Desalination of Shale Gas Wastewater: Thermal and Membrane Applications **3** for Zero-Liquid Discharge Viviani C. Onishi^{*a*, *}, Eric S. Fraga^{*b*}, Juan A. Reyes-Labarta^{*a*}, José A. Caballero^{*a*} ^a Institute of Chemical Process Engineering, University of Alicante, Ap. Correos 99, Alicante 03080, Spain ^b Centre for Process Systems Engineering, Department of Chemical Engineering, University College London, London WC1E 7JE, UK * Corresponding author at. Institute of Chemical Process Engineering, University of Alicante, Ap. Correos 99, Alicante 03080, Spain. Phone: +34 965903400. E-mail addresses: viviani.onishi@ua.es / viviani.onishi@pq.cnpq.br (Viviani C. Onishi).

22 ABSTRACT

Natural gas exploration from unconventional shale formations, known as "shale gas", 23 has recently arisen as an appealing energy supply to meet the increasing worldwide 24 demand. During the last decade, development of horizontal drilling and hydraulic 25 fracturing ("fracking") technologies have allowed the cost-effective gas exploration from 26 previously inaccessible shale deposits. In spite of optimistic expansion projections, 27 28 natural gas production from tight shale formations has social and environmental implications mainly associated with the depletion of freshwater resources and polluting 29 wastewater generation. In this context, the capability of desalination technologies to allow 30 water recycling and/or water reuse is crucial for the shale gas industry. Advances in zero-31 liquid discharge (ZLD) desalination processes for treating hypersaline shale gas 32 33 wastewater, can play a key role in the mitigation of public health and environmental impacts, and improvement of overall process sustainability. This chapter outlines the 34 most promising thermal and membrane-based alternatives for ZLD desalination of shale 35 gas wastewater. 36

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Keywords: Shale gas wastewater; Zero-liquid discharge (ZLD); Thermal desalination;
Membrane desalination; Water treatment; Water reuse and water recycling.

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43 **1 Introduction**

Shale gas is currently the natural gas resource whose production exhibits the largest 44 45 worldwide growth. Especially in the last decade, technological developments in 46 horizontal drilling and hydraulic fracturing ("fracking") have boosted large-scale gas extraction from previously inaccessible unconventional shale reservoirs. Recent 47 48 projections from the U.S. Energy Information Administration (EIA) (1,2) draw attention to the global increase in natural gas exploitation from 342 in 2015 to 554 billion cubic 49 feet per day (Bcf day⁻¹) by 2040. The almost 62% rise in total natural gas production is 50 mainly due to the intensification in shale gas exploration. Actually, shale gas production 51 is expected to grow by more than 125 Bcf day⁻¹ over the forecast period, reaching 30% 52 53 of all natural gas produced in the world by 2040(1,2).

Along with the depletion of conventional natural gas reserves, supply reliability 54 and energy independence have emerged as driving forces for further development of shale 55 gas exploration (3). Notwithstanding, the latent advance of shale gas production around 56 the globe, notably in the United Kingdom, Argentina, Brazil, Australia, Algeria and 57 58 Poland, to name a few (4); has also prompted serious concerns about environmental and social implications associated with greenhouse gas (GHG) emissions (5,6), induced 59 seismic events (7), and quantity and quality of natural water resources and wastewater 60 61 discharges (8-11). Regarding water-related impacts alone, shale gas production from tight shale formations usually requires impressive freshwater volumes and generates large 62 63 amounts of polluting hypersaline wastewater. Consequently, water management is 64 nowadays one of the biggest challenges faced by the shale gas industry for maintaining process cost-effectiveness, while accounting for environmental and human health 65 protection (12,13). 66

67	Environmental, public health and socioeconomic risks can be significantly
68	reduced by adequate high-salinity wastewater treatment for allowing water reuse (i.e.,
69	water reinjection in new wells or existing ones), water recycling (i.e., water reuse in other
70	activities not related to hydraulic fracking operations) or safe disposal. Decreasing total
71	dissolved solids (TDS) is the key consideration to attain water quality required for internal
72	and/or external reuse or discharge (13). Within this framework, the application of
73	effective desalination technologies is imperative to enhance overall shale gas process
74	efficiency and sustainability (14,15). The main strength of desalination resides in its
75	ability to achieve salt concentrations that comply with strict regulations, promoting
76	cleaner shale gas production (16,17). In this chapter, the most promising thermal and
77	membrane-based desalination alternatives for shale gas wastewater management are
78	summarized and examined in detail. Energy and economic analyses of potential zero-
79	liquid discharge (ZLD) processes are presented as well, to evaluate the best desalination
80	options for more sustainable shale gas development.

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2 Water Consumption, Wastewater Generation and Management Options

85 2.1 Water Consumption in Shale Gas Operations

Contrarily to conventional natural gas production from geological formations such as porous sandstone and carbonate reservoirs, shale gas extraction is strongly impaired by the low shale rock permeability that compels the use of additional engineered solutions for attaining cost-effective production rates (9,18). Economically viable gas exploitation

from shale reservoirs is facilitated through the combined application of horizontal drilling 90 91 and fracking processes (19). These techniques together have allowed access to major 92 shale deposits and have improved permeability for releasing natural gas entrapped into tight rock formations (13). 93

In shale gas production, water-based fracturing fluid at very high pressure (about 94 480-680 bar) is injected in the shale well to unlock the existing fissures and create new 95 96 artificial fracture networks, increasing the contact surface between reservoir and wellbore (20,21). The chemical composition of the hydrofracturing fluid is conditioned by 97 geological shale formations and water supply features, as well as the gas extraction 98 99 operators (20,22). Recent reports suggest that horizontal drilling and well-completion technologies demand about 7,570–30,000 m^3 (~2–8 million U.S. gallons) of water per 100 101 well operation (23,24). The hydraulic fracturing process requires approximately 90% of 102 the total water amount, while the remaining $(\sim 10\%)$ is used for horizontal drilling (25). As a result of the exhaustive water consumption, progress in shale gas industry is greatly 103 104 restricted by water availability, particularly in water-stressed regions (26,27). In these 105 areas, the effects of water shortages can be controlled by enhancing water usage 106 efficiency in the shale gas process. The latter is achieved via more rigorous regulations 107 on water conservation and reuse and, finally, through the implementation of effective 108 desalination plants.

through 109

2.2 Wastewater Generation in Shale Gas Operations 110

Shale gas wastewater encompasses both flowback water and produced water (also 111 112 referred as formation water). Depending on the geologic setting and the well characteristics, U.S. shale basins exploration indicates that around 10-80% of the 113 injection fluid returns to the surface as flowback water within the first two weeks 114

115	following the hydraulic fracturing operation (23,28). Afterwards, with the beginning of
116	gas production, flowback water gradually decreases—usually, it remains in a range from
117	~210 to 420 U.S. gallons h^{-1} , as has been observed in prominent shale plays from North
118	America, including Marcellus, Fayetteville, Haynesville, and Barnett (29)-and high-
119	salinity produced water is recovered over the well lifetime ($\sim 20-40$ years) (30). Recently,
120	Kondash et al. (31) have estimated wastewater quantities ranging from 0.5 to 3.8 million
121	U.S. gallons per well over a period of 5-10 years of shale gas production. Among other
122	pollutants, the high-salinity nature (average values typically higher than 100,000 ppm
123	TDS) of shale gas wastewater is extremely hazardous to the environment and human
124	health (32), and demands energy-intensive desalination processes. Table 1 displays the
125	average water amounts required for horizontal drilling and fracking operations, and shale
126	gas wastewater data from important U.S. shale plays.

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[INSERT TABLE 1 HERE]

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130 2.3 Wastewater Management Options

Different management options are available for dealing with the wastewater from shale 131 132 gas operations. In the U.S., it is estimated that almost 95% of all wastewater generated in shale gas industry is currently disposed in Class II salt water disposal wells through deep 133 underground injection (22,33). Concerning the latter procedure, waste brine can be 134 135 released to the environment with or without water treatment (34). Although underground 136 injection is the preferred practice for managing wastewater due to its economic benefits, it has lately been associated with potential induced seismic activity, and groundwater and 137 soil contamination (7,33). Moreover, capacity of Class II disposal wells is becoming 138 progressively more limited and, consequently, it might not be able to accommodate all 139

produced shale gas wastewater (35). Besides water conservation policies and severe environmental regulations on discharges quality, disposal capacity constraints have also emphasized the importance of developing new alternatives for high-salinity wastewater desalination, mainly to allow its reuse or recycling (36). Figure 1 presents the main options for wastewater management in shale gas industry.

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[INSERT FIGURE 1 HERE]

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Reusing wastewater in hydraulic fracking operations, commonly classified as 148 149 "internal reuse" (13), is an economically advantageous management strategy to address 150 current concerns about the considerable freshwater consumption and wastewater 151 pollution risks. However, direct water reuse is unsuitable due to the high concentration of 152 contaminants that can compromise the well exploration (37). For this reason, onsite portable units for wastewater pretreatment-which comprises primary and secondary 153 treatment options such as filtration, physical and chemical precipitation, flotation, 154 155 sedimentation, and softening—are generally used to avoid operational problems (35).

156 Onsite treatment plants usually include established technologies to remove total 157 suspended solids (TSS), oil and greases and scaling materials (38). Typically, the onsite treated wastewater is blended with freshwater to reduce the high TDS contents (which 158 are responsible for negative viscosity effects on the hydraulic fluid), allowing its reuse in 159 160 hydraulic fracturing operations (13). Nevertheless, even if transportation costs are not considered in onsite plants, capacity and practical constraints alone restrict the application 161 162 of this treatment alternative (35). It is also worth noting that wastewater composition and water treatment technologies employed in the corresponding system are crucial to the 163

process cost-effectiveness. Moreover, internal reuse practice is dependent on the demandfor new well exploitation and ultimately, on the industry expansion.

