# Discovery of potent myeloid cell leukemia 1 (Mcl 1) inhibitors using fragment based methods and structure based design 

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#### Abstract

Myeloid cell leukemia-1 (Mcl-1), a member of the Bcl-2 family of proteins, is overexpressed and amplified in various cancers and promotes the aberrant survival of tumor cells that otherwise would undergo apoptosis. Here we describe the discovery of potent and selective Mcl-1 inhibitors using fragment-based methods and structure-based design. NMR-based screening of a large fragment library identified two chemically distinct hit series that bind to different sites on Mcl-1. Members of the two fragment classes were merged together to produce lead compounds that bind to Mcl-1 with a dissociation constant of $<100 \mathrm{nM}$ with selectivity for Mcl-1 over Bcl-xL and Bcl-2. Structures of merged compounds when complexed to Mcl-1 were obtained by X-ray crystallography and provide detailed information about the molecular recognition of smallmolecule ligands binding Mcl-1. The compounds represent starting points for the discovery of clinically useful Mcl-1 inhibitors for the treatment of a wide variety of cancers.


## Keywords

Fragment-based screening; apoptosis; cancer; Mcl-1; drug discovery

## INTRODUCTION

Mcl-1 (Myeloid cell leukemia 1) is a member of the $\mathrm{Bcl}-2$ family of proteins that when dysregulated prevents cancer cells from undergoing programmed cell death, a hallmark of cancer. ${ }^{1}$ By overexpressing the Mcl-1 protein or amplifying the Mcl-1 gene, a cancerous cell can avoid death, the normal fate for cells exhibiting abnormal and deregulated growth. ${ }^{2,3}$ Indeed, amplification of Mcl-1 is one of the most common genetic aberrations observed in human cancer, ${ }^{4,5}$ including lung, breast, prostate, pancreatic, ovarian, and cervical cancers, as well as melanoma and leukemia. ${ }^{6-13}$ Moreover, Mcl-1 overexpression has emerged as a resistance mechanism against a number of anti-cancer therapies including the widely prescribed microtubule-targeted agents paclitaxel and vincristine ${ }^{14}$ as well as gemcitabine ${ }^{15}$, a first-line treatment option for pancreatic cancer. Mcl-1 overexpression also confers resistance to ABT-263, a Bcl-2/Bcl-xL inhibitor currently in clinical trials. ${ }^{16-18}$ Not

[^0]surprisingly, specific down-regulation of Mcl-1 by RNA interference inhibits cell growth, colony formation, and induces apoptosis in pancreatic cancer cells in vitro, and markedly decreases tumorigenicity in mouse xenograft models. ${ }^{15}$ Silencing of Mcl-1 also potently kills particular subgroups of non-small-cell lung cancer (NSCLC) cell lines. ${ }^{19}$ Together, these data suggest that direct inhibition of Mcl-1 could be an effective therapeutic option for a wide variety of cancers. In an attempt to target Mcl-1 for the treatment of cancer, smallmolecule inhibitors ${ }^{20-22}$ as well as stapled helix peptides ${ }^{23,24}$ that bind to $\mathrm{Mcl}-1$ have been reported. However, no Mcl-1 inhibitors have entered clinical trials. Indeed, targeting Mcl-1 is extremely difficult, since Mcl-1 exerts its activity through protein-protein interactions involving a large binding interface. To overcome the challenges associated with targeting Mcl-1, we applied a combination of fragment-based methods ${ }^{25}$ and structure-based design ${ }^{26}$. A similar approach was successfully utilized in the discovery of ABT-263 ${ }^{27,28}$ and in the discovery of inhibitors against other challenging protein targtets, such as $\mathrm{Hsp} 90^{29}$ and BACE ${ }^{30}$. Fragment-based drug discovery (FBDD) takes advantage of sensitive biophysical methods to detect the binding of molecular fragments (MW <300 Da) that bind weakly to their target proteins. This approach leads to a more efficient sampling of chemical space and can provide hits with higher ligand efficiencies. ${ }^{31,32}$ Furthermore, the affinity of the initial fragment hits can be markedly increased by growing, linking, or scaffold merging to obtain potent lead molecules.

Here, we describe the discovery of potent small-molecule Mcl-1 inhibitors ( $\mathrm{K}_{\mathrm{i}}<100 \mathrm{nM}$ ) that inhibit BH3-containing peptides from binding to Mcl-1. An NMR-based fragment screen yielded two distinct classes of hits that were shown by NMR to bind to two different regions of Mcl-1. Based on NMR-derived structural information, the hits from the two classes were merged together to obtain potent Mcl-1 inhibitors that exhibited $>100$-fold enhanced binding affinity over each component separately. Improved analogs were rapidly optimized by exploiting the SAR observed for the initial fragment hits. Crystal structures of inhibitor-bound Mcl-1 complexes validate the design strategy and also provide detailed structural information for the discovery of more potent inhibitors.

## RESULTS

## Hit Identification

Recombinant, isotopically labeled Mcl-1 (residues 172-327) was used to screen our curated fragment library ( $>13,800$ compounds) by recording SOFAST ${ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N}$ HMQC spectra of Mcl-1 (Figure 1) incubated with mixtures of 12 fragments. Deconvolution of the mixtures by screening individual compounds yielded 132 hits ( $0.95 \%$ hit rate) of eleven distinct chemical classes. Of the initial hits, 93 inhibited Mcl-1 with a $\mathrm{K}_{\mathrm{i}}<500 \mu \mathrm{M}$, and more than a quarter of the hits demonstrated a ligand efficiency (LE) of 0.25 or greater. Analogs were tested at a single concentration $(400 \mu \mathrm{M})$ for their ability to displace a FITC-labeled Mcl-1derived BH 3 peptide from binding to Mcl-1 in a fluorescence polarization anisotropy (FPA) assay.

Based on their affinity and distinct chemical characteristics, two classes of compounds were selected for follow-up experiments (Table 1). $\mathrm{K}_{\mathrm{i}}$ values were derived from competition FPA assays measuring the disruption of Mcl-1 and a FITC-Mcl-1-BH3 peptide, or $\mathrm{Bcl}-\mathrm{xL}$ and Bcl-2 and a FITC-BAK-BH3 peptide. Class I fragment hits contained 6,5-fused heterocyclic carboxylic acids. Analogs of the hits were obtained to explore the SAR. Parent benzothiophene-, benzofuran- and indole-2-carboxylic acids were not identified in our initial screen because of their weak affinity ( $\mathbf{1}, \mathbf{6}$ and $\mathbf{1 2 )}$. The Class I fragment hits identified in the screen contained one or two Cl and/or Me substitutions which significantly improved their binding affinities. Similar trends in the SAR for all three core series suggest that they may display similar binding modes. The fragment SAR also suggests that the 3-, 4- and 6-
positions are preferred sites for substitutions, and that multiple substitutions can be tolerated or even beneficial in some combinations. For the indole series, a 1-Me substitution exerted similar effects as 3-Cl or 3-Me on benzothiophene and benzofuran scaffolds, suggesting that this core unit could be flipped along its long axis when bound. Interestingly, Mcl-1 inhibitors containing indole acids have also been recently reported in the patent literature. ${ }^{33,34}$

Another structurally distinct group of fragment hits (Class II) consists of a hydrophobic aromatic system tethered by a linker to a polar functional group, most often a carboxylic acid (Table 1). Of the hits in this series, 4-chloro-3,5-dimethylphenyl and 1-naphthyl groups as contained in compounds $\mathbf{1 7}$ and $\mathbf{1 8}$, respectively, were found to be preferred for binding to Mcl-1.

## NOE-guided fragment docking

To determine how Class I and II fragments bind to Mcl-1, we performed NMR-based structural studies of Mcl-1/fragment hit complexes. Using double-labeled ( $\left.{ }^{15} \mathrm{~N},{ }^{13} \mathrm{C}\right) \mathrm{Mcl}-1$ protein, we acquired NOE-derived distance restraints and used these to dock representative fragments into a previously determined Mcl-1/Bim BH 3 peptide complex (PDB: 3KJ2 ${ }^{35}$ ). Compounds were docked using a simulated annealing protocol in XplorNIH, followed by energy minimization of such models in MOE 2011.10 (Chemical Computing Group Inc.). Fragments representing Class I and II exhibited different patterns of protein/ligand NOEs confirming that they bind to different regions of Mcl-1. The model structures obtained by this method were consistent with the observed NOEs. A model for Mcl-1 complexed to a compound representative of Class I (compound 5) is shown in Figure 2A. Compound 5 displayed NOEs from A227, M231, and F270 to the six-membered ring of the ligand. In this model the acid at the 2-position points towards R263. An NOE-derived model of a fragment hit representing Class II (compound 17) is depicted in Figure 2B. In this case, NOEs from M250, F270, and V249 to the two methyl groups of the 4-chloro-3,5-dimethylphenyl indicate that this aromatic group sits deep in the pocket. Additional NOEs from V253 to the aliphatic tether show that the carboxylic acid sits at the surface of the pocket. The NMRderived models demonstrate that the hydrophobic portions of the two classes of molecules bind to adjacent parts of a large pocket and that the carboxylic acids of both series point towards and likely interact with R263. These results explain the observation that Class I and Class II compounds are not able to bind simultaneously to Mcl-1.

## Fragment merging and linker optimization

NOE-guided molecular modeling positioned the Class I heterocycles above the hydrophobic pocket utilized by the Class II compounds. This arrangement suggests that the attachment of Class II aromatic groups with a 2-4 atom linker to the 3-position of the Class I 6,5-fused heterocycles would lead to merged compounds that maintain the favorable hydrophobic contacts of both fragments to Mcl-1 as well as the interaction between the common carboxylic acid and R263 of Mcl-1. Support for this linking strategy was also obtained from compounds identified and tested from the Vanderbilt Institute of Chemical Biology chemical collection, compounds $\mathbf{1 9}$ and $\mathbf{2 0}$ (Table 1), that exhibited Mcl-1 binding affinities of $62 \mu \mathrm{M}$ and $87 \mu \mathrm{M}$, respectively. In order to test this hypothesis, the fragment hits for which we obtained NMR-derived model structures (Figure 2) were linked together. Scaffold merging produced a compound with greater than two orders of magnitude improved binding affinity (Figure 3). To determine the optimum linker length, a 1-naphthyl group was connected to the benzothiophene- and benzofuran-2-carboxylic acid cores at the 3-position using two to four atom linkers. Table 2 summarizes the Mcl-1 binding affinities for merged compounds with varying linker lengths. The two series of compounds containing different core units showed a parallel SAR, and all merged compounds 21-26 exhibited markedly enhanced
binding affinity compared to the unsubstituted cores $\mathbf{1}$ and $\mathbf{6}$ confirming the validity of the linking strategy. Although the 2- and 3-atom linked compounds led to a significant increase in potency, the 4 -atom linked compounds (e.g., 23 and 26) yielded the most potent Mcl-1 inhibitors in the series, displaying sub-micromolar dissociation constants. Moreover, these compounds exhibited selectivity for binding to Mcl-1 over Bcl-xL and Bcl-2 (Table 2).

## SAR of the merged compounds

To further improve the binding affinity to Mcl-1, the binding contribution of the hydrophobic group of the Class II fragments was investigated (Table 3). A library of indoles (27-48) tethered by the optimized 4 -atom linker to various aromatic groups containing different substitution patterns and sizes was prepared. The parent indole 27 with an unsubstituted phenyl group $\left(\mathrm{K}_{\mathrm{i}}=35 \mu \mathrm{M}\right)$ showed a marginal increase in affinity over the initial fragment hits in the class (13-15). Substitutions at the $2^{\prime}$-position $(\mathbf{2 8}, \mathbf{2 9})$ on the phenyl group yielded only a minor increase in potency. However, potency was remarkably enhanced ( 20 -fold) by mono-substitution of small groups at the $3^{\prime}$-position as demonstrated by compounds $\mathbf{3 0}$ and $\mathbf{3 1}$ containing $3^{\prime}-\mathrm{Me}$ and $3^{\prime}-\mathrm{CF}_{3}$ substitutions, respectively. An increase in affinity was also observed for $4^{\prime}-\mathrm{Cl}$ substitutions as exemplified by compound 33. When these beneficial groups were combined, the effect was additive. For example, compound 35 which contains a $3^{\prime}-\mathrm{Me}-4^{\prime}-\mathrm{Cl}-\mathrm{phenyl}$ group displays a 90 -fold improved potency over the parent molecule 27 . Indole 37 containing a $3^{\prime}, 5^{\prime}$-diMe- $4^{\prime}$ - Cl -phenyl also exhibited a similar potency as $\mathbf{3 5}$; whereas, increasing the size of the $3^{\prime}$-substitution resulted in a slight reduction in affinity as evidenced by compound $\mathbf{3 6}$.

