

## Computational Design of Thermally Activated Delayed Fluorescence Materials: The Challenges Ahead

Yoann Olivier, Juan Carlos Sancho-García, Luca Muccioli, Gabriele D'Avino, and David Beljonne

*J. Phys. Chem. Lett.*, **Just Accepted Manuscript** • DOI: 10.1021/acs.jpcllett.8b02327 • Publication Date (Web): 28 Sep 2018

Downloaded from <http://pubs.acs.org> on September 29, 2018

### Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

# Computational Design of Thermally Activated Delayed Fluorescence Materials: The Challenges Ahead

Y. Olivier,<sup>1</sup> J.-C. Sancho-Garcia,<sup>2</sup> L. Muccioli,<sup>3</sup> G. D'Avino,<sup>4</sup> D. Beljonne<sup>1</sup>

<sup>1</sup>Laboratory for Chemistry of Novel Materials, University of Mons, Place du Parc 20, B-7000 Mons, Belgium.

<sup>2</sup>Departamento de Química Física, Universidad de Alicante, E-03080 Alicante, Spain.

<sup>3</sup>Dipartimento di Chimica Industriale "Toso Montanari", Università di Bologna, I-40136 Bologna, Italy, and Institut des Sciences Moléculaires, UMR 5255, University of Bordeaux, F- 33405 Talence, France.

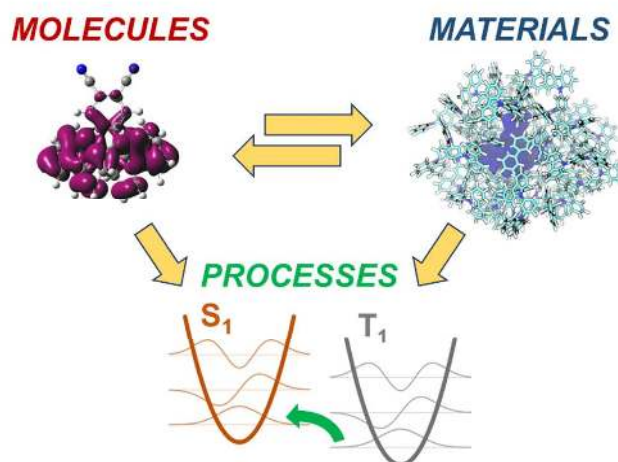
<sup>4</sup>Institut Néel, CNRS and Grenoble Alpes University, F-38042 Grenoble, France.

**Corresponding author:** David.beljonne@umons.ac.be

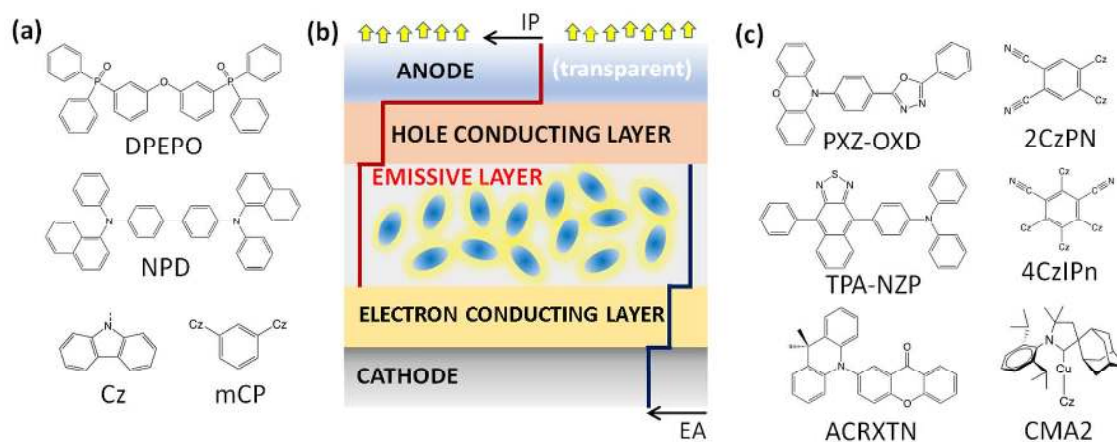
## Abstract

Thermally activated delayed fluorescence (TADF) offers the premise for all-organic light emitting diodes with quantum efficiencies competing those of transition metal-based phosphorescent devices. While computational efforts have so far largely focused on gas-phase calculations of singlet and triplet excitation energies, the design of TADF materials requires multiple methodological developments targeting among others a quantitative description of electronic excitation energetics, fully accounting for environmental electrostatics and molecular conformational effects, the accurate assessment of the quantum-mechanical interactions that trigger the elementary electronic processes involved in TADF, as well as a robust picture for the dynamics of these fundamental processes. In this perspective, we describe some recent progress along those lines and highlight the main challenges ahead for modeling, which we hope will be useful to the whole TADF community.

## TOC Figure: Multiscale TADF molecular-based design



Organic light-emitting diodes (OLEDs) have emerged as a mature technology, reaching commercial applications in lighting and full-color displays. Delayed emission with characteristic spectra coinciding with prompt fluorescence but differing in the emission lifetimes has been observed in eosin solutions and has since been known as E-type fluorescence<sup>1</sup>. This is also referred to as Thermally Activated Delayed Fluorescence (TADF), a process that offers the premise to boost internal quantum efficiency (IQE) of electroluminescence beyond the 25% spin statistical limit. The demonstration of TADF in OLEDs has shifted the material design paradigm from phosphors containing rare and expensive transition metals towards all-organic compounds with reduced singlet-triplet exchange interactions.



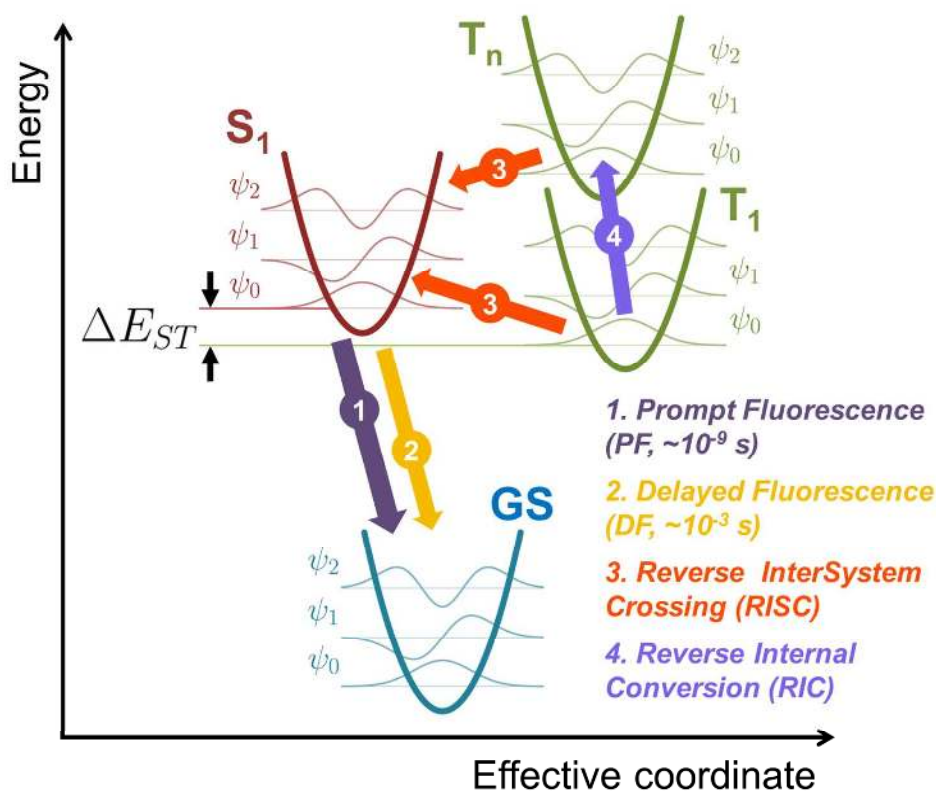
**Figure 1:** (a) Chemical structures of some relevant host materials: Bis[2-(diphenylphosphino)phenyl] ether oxide (DPEPO), N,N'-Bis(naphthalen-1-yl)-N,N'-bis(phenyl)benzidine, (NPD) and 1,3-Di(9H-carbazol-9-yl)benzene, 9,9'-(1,3-Phenylene)bis-9H-carbazole (mCP). Chemical structure of the carbazole (Cz) group. (b) Minimal sketch of a TADF-doped OLED, where electron and hole conducting layers serve also as hole and blocking layers, respectively, and as exciton blocking layer. Dark blue and red lines indicate typical levels for electron affinities and ionization potentials. This perspective focuses only on the emissive layer, where electron and holes recombine to form an exciton. The layer is constituted of a hole and electron conducting matrix (host), containing a few percent of TADF emitters (guest). The ideal host should possess EA (IP) values lower (higher) than the emitter. (c) Chemical structures of emitters discussed in the text: phenoxazine-2,5-diphenyl-1,3,4-oxadiazole (PXZ-OXD), 4,5-di(9H-carbazol-9-yl)phthalonitrile (2CzPN), N,N-diphenyl-4-(9-phenylnaphtho-[2,3-c][1,2,5]-thiadiazol-4-yl)aniline (TPA-NZP), 2,4,5,6-Tetra(9H-carbazol-9-yl)isophthalonitrile (4CzIPn), 3-(9,9-dimethylacridin-10(9H)-yl)-9H-xanthen-9-one (ACRXTN) and Cyclic (alkyl)(amino)carbene ligands copper-carbazole (CMA2).

1  
2  
3 TADF emission is triggered by Reverse InterSystem Crossing (RISC) from the non-radiative triplet  
4 'reservoir' states to radiative singlet states<sup>2</sup>, a process that is facilitated by a small energy splitting  
5  $\Delta E_{ST}$  between the lowest singlet ( $S_1$ ) and triplet ( $T_1$ ) excited states, and possibly assisted by the  
6 manifold of higher-lying triplet states ( $T_n$ , see Figure 2)<sup>3-5</sup>. Hence, both prompt (from the lowest  
7 singlet) and delayed (from upconverted triplets) fluorescence decay channels add up in converting  
8 into light potentially all electrically generated excitons, irrespective of spin. The most common design  
9 strategy of TADF emitters consists in partitioning hole and electron densities over different spatial  
10 regions via electron donating (D) and accepting (A) units, often connected in a twisted conformation,  
11 hence reducing exchange interactions splitting singlets from triplets<sup>6-9</sup>. In actual cases, the excited  
12 states involved in TADF often turn out to be hybrid mixtures of charge transfer (CT) and local  
13 excitation (LE) diabatic states, with the amount of mixing prompted by vibronic coupling<sup>10,11</sup>. We  
14 refer the interested reader to a recently published review<sup>12</sup> for a critical discussion of the chemical  
15 design rules that have emerged so far through the fruitful interplay between theoretical and  
16 experimental chemists. Despite some progress has been achieved, molecular design has largely relied  
17 on the calculation of excitation energies from the optimized ground-state geometry using time-  
18 dependent density functional theory (TD-DFT), either for isolated molecules or assuming a  
19 continuum embedding (as for instance described using polarizable continuum models, PCM)<sup>13-15</sup>. The  
20 sole criterion for the selection of potential candidates for TADF is then the calculated exchange gap,  
21  $\Delta E_{ST}$ , vertical or adiabatically, between the lowest singlet and triplet excitations. Besides raising a  
22 number of important technical questions regarding the accuracy of the predictions, this simplified  
23 view of the TADF mechanism might considerably bias the engineering rules, as it neglects potentially  
24 important effects such as the specifics of intramolecular conformation and intermolecular (host-  
25 guest) interactions on spin conversion dynamics in the solid state. This perspective aims at bringing a  
26 comprehensive view on the modeling of TADF emitters and at highlighting the challenges ahead. For  
27 being predictive, we argue that TADF-oriented computational models should include in an integrated  
28 framework<sup>10</sup>:

- 29 1) An accurate description of the singlet and triplet excited-state manifolds in terms of their  
30 relative energy and detailed nature, including coupling to relevant vibrational degrees of  
31 freedom;
- 32 2) A reliable evaluation of spin-orbit and hyperfine couplings mediating (reverse) singlet-triplet  
33 intersystem crossing;
- 34 3) A proper embedding scheme to account for both structural and electrostatic effects in the  
35 emitting layer;

4) A solver for excited-state dynamics that includes electron-phonon and spin-mixing interactions on an equal footing.

In the following, the needed methodologies to reach these ambitious targets are critically reviewed and discussed, going all the way from a quantum-chemical description of the molecular TADF building blocks to the simulation of the corresponding solid-state materials using combined quantum-classical methods.



**Figure 2 :** Illustration of the electronic states involved in TADF, i.e. ground (GS), singlet (S<sub>0</sub>) and triplet (T<sub>0</sub>, T<sub>n</sub>) excited states along an effective vibrational coordinate. The elementary steps leading to prompt (PF) and delayed (DF) fluorescence are illustrated as arrows. Low-quanta vibrational wavefunctions, which promote reverse intersystem crossing (RISC) and reverse internal conversion (RIC), are also shown.

