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Effect of buffer composition on PNA-RNA hybridization studied in the microfluidic microarray chip

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Complete List of Authors:	Chim, Wilson; Simon Fraser University Sedighi, Abootaleb; Simon Fraser University Brown, Christopher; Griffith University Pantophlet, Ralph; Simon Fraser University Li, Paul; Simon Fraser University
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3	Wilson Chim, Abootaleb Sedighi, Christopher L. Brown, Ralph Pantophlet, Paul C.H. Li
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5	Wilson Chim, Abootaleb Sedighi, and Paul C.H. Li. Department of Chemistry, Simon Fraser
6	University, Burnaby, British Columbia, Canada, V5A 1S6.
7	Christopher L. Brown. School of Natural Sciences and Queensland Micro- and Nanotechnology
8	Centre, Griffith University, Brisbane, Queensland, Australia.
9	Ralph Pantophlet. Faculty of Health Sciences; Department of Molecular Biology and Biochemistry,
10	Simon Fraser University, Burnaby, British Columbia, Canada, V5A 1S6.
11	Corresponding author: Paul C.H. Li (e-mail: paulli@sfu.ca)
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Abstract: Herin we report that Peptide Nucleic Acid sequences (PNAs) can be used as the probe 23 species for detection of RNA and that a microfluidic microarray (MMA) chip can be used as the 24 platform for detection of hybridizations between immobilized PNA probes and RNA targets. The 25 RNA targets used are derived from influenza A sequences. This paper discusses the optimization of 26 the two probe technologies used for RNA detection and investigates how the composition of the probe 27 buffer or the content of the hybridization solution can influence the overall results. Our data shows 28 29 that the PNA probe is a better choice over the DNA probe when there is low salt in the probe buffer composition. Furthermore, we have shown that the absence of salt (NaCl) in the hybridization buffer 30 does not hinder the detection of RNA sequences. The results conclude that PNA probes are superior 31 to DNA probes in term of sensitivity and adaptability, asPNA immobilization and PNA-RNA 32 hybridization are less affected by salt content in the reaction buffers unlike DNA probes. 33 Key words: peptide nucleic acid (PNA), influenza viral RNA, probe buffer, hybridization buffer, salt, 34 formamide, microfluidic microarray (MMA). 35

36

37 Introduction

Peptide nucleic acids (PNAs) are DNA analogs in which the negatively charged deoxyribose phosphate backbone is replaced by an electrically neutral peptide-like backbone.¹⁻² Since the PNA probes are not negatively charged, unlike traditional DNA probes, they should be less affected by the salt content of the buffer in which these probes are used.³ When PNAs are used in the assembly of microarrays, it would be useful to better understand the effects of variation in salt concentration, in both the probe buffer used for printing the PNA probes on solid substrates and the hybridization buffer used for reaction between the probes and the target nucleic acids.

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In this paper, PNA probes designed to detect a RNA sequence related to influenza A are
employed in the construction of a microfluidic microarray (MMA) chip. The influenza A virus is the
causative agent of yearly epidemics and occasional pandemics ⁴ and numerous microfluidic
technologies have been developed to detect the presence of influenza A viral RNAs in clinical
samples. Devices employed in these technologies include microfluidic chips, ⁵⁻¹⁵ integrated
Microsystems, ¹⁶⁻²⁰ and lateral flow strips. ^{8, 21-22} The detection methodologies include RT-qPCR, ^{15, 17,}
^{20, 23-26} electrophoretic immunoassays, ²⁷ PCR-capillary electrophoresis, ^{11-14, 18, 29}
immunochromatography, ^{8, 19, 21-22} microarray, ^{5, 9, 16, 30-32} and electrochemical approaches. ^{4, 7, 28, 33}
Among the microarray detection methods, oligonucleotide probes are most commonly used. ^{5, 9, 16, 30-32}
Whilst PNA probes have been employed to detect various influenza virus strains, ³⁴⁻³⁶ and one group
has employed PNA probes in the microfluidic chip to detect the viral DNA ⁷ to date, there are no
details provided about the buffer compositions in the use of the PNA probes. Therefore, we conducted
a study to investigate the effect of buffer composition on PNA probe immobilization and on PNA-
RNA hybridization. We believe it is informative to report this study even though, generally, influenza
A viral RNAs are detected indirectly after their conversion to cDNA.
In our group, we have previously employed the MMA chip for detection of DNA ³⁷⁻³⁹ and
RNA targets. ⁴⁰ The microfluidic microarray chip offers several advantages to other designs (e.g. plain
microarrays) in that it can be constructed without an expensive robotic spotting system, and used for
multiple samples. The use of the chip can also offer fast hybridization rates, and achieve cost-
effectiveness. ⁴¹⁻⁴⁵ The schematic diagrams for the use of the MMA chip are shown in Fig. 1 A-C. The
construction of the device follows several steps. Firstly, a PDMS channel plate is sealed with a glass
slide that is arrayed with lines consisting the probe molecules (Fig. 1A). Secondly, after removing the

