

Direct Protection Against Acetaminophen Hepatotoxicity by Propylthiouracil

IN VIVO AND IN VITRO STUDIES IN RATS AND MICE

TADATAKA YAMADA, SHELLY LUDWIG, JOHN KUHLENKAMP, and NEIL KAPLOWITZ,
*Gastroenterology Section, Medicine and Research Services, Veterans
Administration Wadsworth Medical Center, and University of California, Los
Angeles, California 90073*

ABSTRACT Hepatotoxicity caused by acetaminophen can be prevented by enzyme-catalyzed conjugation of its reactive metabolite with glutathione (GSH). Since we have shown in previous studies that 6-*N*-propyl-2-thiouracil (PTU) can substitute for GSH as a substrate for the GSH *S*-transferases, we examined the possibility that PTU might also protect against acetaminophen hepatotoxicity by direct chemical interaction with the reactive metabolite of acetaminophen. In an *in vitro* system consisting of [³H]acetaminophen, liver microsomes from phenobarbital-pretreated rats, and an NADPH-generating system, we found that PTU had a dose-dependent additive effect with GSH on inhibition of acetaminophen covalent binding. PTU administration also resulted in a dose-dependent decrease in both GSH depletion and covalent binding *in vivo* in acetaminophen-treated mice. To examine the possible mechanisms by which PTU exerts its protective effect, we studied the action of PTU on both acetaminophen conjugation and metabolic activation. PTU had no effect upon acetaminophen pharmacokinetics in phenobarbital-pretreated rats, as examined by measuring acetaminophen concentration in bile, urine, and blood after an intraperitoneal dose, nor did it alter the total amount of polar conjugates formed. Microsomes from PTU-treated rats were unaltered in cytochrome P-450 concentrations and *p*-nitroanisole-*O*-demethylase, benzo- α -pyrene hydroxylase, and cytochrome *c*-reductase activities. Furthermore PTU did not decrease acetaminophen-GSH adduct formation *in vitro*, sug-

gesting that there was no reduction in drug activation. However, in bile from [³⁵S]PTU and [³H]acetaminophen treated rats, as well as in incubates of the two drugs with liver microsomes, a new ³⁵S- and ³H-containing product could be identified. By both thin layer chromatography and high pressure liquid chromatography this new product, which co-eluted with [³H]acetaminophen, was separated from unreacted [³⁵S]-PTU. The formation of this product *in vitro* was a function of PTU concentration and reached a maximum of 0.06 μ mol/min per mg protein at 0.5 mM PTU. *In vivo*, the total biliary excretion of this product over 4 h (116 nmol) equaled the net reduction in acetaminophen metabolite covalent binding in the liver of phenobarbital-pretreated rats (108 nmol). We conclude that PTU, independent of its antithyroid effect, diminishes hepatic macromolecular covalent binding of acetaminophen reactive metabolite both *in vivo* and *in vitro*, and it does so by detoxifying the reactive metabolite through direct chemical interaction in a manner similar to GSH. These observations may define the mechanism by which PTU is protective against liver injury caused by acetaminophen.

INTRODUCTION

In recent years, acetaminophen hepatotoxicity has become a clinical problem of increasing frequency (1). The biochemical mechanism by which acetaminophen induces liver injury has been elucidated in detail (2-5) and has served as an important model in understanding the mechanisms of a wide variety of toxic drug hepatopathies. Although the bulk of ingested acetaminophen is metabolized in the liver by sulfation or glucuronidation to polar nontoxic metabolites (6), a minor metabolic pathway involves oxidation by cytochrome P-450 mixed function oxidases, resulting in the formation of a reactive metabolite which is detoxified

This work was presented in part to the American Association for the Study of Liver Diseases, 23 May 1979, New Orleans, La.

Dr. Yamada is a recipient of a Veterans Administration Research Associate Award.

Received for publication 15 April 1980 and in revised form 12 November 1980.

by glutathione (GSH).¹ The conjugation of this reactive metabolite of acetaminophen by GSH appears to be catalysed by an enzyme from liver cytosol, possibly a glutathione S-transferase (7). In the event of GSH depletion, such as might occur after a massive overdose of acetaminophen, the reactive metabolite may bind covalently to hepatic macromolecules with resultant hepatocellular necrosis. Treatment of acetaminophen hepatotoxicity has involved, therefore, the administration of GSH precursors such as cysteamine (8) and *N*-acetylcysteine (9).

6-*N*-Propyl-2-thiouracil (PTU) also has been shown to be protective against acetaminophen-induced liver necrosis (10) in addition to numerous other forms of experimental (11, 12) and clinical (13) toxic hepatopathy. The mechanism of this protection has been ascribed to the drug's antithyroid properties. Recently, however, we have found that PTU can substitute for GSH as a substrate in glutathione S-transferase catalyzed reactions (14). This observation has prompted us to study whether PTU might protect against acetaminophen hepatotoxicity through a direct biochemical mechanism rather than indirectly through its antithyroid activity. Our results suggest that PTU protects by forming a conjugate with the reactive metabolite of acetaminophen, much in the same manner as GSH.

METHODS

Chemicals. NADP, glucose-6-phosphate, glucose-6-phosphate dehydrogenase, GSH, acetaminophen, and benzo- α -pyrene were obtained from Sigma Chemical Co. (St. Louis, Mo.). PTU was a gift of Eli Lilly and Co. (Indianapolis, Ind.). *p*-Nitroanisole was purchased from Aldrich Chemical Co., Inc. (Milwaukee, Wis.). [³H]acetaminophen (hydroxyacetanilide, *p*-[³H(G)], 3.54 Ci/mmol) was obtained from New England Nuclear (Boston, Mass.) and [³⁵S]PTU (6-*N*-propyl-2-[³⁵S]thiouracil, 109 mCi/mmol) from Amersham Corp. (Arlington Heights, Ill.). All other chemicals used in this study were readily available commercial products.