1  
2  
3 Generally speaking, the set of molecules acting as hosts or emitters share a common feature, that is,  
4 a relatively large size compared to those molecules constituting the datasets used for historically  
5 benchmarking *ab initio* methods<sup>16–18</sup>. We focus here on modern variants of *ab initio* theories for  
6 excited states, particularly for calculating vertical transition energies to the lowest excited-state of  
7 singlet and triplet spin symmetry,  $\Omega(S_1)$  and  $\Omega(T_1)$  respectively, also applicable to the whole manifold  
8 of  $S_n$  and  $T_n$  states. Although in principle superior to all other amenable approaches, due to the  
9 unfavorable scaling with system size ( $N$ ) of the family of methods based on iterative coupled-cluster  
10 (CC) equations ( $\Omega$ ), their application to TADF compounds have been severely limited and only recently  
11 some examples applying the hierarchy of CC-based methods to small molecules have appeared<sup>19</sup>.  
12 Actually, for CC-based methods, the hierarchy is  $CC2 (N^5) < CCSD (N^6) < CC3 (N^7) < CCSDT (N^8)$ ,  
13 concerning both computational scaling with system size ( $N$ ) and expected accuracy. Note also that  
14  $CC2 (CC3)$  is an approximation to  $CCSD (CCSDT)$  that has been developed with the prediction of  
15 excited-state energies as the main focus, and that all these variants are based on linear-response  
16 theory. On the other hand,  $EOM-CCSD (N^6)$  is based on a slightly different theoretical frame, though  
17 it leads to similar excitation energies (yet different transition dipole moments).  
18  
19

20  
21  
22  
23  
24  
25  
26  
27 The accuracy of the methods is very high, e.g.  $CCSDT$  excitation energies are within a few tens of meV  
28 w.r.t. FCI, but they are only amenable to very small molecules, hence not applicable for most TADF  
29 chromophores. Yet, simplified versions of second-order wavefunction methods, and thus with a  
30 reduced formal scaling of  $O(N^5)$  with respect to both canonical equations and higher-order methods  
31 such as  $CCSDT$ , are available. We mention, among them, linear-response  $CC2$  and  $ADC(2)$  as emerging  
32 cost-effective and accurate methods<sup>20</sup>. Note that in the case of  $CC2$ , due to the fact that the full  
33 expression for double amplitudes is retained only at first-order, the method behaves better for single  
34 excitations and tends to overestimate excitations energies for doubly excited states. The  $ADC(2)$   
35 method suffers from the same inherent limitation for double excitations, and the corresponding  
36  $ADC(3)$  is too costly for routine calculations, although it leads to reliable benchmarks. However, the  
37 extended  $ADC(2)-x$  version provides an overall improvement for the description of excited states with  
38 double-excitation character. An interesting compromise between accuracy and computational cost is  
39 provided by the spin-component-scaling (SCS-) $CC2$  method, through the introduction of scaling  
40 factors to the same-spin and opposite-spin contributions of the second-order correlation energy,  
41 leading thereby a better performance for excited states<sup>21</sup>. There are other methods that allow, by  
42 construction,  $n$ - or multi-electron excitations, that is by promoting up to  $n$  electrons into an active  
43 window of  $m$  molecular orbitals, as in RAS- or CASSCF/CASPT2 methods. However, computational  
44 limitations preclude the use of large  $(n,m)$  active spaces ideally including all relevant  $\pi$  and  $\pi^*$   
45 orbitals. We finally mention many-body Green's function methods such as the  $GW$  plus Bethe-  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57

1  
2  
3 Salpeter equation (BSE) formalisms, also explored recently within this context with some success<sup>22,23</sup>.  
4 We shall, however, remark that full BSE suffers the issue of triplet instability, much as TD-DFT does<sup>24</sup>  
5 (see below).  
6  
7

8 As other possible caveats for these calculations, we here mention: (i) the very costly calculation of  
9 excited-state geometries and adiabatic values for  $\Omega(S_1)$  and  $\Omega(T_1)$ , which are important figures when  
10 dealing with light emission; and (ii) the marked dependence (i.e. slow convergence) with basis sets  
11 size expected for these methods and the possible need to include diffuse functions. This has  
12 historically prompted the use of more cost-effective methods, despite the fact that the  
13 computational efforts needed for *ab initio* calculations can be reduced by using resolution-of-the-  
14 identity<sup>25,26</sup>, or density-fitting techniques<sup>27</sup>. To conclude, so far, wavefunction-based methods are not  
15 routinely used for TADF applications and TD-DFT is usually preferred. However, they offer a robust  
16 framework to tackle challenging issues, such as assessing the contribution to RISC of higher-lying  
17 singlet and triplet excited states<sup>26</sup> or modeling multiple resonance effects<sup>28,29</sup>.  
18  
19  
20  
21  
22  
23

24 Due to the need to handle large molecular sizes and to achieve a good trade-off between accuracy,  
25 scaling, and feasibility of the calculations for the fast screening of compounds, (Time-Dependent)  
26 Density Functional Theory, (TD-)DFT methods are by far the most employed computational tools in  
27 the TADF community. However, contrarily to what was initially thought, the reliability of the results  
28 depends not only on the functional choice for the time-dependent part, but also on the whole  
29 computational protocol employed including geometrical and basis sets issues, as well as on the  
30 chemical nature of the target molecule. We focus next on excited-state properties of emitters, for  
31 which a relatively wide body of information is available from the recent literature.  
32  
33  
34  
35  
36  
37

38 Earlier calculations combined the use of the B3LYP functional with a moderate basis set such as 6-  
39 31G\* to obtain ground-state geometries and dissect the spatial shape and energy location of the  
40 frontier molecular orbitals<sup>30,31</sup>. When applied to the calculation of vertical transition energies to the  
41 lowest excited state of singlet and triplet spin symmetry and their difference dubbed as  $\Delta E_{ST}^V$ , the  
42 results were found to critically depend on the weight, or proportion, of the exact-like exchange  
43 introduced into the hybrid functional (typically ranging from 5 to 40%, namely 20% for the widely  
44 used B3LYP model)<sup>32</sup>. Note that any general protocol should give accurate excitation energies for the  
45 lowest singlet and triplet excited states, without relying on any error cancellation and without  
46 limiting the focus to the magnitude of  $\Delta E_{ST}^V$  as unique target. Namely, a system- and state-dependent  
47 procedure should be avoided, as extension to excited states beyond  $S_1$  and  $T_1$  might be problematic.  
48 Since organic molecules are known to benefit from higher-than-defaults (e.g. default values for  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3 B3LYP and PBE0 are 20% and 25% respectively) weights of the exact-like exchange, other functionals  
4 like M06-2X have also been used with some success up to now<sup>33</sup>.  
5

6  
7 A major breakthrough was found after imposing the Tamm-Dancoff approximation (TDA) for the  
8 solution of the full TD-DFT equation<sup>34</sup>, which improves the accuracy of routine calculations. This is  
9 especially true for  $T_1$  states where TDA cures for the triplet instability problem and, as a result, yields  
10 improved  $\Delta E_{ST}^V$  values. Note that the Thomas-Reiche-Kuhn sum rule for oscillator strengths (i.e. the  
11 sum of all oscillator strengths from a particular state to all other equals the number of electrons in  
12 the system) is no longer fulfilled with TDA-DFT, as it happened with CIS too. That violation can lead to  
13 inaccurate oscillator strength distribution, and thus precludes a state-by-state quantitative  
14 comparison of oscillator strength values between theories fulfilling (i.e. TD-HF and TD-DFT) or  
15 violating the rule. The PBE0 functional was also applied with some correction for dispersion, i.e.  
16 D3(BJ), intended mostly to provide more accurate ground- and excited-state geometries after  
17 including non-covalent (intramolecular) effects, and used with large basis sets, i.e. def2-TZVP, to  
18 estimate  $\Omega(S_1)$  and  $\Omega(T_1)$  energies at the (nearly) complete basis set limit. However, while accuracy  
19 reaching 0.1-0.2 eV can be achieved for  $\Delta E_{ST}^V$  values<sup>13</sup>, studies using this computational protocol have  
20 not yet been extended beyond  $S_1$  and  $T_1$  states, and thus further efforts are still needed in this  
21 direction.  
22  
23  
24  
25  
26  
27  
28  
29  
30

31 Another strategy is provided by range-separated hybrid functionals, i.e. CAM-B3LYP or  $\omega$ B97X as  
32 paradigmatic examples, in which the range-separation parameter may be fine-tuned for each  
33 compound<sup>14,35</sup>, as well as for isolated or host-embedded emitters. In these range-separation models,  
34 the electron-electron interaction is split into two contributions, short- and long-range, treating each  
35 one at a different theoretical level (short-range often with a semi-local GGA exchange functional and  
36 long-range with exact-like exchange to obtain the desired correct asymptotic behavior). The first  
37 mention in literature was due to A. Savin et al. in 1997,<sup>36</sup> but for the coupling of multiconfigurational  
38 ab initio with DFT methods, with probably the first application to GGA exchange functionals by K.  
39 Hirao et al. in 2001,<sup>37</sup> and popularized later by N.C. Handy et al. with the CAM-B3LYP method.<sup>38</sup> Since  
40 the tuning of the  $\omega$  parameter aims at accurately reproducing one-electron attachment/detachment  
41 energies, it also naturally leads to accurate HOMO-LUMO gaps and corresponding excited-state  
42 energies (relying heavily on that energy difference, to first order). However, more work is still needed  
43 to confirm the accuracy of these methods for higher-lying singlets and triplets.  
44  
45  
46  
47  
48  
49  
50

51  
52  $S_1$  and  $T_1$  optimized excited-state geometries are now routinely computed from linear-response  
53 TD(A)-DFT. However, different options such as the use of UKS (for  $T_1$ ) or ROKS (for  $S_1$  and  $T_1$  states)  
54 exist<sup>39</sup>. Still, we stress that mixing different levels of theory such as TD(A)-DFT (for  $S_1$ ) and UKS (for  $T_1$ )  
55  
56  
57  
58  
59  
60



1  
2  
3 can lead, in some instances, to spurious negative values of  $\Delta E_{ST}^{14,40}$ . The fast screening of TADF  
4 molecules would also benefit of low-cost TD-DFT based methods, such as e.g. sTDA-DFT or sTDA-  
5 xTB<sup>41</sup>. It would also be desirable to extend these methods beyond one-electron effects. The use of a  
6 (D)-like correction introduced by double-hybrid methods, namely a MP2-like term added non self-  
7 consistently to the standard TD-DFT treatment, or the coupling of DFT with MRCI (DFT/MRCI), are  
8 among the envisioned possibilities, although the price to pay is a higher formal scaling,  $O(N^5)$  in the  
9 case of double-hybrid functionals. While double-hybrid models appear robust for  $\Omega(S1)$  values, the  
10 lack of implementations for the calculation of the corresponding  $\Omega(T1)$  values has precluded so far  
11 the evaluation of  $\Delta E_{ST}^V$ <sup>42</sup>. DFT/MRCI has been recently extended to deal with higher-order  
12 excitations in multichromophoric systems<sup>43</sup>, opening new possibilities for exploring light-emission  
13 mechanisms for which both singlet and triplet states are of importance, as well as their possible  
14 modulation by environmental effects<sup>44</sup>.

21 The need for quantifying the nature of the excited states involved in the TADF process has led to the  
22 development of a set of metrics able namely to gauge the CT versus LE character of the excitations,  
23 which is critically entangled with the molecular geometry and its fluctuations around  
24 equilibrium<sup>10,43,45</sup>. The existing metrics can be classified into two groups, those based on molecular  
25 orbitals, be them KS or NTO, and those based on density differences. Note that the metrics are  
26 intended to be: (i) generally applicable, in the sense that they can be applied to any of the desired  
27 singlet or triplet excited states; and (ii) easily transferable, with the results not expected to heavily  
28 depend on the functional and/or basis set choice. However, the use of relaxed and unrelaxed density  
29 might introduce some difference from a quantitative standpoint<sup>46</sup> as it was recently disclosed.

36 Table 1 summarizes the observables developed so far and their typical values for the limiting cases of  
37 pure CT and LE. As regards the first category of tools, the metrics  $\Lambda$  roughly measures the overlap  
38 between a pair of occupied (*i*) and virtual (*a*) orbitals involved in the ground-to-excited state  
39 transition<sup>47,48</sup>; other approaches, such as  $\Delta r$ <sup>49,50</sup>, aim at giving an effective hole-electron separation  
40 during the excitation. On the other hand, for density-based descriptors, the  $D_{CT}$  index is based on the  
41 barycenters of densities associated with an electronic transition<sup>51</sup>, while  $\Phi_S$  refers to the overlap  
42 between the attachment/detachment densities, that is, the electron density removed/rearranged  
43 during the excitation<sup>52</sup>. In some cases, due to a spurious behavior of density functionals for CT states,  
44 low-lying but unreal (intruder) CT states can appear, which can be also identified and discarded on  
45 the basis of the evaluation of intermolecular electrostatic interactions, e.g. by the  $M_{AC}$  metrics<sup>53</sup>.  
46 Despite the appearance of a few topological metrics in the last years, there is also convincing  
47 evidence that these are highly correlated, as recently shown for a large sample of molecules  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

displaying CT excitations,<sup>54</sup> which should facilitate comparison between different existing or future studies.

**Table 1.** Summary of the metrics most used and their limiting values in the case of pure CT and LE excitations.

Metrics	Description	Distinction between the nature of excited-states	
		CT	LE
$\Lambda_{ia}$	Overlap between the <i>ia</i> pair of the norms of molecular orbitals	$\Lambda \sim 0$	$\Lambda \sim 1$
$\Delta r$	Coefficient-weighted hole-electron distance between a set of orbital centroids	$\Delta r > 2 \text{ \AA}$	$\Delta r < 2 \text{ \AA}$
$D_{CT}$	Distance between barycenters of density variations (average between the corresponding barycenters)	$t - D_{CT} > 1.6 \text{ \AA}$	$t - D_{CT} < 1.6 \text{ \AA}$
$\Phi_s$	Normalized overlap between attachment/detachment densities	$\Phi_s \sim 0$	$\Phi_s \sim 1$

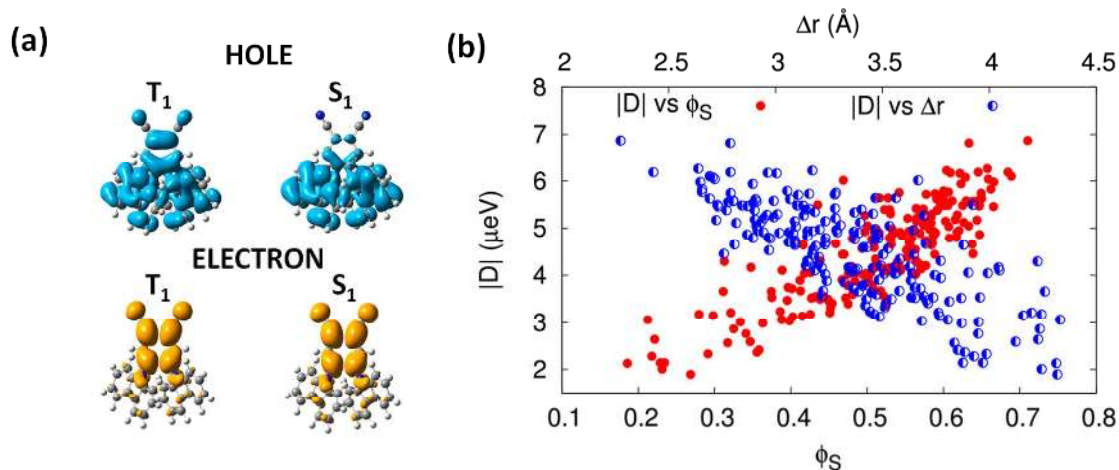
Metrics based exclusively on orbitals raise some concern. For instance, many-electron excited states might involve multiple pairs of orbitals, thus casting doubt on the meaning of the  $\Lambda$  value. Another example is the case of a symmetric D-A-D triad, where the hole and electron centroids can spuriously occupy similar spatial positions, hence leading to a vanishing value of  $\Delta r$ , despite a strong CT excited-state character.  $D_{CT}$  and  $\Phi_s$  overcome these limitations and appear as more universal metrics<sup>55</sup>.