67 first PDMS plate, the glass slide is sealed against a second PDMS plate to admit target samples for

68	hybridization (Fig. 1B). Thirdly, the PDMS plate is removed and the glass slide is fluorescently
69	scanned for detection (Fig. 1C). An image of the actual experiment used in this study and consisting
70	of 16 probes hybridized with 16 targets is shown in Fig. 1D.
71	In this paper, we describe the principle of detection of PNA-RNA hybridizations in the MMA
72	chip. This chip has the capability of studying various buffer conditions simultaneously using multiple
73	samples. We report the effect of compositions of reaction buffers (<i>i.e.</i> probe buffer and hybridization
74	buffer) on PNA-RNA hybridization results.
75	
76	Experimental
77	Nucleic acid sequences
78	RNA target and DNA probe sequences were synthesized by Integrated DNA Technologies
79	(Coralville IA). PNA probes were purchased from PNA Bio, Inc. (Thousand Oaks, CA). The
80	sequences of the nucleic acid probes as well as RNA targets utilized in this study are shown in Table
81	1.
82	Three probes were used: two were PNA probes (P-PNA, P-PNA-inf) and one was DNA probe
83	(P-DNA). Both PNA probes are 13-mer PNA probe sequences utilizing an 'O monomer' spacer
84	molecule (2-(2-aminoethylethoxy) ethoxy acetic acid) located at the N-terminal position for
85	attachment to the chip surface. The P-PNA probe has the same base sequence as the central region of
86	a 30-mer DNA probe (P-DNA), utilising an amino group with a C6 linker located at the 5' end for
87	surface attachment. This DNA probe was previously used for analyzing RNA samples ⁴⁰ . The P-PNA-
88	inf probe is a 13-mer PNA probe that was previously reported to probe influenza A viral RNA. ^{34, 50}
89	Two target RNA sequences were used. One is complementary to the PNA probe (T-RNA),
90	whilst the other is designed to mimic an influenza A viral RNA target sequence (T-RNA-inf). Both

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RNA target sequences were labeled by biotin at the 5' termini. The sequence of T-RNA was so

92	designed that the binding region (section underlined in the T-RNA sequence in Table 1) is
93	complementary to both the PNA probe (P-PNA) and the central region of the oligonucleotide probe
94	(underlined in P-DNA).
95	The influenza A virus consists of single-stranded RNAs ⁴⁶⁻⁴⁷ . The RNA molecule comprises
96	eight segments that encode for different proteins necessary for transcription, replication and viral
97	assembly ⁴⁸⁻⁴⁹ . The sequence of T-RNA-inf was obtained from the analysis of the 5'untranslated
98	region (UTR) which was conserved among each of the eight RNA segments. ^{34, 50}
99	PNA-RNA or DNA-RNA hybridizations were determined by imaging of theCy5 tag (emits at
100	670 nm when excited at 633 nm using fluorescence spectroscopy) incorporated by conjugation of
101	streptavidin-Cy5 to the 5' biotin moiety on the RNA target sequences.
102	
103	Materials
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The plain glass microscope slides (3" x 2") were purchased from Fisher Scientific (Ottawa, ON).

