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Decline in Coccolithophore Diversity and Impact on Coccolith Morphogenesis Along a Natural CO₂ Gradient

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A natural pH gradient caused by marine CO₂ Abstract. seeps off Vulcano Island (Italy) was used to assess the effects of ocean acidification on coccolithophores, which are abundant planktonic unicellular calcifiers. Such seeps are used as natural laboratories to study the effects of ocean acidification on marine ecosystems, since they cause longterm changes in seawater carbonate chemistry and pH, exposing the organisms to elevated CO₂ concentrations and therefore mimicking future scenarios. Previous work at CO₂ seeps has focused exclusively on benthic organisms. Here we show progressive depletion of 27 coccolithophore species, in terms of cell concentrations and diversity, along a calcite saturation gradient from Ω_{calcite} 6.4 to <1. Water collected close to the main CO2 seeps had the highest concentrations of malformed Emiliania huxleyi. These observations add to a growing body of evidence that ocean acidification may benefit some algae but will likely cause marine biodiversity loss, especially by impacting calcifying species, which are affected as carbonate saturation falls.

Introduction

Human-induced CO₂ emissions are causing rapid changes in surface ocean carbonate chemistry. These

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changes correspond to increases of both $[CO_2]$ and $[H^+]$, and decreases of pH, $[CO_3^{2-}]$, and of carbonate saturation states (Ω) . This process, called ocean acidification, has raised concerns about long-term effects on marine ecosystems (Caldeira and Wickett, 2003; Kroecker *et al.*, 2013a). High CO_2 concentrations affect the metabolism, growth, calcification, and behavior of many marine organisms (Rodolfo-Metalpa *et al.*, 2011; Kroeker *et al.*, 2013a). Some photosynthetic organisms benefit from increased levels of CO_2 , although many calcified species are adversely affected by corrosion or competition from non-calcified forms (Porzio *et al.*, 2011; Johnson *et al.*, 2013; Kroeker *et al.*, 2013b).

Coccolithophores are a major group of unicellular marine calcifying algae secreting calcite plates (coccoliths) covering the cell (coccosphere). Coccolithophores have been present since the Mesozoic (Bown, 1998). These phytoplankton are at the base of the food web and have been a focus of numerous ocean acidification studies because they are a major carbonate producer (Beare et al., 2013). They are an important component of the marine carbon cycle, contributing to carbonate and carbon export, acting as "ballast" by increasing particle density and sinking speeds, and providing protection from remineralization (Ziveri et al., 2007; Wilson et al., 2012). The vast majority of ocean acidification studies on plankton have been performed in culture and mesocosm experiments, in which the carbonate system has been manipulated by the addition of acid (e.g., HCl) or by bubbling of CO₂ (Riebesell et al., 2008). Results of these short-term studies show a range of coccolithophore calcification (particulate inorganic carbon) and productivity (particulate organic carbon) responses to changes in seawater carbonate chemistry (Riebesell *et al.*, 2000; Zondervan, 2007; Iglesias-Rodriguez *et al.*, 2008; Langer *et al.*, 2009). It is not known whether these organisms will evolve or adapt to long-term acidification of the oceans (Beare *et al.*, 2013)

An interest in the long-term effects of elevated pCO_2 has prompted *in situ* studies. Recent studies in the Mediterranean demonstrated that coccolithophore distribution and mass are strongly related to the carbonate chemistry of the seawater (Meier *et al.*, 2014; Oviedo *et al.*, 2014). Tyrrell *et al.* (2008) reported that coccolithophores are common in the Black Sea, where surface waters are saturated with carbonate year-round, but absent from the Baltic Sea, which has seasonal carbonate undersaturation. However, because salinity, temperature, and light vary with carbonate saturation in these two seas it is difficult to disentangle the effects of these potentially confounding factors. Other studies have highlighted the complexity of decoupling multiple environmental changes on coccolithophore calcification (*e.g.*, Smith *et al.*, 2009; Horigome *et al.*, 2014).

A comparison of water samples and sediments across a variety of locations has shown that the weight of individual coccoliths (mainly those of *Emiliania huxleyi*, the most common coccolithophore species, and *Gephyrocapsa*) increases as seawater [CO₃²⁻] increases (Beaufort *et al.*, 2011). However, highly variable coccolith weight reflects extensive *E. huxleyi* genome variability, which is why this species can thrive in very diverse habitats and under a wide variety of environmental conditions (Read *et al.*, 2013).

Natural CO₂ seeps are being used to examine the response of coastal ecosystems to ocean acidification (Hall-Spencer et al., 2008). Hundreds of benthic species have been investigated in these settings (Fabricius et al., 2011; Porzio et al. 2011; Inoue et al., 2013; Kroeker et al., 2013b). Meroplanktonic calcified organisms such as juvenile gastropods and foraminifera are adversely affected as CO₂ levels increase (Cigliano et al., 2010), but carbon dioxide seeps have not so far been used for the study of ocean acidification on holoplankton. Here, we analyzed phytoplankton along a natural pH gradient near underwater CO₂ seeps. Several papers have been published on the acidification issue in the Levante Bay (Boatta et al., 2013; Milazzo et al., 2014) and show that there is a clear pH gradient in a consistent coastline direction, exactly like during our survey. We examined coccolithophore cell concentration, diversity, and coccolith morphology as well as physicochemical parameters with an emphasis on seawater carbonate chemistry. Our aims were to assess whether CO2-driven acidification gradients revealed shifts in the diversity, abundance, and calcification of a natural coccolithophore assemblage.