Interestingly, the use of these metrics has clearly evidenced that the excited states involved in the TADF process are neither full CT or LE, as abusively claimed when discussing the photophysics of TADF emitters<sup>3,5</sup>, but most often feature a rather mixed CT-LE character<sup>7,10,56</sup> (see Figure 3a). The amount of CT-LE mixing is controlled by the magnitude of the D-A electronic coupling, which is in turn governed by structural parameters (such as the torsion angle between the two moieties)<sup>7,10</sup>. While the  $S_1$  excited state often displays a large CT character (small  $\Phi_s$ <sup>7</sup> and large  $\Delta r$ <sup>13</sup>, see Figure 3a), especially in conformations with near orthogonal D and A moieties<sup>6</sup>, this is less true for the  $T_1$  excited state. This arises from exchange interactions that stabilize localized triplets more than their singlet counterparts, thereby prompting a more intimate LE-CT mixing in the triplet manifold<sup>7,57</sup>.

From a practical point of view, the metrics presented in table 1 have been applied as a tool to rationalize the  $\Delta E_{ST}$  and oscillator strength values in TADF compounds. Especially, it has been evidenced that the nature of  $T_1$  is the limiting factor in order to minimize  $\Delta E_{ST}$ . The nature of  $T_1$  can

be directly probed experimentally by extracting the Zero-Field Splitting (ZFS) parameters from Electron Spin Resonance (ESR) spectroscopy studies. The dominant dipole-dipole component to the ZFS parameters is indeed inversely proportional to the third power of the interspin distance<sup>58</sup>. The larger the CT character of the triplet excited state, the larger the effective electron-hole radius and spin-spin distance and the lower the ZFS parameter value. From a computational point of view, the ZFS parameter can be evaluated through unrestricted DFT calculations using EPR-II and EP-III basis sets that are optimized for the computation of hyperfine coupling constants<sup>59</sup>. Interestingly, a recent study of carbazolyl-dicyanobenzene based TADF emitters (2CzPN and 4CzIPN) has demonstrated that ZFS (as calculated at the UKS level) and  $\Phi_S$  (obtained from TDA-DFT) values go in par, offering the possibility to confront experimental and theoretical metrics of the triplet excitations<sup>60</sup> (see Figure 3b). As for singlet excitations, calculated oscillator strengths have been shown to correlate very well to  $\Delta r^{11}$  and overlap metrics such as  $\Lambda_{ig}$ <sup>61</sup> and  $\Phi_S$ <sup>7</sup>.

Finally, we mention that only some of the metrics above can be calculated with common quantum chemistry codes, while others require post-processing after the TD(A)-DFT run. More automated procedures, and further benchmarking, are definitively needed, as well as their integration in large-scale computational codes.



**Figure 3:** (a) Hole and electron densities calculated in the attachment/detachment formalism for  $T_1$  and  $S_1$  excited states for 2CzPN. (b) Absolute value of the  $T_1$  Zero-Field splitting parameter as a function of the overlap between the hole and electron densities  $\phi_S$  and the distance between the hole and electron densities centroids  $\Delta r$ , as calculated in the attachment/detachment formalism for 2CzPN  $T_1$  excited state. Reprinted and adapted with permission from reference<sup>10</sup>. Copyright 2017 by the American Physical Society

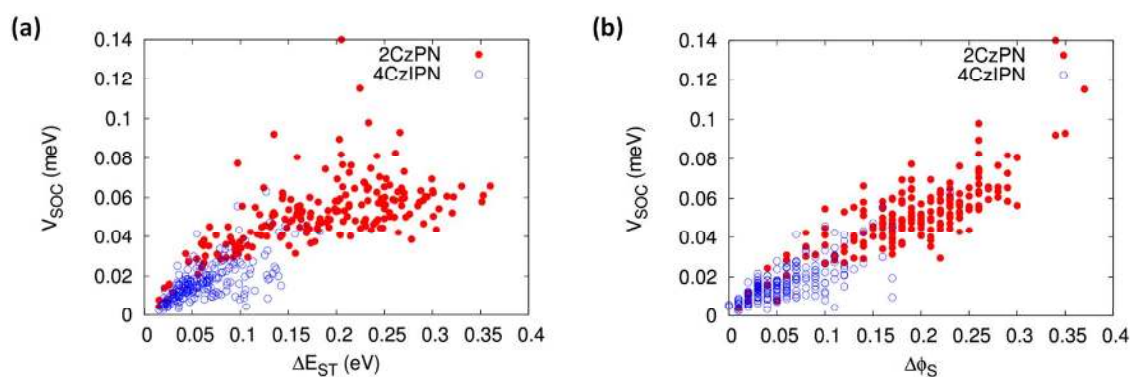
Non-radiative processes are playing a key role in determining the internal quantum efficiency of TADF-based devices. On the one hand, reverse intersystem crossing (RISC) and reverse internal

1  
2  
3 conversion (RIC) are promoting the upconversion from the triplet to the singlet excited-state  
4 manifold. On the other hand, ISC favors the generation of triplets that could subsequently be  
5 recycled to  $S_1$  or, because of their long lifetime, could decay through triplet-triplet or triplet-polaron  
6 annihilation, processes giving rise to device efficiency roll-off at high luminance. Of course, also non-  
7 radiative decays to the ground state from both the lowest singlet and triplet states are competing  
8 pathways reducing IQE. In the following, we distinguish between spin-conserving, namely (R)IC,  
9 triplet-triplet and triplet-polaron annihilation, and spin-non-conserving processes, namely ISC and  
10 RISC. Conformational effects on light emission and ISC and RISC are discussed thoroughly below.

11  
12 Spin-orbit is a relativistic effect that is responsible for mixing orbital and spin degrees of freedom,  
13 thus allowing electronic states of different multiplicities to couple. Spin-orbit coupling naturally arises  
14 from the one-electron Dirac equation. Even though being of a fundamental interest, it cannot be  
15 solved exactly when considering large many-electron systems such as TADF molecules. Practical  
16 applications thus rely on perturbative approaches to non-relativistic electronic structure calculations.  
17 Among the most commonly reported approaches, we distinguish the zero-order relativistic  
18 approximation (ZORA) from the full Breit-Pauli Hamiltonian (including relativistic mass corrections  
19 and spin-orbit effects) and its mean-field approximation that are implemented in a number of  
20 softwares based on CASSCF<sup>62</sup>, TD-DFT and DFT-MRCI<sup>43</sup> excited-state descriptions. The interest in spin-  
21 orbit coupling calculations in the field of OLEDs dates back from the emergence of phosphorescent  
22 emitters containing heavy metal centers allowing for rapid ISC from  $S_1$  to  $T_1$  and radiative decay from  
23  $T_1$  to the ground state (usually in the microsecond timescale regime). Because of the presence of  
24 these heavy atoms, spin-orbit coupling is usually large in phosphorescent emitters allowing for strong  
25 mixing between singlet and triplet excited states, while it is relatively weak (on the order of tenths or  
26 hundredths of meV) for TADF emitters made only of light elements. Still, the RISC mechanism is  
27 expected to be mainly driven by spin-orbit coupling with a small contribution of hyperfine coupling,  
28 as evidenced by electron paramagnetic resonance measurements<sup>9</sup>. This is further supported by  
29 transient electron spin resonance (TrESR) on 2CzPN and 4CzIPN carbazoyl-dicyanobenzene based  
30 TADF emitters, for which the observed absorption and emission patterns are characteristic of a spin-  
31 orbit-driven ISC mechanism. Up to now, a systematic comparison between the different software  
32 and models has not been carried out yet, although it was shown that the full Breit-Pauli model and  
33 its mean field approximation lead to very similar spin-orbit coupling matrix elements for organic  
34 dyes<sup>63</sup>. Already in the 60s, Mostafa El-Sayed<sup>64</sup> highlighted the fact that spin-orbit coupling is  
35 vanishing when (R)ISC takes place between excited states with identical  $\pi$ - $\pi^*$  character. This is  
36 because the change in spin angular momentum must be compensated by a corresponding change in  
37 angular orbital momentum so that total angular momentum is conserved. As a result, only excited  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

states with different spatial wavefunctions couple through spin-orbit coupling. A textbook example is benzophenone that sustains  $n\text{-}\pi^*$  and  $\pi\text{-}\pi^*$  electronic excitations coupled through large spin-orbit matrix elements<sup>65</sup>. Recently, this concept was put in practice in the context of TADF materials by associating the magnitude of the spin-orbit coupling to the difference in the nature of these excited states calculated based on the difference of normalized overlap between attachment/detachment densities,  $\Delta\Phi_S$ , between  $S_1$  and  $T_1$ .  $\Delta\Phi_S$  was calculated on molecular conformations taken from an amorphous morphology of pure films of 2CzPN and 4CzIPN. Spin-orbit matrix elements are found to correlate approximately linearly with  $\Delta\Phi_S$ , becoming vanishingly small for  $\Delta\Phi_S=0$ , namely in the case where  $S_1$  and  $T_1$  are nearly degenerate ( $\Delta E_{ST}\sim 0$ ) and exhibit both a strong and identical CT character, in line with the El-Sayed rules. Interestingly, because of its overall larger  $\Delta\Phi_S$  offset, spin-orbit coupling is larger for 2CzPN than for 4CzIPN (see Figure 4b). Most importantly, we note that in the case of the carbazole derivatives studied in <sup>7</sup> spin-orbit coupling and  $\Delta E_{ST}$  have antagonistic evolution, such as a trade-off has to be found in order to maximize the rate of RISC (see Figure 4a).

Most importantly, we note that spin-orbit coupling and  $\Delta E_{ST}$  have antagonistic evolution, such as a trade-off has to be found in order to maximize the rate of RISC (see Figure 3a).



**Figure 4:** Spin-orbit coupling as a function of **a)**  $\Delta E_{ST}$  and **b)**  $\Delta\Phi_S$  in 2CzPN (blue data) and 4CzIPN (red data), as sampled from a realistic amorphous morphology simulated with molecular dynamics. The approximately linear correlation between  $V_{SOC}$  and  $\Delta\Phi_S$  follows from the El Sayed rules, while the relationship between  $V_{SOC}$  and  $\Delta E_{ST}$  highlights the need for a trade-off between different parameters for the optimization of TADF performances. Reprinted with permission from reference<sup>10</sup>. Copyright 2017 by the American Physical Society.

Two different internal conversion mechanisms with contrasting effects are considered here. On the one hand, the reverse IC in the triplet manifold of excited states which is a thermally-activated process, is expected to assist RISC by promoting the formation of higher-lying triplet excited states, from which the conversion to the singlet manifold might occur more efficiently because it is

1  
2  
3 associated with a larger exergonic character. This interconversion channel is expected to compete  
4 with the direct conversion from  $T_1$  to  $S_1$  and features smaller activation energies<sup>66</sup>. On the other  
5 hand, non-radiative recombination to the ground state should take place mainly from  $S_1$  and  
6 contribute as the main monomolecular pathway to molecular excitation loss. Even though the two IC  
7 processes described above involve excited states of different spin multiplicities, the initial and final  
8 states are in both cases coupled through non-adiabatic couplings. The rate of internal conversion has  
9 been calculated in triphenylamine-thiadiazole molecule using a Fermi Golden rule formalism and  
10 considering non-adiabatic couplings for all relevant vibrational normal modes<sup>67</sup>. This approach,  
11 however, breaks down in the case internal conversion occurs at conical intersections, points of  
12 degeneracy between electronic states acting as dynamic funnels for radiationless transitions<sup>68,69</sup>.

13  
14  
15  
16  
17  
18  
19 In an OLED device, at high current, i.e. high hole and electron densities, triplets start accumulating in  
20 the device. Bimolecular processes such as triplet-triplet or triplet-polaron annihilation might occur,  
21 leading to an undesired roll-off behavior<sup>70</sup>. So far, no mechanistic picture has been proposed for  
22 these processes in the context of TADF materials. Triplet-triplet annihilation can proceed either via a  
23 virtual CT state or through a two-electron exchange mechanism, similarly to its reverse process,  
24 singlet fission<sup>71</sup>. The first attempts to include both processes in a device-like Kinetic Monte Carlo  
25 simulation have been based on phenomenological grounds considering that triplet-triplet (triplet-  
26 polaron) annihilation takes place when two triplet excitations (a triplet excitation and a charge)  
27 occupy neighboring sites<sup>72</sup>. Assessing annihilation rates from first principles remains very challenging  
28 since these processes involve the transient formation of high-lying electronic excitations with  
29 multiple-excitation character, the description of which demands highly correlated quantum-chemical  
30 methods often difficult to handle for large-size TADF emitters. A further challenge is that, since  
31 annihilation is a bimolecular process, its theoretical investigation requires the knowledge of the  
32 microscopic arrangement of emitters, so a further layer of calculations, as detailed in the next  
33 paragraphs.