116 Functionalization of Glass Slides

Four glass slides were washed with detergent (Sparkleen) and rinsed with pure water, as 117 previously described ⁵¹⁻⁵². In summary, the glass slides were submerged in piranha solution (3:1 118 mixture of concentrated H₂SO₄, 30% H₂O₂) for 15 min. at 80 °C. After being rinsed with distilled 119 water, ethanol 95%, then distilled water, the glass slides were dried under N₂. They were then 120 submerged in a 2% APTES solution (under a N₂ gas environment) for 20 min. and then rinsed with 121 100% ethanol. Dispensing of APTES liquid was performed by a 10"-long needle together with a 122 compensating nitrogen gas balloon. One observation is that the liquid should be yellow or light brown 123 at this point and if the color turns dark and/or particles are found in the liquid it should be discarded. 124 The glass slides were subsequently baked in the oven for 1 h at 120 °C in air. Thereafter, the glass 125 slides were submerged in a 5% glutaraldehyde solution in the refrigerator for 1 h and then rinsed with 126 pure water. Finally, the glass slides were N₂-dried and stored in the dark at 4 °C before probe printing. 127 128

129 **Probe Immobilization**

A 2" x 2" PDMS channel plate consisting of microchannels (200- μ m wide and 35- μ m deep) was fabricated as described elsewhere.^{37, 52} The 1.5-mm thick channel plate was sealed against the aldehyde-functionalized glass slide to form the MMA chip. Then, 0.6 μ L of nucleic acid probes prepared in the probe buffer (0.15 M sodium bicarbonate with 1 M NaCl, unless noted otherwise) were added into the inlet reservoirs of the MMA chip using a micropipette. The probe solutions were filled through the channels by applying suction at the outlets followed by incubation at room temperature for 1 h. After the microchannels had been rinsed with 1 μ L of a wash solution (0.15 M sodium bicarbonate with 1 M NaCl), the PDMS channel plate was peeled away from the surface.
Afterward, the glass slide was treated with the sodium borohydride solution (50mg NaBH₄, 15mL
PBS 1X, 5mL ethanol 95%, 150 µL SDS 1.5%) for 20 min. The glass chip was then rinsed with pure
water for 2 min. and then subsequently dried under N₂. The prepared glass slide was submerged in the
blocking solution (10 mL 10X SSC, 150 µL SDS 1.5%, 20mg BSA and made up to 20mL with pure
water) to inhibit non-specific binding. The glass chip incorporating the printed probe was then rinsed
with deionized water for 2 min. and finally dried as above.

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145 Hybridization and Detection

The glass slide with probe line arrays was covered with a second 2" x 2" PDMS channel plate 146 to form the MMA chip assembly. The straight channels were aligned orthogonal to the printed probe 147 148 lines on the slide. Hybridization reactions occur at the intersections between the channels and the probe lines. The target RNA samples were diluted in the hybridization buffer (1X SSC, unless noted 149 otherwise, with 0.15% SDS). After the target solutions were denatured at 95 °C for 5 min, they were 150 151 quickly cooled in ice just prior to their hybridization to immobilized probes. The target solutions (0.6 μ L) were added to the inlet reservoirs of the MMA chip using a micropipette. Target solutions in 152 different reservoirs were then drawn into the channels simultaneously by suction. Probe-target 153 hybridization was carried out in an oven at 40 °C for incubation for 30 min. The microchannels were 154 rinsed with 1µL of hybridization buffer immediately after incubation. SA-Cy5 solutions (0.6 µL, 1 155 mg/mL) were added to the inlet reservoirs using a micropipette, and the solutions were then drawn 156 into the channels by suction simultaneously. After a further incubation step (30 min. at room 157 temperature), the microchannels were rinsed immediately with 1 µL of wash buffer (1X PBS and 158 159 0.2% Tween 20)