Bicyclic aromatic groups, such as $1^{\prime}$-naphthyl, $1^{\prime}-\left(4^{\prime}-\mathrm{Cl}-\right.$ naphthyl $)$, and $1^{\prime}-(5,6,7,8-$ tetrahydronaphthyl) as exemplified in compounds $\mathbf{4 2}, \mathbf{4 4}$, and $\mathbf{4 5}$, exhibited comparable submicromolar binding affinities compared to compound 37 . Interestingly, compound 43 containing a $2^{\prime}$-naphthyl moiety was 20 -fold less potent than its regioisomer $\mathbf{4 2}$, indicating that the attachment position to this group is important for optimal binding to Mcl-1. The largest reduction in potency was observed for compounds 47 and 48 with nitrogencontaining heterocycles ( $6^{\prime}$-quinolinyl and $4^{\prime}$-indole) showing a 190 - and 40 -fold lower affinity compared to $\mathbf{4 2}$. These observations indicate that binding in the lower part of the pocket is dictated by hydrophobic interactions and that polar moieties are not accommodated in the binding pocket of Mcl-1. Additional substitutions of the core indole nitrogen are also tolerated as shown in compounds 56-59. This SAR trend strongly suggests that size, substitution pattern, and hydrophobicity of the aromatic group are all essential for binding to Mcl-1. Interestingly, most of the optimal groups identified from the data set were identical to the initial Class II fragment hits, demonstrating that the preliminary fragment SAR can guide the optimization of the linked compounds.

The initial fragment SAR indicated that substitutions at the 4- or 6-position are beneficial. The same principles were applied to further optimize the potency of the linked compounds. As expected from the SAR of the fragment hits, $6-\mathrm{Cl}$ adducts $(\mathbf{5 2 - 5 5})$ displayed a potency increase. In contrast, 4-Cl analogs $\mathbf{( 4 9 - 5 1 )}$ exhibited mixed results, suggesting that the anchoring unit might cause a subtle alteration in the binding conformation of the core unit. Merged compounds comprising benzothiophene and benzofuran scaffolds revealed similar SAR patterns as the indole series.

An additional linking strategy was explored by attaching compounds to the 1-position of the indole core (Table 4). Compounds 69-71 containing the optimized linker and anchor units retained comparable potency as the corresponding 3 -substituted indoles. This demonstrates that the core indole can bind in a conformation flipped along its long axis as previously predicted from the SAR of the fragment hits. Finally, the importance of the 2-carboxylic
acid was evaluated by converting 71 to an amide as in compound 72. This single modification resulted in a dramatic reduction in potency of at least 3-orders of magnitude. In line with previous results, merged compounds showed selectivity over $\mathrm{Bcl}-2$ and $\mathrm{Bcl}-\mathrm{xL}$, respectively. Compound 53, for example, is one of the most potent inhibitors of Mcl-1 ( $\mathrm{K}_{\mathrm{i}}=$ 55 nM , Supporting Information, Figure S1) and displays a 16 -fold selectivity over Bcl-2 and 270-fold over Bcl-xL.

These results clearly demonstrate the effectiveness of fragment-based methods and a structure-guided fragment merging strategy as a very powerful approach for rapidly obtaining potent binders against target proteins. Additionally, the ability to translate the lessons learned from the SAR of the fragment hits to our merged compounds aided the optimization of our lead compounds.

## X-ray structures of merged compounds complexed to Mcl-1

The three-dimensional structures of compounds representing different series of merged compounds in complex with Mcl-1 were obtained by X-ray crystallography (Table 5). As shown in Figure 4 for a representative member of both the benzothiophene and the indole series, the merged compounds occupy both pockets identified by NMR. The Mcl-1 inhibitors bind in the BH 3 binding groove and explain why our compounds compete with BH3-peptides for binding to Mcl-1. The overall secondary structures of Mcl-1 in the Mcl-1/ inhibitor complexes did not change upon binding the small molecules. ${ }^{23,36}$ However, ligand binding does result in several important structural changes near the binding pocket (Figure 5A,B). In compound-bound structures, a pronounced bend was observed in helix 4 (a4) which helps accommodate the anchor unit of the ligand in the deep hydrophobic pocket, and places the top of a 4 closer to the ligand. In fact, the top of a 4 protrudes slightly into the BH3 binding cleft. This causes the loop connecting helix 4-5 to be pulled towards the ligand and allows changes in the charge-charge interactions of R263 (Figure 5C,D). The guanidinium group of R263 can now interact with the carboxylate moiety of the merged compounds. These changes are not easily anticipated from the inspection of Mcl-1 complexed with a peptide. In peptide-bound structures, a4 is straight which keeps the BH3 binding cleft and the connecting loop open to better accommodate the peptide. The side chain of R263 also adopts a different conformation so that it can interact with D218 of the peptide (Figure 5C). It is also evident that the peptide does not utilize the full capacity of the hydrophobic pocket. In the peptide structure, L213 is only occupying the very upper part of this deep pocket (Figure 6A).

In all of the structures obtained thus far, the hydrophobic unit of the Class II fragments binds in the lower part of the pocket (Figure 4). Furthermore, all core units derived from Class I fragments are positioned in the upper part of the pocket, directing the carboxylic acid towards R263, and forming an interaction with M231, A227, F228, and F270 via the sixmembered ring (Figure 4B).

The SAR of our merged compounds can now be rationalized from a more detailed analysis of the X-ray structures. Positions 1 and 7 of the core unit are surface exposed and, as shown, substitutions here have little or no effect on compound binding. In contrast, there is minimal space for any substitution at the 5-position which is consistent with results of the observed SAR. The results obtained when adding groups at the 4-position could be explained by slight differences in how the hydrophobic group and the linker are positioned. These subtle changes may affect how much room there is at the 4-position or how the entire core unit is positioned in the binding pocket. Insights regarding the improved affinity due to substitutions at the 6 -position can be derived from the X-ray structures. The up to 5 -fold increase in binding may be explained by the addition of a hydrophobic group, which fits between M231 and A227 in a previously unoccupied pocket (Figure 4B). Presumably the
chlorine keeps the core unit in a more favorable position so that the carboxylic acid can make a better interaction with R263 (Figure 5D). The latter interaction also explains the requirement of a negatively charged acidic group in our fragment hits as well as the loss of affinity upon amidation of the carboxylic acid (71, 72).

The SAR of the Class II hits was also analyzed in terms of the X-ray structures. From the structures, it is evident why hydrophobic groups are preferred over more hydrophilic aromatic ring systems containing a nitrogen atom (compound 47 and 48). The pocket is lined with numerous non-polar side chains, namely M231, L235, L246, V249, M250, L267, F270, V273, L290, and I294. An overlay of compounds $\mathbf{5 3}$ and $\mathbf{6 0}$ when bound to Mcl-1 show a precise overlap of the anchoring $3^{\prime}, 5^{\prime}$-diMe-4-Cl-phenyl group (Figure 4A), suggesting that this unit anchors the rest of the ligand in the binding pocket. This binding pocket is present in both $\mathrm{Bcl}-\mathrm{xL}$ and $\mathrm{Bcl}-2$; however, it is neither as deep nor as large as it is in Mcl-1. Accessing parts of this pocket not present in other members of the Bcl-2 family provides a possible explanation for the selectivity of our compounds for binding to specific Bcl-2 family proteins.

## SYNTHESIS

Compounds 21-26 and 60-68 were prepared as outlined in Scheme 1. Selective lithiation at the 3 -Me position of compounds $\mathbf{8 0 a}$-b with LDA at $-10^{\circ} \mathrm{C}$ followed by nucleophilic substitution with 1-(bromoalkyl)naphthylenes or 1-(2-bromoethoxy)naphthalene gave compounds 21-26 with varying linker length to the terminal 1-naphthyl group. 3-Methylbenzothiophene-2-carboxylic acid cores 82a-c containing 4-, 6- or 7-Cl substitutions were synthesized from the corresponding chloro-(2-fluorophenyl)ethanones (81a-c) by reacting with methyl thioglycolate under DBU followed by saponification of the produced esters. Compounds 82a-c and 80a were then subjected to the alkylation condition described above to yield compounds $\mathbf{6 0 - 6 6}$. Unlike unsubstituted benzothiophene 80a, it is necessary to employ alkyl iodides as substrates to increase yields due to reduced nucleophilicity of cores 82a-c by the substituted chloride. 5-Cl-Benzofuran 84 was prepared from 83 by reacting with methyl bromoacetate. Using similar conditions, compounds 67 and 68 were obtained.

Mcl-1 inhibitors containing indole core 27-59 were synthesized by the route shown in Scheme 2. Indoles 85a-c were prepared utilizing the modified Japp-Klingemann reaction. ${ }^{37}$ Selective reduction using excess $\mathrm{BH}_{3}$ gave the corresponding alcohols which were then treated with a variety of aromatic alcohols under Mitsunobu conditions to afford the corresponding ether 86. The ester was then saponified to give indoles 27-55. N -substituted indoles 56-59 were produced through alkylations of selected esters $\mathbf{8 6}$ followed by hydrolysis.

Class II fragments, binding in the deep hydrophobic pocket, can also be linked to the indole nitrogen as depicted in Scheme 3. Ethyl indole-2-carboxylate $\mathbf{8 8}$ was substituted at the 1position with ethyl 3-bromopropionate using $\mathrm{K}_{2} \mathrm{CO}_{3}$ as a base under reflux in good yield. The adduct 89 then underwent the same series of reactions to give compounds 69-71. Compound $\mathbf{7 1}$ was further functionalized by amidation to produce $\mathbf{7 2}$.

## DISCUSSION

Mcl-1 is one of the most frequently amplified genes in human cancers and a common resistance factor to treatment with current chemotherapeutic agents and existing $\mathrm{Bcl}-2$ family inhibitors (e.g., ABT-263 and ABT-199). ${ }^{4,38}$ Here, we describe the discovery of small-molecules that potently inhibit Mcl-1 using fragment-based methods and structure-
based design. The X-ray structures of our merged compounds complexed with Mcl-1 identify several opportunities to significantly improve the affinity of the current lead molecules.

Although current lead compounds are able to compete with BH3-containing peptides for binding to Mcl-1, crystal structures reveal that the compounds may not be maximally exploiting the available binding opportunities. For example, pockets occupied by the peptide residues V216, D218, and V220 are currently not occupied by the compounds (Figure 6). Mutation of V216, D218, and V220 to alanine reduces peptide binding affinity 50-, 5-, and 45 -fold, respectively, demonstrating the importance of these additional binding pockets. ${ }^{23}$ Expanding our lead compounds to fill more of the peptide interface could yield more potent Mcl-1 inhibitors. In support of this hypothesis is a comparison of one of our current lead molecules bound to Mcl-1 with ABT-737 complexed to Bcl-xL (Figure 6B). Our lead compounds are much smaller than ABT-737, and only use a fraction of the binding pockets.

To identify small molecules that bind to these additional pockets, fragment-based screens of Mcl-1 in the presence of one of our nanomolar leads represents a possible strategy for improving potency. By blocking the initial site, fragments could be identified that bind to these additional pockets of the protein. These fragments could be rapidly incorporated into new molecules to markedly improve binding affinity.

