JNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE

AD-A217 792 REPORT DOCUMENTATION PAGE

Form Approved OME No. 0704-018

A. REPORT SECURITY CLASSIFICATION Unclassified Ib. RESTRICTIVE MARKINGS Ib. RESTRICTIVE MARKENS Ib. RESTRICTION NUMBER Ib. RESTRICTION NUMBER Ib. RESTRICTION Ib. RESTRICTION Ib. RESTRICTI						
SECURITY CLASSIFICATION AUTHORITY S. DISTRIBUTION/AVAILABILITY OF REPORT Approved for public release; distribution unlimited PERFORMING ORGANIZATION REPORT NUMBER(S) HDL-TR-2169 S. MONITORING ORGANIZATION REPORT NUMBER(S) HDL-TR-2169 S. MONITORING ORGANIZATION REPORT NUMBER(S) HArry Diamond Laboratories SLCHD-ST-AP S. ADDRESS (Chy, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 AMME OF FUNDING/SPONSORING To. ADDRESS (Chy, State, and ZIP Code) SICCHD-ST-AP S. ADDRESS (Chy, State, and ZIP Code) SOURCE OF FUNDING/SPONSORING Mapplicable) J. SOURCE OF FUNDING NUMBERS SOURCE OF Source OF Code) SOURCE OF FUNDING NUMBERS SOURCE OF Source OF Code SOURCE OF FUNDING NUMBERS SOURCE OF Source OF Code SOURCE OF FUNDING NUMBERS SOURCE OF CODE SO						
Approved for public release; distribution unlimited PERFORMING ORGANIZATION REPORT NUMBER(S) HDL-TR-2169 AMME OF PERFORMING ORGANIZATION Harry Diamond Laboratories Bb. OFFICE SYMBOL (# applicable) SLCHD-ST-AP ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 AMME OF FUNDING/SPONSORING City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 AMSLC DROAMIZATION AMSLC ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 AMSLC DROAMIZATION AMSLC ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 AMSLC ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1145 AMSLC ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1145 AMSLC DI SOURCE OF FUNDING NUMBERS PROGRAM PROJECT TASK NO. Adelphi, MD 20783-1145 I.TITLE (include Security Classification) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta						
FERFORMING ORGANIZATION REPORT NUMBER(S) S. MONITORING ORGANIZATION REPORT NUMBER(S) HDL-TR-2169 S. MONITORING ORGANIZATION BEPORT NUMBER(S) HDL-TR-2169 S. MONITORING ORGANIZATION B. Gruper Stress (City, State, and ZIP Code) SLCHD-ST-AP Sc ADDRESS (City, State, and ZIP Code) Z800 Powder Mill Road Adelphi, MD 20783-1197 S. AMME OF FUNDING/SPONSORING To. OFFICE SYMBOL (# applicable) SLCHD-ST-AP Sc ADDRESS (City, State, and ZIP Code) SOUPCOMENT MILLION SUBJECT TASK AME OF FUNDING/SPONSORING SC OFFICE SYMBOL (# applicable) SLCHD-ST-AP Sc ADDRESS (City, State, and ZIP Code) SOUPCOMENT MILLION SOURCE OF FUNDING/SPONSORING SC OFFICE SYMBOL (# applicable) SLCHD-ST-AP Sc ADDRESS (City, State, and ZIP Code) SOUPCOMENT INSTRUMENT IDENTIFICATION NUMBER SC ADDRESS (City, State, and ZIP Code) SOUPCOMENT MILLION SOURCE OF FUNDING NUMBERS SC ADDRESS (City, State, and ZIP Code) SOUPCOMENT MILLION SOURCE OF FUNDING NUMBERS SC ADDRESS (City, State, and ZIP Code) SOUPCOMENT MILLION SOURCE MILLION SOURCE OF FUNDING NUMBERS SC ADDRESS (City, State, and ZIP Code) SOUPCOMENT MILLION SOURCE MILLION SOU						
HDL-TR-2169 a. NAME OF PERFORMING ORGANIZATION Harry Diamond Laboratories b. OFFICE SYMBOL (# applicable) SLCHD-ST-AP b. ADDRESS (Chy, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 a. NAME OF FUNDING/SPONSORING U.S. Army Laboratory Command AMSLC b. OFFICE SYMBOL (# applicable) AMSLC b. OFFICE SYMBOL (# applicable) b. OFFICE S						
Barry Diamond Laboratories Bb. OFFICE SYMBOL (If applicable) SLCHD-ST-AP 7a. NAME OF MONITORING ORGANIZATION Harry Diamond Laboratories SLCHD-ST-AP 7b. ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 7b. OFFICE SYMBOL (If applicable) 7b. ADDRESS (City, State, and ZIP Code) 3a. NAME OF FUNDING/SPONSORING ORGANIZATION 7b. OFFICE SYMBOL (If applicable) 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 3a. NAME OF FUNDING/SPONSORING ORGANIZATION 7b. OFFICE SYMBOL (If applicable) 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 3a. NAME OF FUNDING/SPONSORING ORGANIZATION 7b. OFFICE SYMBOL (If applicable) 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 3b. ADDRESS (City, State, and ZIP Code) 10. SOURCE OF FUNDING NUMBERS 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 2800 Powder Mill Road Adelphi, MD 20783-1145 7b. OFFICE SYMBOL (If applicable) 10. SOURCE OF FUNDING NUMBERS 2800 Powder Mill Road Adelphi, MD 20783-1145 6.2120.A H125 MORH NO. 11. TITLE (include Security Classification) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ 15. PAGE COUNT 2. PERSONAL AUTHOR(S) TOOMAS H. Allik, Clyde A. MORTISON, John B. Gruber, and Milan R. Kokta 15. PAGE COUNT 15. PAGE COUNT						
Harry Diamond Laboratories (* application) SLCHD-ST-AP G. ADORESS (Chy, State, and ZIP Code) 7b. ADDRESS (Chy, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197 7b. OFFICE SYMBOL (# applicable) 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER "A. NAME OF FUNDING/SPONSORING ORGANIZATION U.S. Army Laboratory Command 7b. OFFICE SYMBOL (# applicable) 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER "A. ADDRESS (Chy, State, and ZIP Code) 10. SOURCE OF FUNDING NUMBERS PROJECT "A. ADDRESS (Chy, State, and ZIP Code) 10. SOURCE OF FUNDING NUMBERS 2800 Powder Mill Road Adelphi, MD 20783-1145 PROGRAM ELEMENT NO. 6.2120.A PROJECT NO. AH25 "1. TITLE (Include Security Classification) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ "2. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta "1. TITLE (Include Security Classification) "2. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta						
ADDRESS (Chy, State, and ZIP Code) 2800 Powder Mill Road 7b. ADDRESS (Chy, State, and ZIP Code) 2800 Powder Mill Road 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 29. NAME OF FUNDING/SPONSORING 7b. OFFICE SYMBOL (# applicable) 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 2800 Powder Mill Road AMSLC 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER 2800 Powder Mill Road AMSLC 10. SQURCE OF FUNDING NUMBERS 2800 Powder Mill Road PROJECT TASK NO. Adelphi, MD 20783-1145 10. SQURCE OF FUNDING NUMBERS 11. TITLE (Include Security Classel/cation) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ 2. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta 10. TOWE OF BEPORT 112. TAK E COVERED						
2800 Powder Mill Road Adelphi, MD 20783-1197 7b. OFFICE SYMBOL (# applicable) AGANIZAT::::N 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER ''.A.NAME OF FUNDING/SPONSORING ORGANIZAT::::N 7b. OFFICE SYMBOL (# applicable) AMSLC 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER ''.A.DDRESS (Chy, State, and ZIP Code) 10. SOURCE OF FUNDING NUMBERS PROJECT NO. NO. 2800 Powder Mill Road Adelphi, MD 20783-1145 10. SOURCE OF FUNDING NUMBERS WORK NO. WORK ACCE ''. TITLE (Include Security Classification) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ Very Month Day 15. PAGE COUNT ''. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta 114. DATE OF REPORT (Year, Month Day) 115. PAGE COUNT						
Adelphi, MD 20783-1197 Ia. NAME OF FUNDING/SPONSORING ORGANIZATION U.S. Army Laboratory Command 7b. OFFICE SYMBOL (# applicable) AMSLC 9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER C. ADDRESS (Chy, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1145 10. SOURCE OF FUNDING NUMBERS PROGRAM ELEMENT NO. 6.2120.A PROJECT NO. AH25 TASK NO. WORK ACCE 11. TITLE (Include Security Classification) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ Y ₃ Sc ₂ Al ₃ O ₁₂ 22. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta 14. DATE OF REPORT (Year, Month, Day) 15. PAGE COUNT						
Image: Second Control of Second Contervice Contervice Contervice Control of Second Contro						
ORGANIZATION (# applicable) U.S. Army Laboratory Command AMSLC ADDRESS (City, State, and ZIP Code) 10. SOURCE OF FUNDING NUMBERS 2800 Powder Mill Road PROGRAM Adelphi, MD 20783-1145 PROGRAM *1. TITLE (include Security Classification) 6.2120.A Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ *2. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta 113. TIME Coverage 113. TIME Coverage						
C. ADDRESS (City, State, and ZiP Code) 10. SOURCE OF FUNDING NUMBERS 2800 Powder Mill Road PROGRAM Adelphi, MD 20783-1145 PROGRAM 11. TITLE (Include Security Classification) 6.2120.A Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ 22. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta 113. TIME Coverage 114. DATE OF REPORT 115. PAGE COUNT						
2800 Powder Mill Road PROGRAM PROGRAM PROJECT TASK WORM Adelphi, MD 20783-1145 6.2120.A AH25 AH25 ACCE 11. TITLE (Include Security Classification) 6.2120.A AH25 AH25 Comparison Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ Y ₃ Sc ₂ Al ₃ O ₁₂ Environmental Authors 2. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta Lis. PAGE COUNT Lis. PAGE COUNT						
Adelphi, MD 20783-1145 6.2120.A AH25 11. TITLE (Include Security Classification)						
 TITLE (Include Security Classification) Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd³⁺: Y₃Sc₂Al₃O₁₂ PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta TYPE OF REPORT (Year, Month, Day) 15. PAGE COUNT 						
Crystallography, Spectroscopic Analysis, and Lasing Properties of Nd ³⁺ : Y ₃ Sc ₂ Al ₃ O ₁₂ 2. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta 114. DATE OF REPORT (Year, Month Day) 15. PAGE COUNT						
22. PERSONAL AUTHOR(S) Toomas H. Allik, Clyde A. Morrison, John B. Gruber, and Milan R. Kokta						
TYPE OF REPORT 112h TIME COVERED 14. DATE OF REPORT (Year, Month, Day) 15. PAGE COUNT						
Final FROM Aug 88 to Dec 88 December 1989 24						
AMS code: AH25, HDL project: R8A951						
17. COSATI CODES 18. SUBJECT TERMS (Continue on reverse if necessary and identity by block number)						
BELD GROUP SUB-GROUP Yttrium scandium aluminum garnet, neodymium, crystallography, retrac						
20 02 index, diode pumped laser, crystal-field parameters, Judd-Oren parameter						
20 05 [Dranching railos: (Ve.Od. 101. Un. (3F)						
¹⁹ ABSTRACT (Continue on revene if necessary and Identify by block number) The crystallographic, optical, and spectroscopic properties of Nd ³ (Y, Sc ₂ Al ₂ O ₁) are reported from which an assessment can be made regarding the material's potential as a laser. Individual Stark levels for many of the (2 ^{s+1}), manifolds of Nd ³ (4f. ³) in the crystal have been identified from emission and absorption data up to 17,600 cm ⁻² at 14 K. The observed crystal-field splitting and the measured cross sections (intensities) associated with manifold-to-manifold transitions are compared with calculated splittings and calculated intensities. Branching ratios and diode-array-pumped laser slope efficiencies are also reported. We conclude that Nd ³ ? YSAG has potential as a diode-pumped 1 tim laser material.						
reported. We conclude that Ndy? YSAG has potential as a diode-pumped laum laser material. Ke uword						
reported. We conclude that Nd?? YSAG has potential as a diode-pumped I tum jaser material. Ke yword						
20. DISTRIBUTIONVAVAILABILITY OF ABSTRACT						
20. DISTRIBUTION/AVAILABILITY OF ABSTRACT 20. DISTRIBUTION/AVAILABILITY OF ABSTRACT 21. ABSTRACT SECURITY CLASSIFICATION Unclassified 22a. NAME OF RESPONSIBLE INDIVIDUAL						