***In vitro* covalent binding studies.** Six male Sprague-Dawley rats weighing 250 g each were pretreated with phenobarbital 80 mg/kg i.p. daily in three divided doses for 3 d. Phenobarbital has been shown in rats to be a necessary pretreatment to induce sufficient acetaminophen activation for hepatic necrosis to occur (2). The animals were killed, their livers excised, and their microsomes prepared by the method of Potter et al. (4). The protein concentration of the microsomes was assayed by the Lowry technique (15) using bovine serum albumin as a standard. Aliquots of microsomal suspension containing 2 mg protein were incubated with 0.5 μ Ci [³H]acetaminophen (0.04 μ Ci/ μ mol) in a final vol of 2.5 ml of ice-cold 0.12 M potassium phosphate pH 7.4. The incubations were performed in the presence of 0.001, 0.010, 0.100, and 1.000 mM GSH and in the presence or absence of 0.5 mM PTU. After the incubation mixture was equilibrated at 37°C for 2 min, 0.5 ml of a solution containing 1.15% KCl, 30 mM MgCl₂, 50 mM glucose-6-phosphate, 1 U/ml glucose-6-phos-

phate dehydrogenase, and 2 mM NADP was added and the samples were mixed in a Dubnoff shaking incubator for 30 min at 37°C. The reactions were stopped by adding 0.8 ml of trichloroacetic acid and vortexing. The samples were centrifuged at 2,000 g for 15 min at 4°C, the supernates were decanted, and the pellets were washed sequentially with 2 ml of 0.6 M trichloroacetic acid once and 2 ml of absolute methanol until no further radioactivity could be detected in the supernates when added to 10 ml Aquasol-II (New England Nuclear) and counted in a liquid scintillation spectrometer (an average of nine washes). The remaining radioactivity was considered covalently bound to the microsomal protein and was counted after dissolving the pellet in 1 ml of 1N NaOH at 25°C for 16 h. To study the effect of PTU concentration on microsomal covalent binding by acetaminophen-reactive metabolite, liver microsomes from phenobarbital pretreated rats that had been dosed with 0 or 150 mg/kg PTU 30 min before killing (*n* = 4 in each group) were incubated and processed as detailed above except in the presence of one concentration of GSH (0.01 mM) and four different concentrations of PTU (0, 0.1, 0.3, and 0.5 mM). The initial trichloroacetic acid supernates from the samples were analyzed by high pressure liquid chromatography to determine the influence of PTU on the formation of GSH-acetaminophen conjugate as well as to measure the formation of PTU-acetaminophen conjugate as a function of PTU concentration. GSH-acetaminophen conjugate was assayed using the method of Buckpitt et al. (16) and details of the PTU-acetaminophen conjugate chromatography are given below.

***In vivo* studies.** Mice were used for *in vivo* studies because of their greater susceptibility to the hepatotoxic effect of acetaminophen as compared to rats (2). Three groups of 32 male Swiss Weiss mice (weighing 20–30 g each) were injected intraperitoneally with 0, 3, or 30 mg/kg PTU dissolved in 0.1 ml ethanol, and 30 min later, each group was divided into four subgroups of eight mice each to which 0, 250, 500, or 750 mg/kg acetaminophen i.p. was administered. The acetaminophen was first dissolved in water at 70°C then cooled to 37°C for administration. After 4 h the animals were killed and their livers homogenized in a motor-driven glass-Teflon tissue homogenizer (Dupont Instruments, Wilmington, Del.) with two parts (vol/wt) 0.1 M sodium phosphate, 0.25 M sucrose buffer, pH 7.4. The homogenates were assayed for GSH concentration by reacting with 5,5'-dithiobis (2-nitrobenzoic acid) and measuring absorbance at 412 nm according to the method of Owens and Belcher (17) as modified by Kaplowitz (18). PTU itself did not react with the reagent.

Three additional groups of eight mice each were pretreated with 0, 3, or 30 mg/kg PTU followed in 30 min by 750 mg/kg [³H]acetaminophen (4 μ Ci/mmol). 4 h later, at the time of peak covalent binding (3), the animals were killed after withdrawing a sample of blood. The blood was assayed for ³H, liver were homogenized, and assayed for [³H]acetaminophen metabolite covalent bindings as described above and for unconjugated acetaminophen by measuring ethylacetate-extractable ³H (19).

The effect of PTU on the pharmacokinetics of acetaminophen was studied by measuring the concentration of [³H]-acetaminophen in the blood, urine, and bile of two groups of six phenobarbital pretreated rats (80 mg/kg per d for 3 d). The rats were anesthetized with pentobarbital 50 mg/kg, placed on heated operating boards, and their left external jugular veins, left femoral arteries, common bile ducts, and urinary bladders were cannulated. Sodium chloride (0.15 M) was infused via the jugular vein at a rate of 2 ml/kg per h. Six rats were injected with 150 mg/kg PTU in ethanol (0.1 ml) and six were treated with vehicle alone. After 30 min, each rat was injected with 500 mg/kg [³H]acetaminophen (8 μ Ci/mmol).

¹ Abbreviations used in this paper: GSH, glutathione; PTU, 6-*N*-propyl-2-thiouracil.