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For analysis, the glass slide was imaged on a confocal laser fluorescent scanner (Typhoon
Trio+ variable mode imager) purchased from GE Healthcare (Baie d'Urfé, QC). The scan conditions
were spatial resolution: 25 μm, excitation wavelength: 633 nm, emission wavelength: 670 nm, and
photomultiplier tube voltage: 600V. The scanned image was analyzed by the ImageQuant 5.2
software. In the data quantification procedure, lines were drawn manually to cover the rectangular
hybridization patches on the glass slide, and the average fluorescent intensity of the patch was
recorded in the relative fluorescent unit (RFU).

167

168 **Results and Discussion**

169 Effect of probe buffer composition

RNA targets were hybridized to DNA and PNA probes in the presence of different amounts of salt 170 in the probe buffers. The hybridization results are shown in Figure 2A. Here, the RNA targets were 171 applied along the vertical direction, intersecting with the probe lines created in the horizontal 172 orientation. Experiments were conducted using two concentrations of RNA targets (T-RNA). Stronger 173 fluorescent signals were observed for the DNA-based probes (P-DNA) when prepared in the probe 174 buffer containing the high salt concentration (1 M NaCl) rather than the low salt one (0.1 M NaCl). It 175 176 is assumed this observation arises as the DNA probe possesses a negatively charged backbone and accordingly it requires the high ionic strength conditions provided by the 1M NaCl solution to reduce 177 the Columbic repulsion between the negatively charged backbone and the glass slide during the probe 178 immobilization step as previously reported. 55-56 Another observation is that much stronger signals are 179 observed for the PNA-based probes (P-PNA) rather than the DNA-based probes (P-DNA). As shown 180 in Fig. 2B, enhancement in signal intensity of 521% was found for the 10-nM RNA target, and 317% 181

182 for the 20-nM target, when P-PNA rather than P-DNA was the probe using 0.1 M NaCl as the probe183 buffer.

Furthermore, there was considerably less influence of salt concentration on RNA hybridization 184 efficiencies when using the PNA-based probes (P-PNA). For instance, the hybridization signals 185 (depicted in Fig. 2B) were similar for the 10-nM RNA target when the PNA probes were immobilized 186 using buffers of different salt contents (i.e. the signal was decreased by just 28.8% when 0.1 M 187 instead of 1 M NaCl was used). On the other hand, the DNA-based probes were less robust in which 188 the signal decreases were higher when the probe buffer contained 1 M instead of 0.1 M NaCl (i.e. the 189 signals were decreased by 83.9% and 66.4% for 20-nM and 10-nM RNA, respectively). These data 190 supports the notion that PNA probes are more robust than DNA probes in terms of sensitivity and 191 adaptability.34-36 192

193

194 Effect of hybridization buffer composition

Hybridizations involving RNA targets generally require a high temperature that functions to remove secondary structures of the single-stranded nucleic acid molecules. Nevertheless, formamide can also be used to achieve the same function allowing hybridization to be conducted at more reasonable temperatures.⁵³⁻⁵⁴

The results of RNA hybridizations conducted in formamide-containing hybridization buffer are shown in Fig.3A. Stronger signal intensities are observed for T-RNA targets when prepared in the hybridization buffer containing 30% formamide (Fig. 3B) indicating the utility of formamide to minimize any secondary structural motifs inherent in the target RNA sequences. Such a signal enhancement was observed even when the PNA probes (P-PNA) were printed at a low-salt probe buffer containing 0.1 M NaCl. Similar observations were obtained in Fig. 3A for the RNA target (TRNA-inf) which is related to the influenza A viral RNA sequence.

