

## Crystal and Molecular Structure of 5-(Bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene

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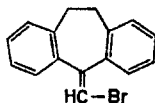
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5-(Bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene has been studied in order to determine the molecular conformation of the pharmacologically active dibenzocycloheptene derivatives. The crystals are orthorhombic with  $a=6.270 \text{ \AA}$ ,  $b=9.288 \text{ \AA}$ ,  $c=22.46 \text{ \AA}$ . The unit cell contains four molecules, and the space group is  $P2_12_12_1$ . The seven-membered ring exists in a boat conformation, the long sides of which are formed by the bonds common with the benzene rings. The planes of the two benzene rings form a dihedral angle of  $76^\circ$ . A possible mechanism of action on the neuronal membrane is discussed.

Numerous tricyclic compounds are used as psychoactive agents. Certain dibenzoheptene derivatives are utilized in the therapy of depressions, whereas phenothiazines of various kinds are used as tranquillizers. The molecules of these two types of tricyclic compounds, which are relatively similar, can thus exhibit opposite pharmacological effects. This has been a reason for extensive speculations on their molecular geometry.

A crystal structure study was undertaken by the present author in order to provide information on the detailed molecular conformation in these active tricyclic compounds. As will be discussed below it is believed that drugs of this type act on the biological membranes, and knowledge on the structure of the drug as well as of the membrane is needed in order to understand their function on a molecular level.

The structural formula of 5-(bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]cycloheptene is:



If the bromine atom is replaced by a  $-\text{CH}_2-\text{CH}_2-\text{N}(\text{CH}_3)_2$  group a wellknown antidepressive agent called amitriptylin is obtained.

## PREPARATION OF CRYSTALS

5-(Bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene was prepared according to a method described by Ebnöther *et al.*,<sup>1</sup> and crystals for X-ray work were grown from ethanol at room temperature. The crystals form needles parallel to the *a*-axis with a cross-section of hexagonal appearance (edges parallel to the *b*-axis and two *bc*-diagonals).

## X-RAY DATA

Rotation and Weissenberg photographs were taken about the *a*- and *b*-axis with a calibrated camera using  $\text{CuK}\alpha$  radiation. The following X-ray data were obtained:

Unit cell: orthorhombic

$a = 6.270 \pm 0.009 \text{ \AA}$ ,  $b = 9.288 \pm 0.012 \text{ \AA}$ ,  $c = 22.46 \pm 0.03 \text{ \AA}$ .

Space group:  $P2_12_12_1$ .

Four molecules per unit cell.

Density calculated:  $1.45 \text{ g.cm}^{-3}$ .

Density measured:  $1.43 \text{ g.cm}^{-3}$ .

The reflection intensities were measured using an on-line automatic scanner developed by Abrahamsson.<sup>2</sup> The corresponding operation programs produce a paper tape with indices and integrated intensity values of the spots on the films which can be used directly as input to the data reduction program. The film factors were plotted against the intensity values for each film pack, and from this plot the most reliable intensity interval was estimated. The final intensity values were then selected from these intervals.

The reflection intensities were corrected for the Lorentz and polarization factors but not for absorption. Absolute values were later obtained by comparison with calculated structure factors.

## STRUCTURE DETERMINATION

The position of the bromine atom in the asymmetric unit was derived from the Patterson function, which was sharpened to correspond to point atoms at rest. The carbon atoms were then located by successive cycles of structure factors calculations and Fourier syntheses.

The structure was refined by block-diagonal and full matrix least-squares treatment using anisotropic temperature factors. When an *R* value of 0.20 was reached a difference synthesis was calculated. All hydrogen atoms could then be located. The hydrogen atoms were included in the following least-squares refinement with calculated positions, as the corresponding peaks in the Fourier maps were too broad to give positions of reasonable accuracy. The hydrogen positions were not refined but recalculated after each round of least-squares refinement. Isotropic temperature factors were used for the hydrogen atoms calculated from the mean-square amplitude tensor of the corresponding hydrogen-carrying atom. When all shifts were less than one-third of the standard deviations the refinement was stopped. The final *R* value for the 611 observed reflections was 0.12.

The scattering curves given in the *International Tables for X-ray Crystallography*, Vol. III (1962), were used. The calculations were performed on the Datasaab D21 computer using a program system developed at this Institute.<sup>3</sup> The weighting scheme applied in the refinement was:

$$w = \frac{1}{1 + [(|F_o| - 8|F_{\min}|)/5|F_{\min}|]^2}$$

## RESULTS AND DISCUSSION

Observed and calculated structure factors are listed in Table 1. Final atomic positions are given in Table 2. The vibration tensor elements for non-hydrogen atoms are given in Table 3. Isotropic temperature factors were used for hydrogen atoms averaged from the mean-square amplitude tensors of the corresponding hydrogen-carrying atom.

