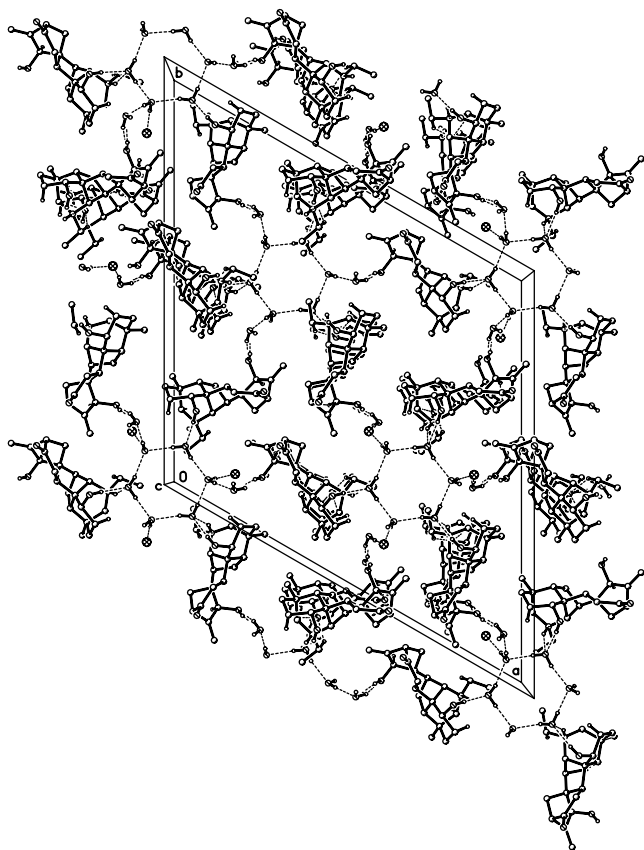
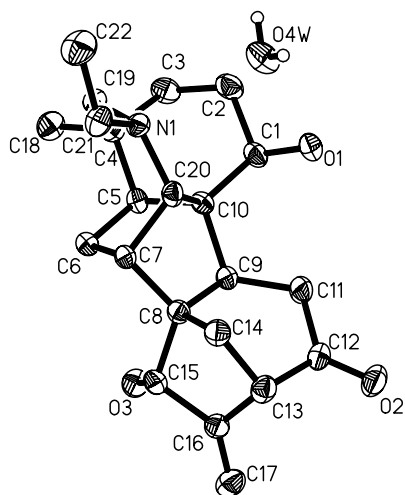


Crystal structure of songorine chloride trihydrate, $[C_{22}H_{32}NO_3]Cl \cdot 3 H_2O$

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Abstract

$C_{22}H_{38}ClNO_6$, trigonal, $R3$ (no. 146), $a = 28.293(2) \text{ \AA}$, $c = 7.4204(7) \text{ \AA}$, $V = 5144.3 \text{ \AA}^3$, $Z = 9$, $R_{gt}(F) = 0.051$, $wR_{ref}(F^2) = 0.137$, $T = 296 \text{ K}$.

Source of material

Songorine, a C_{20} -diterpenoid alkaloid is isolated from the chloroform extract of the roots of *Aconitum szechenyianum* Gay, wetted by 10% ammonium hydroxide solution before. The purified compound is dissolved in a 1% hydrochloric acid solution at first. Then the solution is adjusted to pH 6.5 with a 10% NaOH solution. Colorless single crystals were obtained at room temperature for days by slowly volatilizing the solvent.

Experimental details

Water H atoms were located in a difference Fourier map and refined with restrained O—H bond lengths (0.85(1) \AA) and fixed isotropic displacement parameters (0.08 \AA^2). Other H atoms were placed at calculated positions and refined using a riding model with O—H distances restrained to 0.82 \AA , C—H distances restrained to 0.93 \AA and N—H distance restrained to 0.91 \AA .

Discussion

Songorine belongs to the class of C_{20} -diterpenoid alkaloids, a large group of biologically active compounds of naturally occurrence. As a diterpenoid alkaloid, songorine has been found to enhance both the orthodromic population spike and field excitatory postsynaptic potential (EPSP) in CA1 region of hippocampal slices [1]. Recently, it was indicated that songorine can enhance the excitatory synaptic transmission in rat hippocampus, and was proved that songorine is a novel non-competitive antagonist at the GABA_A receptor in rat brain [2].