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Although 1X SSC (saline-sodium citrate) is the buffer commonly used for hybridization 207 experiments, we tested the capacity of our system to deliver reasonable results when using low-salt 208 hybridization buffers such as C (15 mM sodium citrate) in the presence of 30% formamide. A 209 comparison of hybridizations obtained in SSC and C buffers is illustrated in Fig. 4A. It is found that 210 intensities with signal-to-noise ratio greater than 2 are still observed using hybridization buffers 211 prepared in the low-salt conditions such as C buffer without NaCl added, suggesting that the need of 212 high salt for reducing Columbic repulsion is not as critical when PNA probes, rather than DNA 213 probes, are used. In fact, using PNA probes signals are still observable against the T-RNA-inf target 214 215 sequence even at 5nM using low-salt hybridization buffers (i.e. C buffer) are used (Fig. 4B).

216

217 Conclusion

The microfluidic microarray (MMA) chip has been employed to study the effects of buffer 218 compositions on PNA-RNA hybridization experiments. The presence of salt in the probe buffer and 219 hybridization buffer does result in better signals for the detection of RNA targets, although 220 hybridization still occurred to a reasonable extent in the absence of NaCl. Moreover, the presence of 221 formamide in the hybridization buffer offers better RNA hybridization when conducted at a relatively 222 low temperature of 40 °C. The MMA chip results support the notion that the PNA probe is more 223 useful than the DNA probe in low ionic strength conditions with respect to assay sensitivity and 224 experimental adaptability for RNA detection. 225

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231 **References**

232 (1) Nielsen, P. E. Current Opinion in Biotechnology. 1999, 10, 71. DOI:10.1016/S0958-

233 1669(99)80013-5

- (2) Brandt, O.; Hoheisel, J. D. *TRENDS in Biotechnology*, **2004**, 22, 12, 617.
- 235 DOI:10.1016/j.tibtech.2004.10.003
- 236 (3) Weiler, J.; Gausepohl, H.; Hauser, N.; Jensen, O. N.; Hoheisel, J. D. Nucleic Acids Research,
- **1997**, 25, 14. DOI:10.1093/nar/25.14.2792
- (4) Kilbourne, E. D. *Emerg Infect Dis.* **2006**, 12, 1, 9. DOI: 10.3201/eid1201.051254.
- (5) Teo, J.; Di Pietro, P.; San Biagio, F.; Capozzoli, M.; Deng, Y. M.; Barr, I.; Lin, R. Arch. Virol.
- **240 2011**, 156, 1371. DOI 10.1007/s00705-011-0999-7
- 241 (6) Jung, J.H.; Park, B.H.; Oh, S.J.; Choi, G.; Seo, T.S. *Biosen. Bioelectron.* 2015, 68, 218.
- 242 DOI:10.1016/j.bios.2014.12.043
- 243 (7) Kao, L. T. H.; Shankar, L.; Kang, T. G.; Zhang, G.; Tay, G. K. I.; Rafei, S. R. M.; Lee, C. W. H.
- 244 Biosen. Bioelectron. 2011, 26, 2006. DOI:10.1016/j.bios.2010.08.076
- 245 (8) Kim, Y. T.; Chen, Y.; Choi, J. Y.; Kim, W. J.; Dae, H. M.; Jung, J.; Seo, T. S. Biosen. Bioelectron.
- **246 2012**, 33, 88. DOI:10.1016/j.bios.2011.12.024
- 247 (9) Sun, Y.; Dhumpa, R.; Bang, D. D.; Høgberg, J.; Handberg, K.; Wolff, A. Lab Chip. 2011, 11,
- 248 1457. DOI: 10.1039/c0lc005628b

