



# Design and tailoring of patterned ZnO nanostructures for energy conversion applications

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**ABSTRACT** ZnO is a typical direct wide-bandgap semiconductor material, which has various morphologies and unique physical and chemical properties, and is widely used in the fields of energy, information technology, biomedicine, and others. The precise design and controllable fabrication of nanostructures have gradually become important avenues to further enhancing the performance of ZnO-based functional nanodevices. This paper introduces the continuous development of patterning technologies, provides a comprehensive review of the optical lithography and laser interference lithography techniques for the controllable fabrication of ZnO nanostructures, and elaborates on the potential applications of such patterned ZnO nanostructures in solar energy, water splitting, light emission devices, and nanogenerators. Patterned ZnO nanostructures with highly controllable morphology and structure possess discrete three-dimensional space structure, enlarged surface area, and improved light capture ability, which realize the efficient carrier regulation, achieve highly efficient energy conversion, and meet the diverse requirements of functional nanodevices. The patterning techniques proposed for the precise design of ZnO nanostructures not only have important guiding significance for the controllable fabrication of complex nanostructures of other materials, but also open up a new route for the further development of functional nanostructures.

**Keywords:** patterned ZnO nanorod arrays, laser interference lithography, optical lithography, energy conversion devices

## INTRODUCTION

Semiconductor materials have attracted considerable re-

search interest because of their broad applications in information transmission, detectors, monitors, lasers, and many other fields [1–3]. The first and second generations of semiconductor materials, based on silicon and gallium arsenide, respectively, have contributed to the rapid development of the computer electronics and mobile communication industries [4–6]. However, owing to the limitation of material properties, the devices based on these semiconductor materials cannot operate at high temperatures [7]. Their anti-radiation performance and emission wavelength range are also unable to meet the increasing needs of societal development [8]. As a representative of the third generation of semiconductor materials, gallium nitride (GaN) has been widely used in short wavelength optoelectronic devices and high-power/high-frequency electronic devices, owing to its wide bandgap, high breakdown voltage, and strong radiation resistance [9,10]. Zinc oxide (ZnO) is another wide-bandgap semiconductor, which has a similar crystal structure to GaN and has attracted considerable interest over the past years [11,12].

As a typical direct bandgap semiconductor, ZnO is a potential candidate material for photoelectronic applications, owing to its excellent physical and chemical properties, such as high electron mobility, excellent environmental stability, and high thermal conductivity [13,14]. ZnO exhibits the piezoelectric effect and excellent photocatalytic performance [15,16]. The large exciton binding energy of ZnO facilitates efficient exciton emission at room temperature or higher temperatures. Generally, the properties of ZnO can be further optimized by

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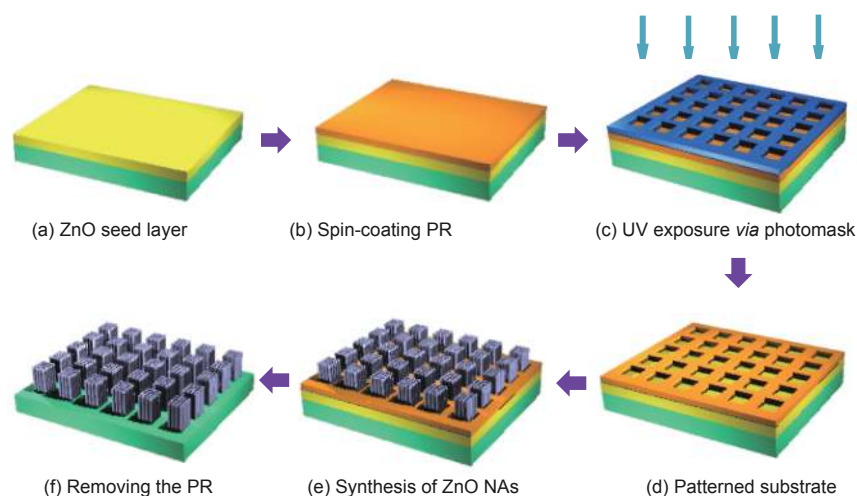
doping with various elements [17,18]. The incorporation of Al, Ga, and In reduces the lattice energy of ZnO [19,20]. In particular, the addition of Ga and In can improve the conductivity of ZnO and enhance the stability of its crystal structure [21]. The addition of Al, Ga, and In can further increase the bandgap and conductivity of ZnO without sacrificing its transmittance. The mechanical and electrical properties can be enhanced by Sb doping. The incorporation of transition-metal elements (Sc, Mn, Ti, Co, Ni, Cu, etc.) into ZnO will form diluted magnetic semiconductors or semimagnetic semiconductors [22–24]. The electrical properties of ZnO can be controlled by precisely tuning the dopant concentration [25]. Additionally, the elements of N, P, and others have been used as effective dopants to obtain p-type semiconductors [26]. However, it is generally difficult to achieve bipolar doping for wide-bandgap semiconductors. Furthermore, a variety of ZnO nanostructures have been synthesized by low-cost and low-temperature methods [27]. Among the obtained nanostructures, ZnO nanorods (NAs) can provide an electron transport pathway for the efficient separation of photogenerated electron–hole pairs [28–30]. ZnO NAs have a large surface area, which is conducive to compositing with other materials [17]. The light absorption performance is higher than other structures, because of the light scattering effect. Therefore, ZnO NAs have been widely used in optoelectronic, electrochemical, and electronic devices, such as nanogenerators [31–34], sensors [35–38], light-emitting diodes [39,40], ultraviolet (UV) detectors [41–45], solar cells [46–49], field emission devices [50,51], and biosensors [52–55].

ZnO materials are easy to synthesize and have plentiful preparation methods, which can be divided into the vapor-phase, solution-phase, and solid-phase growth methods, according to the state of the synthetic phase [56–59]. Regarding the ZnO NAs, a wide range of fabrication techniques, such as metal-organic chemical vapor deposition, chemical vapor deposition, pulsed laser deposition, electrochemical deposition, and hydrothermal methods, have been developed [60–62]. However, the ZnO NAs synthesized by traditional vapor-phase and solution-phase growth methods have several drawbacks, such as poor orientation, different space and length of nanorods, and less morphology homogeneity. These ZnO NAs resulted in poor electrode contact, leakage current, large reverse current, and poor stability, which limit the further improvement of the performance and lifetime of the devices [63]. In order to solve the above problems, patterning technology was widely used in the preparation of ZnO NAs.

Recently, with the continuous development of patterning technology, the precise control of the arrays can be realized gradually. ZnO NAs with controllable diameter and period, uniform distribution, and high orientation have been widely used in various energy devices. In this review, we focus on the fabrication techniques of patterned ZnO NAs and their applications in the field of energy conversion.