In the title crystal structure, there are one songorine cation, one chloride ion and 3 crystal water molecules (figure top). The bond length of C12—O2 is 1.202 \AA , and the angles of O2—C12—C11, O2—C12—C13 and C11—C12—C13 are 121.8(4) $^\circ$, 122.8(4) $^\circ$ and 115.4(3) $^\circ$, respectively, which indicates that the C12 and O2 are sp^2 hybridized, and C11, C12, C13 and O2 atoms are in a plane. The bond length of C17—C16 is 1.322(6) \AA . The angles of C13—C16—C15, C17—C16—C15 and C17—C16—C13 are 108.0(3) $^\circ$, 126.2(4) $^\circ$ and 125.5(4) $^\circ$, respectively, which show an exomethylene attached to C16.

The songorine cations are interlinked by hydrogen bonds (figure bottom). Intra-molecular hydrogen bonds O6W—H6WA...C11 and O5W—H5WA...C11 link the Cl^- , crystal water O5W and O6W together. The hydrogen bonds C2—H2A...O4W, O1—H1...O4W, N1—H1N...O4W fix the crystal water O4W to songorine. The intermolecular hydrogen bonds O6W—H6WB...O5W (symmetric code: $x, y, z-1$) and C21—H21A...O1, C19—H19B...O1, C15—H15...O2 (symmetric code: $x, y, z+1$), link the songorine molecules along [001]. The hydrogen bonds O5W—

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H5WB...O3 (symmetric code: $x, y+1, z$) and O3–H3...O5W (symmetric code: $x, y-1, z$) connect the molecules along [010]. The hydrogen bond O4W–H4WA...O6W (symmetric code: $-y+2/3, x-y+1/3, z-2/3$) links the molecules into triangles at the ab plane. Thus the hydrogen bonds form a network structure (figure, bottom). Based on the parameter Flack [3], $x = 0.00(9)$, the absolute configuration of the ten chiral carbons in the title crystal structure are identified as 1*R*, 4*S*, 5*S*, 7*S*, 8*S*, 9*S*, 10*R*, 13*S*, 15*R* and 20*S*.

Table 1. Data collection and handling.

Crystal:	colorless prism, size 0.14 × 0.21 × 0.40 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	2.05 cm ⁻¹
Diffractometer, scan mode:	Bruker SMART, φ/ω
$2\theta_{\max}$:	50.18°
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}$:	8777, 4031
Criterion for $I_{\text{obs}}, N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 2777
$N(\text{param})_{\text{refined}}$:	321
Programs:	SHELXS-97 [4], SHELXL-97 [5], SHELXTL [6]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(1N)	9 <i>b</i>	0.3836	0.8233	0.0177	0.051
H(1)	9 <i>b</i>	0.4050	0.8492	0.3662	0.082
H(3)	9 <i>b</i>	0.5564	1.0917	0.0285	0.078
H(1A)	9 <i>b</i>	0.4001	0.9386	0.3521	0.044
H(2A)	9 <i>b</i>	0.3313	0.8353	0.2019	0.064
H(2B)	9 <i>b</i>	0.3152	0.8667	0.3401	0.064
H(3A)	9 <i>b</i>	0.3196	0.9261	0.1231	0.056
H(3B)	9 <i>b</i>	0.2827	0.8670	0.0482	0.078
H(5)	9 <i>b</i>	0.4138	0.9857	0.0073	0.046
H(6A)	9 <i>b</i>	0.4360	0.9465	-0.2981	0.050
H(6B)	9 <i>b</i>	0.4769	1.0035	-0.2100	0.039
H(7)	9 <i>b</i>	0.5051	0.9286	-0.2143	0.057
H(9)	9 <i>b</i>	0.4942	1.0142	0.1701	0.044
H(11A)	9 <i>b</i>	0.5039	0.9931	0.4497	0.052
H(11B)	9 <i>b</i>	0.5013	0.9386	0.3939	0.105
H(13)	9 <i>b</i>	0.6433	1.0091	0.2291	0.074
H(14A)	9 <i>b</i>	0.5912	0.9574	-0.0169	0.084
H(14B)	9 <i>b</i>	0.5520	0.9269	0.1474	0.059
H(15)	9 <i>b</i>	0.5899	1.0371	-0.1566	0.053

Table 2. Continued.

