Effect of Artificial Shading and Temperature on Radical Scavenging Activity and Polyphenolic Composition in Sweetpotato (*Ipomoea batatas* L.) Leaves

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ABSTRACT. The phenolic content and the radical scavenging activity were compared in leaves of sweetpotato (*Ipomoea batatas* L.) cultivars Shimon-1, Kyushu-119 and Elegant Summer grown under different temperature and shading conditions. Compared to cultivar differences, there was less effect of temperature and shading on the total phenolic content in sweetpotato leaves, however certain polyphenolic components differed widely among the treatments. The positive correlation between the radical scavenging activity and the level of total phenolics (r = 0.62) suggests that phenolic compounds are important antioxidant components of sweetpotato leaves. All the reverse-phase high-performance liquid chromatography (RP-HPLC) profiles of the cultivars tested showed peaks at the same retention times but peak areas of individual phenolic compounds differed with respective temperature and shading treatments. The phenolic compounds identified in the sweetpotato leaf were caffeic acid, chlorogenic acid, 4,5-di-*O*-caffeoylquinic acid, 3,5-di-*O*-caffeoylquinic acid, 3,4-di-*O*-caffeoylquinic acid, and 3,4,5-tri-*O*-caffeoylquinic acid. Most of the phenolic compounds were highest in leaves from plants grown at 20 °C without shading except 4,5-di-*O*-caffeoylquinic acid. The results indicate that growing leaves under moderately high temperatures and in full sun enhances the accumulation of phenolic components. These phenolic components have possible value in enhancing human health.

The consumption of sweetpotato greens as a fresh vegetable in some parts of the world (As-Saqui, 1982; Nwinyi, 1992; Villareal et al., 1982) indicates that they are acceptable as a leafy vegetable. Horticulturists and food scientists, faced with the problem of feeding the world's hungry, are becoming increasingly interested in previously neglected tropical green leafy vegetables like sweetpotato greens. Since sweetpotato leaves can be harvested several times in a year their annual yield is ultimately higher than many other green vegetables. Furthermore, sweetpotato leaves are one of the few vegetables that can be grown easily during the monsoon seasons of the tropics, and are usually the only greens available in some countries after a flood or typhoon. They are rich in vitamin B, iron, calcium, zinc and protein (Pace et al., 1985), and are more tolerant of diseases, pests (Asian Vegetable Research and Development Center, 1985), and high moisture than many other leafy vegetables grown in the tropics (Woolfe, 1992). Our previous reports revealed that sweetpotato leaves are an excellent source of antioxidative compounds, namely polyphenolics (Islam et al., 2002a; Yoshimoto, 2001). In addition, sweetpotato leaves can also be used in noodles, breads, drinks, and confectioneries.

Phenolic compounds are a diverse group of secondary metabolites present in higher plants that play important roles in the structure of plants and are involved in a number of metabolic pathways (Harborne, 1980). Plant phenolics, because of their diversity and extensive distribution, can be argued to be an important group of natural antioxidants, and contribute to organoleptic and nutritional qualities of fruit and vegetables. Phenolic compounds recently have

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attracted special attention in that they are thought to protect the human body from oxidative stress, which may cause the onset of cancer, aging, and cardiovascular diseases (Huang and Ferraro, 1991; Kaul and Khanduja, 1998; Peluso et al., 1995; Robards et al., 1999; Shahrzed and Bitsch, 1996; Shimozono et al., 1996; Yoshimoto et al., 1999). Antioxidative capacity of fruit and vegetables is influenced by genetic as well as environmental factors. The measurement of total antioxidant capacity may be an effective tool for use in fruit and vegetable breeding programs designed to increase antioxidant components available for human consumption. Since the natural production environment is variable, controlled growing conditions are needed to ascertain environmental factors affecting antioxidant properties. The investigation of the effects of temperature and shading on phenolic content and radical scavenging activity (RSA) of sweetpotato leaves may identify new methods to enhance its nutritional value. The aim of the present study is to identify suitable growing conditions to enhance desired bioactive compounds.