166 With the maturity of shale gas industry, drilling and fracking operations will eventually decrease, transforming the activity in a potential wastewater producer. At this 167 point, the application of effective desalination processes will become inevitable (9,39). 168 169 In this context, centralized (offsite) plants for wastewater pretreatment followed by 170 effective desalination emerge as other options for water management. In fact, they are appealing alternatives to achieve high water quality, permitting its reuse for other 171 172 beneficial purposes-for instance, water recycling for agricultural activities (40)-or 173 even safe release to surface water bodies.

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3 Challenges of Shale Gas Wastewater Desalination

Shale gas wastewater produced by hydraulic fracturing operations present physical and 177 chemical properties that varies according to different factors, including formation 178 179 geology and geographic location, fracking fluid composition, and the water's time of 180 contact with shale deposits (13,41,42). Note that the fracturing fluid is a complex mixture, predominantly composed by proppant (sand suspension ~99.5% v/v) and chemical 181 182 enhancers that embrace surfactants, inorganic acids, biocides, friction reducers, scale and corrosion inhibitors, flow improvers, etc. (20,43,44). Furthermore, chemical contents in 183 shale gas wastewater may also vary throughout the time of well exploitation (13). 184

The selection of most suitable treatment alternatives is strongly influenced by the physicochemical composition of the wastewater (42). Apart from the chemical additives utilized within hydrofracturing fluids, shale gas wastewater is also composed by formation-based constituents, which comprises salt and other minerals (*i.e.*, scale-

189	forming ions: Ba ²⁺ , Ca ²⁺ and Mg ²⁺), organic matter and naturally occurring radioactive
190	materials (NORM) (45-48). Table 2 shows the typical composition ranges for critical
191	components in shale gas wastewater from Marcellus play.
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193	[INSERT TABLE 2 HERE]
194	
195	Among all contaminants, removal of the high TDS contents from shale gas
196	wastewater is especially challenging due to the intensive energy consumption needed to
197	accomplish with the severe regulations on water quality (particularly on water recycling
198	and safe disposal). Additionally, besides the variations in wastewater compositions
199	throughout the well lifetime, another complicating factor is associated with the
200	considerable differences observed in wastewater from distinct shale basins, and even in
201	different wellbores from the same well pad (see Table 1) (30). Figure 2 displays
202	conceptual profiles for TDS concentration and wastewater flowrate after hydraulic
203	fracturing operations.
204	
205	[INSERT FIGURE 2 HERE]
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207	Hayes and Severin (37) have showed TDS contents in wastewater samples from
208	Barnett shale play ranging from 5,850–31,400 ppm (average value of 25,050 ppm) in day
209	1 following hydraulic fracturing; and, values between 16,400–97,800 ppm (average value
210	of 50,550 ppm) for 10–12 th days from the beginning of well exploration. As reported by
211	Acharya et al. (49), TDS concentrations in shale gas wastewater can widely vary from
212	average values of 13,000 ppm for Fayetteville shale play (maximum value of 20,000
213	ppm), to 120,000 ppm for Marcellus shale play (maximum value >280,000 ppm TDS).

Also, chemical composition analyses performed by Thiel and Lienhard (30) have indicated TDS amounts in wastewater from Permian and Marcellus basins ranging from 120,000 ppm to approximately 250,000 ppm. Results presented by Barbot et al. (20) reveal even higher maximum TDS concentrations of 345,000 ppm (data from Northeast Pennsylvania basins).

219 Several desalination processes can be applied to treat the hypersaline shale gas 220 wastewater, for ensuring the strict composition constraints in accordance with specific 221 wastewater-desired destinations (*i.e.*, water reuse, water recycling or disposal). Desalination technologies include thermal and membrane-based desalination processes. 222 223 Thermal technologies comprise multistage flash distillation (MSF), multi-effect 224 distillation (MED), and single or multiple-effect evaporation (SEE/MEE) systems, which 225 can be coupled to mechanical or thermal vapor compression (MVC/TVC); the membrane-226 based group includes processes such as membrane distillation (MD), forward osmosis (FO), reverse osmosis (RO) and electrodialysis (ED). Figure 3 displays the schematic 227 228 representation of main thermal and membrane-based processes for shale gas wastewater 229 desalination.

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[INSERT FIGURE 3 HERE]

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High TDS contents in shale gas wastewater pose specific desalination challenges, mostly related to high energy consumption and operational problems produced by scaling, fouling and corrosion (50,51). Actually, deposition of scale forming ions on the equipment surface can compromise the system energy performance of both thermal and membrane-based technologies. Due to changes in process conditions (*i.e.*, composition, pH and temperature) during desalination, fouling and scaling surface-growth phenomena can reduce heat transfer in thermal evaporation technologies and mass transfer in
membrane-based systems (51). In the last case, the presence of scaling compounds in the
wastewater can severely decrease permeate flux across the membrane (52,53).

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4 ZLD Desalination for Wastewater Management

245 4.1 Drivers and Benefits of ZLD Systems

In recent years, ZLD desalination has attracted increased interest by the scientific community and industry as a strategy for wastewater management. This is mainly due to its ability to enhance water usage efficiency, while reducing brine discharges and water and disposal-related environmental impacts (54,55). From general efficiency and environmental protection viewpoints, the ambitious goal of zero-emission desalination could be a game changer for the entire shale gas industry.

252 ZLD desalination systems are high-recovery processes that allow the production of high-quality treated water (*i.e.*, freshwater) and concentrate brine, by decreasing liquid 253 254 contents present in the brine waste (56). Here, brine discharges salinity near to salt 255 saturation conditions is considered as ZLD operation. Thus, ZLD alternatives are usually 256 operated to recover $\sim 75-90\%$ of the total amount of water from the wastewater. The remaining water contents can be eliminated by including brine crystallizers or 257 258 evaporation ponds into the system. Consequently, almost the water totality in the wastewater can be reclaimed for internal reuse in shale gas operations. In this way, ZLD 259 260 desalination enhances water sustainability and diminishes the environmental pollution 261 and social risks related to wastewater and brine disposals, as well as depletion of 262 freshwater resources (14,54).

Although widely recognized as an important approach for reducing water impacts 263 264 and improving water supply sources, the implementation of ZLD desalination systems is still limited by intensive energy consumption and high associated processing costs 265 266 (54,57). However, recent studies have demonstrated the economic viability of thermalbased ZLD desalination systems applied to shale gas wastewater treatment (3,14,15,17). 267 268 In Onishi et al. (15), for instance, electric-driven SEE/MEE-MVC technologies for ZLD desalination (by considering brine discharges at 300,000 ppm or 300 g kg⁻¹) have 269 presented specific energy consumption in a range of 28.12-50.47 kWh_e m⁻³, with 270 operational expenses estimated between 2.73-4.90 US\$ m⁻³ for 77% conversion ratio 271 (freshwater production ratio at 7.99 kg s⁻¹). Also, the authors have shown freshwater 272 production costs ranging from 6.7 US\$ m⁻³ (MEE-MVC with thermal integration) to 10.9 273 US\$ m⁻³ (SEE-MVC with thermal integration). It should be noted that disposal costs in 274 275 Class II saline water injection wells (i.e., conventional deep-well injection) are estimated to be in the range of ~8-25 US\$ m⁻³ (~0.03-0.08 US\$ gallon⁻¹)—water disposal cost for 276 277 locally available injection sites in Barnett shale play-(49). These results emphasize the 278 need for developing more realistic energy performance and cost analysis for ZLD desalination systems, to evaluate the best trade-off between their benefits, energy 279 280 consumption and capital and operating costs.

Future progress in ZLD applications to shale gas wastewater will ultimately be achieved by stricter regulations on water quality and brine discharges, as well as by incrementing regulatory incentives to compensate eventual economic shortcomings (54). These factors, allied to the rising in wastewater disposal costs, will drive shale gas industry towards the implementation of cleaner ZLD desalination systems. **Table 3** and **Table 4** present the freshwater production cost and energy consumption of promising thermal and membrane-based ZLD desalination technologies for shale gas wastewater.

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[INSERT TABLE 3 HERE]

[INSERT TABLE 4 HERE]

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292 4.2 Environmental Impacts

Since both thermal and electric power used in desalination systems are usually produced 293 from fossil fuel energy sources, the elevated energy consumption related to ZLD systems 294 295 is also responsible for significant pollutant emissions to the atmosphere. These emissions 296 are predominantly composed by GHG (Carbon dioxide), acid rain gases (Nitrogen oxides and Sulphur dioxide) and fine particulate matter (58). According to EIA (59), around 939 297 298 g of CO₂ per kWh_e are generated from burning coal. Under the latter consideration, MEE-299 MVC systems operating at ZLD conditions will produce approximately 26.4-47.4 kg of CO2 per cubic meter of treated water-considering an energy consumption in a range of 300 28.12–50.47 kWhe m⁻³ (15) —. Carbon footprint and other air pollutant releases directly 301 (e.g., thermal sources as steam) or indirectly (e.g., energy from electricity grids) 302 303 associated with ZLD schemes can be mitigated by developing higher energy efficiency 304 technologies, and incorporating renewable (*e.g.*, solar, wind and geothermal energy) and/or low-grade energy sources (17,54). 305

Additional polluting risks linked to ZLD systems are connected to brine waste production. Concentrate management strategies can include brine disposal in landfills and evaporation ponds. Apart from soil contamination possibility, the deposition of solid wastes in landfills can also compromise groundwater by leaching chemicals through the soil matrix. Likewise, brine storage in evaporation ponds can cause environmental and social impacts, due to leakage risks, odors generation and wildlife depletion (60). These negative effects on water and soil and their consequences can be prevented by the implementation of reliable monitoring systems, as well as the use of impermeable liningsto isolate surface zones (54).