An important lesson learned from these studies is that the SAR observed with the fragment hits is mirrored in the merged compounds and quickly led to the design and synthesis of potent lead molecules. In addition, the selectivity profile of the fragment hits for binding to Bcl-2 family members is mimicked by the merged compounds. The affinity of initial Mcl-1 fragment hits against the closely related family members $\mathrm{Bcl}-\mathrm{xL}$ and $\mathrm{Bcl}-2$ revealed a 5- to 50 -fold selectivity for binding to Mcl-1 which is retained in the lead compounds.

In summary, this study demonstrates the successful targeting of Mcl-1 by fragment-based and structure-guided methods to quickly produce an initial set of high-affinity and Mcl-1 selective inhibitors. The compounds described here serve as starting points for the discovery of clinically useful Mcl-1 inhibitors for the treatment of cancer.

## EXPERIMENTAL SECTION

## Protein expression and purification

A codon-optimized gene sequence encoding residues 172-327 of human Mcl-1 (Uniprot: Q07820) was purchased (Genscript) and cloned into a Gateway entry vector (pDONR-221, Invitrogen) using the protocols provided (Supporting Information, Table S1). This construct was further sub-cloned into an expression vector (pDEST-HisMBP) containing a maltose binding protein (MBP) tag for increased solubility, a tobacco etch virus (TEV) protease recognition site for tag-removal and an N-terminal His-tag to facilitate purification. The integrity of all plasmids was checked by sequencing. Soluble Mcl-1 protein was expressed in Escherichia coli BL21 CodonPlus (DE3) RIL (Stratagene) using ampicillin and chloramphenicol for selection.

In brief, a colony from a fresh transformation plate was picked to inoculate 100 mL of LB medium $\left(37^{\circ} \mathrm{C}\right)$. The overnight culture was used to start a 10 L fermentation (BioFlo 415, New Brunswick Scientific) grown at $37^{\circ} \mathrm{C}$. For NMR studies, uniformly ${ }^{15} \mathrm{~N}$ and ${ }^{15} \mathrm{~N} /{ }^{13} \mathrm{C}$ isotopically labeled protein samples were produced in minimal M9 media, where ${ }^{15} \mathrm{NH}_{4} \mathrm{Cl}$ and $\left[\mathrm{U}-{ }^{13} \mathrm{C}\right]$-D-glucose were used as sole nitrogen and carbon sources (Cambridge Isotope Laboratories). When the cell density corresponded to $\mathrm{OD}_{600}=2$, the temperature was lowered to $20^{\circ} \mathrm{C}$. After one hour, protein expression was induced with 0.5 mM IPTG. Cells
were harvested after 16 h by centrifugation. Pellets were frozen and re-dissolved in lysis buffer ( 20 mM TRIS pH 7.5, $300 \mathrm{mM} \mathrm{NaCl}, 20 \mathrm{mM}$ imidazole, 5 mM BME), approximately $100 \mathrm{~mL} / 10 \mathrm{~g}$ pellet, before the cells were broken by homogenization (APV-2000, APV). Prior to application to an affinity column ( 140 mL , ProBond, Invitrogen), lysate was cleared by centrifugation $(18,000 \mathrm{rpm})$ and filtration $(0.44 \mu \mathrm{M})$. Bound protein was washed on the column and then eluted by a gradient ( 20 mM TRIS $\mathrm{pH} 7.5,300 \mathrm{mM} \mathrm{NaCl}, 500 \mathrm{mM}$ imidazole, 5 mM BME). To enhance TEV protease cleavage, samples were buffer exchanged ( 50 mM TRIS $\mathrm{pH} 7.5,100 \mathrm{mM} \mathrm{NaCl}, 5 \mathrm{mM} \mathrm{BME}$ ) on three serially connected columns (HiPrep 26/10 Desalting, GE Healthcare). TEV protease was added to a molar ratio of 1:10 (TEV:Mcl-1) and incubated at room temperature until cleavage was complete. After adding 20 mM imidazole to the samples, they were passed over a subtractive second nickelcolumn ( 120 mL , Ni-NTA Superflow, Qiagen) to remove the MBP-tag, non-cleaved protein, and TEV protease. Mcl-1 protein for NMR screening was buffer exchanged into an optimized NMR buffer ( 25 mM sodium phosphate $\mathrm{pH} 6.3,25 \mathrm{mM} \mathrm{NaCl}, 1 \mathrm{mM}$ DTT, $0.01 \%$ $\mathrm{NaN}_{3}$ ). To achieve highly pure samples (e.g. for crystal screening), a supplementary step of size-exclusion chromatography (HiLoad 26/60, Superdex 75, GE Healthcare) was implemented. The running buffer also acted as the Mcl-1 storage and crystallography buffer ( 20 mM HEPES $\mathrm{pH} 6.8,50 \mathrm{mM} \mathrm{NaCl}, 3 \mathrm{mM}$ DTT, $0.01 \% \mathrm{NaN}_{3}$ ). Purifications were done at $4^{\circ} \mathrm{C}$, and concentration steps were performed in stirred ultrafiltration cells (Amicon, Millipore).

To improve protein sample quality and to increase crystal diffraction, protein mutants were created by site-directed mutagenesis (QuikChange, Agilent Technologies). Primers for a Cterminal deletion $(\Delta 5)$ were designed using their online tool and ordered from Eurofins MWG Operon. Mutations were made on the entry vector above, analyzed by in-house sequencing, and subsequently transferred into the pDEST-HisMBP expression vector. Mutant proteins were purified in the same way as wild-type (WT) proteins.

## NMR experiments: Fragment screening and NOE-guided fragment docking

Nuclear magnetic resonance (NMR) screening experiments were performed at $30^{\circ} \mathrm{C}$ using either a Bruker Avance III $500-\mathrm{MHz}$ or $600-\mathrm{MHz}$ NMR spectrometer equipped with a $5-\mathrm{mm}$ single-axis z-gradient cryo-probe and a Bruker Sample Jet sample changer. Twodimensional, gradient-enhanced ${ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}$ heteronuclear multiple-quantum coherence (SOFAST-HMQC) spectra ( 32 scans, $\sim 12 \mathrm{~min}$ ) were used to track shift changes upon ligand binding. ${ }^{39}$ Spectra were processed and analyzed in Topspin v3.1 (Bruker BioSpin). A fragment library comprising $\sim 13,800$ compounds which generally satisfy the "Rule of Three" (MW $\leq 300$, cLogP $\leq 3.0$, no more than 3 hydrogen bond donors) ${ }^{40}$, and having no more than 4 rotatable bonds was screened as mixtures of 12 fragments. Samples ( $500 \mu \mathrm{~L}$ ) contained $50 \mu \mathrm{M}{ }^{15} \mathrm{~N}$-labeled Mcl-1, $400 \mu \mathrm{M}$ of each fragment and $5 \%$ d6-DMSO. Deconvolution of hit mixtures was performed as single fragments. Recycling of the protein used for screening was accomplished by a simple buffer exchange with a yield of 50-80\%.

NOE-derived distance restraints were acquired to enable NMR-based docking of fragments into a previously determined X-ray structure of a Mcl-1/peptide complex. Spectra were recorded on a Bruker $800-\mathrm{MHz}$ spectrometer equipped with a cryo-probe and pulsed field gradients. $300 \mu \mathrm{M}{ }^{15} \mathrm{~N} /{ }^{13} \mathrm{C}$-labeled samples of Mcl-1 was prepared in a $\mathrm{D}_{2} \mathrm{O}$-based NMR buffer and mixed with selected fragments at a concentration of 1 mM . Side-chain ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR signals were assigned from ${ }^{13} \mathrm{C}$-edited NOE and HCCH-TOCSY experiments. ${ }^{41}$ NOE distance restraints were obtained from three-dimensional ${ }^{13} \mathrm{C}$-edited NOESY spectra, as well as three-dimensional ${ }^{15} \mathrm{~N}$ - and ${ }^{13} \mathrm{C}$-filter/edited NOESY spectra acquired with a mixing time of 80 ms . Compounds were docked into a X-ray structure of a Mcl-1/BIM peptide complex (PDB: 3KJ2) using the NMR-derived restraints and a simulated annealing protocol using the program Xplor-NIH. ${ }^{35,42}$ A square-well potential $\left(\mathrm{F}_{\mathrm{NOE}}=50\right.$
kcal $\mathrm{mol}^{-1}$ ) was employed to constrain NOE-derived distances. Five low energy models from this process were energy minimized using MOE 2011.10 (Chemical Computing Group Inc., Montreal). The lowest energy structures obtained by this method were consistent with the observed NOEs.

## Protein crystallization, data collection and structure refinement

Fresh batches of Mcl-1 proteins, WT and $\Delta 5$, were concentrated to $600 \mu \mathrm{M}(10.7 \mathrm{mg} / \mathrm{mL})$ and $1 \mathrm{mM}(17.4 \mathrm{mg} / \mathrm{mL})$, respectively, and screened for crystallization conditions with a 1.2 x excess of ligand. Crystals were obtained by mixing $1 \mu \mathrm{~L}$ protein with $1 \mu \mathrm{~L}$ reservoir solution ( $25-30 \%$ PEG $3350,0.1 \mathrm{M}$ Bis-TRIS pH $6.5,0.2 \mathrm{M} \mathrm{MgCl}_{2}$ ) as a hanging drop at $4^{\circ} \mathrm{C}$ or $18^{\circ} \mathrm{C}$. Crystals appeared within the first week and were flash frozen in liquid nitrogen after cryoprotection using 10-20\% glycol.

Data were collected on the Life Sciences Collaborative Access Team (LS-CAT) 21-ID-D beamline at the Advanced Photon Source (APS), Argonne National Laboratory. Indexing, integration and scaling was performed with HKL2000. ${ }^{43}$ Using a previously determined peptide-bound structure (PDB: 3MK8), phasing was done by molecular replacement with Phaser ${ }^{44}$ as implemented in CCP4 ${ }^{45}$. Refinement of the structural models were performed with Phenix ${ }^{46}$ and Refmac ${ }^{47}$, and included rounds of manual model building in COOT $^{48}$. Figures were prepared in PyMOL ${ }^{49}$ and MOE.

## FPA Competition Assays

Fluorescein isothiocyanate (FITC)-labeled Mcl-1-BH3 peptide (FITC-AHx-KALETLRRVGDGVQRNHETAF-NH2) ${ }^{23}$ and FITC-Bak-BH3 peptide (FITC-AHx-GQVGRQLAIIGDDINR-NH2) were purchased from GenScript and used without further purification. FPA measurements were carried out in 384-well, black, flat-bottom plates (Greiner Bio-One) using the EnVision plate reader (PerkinElmer). All assays were conducted in assay buffer containing 20 mM TRIS $\mathrm{pH} 7.5,50 \mathrm{mM} \mathrm{NaCl}, 3 \mathrm{mM}$ DTT, and $2 \%$ DMSO. To measure inhibition of the Mcl-1/FITC-Mcl-1-BH3 interaction, Mcl-1 and FITC-Mcl-1-BH3 peptide were each added at 250 nM . To measure inhibition of the FITC-Bak-BH3 interaction with Bcl-2 family members, 10nM FITC-Bak-BH3 peptide was incubated with either 14 nM Mcl-1, $4 \mathrm{nM} \mathrm{Bcl-xL}$, or $40 \mathrm{nM} \mathrm{Bcl-2} .\mathrm{For} \mathrm{IC}_{50}$ determination, compounds were diluted in DMSO in a 16-point, 2 -fold serial dilution scheme, added to assay plates, and incubated for 1 hour at room temperature. The change in anisotropy was measured and used to calculate an $\mathrm{IC}_{50}$ (inhibitor concentration at which $50 \%$ of bound peptide is displaced), by fitting the inhibition data using XLFit software (Guildford, UK), to a single-site binding model. This was converted into a binding dissociation constant $\left(\mathrm{K}_{\mathrm{i}}\right)$ according to the formula ${ }^{50}$ :

$$
\mathrm{K}_{\mathrm{i}}=[\mathrm{I}]_{50} /\left([\mathrm{L}]_{50} / \mathrm{K}_{\mathrm{d}}^{\mathrm{pep}}+[\mathrm{P}]_{0} / \mathrm{K}_{\mathrm{d}}^{\mathrm{pep}}+1\right)
$$

where $[\mathrm{I}]_{50}$ is the concentration of the free inhibitor at $50 \%$ inhibition, $[\mathrm{L}]_{50}$ is the concentration of the free labeled ligand at $50 \%$ inhibition, $[\mathrm{P}]_{0}$ is the concentration of the free protein at $0 \%$ inhibition and $\mathrm{K}_{\mathrm{d}}{ }^{\text {pep }}$ represents the dissociation constant of the FITClabeled peptide probe.