Samples of urine, blood, and bile were extracted with ethyl acetate and the extracts were assayed for ^3H 60, 120, 180, and 240 min later. The bile and urine samples were later assayed for PTU-acetaminophen conjugate concentration as noted below.

In another study, two groups of 10 phenobarbital pretreated rats were administered 0 or 150 mg/kg [^{35}S]PTU (3.4 $\mu\text{Ci}/\text{mmol}$). After 30 min, the groups were divided into subgroups of five rats treated with 0 or 750 mg/kg [^3H]acetaminophen (8 $\mu\text{Ci}/\text{mmol}$). Bile was collected for 4 h, after which the animals were killed and [^3H]acetaminophen covalent binding to whole liver homogenate was assayed. Bile samples were applied to thin layer chromatography as described below and [^{35}S]PTU eluting as new product was estimated. The cytochrome P-450 concentration of microsomes prepared from these rats was measured by the method of Omura and Sato (20) and cytochrome P-450 activity was quantified by measuring *p*-nitroanisole-*O*-demethylase activity (21). Cytochrome P₁-450 activity was assayed by measuring benzo- α -pyrene-hydroxylase activity (22) and cytochrome *c*-reductase activity was measured by the method of Williams and Kamin (23).

Identification of PTU-acetaminophen conjugate. Bile and liver homogenate samples from the rats used in the pharmacokinetic studies as well as the initial trichloroacetic acid supernates from in vitro covalent binding studies using [^3H]acetaminophen and [^{35}S]PTU were analyzed for the presence of a PTU-acetaminophen conjugate. Aliquots (50 μl) of sample, [^3H]acetaminophen, and [^{35}S]PTU were spotted on silica gel thin layer chromatography plates (20 \times 20 cm), air-dried, and chromatographed with a chloroform/benzene/methanol (3:1:1) solvent system. Sections of the plates (1 cm) were scraped into test tubes, vortexed with hot distilled water, and centrifuged at 2,000 *g* for 5 min. Aliquots (0.5 ml) of the supernates were added to 10 ml Aquasol-II and counted for both ^3H and ^{35}S simultaneously in a liquid scintillation spectrometer. In addition 20- μl aliquots of trichloroacetic acid supernate from in vitro covalent binding studies using [^3H]acetaminophen and [^{35}S]PTU were neutralized and applied to a $\mu\text{Bondapak C-18}$ high pressure liquid chromatography column (0.4 \times 25 cm, Altex Scientific, Inc., Berkeley, Calif.) which was equilibrated with 30% acetonitrile and eluted according to the following protocol: 30% acetonitrile for 4 min, 30–100% over 3 min, and 100% for 5 min. The flow rate was 2 ml/min and eluted fractions were collected at 0.5-min intervals and assayed for both ^3H and ^{35}S . [^3H]Acetaminophen standard had a retention time of 5.5 min and [^{35}S]PTU standard eluted at 8.0 min. All liquid scintillation counting was performed in a Beckman LS-3150T liquid scintillation spectrometer (Beckman Instruments, Inc., Fullerton, Calif.) using external standardization for quench correction. Double isotope (^3H and ^{35}S) counting was performed by setting windows so that ^3H had no crossover into the ^{35}S channel. ^{35}S crossover into the ^3H channel was corrected for by using appropriate control standards. The automatic quench compensator was used to minimize crossover of ^{35}S .

Statistics. All statistical comparisons were made using Student's *t* test for analysis. Unless otherwise noted, a *P* value of <0.05 was accepted as significant.

RESULTS

Covalent binding of [^3H]acetaminophen reactive metabolite to rat liver microsomes in vitro decreased with increasing GSH concentration (Fig. 1) as noted previously by Mitchell et al. (5). However, at any given GSH concentration the presence of PTU (0.5 mM) significantly decreased the amount of covalent binding

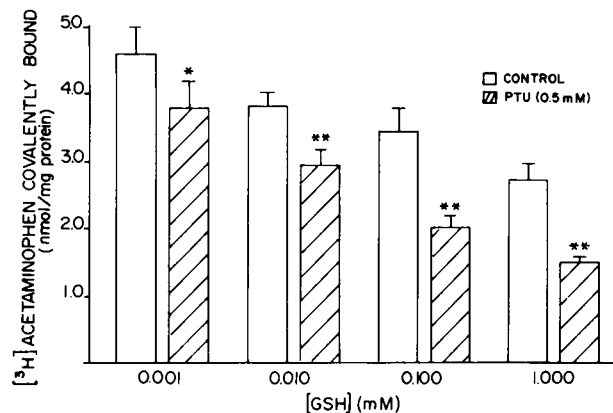


FIGURE 1 Inhibition of [^3H]acetaminophen covalent binding to rat liver microsomes in vitro by PTU at varying GSH concentrations. Microsomes (2 mg protein) from each of six phenobarbital pretreated rats were incubated for 30 min at 37°C with 0.5 μCi [^3H]acetaminophen (0.04 $\mu\text{Ci}/\mu\text{mol}$), GSH (0.001–1.000 mM), PTU (0 or 0.5 mM), and an NADPH generating system. The reaction was stopped and microsomal proteins were precipitated by the addition of trichloroacetic acid. After a subsequent wash with trichloroacetic acid, the precipitates were washed with methanol repeatedly until no further radioactivity could be eluted. The remaining radioactivity was considered covalently bound and was counted after dissolving the precipitate in 1 N NaOH. Results are depicted as mean \pm SE. A progressive decline was noted in covalently bound [^3H]acetaminophen with increasing GSH concentration. At each GSH concentration, PTU significantly decreased covalent binding. **P* < 0.05; ***P* < 0.01.

