

# Comparison of Ti/Pd/Ag, Pd/Ti/Pd/Ag and Pd/Ge/Ti/Pd/Ag contacts to n-type GaAs for electronic devices handling high current densities

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## Abstract

In the quest for metal contacts for electronic devices handling high current densities, we report the results of Pd/Ti/Pd/Ag and Pd/Ge/Ti/Pd/Ag contacts to n-GaAs and compare them to Ti/Pd/Ag and AuGe/Ni/Au. These metal systems have been designed with the goal of producing an electrical contact with (a) low metal–semiconductor specific contact resistance, (b) very high sheet conductance, (c) good bondability, (d) long-term durability and (e) cost-effectiveness. The structure of the contacts consists of an interfacial layer (either Pd or Pd/Ge) intended to produce a low metal–semiconductor specific contact resistance; a diffusion barrier (Ti/Pd) and a thick top layer of Ag to provide the desired high sheet conductance, limited cost and good bondability. The results show that both systems can achieve very low metal resistivity ( $\rho_M \sim 2 \times 10^{-6} \Omega \text{ cm}$ ), reaching values close to that of pure bulk silver. This fact is attributed to the Ti/Pd bilayer acting as an efficient diffusion barrier, and thus the metal sheet resistance can be controlled by the thickness of the deposited silver layer. Moreover, the use of Pd as interfacial layer produces contacts with moderate specific contact resistance ( $\rho_C \sim 10^{-4} \Omega \text{ cm}^2$ ) whilst the use of Pd/Ge decreases the specific contact resistance to  $\rho_C \sim 1.5 \times 10^{-7} \Omega \text{ cm}^2$ , as a result of the formation of a Pd<sub>4</sub>(GaAs, Ge<sub>2</sub>) compound at the GaAs interface.

Keywords: ohmic contact, n-GaAs, high conductivity

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## Introduction

Ohmic contacts are key components for high current density devices [1, 2]. Moreover, in the case of LEDs and solar cells this problem is specially demanding since the layout of the front contact has the shape of a grid (i.e. does not fully cover the front side of the device) and thus the problem of high current densities is aggravated by an electrode with limited contact area [3–5]. In such devices, the minimization of

ohmic losses needs to be addressed not only at the semiconductor-metal interface, but also in the top metal layer itself where large current densities flow laterally. Therefore, the search for metallization systems with low contact resistance, high metal sheet conductivity, good bondability, reliability and cost-effectiveness continue to be an interesting and longstanding research topic [6–10]. For ohmic contacts on GaAs, the AuGe/Ni/Au contact has been extensively used for several decades since its inception in 1967 [11–13] due to its low contact resistance and good adherence caused by the alloying process with controllable roughness [14]. However,

**Table 1.** Comparison of contact properties of Ti/Pd/Ag and Pd/Ti/Pd/Ag on n-type GaAs with different doping and annealing conditions. The results of the classic AuGe/Ni/Au metallization have been included for reference in the last row.

System	Doping concentration $N_D$ ( $\text{cm}^{-3}$ )	RTA	Specific contact resistance $\rho_C$ ( $\Omega \text{ cm}^2$ )	Normalized contact resistance $r_C$ ( $\Omega \text{ mm}$ )	Metal resistivity $\rho_M$ ( $\Omega \text{ cm}$ )
Ti/Pd/Ag (50/50/1000 nm)	$3.1 \times 10^{18}$	No	Non-ohmic	Non-ohmic	$2.0 \times 10^{-6}$
		400 °C, 100 s	Non-ohmic	Non-ohmic	$2.5 \times 10^{-6}$
Pd/Ti/Pd/Ag (50/50/50/1000 nm)	$1.6 \times 10^{19}$	No	$1.9 \times 10^{-3}$	1.80	$2.0 \times 10^{-6}$
		400 °C, 100 s	$1.5 \times 10^{-3}$	1.65	$2.2 \times 10^{-6}$
		No	Non-ohmic	Non-ohmic	$1.9 \times 10^{-6}$
AuGe/Ni/Au (200/60/500 nm)	$3.1 \times 10^{18}$	400 °C, 100 s	Non-ohmic	Non-ohmic	$2.4 \times 10^{-6}$
		No	$5.8 \times 10^{-4}$	0.91	$2.0 \times 10^{-6}$
		400 °C, 100 s	$9.7 \times 10^{-5}$	0.39	$2.3 \times 10^{-6}$
	$1.6 \times 10^{19}$	375 °C, 180 s	$2.9 \times 10^{-6}$	0.08	$2.2 \times 10^{-5}$

**Table 2.** Comparison of contact properties of Pd/Ge/Ti/Pd/Ag on n-type GaAs with different doping level, Pd/Ge metal bilayer thickness and annealing time. The results of the classic AuGe/Ni/Au metallization have been included for reference in the last row.