Atom	Site	x	y	z	U_{iso}
H(17A)	9 <i>b</i>	0.6734	1.1170	0.2481	0.080
H(17B)	9 <i>b</i>	0.6528	1.1373	0.0779	0.091
H(18A)	9 <i>b</i>	0.3295	0.9579	-0.1888	0.089
H(18B)	9 <i>b</i>	0.2957	0.8979	-0.2603	0.110
H(18C)	9 <i>b</i>	0.3544	0.9370	-0.3362	0.074
H(19A)	9 <i>b</i>	0.3229	0.8265	-0.1419	0.049
H(19B)	9 <i>b</i>	0.3654	0.8639	-0.2861	0.038
H(20)	9 <i>b</i>	0.4708	0.8827	0.0840	0.040
H(21A)	9 <i>b</i>	0.4321	0.8393	-0.3130	0.080
H(21B)	9 <i>b</i>	0.4471	0.8137	-0.1512	0.082
H(22A)	9 <i>b</i>	0.3845	0.7451	-0.3367	0.075
H(22B)	9 <i>b</i>	0.3431	0.7660	-0.3087	0.100
H(22C)	9 <i>b</i>	0.3582	0.7404	-0.1470	0.081
H(4WA)	9 <i>b</i>	0.395(2)	0.760(2)	0.266(8)	0.080
H(4WB)	9 <i>b</i>	0.3455(4)	0.759(2)	0.24(1)	0.080
H(5WA)	9 <i>b</i>	0.575(2)	0.147(2)	0.319(5)	0.080
H(5WB)	9 <i>b</i>	0.549(2)	0.118(1)	0.163(5)	0.080
H(6WA)	9 <i>b</i>	0.613(2)	0.205(2)	0.861(3)	0.080
H(6WB)	9 <i>b</i>	0.604(2)	0.203(2)	1.048(4)	0.080