Material and Methods

Seawater samples were collected on 23-26 November 2010 at eight stations along two transects running parallel to the shore off Vulcano Island in the Mediterranean Sea (Fig. 1). All stations were >200 m from a main area of CO₂ bubbling to avoid confounding factors such as the presence of H₂S (Boatta et al., 2013). The methods used to count coccolithophores in water samples followed standard procedures as detailed in several recent studies (Ziveri et al., 1995; Oviedo et al., 2014). All samples were treated and analyzed following these procedures. For each sample, 21 of seawater was collected in plastic or Niskin bottles from the surface 1 m of water. This was immediately passed through cellulose acetate filters (0.45-µm pore size, 47-mm diameter) using a low-pressure vacuum pump (<200 mm Hg). Each filter membrane was rinsed with distilled water buffered with ammonia after filtration to remove sea-salt, oven-dried at 40 °C, and stored in plastic petri dishes.

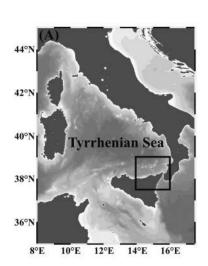
The samples were analyzed using polarized light microscopy for cell concentrations, and scanning electronic microscopy (SEM) for species identification and morphological examination of *E. huxleyi* specimens.

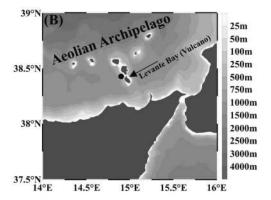
For light microscopy, a portion of filter was mounted onto a glass slide, made transparent by immersion oil, and fixed beneath a cover slip. Coccosphere counts were made using a Zeiss polarized light microscope at $1000\times$. The concentration of coccospheres was obtained by quantifying their number on a 5-mm² surface of the filtered area. The coccosphere abundance was calculated using the following equation:

$$CD = (A \times N)/(a \times V)$$

where CD = coccosphere density (number of coccospheres 1^{-1}); $A = \text{effective filtration area (mm}^2)$; N = total numberof coccospheres counted; $a = \text{analyzed area (mm}^2)$ and $V = \text{analyzed area (mm}^2)$ volume of filtered water (1). The methods for quantifying the species using SEM followed standard procedures (Ziveri et al., 1995, Oviedo et al., 2014). SEM analyses were performed by Zeiss EVO MA 10: a portion of filter membrane was mounted onto a stub for quantitative analysis of coccolith morphology of selected species in 50 coccospheres and for taxonomic identification following the concepts of Young et al. (2003) and Jordan et al. (2004). A heteromorphic life cycle including diploid cells producing heterococcoliths and haploid cells holococcoliths has been described for coccolithophores (Billard and Inouye, 2004, and reference therein). Holococcolithophore species have been clustered in a single group called Holococcolithophore spp.

Temperature, salinity, and pH measurements were acquired *in situ* using a YSI 556 MPS probe with an accuracy of \pm 0.1 °C, 0.1 salinity, and \pm 0.05 pH. Additional 50-ml seawater samples were collected for total alkalinity analyses; these were pumped through 0.2- μ m pore size filters,





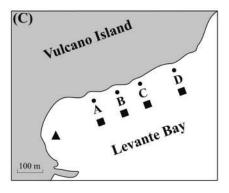


Figure 1. Bathymetry of (A) the central Mediterranean Sea and (B) the southern Tyrrhenian Sea. Black circle shows location of living coccolithophores collected in summer north of Vulcano and south of Lipari island (Bonomo, S., CNR, Consiglio Nazionale delle Ricerche, Istituto per l'Ambiente Marino Costiero, Naples, Italy unpubl. data). (C) Levante Bay, inshore (circles) and offshore (squares) sample stations and the main vent area (triangle).

poisoned with 0.05 ml of 50% $\rm HgCl_2$ to avoid biological alteration, and then stored in the dark at 4 °C. Three replicate subsamples were analyzed at 25 °C using a titration system (Mettler Toledo, Inc.). The pH was measured at 0.02-ml increments of 0.1 N HCl. Total alkalinity was calculated from the Gran function applied to pH variations from 4.2 to 3.0, as mEq $\rm Kg^{-1}$ from the slope of the curve HCl volume *versus* pH. The $\rm pCO_2$ and the saturation states of calcite and aragonite were calculated from pH $_T$, total alkalinity, temperature, and salinity, using the free-access $\rm CO_2$ SYS package ver. 2.1 (Dickson *et al.*, 2007).

We tested for differences in the physicochemical parameters recorded on different sampling dates along the inshore transect using permutational analysis of variance (Permanova; Anderson and Ter Braak, 2003). The four sampling stations were considered as a fixed factor. The analyses were performed on Euclidian distance of untransformed data using 9999 permutations. The software Primer 6-Permanova +B20 was used to perform these analyses (Clarke and Gorley, 2006). Linear regression analyses were also run to assess relationships between pH and $\Omega_{\rm calcite}$ (independent variables) and coccolithophore concentration, number of species, and percentage of malformed and corroded coccoliths (dependent variables).

