M}+\mathrm{H}-4^{*} 113-57\right]^{+}, \quad 340.43 \quad[\mathrm{M}+\mathrm{H}-4 * 113-57-113]^{+}$, $227.29\left[\mathrm{M}+\mathrm{H}-4^{*} 113-57-2^{*} 113\right]^{+}$in the positive ESIMSMS.

The absolute configuration of the amino acids of $\mathbf{1}$ was determined using the advanced Marfey's method and LCMS analysis [11, 12]. The results indicated that the absolute configurations of the amino acid residues (Leu and Ile) in 1 were the l-configuration (Table S1; Fig. 3). Therefore the structure of $\mathbf{1}$ is determined as cyclo-(Gly ${ }^{1}$-L-Leu ${ }^{2}$-L-


Clausenlanin B (2) was obtained as an amorphous solid. Its molecular formula was shown as $\mathrm{C}_{49} \mathrm{H}_{89} \mathrm{~N}_{9} \mathrm{O}_{9}$ by its negative HRESIMS ([M-H $]^{-}, 946.6723$, calcd 946.6710), indicating the $10^{\circ}$ of unsaturation. The IR spectrum exhibited the absorption bands at 3430 and $1661 \mathrm{~cm}^{-1}$ ascribable to NH and CO groups. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 2 in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}$ (Table 1) displayed the characteristic signals of typical CPs.

The ${ }^{1} \mathrm{H}$ NMR signals of the amino acid residues of $\mathbf{2}$, especially the signals of NH and $\alpha-\mathrm{H}$, were severely
overlapped taken at room temperature. The significant improvement of the ${ }^{1} \mathrm{H}$ NMR signals was observed by increasing the temperatures from 243 to 338 K. Finally a well-resolved ${ }^{1} \mathrm{H}$ NMR spectrum with sharp proton signals (Fig. 2b) was obtained at 303 K in pyridine $-\mathrm{d}_{5}$. After compared all data of $\mathbf{2}$ with those of $\mathbf{1}$, the results indicated that $\mathbf{2}$ and $\mathbf{1}$ are very similar, and $\mathbf{2}$ might also be a cyclic nonapeptide too. The only difference is to be replaced the isoleucine residue in $\mathbf{1}$ by valine residue in $\mathbf{2}$. The assignment of the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR signals of the valine residue was obtained by analyzing the HSQC, HMBC and COSY spectra, i.e. $\delta_{\mathrm{H}} 4.42(\alpha-\mathrm{CH}), 2.57(\beta-\mathrm{CH}), 1.18$ and $1.19\left(2^{*} \gamma-\right.$ $\left.\mathrm{CH}_{3}\right), 9.22(\mathrm{NH}) ; \delta_{\mathrm{C}} 173.9(\mathrm{CO}), 62.6(\alpha-\mathrm{CH}), 30.6(\beta-\mathrm{CH})$, 20.3 and $20.2\left(2 * \gamma-\mathrm{CH}_{3}\right)$. Therefore 2 consisted of seven leucines, one glycine, and one valine (Table 1; Fig. 3).