Doping concentration $N_D$ ( $\text{cm}^{-3}$ )	System	RTA	Specific contact resistance $\rho_C$ ( $\Omega \text{ cm}^2$ )	Normalized contact resistance $r_C$ ( $\Omega \text{ mm}$ )	Metal resistivity $\rho_M$ ( $\Omega \text{ cm}$ )	No.
$1.3 \times 10^{18}$	Pd/Ge/Ti/Pd/Ag (50/100/50/50/500 nm)	300 °C, 20 s	$8.7 \times 10^{-7}$	$6.7 \times 10^{-2}$	$1.9 \times 10^{-6}$	<i>a</i>
		300 °C, 60 s	$1.8 \times 10^{-6}$	$9.5 \times 10^{-2}$	$1.9 \times 10^{-6}$	<i>b</i>
$2.5 \times 10^{18}$	Pd/Ge/Ti/Pd/Ag (50/100/50/50/500 nm)	300 °C, 20 s	$1.7 \times 10^{-7}$	$2.4 \times 10^{-2}$	$2.0 \times 10^{-6}$	<i>c</i>
		300 °C, 20 s	$1.5 \times 10^{-7}$	$2.2 \times 10^{-2}$	— <sup>a</sup>	<i>d</i>
	Pd/Ge/Ti/Pd/Ag (15/30/50/50/1000 nm)	300 °C, 20 s	$1.1 \times 10^{-5}$	$1.8 \times 10^{-1}$	$1.9 \times 10^{-6}$	<i>e</i>
$1.6 \times 10^{19}$	Pd/Ge/Ti/Pd/Ag (50/100/50/50/500 nm)	300 °C, 20 s	$2.5 \times 10^{-6}$	$7.4 \times 10^{-2}$	$1.9 \times 10^{-6}$	<i>f</i>
$1.6 \times 10^{19}$	AuGe/Ni/Au (200/60/500 nm)	375 °C, 180 s	$2.9 \times 10^{-6}$	$8.2 \times 10^{-2}$	$2.2 \times 10^{-5}$	<i>g</i>

<sup>a</sup> The value is not available.

the interdiffusion that takes place between Au and GaAs, and Ni being an ineffective barrier layer during the annealing process, bring about the intermixing of the metals and even may produce Au spiking into the GaAs layer. Both these facts impact the quality and reliability of the contact since the contamination of the top Au layer deteriorates the sheet conductance of the contact whilst Au spikes can produce short circuits or can cause thermal instability in the contact.

An alternative line of investigation has been the quest for new metallization systems that include robust diffusion barriers between the GaAs and the contact metal stack. For

example, contacts using Ti/Pt barriers on GaAs have been intensively studied [15–19]. In addition to being a very effective barrier, Ti promotes adhesion to GaAs, resulting in a contact with good lateral homogeneity, no spiking and good bondability. On the other hand, a key limitation of this contact is that Ti forms a Schottky barrier to GaAs of  $\sim 0.8$  eV, and thus very high doping levels ( $N_D \gg 10^{19} \text{ cm}^{-3}$ ) are required to increase the tunneling probability and reach a virtually ohmic behavior [20].

When it comes to minimizing the specific metal–semiconductor contact resistance, another interesting metallization

system that has been widely studied lately is that based on Pd/Ge, which has been reported to show very low metal–semiconductor specific contact resistance ( $\sim 10^{-7}$ – $10^{-6} \Omega \text{cm}^2$ ) [21–26]. Since Pd/Ge-based contacts show no spiking and have smooth surfaces comparable to unalloyed contacts [24–26], they can be combined with diffusion barriers such as TiPt [27] or W [28], with a highly conductive metal on top to increase its bondability and sheet conductivity.

In previous work, we studied Ti/Pd/Ag contacts to n-GaAs as a possible low cost and high reliability alternative to Ti/Pt/Au (and alike) contacts [20]. The combination of Ti/Pd acted as an excellent diffusion barrier that preserved the sheet conductivity of the top Ag layer, thus yielding excellent lateral metal resistivity values (i.e. virtually those of bulk silver). In order to improve the metal–semiconductor specific contact resistance ( $\rho_C$ ) of Ti/Pd/Ag based contacts, in this paper two interfacial metal layers, namely, Pd and Pd/Ge, have been explored. The advantages of Pd/Ge have already been discussed. Regarding the use of Pd, according to some studies [19, 29], a Pd interfacial layer would allow the achievement of a low  $\rho_C$  to GaAs by (1) the absence of oxide on the metal–semiconductor interface (caused by Pd penetrating native oxides and dispersing them uniformly) and (2) the formation of PdGaAs phases with low Schottky barrier height to GaAs. In both cases, the impact of such interfacial layers on the robustness of the Ti/Pd diffusion barrier is anticipated to be low, preserving the quality of the sheet resistance attained with Ti/Pd/Ag.

Consequently, in this work we analyze the performance of Pd/Ti/Pd/Ag and Pd/Ge/Ti/Pd/Ag contacts to n-GaAs with different doping levels, in the quest for contacts that provide both minimum metal–semiconductor specific contact resistance and excellent lateral conductivity. These contacts are intended to be used in electronic devices that handle large current densities, especially for those devices with grid-like contacts and limited surface coverage such as solar cells, lasers or light emitting diodes. In order to establish a clear reference, the performance of these contacts will be compared to the simpler Ti/Pd/Ag and to the widely used AuGe/Ni/Au system.