Results

The CO₂ seeps did not affect temperature (range: 17.14– 18.52 °C; Permanova, pseudo- $F_{3.8} = 0.3859$, P = 0.7842), total alkalinity (2516-2563 µmol kg⁻¹; Permanova, pseudo- $F_{3.8} = 2.373$, P = 0.174), and water salinity (37.15– 37.66; Permanova, pseudo- $F_{3,8} = 0.704$, P = 0.565), across our inshore sampling stations (Table 1). In contrast, we recorded strong gradients in carbonate chemistry along the inshore transect; pH values increased significantly from station A (range: pH_T 6.84-7.32) at >200 m distance from the vents to station D (8.02-8.19 pH_T) farthest from the vents (Permanova, pseudo- $F_{3,8} = 10.3$, P = 0.007) (Table 1). There were corresponding gradients along the inshore transect for pCO_2 (Permanova, pseudo- $F_{3,8} = 5.398$, P =0.039) and $\Omega_{\rm calcite}$ levels (Permanova, pseudo- $F_{\rm 3,8}=6.975,$ P = 0.033). All the offshore stations had similar pCO_2 levels (400–501 μ atm) and pH values (pH_T 8.00–8.08). The Ω_{calcite} was >4.6 at the farthest inshore station and at all the offshore stations, with values decreasing closer to the seeps, sometimes falling to $\Omega_{\rm calcite}$ <1 at inshore stations A and B.

Coccolithophore concentrations ranged between 2.7 $\times 10^3$ and 2.8 $\times 10^4$ coccospheres 1^{-1} (Table 1) and were

Table 1.	Sampling sites,	dates, seawater	properties, and	coccolithophore	analysis of the	e studied samples
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					Total	$\Omega_{ m calcite} \ m calcite$	Coccolithophore		Emiliana huxleyi (% coccospheres)				
Sample	Temp (°C)	Salinity	pH _T (total scale)	pCO ₂ (μatm)	alkalinity (μkg ⁻¹)	saturation state	concentration (coccospheres l ⁻¹)	Number of species	Malformed	Corroded			
23 Nov 20	010 Inshore												
A	18.43	37.41	7.32	2822.7	2526.5	1.13	7384	9	19.0	7.1			
В	18.30	37.24	7.47	1963.5	2530.5	1.56	7030						
C	18.48	37.34	8.16	321.3	2535.9	6.19	9274	11	14.3	0.0			
D	18.01	37.57	8.19	292.3	2516.1	6.38	14378	7					
24 Nov 20	010 Inshore												
A	17.80	37.53	6.94	6971.1	2539.8	0.47	7633	5	11.4	2.3			
C	18.26	37.15	7.66	1222.5	2525.3	2.33	18621	5	9.5	0.0			
D	18.32	37.16	8.04	448.3	2517.9	4.94	14948	8	0.0	0.0			
25 Nov 20	010 Inshore												
A	17.18	37.66	6.84	8828.8	2554.3	0.37	2740	6	94.9	51.3			
В	17.14	37.49	7.01	5947.1	2563.8	0.55	6180						
C	17.80	37.57	7.56	1563.0	2520.3	1.86	12909	7	9.1	0.0			
D	18.28	37.47	8.02	474.7	2526.5	4.78	17011	11	5.9	0.0			
26 Nov 20	010 Offshore												
A	18.51	37.35	8.00	501.5	2525.9	4.63	19487	10	2.1	0.0			
В	18.52	37.34	8.02	474.1	2522.7	4.80	15818	9	10.0	0.0			
C	18.54	37.26	8.07	412.1	2519.0	5.26	20964	8	2.5	0.0			
D	18.52	37.31	8.08	400.3	2517.1	5.35	28306	12	6.3	0.0			

significantly correlated with increasing pH ($R^2 = 0.54$, P = 0.002, n = 15) (Fig. 2A), and with increasing Ω_{calcite} ($R^2 = 0.42$, P = 0.006, n = 15). Maximum cell density was usually higher at offshore stations, especially at station D (Fig. 2A). Our Nov 2010 sample from inshore station B 24 was lost, and SEM analysis was not conducted on 2 out of 15 samples (B inshore stations recovered on 23 and 25 November 2010), since inorganic nannoparticle material covered the filters, preventing taxonomic and morphological evaluation by SEM. These samples were, however, used for identification and quantification of coccospheres by polarized light microscopy. SEM observations of the D inshore 23 Nov 2010 sample allowed species identification but not detailed morphological examinations of *E. huxleyi* specimens.

We found 27 coccolithophore taxa in Levante Bay (Fig. 3). Species diversity decreased significantly as pH ($R^2=0.39$, P=0.016, n=13) (Fig. 2B) and $\Omega_{\rm calcite}$ ($R^2=0.39$, P=0.016, n=13) fell. *Emiliania huxleyi* was always dominant although *Umbellosphaera tenuis*, *Syracosphaera molischii*, and *Syracosphaera tumularis* were major components of the coccolithophore assemblages (Figs. 3 and 4). All coccospheres of *E. huxleyi* we sampled were Type A (*sensu* Young *et al.*, 2003), being 3–4 μ m wide with robust distal shield elements and curved central area elements. The diverse coccolithophore assemblages of Levante Bay (Fig. 4) are similar to those described in more open oligotrophic Mediterranean seawater settings (Knappertsbusch, 1993; Malinverno *et al.*, 2003; Bonomo *et al.*, 2012). These are the first coccolithophore assemblage data for the Tyrrhenian