even further. This protective effect of PTU, when studied at a single GSH concentration (0.1 mM) was dose dependent, as is shown in Fig. 2. Thus, in this system PTU appears to have a dose-dependent effect

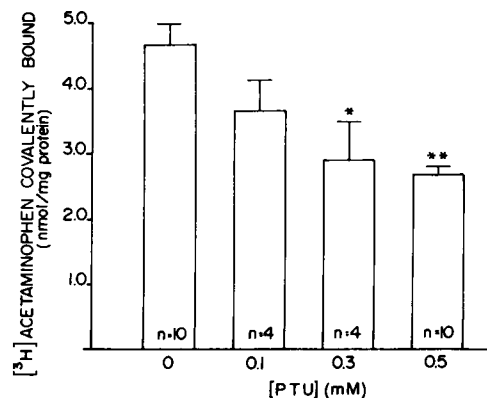


FIGURE 2 Dose-dependent inhibition of [^3H]acetaminophen covalent binding to rat liver microsomes in vitro by PTU. Using the same microsomal incubation system described in Fig. 1 but at only one concentration of GSH (0.01 mM) and varying concentrations of PTU (0, 0.1, 0.3, and 0.5 mM) the covalent binding of [^3H]acetaminophen to microsomal proteins was measured. Results are depicted as mean \pm SE. PTU had a dose-dependent inhibitory effect on covalent binding that was significant at the two highest concentrations of drug. **P* < 0.02; ***P* < 0.001.

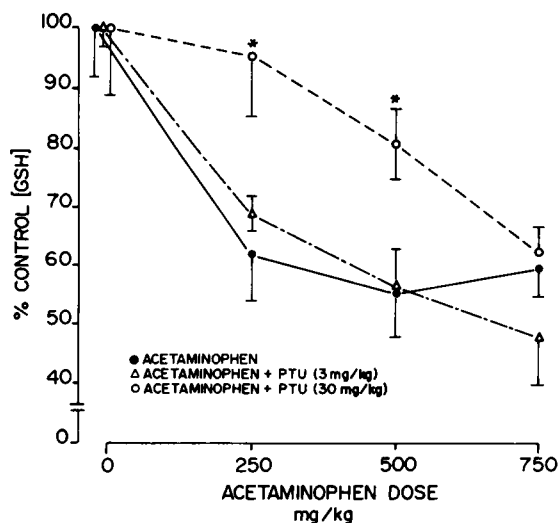


FIGURE 3 Inhibition of acetaminophen-induced mouse hepatic GSH depletion by pretreatment with PTU. Groups of 32 mice were treated with 0, 3, or 30 mg/kg PTU. 30 min later, the groups were divided into subgroups of eight mice each, which were treated with 0, 250, 500, or 750 mg/kg acetaminophen. After 4 h, the mice were killed and their livers homogenized and assayed for GSH. Results are depicted as mean \pm SE. PTU at the highest dose inhibited acetaminophen-induced GSH depletion at the 250 and 500 mg/kg doses of acetaminophen. This GSH-sparing effect of PTU was overcome at the 750 mg/kg dose of acetaminophen. * $P < 0.03$.

which is additive with respect to the effect of GSH in protecting against covalent binding of acetaminophen metabolite to liver microsomes.

This interaction between PTU and GSH was further studied in vivo in mice as shown in Fig. 3. At the low dose (3 mg/kg) PTU was ineffective in preventing acetaminophen-induced GSH depletion. However, PTU at the high dose (30 mg/kg) prevented GSH depletion induced by acetaminophen at the 250 and 500 mg/kg dose. Even this protective effect could be overcome if a large enough dose of acetaminophen (750 mg/kg) was administered. Nevertheless PTU protection against covalent binding of acetaminophen administered at this dose was observed as shown in Fig. 4. Although low dose PTU was without measurable effect, high dose PTU significantly diminished acetaminophen covalent binding, thus confirming in vivo the in vitro observations noted above.

We next directed our efforts at determining the mechanism by which PTU exerts this protection against acetaminophen covalent binding. One possible explanation for this phenomenon might be interference with metabolic activation of acetaminophen. Table I summarizes the experimental evidence that we have derived to examine this possibility. In short-term acute studies, neither PTU nor acetaminophen had significant effects on cytochrome P-450 concentration or activity, cytochrome P₁-450 activity, or cytochrome c-re-

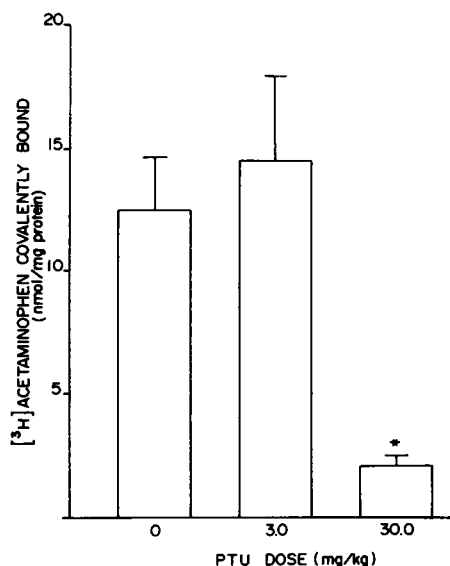


FIGURE 4 Inhibition of [³H]acetaminophen covalent binding to mouse liver proteins in vivo by treatment with PTU. Three groups of eight mice each were treated with 0, 3, or 30 mg/kg PTU followed in 30 min by 750 mg/kg [³H]acetaminophen. 4 h later, the mice were killed and their liver homogenates were assayed for [³H]acetaminophen covalent binding as described in Fig. 1. Results are depicted as mean \pm SE. PTU at the 30 mg/kg dose significantly inhibited covalent binding. * $P < 0.001$.

ductase activity. Furthermore, PTU had no effect on GSH-acetaminophen conjugate formation in vitro using the microsomal test system, implying that it had no inhibitory effect on acetaminophen reactive metabolite formation specifically (Table II). PTU exerted no influence on serum or hepatic ethylacetate-extractable acetaminophen concentrations in the mice administered 750 mg/kg acetaminophen described above (Table III), suggesting that PTU did not alter overall acetaminophen metabolism or acetaminophen conjugate formation. Lastly, PTU had no influence on the pharmacokinetics of acetaminophen in blood, urine, or bile of rats, as depicted in Fig. 5.