The findings in this report are not to be construed as an official Department of the Army position unless so designated by other authorized documents.

Citation of manufacturer's or trade names does not constitute an official endorsement or approval of the use thereof.

Destroy this report when it is no longer needed. Do not return it to the originator.

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE

		REPO	ORT DOCUMENTA	TION PAGE	· · · · · · · · · · · · · · · · · · ·		Form Approved OMB No. 0704-0188
1a. REPORT SE Unclassifi	ECURITY CLASS			1b. RESTRICTIVE	MARKINGS		
2a. SECURITY	CLASSIFICATION	AUTHORITY	<u> </u>	3. DISTRIBUTION	AVAILABILITY OF RE	PORT	
26. DECLASSIF		GRADING SCHEDULE	,,,,,,,,,,_,_,	Approved fo	r public release	; distribu	tion unlimited.
4. PERFORMIN	G ORGANIZATIO	N REPORT NUMBER(S)	5. MONITORING C	DRGANIZATION REPO		R(S)
HDL-TR-	2169						
6a. NAME OF P	ERFORMING OR	GANIZATION	6b. OFFICE SYMBOL (Il applicable)	7aL NAME OF MOR	NITORING ORGANIZA	TION	
Harry Diamond Laboratories SLCHD-ST-AP							
6c. ADDRESS (City, State, and ZIP Code)				7b. ADDRESS (CA	ty, State, and ZIP Code	e)	
2800 Pow Adelphi, l	der Mill Ro MD 20783-1	ad 197					
8a. NAME OF FI	UNDING/SPONS	ORING	76. OFFICE SYMBOL	9. PROCUREMEN	T INSTRUMENT IDEN		NUMBER
U.S. Arm	y Laboratory	y Command	AMSLC				
8c. ADDRESS	City, State, and Z	(IP Code)	L	10. SOURCE OF FU	INDING NUMBERS		
2800 Pow	der Mill Ro	ad		PROGRAM	PROJECT	TASK	WORK UNIT
Adelphi, 1	MD 20783-1	145		6.2120.A	AH25	140.	
11. TITLE (Inck	ide Security Class	silication)		L			
Crystallog	graphy, Spec	ctroscopic Analys	is, and Lasing Propert	ties of Nd ³⁺ : Y ₃	Sc ₂ Al ₃ O ₁₂		
12. PERSONAL Toomas H	AUTHOR(S)	de A Morrison I	obn B. Gruber and M	filan R. Kokta			
13a. TYPE OF F	EPORT	Lisb. TIME CO	VERED	14. DATE OF REPOR	T (Year Month Dav)		15. PAGE COUNT
Final		FROM Au	<u>g 88</u> то <u>Dec 88</u>	December 1	989		24
16. SUPPLEME AMS cod	e: AH25, H	DL project: R8A9	251				
17.	COSATI	CODES	18. SUBJECT TERMS (C	ontinue on reverse If n	ecessary and identify i	by block num	ber) CV
FIELD	GROUP	SUB-GROUP	Yttrium scandium a	luminum garne), neodymium,	crystallog	graphy, fefractive
20	02		index, diode pumpe	d laser, crystal-	tield parameter	s, Judd-O	telt parameters,
19. ABSTRACT	US (Cantinue on rev	area il nacessary ant ide	dily by block number)	Roding		<u></u>	<u></u>
The crysta	allographic, o	optical, and spectr	oscopic properties of	Ndy Y.Sc. ALO	are reported fi	rom whic	h an assessment can
be made r	egarding the	material's potent	ial as a laser. Individu	al Stark levels f	or many of the	s+12 man	ifolds of $Nd^{(4f^3)}$
in the crys	stal have bee	n identified from	emission and absorpti	on data up to 17	,600 cm ²² at 141	K. The ob	served crystal-field
splitting a	nd the measu	ured cross section	s (intensities) associat	ed with manifol	ld-fo-manifold t	ransitions	are compared with
calculated	splittings a	nd calculated inte	nsities. Branching rati	ios and diode-a	ray-pumped las	er slope e	efficiencies are also
reponed.	we conclude	e mat NG2. I SAC	a nas potential as a die	bae-pumpea 14	iaser materia	ai. K. <i>ş</i>	un alti
					1. compt -1	-	
				۰.	ne i ser e ser a ser Norm	L.	
20. DISTRIBUTI	ONVAVAILABILIT	Y OF ABSTRACT		21. ABSTRACT SE	ECURITY CLASSIFICA	TION	<u></u>
22a. NAME OF	BIFIED/UNLIMITE RESPONSIBLE I	D LI SAME AS RP		22b. TELEPHONE	(Include Area Code)		22c. OFFICE SYMBOL
Clyde A.	Morrison	-		(202) 394-20)42	SL	CHD-ST-AP
DD Form 147	3, JUN 86		Previous editions are o	beolete.	SE	CURITY CLA	UNCLASSIFICATION OF THIS PAGE

Contents

	Pag
1. INTRODUCTION	5
2. EXPERIMENTAL RESULTS AND DISCUSSION	5
2.1 Crystal Growth and Structure	5
2.2 Index of Refraction	/ 7
2.4 Nd ³⁺ Fluorescence	
2.5 Judd-Ofelt Theory	
2.6 Crystal-Field Calculations	12
2.7 Laser Experiments	16
3. SUMMARY AND CONCLUSIONS	16
Acknowledgements	
References	17
Distribution	21
Figures	
1. Room-temperature absorption spectrum of Nd ³⁺ :YSAG	8
2. ${}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2}$ fluorescence spectra of Nd ³⁺ doped in YSAG and YAG at room temperature	re11
Tables	
1. Atom coordinates and thermal coefficients of Y ₃ Sc ₂ Al ₃ O ₁₂	7
2. Measured and calculated indices of refraction of Nd ³⁺ :Y ₃ Sc ₂ Al ₃ O ₁₂ at 298 K	7
	9
3. Experimental and theoretical crystal-field splittings of Nd ³⁺ ion manifolds in YSAG	
 Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG Absorption intensities for Nd:YSAG at 298 K 	12
 Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG Absorption intensities for Nd:YSAG at 298 K Experimental and calculated Judd-Ofelt parameters and predicted branching ratios for Nd³⁺:YAG and Nd³⁺:YAG 	12
 Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG Absorption intensities for Nd:YSAG at 298 K	12
 Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG Absorption intensities for Nd:YSAG at 298 K	
 3. Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG 4. Absorption intensities for Nd:YSAG at 298 K	

A-I v Codes nd/or al

1. Introduction

Increasing demands placed on solid-state lasers in applications ranging from communications to medicine highlight the need to develop new materials that have better diode pump laser characteristics than the standard laser material Nd:YAG [1–9]. The challenge is made clear with the present availability of single laser diodes with powers exceeding 1 W and two-dimensional arrays producing fluxes of more than 4 kW/cm² at the required wavelength. Desirable properties of new diodepumped Q-switched solid-state lasers include a longer fluorescent lifetime and a larger absorption coefficient than is possible with Nd:YAG. In addition, the optical, mechanical, and thermal crystal properties of the host must be competitive with Nd:YAG to permit highrepetition-rate applications.

There are several reasons for examining the laser properties of Nd:Y₃Sc₂Al₃O₁₂ (YSAG) in greater detail [10–12]. The distribution coefficient for Nd³⁺ in YSAG is roughly twice that for YAG [13,14], making it possible to increase the Nd³⁺ concentration in YSAG over that in YAG. Replacing Al³⁺ ions with larger Sc³⁺ ions increases the distance between dodecahedral lattice sites (substitutional sites for Nd³⁺ ions in the garnet structure). Any increase in separation between neighboring Nd³⁺ ions, especially with increasing concentration, tends to reduce the relatively strong ion/ion interaction in YAG, which leads to concentration quenching of the Nd³⁺ fluorescence [10,15,16]. In addition, the aluminum-based systems, such as YAG, YSAG, or gadolinium scandium aluminum garnet (GSAG), are formed from more stable constituent oxides than gallium-containing materials, such as gadolinium scandium gallium garnet (GSGG). The tendency for color center formation in gallium-containing garnets is due to oxidation state variation or oxygen vacancies, and this problem is greatly reduced in aluminate systems.

Only some of the spectroscopic properties of Nd:YSAG have been reported in the open literature [10,17]. Kaminskii reports energy levels up to the ${}^{4}F_{3/2}$ manifold only [10]. Most of the literature concentrates on the empirical evaluation of Nd:YSAG and Cr³⁺ sensitized Nd:YSAG as a laser [11,12,18,19]. However, to fully assess the potential of this material, it is important to study the spectroscopic properties in greater detail. The individual experimental Stark levels and the measured cross sections and lifetimes of transitions between these levels should be compared with theoretical predictions based on lattice-sum calculations, crystal-field splitting, and the predicted cross sections and lifetimes based on the Judd-Ofelt model for rare-earth ions in solids [16,20-22].

We report here the results of crystal growth and x-ray diffraction studies, along with measurements on the index of refraction of Nd: YSAG. The experimental Stark levels for many of the $^{25+1}L_1$ manifolds of Nd³⁺ (4f³) deduced from both absorption and emission data are tabulated up to 17,600 cm⁻¹ and compared with a theoretical crystal-field splitting calculation. A survey spectrum of Nd:YSAG between 300 and 1000 nm and the fluorescence from ${}^{4}F_{3/2}$ to ${}^{4}I_{11/2}$, both obtained at room temperature, provide a general overview of observed optical properties of Nd³⁺. Absorption intensities from the groundstate manifold of Nd³⁺ (${}^{4}I_{g/2}$) to excited manifolds observed in the survey spectrum are compared with calculated intensities based on the Judd-Ofelt theory [20–22]. Branching ratios and slope efficiencies are also reported from which an assessment can be made regarding Nd:YSAG as a laser material.