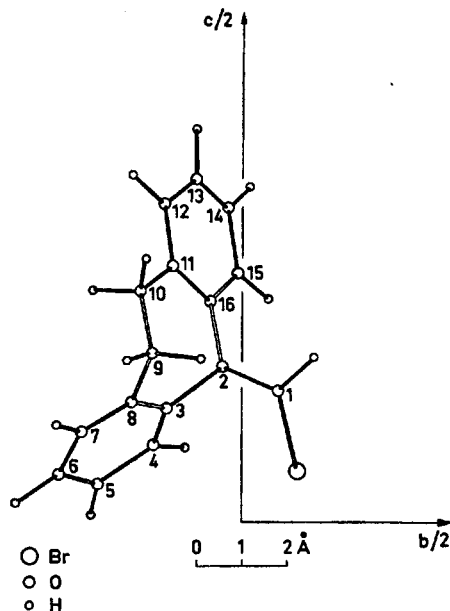


Fig. 1. Stereochemistry of 5-(bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene.

The stereochemistry of the molecule is shown in Fig. 1. The seven-membered ring exists in a boat form although irregular as will be described below, and the long sides are formed by the bonds which are common with the benzene rings. Bond distances and angles are given in Tables 4 and 5. The  $sp^3$ -carbon atoms in the seven-membered ring have an average bond angle of  $114.5^\circ$ . Enlarged angles in seven-membered rings have been discussed by Asher and Sim.<sup>4</sup> The planes of the two benzene rings and adjacent bonds form a dihedral angle of  $76^\circ$ . The geometry of the seven-membered ring is

Table 1. Observed and calculating structure factors ( $\times 100$ ) and phase angles as fraction of one revolution.