- 249 (10) Abe, T.; Segawa, Y.; Watanabe, H.; Yotoriyama, T.; Kai, S.; Yasuda, A.; Tojo, N. Lab Chip.
- **250 2011**, 11, 1166. DOI: 10.1039/c0lc00519c
- 251 (11) Jung, J. H.; Choi, S. J.; Park, B. H.; Choi, Y. K.; Seo, T. S. Lab Chip. 2012, 12, 1598. DOI:
- 252 10.1039/c2lc21269b
- 253 (12) Park, B. H.; Jung, J. H.; Zhang, H.; Lee, N. Y.; Seo, T. S. Lab Chip. 2012, 12, 3875. DOI:
- 254 10.1039/c2lc40487g
- 255 (13) Hagan, K. A.; Reedy, C. R.; Uchimoto, M. L.; Basu, D.; Engel, D. A.; Landers, J. P. Lab Chip.
- **256 2011**, 11, 957. DOI: 10.1039/c0lc00136h
- 257 (14) Cao Q. PLOS One. 2012, 7, e33176. DOI:10.1371/journal.pone.0033176
- 258 (15) Song, H. O.; Kim, J. H.; Ryu, H. S.; Lee, D. H.; Kim, S. J.; Kim, D. J.; Park, H. PLOS One.
- **259 2012**, 7, e53325. DOI:10.1371/journal.pone.0053325
- 260 (16) Liu, R. H.; Lodes, M. J.; Nguyen, T.; Siuda, T.; Slota, M.; Fuji, H. S.; McShea, A. Anal. Chem.
- **261 2006**, 78, 4184. DOI: 10.1021/ac060450v
- 262 (17) Xu, G.; Hsieh, T. M.; Lee, D. Y.; Ali, E. M.; Xie, H.; Looi, X. L.; Ying, J. Y. Lab Chip. 2010,
- 263 10, 3103. DOI: 10.1039/c005265e/
- 264 (18) Pal, R.; Yang, M.; Lin, R.; Johnson, B. N.; Srivastava, N.; Razzacki, S. Z.; Thwar, P. K. Lab
- 265 *Chip.* **2005**, 5, 1024. DOI: 10.1039/b505994a
- 266 (19) Jung, J. H.; Park, B. H.; Oh, S. J.; Choi, G.; Seo, T. S. Lab Chip. 2015, 15, 718. DOI:
- 267 10.1039/c4lc01033g
- 268 (20) Chih Wang, C. H.; Lien, K. Y.; Hung, L. Y.; Lei, H. Y.; Lee, G. B. Microfluid. Nanofluid. 2012,
- 269 13, 113. DOI: 10.1007/s10404-012-0947-1
- 270 (21) Rodriguez, N. Anal. Chem. 2015, 87, 7872. DOI: 10.1021/acs.analchem.5b01594

- 271 (22) Nagatani, N.; Yamanaka, K.; Ushijima, H.; Koketsu, R.; Sasaki, T.; Ikuta, K.; Tamiya, E.
- 272 *Analyst.* **2012**, 137, 3422. DOI: 10.1039/c2an16294f
- 273 (23) Prakash, R.; Pabbaraju, K.; Wong, S.; Wong, A.; Tellier, R.; Kaler, K. V. I. S. J Electrochem.
- 274 Soc. 2014, 161, B3083. DOI: 10.1149/2.013402jes
- 275 (24) Jung, J. H.; Park, B. H.; Choi, Y. K.; Seo, T. S. Lab Chip. 2013, 13, 3383. DOI:
- 276 10.1039/c3lc50266j
- 277 (25) Neuzil, P.; Novak, L.; Pipper, J.; Lee, S.; Ng, L. F.; Zhang, C. Lab Chip. 2010, 10, 2632. DOI:
- 278 10.1039/c004921b
- 279 (26) Bhattacharyya, A.; Klapperich, C. M. Sens. Actuat. B 2008, 129, 693.
- 280 DOI:10.1016/j.snb.2007.09.057
- (27) Reichmuth, D. S.; Wang, S. K.; Barrett, L. M.; Throckmorton, D. J.; Einfeld, W.; Singh, A. K.
- 282 *Lab Chip.* **2008**, 8, 1319. DOI: 10.1039/b801396a
- 283 (28) Yamanaka, K.; Saito, M.; Kondoh, K.; Hossain, M. M.; Koketsu, R.; Sasaki, T.; Tamiya, E.
- 284 Analyst. 2011, 136, 2064. DOI: 10.1039/C1AN15066A
- 285 (29) Wojciech Kubicki, W.; Pajak, B.; Kucharczyk, K.; Walczak, R.; Dziuban, J. A. Sens. Actuat. B.
- **286 2016**, 236, 926. DOI: 10.1016/j.snb.2016.03.083
- 287 (30) Dawson, E. D.; Moore, C. L.; Smagala, J. A.; Dankbar, D. M.; Mehlmann, M.; Townsend, M. B.;
- 288 Rowlen, K. L. Anal. Chem. 2006, 78, 7610. DOI: 10.1021/ac061739f
- 289 (31) Kessler, N.; Ferraris, O.; Palmer, K.; Marsh, W.; Steel, A. J Clin. Microbiol. 2004, 2173. DOI:
- 290 10.1128/JCM.42.5.2173–2185.2004
- (32) Townsend, M. B.; Dawson, E. D.; Mehlmann, M.; Smagala, J. A.; Dankbar, D. M.; Moore, C. L.;
- 292 Rowlen, K. L. J Clin. Microbiol. 2006, 44, 2863. doi:10.1128/JCM.00134-06