We studied the remaining possibility that PTU forms a direct conjugate with acetaminophen reactive metabolite. Using both a thin layer chromatography system (Fig. 6) and a high pressure liquid chromatography system (data not shown) we were able to clearly separate [³H]acetaminophen from [³⁵S]PTU. After reaction of [³H]acetaminophen with [³⁵S]PTU in vivo or in vitro in the presence of liver microsomes and an NADPH-generating system, a new peak of ³⁵S radioactivity could be found that coincided with the unreacted [³H]acetaminophen peak (seen in both chromatography systems), suggesting the formation of a [³H]acetaminophen-³⁵S]PTU conjugate. This new product was never observed in the absence of acetaminophen or liver microsomes. The amount of new

TABLE I
Effect of PTU and Acetaminophen Administration on Microsomal Drug-metabolizing Enzymes in Rats

Treatment group	Cytochrome P-450	<i>p</i> -Nitroanisole- <i>O</i> -demethylase	Cytochrome <i>c</i> reductase	Benzo- α -pyrene hydroxylase
	nmol/mg protein	nmol/min/mg	nmol/min/mg	relative activity/mg
0 PTU				
0 Acetaminophen	0.86 \pm 0.07	0.90 \pm 0.13	71.0 \pm 4.6	0.44 \pm 0.02
150 PTU				
0 Acetaminophen	0.73 \pm 0.09	0.90 \pm 0.12	70.9 \pm 4.5	0.41 \pm 0.03
0 PTU				
750 Acetaminophen	0.83 \pm 0.05	1.06 \pm 0.16	67.6 \pm 6.5	0.46 \pm 0.02
150 PTU				
750 Acetaminophen	0.86 \pm 0.06	0.87 \pm 0.04	80.4 \pm 3.3	0.40 \pm 0.02

Two groups of 10 phenobarbital pretreated rats were administered 0 or 150 mg/kg PTU. After 30 min the rats were divided into two subgroups of five rats which were treated with 0 or 750 mg/kg acetaminophen. 4 h later the rats were killed and their liver microsomes assayed for cytochrome P-450 content and *p*-nitroanisole-*O*-demethylase, cytochrome *c*-reductase, and benzo- α -pyrene hydroxylase activities. Means \pm SE are recorded in the table.

product formed in vitro was a function of PTU concentration, reaching a maximum of 0.06 μ mol/min per mg protein at 0.5 mM PTU. To determine the physiologic significance of this possible new reaction product, the net reduction in acetaminophen covalent binding to whole liver homogenate in vivo produced by treatment with [35 S]PTU was compared to the total excretion of [35 S]PTU as "new" reaction product separated by thin layer chromatography. As shown in Fig. 7, the reduction in covalent binding in PTU-treated rats (108 nmol)

closely approximates the total biliary excretion of [35 S]-PTU as new product (116 nmol). No new product was observed in the urine. These data suggest that formation of a PTU-acetaminophen conjugate may account quantitatively for the decreased hepatic covalent bindings of acetaminophen in PTU-treated rats.

DISCUSSION

The mechanism of acetaminophen hepatotoxicity has been ascribed to the covalent binding of hepatic macro-

TABLE II
Effect of PTU on Acetaminophen-GSH Adduct Formation in vitro

Untreated rat microsomes		PTU pretreated rat microsomes	
[PTU]	GSH-acetaminophen adduct formed	[PTU]	GSH-acetaminophen adduct formed
mM	nmol	mM	nmol
0	21 \pm 5	0	18 \pm 3
0.5	16 \pm 3	0.5	17 \pm 4

Liver microsomes (2 mg protein) from phenobarbital pretreated rats that had been administered 0 or 150 mg/kg PTU ($n = 4$ in each group) 30 min before killing were incubated for 60 min at 37°C with 0.5 μ Ci [3 H]acetaminophen (0.04 μ Ci/ μ mol), 0.01 mM GSH, and PTU (0 or 0.5 mM) in the presence of an NADPH generating system. The reaction was stopped and microsomal proteins were precipitated for the addition of trichloroacetic acid. The supernates were assayed for acetaminophen-GSH conjugate formation by the method of Buckpitt (16). Results are expressed as mean \pm SE.

TABLE III
Effect of PTU on Metabolism of Acetaminophen in Mice

PTU dose	[3 H]Acetaminophen concentration	
	Serum	Liver
mg/kg	μ mol/ml	μ mol/mg protein
0	252 \pm 48	22.0 \pm 4.6
3	261 \pm 42	25.2 \pm 5.4
30	254 \pm 38	29.6 \pm 6.7

Three groups of eight mice each were treated with 0, 3, or 30 mg/kg PTU followed in 30 min by 750 mg/kg [3 H]acetaminophen. 4 h later the mice were killed after withdrawing a serum sample and their livers were homogenized. Ethyl acetate-extractable (unmetabolized) [3 H]acetaminophen concentrations in both serum and liver homogenate samples were assayed. PTU had no significant effect, indicating that it did not alter overall metabolism of acetaminophen. Means \pm SE are recorded on the table.