Sea, but an unpublished survey between Vulcano and Lipari Islands (Fig. 1) also had a coccolithophore assemblage similar to data collected in Levante Bay in normal pH conditions: $E.\ huxleyi$ was dominant amongst the about 30 coccolithophore species present; a few more coccolithophore species were found offshore because the assemblage included deep photic zone taxa such as $Florisphaera\ profunda$ and $Gladiolithus\ flabellatus$, which were not found in the shallow waters of Levante Bay. The coccolithophore concentration at the offshore site was 3.3×10^4 coccosphere/l in the uppermost 25 m (samples collected at depths of 5 and 25 m) of the water column (Bonomo, S., CNR, Consiglio Nazionale delle Ricerche, Istituto per l'Ambiente Marino Costiero, Naples, Italy; unpubl. data).

Coccoliths and coccospheres were neither malformed nor corroded at the offshore site, whereas this was common in E. huxleyi specimens collected close to the CO₂ seeps (Fig. 4). Malformation was positively related to falling pH (R^2 = 0.45, P = 0.016, n = 12), as was dissolution ($R^2 = 0.44$, P = 0.018, n = 12) (Fig. 2C, D). However, malformation and dissolution were common only at the lowest pH and $\Omega_{\rm calcite}$ station (Fig. 2C), and this relationship needs to be carefully interpreted. Emiliania huxleyi coccospheres had malformed T-shaped elements on the distal shield that were not connected with each other, were twisted, were not completely calcified, or had some combination of those malformations; while signs of dissolution were evident in the central area (Fig. 4). Some E. huxleyi coccospheres had corroded coccoliths at stations with low calcite saturation $(\Omega_{\rm calcite} <$ 1.13). On coccospheres with their interlocked

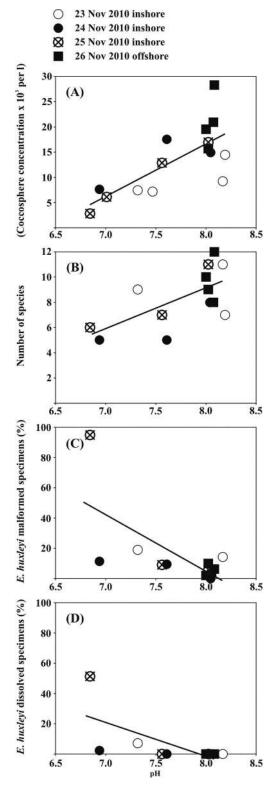


Figure 2. Coccosphere concentrations (A), species diversity (B), number of *Emiliana huxleyi* malformed (C) and corroded (D) coccoliths plotted *versus* pH; black, white, and crossed circles indicate samples collected from the inshore transect on 23–25 November 2010, respectively. Black squares indicate samples collected from the offshore transect on 26 November 2010. Linear regression lines are also shown.

liths still attached, the dissolution was most evident around the central area, where the central tube was partly corroded.

Discussion

Previous studies along pH gradients near Mediterranean volcanic seeps have demonstrated that although seawater acidification with CO₂ can benefit some organisms, such as heterokont algae, it is detrimental to most sessile calcified species and leads to a reduction in biodiversity (Porzio et al., 2011; Johnson et al., 2013; Kroeker et al., 2013b). Here we show that CO₂ seeps may also be useful for studying the effects of ocean acidification on plankton. We found a diverse coccolithophore assemblage that was typical of oligotrophic conditions in the Mediterranean. Coccolithophore abundance decreased significantly with decreasing pH; water near to the seeps (pH 6.84) contained far fewer species and had malformed and corroded specimens of E. huxleyi. Transplant experiments along the same CO₂ gradient have shown that recruitment of reef-building gastropods was adversely affected and that exposure to acidified conditions predicted for the year 2100 and beyond caused dissolution and a significant difference in the mineralogical composition of the recruited shells (Milazzo et al., 2014).

We found a progressive decrease in the diversity of coccolithophore species as ${\rm CO_2}$ levels increased and $\Omega_{\rm calcite}$ values fell. Coccolith malformation and dissolution occurs when pH is lowered in culture experiments (Bach et al., 2012; Kottmeier et al., 2014), and our results are the first evidence of this effect in natural coccolithophore assemblages. The fall in coccolithophore diversity and cell concentration pattern as CO2 levels increased is not explained by dissolution of the assemblage at the seep. Although cell abundance and species diversity increased toward offshore and reference (D inshore) stations, at the seep site we did not detect any evidence of species selection based on lightly versus heavily calcified species. For example, delicate species of Discophaera tubifera, Syracosphaera sp., and Holococcolithophores (Fig. 4) were present at the lowest pH stations (Table 2). Coccolithophore diversity and abundance within Levante Bay may be due to (1) slow surface currents that carry upper photic zone coccolithophores into the bay, with detrimental conditions resulting in assemblage depauperization toward the seeps; or (2) almost complete dissolution of the community at the seep and subsequent dilution further from this area, coupled with the mixing of new surface water from offshore. Future plankton surveys around CO2 seeps would benefit from the use of current meters to monitor water residence time.

