

Determining the Chemical Compositions of Garlic Plant and its Existing Active Element

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Abstract: In this research, we investigate the process of producing allicin in garlic. With regard to the chemical compositions of garlic (*Allium Sativum L.*), allicin is among the active sulfuric materials in garlic that has a lot of benefits such as anti-bacterial, anti-oxidant and deradicalizing properties.

Keywords: the chemical compositions, Garlic, Allicin

I. Introduction

Alliin was first reported in 1944 by Cavallito and Bailey, who described it as a colorless oil with low solubility in water that was relatively unstable. Alliin derives from the precursor alliin through the enzymatic activity of alliinase. According to Murray, Garlic is one of the most widely used medicinal plants since ancient times. It was used by the Egyptians, Greeks and Romans.

The researches performed on garlic plant represent that the key element and composition of garlic is constituted of alliin element. Name of garlic is most probably extracted from celtic i.e. harsh.

Amino acid alliin is an alkyl amino extracted of Cysteine Sulfoxide. Cutting or extracting garlic frees its alliinase enzyme and rapidly turns it into a Sulfonic acid named allicin or de allyl thiosulfinate. If another insect or animal reaches garlic, just by cracking its skin, they are forced to go back because of the taste and smell of allicin and its toxic effect. Allicin is a strong insecticide.

Antibacterial effect of garlic because of different materials such as Alliin, Ahoin, allicin and Allistain 1 and 2, and for this reason is used for gram-positive antibacterial and gram-negative antibacterial, fungi, parasites and viruses. The best way to extract allicin from garlic is cutting and smashing garlic and putting it in water and alcohol for several times (3). The composition of water and garlic extract can be used in a form of spray to repel insects.

Researchers have found in their researches that the chemical contents of garlic having allicin, de allyl de sulfide and de allyl trisulfide, all have volatile oils and s-allyl Cysteine (SAC) and these compositions include sulfur element.

Sulfur compositions are the factor of creating bad smell in garlic. Allicin is very unstable and after creating harsh smell of garlic is converted to other sulfur compositions including diallyl disulfides, methyl allyl sulfides and dimethyl sulfides and diins and Ajoieins. Garlic has at least 100 types of sulfuric compositions and 70-80% of them are constituted of allicins (2).

4 g of fresh garlic contains approximately 10 mg de alliin and can be converted into at least 4 mg of allicin. Alliin and alliinase are located in separate regions of garlic cloves, and the alliinase reaction is initiated only after the cells have been crushed. Allicin is completely formed in 0.3 min at 310 K and, at room temperature, its half-life ranges from 10 days to few hours depending in the solvent in which it is dissolved. Despite allicin being a low-polarity molecule, it is often extracted using polar solvents, such as water and ethanol, at room pressure (0.1 MPa) because it is very unstable in non-polar organic solvents (10). Fujisawa et al..

Fresh Garlic cloves contain about 2-6 mg/g of γ -glutamyl-S-allylcysteine (0.2 – 0.6 % fresh weight) and 6-14 mg/g of alliin (0.6 – 14 % fresh weight). Garlic cloves yield about 2.500 – 4.500 mg of allicin per gram of fresh weight when crushed. One fresh garlic clove weighs 2 – 4 g.

Non-fresh no-alliin garlic, 1-7g consumption per day is also having the benefits of fresh garlic. The aim of this research is to determine the chemical compositions of garlic plant and investigate the process of producing allicin of garlic.

THE CHEMISTRY OF GARLIC

WHAT CAUSES GARLIC BREATH?

There are four major volatile organic compounds responsible for 'garlic breath': diallyl disulfide, allyl methyl sulfide, allyl mercaptan, and allyl methyl disulfide. None of these compounds are present in garlic until it is crushed or chopped.


C=CC(=O)S=C
ALLICIN

C=CCSC
ALLYL METHYL SULFIDE

When garlic is mechanically damaged, enzymes convert the compound alliin to alliin (which gives chopped garlic its aroma). This is broken down further into the afore-mentioned volatile compounds.

Allyl methyl sulfide is broken down in the body more slowly than the other three compounds, so it is the primary volatile responsible for garlic breath. It is excreted via sweating, breathing, and through the urine, and its effects can last up to a day!

A few foods have been shown to have some effect on reducing garlic breath, including parsley & milk.



GARLIC'S ANTIBACTERIAL PROPERTIES

C=CCSC
DIALLYL SULFIDE

C=CCSCC=C
DIALLYL DISULFIDE

C=CCSCC=C
DIALLYL TRISULFIDE

Sulfur-containing organic compounds give garlic antibacterial properties. Antimicrobial effects have been shown to increase as the number of sulfur atoms in the compounds increases.

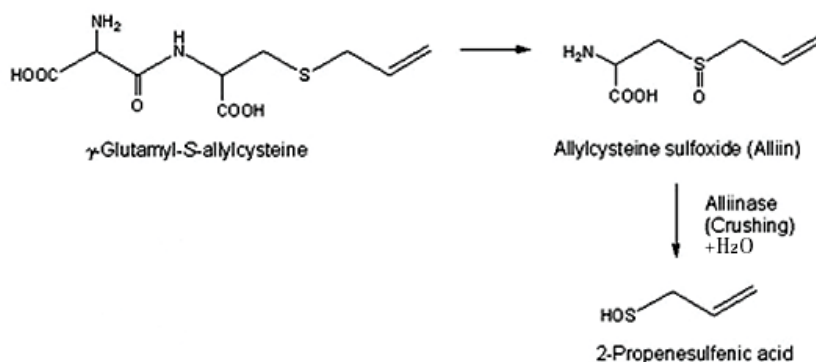
The organosulfur compounds can penetrate the cell membranes of bacteria cells, and combine with certain enzymes or proteins to alter their structure, injuring the cells. Alliin, formed initially when garlic is crushed, also has antibacterial properties.

