

Efficacy and safety of iron-chelation therapy with deferoxamine, deferiprone, and deferasirox for the treatment of iron-loaded patients with non-transfusion-dependent thalassemia syndromes

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Abstract: The prevalence rate of thalassemia, which is endemic in Southeast Asia, the Middle East, and the Mediterranean, exceeds 100,000 live births per year. There are many genetic variants in thalassemia with different pathological severity, ranging from a mild and asymptomatic anemia to life-threatening clinical effects, requiring lifelong treatment, such as regular transfusions in thalassemia major (TM). Some of the thalassemias are non-transfusion-dependent, including many thalassemia intermedia (TI) variants, where iron overload is caused by chronic increase in iron absorption due to ineffective erythropoiesis. Many TI patients receive occasional transfusions. The rate of iron overloading in TI is much slower in comparison to TM patients. Iron toxicity in TI is usually manifested by the age of 30–40 years, and in TM by the age of 10 years. Subcutaneous deferoxamine (DFO), oral deferiprone (L_1), and DFO- L_1 combinations have been effectively used for more than 20 years for the treatment of iron overload in TM and TI patients, causing a significant reduction in morbidity and mortality. Selected protocols using DFO, L_1 , and their combination can be designed for personalized chelation therapy in TI, which can effectively and safely remove all the excess toxic iron and prevent cardiac, liver, and other organ damage. Both L_1 and DF could also prevent iron absorption. The new oral chelator deferasirox (DFX) increases iron excretion and decreases liver iron in TM and TI. There are drawbacks in the use of DFX in TI, such as limitations related to dose, toxicity, and cost, iron load of the patients, and ineffective removal of excess iron from the heart. Furthermore, DFX appears to increase iron and other toxic metal absorption. Future treatments of TI and related iron-loading conditions could involve the use of the iron-chelating drugs and other drug combinations not only for increasing iron excretion but also for preventing iron absorption.

Keywords: thalassemia intermedia, chelation therapy, deferoxamine, deferiprone, deferasirox, iron overload, iron absorption, efficacy, safety

Introduction

Iron is an essential metal found in all living organisms, and is involved in many physiological processes, such as oxygen transport and energy transduction.^{1,2} It is estimated that more than a quarter of the human population is affected at some stage of their life by abnormalities of iron metabolism, in particular iron-deficiency anemia. Similarly, many millions of people suffer from iron overload, including hereditary hemochromatosis and thalassemia intermedia (TI), which are caused by increased iron absorption.^{3–6} In these and other iron-loading conditions, excess iron is toxic, causing increased morbidity and mortality as a result of tissue damage. Iron plays an important catalytic role in free radical pathology and oxidative stress damage, which is not only observed in iron-overload

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diseases but also in non-iron-loaded diseases of focal or labile iron toxicity, such as cardiovascular, neurodegenerative, hepatic, and renal diseases, as well as in cancer and aging.^{7,8}

A major group of diseases related to iron abnormalities involving millions of people are the inherited hemoglobinopathies, especially thalassemia and sickle-cell disease. It is estimated that the prevalence of these two inherited hemoglobinopathies exceeds 200,000 live births per year, with approximately 100,000 for each disease.^{3,4} Sickle-cell disease is endemic in Africa, Central and South America, and the Afro-American population of North America, and is also distributed with lower frequency in Middle East and the Mediterranean countries. Thalassemia is endemic mainly in Southeast Asia, the Middle East, and Mediterranean countries.

There are many genetic variants of thalassemia related to the hemoglobin (Hb) chains (α , β , γ , δ , and ϵ). Many of the thalassemia variants can cause pathological effects of different severity, ranging from a mild and asymptomatic anemia to death, eg, in hydrops fetalis in α -thalassemia (4/4 α -globin gene deletion).⁹ A major group of thalassemia patients with life-threatening clinical effects requiring life long treatments, such as regular red blood cell transfusions within a few years of life, is β thalassemia major (TM) (2/2 β -globin gene deletion).^{3,4} Iron overload in TM is caused mainly from chronic transfusions and to a lesser extent by increased iron absorption. Another major group of thalassemias of milder severity than TM are the iron-loaded non-transfusion-dependent thalassemias (IL-NTDTs), which include many TI variants where iron overload is caused by chronic increase in iron absorption due to ineffective erythropoiesis.^{3,4,9–12} The same mechanism, but with a higher rate of iron absorption, is also observed in nontransfused TM patients in developing countries, where no transfusion or chelation therapy is available.

Many patients with IL-NTDT may have a number of red blood-cell transfusions because of low Hb levels, infections, and other causes. The rate of transfusion increases with age in most cases of IL-NTDT, eg, many TI patients at the age of 30–50 years may be regularly transfused at a rate similar to that of TM patients.^{9–12} The rate of iron overload from increased iron absorption in IL-NTDT is generally much slower in comparison to iron accumulated from transfusions in regularly transfused TM patients.^{12,13} However, the rate of iron absorption in IL-NTDT is generally similar to that observed in hereditary hemochromatosis, and in both conditions iron overload is manifested in many cases very early in life, even from early childhood.^{9–15}

Genetic, dietary, immunological, and other factors influence the rate of iron-overload and related toxicity in both IL-NTDT

and TM. Iron-overload toxicity is manifested in many organs, including the heart, liver, spleen, and pancreas.^{11,16–19} Iron overload in IL-NTDT can lead progressively to cardiomyopathy, liver fibrosis, cirrhosis, hepatocellular carcinoma, and diabetes.^{11,16–21} In addition to iron overload-related toxicity, ineffective erythropoiesis, and chronic anemia, there are also many other abnormalities related to the underlying condition in TI and other similar IL-NTDTs, such as hypothyroidism, hypogonadism, hypersplenism, pulmonary hypertension, and osteoporosis.^{11,16–21}

Different clinical strategies are generally used for the treatment of TI and other similar IL-NTDT patients, which mainly include red blood-cell transfusions, increase in HbF production using hydroxycarbamide (hydroxyurea), splenectomy, and iron-chelation therapy.^{9–11}

Iron-chelation therapy can decrease and in many cases eliminate the morbidity and mortality rates associated with iron toxicity in IL-NTDT syndromes. Iron-chelation strategies in patients with this syndrome are similar to those in TM, and are generally based on the early diagnosis and prevention of iron overload and toxicity and also the rapid normalization of the iron stores.^{14,15} The chelating drugs and chelation protocols used in IL-NTDT are similar to those in TM. Most of these protocols involve the administration of subcutaneous deferoxamine (DFO), oral deferiprone (L_1), and DFO- L_1 combinations, which have been effectively used for more than 20 years for the treatment of iron overload in TM and TI patients (Figure 1).^{14,15} In addition, the recent introduction of oral deferasirox (DFX) has increased the options for the treatment of iron overload in TM and other conditions, including IL-NTDT (Figure 1). Several studies sponsored by the manufacturers of DFX have been recently published on the effect of the drug on IL-NTDT.²²