## SYNTHETIC METHODOLOGIES AND PROPERTIES FOR PATTERNED ZnO NAs

Several strategies have been proposed to fabricate highly ordered ZnO NAs, such as nanosphere lithography (NSL) [64–68], electron beam lithography (EBL) [69–72], optical lithography (OL), and laser interference lithography (LIL) techniques [73–80]. Among these, self-assembled NSL is a simple and economical technique, which employs two-dimensional (2D) self-assembled nanometer-sized polystyrene spheres as lithography masks to fabricate patterned arrays [64]. The period and size of the patterned ZnO NAs are dependent on the size of nanospheres [65]. In order to overcome the limitation of period control of this method, heterogeneous NSL was proposed by using two different particles [66]. The conventional NSL required high-temperature post-treatment and had high requirement on the substrate flatness and hydrophobicity. On the other hand, the fabrication of defect-free and period-adjustable ZnO NAs was difficult, owing to the introduction of metal catalyst dots and the inhomogeneous distribution of polystyrene nanospheres [67]. Recently, patterned ZnO NAs were fabricated by an NSL technique that patterns a thickness variation on a polymethyl methacrylate layer [68]. The size of the ZnO NAs was successfully controlled by tuning the etching time under oxygen plasma. The morphology and orientation of ZnO NAs can be effectively controlled by tailoring the thickness of the seed layer and solution concentration. EBL is a high-resolution maskless lithography technique, in which a focused electron beam is used to record predetermined shapes on photoresist (PR) materials [70]. Through this method, highly oriented ZnO NAs with controllable diameter and tunable pattern density were obtained. However, the EBL method was too costly and the processing speed was slow, which was not suitable for large-area periodic ZnO NAs fabrication [72]. Compared with the above methods, OL and LIL techniques are precision processing methods, which combine chemical and physical etching techniques to form the patterned geometry. The OL method uses UV light to transfer a geometric pattern from the photomask to the



**Figure 1** Schematic fabrication process of patterned ZnO NAs by OL method.

PR material. By the OL technique, patterns of various size and shapes on different substrates have been obtained. However, owing to the diffraction effect, the ZnO NAs are usually distributed in clusters instead of individual nanorods, and the nanorod diameter is limited by the incident wavelength [73]. Compared with OL, the LIL technique is a reliable and fast method, which has been widely utilized to obtain nanoscale-patterned ZnO NAs without using a photomask [78]. Generally, an interference pattern is produced by two or multiple coherent lights under one or more exposures, and the irradiated laser energy is recorded on the PR [79]. Then such a pretreated substrate is used to fabricate patterned NAs with controllable density and diameter. Thus, this method allows not only the precise control of the growth cycles and size of the nanorods, but also the successful fabrication of large-area highly oriented ZnO NAs.

#### Fabrication of patterned ZnO NAs by OL technique

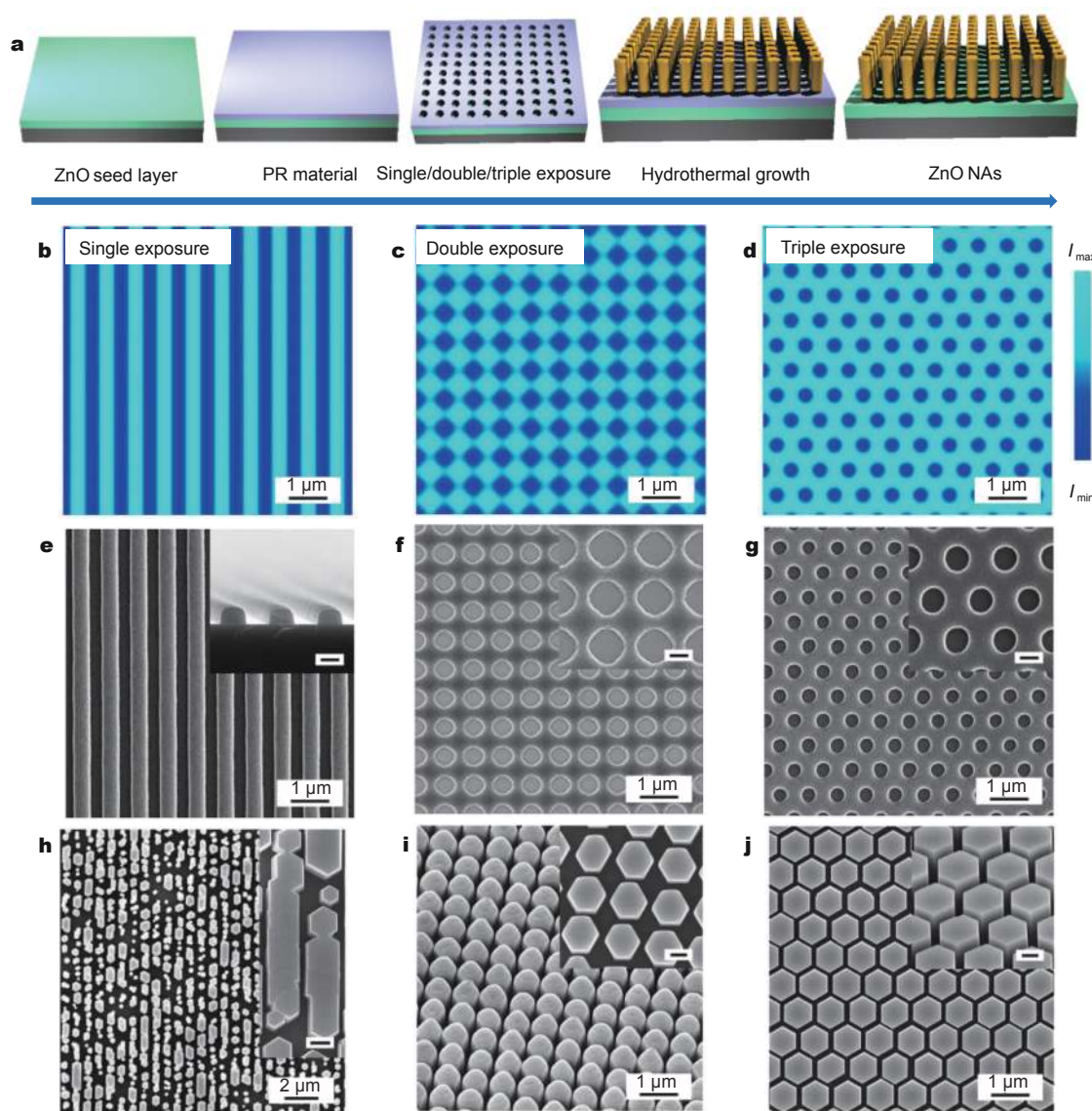
The OL technique is a microfabrication method used to form geometric patterns on substrates. Generally, geometric patterns on masks are transferred to PR materials by light treatment. Fig. 1 shows the typical fabrication processes. First, the PR materials are spin-coated onto the substrate. Second, intense UV light is irradiated onto the substrate with a mask. The light treatment causes a chemical change in positive PR materials, which allows the exposed regions of PR to be removed by solution. For negative PR materials, the unexposed regions become soluble. Thus, the geometric patterns are recorded on the substrate from the mask. Finally, ZnO NAs with different geometric patterns are fabricated. The PR materials are

not essential in the OL technique. The geometric patterns can be observed on a photosensitive surface instead of a PR layer [81]. The UV exposure could result in carboxylic acid groups [82] or sulfate anion groups [83] on polycarbonate (PC) films and polyethylene terephthalate (PET) filaments. The acid groups form an acidic environment, which is not conducive to the nucleation and growth of ZnO NAs. Thus, the polymer successfully records the patterns of the mask and is used to grow patterned ZnO NAs. This method enables the direct growth of well-organized ZnO NAs with different patterns.

The OL technique is a simple and effective method for preparing patterned ZnO NAs with different periodicities. The shape, size, and number of ZnO NAs on the substrate can be controlled by changing the fabrication conditions and the photomask.