Malformed coccoliths are almost absent in coccolithophores from the Mediterranean Sea (Cros and Fortuño, 2002; Malinverno *et al.*, 2008; Oviedo *et al.*, 2014), and the presence of malformed *E. huxleyi* coccoliths at the Vulcano

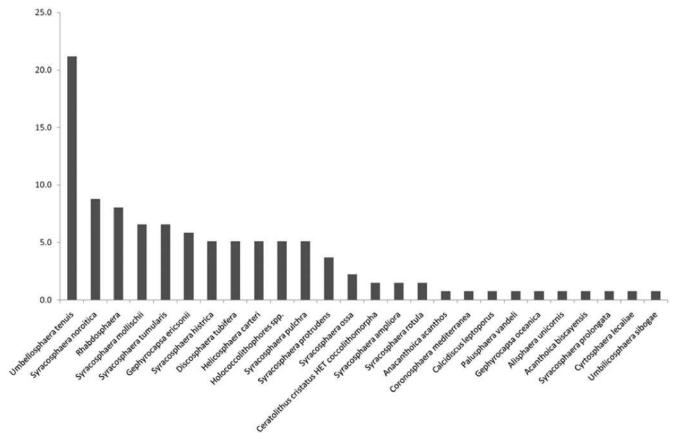


Figure 3. Percentage contribution of the diverse coccolithophore community plotted as average of the whole Levante Bay data set and ruling out the dominant (79%) species, *Emiliana huxleyi*.

seep site waters suggests anomalous environmental conditions affecting morphogenesis. The significant statistical relationship between pH values and malformed and corroded *E. huxleyi* coccoliths (Fig. 2C, D) is due to the massive malformation and dissolution in the A inshore (25NOV10) sample. Such a significant perturbation in the coccolithophore community did not occur the day before (sample A inshore 24NOV10), despite the fact that pH and carbonate chemistry seawater values were similar. Information on water residence time may explain this discrepancy. In any case, it seems that very low pH water was the primary cause of *E. huxleyi* malformation.

Strain-specific responses of *E. huxleyi* calcification rate under elevated CO₂ conditions have been reported in culture experiments and in upwelling regions, possibly due to the sensitivity of genetically different *E. huxleyi* strains to acidification (Langer *et al.*, 2009; Beaufort *et al.*, 2011; Read *et al.*, 2013). The genomic plasticity of *E. huxleyi* has recently been confirmed and may explain much of the phenotypic variation, ecological dynamics, and physiological heterogeneity observed in the geological record (Read *et al.*, 2013). Here we highlight the vulnerability of *E. huxleyi* type A specimens to lowered pH, which is expressed by reduced

cell density and abnormal morphogenesis, suggesting that H^+ is a major factor causing malformation, as demonstrated in laboratory experiments (Bach *et al.*, 2012)

An important consideration when working near CO₂ seeps is the possible influence of confounding factors such as dissolved metals and H₂S (Hall-Spencer *et al.*, 2008). At Vulcano, rocky shore transects have been established that are not contaminated with H₂S; these can, however, have high seawater iron concentrations (Boatta *et al.*, 2013) that can affect CaCO₃ production rate in *E. huxleyi*.

The seep region had elevated levels of some trace elements in the sediments (Vizzini et al., 2013). Malformed E. huxleyi specimens collected at high CO₂ levels had lith dissolution (Fig. 4) likely due to low carbonate saturation rather than to elevated levels of trace metals, but it would be of interest in future studies to determine whether ocean acidification can act synergistically with Fe or other metals to alter marine ecosystems, since many coastal regions have elevated trace metal concentrations.

Considerable spatial and temporal variations in ${\rm CO_2}$ levels around the seeps make it difficult to determine doseresponse relationships for pelagic organisms; this is why most work to date has focused on sessile benthic organisms

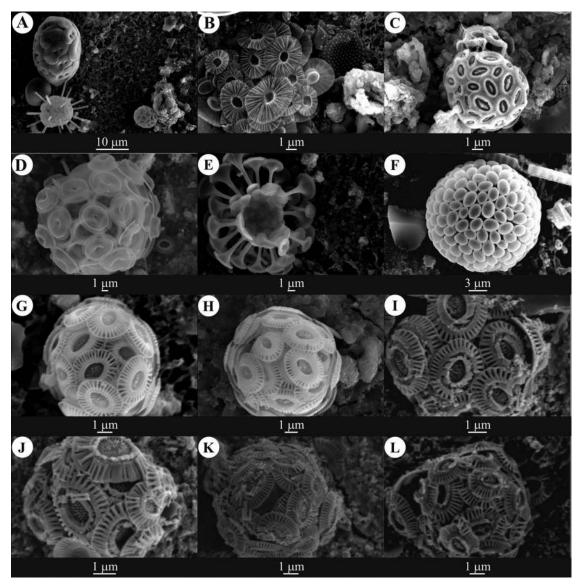


Figure 4. Scanning electron microscope images of coccolithophores collected in Levante Bay, November 2010. Coccospheres of (A) *Helicosphaera carteri*, *Rhabdosphaera clavigera*, and *Syracosphaera mollischii*; (B) *Umbellosphaera tenuis*; (C) *Syracosphaera mollischii*; (D) *Syracosphaera pulchra*; (E) *Discosphaera tubifera*; (F) *Syracosphaera pulchra* HOL *oblonga* (*Calyptrosphaera oblonga*); (G) and (H) Perfectly preserved *Emiliania huxley*; (I)–(L) Malformed specimens of *E. huxleyi* with dissolution in the central area.