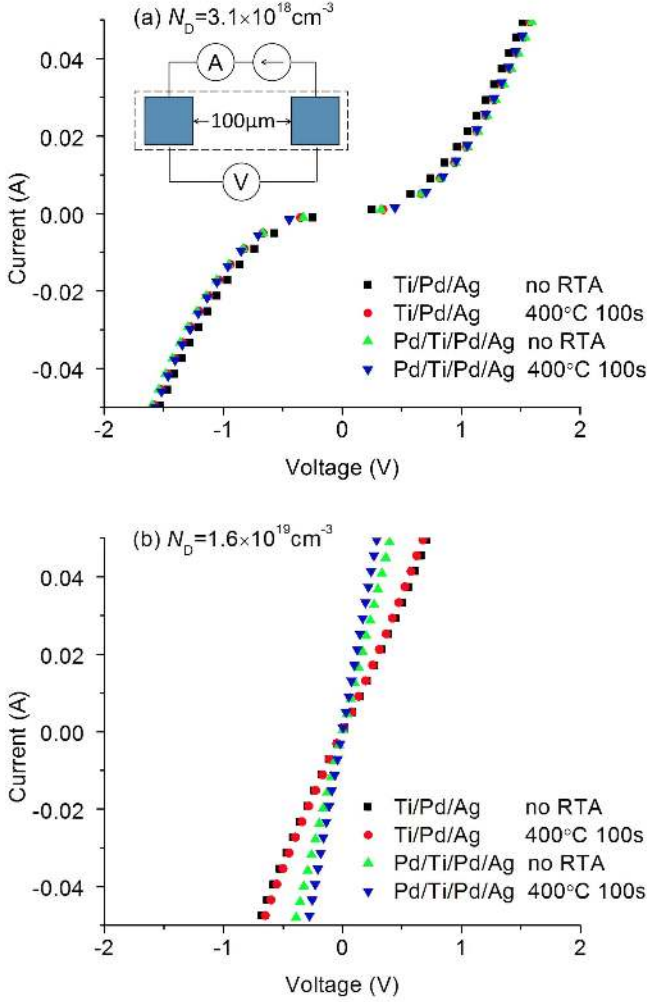
## Experimental procedures

A set of n-GaAs layers were grown by MOVPE on semi-insulating (100) GaAs wafers with a miscut of  $2^\circ$  towards the nearest (111)A plane. The epilayer thickness was of 400 nm and four different doping concentrations of  $1.3 \times 10^{18}$ ,  $2.5 \times 10^{18}$ ,  $3.1 \times 10^{18}$ , and  $1.6 \times 10^{19} \text{cm}^{-3}$  were fabricated to observe the influence of doping level on the contact quality. This doping range was chosen to sweep typical values used in contact layers of lasers, LEDs or solar cells. The dopant used for doping concentrations of  $1.3 \times 10^{18}$  and  $3.1 \times 10^{18}$  was Si, whilst the dopant for doping concentrations of  $2.5 \times 10^{18}$  and  $1.6 \times 10^{19}$  was Te. Contact areas were defined using conventional photolithography techniques. Prior to contact deposition, the substrates were cleaned

using  $\text{H}_2\text{SO}_4:\text{H}_2\text{O}_2:\text{H}_2\text{O}$  (2:1:50) and  $\text{HCl}:\text{H}_2\text{O}$  (1:1) to remove the native oxide layer until a completely hydrophobic surface was obtained; DI water rinsing and blow drying with nitrogen followed. Ti/Pd/Ag, Pd/Ti/Pd/Ag, and Pd/Ge/Ti/Pd/Ag metal stacks were deposited using a multi-pocket electron beam evaporator at a base vacuum of  $1 \times 10^{-6}$  mbar or lower. The thickness of the different metal layers will be indicated in tables 1 and 2 in the following section. Immediately after evaporation, the patterns went through a lift-off process to take away the metal from unwanted areas. The samples were separately annealed by rapid thermal annealing (RTA) in forming gas ( $\text{H}_2:\text{N}_2$ , 1:9) at different temperatures ( $300^\circ\text{C}$ – $400^\circ\text{C}$ ) and times (20–100 s), as will be summarized in tables 1 and 2. In order to compare the quality of the metallizations obtained, samples with the classic contact structure AuGe/Ni/Au (200/60/500 nm) were also fabricated on the highest doped n-GaAs sample ( $1.6 \times 10^{19} \text{cm}^{-3}$ ) and processed in the RTA at  $375^\circ\text{C}$  for 180 s. For electrical characterization, the transmission line model (TLM) was applied to measure the specific contact resistance and the Van der Pauw method [30] together with line shaped patterns were used to measure the metal layer sheet resistance. In the latter case, the metal resistivity was calculated as the sheet resistance times the nominal thickness of the metal layer. To isolate the TLM and Van der Pauw patterns a mesa etching was conducted with  $\text{NH}_4\text{OH}:\text{H}_2\text{O}_2:\text{H}_2\text{O}$  (2:1:10). The electrical characterization was carried out using the 4-wire method by sweeping current and measuring voltage with a Keithley 2062 programmable power supply. Additionally, structural and chemical measurements were carried out on a Pd/Ge/Ti/Pd/Ag sample by means of transmission electron microscopy (TEM) and energy-dispersive x-ray spectroscopy (EDX). Cross section lamellae for electron transparency were prepared using a focused ion beam, protecting the sample surface through pre-deposition of a Pt layer, and milling with  $\text{Ga}^+$  ions close to normal incidence. TEM measurements were carried out with a Philips Tecnai 20F FEG transmission electron microscope.

## Results and discussion

As discussed above, the Ti/Pd/Ag metal system [20] produces contacts with very good metal sheet conductivity; however, its metal–semiconductor specific contact resistance ( $\sim 10^{-3} \Omega \text{cm}^2$ ) is far from state of the art values. In order to improve the specific contact resistance, a 50 nm Pd layer was introduced between GaAs and Ti/Pd/Ag. Figure 1 shows the comparison of  $I$ – $V$  curves of representative samples of Ti/Pd/Ag and Pd/Ti/Pd/Ag with and without RTA on n-GaAs layers with different doping levels. As illustrated in figure 1(a), when the n-GaAs doping level is  $3.1 \times 10^{18} \text{cm}^{-3}$ , both types of metal systems show Schottky behavior before annealing, with the contact with the Pd/n-GaAs interface (green triangles in figure 1(a)) exhibiting a slightly higher barrier height than the contact with the Ti/n-GaAs interface (black squares in figure 1(a)). This is in good agreement with the expected Schottky barrier heights for these two contacts, namely,  $\phi_B \sim 0.8$ – $0.85$  eV for Ti/n-GaAs



**Figure 1.** Comparison of  $I$ - $V$  curves of Ti/Pd/Ag and Pd/Ti/Pd/Ag with and without RTA on n-GaAs. The n-GaAs layer doping concentration is: (a)  $N_D = 3.1 \times 10^{18} \text{ cm}^{-3}$ ; (b)  $N_D = 1.6 \times 10^{19} \text{ cm}^{-3}$ . As sketched by the inset,  $I$ - $V$  curves were measured between pads separated  $100 \mu\text{m}$ .