Materials and Methods

PLANT MATERIALS AND CULTURAL METHODS. Sweetpotato tips (top 10cm) of 'Shimon-1' (S-1), 'Kyushu-119' (K-119), and 'Elegant Summer' (ES), which were developed for leaf consumption, are harvested from plants grown at the Department of Upland Farming Research of the National Agricultural Research Center for Kyushu Okinawa Region at Miyazaki, Japan. The average air temperature (May to October) of the experimental location was 25 °C. Cuttings were planted in sterilized soil in 12 cm vinyl pots and placed in a greenhouse. Six weeks after planting, pots were transferred into a greenhouse having a constant temperature of 25 ± 2 °C. There were five pots, which each served as a replication per treatment and they were arranged in a randomized block design. Temperature and shading experiments were done in separate greenhouses using the above three cultivars. After seven days, the pots were transferred

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Fig. 1. Chemical structures of the hydroxycinnamate esters in sweetpotato leaves.

into individual greenhouses having temperatures of 20 ± 1.5 °C, 25 ± 2 °C and 30 ± 2.5 °C. The shading experiment was conducted using 80%, 40% and 0% shading. Shading was achieved using commercial shading materials purchased from Nippon Wide Cloth Co. Ltd, Osaka, Japan (for 40% shading), and Dio Kasei Co., Ltd., Tokyo, Japan (for 80% shading). The light intensity was checked using a illuminance meter (model T-1H; Minolta Co. Ltd., Osaka, Japan). After 7 d of continuous treatment, all the leaves (20 to 25 leaves) from each plant were harvested, washed gently, put into prelabeled vinyl bags, and immediately frozen at -70 °C. The following day the frozen samples were freeze-dried for 48 h in a vacuum freeze dryer (model TR-PK-3-80; Trio Sciences Co, Ltd., Tokyo, Japan) with a plate temperature of 27 to 30 °C. The freeze-dried samples were powdered using a blender prior to analysis.

EXTRACTION AND MEASUREMENT OF TOTAL PHENOLICS. Total phenolics were measured by the procedure described by Coseteng and Lee (1987) with a slight modification. The lyophilized powdered sweetpotato leaf tissue (10 mg) was vigorously mixed with 1:10 (mg·mL⁻¹) 80% aqueous ethanol solution. The mixture was boiled for 5 min under a hood, centrifuged at $5000 g_n$ for 10 min, and the supernatant was collected. The residue was mixed with 5 mL of 80% aqueous ethanol, boiled for 5 min to reextract the phenolics and centrifuged under the same conditions. The extracts were combined, made up to 10 mL, and used for the measurement of total phenolics. The alcohol extract was diluted to obtain an absorbance reading within the range of the standards (40 to 800 mg chlorogenic acid/ mL). The absorbance was measured at 600 nm with a dual wavelength flying spot scanning densitometer (Shimadzu Co., Kyoto, Japan), with a microplate system. The results were expressed as g/ 100 g dry weight (DW).

IDENTIFICATION OF ISOLATED SWEETPOTATO LEAF PHENOLICS. The dried leaves of sweetpotato (150 g) were extracted twice by shaking with 2 L of 100% methanol at room temperature. The dried extract (17 g) was partitioned between benzene and water (50:50). The water layer (8 g) was fractionated on a adsorption chromatography using MCI gel CHP20P column (50×350 mm i.d., 75 to 150µm, styrene polymer, Mitsubishi Chemical Ind. Ltd., Tokyo, Japan) equilibrated with deionized water and adsorbed components were eluted with 20%, 40%, 60%, 80% and 100% methanol successively. The 40% methanol eluate contained mainly caffeic acid (CA) and chlorogenic acid (ChA), while the 60% methanol eluate mainly di-*O*-caffeoylquinic acids and 3,4,5-tri-*O*-caffeoylquinic acid. These eluates were each further fractionated on a reversed-phase chroma-

