# Colorimetric detection of DNA, small molecules, proteins, and ions using unmodified gold nanoparticles and conjugated polyelectrolytes

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We have demonstrated a novel sensing strategy employing single-stranded probe DNA, unmodified gold nanoparticles, and a positively charged, water-soluble conjugated polyelectrolyte to detect a broad range of targets including nucleic acid (DNA) sequences, proteins, small molecules, and inorganic ions. This nearly "universal" biosensor approach is based on the observation that, while the conjugated polyelectrolyte specifically inhibits the ability of single-stranded DNA to prevent the aggregation of gold-nanoparticles, no such inhibition is observed with double-stranded or otherwise "folded" DNA structures. Colorimetric assays employing this mechanism for the detection of hybridization are sensitive and convenient—picomolar concentrations of target DNA are readily detected with the naked eye, and the sensor works even when challenged with complex sample matrices such as blood serum. Likewise, by employing the binding-induced folding or association of aptamers we have generalized the approach to the specific and convenient detection of proteins, small molecules, and inorganic ions. Finally, this new biosensor approach is quite straightforward and can be completed in minutes without significant equipment or training overhead.

biosensor | aptamer | visual detection | thrombin detection |

old nanoparticle colorimetric biosensors have seen signifi-Cant applications in diagnostics, environmental monitoring, and antibioterrorism supporting unaided, visual readout (1–12). Commonly, the relevant nanoparticles are covalently modified with either a probe DNA or an aptamer such that hybridization (13–16) or aptamer-target interactions (17–27), for example the scanometric method developed by Mirkin (25), which is a very sensitive and specific tool, crosslink them, inducing aggregation. The second broad approach utilizes unmodified nanoparticles. (28–30) These two approaches, however, suffer from timeconsuming (20-40 h of assembly) and relatively poor (low nanomolar) detection limits, respectively. Here, a unique, colorimetric sensing strategy employing a simple but selective combination of a single-stranded DNA probe, a positively charged, water-soluble conjugated polyelectrolyte, and unmodified gold nanoparticles is demonstrated. The universality of this method allows detection of a broad range of targets, including nucleic acid (DNA) sequences, proteins, small molecules, and inorganic ions. Our approach is rapid (turnaround time is 5–10 min) and sensitive (picomolar concentrations of target DNA are readily detected with the naked eye, even in complex sample matrices like blood serum). Hence, an operator with minimum scientific overhead can easily employ this technique.

Generally, the gold nanoparticle applications typically rely on a quantitative coupling between target recognition and the aggregation of the nanoparticles, which, in turn, leads to a dramatic change in the photonic properties—and thus the color—of the nanoparticle solution. This colorimetric "readout" avoids the relative complexity inherent in optical imaging/detection meth-

odologies and thus may prove suitable for point-of-care and developing world applications (9–12). Examples for the covalently modified nanoparticles include the work of Mirkin and collaborators, who pioneered the use of gold nanoparticle-DNA conjugates (22, 23) for the sensitive detection of DNA and proteins (24–27). For the unmodified nanoparticles, Li and Rothberg have shown that, at high ionic strength, single-stranded—but not double-stranded—DNA protects unmodified gold nanoparticles from aggregating, thus modifying their color (28–30). Employing this mechanism, these authors have demonstrated a colorimetric assay for specific DNA sequences.

Our approach to the application of gold nanoparticles in optical biosensing relies on the recent observations that (i) both single-stranded DNA and double-stranded DNA prevent aggregation of gold nanoparticles at low salt concentrations (28–30); (ii) conjugated polyelectrolytes lead to the ready aggregation of such nanoparticles (31), and (iii) previous work has demonstrated that, at certain conditions, the cationic conjugated polyelectrolyte poly [(9,9-bis (6'-N,N,N-trimethylammonium) hexyl) fluorene-alt-1,4phenylene] bromide (PFP-Br) binds single-stranded DNA preferentially to double-stranded or otherwise "folded" DNA, (32–34). This appears to arise due to the greater hydrophobicity of singlestranded DNA. Upon reducing the strength of the hydrophobic interactions, the electrostatic attraction becomes the important interaction that regulates the binding between the water-soluble conjugated polymer and DNA (33). The different affinities between the cationic conjugated polymer and various forms of DNA (single-stranded DNA and double-stranded DNA and single-stranded DNA and complex DNA folds) can be used to design a variety of biosensors (32–34).