## CHEMISTRY

General
All NMR spectra were recorded at room temperature on a 400 MHz AMX Bruker spectrometer. ${ }^{1} \mathrm{H}$ chemical shifts are reported in $\delta$ values in ppm downfield with the
deuterated solvent as the internal standard. Data are reported as follows: chemical shift, multiplicity ( $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{br}=$ broad, $\mathrm{m}=$ multiplet ), integration, coupling constant $(\mathrm{Hz})$. Low resolution mass spectra were obtained on an Agilent 1200 series 6140 mass spectrometer with electrospray ionization. All samples were of $\geqslant 95 \%$ purity as analyzed by LC-UV/vis-MS. Analytical HPLC was performed on an Agilent 1200 series with UV detection at 214 and 254 nm along with ELSD detection. LC/ MS parameters were as follows: Phenomenex-C18 Kinetex column, $50 \times 2.1 \mathrm{~mm}, 2 \mathrm{~min}$ gradient, $5 \%(0.1 \% \mathrm{TFA} / \mathrm{MeCN}) / 95 \%\left(0.1 \% \mathrm{TFA} / \mathrm{H}_{2} \mathrm{O}\right)$ to $100 \%(0.1 \% \mathrm{TFA} / \mathrm{MeCN})$. Preparative purification was performed on a Gilson HPLC (Phenomenex-C18, $100 \times 30$ $\mathrm{mm}, 10 \mathrm{~min}$ gradient, $5 \rightarrow 95 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$ with $0.1 \% \mathrm{TFA}$ ) or by automated flash column chromatography (Isco, Inc. 100sg Combiflash). Solvents for extraction, washing, and chromatography were HPLC grade. All reagents were purchased from chemical suppliers and used without purification.

## 3-(2-(naphthalen-1-yl)ethyl)benzo[b]thiophene-2-carboxylic acid (21): General procedure for alkylation of 3-methylbenzothiophene or 3-methylbenzofuran-2-carboxylic acid

To a stirred solution of 3-methylbenzo[b]thiophene-2-carboxylic acid ( $101 \mathrm{mg}, 0.53 \mathrm{mmol}$ ) in anhydrous THF ( 5.0 mL ) was added LDA ( $0.58 \mathrm{~mL}, 1.17 \mathrm{mmol}, 2 \mathrm{M}$ in THF) dropwise under Ar at $-10^{\circ} \mathrm{C}$. The reaction mixture was stirred for 30 min then a solution of 1(bromomethyl)naphthalene ( $116 \mathrm{mg}, 0.525 \mathrm{mmol}$ ) in THF ( 2 mL ) was added dropwise. The reaction mixture was stirred for an additional 1 h at $-10^{\circ} \mathrm{C}$ then warm to room temperature and stirred overnight. The reaction was quenched by addition of saturated $\mathrm{NH}_{4} \mathrm{Cl}$ aq. solution and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{~mL})$. The organic extracts were combined, concentrated and the residue was purified by reverse phase prep HPLC to give the product as a solid $(103 \mathrm{mg}, 0.310 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.37(\mathrm{~d}, J=8.2$ $\mathrm{Hz}, 1 \mathrm{H}), 8.02(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.93(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{dd}, J=7.2,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $7.53(\mathrm{~m}, 3 \mathrm{H}), 7.44(\mathrm{~m}, 3 \mathrm{H}), 3.61(\mathrm{~m}, 2 \mathrm{H}), 3.33(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=$ $333.3(\mathrm{M}+\mathrm{H})^{+}$.

## 3-(3-(naphthalen-1-yl)propyl)benzo[b]thiophene-2-carboxylic acid (22)

The general procedure for alkylation was followed using 3-methylbenzo[b]thiophene-2carboxylic acid ( $86 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and 1-(2-bromoethyl)naphthalene ( $105 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) to yield the title compound ( $78 \mathrm{mg}, 0.23 \mathrm{mmol}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}-\mathrm{DMSO}$ ): $\delta(\mathrm{ppm})$ $7.96(\mathrm{~m}, 2 \mathrm{H}), 7.88(\mathrm{~m}, 2 \mathrm{H}), 7.75(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~m}, 3 \mathrm{H}), 7.40(\mathrm{~m}, 3 \mathrm{H}), 3.41(\mathrm{t}, J=$ $7.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.15(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 1.99(\mathrm{p}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}) ;>98 \%$ @ 215 nm , MS (ESI) $\mathrm{m} / \mathrm{z}=369.1(\mathrm{M}+\mathrm{Na})^{+}$.

## 3-(3-(naphthalen-1-yloxy)propyl)benzo[b]thiophene-2-carboxylic acid (23)

The general procedure for alkylation was followed using 3-methylbenzo[b]thiophene-2carboxylic acid ( $100 \mathrm{mg}, 0.52 \mathrm{mmol}$ ) and 1-(2-bromoethoxy)naphthalene ( $131 \mathrm{mg}, 0.52$ $\mathrm{mmol})$ to yield the title compound $(101 \mathrm{mg}, 0.28 \mathrm{mmol}){ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{d}_{6}-\mathrm{DMSO}\right): ~ \delta$ $(\mathrm{ppm}) 8.18(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.01(\mathrm{t}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.85(\mathrm{~m}, 1 \mathrm{H}), 7.48(\mathrm{~m}, 4 \mathrm{H}), 7.37(\mathrm{~m}$, $2 \mathrm{H}), 6.89(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.21(\mathrm{t}, J=5.9,2 \mathrm{H}), 3.54(\mathrm{t}, J=7.6,2 \mathrm{H}), 2.20(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=363.1(\mathrm{M}+\mathrm{H})^{+}$.

## 3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)benzo[b]thiophene-2-carboxylic acid (60)

The general procedure for alkylation was followed using 3-methylbenzo[b]thiophene-2carboxylic acid ( $97 \mathrm{mg}, 0.50 \mathrm{mmol}$ ) and 5-(2-bromoethoxy)-2-chloro-1,3-dimethylbenzene $(133 \mathrm{mg}, 0.50 \mathrm{mmol})$ to yield the title compound ( $89 \mathrm{mg}, 0.24 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{d}_{6}$-DMSO): $\mathcal{S}(\mathrm{ppm}) 8.00(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.96(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{t}, J=7.5 \mathrm{~Hz}$,
$1 \mathrm{H}), 7.42(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~s}, 2 \mathrm{H}), 3.98(\mathrm{t}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.38(\mathrm{t}, J=7.5,2 \mathrm{H}), 2.26$ $(\mathrm{s}, 6 \mathrm{H}), 2.01(\mathrm{p}, J=7.2,2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=375.1(\mathrm{M}+\mathrm{H})^{+}$.

## 4-chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)benzo[b]thiophene-2-carboxylic acid

 (61)The general procedure for alkylation was followed using 4-chloro-3-methylbenzo[b]thiophene-2-carboxylic acid ( $57 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 2-chloro-5-(2-iodoethoxy)-1,3-dimethylbenzene ( $78 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) to yield the title compound ( 25.6 mg , $0.063 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.01(\mathrm{dd}, J=7.7,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.49$ $(\mathrm{m}, 2 \mathrm{H}), 6.73(\mathrm{~s}, 2 \mathrm{H}), 4.05(\mathrm{t}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.70(\mathrm{~m}, 2 \mathrm{H}), 2.26(\mathrm{~s}, 6 \mathrm{H}), 2.05(\mathrm{~m}, 2 \mathrm{H})$; $>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=409.1(\mathrm{M}+\mathrm{H})^{+}$.

## 4-chloro-3-(3-(naphthalen-1-yloxy)propyl)benzo[b]thiophene-2-carboxylic acid (62)

The general procedure for alkylation was followed using 4-chloro-3-methylbenzo[b]thiophene-2-carboxylic acid ( $57 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 1-(2iodoethoxy)naphthalene ( $75 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) to yield the title compound $(9.9 \mathrm{mg}, 0.025$ mmol). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.15(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.02(\mathrm{~m}, 1 \mathrm{H})$, $7.85(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~m}, 6 \mathrm{H}), 6.96(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.28(\mathrm{t}, J=5.9,2 \mathrm{H}), 3.87$ $(\mathrm{m}, 2 \mathrm{H}), 2.23(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=419.1(\mathrm{M}+\mathrm{Na})^{+}$.

6-chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)benzo[b]thiophene-2-carboxylic acid (63)

The general procedure for alkylation was followed using 6-chloro-3-methylbenzo[b]thiophene-2-carboxylic acid ( $57 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 2-chloro-5-(2-iodoethoxy)-1,3-dimethylbenzene ( $78 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) to yield the title compound ( 36 mg , $0.087 \mathrm{mmol})^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.18(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}) 7.97(\mathrm{~d}, J=$ $8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.45(\mathrm{dd}, J=8.7,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~s}, 2 \mathrm{H}), 3.96(\mathrm{t}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.36$ (m, $2 \mathrm{H}), 2.26(\mathrm{~s}, 6 \mathrm{H}), 2.00(\mathrm{p}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=431.1$ (M $+\mathrm{Na})^{+}$.

6-chloro-3-(3-(naphthalen-1-yloxy)propyl)benzo[b]thiophene-2-carboxylic acid (64)
The general procedure for alkylation was followed using 6-chloro-3-methylbenzo[b]thiophene-2-carboxylic acid ( $57 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 1-(2iodoethoxy)naphthalene ( $75 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) to yield the title compound $(9.9 \mathrm{mg}, 0.025$ mmol). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.17$ (d, $\left.J=1.9 \mathrm{~Hz}, 1 \mathrm{H}\right), 8.12$ (d, $J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 8.02(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.84(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~m}, 3 \mathrm{H}), 7.38(\mathrm{~m}, 2 \mathrm{H}), 6.89$ (d, $J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{t}, J=5.9,2 \mathrm{H}), 3.52(\mathrm{t}, J=7.6,2 \mathrm{H}), 2.18(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ 215 $\mathrm{nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=419.1(\mathrm{M}+\mathrm{Na})^{+}$.

7-chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)benzo[b]thiophene-2-carboxylic acid (65)

The general procedure for alkylation was followed using 7-chloro-3-methylbenzo[b]thiophene-2-carboxylic acid ( $57 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 2-chloro-5-(2-iodoethoxy)-1,3-dimethylbenzene ( $78 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) to yield the title compound ( 53 mg , $0.13 \mathrm{mmol}){ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 7.97(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}) 7.64(\mathrm{~d}, J=$ $7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~s}, 2 \mathrm{H}), 3.97(\mathrm{t}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.38(\mathrm{~m}, 2 \mathrm{H})$, $2.25(\mathrm{~s}, 6 \mathrm{H}), 2.02(\mathrm{p}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}) ; \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=409.1(\mathrm{M}+\mathrm{H})^{+}$.

7-chloro-3-(3-(naphthalen-1-yloxy)propyl)benzo[b]thiophene-2-carboxylic acid (66)
The general procedure for alkylation was followed using 7-chloro-3-methylbenzo[b]thiophene-2-carboxylic acid ( $57 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 1-(2-
iodoethoxy)naphthalene ( $75 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) to yield the title compound ( $25 \mathrm{mg}, 0.062$ mmol) ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.05(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.02(\mathrm{~d}, J=8.1$ $\mathrm{Hz}, 1 \mathrm{H}), 7.84(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~m}, 1 \mathrm{H}), 7.41(\mathrm{~m}, 4 \mathrm{H}), 6.88$ $(\mathrm{d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{t}, J=5.8,2 \mathrm{H}), 3.54(\mathrm{t}, J=7.5,2 \mathrm{H}), 2.21(\mathrm{p}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}) ; \mathrm{MS}$ $(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=397.0(\mathrm{M}+\mathrm{H})^{+}$.