2. Experimental Results and Discussion

2.1 Crystal Growth and Structure

Yttrium scandium aluminum garnet belongs to the group of oxide compounds crystallizing in garnet structure. The first garnet containing scandium was synthesized by Moronova and Feofilov [23], and a systematic study of Sc incorporation into aluminum garnets was made by Kokta [13] in 1973. Subsequently, a scandium-substituted rare-earth aluminum garnet (GSAG) was grown by Brandle and Vanderleeden [24]. An interest in scandiumsubstituted garnets was revived a decade later when their usefulness as tunable solid-state laser hosts was demonstrated with Cr^{3+} doped in GSGG [25].

The first crystals of yttrium scandium aluminum garnets doped with either neodymium or chromium were grown in a 2 in. \times 2 in. crucible. They were approximately 0.9 in. in diameter and 2 in. long. These crystals were used to fabricate spectroscopic samples as well as seeds for further crystal growth.

For laser application, a 5-in.-long Nddoped crystal of 1.5-in. diameter was grown. The furnace used to grow this material was built from a silica sleeve inserted in an rf coil. A 3 in. $\times 3$ in. iridium crucible was used which was surrounded by a 3.5-in. I.D. zirconium oxide liner. The space between the ZrO, liner and the SiO, sleeve was filled with insulation consisting of zirconium oxide bubbles (grog). The induction coil, which was made from 3/8-in.diameter copper tubing, was wound into 12 turns around the growth furnace, and powered by a 50-kW motor generator operating at a 10kHz frequency. The crucible was filled in the 3:2:3 molar ratio for Y_2O_3 , Sc_2O_3 , and Al_2O_3 . The amount of Nd₂O₃ was calculated for substitution of 1.5-percent Nd into eightfold coordination sites, under the assumption that the Nd distribution coefficient, k_{Nd} , approached 0.4 in this system. However, the Nd concentration of a spectroscopic sample from the boule was determined by x-ray fluorescence to be $1.76 \pm$ 0.10 at. wt.%, which corresponds to an Nd density of $(3.33 \pm 0.07) \times 10^{19}$ cm⁻³ [26]. The deviation between the measured and calculated Nd concentration is not surprising, since the exact value of k_{N} is a growth-dependent parameter (rotation rate, pull rate). More growth runs would be required to determine k_{NA} precisely for given growth conditions.

The crystals were grown along the <111> orientation, at a rate of 0.015 in./hour, and were rotated at 15 rpm. They were grown under an ambient atmosphere of nitrogen containing 800 ppm by volume of O₂. The melting point was determined with an optical pyrometer to be $1900 \pm 25^{\circ}$ C, uncorrected for emissivity. YSAG showed typical garnet faceting as observed in YAG crystals. The interface shape was convex, and strain was observed in the "core" area. No attempts were made to change the interface shape. The strain pattern is significantly more pronounced in YSAG than is the strain in YAG. YSAG crystals have a much higher tendency to crack, and therefore extreme caution must be exercised during their fabrication. Contrary to the findings of Brandle [24], a slower pull rate seems to ease this problem, and rates even lower than 0.015 in./hr may be well justified, especially for crystals doped with Nd.

The crystal structure analysis was performed on an automated Nicolet $R3m/\mu$ diffractometer equipped with an incident-beam graphite monochromator and Mo K_{α} radiation $(\lambda = 0.7107 \text{ Å})$. Single-crystal diffraction patterns of the crystal showed that the crystals were cubic, belonging to the space group la3d (No. 230), with a unit cell axis length of a =12.271 Å (V = 1847.6 Å³). The lattice parameter differs from that of Kokta [13] (a = 12.324 Å) and Bogomolova [27] (a = 12.251 Å); this difference is attributed to the distribution coefficient for Sc being less than unity, which allows for mixed occupancy between Sc and Al in the octahedral site. This should allow ranges in lattice parameters from stoichiometric YSAG (a = 12.32 Å) to YAG (a = 12.00 Å). Elemental analysis performed on the sample by x-ray fluorescence did indeed show lower Sc than expected in the crystal [26]. The 191 independent single-crystal reflections recorded were used to refine the structure by least squares to residuals of R =0.0342 and wR = 0.0502. Positional and thermal parameters are listed in table 1. Further details on the data collection and on the crystal structure are given by Campana [28].

Table 1. Atom coordinates (×10⁴) and thermal coefficients ($Å^2 \times 10^3$) of Y₃Sc₂Al₃O₁₂ Parenthetical values are estimated standard deviations.

Atom	x	y	z	U*
Y	0	0	0	74 (4)
Sc	0	2500	1250	51 (3)
Al	0	2500	3750	41 (7)
0	309 (3)	562 (3)	6562 (3)	67 (9)

^aEquivalent isotropic *U* defined as one third of the trace of the orthogonalized U_{ii} tensor.

2.2 Index of Refraction

The refractive indices of Nd:YSAG were measured using the method of minimum deviation [29]. A polished prism of Nd:YSAG was fabricated to a height of 5 mm and had faces of 12 and 17 mm. The prism angle was $44^{\circ}55'$. A Spencer 2754 Spectrometer (American Optical Company) was used to make all angular measurements, and multiline argon ion and helium neon lasers were used as light sources between 457.9 and 632.8 nm. The measured refractive indices are given in table 2. The accuracy of these measurements was ± 0.002 because of the poor optical quality of the sample.

These experimental data were leastsquares fit to Sellmeier's dispersion equation

$$[n(\lambda)]^2 = 1 + \frac{A\lambda^2}{\lambda^2 - B}$$
(1)

where $A = 2.420 \pm 0.008$ and $B = 0.01520 \pm 0.00064 \,\mu\text{m}^2$. These results agree well with the results of Wemple and Tabor for undoped YSAG [30]. The refractive indices for the doped sample are higher than the ones for the undoped.

Table 2. Measured and calculated inc	lices of
refraction of Nd ³⁺ :Y ₃ Sc ₂ Al ₃ O ₁₂ at 298 l	K

Wavelength	n. meas	n _{calc}	
457.9	1.900	1.900	
476.5	1.895	1.896	
488.0	1.893	1.893	
496.5	1.891	1.892	
514.5	1.889	1.889	
594.5	1.880	1.878	
611.9	1.878	1.877	
632.8	1.873	1.875	
Crystal	Sellmeier	coefficients	
	A	В	
Nd:YSAG	2.420	0.01520	
YSAG ^a	2.4118	0.01477	
Defense 20			

*Reference 30.

2.3 Nd³⁺ Absorption

The absorption spectrum of neodymiumdoped YSAG was investigated in the range from 1,500 to 40,000 cm⁻¹. These data were recorded in the ultraviolet, visible, and infrared on Perkin-Elmer Lambda 9 and 983G spectrometers interfaced to the Perkin-Elmer 7500 computer. Figure 1 shows the room-temperature absorption spectrum between 300 and 1000 nm of a 2.95-mm-long, Nd³⁺:YSAG sample with the Fresnel reflection losses removed.

Determination of the individual Stark levels of the Nd³⁺ ions in the dodecahedral sites (D, symmetry) was accomplished by cooling the sample to cryogenic temperatures. A closedcycle refrigerator, CTI-Cryogenics Model 21, was used to obtain spectra at 14 K. Table 3 lists the 60 lowest experimentally determined energy levels (up to 17,600 cm⁻¹). Energy levels up to 40,000 cm⁻¹ have been determined and are currently being fit to a theoretical crystal-field calculation which includes spin-correlation effects; this calculation will be reported at a later date [31]. The low-lying energy levels, up to ${}^{4}F_{3/2}$, agree very well with those of Kaminskii [10]. The overall accuracy of the measurements is $\leq 5 \text{ cm}^{-1}$.



Figure 1. Room-temperature absorption spectrum of Nd³⁺:YSAG. Nd concentration is 3.3 × 10¹⁹ cm⁻³.

2.4 Nd³⁺ Fluorescence

The fluorescence spectrum of Nd³⁺:YSAG was recorded with a Spex F222 spectrometer equipped with a North Coast model EO-817L Ge detector. Figure 2 shows the fluorescence of Nd³⁺:YSAG and, for comparison, Nd³⁺:YAG in the region of the ${}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2} (R_{1,2} \rightarrow Y_{1-6})$ transitions. In general, the fluorescence lines of Nd:YSAG show a broadening versus YAG. At room temperature, the two most intense lines for Nd:YSAG appear at 1.0622 and 1.0595 μm. These wavelengths correspond to the $R_2 \rightarrow Y_2$ and $R_1 \rightarrow Y_1$ transitions, respectively. In addition to these two prominent lines, the $R_1 \rightarrow Y_2$ and $R_2 \rightarrow Y_4$ transitions appear as shoulders on the long-wavelength side of the band. The individual Stark-level branching ratios were estimated from the peak heights to be 15 percent

for $R_2 \rightarrow Y_3$ and 13 percent for $R_1 \rightarrow Y_1$. Accurate determinations of the distributions were not possible because of the limited resolution of the monochromator.

The fluorescence lifetime of the ${}^4F_{3/2}$ state was measured using a GaAlAs laser diode as the excitation source. The diode emits radiation at 805 nm at room temperature, and the pulse duration was 2 μ s. Fluorescence detection was viewed through an 850-nm long-pass filter into an Si detector. Signals were processed by a Stanford Research Systems boxcar integrator and stored in a computer. The fluorescence lifetime at room temperature was 208 ± 5 μ s at an Nd concentration of 1.76 at. wt.%. A comparable Nd concentration in YAG would have a lifetime of 160 μ s [10].