315 Major thermal and membrane-based process for ZLD desalination of shale gas316 wastewater are presented in the following sections.

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319 5 ZLD Desalination Technologies for Shale Gas Wastewater

321 Desalination systems for the ZLD treatment of high-salinity shale gas wastewater can comprise thermal and membrane-based technologies such as SEE/MEE (with MVC or 322 TVC), MD, FO and RO (see Figure 3). As described before, these technologies are able 323 to produce high-quality water by accomplishing with the severe regulations on salt 324 contents required for recycling opportunities (e.g., irrigation, livestock watering or 325 326 industrial uses). In addition, their modular feature and simple scale-up are propitious for 327 the implementation of onsite treatment plants at shale plays constrained by infrastructure limitations (13). Thermal-based evaporation systems coupled to MVC are comparatively 328 329 well-established processes, whereas MD, FO, RO and ED are promising technologies for high-salinity shale gas wastewater applications. Table 5 shows the main advantages and 330 limitations of thermal and membrane-based ZLD desalination processes. 331

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[INSERT TABLE 5 HERE]

335 5.1 Thermal-based ZLD Processes

336 **ZLD** evaporation systems

337 Despite the significant research efforts on the development of thermal-based MSF and MED processes for seawater desalination (61–63), their application in ZLD systems for 338 339 shale gas wastewater has not been reported in the literature to date. In general, thermal 340 evaporation systems with MVC can be more advantageous than membrane technologies for shale gas wastewater treatment (13). Due to lower susceptibility to rusting and fouling 341 342 problems, MEE-MVC demands lesser energy-intensive pretreatment processes than those 343 required prior to membrane desalination. Furthermore, thermal systems are generally more robust and require lower cleaning frequency and intensity than membrane ones (64). 344 345 On the other hand, while low-grade thermal energy can be used in membrane systems 346 (65,66), typical thermal evaporation schemes with MVC are driven by high-grade 347 electrical energy. Besides the related high operating costs and GHG emissions, this is also 348 a barrier for their operation in remote areas without easy access to power grids. To surpass 349 these limitations, geothermal or other renewable energy sources can be incorporated into the thermal systems. 350

351 ZLD thermal evaporation systems for the desalination of hypersaline shale gas 352 wastewater have been addressed by Onishi et al. (3,14-17). In Onishi et al. (15), the 353 authors have developed a mathematical optimization model for SEE/MEE systems 354 design, considering single and multistage MVC and heat integration. Figure 4 displays 355 the MEE-MVC system proposed by Onishi et al. (15) for the ZLD desalination of shale 356 gas wastewater. Their modelling approach is aimed at enhancing process energy efficiency, while reducing polluting brine discharges. The authors have performed a 357 thorough comparison between the optimal systems configurations found (SEE/MEE with 358

359	single or multistage mechanical compression) under a wide range of inlet wastewater
360	salinities (10,000-220,000 ppm TDS), to evaluate their ability to achieve high water
361	recovery ratios and ZLD operation. Energy and economic analyses have revealed the
362	MEE process with single-stage MVC as the most cost-effective system for treatment of
363	shale gas wastewater. Further information on ZLD desalination process of shale gas
364	wastewater via SEE/MEE-MVR systems can be found in references (14,15).
365	
366	[INSERT FIGURE 4 HERE]
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368	Based on the latter result, Onishi et al. (14) have proposed a new rigorous
369	optimization approach for MEE-MVC systems design, by considering more precise
370	estimation of the global heat transfer coefficient to minimize process costs. Furthermore,
371	their method considers the modelling of major equipment features, including optimal
372	number and length of tubes, and evaporator diameter. Their results indicate that the MEE-
373	MVC system can be almost 35% less expensive than SEE-MVC for recovering 76.7% of
374	freshwater (brine discharge salinity at 300,000 ppm TDS). Afterwards, Onishi et al. (3)
375	have focused on the high uncertainty related to well data (wastewater flowrates and
376	salinities) from shale plays to support decision-makers in the implementation of more
377	robust MEE-MVC systems. Distributions of energy consumption and operating expenses
378	throughout different feeding scenarios are shown in Figure 5.
379	
380	[INSERT FIGURE 5 HERE]
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382	Lastly, Onishi et al. (17) have developed a mathematical modelling approach for

the optimization of solar energy-driven MEE-MVC systems. The authors have considered

an integrated process composed by a solar assisted Rankine cycle and a MEE-MVC 384 385 desalination plant. The multi-objective optimization model allows to minimize 386 environmental impacts, and investment and operating costs. Their trade-off Paretooptimal solutions (especially intermediate solutions containing hybrid solar and 387 388 electricity energy sources) reveal that renewable energy co-generation in desalination ZLD plants can promote significant environmental and cost savings for shale gas 389 390 industry. Figure 6 presents the zero-discharge MEE-MVC system driven by solar energy proposed by Onishi et al. (17) for the desalination of high-salinity shale gas wastewater. 391

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[INSERT FIGURE 6 HERE]

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395 Crystallizers

Solid waste produced by thermal evaporation systems can be further concentrated in brine 396 397 crystallizers. In this case, all remaining water can be recovered from waste brine. Analogously to SEE/MEE-MVC concentrators, electrically driven mechanical 398 399 compressors are used in large-scale crystallizers (*i.e.*, for treating brine flows higher than 400 6 gallons per minute) to superheat vapor and supply heat required for driving the 401 evaporation process. For lower brine flows ranging 2–6 gallons per minute, steam-driven crystallizers are generally more economical (67). While horizontal-tube falling film 402 403 evaporators are preferred in SEE/MEE-MVC schemes, thermal evaporative crystallizers 404 are generally operated thru forced-circulation. Crystallization of concentrate brines is an 405 energy intensive process, which usually demands a range of 52–66 kWhe per cubic meter 406 of treated water (54,60). This is mainly due to the higher salt concentration and viscosity 407 that characterize brine wastes. However, crystallizer technology can be especially 408 appropriate for shale gas exploration areas in which deep-well injection is not allowed or 409 costly, the solar irradiance is low, or cost of evaporation ponds construction is excessively410 high (68).

411

412 Evaporation Ponds

413 Evaporation ponds are competitive disposal alternatives to thermal brine crystallizers. This technology uses natural solar irradiance to drive the evaporation process and 414 415 eliminate the water contents from brine waste. Although the operational expenses are low, 416 evaporations ponds implementation is constrained by its high capital investment and 417 environmental concerns related to brine waste leakage risks (54). Additionally, since the 418 process allows to recover only solid wastes, water cannot be reclaimed for recycling or reuse in shale gas operations. As a consequence, water usage efficiency in shale gas 419 420 industry cannot be improved by evaporation ponds. Also, evaporation ponds coupled to 421 ZLD desalination systems should be designed to ensure the deposition of all precipitated 422 solids over the zero-discharge plant lifetime, or even the construction of new ponds (67). 423 Figure 7 depicts the schematic representation of a thermal-based ZLD evaporation plant 424 coupled to the pretreatment system and crystallization or evaporation ponds.

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[INSERT FIGURE 7 HERE]

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428 5.2 Membrane-based ZLD Processes

429 Membrane-based technologies have recently arisen as promising alternatives for ZLD 430 desalination of high-salinity wastewater from shale gas production. Membrane systems 431 usually present great potential for shale gas wastewater applications due to their high 432 efficiency, operational and control simplicity, elevated permeability and selectivity for

some critical components, simple scale-up and possibility of using low-grade waste 433 434 energy (69,70). Table 6 presents process characteristics and applications of major 435 membrane-based systems for ZLD desalination of shale gas wastewater. 436 437 [INSERT TABLE 6 HERE] 438 439 Analogously to MVC concentrators, membrane-based technologies are able to achieve (near-)ZLD conditions with brine discharge salinity higher than 300,000 ppm or 440 441 30% weight-to-volume fraction (w/v) (65,71,72). Note that, although these systems can 442 theoretically concentrate the feed stream until the salt saturation conditions (~350,000 443 ppm or 35% w/v), near-ZLD operation is preferable to prevent operational difficulties related to salt crystallizing in the system (66)-In this case, crystallizer units or 444 445 evaporation ponds can be considered to recover the remaining water and valuable byproducts (54)-... Also, recent studies indicate that the energy requirements and 446 447 associated capital and operating costs of membrane technologies are competitive when compared to more conventional thermal ZLD desalination systems and disposal 448

450 broad application of membrane-based schemes in shale gas industry (64).

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453 6 Outlook and Future Directions

454 Shale gas industry is responsible for elevated freshwater consumption and generation of 455 large amounts of hazardous wastewater, which is comprised by flowback and produced 456 waters. Developing more effective desalination systems for the treatment of high-salinity 457 wastewater to allow its reuse and/or recycling is critical to alleviate environmental and

alternatives (71,72). However, the elevated pretreatment costs are still an obstacle for the

public health impacts, and enhance the overall sustainability of shale gas process. Among
all pollutants in shale gas wastewater, removal of high TDS contents (usually >100,000
ppm) is particularly challenging due to the intensive energy consumption needed to
comply with strict regulations on water quality (especially on water recycling and safe
disposal).

ZLD desalination systems have recently emerged as an interesting alternative for shale gas wastewater management. The main advantages of ZLD processes relies in their ability to enhance water usage efficiency in shale gas production, while reducing brine discharges and water-related environmental impacts. As ZLD desalination systems are typically able to achieve water recovery ratios up to 90% (note that the remaining water contents can be eliminated by crystallizers or evaporation ponds), almost the totality of water from wastewater can be reclaimed for internal reuse or recycling opportunities.