## 3-(3-(naphthalen-1-yl)propyl)benzofuran-2-carboxylic acid (24)

The general procedure for alkylation was followed using 3-methylbenzofuran-2-carboxylic acid and 1-(bromomethyl)naphthalene to yield the title compound. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 8.26(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.91(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.75(\mathrm{~d}, 1 \mathrm{H}, J=8.2 \mathrm{~Hz})$, 7.32-7.62 (m, 8H), 3.48-3.59 (m, 4H); >98\% @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=317.0(\mathrm{M}+\mathrm{H})^{+}$.

3-(2-(naphthalen-1-yl)ethyl)benzofuran-2-carboxylic acid (25)
The general procedure for alkylation was followed using 3-methylbenzofuran-2-carboxylic acid and 1-(2-bromoethyl)naphthalene to yield the title compound. ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 8.26(\mathrm{~d}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}), 7.32-7.97(\mathrm{~m}, 10 \mathrm{H}), 3.42-3.59(\mathrm{~m}, 2 \mathrm{H}), 3.17-3.36$ $(\mathrm{m}, 2 \mathrm{H}), 2.18-2.30(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=331.0(\mathrm{M}+\mathrm{H})^{+}$.

## 3-(3-(naphthalen-1-yloxy)propyl)benzofuran-2-carboxylic acid (26)

The general procedure for alkylation was followed using 3-methylbenzofuran-2-carboxylic acid and 1-(2-bromoethoxy)naphthalene to yield the title compound. ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 8.26(\mathrm{~d}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}), 7.32-7.97(\mathrm{~m}, 10 \mathrm{H}), 4.01(\mathrm{~m}, 2 \mathrm{H}), 3.32(\mathrm{~m}, 2 \mathrm{H})$, $2.28(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=347.2(\mathrm{M}+\mathrm{H})^{+}$.

## 3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)benzofuran-2-carboxylic acid (67)

The general procedure for alkylation was followed using 3-methylbenzofuran-2-carboxylic acid and 2-chloro-5-(2-bromoethoxy)-1,3-dimethylbenzene to yield the title compound. ${ }^{1} \mathrm{H}$ NMR ( $\left.400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 7.71(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.49-7.54(\mathrm{~m}, 1 \mathrm{H}), 7.28-7.34(\mathrm{~m}, 1 \mathrm{H}), 6.62(\mathrm{~s}, 2 \mathrm{H}), 3.99(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.32(\mathrm{t}, J=7.4$ $\mathrm{Hz}, 2 \mathrm{H}), 2.34(\mathrm{~s}, 6 \mathrm{H}), 2.23(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=359.1(\mathrm{M}+\mathrm{H})^{+}$.

5-chloro-3-(3-(naphthalen-1-yloxy)propyl)benzofuran-2-carboxylic acid (68)
The general procedure for alkylation was followed using 5-chloro-3-methylbenzofuran-2carboxylic acid ${ }^{51}$ and 1-(2-bromoethoxy)naphthalene to yield the title compound. $>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=381.1(\mathrm{M}+\mathrm{H})^{+}$.

## 3-(3-phenoxypropyl)-1H-indole-2-carboxylic acid (27): General procedure for phenol analogue coupling and saponification

To a solution of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate ( $70 \mathrm{mg}, 0.28 \mathrm{mmol}$ ), $\mathrm{PPh}_{3}(110 \mathrm{mg}, 0.51 \mathrm{mmol})$ and phenol ( $49 \mathrm{mg}, 0.52 \mathrm{mmol}$ ) in THF ( 3.5 mL ) was added Dt$\mathrm{BuAD}(99 \mathrm{mg}, 0.51 \mathrm{mmol})$ at $20^{\circ} \mathrm{C}$. The reaction mixture was stirred for 15 h at $20^{\circ} \mathrm{C}$ then concentrated in vacuo. The residue was purified by flash chromatography (Combi-flash Rf Hexane/EtOAc gradient $0-10 \%$ ) to give ethyl 3-(3-phenoxypropyl)-1 $H$-indole-2-carboxylic acid the title compound ( $92 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) as a colorless oil. $>98 \%$ @ 215 nm , MS (ESI) $\mathrm{m} / \mathrm{z}=324.1(\mathrm{M}+\mathrm{H})^{+}$.

To a solution of the ethyl ester ( $28 \mathrm{mg}, 0.085 \mathrm{mmol}$ ) in $\mathrm{EtOH}(1.0 \mathrm{~mL})$ was added $50 \%$ $\mathrm{NaOH} \mathrm{H} \mathrm{H}_{2} \mathrm{O}$ solution $(50 \mu \mathrm{~L})$ at $20^{\circ} \mathrm{C}$. The reaction mixture was stirred for 15 h at $20^{\circ} \mathrm{C}$. The reaction mixture was acidified with 1 N HCl solution, extracted with EtOAc, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. The crude product was purified by reverse phase prep. HPLC $\left(\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}\right.$ gradient to $\left.95 \% \mathrm{CH}_{3} \mathrm{CN} 0.5 \% \mathrm{TFA}\right)$ to yield the title compound
$(23 \mathrm{mg}, 0.078 \mathrm{mmol})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.43(\mathrm{~s}$, $1 \mathrm{H}), 7.64$. (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.20-7.31(\mathrm{~m}, 3 \mathrm{H}), 7.00(\mathrm{t}, J=7.2$ $\mathrm{Hz}, 1 \mathrm{H}), 6.89-6.99(\mathrm{~m}, 3 \mathrm{H}), 3.95(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.21(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.01-2.10$ (m, 2H) ; >98\% @215nm, MS (ESI) m/z = $296.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compounds 28-55 were prepared following the general procedure outlined above in library format. Purity of all final compounds was determined by HPLC analysis is $>95 \%$. All compounds were isolated as solids.

## 3-(3-(o-tolyloxy)propyl)-1 H-indole-2-carboxylic acid (28)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and $o$-cresol yielded $\mathbf{2 8} .{ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.39(\mathrm{~s}, 1 \mathrm{H}), 7.64 .(\mathrm{d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=$ $8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.13(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.67-$ $6.76(\mathrm{~m}, 3 \mathrm{H}), 3.94(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.20(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.25(\mathrm{~s}, 3 \mathrm{H}), 2.01-2.11(\mathrm{~m}$, $2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=310.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(2-(trifluoromethyl)phenoxy)propyl)-1H-indole-2-carboxylic acid (29)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H-indole-2-carboxylate and 2(trifluoromethyl)phenol yielded 29. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.43(\mathrm{~s}, 1 \mathrm{H})$, 7.64. (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.16-7.35(\mathrm{~m}$, $4 \mathrm{H}), 6.98(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.22(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.01-2.12$ $(\mathrm{m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=364.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-( $m$-tolyloxy)propyl)-1 H-indole-2-carboxylic acid (30)

Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and m-cresol yielded 30. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.39(\mathrm{~s}, 1 \mathrm{H}), 7.64$. (d, $\left.J=8.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 7.40$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.13(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, $6.67-6.76(\mathrm{~m}, 3 \mathrm{H}), 3.94(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.20(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.25(\mathrm{~s}, 3 \mathrm{H}), 2.01-$ $2.11(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=310.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(3-(trifluoromethyl)phenoxy)propyl)-1H-indole-2-carboxylic acid (31)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and 3(trifluoromethyl)phenol yielded 31. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.43(\mathrm{~s}, 1 \mathrm{H})$, 7.64. (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.16-7.35$ (m, $4 \mathrm{H}), 6.98(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.22(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.01-2.12$ (m, 2H) ; >98\% @215nm, MS (ESI) m/z=364.1 [M+H].

## 3-(3-(p-tolyloxy)propyl)-1H-indole-2-carboxylic acid (32)

Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and p-cresol yielded 32. ${ }^{1} \mathrm{H}$ NMR (400MHz, d ${ }_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.41(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.06(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.91(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.19(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.22(\mathrm{~s}, 3 \mathrm{H}), 1.98-2.09(\mathrm{~m}$, $2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=310.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-chlorophenoxy)propyl)-1H-indole-2-carboxylic acid (33)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and 4-chlorophenol yielded 33. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}$, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.99(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, $6.93(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.94(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.19(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.00-2.10(\mathrm{~m}$, $2 \mathrm{H}) ;>98 \%$ @ 215 nm, MS (ESI) m/z $=330.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-(trifluoromethyl)phenoxy)propyl)-1H-indole-2-carboxylic acid (34)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H-indole-2-carboxylate and 4(trifluoromethyl)phenol yielded 34. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.43(\mathrm{~s}, 1 \mathrm{H})$, $7.60-7.69(\mathrm{~m}, 3 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.08(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 6.99(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.22(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.03-2.13$ $(\mathrm{m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=364.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-chloro-3-methylphenoxy)propyl)-1H-indole-2-carboxylic acid (35)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and 4-chloro-3methylphenol yielded 35. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J$ $=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.18-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.00(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.85$ $(\mathrm{s}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.94(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.19(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.24(\mathrm{~s}$, 3H), 1.95-2.10 (m, 2H); >98\% @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=344.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-chloro-3-ethylphenoxy)propyl)-1H-indole-2-carboxylic acid (36)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H-indole-2-carboxylate and 4-chloro-3-ethylphenol yielded 36. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.26(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, \mathrm{J}=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{t}, \mathrm{J}=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 6.87(\mathrm{~s}, 1 \mathrm{H}), 6.76(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.95(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.20(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $2 \mathrm{H}), 2.61(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.95-2.10(\mathrm{~m}, 2 \mathrm{H}), 1.15(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}$, MS (ESI) m/z $=358.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1H-indole-2-carboxylic acid (37)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H-indole-2-carboxylate and 4-chloro-3,5dimethylphenol yielded 37. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}$, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.74(\mathrm{~s}, 2 \mathrm{H}), 3.93(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.19(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}), 1.95-2.10(\mathrm{~m}$, 2 H ) $;>98 \%$ @ 215 nm, MS (ESI) m/z $=358.1[\mathrm{M}+\mathrm{H}]^{+}$.

3-(3-([1,1'-biphenyl]-3-yloxy)propyl)-1H-indole-2-carboxylic acid (38)
Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and [1,1'-biphenyl]-3-ol yielded 38. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.43(\mathrm{~s}, 1 \mathrm{H}), 7.55-7.75(\mathrm{~m}, 3 \mathrm{H})$, $7.31-7.52(\mathrm{~m}, 5 \mathrm{H}), 7.25-7.46(\mathrm{~m}, 3 \mathrm{H}), 6.98-7.03(\mathrm{~m}, 1 \mathrm{H}), 6.91(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.05$ (t, $J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.23(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.01-2.10(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}$ (ESI) $\mathrm{m} / \mathrm{z}=372.1[\mathrm{M}+\mathrm{H}]^{+}$.

3-(3-([1,1'-biphenyl]-4-yloxy)propyl)-1H-indole-2-carboxylic acid (39)
Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and [1, $1^{\prime}$-biphenyl]-4-ol yielded 39. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.66(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.63-7.52(\mathrm{~m}, 4 \mathrm{H}), 7.35-7.45(\mathrm{~m}, 4 \mathrm{H}), 7.32(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, 6.95-7.05 (m, 2H), $4.01(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.23(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.03-2.13(\mathrm{~m}, 2 \mathrm{H})$; $>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}$ (ESI) m/z=372.2[M+H] ${ }^{+}$.