The sequence of the nine amino acid residues in 2 was determined by analyzing NOESY correlations between the $\alpha-$ H of one amino acid residue and the amide proton NH of the next amino acid residue (Fig. 3). The NOESY correlations of $\mathrm{Gly}^{1}-\alpha \mathrm{H} / \mathrm{Leu}^{2}-\mathrm{NH}, \mathrm{Leu}^{2}-\alpha \mathrm{H} / \mathrm{Val}^{3}-\mathrm{NH}, \mathrm{Val}^{3}-\alpha \mathrm{H} / \mathrm{Leu}^{4}-\mathrm{NH}$, Leu $^{4}-\alpha H /$ Leu $^{5}-\mathrm{NH}$, Leu $^{5}-\alpha H /$ Leu $^{6}-\mathrm{NH}$, Leu $^{6}-\alpha H /$ Leu $^{7}-$ NH, Leu $^{7}-\alpha \mathrm{H} /$ Lue $^{8}-\mathrm{NH}, \mathrm{Leu}^{8}-\alpha \mathrm{H} / \mathrm{Leu}^{9}-\mathrm{NH}, \mathrm{Leu}^{9}-\alpha \mathrm{H} / \mathrm{Gly}^{1}-\mathrm{NH}$ indicated that the structure of $\mathbf{2}$ is cyclo-(Gly ${ }^{1}$-Leu ${ }^{2}$ - $\mathrm{Val}^{3}$ -$\left.\mathrm{Leu}^{4}-\mathrm{Leu}^{5}-\mathrm{Leu}^{6}-\mathrm{Leu}^{7}-\mathrm{Leu}^{8}-\mathrm{Leu}^{9}\right)$. This sequence of $\mathbf{2}$ was confirmed by the fragment ion peaks at $948.86[\mathrm{M}+\mathrm{H}]^{+}$, $835.70[\mathrm{M}+\mathrm{H}-113]^{+}, 722.56[\mathrm{M}+\mathrm{H}-2 * 113]^{+}, 609.54$ $[\mathrm{M}+\mathrm{H}-3 * 113]^{+}, \quad 496.39 \quad[\mathrm{M}+\mathrm{H}-4 * 113]^{+}, \quad 383.41$ $\left[\mathrm{M}+\mathrm{H}-5^{*} 113\right]^{+}, 270.24\left[\mathrm{M}+\mathrm{H}-6^{*} 113\right]^{+}$in the positive ESIMSMS.

The absolute configuration of 2 was determined using the advanced Marfey's method and LC-MS analysis too $[11,12]$. The results indicated that the absolute
configurations of the amino acid residues (Leu and Val) in 2 were the l-configuration (Table S1; Fig. 3). Therefore the structure of 2 is determined as cyclo-(Gly ${ }^{1}-\mathrm{L}-\mathrm{Leu}^{2}-\mathrm{L}-\mathrm{Val}^{3}$ -L-Leu $^{4}$-L-Leu ${ }^{5}$-L-Leu ${ }^{6}$ - - Leu $^{7}$-L-Leu ${ }^{8}$-L-Leu ${ }^{9}$ ).

## 3 Experimental

### 3.1 General Experimental Procedures

Optical rotations were obtained on a Jasco P-1020 polarimeter. IR spectra were measured on a Tensor 27 spectrometer with KBr pellets. UV spectra were obtained using a Shimadzu UV-2401PC spectrophotometer. 1D and 2D NMR spectra were performed on a Bruker AM-400 $\left({ }^{1} \mathrm{H}\right.$ : $\left.400 \mathrm{MHz},{ }^{13} \mathrm{C}: 100 \mathrm{MHz}\right)$ or Bruker AVANCE III-800 $\left({ }^{1} \mathrm{H}\right.$ : $800 \mathrm{MHz},{ }^{13} \mathrm{C}: 200 \mathrm{MHz}$ ). Chemical shifts were expressed in ppm with reference to the solvent signals. Mass spectra were measured on a Waters XEVO-TQD spectrometer or an Agilent 1290 UPLC/6540 Q-TOF spectrometer. Analytical or semi-preparative HPLC was performed on Agilent 1100 apparatus equipped with a UV detector and a SunFire OBD (Waters, $1.9 \times 25 \mathrm{~cm}, 5 \mu \mathrm{~m}$ ). Column chromatography was performed with silica gel (100-200 mesh and 200-300 mesh, Qingdao Yu-Min-Yuan Chemical Co. Ltd., Qingdao, P.R. China), MCI gel (CHP-20P, 70-150 $\mu \mathrm{m}$, Mitsubishi Chemical Co., Japan) or Lichroprep RP-18 gel (40-63 mm, Merck, Darmstadt, Germany). Fractions were monitored by TLC (GF254, Qingdao Yu-Min-Yuan Chemical Co. Ltd., Qingdao, P.R. China), and the orange spots were visualized on the plate by spraying with $2 \%$ ninhydrin reagent, after hydrolyzed in an incubator ( $110^{\circ} \mathrm{C}$ ) for 30 min by concentrated HCl [13].