tography using an ODS column ($25 \times 140 \text{ mm i.d.}$, $30 \text{ to } 50 \mu\text{m}$, Fuji Silisia Ltd., Nagoya, Japan) using 20% to 70% methanol give CA (15 mg/150 g DW), ChA (400 mg/150 g DW), 3,4-di-Ocaffeoylquinic acid (3,4-diCQA) (2 mg/150 g DW), 3,5-di-Ocaffeoylquinic acid (4,5-diCQA) (60 mg/150 g DW), 4,5-di-Ocaffeoylquinic acid (4,5-diCQA) (21 mg/150 g DW) and 3,4,5-tri-O-caffeoylquinic acid (3,4,5-triCQA) (2 mg/150 g DW). The above phenolics were identified as described in a previous paper (Islam et al., 2002b). The chemical structures of these phenolic compounds are shown in Fig. 1.

QUANTIFICATION OF PHENOLIC ACIDS BY RP-HPLC. The lyophilized, powdered sweetpotato leaf tissue (50 mg) was vigorously mixed with 4 mL of 80% ethanol in a capped centrifuge tube. The mixture was boiled for 5 min in a hood and centrifuged at 3000 g_n for 10 min. The supernatant was filtered through a cellulose acetate membrane filter (0.20 mm, Advantec, Tokyo, Japan). A 5-µL portion of the filtrate was injected into the HPLC system and eluted as described below. The HPLC system consisted of two model LC-10AT pumps, a model SIL-10AXL autoinjector, a model CTO-10AC column oven, and a model SPD-M10AVP photodiode array UV-VIS detector (Shimadzu, Co., Kyoto, Japan). The column was a YMC-Pack ODS-AM AM-302 (150×4.6 mm i.d., 5-µm particles; YMC, Kyoto, Japan). The column oven temperature was set at 40 °C. The mobile phase consisted of water containing 0.2% (v/v) formic acid (A) and methanol (B). Elution was performed with a linear gradient as follows: 2% B from 0 to 15 min, 2% to 45% B from 15 to 50 min, and 45% B from 50 to 65 min. The flow rate was 1 mL·min⁻¹. The phenolics were detected at 326 nm. The retention times (t_R) of the phenolics compounds were compared with those of purified phenolics from sweetpotato leaves used as authentic standard (purity >97% estimated by the HPLC analysis) (Islam et al., 2002b).

MEASUREMENT OF RADICAL SCAVENGING ACTIVITY USING THE DPPH METHOD. The radical scavenging activity (RSA) as measured according to Brand-Williams et al. (1995) with slight modifications. 1,1-Diphenyl-2-picrylhydrazyl (DPPH) was used as a stable radical. All reactions were run in a 96 well microplate with a total volume of 300 µL. A 75-µL sample (80% ethanol extract) was combined with 150 µL 0.1 M 2-(N-morpholino) ethanesulfonic acid (MES) (pH 6.0) and 75 µL DPPH in 50% ethanol in the microplate well and mixed. For the control, 75 µL of 80% ethanol was used in place of the sample. The reaction mixtures were shaken and held for 2 min at room temperature in the dark. Trolox (6-hydroxy-2, 5,7,8tetramethyl-chroman-2-carboxylic acid) was used as the reference antioxidant compound and 80% ethanol was used as the blank solution (without DPPH). The decrease in absorbance of DPPH at 520 nm was measured within 2 min. All samples were analyzed in triplicate. The RSA of samples (antioxidants) was expressed in terms of IC₅₀ (concentration in µmol Trolox/mg DW required for a 50% decrease in absorbance of the DPPH radical). A plot of absorbance vs. concentration was made to calculate IC₅₀. The results were expressed as µmole Trolox/mg DW.

CHEMICALS. ChA was purchased from Sigma Chemical (St. Louis, Mo.). Trolox was obtained from Aldrich Chemical Co. (Milw., Wis.). CA, DPPH and other chemicals used were the highest grade available supplied by Wako Pure Chemicals Industries Ltd., Osaka, Japan.

STATISTICS. A randomized complete block design with five replications was used. Data for the different parameters were analyzed by analysis of variance (ANOVA) procedure, and the level of significance was calculated from the F value of ANOVA using the Excel Statistics 2000 program package for Windows (Social

Table 1. Effect of temperature on the total leaf phenolic content of sweetpotato cultivars. Each value is the mean of five replications \pm standard error.