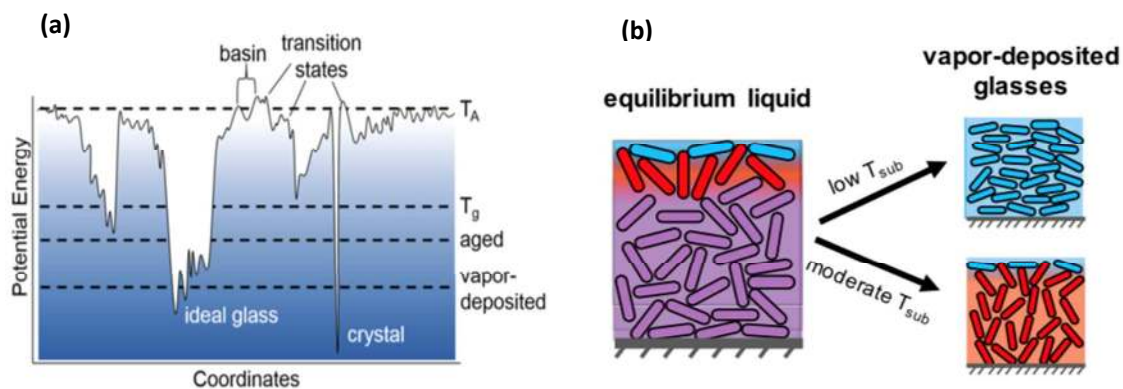
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44 With respect to the gas phase or implicit solvent calculations described so far, a step further towards  
45 the realistic modeling of TADF materials consists in taking into account explicitly the presence of  
46 surrounding host molecules. This is possible via a simple two-step multiscale scheme, by first  
47 employing classical (i.e. molecular mechanics) force fields and Molecular Dynamics (MD) or Monte  
48 Carlo (MC) simulations to produce a realistic guess of the emitting layer morphology (see Figure 1 for  
49 a simplified scheme of a multilayer OLED architecture),<sup>73</sup> and then performing electronic structure  
50 calculations on single molecules or clusters extracted from the simulated trajectories. This approach  
51 is considerably more expensive than a pure quantum mechanical-(QM-)based study, but its higher  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3 computational cost is compensated by the number of additional important effects occurring in real  
4 devices that can be accounted for, namely: i) sampling of many molecular geometries with a  
5 probability of occurrence depending on temperature (Boltzmann-weighted); ii) conformational  
6 changes and freezing induced by the solid matrix; and iii) inclusion of polarization and electrostatic  
7 effects. Besides, simulations predict the orientation distribution of the emitters and, more generally,  
8 provide the molecular positions of carriers and emitters, which are fundamental ingredients for  
9 device modeling, and in particular for kinetic simulations of electronic processes.<sup>74</sup>

10  
11  
12  
13  
14  
15 Note that emitting layers are typically (meta)stable, disordered or partially ordered glasses, and host  
16 and emitter molecules themselves (Figure 1) have characteristic of organic glass-formers, being often  
17 constituted of a few, nonlinear rigid aromatic units interconnected by single rotatable bonds,  
18 allowing thus for a relatively large number of stable conformations. There are at least three routes to  
19 simulate amorphous morphologies of emissive layers, all of them essentially out of thermodynamic  
20 equilibrium. The simplest scheme starts with an NPT-ensemble MD simulation at temperatures and  
21 pressure high enough to obtain a fluid mixture of the desired composition, and subsequently cools  
22 the sample at room temperature and equilibrates it until average volume and total energy appear  
23 stable in time.<sup>10,75</sup> A second, more costly approach imitates solvent annealing process, by starting  
24 from a concentrated solution of the target materials equilibrated at ambient conditions, from which  
25 molecules of solvent are progressively removed until a dry film is obtained. More recently, MD or MC  
26 simulations in which molecules are progressively inserted in the system and landed on a substrate, a  
27 scheme pioneered by some of us for the vapor deposition of organic crystalline semiconductors,<sup>76,77</sup>  
28 have gained increasing popularity in the OLED research field.<sup>78-81</sup> The origin of this trend is two-fold:  
29 on the one hand, real TADF-based active layers are in fact prepared most often by co-deposition of  
30 (at least) one host semiconductor and a guest TADF emitter,<sup>82</sup> and on the other hand, since these  
31 films are amorphous, it is difficult to validate the simulation results versus experimental structural  
32 data, and it is tempting to believe that mimicking the experimental process could improve the quality  
33 of the predictions. Indeed, one of the open questions in the field is how important is the simulation  
34 procedure in determining the final morphology, and in turn how does it influence the calculated  
35 electronic properties. Experimentally, there is a large body of evidence suggesting that vapor-  
36 deposited systems are in general more stable and dense than spin-coated ones prepared by solvent  
37 evaporation<sup>82</sup> or freezing from the liquid phase;<sup>83</sup> however the exact shape of the potential energy  
38 landscape (see scheme in Figure 5a for a pure material) not only does depend on the chemical nature  
39 of the host-guest system but also on their relative concentration.

40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

A further important result has recently emerged both from experiments<sup>78,84,85</sup> and simulations:<sup>78–81</sup> vapor deposited glasses can be to some extent orientationally anisotropic or, in other words, the orientation of the emitters in the active layer is not completely random. Since the specific orientation of emitter transition dipoles can strongly impact the outcoupling efficiency,<sup>86,87</sup> it would be desirable to employ computer simulations in designing materials and setups apt at precisely controlling the orientation of the dyes, and then the direction of emitted light, in order to maximize the light output. The resurgence of anisotropy has been attributed to multiple factors, ranging from van der Waals interactions to dipole-dipole interactions and kinetic effects. These may be again system-dependent, but regardless of specific effects, it appears clear that a critical role is played by the orientation of the molecules at the growing interface with vacuum, as depicted in Figure 5b. Unfortunately, the current lack of knowledge about the relationship between all the simulation parameters (e.g. models, rates, temperatures, times, etc.) and the morphology obtained, so far does not allow to establish how general are the above-mentioned effects and to which extent they could be applicable to experimental results. Therefore, more systematic studies in this direction are urgent.



**Figure 5:** a) Schematic representation of the potential energy landscape of a glass forming system.  $T_A$  is the temperature where non-Arrhenius dynamics are first observed in the liquid. Upon cooling at a constant rate, a temperature is reached at which molecular motions freeze and a glass is formed (glass transition temperature  $T_g$ ). Aging slightly below  $T_g$  allows some equilibration and lowers the potential energy. For some systems, much lower energies can be reached by physical vapor deposition with respect to aging or slow cooling. b) Possible origin of anisotropic molecular packing in vapor-deposited glasses of a rod-like molecule for which the free surface of the equilibrium liquid is anisotropic.<sup>78,88</sup> The substrate temperature  $T_{sub}$  determines the depth to which structure at the surface can equilibrate during deposition. The lowest portion of the growing film becomes trapped by further deposition. Also the interface with the solid substrate can be in principle anisotropic. Reprinted from reference<sup>83</sup>, with the permission of AIP Publishing.



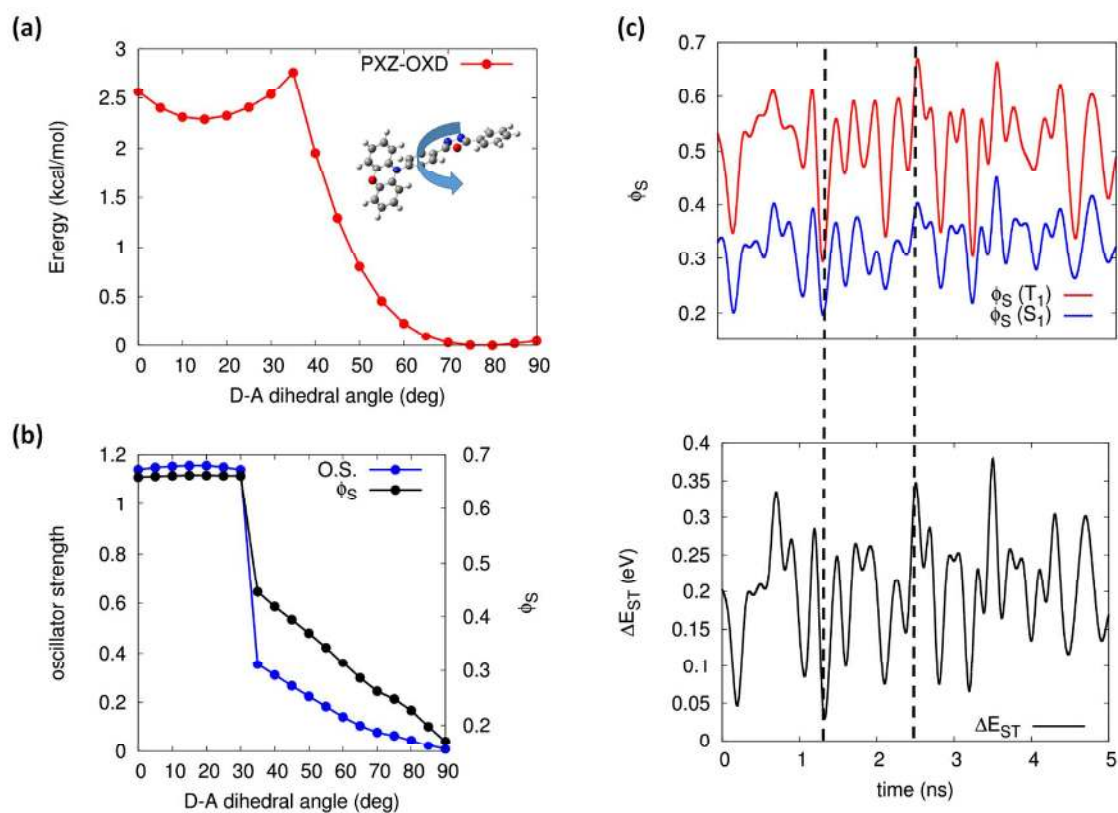
1  
2  
3 Another, maybe unavoidable, problem in comparing simulations and experiments is the enormous  
4 gap between the time scales accessible to simulations (microseconds) and the ones employed in the  
5 lab (minutes, hours). This difference might shift the balance between kinetic and thermodynamic  
6 effects, and may hamper the successful application of computer simulations to systems in which this  
7 balance is delicate. Concerning instead length scales, the mismatch between reality and atomistic  
8 models is much narrower, hundreds of nanometers versus tens or so. To reach the device scale,  
9 coarse-grained models can be employed,<sup>78,89</sup> with the main disadvantage of some extra effort for the  
10 parameterization and complicated backmapping schemes to revert to the fully atomistic morphology,  
11 needed for electronic structure calculations. United-atom force fields can be an effective  
12 compromise, since the reduced number of centers allows a CPU time speed-up of about one order of  
13 magnitude<sup>90</sup> and the backmapping to a full atom model is straightforward. Focusing on full-atom  
14 models, there is a raising awareness that the use of classical force field geometries for QM  
15 calculations may lead to uncontrolled approximations and systematic errors in the evaluation of  
16 electronic properties, e.g.  $S_0$  and  $T_1$  energies.<sup>10,75</sup> In order to minimize this source of error, it would  
17 be beneficial to employ, in future studies, non-transferable force fields specifically tailored for  
18 reproducing not only the QM optimized geometry, but also the vibrational frequencies in the  
19 electronic state of interest.<sup>91</sup>

20  
21 Radiative emission cross-sections, also quantified through the oscillator strength, are often very  
22 small in TADF molecules with close to orthogonal donor and acceptor moieties in their ground- or  
23 excited-state optimized geometries (see Figure 6b). Unless counterbalanced by specific  
24 intramolecular interactions such as hydrogen-bonds,<sup>61</sup> most TADF molecules usually feature a  
25 twisted equilibrium conformation, which inevitably leads to a small overlap between the frontier  $\pi$ -  
26 conjugated orbitals of the donor and acceptor moieties. Fortunately, in TADF emitters based on D  
27 and A moieties connected through single bond, soft torsional modes are easily activated, generating  
28 a large spectrum of conformations at room temperature<sup>4,7</sup> (see Figure 6a). These display significantly  
29 different absorption/emission energies and associated  $\Delta E_{ST}$  values as the nature of the lowest singlet  
30 and triplet excited states (as probed through  $\Phi_S$ ) varies in time and space (see Figures 6c and 6d).  
31 Thermal excitation of these soft vibrational modes also results in broad and unstructured absorption  
32 and emission spectra. Thermal motion around the twisted equilibrium structures has a positive  
33 impact on luminescence, as it allows to sample conformations with larger overlap between the  
34 electron and hole wavefunctions, *i.e.* also larger oscillator strengths and radiative decay rates.<sup>20</sup> In  
35 this case, emission of the TADF materials appears as a vibrationally-assisted process strongly coupled  
36 to soft (low-frequency) torsional modes.<sup>7</sup>

1  
2  
3 Also the RISC process, as claimed by Monkman and coworkers,<sup>3</sup> can be seen as a spin-vibronic  
4 mechanism where spin mixing is dynamically gated by conformational fluctuations triggered by low-  
5 energy torsional modes. The influence of torsional modes on excited-state dynamics is two-fold: (i) it  
6 brings excited states of the same spin multiplicity closer to each other, enhancing non-adiabatic  
7 coupling between the lowest singlet (triplet) excited states;<sup>92</sup> (ii) it affects dynamically the nature of  
8 the singlet and triplet excited states involved in the TADF process in a way that the thermally-  
9 averaged spin-orbit coupling is enhanced compared to its value at the equilibrium geometry.<sup>67,92</sup> The  
10 role of vibrations on TADF dynamics can be modeled using different formalisms, either based on non-  
11 adiabatic molecular dynamics, such as mixed quantum-classical surface hopping methods<sup>93</sup> or full  
12 quantum wavepacket propagation simulations,<sup>94</sup> or relying on rate expressions derived from time-  
13 dependent second-order perturbation theory (Fermi Golden Rule).<sup>67,92</sup> These approaches usually  
14 include a set of preselected intramolecular vibrational modes for which the frequencies and  
15 displacements (electron-phonon couplings) are computed from first-principles. Alternatively, one can  
16 resort to MD simulations that allow sampling all vibrational modes classically, being intra- or inter-  
17 molecular in origin, at once. In the case of low-frequency vibrations, classical and quantum  
18 approximations yield similar results for thermally averaged spin-orbit coupling and  $\Delta E_{ST}$  values.<sup>95</sup> The  
19 reorganization energy associated with high-frequency vibrations, *i.e.* mostly bond stretching, can be  
20 obtained from ground- and excited-state geometry optimization of the isolated molecules. These  
21 modes steer temperature-independent quantum tunneling effects and can be easily incorporated  
22 into rate expressions for ISC and RISC. For instance, by a careful analysis of the time evolution of the  
23 dihedral angles in 2CzPN and 4CzIPN, a characteristic time scale of about 1 ps has been inferred. This  
24 is fast compared to RISC, hence a large portion of the dihedral angles distribution is explored by the  
25 molecule before upconversion takes place, which confirms the truly dynamic nature of the RISC  
26 process. Considering thermally-averaged spin-orbit coupling and  $\Delta E_{ST}$  for the specific case of 4CzIPN,  
27 (R)ISC rates calculated within the semi-classical Marcus theory were found to be in excellent  
28 agreement with experimental data.<sup>10</sup>