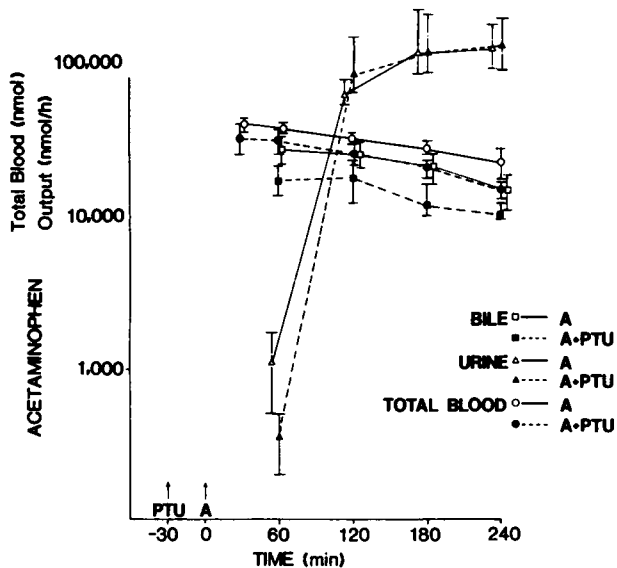


FIGURE 5 Effect of PTU on the concentrations of [^3H]acetaminophen (A) in the bile, urine, and blood of phenobarbital pretreated rats. Two groups of six pentobarbital-anesthetized rats each were treated with 0 or 150 mg/kg PTU followed in 30 min by [^3H]acetaminophen 500 mg/kg. Samples of bile, urine, and blood were collected at 60, 120, 180, and 240 min, extracted in ethyl acetate, and the extracts were counted for ^3H . Results are plotted on a semilogarithmic scale and depicted as mean \pm SE. PTU had no significant effect on [^3H]acetaminophen concentrations.

molecules by a reactive metabolite of the drug (3). We have confirmed both *in vitro* and *in vivo* the observations of Mitchell et al. (5) that the amount of covalent binding of acetaminophen metabolite is inversely related to the concentration of GSH, a potent nucleophile that detoxifies the metabolite by forming a conjugate with it. This conjugation *in vitro* has been shown to be catalyzed by a cytosolic enzyme, presumably a glutathione S-transferase (7). Our previous studies showing that PTU is a substrate for the glutathione S-transferases that substitutes for GSH (14) prompted us to examine the effect of PTU on hepatocyte macromolecular covalent binding by acetaminophen reactive metabolite. We found *in vitro* that PTU had a dose-dependent protective effect that was additive with the effect of GSH. *In vivo*, PTU not only decreased covalent binding but also prevented hepatic GSH depletion by acetaminophen.

Previous *in vivo* studies of acetaminophen toxicity by Linscheer et al. (10) have suggested that the protective effect of PTU is mediated by its antithyroid activity, much in the same manner as has been postulated for its beneficial effect in alcoholic liver disease (24). However, both our *in vitro* and short term *in vivo* studies showing the protective effect of PTU against hepatocyte macromolecular covalent binding by acetaminophen reactive metabolite indicate that PTU acts, at

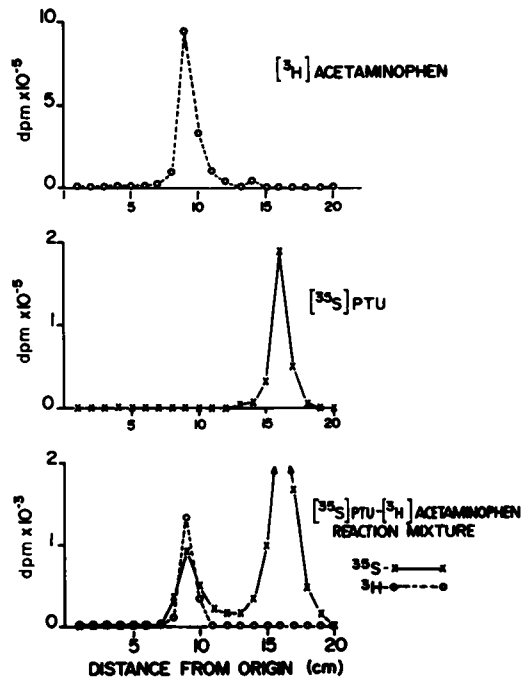


FIGURE 6 Thin layer chromatography of [^3H]acetaminophen, [^{35}S]PTU, and [^{35}S]PTU- ^3H]acetaminophen reaction mixture using a chloroform/benzene/methanol (3:1:1) solvent system for elution. Sections (1 cm) were scraped from the 20 \times 20 plates, eluted with hot water, and counted for ^3H , ^{35}S , or both. A new ^{35}S peak could be separated in the reaction mixture that was separate from the [^{35}S]PTU peak but that coeluted with the [^3H]acetaminophen peak.

least in part, independently of its antithyroid effect. More recent studies, also by Linscheer et al. (25), showing that triiodothyronine treatment had no effect upon the protective action of PTU against acetaminophen hepatotoxicity confirm our observations.