- 293 (33) Nagatani, N.; Yamanaka, K.; Saito, M.; Koketsu, R.; Sasaki, T.; Ikuta, K.; Tamiya, E. Analyst.
- **294 2011**, 136, 5143. DOI: 10.1039/c1an15638a
- (34) Silvester, N. C.; Bushell, G. R.; Searles, D. J.; Brown, C. L. Org. Biomol. Chem. 2007, 5, 917–
 923
- (35) Kim, H.; Choi, J. J.; Cho, M.; Park, H. *BioChip Journal*. **2012**, 6, 1, 25. DOI:10.1007/s13206-
- **298** 012-6104-9
- (36) Arlinghaus, H. F.; Kwoka, M. N.; Jacobson, K. B. *Anal. Chem.* 1997, 69, 3747. DOI:
 10.1021/ac970267p
- 301 (37) Wang, L.; Li, P. C. J. Agri. Food Chem, 2007, 55, 10509. DOI: 10.1021/jf0721242
- 302 (38) Wang, L.; Li, P. C. *Biomicrofluidics*, **2010**, *4*, 032209. DOI: 10.1063/1.3463720
- 303 (39) Sedighi, A.; Li, P. C. Anal. Biochem. 2014, 448, 58. DOI: 10.1016/j.ab.2013.11.019
- 304 (40) Chen, H.; Li, P.C.; Yu, H.C.; Parameswaran, M.; Yoganathan, N. Proc. 11th Conf. Miniaturized
- Chem. Biochem. Analysis Systems, Paris, France, Oct. 7-11, 2007, 1282
- 306 (41) Benn, J. A.; Hu, J.; Hogan, B. J.; Fry, R. C.; Samson, L. D.; Thorsen, T. Analytical Biochemistry.
- **2006**, 348, 284. DOI: 10.1016/j.ab.2005.10.033
- 308 (42) Howbrook, D. N.; van der Valk, A. M.; O'Shaughnessy, M. C.; Sarker, D. K.; Baker, S. C.;
- 309 Lloyd, A. W. DDT. 2003, 8, 14, 642. DOI: 10.1016/S1359-6446(03)02773-9
- 310 (43) Abruzzo, L. V.; Lee, K. Y.; Fuller, A.; Silverman, A.; Keating, M. J.; Medeiros, L. J.; Coombes,
- 311 K. R. BioTechniques. 2005, 38, 785
- 312 (44) Livak-Dahl, E.; Sinn, I.; Burns, M. Annu. Rev. Chem. Biomol. Eng. 2011, 2, 325. doi:
- 313 10.1146/annurev-chembioeng-061010-114215
- 314 (45) Vorwerk, S.; Ganter, K.; Cheng, Y.; Hoheisel, J.; Stähler, P. F.; Beier, M. New Biotechnology.
- **2008**, 25, 142. doi: 10.1016/j.nbt.2008.08.005