### 3.2 Plant Material

The roots and rhizomes of C. lansium were collected in Hekou, Yunnan Province, P. R. China, in September 2010, and identified by Prof. Yu-Min Shui, Kunming Institute of Botany, Chinese Academy of Sciences. A voucher specimen (No. 0599043) was deposited at the Herbarium of Kunming Institute of Botany, Chinese Academy of Sciences.

### 3.3 Extraction and Isolation

Air dried, powdered roots and rhizomes of C. lansium $(27 \mathrm{~kg})$ were extracted and refluxed with MeOH for three times each for $4 \mathrm{~h}(\mathrm{MeOH}, 3 * 50 \mathrm{~L})$. The extract was evaporated under reduced pressure to yield a dark brown residue ( 0.9 kg ). The residue was suspended in $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ (7:3, 3 L ) and then partitioned with EtOAc ( $3 * 2 \mathrm{~L}$ ). After removing solvent, the EtOAc-soluble part ( 406 g ) was
fractionated by silica gel (200-300 mesh) column chromatograph (CC) and eluted with $\mathrm{CHCl}_{3} / \mathrm{MeOH}(30: 1-4: 1)$ to afford six fractions (Fr.1-Fr.6), on the basis of TLC detection.

Fr. $6(77 \mathrm{~g})$ was subjected to silica gel $\mathrm{CC}\left(\mathrm{CHCl}_{3} /\right.$ acetone $15: 1-7: 3$ ) to afford Fr.6.1-Fr.6.4. Fr.6.2 (14.3 g) was subjected to silica gel CC (PE/acetone 5:1), MPLC with MCI ( $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ 10:90-60:40), and MPLC with RP-18 ( $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O} 5: 95-70: 30$ ). Then, the fractions was further purified by silica gel $\mathrm{CC}\left(\mathrm{CHCl}_{3} / \mathrm{MeOH} 50: 1\right)$, subsequently to afford $1(308 \mathrm{mg})$ and $2(47 \mathrm{mg})$.

### 3.4 Clausenlanin A (1)

Amorphous powder; $[\alpha]_{\mathrm{D}}^{20.7}-154.6$ (c $\left.0.10, \mathrm{MeOH}\right)$; UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 203.0(4.61) \mathrm{nm} ; \mathrm{CD}(\mathrm{MeOH}) 203$ ( $\Delta \varepsilon-30.8$ ); IR (KBr) $v_{\max } 3429,2960,1661,1528$, $584 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}(400 \mathrm{MHz})$ and ${ }^{13} \mathrm{C}(100 \mathrm{MHz}) \mathrm{NMR}$ data, see Table 1; positive ESIMSMS $m / z 962.96[\mathrm{M}+\mathrm{H}]^{+}$, $849.83[\mathrm{M}+\mathrm{H}-113]^{+}, 736.72[\mathrm{M}+\mathrm{H}-2 * 113]^{+}, 623.64$ $[\mathrm{M}+\mathrm{H}-3 * 113]^{+}, \quad 510.52 \quad[\mathrm{M}+\mathrm{H}-4 * 113]^{+}, \quad 453.49$ $[\mathrm{M}+\mathrm{H}-4 * 113-57]^{+}, 340.43[\mathrm{M}+\mathrm{H}-4 * 113-57-113]^{+}$, $227.29\left[\mathrm{M}+\mathrm{H}-4^{*} 113-57-2^{*} 113\right]^{+}$; negative HRESIMS $\mathrm{m} / \mathrm{z} 960.6876[\mathrm{M}-\mathrm{H}]^{-}$, calcd for $\mathrm{C}_{50} \mathrm{H}_{91} \mathrm{~N}_{9} \mathrm{O}_{9}, 960.6867$.