We reasoned that the mechanism of PTU action could be explained by (a) increased conjugation of acetaminophen by the usual mechanisms of sulfation or glucuronidation, (b) decreased metabolic activation by microsomal drug metabolizing enzymes, or (c) substitution of PTU for GSH in detoxification of acetaminophen reactive metabolite. Slattery and Levy (26) have suggested that sulfation and glucuronidation of acetaminophen are capacity-limited processes, whereas formation of reactive metabolite follows first-order kinetics. Thus, thiols such as *N*-acetylcysteine, which have been shown to be effective against acetaminophen toxicity, may act by serving as sources of inorganic sulfate, the availability of which is rate-limiting, rather than as GSH precursors (27). Although it is conceivable that PTU might be acting in a similar manner, our data suggest the contrary since PTU had no effect upon overall pharmacokinetics, metabolism, or conjugation of acetaminophen. Certain thionosulfur-containing compounds, including PTU, have been shown to inhibit

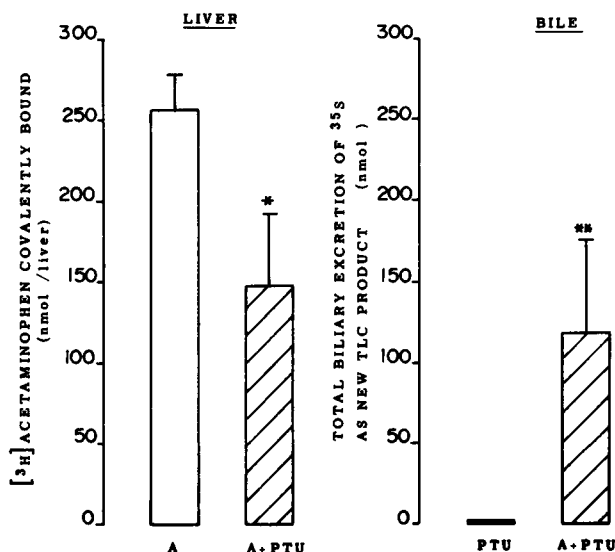


FIGURE 7 Comparison of PTU induced inhibition of total [^3H]acetaminophen covalently bound to rat whole liver homogenates vs. total biliary excretion of [^{35}S]PTU product as defined and separated by thin layer chromatography (TLC). Two groups of 10 phenobarbital pretreated rats each were administered 0 or 150 mg [^{35}S]PTU(P). 30 min later, the groups were each divided into two subgroups of five rats and treated with 0 or 750 mg/kg [^3H]acetaminophen (A). Bile was collected from each rat for 4 h, after which the animals were killed and their whole liver homogenates were assayed for [^3H]acetaminophen covalent binding. Bile samples were applied to thin layer chromatography as noted in Fig. 6 and the total amount of ^{35}S eluting as new product was counted. Results are depicted as mean \pm SE. The reduction in covalent binding resulting from PTU treatment (108 nmol), which is the difference between the two bars on the left, approximates total biliary excretion of new [^{35}S]PTU product (116 nmol). This product was not found in rats treated with PTU alone. * $P < 0.05$; ** $P < 0.001$.

microsomal drug metabolizing enzymes (28). Thus, the apparent protective effect of PTU could be ascribed to inhibition of reactive metabolite formation. However, we were unable to observe in vivo any effect of short-term PTU treatment on hepatocyte microsomal drug metabolizing enzyme concentration, activity, or regenerating capacity, nor did we observe any inhibition of acetaminophen metabolic activation by PTU in vitro.

Thus, we examined the remaining possibility that PTU forms a conjugate with the reactive metabolite of acetaminophen. We were able to identify in liver homogenates from acetaminophen-treated mice, as well as in incubates of PTU with acetaminophen in the presence of liver microsomes and an NADPH-generating system, a new PTU product that could be distinguished clearly from unreacted PTU. The amount of new product formation in a microsomal incubation system was a function of PTU concentration but was not observed in the absence of acetaminophen. The physiologic significance of this new product was confirmed by

our observation that the amount of PTU excreted as new product in vivo, presumably acetaminophen-PTU conjugate, correlated quite closely with the net decrease in acetaminophen covalent binding of hepatocyte macromolecules. We conclude therefore that PTU exerts a protective effect against acetaminophen toxicity by direct chemical reaction with its reactive metabolite, thereby preventing its covalent binding to macromolecules and resultant hepatocellular injury.

ACKNOWLEDGMENTS

We are grateful for the secretarial assistance of Anita Boesman, and technical advice of Dr. Joseph Reeve.

This work was supported by U. S. Public Health Service grant GM/CA 24987 and Veterans Administration Medical Research funds.