[31] and  $\phi_B \sim 0.85$ – $0.9 \text{ eV}$  for Pd/n-GaAs [32]. After RTA no big changes are observed in either contact since the  $I$ - $V$  curves of all samples in figure 1(a) almost overlap. This result was surprising since it was expected that the widely reported reactions that occur between Pd and GaAs producing PdGaAs phases [29] would more significantly impact the barrier height. However, similar results have been reported in Pd/n-GaAs contacts made on lightly doped GaAs ( $N_D = 3 \times 10^{16} \text{ cm}^{-3}$ ) where a reduction of only  $\sim 0.02 \text{ eV}$  was observed on the barrier height after annealing under similar conditions [33]. So, in essence, our results indicate that the introduction of a Pd layer between n-GaAs and the Ti/Pd/Ag has little influence on the contact resistance when deposited on lightly doped n-GaAs.

On the contrary, the situation on the highly doped samples changes notably as shown in figure 1(b). As illustrated in this figure, all the contacts made on n-GaAs with  $N_D = 1.6 \times 10^{19} \text{ cm}^{-3}$  have ohmic behavior and, as compared to the  $I$ - $V$  curves of Ti/Pd/Ag, the  $I$ - $V$  curves of Pd/Ti/Pd/Ag exhibit lower contact resistance. Hence, the

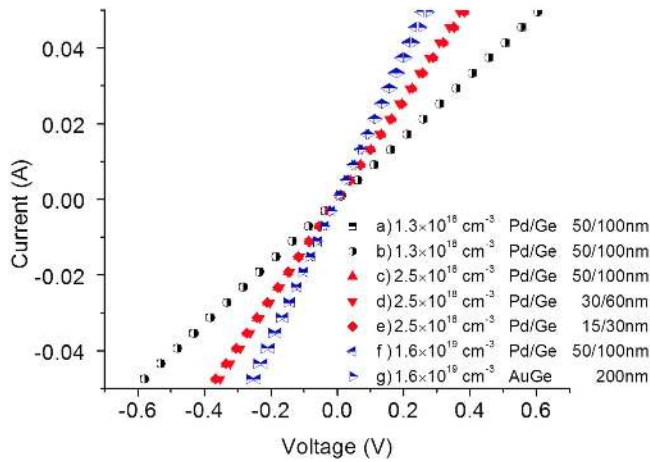
specific contact resistance and sheet metal resistivity were calculated and listed in table 1. As shown in this table, the specific contact resistance of Pd/Ti/Pd/Ag on highly doped n-GaAs ( $1.6 \times 10^{19} \text{ cm}^{-3}$ ) is  $5.8 \times 10^{-4} \Omega \text{ cm}^2$  and, after RTA, it decreases to  $9.7 \times 10^{-5} \Omega \text{ cm}^2$ . This value is lower than the Ti/Pd/Ag specific contact resistance after RTA by a factor of 15. In both cases, it seems likely that the high doping in the GaAs thins the barrier to a point where both contacts behave as ohmic. However, the notable difference between them suggests a more efficient tunneling—i.e. a thinner or lower barrier—in the case of Pd/Ti/Pd/Ag contacts. The physical processes behind this improvement could be the ability of Pd to dissolve surface native oxides of GaAs yielding to thinner barriers. Also the possibility of a lower barrier cannot be ruled out since the high doping in the GaAs could enhance somehow the formation of PdGaAs phases during RTA [19, 29].

In regard to the metal sheet resistivity, all the samples of Ti/Pd/Ag and Pd/Ti/Pd/Ag have similar performance. Before RTA, the metal resistivity is  $\sim 2.0 \times 10^{-6} \Omega \text{ cm}$ , and after RTA, it increases slightly to  $\sim 2.4 \times 10^{-6} \Omega \text{ cm}$ . The small difference may be due to deterioration of Ag conductivity caused by limited diffusion from the GaAs to the Ag layer during the RTA. Nevertheless, these values are still nearly one order of magnitude better than the classic metallization AuGe/Ni/Au, which shows a sheet resistivity of  $2.2 \times 10^{-5} \Omega \text{ cm}$ . These results confirm that introducing a Pd interfacial layer has minimal influence on the performance of the Ti/Pd bilayer that works as an effective barrier to restrict the contamination of Ag by GaAs.