#### Results

Inspired by the above, we have developed a unique colorimetric assay for the detection of nucleic acids, small-molecules, proteins, and inorganic ions that relies on unmodified gold nanoparticles but that employs the conjugated, positively charged, water-soluble polymer poly [(9,9-bis (6'-N,N,N-trimethylammonium) hexyl) fluorene-alt-1,4-phenylene] bromide). We have employed cationic conjugated polyelectrolytes to efficiently sequester single-stranded—but not double-stranded or otherwise "folded"—DNA from nanoparticles, leading to their aggregation and color change (Fig. S1) (32–34). We have then used this effect as the

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The authors declare no conflict of interest.

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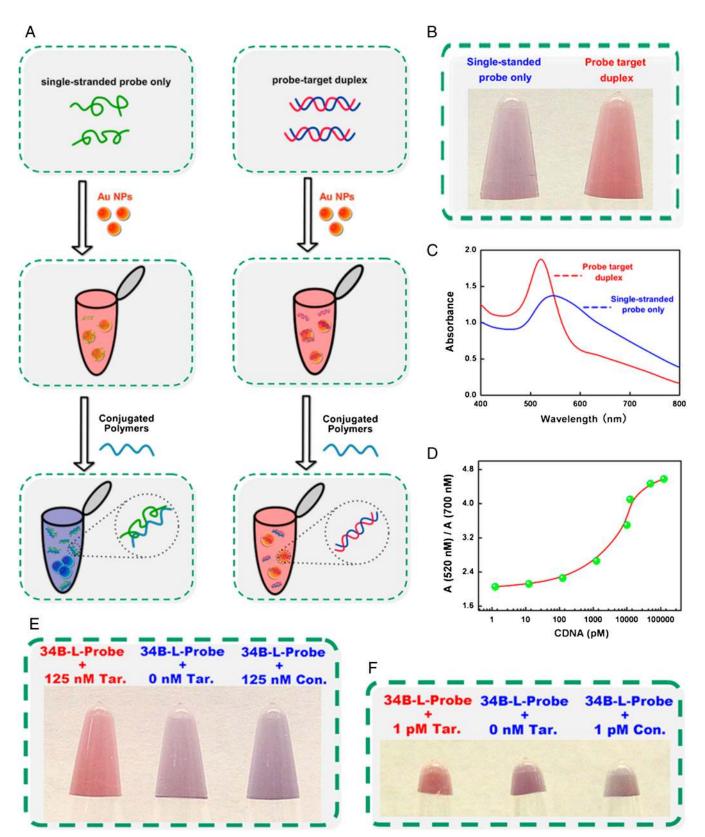


Fig. 1. Here we demonstrate a sensitive colorimetric assay for the detection of DNA. We find that, whereas a mixture of a positively charged, water-soluble, conjugated polyelectrolyte and single-stranded DNA leads to the aggregation of gold nanoparticles (and, consequently, a readily detectable change in their color), admixtures of this polymer with double-stranded DNA does not similarly produce a color change. (A) Here we have used this effect as the basis of a ready  $color imetric assay for the \ detection of specific oligonucle otides. The \ assay is \ rapid and \ sensitive: (<125 \ attomoles of \ target in 10 \ \mu L \ solution \ differentiates \ a \ color \ and \ sensitive \ a \ color \ a \ col$ change observable with the naked eye in less than 10 min. (B) According to the scheme, the solution only contains a single-stranded probe that makes the gold  $nanoparticles\ aggregate\ (blue), while\ the\ probe-target\ duplex\ keeps\ the\ gold\ nanoparticles\ stable\ (red).\ (C)\ The\ ratio\ A_{520}/A_{700}\ for\ a\ perfect\ match\ target\ is\ much$ larger than no target indicating the specificity of our assay. (D) By calculating the ratio  $A_{520}/A_{700}$  form the UV-Vis test we could measure the target concentration, which is also crucial for detection. Shown is our ability to specifically detect target DNA molecules at both E, high (125 nM), and F, (1.25 pM) concentrations.

basis for an assay for the sensitive, colorimetric detection of a wide range of molecular analytes.