Personalized medicine is designed and appropriately used in each case for optimizing chelation therapy in patients with IL-NTDT and TM. This approach is generally based on the risk–benefit assessment of each chelator and chelation protocol, and also the idiosyncratic clinical profile of the patients, as determined by drug absorption, distribution, metabolism, excretion, and toxicity effects and other relevant parameters, such as organ function.²³ However, other factors are also considered in the selection criteria of chelating drugs, including costs, influence of the pharmaceutical companies, and drug policies of the regulatory authorities.^{24,25}

Epidemiology and severity of iron-loaded non-transfusion-dependent thalassemia syndromes

IL-NTDT is a syndrome characterized by excess iron deposition in organs similar to hereditary hemochromatosis,

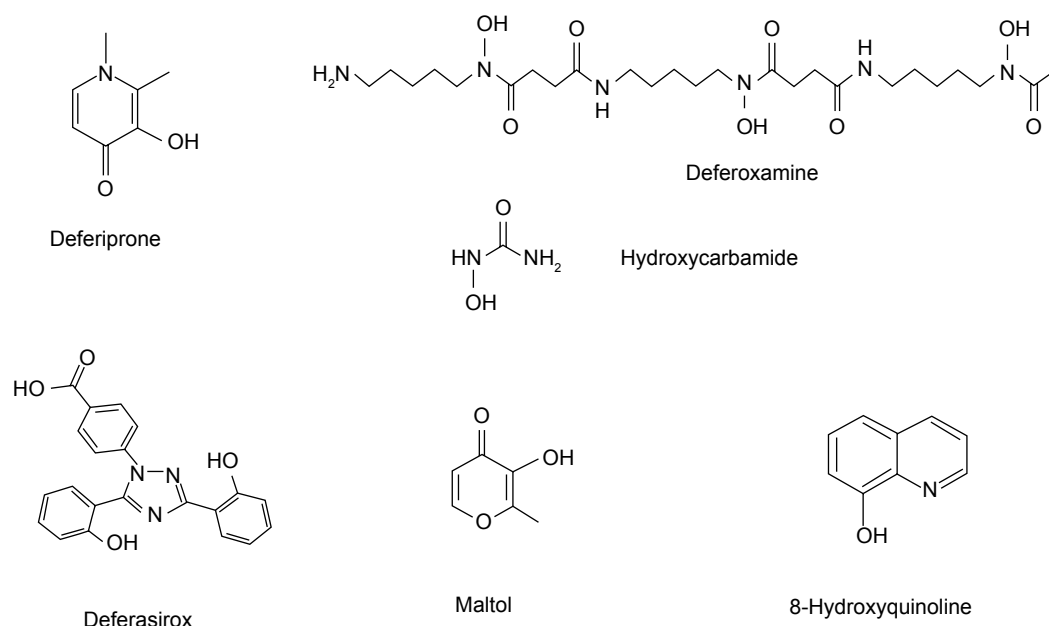


Figure 1 Structure of the chelating drugs.

Notes: The chemical and physicochemical properties of chelating drugs influence their clinical activity, including their mode of action, organ targeting, efficacy, and toxicity. Deferiprone and deferoxamine are hydrophilic chelators that increase iron excretion and decrease iron absorption. Maltol, deferasirox, and 8-hydroxyquinoline are lipophilic chelators that form lipophilic metal complexes, and can cause an increase in iron and other metal absorption. Orally absorbed, nonmetal-bound deferasirox mobilizes excess iron, mainly from the liver, and causes an increase in iron excretion. The pharmacological effects of different drugs, eg, hydroxycarbamide (hydroxyurea) are affected by iron binding.

which is caused by increased body-iron intake as a result of an increase in net body-iron intake from gastrointestinal absorption in comparison to total body-iron excretion and other losses.^{12–15,26,27} The level of iron overload and its severity in this syndrome is related primarily to hereditary abnormalities, but also a number of metabolic, environmental, dietary, aging, and other factors influencing the ferrokinetics of iron metabolism through different mechanisms.^{12–15,26,27}

In relation to the hereditary factors related to IL-NTDT syndromes, β -TI is the best-characterized genetic disorder, which is associated with a variety of gene mutations involving the β -globin chain of Hb and is predominantly found in Southeast Asia, Mediterranean, and Middle Eastern countries, similar to that of TM patients (Table 1).^{3,4} In TI patients, the normal production of Hb is affected to a variable extent depending on the genotype and age of the patient. Thousands of TI patients diagnosed with Hb variants of the α -, β -, γ -, δ -, and ϵ -globin genes exhibit milder forms of anemia than TM patients (Table 1).^{3,4}

There are several categories of TI patients where red blood-cell transfusions may be initiated to compensate for the decreasing level of Hb production. In some cases, the rate of transfusions may increase to levels similar to those observed in TM patients. There are also other groups of TI patients who may receive less frequent transfusions and some where transfusions are not necessary. In the latter group,

TI patients usually maintain Hb levels at 6–10 g/dL, which in many cases is sufficient in younger patients to be maintained without complications, and transfusion can be avoided.^{3,4} For example, increased production of HbF is associated with milder TI disease. In general, Hb production in TI decreases with age, and the general health of TI patients also deteriorates as they grow older.^{10,11,15}

Similar variations are observed in HbE β -thalassemia patients, most of whom live in developing countries of Southeast Asia, and also in immigrants from this region.^{3,4,17,27,28} It is estimated that the HbE β -thalassemia genotype is responsible for approximately 50% of all severe β -thalassemias worldwide (Table 1). The disorder is characterized by marked clinical variability and Hb production, ranging from a mild and asymptomatic anemia to a life-threatening disorder requiring transfusions from infancy. For example, no treatment is required for mild HbE β -thalassemia that is characterized by normal Hb levels (9–12 g/dL). Patients in this category do not usually develop clinically significant symptoms. However, more severe forms are characterized by decreased Hb levels (6–8 g/dL) and the clinical manifestations are similar to those of TI, where transfusions are not required unless infections or other complications increase the anemia further. Most patients of HbE β -thalassemia (~50%) are thought to be included in this category of IL-NTDT, where iron overload from progressive

Table 1 Examples of thalassemia intermedia and other conditions where iron overload can be caused from the increased gastrointestinal absorption of iron*