29  
30  
31 To close this discussion, we would like to briefly refer to studies pointing to the role of 'hard' modes.  
32 For instance, it has been shown that the displacement along a C=O stretching mode on the donor  
33 moiety of a xanthone-acridine D-A complex is able to bring in near resonance triplet <sup>3</sup>CT and <sup>3</sup>LE  
34 states, from which efficient RISC to <sup>1</sup>CT proceeds.<sup>96</sup> Similarly, highly correlated wavefunction based  
35 calculations in carbene-metal-amide complexes suggest that the dynamic red shift observed  
36 experimentally is associated with changes in the carbon-nitrogen bond length and metal-carbon-  
37 nitrogen bond angle within the carbene-metal-amide three-center core. Very interestingly, these  
38 changes reduce  $\Delta E_{ST}$ , while keeping unaffected the spin-orbit coupling and emission transition dipole  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57

moment from the singlet excited state, at odds with the initially proposed rotationally-assisted upconversion mechanism.<sup>45</sup>



**Figure 6:** (a) Torsional energy profiles calculated at the PBE0-D3(BJ)/6-31G(d,p) level of theory with the PCM module for solvent (toluene) for PXZ-OXD. (b) Variation of  $\Delta E_{ST}$  and oscillator strength (O.S) as a function of the D-A torsion angle. (c) Time evolution of the  $\Phi_S(S_1)$  and  $\Phi_S(T_1)$  from electronic structure calculations performed along a molecular dynamics trajectory for 2CzPN. (d) Time evolution of the  $\Delta E_{ST}$  for 2CzPN. Vertical dashed lines highlight that  $\Delta E_{ST}$  is the largest (smallest) when the difference in  $\Phi_S(S_1)$  and  $\Phi_S(T_1)$  is the largest (the smallest). Reproduced from Ref.<sup>7</sup> with permission from The Royal Society of Chemistry. Reprinted and adapted with permission from reference<sup>10</sup>. Copyrights 2017 by the American Physical Society.

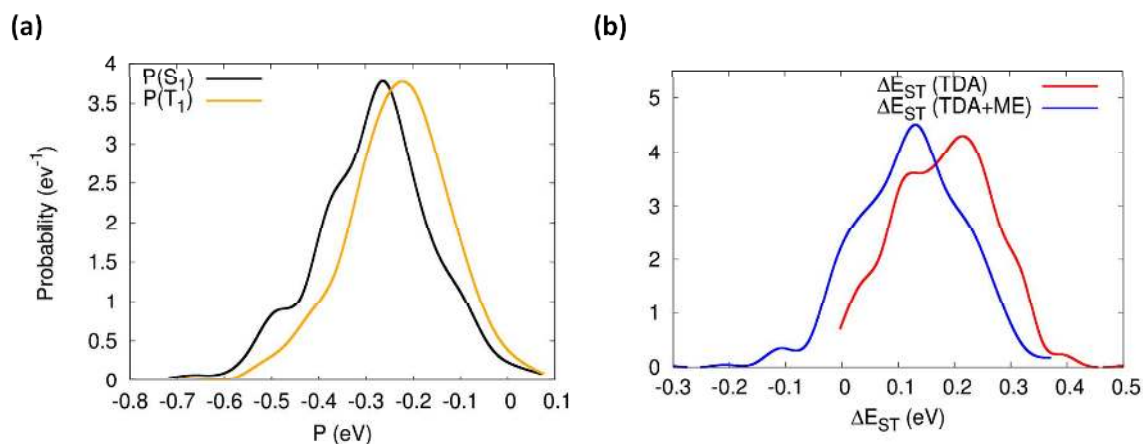
Environmental effects are known to have a major impact on molecular excitations of charge transfer character, yet the implications on  $\Delta E_{ST}$  in TADF emitters has started to be appreciated only recently. The role of the environment is particularly subtle and important in systems where singlet and triplet excitations of CT and/or LE character are all close in energy and compete, while the energy window interesting for applications is of few tens of meV. Viable computational strategies consist in QM/MM approaches, where the TADF emitter (QM system) is embedded in a classical medium that can be

1  
2  
3 described either with PCM or with discrete schemes of atomistic resolutions. Both approaches have  
4 advantages and limitations that we briefly discuss in the following.

5  
6 In this context, Brédas and coworkers systematically employed the PCM in their TD-DFT  
7 investigations of singlet-triplet splitting and spin-orbit matrix elements for different TADF  
8 molecules.<sup>14,56</sup> Sun et al. specifically addressed the effect of the dielectric constant ( $\epsilon$ ) on the nature  
9 of the excited states and on the single-triplet splitting by using optimally tuned range-separated  
10 hybrid functionals, whose range separation parameter  $\omega$  was optimized for each value of  $\epsilon$ .<sup>97</sup> These  
11 calculations highlighted the role of the medium polarizability in stabilizing electronic configurations  
12 with large CT character, in turn affecting the CT-LE hybridization in singlet and triplet excitations. For  
13 instance, dipolar D-A molecules such as TXO-TPA and TXO-PhCz, which in the gas phase are  
14 prescribed to feature large  $\Delta E_{ST}$  (ca. 0.5 eV) as a result of the large LE character of  $T_1$ , become  
15 interesting for TADF applications in a typical organic matrix with  $\epsilon \sim 3$ . Indeed the medium polarization  
16 leads to excitations with large CT character and nearly non-overlapping hole and electron clouds and  
17  $\Delta E_{ST}$  below 0.1 eV.<sup>97</sup> Marian *et al.*<sup>44</sup> proposed a reinterpretation of the emission observed for  
18 carbene-metal-amide in both solution and film based on a combination of DFT/MRCI and PCM.  
19 Especially, they showed that solvent reorganization has to be taken into account when computing  
20 the delayed fluorescence spectrum, while it is not required when discussing prompt fluorescence  
21 occurring at early-timescale. Practically speaking, one needs to consider the relaxed density matrix of  
22 the excited state to calculate the solvent reaction field. Delayed fluorescence appears to be red-  
23 shifted in comparison the prompt one in agreement with experiment. However, in glassy films, such  
24 a treatment is not needed because the medium reorientation is sterically hindered so that delayed  
25 fluorescence appears to be blue-shifted compared to solution. Along the same line, Penfold *et al.*<sup>75</sup>  
26 have highlighted through a combination of MD and TD-DFT calculations that a blue shift in the  
27 delayed emission of D-A TADF emitters in films at the longest timescale does not result from host  
28 reorganization, and thus on specific host-guest interaction, but rather from a distribution of CT states  
29 with different emission energies. The prompt fluorescence is essentially governed by the higher-  
30 energy CT states that exhibit the largest hole-electron wavefunction overlap and therefore also  
31 oscillator strength. As for the delayed fluorescence, the early part of the signal appears to be red-  
32 shifted in comparison to the prompt fluorescence since RISC occurs first through the lower-energy CT  
33 states that minimize  $\Delta E_{ST}$ . The late delayed fluorescence component then occurs through higher-lying  
34 but more emissive CT states, thereby rationalizing the blue-shift observed.

35  
36  
37 Beyond PCM, atomistic polarizable models aim at describing excitations of molecules in their specific  
38 environment, usually an amorphous matrix blending the charge transporting material and other  
39 emitters. Atomistic simulations of such blends are a prerequisite for these approaches, which also  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57

allow for the sampling over a statistical collection of molecule-environment configurations. Our group adopted this route to study environmental effects in carbazoyl-dicyanobenzene based TADF emitters, 2CzPN and 4CzIPN, employing atomistic microelectrostatic models parameterized from first principles.<sup>10</sup> These calculations showed that the medium polarization, modeled with mutually interacting anisotropic polarizabilities, reacts differently to states of different CT character (see Figure 7a). For instance, 2CzPN shows a singlet excitation  $S_1$  that has larger CT character than the triplet  $T_1$ , and thus a larger electrical dipole, resulting in a stronger stabilization of the former by the environment polarization, finally leading to a reduction in  $\Delta E_{ST}$  (see Figure 7b). This provides a general mechanism through which the embedding medium compensates for large  $\Delta E_{ST}$  established at the molecular level, leading in some cases to negative  $\Delta E_{ST}$  values. In addition, atomistic models do also account for the fact that molecules are distorted in real morphologies and experience the inhomogeneous electrostatic potential of the neighborhood. These phenomena affect the energies of the excited states within both the singlet and the triplet manifold of states, leading to inhomogeneous broadening in the solid matrix and to broad distributions of  $\Delta E_{ST}$  values. Such a disorder can also break the symmetry of the molecule (e.g. localizing the hole of a CT excitation on a given D carbazole unit) and, most interestingly, have a dynamic nature, i.e. lead to a modulation in time of the nature of the relevant states (CT-LE hybridization) and of the  $\Delta E_{ST}$ .<sup>10</sup>



**Figure 7:** (a) Polarization energy distributions associated to  $S_1$  and  $T_1$  excited states. (b) Distributions of  $\Delta E_{ST}$ . Red and blue lines correspond to TDA-PBE0 results in the vacuum and accounting for local dielectric effects, respectively. All data provided in this Figure refer to 2CzPN. Reprinted and adapted with permission from reference<sup>10</sup>. Copyright 2017 by the American Physical Society.

1  
2  
3 In summary, predictive modeling of TADF emitters requires an integrated multiscale approach able to  
4 capture the energetics and the dynamics of electronic excitations in realistic morphologies of OLEDs  
5 emitting layers. In combination with state-of-the-art experimental investigations (optical  
6 spectroscopy and device characterization), computational studies have already shed light on some  
7 key features of TADF, including:  
8  
9

- 10 - The excited states involved in TADF often feature mixed CT and LE character.
- 11 - The host influences  $\Delta E_{ST}$  through conformational and dielectric effects.
- 12 - RISC and light emission are dynamic processes assisted by intramolecular vibrations.

13  
14 While some of the items above remain open to discussion, as they are likely material-specific, they  
15 also prompt new questions and challenges that need to be addressed when designing the next  
16 generation of TADF emitters. For instance, can one tune the nature of the excited states involved in  
17 the TADF process in order to speedup RISC? Can one take advantage of environmental effects to  
18 design molecules with negative  $\Delta E_{ST}$ ? Can we expand the modeling approaches in order to account  
19 for all the mono- and bi-molecular radiative and non-radiative processes taking place in TADF-based  
20 OLEDs and identify host-guest combinations that would minimize annihilation and maximize pure  
21 and color-tunable light emission?  
22  
23  
24  
25  
26  
27

28  
29 Clearly the answer to these questions can only be obtained in the scope of a multifaceted theoretical  
30 framework, where molecular and material properties are conjointly addressed and optimized. More  
31 specifically, we would like to end this perspective with modeling challenges inspired by a few  
32 selected opportunities from recent experimental investigations:  
33  
34

- 35 - Hyperfluorescent OLEDs.<sup>98</sup> Here, TADF molecules act mostly as assistant dopants that drive  
36 the excitations towards a dye with narrow-line singlet emission. To further improve what is  
37 referred to as the 4<sup>th</sup> generation OLEDs, a microscopic picture for the diffusion of singlet and  
38 triplet excitations relevant to TADF is definitively needed. This would require going beyond  
39 the widely used Förster model for weakly dipole-allowed CT singlets and including both  
40 exchange and superexchange interactions for triplets.<sup>99–101</sup>
- 41 - Highly emissive TADF emitters. Architectures sustaining multi-resonance effects<sup>28,102,103</sup> have  
42 the potential to solve the conundrum of large singlet radiative decay rates despite small  
43 exchange energies. A proper description of the singlet and triplet excitations in these  
44 molecules calls for the inclusion of high-order electronic correlation effects, difficult to  
45 capture using conventional TD-DFT methods.
- 46 - Exciplexes. Despite considerable efforts to establish them as viable technology for OLED  
47 applications, exciplexes, namely emissive D-A intermolecular CT states, have been  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57

1  
2 investigated in only very few computational studies.<sup>104,105</sup> This is surely related to the  
3 weakness of intermolecular interactions in organics resulting in the large configurational  
4 space explored by the D and A molecules that can adopt multiple relative orientations.  
5 Besides sampling issues, another timely question relates to the quantum-mechanical effect  
6 driving RISC in exciplexes, with scenarios based on either spin-orbit or hyperfine field  
7 couplings proposed in the literature.<sup>104,106</sup> Note that recent studies of crystalline  
8 multichromophoric materials, composed on a mixed stack of 1:1 D-A molecules, are  
9 particularly interesting in this context, too.<sup>107,108</sup>  
10  
11  
12  
13  
14  
15

16 As a last note, we would like to stress that, as highlighted in this perspective, modeling of TADF is a  
17 complex endeavor and we warn the community about the potential pitfalls of 'black box' calculations  
18 using standard approaches. While these might provide a useful first screening, we believe that the  
19 field of computational modeling has now reached a maturity level that allows for a truly first-  
20 principles description of TADF emitters and hope this perspective will guide modelers and  
21 experimentalists on finding their way to best practices in TADF.  
22  
23  
24  
25  
26  
27

### 28 **Biographical Sketches**

29  
30  
31 **Yoann Olivier** obtained a Ph.D. from the University of Mons in 2008. From 2009 to 2013, he held a  
32 postdoctoral fellowship from Belgian National Fund for Scientific Research (FNRS) and went on  
33 postdoctoral stays with Prof. Claudio Zannoni at the University of Bologna and Prof. Henning  
34 Siringhaus at the University of Cambridge. He is currently a research associate at the University of  
35 Mons. His research interests deal with the understanding of electronic processes in organic  
36 conjugated and 2D materials, using a multiscale approach combining quantum-chemical methods,  
37 Monte Carlo approach and molecular dynamics simulations.  
38  
39  
40  
41  
42  
43

44 **Juan-Carlos Sancho-Garcia** obtained a Ph.D. in Quantum Chemistry in 2001 at the University of  
45 Alicante, followed by a postdoctoral stay (2002-2004) at the Laboratory for Chemistry of Novel  
46 Materials in Mons, Belgium. He returned to Alicante as a 'Ramon y Cajal' research fellow and  
47 received a permanent position in 2010 in the Dept. of Physical Chemistry. His work integrates the  
48 development of more accurate DFT methods with applications to the field of Organic Electronics to  
49 elucidate structure-property relationships."  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3 **Luca Muccioli** is assistant professor of Physical Chemistry at the University of Bologna (Italy). He  
4 earned a Ph.D. in Chemical Sciences from University of Bologna in 2003, under the supervision of  
5 Prof. Claudio Zannoni, with a thesis on atomistic computer simulations of liquid crystals.