h	k	l	Fobs	Fcalc	F1	h	k	l	Fobs	Fcalc	F1	h	k	l	Fobs	Fcalc	F1
0	0	4	1834	1883	1.0000	0	6	6	2240	2110	0.5000	1	4	2	5254	5509	0.0332
0	0	6	9680	9916	0.5000	0	6	7	5023	4791	1.0000	1	4	3	2149	1783	0.6874
0	0	8	7411	7669	0.5000	0	6	8	2362	2153	0.5000	1	4	4	7398	8293	0.9823
0	0	10	10073	10698	0.5000	0	6	10	2266	2266	0.5000	1	4	5	4466	3909	0.0626
0	0	12	5372	5301	0.5000	0	6	11	2350	2365	0.5000	1	4	6	6882	1841	0.3936
0	0	14	3747	3268	0.5000	0	6	13	3344	3479	0.5000	1	4	7	3299	3127	0.6257
0	0	16	2798	1364	1.0000	0	6	14	2486	2360	0.5000	1	4	8	3817	4192	0.9979
0	0	18	1370	3649	1.0000	0	6	15	3052	3298	0.5000	1	4	9	3308	3092	0.4896
0	0	20	4262	3516	1.0000	0	6	20	2161	2489	1.0000	1	4	10	3705	3670	0.1020
0	1	1	868	467	0.2500	0	7	1	2381	2652	0.7500	1	4	11	4955	4197	0.4554
0	1	2	5229	4873	0.2500	0	7	2	1987	1901	0.2500	1	4	12	4203	4395	0.5064
0	1	3	5921	4218	0.2500	0	7	3	1784	1633	0.7500	1	4	13	2614	2513	0.2486
0	1	4	5368	5295	0.2500	0	7	4	6198	6261	0.2500	1	4	14	2629	2487	0.5321
0	1	5	4013	3659	0.2500	0	7	6	4337	5096	0.2500	1	4	15	1860	1571	0.6988
0	1	6	2076	2379	0.2500	0	7	7	1720	958	0.7500	1	5	0	4103	3669	0.7500
0	1	7	2041	2120	0.7500	0	7	8	2183	2270	0.2500	1	5	1	2417	2058	0.3138
0	1	8	2635	2528	0.2500	0	7	9	2031	2125	0.2500	1	5	2	2680	2949	0.6827
0	1	9	8265	9094	0.7500	0	7	11	2489	2335	0.2500	1	5	3	6439	6318	0.2522
0	1	11	2397	2901	0.2500	0	7	12	2170	2411	0.7500	1	5	4	2119	1404	0.8762
0	1	11	3951	4230	0.7500	0	7	14	4179	4179	0.2500	1	5	5	6600	6362	0.2666
0	1	12	6596	6370	0.7500	0	7	16	3644	3696	0.7500	1	5	6	6042	5813	0.2374
0	1	13	5978	4439	0.7500	0	8	0	3988	3613	1.0000	1	5	8	1787	1483	0.2452
0	1	14	3660	3424	0.7500	0	8	2	3469	3672	1.0000	1	5	9	1937	1321	0.1936
0	1	15	2349	2181	0.2500	0	8	3	2859	2370	0.5000	1	5	10	2693	2636	0.2992
0	1	16	3144	3730	0.7500	0	8	4	1955	1794	1.0000	1	5	11	2609	2652	0.7830
0	1	17	2251	2650	0.2500	0	8	5	3184	3099	0.5000	1	5	12	2614	2934	0.2725
0	1	17	1475	1406	0.2500	0	8	6	1987	1881	0.5000	1	5	13	5915	5501	0.7614
0	1	19	2981	2440	0.2500	0	8	7	3312	3312	0.2500	1	5	14	2889	2487	0.2826
0	1	20	1571	1324	0.2500	0	8	8	2252	2262	0.5000	1	5	15	3705	4480	0.7878
0	1	21	4088	4577	0.2500	0	9	1	3466	3956	0.2500	1	6	1	7409	6924	0.0297
0	1	22	871	3939	0.5000	0	10	2	2010	2271	0.5000	1	6	2	3940	3317	0.0498
0	1	22	1	2534	1.0000	1	0	2	1886	1886	0.5000	1	6	3	2771	2836	0.8436
0	1	22	3369	2194	0.5000	1	0	2	13490	13490	0.5000	1	6	7	2725	3343	0.5255
0	1	22	15104	15634	0.5000	1	0	3	1569	1043	0.7499	1	6	9	4453	4998	0.4906
0	1	22	3027	2199	0.2500	1	0	4	15971	16631	0.2500	1	6	11	5195	4850	0.4674
0	1	22	17823	15823	0.5000	1	0	5	8023	8023	0.5000	1	6	11	1979	1907	0.1919
0	1	22	3201	2197	1.0000	1	0	6	8431	8849	0.5000	1	6	19	1047	3161	0.9854
0	1	22	9634	8297	0.5000	1	0	8	2622	4465	0.5000	1	6	21	1707	1899	0.9950
0	1	22	1118	1323	0.2500	1	0	10	4927	4927	0.5000	1	6	21	5633	5096	0.2500
0	1	22	3451	3639	0.5000	1	0	12	6389	5757	1.0000	1	7	0	2649	2095	0.2500
0	1	22	1519	990	1.0000	1	0	14	5810	6709	1.0000	1	7	2	5478	4842	0.2070