	E	nergy	²⁵⁺¹ L _j (centroid	Free ion mixture (%)
	Exp	Theo	cm ⁻¹)	
1	0 114	-11 120	41	98.18 ${}^{4}I_{9/2}$ + 1.40 ${}^{4}I_{11/2}$ + 0.27 ${}^{4}I_{13/2}$ 98.08 ${}^{4}I_{1}$ + 1.38 ${}^{4}I_{1}$ + 0.39 ${}^{4}I_{13/2}$
2	192	120	¹ 9/2 (242)	$95.00 I_{9/2} + 1.50 I_{11/2} + 0.59 I_{13/2}$ 06.69.41 + 2.06.41 + 0.00.41
1	301	300	(302)	95.06 $I_{9/2}$ + 5.06 $I_{11/2}$ + 0.05 $I_{13/2}$
5	802	300 934		$95.70 \ 1_{9/2} + 4.00 \ 1_{11/2} + 0.07 \ T_{3/2}$
5	025	024		$97.08 \ r_{9/2} + 2.04 \ r_{11/2} + 0.20 \ r_{13/2}$
6	1979	1982		96.98 ${}^{4}I_{11/2}$ + 2.31 ${}^{4}I_{13/2}$ + 0.30 ${}^{4}I_{15/2}$
7	2022	2016	4 <i>I</i> _{11/2}	$95.12 {}^{4}I_{11/2} + 2.77 {}^{4}I_{13/2} + 1.81 {}^{4}I_{9/2}$
8	2101	2104	(2222)	96.78 $I_{11/2}$ + 2.07 $I_{13/2}$ + 0.77 $I_{9/2}$
9	2136	2131		96.82 ${}^{4}I_{11/2}$ + 2.47 ${}^{4}I_{13/2}$ + 0.42 ${}^{4}I_{9/2}$
10	2437	2442		$93.83 {}^{4}I_{11/2} + 4.38 {}^{4}I_{9/2} + 1.58 {}^{4}I_{13/2}$
11	2495	2495		$95.11 {}^{4}I_{11/2}^{1/2} + 4.10 {}^{4}I_{9/2}^{1/2} + 0.61 {}^{4}I_{13/2}^{1/2}$
12	3905	3906		$97.08 {}^{4}I_{,0} + 2.48 {}^{4}I_{,0} + 0.14 {}^{4}I_{,0}$
13	3929	3923	41	$95.92 {}^{4}I_{12}^{13/2} + 2.34 {}^{13/2}I_{12}^{11/2} + 1.42 {}^{4}I_{12}^{11/2}$
14	4029	4036	(4188)	$97.90 \frac{41.5}{2} + 1.33 \frac{41.5}{2} + 0.35 \frac{41.7}{2}$
15	4044	4042		$96.58 {}^{4}I_{.5,0} + 2.36 {}^{4}I_{.5,0} + 0.67 {}^{4}I_{.5,0}$
16	4057ª	4411		$96.20^{4}I_{1.5/2} + 2.88^{4}I_{1.5/2} + 0.53^{4}I_{1.5/2}$
17	4419	4420		$94.25^{4}J_{max} + 3.25^{4}J_{max} + 2.16^{4}J_{max}$
18	4478	4478		95.96 ${}^{13/2}_{1_{3/2}}$ + 2.76 ${}^{11/2}_{1_{1/2}}$ + 0.91 ${}^{4}I_{1_{5/2}}$
19	5766	5772		97.68 4
20	5797	5794		$99.08^{4}I_{res} + 0.51^{4}I_{res} + 0.13^{4}F_{res}$
21	5927	5924	41	$98.75^{4}I_{res} + 0.77^{4}I_{res} + 0.14^{4}F_{res}$
22	5981	5988	(6221)	$98.57^{4}I_{max} + 0.85^{4}I_{max} + 0.21^{4}F_{max}$
23	6544	6539	()	$97.20^{4}I_{1} + 2.48^{4}I_{1} + 0.08^{4}I_{1}$
24	6560	6563		98484I + 1 03 $4I$ + 0 43 $4I$
25	6622	6625		97564I + 2124I + 0134I
26	6711	6704		97.42 ${}^{4}I_{15/2}$ + 2.24 ${}^{4}I_{13/2}$ + 0.25 ${}^{4}F_{7/2}$
27	11 400	11 431	45	
2/	11,423	11,431	(11522)	$93.77 F_{3/2} + 2.00 F_{5/2} + 1.35 F_{9/2}$
20	11,525	11,515	(11,523)	$93.44 r_{3/2} + 3.54 r_{5/2} + 1.45 r_{7/2}$
29	12,382	12,367	⁴ F _{5/2}	$77.45 {}^{4}F_{5/2} + 13.84 {}^{2}H_{9/2} + 3.96 {}^{4}F_{7/2}$
30	12,441	12,435	(12,524)	$61.74 {}^{2}H_{9/2} + 32.80 {}^{4}F_{5/2} + 3.17 {}^{4}F_{3/2}$
31	12,538*	12,456		$65.56{}^{2}H_{g/2} + 29.23{}^{4}F_{5/2} + 2.90{}^{4}F_{3/2}$
32	12,583	12,590		$75.52 {}^{2}\text{H}_{9/2} + 23.52 {}^{4}\text{F}_{5/2} + 0.20 {}^{4}\text{F}_{3/2}$
33	12,621	12,633		$95.97 {}^{4}F_{5/2} + 2.41 {}^{2}H_{9/2} + 0.62 {}^{4}F_{7/2}$
34	12,637ª	12,690	${}^{2}H_{9/2}$	$88.30^{2}H_{9/2}^{-} + 10.59^{4}F_{5/2}^{-} + 0.29^{2}H_{11/2}^{-}$
35	12,825	12,817	(12,664)	92.38 ${}^{2}H_{9/2}$ + 6.91 ${}^{4}F_{5/2}$ + 0.17 ${}^{4}F_{3/2}$
36	12,860	12,869		$93.27 {}^{2}H_{9/2}^{-} + 5.82 {}^{4}F_{5/2}^{-} + 0.18 {}^{2}H_{11/2}^{-}$

Table 3. Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG

*Expermental energy levels not used in the crystal-field calculations.

_	Energy		$\frac{2S+1}{L_j}$	Free ion mixture (%)
	Exp	Theo	(cm ⁻¹)	
37	13,367	13,361		$88.58 {}^{4}F_{7/2} + 4.65 {}^{4}F_{5/2} + 2.21 {}^{4}S_{3/2}$
38	13,441	13,451	4F7/2	$87.90 {}^{4}F_{7/2} + 4.68 {}^{4}S_{3/2} + 3.09 {}^{4}F_{5/2}$
39	13,570	13,562	(13,490)	$58.76 {}^{4}S_{3/2}^{1/2} + 39.38 {}^{4}F_{7/2} + 0.58 {}^{4}G_{5/2}$
40	13,580	13,588		$95.55 {}^{4}S_{3/2} + 2.90 {}^{4}F_{7/2} + 0.41 {}^{2}H_{11/2}$
41	13,602	13,594	4S3/2	$63.68 {}^{4}F_{7/2}^{3/2} + 35.15 {}^{4}S_{3/2}^{2} + 0.28 {}^{2}H_{11/2}^{1/2}$
42	13,642	13,647	(13,588)	$98.04 {}^{4}F_{7/2}^{7/2} + 0.58 {}^{4}S_{3/2}^{5/2} + 0.36 {}^{4}I_{15/2}^{5/2}$
43	14,630	14,650		$98.65{}^{4}F_{9/2} + 1.60{}^{4}F_{7/2} + 0.57{}^{4}F_{5/2}$
44	14,695	14,696	4F.,,	96.21 ${}^{4}F_{9/2} + 2.28 {}^{4}F_{7/2} + 0.60 {}^{4}F_{5/2}$
45	14,786	14,794	(14,756)	$97.14 {}^{4}F_{0/2} + 0.97 {}^{2}H_{11/2} + 0.63 {}^{2}G_{7/2}$
46	14,834	14,820		$97.51 {}^{4}F_{0/2} + 0.99 {}^{2}H_{11/2} + 0.59 {}^{2}G_{7/2}$
47	14,939	14,924		$98.18 {}^{4}F_{9/2}^{1/2} + 1.23 {}^{2}G_{7/2}^{1/2} + 0.16 {}^{4}F_{7/2}^{1/2}$
48	15 ,770 •	15,892		97.34 ${}^{2}H_{11/2}$ + 2.11 ${}^{2}G_{7/2}$ + 0.27 ${}^{4}G_{5/2}$
49	15,860*	15,930	${}^{2}H_{11/2}$	$97.76^{2}H_{11/2}^{11} + 1.76^{2}G_{7/2}^{11} + 0.19^{4}F_{9/2}^{11}$
50	15,886ª	15 <i>,</i> 959	(15,971)	$98.95 {}^{2}H_{11/2} + 0.39 {}^{2}G_{7/2} + 0.21 {}^{2}H_{9/2}$
51	15,964	15,964		$98.68{}^{2}H_{11/2}^{11/2} + 0.53{}^{2}G_{7/2}^{1/2} + 0.32{}^{4}F_{9/2}^{1/2}$
52	16,093°	16,022		$96.11 {}^{2}H_{11/2} + 1.97 {}^{2}G_{7/2} + 1.05 {}^{4}F_{9/2}$
53	16,12 4 ª	16,067		96.69 ${}^{2}H_{11/2}^{1/2} + 1.30 {}^{4}F_{9/2}^{1/2} + 1.15 {}^{2}G_{7/2}^{1/2}$
54	16,880	16,893		56.29 ${}^{4}G_{5,2}$ + 41.53 ${}^{2}G_{7,2}$ + 1.34 ${}^{2}H_{11/2}$
55	17,010	17,000	4G _{5/2} -	$86.51 {}^{4}\text{G}_{5/2} + 10.87 {}^{2}\text{G}_{7/2} + 0.74 {}^{4}\text{S}_{3/2}$
56	17,065	17,067	(17,090)	$73.89 {}^{4}G_{5/2} + 22.91 {}^{2}G_{7/2} + 0.98 {}^{2}H_{11/2}$
57	17,262	17,231		$93.32 {}^{2}G_{7/2}^{-} + 2.87 {}^{4}G_{5/2}^{-} + 2.14 {}^{2}H_{11/2}^{-}$
58	17,286	17,303	${}^{2}G_{7/2}$	$95.89 {}^{2}G_{7/2}^{7/2} + 1.81 {}^{2}H_{11/2}^{7/2} + 1.56 {}^{4}G_{5/2}^{7/2}$
59	17,331	17,341	(17,192)	$88.78 {}^{2}\text{G}_{7/2}^{7} + 8.22 {}^{4}\text{G}_{5/2}^{7} + 1.91 {}^{2}\text{H}_{11/2}^{7}$
60	17,587	17,664		$65.32 {}^{4}G_{5/2}^{''} + 34.12 {}^{2}G_{7/2}^{''} + 0.16 {}^{2}H_{11/2}^{''}$

⁻⁻ able 3 (cont'd). Experimental and theoretical crystal-field splittings of Nd³⁺ ion manifolds in YSAG

*Experimental energy levels not used in the crystal-field calculations.

2.5 Judd-Ofelt Theory

Application of Judd-Ofelt (JO) theory [20,21] has become a valuable model in predicting rare-earth laser performance. The model was first successfully applied to individual Stark levels in the ethylsulfate system by Axe [32] (Eu³⁺) and by Krupke and Gruber [33] (Tm³⁺). Since then JO theory has been used by numerous laboratories to calculate the branching ratios, radiative lifetimes, and eventually stimulated emission cross sections of the ${}^{4}F_{3/2} \rightarrow {}^{4}I_{1}(J = 9/2, 11/2, 13/2, 15/2)$ transitions. Detailed theoretical and experimental procedures are contained in works by Krupke [34,35], Weber [36], DeShazer [37,38], and Kaminskii [10,16]. The JO model is based on the following relationship: the line strength, S(J,J'), for a transition between an initial J manifold $|4f^{n}[SL]J\rangle$ and final J' manifold $|4f^{n}[S'L']J'\rangle$ can be written in the form

$$S(J,J') = \sum_{i=2,4,6} \Omega_i \left| \langle 4f^{*}[SL]JW^{(0)}Hf^{*}[S'L']J' \rangle \right|^2$$
(2)

where $<||U^{(t)}||>^2$ are the squares of the transitionmatrix elements for intermediate coupling from the ground state to the excited manifold, and Ω_t are the three phenomenological JO parameters.