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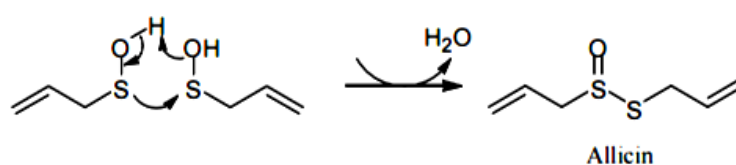
Table 1. Scientific Classification Garlic

Kingdom:	Plantae
Clade:	Angiosperms
Clade:	Monocots
Order:	Asparagales
Family:	Amaryllidaceae
Subfamily:	Allioideae
Tribe:	Allieae
Genus:	Allium
Species:	A. Sativum
Scientific noun:	Allium Sativum L.

Smashing garlic grain releases some enzymes called alliinase that converts sulfoxide materials due to sulfonic acid. Then, these acids are converted to aromatic compounds such as alliin and dimmer by losing water .



Allyl Sulfenic acid is unstable and very reactive at room temperature. With the elimination of water, two molecules allyl Sulfenic acid condenses spontaneously to alliin.



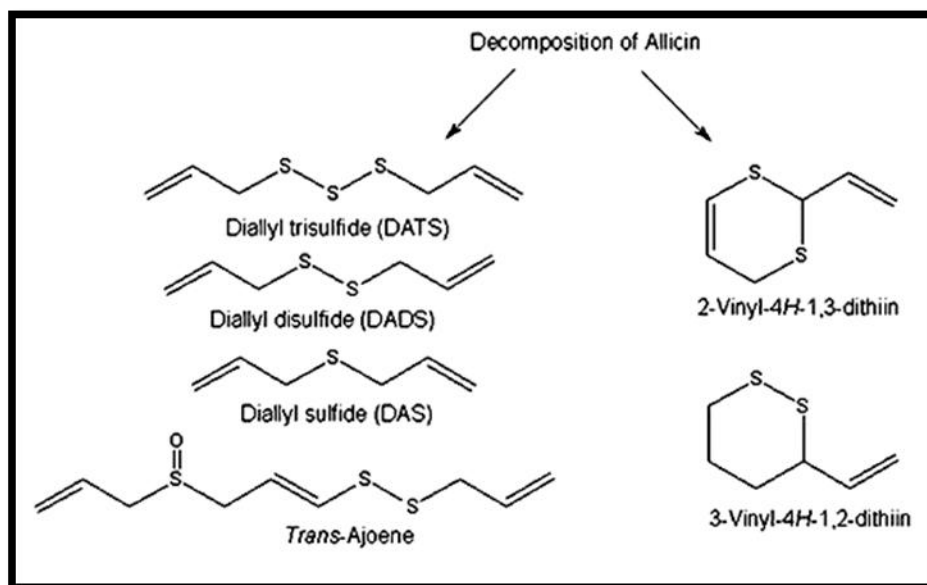


Figure 3: some organosulfur Compounds Derived from Garlic

II. Components of garlic

- Amino acids: Glutamic acid, arginine, aspartic acid, leucine, lysine, valine ... etc
- Minerals: Mainly: manganese, potassium, calcium, phosphorus and in minor quantities: magnesium, selenium. Sodium, iron, zinc and copper.
- Vitamins: Mainly: vitamin B6, also Vitamin C and in minor quantities, folic acid, panthotenic acid and niacin.
- Essential oil with many sulphur-containing components: allyl disulfide, allyl trisulfide, etc
- Alliin, which, by means of enzyme alliinase, is converted into allicin.
- Ajoene, produced by allicin condensation.
- Quercetin.
- Sugars: fructose and glucose.

Table 2. Nutritional composition of Garlic

Nutritional composition of garlic 100 gr	
Water (g)	59
Calories (kcal)	149
Lipids (g)	0,5
Carbohydrates (g)	33,07
Fiber (g)	2,1
Manganese (mg)	1672
Potassium (mg)	401
Sulphur (mg)	70
Calcium (mg)	181
Phosphorus (mg)	153
Magnesium (mg)	25
Sodium (mg)	17
Vitamin B6 (mg)	1235 ↑
Vitamin C (mg)	31
Glutamic acid (g)	0,805
Arginine (g)	0,634
Aspartic acid (g)	0,489
Leucine (g)	0,308
Lysine (g)	0,273

III. Conclusion

In this research, we investigate the chemical compositions in garlic and also the process of producing alliin inside garlic. With regard to the chemical compositions obtained of garlic, it can be concluded that sulfuric compositions are among active and applicable elements in garlic and alliin is one important sulfuric composition in garlic.

The alliin inside garlic having anti-oxidant feature is useful for human diseases and has antibacterial properties that is used for repelling insects and/or removing insects and is used as one plant agrochemicals as pesticides that is safe for human and environment.

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