6  
7 From 2003 to 2014 he was postdoctoral researcher at the University of Bologna, and from 2014 to  
8 2016 assistant professor at the Institut des Sciences Moléculaires of the University of Bordeaux  
9 (France). His research interests concern the application of computational chemistry techniques to the  
10 study of the physico-chemical properties of organic materials, with particular focus on the prediction  
11 of structural and electronic properties of liquid crystals and organic semiconductors through  
12 multiscale approaches  
13  
14  
15  
16  
17

18 **Gabriele D'Avino** received his Ph.D. in Materials Science at the University of Parma in 2010. After a  
19 postdoctoral research position at the University of Bologna, and two Marie Curie fellowships at the  
20 University of Liege and at the University of Mons, he is currently CNRS associate researcher at the  
21 Institut Institut Néel, Grenoble. His research focuses on the multiscale modeling of organic functional  
22 materials, merging quantum and classical techniques.  
23  
24  
25  
26  
27

28 **David Beljonne** got his PhD in Chemistry at the University of Mons-Hainaut in 1994. After post-  
29 doctoral stays at the Universities of Cambridge (with Prof. Friend) and Rochester (with Prof.  
30 Mukamel), he is now a Research Director of the Belgian National Science Foundation (FNRS) and  
31 Professor at the University of Mons. He is also a Visiting Principal Research Scientist at the Georgia  
32 Institute of Technology in Atlanta. His research activities deal with the modeling of semiconducting  
33 materials for energy applications.  
34  
35  
36  
37  
38

### 39 **PULL QUOTES**

40  
41

- 42 (i) Can we predict the nature of the excited states and tune the primary chemical structure of  
43 TADF emitters in order to maximize the efficiency of upconversion and light emission in  
44 TADF-based OLEDs?  
45  
46  
47  
48 (ii) How important is the simulation procedure in determining the final morphology, and in turn  
49 how does it influence the calculated electronic properties?  
50  
51  
52  
53 (iii) How can we design *in silico* TADF emitters that yield the right orientation in the solid-state  
54 matrix to maximize light outcoupling?  
55  
56  
57  
58  
59  
60



- 1  
2  
3 (iv) How much torsional dynamics is affected by the host-guest interaction? How does it affect  
4 the efficiency of TADF?  
5  
6  
7 (v) Polarization medium effects are stabilizing CT states with respect to LE states. Could we  
8 possibly target the right combination of host and guest in order to get negative  $\Delta E_{ST}$ ?  
9  
10  
11 (vi) Should we freeze-out the motion of triplet excitations in TADF materials in order to reduce  
12 annihilation processes?  
13  
14  
15

### 16 Acknowledgments

17  
18 The work in Mons was supported by the Belgian National Science Foundation, F.R.S.-FNRS.  
19 Computational resources have been provided by the Consortium des Équipements de Calcul Intensif  
20 (CÉCI), funded by F.R.S.-FNRS under Grant No. 2.5020.11 as well as the Tier-1 supercomputer of the  
21 Fédération Wallonie-Bruxelles, infrastructure funded by the Walloon Region under the grant  
22 agreement n1117545. The research in Bologna, Grenoble and Mons is also through the European  
23 Union's Horizon 2020 research and innovation program under Grant Agreement No. 646176  
24 (EXTMOS project). GD would like to thank Prof. Xavier Blase for discussions. JC, LM and YO  
25 acknowledge discussion with Dr. Mónica Moral. DB and YO would like to thank Prof. Thuc Quyen  
26 Nguyen and Brett Yurash for the fruitful collaboration on the 2CzPN and 4CzIPN study. DB is a FNRS  
27 Research Director.  
28  
29  
30  
31  
32  
33

### 34 References

- 35  
36  
37 (1) Parker, C. A.; Hatchard, C. G. Triplet-Singlet Emission in Fluid Solutions. Phosphorescence of  
38 Eosin. *Trans. Faraday Soc.* **1961**, *57*, 1894-1904.  
39  
40 (2) Uoyama, H.; Goushi, K.; Shizu, K.; Nomura, H.; Adachi, C. Highly Efficient Organic Light-  
41 Emitting Diodes from Delayed Fluorescence. *Nature* **2012**, *492*, 234–238.  
42  
43 (3) Etherington, M. K.; Gibson, J.; Higginbotham, H. F.; Penfold, T. J.; Monkman, A. P. Revealing  
44 the Spin–vibronic Coupling Mechanism of Thermally Activated Delayed Fluorescence. *Nat.*  
45 *Commun.* **2016**, *7*, 13680.  
46  
47 (4) Dias, F. B.; Santos, J.; Graves, D. R.; Data, P.; Nobuyasu, R. S.; Fox, M. A.; Batsanov, A. S.;  
48 Palmeira, T.; Berberan-Santos, M. N.; Bryce, M. R.; et al. The Role of Local Triplet Excited  
49 States and D-A Relative Orientation in Thermally Activated Delayed Fluorescence:  
50 Photophysics and Devices. *Adv. Sci.* **2016**, *3*, 1600080.  
51  
52 (5) Hosokai, T.; Matsuzaki, H.; Nakanotani, H.; Tokumaru, K.; Tsutsui, T.; Furube, A.; Nasu, K.;  
53 Nomura, H.; Yahiro, M.; Adachi, C. Evidence and Mechanism of Efficient Thermally Activated  
54  
55  
56  
57  
58  
59  
60

- 1  
2  
3 Delayed Fluorescence Promoted by Delocalized Excited States. *Sci. Adv.* **2017**, *3*, e1603282.
- 4 (6) Milián-Medina, B.; Gierschner, J. Computational Design of Low Singlet–triplet Gap All-Organic  
5 Molecules for OLED Application. *Org. Electron.* **2012**, *13*, 985–991.
- 6  
7 (7) Olivier, Y.; Moral, M.; Muccioli, L.; Sancho-García, J.-C. Dynamic Nature of Excited States of  
8 Donor-Acceptor TADF Materials for OLEDs: How Theory Can Reveal Structure-Property  
9 Relationships. *J. Mater. Chem. C* **2017**, *5*, 5718–5729.
- 10  
11 (8) Wong, M. Y.; Zysman-Colman, E. Purely Organic Thermally Activated Delayed Fluorescence  
12 Materials for Organic Light-Emitting Diodes. *Adv. Mater.* **2017**, *29*, 1605444.
- 13  
14 (9) Schott, S.; McNellis, E. R.; Nielsen, C. B.; Chen, H. Y.; Watanabe, S.; Tanaka, H.; McCulloch, I.;  
15 Takimiya, K.; Sinova, J.; Sirringhaus, H. Tuning the Effective Spin-Orbit Coupling in Molecular  
16 Semiconductors. *Nat. Commun.* **2017**, *8*, 15200.
- 17  
18 (10) Olivier, Y.; Yurash, B.; Muccioli, L.; D’Avino, G.; Mikhnenko, O.; Sancho-García, J. C.; Adachi, C.;  
19 Nguyen, T.-Q.; Beljonne, D. Nature of the Singlet and Triplet Excitations Mediating Thermally  
20 Activated Delayed Fluorescence. *Phys. Rev. Mater.* **2017**, *1*, 75602.
- 21  
22 (11) Gómez-Bombarelli, R.; Aguilera-Iparraguirre, J.; Hirzel, T. D.; Duvenaud, D.; Maclaurin, D.;  
23 Blood-Forsythe, M. A.; Chae, H. S.; Einzinger, M.; Ha, D.-G.; Wu, T.; et al. Design of Efficient  
24 Molecular Organic Light-Emitting Diodes by a High-Throughput Virtual Screening and  
25 Experimental Approach. *Nat. Mater.* **2016**, *15*, 1120–1127.
- 26  
27 (12) Chen, X.-K.; Kim, D.; Brédas, J.-L. Thermally Activated Delayed Fluorescence (TADF) Path  
28 toward Efficient Electroluminescence in Purely Organic Materials: Molecular Level Insight.  
29 *Acc. Chem. Res.* **2018**, *51*, 2215–2224.
- 30  
31 (13) Moral, M.; Muccioli, L.; Son, W.-J.; Olivier, Y.; Sancho-García, J. C. Theoretical Rationalization  
32 of the Singlet–Triplet Gap in OLEDs Materials: Impact of Charge-Transfer Character. *J. Chem.*  
33 *Theory Comput.* **2015**, *11*, 168–177.
- 34  
35 (14) Sun, H.; Zhong, C.; Brédas, J. L. Reliable Prediction with Tuned Range-Separated Functionals of  
36 the Singlet-Triplet Gap in Organic Emitters for Thermally Activated Delayed Fluorescence. *J.*  
37 *Chem. Theory Comput.* **2015**, *11*, 3851–3858.
- 38  
39 (15) Peng, Q.; Fan, D.; Duan, R.; Yi, Y.; Niu, Y.; Wang, D.; Shuai, Z. Theoretical Study of Conversion  
40 and Decay Processes of Excited Triplet and Singlet States in a Thermally Activated Delayed  
41 Fluorescence Molecule. *J. Phys. Chem. C* **2017**, *121*, 13448–13456.
- 42  
43 (16) Schreiber, M.; Silva-Junior, M. R.; Sauer, S. P. A.; Thiel, W. Benchmarks for Electronically  
44 Excited States: CASPT2, CC2, CCSD, and CC3. *J. Chem. Phys.* **2008**, *128*, 134110.
- 45  
46 (17) Silva-Junior, M. R.; Schreiber, M.; Sauer, S. P. A.; Thiel, W. Benchmarks for Electronically  
47 Excited States: Time-Dependent Density Functional Theory and Density Functional Theory  
48 Based Multireference Configuration Interaction. *J. Chem. Phys.* **2008**, *129*, 104103.
- 49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

- 1  
2  
3 (18) Jacquemin, D.; Wathelet, V.; Perpète, E. A.; Adamo, C. Extensive TD-DFT Benchmark: Singlet-  
4 Excited States of Organic Molecules. *J. Chem. Theory Comput.* **2009**, *5*, 2420–2435.  
5  
6 (19) Loos, P.-F.; Scemama, A.; Blondel, A.; Garniron, Y.; Caffarel, M.; Jacquemin, D. A  
7 Mountaineering Strategy to Excited States: Highly Accurate Reference Energies and  
8 Benchmarks. *J. Chem. Theory Comput.* **2018**, *14*, 4360–4379.  
9  
10 (20) Mewes, J.-M. Modeling TADF in Organic Emitters Requires a Careful Consideration of the  
11 Environment and Going beyond the Franck–Condon Approximation. *Phys. Chem. Chem. Phys.*  
12 **2018**, *20*, 12454–12469.  
13  
14 (21) Hellweg, A.; Grün, S. A.; Hättig, C. Benchmarking the Performance of Spin-Component Scaled  
15 CC2 in Ground and Electronically Excited States. *Phys. Chem. Chem. Phys.* **2008**, *10*, 4119-  
16 4127.  
17  
18 (22) Jacquemin, D.; Duchemin, I.; Blondel, A.; Blase, X. Benchmark of Bethe-Salpeter for Triplet  
19 Excited-States. *J. Chem. Theory Comput.* **2017**, *13*, 767–783.  
20  
21 (23) Noguchi, Y.; Sugino, O. High-Lying Triplet Excitons of Thermally Activated Delayed  
22 Fluorescence Molecules. *J. Phys. Chem. C* **2017**, *121*, 20687–20695.  
23  
24 (24) Rangel, T.; Hamed, S. M.; Bruneval, F.; Neaton, J. B. An Assessment of Low-Lying Excitation  
25 Energies and Triplet Instabilities of Organic Molecules with an *Ab Initio* Bethe-Salpeter  
26 Equation Approach and the Tamm-Dancoff Approximation. *J. Chem. Phys.* **2017**, *146*, 194108.  
27  
28 (25) Weigend, F.; Häser, M.; Patzelt, H.; Ahlrichs, R. RI-MP2: Optimized Auxiliary Basis Sets and  
29 Demonstration of Efficiency. *Chem. Phys. Lett.* **1998**, *294*, 143–152.  
30  
31 (26) Weigend, F. A Fully Direct RI-HF Algorithm: Implementation, Optimised Auxiliary Basis Sets,  
32 Demonstration of Accuracy and Efficiency. *Phys. Chem. Chem. Phys.* **2002**, *4*, 4285–4291.  
33  
34 (27) Werner, H.-J.; Manby, F. R.; Knowles, P. J. Fast Linear Scaling Second-Order Møller-Plesset  
35 Perturbation Theory (MP2) Using Local and Density Fitting Approximations. *J. Chem. Phys.*  
36 **2003**, *118*, 8149–8160.  
37  
38 (28) Hatakeyama, T.; Shiren, K.; Nakajima, K.; Nomura, S.; Nakatsuka, S.; Kinoshita, K.; Ni, J.; Ono,  
39 Y.; Ikuta, T. Ultrapure Blue Thermally Activated Delayed Fluorescence Molecules: Efficient  
40 HOMO-LUMO Separation by the Multiple Resonance Effect. *Adv. Mater.* **2016**, *28*, 2777–2781.  
41  
42 (29) Brückner, C.; Engels, B. Benchmarking Singlet and Triplet Excitation Energies of Molecular  
43 Semiconductors for Singlet Fission: Tuning the Amount of HF Exchange and Adjusting Local  
44 Correlation to Obtain Accurate Functionals for Singlet–triplet Gaps. *Chem. Phys.* **2017**, *482*,  
45 319–338.  
46  
47 (30) Nakagawa, T.; Ku, S.-Y.; Wong, K.-T.; Adachi, C. Electroluminescence Based on Thermally  
48 Activated Delayed Fluorescence Generated by a Spirobifluorene Donor–acceptor Structure.  
49 *Chem. Commun.* **2012**, *48*, 9580–9582.  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