0	1	22	1373	7129	1.0000	1	0	16	3788	4762	1.0000	1	7	4	2968	2407	0.1909
0	1	22	1644	3359	1.0000	1	1	0	1212	1341	0.2500	1	7	6	2417	1972	0.7480
0	1	22	1714	1332	0.5000	1	1	1	2511	2102	0.1917	1	7	6	4335	4282	0.7849
0	1	22	4188	3972	1.0000	1	1	2	3777	4318	0.4532	1	7	9	1645	1959	0.2759
0	1	22	2156	1704	0.5000	1	1	2	4603	6282	0.2237	1	7	10	5039	5056	0.7598
0	1	22	2205	1934	1.0000	1	1	3	2213	3319	0.2500	1	7	12	2645	2400	0.7646
0	1	22	1310	1112	1.0000	1	1	5	5604	6579	0.6780	1	7	12	2095	2095	0.7985
0	1	22	3329	3212	0.5000	1	1	6	2842	1883	0.7764	1	7	15	2124	1859	0.7067
0	1	22	6077	7138	0.2500	1	1	6	6029	7161	0.6797	1	7	16	1840	2084	0.3004
0	1	22	1216	213	0.7497	1	1	9	3190	2607	0.7905	1	7	16	2319	2628	0.2691
0	1	22	9731	9216	0.7500	1	1	10	4811	5749	0.7247	1	8	3	2517	2622	0.5355
0	1	22	1514	1831	0.7500	1	1	11	3988	3617	0.2963	1	8	4	4057	3874	0.5053
0	1	22	2743	2860	0.7500	1	1	12	2674	3640	0.7644	1	8	6	4291	4029	0.2500
0	1	22	5873	3451	0.7500	1	1	13	5505	5392	0.1987	1	8	8	2850	3074	0.0396
0	1	22	1644	3636	1.0000	1	1	15	3330	3338	0.2800	1	8	8	1997	2346	0.5163
0	1	22	4348	4057	0.7500	1	1	16	2456	1859	0.1287	1	8	9	2048	2337	0.0443
0	1	22	3664	3957	0.2500	1	1	18	3149	2852	0.2235	1	8	16	2539	2970	0.0038
0	1	22	2462	2987	0.7500	1	1	20	3068	2646	0.2714	1	9	1	2338	1622	0.6208
0	1	22	5671	5323	0.2500	1	1	23	2826	2693	0.7910	1	9	2	2460	2141	0.3232
0	1	22	3787	4034	0.2500	1	1	25	2240	2999	0.7286	1	9	3	2303	3293	0.7827
0	1	22	3330	3490	0.2500	1	2	0	1967	1938	0.7500	1	9	4	1545	1321	0.2735
0	1	22	1833	1438	0.2500	1	2	1	15171	15180	0.4199	1	9	5	4697	4798	0.7374
0	1	22	9138	8731	0.5000	1	2	2	1778	1925	0.3754	1	9	7	3088	3137	0.7232
0	1	22	1573	909	0.5000	1	2	4	6001	6749	0.0890	1	9	13	2018	2522	0.2681
0	1	22	10908	10032	0.5000	1	2	5	2135	1168	0.5945	1	10	6	1598	1608	0.0315
0	1	22	2569	2300	0.7500	1	2	6	2507	1944	0.3479	1	10	6	2337	1608	0.2500
0	1	22	1419	1306	1.0000	1	2	7	4623	4807	0.2418	1	10	8	1504	376	0.2500
0	1	22	2965	2390	1.0000	1	2	8	2800	1904	0.4150	1	10	8	3504	3435	0.2500
0	1	22	5872	3636	1.0000	1	2	9	10683	10643	0.0272	1	10	6	1158	1686	0.5000
0	1	22	7050	7131	1.0000	1	2	10	2454	2957	0.7365	1	10	9	2112	2634	0.7500
0	1	22	7245	6888	1.0000	1	2	11	6511	6877	0.0036	1	10	8	3725	4699	1.0000
0	1	22	897	898	0.5000	1	2	12	2263	1664	0.7609	1	10	9	2456	2493	0.7500
0	1	22	4923	5069	1.0000	1	2	13	4445	5055	0.0280	1	10	10	7333	7641	1.0000
0	1	22	2526	3038	0.5000	1	2	14	2658	1939	0.6486	1	10	11	1516	1619	0.7500
0	1	22	2173	2049	1.0000	1	2	17	3636	3283	0.5171	1	10	12	7800	7233	1.0000
0	1	22	2303	2295	0.5000	1	2	19	2134	2307	0.4734	1	10	14	1629	1707	1.0000
0	1	22	3136	3014	0.5000	1	2	21	4525	4059	0.4887	1	10	18	3704	3996	0.5000
0	1	22	4004	3410	0.5000	1	3	0	10738	11141	0.7800	1	10	20	3530	3768	0.5000
0	1	22	8254	7926	0.7500	1	3	1	1421	982	0.4888	1	10	20	2106	1236	1.0000
0	1	22	1114	782	0.7499	1	3	2	5186	5531	0.7173	1	10	1	9268	8865	0.7180
0	1	22	1790	2129	0.7500	1	3	3	6499	6514	0.7547	1	10				