Figure 2. ${}^{4}F_{32} \rightarrow {}^{4}I_{112}$ fluorescence spectra of Nd³⁺ doped in YSAG and YAG at room temperature.

The numerical values of the transition-matrix elements for Nd³⁺ were taken from DeShazer [37].

In practice, the integrated absorption coefficient, $\Gamma = \int \alpha(\lambda) d\lambda$ emanating from the ground state (the ${}^{4}I_{9/2}$ manifold) was measured for 11 absorption bands using figure 1. The integrated absorption coefficient in turn is related to the line strength *S* by equation (3):

$$\Gamma = \frac{8\pi^3 N \overline{\lambda} e^2}{3ch(2J+1)} \frac{(n^2+2)^2}{9n} S(J,J')$$
(3)

where *N* is the Nd³⁺ concentration, *J* is the total angular momentum quantum number of the initial level $\overline{\lambda}$ is the mean wavelength, and *n* is

the index of refraction. The values for *n* were taken from Sellmeier's dispersion equation, equation (1). When the absorption band was a superposition of lines assigned to several intermultiplet transitions, the matrix element was taken to be the sum of the corresponding squared matrix elements. The JO parameters were obtained by minimizing the sum of the squared differences between S_{meas} and S_{calc} . Table 4 shows Γ , *n*, S_{meas} , and S_{calc} for 11 absorption bands. The rms error of these calculations was 14 percent.

Once the JO parameters are known, S_{calc} was determined for transitions between ${}^{4}F_{3/2}$ and ${}^{4}I_{l}$ using the matrix elements emanating

Table 4. Absorp- tion intensities	Excited state W	avelength	n	Γ (nm/cm)	S _{mean} *	S _{celc} *
for Nd:YSAG at	4F _{3/2}	880	1.862	34.3	0.673	0.921
270 K	⁴ F _{5/2} , ² H _{9/2}	805	1.865	159.2	3.408	3.177
	4F _{7/2} , 4S _{3/2}	747	1.868	132.3	3.046	3.289
	4F _{9/2}	680	1.871	10.3	0.260	0.217
	${}^{2}H_{11/2}$	625	1.876	1.3	0.036	0.058
	⁴ G _{5/2} , ² G _{7/2}	583	1.880	75.8	2.213	2.242
	⁴ G _{7/2} , ⁴ G _{9/2} , ² K _{13/2}	520	1.888	50.9	1.655	1.253
	${}^{2}G_{9/2'} {}^{4}G_{11/2'} {}^{2}K_{15/2'} ({}^{2}D, {}^{2}P)_{3/2}$	47 0	1.897	17.4	0.621	0.310
	${}^{2}P_{1/2'}{}^{2}D_{5/2}$	430	1.907	2.3	0.089	0.127
	² P _{3/2}	385	1.923	0.1	0.004	0.006
	${}^{4}D_{3/2'} {}^{4}D_{1/2'} {}^{4}D_{5/2'} {}^{2}I_{11/2}$	357	1.936	35.0	1.593	1.733

*In units of 10^{-20} cm².

rms line strength of $S = 1.705 \times 10^{-20} \text{ cm}^2$.

rms deviation of line strength (ΔS)_{ms} = 0.239 × 10⁻²⁰ cm²,

 $(\Delta S)_{rms} = {\Sigma(\Delta S^2)/(\text{No. of bands fitted} - \text{No. of parameters})}^{1/2}$

from the metastable ${}^{4}F_{3/2}$ state [34]. The total spontaneous emission probability A(J,J') was calculated from

$$A(JJ') = \frac{64\pi^4 e^2}{3h(2J+1)\overline{\lambda}^3} \frac{n(n^2+2)^2}{9} S(JJ') , \quad (4)$$

and the intermanifold branching ratio $\beta(J,J')$ is given by

$$\beta(JJ) = \frac{A(JJ)}{\Sigma A(JJ)} .$$
 (5)

The JO parameters and the predicted branching ratios are given in table 5 for Nd: YSAG. For comparison, the JO parameters and both the predicted and experimentally determined branching ratios [39] for Nd:YAG are given. It is interesting to note that the experimental JO parameters are virtually the same for both YSAG and YAG. These results in turn yield comparable radiative lifetimes of the ${}^{4}\!F_{3/2}$ upper laser levels and branching ratios to the I_1 states.

Finally, the stimulated emission cross section σ_{2} , for an inhomogeneously broadened linewidth (Gaussian lineshape) can be written as

$$\sigma_{21} = \frac{A_{21}\lambda^2}{4\pi n^2 \Delta v} \left(\frac{\ln 2}{\pi}\right)^{1/2} , \qquad (6)$$

The transition probability for the laser transition $(2 \rightarrow 1)$ is given by

$$A_{21} = \frac{(1+\kappa)}{\kappa} \left(\frac{I_{21}}{I_T}\right) \tau_{nal}^{-1} , \qquad (7)$$

where κ is the Boltzmann factor between the two levels of ${}^{4}F_{3/2}$, and I_{21}/I_{T} is the ratio of the photon rate for the laser transition to the photon rate of all transitions originating from either level of ${}^{4}F_{3/2}$. Using the value $\Delta v = 8 \text{ cm}^{-1}$ from the literature [10] and the experimentally determined values of n = 1.86, $\lambda = 1.0622 \times 10^{-4}$ cm, and $A_{21} = 780 \text{ s}^{-1}$, we determine the value of $\sigma(R_{21})$ \rightarrow Y₃) to be 4.0 × 10⁻¹⁹ cm².

2.6 Crystal-Field Calculations

The analysis of the experimental absorption data on Nd³⁺:YSAG is the same as that of Nd³⁺:LaLuGG given by Allik et al [40]. In these calculations, the experimentally determined Stark-level positions of Nd³⁺ given in table 3

Table 5. Experimental andcalculated Judd-Ofeltparameters and predicted		Values for Nd ³⁺ :YSAG		Values for Nd ³⁺ :YAG		
branching ratios for Nd ³⁺ :YSAG and	Judd-Ofelt parameters	Exp	Theo	Exp*	Exp ^b	Theo
Nd³+:YAG [16,34]	$Q_{10^{-20}}$ cm ²)	0.23	0.16	0.37	0.2	0.35
Note: Experimental	Ω_{10}^{2} (10 ⁻²⁰ cm ²)	2.87	1.79	2.29	2.7	2.36
branching ratios for Nd:YAG from reference 39	Ω_{6}^{4} (10 ⁻²⁰ cm ²)	4.78	10.81	5.97	5.0	13.02
are given in square brackets.	Radiative lifetime ⁴ F _{3/2} (μs)	250	173	259	261	128
	Branching ratios (%)					
	$\beta({}^{4}F_{3/2} \rightarrow {}^{4}I_{9/2})$	37.8	23	32	37 [25]	21
	$\beta({}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2})$	49.4	62	53	50 [60]	62
	$\beta({}^{4}F_{3/2} \rightarrow {}^{4}I_{13/2})$	12.4	15	15	13 [15]	16
	$\beta({}^{4}F_{3/2} \rightarrow {}^{4}I_{15/2})$	0.4	—		_	1

^aReference 16. ^bReference 34.

were used along with the free-ion Russell-Saunders [*SL*]*J* states with the free-ion Hamiltonian containing the Coulomb, spin-orbit, L^2 , $G(G_2)$, and $G(R_7)$ interactions [41]. The phenomenological crystal-field parameters were obtained by a least-squares fit of the calculated energy levels to the experimental energy levels. The theoretical energy levels were obtained using the crystal-field Hamiltonian

$$H_{CEF} = \sum_{ikq} B_{kq}^{\star} C_{kq}(\hat{r}_i) , \qquad (8)$$

with k = 2, 4, 6 and $-k \le q \le k$. The B_{kq} are the crystal-field parameters, and the $C_{kq}(r)$ are spherical tensors. The sum on *i* in equation (8) covers the three electrons in the $4f^3$ electronic configuration of Nd³⁺. Since we assume that the Nd³⁺ ions occupy the dodecahedral site with D_2 symmetry, the crystal-field parameters can be chosen real; thus there is a total of nine even-k B_{kq} . In this fitting, 9 out of 60 experimental levels were discarded because attempts to fit these levels were unsuccessful.

The positions of the energy levels of Nd:YSAG are quite similar to those of Nd:YAG;

consequently, the crystal-field parameters of the latter [17] were chosen as starting parameters in the least-squares fitting. The resulting parameters that gave the best fit in units of cm⁻¹ are

$$B_{20} = 588, \quad B_{22} = 40.8, \quad B_{40} = -192, \\ B_{42} = -1877, \quad B_{44} = -1194, \quad B_{60} = -1620, \\ B_{62} = -805, \quad B_{64} = 797, \quad B_{66} = -612, \\ \text{with an rms deviation of } 9.1 \text{ cm}^{-1}.$$

As pointed out by Leavitt [42], the concept of rotational invariance is a convenient measure of the overall strength of the crystal field for comparison of the resulting B_{kq} for the same ion in different crystals. Here we define the rotational invariants, S_{kr} by

$$S_{k} = \left(\sum_{q=-k}^{k} B_{kq}^{*} B_{kq}\right)^{1/2}$$
(9)

for k = 2, 4, and 6. The values of S_k for Nd:YAG [17] and those computed for Nd:YSAG are $S_2(YAG) = 545 \text{ cm}^{-1}$, $S_2(YSAG) = 591 \text{ cm}^{-1}$, $S_4(YAG) = 3159 \text{ cm}^{-1}$, $S_4(YSAG) = 3152 \text{ cm}^{-1}$, $S_6(YAG) = 2548 \text{ cm}^{-1}$, $S_6(YSAG) = 2437 \text{ cm}^{-1}$.

Initially, these results are rather surprising in view of the difference in cell size of YAG (*a* = 12.000 Å) and YSAG (*a* = 12.271 Å), which would predict that the YAG crystal-field parameters would be much larger than those of YSAG. However, the distances from the yttrium site to the nearest oxygens are, for YAG, 2.303 Å (×4) and 2.432 Å (×4), and for YSAG, 2.338 Å (×4) and 2.440 Å (×4). Since these values are very similar, it is not surprising that the rotational invariants are comparable if it is assumed that the crystal-field parameters are predominately determined by the nearest-neighbor oxygen ions.

