# Desymmetrization of meso-2,5-Diallylpyrrolidinyl Ureas via Asymmetric Pd-Catalyzed Carboamination Reactions. Stereocontrolled Synthesis of Bicyclic Ureas 

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#### Abstract

The stereoselective synthesis of fused bicyclic ureas $\mathbf{8}$ is accomplished via enantioselective Pdcatalyzed desymmetrizing carboamination reactions of meso-2,5-diallylpyrroldinyl urea 7c. The reactions generate a $\mathrm{C}-\mathrm{N}$ bond, a $\mathrm{C}-\mathrm{C}$ bond, and afford products bearing three stereocenters with good diastereoselectivity ( $6-12: 1 \mathrm{dr}$ ) and enantioselectivity (up to $95: 5 \mathrm{er}$ ). The $N-$ ( $p$ chlorophenyl) group can be cleaved in good yield using a two step sequence. In addition, $\mathbf{8 c}$ was transformed to a tricyclic guanidine product using a four-step (two pot) procedure and was converted to 9-epi-batzelladine k in seven steps.


## Keywords

Desymmetrization; Heterocycles; Asymmetric Catalysis; Palladium; Stereoselective

Catalytic asymmetric desymmetrization reactions are powerful and efficient tools for the synthesis of chiral molecules. ${ }^{[i]}$ These transformations convert simple achiral substrates into complex enantioenriched products through the differentiation of two enantiotopic groups, and can generate complex structures bearing multiple stereocenters in a highly controlled fashion. As such, the development of asymmetric desymmetrization reactions that allow for the construction of important structural motifs is of considerable utility.

Tricyclic guanidines are an interesting class of compounds that could potentially be accessed via catalytic asymmetric desymmetrization reactions (Figure 1). These scaffolds are displayed in a wide variety of biologically active natural products, ${ }^{[i i]}$ including the batzelladine alkaloids ${ }^{[i i i]}$ (e.g. batzelladine $\mathrm{K}, \mathbf{1}$ ), ${ }^{[\text {[iic] }]}$ the merobatzelladine alkaloids (e.g., merobatzelladine B, 2), ${ }^{[\mathrm{iv]}}$ and the crambescidin alkaloids ${ }^{[\mathrm{v}]}$ (e.g., crambescidin 359, 3). ${ }^{[\mathrm{vb}]}$ Many synthetic routes to these compounds involve the generation of a fused-bicyclic urea or guanidine derivative (e.g., 4), which is then transformed to the tricyclic guanidine in subsequent steps. ${ }^{\text {[vi,vii,viii] }}$ As such, development of a concise asymmetric synthesis of 4 could provide access to a broad array of interesting alkaloids.

[^0]We recently reported an asymmetric synthesis of the tricyclic guanidine natural product (+)merobatzelladine B(2), which featured a new strategy for the construction of bicyclic ureas and polycyclic guanidines via Pd-catalyzed carboamination reactions of enantiomerically enriched 2-allylpyrrolidine-1-carboxamide derivatives 5 (Scheme 1). ${ }^{[v i i i]}$ These reactions provided bicyclic urea products $\mathbf{6}$ in good yield with excellent diastereoselectivity, but control of absolute stereochemistry required the chiral-auxiliary mediated introduction of the C2 stereocenter during the fairly lengthy asymmetric synthesis of 5 (7-9 steps). ${ }^{[i x]}$

A potentially more attractive route to enantiomerically enriched bicyclic ureas and related bi- and tricyclic guanidines would involve the asymmetric Pd-catalyzed desymmetrization of meso-2,5-diallylpyrrolidnyl urea 7. This approach would allow for facile introduction of different $\mathrm{R}^{1}$-substituents, and the alkene present in product $\mathbf{8}$ provides a convenient handle for further elaboration to tricyclic guanidine products or more highly substituted urea derivatives. In addition, the meso substrate 7 can be prepared in only four steps. Our preliminary studies in this area are described in this communication. These transformations represent the first examples of asymmetric desymmetrizations of bis-alkene substrates in intermolecular Pd-catalyzed alkene carboamination reactions, and also the first examples of six-membered ring formation in an asymmetric Pd-catalyzed alkene carboamination. ${ }^{[x, x i, x i i]}$