rather than fish or plankton that can move in and out of the acidified areas (Fabricius *et al.*, 2011; Rodolfo-Metalpa *et al.*, 2011; Inoue *et al.*, 2013). For phytoplankton, levels of exposure to elevated CO₂ at volcanic seeps will depend upon wind-driven currents. We found that upper photic zone coccolithophore communities in southern Tyrrhenian Sea water and Levante Bay were very similar in areas unaffected by elevated CO₂. Future studies could combine *in situ* mesocosm-type experiments to constrain the movement of plankton at the seeps or use Free Ocean Carbon Experiments to control the CO₂ dose (Arnold *et al.*, 2012). Given the shallow nature of the seeps, such work could also

incorporate investigations of the possible effects of ultraviolet radiation.

Conclusion

Natural assemblages of living coccolithophores were investigated off Vulcano Island, southern Italy, along a pH and $\Omega_{\rm calcite}$ gradient formed by CO₂ seeps. Coccosphere concentrations decreased significantly as pH levels fell ($R^2 = 0.54$, n = 15). Species diversity also declined as pH fell, with the greatest number of malformed and corroded E. huxleyi coccoliths nearest to the seeps where $\Omega_{\rm calcite}$ was

Table 2. Coccolithophore species identified in the Vulcano sampling sites; pH and $\Omega_{calcine}$ are also indicated

Station	рΗ С	$\Sigma_{ m calcite}$	Emiliania huxleyi	Coronosphaera mediterranea	Syracosphaera mollischii	Rhabdosphaera clavigera	Umbellosphaera tenuis	Syracosphaera ossa	Syracosphaera tumularis	Discophaera tubifera	Helicosphaera carteri	Syracosphaera protrudens	Holococcolithophores spp.	Calcidiscus leptoporus	Syracosphaera noroitica	Syracosphaera pulchra	Syracosphaera rotula	Ceratholitus cristatus HET coccoithomorpha	Syracosphaera histrica	Gephyrocapsa ericsonii	Palusphaera vandeli	Syracosphaera prolongata	Cyrtosphaera lecaliae	Umbilicosphaera sibogae	Gephyrocapsa oceanica	Alisphaera unicornis	Acanthoica biscayensis	Syracosphaera ampliora	Anacanthoica acanthos
A inshore (25NOV10)	6.84	0.37	x				X		x				X	x	X														
A inshore (24NOV10)	6.94	0.47	X				X			X	X				X														
A inshore (23NOV10)	7.32	1.13	X	X	X	X	X	X		X	X								X										
C inshore (25NOV10)	7.56	1.86	X			X	X								X	X		X		X									
C inshore (24NOV10)	7.66	2.33	X						X	X						X			X										
A offshore (26NOV10)	8.00	4.63	X		X				X		X		X		X				X	X	X	X							
D inshore (25NOV10)	8.02	4.78	X		X	X	X					X	X		X	X	X			X			X						
B offshore (26NOV10)	8.02	4.8	X		X	X	X			X						X			X	X				X					
D inshore (24NOV10)	8.04	4.94	X		X		X			X		X			X		X											X	
C offshore (26NOV10)	8.07	5.26	X		X	X	X						X			X				X					X				
D offshore (26NOV10)	8.08	5.35	X		X	X	X	X		X	X		X		X					X						X	X		
C inshore (23NOV10)	8.16	6.19	X		X	X	X				X	X				X		X	X									X	X
D inshore (23NOV10)																													

lowest. These field observations show that ocean acidification was correlated with reduced coccolithophore diversity and disruptions in coccolith morphogenesis. Further work is required to assess effects on noncalcified phytoplankton and to check the bioavailability of metals such as iron that increased in concentration near to the CO₂ seeps. Our *in situ* observations add to concerns based on laboratory and mesocosm work that calcifying phytoplankton will be adversely affected by ocean acidification.

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Literature Cited

Anderson, M. J., and C. J. F. Ter Braak. 2003. Permutation tests for multi-factorial analysis of variance. J. Stat. Comput. Simul. 73: 85–113.

- Arnold, T., C. Mealey, H. Leahey, A. W. Miller, J. M. Hall-Spencer, M. Milazzo, and K. Maers. 2012. Ocean acidification and the loss of protective phenolics in seagrasses. *PLoS ONE* 7: e35107.
- Bach, L. T., C. Bauke, K. J. S. Meier, U. Riebesell, and K. G. Schulz. 2012. Influence of changing carbonate chemistry on morphology and weight of coccoliths formed by *Emiliania huxleyi*. *Biogeosciences* 9: 3449–3463.
- Beare, D. J., A. McQuatters-Gollup, T. van der Hammen, M. Machiels,
 S. J. Teoh, and J. M. Hall-Spencer. 2013. Long-term trends in calcifying plankton and pH in the North Sea. *PLoS ONE* 8: e61175.
- Beaufort, L., I. Probert, T. de Garidel-Thoron, E. M. Bendif, D. Ruiz-Pino, N. Metzl, C. Goyet, N. Buchet, P. Coupel, M. Grelaud, B. Rost, R. E. M. Rickaby, and C. de Vargas. 2011. Sensitivity of coccolithophores to carbonate chemistry and ocean acidification. *Nature* 476: 80-83.
- **Billard, C., and I. Inouye. 2004,** What is new in coccolithophore biology? Pp. 2–29 in *Coccolithophores: From Molecular Processes to Global Impact,* H. Thierstein and J. Young, eds. Springer Verlag, Berlin.
- Boatta, F., W. D'Alessandro, A. L. Gagliano, M. Liotta, M. Milazzo, R. Rodolfo-Metalpa, J. M. Hall-Spencer, and F. Parello. 2013. Geochemical survey of Levante Bay, Vulcano Island (Italy), a natural laboratory for the study of ocean acidification. *Mar. Pollut. Bull.* 73: 485–494.
- Bonomo, S., M. Grelaud, A. Incarbona, E. Malinverno, F. Placenti, A. Bonanno, E. Di Stefano, B. Patti, M. Sprovieri, S. Genovese, P.