For the detection of DNA (Fig. 1A), we first prepare a control sample containing a single-stranded probe DNA and a test sample containing the probe DNA and its complementary DNA target. A solution of 20 nm gold nanoparticles is added to both, producing a readily apparent red color due to the intense surface plasmon resonance (SPR) absorption of the nanoparticles at 520 nm. The subsequent addition of the conjugated polyelectro-

lyte sequesters most of single-stranded DNA, leaving it unable to stabilize the nanoparticles against aggregation and thus leading to a characteristic blue color for the control sample (which only contains single-stranded DNA). In the presence of the complementary target sequence a significant concentration of double-stranded DNA is formed, which only weakly binds the conjugated polyelectrolytes and thus remains largely free to stabilize the nanoparticles against aggregation. This, in turn, causes the sample to retain the red color associated with dispersed nanoparticles (Fig. 1B).

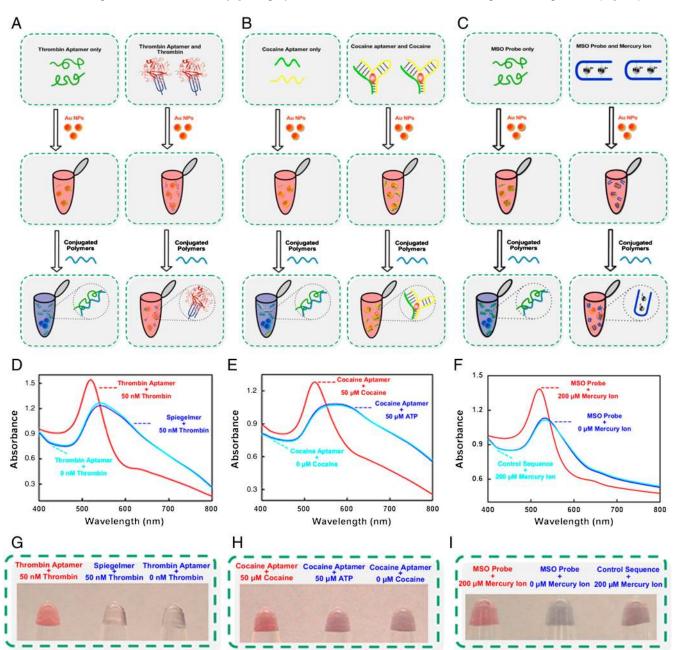


Fig. 2. Our assay also could easily be expanded to the detection of proteins, small molecules, and ions. (A) A thrombin aptamer has been used in our protein detection. (B) We use a sandwich approach based on single aptamer sequences in our cocaine detection. (C) We use a mercury-responsive sequence that folds in the presence of Hg (II) to detect the Hg (II). (D) When the sample contains thrombin, it folded the aptamer into G-quadruplex structure together, which stabilized the gold nanoparticles remaining red. The sample containing control probe or no target induced the aggregation, which shifts the absorption from 520 nM to the longer wavelength, resulting in the characteristic red-blue color change. The detect limit for thrombin is 10 nM. (E) When the sample contains cocaine, it associate two part aptamer together, which stabilize the gold nanoparticles remaining red. The sample containing ATP or no target induced the aggregation, which shifts the absorption from 520 nM to the longer wavelength, resulting in the characteristic red-blue color change. The detect limit for cocaine is  $10 \mu$ M. (F) The sample containing no target induced the aggregation, which shifts the absorption from 520 nM to the longer wavelength, resulting in the characteristic red-blue color change. The detect limit for Hg (II) is  $50 \mu$ M. (G, H, and I) This color change allows us to easily differentiate between target-containing and target-free samples by visual inspection.