However, although the introduction of the Pd layer decreased considerably the metal–semiconductor specific contact resistance of the contact (by a factor of 15), the final value reached is still far from that of the classic metallization AuGe/Ni/Au with  $\rho_C = 2.9 \times 10^{-6} \Omega \text{ cm}^2$ . Therefore, with the target of obtaining lower specific contact resistances, an interfacial Pd/Ge bilayer was introduced between the n-GaAs layer and the Ti/Pd/Ag metal system. According to the literature, in order to obtain the lowest contact resistance, Ge must be in excess in the Pd/Ge bilayer (i.e. the atomic ratio in the layers must be Pd/Ge  $< 1$ , which corresponds to a thickness ratio of Pd/Ge  $< 2/3$ ) [21, 26, 34]. Accordingly, a Pd/Ge thickness ratio around 1/2 was widely used in these works (45/90, 50/100, 60/130, 75/135 nm) [21, 26, 34] and hence all the Pd/Ge layers in our samples were fabricated with such thickness ratio of 1/2. As shown in table 2, the standard thickness combination chosen in our experiments was 50/100 nm. However, for the samples doped  $2.5 \times 10^{18} \text{ cm}^{-3}$  (namely, samples *c*, *d* and *e* in table 2), three different thickness of Pd/Ge (50/100, 30/60, 15/30 nm) were fabricated to study the impact of total thickness for the bilayer, as will be discussed below.

Figure 2 shows the  $I$ - $V$  curves of representative samples of this contact as a function of n-GaAs doping level, Pd/Ge thickness and RTA conditions. Correspondingly, the values obtained from representative samples of each type are listed in

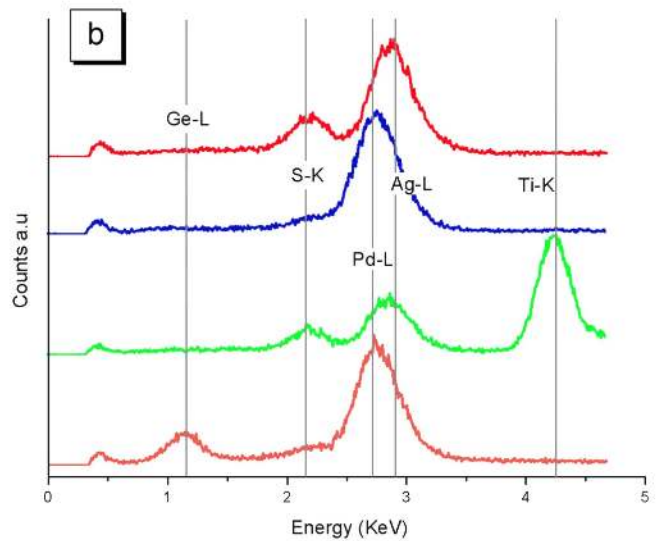
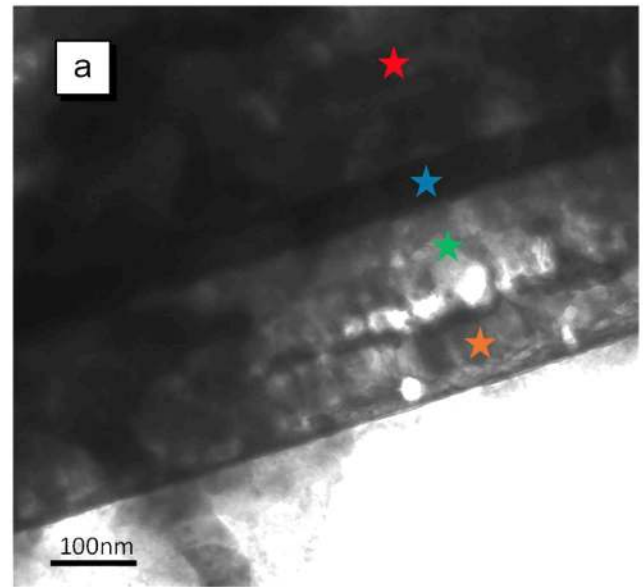




**Figure 2.**  $I$ - $V$  curves of Pd/Ge/Ti/Pd/Ag contact resistance as a function of doping level and Pd/Ge thickness. Pad separation is 100  $\mu\text{m}$  in all cases. The details of doping concentration and Ge/Pd thickness from sample (a)–(f) are listed in the figure whilst other details are listed in table 2. The  $I$ - $V$  curves of the classic AuGe/Ni/Au metallization have also been included as (g).

table 2. As figure 2 shows, the  $I$ - $V$  curves of all samples exhibit ohmic behavior with three slopes that correspond to the three different doping levels used in the n-GaAs contact layer. As shown in table 2, the specific contact resistances of all these samples are much lower than in the case of a single Pd interfacial layer (table 1). In fact, the best values reached for the specific contact resistance ( $1.5$ – $1.7 \times 10^{-7} \Omega \text{cm}^2$ ) are slightly higher than the best reported for Pd/Ge contacts [21, 26, 34]. We interpret this fact as an indirect proof of the lack of influence of the upper layers in the contact region and in turn an indirect proof of the Ti/Pd bilayer working successfully as a diffusion barrier, effectively separating the contact interface region and the sheet conductivity region. This hypothesis has been verified by means of TEM and EDX (see figure 3), and will be discussed later.