Thalassemia genotypes

β -Thalassemia

- Heterozygous mild β^+ /severe β^+
- Homozygous β^0 or β^+ with genetic factors leading to increased γ -chain production
- Homozygous normal HbA₂ β -thalassemia (type I)
- Heterozygous β^0 or β^+ /normal HbA₂ β -thalassemia (type I)
- Severe heterozygous β -thalassemia
- $\delta\beta$ -thalassemia and high persistent fetal hemoglobin (HPFH)
- Homozygous $\gamma\delta$ -thalassemia
- Heterozygous β^0 - or β^+ -thalassemia/ $\delta\beta$ -thalassemia
- Heterozygous Greek HPFH/ β -thalassemia

β or $\delta\beta$ thalassemia with structural hemoglobin variants

- Hb S, C, or E in association with β^0 -, β^+ -, or $\delta\beta$ -thalassemia
- HbE thalassemia
- HbE β -thalassemia

α -Thalassemia

- Homozygous α (0 or $-/-$) hydrops fetalis
- Heterozygous α ($-/-\alpha$) (HbH disease)
- Heterozygous α ($-/\alpha\alpha$)

Interactions of α - and β -thalassemia

Interactions of α -thalassemia and hemoglobin S

Other iron-loading conditions caused from increased iron absorption

Hereditary hemochromatosis

- Related to the *HFE* gene. Not related to *HFE* gene.
- Juvenile hemochromatosis
- Neonatal hemochromatosis

Chronic anemias

- Sideroblastic anemia. Congenital atransferrinemia

Exogenous iron overload

- Oral iron supplements. Accidental iron poisoning

Chronic liver diseases

- Viral hepatitis. Alcoholic liver disease. Nonalcoholic steatohepatitis

Porphyria cutanea tarda

Notes: *Increased gastrointestinal iron absorption is observed in different thalassemia intermedia genotypes, hereditary hemochromatosis, and other conditions, which can lead to iron overload. The main diagnostic criteria for iron overload are increased serum ferritin and transferrin iron-saturation levels and decrease in the signal intensity of magnetic resonance imaging T₂ or T₂* in the liver and heart, due to excess iron accumulation and deposition.

Abbreviation: Hb, hemoglobin.

increases in iron absorption may occur. It is estimated that about a third of the cases of HbE β -thalassemia are present with the severe form of the disease, which is characterized by very low Hb levels (4–5 g/dL), and the clinical symptoms are similar to those of TM patients. Regular red blood-cell transfusions and iron-chelation therapy are required for the survival of this category of HbE β -thalassemia patients similar to TM patients.^{3,4,17,27,28}

There are many other categories of patients with similar rates of body-iron loading to that of IL-NTDT that are due

to increased iron absorption and require chelation therapy (Table 1). These include hereditary hemochromatosis patients who cannot be treated with venesection because of complications, such as severe congestive heart failure, cardiomyopathy, severe anemia, advanced liver cirrhosis, or poor peripheral venous access, and also patients with porphyria cutanea tarda (Table 1).¹⁵ Additionally, there are other categories of patients, such as those with sickle-cell anemia or sickle-cell anemia with β -thalassemia, with similar rates of body-iron loading to that of IL-NTDT, which are caused by occasional red blood-cell transfusions (Table 1).^{10,11,15} In all these cases, the body-iron intake from increased gastrointestinal absorption and occasional red blood-cell transfusions is much lower and iron accumulation is at a slower rate in comparison to regularly transfused TM patients. Accordingly, the toxic side effects of iron overload are manifested at different ages in IL-NTDT, and less intensive chelation therapy may be used at younger ages for the treatment of iron overload and toxicity in comparison to TM.

Molecular mechanisms of increased iron absorption and storage in iron-loaded non-transfusion-dependent thalassemias

Under normal conditions, the regulatory uptake and utilization of iron appear to involve a number of pathways and proteins that control the rate of iron transfer from the gastrointestinal tract to the bloodstream such as ferrireductase(s), DMT1, ferroportin, hepcidin, transferrin, and ferritin (Figure 2).^{1,2} Different changes in the regulatory mechanisms of dietary iron uptake from the gastrointestinal tract are suggested for causing increased iron absorption in hereditary hemochromatosis, TI, and other similar conditions.^{1,2,15}

The mechanism and the regulatory pathway of iron absorption is thought to involve initially the conversion of dietary ferric iron to ferrous iron by a ferrireductase present at the cell surface of the enterocyte and then its intracellular transport by DMT1 (Figure 2). Once inside the cell, iron is partly incorporated into ferritin or transferred into the low-molecular-weight intracellular iron pool. Some of the iron is then transported to ferroportin at the basolateral membrane of the enterocyte, which exports iron into the plasma. The critical role of iron absorption is thought to be played by hepcidin, a peptide hormone produced in the liver. The mode of action of hepcidin is the prevention of iron release into plasma by binding ferroportin, causing its internalization and degradation within the enterocyte. Following the binding of ferroportin by hepcidin, iron is not released into plasma,

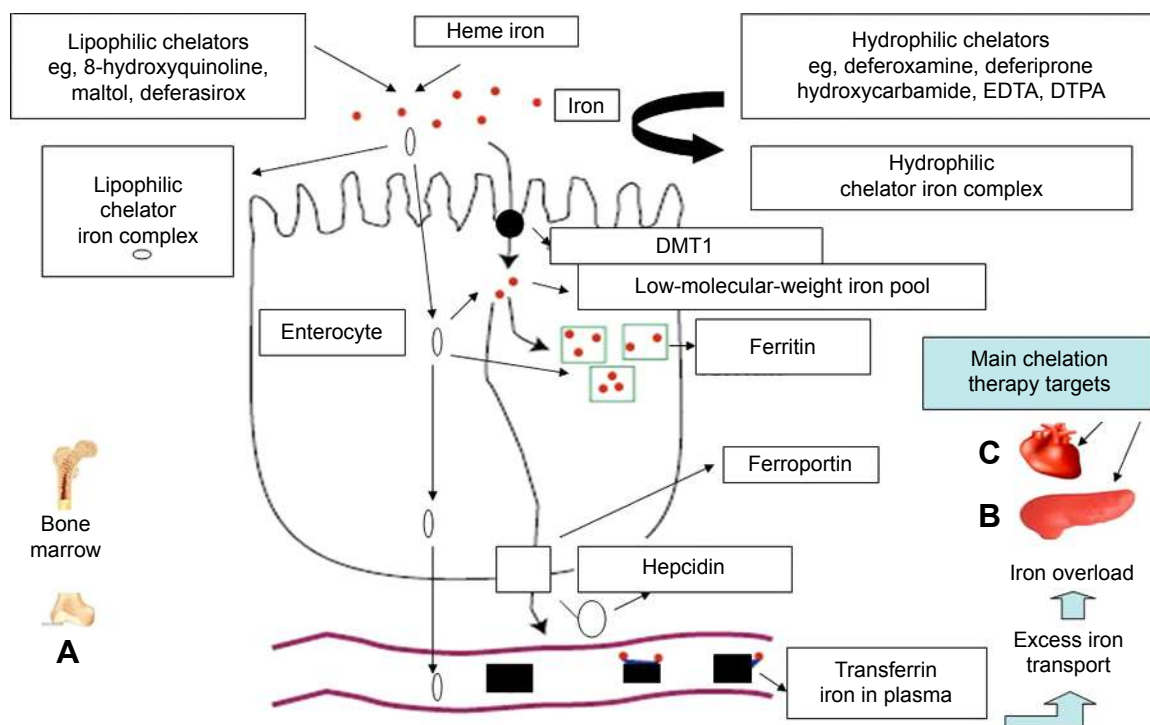


Figure 2 Iron-absorption and iron-overload mechanisms in non-transfusion-dependent thalassemias: the role of chelators and chelating drugs.