In initial experiments we elected to employ a catalyst composed of $\mathrm{Pd}_{2}(\mathrm{dba})_{3} /(S)$-Siphos$\mathrm{PE}^{[\text {xiii] }}$ for desymmetrization reactions of 7 , as we previously illustrated this complex provides good results in related asymmetric carboamination reactions of simple $N$-allyl urea derivatives. ${ }^{[x i v, x v]}$ We decided to first optimize the structure of the urea $N$-aryl group, as prior studies in our lab suggested this group may have a significant influence on the level of asymmetric induction. ${ }^{[\mathrm{xiv}]}$ Thus, we explored the coupling of $Z-1$-bromobutene ${ }^{[\mathrm{xvi}]}$ with ureas 7 bearing different N -aryl substituents. As shown in Table 1, the use of electron-poor $p$-cyanophenyl or $p$-nitrophenyl $N$-aryl groups resulted in the formation of products $\mathbf{8}$ with the highest levels of both diastereoselectivity and enantioselectivity. However, these electron-poor substrates were transformed in modest chemical yield due to competing cleavage of the urea moiety (entries 5-6). Use of the electron-rich p-methoxyphenyl group led to improved yields but with lower levels of stereocontrol. After some exploration we found that a substrate bearing a p-chlorophenyl group was transformed to the desired product with both good chemical yield and stereoselectivity (entry 3). [xvii]

As shown in Table 2, the asymmetric desymmetrization reactions of 7c are effective with a number of different alkenyl and aryl bromide electrophiles. The main side products generated in these reactions were cis-2,5-diallylpyrrolidine (resulting from competing urea cleavage) and an unsaturated bicyclic urea that is generated by competing Ghydride elimination of an intermediate alkylpalladium complex. ${ }^{[x v i i i]}$ In the reaction of $7 \mathbf{c}$ with $E$-1bromohexene a regioisomeric side product bearing a 2-hex-1-enyl group was also generated. ${ }^{[x i x]}$

The best enantioselectivities were obtained when either alkenyl bromides, electron-rich aryl bromides, or electron-neutral aryl bromides were employed as substrates.
Diastereoselectivities were generally higher with the alkenyl electrophiles than with aryl electrophiles. Use of sterically hindered aryl bromides (entries 14-15) or electron-poor aryl bromides (entries 9, 11, and 13) led to lower diastereo- and enantioselectivities. Selectivities improved when NaOMe was used in place of $\mathrm{NaO}^{t} \mathrm{Bu}$ in reactions of electron-poor aryl bromides (entries 10 and 12), although yields decreased in these cases.

To further demonstrate the utility of the asymmetric desymmetrization reactions, we examined the deprotection of $\mathbf{8 c}$ and the conversion of $\mathbf{8 c}$ to tricyclic guanidine derivatives. As shown in equation 1, cleavage of the $N$ - $p$-chlorophenyl group can be accomplished via

Pd-catalyzed $N$-arylation with acetamide ${ }^{[x x, x x i]}$ followed by oxidation of the resulting $N$ aryl amide with ceric ammonium nitrate. This sequence afforded 9 in $65 \%$ yield over two steps.


The conversion of $\mathbf{8 c}$ to tricyclic guanidine $\mathbf{1 2}$ was carried out as shown in Scheme 2. Treatment of $\mathbf{8 c}$ with $\mathrm{POCl}_{3}$ followed by $\mathrm{NH}_{3}$ provided bicyclic guanidine $\mathbf{1 0}$ in $78 \%$ yield. Wacker oxidation of $\mathbf{1 0}$ afforded hemiaminal 11, which was then transformed to tricyclic product $\mathbf{1 2}$ in $70 \%$ yield with $5: 1 \mathrm{dr}$ via reductive amination with $\mathrm{NaBH}_{3} \mathrm{CN}$. ${ }^{[x x i i]}$ Overall, the synthesis of $\mathbf{1 2}$, which is structurally related to the batzelladine and merobatzelladine alkaloids, was accomplished in 5 steps and $41 \%$ yield from meso-2,5-diallylpyrrolidinyl urea $7 \mathbf{7 c}$. In addition, this is the first example of a Wacker oxidation/ring-closure sequence to generate a tricyclic guanidine. ${ }^{\text {[xxiii,xxiv] }}$

Finally, 8c was converted to tricyclic guanidine 16, which is an unnatural stereoisomer of batzelladine k, ${ }^{[x x v, x x v i]}$ as shown in Scheme 3. To avoid problems with base-mediated epimerization of the C 4 stereocenter, the Pd-catalyzed N -arylation with acetamide was carried out prior to Wacker oxidation of the alkene. This two-step sequence provided $\mathbf{1 3}$ in $65 \%$ yield. Reduction of the alkene followed by CAN deprotection generated urea 14, which was converted to guanidine aminal 15 by $O$-methylation and treatment with ammonia. ${ }^{\text {[xxvii] }}$ The reduction of $\mathbf{1 5}$ proceeded with modest diastereoselectivity ( $3: 1 \mathrm{dr}$ ), but upon purification 9-epi-batzelladine k was isolated as a single stereoisomer in $48 \%$ yield over three steps from 14.