Rumolo, S. Mazzola, S. Zgozi, and P. Ziveri. 2012. Living coccolithophores from the Gulf of Sirte (Southern Mediterranean Sea) during the summer of 2008. *Micropaleontology* **58:** 487–503.

- Bown, P. R. 1998. Calcareous Nannofossil Biostratigraphy. British Micropaleontological Society Publication Series, Kluwer Academic Publishers, Dordrecht.
- Caldeira, K., and M. E. Wickett. 2003. Oceanography: anthropogenic carbon and ocean pH. Nature 425: 365.
- Cigliano, M., M. C. Gambi, R. Rodolfo-Metalpa, F. P. Patti, and J. M. Hall-Spencer. 2010. Effects of ocean acidification on invertebrate settlement at volcanic CO₂ vents. *Mar. Biol.* 157: 2489–2502.
- Clarke, K. R., and R. N. Gorley. 2006. PRIMER v6: User Manual/ Tutorial. PRIMER-E, Plymouth, United Kingdom.
- Cros, L., and J. M. Fortuño. 2002. Atlas of northwestern Mediterranean coccolithophores. Sci. Mar. 66: 1–186.
- Dickson, A. G., C. L. Sabine, and J. R. Christian, Eds. 2007. Guide to Best Practices for Ocean CO₂ Measurements, PICES Special Publication 3 [Online]. Available: http://cdiac.ornl.gov/oceans/Handbook_2007.html [2014, June 9].
- Fabricius, K. E., C. Langdon, S. Uthicke, C. Umphrey, S. Noonan, G. De'ath, R. Okazaky, N. Muehllehner, M. S. Glas, and J. M. Lough.
 2011. Losers and winners in coral reefs acclimatized to elevated carbon dioxide concentrations. *Nat. Clim. Change* 1: 165–169.
- Hall-Spencer, J. M., R. Rodolfo-Metalpa, S. Martin, E. Ransome, M. Fine, S. M. Turner, S. J. Rawley, D. Tedesco and M.-C. Buia. 2008. Volcanic carbon dioxide vents show ecosystem effects of ocean acidification. *Nature* 454: 96–99.
- Horigome, M. T., P. Ziveri, M. Grelaud, K.-H. Baumann, G. Marino, and P. G. Mortyn. 2014. Environmental controls on the *Emiliania huxleyi* calcite mass. *Biogeosciences* 11: 2295–2308.
- Iglesias-Rodriguez, M. D., P. R. Halloran, R. E. M. Rickaby, I. R. Hall,
 E. Colmenero-Hidalgo, J. R. Gittins, D. H. R. Green, T. Tyrrell,
 S. J. Gibbs, P. von Dassow et al. 2008. Phytoplankton calcification in a high-CO₂ world. Science 320: 336–340.
- Inoue, S., H. Kayanne, S. Yamamoto, and H. Kurihara. 2013. Spatial community shift from hard to soft corals in acidified water. *Nat. Clim. Change* 3: 683–687.
- Johnson, V. R., C. Brownlee, R. E. M. Rickaby, M. Graziano, M. Milazzo, and J. M. Hall-Spencer. 2013. Responses of marine benthic microalgae to elevated CO₂. Mar. Biol. 160: 1813–1824.
- Jordan, R. W., L. Cros, and J. R. Young. 2004. A revised classification scheme for living haptophytes. *Micropaleontology* 50: 55-79.
- Knappertsbusch, M. 1993. Geographic distribution of living and Holocene coccolithophores in the Mediterranean Sea. *Mar. Micropaleontol.* 21: 219–247.
- Kottmeier, D. M., S. D. Rokitta, P. D. Tortell, and B. Rost. 2014. Strong shift from HCO₃⁻ to CO₂ uptake in *Emiliania huxleyi* with acidification: new approach unravels acclimation versus short-term pH effects. *Photosynth. Res.* doi: 10.1007/s11120-014-9984-9.
- Kroeker, K. J., R. L. Kordas, R. Crim, I. E. Hendriks, L. Ramajo, G. S. Singh, C. M. Duarte, and J.-P. Gattuso. 2013a. Impacts of ocean acidification on marine organisms: quantifying sensitivities and interaction with warming. *Glob. Change Biol.* 19: 1884–1896.
- Kroeker, K. J., M. C. Gambi, and F. Micheli. 2013b. Community dynamics and ecosystem simplification in a high-CO₂ ocean. PNAS 110: 12721–12726.
- Langer, G., G. Nehrke, I. Probert, J. Ly, and P. Ziveri. 2009. Strainspecific responses of *Emiliania huxleyi* to changing seawater carbonate chemistry. *Biogeosciences* 6: 2637–2646.
- Malinverno, E., P. Ziveri, and C. Corselli. 2003. Coccolithophorid distribution in the Ionian Sea and its relationship to eastern Mediter-