The color change associated with the lack of double-stranded DNA allows us to easily differentiate the single-stranded DNA probe and the double-stranded probe-target duplex via visual inspection at concentrations as low as 1.25 pM (Fig. 1 E and F), a detection limit comparable to those of many previously reported fluorescent DNA assays and enzyme-linked amplified assays (35). A more quantitative analysis can be made using UV-Vis spectroscopy. Specifically, the absorbance of the solution at 520 nm increases and the absorbance at 700 nm decreases when the concentration of double-stranded DNA is increased from 1.25 pM to 125 nM (Fig. 1 C and D). A series of control experiments confirm the specificity of the assay: As expected, duplexes between the 34-base pair probe and targets containing 3-, 5- or 7 base-pair mismatches do not stabilize the nanoparticles against aggregation, leading to the characteristic blue color

A key feature of our method is that, unlike approaches that utilize functionalized nanoparticles, hybridization and binding occur under conditions that can be independently regulated and optimized. For example, our approach relies on an ability to detect double-stranded probe-target duplexes. Thus, a potential disturbance is that any double-stranded DNA contaminating the sample will also stabilize the nanoparticles. Because we can optimize test conditions, the pretreatment of the sample with Exonuclease III, which provides a ready and convenient solution degrading any double-stranded DNA initially present in the sample (Fig. S3). Thus our assay can be made quite specific. Likewise the assay is selective and performs well even when challenged directly in complex biological media, such as target-doped blood serum (Fig. S4).

#### Discussion

While the detection of DNA is of significant clinical utility in its own right, by employing aptamers, nucleic acid molecules that have been selected in vitro to bind to specific molecular targets (36-39), our colorimetric assay can also be expanded to the detection of nonnucleic acid targets. Specifically, because it is relatively easy to reengineer aptamers such that they undergo a transition from largely single-stranded to a folded, compact structure upon target binding (40-44), and because the conjugated polyelectrolyte only weakly binds with these compact, folded structures (31–34), we can adapt our assay to the detection of the aptamer's binding partner (Fig. 2).

As a first example of the broader applicability of our approach, we developed an assay for the small molecule cocaine using the anticocaine aptamer of Stojanovic (39–42). To do so we modified the aptamer sequence slightly (see Methods) to prevent interstrand binding and then cut the sequence into two short, single-stranded pieces (40, 41). We then prepared two samples: a control sample containing only these fragments and a test sample containing the two aptamer fragments and cocaine. Upon adding a solution of ~20 nm gold nanoparticles, both samples turn red

due to the intense surface plasmon resonance absorption of the nanoparticles. Upon the addition of the conjugated polyelectrolyte, the nanoparticles in the control solution aggregate and the solution turns blue. The test sample, in contrast, retains its red color, presumably because cocaine binding drives the two pieces to associate (40-42) to form a well-folded, largely double-stranded aptamer that only weakly binds with the conjugated polyelectrolytes and thus remains free to inhibit nanoparticle aggregation (Fig. 2). This color change supports the ready detection of cocaine via visual inspection. As expected, this sensor does not respond when challenged with other small molecules, such as ATP (Fig. 2).

Abundant literature suggests that it is relatively easy to modify aptamers such that they only fold upon target binding (40-45), an effect that also supports our assay. As examples we have used Bock's thrombin-binding aptamer, which folds from a singlestranded state to a largely G-quadruplex structure upon target binding (43, 44) and a mercury-responsive sequence that folds in the presence of Hg (II) (45–49). Once again, we employ control (aptamer-only) and test (aptamer-plus-target) solutions, which, upon mixing with ~20 nm gold nanoparticles, produce the characteristic absorption at 520 nm. In the absence of target, both aptamers are single-stranded and thus do not stabilize the gold nanoparticles against aggregation in the presence of the polymer. Upon target binding, however, the aptamers fold, reducing their interactions with the conjugated polyelectrolytes and allowing them to prevent nanoparticle aggregation (Fig. 2). The resultant color change supports the easy differentiation between targetcontaining and target-free samples via visual inspection.