In conclusion we have developed a concise route to enantiomerically enriched bicyclic ureas via Pd-catalyzed desymmetrizing carboamination reactions of meso-diallylpyrroldinyl ureas. These transformations effect formation of both a $\mathrm{C}-\mathrm{N}$ and a $\mathrm{C}-\mathrm{C}$ bond, and provide products bearing three stereocenters with good levels of diasteroselectivity and enantioselectivity. These reactions illustrate the potential utility of asymmetric Pd-catalyzed alkene carboamination for desymmetrization processes and provide synthetically valuable products in a straightforward manner. Further exploration of enantioselective Pd-catalyzed alkene difunctionalization reactions are currently underway.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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xviii. In some instances (Table 2, entries 2, 3, and 5) modest yields were due to product losses during the chromatographic separation of diastereomers.
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1 Batzelladine K


3 Crambescidin 359



Figure 1.
Bioactive guanidine alkaloids prepared from bicyclic urea precursors

## Prior approach - substrate controlled cyclization



This work - catalytic asymmetric desymmetrization


Scheme 1.
Synthesis of bicyclic ureas via Pd-catalyzed asymmetric desymmetrization



Scheme 2.
Conversion of $\mathbf{8 c}$ to a tricyclic guanidine derivative



## Table 1

$N$-Aryl group effects. ${ }^{[a]}$

|  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Ar | Yield(\%) ${ }^{[b]}$ | dr ${ }^{[c]}$ | er |
| 1 | 4-MeOC $\mathrm{C}_{6} \mathrm{H}_{4}$ | 65 | 7:1 | 86:14 |
| 2 | $3,4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | 41 | 7:1 | 82:18 |
| 3 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | 76 | 12:1 (20:1) ${ }^{[d]}$ | 95:5 |
| 4 | $4-\mathrm{BrC}_{6} \mathrm{H}_{4}$ | $12^{[e]}$ | 18:1 | 94:6 |
| 5 | $4-\mathrm{CNC}_{6} \mathrm{H}_{4}$ | $40^{[e]}$ | 17:1 | 95:5 |
| 6 | $4-\mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4}$ | $22^{[f]}$ | 20:1 | 96:4 |

${ }^{[a]}$ Conditions: 1.0 equiv substrate, 1.5 equiv ( $Z$ )-1-bromobutene, 1.5 equiv $\mathrm{NaO}{ }^{t_{\mathrm{Bu}}}, 2 \mathrm{~mol} \% \mathrm{Pd}_{2}(\mathrm{dba})_{3}, 8 \mathrm{~mol} \%(S)$-Siphos-PE, Toluene ( 0.2 M ), $100^{\circ} \mathrm{C}, 2 \mathrm{~h}$.
${ }^{[b]}$ Isolated yield (average of two or more runs).
${ }^{[c]}$ Diastereomeric ratio of the pure isolated material. Diastereomeric ratios of the isolated materials were identical to those of the crude products except for entry 3 .
${ }^{[d]}$ The diastereomeric ratio of the crude material was $12: 1$. The product was isolated in $76 \%$ yield with $20: 1 \mathrm{dr}$.
${ }^{[e]}$ This material contained a small amount of the corresponding aniline derivative.
$[f]$ The reaction was conducted at $120^{\circ} \mathrm{C}$ for 16 h . The isolated material contained ca. $8 \%$ of unreacted substrate.
Table 2
${ }^{[a]}$ Conditions: 1.0 equiv substrate, 1.5 equiv R-Br, 1.5 equiv NaO ${ }^{t_{\mathrm{Bu}}}, 2 \mathrm{~mol} \% \mathrm{Pd}_{2}(\mathrm{dba}) 3,8 \mathrm{~mol} \%(S)$-Siphos-PE, Toluene $(0.2 \mathrm{M}), 100{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}$.
${ }^{[b]}$ Isolated yield (average of two or more runs).
${ }^{[c]}$ Diastereomeric ratio of the pure isolated material. Diastereomeric ratios of the isolated materials were identical to those of the crude products unless otherwise noted.
${ }^{[d}{ }^{\text {The }}$ The diastereomeric ratio of the crude material was $10-12: 1$.
${ }^{[e]}$ The diastereomeric ratio of the crude material was $6: 1$


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    Supporting information for this article is available on the WWW under http://www.angewandte.org or from the author.