- ranean circulation during late fall to early winter 1997. *J. Geophys. Res.* **108:** doi:10.1029/2002JC001346.
- Malinverno, E., M. D. Dimiza, M. B. Triantaphyllou, M. D. Dermitzakis, and C. Corselli. 2008. Coccolithophores of the Eastern Mediterranean Sea. ION Publishing Group, Athens.
- Meier, K. J. S., L. Beaufort, S. Heussner, and P. Ziveri. 2014. The role of ocean acidification in *Emiliania huxleyi* coccolith thinning in the Mediterranean Sea. *Biogeosciences* 11: 2857–2869.
- Milazzo, M., R. Rodolfo-Metalpa, V. B. S. Chan, M. Fine, C. Alessi, V. Thiyagarajan, J. M. Hall-Spencer, and R. Chemello. 2014. Ocean acidification impairs vermetid reef recruitment. Sci. Rep. 4: 4189.
- Oviedo, A., P. Ziveri, M. Alvarez, and T. Tanhua. 2014. Is coccolithophore distribution in the Mediterranean Sea related to seawater carbonate chemistry? *Ocean Sci. Discuss.* 11: 613–653.
- Porzio, L., M.-C. Buia, and J. M. Hall-Spencer. 2011. Effects of ocean acidification on macroalgal communities. J. Exp. Mar. Biol. Ecol. 400: 278–287.
- Read, B. A., J. Kegel, M. J. Klute, A. Kuo, C. Lefebvre, F. Maumus, C. Mayer, J. Miller, A. Monier, A. Salamov et al. 2013. Pan genome of the phytoplankton *Emiliania* underpins its global distribtuion. *Nature* 499: 209–213.
- Riebesell, U., I. Zondervan, B. Rost, P. D. Tortell, R. E. Zeebe, and F. M. M. Morell. 2000. Reduced calcification of marine plankton in response to increased atmospheric CO₂. *Nature* 407: 364–366.
- Riebesell, U., R. G. J. Bellerby, H.-P. Grossart, and F. Thingstad. 2008.
 Mesocosm CO₂ perturbation studies: from organism to community level. *Biogeosciences* 5: 1157–1164.
- Rodolfo-Metalpa, R., F. Houlbrèque, E. Tambutté, F. Boisson, C. Baggini, F. P. Patti, R. Jeffree, M. Fine, A. Foggo, J.-P. Gattuso, and J. M. Hall-Spencer. 2011. Coral and mollusc resistance to ocean acidification adversely affected by warming. *Nat. Clim. Change* 1: 308–312.
- Smith, H. E. K., T. Tyrrell, A. Charalampopoulou, C. Dumousseaud,
 O. J. Legge, S. Birchenough, L. R. Pettit, R. Garley, S. E. Hartman,
 M. C. Hartman *et al.* 2009. Predominance of heavily calcified coccolithophores at low CaCO₃ saturation during winter in the Bay of Biscay. *PNAS* 109: 8845–8849.
- Tyrrell, T., B. Schneider, A. Charalampopoulou, and U. Riebesell. 2008. Coccolithophores and calcite saturation state in the Baltic and Black Seas. *Biogeosciences* 5: 485–494.
- Vizzini, S., R. Di Leonardo, V. Costa, C. D. Tramati, F. Luzzu, and A. Mazzola. 2013. Trace element bias in the use of CO₂ vents as analogues for low pH environments: implications for contamination levels in acidified oceans. *Estuar. Coast. Shelf Sci.* 134: 19–30.
- Wilson, J. D., S. Barker, and A. Ridgwell. 2012. Assessment of the spatial variability in particulate organic matter and mineral sinking fluxes in the ocean interior: implications for the ballast hypothesis. Global Biogeochem. Cycles 26: doi:10.1029/2012GB004398.
- Young, J. R., M. Geisen, L. Cros, A. Kleijne, C. Sprengel, I. Probert, and J. B. Østergaard. 2003. A guide to extant coccolithophore taxonomy. J. Nannoplankton Res. Special Issue 1: 1–121.
- Ziveri, P., R. Thunell, and D. Rio. 1995. Seasonal changes in coccolithophore densities in the Southern California Bight during the 1991/1992
 El Niño event. Deep-Sea Res. Part 1 Oceanogr. Res. Pap. 42: 1881–1903.
- Ziveri, P., B. De Bernardi, K.-H. Baumann, H. M. Stoll, and P. G. Mortyn. 2007. Sinking of coccolith carbonate and potential contribution to organic carbon ballasting in the deep ocean. *Deep-Sea Res. Part 2 Top. Stud. Oceanogr.* 54: 659–675.
- Zondervan, I. 2007. The effects of light, macronutrients, trace metals and CO₂ on the production of calcium carbonate and organic carbon in coccolithophores—a review. *Deep-Sea Res. Part 2 Top. Stud. Oceanogr.* 54: 521–537.