STRUCTURE OF DIBENZOCYCLOHEPTENES

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Table 1. Continued.

h	k	l	Fobs	Fcalc	F1	h	k	l	Fobs	Fcalc	F1	h	k	l	Fobs	Fcalc	F1	
2	2	2	2198	1356	0.4667	3	1	12	3673	3531	0.2926	4	0	12	4198	3884	0.5000	
2	2	2	6948	7257	0.9915	3	1	13	3924	3604	0.8169	4	0	10	2404	2084	1.0000	
2	2	2	2460	2150	0.8654	3	1	15	2757	2438	0.7278	4	1	1	3113	1950	0.2920	
2	2	2	6556	7563	0.0028	3	1	0	2597	2998	0.7500	4	1	2	3415	2468	0.2082	
2	2	2	2132	2365	0.5761	3	2	1	4307	5225	0.0754	4	1	4	2977	3711	0.1448	
2	2	2	7011	7197	0.0098	3	2	2	2467	1921	0.8633	4	1	3	4258	2454	0.1877	
2	2	2	4276	3760	0.6886	3	2	3	5041	4785	0.9775	4	1	4	3451	3172	0.4207	
2	2	2	4649	4455	0.9959	3	2	4	1897	1516	0.7763	4	1	8	3884	2943	0.3641	
2	2	2	1784	1785	0.8101	3	2	5	2677	2344	0.2091	4	1	10	2650	3259	0.4821	
2	2	2	3502	2662	0.5111	3	2	6	1696	1488	0.4401	4	1	9	4875	4320	0.7518	
2	2	2	5702	5144	0.5310	3	2	7	4310	3712	0.5431	4	1	11	2198	1962	0.7971	
2	2	2	2418	2314	0.4054	3	2	9	4812	4808	0.4950	4	1	12	2594	3028	0.7278	
2	2	2	5916	5209	0.4813	3	2	10	1652	1654	0.2290	4	1	13	3224	2794	0.5954	
2	2	2	2644	2561	0.0428	3	2	11	4324	4637	0.4763	4	2	0	1416	1616	0.4162	
2	2	2	2112	2323	0.5174	3	2	13	3511	3075	0.5416	4	2	2	1352	1490	0.3952	
2	2	2	3452	3592	0.8093	3	2	14	2123	1448	0.3040	4	2	2	3739	3501	0.5170	
2	2	2	4781	4624	0.3174	3	2	15	2140	2124	0.9271	4	2	2	2191	2249	0.2226	
2	2	2	6600	5235	0.8501	3	2	16	2195	1787	0.9802	4	2	2	4541	4230	0.5332	
2	2	2	8629	9100	0.2747	3	2	17	2124	1785	0.0220	4	2	5	1796	1974	0.1944	
2	2	2	4906	4121	0.2796	3	2	18	2007	1291	0.2845	4	2	6	3559	3099	0.4678	
2	2	2	5725	5209	0.2024	3	2	19	1751	1742	0.2500	4	2	7	1935	1298	0.5087	
2	2	2	3551	3724	0.1176	3	2	20	2069	2023	0.1379	4	2	9	825	1161	0.5186	
2	2	2	4120	3834	0.3206	3	2	2	4801	4604	0.2348	4	2	10	2269	2314	0.9927	
2	2	2	2019	1916	0.0916	3	2	3	2948	2432	0.1486	4	2	11	2828	2891	0.8927	
2	2	2	1564	2260	0.2750	3	2	4	2062	2068	0.1048	4	2	13	1314	1616	0.2000	
2	2	2	3137	3253	0.7347	3	2	5	1974	2944	0.2992	4	2	14	2470	1627	0.2237	
2	2	2	1752	2219	0.3205	3	2	6	2189	2834	0.8463	4	2	15	2842	3080	0.8527	
2	2	2	3164	3371	0.7260	3	2	7	2352	2361	0.2517	4	2	16	1814	1972	0.3741	
2	2	2	4474	4997	0.7444	3	2	8	3432	3450	0.8054	4	2	17	4126	3616	0.7338	
2	2	2	9304	9769	1.0000	3	2	9	1637	1501	0.4111	4	2	18	1082	1546	0.4794	
2	2	2	3863	3194	0.9778	3	2	10	3693	3717	0.7657	4	2	19	2508	2872	0.7688	
2	2	2	1956	1956	0.6723	3	2	11	1716	2338	0.6180	4	2	20	2219	2759	0.6560	
2	2	2	4271	3573	0.0422	3	2	12	1716	1716	0.0915	4	2	21	1752	1752	0.5251	
2	2	2	1686	1472	0.4797	3	2	14	3347	2615	0.6920	4	2	22	1190	1537	0.5625	
2	2	2	4446	3919	0.4174	3	2	15	2060	2320	0.6849	4	2	23	1737	2263	0.5686	
2	2	2	1372	961	0.2133	3	2	16	2254	2021	0.3707	4	2	24	1786	2621	0.7507	
2	2	2	4155	3273	0.5133	3	2	17	2213	2072	0.3707	4	2	25	1582	1616	0.2065	
2	2	2	1404	596	0.1601	3	2	18	2109	2253	0.3117	4	2	26	1314	1616	0.2065	
2	2	2	6892	5950	0.4595	3	2	20	1895	2253	0.2916	4	2	27	2815	3257	0.3285	
2	2	2	3658	3658	0.1134	3	2	0	1219	1170	0.2500	4	2	28	2536	2616	0.2265	