Temp	Total phenolic content (g/100 g dry wt)			LSD ^z	LSD
(°C)	Simon-1	Kyushu-119	Elegant Summer	(1%)	(5%)
20	10.1 ± 0.32	7.7 ± 0.33	7.0 ± 0.28	1.52	1.08
25	9.5 ± 0.25	8.2 ± 0.34	7.3 ± 0.27	1.41	1.01
30	9.0 ± 0.19	7.4 ± 0.16	6.6 ± 0.24	0.99	0.71
LSD (1%)	1.29	NS	NS		
(5%)	0.92				

^zFor cultivar.

NSNonsignificant.

Table 2. Effect of shading on the leaf phenolic content of sweetpotato cultivars. Each value is the mean of five replications \pm standard error.

Shading (%)	Total phenolic content (g/100 g dry wt)			LSD ^z	LSD
	Simon-1	Kyushu-119	Elegant Summer	(1%)	(5%)
0	9.82 ± 0.21	8.11 ± 0.34	7.61 ± 0.21	1.27	0.90
40	7.28 ± 0.28	6.87 ± 0.16	6.38 ± 0.16	1.01	0.72
80	6.79 ± 0.12	6.18 ± 0.21	6.42 ± 0.19	0.85	0.61
LSD (1%)	1.02	1.21	0.92		
(5%)	0.73	0.86	0.65		

^zFor cultivar.

Table 3. Effect of temperature on the leaf radical scavenging activity of sweetpotato cultivars. Each value is the mean of five replications \pm standard error.

Temp	Radical scavenging activity (µmol Trolox/mg dry wt)			LSD ^z	LSD
(°C)	Simon-1	Kyushu-119	Elegant Summer	(1%)	(5%)
20	0.99 ± 0.18	0.54 ± 0.04	0.51 ± 0.04	0.52	0.37
25	0.67 ± 0.13	0.57 ± 0.03	0.47 ± 0.03	0.38	0.27
30	0.51 ± 0.04	0.46 ± 0.03	0.42 ± 0.04	0.18	0.13
LSD (1%)	0.63	NS	NS		
(5%)	0.45				

^zFor cultivar.

^{NS}Nonsignificant.

Table 4. Effect of artificial shading on the leaf radical scavenging activity of sweetpotato cultivars. Each value is the mean of five replications \pm standard error.

Shading	Radical scavenging activity (µmol Trolox/mg dry wt)			LSD ^z	LSD
(%)	Simon-1	Kyushu-119	Elegant Summer	(1%)	(5%)
0	0.70 ± 0.06	0.68 ± 0.03	0.57 ± 0.04	0.22	0.15
40	0.57 ± 0.06	0.51 ± 0.05	0.53 ± 0.08	0.26	0.18
80	0.53 ± 0.04	0.52 ± 0.06	0.54 ± 0.06	0.24	0.17
LSD (1%)	0.24	0.23	NS		
(5%)	0.17	0.16			

^zFor cultivar.

^{NS}Nonsignificant.

Information Service Co. Ltd., Tokyo, Japan). The relation between total polyphenols and RSA were described with correlation analysis using the above statistical program.

Results

TEMPERATURE AND SHADING EFFECTS. The effect of temperature on the total leaf phenolic content of sweetpotato cultivars is shown in Table 1. The phenolics content differed (P < 0.01) among the cultivars studied and the leaves from 'S-1' had the highest content followed by 'K-119' and 'ES'. The 'S-1' sweetpotato leaves grown at 20 °C had the highest phenolic content and was greater than that of 'S-1' leaves grown at 30 °C. Although total leaf phenolic content of other cultivars did not differ significantly with growing temperature, levels in leaves grown at 20 and 25 °C were always greater than that of leaves grown at 30 °C. Shading also affected leaf phenolic content. Sweetpotato leaves in the 0% shading treatment had a significantly greater accumulation of total phenolics than the 40% and 80% shading treatments in all cultivars (Table 2). The effect of temperature on the RSA in sweetpotato leaves showed a trend similar to temperature effects on phenolic accumulation (Table 3). The RSA was significantly greater in 20 °C grown 'S-1' leaves than in 30 °C leaves, but the other cultivars did not differ with the temperature treatments. Like polyphenol accumulation, the RSA in leaves from 0% shading were significantly greater than 40% and 80% shading except in 'ES' (Table 4). There was a significant (P <



Fig. 2. Linear correlations between the total polyphenol contents (g/100 g dry weight) and radical scavenging activities (µmol Trolox/mg dry weight) of sweetpotato leaves.