REFERENCES

- Black, M. 1980. Acetaminophen hepatotoxicity. *Gastroenterology*. **78**: 382-392.
- Mitchell, J. R., D. J. Jollow, W. Z. Potter, D. C. Davis, J. R. Gillette, and B. B. Brodie. 1973. Acetaminophen-induced hepatic necrosis. I. Role of drug metabolism. *J. Pharmacol. Exp. Ther.* **187**: 185-194.
- Jollow, D. J., J. R. Mitchell, W. Z. Potter, D. C. Davis, J. R. Gillette, and B. B. Brodie. 1973. Acetaminophen-induced hepatic necrosis. II. Role of covalent binding in vivo. *J. Pharmacol. Exp. Ther.* **187**: 195-202.
- Potter, W. Z., D. C. Davis, J. R. Mitchell, D. J. Jollow, J. R. Gillette, and B. B. Brodie. 1973. Acetaminophen-induced hepatic necrosis. III. Cytochrome P-450 mediated covalent binding in vitro. *J. Pharmacol. Exp. Ther.* **187**: 203-210.
- Mitchell, J. R., D. J. Jollow, W. Z. Potter, J. R. Gillette, and B. B. Brodie. 1973. Acetaminophen-induced hepatic necrosis. IV. Protective role of glutathione. *J. Pharmacol. Exp. Ther.* **187**: 211-217.
- Brodie, B. B., and J. Axelrod. 1948. The estimation of acetanilide and its metabolic products, aniline, *N*-acetyl-*p*-aminophenol and *p*-aminophenol (free and total conjugated) in biological fluids and tissues. *J. Pharmacol. Exp. Ther.* **94**: 22-28.
- Rollins, D. E., and Buckpitt, A. R. 1979. Liver cytosol catalyzed conjugation of reduced glutathione with a reactive metabolite of acetaminophen. *Toxicol. Appl. Pharmacol.* **47**: 331-339.
- Prescott, L. F., R. W. Newton, C. P. Swainson, N. Wright, A. R. W. Forrest, and H. Matthew. 1974. Successful treatment of severe paracetamol overdose with cysteamine. *Lancet*. **I**: 588-592.
- Prescott, L. F., J. Park, A. Ballantyne, P. Adriaenssens, and A. T. Proudfoot. 1977. Treatment of paracetamol (acetaminophen) poisoning with *N*-acetylcysteine. *Lancet*. **II**: 432-434.
- Linscheer, W. G., K. L. Raheja, C. Cho, and N. J. Smith. 1980. Mechanism of the protective effect of propylthiouracil against acetaminophen (Tylenol) toxicity in the rat. *Gastroenterology*. **78**: 100-107.
- Israel, Y., H. Kalant, H. Orrego, J. M. Khanna, L. Videla, and J. N. Phillips. 1975. Experimental alcohol-induced hepatic necrosis: suppression by propylthiouracil. *Proc. Natl. Acad. Sci. U. S. A.* **72**: 1137-1141.
- Orrego, H., F. J. Charnichael, M. J. Phillips, H. Kalant, J. Khanna, and Y. Israel. 1976. Protection by propylthio-

- uracil against carbon tetrachloride-induced liver damage. *Gastroenterology*. **71**: 821–826.
13. Orrego, H., H. Kalant, Y. Israel, J. Blake, A. Medline, J. G. Rankin, A. Armstrong, and B. Kapur. 1979. Effect of short-term therapy with propylthiouracil in patients with alcoholic liver disease. *Gastroenterology*. **76**: 105–115.
 14. Yamada, T., and N. Kaplowitz. 1980. Propylthiouracil: a substrate for the glutathione S-transferases that competes with glutathione. *J. Biol. Chem.* **255**: 3508–3513.
 15. Lowry, O. H., N. J. Rosebrough, A. L. Farr, and R. J. Randall. 1951. Protein measurement with the Folin phenol reagent. *J. Biol. Chem.* **193**: 265–275.
 16. Buckpitt, A. R., D. F. Rollins, S. D. Nelson, R. B. Franklin, and J. R. Mitchell. 1977. Quantitative determination of the glutathione, cysteine, and N-acetyl cysteine conjugates of acetaminophen by high-pressure liquid chromatography. *Anal. Biochem.* **83**: 168–177.
 17. Owens, C. W., and R. V. Belcher. 1965. A colorimetric micromethod for the determination of glutathione. *Biochem. J.* **94**: 705–711.
 18. Kaplowitz, N. 1977. Interaction of azathioprine and glutathione in the liver of rats. *J. Pharmacol. Exp. Ther.* **200**: 279–286.
 19. Fretthold, D., and I. Shunshine. 1978. Analysis of acetaminophen (Tylenol) by HPLC. In *Altex Chromatogram*. Altex Scientific Inc. **1**: 3.
 20. Omura, T., and R. Sato. 1964. The carbon monoxide-binding pigment of liver microsomes. I. Evidence for its hemoprotein nature. *J. Biol. Chem.* **239**: 2370–2378.
 21. Netter, K. J., and G. Seidel. 1964. An adaptively stimulated O-demethylating system in rat liver microsomes and its kinetic properties. *J. Pharmacol. Exp. Ther.* **146**: 61–65.
 22. Dehnen, W., R. Tomingas, and J. Ross. 1973. A modified method for the assay of benzo (a) pyrene hydroxylase. *Anal. Biochem.* **53**: 373–383.
 23. Williams, C. H. Jr., and H. Kamin. 1962. Microsomal triphosphopyridine nucleotide-cytochrome C reductase of liver. *J. Biol. Chem.* **237**: 587–595.
 24. Israel, Y., P. G. Walfish, H. Orrego, J. Blake, and H. Kalant. 1979. Thyroid hormones in alcoholic liver disease. Effect of treatment with 6-n-propylthiouracil. *Gastroenterology*. **76**: 117–122.
 25. Linscheer, W. G., K. L. Raheja, C. Cho, C. Lely, and J. Schmidt. 1980. Propylthiouracil (PTU) detoxification of glutathione depleting hepatotoxins. *Gastroenterology*. **78**: 1209. (Abstr.)
 26. Slattery, J. T., and G. Levy. 1979. Acetaminophen kinetics in acutely poisoned patients. *Clin. Pharmacol. Ther.* **25**: 184–195.
 27. Galinsky, R. E., and G. Levy. 1979. Effect of N-acetylcysteine on the pharmacokinetics of acetaminophen in rats. *Life Sci.* **25**: 693–700.
 28. Hunter, A. L., and R. A. Neal. 1975. Inhibition of hepatic mixed function oxidase activity in vitro and in vivo by various amino-sulfur-containing compounds. *Biochem. Pharmacol.* **24**: 2199–2205.