Notes: Mechanism of iron absorption at the enterocyte using regulatory pathways of iron metabolism involving DMT1, hepcidin, ferroportin, and transferrin. Increased gastrointestinal iron absorption and iron overload is observed in non-transfusion-dependent thalassemias, due to ineffective erythropoiesis in the bone marrow (A). The level of increased iron absorption depends on the form and quantity of iron present in the diet and other factors, such as the presence of natural or synthetic iron chelators in the gastrointestinal tract. Iron-chelating drugs and other chelators have variable pharmacological effects on iron absorption, with lipophilic chelators causing an increase in iron absorption and hydrophilic chelators a decrease in iron absorption. Excess iron absorption causes iron overload and damage in the liver, the heart, and other organs. The liver is the main organ of excess iron deposition, whereas the heart is the most susceptible organ of iron toxicity, as a result of iron overload from increased iron absorption. Iron overload in both the liver (B) and the heart (C) in non-transfusion-dependent thalassemia are the main target sites of chelation therapy.

Abbreviations: DMT1, divalent metal transporter 1; EDTA, ethylenediaminetetraacetic acid; DTPA, diethylenetriaminepentaacetic acid.

but remains within the enterocytes and returns in the gut lumen, since enterocytes are shed every 2–3 days. Within this context, a decrease in plasma hepcidin concentration will increase iron absorption.^{1,2,15}

Iron released into plasma from the basolateral membrane of the enterocyte by ferroportin is mobilized by plasma transferrin and transferred to all the cells of the body. The uptake of transferrin iron by cells is mediated by transferrin receptors that are present at the cell surface. Increased transferrin receptors are present especially in hepatocytes for iron storage and the erythroid cells for Hb synthesis. Intracellular release of iron is mediated by receptor-mediated endocytosis, through the internalization in a vesicle and acidification of the transferrin receptor–transferrin–iron complex.^{1,2}

The expression of hepcidin, DMT1, and ferroportin appears to be affected by a number of factors. In particular, the signal to the intestine to increase iron absorption appears to be influenced by regulatory molecules (iron-responsive elements) which sense the iron stores.^{1,2,29,30} Different iron-responsive element regulatory molecules are also present, which appear to sense the production of Hb or anemia.

In both cases, the influence of the regulatory molecules cause an increase in the expression of DMT1 and a corresponding increase in the uptake of iron. In general, it would appear that in TI and other IL-NTDT, the signal arriving at the enterocytes is a combination of ineffective erythropoiesis, anemia, and hypoxia, which leads to the initiation of compensatory mechanisms, including an increase in plasma levels of erythropoietin, as well as a decrease in plasma levels of hepcidin. The decrease in hepcidin increases the rate of transfer of iron by ferroportin on the intestinal epithelium into the bloodstream, followed by an increase of transferrin iron uptake, transferrin iron delivery, and intracellular storage, mainly in the liver hepatocytes but also in other organs, including the heart and the pancreas.^{12,13,16–19}

The level of iron overload in IL-NTDT patients is mainly characterized by increases in serum ferritin, plasma iron, and transferrin iron saturation, as well as by increases in iron concentration in the liver, heart, and other organs, which can be measured by magnetic resonance imaging (MRI) T_2 and T_2^* relaxation-time values.^{10–12,16,17,31,32} The same methods are also used for monitoring the progress of chelation therapy,

as well as other treatments used for the reduction of iron absorption or the elimination of excess iron in organs of patients with IL-NTDT.

Iron overload has important clinical consequences in both patients with TI and TM. However, unlike TM, where iron overload is mostly caused from transfused red blood cells and is initially stored in the Kupffer cells of the liver, the iron overload caused in IL-NTDT originates from the absorption and intake of excess iron present in the gastrointestinal tract, which is mainly delivered and stored in the hepatocytes. Since iron accumulation primarily occurs in the hepatocytes, iron overload can predispose patients to liver fibrosis and cirrhosis, hepatocellular carcinoma, and other toxic side effects similar to hereditary hemochromatosis patients.^{20,21,30,31} Cardiac complications, including cardiac iron overload, are also very common in TI and other IL-NTDT patients, and in both cases appropriate therapeutic interventions are needed.^{16–20,30,31}

Nongenetic factors influencing the rate of iron absorption and body-iron intake in iron-loaded non-transfusion-dependent thalassemias

The genotype of thalassemia is the major factor affecting the rate of gastrointestinal iron absorption and body-iron loading in IL-NTDT. However, there are many other additional factors that also appear to influence this process, and this is reflected by the wide variation in iron-loading in IL-NTDT patients with the same genotype. Some of these factors may be influenced by changes in relation to the iron-metabolism pathways and related proteins and regulatory molecules, such as hepcidin and ferroportin. Infection, inflammation, and pathological conditions may also influence normal iron metabolism and the rate of iron absorption in IL-NTDT (Table 2).^{1,2,5,6,10,11}

Dietary habits are also important factors affecting the rate of iron absorption in IL-NTDT patients, but also in normal individuals. Within this context, the quantity and quality (ferrous, ferric, heme) of iron in food, as well as other dietary factors or constituents, such as ascorbic acid, polyphenols, inorganic phosphates, and tannins, can also influence the rate of iron absorption.^{15,26} Increased iron absorption is anticipated when dietary iron is in the form of heme, in comparison to ferrous iron and to a lesser extent ferric iron, which is the least soluble and absorbable form.^{15,26} The presence of reducing agents, such as ascorbic acid, could increase the absorption of iron by reducing ferric to ferrous iron. Low pH and dietary factors that increase the solubility of iron can also increase the