#### Fabrication of patterned ZnO NAs by LIL technique

The LIL technique uses a dual-beam laser to design the template patterns [84]. Two-beam laser interference lithography (2BLIL) is used to obtain line template [77]. To obtain various diffraction patterns, multiple exposure processing is employed, as shown in Fig. 2a. Generally, the PR is spin-coated onto a *c*-oriented ZnO seed layer. Then, 2BLIL processes with single exposure, double exposure at  $0^\circ/90^\circ$ , and triple exposure at  $-60^\circ/0^\circ/60^\circ$  are adopted for line, square, and hexagonal lattice arrays, respectively. In 2013, based on the 2BLIL technique, the large-scale patterned ZnO NAs were fabricated on single-crystalline *p*-GaN substrates through three-exposure modes [85,86]. Such three-exposure modes resulted in different intensity distributions (Fig. 2b–d) and PR tem-

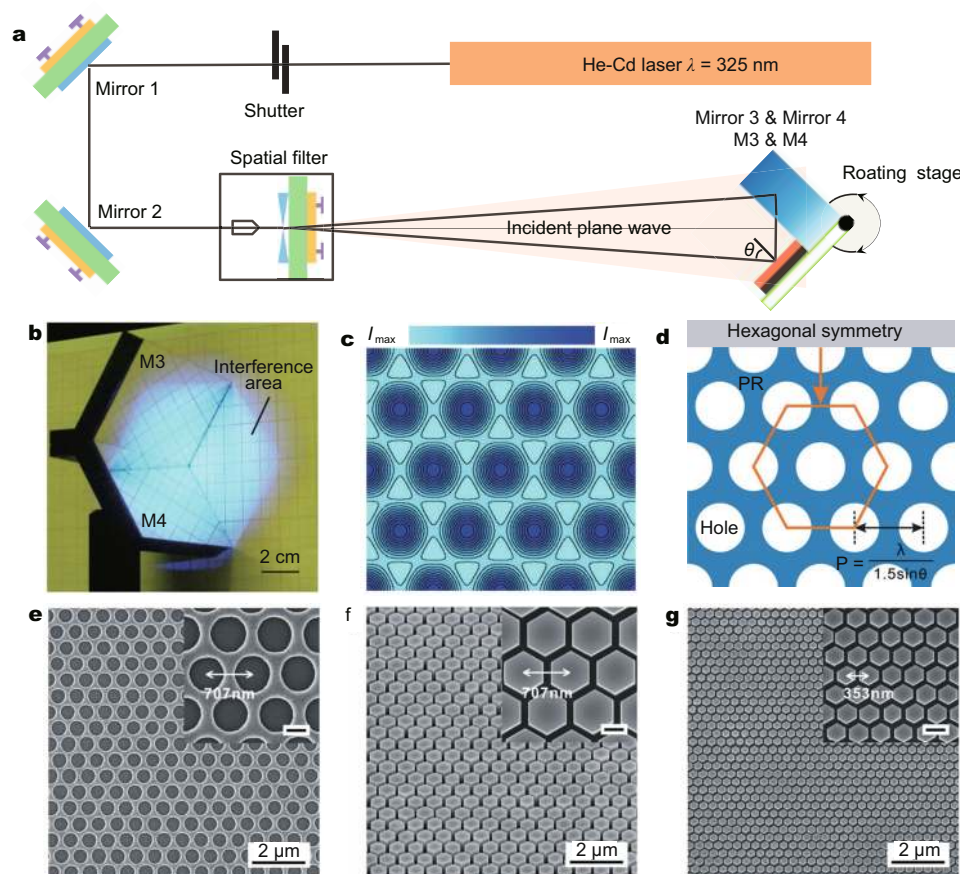


**Figure 2** (a) Schematic diagram of patterned ZnO NAs by LIL technique. Simulated exposure intensity distribution of (b) single exposure, (c) double exposure ( $0^\circ$  and  $90^\circ$ ), and (d) triple exposure ( $-60^\circ$ ,  $0^\circ$ , and  $60^\circ$ ). Such interference patterns can be recorded and the corresponding PR templates are line template (e), hole templates with square symmetry (f), and hexagonal symmetry (g). The corresponding SEM images of patterned ZnO NAs are wall-like ZnO NAs with smooth sidewalls (h), square ZnO NAs with rough-top (i), and hexagonal ZnO NAs with flat-top (j) in a period of 774 nm. All insets show the corresponding magnified SEM images; scale bars are 300 nm. Reprinted with permission from Ref. [85], Copyright 2013, the Royal Society of Chemistry.

plate patterns (Fig. 2e–g). Based on the line templates, the large-scale wall-like patterned ZnO NAs with smooth sidewalls were obtained, as shown in Fig. 2h. The large-scale individual patterned ZnO NAs with square symmetry was also fabricated on a p-GaN substrate by double-exposure processing, as depicted in Fig. 2i. By triple-exposure processing, the ZnO NAs with hexagonal symmetry was prepared on a p-GaN substrate (Fig. 2j). However, the morphology of the patterned ZnO NAs with

hexagonal symmetry was influenced by the orientation of the PR template.

In order to fabricate highly ordered ZnO NAs by single-exposure processing, the three-beam laser interference lithography (3BLIL) was proposed [87]. A schematic diagram of the three-beam interferometer system is illustrated in Fig. 3a. The laser beam with a wavelength of 325 nm was first reflected by two dielectric mirrors (M1, M2), and then diverged by a spatial filter. Finally, the



**Figure 3** (a) Schematic diagram of the three-beam interferometer system. (b) Optical image of the three-beam Lloyd's mirror interferometer. (c) Simulated intensity distribution of the hexagonal interference pattern. (d) Schematic of the corresponding hexagonal PR hole template by 3BLIL with single exposure. (e) Top-view SEM image of the PR hole template in a period of 707 nm. (f) SEM image of ZnO NAs in a period of 707 nm. (g) Top-view SEM image of the ZnO NAs in a smaller period of 353 nm. Insets show the corresponding magnified SEM images; scale bars are 300 nm. Reprinted with permission from Ref. [87], Copyright 2013, the Royal Society of Chemistry.

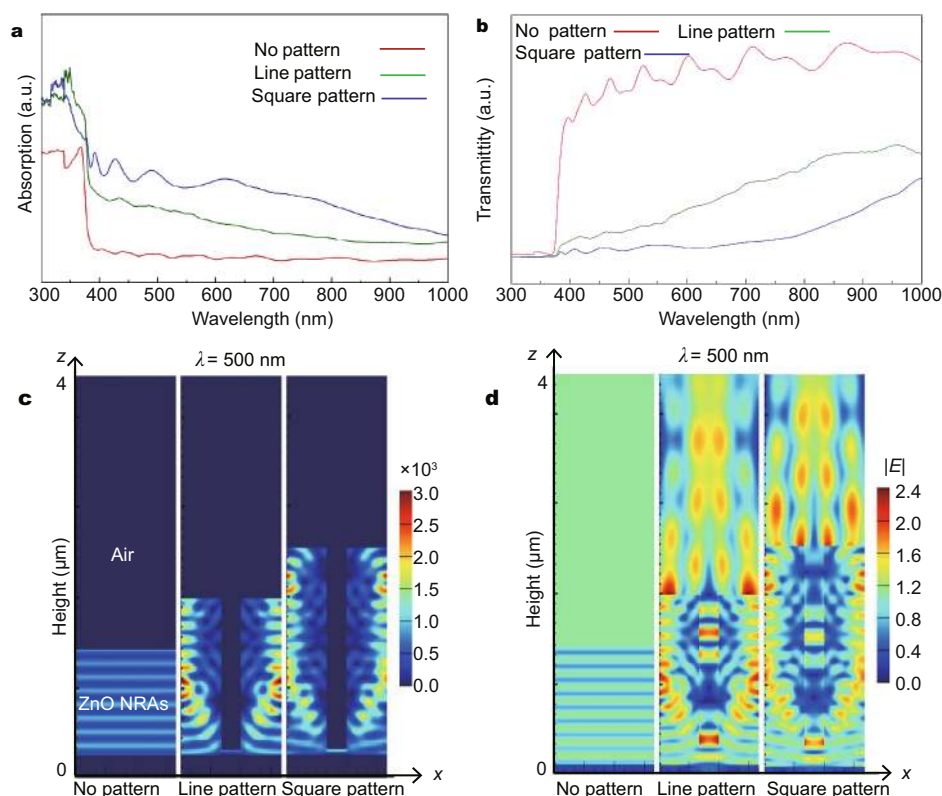
incident laser beam was divided into three parts by a three-beam Lloyd's mirror interferometer, which consists of a sample holder, two dielectric mirrors, and a rotating stage, as shown in Fig. 3b. These three laser beams formed a large diamond-shaped exposure area of approximately  $22 \text{ cm}^2$ , and the irradiated energy was recorded on the PR (Fig. 3d). The corresponding intensity distribution of the hexagonal interference pattern by 3BLIL is illustrated in Fig. 3c. 3BLIL simply fabricated the hole template and the period of ZnO NAs can be continuously controlled. Fig. 3e–g show patterned PR templates and highly ordered ZnO NAs with the different periods. The 3BLIL technique has two advantages compared with the 2BLIL technique. First, 3BLIL is a simple approach to fabricating PR hole templates with hexagonal symmetry through a single exposure without any sample rotation or multiple exposure processes. Second, through 3BLIL, the period of patterned ZnO NAs is successfully decreased from several wave-