### 3.5 Clausenlanin B (2)

Amorphous powder; $[\alpha]_{\mathrm{D}}^{20.6}-73.69$ (c 0.10, MeOH); UV $(\mathrm{MeOH}) \lambda_{\text {max }}(\log \varepsilon) 202.8(4.40) \mathrm{nm} ; \mathrm{CD}(\mathrm{MeOH}) 203(\Delta \varepsilon-$ 26.1); IR (KBr) $v_{\max } 3430,2960,1661,1527,584 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ ( 800 MHz ) and ${ }^{13} \mathrm{C}(200 \mathrm{MHz})$ NMR data, see Table 1; positive ESIMSMS $m / z, 948.86[\mathrm{M}+\mathrm{H}]^{+}, 835.70[\mathrm{M}+\mathrm{H}-113]^{+}$, $722.56[\mathrm{M}+\mathrm{H}-2 * 113]^{+}, 609.54[\mathrm{M}+\mathrm{H}-3 * 113]^{+}, 496.39$ $[\mathrm{M}+\mathrm{H}-4 * 113]^{+}, \quad 383.41 \quad\left[\mathrm{M}+\mathrm{H}-5^{*} 113\right]^{+}, \quad 270.24$ $\left[\mathrm{M}+\mathrm{H}-6^{*} 113\right]^{+}$; negative HRESIMS $\mathrm{m} / \mathrm{z} 946.6723$ $[\mathrm{M}-\mathrm{H}]^{-}$, calcd for $\mathrm{C}_{49} \mathrm{H}_{89} \mathrm{~N}_{9} \mathrm{O}_{9}, 946.6710$.
3.6 Advanced Marfey's Method [11, 12]

The cyclic peptide (about 1.0 mg each) was dissolved in $6 \mathrm{~N} \mathrm{HCl}(1 \mathrm{~mL})$ and heated at $110{ }^{\circ} \mathrm{C}$ for 24 h . The hydrolyzate was evaporated to dryness, and the residue was re-dissolved in $100 \mu \mathrm{~L}$ of acetone. To each a half portion ( $50 \mu \mathrm{~L}$ ) were added $20 \mu \mathrm{~L}$ of $\mathrm{NaHCO}_{3}(1 \mathrm{M})$ and $100 \mu \mathrm{~L}$ of $\mathrm{N}^{\alpha}$-(5-Fluoro-2,4-dinitrophenyl)-L-leucinamide (L-FDLA, $1 \%$ in acetone) or $50 \mu \mathrm{~L}$ of $\mathrm{N}^{\alpha}$-(5-Fluoro-2,4-dinitro-phenyl)-L-leucinamide and $50 \mu \mathrm{~L}$ of $\mathrm{N}^{\alpha}$-(5-Fluoro-2,4-dini-trophenyl)-d-leucinamide (mixture of L-FDLA and D-FDLA, $1 \%$ in acetone), and the mixture was heated at $45^{\circ} \mathrm{C}$ for 1.5 h . Reaction was cooled to room temperature, and then acidified with $2 \mathrm{~N} \mathrm{HCl}(10 \mu \mathrm{~L})$, dried and dissolved in $50 \%$ aqueous MeCN. $5 \mu \mathrm{~L}$ of each solution of FDLA derivatives were analyzed by LC/MS.

The analysis of the $\mathrm{L}-$ and D , L-FDLA (mixture of D - and l-FDLA) derivatives was performed using an Waters Sunfire $\mathrm{C}_{18}$ column $(4.6 * 150 \mathrm{~mm}, 5 \mu \mathrm{~m})$ maintained at $30{ }^{\circ} \mathrm{C}$. Acetonitrile- $0.1 \% \mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}$ was used as the mobile phase under a linear gradient elution mode (acetonitrile, $28-60 \%$, 50 min (compound 1); acetonitrile, $35-60 \%, 50 \mathrm{~min}$ (compound 2)) at a flow rate of $1 \mathrm{~mL} /$ min . A Waters Xevo-TQD mass spectrometer was used for detection in $\mathrm{ESI}^{-}$mode. The capillary voltage was kept at 2.5 kV , and the ion source at $450^{\circ} \mathrm{C}$. Nitrogen gas was used as a sheath gas at $650 \mathrm{~L} / \mathrm{h}$. A mass range of $\mathrm{m} / \mathrm{z}$ $100-2000$ was scanned in 0.2 s .

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## Compliance with Ethical Standards

Conflicts of interest The authors declare no conflict of interest.
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