2	2	2	5356	5013	0.5071	3	2	1	1792	1743	0.2297	4	2	29	1879	2181	0.5000	
2	2	2	2706	2208	0.1415	3	2	2	5366	4233	0.4928	4	2	30	1333	2142	0.1921	
2	2	2	1641	1783	0.4884	3	2	3	1563	1095	0.2858	4	2	31	4390	4708	0.5251	
2	2	2	1949	2492	0.9703	3	2	4	1642	1898	0.2538	4	2	32	1710	2353	0.2810	
2	2	2	1889	2777	0.9592	3	2	5	3664	3679	0.5163	4	2	33	1956	3003	0.0115	
2	2	2	1886	2618	0.1814	3	2	6	2200	2577	0.1416	4	2	34	2069	3068	0.5699	
2	2	2	1177	1633	1.0000	3	2	7	2129	2463	0.3923	4	2	35	2231	2608	0.0708	
2	2	2	6577	6428	0.2367	3	2	8	2267	2625	0.0492	4	2	36	1489	1455	0.6349	
2	2	2	3371	2883	0.2655	3	2	9	2120	1369	0.7095	4	2	37	1510	2208	0.7432	
2	2	2	3368	3336	0.6825	3	2	10	2129	2402	0.9624	4	2	38	2765	2765	0.3466	
2	2	2	2525	2414	0.1591	3	2	11	1693	1693	0.9939	4	2	39	2488	3037	0.2021	
2	2	2	1964	1675	0.1687	3	2	12	3605	4218	0.9837	4	2	40	3697	3950	0.5000	
2	2	2	2950	3019	0.3151	3	2	13	1720	2047	0.5465	4	2	41	52	106	0.2500	
2	2	2	4116	4116	0.1000	3	2	14	2220	2664	0.5997	4	2	42	2912	2912	0.5000	
2	2	2	7330	6828	0.7746	3	2	15	1631	1628	0.9900	4	2	43	2024	1640	0.2500	
2	2	2	4018	4029	0.7258	3	2	16	1897	1957	0.8036	4	2	44	1660	1890	0.2500	
2	2	2	1914	1975	0.8002	3	2	17	2168	2610	0.3354	4	2	45	1950	1693	0.3073	
2	2	2	1740	1740	0.7204	3	2	18	1878	1158	0.7646	4	2	46	2843	3212	0.7627	
2	2	2	1948	1935	0.5000	3	2	19	2148	1265	0.3078	4	2	47	1952	2817	0.5195	
2	2	2	1619	2330	0.3758	3	2	20	5929	5264	0.7899	4	2	48	2807	1789	0.4932	
2	2	2	2839	2937	0.3725	3	2	21	2365	2583	0.6259	4	2	49	2843	3212	0.7627	
2	2	2	4824	4386	0.4743	3	2	22	5743	5158	0.7424	4	2	50	1952	2817	0.5195	
2	2	2	4016	3763	0.5360	3	2	23	1600	1513	0.7877	4	2	51	2063	1699	0.6900	
2	2	2	7781	4206	0.5020	3	2	24	2017	1726	0.6198	4	2	52	1694	2061	0.5851	
2	2	2	1463	1558	0.5383	3	2	25	2267	2708	0.3738	4	2	53	1762	2715	0.7049	
2	2	2	2759	3737	0.3942	3	2	26	2120	2252	0.0168	4	2	54	2912	2912	0.5000	
2	2	2	2301	3030	0.9792	3	2	27	1465	1210	0.8045	4	2	55	1660	1890	0.2500	
2	2	2	3134	3334	0.1939	3	2	28	2265	2258	0.2224	4	2	56	2024	1640	0.2500	
2	2	2	2688	2928	0.5374	3	2	29	3371	3079	0.2762	4	2	57	1660	1890	0.2500	
2	2	2	1481	1483	0.7493	3	2	30	1375	1668	0.2500	4	2	58	1660	1890	0.2500	
2	2	2	4828	5045	0.7426	3	2	31	5454	4174	0.4285	4	2	59	1571	1504	0.3039	
2	2	2	3277	4010	0.5713	3	2	32	2336	2557	0.6515	4	2	60	3643	3009	0.4122	
2	2	2	3401	3996	0.7078	3	2	33	2579	2564	0.0168	4	2	61	1510	845	0.2186	
2	2	2	3964	3709	1.0000	3	2	34	989	411	0.0099	4	2	62	2877	2850	0.6378	
2	2	2	2504	2370	0.7500	3	2	35	1662	1475	0.7464	4	2	63	1842	1812	0.0011	
2	2	2	5077	4441	1.0000	3	2	36	1879	1869	0.8248	4	2	64	2098	1624	0.2230	
2	2	2	6121	5483	1.0000	3	2	37	1863	1078	0.5491	4	2	65	1442	1624	0.5100	
2	2	2	1406	702	0.2500	3	2	38	2296	2805	0.0316	4	2	66	1948	2049	0.9582	
2	2	2	2130	1442	1.0000	3	2	39	2870	3710	0.9681	4	2	67	641	1135	0.8328	
2	2	2	11	1652	1541	0.7500	3	2	40	1947	1608	0.0961	4	2	68	1627	2040	0.9695
2	2	2	4641	4185	0.5000	3	2	41	2461	2948	0.7500	4	2	69	1714	1777	0.9617	
2	2	2	2382	1851	0.2500	3	2	42	1808	1962	0.7422	4	2	70	812	1681	0.5317	