In order to calculate the intensity of the electric dipole transitions, we need the odd-*k* crystal-field components, A_{kq} (cm⁻¹/Å^k). In the point-charge model, the crystal-field components are given by [43]

$$A_{kq} = -e^{2} \sum_{j} \frac{q_{j} C_{kq}(\widehat{R}_{j})}{R_{j}^{k+1}} , \qquad (10)$$

where R_j is the location of the ion with charge q_j (in units of the electronic charge) relative to the rare-earth site. We assume that the charges on the individual ions are $q_Y = 3$, $q_{sc} = 3$, and $q_{Al} = -5$ $-4q_{or}$, with q_o being the charge on the oxygen ions (note that when q_o is taken at the valence value, -2, q_{Al} is at its valence value of 3). The choice of covalency effects between the oxygen and Al site was made based on the fact that the Al-O distance is very small, 1.77 Å, compared to any other inter-ionic distances (the next smallest distance, Sc-O, is 2.07 Å). In the pointcharge model, the crystal-field parameters are given by

$$B_{kq} = \rho_k A_{kq} , \qquad (11)$$

where the ρ_k are radial factors given by Morrison and Leavitt [44]. Using the values of ρ_k for Nd³⁺, a set of experimental A_{kq}^e was obtained from the B_{kq} values; these values are given in the top row of table 6. These experimental A_{kq}^e were used to obtain the best value of q_0 that fit the A_{kq} obtained from equation (10). Based on a value of $q_0 = -1.79$, the odd- $k A_{kq}$ (cm⁻¹/Å^k) from equation (10) are

$$A_{32} = 1102,$$

$$A_{52} = -2179,$$

$$A_{54} = 1211,$$

$$A_{72} = 71.40,$$

$$A_{74} = 152.9, \text{ and}$$

$$A_{76} = -200.3$$

(all these odd- $k A_{ba}$ are imaginary).

Having obtained a set of crystal-field components A_{iq}^{e} by the above procedure, one can obtain a set of crystal-field components for the entire rare-earth series by using equation (11). These results are given in table 6. These B_{iq} can serve as starting parameters for fitting the spectra of any rare-earth ion in YSAG. We refer to the crystal-field parameters obtained by this process as smoothed B_{iq} , since the process is usually used when the experimental data are analyzed on two or more rare-earth ions, in which case the experimental B_{iq} are forced, to a certain degree of consistency, for the entire rare-earth series.

The best-fit B_{kq} and the resulting values of the odd- $k A_{kq}$ were used to calculate the intensity of the electric- and magnetic-dipole transitions for the rare-earth series. A detailed discussion of this calculation is given by Leavitt and Morrison [45]. The resulting theoretical JO intensity parameters are given in table 7 for the rare earths. In addition, the theoretical JO parameters, manifold-to-manifold branching ratios, and radiative lifetimes of the ${}^{4}F_{3/2}$ state for Nd:YSAG and Nd:YAG are given in table 5.

The individual Stark-level line strengths for all the crystal-field split levels of the multiplets ${}^{4}I_{9/2}$ through ${}^{2}G_{9/2}$ were calculated. From these line strengths, the branching ratios for the two levels of the ${}^{4}F_{3/2}$ (E = 11,423 cm⁻¹ (No. 27) and E = 11,523 cm⁻¹ (No. 28)) to the lower ${}^{4}I_{i}(9/2 \le J \le 15/2)$ crystal-field split levels were determined. The line strengths of the magnetic dipole operators were found to be less than the corresponding electric dipole line strengths by two orders of magnitude in almost all the transitions and were ignored in the calculation. In table 8 we give the branching ratios from both levels of the ${}^{4}F_{3/2}$ state to all the levels below. The Sellmeier dispersion equation (eq (1)) was used to obtain these results.

Ion	B ₂₀	B ₂₂	B ₄₀	B ₄₂	B44	B ₆₀	B ₆₂	B ₆₄	B ₆₆
Akq	3447	239	-332	-3250	-2067	-1019	-506	501	-385
Ce	635	44	-251	-2449	-1558	-2386	-1186	1174	-902
Pr	605	42	-215	-2101	-1336	-1911	-950	940	-722
Nd	588	41	-192	-1877	-1194	-1620	-805	797	-612
Pm	579	40	-177	-1735	-1104	-1449	-720	713	-547
Sm	575	40	-168	1641	-1044	-1346	-669	662	-509
Eu	574	40	-161	-1572	-1000	-1274	-633	627	-481
Gd	575	40	-155	-1513	- 96 2	-1210	601	595	-457
Тb	577	40	-149	-1459	-928	-1145	-569	563	-432
Dy	57 9	40	-144	1411	897	-1082	-537	532	-409
Ho	583	40	-140	-1370	872	-1031	-512	507	-390
Er	588	41	-137	-1341	853	-1001	-498	493	-378
Tm	594	41	-135	-1317	-838	-983	-489	484	-371
Yb	5 99	42	-131	-1280	814	-929	-462	457	-351

Table 7. Calculated Judd-Ofelt intensity parametersTable 8. Line-to-line branching ratios (%) of twoTable 7. Calculated Judd-Ofelt intensity parameterslevels of ${}^{4}F_{3/2}$ levels of ${}^{4}F_{3/2}$ levels of ${}^{4}I_{j}$ manifolds Ω_{k} of rare-earth ions in Y site of Y₃Sc₂Al₃O₁₂(j = 1-26).

	JO intens	sity parameters	$(10^{-20}\mathrm{cm}^2)$	Manifolds	j	E (cm ⁻¹)	$\beta_{27 \rightarrow j}$	β _{28→j}
Ion	Ω.	Ω.	Ω,	4I9/2	1	0	5.8	1.2
		4	6	.,_	2	114	2.7	3.4
Ce	0.3031	5.634	46.20		3	183	1.9	5.3
					4	301	11.5	7.9
Pr	0.1696	2.846	19.70		5	823	0.3	0.3
Nd	0.1635	1.789	10.81	4 <i>I</i> _{11/2}	6	1979	22.5	4.6
			-	·	7	2022	23.1	12.9
Pm	0.09461	1.392	8.309		8	2101	2.9	25.6
					9	2136	3.4	9.8
Sm	0.08253	1.172	6.878		10	2437	2.6	5.5
			0.0.0		11	2495	5.5	6.2
Eu	0.06529	0.9122	4.864					
		•••		4I _{13/2}	12	3905	3.8	4.8
Gd	0.05109	0.7031	3.359	, -	13	3929	4.0	2.4
	0.00-07	0	0.007		14	4029	2.2	3.7
ТЪ	0.08921	1.131	8.146		15	4044	2.5	2.9
			0.010		16	4411 *	0.4	1.7
Dv	0.0627	0.8209	5.055		17	4419	1.4	0.4
2,	0.002	0.0207	0.000		18	4478	2.3	0.5
Но	0.05357	0.6590	3.691					
				4I _{15/2}	19	5766	0.1	0.4
Er	0.05137	0.6182	3.481		20	5797	0.0	0.3
					21	5927	0.3	0.1
Tm	0.04996	0.5897	3.363		22	5981	0.4	0.1
					23	6544	0.0	0.0
Yb	0.04291	0.4862	2.562		24	6560	0.1	0.0
					25	6622	0.1	0.1
					26	6711	0.2	0.0

*****Theoretical level

2.7 Laser Experiments

A long-pulse laser performance study of Nd:YSAG at 1.06 μ m was undertaken using diode array excitation in the side-pump configuration. A diode array capable of producing 475 W in a 300- μ s pulse was used as the excitation source. Details of the diode array and experimental procedures have been published previously [1].

One rod and one straight-through slab were fabricated from the same 6.35-mmdiameter "cored out" stock material by Lightning Optical Corp. (Tarpon Springs, FL). Both materials were 15 mm in length and had appropriate HR and AR coatings centered at 1.06 µm applied on opposite ends. The rod was 6.35 mm in diameter with the barrel polished. The slab was 3 mm thick. AR and HR coatings centered at 808 nm were applied on the side surfaces of the slab and the rod barrel to maximize diode absorption.

Of the two samples, the slab yielded the better results. The presence of significant optical (index-of-refraction) distortions in both samples was quite evident when the laser cavity was being aligned with a HeNe laser. The rod had an extremely high threshold, and laser oscillation could only be detected with a 99.9percent output coupler at an input power of 360 W. The improved performance (lower threshold) of the slab may be attributed to the better geometrical coupling of the two-dimensional diode array to the slab than to the rod. The optical slope efficiencies and extrapolated thresholds for various reflectivity output couplers are shown in table 9 for the slab. The round-trip (Findlay-Clay) resonator loss was 20.2 ± 0.2 percent.

3. Summary and Conclusions

The Judd-Ofelt intensity parameters for Nd:YSAG have been established by two different approaches. The first treats the parameters as phenomenological and adjusts them by directly fitting them to the experimentally meas-

excitation						
Output coupler reflectivity	Optical slope efficiency (%)	Extrapolated threshold (mJ)				
0.975 (∞)	4.2	48.1				
0.965 (∞)	4.6	50.0				
0.961 (∞)	5.1	52.0				
0.908 (∞)	5.6	63.2				
0.975 (63.5 cm)	7.6	47.0				
0.950 (63.5 cm)	9.2	53.9				
0.915 (75.0 cm)	8.4	55.1				

 Table 9. Laser slope efficiencies and thresholds

 for Nd:YSAG using side-pump diode array

 excitation

ured line strengths. The second approach uses the results of a point-charge electrostatic model to predict values for the odd- $k A_{kq}$ terms in the crystal-field expansion and then calculates a set of predicted intensities. Through a least-squares fitting subroutine, the predicted and observed intensities are reconciled, and a set of JO parameters is then calculated.

Overall good agreement between observed and calculated intensities eludes both approaches for several reasons. The model does not include dynamic lattice contributions or strain-broadening effects. The measured lifetimes usually include nonradiative contributions in emission. In absorption, multiple (minority) site absorption and phonon sidebands contribute to the measured absorption cross section. For example, in the experimental (first) method, S_{calc} from the ground state to the $F_{3/2}$ manifold is larger than S_{meas} for both YAG and YSAG. This leads to a 20-percent error in the calculated branching ratio to the ${}^{4}I_{1}$ manifolds in Nd:YAG [34]. On the other hand, this method does predict the radiative lifetime of the ${}^{4}F_{3/2}$ state very well, provided the Nd concentration can be accurately determined. The theoretical (second) method predicts too small a radiative lifetime for the metastable state but does predict very well the manifold-to-manifold and line-to-line Stark transitions (see table 6 and compare table 8 to fig. 2). Additional comparisons of these two models have been published for Nd³⁺ in Y₂O₃ [46].