- 1  
2  
3 (31) Cai, X.; Chen, D.; Gao, K.; Gan, L.; Yin, Q.; Qiao, Z.; Chen, Z.; Jiang, X.; Su, S.-J. "Trade-Off"  
4 Hidden in Condensed State Solvation: Multiradiative Channels Design for Highly Efficient  
5 Solution-Processed Purely Organic Electroluminescence at High Brightness. *Adv. Funct. Mater.*  
6 **2018**, *28*, 1704927.  
7  
8  
9 (32) Huang, S.; Zhang, Q.; Shiota, Y.; Nakagawa, T.; Kuwabara, K.; Yoshizawa, K.; Adachi, C.  
10 Computational Prediction for Singlet- and Triplet-Transition Energies of Charge-Transfer  
11 Compounds. *J. Chem. Theory Comput.* **2013**, *9*, 3872–3877.  
12  
13 (33) Jacquemin, D.; Perpète, E. A.; Ciofini, I.; Adamo, C.; Valero, R.; Zhao, Y.; Truhlar, D. G. On the  
14 Performances of the M06 Family of Density Functionals for Electronic Excitation Energies. *J.*  
15 *Chem. Theory Comput.* **2010**, *6*, 2071–2085.  
16  
17 (34) Hirata, S.; Head-Gordon, M. Time-Dependent Density Functional Theory within the Tamm–  
18 Dancoff Approximation. *Chem. Phys. Lett.* **1999**, *314*, 291–299.  
19  
20 (35) Penfold, T. J. On Predicting the Excited-State Properties of Thermally Activated Delayed  
21 Fluorescence Emitters. *J. Phys. Chem. C* **2015**, *119*, 13535–13544.  
22  
23 (36) Leininger, T.; Stoll, H.; Werner, H.-J.; Savin, A. Combining Long-Range Configuration  
24 Interaction with Short-Range Density Functionals. *Chem. Phys. Lett.* **1997**, *275*, 151–160.  
25  
26 (37) Iikura, H.; Tsuneda, T.; Yanai, T.; Hirao, K. A Long-Range Correction Scheme for Generalized-  
27 Gradient-Approximation Exchange Functionals. *J. Chem. Phys.* **2001**, *115*, 3540–3544.  
28  
29 (38) Yanai, T.; Tew, D. P.; Handy, N. C. A New Hybrid Exchange–correlation Functional Using the  
30 Coulomb-Attenuating Method (CAM-B3LYP). *Chem. Phys. Lett.* **2004**, *393*, 51–57.  
31  
32 (39) Hait, D.; Zhu, T.; McMahon, D. P.; Van Voorhis, T. Prediction of Excited-State Energies and  
33 Singlet–Triplet Gaps of Charge-Transfer States Using a Restricted Open-Shell Kohn–Sham  
34 Approach. *J. Chem. Theory Comput.* **2016**, *12*, 3353–3359.  
35  
36 (40) Di, D.; Romanov, A. S.; Yang, L.; Richter, J. M.; Rivett, J. P. H.; Jones, S.; Thomas, T. H.; Abdi  
37 Jalebi, M.; Friend, R. H.; Linnolahti, M.; et al. High-Performance Light-Emitting Diodes Based  
38 on Carbene-Metal-Amides. *Science* **2017**, *356*, 159–163.  
39  
40 (41) Grimme, S.; Bannwarth, C. Ultra-Fast Computation of Electronic Spectra for Large Systems by  
41 Tight-Binding Based Simplified Tamm-Dancoff Approximation (sTDA-xTB). *J. Chem. Phys.* **2016**,  
42 *145*, 054103.  
43  
44 (42) Alipour, M.; Karimi, N. Dissecting the Accountability of Parameterized and Parameter-Free  
45 Single-Hybrid and Double-Hybrid Functionals for Photophysical Properties of TADF-Based  
46 OLEDs. *J. Chem. Phys.* **2017**, *146*, 234304.  
47  
48 (43) Lyskov, I.; Kleinschmidt, M.; Marian, C. M. Redesign of the DFT/MRCI Hamiltonian. *J. Chem.*  
49 *Phys.* **2016**, *144*, 034104.  
50  
51 (44) Föllner, J.; Marian, C. M. Rotationally Assisted Spin-State Inversion in Carbene–Metal–Amides Is  
52  
53  
54  
55  
56  
57  
58  
59  
60

- 1  
2  
3 an Artifact. *J. Phys. Chem. Lett.* **2017**, *8*, 5643–5647.
- 4 (45) Taffet, E. J.; Olivier, Y.; Lam, F.; Beljonne, D.; Scholes, G. D. Carbene–Metal–Amide Bond  
5 Deformation, Rather Than Ligand Rotation, Drives Delayed Fluorescence. *J. Phys. Chem. Lett.*  
6 **2018**, *9*, 1620–1626.
- 7  
8 (46) Maschietto, F.; Campetella, M.; Frisch, M. J.; Scalmani, G.; Adamo, C.; Ciofini, I. How Are the  
9 Charge Transfer Descriptors Affected by the Quality of the Underpinning Electronic Density? *J.*  
10 *Comput. Chem.* **2018**, *39*, 735–742.
- 11  
12 (47) Peach, M. J. G.; Benfield, P.; Helgaker, T.; Tozer, D. J. Excitation Energies in Density Functional  
13 Theory: An Evaluation and a Diagnostic Test. *J. Chem. Phys.* **2008**, *128*, 044118.
- 14  
15 (48) Peach, M. J. G.; Tozer, D. J. Illustration of a TDDFT Spatial Overlap Diagnostic by Basis Function  
16 Exponent Scaling. *J. Mol. Struct. THEOCHEM* **2009**, *914*, 110–114.
- 17  
18 (49) Guido, C. A.; Cortona, P.; Mennucci, B.; Adamo, C. On the Metric of Charge Transfer Molecular  
19 Excitations: A Simple Chemical Descriptor. *J. Chem. Theory Comput.* **2013**, *9*, 3118–3126.
- 20  
21 (50) Guido, C. A.; Cortona, P.; Adamo, C. Effective Electron Displacements: A Tool for Time-  
22 Dependent Density Functional Theory Computational Spectroscopy. *J. Chem. Phys.* **2014**, *140*,  
23 104101.
- 24  
25 (51) Le Bahers, T.; Adamo, C.; Ciofini, I. A Qualitative Index of Spatial Extent in Charge-Transfer  
26 Excitations. *J. Chem. Theory Comput.* **2011**, *7*, 2498–2506.
- 27  
28 (52) Etienne, T.; Assfeld, X.; Monari, A. Toward a Quantitative Assessment of Electronic  
29 Transitions' Charge-Transfer Character. *J. Chem. Theory Comput.* **2014**, *10*, 3896–3905.
- 30  
31 (53) Campetella, M.; Maschietto, F.; Frisch, M. J.; Scalmani, G.; Ciofini, I.; Adamo, C. Charge  
32 Transfer Excitations in TDDFT: A Ghost-Hunter Index. *J. Comput. Chem.* **2017**, *38*, 2151–2156.
- 33  
34 (54) Savarese, M.; Guido, C. A.; Brémond, E.; Ciofini, I.; Adamo, C. Metrics for Molecular Electronic  
35 Excitations: A Comparison between Orbital- and Density-Based Descriptors. *J. Phys. Chem. A*  
36 **2017**, *121*, 7543–7549.
- 37  
38 (55) Etienne, T.; Assfeld, X.; Monari, A. New Insight into the Topology of Excited States through  
39 Detachment/Attachment Density Matrices-Based Centroids of Charge. *J. Chem. Theory*  
40 *Comput.* **2014**, *10*, 3906–3914.
- 41  
42 (56) Samanta, P. K.; Kim, D.; Coropceanu, V.; Brédas, J. L. Up-Conversion Intersystem Crossing  
43 Rates in Organic Emitters for Thermally Activated Delayed Fluorescence: Impact of the Nature  
44 of Singlet vs Triplet Excited States. *J. Am. Chem. Soc.* **2017**, *139*, 4042–4051.
- 45  
46 (57) Köhler, A.; Beljonne, D. The Singlet–Triplet Exchange Energy in Conjugated Polymers. *Adv.*  
47 *Funct. Mater.* **2004**, *14*, 11–18.
- 48  
49 (58) Richert, S.; Tait, C. E.; Timmel, C. R. Delocalisation of Photoexcited Triplet States Probed by  
50 Transient EPR and Hyperfine Spectroscopy. *J. Magn. Reson.* **2017**, *280*, 103–116.
- 51  
52  
53  
54  
55  
56  
57  
58  
59  
60

- 1  
2  
3 (59) Barone, V. Structure, Magnetic Properties and Reactivities of Open-Shell Species From Density  
4 Functional and Self-Consistent Hybrid Methods. In *Recent Advances in Computational*  
5 *Chemistry*; 1995; pp. 287–334.
- 6  
7 (60) Evans, E. W.; Olivier, Y.; Puttisong, Y.; Myers, W. K.; Hele, T. J. H.; Menke, S. M.; Thomas, T. H.;  
8 Credgington, D.; Beljonne, D.; Friend, R. H.; et al. Vibrationally Assisted Intersystem Crossing  
9 in Benchmark Thermally Activated Delayed Fluorescence Molecules. *J. Phys. Chem. Lett.* **2018**,  
10 *9*, 4053–4058.
- 11  
12 (61) Chen, X.-K.; Tsuchiya, Y.; Ishikawa, Y.; Zhong, C.; Adachi, C.; Brédas, J.-L. A New Design  
13 Strategy for Efficient Thermally Activated Delayed Fluorescence Organic Emitters: From  
14 Twisted to Planar Structures. *Adv. Mater.* **2017**, *29*, 1702767.
- 15  
16 (62) Walker, T. E. H.; Richards, W. G. Molecular Spin–Orbit Coupling Constants. The Role of Core  
17 Polarization. *J. Chem. Phys.* **1970**, *52*, 1311–1314.
- 18  
19 (63) Alberto, M. E.; De Simone, B. C.; Mazzone, G.; Quartarolo, A. D.; Russo, N. Theoretical  
20 Determination of Electronic Spectra and Intersystem Spin–Orbit Coupling: The Case of  
21 Isoindole-BODIPY Dyes. *J. Chem. Theory Comput.* **2014**, *10*, 4006–4013.
- 22  
23 (64) Lower, S. K.; El-Sayed, M. A. The Triplet State and Molecular Electronic Processes in Organic  
24 Molecules. *Chem. Rev.* **1966**, *66*, 199–241.
- 25  
26 (65) M. Klessinger, J. M. *Excited States and Photochemistry of Organic Molecules*; VCH Weinheim,  
27 1995.
- 28  
29 (66) Gibson, J.; Penfold, T. J. Nonadiabatic Coupling Reduces the Activation Energy in Thermally  
30 Activated Delayed Fluorescence. *Phys. Chem. Chem. Phys.* **2017**, *19*, 8428–8434.
- 31  
32 (67) Fan, D.; Yi, Y.; Li, Z.; Liu, W.; Peng, Q.; Shuai, Z. Solvent Effects on the Optical Spectra and  
33 Excited-State Decay of Triphenylamine-Thiadiazole with Hybridized Local Excitation and  
34 Intramolecular Charge Transfer. *J. Phys. Chem. A* **2015**, *119*, 5233–5240.
- 35  
36 (68) Levine, B. G.; Martínez, T. J. Isomerization Through Conical Intersections. *Annu. Rev. Phys.*  
37 *Chem.* **2007**, *58*, 613–634.
- 38  
39 (69) Blancafort, L. Photochemistry and Photophysics at Extended Seams of Conical Intersection.  
40 *ChemPhysChem* **2014**, *15*, 3166–3181.
- 41  
42 (70) Murawski, C.; Leo, K.; Gather, M. C. Efficiency Roll-Off in Organic Light-Emitting Diodes. *Adv.*  
43 *Mater.* **2013**, *25*, 6801–6827.
- 44  
45 (71) Smith, M. B.; Michl, J. Singlet Fission. *Chem. Rev.* **2010**, *110*, 6891–6936.
- 46  
47 (72) Coehoorn, R.; van Eersel, H.; Bobbert, P.; Janssen, R. Kinetic Monte Carlo Study of the  
48 Sensitivity of OLED Efficiency and Lifetime to Materials Parameters. *Adv. Funct. Mater.* **2015**,  
49 *25*, 2024–2037.
- 50  
51 (73) Muccioli, L.; D’Avino, G.; Berardi, R.; Orlandi, S.; Pizzirusso, A.; Ricci, M.; Roscioni, O. M.;