lengths  $\lambda$  to  $2\lambda/3$ . Periodic regulation can be easily and precisely achieved by rotating the Lloyd's mirror interferometer.

The LIL technique has the advantages of being a maskless and simple process, which is suitable for the fast and large-area fabrication of patterned ZnO NAs. The pattern shape, period, diameter, and distribution can be flexibly controlled by tuning the exposure dosage, beam power, LIL angle, and exposure times. The LIL method is a powerful technique for producing large-area, periodically patterned ZnO NAs and shows great potential for material fabrication.

#### Optical properties of patterned ZnO NAs

The light absorption properties of ZnO NAs with no pattern, line pattern, and square pattern have been investigated [68,88]. Owing to the light scattering effect, the patterned ZnO NAs show wide absorbance bands com-



**Figure 4** (a) UV-vis-NIR absorption spectra, (b) UV-vis-NIR transmission spectra, (c) simulated cross-sectional optical absorption profiles, and (d) simulated cross-sectional electric-field intensity  $|E|$  distributions of the ZnO NAs with no pattern, line pattern, and square pattern, respectively. Simulated profiles used a 500-nm wavelength planar light from the bottom up. Reprinted with permission from Ref. [89], Copyright 2013, American Chemical Society.

pared with the no-pattern samples. The square-patterned ZnO NAs demonstrate the highest absorption among the three samples, as illustrated in Fig. 4a. This result is in good agreement with the transmission spectra in Fig. 4b. The square-patterned ZnO NAs have the maximum absorption and the minimum transmission, indicating an increased optical path length. Two-dimensional finite difference time domain (FDTD) simulations were also performed to understand the light absorption properties. The FDTD simulations allow solving Maxwell's time-dependent equations on a discrete spatial grid. This can be used to analyze the interaction of electromagnetic waves with complex structures in the UV, visible, infrared, terahertz, and microwave frequencies. Three simulation models with different length (1, 1.5, 2  $\mu\text{m}$ ) and space (0, 200, 200 nm) were established. The simulation regions were  $1 \mu\text{m} \times 4 \mu\text{m}$  in size. The boundary conditions were set as periodic boundary conditions and perfectly matched layer boundary conditions in the  $x$ -axis and  $y$ -axis, respectively. A plane-wave light source irradiated the

device from the substrate side [89]. As can be seen from Fig. 4c, the light absorption in the patterned ZnO NAs has been enhanced. This is consistent with the observation from the simulated electric-field intensity  $|E|$  distributions in Fig. 4d. It revealed that the propagation nature of the plane light in the patterned ZnO NAs was enhanced, because of the light reflection and interference in the periodic ZnO NAs. Additionally, the high-frequency structural simulator was used to obtain the optical properties by solving Maxwell's equations [90].

As mentioned above, patterned ZnO NAs with different sizes and periods have different optical properties. The highly controllable and uniform growth of ZnO NAs demonstrates a high application potential for energy devices.

### APPLICATIONS OF PATTERNED ZnO NAs IN ENERGY CONVERSION DEVICES

The fabrication of ZnO nanostructures with high light-harvesting capability and large charge-transfer efficiency

is important for highly efficient energy devices [91–96]. Patterned ZnO NAs with adjustable height and controllable period effectively promote light scattering and provide charge transport channels, which have broad application prospects in the field of energy devices [97]. In this review, we summarize the application of patterned ZnO *via* OL and LIL techniques for solar energy, water splitting, light emission devices, and nanogenerators.

### Photovoltaic devices

In order to address the global energy crisis, it is vital to exploit abundant, clean, and renewable energy to replace fossil fuels [98–100]. Photovoltaic (PV) technology is able to convert sun's energy to electricity through solar cells [101]. The device performance is closely related to the generation and separation of photogenerated charges [102–105]. To improve the performance of PV solar cells, it is important to fabricate photoanodes with high light-harvesting ability, direct electron transport channel, and efficient carrier collection [106–108].

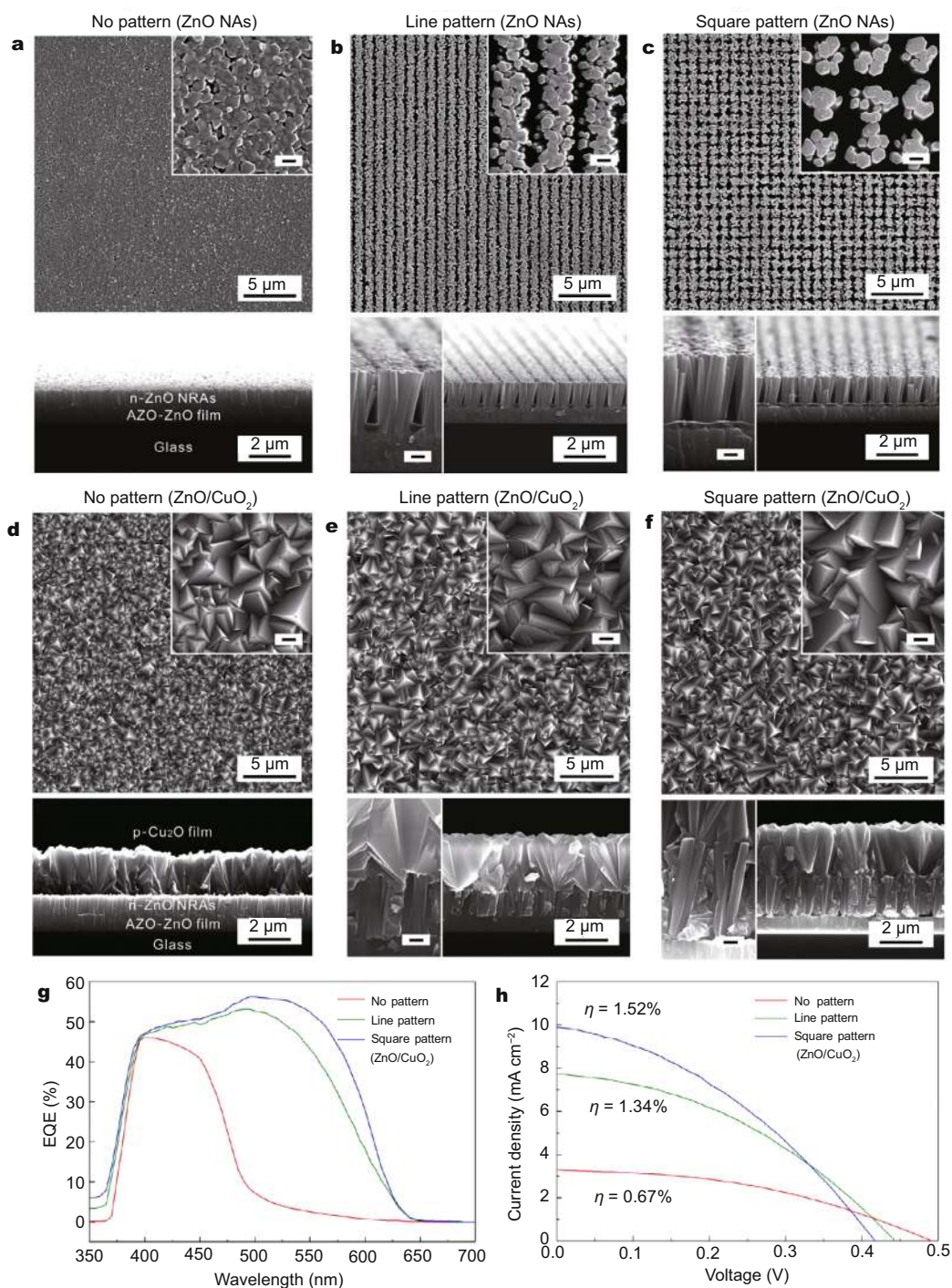
For dye-sensitized solar cells (DSSCs), the large-scale patterned ZnO–ZnS core-shell NAs were fabricated by using 2BLIL, hydrothermal synthesis, and chemical conversion synthesis methods for the first time [109]. First, a negative PR was spin-coated onto a substrate and the substrate was then annealed for 2 min. Second, the above substrates with top anti-reflective coating were exposed by 2BLIL with different patterns. Third, the patterned ZnO NAs were synthesized in mixed solution of zinc nitrate and hexamethylenetetramine. Finally, the ZnO NAs were immersed in thiacetamide solution to form the ZnS. Four exposure modes were used to obtain the PR template with line, long hexagonal, hexagonal, and square patterns. Compared with other patterned NAs, it is clear that the patterned ZnO–ZnS core-shell NAs with hexagonal symmetry have a large unit density and increased surface area. The UV-vis absorption spectra show that the NAs with hexagonal symmetry have the highest absorption. The enhanced light absorption ability is ascribed to the light scattering effect and increased amount of dye because of the enlarged surface area. The power conversion efficiency (PCE) of the DSSCs was enhanced from 1.11% to 2.09%. The NAs with hexagonal symmetry resulted in the highest PCE and circuit current density with approximately 88% and 60% enhancement. Thus, patterned NAs with tunable arrangement and density are beneficial for efficient DSSCs.