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
N(1)	9 <i>b</i>	0.4010(1)	0.8472(1)	-0.0739(4)	0.039(2)	0.035(2)	0.036(2)	0.017(1)	-0.001(1)	-0.005(1)
O(1)	9 <i>b</i>	0.4136(1)	0.8794(1)	0.4073(3)	0.063(2)	0.052(2)	0.039(2)	0.022(1)	0.002(1)	0.010(1)
O(3)	9 <i>b</i>	0.5509(1)	1.0736(1)	-0.0633(4)	0.061(2)	0.044(2)	0.053(2)	0.027(1)	-0.007(1)	-0.001(1)
O(2)	9 <i>b</i>	0.6007(1)	1.0189(1)	0.5226(4)	0.060(2)	0.090(2)	0.043(2)	0.031(2)	-0.012(2)	-0.013(2)
C(1)	9 <i>b</i>	0.3978(2)	0.9076(2)	0.2854(5)	0.047(2)	0.045(2)	0.036(2)	0.021(2)	0.011(2)	0.005(2)
C(2)	9 <i>b</i>	0.3375(2)	0.8711(2)	0.2353(6)	0.043(2)	0.060(3)	0.053(3)	0.022(2)	0.014(2)	0.010(2)
C(3)	9 <i>b</i>	0.3198(2)	0.8937(2)	0.0821(6)	0.041(2)	0.057(3)	0.060(3)	0.026(2)	0.010(2)	0.004(2)
C(4)	9 <i>b</i>	0.3562(2)	0.90780(2)	-0.0828(5)	0.038(2)	0.043(2)	0.043(2)	0.021(2)	-0.001(2)	-0.000(2)
C(5)	9 <i>b</i>	0.4137(1)	0.9523(1)	-0.0295(5)	0.044(2)	0.041(2)	0.033(2)	0.024(2)	0.002(2)	0.005(2)
C(6)	9 <i>b</i>	0.4549(1)	0.9646(2)	-0.1880(5)	0.039(2)	0.042(2)	0.036(2)	0.017(2)	0.003(2)	0.005(2)
C(7)	9 <i>b</i>	0.4900(1)	0.9411(1)	-0.1190(5)	0.036(2)	0.040(2)	0.029(2)	0.018(2)	0.005(2)	-0.000(2)
C(8)	9 <i>b</i>	0.5320(1)	0.9796(2)	0.0173(5)	0.034(2)	0.044(2)	0.034(2)	0.020(2)	0.004(2)	0.000(2)
C(9)	9 <i>b</i>	0.4948(1)	0.9800(1)	0.1764(5)	0.037(2)	0.041(2)	0.031(2)	0.018(2)	0.003(2)	0.000(2)
C(10)	9 <i>b</i>	0.4363(1)	0.9318(1)	0.1235(5)	0.035(2)	0.037(2)	0.035(2)	0.018(2)	0.006(2)	0.003(2)
C(11)	9 <i>b</i>	0.5169(2)	0.9767(2)	0.3621(5)	0.044(2)	0.070(3)	0.032(2)	0.024(2)	0.006(2)	-0.001(2)
C(12)	9 <i>b</i>	0.5782(2)	1.0039(2)	0.3796(5)	0.048(2)	0.053(2)	0.036(2)	0.025(2)	-0.007(2)	-0.002(2)
C(13)	9 <i>b</i>	0.6086(2)	1.0080(2)	0.2069(5)	0.041(2)	0.063(3)	0.043(2)	0.029(2)	-0.002(2)	-0.005(2)
C(14)	9 <i>b</i>	0.5711(2)	0.9612(2)	0.0825(5)	0.040(2)	0.052(2)	0.044(2)	0.025(2)	0.002(2)	0.001(2)
C(15)	9 <i>b</i>	0.5733(2)	1.0385(2)	-0.0424(5)	0.042(2)	0.047(2)	0.035(2)	0.019(2)	0.008(2)	0.003(2)
C(16)	9 <i>b</i>	0.6161(1)	1.0578(2)	0.1053(5)	0.036(2)	0.049(2)	0.040(2)	0.020(2)	0.002(2)	-0.001(2)
C(17)	9 <i>b</i>	0.6509(2)	1.1091(2)	0.1480(6)	0.046(2)	0.064(3)	0.063(3)	0.025(2)	-0.010(2)	-0.006(2)
C(18)	9 <i>b</i>	0.3317(2)	0.9270(2)	-0.2309(6)	0.053(3)	0.060(3)	0.060(3)	0.035(2)	-0.007(2)	0.000(2)
C(19)	9 <i>b</i>	0.3585(2)	0.8586(2)	-0.1576(5)	0.038(2)	0.048(2)	0.039(2)	0.019(2)	-0.004(2)	-0.001(2)
C(20)	9 <i>b</i>	0.4506(2)	0.8955(1)	0.0108(5)	0.041(2)	0.047(2)	0.030(3)	0.024(2)	0.001(2)	-0.002(2)
C(21)	9 <i>b</i>	0.4180(2)	0.8176(2)	-0.2044(6)	0.056(2)	0.053(2)	0.046(3)	0.034(2)	-0.005(2)	-0.010(2)
C(22)	9 <i>b</i>	0.3718(2)	0.7623(2)	-0.2536(8)	0.090(4)	0.056(3)	0.078(4)	0.039(3)	-0.008(3)	-0.021(3)
O(4W)	9 <i>b</i>	0.3800(2)	0.7789(1)	0.2364(5)	0.072(2)	0.063(2)	0.074(2)	0.036(2)	0.014(2)	0.019(2)
O(5W)	9 <i>b</i>	0.5592(2)	0.1469(2)	0.2216(6)	0.116(3)	0.124(4)	0.076(3)	0.087(3)	-0.014(2)	-0.024(2)
Cl(1)	9 <i>b</i>	0.59713(6)	0.15293(6)	0.6269(2)	0.089(1)	0.098(1)	0.0760(9)	0.0408(8)	0.0127(8)	0.0247(8)
O(6W)	9 <i>b</i>	0.629(1)	0.2233(1)	0.9592(6)	0.060(2)	0.073(2)	0.114(3)	0.032(2)	0.000(2)	0.013(2)

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