Several desalination technologies can be used in ZLD systems for high-salinity wastewater application, including thermal and membrane-based processes. While thermal evaporation systems with MVC are relatively well-established processes, membranebased schemes containing MD, FO, RO and ED/EDR technologies are promising desalination systems for high-salinity shale gas wastewater. In general, membrane desalination systems present high efficiency, operational and control simplicity, easy scale-up and possibility of using low-grade waste energy.

Although widely accepted as an important wastewater management option to reduce water-related impacts, the implementation of ZLD systems in shale gas industry is still constrained by high energy demands and associated processing costs. Nevertheless, a critical review of literature has revealed the cost-competitiveness of ZLD thermal evaporation systems for shale gas wastewater desalination. Advances in membrane materials, fouling control and optimization of operating conditions should increase the

- 483 application of membrane-based ZLD systems in the shale gas desalination market. More
- 484 generally, the wide employment of ZLD systems depends on further development of
- 485 effective and sustainable desalination technologies, regulatory incentives to compensate
- 486 economic limitations, and stricter regulations on brine discharges and water quality.

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495 Nomenclature

496	AGMD	Air Gap Membrane Distillation
497	DCMD	Direct Contact Membrane Distillation
498	EC	Evaporative Crystallization
499	ED	Electrodialysis
500	EDR	Electrodialysis Reversal
501	EIA	Energy Information Administration
502	FO	Forward Osmosis
503	GHG	Greenhouse Gas
504	MEE	Multiple-Effect Evaporation
505	MED	Multi-Effect Distillation
506	MD	Membrane Distillation
507	MSF	Multistage Flash Distillation
508	MVC	Mechanical Vapor Compression
509	NF	Nanofiltration
510	NORM	Naturally Occurring Radioactive Materials
511	RO	Reverse Osmosis
512	SEE	Single-Effect Evaporation
513	TDS	Total Dissolved Solids
514	TOC	Total Organic Carbon
515	TSS	Total Suspended Solids
516	TVC	Thermal Vapor Compression
517	VMD	Vacuum Membrane Distillation
518	ZLD	Zero-Liquid Discharge

520 **References**

- EIA. Annual Energy Outlook 2016 with Projections to 2040. Washington, DC:
 U.S. Energy Information Administration, 2016.
- 523 2. EIA. International Energy Outlook 2016. Washington, DC: U.S. Energy
 524 Information Administration, 2016.
- Onishi VC, Ruiz-Femenia R, Salcedo-Díaz R, Carrero-Parreño A, Reyes-Labarta
 JA, Fraga ES, et al. Process optimization for zero-liquid discharge desalination of
 shale gas flowback water under uncertainty. J Clean Prod. 2017 Oct;164:1219–38.
- 528 4. Cooper J, Stamford L, Azapagic A. Shale gas: A review of the economic,
- 529 Environmental, and Social Sustainability. Energy Technol. 2016 Jul;4(7):772–92.
- 5. Staddon PL, Depledge MH. Fracking cannot be reconciled with climate change
 citigation policies. Environ Sci Technol. 2015 Jul 21;49(14):8269–70.
- 532 6. Stephenson T, Valle JE, Riera-Palou X. Modeling the relative GHG emissions of
 533 conventional and shale gas production. Environ Sci Technol. 2011 Dec
 534 15;45(24):10757–64.
- 535 7. NRC. Induced Seismicity Potential in Energy Technologies. Washington, D.C.:
 536 National Academies Press; 2013. 300 p.
- 537 8. Thomas M, Partridge T, Harthorn BH, Pidgeon N. Deliberating the perceived risks,
 538 benefits, and societal implications of shale gas and oil extraction by hydraulic
 539 fracturing in the US and UK. Nat Energy. 2017 Apr 10;2(5):17054.
- 540 9. Vidic RD, Brantley SL, Vandenbossche JM, Yoxtheimer D, Abad JD. Impact of
 541 shale gas development on regional water quality. Science (80). 2013 May
 542 17;340(6134):1235009–1235009.
- 543 10. Prpich G, Coulon F, Anthony EJ. Review of the scientific evidence to support
 544 environmental risk assessment of shale gas development in the UK. Sci Total

545 Environ. 2016 Sep;563–564:731–40.

- 546 11. Warner NR, Kresse TM, Hays PD, Down A, Karr JD, Jackson RB, et al.
 547 Geochemical and isotopic variations in shallow groundwater in areas of the
 548 Fayetteville Shale development, north-central Arkansas. Appl Geochemistry. 2013
 549 Aug;35:207–20.
- Kausley SB, Malhotra CP, Pandit AB. Treatment and reuse of shale gas
 wastewater: Electrocoagulation system for enhanced removal of organic
 contamination and scale causing divalent cations. J Water Process Eng. 2017
 Apr;16:149–62.
- Shaffer DL, Arias Chavez LH, Ben-Sasson M, Romero-Vargas Castrillón S, Yip
 NY, Elimelech M. Desalination and reuse of high-salinity shale gas produced
 water: Drivers, technologies, and future directions. Environ Sci Technol. 2013 Sep
 3;47(17):9569–83.
- 558 14. Onishi VC, Carrero-Parreño A, Reyes-Labarta JA, Fraga ES, Caballero JA.
 559 Desalination of shale gas produced water: A rigorous design approach for zero560 liquid discharge evaporation systems. J Clean Prod. 2017 Jan;140:1399–414.
- 561 15. Onishi VC, Carrero-Parreño A, Reyes-Labarta JA, Ruiz-Femenia R, Salcedo-Díaz
 562 R, Fraga ES, et al. Shale gas flowback water desalination: Single vs multiple-effect
 563 evaporation with vapor recompression cycle and thermal integration. Desalination.
 564 2017 Feb;404(C):230–48.
- 565 16. Onishi VC, Ruiz-Femenia R, Salcedo-Díaz R, Carrero-Parreño A, Reyes-Labarta
 566 JA, Caballero JA. Optimal Shale Gas Flowback Water Desalination under
 567 Correlated Data Uncertainty. In: Computer Aided Chemical Engineering. 2017. p.
 568 943–48.
- 569 17. Onishi VC, Ruiz-femenia R, Salcedo-díaz R, Carrero-Parreño A, Reyes-Labarta

570		JA, Caballero JA. Multi-Objective Optimization of Renewable Energy-Driven
571		Desalination Systems. In: Computer Aided Chemical Engineering. 2017. p. 499-
572		504.
573	18.	Clark CE, Horner RM, Harto CB. Life cycle water consumption for shale gas and
574		conventional natural gas. Environ Sci Technol. 2013 Oct 15;47(20):11829-36.
575	19.	Bilgen S, Sarıkaya İ. New horizon in energy: Shale gas. J Nat Gas Sci Eng. 2016
576		Sep;35(A):637–45.
577	20.	Barbot E, Vidic NS, Gregory KB, Vidic RD. Spatial and temporal correlation of
578		water quality parameters of produced waters from devonian-age shale following
579		hydraulic fracturing. Environ Sci Technol. 2013 Mar 19;47(6):2562–9.
580	21.	Ghanbari E, Dehghanpour H. The fate of fracturing water: A field and simulation
581		study. Fuel. 2016 Jan;163:282–94.
582	22.	Rosenblum J, Nelson AW, Ruyle B, Schultz MK, Ryan JN, Linden KG. Temporal
583		characterization of flowback and produced water quality from a hydraulically
584		fractured oil and gas well. Sci Total Environ. 2017 Oct;596–597:369–77.
585	23.	Hammond GP, O'Grady Á. Indicative energy technology assessment of UK shale
586		gas extraction. Appl Energy. 2017 Jan;185:1907–18.
587	24.	Zammerilli A., Murray RC., Davis T., Littlefield J. Environmental impacts of
588		unconventional natural gas development and production. 2014;800:553-7681.
589	25.	Yang L, Grossmann IE, Manno J. Optimization models for shale gas water
590		management. AIChE J. 2014 Oct;60(10):3490-501.
591	26.	Wan Z, Huang T, Craig B. Barriers to the development of China's shale gas
592		industry. J Clean Prod. 2014 Dec;84:818–23.
593	27.	Nicot J-P, Scanlon BR. Water use for shale-gas production in Texas, U.S. Environ
594		Sci Technol. 2012 Mar 20;46(6):3580–6.

- Slutz, J.; Anderson, J.; Broderick, R.; Horner P. Key shale gas water management
 strategies: an economic assessment tool. SPE/ APPEA Int Conf Heal Safety,
 Environ Oil Gas Explor Prod Perth, Aust Sept 2012.
- 598 29. Takahashi S, Kovscek AR. Spontaneous countercurrent imbibition and forced
 599 displacement characteristics of low-permeability, siliceous shale rocks. J Pet Sci
 600 Eng. 2010 Mar;71(1–2):47–55.
- 30. Thiel GP, Lienhard JH. Treating produced water from hydraulic fracturing:
 Composition effects on scale formation and desalination system selection.
 Desalination. 2014 Aug;346:54–69.
- Kondash AJ, Albright E, Vengosh A. Quantity of flowback and produced waters
 from unconventional oil and gas exploration. Sci Total Environ. 2017
 Jan;574:314–21.
- Vengosh A, Jackson RB, Warner N, Darrah TH, Kondash A. A critical review of
 the risks to water resources from unconventional shale gas development and
 hydraulic fracturing in the United States. Environ Sci Technol. 2014 Aug
 5;48(15):8334–48.
- 33. Jang E, Jeong S, Chung E. Application of three different water treatment
 technologies to shale gas produced water. Geosystem Eng. 2017 Mar 4;20(2):104–
 10.
- 814 34. Rahm BG, Bates JT, Bertoia LR, Galford AE, Yoxtheimer DA, Riha SJ.
 815 Wastewater management and Marcellus Shale gas development: Trends, drivers,
 816 and planning implications. J Environ Manage. 2013 May;120:105–13.
- Gao J, You F. Design and optimization of shale gas energy systems: Overview,
 research challenges, and future directions. Comput Chem Eng. 2017 Jan;1–20.
- 619 36. Zendehboudi S, Bahadori A. Shale Oil and Gas. In: Shale Oil and Gas Handbook.