Fig. 3. A representative reverse-phase HPLC (C_{18}) chromatogram for polyphenolic compounds in sweetpotato leaf extracts. Detector absorbance was at 326 nm. CA = caffeic acid (32.27 ± 0.12 min); ChA = chlorogenic acid (33.42 ± 0.18 min); 4,5-diCQA = 4,5-di-O-caffeoylquinic acid (46.72 ± 0.14 min); 3,5-diCQA = 3,5-di-O-caffeoylquinic acid (47.34±0.17 min); 3,4-diCQA = 3,4-di-O-caffeoylquinic acid (50.55 ± 0.22 min); 3,4,5-triCQA = 3,4,5-tri-O-caffeoylquinic acid (57.25 ± 0.13 min).

0.001) linear correlation (r = 0.62, n = 90) observed between RSA and total phenolics accumulation in sweetpotato leaves (Fig. 2).

IDENTIFICATION OF PHENOLIC COMPOUNDS AND THE EFFECT OF TEMPERATURE AND SHADING. Our extraction method and RP-HPLC analysis identified six different phenolic compounds in sweetpotato leaves (Fig. 3). All the HPLC profiles of the treatments displayed the same peaks indicating no qualitative difference among cultivars (data not shown). However, the peak areas differed with cultivars and experimental treatments. Peaks were identified as CA (retention time (t_R) = 32.27 ± 0.12 min), ChA (t_R = 33.42 ± 0.18 min), 4,5diCQA (t_R = 46.72 ± 0.14 min), 3,5-diCQA (t_R = 47.34 ± 0.17 min), 3,4-diCQA (t_R = 50.55 ± 0.22 min) and 3,4,5-triCQA (t_R = 57.25 ± 0.13 min). All the cultivars studied showed a similar chromatographic pattern for the above caffeic acid derivatives and 3,5diCQA and 4,5-diCQA were the dominant phenolics. Temperature

affected phenolic composition in sweetpotato leaves (Fig. 4). In all three cultivars there were no significant differences observed in the concentration of CA, ChA, 3,4-diCQA and 3,4,5-triCOA among the temperature treatments. However, 3,5-diCOA content was lower in leaves grown at 30 °C as compared with those grown at 20 and 25 °C, and the content of 4,5-diCQA was highest in leaves grown at 30 °C. The shading treatment also influenced the phenolic composition of sweetpotato leaves (Fig. 5). The highest content of phenolic compounds from the leaves of all 3 cultivars was in plants grown without shading. Similar to the temperature responses, the 3,5-diCQA and 4,5-diCQA were also the phenolics most affected by shading treatments in all the cultivars studied. In the case of 'S-1', there was little effect of the shading treatments. However, in 'K-119' and 'ES',

shading decreased the concentration of phenolic compounds with the exception of CA and 3,4,5-triCQA.

Discussion

Our recent studies revealed that sweetpotato leaves were an excellent source of antioxidative phenolic compounds (Islam et al., 2002a; Yoshimoto, 2001). In the present study, total phenolic concentration in sweetpotato leaves ranged from 6.18 g/100 g DW to 10.1 g/100 g DW (Table 1 and 2), which is high when compared to sweetpotato storage roots (Walter et al., 1979) and potato tubers (Lugasi et al., 1999). The results of this study suggest that plants grown under moderate temperature and without shading accumulate a higher concentration of leaf phenolics than plant grown under high temperatures and shaded conditions (Tables 1-4). The decrease in phenolic content in leaves grown under shade is probably due to the reduced stimulation of phenolic production by light. Moriyama et al. (1999) reported that 60% and 80% shading decreased the phenolic content of tea leaves compared to 0% shading. The present results also indicate that moderate temperatures (20 to 25 °C) enhanced synthesis and/or warm temperature (30 °C) enhanced degradation of sweetpotato leaf phenolics.