## 3-(3-(3-phenoxyphenoxy)propyl)-1H-indole-2-carboxylic acid (40)

Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and 3-phenoxyphenol yielded 40. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.61(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.33-7.42(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.31(\mathrm{~m}, 4 \mathrm{H}), 7.16(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.95-7.05(\mathrm{~m}, 3 \mathrm{H}), 6.69$ $(\mathrm{d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{~s}, 1 \mathrm{H}), 3.93(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.19(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.95-2.10$ (m, 2H); >98\% @215 nm, MS (ESI) m/z = $388.2[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-phenoxyphenoxy)propyl)-1H-indole-2-carboxylic acid (41)

Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and 4-phenoxyphenol yielded 41. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.64(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.32-7.45(\mathrm{~m}, 3 \mathrm{H}), 7.23(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.09(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.90-7.03(\mathrm{~m}, 7 \mathrm{H})$, 3.95 (t, $J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.21$ (t, $J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.02-2.10(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}$, MS $(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=388.2[\mathrm{M}+\mathrm{H}]^{+}$.

3-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylic acid (42)
Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and 1-naphthol yielded 42. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.60(\mathrm{~s}, 1 \mathrm{H}), 8.24(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.87(\mathrm{~d}$, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.30-7.55 .(\mathrm{m}, 5 \mathrm{H}), 7.25 .(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.95$ $(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.16(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.30-3.40(\mathrm{~m}, 2 \mathrm{H})$, 2.15-2.25 (m, 2H); >98\% @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=346.1[\mathrm{M}+\mathrm{H}]^{+}$.

3-(3-(naphthalen-2-yloxy)propyl)-1H-indole-2-carboxylic acid (43)
Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and 2-naphthol yielded 43. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.82(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.73(\mathrm{~d}$, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.67(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.35-7.52(\mathrm{~m}, 3 \mathrm{H}), 7.34(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.17$ $-7.27(2 \mathrm{H}, \mathrm{m}), 6.98(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.26(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H})$, 2.10-2.18 (m, 2H); >98\% @ 215 nm, MS (ESI) m/z=346.1[M+H] ${ }^{+}$.

3-(3-((4-chloronaphthalen-1-yl)oxy)propyl)-1H-indole-2-carboxylic acid (44)
Coupling of ethyl 3-(3-hydroxypropyl)-1H-indole-2-carboxylate and 4-chloronaphthalen-1ol yielded 44. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.47(\mathrm{~S}, 1 \mathrm{H}), 8.30(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $1 \mathrm{H}), 8.11$ (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.73$ (t, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.62-7.69$ (m, 2H), 7.56 (d, $J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.88(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.17(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.28-2,35(\mathrm{~m}, 2 \mathrm{H}), 2.17-2.27(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=380.1[\mathrm{M}+\mathrm{H}]^{+}$.

3-(3-((5,6,7,8-tetrahydronaphthalen-1-yl)oxy)propyl)-1H-indole-2-carboxylic acid (45)
Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and 5,6,7,8-tetrahydronaphthalen-1-ol yielded $45 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}$, $1 \mathrm{H}), 7.63$ (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.20-7.28(\mathrm{~m}, 1 \mathrm{H}), 6.92-7.07$ (m, $2 \mathrm{H}), 6.60-6.70(\mathrm{~m}, 2 \mathrm{H}), 3.94(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.23(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.69(\mathrm{t}, J=0.60$ $\mathrm{Hz}, 2 \mathrm{H}), 2.63(\mathrm{t}, J=0.60 \mathrm{~Hz}, 2 \mathrm{H}), 2.01-2.10(\mathrm{~m}, 2 \mathrm{H}), 1.63-1.80(\mathrm{~m}, 4 \mathrm{H}) ;>98 \%$ @215 $\mathrm{nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=350.2[\mathrm{M}+\mathrm{H}]^{+}$.

3-(3-((2,3-dihydro-1H-inden-5-yl)oxy)propyl)-1H-indole-2-carboxylic acid (46)
Coupling of ethyl 3-(3-hydroxypropyl)-1 H -indole-2-carboxylate and 2,3-dihydro-1 H -inden-5-ol yielded 46. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.42(\mathrm{~s}, 1 \mathrm{H}), 7.64(\mathrm{~d}, J=$ $8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, \mathrm{J}=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.08(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.03$ $(\mathrm{t}, \mathrm{J}=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~s}, 1 \mathrm{H}), 6.65-6.70(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.91(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.33$ (t, $J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.70-2.83(\mathrm{~m}, 4 \mathrm{H}), 1.90-2.10(\mathrm{~m}, 4 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}$ $=336.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(quinolin-6-yloxy)propyl)-1 H-indole-2-carboxylic acid (47)

Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and quinolin-6-ol yielded 47. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.47(\mathrm{~s}, 1 \mathrm{H}), 8.92(\mathrm{~d}, J=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.53(\mathrm{~d}$, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.04(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.61(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~s}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.21$

$$
(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.26(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}),
$$ 2.14-1.93 (m, 2H); >98\% @215 nm, MS (ESI) m/z = $347.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-((1H-indol-4-yl)oxy)propyl)-1H-indole-2-carboxylic acid (48)

Coupling of ethyl 3-(3-hydroxypropyl)-1 $H$-indole-2-carboxylate and quinolin-6-ol yielded 48. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.45(\mathrm{~s}, 1 \mathrm{H}), 11.04(\mathrm{~S}, 1 \mathrm{H}), 7.68(\mathrm{~d}, J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.15-7.30(\mathrm{~m}, 2 \mathrm{H}), 6.91-7.07(\mathrm{~m}, 3 \mathrm{H}), 6.48(\mathrm{~s}, 1 \mathrm{H}), 6.39$ (d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.26(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.09-2.20(\mathrm{~m}, 2 \mathrm{H}) ;$ $>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}$ (ESI) m/z = $335.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-chloro-3-(3-(4-chloro-3-methylphenoxy)propyl)-1 H-indole-2-carboxylic acid (49)

Coupling of ethyl 4-chloro-3-(3-hydroxypropyl)-1 H -indole-2-carboxylate ${ }^{37}$ and 4-chloro-3methylphenol yielded 49. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}){ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}{ }^{-}$ DMSO): $\delta 11.68$ (s, 1H), 7.66. (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.26$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.19$ (t, $J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.07(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.88-6.94(\mathrm{~m}, 1 \mathrm{H}), 6.69-6.78(\mathrm{~m}, 1 \mathrm{H}), 4.03(\mathrm{t}, J=6.4$ $\mathrm{Hz}, 2 \mathrm{H}), 3.45(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 3 \mathrm{H}), 2.01-2.08(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}$ $(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=378.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1 H-indole-2-carboxylic acid (50)

Coupling of ethyl 4-chloro-3-(3-hydroxypropyl)-1 H-indole-2-carboxylate ${ }^{37}$ and 4-chloro-3,5-dimethylphenol yielded 50. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.70(\mathrm{~s}$, $1 \mathrm{H})$, 7.67. (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.19(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~s}, 2 \mathrm{H})$, $4.20(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.44(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.28(\mathrm{~s}, 6 \mathrm{H}), 1.97-2.08(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=392.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-chloro-3-(3-(naphthalen-1-yloxy)propyl)-1 H-indole-2-carboxylic acid (51)

Coupling of ethyl 4-chloro-3-(3-hydroxypropyl)-1 H -indole-2-carboxylate ${ }^{37}$ and 1-naphthol yielded 51. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.68(\mathrm{~s}, 1 \mathrm{H}), 8.18(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, 7.86. (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.35-7.57(\mathrm{~m}, 5 \mathrm{H}), 7.21(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.09(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.94(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.69-6.78(\mathrm{~m}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.25(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.60(\mathrm{t}$, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.16-2.25(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=380.1[\mathrm{M}+\mathrm{H}]^{+}$.

6-chloro-3-(3-(4-chloro-3-methylphenoxy)propyl)-1H-indole-2-carboxylic acid (52)
Coupling of ethyl 6-chloro-3-(3-hydroxypropyl)-1 H-indole-2-carboxylate ${ }^{37}$ and 4-chloro-3methylphenol yielded 52. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.64(\mathrm{~s}, 1 \mathrm{H}), 7.66 .(\mathrm{d}, J$ $=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~s}, 1 \mathrm{H}), 7.26(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{~d}, J=$ $7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{~s}, 1 \mathrm{H}), 6.80(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.93(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.17(\mathrm{t}, J=7.2$ $\mathrm{Hz}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 3 \mathrm{H}), 1.95-2.08(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=378.1[\mathrm{M}+\mathrm{H}]^{+}$.

6-chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1H-indole-2-carboxylic acid (53)
Coupling of ethyl 6-chloro-3-(3-hydroxypropyl)-1 H-indole-2-carboxylate ${ }^{37}$ and 4-chloro-3,5-dimethylphenol yielded 53. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.60(\mathrm{~s}$, $1 \mathrm{H}), 7.66(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~s}, 1 \mathrm{H}), 7.02(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~s}, 2 \mathrm{H}), 3.92(\mathrm{t}, J=$ $6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.17(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}), 1.95-2.05(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}$ $(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=392.1[\mathrm{M}+\mathrm{H}]^{+}$.

6-chloro-3-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylic acid (54)
Coupling of ethyl 6-chloro-3-(3-hydroxypropyl)-1 H -indole-2-carboxylate ${ }^{37}$ and 1-naphthol yielded 54. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 11.62(\mathrm{~s}, 1 \mathrm{H}), 8.19(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, 7.86. (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.69$. (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.57(\mathrm{~m}, 5 \mathrm{H}), 6.97(\mathrm{~d}, J=8.0 \mathrm{~Hz}$,
$1 \mathrm{H}), 6.87(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.15(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.27-3-37(\mathrm{~m}, 2 \mathrm{H}), 2.16-2.25(\mathrm{~m}$, $2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=380.1[\mathrm{M}+\mathrm{H}]^{+}$.

6-chloro-3-(3-((5,6,7,8-tetrahydronaphthalen-1-yl)oxy)propyl)-1 H-indole-2-carboxylic acid (55)

Coupling of ethyl 6-chloro-3-(3-hydroxypropyl)-1 H-indole-2-carboxylate ${ }^{37}$ and 5,6,7,8-tetrahydronaphthalen-1-ol yielded 55. >98\% @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=384.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1-methyl-1H-indole-2-carboxylic acid (56): General Procedure for alkylation of indole NH

To a solution of ethyl 3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1H-indole-2-carboxylate $(15 \mathrm{mg}, 0.04 \mathrm{mmol})$ in DMF $(0.8 \mathrm{~mL})$ was added $\mathrm{NaH}(3.0 \mathrm{mg}, 0.075 \mathrm{mmol}, 60 \%$ in mineral oil) at $20^{\circ} \mathrm{C}$ under Ar. The reaction mixture was stirred for 20 min at $20^{\circ} \mathrm{C}$ then MeI ( $34 \mathrm{mg}, 0.24 \mathrm{mmol}$ ) was added. The reaction mixture was stirred for additional 1 h then quenched by addition of water. The reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. Combined organic layer was concentrated in vacuo, and the residue was purified by reverse phase prep HPLC to give the product as a solid ( $12 \mathrm{mg}, 0.03 \mathrm{mmol}$ ). $>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=$ $400.2[\mathrm{M}+\mathrm{H}]^{+}$.

The title compound was obtained by following the general procedure for saponification outlined above as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 7.67$ (d, $J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.52 .(\mathrm{d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.32$. (t, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~s}$, 2H), $3.91-3.99(\mathrm{~m}, 5 \mathrm{H}), 3.18$ (t, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.23(\mathrm{~s}, 6 \mathrm{H}), 1.96-2.05(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=372.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compounds 57-59 were prepared following the general procedure outlined above in library format. Purity of all final compounds was determined by HPLC analysis is $>95 \%$. All compounds were isolated as solids.

## 1-methyl-3-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylic acid (57)

Methylation of ethyl 3-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylate by MeI yielded 57. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta 8.22$ (d, $J=6.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.87(\mathrm{~d}, J=6.4 \mathrm{~Hz}$, $1 \mathrm{H}), 7.70(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.36-7.58 .(\mathrm{m}, 5 \mathrm{H}), 7.30 .(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.06(\mathrm{t}, J=7.6$ $\mathrm{Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.16(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.30-3.40(\mathrm{~m}, 2 \mathrm{H})$, 2.15-2.25 (m, 2H); >98\% @ $215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=360.1[\mathrm{M}+\mathrm{H}]^{+}$.