Patterned ZnO NAs were also applied in the ZnO/Cu<sub>2</sub>O heterojunction solar cells [89]. Generally, the contact area between ZnO and Cu<sub>2</sub>O and the carrier recombination in

the materials greatly limited the current density, which further reduced the performance of the solar cells [110–112]. The vertically aligned highly ordered ZnO NAs not only enhanced the light harvesting owing to the light scattering effect, but also increased the carrier collection efficiency owing to the increased heterojunction area [113]. Fig. 5a–c show the top-view and cross-section scanning electron microscopy (SEM) images of three kinds of ZnO NAs. The density clearly varies with the patterns of the NAs. The density of ZnO NAs without pattern has the maximum density of approximately  $3.8 \times 10^9$  rod cm<sup>-2</sup>. The density of the ZnO NAs with line and hole patterns are reduced to approximately  $2.7 \times 10^9$  and  $1.2 \times 10^9$  rod cm<sup>-2</sup>, respectively. The corresponding ZnO/Cu<sub>2</sub>O heterojunctions are shown in Fig. 5d–f. The Cu<sub>2</sub>O films based on the no pattern, line-patterned, and square-patterned ZnO NAs have different grain size, thickness, and heterojunction area. The ordered space of the patterned ZnO NAs was completely filled by Cu<sub>2</sub>O, which promotes the generation and separation of carriers in the cells [89]. The light absorption ability has been effectively enhanced owing to the strong light scattering effect. The light-harvesting capability will be further enhanced by increasing the ZnO periods. As illustrated in Fig. 5g, the external quantum efficiency of ZnO/Cu<sub>2</sub>O heterojunction cells based on patterned ZnO NAs was significantly increased at 380 nm and reached its peak value at 490 nm. It is clear that the heterojunction based on the square pattern has the highest light absorption. The PV characteristics of the ZnO/Cu<sub>2</sub>O solar cells are shown in Fig. 5h. The PCE of the ZnO/Cu<sub>2</sub>O heterojunction solar cell based on the square pattern was 2.3 times higher than the reference cells. In particular, the current density increased from 3.29 to 9.89 mA cm<sup>-2</sup>, which was attributed to the enhanced light harvesting and the efficient carrier collection. Obviously, the period-adjustable ZnO NAs can simultaneously enhance the light absorption and photo-carrier collection capability, which is beneficial for efficient solar cells.

### Photoelectrochemical devices

The photoelectrochemical (PEC) cell is one of the most promising devices for hydrogen generation, owing to the high theoretical PCE, low cost, and nonpolluting operation [114–119]. PEC cells have two key processes: light trapping and the separation and transport of photogenerated carriers [120,121]. In order to improve the light-harvesting capability, highly ordered square-patterned ZnO NAs were used as the PEC photoanode [122–124]. For no-patterned ZnO NAs, the substrate was di-



**Figure 5** Top and cross-sectional SEM images of the ZnO NAs with no pattern (a), line pattern (b), and square pattern (c). Top and cross-sectional SEM of the ZnO/Cu<sub>2</sub>O heterojunctions based on the ZnO NAs with no pattern (d), line pattern (e), and square pattern (f). Insets are the corresponding magnified SEM images with scale bars of 300 nm. External quantum efficiency spectra (g) and current density curves (h) of ZnO/Cu<sub>2</sub>O heterojunction cells based on ZnO NAs with different patterns. Reprinted with permission from Ref. [89], Copyright 2013, American Chemical Society.



rectly immersed in the solution with 2.5 mmol zinc nitrate, 2.5 mmol hexamethylenetetramine, and 100 mL deionized water at 95°C for 3 h. For the synthesis of square-patterned ZnO NAs, the PR hole template was first fabricated on the seed layer by the 2BLIL technique. The patterned ZnO NAs were then synthesized by the mentioned hydrothermal method. The patterned ZnO NAs exhibited light scattering properties and induced a large surface area, which simultaneously improved the light-trapping ability and the carrier transport efficiency. The solar to hydrogen conversion efficiency was increased to 0.18%, representing an improvement of 135% [125].