The strategy described in this report offers advantages over many existing sensing methodologies. First, our approach is likely universal; here, for example, we have shown that it is applicable to the detection of nucleic acids, small molecules, proteins, and inorganic ions. Second, our approach is convenient, requiring only the mixing of several solutions at room temperature to achieve rapid, semiquantitative detection via visual inspection or quantitative detection via visible light absorbance spectroscopy. Third, the detection of DNA using this approach is sensitive. We achieve, for example, low picomolar sensitivity for DNA detection, comparing very favorably with previously reported methods (Table 1) (28, 35, 50–52). Fourth, each step in the process (pretreatment with Exonuclease, the binding of the target to the DNA probe, the addition of conjugated polyelectrolytes) is performed separately, allowing each step to be optimized independently. Finally, the detection process is rapid and convenient.

#### Methods

In a typical gold nanoparticles/CP assay, the DNA probes were dissolved in a salt solution (112.5 mM NaCl) to form 1.25  $\mu M$  probe solution at room temperature. Such solution (2.5  $\mu$ L) was added to gold nanoparticles solution (25 LL), and the obtained solution was incubated for 1 min at room temperature. The concentrations of probes are 125 nM in this solution unless specially indicated. After that CP (5  $\mu$ L, 1  $\mu$ M) was added to the above solution,

Table 1. Comparison of sensors for DNA detection

Detection method	Strategy	Detection limit	Existing problems
Colorimetric (this work)	AuNP & conjugated polyelectrolyte	1 pM (naked eyes)	Not applicable for colored samples
Colorimetric (28)	AuNP & salt	4.3 nM (naked eyes)	Detect limit not applicable for colored samples
Colorimetric (50)	Enzyme amplification magnetic separation	100 pM (naked eyes)	Multistep process, cumbersome preparation of functionalized gold nanoparticles and magnetic bead. Time-consuming (>1 day), not applicable for colored samples
Electrochemistry (51)	Electron transfer	10 pM	Dual-labeled oligos. Time-consuming (≈5 hours)
Electrochemistry (52)	Enzyme Amplification & PNA Probe & AgNP	10 fM	Multistep process, stability of enzyme, dual label of DNA, reagent intensive. Time-consuming (≈5 hours)
Fluorescence (35)	Conjugated Polyelectrolyte FRET	2.14 pM	Using expensive and complicated instruments for detection. Time-consuming ( $\approx$ 3 hours)

followed by either visual observation or UV/Vis characterization. The total volume of the final solutions in this work may differ, but the volume ratio of the above DNA, gold nanoparticles solution, and CPs is 1:10:2. In cocaine assays, the 3' fragment and 5' fragment were dissolved in salt solution (112.5 mM NaCl) to form 0.625 uM 3' fragment and 0.625 uM 5' fragment probe solution at room temperature. Then cocaine was added to the above solution. This solution (2.5  $\mu$ L) was added to gold nanoparticles solution (25  $\mu$ L), and the obtained solution was incubated for 2 min at room temperature. After that CP (5  $\mu$ L, 1  $\mu$ M) was added, followed by either visual observation or UV/Vis characterization. For the thrombin assay, the thrombin aptamer (or spiegelmer) was dissolved in a salt solution (112.5 mM NaCl)

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to form 1.25  $\mu$ M probe solution at room temperature. Then thrombin was added followed by incubation for 20 min. After that CP (5  $\mu$ L, 1  $\mu$ M) was added, followed by either visual observation or UV/Vis characterization.

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# **Supporting Information**

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#### SI Text

Mechanism Discussion. The mechanism underlying this detection method appears to be linked to the different affinities with which double- and single-stranded DNA bind to the conjugated polymer (1–3). Because of this difference, the polymer does not bind all double-stranded DNA to prevent aggregation but efficiently removes single-stranded DNA from the gold nanoparticles, which, in turn, allows the nanoparticles to aggregate and change color. In support of this argument, dye-modified single-stranded DNA is quenched far more effectively by conjugated polyelectrolytes than is the equivalent dye-modified double-stranded DNA (Fig. S1 A and B), indicating that the polymer binds to singlestranded DNA binds more tightly than it binds double-stranded DNA. Although the single-stranded DNA has less charge density than double-stranded DNA, the additional hydrophobic interactions in ssDNA may compensate for the weaker electrostatic interactions due to its flexible structure and more exposed hydrophobic region (1–6).