- 620 Elsevier; 2017. p. 357–404.
- 37. Hayes T, Severin BF. Barnett and Appalachian Shale Water Management and
 Reuse Technologies RPSEA Report No 08122-05. In: Project report by Gas
 Technology Institute for Research Partnership to Secure Energy for America
 (RPSEA). 2012. p. 1–125.
- 625 38. Carrero-Parreño A, Onishi VC, Salcedo-Díaz R, Ruiz-Femenia R, Fraga ES,
 626 Caballero JA, et al. Optimal pretreatment system of flowback water from shale gas
 627 production. Ind Eng Chem Res. 2017 Apr 19;56(15):4386–98.
- 628 39. Estrada JM, Bhamidimarri R. A review of the issues and treatment options for
 629 wastewater from shale gas extraction by hydraulic fracturing. Fuel. 2016
 630 Oct;182:292–303.
- 40. Camarillo MK, Domen JK, Stringfellow WT. Physical-chemical evaluation of
 hydraulic fracturing chemicals in the context of produced water treatment. J
 Environ Manage. 2016 Dec;183:164–74.
- GWPC. Ground Water Protection Council, ALL Consulting. Modern Shale Gas
 Development in the United States: A Primer; United States Department of Energy,
 Office of Fossil Energy: Washington, D.C., 2009; p 96.
- 42. Lester Y, Ferrer I, Thurman EM, Sitterley KA, Korak JA, Aiken G, et al.
 Characterization of hydraulic fracturing flowback water in Colorado: Implications
 for water treatment. Sci Total Environ. 2015 Apr;512–513:637–44.
- 640 43. Stringfellow WT, Domen JK, Camarillo MK, Sandelin WL, Borglin S. Physical,
- 641 chemical, and biological characteristics of compounds used in hydraulic fracturing.
- 642 J Hazard Mater. 2014 Jun;275:37–54.
- 643 44. Gregory KB, Vidic RD, Dzombak DA. Water management challenges associated644 with the production of shale gas by hydraulic fracturing. Elements. 2011 Jun

- 645 1;7(3):181–6.
- 45. Vengosh A, Warner N, Jackson R, Darrah T. The effects of shale gas exploration
 and hydraulic fracturing on the quality of water resources in the United States.
 Procedia Earth Planet Sci. 2013;7:863–6.
- 649 46. Dale AT, Khanna V, Vidic RD, Bilec MM. Process based life-cycle assessment of
 650 natural gas from the Marcellus shale. Environ Sci Technol. 2013 May
 651 21;47(10):5459–66.
- 47. Zhang T, Gregory K, Hammack RW, Vidic RD. Co-precipitation of Radium with
 Barium and Strontium sulfate and its impact on the fate of Radium during treatment
 of produced water from unconventional gas extraction. Environ Sci Technol. 2014
 Apr 15;48(8):4596–603.
- 48. Rahm BG, Riha SJ. Toward strategic management of shale gas development:
 Regional, collective impacts on water resources. Environ Sci Policy. 2012
 Mar;17:12–23.
- 49. Acharya HR, Henderson C, Matis H, Kommepalli H, Moore B, Wang H. Cost
 effective recovery of low-TDS frac flowback water for re-use. Glob Res.
 2011;(June):1–100.
- 50. Xiong B, Zydney AL, Kumar M. Fouling of microfiltration membranes by
 flowback and produced waters from the Marcellus shale gas play. Water Res. 2016
 Aug;99:162–70.
- 51. Kaplan R, Mamrosh D, Salih HH, Dastgheib SA. Assessment of desalination
 technologies for treatment of a highly saline brine from a potential CO2 storage
 site. Desalination. 2017 Feb;404:87–101.
- 668 52. Lee S, Kim YC. Calcium carbonate scaling by reverse draw solute diffusion in a
 669 forward osmosis membrane for shale gas wastewater treatment. J Memb Sci. 2017

670		Jan;522:257–66.
671	53.	Miller DJ, Huang X, Li H, Kasemset S, Lee A, Agnihotri D, et al. Fouling-resistant
672		membranes for the treatment of flowback water from hydraulic shale fracturing: A
673		pilot study. J Memb Sci. 2013 Jun;437:265–75.
674	54.	Tong T, Elimelech M. The global rise of zero liquid discharge for wastewater
675		management: drivers, technologies, and future directions. Environ Sci Technol.
676		2016 Jul 5;50(13):6846–55.
677	55.	Loganathan K, Chelme-Ayala P, Gamal El-Din M. Treatment of basal water using
678		a hybrid electrodialysis reversal-reverse osmosis system combined with a low-
679		temperature crystallizer for near-zero liquid discharge. Desalination. 2015
680		May;363:92–8.
681	56.	López DE, Trembly JP. Desalination of hypersaline brines with joule-heating and
682		chemical pre-treatment: Conceptual design and economics. Desalination. 2017
683		Aug;415:49–57.
684	57.	Xu P, Cath TY, Robertson AP, Reinhard M, Leckie JO, Drewes JE. Critical review
685		of desalination concentrate management, treatment and beneficial use. Environ
686		Eng Sci. 2013 Aug;30(8):502–14.
687	58.	Ghaffour N, Lattemann S, Missimer T, Ng KC, Sinha S, Amy G. Renewable
688		energy-driven innovative energy-efficient desalination technologies. Appl Energy.
689		2014 Dec;136:1155–65.
690	59.	EIA. U.S. Energy Information Administration. How much carbon dioxide is
691		produced per kilowatthour when generating electricity with fossil fuels? 2016.
692	60.	Burbano A, Brandhuber P. Demonstration of Membrane Zero Liquid Discharge
693		for Drinking Water Systems: A Literature Review. Water Environment Research
694		Foundation, 2012. 72 p.

- 695 61. Lappalainen J, Korvola T, Alopaeus V. Modelling and dynamic simulation of a
 696 large MSF plant using local phase equilibrium and simultaneous mass, momentum,
 697 and energy solver. Comput Chem Eng. 2017 Feb;97:242–58.
- 698 62. Piacentino A. Application of advanced thermodynamics, thermoeconomics and
 699 exergy costing to a multiple effect distillation plant: In-depth analysis of cost
 700 formation process. Desalination. 2015 Sep;371:88–103.
- Gabriel KJ, Linke P, El-Halwagi MM. Optimization of multi-effect distillation
 process using a linear enthalpy model. Desalination. 2015 Jun;365:261–76.
- 703 64. Thiel GP, Tow EW, Banchik LD, Chung HW, Lienhard JH. Energy consumption
 704 in desalinating produced water from shale oil and gas extraction. Desalination.
 705 2015 Jun;366:94–112.
- Lokare OR, Tavakkoli S, Rodriguez G, Khanna V, Vidic RD. Integrating
 membrane distillation with waste heat from natural gas compressor stations for
 produced water treatment in Pennsylvania. Desalination. 2017 Jul;413:144–53.
- Tavakkoli S, Lokare OR, Vidic RD, Khanna V. A techno-economic assessment of
 membrane distillation for treatment of Marcellus shale produced water.
 Desalination. 2017 Aug;416:24–34.
- Mickley M. Survey of High-Recovery and Zero Liquid Discharge Technologies
 for Water Utilities. WateReuse Foundation; 2008. 156 p.
- US Department of the Interior Bureau of Reclamation. Evaluation and Selection
 of Available Processes for a Zero-Liquid Discharge System for the Perris,
 California, Ground Water Basin. Desalin Water Purif Res Dev Progr Rep No 149.
 2008;(149).
- 718 69. Drioli E, Ali A, Lee YM, Al-Sharif SF, Al-Beirutty M, Macedonio F. Membrane
 719 operations for produced water treatment. Desalin Water Treat. 2016 Jul

720 2;57(31):14317–35.

- 721 70. Tufa RA, Curcio E, Brauns E, Van Baak W, Fontananova E, Di Profio G.
 722 Membrane distillation and reverse electrodialysis for near-zero liquid discharge
 723 and low energy seawater desalination. J Memb Sci. 2015 Dec;496:325–33.
- 724 71. Carrero-Parreño A, Onishi VC, Ruiz-Femenia R., Salcedo-Díaz R, Caballero JA,
 725 Reyes-Labarta JA. Multistage Membrane Distillation for the Treatment of Shale
 726 Gas Flowback Water: Multi-Objective Optimization under Uncertainty. In:
 727 Computer Aided Chemical Engineering. 2017. p. 571–76.
- 728 72. Salcedo-Díaz R, Ruiz-femenia R, Carrero-parreño A, Onishi VC, Reyes-Labarta
 729 JA, Caballero JA. Combining Forward and Reverse Osmosis for Shale Gas
 730 Wastewater Treatment to Minimize Cost and Freshwater Consumption. In:
 731 Computer Aided Chemical Engineering. 2017. p. 2725–30.
- 732 73. Hayes T. Sampling and analysis of water streams associated with the development
 733 of Marcellus shale gas. Rep by Gas Technol Institute, Des Plaines, IL, Marcellus
 734 Shale Coalit. 2009;10.
- 735 74. Galusky P, Hayes TD. Feasibility Assessment of Early Flowback Water Recovery
- for Reuse in Subsequent Well Completions RPSEA Report No 08122-05.07. 2011;
- 737 75. Ahmad M, Williams P. Assessment of desalination technologies for high saline
 738 brine applications Discussion Paper. Desalin Water Treat. 2011 Jun 3;30(1–
 739 3):22–36.
- 740 76. Hao H, Huang X, Gao C, Gao X. Application of an integrated system of
 741 coagulation and electrodialysis for treatment of wastewater produced by fracturing.
 742 Desalin Water Treat. 2015 Aug 21;55(8):2034–43.
- 743 77. Gude VG. Energy storage for desalination processes powered by renewable energy
 744 and waste heat sources. Appl Energy. 2015 Jan;137:877–98.