Slope efficiencies of 47 percent have been obtained for Nd:YAG using a diode array in the side-pump configuration with thresholds of approximately 20 mJ [1]. Under similar conditions, our present Nd:YSAG crystal obtained a best slope efficiency of only 9.2 percent. This is due to the much poorer optical quality of the crystal than is found for Nd:YAG. At high Nd³⁺ concentrations, Nd:YSAG has the advantage over Nd:YAG because the fluorescence lifetime is longer. The lower nonradiative transition rate of Nd:YSAG versus Nd:YAG can be attributed to greater distance between Nd ion pairs in YSAG. This yields fewer ion/ion interactions which quench the fluorescence. Thus, if more effort can be given to improving the optical quality of YSAG, as has been done for YAG, the Nd:YSAG crystal is potentially a better Q-switch laser than Nd:YAG.

Acknowledgements

SAIC gratefully acknowledges financial support from the Center for Night Vision and Electro-Optics. The authors thank C. F. Campana for the crystal structure solution, R. Phillips and W. Hovis for the elemental analysis on YSAG, V. King for technical assistance in the lifetime determination, L. Thompson for polishing samples, and L. Merkle for reviewing the manuscript. JBG wishes to thank the American Society for Engineering Education for their support, and M. E. Hills, Chemistry Division, Naval Weapons Center, China Lake, CA, for many helpful discussions and encouragement.

References

- 1. T. H. Allik, W. W. Hovis, D. P. Caffey, and V. King, Opt. Lett. **14** (1989), 116.
- J. Berger, D. F. Welch, D. F. Scifres, W. Streifer, and P. Cross, Electron. Lett. 23 (1987), 669.
- B. Zhou, T. J. Kane, G. J. Dixon, and R. L. Byer, Opt. Lett. 10 (1985), 62.
- 4. R. A. Fields, M. Birnbaum, and C. L. Fincher, Appl. Phys. Lett. **51** (1987), 1885.
- 5. F. Hanson and D. Haddock, Appl. Opt. 27 (1988), 80.
- 6. F. Hanson and G. Imthurn, IEEE J. Quantum Electron. **QE-24** (1988), 1811.
- J. B. Gruber, M. E. Hills, C. A. Morrison, G. A. Turner, and M. R. Kokta, Phys. Rev. B2 37 (1988), 8564.

- R. Burnham and A. D. Hayes, Opt. Lett. 14 (1989), 27.
- A. L. Denisov, E. V. Zharikn, A. I. Zagumennyi, S. P. Kalitin, V. A. Smirnov, A. I. Talybov, and I. A. Shcherbakov, Zh. Prikl. Spektrosk. 49 (1988), 430.
- A. A. Kaminskii, *Laser Crystals*, Springer, New York (1981).
- A. G. Avanesov, A. A. Danilov, A. L. Denisov, E. V. Zharikov, A. I. Zagumennyi, O. V. Kuz'min, M. Yu. Nikol'skii, V. G. Ostroumov, V. F. Pisarenko, Academician A. M. Prokhorov, V. A. Smirnov, I. T. Sorokina, E. V. Tumaev, and I. A. Shcherbakov, Sov. Phys. Dokl. 32 (1987), 665.

- 12. Kh. S. Bagdasarov, A. A. Kaminskii, A. M. Kevorkov, and A. M. Prokorov, Sov. Phys. Dokl. **19** (1975), 671.
- 13. M. Kokta, J. Solid State Chem. 8 (1973), 39.
- 14. C. D. Brandle and R. L. Barns, J. Crystal Growth 20 (1973), 1.
- 15. V. F. Kitaeva, E. V. Zharikov, and I. L. Chistyi, Phys. Status Solidi **a92** (1985), 475.
- 16. A. A. Kaminskii and L. Li, Phys. Status Solidi **a26** (1974), K21.
- C. A. Morrison and R. P. Leavitt, "Spectroscopic Properties of Triply Ionized Lanthanides in Transparent Host Materials," in Handbook of the Physics and Chemistry of Rare Earths, Vol. 5, eds. K. A. Gschneidner, Jr., and L. Eyring, North-Holland, New York (1982), 461–684.
- 18. G. Huber, E. W. Duczynski, P. Mitzscherlich, and H. O. Teichmann, J. Phys. Paris 48 (1987), C7-309.
- E. W. Duczynski, H. J. v.d. Heide, G. Huber, P. Mitzscherlich, K. Petermann, and H. O. Teichmann, in Conference on Lasers and Electro-Optics, Technical Digest Series 1989, Optical Society of America, Washington, DC (1989), paper TuJ58.
- 20. B. R. Judd, Phys. Rev. 127 (1962), 750.
- 21. G. S. Ofelt, J. Chem. Phys. 37 (1962), 511.
- C. A. Morrison, N. Karayianis, and D. E. Wortman, Rare-Earth Ion-Host Lattice Interactions, 4.—Predicting Spectra and Intensities of Lanthanides in Crystals, Harry Diamond Laboratories, HDL-TR-1816 (June 1977).
- 23. L. G. Morozova and P. P. Feofilov, Izv. Akad. Nauk. SSSR, Neorg. Mater. 4 (1968), 1738.
- 24. C. D. Brandle and J. C. Vanderleeden, IEEE J. Quant. Electron. **QE-10**, No. 2 (1974), 67.
- D. Pruss, G. Huber, A. Belmowski, V. V. Laptev, I. A. Shcherbakov, and Y. V. Zharikov, J. Appl. Phys. B28 (1982), 355.
- R. Phillips, Kevex Instruments, 50 Valley Stream Parkway, Malvern, PA, 19355 (unpublished).

- 27. G. A. Bogomolova, L. A. Bumagina, A. A. Kaminskii, and B. Z. Malkin, Sov. Phys. Solid State 19 (1977), 1428.
- C. F. Campana, Nicolet X-ray Division, 5225-5 Verona Road, Madison, WI 53711 (unpublished).
- 29. W. L. Bond, J. Appl. Phys. 36 (1965), 1674.
- S. H. Wemple and W. J. Tabor, J. Appl. Phys. 44 (1973), 1395.
- J. B. Gruber, M. E. Hills, C. K. Jayasankar, F. S. Richardson, and T. H. Allik, Energy Levels and Spin-Correlation Crystal Field Effects: Nd³⁺ (4f³) in Y₃Al₅O₁₂, Y₃Sc₂Al₃O₁₂, Gd₃Sc₂Ga₃O₁₂, and La₃Lu₂Ga₃O₁₂, manuscript in preparation.
- 32. J. D. Axe, J. Chem. Phys. 39 (1963), 1154.
- W. F. Krupke and J. B. Gruber, Phys. Rev. 139 (1965), A2008.
- 34. W. F. Krupke, IEEE J. Quantum Electron. QE-7 (1971), 153.
- 35. W. F. Krupke, IEEE J. Quantum Electron. QE-10 (1974), 450.
- 36. M. J. Weber, T. E. Varitmos, and B. M. Matsinger, Phys. Rev. B8 (1973), 47.
- T. S. Lomheim and L. G. DeShazer, J. Appl. Phys. 49 (1978), 5517.
- T. S. Lomheim and L. G. DeShazer, Phys. Rev. B20 (1979), 4343.
- E. Comperchio, M. Weber, and R. Monchamp, High Quality Nd:YAG Laser Materials, U.S. Army Electronics Command, Fort Monmouth, NJ, Final Report, Contract DAAB07-69-C-0227 (1970).
- T. H. Allik, S. A. Stewart, D. K. Sardar, G. J. Quarles, R. C. Powell, C. A. Morrison, G. A. Turner, M. R. Kokta, W. W. Hovis, and A. A. Pinto, Phys. Rev. B37 (1988), 9129.
- B. G. Wybourne, Spectroscopic Properties of Rare Earths, Wiley, New York, (1965). [The free-ion parameters E^k, α, β, γ, and ζ are given in Ref. 40.]
- 42. R. P. Leavitt, J. Chem. Phys. 77 (1982), 1661.

- 43. C. A. Morrison, Angular Momentum Theory Applied to Interactions in Solids, Lecture Notes in Chemistry 47, Springer-Verlag, New York (1988).
- 44. C. A. Morrison and R. P. Leavitt, J. Chem. Phys. **71** (1979), 2366.
- 45. R. P. Leavitt and C. A. Morrison, J. Chem. Phys. **73** (1980), 749.
- 46. C. A. Morrison, R. P. Leavitt, J. B. Gruber, and N. C. Chang, J. Chem. Phys. **79** (1983), 4758.

DISTRIBUTION

ADMINISTRATOR DEFENSE TECHNICAL INFORMATION CENTER ATTN DTIC-DDA (12 COPIES) CAMERON STATION, BUILDING 5 ALEXANDRIA, VA 22314-6145 DIRECTOR NIGHT VISION & ELECTRO-OPTICS CENTER ATTN TECHNICAL LIBRARY ATTN R. BUSER ATTN A. PINTO ATTN J. HABERSAT ATTN R. RHODE ATTN W. TRESSEL FT BELVOIR. VA 22060 DIRECTOR DEFENSE NUCLEAR AGENCY ATTN TECH LIBRARY WASHINGTON, DC 20305 UNDER SECRETARY OF DEFENSE RES & ENGINEERING ATTN TECHNICAL LIBRARY, 3C128 WASHINGTON, DC 20301 OFFICE OF THE DEPUTY CHIEF OF STAFF, FOR RESEARCH, DEVELOPMENT, & ACQUISITION DEPARTMENT OF THE ARMY ATTN DAMA-ARZ-B, I. R. HERSHNER WASHINGTON, DC 20310 COMMANDER US ARMY ARMAMENT MUNITIONS & CHEMICAL COMMAND (AMCCOM) US ARMY ARMAMENT RESEARCH & DEVELOPMENT CENTER ATTN DRDAR-TSS, STINFO DIV DOVER, NJ 07801 COMMANDER ATMOSPHERIC SCIENCES LABORATORY ATTN TECHNICAL LIBRARY WHITE SANDS MISSILE RANGE, NM 88002 ATTN TECH LIBRARY ADEDDEEN BROWING (DIRECTOR US ARMY BALLISTIC RESEARCH LABORATORY ATTN SLCBR-DD-T (STINFO) ABERDEEN PROVING GROUND, MD 21005 DIRECTOR US ARMY ELECTRONICS WARFARE LABORATORY ATTN J. CHARLTON ATTN DELET-DD FT MONMOUTH, NJ 07703