quite unsymmetrical compared to those found in other compounds (*cf.* Ref. 5). This is a consequence of the steric requirements due to the bond situation in the ring. The carbon atom C2 is situated within 0.02 Å in the intersection of the planes of the two benzene rings, whereas the carbon atoms C9 and C10 are situated in the planes of the benzene rings C3–C8 and C11–C16, respectively, (distances 0.01 Å and 0.04 Å). The carbon atom C9 is located 0.95 Å from the best least-squares plane through the four carbon atoms of the seven-membered ring which are common to the benzene rings, and the corresponding distance for C10 is 0.19 Å.

*Table 2.* Fractional coordinates for the atoms of one molecule. The numbering for non-hydrogen atoms is shown in Fig. 1. The hydrogen atoms are numbered in the same way as their parent atom with the addition of one last digit to indicate the number attached to the same parent atom.

	<i>x</i>	<i>y</i>	<i>z</i>
Br	0.2312	1.1352	0.0505
C(1)	0.2753	1.0849	0.1284
C(2)	0.2529	0.9502	0.1515
C(3)	0.2454	0.8176	0.1113
C(4)	0.0939	0.7873	0.0754
C(5)	0.0995	0.6593	0.0363
C(6)	0.2998	0.5666	0.0442
C(7)	0.4602	0.6159	0.0862
C(8)	0.4312	0.7422	0.1178
C(9)	0.6183	0.7849	0.1653
C(10)	0.5312	0.7556	0.2254
C(11)	0.3426	0.8351	0.2501
C(12)	0.2599	0.8151	0.3108
C(13)	0.0993	0.8915	0.3338
C(14)	-0.0228	0.9701	0.3040
C(15)	0.0144	0.9932	0.2444
C(16)	0.1889	0.9248	0.2169
H(11)	0.326	1.169	0.158
H(41)	-0.043	0.860	0.073
H(51)	-0.026	0.638	0.004
H(61)	0.324	0.465	0.021
H(71)	0.605	0.553	0.092
H(91)	0.758	0.721	0.157
H(92)	0.654	0.899	0.161
H(101)	0.484	0.640	0.225
H(102)	0.661	0.769	0.256
H(121)	0.337	0.733	0.338
H(131)	0.071	0.884	0.381
H(141)	-0.156	1.021	0.325
H(151)	-0.090	1.061	0.218

The C–Br bond seems to be significantly shorter than olefinic carbon-bromine distances in general ( $1.89 \pm 0.01$  Å according to *The International Tables for X-Ray Crystallography*, Vol. III (1962) p. 273). This might be an inductive effect due to the vicinity of the  $\pi$ -electron orbitals of the conjugated systems of the molecule.

Table 3. Vibration parameters. Anisotropic temperature factors were used in the refinement according to  $\exp[-2^2(h^2a^2U_{11} + k^2b^2U_{22} + l^2c^2U_{33} + 2klbcU_{23} + 2hklacU_{13} + 2hkabU_{12})]$ . The tensor elements in  $\text{\AA}^2$  have been multiplied by 1000.

	$U_{11}$	$U_{22}$	$U_{33}$	$U_{23}$	$U_{31}$	$U_{12}$
Br	712	257	417	109	81	17
C(1)	1108	156	401	95	-1064	13
C(2)	820	342	233	-69	-304	663
C(3)	641	227	-68	38	279	21
C(4)	725	491	19	206	-23	34
C(5)	889	219	402	-86	-571	7
C(6)	1335	295	1357	301	1880	457
C(7)	341	572	327	-110	13	-28
C(8)	97	187	534	142	-198	525
C(9)	809	449	486	44	8	-92
C(10)	-350	1226	389	487	149	307
C(11)	1067	362	285	-77	203	-259
C(12)	1207	652	197	7	964	95
C(13)	430	860	751	650	-98	-106
C(14)	-101	973	502	-336	123	112
C(15)	-266	304	334	-199	-174	37
C(16)	1388	271	-16	-11	-517	-94

Table 4. Bond lengths in  $\text{\AA}$  (with standard deviations  $\times 100$ ). Atom numbering according to Fig. 1.