The RSA of sweetpotato leaves exhibited similar trends as the phenolic content in all the cultivars studied. Also there was a strong relationship (P < 0.001) between RSA and total polyphenol content of sweetpotato leaves (Fig. 2). Thus, the RSA of sweetpotato leaves is due primarily to their phenolic contents. Although different among the treatments studied, sweetpotato leaves had a high RSA. The present data and our recent report suggests that sweetpotato leaves by showing strong antimutagenicity against Trp-P-1 using *Salmonella typhimurium* TA 98 (Yoshimoto et al., 2002).

We have analyzed the phenolic composition of sweetpotato leaves by using RP-HPLC. Although HPLC profiles of the treatment tested showed the peaks at the same t_{R} there was wide variation in quantities of individual phenolic compounds in sweetpotato leaves in relation to cultivars and growing conditions. These results suggest that the phenolic compounds in sweetpotato leaves are distributed as per the following order: 3,5-diCQA > 4,5-diCQA > ChA > 3,4-diCQA > CA > 3,4,5-triCQA (Figs. 4 and 5). The leaves grown under moderate temperature and without shading accumulated the highest concentration of all the phenolic compounds with the exception of 4,5-diCQA. Therefore, the high temperature treatment (30 °C) reduced the phenolic compounds in sweetpotato leaves, but moderate temperatures namely 20 °C enhanced the accumulation of the individual polyphenolics. Chlorogenic acid content of sweetpotato leaves was much higher than that of sweetpotato storage roots (Walter et al., 1979), potato tubers (Lugasi et al., 1999), and apple fruit (Murata et al., 1995) and it has various physiological functions in both animal and humans (Kapil et al., 1995; Tsuchiya et al., 1996; Yagasaki et al., 2000). Among other phenolic compounds, CA was the most effective inhibitor of tumor promotion in mice skin and, ChA, 3,4-diCQA, 3,5-diCQA, and 4,5diCQA, which were extracted from steamed sweetpotato suppressed the melanogenesis equally (Kaul and Kkanduja, 1998; Shimozono et al., 1996). Furthermore, the 3,4,5-triCQA exhibited a greater selective inhibition of HIV replication than 4,5-diCQA,



and CA had only slight anti HIV activity (Mahmood et al., 1993).

These results may help to get the desired phenolic compounds in sweetpotato leaves by manipulating the growing conditions, and also be of use in future breeding programs designed to increase antioxidant components for human health. The results also indicate that growth of sweetpotato leaves under moderate temperatures and in full sun enhances the accumulation of phenolic components.

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Fig. 4. Effect of temperature on the concentration of polyphenolic compounds in the leaves of three sweetpotato cultivars. Bars indicate the standard error of the mean of five replications, when absent it falls under the symbol. CA = caffeic acid, ChA = chlorogenic acid, 4,5-diCQA = 4,5-di-*O*-caffeoylquinic acid, 3,5-diCQA = 3,5-di-*O*-caffeoylquinic acid, 3,4-diCQA = 3,4-di-*O*-caffeoylquinic acid, 3,4-5-triCQA = 3,4,5-tri-*O*-caffeoylquinic acid.

Fig. 5. Effect of artificial shading on the concentration of polyphenolic compounds in the leaves of three sweetpotato cultivars. Bars indicate the standard error of the mean of five replications, when absent it falls under the symbol. CA = caffeic acid, ChA = chlorogenic acid, 4,5-diCQA = 4,5-di-*O*-caffeoylquinic acid, 3,5diCQA = 3,5-di-*O*-caffeoylquinic acid, 3,4-diCQA = 3,4-di-*O*-caffeoylquinic acid, 3,4-5-triCQA = 3,4,5-tri-*O*-caffeoylquinic acid.

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