- 745 78. Lu Y, Chen J. Integration design of heat exchanger networks into membrane
 746 distillation systems to save energy. Ind Eng Chem Res. 2012 May 16;51(19):6798–
 747 810.
- 748 79. Cho H, Choi Y, Lee S, Sohn J, Koo J. Membrane distillation of high salinity
 749 wastewater from shale gas extraction: effect of antiscalants. Desalin Water Treat.
 750 2016 Nov 25:57(55):26718–29.
- 80. Cho H, Jang Y, Koo J, Choi Y, Lee S, Sohn J. Effect of pretreatment on fouling
 propensity of shale gas wastewater in membrane distillation process. Desalin
 Water Treat. 2016 Nov 15;57(51):24566–73.
- 81. Boo C, Lee J, Elimelech M. Omniphobic polyvinylidene fluoride (PVDF)
 membrane for desalination of shale gas produced water by membrane distillation.
 Environ Sci Technol. 2016 Nov 15;50(22):12275–82.
- Singh D, Sirkar KK. Desalination of brine and produced water by direct contact
 membrane distillation at high temperatures and pressures. J Memb Sci. 2012
 Feb;389:380–8.
- Kim J, Kwon H, Lee S, Lee S, Hong S. Membrane distillation (MD) integrated
 with crystallization (MDC) for shale gas produced water (SGPW) treatment.
 Desalination. 2017 Feb;403:172–8.
- 763 84. Chung HW, Swaminathan J, Warsinger DM, Lienhard V JH. Multistage vacuum
 764 membrane distillation (MSVMD) systems for high salinity applications. J Memb
 765 Sci. 2016 Jan;497:128–41.
- McGinnis RL, Hancock NT, Nowosielski-Slepowron MS, McGurgan GD. Pilot
 demonstration of the NH3/CO2 forward osmosis desalination process on high
 salinity brines. Desalination. 2013 Mar;312:67–74.
- 769 86. Chen G, Wang Z, Nghiem LD, Li X-M, Xie M, Zhao B, et al. Treatment of shale

770	gas drilling flowback fluids (SGDFs) by forward osmosis: Membrane fouling and
771	mitigation. Desalination. 2015 Jun;366:113–20.

- Hickenbottom KL, Hancock NT, Hutchings NR, Appleton EW, Beaudry EG, Xu
 P, et al. Forward osmosis treatment of drilling mud and fracturing wastewater from
 oil and gas operations. Desalination. 2013 Mar;312:60–6.
- Yun T, Koo J-W, Sohn J, Lee S. Pressure assisted forward osmosis for shale gas
 wastewater treatment. Desalin Water Treat. 2015 May 18;54(4–5):829–37.
- 89. Elimelech M, Phillip WAA. The Future of Seawater Desalination: Energy,
 Technology, and the Environment. Science (80-). 2011 Aug 5;333(6043):712–7.
- 90. Greenlee LF, Lawler DF, Freeman BD, Marrot B, Moulin P. Reverse osmosis
 desalination: Water sources, technology, and today's challenges. Water Res. 2009
 May;43(9):2317–48.
- 91. Salvador Cob S, Yeme C, Hofs B, Cornelissen ER, Vries D, Genceli Güner FE, et
 al. Towards zero liquid discharge in the presence of silica: Stable 98% recovery in
 nanofiltration and reverse osmosis. Sep Purif Technol. 2015 Jan;140:23–31.
- Michel MM, Reczek L, Granops M, Rudnicki P, Piech A. Pretreatment and
 desalination of flowback water from the hydraulic fracturing. Desalin Water Treat.
 2016 May 8;57(22):10222–31.
- McGovern RK, Weiner AM, Sun L, Chambers CG, Zubair SM, Lienhard V JH.
 On the cost of electrodialysis for the desalination of high salinity feeds. Appl
 Energy. 2014 Dec;136:649–61.
- 94. Peraki M, Ghazanfari E, Pinder GF, Harrington TL. Electrodialysis: An application
 for the environmental protection in shale-gas extraction. Sep Purif Technol. 2016
 Mar;161:96–103.
- 794 95. Camacho L, Dumée L, Zhang J, Li J, Duke M, Gomez J, et al. Advances in

- membrane distillation for water desalination and purification applications. Water.
 2013 Jan 25;5(1):94–196.
- 797 96. Lee S, Boo C, Elimelech M, Hong S. Comparison of fouling behavior in forward
 798 osmosis (FO) and reverse osmosis (RO). J Memb Sci. 2010 Dec;365(1–2):34–9.
- 799 97. Fakhru'l-Razi A, Pendashteh A, Abdullah LC, Biak DRA, Madaeni SS, Abidin
- ZZ. Review of technologies for oil and gas produced water treatment. J Hazard
 Mater. 2009 Oct 30;170(2–3):530–51.
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Figure Captions

- **Figure 1.** Wastewater management alternatives for shale gas industry.
- **Figure 2.** Conceptual profiles for total dissolved solids (TDS) concentration and
- 808 wastewater flowrate in function of time from hydraulic fracturing operations.
- **Figure 3.** Schematic representation of major thermal and membrane-based processes for
- 810 shale gas wastewater desalination.
- 811 Figure 4. Multiple-effect evaporation system with mechanical vapor compression (MEE-
- 812 MVC) for the zero-liquid discharge (ZLD) desalination of shale gas wastewater as
- 813 proposed by Onishi et al. (15).
- 814 Figure 5. Distributions throughout different feeding scenarios of zero-discharge MEE-
- 815 MVC system for: (*a*) energy consumption; and, (*b*) operational expenses. Data retrieved
- 816 from Onishi et al. (14).
- **Figure 6.** Zero-discharge MEE-MVC system driven by solar energy for the desalination
- 818 of high-salinity shale gas wastewater.
- **Figure 7.** Schematic representation of a thermal-based ZLD evaporation plant coupled to
- the pretreatment system and crystallization or evaporation ponds.
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Water Recycling or Discharge

826

828 Figure 1. Wastewater management alternatives for shale gas industry.



Days following hydraulic fracturing

829

Figure 2. Conceptual profiles for total dissolved solids (TDS) concentration and wastewater flowrate in function of time from hydraulic fracturing 830

operations. 831



Figure 3. Schematic representation of major thermal and membrane-based processes for shale gas wastewater desalination.



Figure 4. Multiple-effect evaporation system with mechanical vapor compression (MEE-MVC) for the zero-liquid discharge (ZLD) desalination

of shale gas wastewater as proposed by Onishi et al. (15).



Figure 5. Distributions throughout different feeding scenarios of zero-discharge MEE-MVC system for: (a) energy consumption; and, (b)

operational expenses. Data retrieved from Onishi et al. (14).







845 Figure 7. Schematic representation of a thermal-based ZLD evaporation plant coupled to the pretreatment system and crystallization or evaporation

846 ponds.

847	Table 1. Water amount required per well for drilling and hydrofracturing processes, and
848	shale gas wastewater information from prominent U.S. shale plays.

	U.S. shale	Water	Wastewater	Average TDS		
Data source	play	amount (m ³)	recovery (%)	(k ppm)		
Hayes (73)	Marcellus	11,356-1,5142	25%	157 ²		
	Fayetteville	11,368		13		
	Woodford	-		30		
Acharya et al.	Barnett	12,719	15-40% 1	80		
(12)	Marcellus	14,627		120		
	Haynesville	14,309		110		
Galusky and		11 256 10 025	25 400/	02		
Hayes (74)	Barnett	11,356-18,927	25-40%	~92		
Hayes and	Marcellus			120 ²		
Severin (37)	Barnett	-	-	50.55 ³		
Slutz et al. (28)	-	12,700-19,000	10-40%	-		
Vidic et al. (9)	Marcellus	7,570-26,500	9-53%	-		
Zammerilli et al.	Magaallug	7 570 22 712	20 700/	70		
(24)	Marcellus	7,570-22,712	30-70%			
Rosenblum et al.	NT: - h	11,000	20/ 200/ 4	10 (10 0 4		
(22)	INIODIATA	11,000	~3%0-30%0	18.0-18.8		
Hammond and		10 000-20 000	40-200/			
O'Grady (23)	-	10,000 -30,000	40 0070	-		

¹ Overall produced water recovery after 90 days.

850 2 TDS average values for the shale gas flowback water in 14th day following hydraulic fracturing.

851 ³ TDS average values for the shale gas flowback water in 10th to 12th day following hydraulic fracturing.

852 ⁴ Average values in 15th and 220th days following hydraulic fracturing.

853	Table 2.	Typical	concentration	ranges	for	critical	constituents	found	in	shale	gas
854	wastewater from Marcellus play ¹ .										