Br	-C(1)	1.83(2)
C(1)	-C(2)	1.36(3)
C(2)	-C(3)	1.53(3)
C(2)	-C(16)	1.54(3)
C(3)	-C(4)	1.28(4)
C(4)	-C(5)	1.48(3)
C(5)	-C(6)	1.53(4)
C(6)	-C(7)	1.45(4)
C(7)	-C(8)	1.38(4)
C(3)	-C(8)	1.37(4)
C(8)	-C(9)	1.63(4)
C(9)	-C(10)	1.48(3)
C(10)	-C(11)	1.50(4)
C(11)	-C(12)	1.47(3)
C(12)	-C(13)	1.34(4)
C(13)	-C(14)	1.26(4)
C(14)	-C(15)	1.37(4)
C(15)	-C(16)	1.41(4)
C(11)	-C(16)	1.47(3)

The molecular packing viewed along the  $a$ -axis is shown in Fig. 2. Only forces of van der Waals type determine the intermolecular packing. There are eight hydrogen-hydrogen distances less than 3  $\text{\AA}$  per molecule.

It is believed that the effect of this type of tricyclic compounds is due to their ability to inhibit the uptake in the neuronal membrane of nor-

Table 5. Bond angles in degrees (with standard deviations). Atom numbering according to Fig. 1.

Br—C(1) —C(2)	125.7(1.7)
C(1) —C(2) —C(3)	121.2(1.3)
C(1) —C(2) —C(16)	122.1(2.0)
C(3) —C(2) —C(16)	115.6(1.8)
C(2) —C(3) —C(4)	125.0(2.9)
C(2) —C(3) —C(8)	108.8(2.6)
C(4) —C(3) —C(8)	126.1(2.1)
C(3) —C(4) —C(5)	122.3(2.7)
C(4) —C(5) —C(6)	113.7(2.4)
C(5) —C(6) —C(7)	117.8(2.3)
C(6) —C(7) —C(8)	120.6(2.5)
C(7) —C(8) —C(9)	116.5(2.1)
C(7) —C(8) —C(3)	119.5(2.3)
C(3) —C(8) —C(9)	123.9(2.1)
C(8) —C(9) —C(10)	106.6(2.4)
C(9) —C(10) —C(11)	122.5(2.5)
C(10) —C(11) —C(12)	123.9(2.5)
C(10) —C(11) —C(16)	127.3(2.1)
C(16) —C(11) —C(12)	108.0(2.8)
C(11) —C(12) —C(13)	123.9(2.9)
C(12) —C(13) —C(14)	124.2(2.9)
C(13) —C(14) —C(15)	120.6(2.6)
C(14) —C(15) —C(16)	119.3(2.2)
C(15) —C(16) —C(2)	123.5(2.5)
C(15) —C(16) —C(11)	122.7(2.9)
C(11) —C(16) —C(2)	113.4(2.8)

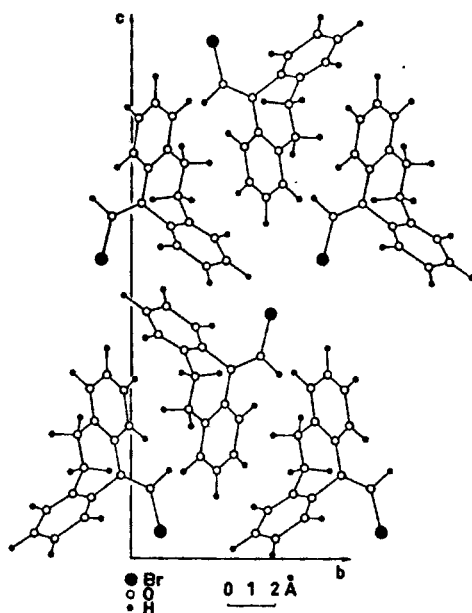


Fig. 2. Molecular packing of 5(bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]cycloheptene as seen along the *a*-axis.



epinephrine (*cf.* Ref. 6). Furthermore a study of penetration of lipid monolayers by psychoactive drugs indicate that the drug molecules are attached to the lipids of natural membranes. On the assumption that the lipids in the membrane form a bimolecular layer where the chains possess a liquid type of disorder a possible mechanism might be that the rigid skeleton of the drug molecules promote the formation of a more crystalline structure of the lipids and thus reduce the membrane permeability. The present author has started measurements of the effect of this compound with known molecular dimensions on the structure of lipid surface films.

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