**Test of Bio-Specificity.** A series of control experiments confirm the specificity of the assay. As expected, duplexes between the 34-base pair probe and targets containing 3-, 5 or 7 base-pair mismatches (Fig. S2) produce the same blue color as control samples lacking any target.

Optimizing the Conditions. The color change, induced by bio-specific binding to the unmodified gold nanoparticles, occurs under predetermined and separately optimized conditions. For example, our approach relies on the ability to detect double-stranded probetarget duplexes. A potential disturbance is thus that any doublestranded DNA contaminating the sample will stabilize the nanoparticles. However we could optimize test conditions, the pretreatment of the sample with Exonuclease III provides a ready and convenient solution to this problem. Specifically, this enzyme catalyzes the removal of mononucleotides from the 3'-end of double-stranded DNA but does not degrade single-stranded DNA. For example, an actual sample could contain some kind of double-stranded DNA (here, 27 bases double-stranded DNA). If so, when the sample contains no perfect DNA (here, 7-mismatch-DNA) to the probe, the double-stranded DNA could stabilize AuNPs using the above described assay (Fig. S4A). If, however, we first treat the sample with Exonuclease III followed by heat deactivation of the enzyme (via immersion for 5 min in a 65 °C water bath), the problem is solved without disrupting the method's ability to detect the authentic target (Fig. S3 B, C, and D).

**Recognition of dsDNA in Serum.** To demonstrate that the new assay is able to recognize dsDNA in biological media, we prepared the ssDNA and dsDNA ample in 56 mM NaCl solution containing 30 vol% serum. Serum is plasma without fibrinogen or other clotting factors; serum contains proteins, glucose, mineral ions, hormones, and other biological substances. We added the above solution into the AuNPs and followed added CP. The final con-

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centration for the serum is 3 vol%. And both of the solutions are red at this stage. Then we centrifuged the two samples for 2 min at 14,100 rcf. Finally the sample contains dsDNA retains red, while the sample with ssDNA turns to very light blue (Fig. S4).

**Details of the Assay.** Tables S1, S2, S3, and S4 show the details of the assay.

Materials. Gold nanoparticles (20 nm) were obtained from Sigma-Aldrich. The AuNP solution was centrifuged to a concentrated solution; the optimum conditions were centrifuging for 15 min at 4 °C, with a RCF of 16100 g. Exonuclease III was obtained from Sigma-Aldrich. DNase I (RNase-free) was obtained from New England Biolabs. Human R-thrombin was obtained from Haematologic Technologies Inc.; Essex Junction, VT. Cocaine, mercury ion and serum were used as received. The sequences of the involved oligonucleotides, which were all purchased from Biosearch Technologies, Inc., are listed in following. UV/Vis absorption spectroscopy was performed with a Beckman Coulter DU 800 spectrophotometer and photographs were taken with a Nikon digital camera. The Polymer, poly[(9,9-bis(6'-N,N,N-trimethylammonium)hexyl)fluorene-alt-1,4-phenylene] bromide (PFP-Br) was prepared according to the published references (7, 8).

The sequences of the DNA probes and targets used in this work. 34-mer-Linear-Probe: 5'-TGGATCGGCGTTTTATTCTTGTTCA-GATATTCAA-3'

34-mer-Linear-Target 0 mismatch: 5'-TTGAATATCTGAACA-AGAATAAAACGCCGATCCA-3'

34-mer-Linear-Target 3 mismatch: 5'-TTGAATATCTGAACA-ACTTTAAAACGCCGATCCA-3'

34-mer-Linear-Target 5 mismatch: 5'-TTGAATATCTGAACA-TCTTAAAAACGCCGATCCA-3'

34-mer–Linear-Target 7 mismatch: 5'-TTGAATATCTGAAC-TTCTTATAAACGCCGATCCA-3'