- Zannoni, C. Supramolecular Organization of Functional Organic Materials in the Bulk and at Organic/Organic Interfaces: A Modeling and Computer Simulation Approach. *Top. Curr. Chem.* **2013**, *352*, 39–101.
- (74) Groves, C. Simulating Charge Transport in Organic Semiconductors and Devices: A Review. *Rep. Prog. Phys.* **2017**, *80*, 026502.
- (75) Northey, T.; Stacey, J.; Penfold, T. J. The Role of Solid State Solvation on the Charge Transfer State of a Thermally Activated Delayed Fluorescence Emitter. *J. Mater. Chem. C* **2017**, *5*, 11001–11009.
- (76) Muccioli, L.; D'Avino, G.; Zannoni, C. Simulation of Vapor-Phase Deposition and Growth of a Pentacene Thin Film on C60 (001). *Adv. Mater.* **2011**, *23*, 4532–4536.
- (77) D'Avino, G.; Muccioli, L.; Zannoni, C. From Chiral Islands to Smectic Layers: A Computational Journey Across Sexithiophene Morphologies on C 60. *Adv. Funct. Mater.* **2015**, *25*, 1985–1995.
- (78) Dalal, S. S.; Walters, D. M.; Lyubimov, I.; de Pablo, J. J.; Ediger, M. D. Tunable Molecular Orientation and Elevated Thermal Stability of Vapor-Deposited Organic Semiconductors. *Proc. Natl. Acad. Sci. USA* **2015**, *112*, 4227–4232.
- (79) Friederich, P.; Coehoorn, R.; Wenzel, W. Molecular Origin of the Anisotropic Dye Orientation in Emissive Layers of Organic Light Emitting Diodes. *Chem. Mater.* **2017**, *29*, 9528–9535.
- (80) Tonnelé, C.; Stroet, M.; Caron, B.; Clulow, A. J.; Nagiri, R. C. R.; Malde, A. K.; Burn, P. L.; Gentle, I. R.; Mark, A. E.; Powell, B. J. Elucidating the Spatial Arrangement of Emitter Molecules in Organic Light-Emitting Diode Films. *Angew. Chemie Int. Ed.* **2017**, *56*, 8402–8406.
- (81) Youn, Y.; Yoo, D.; Song, H.; Kang, Y.; Kim, K. Y.; Jeon, S. H.; Cho, Y.; Chae, K.; Han, S. All-Atom Simulation of Molecular Orientation in Vapor-Deposited Organic Light-Emitting Diodes. *J. Mater. Chem. C* **2018**, *6*, 1015–1022.
- (82) Shibata, M.; Sakai, Y.; Yokoyama, D. Advantages and Disadvantages of Vacuum-Deposited and Spin-Coated Amorphous Organic Semiconductor Films for Organic Light-Emitting Diodes. *J. Mater. Chem. C* **2015**, *3*, 11178–11191.
- (83) Ediger, M. D. Perspective: Highly Stable Vapor-Deposited Glasses. *J. Chem. Phys.* **2017**, *147*, 210901.
- (84) Komino, T.; Tanaka, H.; Adachi, C. Selectively Controlled Orientational Order in Linear-Shaped Thermally Activated Delayed Fluorescent Dopants. *Chem. Mater.* **2014**, *26*, 3665–3671.
- (85) Hasegawa, Y.; Yamada, Y.; Sasaki, M.; Hosokai, T.; Nakanotani, H.; Adachi, C. Well-Ordered 4CzIPN ((4s,6s)-2,4,5,6-Tetra(9-H-Carbazol-9-Yl)isophthalonitrile) Layers: Molecular Orientation, Electronic Structure, and Angular Distribution of Photoluminescence. *J. Phys. Chem. Lett.* **2018**, *9*, 863–867.
- (86) Schmidt, T. D.; Lampe, T.; Sylvinson M. R., D.; Djurovich, P. I.; Thompson, M. E.; Brütting, W.

- 1  
2  
3 Emitter Orientation as a Key Parameter in Organic Light-Emitting Diodes. *Phys. Rev. Appl.*  
4 **2017**, *8*, 037001.  
5  
6 (87) Mayr, C.; Lee, S. Y.; Schmidt, T. D.; Yasuda, T.; Adachi, C.; Brütting, W. Efficiency Enhancement  
7 of Organic Light-Emitting Diodes Incorporating a Highly Oriented Thermally Activated Delayed  
8 Fluorescence Emitter. *Adv. Funct. Mater.* **2014**, *24*, 5232–5239.  
9  
10 (88) Lyubimov, I.; Antony, L.; Walters, D. M.; Rodney, D.; Ediger, M. D.; de Pablo, J. J. Orientational  
11 Anisotropy in Simulated Vapor-Deposited Molecular Glasses. *J. Chem. Phys.* **2015**, *143*,  
12 094502.  
13  
14 (89) Alessandri, R.; Uusitalo, J. J.; de Vries, A. H.; Havenith, R. W. A.; Marrink, S. J. Bulk  
15 Heterojunction Morphologies with Atomistic Resolution from Coarse-Grain Solvent  
16 Evaporation Simulations. *J. Am. Chem. Soc.* **2017**, *139*, 3697–3705.  
17  
18 (90) Moral, M.; Son, W.-J.; Sancho-García, J. C.; Olivier, Y.; Muccioli, L. Cost-Effective Force Field  
19 Tailored for Solid-Phase Simulations of OLED Materials. *J. Chem. Theory Comput.* **2015**, *11*,  
20 3383–3392.  
21  
22 (91) Andreussi, O.; Prandi, I. G.; Campetella, M.; Prampolini, G.; Mennucci, B. Classical Force Fields  
23 Tailored for QM Applications: Is It Really a Feasible Strategy? *J. Chem. Theory Comput.* **2017**,  
24 *13*, 4636–4648.  
25  
26 (92) Chen, X. K.; Zhang, S. F.; Fan, J. X.; Ren, A. M. Nature of Highly Efficient Thermally Activated  
27 Delayed Fluorescence in Organic Light-Emitting Diode Emitters: Nonadiabatic Effect between  
28 Excited States. *J. Phys. Chem. C* **2015**, *119*, 9728–9733.  
29  
30 (93) Wang, L.; Prezhdo, O. V.; Beljonne, D. Mixed Quantum-Classical Dynamics for Charge  
31 Transport in Organics. *Phys. Chem. Chem. Phys.* **2015**, *17*, 12395–12406.  
32  
33 (94) Meyer, H.-D.; Manthe, U.; Cederbaum, L. S. The Multi-Configurational Time-Dependent  
34 Hartree Approach. *Chem. Phys. Lett.* **1990**, *165*, 73–78.  
35  
36 (95) Coropceanu, V.; Sánchez-Carrera, R. S.; Paramonov, P.; Day, G. M.; Brédas, J.-L. Interaction of  
37 Charge Carriers with Lattice Vibrations in Organic Molecular Semiconductors: Naphthalene as  
38 a Case Study. *J. Phys. Chem. C* **2009**, *113*, 4679–4686.  
39  
40 (96) Marian, C. M. Mechanism of the Triplet-to-Singlet Upconversion in the Assistant Dopant  
41 ACRXTN. *J. Phys. Chem. C* **2016**, *120*, 3715–3721.  
42  
43 (97) Sun, H.; Hu, Z.; Zhong, C.; Chen, X.; Sun, Z.; Brédas, J.-L. Impact of Dielectric Constant on the  
44 Singlet–Triplet Gap in Thermally Activated Delayed Fluorescence Materials. *J. Phys. Chem.*  
45 *Lett.* **2017**, *8*, 2393–2398.  
46  
47 (98) Nakanotani, H.; Higuchi, T.; Furukawa, T.; Masui, K.; Morimoto, K.; Numata, M.; Tanaka, H.;  
48 Sagara, Y.; Yasuda, T.; Adachi, C. High-Efficiency Organic Light-Emitting Diodes with  
49 Fluorescent Emitters. *Nat. Commun.* **2014**, *5*, 4016.  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60



- 1  
2  
3 (99) Dexter, D. L. A Theory of Sensitized Luminescence in Solids. *J. Chem. Phys.* **1953**, *21*, 836–850.  
4  
5 (100) McConnell, H. M. Intramolecular Charge Transfer in Aromatic Free Radicals. *J. Chem. Phys.*  
6 **1961**, *35*, 508–515.  
7  
8 (101) Beljonne, D.; Curutchet, C.; Scholes, G. D.; Silbey, R. J. Beyond Förster Resonance Energy  
9 Transfer in Biological and Nanoscale Systems. *J. Phys. Chem. B* **2009**, *113*, 6583–6599.  
10  
11 (102) Nakatsuka, S.; Gotoh, H.; Kinoshita, K.; Yasuda, N.; Hatakeyama, T. Divergent Synthesis of  
12 Heteroatom-Centered 4,8,12-Triazatriangulenes. *Angew. Chemie Int. Ed.* **2017**, *56*, 5087–  
13 5090.  
14  
15 (103) Matsui, K.; Oda, S.; Yoshiura, K.; Nakajima, K.; Yasuda, N.; Hatakeyama, T. One-Shot Multiple  
16 Borylation toward BN-Doped Nanographenes. *J. Am. Chem. Soc.* **2018**, *140*, 1195–1198.  
17  
18 (104) Hontz, E.; Chang, W.; Congreve, D. N.; Bulović, V.; Baldo, M. A.; Van Voorhis, T. The Role of  
19 Electron–Hole Separation in Thermally Activated Delayed Fluorescence in Donor–Acceptor  
20 Blends. *J. Phys. Chem. C* **2015**, *119*, 25591–25597.  
21  
22  
23 (105) Huang, Y.; Westenhoff, S.; Avilov, I.; Sreearunothai, P.; Hodgkiss, J. M.; Deleener, C.; Friend, R.  
24 H.; Beljonne, D. Electronic Structures of Interfacial States Formed at Polymeric Semiconductor  
25 Heterojunctions. *Nat. Mater.* **2008**, *7*, 483–489.  
26  
27  
28 (106) Wang, Y.; Sahin-Tiras, K.; Harmon, N. J.; Wohlgenannt, M.; Flatté, M. E. Immense Magnetic  
29 Response of Exciplex Light Emission due to Correlated Spin-Charge Dynamics. *Phys. Rev. X*  
30 **2016**, *6*, 11011.  
31  
32  
33 (107) Wykes, M.; Park, S. K.; Bhattacharyya, S.; Varghese, S.; Kwon, J. E.; Whang, D. R.; Cho, I.;  
34 Wannemacher, R.; Lüer, L.; Park, S. Y.; et al. Excited State Features and Dynamics in a  
35 Distyrylbenzene-Based Mixed Stack Donor–Acceptor Cocrystal with Luminescent Charge  
36 Transfer Characteristics. *J. Phys. Chem. Lett.* **2015**, *6*, 3682–3687.  
37  
38  
39 (108) Wykes, M.; Parambil, R.; Beljonne, D.; Gierschner, J. Vibronic Coupling in Molecular Crystals: A  
40 Franck-Condon Herzberg-Teller Model of H-Aggregate Fluorescence Based on Quantum  
41 Chemical Cluster Calculations. *J. Chem. Phys.* **2015**, *143*, 114116.  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

**ACRONYMS**

OLED:	Organic Light Emitting Diode
TADF:	Thermally Activated Delayed Fluorescence
IQE:	Internal Quantum Efficiency
$T_n$ :	n-th triplet excited state
$S_n$ :	n-th singlet excited state
(R)ISC:	(Reverse) InterSystem Crossing
D:	Electron donating moiety/unit
A:	Electron accepting moiety/unit
CT:	Charge Transfer
LE:	Local Excitation
DFT:	Density Functional Theory
TD-DFT:	Time-Dependent Density Functional Theory
PCM:	polarizable continuum models <sup>^</sup>
$\Delta E_{ST}^V$ :	vertical exchange gap/vertical singlet-triplet energy gap
$\Omega(S_1)$ :	Lowest singlet excited state energy
$\Omega(T_1)$ :	Lowest triplet excited state energy
CCSDT:	Coupled-Cluster with Single, Double, and (iterative) Triple substitutions
CC3:	Third-Order Approximate Coupled-Cluster
EOM-CCSD:	Equation-Of-Motion Coupled-Cluster with Single and Double substitutions
FCI:	Full Configuration Interactions
CC2:	Second-Order Approximate Coupled-Cluster
ADC(2):	Algebraic Diagrammatic Construction at second-order
ADC(3):	Algebraic Diagrammatic Construction at third-order
ADC(2)-x:	eXtended Algebraic Diagrammatic Construction at second-order
RAS:	Restricted Active Space
CASSCF:	Complete Active Space Self-Consistent-Field
CASPT2:	Complete Active Space Perturbation Theory at second-order
GW+BSE:	Green's functions with Bethe-Salpeter Equation
B3LYP:	Becke three-parameter hybrid Lee-Yang-Parr exchange-correlation functional
6-31G*:	Pople's double- $\xi$ basis set with polarization functions on 2nd row atoms
M06-2X:	Minnesota exchange-correlation functional in its 2006 version
TDA-DFT:	Time-Dependent Density Functional Theory in the Tamm-Dancoff approximation
CIS:	Configuration Interaction with Single substitutions

1		
2		
3	TD-HF:	Time-Dependent Hartree-Fock
4	PBE0:	Perdew-Burke-Ernzerhof one-parameter hybrid exchange-correlation functional
5		
6	D3(BJ):	Dispersion correction (third-generation) with Becke-Johnson attenuation function
7		
8	def2-TZVP:	Alhrichs' triple- $\xi$ valence polarization basis set extended with diffuse function
9		
10	CAM-B3LYP:	Coulomb-attenuating method B3LYP exchange-correlation functional
11	$\omega$ B97X:	$\omega$ -dependent range-separated Becke'97 exchange-correlation functional
12		
13	HOMO:	Highest-Occupied Molecular Orbital
14		
15	LUMO:	Lowest-Unoccupied Molecular Orbital
16		
17	UKS:	Unrestricted Kohn-Sham
18	ROKS:	Restricted Open-Shell Kohn-Sham
19		
20	sTDA-DFT:	Simplified Tamm-Dancoff Time-Dependent Density Functional Theory
21		
22	sTDA-xTB:	Simplified Tamm-Dancoff Extended Tight-Binding Hamiltonian
23		
24	MP2:	Second order Møller–Plesset perturbation method
25		
26	MRCI:	Multi-Reference Configuration Interaction
27	KS:	Kohn-Sham orbitals
28		
29	NTO:	Natural Transition Orbitals
30		
31	$\Delta E_{ST}$ :	Singlet-triplet energy gap
32	ZFS:	Zero-Field Splitting
33		
34	ESR:	Electron Spin Resonance
35		
36	2CzPN:	4,5-di(9h-carbazol-9-yl)phthalonitrile
37		
38	4CzIPN:	1,2,3,5-tetrakis(carbazol-9-yl)-4,6-dicyanobenzene
39	(R)IC:	(Reverse) Internal Conversion
40		
41	MD:	Molecular Dynamics
42		
43	MC:	Monte Carlo
44		
45	QM:	Quantum mechanical
46	$S_0$ :	Ground state
47		
48	$^3CT$ :	Triplet charge transfer state
49		
50	$^3LE$ :	Triplet local excitation
51		
52	$\epsilon$ :	Dielectric constant
53	PXZ-OXD :	Phenoxazine- 2,5-diphenyl-1,3,4-oxadiazole
54		
55	TXO-TPA:	2- [4- (diphenylamino) phenyl] - 10, 10- dioxide-9H - thioxanthen-9- one
56		
57		
58		
59		
60		

1  
2  
3 TXO-PhCz: 2- (9- phenyl- 9H-carbazol-3-yl)-10,10- dioxide-9H-thioxanthen-9- one  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60