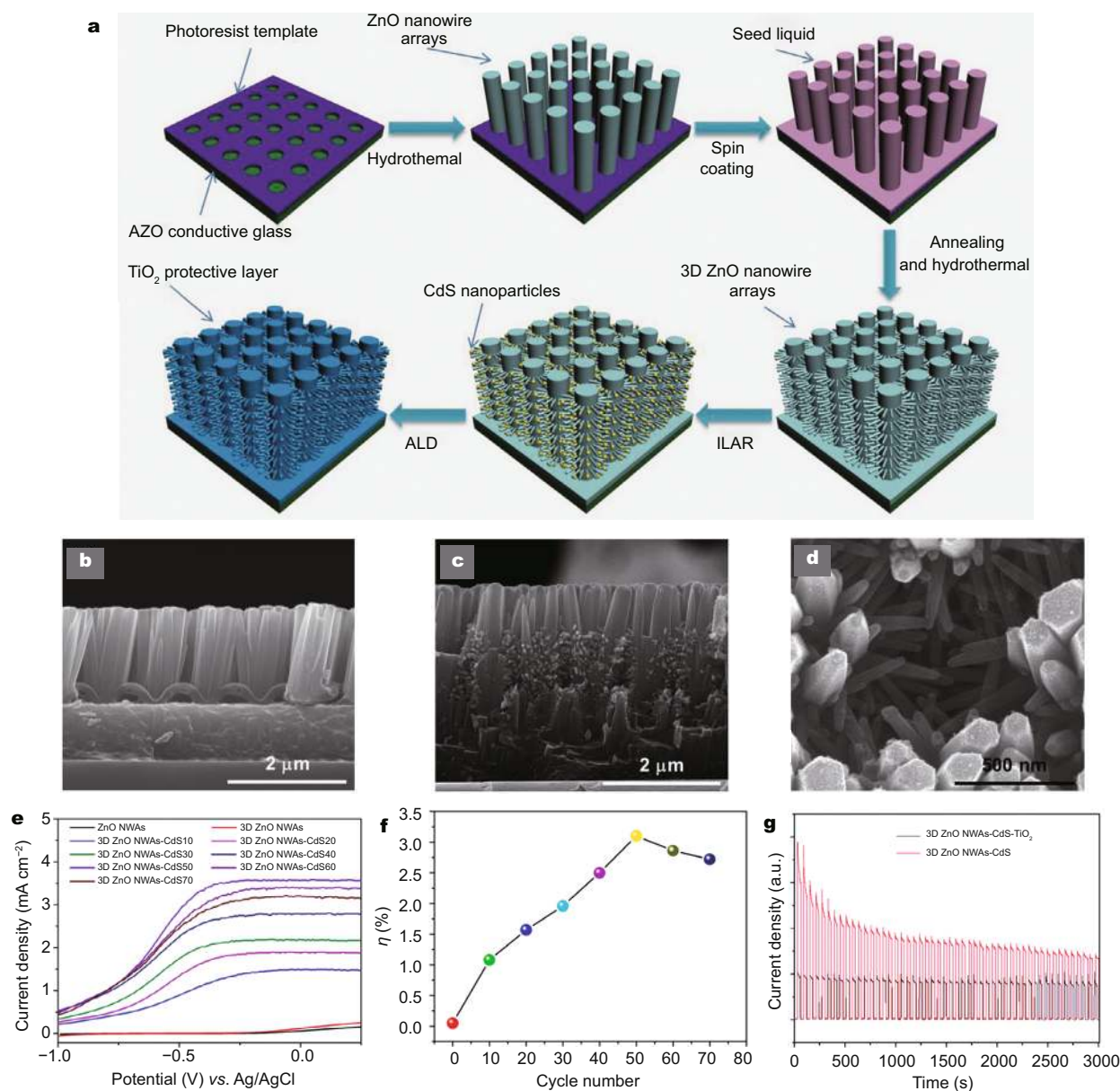
A tree-like three-dimensional (3D) nanowire structure based on patterned ZnO NAs was used as the photoanode, which lengthened the optical paths for light absorption and increased the surface areas for electrochemical reactions [126–129]. The fabrication process of 3D ZnO NAs-CdS is illustrated in Fig. 6a. A patterned PR template fabricated by the LIL technique was used to grow the ZnO NAs. The 3D ZnO NAs were then synthesized by secondary hydrothermal growth. CdS nanoparticles were deposited on the 3D ZnO NAs *via* the ionic layer adsorption reaction (ILAR) method. At last, a thin TiO<sub>2</sub> layer was coated on the above structure by atomic layer deposition (ALD). The CdS nanoparticles and thin TiO<sub>2</sub> layer were deposited onto the 3D structure to increase the light absorption and enhance the photo-corrosion resistance. Fig. 6b shows a cross-section SEM image of the patterned ZnO NAs with square symmetry. After the secondary synthesis, the branched ZnO nanowires grew on such patterned ZnO NAs, which contributed to the 3D ZnO nanostructure, as shown in Fig. 6c. Then, the CdS nanoparticles were deposited on the 3D structure. A top-view SEM image of the complex 3D ZnO NAs-CdS is shown in Fig. 6d. This structure has an enlarged roughness factor and improves the light-trapping capacity, which in turn enhances the device performance. Fig. 6e shows that the current density of the 3D ZnO NAs is approximately twice that of the ZnO NAs, owing to the enlarged surface area of the 3D structure. With the CdS layer, the photocurrent density exhibits a significant increase, because of the enhanced visible light absorption. The PCE of the 3D ZnO NAs-CdS devices varied with the deposition thickness of the CdS nanoparticles and achieved a maximum photon-to-hydrogen efficiency of 3.1% [130]. As shown in Fig. 6f, the introduction of the CdS nanoparticles boosts the conversion efficiency, and the highest efficiency is obtained with fifty deposition cycles. The loaded CdS effectively increased the light absorption in visible region. With the CdS nanoparticles,

type-II band alignment effectively promoted the carrier separation at the interface [130]. The patterned 3D ZnO NAs offer a large surface area, which is beneficial for loading other materials. Additionally, the tree-like structure provides direct electron transport channels for rapid photogenerated carrier separation and charge transport. Thus, the PEC efficiency was dramatically improved. With respect to device stability, the coated TiO<sub>2</sub> layer protected the photoanode from corrosion [131,132]. As shown in Fig. 6g, the current density of the photoanode with TiO<sub>2</sub> was maintained over 80% of the initial level. Using such 3D structures based on patterned ZnO NAs is considered as a promising protocol to design high-performance PEC devices.

### Light emission devices

Light emitting diodes (LEDs) are one of the most promising lighting technologies, and have attracted great attention owing to the demand for displays and illumination [133–135]. ZnO-based solid-state LEDs have been considered as a promising candidate for the next generation of high-efficiency light emitters [136–138]. However, the light extraction efficiency is low [139,140]. Many geometric structures have been used to enhance the light extraction. Periodically aligned ZnO NAs show a periodic modulation of dielectric constant, which could regulate the luminescence property [141]. In 2014, a LIL-patterned PR was used to grow flower-like ZnO NAs by two-phase hydrothermal growth [142]. The patterned ZnO hemisphere array was synthesized by two-step hydrothermal growth. First, a UV-O<sub>3</sub> treatment was performed on the LED substrates with the PR holes, which were then immersed in a mixture of zinc sulfate heptahydrate and ammonium chloride for 15 min to fabricate the flower-like ZnO NAs. Second, the structure was immersed in a solution of zinc sulfate heptahydrate, ammonium chloride and cadmium sulfate hydrate for a second growth. The two-step growth effectively enhanced the lateral directional growth of the ZnO NAs. Compared to a reference LED, the light output of the LED based on the patterned structure was enhanced by 20% at 20 mA. This elevated light extraction efficiency was attributed to the scattering effect of the patterned ZnO NAs.