27-mer–Linear-Probe: 5'-GACACTGGATCGGCGTTTTATT-CACAG-3'

27-mer–Linear-Target 0 mismatch: 5'-CTGTGAATAAAACG-CCGATCCAGTGTC-3'

27-mer–Linear-Target 7 mismatch: 5'-CTGTGAATAA<u>TTGCG-GC</u>ATCCAGTGTC-3'

3 fragment for anticocaine aptamer: 5'-AGACAAGGAAAA-3' 5 fragment for anticocaine aptamer: 5'-TCCTTCAATGAAG-TGGGTCG-3'

Thrombin Aptamer: 5'-TAAGTTCATCTCCCCGGTTGGTGTGGTTGGTTGGT-3'

Spiegelmer: 5'-TAAGTTCATCTCCCCGGTTGGTGTGGTT-GGT-3'

Mercury-Specific Oligonucleotide (MSO) probe: 5'-TGTTTCTT-TCTTCCCCTTGTTTGTTTCA-3'

Mercury-Control sequence: 5'-TAGCTATGGAATTCCTCG-TAGGCA-3'

- Liu B, Gaylord BS, Wang S, & Bazan GC (2003) Effect of chromophore-charge distance on the energy transfer properties of water-soluble conjugated oligomers. J Am Chem Soc 125(22):6705–6714.
- Gaylord BS, Heeger AJ, Bazan GC (2003) DNA hybridization detection with watersoluble conjugated polymers and chromophore-labeled single-stranded DNA. J Am Chem Soc 125(4):896–900.
- Xu Q-H et al. (2004) Time-resolved energy transfer in DNA sequence detection using water-soluble conjugated polymers: The role of electrostatic and hydrophobic interactions. Proc Natl Acad Sci USA 101(32):11634–11639.

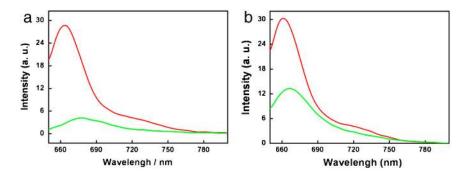


Fig. S1. Experiments for preferential affinity between ssDNA with PFP-Br. (A) Fluorescence emitted from Cy5 attached to 34-mer-DNA probe (red curve) and after adding PFP-Br (green curve). (B) Fluorescence emitted from Cy5 attached to 34-mer-DNA probe with target DNA (red curve) and after adding PFP-Br (green curve).

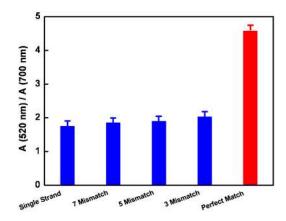


Fig. S2. The absorbance ratio  $(A_{520}/A_{700})$  versus each corresponding DNA/AuNPs/CPs solution.

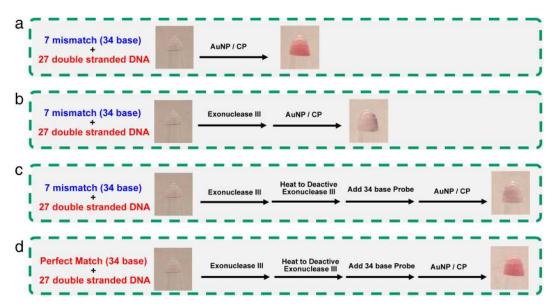


Fig. S3. Optimizing the conditions in the DNA Test via pretreatment with Exonuclease III. (A) A test sample containing a contaminating double-stranded DNA adopts the characteristic red color associated with the presence of no-authentic sample. (B) If the sample is pretreated with Exonuclease III this sample could not stabilize AuNPs (see C). (D) Finally, a sample containing both contaminating double-stranded DNA and authentic, single-stranded target provides a correct, positive response after this treatment protocol.

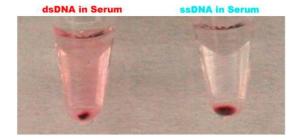


Fig. 54. Colorimetric detection of oligonucleotide hybridization in serum. 34-mer-linear-probes detect the target (dsDNA) and control (ssDNA).