	Minimum	Maximum	Average
Constituent	(mg L ⁻¹)	(mg L ⁻¹)	(mg L ⁻¹)
Total Dissolved Solids (TDS)	680	345,000	106,390
Total Suspended Solids (TSS)	4	7,600	352
Total Organic Carbon (TOC)	1.2	1530	160
Chloride	64.2	196,000	57,447
Sulfate	0	763	71
Sodium	69.2	117,000	24,123
Calcium	37.8	41,000	7,220
Barium	0.24	13,800	2,224
Strontium	0.59	8,460	1,695
Iron, total	2.6	321	76
Alkalinity (as CaCO ₃)	7.5	577	165
Bromide	0.2	1,990	511
Magnesium	17.3	2,550	632
Oil and grease	4.6	802	74

855 ¹ Data compiled from Barbot et al. (20) for flowback water samples collected between day 1 and day 20

856 *following hydraulic fracturing.*

Desalination system	ZLD operation	Freshwater production cost	Specific energy consumption	Reference
SEE-MVC (electric-driven system	Brine salinity at 300k ppm and	10.00 LISC3	50.47 kWh m ⁻³	$O_{1}:=1:=1:(15)$
with single-stage compression)	76.7% of conversion ratio	10.90 US\$ m ³	(4.90 US\$ m ⁻³)	Unishi et al. (15)
SEE-MVC (electric-driven system	Brine salinity at 300k ppm and	10.0 5 LIO C -3	49.85 kWh m ⁻³	0, 1, 1, 1, (15)
with multi-stage compression)	76.7% of conversion ratio	10.85 US\$ m ³	$(4.84 \text{ US} \$ \text{ m}^{-3})$	Unishi et al. (15)
SEE-MVC (rigorous heat transfer	Brine salinity at 300k ppm and	10.07 LIO0 -3	49.78 kWh m ⁻³	
coefficients estimations)	76.7% of conversion ratio	10.07 US\$ m ⁻⁹	(4.83 US\$ m ⁻³)	Unishi et al. (14)
SEE-MVC	Not ZLD, 26% of brine salinity	-	$23 - 42 \text{ kWh m}^{-3}$	Thiel et al. (64)
	Brine salinity at 300k ppm and	10.05 H0¢ -3	214.19 kWh m ⁻³	
MEE (steam-driven system)	76.7% of conversion ratio	12.85 US\$ m ⁻⁵	(10.24 US\$ m ⁻³)	Onishi et al. (15)
MEE-MVC (electric-driven system	Brine salinity at 300k ppm and		28.63 kWh m ⁻³	
with single-stage compression)	76.7% of conversion ratio	6.70 US\$ m ⁻³	(2.78 US\$ m ⁻³)	Onishi et al. (15)
MEE-MVC (electric-driven system	Brine salinity at 300k ppm and		28.84 kWh m ⁻³	
with multi-stage compression)	76.7% of conversion ratio	6.83 US\$ m ⁻³	(2.80 US\$ m ⁻³)	Onishi et al. (15)
MEE-MVC (rigorous heat transfer	Brine salinity at 300k ppm and		28.33 kWh m ⁻³	
coefficients estimations)	76.7% of conversion ratio	6.55 US\$ m ⁻³	(2.75 US\$ m ⁻³)	Onishi et al. (14)

Table 3. Freshwater production cost and specific energy consumption of thermal-based systems for shale gas wastewater desalination.

MEE-MVC (hybrid steam and	Brine salinity at 300k ppm and	5 25 LIS¢ m ⁻³	23.25 kWh m ⁻³	Onichi et al. (2)
electricity energy sources)	73.3% of conversion ratio	5.25 US\$ III	$(2.26 \text{ US} \$ \text{ m}^{-3})$	OIIISIII et al. (3)
MEE-MVC	Not ZLD, 26% of brine salinity	-	20 kWh m ⁻³	Thiel et al. (64)

Table 4. Freshwater production cost and specific energy consumption of membrane-based systems for shale gas wastewater desalination.

Desalination system	ZLD operation	Freshwater production cost	Specific energy consumption	Reference
Direct contact MD system (waste heat energy source)	Brine salinity at 300k ppm or 30% (<i>w/v</i>), water recovery ratio of 66 7%	_	527 – 565 kWh m ⁻³ (depending on feed temperature)	Lokare et al. (65)
Direct contact MD system (waste heat and electricity heat energy sources)	Brine salinity at 300k ppm or 30% (<i>w/v</i>), water recovery ratio of 66.7%	0.74 - 5.70 US\$ m ⁻³ and 61 - 66 US\$ m ⁻³ (with transportation costs) ¹	-	Tavakkoli et al. (66)
Two-stage RO system	Not ZLD, 26% of brine salinity	-	$4 - 16 \text{ kWh m}^{-3}$	Thiel et al. (64)
Hybrid EDR-RO with	Brine salinity at 239k ppm,	_	$10 - 17 \text{ kWh}_{e} \text{ m}^{-3} \text{ (EDR-RO)}$	Loganathan et al.
crystallizer system	water recovery ratio of ~77%	-	and 40 kWhe m^{-3} (crystallizer)	(55)
ED system	Not ZLD	-	49.7 kWh _e m ⁻³ (wastewater with 70k ppm TDS) and 175.7 kWh _e m ⁻³ (wastewater with 250k ppm TDS)	Ahmad and Williams (75)
Integrated coagulation and	Not ZLD, 91% of salt		\sim 7 – 14 kWh m ⁻³ (depending	Hao et al. (76)
ED system	removal	-	on the ED voltage)	11a0 Ct al. (70)

861 ¹ Values estimated based on cubic meter of feed water (with salinity of 100k ppm).

Desalination	Adventages	Dreamh a cha	Defenence
technology	Advantages	Drawbacks	Reference
Multistage flash	- Well-stablished technology with application	- Cost and energy-intensive process, not	NA
distillation (MSF)	to shale gas wastewater with large range of	suitable for small scale operations (77)	
	TDS contents	- Intensive use of scale inhibitors and cleaning	
	- High-quality water product (ultrapure water	agents	
	or freshwater)		
	- Technical maturity		
	- Possibility of using geothermal or solar		
	energy sources		
Single/multiple-	- Well-stablished technology with Application	- Energy-intensive process	Onishi et al.
effect evaporation	to shale gas wastewater with large range of	- Usually operated by high-grade electric	(3,14–17)
with mechanical	TDS contents $(10 - 220 \text{k ppm})$	energy (for this reason, these systems present	
vapor compression	- Brine discharge salinity up to 300k ppm TDS	high operating expenses and indirect GHGs	
(SEE/MEE-MVC)	- Use of less intensive pretreatment processes,	emissions)	
	when compared to membrane-based		
	technologies		

Table 5. Comparison between thermal and membrane-based technologies for ZLD desalination of shale gas wastewater.

	- High energy efficiency	- High capital costs, due to the expensive	
	- High-quality water product (ultrapure water	materials (stainless steel or titanium) required	
	or freshwater)	to prevent rusting	
	- Technical maturity		
	- Modular feature		
	- Heat exchangers and flashing tanks can be		
	used to further enhance energy recovery,		
	reducing energy consumption		
	- Possibility of using geothermal or other		
	renewable energy sources, which allows to		
	reduce carbon footprint		
Membrane	- Application to shale gas wastewater with high	- Energy-intensive process with energy	Carrero-Parreño
distillation (MD)	TDS contents	consumption higher than RO and ED/EDR	et al. (71)
	- Brine discharge salinity higher than 200k ppm	(DCMD requires $40 - 45$ kWht m ⁻³ for	Boo et al. (81)
	TDS	seawater desalination (54))	Singh and Sirkar
	- Modular feature and operation at low	- Heat integration (by using heat exchangers	(82)
	temperature and pressure	and brine recycling) is critical to enhance	Kim et al. (83)
	- Low fouling propensity	energy efficiency to competitive levels with	Chung et al. (84)
		thermal systems (78)	Lokare et al. (65)

	- Possibility of using low-grade thermal energy,	- Membrane wetting potential	
	including geothermal or waste heat, which	- Intensive pretreatment and use of cleaning	
	allows to reduce operating costs and carbon	agents and scale inhibitors (79,80)	
	footprint	- Limited to commercial applications	
Forward osmosis	- Application to shale gas wastewater with TDS	- Intensive pretreatment processes (softening,	Salcedo-Díaz et
(FO)	contents up to 180k ppm (85)	pH adjustment, ultrafiltration, ion exchange,	al.(72)
	- Brine discharge salinities higher than 220k	etc.) to prevent operating problems related to	McGinnis et al.
	ppm TDS	fouling and scaling (however, these processes	(85)
	- Modular feature	are less intensive and more economical than	Chen et al. (86)
	- Can be used for pre-concentrating and	those required prior RO)	Hickenbottom et
	pretreating wastewater prior RO process	- Regular membrane cleaning	al. (87)
	- High rejection of many contaminants		Yun et al. (88)
	- Propensity to membrane fouling and scaling		
	lower than RO process (with reversible		
	membrane fouling)		
	- Low electricity consumption		
	- Possibility of using low-grade thermal energy,		
	including geothermal or waste heat, which		
	allows to reduce operating costs and carbon		
	footprint		

Reverse osmosis

(RO)

- Application to shale gas wastewater with TDS	- High propensity to membrane fouling and	Salcedo-Díaz et
contents up to $40 - 45$ k ppm (38,72)	scaling, which requires intensive pretreatment	al.(72)
- High energy efficiency	processes (softening, pH adjustment,	Miller et al. (53)
- Technical maturity	coagulant/flocculant addition, ultrafiltration,	
- Modular feature and great adaptability to	ion exchange, etc.) to prevent operating	
wastewater treatment plants with other	problems (90)	
technologies, including water pretreatment	- Intensive use of antiscalants (91)	
processes (38)	- Inability to operate at high hydraulic pressure	
- Can be used for pre-concentrating wastewater	- Stand-alone RO systems are not able to	
prior energy-intensive thermal processes (54)	operate at ZLD conditions: brine discharge	
- Low energy consumption of ~2 kWh _e m ⁻³ , for	salinity up to 70k ppm TDS	
seawater desalination (89)	(crystallizer/evaporator should be included in	
	the system) (54)	