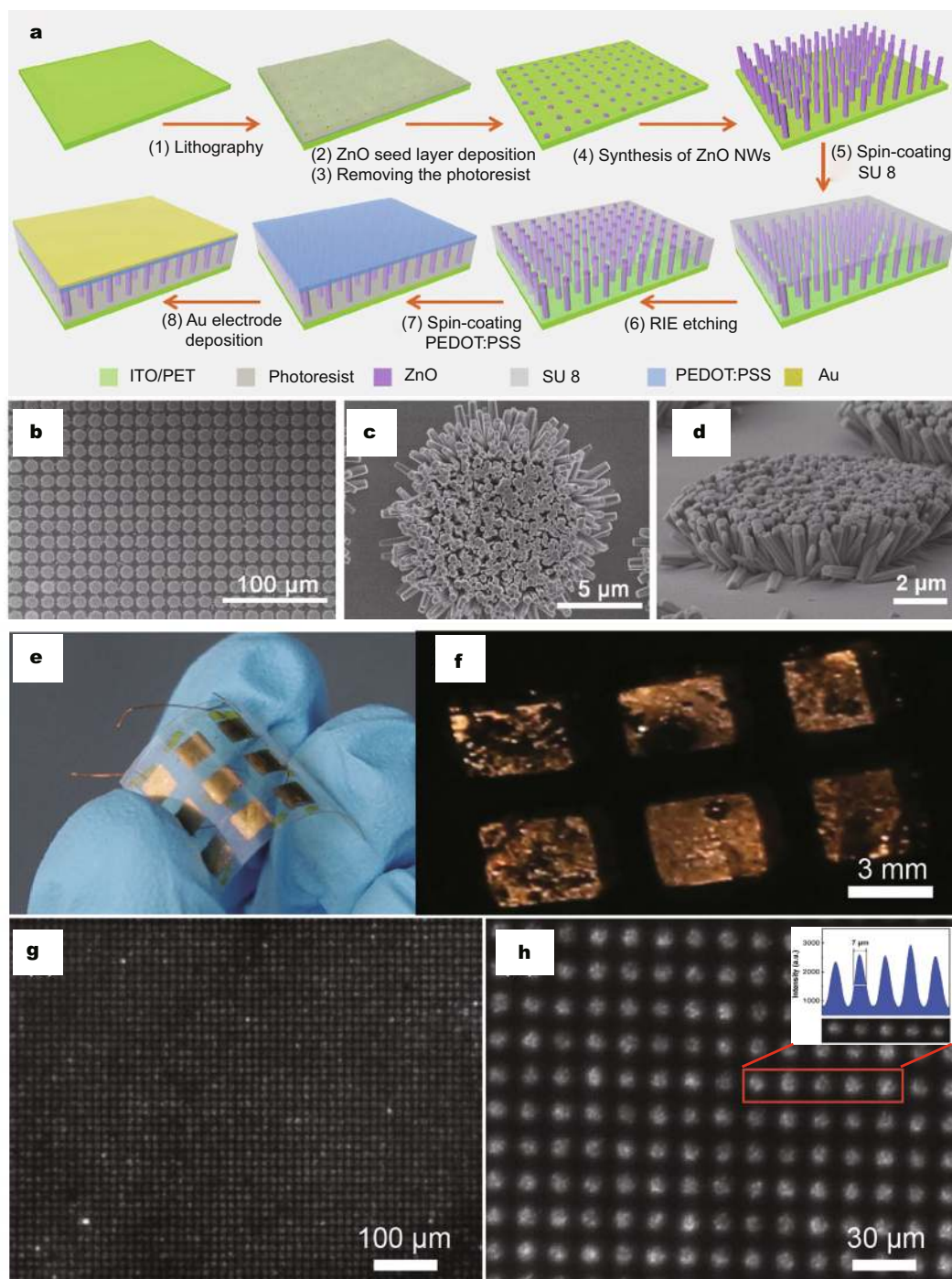
Generally, the spatial resolution of LEDs is related to the size of the lithography patterns. The patterned ZnO NAs with a diameter of 100 nm derived a resolution of 1.3 μm. Considering the cost, the pore pattern with a diameter of 5 μm was used in flexible LED devices, which yielded a spatial resolution of 7 μm [143]. As illustrated in Fig. 7a, a flexible indium tin oxide (ITO)/PET substrate



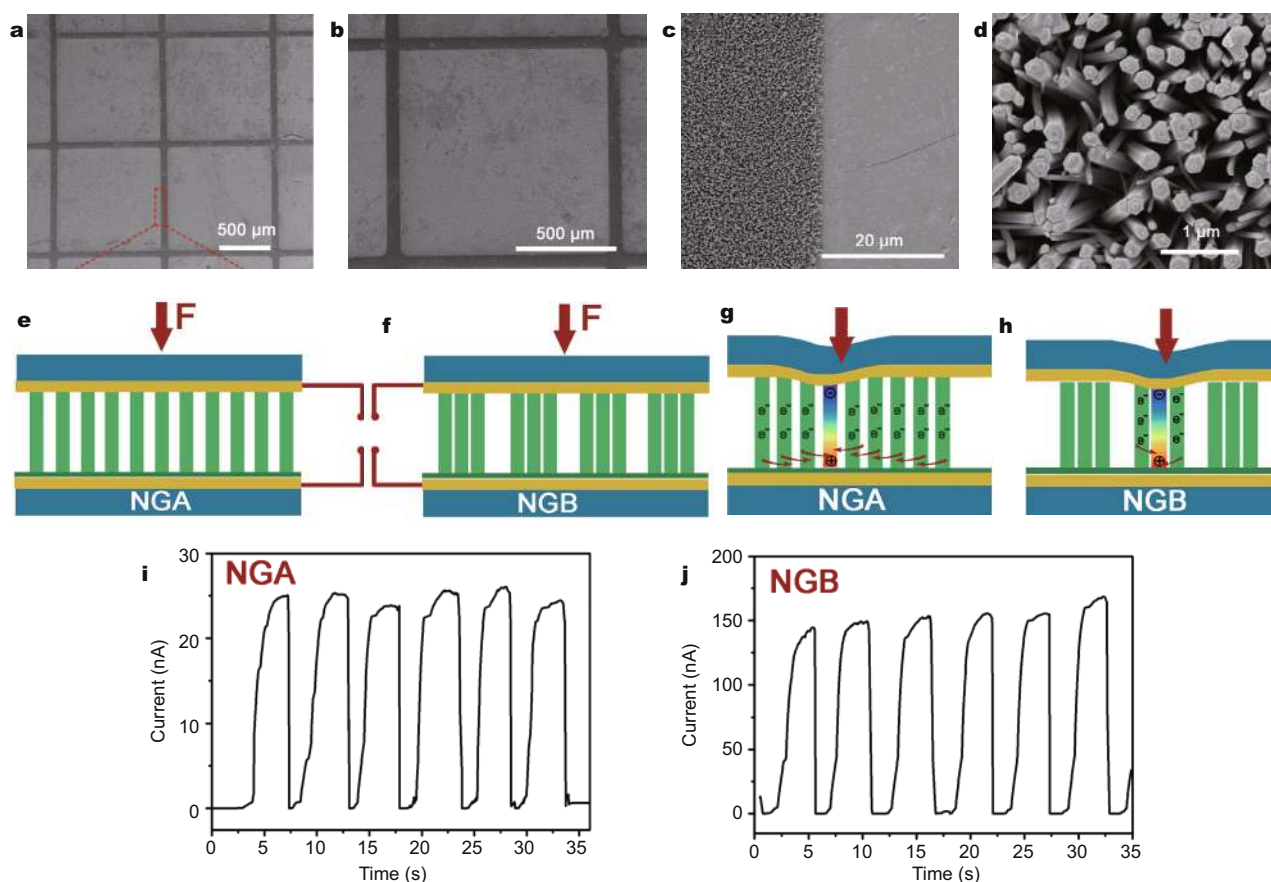
**Figure 6** (a) Schematic diagram of the 3D structure fabrication processes. (b) Cross-section SEM image of the patterned ZnO NAs and (c) 3D ZnO NAs. (d) Top-view SEM image of 3D ZnO NAs-CdS. (e) Current density of different ZnO structures with varied CdS deposition cycles. (f) Photon-to-hydrogen efficiency at different CdS deposition cycle numbers. (g) Stability test of the 3D ZnO NAs-CdS and 3D ZnO NAs-CdS-TiO<sub>2</sub> structures. Reprinted with permission from Ref. [130], Copyright 2015, WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim.