6-Chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1-methyl-1 H-indole-2-carboxylic acid (58)

Methylation of ethyl 6-chloro-3-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1 H -indole-2carboxylate by MeI yielded 58. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta 7.79-7.57(\mathrm{~m}, 2 \mathrm{H}), 7.07$ (dd, $J=8.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~s}, 2 \mathrm{H}), 3.97-3.89(\mathrm{~m}, 5 \mathrm{H}), 3.20-3.10(\mathrm{~m}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H})$, 2.06-1.93 (m, 2H); >98\% @ 215 nm , MS (ESI) $1.83 \mathrm{~min}, m / z=428[\mathrm{M}+\mathrm{Na}]^{+}$.

1-benzyl-3-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylic acid (59)
Benzylation of ethyl 3-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylate by and benzyl bromide yielded 59. ${ }^{1} \mathrm{H}$ NMR (400MHz, $\mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.18-8.24(\mathrm{~m}, 1 \mathrm{H})$, $7.87(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.72-7.80(\mathrm{~m}, 1 \mathrm{H}), 7.42-7.56(\mathrm{~m}, 3 \mathrm{H}), 7.35-7.40 .(\mathrm{m}, 1 \mathrm{H}), 7.15-$ 7.30. (m, 5H), 6.95-7.10 (m, 3H), 6.83-6.90(m, 1H), $5.84(\mathrm{~s}, 2 \mathrm{H}), 4.17(\mathrm{t}, J=6.0 \mathrm{~Hz}$, 2H), $3.30-3.40(\mathrm{~m}, 2 \mathrm{H}), 2.15-2.25(\mathrm{~m}, 2 \mathrm{H}) ;>98 \% @ 215 \mathrm{~nm}, \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}=436.2$ [M $+\mathrm{H}]^{+}$.

## 1-(3-(4-chloro-3-methylphenoxy)propyl)-1H-indole-2-carboxylic acid (69)

The title compound was obtained by following the general procedure for phenol analogue coupling and saponification outlined above using ethyl 1-(3-hydroxypropyl)-1H-indole-2carboxylate and 4-chloro-3-methylphenol as a white solid. >98\% @215nm, MS (ESI) m/z= $344.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 1-(3-(4-chloro-3,5-dimethylphenoxy)propyl)-1H-indole-2-carboxylic acid (70)

The title compound was obtained by following the general procedure for phenol analogue coupling and saponification outlined above using ethyl 1-(3-hydroxypropyl)-1H-indole-2carboxylate and 4-chloro-3,5-dimethylphenol as a white solid. $>98 \%$ @ 215 nm , MS (ESI) $\mathrm{m} / \mathrm{z}=357.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 1-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxylic acid (71)

The title compound was obtained by following the general procedure for phenol analogue coupling and saponification outlined above using ethyl 1-(3-hydroxypropyl)-1H-indole-2carboxylate and 1-naphthol as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}) 8.26$ (d, $J=7,2 \mathrm{~Hz}, 1 \mathrm{H}), 7.87(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.69(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.37-7.57$. (m, 4H), $7.35(\mathrm{t}, \mathrm{J}=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.15-7.25(\mathrm{~m}, 2 \mathrm{H}), 7.07 .(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 4.89(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.23(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.15-2.25(\mathrm{~m}, 2 \mathrm{H}) ;>98 \%$ @ 215 nm , MS (ESI) m/z = $346.1[\mathrm{M}+\mathrm{H}]^{+}$.

## 1-(3-(naphthalen-1-yloxy)propyl)-1H-indole-2-carboxamide (72)

To a solution of 1-(3-(naphthalen-1-yloxy)propyl)-1 H -indole-2-carboxylic acid ( 15 mg , 0.043 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.0 \mathrm{~mL})$ was added oxalyl chloride ( $37 \mu \mathrm{~L}, 0.43 \mathrm{mmol}$ ) followed by a drops of $\mathrm{N}, \mathrm{N}$-dimethylformamide at room temperature under Ar. The reaction mixture was stirred for 3 h at room temperature then concentrated under vacuum. The residue was redissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{ml})$ and conc. $\mathrm{NH}_{4} \mathrm{OH}$ aqueous solution $(0.67 \mathrm{ml})$ was added. The reaction mixture was stirred for 2 h then extracted with ethyl acetate. The organic layer was concentrated in vacuo, and the residue was purified by reverse phase prep HPLC to give the product as a white solid ( $11 \mathrm{mg}, 0.032 \mathrm{mmol}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO): $\delta(\mathrm{ppm}$ ) $8.24-8.30(\mathrm{~m}, 1 \mathrm{H}), 7.99(\mathrm{br}, 2 \mathrm{H}), 7.82-7.90(\mathrm{~m}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=8.0,1 \mathrm{H}), 7.47-7.60(\mathrm{~m}$, $3 \mathrm{H}), 7.42-7.46(\mathrm{~m}, 1 \mathrm{H}), 7.35-7.40 .(\mathrm{m}, 1 \mathrm{H}), 7.19(\mathrm{~s}, 1 \mathrm{H}), 7.14(\mathrm{t}, J=7,2 \mathrm{~Hz}, 1 \mathrm{H}), 7.06$ (t, $J=7,2 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{t}, J=7,2 \mathrm{~Hz}, 1 \mathrm{H}), 4.86(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.12(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H})$, 2.25-2.35 (m, 2H); >98\% @215nm, MS (ESI) m/z=345.1[M+H] .

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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## ABBREVIATIONS

| FBDD | fragment-based drug discovery |
| :--- | :--- |
| Mcl-1 | Myeloid cell leukemia 1 |
| Bcl-2 | B-cell lymphoma 2 |
| Bcl-xL | B-cell lymphoma-extra large |
| BH3 | Bcl-2 homology domain 3 |
| CSP | chemical shift perturbation |
| FITC | fluorescein isothiocyanate |
| FPA | fluorescence polarization anisotropy |

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Figure 1.
${ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N}$ HMQC spectra of Mcl-1 with (red) and without (black) fragment 3. The NMR sample contained $50 \mu \mathrm{M}{ }^{15} \mathrm{~N}$-labeled protein and 800 HM ligand.


Figure 2.
Model structures of Mcl-1 complexed to two different classes of fragment hits obtained using NMR-derived distance restraints. Residues (labeled) with NOEs to the fragments (blue) are rendered as sticks (gray). R263 is also labeled. (A) Class I: benzothiophene 5, and (B) Class II: tethered aromatic 17.


Figure 3.
Schematic illustrating the merging of fragment $\mathbf{2}$ and $\mathbf{1 7}$ to produce compound $\mathbf{6 0}$. The affinity of each molecule to Mcl-1 is shown as well as the ligand efficiencies.


Figure 4.
X-ray structures of merged compounds bound to Mcl-1. Compounds incorporating different scaffolds (benzothiophene and indole) occupy the same binding pocket and adopt very similar binding poses (inset). (A) Compounds 53 and 60 interact with Mcl-1 in the BH3peptide binding cleft between helices 3, 4, and 5. Surface depiction of Mcl-1 when complexed to (B) $\mathbf{6 0}$ and (C) $\mathbf{5 3}$ that illustrates how the merged compounds fill the pockets as predicted by NMR and interact with R263.


Figure 5.
Several structural differences between peptide-bound (red) and compound-bound (gray) Mcl-1 complexes were identified. The Mcl-1-derived BH3-peptide is depicted in green and compound 60 in blue. (A) To accommodate the hydrophobic fragments into the deep pocket, helix 4 (a4) is slightly kinked in the structures of Mcl-1/compound complexes compared to the Mcl-1/peptide structures. This leads to further rearrangements of the loop, pulling it towards the merged compound. (B) Close-up of the small-molecule binding-site in Mcl-1. Compared to the peptide-bound state, our merged compounds induce changes in the position and/or in rotamer states of nearby residues. (C-D) The network of charge-charge interactions and polar contacts involving R263 are dramatically different in the two types of complexes. (C) While bound to the peptide, R263 of Mcl-1 interacts closely with D218 of the 16 mer , a residue shown important by alanine-scanning. ${ }^{23}$ Moreover, R263 interacts with the sidechain of D256 and the backbone (bb) oxygen of V258 in Mcl-1. (D) When complexed with merged compounds (e.g. 60), this network is replaced by an interaction to the carboxylic acid of the small-molecule ligand and additional polar contacts to the backbone oxygen atoms of S255 and D256.


Figure 6.
(A) Comparison of Mcl-1 (gray surface) when complexed to compound 60 (blue) and a 16 mer BH3-peptide derived from Mcl-1 (green), (B) Overlay of the X-ray structures of Mcl-1/60 complex and Bcl-xL/ABT-737 (PDB: 2YXJ). The Mcl-1 protein is shown in gray, 60 in blue, and ABT-737 in green.



 $81 \mathrm{~b} R=4-\mathrm{Cl}$

2. LiOH
$\mathrm{H}_{2} \mathrm{O} / \mathrm{MeOH}$

82a $R=4-\mathrm{Cl}$
82b $R=6-\mathrm{Cl}$
82c $R=7-\mathrm{Cl}$
82c $R=7-C$

$X=\operatorname{Br}$ or $I$
-


Scheme 1.
Synthesis of benzothiophene- and benzofuran-containing compounds


Scheme 2.
Synthesis of indole-based compounds

88
89
90




Scheme 3.
Synthesis of compounds 69-72

Table 1
SAR of Class I and II fragment hits for binding to Bcl-2 family proteins

| Class I |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Structure | Comp. Substitutions |  | $\mathbf{K}_{\mathbf{i}}(\mu \mathbf{M})^{*}$ |  |  | LE (Mcl-1) |
|  |  |  | Mcl-1 | Bcl-xL | Bcl-2 |  |
|  | 1 | H | >1000 |  |  |  |
|  | 2 | $3-\mathrm{Cl}$ | 131 |  |  | 0.30 |
|  | 3 | $3-\mathrm{Cl}, 4-\mathrm{Cl}$ | 22 | >1000 | >1000 | 0.33 |
|  | 4 | $3-\mathrm{Cl}, 6-\mathrm{Cl}$ | 59 |  |  | 0.30 |
|  | 5 | 3-Cl, 6-Me | 40 | 230 | >1000 | 0.31 |
|  | 6 | H | >1000 |  |  |  |
|  | 7 | $3-\mathrm{Me}, 5-\mathrm{Cl}$ | 52 |  |  | 0.31 |
|  | 8 | $3-\mathrm{Me}, 6-\mathrm{Cl}$ | 23 |  |  | 0.33 |
|  | 9 | $3-\mathrm{Me}, 7-\mathrm{Me}$ | 102 | >1000 | >1000 | 0.29 |
|  | 10 | 3-Me, 6-Me, 7-Me | 214 | >1000 | >1000 | 0.24 |
|  | 11 | $4-\mathrm{Br}, 6-\mathrm{Cl}$ | 90 |  |  | 0.29 |
|  | 12 | H | >1000 |  |  |  |
|  | 13 | 1-Me | 160 |  |  | 0.29 |
|  | 14 | $1-\mathrm{Me}, 5-\mathrm{Br}$ | 81 |  |  | 0.29 |
|  | 15 | 3-Ph, 7-Me | 136 | $>1000$ | $>1000$ | 0.20 |


| Class II |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Comp. | Structure | $\mathbf{K}_{\mathbf{i}}(\mu \mathbf{M})^{*}$ |  |  | LE (Mcl-1) |
|  |  | Mcl-1 | Bcl-xL | Bcl-2 |  |
| 16 |  | 103 | >1000 | >1000 | 0.31 |

17

$60>1000>1000 \quad 0.26$

${ }^{*} \mathrm{~K}_{\mathrm{i}}$ values are measured in duplicate and average values are reported.
Analogs from the Vanderbilt compound library