As shown in table 2, when the doping level is low ( $1.3 \times 10^{18} \text{cm}^{-3}$ ), the specific contact resistance is  $8.7 \times 10^{-7} \Omega \text{cm}^2$  after RTA at 300  $^\circ\text{C}$ , 20 s. Extending the RTA time to 60 s does not improve the results, which are in fact a factor of 2 larger. For this reason, we decided to fix our alloying times to 20 s [26, 35]. When the doping level is  $2.5 \times 10^{18} \text{cm}^{-3}$ , the results are much better: sample *d* (30/60 nm) has a specific contact resistance value as low as  $1.5 \times 10^{-7} \Omega \text{cm}^2$ , whilst sample *c* (50/100 nm) reaches virtually the same value ( $1.7 \times 10^{-7} \Omega \text{cm}^2$ ). These values are an order of magnitude lower than those attained with the classic AuGe/Ni/Au (sample *g* with  $\rho_c = 2.9 \times 10^{-6} \Omega \text{cm}^2$ ). However, for sample *e* with Pd/Ge thickness of 15/30 nm, the results deteriorate to  $1.05 \times 10^{-5} \Omega \text{cm}^2$ . This increase of two orders of magnitude in the specific contact resistance seems to suggest that the Pd/Ge interfacial bilayer is not thick enough in this case (15/30 nm) for the solid phase regrowth process to proceed appropriately. Finally, for the highly doped sample ( $1.6 \times 10^{19} \text{cm}^{-3}$ ), the value of the specific contact resistance obtained ( $2.5 \times 10^{-6} \Omega \text{cm}^2$ ) is an order of magnitude higher than that of samples *c* or *d*, albeit still very low and good enough



**Figure 3.** (a) Cross-sectional TEM image of sample *c* in table 2; (b) EDX spectra taken from the different points marked with a colored star in figure 3(a). The star and the related EDX spectrum share the same color. The peaks corresponding to the main elements detected in EDX—from left to right Ge, S, Pd, Ag and Ti—are labeled in the plot as vertical lines. The leftmost unlabeled peak corresponds to adventitious C contamination.

for the requirement of most high current density devices [20]. In this case, the ultra-high doping in the n-GaAs does not seem to yield the lowest specific contact resistance. As will be shown later, when discussing the microstructure of the contact, the very low metal–semiconductor specific contact resistance in sample *c* can be related to the formation of complex PdGeGaAs phases at the interface. Therefore, at this point we speculate that the composition and concentration of such phases might be impacted by the doping of the n-GaAs layer, and subsequently influence the contact performance. However, this is just one among several possibilities and still no conclusive explanation

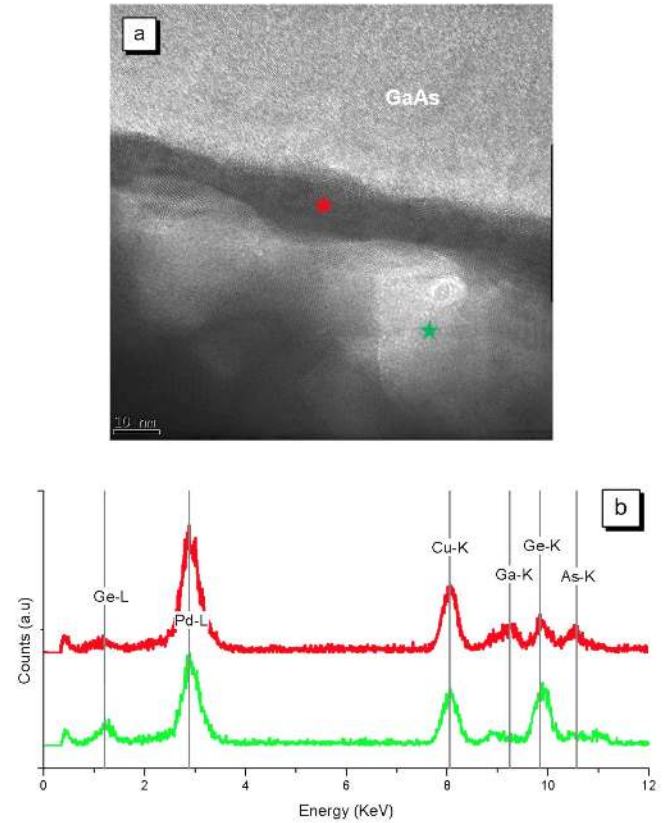


can be given to this phenomenon. Further investigations are underway to clarify this issue.

Regarding to the contact sheet resistivity, as shown in table 2, all the Pd/Ge/Ti/Pd/Ag samples have a metal resistivity around  $1.9 \sim 2.0 \times 10^{-6} \Omega \text{ cm}$ , which is almost the same as Ti/Pd/Ag without RTA and close to the tabulated value for pure bulk Ag ( $1.6 \times 10^{-6} \Omega \text{ cm}$ ). It should be also noted that these resistivities are, on average, around 20% lower than those measured in the Ti/Pd/Ag and Pd/Ti/Pd/Ag systems (see table 1). We link this result to the fact that the Pd/Ge contact demands lower RTA temperatures ( $300^\circ\text{C}$  versus  $400^\circ\text{C}$ ) and therefore contamination of the top metal layer is less likely to occur.

In order to further analyze the microstructure of Pd/Ge/Ti/Pd/Ag contacts and understand the mechanism behind the low contact resistance and low metal resistivity obtained, TEM and EDX measurements were carried out. Figure 3(a) shows a cross section TEM image of sample *c* in table 2. The chemical composition along the metallization has been measured by means of EDX in figure 3(b), where the spectra for the different layers identified in figure 3(a) are gathered. The main elements detected in EDX—from left to right in figure 3(b), Ge, S, Pd, Ag and Ti—are marked in the plot in order to compare the composition throughout the contact. As shown in figures 3(a) and (b), a uniform dark 50 nm layer (marked with a blue star) is clearly visible in the middle of the image. This band can be identified as a Pd layer, according to the corresponding EDX spectrum (blue line in figure 3(b)). The EDX analysis of the layer above (red line figure 3(b)) shows an Ag layer with no signal of Ga or As (Ga and As should appear at energies  $>9 \text{ keV}$ , not shown in the figure to enlarge the relevant portion of the x-axis). This proves that neither Ga nor As significantly diffuse into the top Ag layer, and thus indicates that the Ti/Pd barrier layer works fine with the moderate RTA process needed for this contact ( $300^\circ\text{C}$  and 20 s). The sulfur detected in the Ag layer is related to contamination from the laboratory ambient where the samples were stored (i.e. silver sulfuration). In the green star region of figure 3(a), below the Pd layer, the Ti layer is identified. Some amount of Ag is also detected in this layer illustrating that Ag diffuses into the Ti layer to a certain extent. However, in the next layer (orange star region), only Pd and Ge are measured by the EDX analysis (no significant Ag is detected), suggesting that a 50 nm Ti layer is thick enough to stop the Ag diffusion for our RTA conditions. Finally, in the contact region close to the interface (marked with an orange star) both Pd and Ge are detected evidencing a clear intermixing of the Pd and Ge layers.