Table 2 Examples of nonregulatory factors affecting iron absorption*

Quantity of iron present in the diet, eg, vegetarian meals contain less iron
Quality of iron (ferrous, ferric, heme, ferritin, hemosiderin) in the diet
Presence of dietary reducing agents, eg, ascorbic acid
Presence of dietary molecules with chelating properties, eg, tannins and polyphenols
Presence of oral drugs with chelating properties, eg, tetracycline and hydroxycarbamide
The quantity of water, alcohol, and other fluid intake in the gastrointestinal tract
Drugs and dietary molecules affecting iron transport across the enterocyte (eg, nifedipine, which is an L-type calcium-channel blocker)
Dietary factors affecting the solubilization of iron
pH of the stomach and intestine
Modulators of DMT1, hepcidin, ferroportin, and transferrin activity
Gastrectomy and other interventions where gastrointestinal function is affected
Infectious, inflammatory, chronic liver disease, and other diseases
Aging, sports
Exogenous iron supplementation
Chronic iron supplementation. Oral iron supplements. Use of iron cooking utensils

Notes: *Many dietary, nonregulatory, and other factors such as those described in this table can affect the absorption of iron under normal conditions. The same and similar factors, as well as other regulatory factors, such as the erythropoietic activity of the bone marrow in thalassemia intermedia and other iron-loaded non-transfusion-dependent thalassemias where ineffective erythropoiesis is observed, can also affect the rate of iron absorption and can lead to variable levels of iron deposition and overload in the liver, the heart, and other organs of the affected patients.

Abbreviation: DMT1, divalent metal transporter 1.

absorption of iron (Table 2).^{15,26} In contrast, decrease in iron intake and iron deficiency is generally observed in vegetarians, where absorbable dietary iron is low in concentration. Consequently, vegetarian meals could decrease the rate of iron overload in IL-NTDT in comparison to meals with increased and more absorbable forms of iron from animal food products, such as red meat.^{15,26}

Drugs may also influence the rate of iron absorption. Within this context, increased absorption of iron can be facilitated by natural or synthetic lipophilic chelators, such as maltol, 8-hydroxyquinolone, and DFX, all of which form lipophilic iron complexes.^{14,15,26,33} In contrast, decreased absorption of iron could be caused using hydrophilic chelators, such as DFO and L₁, which form hydrophilic and/or charged iron complexes, as well as by chelators forming insoluble iron complexes (Figure 2, Table 2).^{14,15,26,33}

The ferrokinetic, tissue-distribution, and toxicity profiles vary among the different categories of iron-loading conditions, including TM, IL-NTDT, and hereditary hemochromatosis. In normal individuals and under normal conditions, the amount of iron absorbed from the diet is equivalent to the amount of iron excreted (1–4 mg).^{12,15,26} Different levels

of increased iron absorption are observed (2.6–8.6 mg/day) in nontransfused TI patients, which in most cases can progressively over many years lead to iron overload and subsequently to tissue and organ damage.^{12–15} This pattern of increased rate of iron absorption from the diet, which is estimated to be up to approximately 6 mg/day, is similar to other IL-NTDT and also in hereditary hemochromatosis patients.^{12–15} In comparison, it is estimated that the daily body-iron intake in TM from red blood-cell transfusions for maintaining Hb levels at approximately 9–10 g/dL is approximately 20 mg/day and from increased gastrointestinal absorption 1–6 mg/day.^{12–15,26}

In TM patients, organ damage due to iron overload toxicity in the absence of chelation therapy appears within a few years of initiating red blood-cell transfusions or when approximately 50 units of packed red blood cells (equivalent to 10.0–12.5 g of iron) have been transfused.^{14,15} Similar levels of iron overload appear to cause organ toxicity in TI and other related IL-NTDT conditions, but at an older age.^{14,15}

Alternative treatments to red blood-cell transfusions are generally used in early diagnosed TI and other related IL-NTDT conditions for increasing Hb levels. Within this context, early diagnosed nontransfused TI patients with low Hb levels are currently treated with hydroxyurea, which appears to be effective in increasing HbF levels in some TI genotypes.^{28,34,35} The response to hydroxyurea is variable, and iron-loading, pharmacogenomic, and other factors appear to influence its efficacy.^{28,34–36} However, in many cases, and especially in older TI patients, transfusions may be initiated because of ineffective hydroxyurea treatment and increasing anemia. Erythropoietin has also been tested for elevating Hb levels, but the results were not satisfactory and further development in this area was not pursued.³⁷

In undiagnosed and untreated IL-NTDT patients, excess toxic iron is accumulated over the years, leading to organ damage, including mainly liver cirrhosis, hepatocellular carcinoma, cardiomyopathy, and diabetes. During the progress of the disease, excess iron gradually accumulates initially in the liver and later to other organs, leading to an increase in serum iron, transferrin iron saturation, and serum ferritin and hepatic iron concentration, usually before the age of 30 years. In many cases, tissue damage can be diagnosed by the age of 30 years and liver cirrhosis by the age of 40 years. However, there is great variability in the progress and development of iron overload and organ damage among IL-NTDT patients and other related conditions.³⁸ For example, Hb levels, dietary habits, hydroxyurea treatment, and many other factors can lead to different rates of iron absorption and iron overload-related tissue damage,

which overall can influence the progress of the disease in IL-NTDT.

Within this context, the early diagnosis and treatment of TI and related IL-NTDT could prevent iron-overload toxicity, related tissue and organ damage, and overall decrease the morbidity and mortality in these conditions. Normal life expectancy is expected for early diagnosed and treated IL-NTDT patients who have not developed permanent organ damage, such as liver cirrhosis, cardiomyopathy, and diabetes.²⁴

Reduction of body-iron load in IL-NTDT and other related conditions of iron overload due to increased gastrointestinal iron absorption, as well as transfusional iron overload in TM, can only be accomplished using chelation therapy and specifically the iron-chelating drugs DFO, L₁, and DFX (Figure 1).^{14,24}

Properties and effects of chelators and chelating drugs in relation to iron absorption

Chelating drugs and chelators in general have different properties that can affect iron excretion as well as absorption, and can play a major role in the management of many iron-metabolism disorders, including TM, TI, and other IL-NTDT. In each case, a risk–benefit assessment is required for the selected therapies, based on individual variations and the long-term efficacy, safety, and cost of the chelating drug.

The pharmacological, toxicological, biochemical, chemical, and iron-binding properties of different chelators and the chelating drugs DFO, L₁, and DFX, including their effects on iron absorption and excretion, have been previously reviewed (Table 3).^{14,26,39} L₁ is a bidentate chelator forming a 3L₁–1Fe complex, DFX is a tridentate chelator forming a 2DFX–1Fe complex, and DFO is a hexadentate chelator forming a 1DFO–1Fe complex at physiological pH.