- 316 (46) Resa-Infante, P.; Jorba, N.; Coloma, R.; Ortín, J. *RNA Biol.* **2011**, 8, 2, 207. DOI:
- 317 10.4161/rna.8.2.14513
- 318 (47) Webster, R. G.; Bean, W. J.; Gorman, O. T.; Chambers, T. M.; Kawaoka, Y. Microbiol. Mol.
- 319 *Biol. Rev.* **1992**, 56, 1, 152.
- 320 (48) Arjona, A.; Ledizet, M.; Anthony, K.; Bonafé, N.; Modis, Y.; Town, T.; Fikrig, E. J. Immunol.,
- **2007**, 179, 8403. DOI: 10.4049/jimmunol.179.12.8403
- 322 (49) Mukhopadhyay, S.; Kuhn, R. J.; Rossmann, M. G. *Nature Reviews Microbiology*. 2005, 3, 1, 13.
- 323 DOI:10.1038/nrmicro1067
- 324 (50) Silvester, N.C. *Griffith University PhD thesis*. **2007**, 25.
- 325 (51) Peng, X. Y. L.; Li, P. C.; Yu, H. Z.; Ash, M. P.; Chou, W. L. J. Sens. Actuators B, 2007, 128, 64.
- 326 DOI: 10.1016/j.snb.2007.05.038
- 327 (52) Chen, H., Li, X.; Wang, L.; Li, P. C. *Talanta*, **2010**, *81*, 1203. DOI:
- 328 10.1016/j.talanta.2010.02.011
- 329 (53) Eun, H.M. Academic Press, **1996**, 67.
- 330 (54) Lehmann, R.; Tautz, D. Methods in Cell Biology. 1994, 44, 575. DOI: 10.1016/S0091-
- 331 679X(08)60933-4
- 332 (55) Demers, L. M.; Mirkin, C. A.; Mucic, R. C.; Reynolds, R. A.; Letsinger, R. L.; Elghanian, R.;
- 333 Viswanadham, G. *Analytical chemistry*. **2000**, 72, 22, 5535-5541. DOI: 10.1021/ac0006627
- 334 (56) Erickson, D., Liu, X.; Krull, U., ; Li, D. Analytical Chemistry. 2004, 76, 24. DOI:
- 335 10.1021/ac049396d
- 336
- 337
- 338

Table 1. Sequences of probes (P) and targets (T) used to study PNA-RNA hybridizations

Identifier	Sequence
	Probes
P-DNA	$H_2N\text{-}C6\text{-}CTGTATTG\textbf{A}GTTGTATCGTGTGTGTGTATTT$
P-PNA	H ₂ N-O-O-AGTTGTATCGTGT
P-PNA-inf	H ₂ N-O-O-CCTTGTTTCTACT
	Targets
T-RNA b	viotin-ACGGAGCGCAAAAUACACCACCACGAUACAACUCAAUACAGUCGACGCCUA

T-RNA-inf biotin-AGUAGAAACAAGGCCUGCUUUUG

Fig. 1. Detection procedure using the microfluidic microarray (MMA) chip. The schematic diagrams of printing the probe array (A), hybridization with the RNA targets (B), detection of fluorescent results on the glass slide surface (C); an image of the experimental results (D).



Fig. 2. (A) Image and (B) background-subtracted intensity of hybridization of T-RNA with the
complementary DNA probe (P-DNA) and PNA probe (P-PNA) in different low-salt buffer (0.1 M
NaCl) and high-salt spotting buffer (1 M NaCl). The probe solutions contain P-DNA (20 μM), 0.15 M
sodium bicarbonate, and low/high salt content. The target solutions contain T-RNA (20 nM or 10
nM), 1X SSC (saline-sodium citrate) buffer, and 0.15% SDS.









Ster C