The interface of Pd/Ge/Ti/Pd/Ag has been measured in more detail by high resolution TEM (HRTEM) and key results are summarized in figure 4(a). Additionally, chemical composition has been analyzed by taking punctual EDX spectra (figure 4(b)). The green spectrum in figure 4(b) (corresponding to the location marked with a green star in figure 4(a)) corroborates the existence of a Pd/Ge layer with a very low Ga and As detection. Between the GaAs and the Pd/Ge, a 10 nm darker band is observed (marked with the red star in figure 4(a)). Jong-Lam Lee *et al* [26] reported that this band

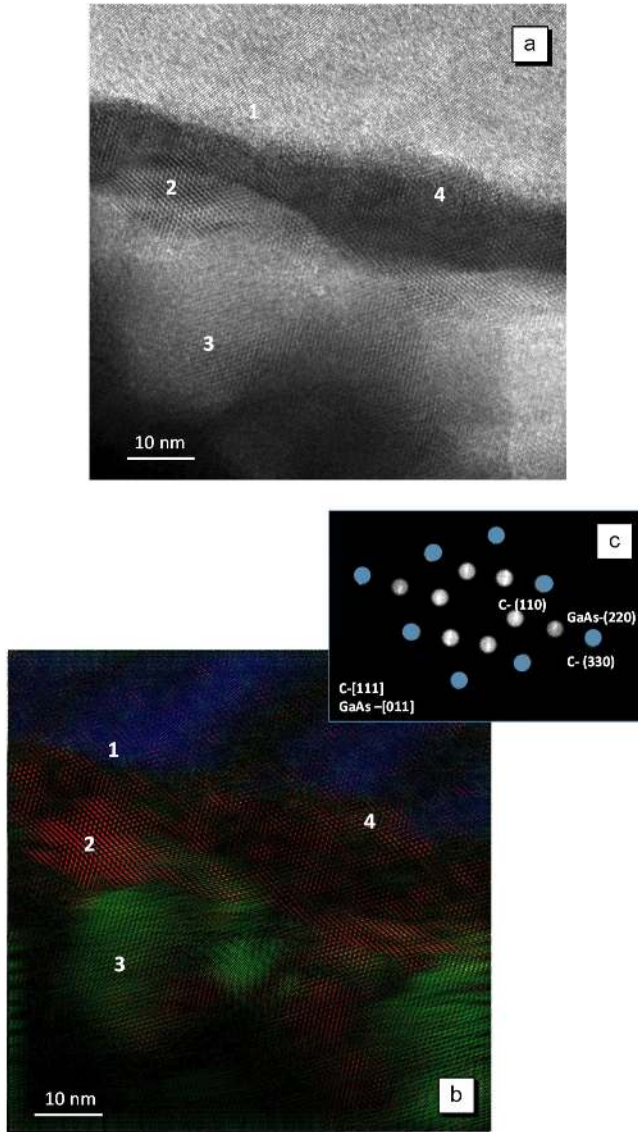


**Figure 4.** (a) Cross-sectional high resolution TEM image of the contact interface region of sample *c* in table 2; (b) EDX spectrum taken from the points marked with a red and green star. The star and the related EDX spectrum share the same color. The peaks corresponding to the main elements detected in EDX—from left to right Ge, Pd, Cu, Ga, Ge and As—are labeled in the plot as vertical lines. The leftmost unlabeled peak corresponds to adventitious C contamination. The presence of Cu represents contamination from the sample holder.

was a Ge epitaxial layer resulting from the regrowth of excess amorphous Ge. However, our EDX results show that this layer contains Pd, Ga, Ge and As (red spectrum in figure 4(b)) that can be associated to a  $\text{Pd}_4(\text{GaAs}, \text{Ge}_2)$  intermediate compound previously reported [36]. The detection of Cu in this spectrum should be neglected since it is attributed to contamination from the sample holder.

Figure 5(a) shows a higher magnification HRTEM image of the interface. This figure was analyzed with the inverse fast Fourier transform method (FFT) and the different crystallographic structures have been identified and isolated. Figure 5(b) represents a false color plot of figure 5(a) that synthesizes this crystallographic information by assigning a different color to a range of nearest-neighbor bond lengths. Different control points, labeled from 1 to 4, have been added in both figures 5(a) and (b) to identify different regions. The blue zone in figure 5(b) (i.e. point 1) corresponds to the zone axis [011] of the GaAs substrate. The green region (i.e. point 3) is formed by interplanar distances of 0.275 nm and 0.224 nm that can be associated with the planes (111) and (110) of compound  $\text{Ge}_{0.4}\text{Pd}_{3.6}$  whose lattice constant is 0.388 nm [37]. This hypothesis is coherent with the EDX