The mode of action and effects of chelators are variable in relation to different metabolic pathways, which is a reflection of the differences in their physicochemical and other properties (Table 3). This variability has been shown in many in vitro, in vivo, and clinical studies. For example, the lipid–water partition coefficient of the chelators and their iron complexes has been shown to affect the transfer of iron across the cell membrane in different cell types, as well as to cause variable effects related to the intracellular iron-metabolism pathways in these cells.^{40–42}

In studies on the effect of chelators on iron absorption, it was observed that the long-term oral administration of lipophilic chelators, such as 8-hydroxyquinoline, appears to cause iron overload in rats, by gradually increasing the absorption

Table 3 Mode of action and property differences of the chelating drugs deferoxamine, deferiprone, and deferasirox and other chelators***Clinical and biological effects**

Recommended doses of the chelating drugs in non-transfusion-dependent thalassemias

DFO subcutaneously or intravenously 10–60 mg/kg/day. Oral L_1 10–100 mg/kg/day. Oral DFX 10–40 mg/kg/day. Combination of chelating drugs at different doses

Iron-loaded patient compliance with chelating drugs

Low compliance with DFO in comparison to oral L_1 and DFX

Effect of chelating drugs on iron absorption

Increase of iron absorption by the lipophilic chelators maltol, 8-hydroxyquinoline, and DFX. Decrease of iron absorption by the hydrophilic chelators DFO, EDTA, DTPA, and L_1

Iron removal from diferric transferrin in iron-loaded patients

Removal of approximately 40% of iron at L_1 concentrations >0.1 mM, but no removal of iron by DFO or DFX at even 4 mM concentrations

Differential iron removal from various organs of iron-loaded patients

Efficacy is related to dose for all chelators. L_1 preferential iron removal from the heart and DFX from the liver. DFO iron removal from the liver and/or heart (especially following intravenous administration)

Iron redistribution in diseases of iron metabolism by chelating drugs

L_1 and to a lesser extent DFO can cause iron redistribution from iron deposits and also through transferrin (only in the case of L_1), eg, from the reticuloendothelial system to the erythron in the anemia of chronic disease. DFX may cause redistribution of iron from the liver to other organs in iron-loaded patients.

Enterohepatic circulation by DFX and DFX metabolites

Increase excretion of metals other than iron, eg, Zn and Al

DTPA $> L_1 > DFO$ (order of increased Zn excretion in iron-loaded patients)

DFO and L_1 cause increased Al excretion in renal dialysis patients

DFX causes increases in Al and other toxic metal absorption

Iron mobilization and excretion of chelator metabolite iron complexes

Several DFO metabolites have iron-chelation potential and increase iron excretion, but not the L_1 glucuronide or the DFX glucuronide metabolites

Combination chelation therapy

L_1 , DFO, and DFX combinations are more effective in iron excretion than monotherapy. The ICOC L_1 and DFO combination protocol causes normalization of the iron stores in thalassemia major and IL-NTDT patients

Chelating drug synergism with reducing agents

Ascorbic acid acts synergistically with DFO but not with L_1 or DFX for increasing iron excretion

Metabolism and pharmacokinetics

Metabolite(s)

DFO: a number of metabolites, including some with chelating properties, which are cleared mainly through the urine. L_1 : the L_1 -glucuronide conjugate is cleared through the urine, but has no iron-chelation properties. DFX: the DFX-glucuronide conjugates are cleared through the feces and have no iron-chelation properties

$t_{1/2}$ absorption following oral administration

L_1 : 0.7–32 minutes. DFX: 1–2 hours

T_{max} of the chelator

L_1 : mostly within 1 hour. DFX: mostly 4–6 hours

$t_{1/2}$ elimination of chelator

Intravenous DFO: 5–10 minutes. Oral L_1 : 47–134 minutes at 35–71 mg/kg. Oral DFX: 19±6.5 hours at 20 and 40 mg/kg

(Continued)

Table 3 (Continued)

$t_{1/2}$ elimination of the iron complex

DFO: 90 minutes. L_1 : estimated within 47–134 minutes. DFX: 17.2±7.8 hours at 20 mg/kg and 17.7±5.1 hours at 40 mg/kg

T_{max} of the iron complex

L_1 : estimated within 1 hour. DFX: 1–6 hours at 20 mg/kg and 4–8 hours at 40 mg/kg

T_{max} of the metabolite

L_1 -glucuronide 1–3 hours

Route of elimination of chelator and its iron complex

DFO: urine and feces. L_1 : urine. DFX: almost exclusively in feces, and less than 0.1%–8% in urine

Enterohepatic circulation

DFX and metabolites, but not DFO or L_1

Chemical and physicochemical properties

Molecular weight of chelators (g/mole)

DFO: 561. L_1 : 139. DFX: 373

Molecular weight of iron complexes (g/mole)

DFO: 617. L_1 : 470. DFX: 798

Charge of chelators at pH 7.4

DFO: positive. L_1 : neutral. DFX: negative

Charge of iron complexes at pH 7.4

DFO: positive. L_1 : neutral. DFX: negative

Partition coefficient of chelators (n-octanol/water)

DF: 0.02. L_1 : 0.19. DFX: 6.3

\log_{10} of chelator iron complexes

DFO: 31. L_1 : 35. DFX: 27

Notes: *The clinical, biological, toxicological, and pharmacological effects of iron-chelating drugs and other chelators depend on their chemical and physicochemical properties, as well as the properties of their iron and other metal complexes, and also on their metabolites. There is wide variation in the mode of action of the three chelating drugs DFO, L_1 , and DFX, including their efficacy in iron removal from different organs and also in the toxic side effects in iron-loaded patients, including IL-NTDT patients.

Abbreviations: DFO, deferoxamine; L_1 , deferiprone; DFX, deferasirox; EDTA, ethylenediaminetetraacetic acid; DTPA, diethylenetriaminepentaacetic acid; ICOC, International Committee on Chelation; IL-NTDT, iron-loaded non-transfusion-dependent thalassemia.

of iron present in the diet.⁴³ An increase in iron absorption in mice has also been shown by other lipophilic chelators, such as maltol and 2-hydroxy-4-methoxypyridine-1-oxide.³³ The maltol-iron complex is currently undergoing clinical development for general use in iron deficiency and especially in patients with inflammatory bowel disease.^{44–46} The same effect is anticipated by the chronic administration of other lipophilic oral iron chelators and especially the chelating drug DFX.^{47,48} DFX is also anticipated to increase aluminum and other toxic metal absorption (Figure 2, Table 3).^{47,48}

In contrast to maltol, DFX, and other lipophilic chelators, many hydrophilic chelators, including DFO and L_1 , appear to cause a decrease in iron absorption under similar conditions.³³ Because of these differences in properties, DFO and L_1 could also be used in the treatment of acute oral iron poisoning.^{49,50} In contrast, the use of DFX in the treatment of acute iron poisoning is inappropriate and not recommended, because it may increase the absorption of iron and increase further iron toxicity in this condition.^{47,48}

Inhibition of iron absorption has also been observed by such chelators as 2-hydroxypyridine-1-oxide, which form

complexes with iron that precipitate and become insoluble.³³ Increase or inhibition of iron absorption can also be observed with commonly used drugs with chelating properties, such as tetracycline and hydroxyurea.^{51,52} It is anticipated that similar chelators could be used to decrease the rate of gastrointestinal iron absorption in TI, other IL-NTDT, hereditary hemochromatosis, and TM.