Nanofiltration (NF) - Effective as softening for subsequent

- wastewater treatment processes
- High water recovery
- Energy consumption lower than RO
- Mature technology

- Not effective as stand-alone process for shale Michel et al. (92) gas wastewater treatment
- Intensive pretreatment and scale inhibitors

Electrodialysis (ED)
and electrodialysis

reversal (EDR)

- Ability to achieve high brine salinities (TDS >
100k ppm)
- Salt removal rate ~91% (product water meets
the requirements on water reclamation)
- Relatively simple operation and maintenance
- Low propensity to fouling (especially with
coagulation pretreatment)
- Long-term operation
- Modular feature

- Application to high-salinity wastewater

- High energy consumption and related	Loganathan et al.
operating costs when coupled to	(55)
crystallizers/evaporators to achieve ZLD	McGovern et al.
conditions	(93)
- Regular membrane cleaning to maintain	Peraki et al.
operational production ratios	(94)

- Inability to remove non-charged contaminants

Desalination technology	Driving force and process characteristics	High-salinity application
	MD is a thermal-driven membrane desalination process,	Singh and Sirkar (82) have performed an experimental
	in which vapor pressure difference across the membrane	study on the desalination of shale gas wastewater through
	acts as driving force. The vapor pressure gradient is	direct contact membrane distillation (DCMD) at high
	caused by the temperature difference between the hot	temperature and above-ambient pressure, using hollow
	wastewater stream (feed stream) and the cold permeate	fibers membranes. Their results emphasize that DCMD is
	stream (distillate) (81). In recent years, MD has gained	a cost-competitive desalination process for high-salinity
	increased attention by the literature due to its potential to	shale gas wastewater, especially when compared to
Membrane	efficiently deal with high-salinity wastewater from shale	conventional RO. This is because the DCMD process does
distillation (MD)	gas production. High purity water can be expected by	not require feed cooling at the operating conditions
	applying MD treatment to the shale gas wastewater. This	considered by the authors. Chung et al. (84) have proposed
	is due its high removal rate of salts, metals and non-	a multistage vacuum membrane distillation (VMD) for
	volatile components. Also, MD systems present several	ZLD ¹ desalination of high-salinity wastewater
	advantages over standard thermal and pressure-based	applications. The latter authors have used a finite
	membrane processes, including their ability to achieve	differences-based method for numerical process
	higher brine concentrations (ZLD operation) and potential	simulations, by allowing brine discharge salinity near to
	use of low-grade waste heat or renewable energy sources	saturation conditions. Their results indicate that

Table 6. Process characteristics and applications of membrane-based technologies for ZLD desalination of shale gas wastewater.

(e.g., wind, solar, geothermal, wave, etc.) (69). Typically, MD processes can be operated at temperatures ranging 40 - 80°C (at atmospheric pressure) and driven by a low temperature difference of 20°C between the feed and distillate streams. For these reasons, waste grade heat can provide the thermal energy required by the MD desalination process (95).

in which a chemical potential difference between the for the treatment of fracturing wastewater from shale gas concentrated draw solution and a wide range of solutions operations. Bench-scale experiments performed by the (e.g., shale gas wastewater) acts as driving force for salt authors reveal that the FO system can achieve a water separation (87). FO is a promising membrane process for recovery efficiency of ~80%, with high rejection of the desalination of high-salinity shale gas wastewater. In organic and inorganic contaminants. Yun et al. (88) have fact, this technology presents several advantages over investigated the application of pressure assisted FO and

multistage VMD systems can be as cost-efficient as MSF schemes for a large range of feed water salinities. Tavakkoli et al. (66) have studied the techno-economic suitability of MD at ZLD operation (brine discharge salinity at 30% w/v) for desalinating produced water from Marcellus shale play. Their results reveal that the freshwater production cost is significantly affected by the initial TDS contents on wastewater, as well as by the thermal energy prices. Lastly, Carrero-Parreño et al. (71) have successfully reach ZLD operation (brine discharge salinities) by applying both DCMD and VMD systems for the shale gas wastewater desalination.

FO is an osmotically driven membrane-based technology, Hickenbottom et al. (87) have studied the suitability of FO

Forward osmosis

(FO)

other membrane alternatives, such as its ability to operate air gap membrane distillation (AGMD) for the

at higher salt concentrations (mainly when draw solutes desalination of shale gas wastewater. Their experimental regeneration is considered) (85), and easier fouling results indicate that the water flux across the membrane reversibility when compared to RO treatment (96). FO can be increased to 10 - 15% for wastewaters with low systems can also be operated at low pressure, which can and medium TDS contents, by considering an external prevent fouling and reduce pre-treatment requirements pressure of 10 bar. However, the effect of the external and maintenance. In this process, concentrate brine can be pressure is considerably reduced for high-salinity sent to a crystallizer (or evaporation ponds) to achieve wastewaters. Also, the authors have shown that AGMD ZLD operation, while treated water is separated from can be an effective process to re-concentrate draw solutes. draw solutes to regenerate the draw solution (54). For McGinnis et al. (85) have tested a pilot-scale FO system shale gas wastewater desalination, RO and MD can be for the desalination of high-salinity shale gas wastewater coupled to the FO system to re-concentrate the draw from Marcellus shale play. The authors have considered a solution and produce high quality water. Despite recent NH_3/CO_2 draw solution to treat wastewaters with ~73k advances, further improvement in the development of ppm TDS (and hardness of 17k ppm CaCO₃). The process membrane materials and draw solutions, as well as proposed by the authors include pretreatment (softening, operating conditions optimization, will be critical to media filtration, activated carbon and cartridge filtration), enhance process cost-effectiveness, and make FO a post-FO thermal desalination, RO and brine stripper. competitive alternative for high-salinity applications (39). Their results indicate water recovery of ~64% (brine discharge salinity of ~180k ppm), with an energy consumption 42% lower than conventional MVC process.

by the separation of dissolved salts from a (pressurized) applicability of three different techniques for the saline water solution through a semi-permeable desalination of high-salinity shale gas wastewater: MD, membrane. In this way, the flow across the membrane RO and evaporative crystallization (EC). Their results occurs due to a pressure differential established between indicate relatively higher efficiencies for MD and EC the high-pressure feed water and the low-pressure (>99.9%) than the RO technology (97.1–99.7%). Despite permeate. In the RO process, water molecules are the elevated removal rates presented by the RO process, transferred from a high salt concentration region to the the latter has been significantly affected by the TDS levels permeate side owed to an osmosis pressure. For this on the wastewater, requiring four times more dilution reason, feed water should be pressurized above osmotic before operation than MD and EC. In a recent study, condition, whilst the permeate should be at near- Salcedo-Díaz et al. (72) have proposed a ZLD desalination atmospheric pressure (90). RO is an energy-intensive system composed RO and FO technologies for shale gas process, in which the major energy requirement is related wastewater application. The authors have developed a to the feed water mechanical pressurization. The mathematical model for the optimal design of onsite ROefficiency of RO separation process can severely be FO systems, to minimize freshwater consumption and impaired by membrane fouling and scaling. These specific fracturing water cost. Their results show that is problems can be prevented by effective wastewater technically possible to reduce to zero the amount of pretreatments and the consideration of different freshwater used in shale gas operations. However, due to

RO is a pressure-driven desalination process characterized Jang et al. (33) have experimentally evaluated the membrane processes in the system (69). Salt the high freshwater production cost presented by the

Reverse osmosis

(RO)

concentrations in shale gas wastewater are critical for RO desalination system—in which, the cost of the cubic meter wastewaters with TDS contents lower than 30k ppm (39). In addition, RO can be included into ZLD desalination more affordable for shale gas industry. systems to enhance process cost-effectiveness. Almost 80% of wastewater volume can be reduced by using RO technology (44). Usually, RO processes are operated at low temperatures $<45^{\circ}$ C (at 20 – 60 atm).

desalination (33). RO systems are cost-effective for of treated water is about 100 times higher than the same amount of freshwater-, an intermediate solution can be

ED membrane-based processes for the desalination of highsalinity shale gas wastewater. These technologies are The authors have experimentally evaluated the optimal characterized by dissolved ions separation across ionselective membranes, in which the electrical potential gradient works as driving force (69,94). In EDR process, membranes polarity is changed to fouling and scaling control (69). ED and EDR systems can be used for (stack voltage) to minimize desalination costs. Hao et al. removing salts from RO treated waters (97). The (76) have developed an integrated process of coagulation performance of ED and EDR processes is significantly and ED for the treatment of fracturing wastewater. The affected by several factors, including applied voltage, coagulation is used for removing organic contaminants

and EDR are electrochemical charge-driven McGovern et al. (93) have proposed a 10-stage ED system for the treatment of high-salinity shale gas wastewater. equipment size and energy requirements to desalinate wastewater with salinities up to 192k ppm TDS. Their results emphasize the process effectiveness and the need for further investigating fouling and operating conditions

Electrodialysis (ED) /

Electrodialysis reversal (EDR) wastewater flowrate and ions concentration, membrane from the wastewater, while its desalination is performed density, diffusion, etc. The main disadvantages are related by the ED system. Their results show ion removal rates up to high energy consumption and water production costs, and fouling propensity (75). In addition, these processes al. (94) have investigated the ED efficiency as a require regular membrane cleaning (alkalis or dilute acidic solutions) to keep operating conditions. The latter drawbacks must be addressed to improve competitiveness of ED/EDR for the industrial scale application to high-salinity shale gas wastewaters (69).

¹ Although evaporation ponds or crystallizers are required to literally achieve zero-discharge operation, brine discharges salinities near to salt saturation conditions are
 considered as ZLD operation in this work.