**Figure 5.** (a) Cross-sectional high resolution TEM image of the contact interface region sample *c* of table 2; (b) false color plot of figure 5(a), where different lattice parameters are indicated with different colors; (c) diffraction pattern of points 1 and 4 (details in the text).

spectrum present in figure 4(b), where we observed GePd regions just above the dark band layer. Finally, the FFT associated with the red region (i.e. point 4) is included in figure 5(c) with white dots. In this figure, the prefix C- in the labels has been used to denote the planes related to this compound. Additionally, the resultant FFT for the GaAs substrate (blue region in figure 5(b)) has been overlapped in the same figure 5(c) using blue dots, and has been labeled as GaAs-[011] zone axis. The analysis of this diffraction pattern suggests that the interfacial layer is a cubic structure oriented in the [111] zone axis with a calculated lattice parameter of around 0.848 nm. As it has been mentioned, this layer is most likely Pd<sub>4</sub>(GaAs, Ge<sub>2</sub>) as previously reported, since the EDX spectra measured at that position revealed the existence of Ge, Pd, As and Ga. It is important to note that the plane (330) of this layer has almost the same value as GaAs (220), i.e.:

$d = 0.199$  nm. This fact is an evidence of epitaxial growth, as can be observed in the HRTEM of figure 5(b).

In the literature several interpretations exist to explain the good performance of Pd/Ge contacts on GaAs. In short, after the initial formation of intermediate PdGaAs phases, Ge diffuses inwards and drives the decomposition of PdGaAs forming GePd [21, 24–26, 36]. At this point, some works claim that the excess Germanium creates an epitaxial layer on the GaAs (i.e. the low specific contact resistance is the result of surface bandgap lowering by Ge) [21, 25]; other works suggest that excess Ge and dissolution of the PdGaAs phases create a newly regrown epitaxial n<sup>++</sup>GaAs layer (i.e.  $\text{Ge} + \text{PdGaAs} \rightarrow \text{n}^{\text{++}}\text{GaAs} + \text{GePd}$ ) [24, 36]; and finally other works simply claim that excess Ge diffuses into the GaAs to form a n<sup>++</sup> superficial layer, which is responsible for the low specific contact resistance by enhancing the tunneling probability [24, 26].

However, in our HRTEM analysis we did not detect any Ge epitaxial layer at the contact interface. The complete decomposition of the PdGaAs phase was not confirmed either. On the contrary, we found an epitaxial Pd<sub>4</sub>(GaAs, Ge<sub>2</sub>) compound in contact to GaAs. Therefore, our vision of the process is that this intermediate compound changes the metal/semiconductor barrier height, and thus a very low specific contact resistance is obtained due to the enhanced tunneling. It is also likely that a very thin n<sup>++</sup>GaAs region exists at the interface since out-diffusion of Ga and As into the Pd<sub>4</sub>(GaAs, Ge<sub>2</sub>) will favor the in-diffusion of Ge into the semiconductor.

## Summary and conclusion

Pd/Ti/Pd/Ag and Pd/Ge/Ti/Pd/Ag metallizations on n-GaAs have been studied in the quest for a metal system that can provide (1) low metal/semiconductor specific contact resistance; (2) high metal conductivity; (3) long term stability; (4) good bondability; and (5) low cost (as compared to other systems including Au or Pt). This study has been conducted analyzing the contact performance on n-GaAs of various doping concentrations and comparing the results to Ti/Pd/Ag and Au/Ge/Ni/Au metal systems.

The results show that Pd/Ti/Pd/Ag produced Schottky-like *I-V* curves when deposited on n-GaAs doped  $3 \times 10^{18} \text{ cm}^{-3}$ , and became ohmic when deposited on n-GaAs doped  $1.6 \times 10^{19} \text{ cm}^{-3}$ . In this latter case, the metal-semiconductor specific contact resistance obtained was of  $\rho_c \sim 10^{-4} \Omega \text{ cm}^2$ , an order of magnitude lower than those formed using Ti/Pd/Ag ( $\rho_c \sim 10^{-3} \Omega \text{ cm}^2$ ). On the other hand, Pd/Ge/Ti/Pd/Ag contacts achieved a very low  $\rho_c = 1.5 \times 10^{-7} \Omega \text{ cm}^2$  with  $N_D = 2.5 \times 10^{18} \text{ cm}^{-3}$ . Microstructure analysis on the contact interface carried out by HRTEM reveals that this low specific contact resistance can be attributed to the formation of a Pd<sub>4</sub>(GaAs, Ge<sub>2</sub>) compound in contact to n<sup>++</sup>GaAs.

In terms of metal sheet conductance, both Pd/Ti/Pd/Ag and Pd/Ge/Ti/Pd/Ag contacts on n-type GaAs present very low sheet resistivities. In fact, our measurements show that



the metal resistivities of both layers ( $\rho_M \sim 2 \times 10^{-6} \Omega \text{ cm}$ ) are similar to the values attained with Ti/Pd/Ag and quite close that of pure bulk Ag. These results indicate that the presence of Pd or Pd/Ge as interfacial layers does not significantly interfere with the diffusion barrier effect of Ti/Pd, which prevents the contamination of Ag from GaAs diffusion. This was confirmed by the microstructure analysis by TEM and EDX, that measured negligible contamination of Ga and As into the Ag layer and vice versa.

Finally, as compared to the classic AuGe/Ni/Au contact layer, Pd/Ge/Ti/Pd/Ag shows a factor 20 lower specific contact resistance and one order of magnitude lower metal resistivity. As a result of such good electrical properties, plus its cost-effectiveness, it exhibits high potential for applications on electronic devices that need to handle high current densities.

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