with PR was patterned by the OL technique and then the ZnO NAs were fabricated. The space between the ZnO NAs was infiltrated with SU8, but the heads of the ZnO NAs were exposed by oxygen plasma etching. Subsequently, poly(3,4-ethylenedioxythiophene)/polystyrene sulfonate (PEDOT:PSS) was spin-coated onto the ZnO NAs. Finally, an Au film was deposited as a positive electrode of the LED. The OL-patterned ZnO NAs are

shown in Fig. 7b–d. The diameter and length of the ZnO nanowires are 300 nm and 4 μm, respectively. Based on such patterned ZnO NAs, a flexible pressure mapping device was fabricated and the corresponding optical images are illustrated in Fig. 7e, f. The electroluminescence performance of the flexible LED based on patterned ZnO NAs is shown in Fig. 7g. The controllable fabrication of patterned ZnO NAs with various periods



**Figure 7** (a) Fabrication process of the patterned ZnO NWs with p-type polymer. (b-d) SEM images of patterned ZnO NWs with different magnification. (e, f) Optical images of the flexible pressure mapping device. (g, h) Optical images of ZnO NWs/p-type polymer LEDs. The inset image shows the five typical LEDs with their emission intensity. Reprinted with permission from Ref. [143], Copyright 2015, WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim.



**Figure 8** Images showing the surface morphologies of patterned ZnO NAs at low magnification (a, b) and high magnification (c, d). Schematic diagrams of the NGs based on the pristine ZnO NAs (e) and patterned ZnO NAs (f). Working mechanism diagrams (g, h) and current outputs (i, j) of the corresponding NGs. Reprinted with permission from Ref. [158], Copyright 2017, AIP Publishing LLC.

and sizes is beneficial for efficient ZnO-based LED devices. Therefore, the rational design of the pattern structure is essential to improve the efficiency of light extraction [144].

### Nanogenerator devices

Nanogenerators (NG) are among the most common devices used to convert mechanical energy produced by mechanical vibration, body movement, airflow, and hydraulic energy into electrical energy at the nanometer scale [145–148]. Vertically aligned ZnO NAs have been widely used in NG devices, by virtue of the piezoelectric effect [149–151]. However, the screening effect on the piezoelectric field induced by the free carriers in ZnO seriously inhibited the output performance of ZnO NAs-based NGs [152–155]. Several strategies have been proposed to reduce the surface free-charges on ZnO NAs, such as the introduction of ZnO p-n homojunctions [156] and doping with lithium [157]. Although the potential

screening effect was reduced, the free charges in the unstrained ZnO NAs drifted to neutralize the piezoelectric charges in the strained ZnO NAs, which seriously reduced the device performance. Patterned ZnO NAs provide independent working areas, which effectively inhibit the charge neutralization [158].

Patterned ZnO NAs were fabricated by the OL technique. The surface morphologies of the patterned ZnO NAs are illustrated in Fig. 8a–d. It is clearly evident that the ZnO NAs with a diameter of 50–100 nm were divided to squares with an interval of 20 μm. These patterned ZnO NAs were used to fabricate the NG, which showed a higher output current of 150 nA compared with the pristine ZnO NAs-based NG. Fig. 8e, f show that the space of the patterned ZnO NAs is larger than that of the pristine ZnO NAs. When a compressive strain is applied on the ZnO NAs, positive and negative potentials are produced at the stretched and compressed sides, respectively [159]. The electrons then flow from the ZnO NAs to

the electrode. However, the free charges within the unstrained ZnO NAs will neutralize the piezoelectric charges, as shown in Fig. 8g. The patterned ZnO NAs with independent working areas effectively reduced the screening effect and in turn optimized the output performance with an improvement of six times, as shown in Fig. 8h, j. Therefore, the patterned ZnO NAs with independent working areas provide an effective protocol to enhance the performance of NGs.

## CONCLUSIONS AND OUTLOOKS

ZnO has received increasing attention over the past few years because of its excellent properties. Various ZnO nanostructures have been widely used in nanoscale devices. For the improvement of device performances, the accurate control of ZnO structure is particularly important. Several synthetic strategies have been exploited to obtain highly ordered ZnO NAs. The detailed fabrication processes and the advantages of the OL and LIL techniques could offer platforms for the precise control of nanostructure fabrication. Furthermore, we highlight a range of excellent properties associated with different sizes and periods. The relationship between the patterned ZnO NAs and the material properties provides a guide for the selection of nanostructures. In terms of energy conversion applications, the successful integration of patterned ZnO NAs into solar energy, water splitting, light emission devices, and nanogenerators opens a new route to improve the device performances. It is of universal significance for further devotion in the energy conversion device evolution.

Overall, a comprehensive review of the controllable fabrication and functional applications of ZnO NAs by OL and LIL technologies is given. The use of patterning technology is an effective approach for fabricating large-scale highly ordered ZnO NAs with controllable period and diameter, and uniform distribution. Patterned ZnO NAs have direct electron pathways, excellent light-scattering, and enlarged surface area, which are conducive to enhanced light-harvesting ability and accelerated charge transfer in energy devices. Additionally, the segmentation of the patterned ZnO NAs significantly reduced the screening effect in ZnO-based NGs. The applications of the patterned ZnO NAs in energy devices such as solar energy, water splitting, LEDs, and NGs have been summarized. All of the advantages contribute to the enhanced performance of these energy devices. Therefore, the preparation of large-area, low-cost, and high-precision ZnO NAs based on OL and LIL technologies has laid a solid foundation for the development of efficient functional

nanodevices. Patterning technology provides an effective approach to fabricating highly efficient devices, which has huge market value and research significance.

Received 2 July 2017; accepted 25 August 2017;  
published online 5 September 2017

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**Acknowledgements** This work was supported by the National Key Research and Development Program of China (2013CB932602 and 2016YFA0202701), the Program of Introducing Talents of Discipline to Universities (B14003), the National Natural Science Foundation of China (51527802, 51232001, 51372020 and 51602020), Beijing Municipal Science & Technology Commission (Z151100003315021), and China Postdoctoral Science Foundation (2016M600039).

**Author contributions** Si H and Kang Z contributed equally to this work. Si H and Kang Z wrote the manuscript under the guidance of Zhang Y. Zhang Y, Liao Q and Zhang Z investigated and analyzed the literatures. Wang L and Zhang Y proposed and supervised the review designing, and contributed to the general discussion and article revision. All authors discussed and provided feedback on the manuscripts structure, arguments and conclusions.

**Conflict of interest** The authors declare that they have no conflict of interest.



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## 图案化ZnO纳米结构的设计调控及其在能量转换器件中的应用

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**摘要** ZnO作为典型的直接带隙宽禁带半导体材料具有丰富的形貌结构和独特的物理、化学性能, 被广泛应用于能源、信息技术、生物医学等领域. 目前纳米结构的精确设计与可控制备已成为改善ZnO基功能型纳器件性能的重要手段. 本论文介绍了利用图案化技术对ZnO纳米结构进行限域生长的技术手段, 重点综述了光刻技术和激光干涉模板法在精细ZnO纳米结构制备方面的研究进展, 及其在光伏电池、光电化学电池、发光器件和纳米发电机四种能量转换器件中的应用. 形貌结构可调的ZnO纳米结构具有分立的高精度空间纳米结构、增大的比表面积、提升的光子捕获能力, 在与其他材料复合时利于实现高效的载流子行为调控, 获得了高效的能量转换, 满足了不同功能型纳器件对材料结构的需求. 针对ZnO纳米结构精确设计所发展的一系列图案化技术对其他材料的复杂纳米结构可控制备具有重要的指导意义, 亦为功能型纳器件的进一步发展开辟了一个全新的途径.