Clinical use and efficacy of deferoxamine, deferiprone, and deferasirox in iron overload

The primary use of DFO, L_1 , and DFX is the treatment of transfusional iron overload in TM, other transfusional iron-loaded conditions, and IL-NTDT. Iron-overload toxicity from chronic transfusions and/or increased gastrointestinal iron absorption leads to progressive multiorgan damage and associated increase in morbidity and mortality, which in most cases is directly related to the level of excess iron load. For example, in the absence of chelation therapy, regularly transfused TM patients die by the age of 20 years, mainly from congestive cardiac failure caused by cardiac iron-overload toxicity.^{53,54} In contrast, the mean survival of TM patients in the UK treated with DFO was estimated to be 35 years.⁵⁴ However, the use of specific chelation-therapy protocols with L_1 and the L_1 -DFO combination appear to increase life expectancy and reduce the morbidity of iron-loaded patients to levels approaching those of normal individuals.^{24,55}

DFO has been used parenterally for the treatment of iron-overloading conditions, including TI, other IL-NTDT, and especially TM in the last 50 years. Similarly, oral L_1 , which is a generic drug like DFO, has been approved and used in India since 1995 and in Europe and other regions since 1999, but in the US since only recently – in 2011. DFX is a relatively new oral but very expensive patented drug that has been used worldwide since 2005, and the patent expires in 2017.

The three chelating drugs have major differences in the efficacy, tolerance, site of action, toxicity profile, and cost, all of which affect the morbidity and mortality of iron-overloaded patients in both developed and developing countries (Table 3).^{14,24,25} Because of the high cost of the transfusions and chelating drugs, most of the TM and IL-NTDT patients in developing countries are left to die untreated. The local manufacturing of cheaper formulations for L_1 and DFX in developing countries, such as India, Thailand, and Iran, has substantially reduced the cost of chelating drugs and increased the number of patients receiving chelation therapy in these countries.^{14,24,25,56,57}

There are general variations among patients not only related to the level of iron overload and organ distribution

but also in the response and pharmacology to each of the chelating drugs. These responses are also related to individual profile differences in the absorption, distribution, metabolism, excretion, and toxicity of the drugs.^{24,25,58,59} For example, L_1 causes an increase in iron excretion exclusively in the urine, DFX exclusively in the feces, and DFO mostly in the urine and also the feces (Table 3).^{14,24,25,59}

The general recommended doses of the chelating drugs in TM and IL-NTDT are 40–60 mg/kg/day for subcutaneous DF, 75–100 mg/kg/day for oral L_1 , and 20–40 mg/kg/day for oral DFX. However, variations in iron overload and targeted therapies can result in the application of a range of therapeutic protocols from an intensive chelation, eg, combinations of DF (40–60 mg/kg/day) and L_1 (75–100 mg/kg/day) in heavily transfused iron-loaded patients, to much lower doses or intermittent withdrawal of chelation in TM and IL-NTDT patients whose normal iron-store levels have been achieved and further chelation may cause iron deficiency and other complications.^{24,60}

In general, the selection of the chelation-therapy protocol in each iron-loaded patient depends on several other factors in addition to the iron load, such as drug efficacy, organ targeting, toxicity, cost, and compliance. However, there are many controversies in the selection and use of chelating drugs in TM and IL-NTDT that arise mainly from commercial influence, the lack of consensus in the ultimate goal or aim of the chelation therapy, and the selective use of each of the chelating drugs for optimal therapy.^{23,25} Furthermore, there is no consensus in the evaluation criteria and risk–benefit assessment for the use of each of the chelating drugs in personalized medicine.^{23,25} In most Western and other developed countries, the selection of the chelating drug for the treatment of TM and IL-NTDT depends on the choice of the physician and the commercial influence of pharmaceutical companies.^{24,25} In developing countries, where most patients live, the main issue is the cost of the chelating drug and its availability and not necessarily risk–benefit assessment issues. These issues have been recently highlighted within the broader context of the use of orphan drugs, which include DFO, L_1 , and DFX, and also of orphan and rare diseases, which include TM and IL-NTDT.^{24,25}

The monitoring of efficacy of different chelation-therapy protocols on the gross body as well as individual organ iron-load levels has been recently achieved following the introduction of new diagnostic techniques, such as MRI T_2 and T_2^* . These methods have increased our understanding of iron-metabolism and chelation pathways of iron removal and resulted in improved drug-targeting therapies of the gross body, as well as focal iron deposits in organs, which

are the cause of toxic side effects.^{61–64} Furthermore, these developments have increased the prospects of the introduction of personalized medicine in TM, IL-NTDT, and other iron-metabolism disorders.²³ The use of these new diagnostic methods in combination with serum ferritin monitoring made it possible to introduce chelation-therapy protocols of L_1 or the L_1 -DFO combination that can achieve the complete elimination of iron overload in the heart and liver of almost all categories of iron-loaded patients.^{65–67}

The clearance of excess iron in both organs can be achieved in most TM cases in 0.5–1.5 years using the International Committee on Chelation protocol on L_1 (75–100 mg/kg/day) and DFO (40–60 mg/kg/day at least 3 days per week). The clearance of excess iron from these organs is faster when using higher overall doses of L_1 and DFO and in patients who are less heavily iron-loaded.^{65–69} The removal of excess toxic iron in TM by L_1 and the L_1 -DFO combination is accompanied by improvement in organ function, such as elevation of left ventricular ejection fraction (LVEF) and other cardiac and endocrinological benefits.^{24,65,66,69} In contrast, excess iron removal from the heart by DFX is not effective and in most cases very slow, with no improvement in LVEF, especially in heavily iron-loaded TM patients.⁷⁰ It should be noted that even after 5 years of treatment with DFX (30–40 mg/kg/day), excess toxic amounts of iron remain in the heart of heavily iron-loaded TM patients and with no improvement in LVEF.^{70,71}

Excess iron removal from the liver by DFX appears to be effective usually at high doses and equivalent to that caused by DFO, L_1 , and their combination at effective doses.^{65–72} The most effective and rapid clearance of excess liver iron, which is also used for intensive chelation in heavily iron-loaded patients, is intravenous DFO in combination with L_1 . Following the clearance of excess iron from the liver and heart and the normalization of the iron stores, which is characterized by normal levels of MRI T_2 and T_2^* and serum ferritin, normal levels of iron are usually maintained using much lower overall chelation doses and in particular L_1 monotherapy.^{65–67}