COMMANDING OFFICER **USA FOREIGN SCIENCE & TECHNOLOGY CENTER** FEDERAL OFFICE BUILDING ATTN DRXST-BS, BASIC SCIENCE DIV CHARLOTTESVILLE, VA 22901 COMMANDER US ARMY MATERIALS & MECHANICS RESEARCH CENTER ATTN DRXMR-TL, TECH LIBRARY WATERTOWN, MA 02172 US ARMY MATERIEL COMMAND 5001 WISENHOWER AVE ALEXANDRIA, VA 22333-0001 US ARMY MATERIEL SYSTEMS ANALYSIS ACTIVITY ATTN DRXSY-MP (LIBRARY) ABERDEEN PROVING GROUND, MD 21005 COMMANDER US ARMY MISSILE & MUNITIONS CENTER & SCHOOL ATTN ATSK-CTD-F ATTN DRDMI-TB, REDSTONE SCI INFO CENTER REDSTONE ARSENAL, AL 35809 COMMANDER US ARMY RESEARCH OFFICE (DURHAM) ATTN J. MINK ATTN M. STROSIO ATTN M. CIFTAN ATTN B. D. GUENTHER PO BOX 12211 RESEARCH TRIANGLE PARK, NC 27709 COMMANDER US ARMY RSCH & STD GRP (EUROPE) FPO NEW YORK 09510 COMMANDER US ARMY TEST & EVALUATION COMMAND ABERDEEN PROVING GROUND, MD 21005 COMMANDER US ARMY TROOP SUPPORT COMMAND ATTN DRXRES-RTL, TECH LIBRARY NATICK, MA 01762 OFFICE OF NAVAL RESEARCH ATTN J. MURDAY ARLINGTON, VA 22217

DISTRIBUTION (cont'd)

DIRECTOR NAVAL RESEARCH LABORATORY ATTN CODE 2620, TECH LIBRARY BR ATTN G. QUARLES ATTN G. KINTZ ATTN A. ROSENBAUM ATTN G. RISENBLATT ATTN CODE 5554, F. BARTOLI ATTN CODE 5554, L. ESTEROWITZ ATTN CODE 5554, R. E. ALLEN WASHINGTON, DC 20375 COMMANDER NAVAL WEAPONS CENTER NAVAL WEAPONS CENTERATTN CODE 3854, R. SCHWARTZATTN CODE 3854, M. HILLSATTN CODE 3844, M. NADLERATTN CODE 385, R. L. ATKINSATTN CODE 343, TECHNICAL INFORMATION9700 SOUTH CASS AVENUEARGONNE, IL 60439 DEP ARTMENT CHINA LAKE, CA 93555 AIR FORCE OFFICE OF SCIENTIFIC RESEARCH ATTN W. DROZDOSKI ATTN MAJOR H. V. WINSOR, USAF BOLLING AFB WASHINGTON, DC 20332 HQ, USAF/SAMI WASHINGTON, DC 20330 DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS ATTN LIBRARY WASHINGTON, DC 20234 NASA LANGLEY RESEARCH CENTER ATTN N. P. BARNES (2 COPIES) ATTN G. ARMAGAN ATTN P. CROSS ATTN D. GETTENY ATTN J. BARNES ATTN E. FILER ATTN C. BAIR ATTN N. BOUNCRISHANI HAMPTON, VA 23665 DIRECTOR ADVISORY GROUP ON ELECTRON DEVICES ATTN SECTRY, WORKING GROUP D 210 VARICK STREET NEW YORK, NY 10013 AEROSPACE CORPORATION ATTN M. BIRNBAUM ATTN N. C. CHANG ATTN T. S. ROSE PO BOX 92957

ALLIED ADVANCED APPLICATION DEPT ATTN A. BUDGOR 31717 LA TIEMDA DRIVE WESTLAKE VILLAGE, CA 91362 ALLIED SIGNAL INC. ATTN I. BAND ATTN R. MORRIS POB 1021R ATTN Y. BAND MORRISTOWN, NJ 07960 AMES LABORATORY DOE IOWA STATE UNIVERSITY ATTN K. A. GSCHNEIDNER, JR. (2 COPIES) ARGONNE NATIONAL LABORATORY ARGONNE, IL 60439 BOOZ, ALLEN AND HAMILTON 4330 EAST WEST HWY BETHESDA, MD 20814 BRIMROSE CORP OF AMERICA ATTN R. G. ROSEMEIER 7527 BELAIR ROAD BALTIMORE, MD 21236 DRAPER LAB ATTN F. HAKIMI MS 53 555 TECH. SO CAMBRIDGE, MA 02139 ENGINEERING SOCIETIES LIBRARY ATTN ACQUISITIONS DEPT 345 EAST 47TH STREET NEW YORK, NY 10017 FIBERTECH INC. ATTN H. R. VERDIN (3 COPIES) 510-A HERNDON PKWY HERNDON, VA 22070 HUGHES AIRCRAFT COMPANY ATTN D. SUMIDA 3911 MALIBU CANYON RD MALIBU, CA 90265

LOS ANGELES, CA 90009

DISTRIBUTION (cont'd)

IBM RESEARCH DIVISION ALMADEN RESEARCH CENTER ATTN R. M. MACFARLANE MAIL STOP K32 802(D) 650 HARRY ROAD SAN JOSE, CA 95120 DIRECTOR LAWRENCE RADIATION LABORATORY ATTN M. J. WEBER ATTN H. A. KOEHLER ATTN W. KRUPKE LIVERMORE, CA 94550 MARTIN MARIETTA ATTN F. CROWNE ATTN J. LITTLE ATTN WORCHESKY ATTN D. WORTMAN 1450 SOUTH ROLLING ROAD BALTIMORE, MD 21227 MIT LINCOLN LAB ATTN B. AULL PO BOX 73 LEXINGTON, MA 02173 DEPARTMENT OF MECHANICAL, INDUSTRIAL, & AEROSPACE ENGINEERING PO BOX 909 ATTN S. TEMKIN PISCATAWAY, NJ 08854 NATIONAL OCEANIC & ATMOSPHERIC ADM ENVIRONMENTAL RESEARCH LABS ATTN LIBRARY, R-51, TECH RPTS BOULDER, CO 80302 OAK RIDGE NATIONAL LABORATORY ATTN R. G. HAIRE OAK RIDGE, TN 37830 W. J. SCHAFER ASSOC. ATTN J. W. COLLINS 321 BILLERICA ROAD HELMSFORD, MA 01824 SCIENCE APPLICATIONS, INTERNATIONAL CORP ATTN T. ALLIK (10 COPIES) 1710 GOODRIDGE DRIVE McLEAN, VA 22102 UNION CARBIDE CORP ATTN M. R. KOKTA (10 COPIES) ATTN J. H. W. LIAW 750 SOUTH 32ND STREET WASHOUGAL, WA 98671

ARIZONA STATE UNIVERSITY DEPT OF CHEMISTRY ATTN L. EYRING TEMPE, AZ 85281 CARNEGIE MELLON UNIVERSITY SCHENLEY PARK ATTN PHYSICS & EE, J. O. ARTMAN PITTSBURGH, PA 15213 COLORADO STATE UNIVERSITY PHYSICS DEPARTMENT ATTN S. KERN FORT COLLINS, CO 80523 UNIVERSITY OF CONNECTICUT DEPARTMENT OF PHYSICS ATTN R. H. BARTRAM STORRS, CT 06269 UNIVERSITY OF SOUTH FLORIDA PHYSICS DEPT ATTN R. CHANG ATTN SENGUPTA TAMPA, FL 33620 JOHNS HOPKINS UNIVERSITY DEPT OF PHYSICS ATTN B. R. JUDD BALTIMORE, MD 21218 KALAMAZOO COLLEGE DEPT OF PHYSICS ATTN K. RAJNAK KALAMAZOO, MI 49007 MASSACHUSETTS INISTITUTE OF TECHNOLOGY CRYSTAL PHYSICS LABORATORY ATTN H. P. JENSSEN ATTN A. LINZ CAMBRIDGE, MA 02139 MASSACHUSETTS INSTITUTE OF TECHNOLOGY ATTN V. BAGNATO ROOM 26-251 77 MASS AVE CAMBRIDGE, MA 02139 UNIVERSITY OF MINNESOTA, DULUTH DEPARTMENT OF CHEMISTRY ATTN L. C. THOMPSON DULUTH, MN 55813

OKLAHOMA STATE UNIVERSITY DEPT OF PHYSICS

DEPARTMENT OC CHEMISTRY ATTN D. S. McCLURE PRINCETON, NJ 08544

SAN JOSE STATE UNIVERSITY DEPARTMENT OF PHYSICS ATTN J. B. GRUBER (10 COPIES) SAN JOSE, CA 95192

SETON HALL UNIVERSITY CHEMISTRY DEPARTMENT ATTN H. BRITTAIN SOUTH ORANGE, NJ 07099

UNIVERSITY OF VIRGINIA DEPT OF CHEMISTRY ATTN DR. F. S. RICHARDSON (2 COPIES) ATTN BAHDER, T., SLCHD-ST-AP ATTN DR. M. REID CHARLOTTESVILLE, VA 22901

UNIVERSITY OF WISCONSIN CHEMISTRY DEPARTMENT ATTN J. WRIGHT ATTN B. TISSUE MADISON, WI 62705

INSTALLATION SUPPORT LABORATORY ATTN LEGAL OFFICE, SLCIS-CC ATTN S. ELBAUM, SLCIS-CC

USAISC ATTN TECHNICAL REPORTS BRANCH, AMSLC-IM-TR (2 COPIES)

HARRY DIAMOND LABORATORIES PENNSYLVANIA STATE UNIVERSITY MATERIALS RESEARCH LABORATORY ATTN W. B. WHITE UNIVERSITY PARK, PA 16802 PRINCETON UNIVERSITY DEPARTMENT ATTN CHIEF, SLCHD-NW-CS ATTN CHIEF, SLCHD-NW-TS ATTN CHIEF, SLCHD-NW-RS ATTN CHIEF, SLCHD-NW-P ATTN CHIEF, SLCHD-NW-PO ATTN CHIEF, SLCHD-ST-C ATTN CHIEF, SLCHD-ST-RS ATTN CHIEF, SLCHD-TT ATTN KENYON, C. S., SLCHD-NW-EP ATTN MILETTA, J. R., SLCHD-NW-EP ATTN MCLEAN, F. B., SLCHD-NW-RP ATTN LIBELO, L., SLCHD-ST-MW ATTN BENCIVENGA, A. A., SLCHD-ST-SP ATTN SATTLER, J., SLCHD-CS ATTN NEMARICH, J., SLCHD-ST-SP ATTN WEBER, B., SLCHD-ST-CB ATTN BENCIVENGA, B., SLCHD-TA-AS ATTN BRUNO J., SLCHD-ST-AP ATTN DROPKIN. H., SLCHD-ST-AP ATTN EDWARDS A., SLCHD-ST-AP ATTN HAY G., SLCHD-ST-AP ATTN LEAVITT, R., SLCHD-ST-AP ATTN PHAM, J., SLCHD-ST-AP ATTN SIMONIS, G., SLCHD-ST-AP ATTN STEAD, M., SLCHD-ST-AP US ARMY LABORATORY COMMAND ATTN TECHNICAL DIRECTOR, AMSLC-CT ATTN TECHNICAL DIRECTOR, AMSLC-CT ATTN STELLATO, J., SLCHD-ST-AP ATTN TOBER, R., SLCHD-ST-AP ATTN TURNER, G., SLCHD-ST-AP ATTN WORTMAN, D., SLCHD-ST-AP ATTN GARVIN, C., SLCHD-ST-SS ATTN GOFF, J., SLCHD-ST-SS ATTN MORRISON C., SLCHD-ST-AP (10 COPIES)