19


20

Table 2
Optimization of linker length in the merged compounds

|  | Comp. | A | n | X | $\mathrm{K}_{\mathrm{i}}(\mu \mathrm{M})$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | Mcl-1 ${ }^{\text {a }}$ | Bcl-xL ${ }^{\text {b }}$ | Bcl-2 ${ }^{\text {b }}$ |
|  | 21 | S | 1 | $\mathrm{CH}_{2}$ | $12.5 \pm 5.6$ |  |  |
|  | 22 | S | 2 | $\mathrm{CH}_{2}$ | $9.6 \pm 3.0$ |  |  |
|  | 23 | S | 3 | O | $0.37 \pm 0.03$ | 15 | 9.7 |
|  | 24 | O | 1 | $\mathrm{CH}_{2}$ | $9.5 \pm 2.9$ |  |  |
|  | 25 | O | 2 | $\mathrm{CH}_{2}$ | $14.6 \pm 7.1$ |  |  |
|  | 26 | O | 3 | O | $1.0 \pm 0.44$ | >15 | 4.3 |

${ }^{a} K_{i}$ values are measured in triplicate
$b_{\mathrm{K}_{\mathrm{i}}}$ values are measured in duplicate

## Table 3

SAR of merged compounds containing an optimized 4-atom linker


| Comp. | A | R | Ar | $\mathrm{K}_{\mathrm{i}}(\mu \mathrm{M})$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | Mcl- ${ }^{\text {a }}$ | Bcl-xL ${ }^{\text {b }}$ | Bcl-2 ${ }^{\text {b }}$ |
| 27 | NH | H | Ph | $35 \pm 7.1$ |  |  |
| 28 | NH | H | 2-Me-Phenyl | $15 \pm 0.7$ |  |  |
| 29 | NH | H | 2- $\mathrm{CF}_{3}$-Phenyl | $8.8 \pm 1.7$ |  |  |
| 30 | NH | H | 3-Me-Phenyl | $1.9 \pm 0.6$ |  |  |
| 31 | NH | H | 3-CF3-Phenyl | $1.7 \pm 0.08$ |  |  |
| 32 | NH | H | 4-Me-Phenyl | $16 \pm 3.1$ |  |  |
| 33 | NH | H | 4-Cl-Phenyl | $9.8 \pm 2.9$ |  |  |
| 34 | NH | H | 4- $\mathrm{CF}_{3}$-Phenyl | $9.9 \pm 3.2$ |  |  |
| 35 | NH | H | 3-Me-4-Cl-Phenyl | $0.38 \pm 0.14$ | >15 | 9.1 |
| 36 | NH | H | 3-Et-4-Cl-Phenyl | $1.1 \pm 0.1$ |  |  |
| 37 | NH | H | 3,5-diMe-4-Cl-Phenyl | $0.30 \pm 0.15$ | >15 | 5.3 |
| 38 | NH | H | 3-(1,1'-biphenyl) | $7.7 \pm 2.1$ |  |  |
| 39 | NH | H | 4-(1,1'-biphenyl) | $7.6 \pm 0.7$ |  |  |
| 40 | NH | H | 3-Phenoxy-Phenyl | $5.2 \pm 0.3$ |  |  |
| 41 | NH | H | 4-Phenoxy-Phenyl | $6.4 \pm 0.6$ |  |  |
| 42 | NH | H | 1-Naphthyl | $0.33 \pm 0.10$ | >15 | 2.7 |
| 43 | NH | H | 2-Naphthyl | $7.5 \pm 0.6$ |  |  |
| 44 | NH | H | 1-(4-Cl-Naphthyl) | $0.48 \pm 0.01$ |  |  |
| 45 | NH | H | 1-(5,6,7,8-tetrahydronaphthyl) | $0.30 \pm 0.10$ |  |  |
| 46 | NH | H | 5-(2,3-dihydroindenyl) | $2.9 \pm 0.1$ |  |  |
| 47 | NH | H | 6-Quinolinyl | $62 \pm 17$ |  |  |
| 48 | NH | H | 4-Indolyl | $14 \pm 0.3$ |  |  |
| 49 | NH | 4-Cl | 3-Me-4-Cl-Phenyl | $0.81 \pm 0.16$ |  |  |
| 50 | NH | $4-\mathrm{Cl}$ | 3,5-diMe-4-Cl-Phenyl | $0.16 \pm 0.04$ |  |  |
| 51 | NH | 4-Cl | 1-Naphthyl | $0.70 \pm 0.24$ |  |  |
| 52 | NH | 6-Cl | 3-Me-4-Cl-Phenyl | $0.19 \pm 0.02$ | >15 | 1.4 |
| 53 | NH | 6-Cl | 3,5-diMe-4-Cl-Phenyl | $0.055 \pm 0.018$ | $>15$ | 0.87 |
| 54 | NH | $6-\mathrm{Cl}$ | 1-Naphthyl | $0.075 \pm 0.025$ | 6.8 | 0.96 |
| 55 | NH | 6-Cl | 1-(5,6,7,8-tetrahydronaphthyl) | $0.066 \pm 0.028$ |  |  |
| 56 | NMe | H | 3,5-diMe-4-Cl-Phenyl | $0.18 \pm 0.01$ | >15 | 7.4 |
| 57 | NMe | H | 1-Naphthyl | $0.26 \pm 0.04$ | >15 | >15 |
| 58 | NMe | 6-Cl | 3,5-diMe-4-Cl-Phenyl | $0.14 \pm 0.01$ |  |  |


| Comp. | A | R | Ar | $\mathbf{K}_{\mathbf{i}}(\mu \mathbf{M})$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | Mcl- ${ }^{\text {a }}$ | Bcl-xL ${ }^{\text {b }}$ | Bcl-2 ${ }^{\text {b }}$ |
| 59 | NBn | H | 1-Naphthyl | $0.29 \pm 0.17$ | >15 | >15 |
| 60 | S | H | 3,5-diMe-4-Cl-Phenyl | $0.32 \pm 0.01$ | >15 | 5.3 |
| 23 | S | H | 1-Naphthyl | $0.37 \pm 0.07$ | 15 | 9.7 |
| 61 | S | 4-Cl | 3,5-diMe-4-Cl-Phenyl | $1.3 \pm 0.52$ |  |  |
| 62 | S | 4-Cl | 1-Naphthyl | $1.6 \pm 0.41$ |  |  |
| 63 | S | 6-Cl | 3,5-diMe-4-Cl-Phenyl | $0.25 \pm 0.04$ | 4.1 | 0.80 |
| 64 | S | 6-Cl | 1-Naphthyl | $0.12 \pm 0.03$ | 6.8 | 0.96 |
| 65 | S | 7-Cl | 3,5-diMe-4-Cl-Phenyl | $0.75 \pm 0.24$ |  |  |
| 66 | S | 7-Cl | 1-Naphthyl | $0.72 \pm 0.15$ |  |  |
| 67 | O | H | 3,5-diMe-4-Cl-Phenyl | $3.0 \pm 0.10$ |  |  |
| 26 | O | H | 1-Naphthyl | $1.0 \pm 0.44$ | >15 | 4.3 |
| 68 | O | 5-Cl | 1-Naphthyl | $2.5 \pm 1.1$ |  |  |

${ }^{a} \mathrm{~K}_{\mathrm{i}}$ values are measured in triplicate
$b_{\mathrm{K}_{\mathrm{i}}}$ values are measured in duplicate

Table 4
SAR of 2-carboxylate analogs on 1-substituted indoles

|  | Comp. | A | Ar | Mcl-1 $\mathrm{K}_{\mathrm{i}}(\mu \mathrm{M}){ }^{*}$ |
| :---: | :---: | :---: | :---: | :---: |
|  | 69 | OH | 3-Me-4-Cl-Phenyl | $1.2 \pm 0.43$ |
|  | 70 | OH | 3,5-diMe-4-Cl-Phenyl | $0.41 \pm 0.07$ |
|  | 71 | OH | 1-Naphthyl | $0.47 \pm 0.08$ |
|  | 72 | $\mathrm{NH}_{2}$ | 1-Naphthyl | >500 |

*All Ki values are measured in triplicate

## Table 5

X-ray data collection and refinement statistics. Values in parenthesis represent the highest resolution shell

| Ligand | 53 | 60 | 16 mer peptide ${ }^{a}$ |
| :---: | :---: | :---: | :---: |
| PDB ID code | 4HW2 | 4HW3 | 4HW4 |
| No. chains | 6 | 12 | 2 |
| Protein construct | Mcl-1 $\Delta 5$ | Mcl-1 $\Delta 5$ | Mcl-1 WT |
| Data collection |  |  |  |
| Space group | $\mathrm{P} 2{ }_{1} 2_{1} 2$ | P 21 | P 21 |
| Cell dimensions |  |  |  |
| a, b, c ( $\AA$ ) | $\begin{gathered} 121.99 \\ 134.32,62.04 \end{gathered}$ | $\begin{gathered} 139.44,58.76 \\ 140.67 \end{gathered}$ | $\begin{gathered} 50.07,48.35, \\ 68.18 \end{gathered}$ |
| a, $\beta, \gamma\left({ }^{\circ}\right)$ | $\begin{gathered} 90.00,90.00 \\ 90.00 \end{gathered}$ | $\begin{gathered} 90.00,90.71, \\ 90.00 \end{gathered}$ | $\begin{gathered} 90.00,93.83, \\ 90.00 \end{gathered}$ |
| Resolution (A) | $\begin{aligned} & 50.00-2.80 \\ & (2.85-2.80) \end{aligned}$ | $\begin{aligned} & 50.00-2.40 \\ & (2.44-2.40) \end{aligned}$ | $\begin{aligned} & 50.00-1.53 \\ & (1.56-1.53) \end{aligned}$ |
| $\mathrm{R}_{\text {sym }}$ | 5.7(29.7) | 5.3(29.9) | 4.7 (40.8) |
| $\mathrm{R}_{\text {merge }}$ | 4.5 (41.1) | 4.5 (36.4) | 4.2 (31.3) |
| I / GI | 5.7 (1.9) | 12.5 (4.9) | 25.0 (2.5) |
| Completeness (\%) | 97.7 (95.0) | 98.2 (96.3) | 100 (100) |
| Redundancy | 4.1 (3.4) | 3.5 (2.7) | 4.1 (3.4) |
| Structure Refinement |  |  |  |
| No. reflections | 25097 | 88346 | 49292 |
| $\mathrm{R}_{\text {work }} / \mathrm{R}_{\text {free }}$ | 21.69 / 24.52 | 21.51 / 26.28 | 13.74 / 18.40 |
| No. atoms |  |  |  |
| Protein | 7181 | 14259 | 4823 b |
| Ligand | 190 | 300 | 520 |
| Water | - | 252 | 387 |
| Ramachandran (\%) ${ }^{\text {c }}$ |  |  |  |
| Preferred regions | 90.1 | 89.9 | 94.5 |
| Allowed regions | 7.4 | 8.4 | 5.1 |
| Generously allowed | 1.5 | 1.2 | 0.3 |
| Disallowed region | 1.0 | 0.6 | 0.0 |
| RMS deviations |  |  |  |
| Bond lengths ( $\AA$ ) | 0.026 | 0.010 | 0.008 |
| Bond angles ( ${ }^{\circ}$ ) | 1.993 | 1.411 | 1.005 |

[^1]
[^0]:    *Corresponding authorstephen.fesik@vanderbilt.edu; Phone, +1 (615) 322 6303; Fax, +1 (615) 8753236.
    Supporting Information Mcl-1 protein sequence used for X-ray crystallography, $1 \mathrm{H}-15 \mathrm{~N}$ HMQC spectra of Mcl-1 with and without compound 53, FPA dose-response curve of compound 53 displacing a FITC-labeled BAK peptide from Mcl-1 and synthesis of intermediates. This material is available free of charge via the Internet at http://pubs.acs.org.
    ACCESSION CODES Atom coordinates and structure factors for Mcl-1/ligand complexes have been deposited at the Protein Data Bank (4HW2, 4HW3 and 4HW4).

[^1]:    ${ }^{a}{ }_{16 m e r ~ M c l-1 ~ B H 3-p e p t i d e: ~ A c-A L E T L R R V G D G V Q R N H-N H 2 ~}^{2}$
    $b$ Including hydrogen atoms
    $c_{\text {Backbone conformation was analyzed with PROCHECK }} 52$