### Table S1. The assay of 34 base probe

#### Step 1. Solution 1

			•		
Color	34 base DNA Probe Colorless	34 base DNA Probe+34 base DNA target (perfect match to probe) Colorless	34 base DNA Probe+34 base DNA control (3 mismatch to probe) Colorless	34 base DNA Probe+34 base DNA control (5 mismatch to probe) Colorless	34 base DNA Probe+34 base DNA control (7 mismatch to probe) Colorless
	Step 2. Adding solution 1 to AuNPs solutions to make solution 2				
Color	34 base DNA Probe+AuNPs				e 34 base DNA Probe+34 base DNA control (7 mismatch to probe)+AuNPs Red
COIOI	iteu				Neu
	Step 3. Adding conjugated polyelectrolyte (CP) into solution 2				
	34 base DNA Probe+AuNPs +CP	34 base DNA Probe+34 base DNA target (perfect match to probe)+AuNPs+CP		234 base DNA Probe+34 base DNA control (5 mismatch to probe)+AuNPs+CP	e 34 base DNA Probe+34 base DNA control (7 mismatch to probe)+AuNPs+CP
Color	Blue	Red	Blue	Blue	Blue
20101	Probe	Probe+Target	Probe+Control	Probe+Control	Probe+Control

#### Table S2. The assay for cocaine aptamer

#### Step 1. Solution 1

		Step 1. Solution 1	
Color	Cocaine aptamer 3' fragment and 5' fragment Colorless	Cocaine aptamer 3' fragment and 5' fragment+Cocaine Colorless	Cocaine aptamer 3' fragment and 5' fragment+ATP Colorless
	Step 2. Addin	g solution 1 to AuNPs solutions to make solution	2
Color	Cocaine aptamer 3' fragment and 5' fragment+AuNPs Red	Cocaine aptamer 3' fragment and 5' fragment+Cocaine+AuNPs Red	Cocaine aptamer 3′ fragment and 5′ fragment+ATP +AuNPs Red
	Step 3. Addi	ng conjugated polyelectrolyte (CP) into solution 2	2
Color	Cocaine aptamer 3' fragment and 5' fragment+AuNPs+CP Blue	Cocaine aptamer 3' fragment and 5' fragment+Cocaine+AuNPs+CP Red	Cocaine aptamer 3' fragment and 5' fragment+ATP+AuNPs+CP Blue
20.01	Probe	Probe+Target	Probe+Control

#### Table S3. The assay of Thrombin aptamer

#### Step 1. Solution 1

Color	Thrombin aptamer Colorless	Thrombin aptamer+Thrombin Colorless	Spiegelmer+Thrombin Colorless
	Step 2. Addi	ng solution 1 to AuNPs solutions to make so	lution 2
Color	Thrombin aptamer+AuNPs Red	Thrombin aptamer+Thrombin+AuNPs Red	Spiegelmer+Thrombin+AuNPs Red
	Step 3. Add	ling conjugated polyelectrolyte (CP) into solu	ition 2
Color	Thrombin aptamer+AuNPs+CP Blue Probe	Thrombin aptamer+Thrombin+AuNPs+CP Red Probe+Target	Spiegelmer+Thrombin+AuNPs+CP Blue Control+Target

# Table S4. The assay for mercury ions

# Step 1. Solution 1

Color	MSO probe Colorless	MSO probe+Mercury ion Colorless	Control sequence+Mercury ion Colorless	
	Step 2.	Adding solution 1 to AuNPs solutions to	o make solution 2	
Color	MSO probe+AuNPs Red	MSO probe+Mercury ion+AuNPs Red	Contro sequence+Mercury ion+AuNPs Red	
	Step 3. Adding conjugated polyelectrolyte (CP) into solution 2			
Color	MSO probe+AuNPs+CP Blue Probe	MSO probe+Mercury ion+AuNPs+CP Red Probe+Target	Contro sequence+Mercury ion+AuNPs+CP Blue Control+Target	