Toxicity aspects and limitations in the use of chelating drugs in iron-loaded non-transfusion-dependent thalassemias and other patient categories

Patients with IL-NTDT and TM are regularly monitored for iron toxicity and other effects caused by their underlying condition.^{9–11,24,73} Clinical examination and biochemical

tests are regularly carried out for organ function, including liver-enzyme levels, echocardiography, serum ferritin, and serum Zn levels.^{9–11,16,17,24,73}

The task of treating iron overload in IL-NTDT is generally easier in comparison to TM, since IL-NTDT patients are not as heavily iron-loaded as TM patients and also the rate of iron loading in IL-NTDT in general is much slower than that in TM. However, the prospect of chelator toxicity is a major issue, and increases especially in patients with low iron stores.^{24,60} Regular monitoring and prophylactic measures for chelating drug toxicity are generally recommended for these categories of patients.^{24,74}

The treatment of patients with TM, IL-NTDT, and other categories of iron-loading conditions with normal iron stores or not heavily iron-loaded is restricted for DFX, due to toxicity implications, as underlined by the labeling information for its use by the manufacturers of the drug.^{75,76} In particular, the use of DFX is not recommended for iron-loaded patients with serum ferritin lower than 500 $\mu\text{g/L}$.^{75,76} This limitation is important for the safety of many patients and in particular the vast majority of IL-NTDT, who are not heavily iron-loaded and have serum ferritin lower than 500 $\mu\text{g/L}$.^{75–76}

High morbidity and mortality have been reported in different categories of patients, and especially non-iron-loaded categories of patients who have been treated with DFX.^{77,78} Many toxic side effects, such as fatal renal, liver, and bone marrow failure, including agranulocytosis, as well as renal toxicity, skin rashes, and gastric intolerance, have also been reported in iron-loaded patients treated with DFX.^{74–78}

Patients treated with DFX are regularly monitored for kidney function, and withdrawal of the drug is recommended for patients with persistent increase in serum creatinine levels. The same limitations and restrictions that apply in the use of DFX in patients with low iron stores also apply for the use of DFO, despite the incidence of serious toxic side effects being much lower in the latter. Within this context, the use of DFO in IL-NTDT and other similar categories of not heavily iron-loaded patients, as well as other categories of patients with normal iron stores, is restricted, due to toxicity implications. Fatal cases of mucormycosis, acute respiratory distress syndrome, and *Yersinia enterocolitica* was reported in different categories of patients treated with DFO.^{39,79} Furthermore, ocular and auditory toxicity has also been reported in non-heavily iron-loaded TM patients using DFO.^{39,80,81}

In contrast, the safety of L_1 in TM, IL-NTDT, and other categories of patients with low or normal iron stores has been shown during the regular use of the drug in thousands of patients in the last 20 years and also in studies in TM patients

with normal iron stores exceeding 100 patient-years.^{24,65,67,74} The prospect of wider clinical use of L_1 as a universal antioxidant in non-iron-overloading diseases, such as neurodegenerative, cardiovascular, renal, and infectious diseases, as well as other diseases, including cancer and aging diseases, has been investigated in clinical trials involving many of these categories of patients.^{24,65,67,74} In particular, the introduction of L_1 for the treatment of non-iron-loaded patients by targeting focal toxic iron deposits, eg, in Friedreich's ataxia, and also of toxic labile iron, eg, in diabetic and nondiabetic glomerular disease, is a reflection of the antioxidant and safety potential of this drug.^{24,82,83} In these two non-iron-loaded conditions, L_1 was used at doses of 50–75 mg/kg/day for 6–9 months with no serious toxic side effects.^{82,83} Similarly, the safety of L_1 in many other categories of non-iron-loaded diseases has also been confirmed in clinical trials involving patients with normal iron stores, including the anemia of chronic disease, renal dialysis, infections, Parkinson's, and other neurodegenerative diseases.^{24,65,67,74}

The most serious toxic side effects of L_1 are reversible agranulocytosis (<1%) and neutropenia (<5%).^{24,84} Weekly or fortnightly mandatory blood count is recommended for patients using L_1 for prophylaxis against this toxicity. A few fatal cases of agranulocytosis in TM have been reported for patients treated with L_1 who did not follow this safety procedure.^{24,39,84} Less serious toxic side effects include gastric intolerance, joint pains, and Zn deficiency. Zn supplements are used for prophylaxis for patients on long-term treatment with L_1 and DFO, including TM and IL-NTDT patients.^{24,84}

It should be noted that the rate of fatalities associated with the use of iron-chelating drugs is different. An estimated 11.7% mortality (1,935 of 16,514 patients) was reported by the European Medicines Agency for DFX, whereas for DFO and L_1 , the rate of mortality is less than 0.1%.^{77,78} Toxicity-vigilance and prophylactic measures are important steps for ensuring the safety of TM, IL-NTDT, and other iron-loaded patients treated with chelation therapy, especially when using high doses in patients with low or normal iron stores.

Effective iron-chelation therapy in thalassemia intermedia and in other iron-loaded non-transfusion-dependent thalassemias

Iron overload and related toxicity as a result of increased gastrointestinal absorption in TI and other IL-NTDT conditions was identified more than 40 years ago.^{12,13}

Diagnostic methods, such as serum ferritin, serum iron, increased urinary excretion in response to DFO, and post-mortem examination, revealed the extent of the iron load and toxicity in these categories of patients.^{12–18,85}

Iron-chelation therapy has been widely and successfully used from the early 1980s for the treatment of iron overload in TI, other IL-NTDT, and other similar conditions.⁸⁶ Initial reports indicated the efficacy of DFO in increasing iron excretion in these categories of patients. For example, increased urinary iron excretion in response to continuous 12-hour subcutaneous DFO (20–100 mg/kg/day, maximum 2 g/day) and decrease in serum ferritin (initial 68–940 μ g/L and final 48–420 μ g/L) after 6 months of treatment was observed in ten TI patients (age 1–17 years, five males, two splenectomized) with different Hb levels (7–10.5 g/dL) and variable iron load.⁸⁶ Four patients participating in the study had previous transfusions to a total in each case of 0.5, 1.5, 2.0, and 10.0 red blood-cell units, respectively. Dose adjustment was carried out by assessing 24-hour urinary iron excretion in response to the different doses of DFO, which were administered over 3 days. Different levels of iron load were detected in this cohort of young TI patients, including a 5-year-old patient. A tailored chelation therapy was recommended using this DFO-response method for reducing chelator toxicity. It was also suggested that iron-chelation treatment should begin from an early age to prevent iron accumulation.⁸⁶

Many nontransfused iron-loaded patients have also been treated with DF. In one case, subcutaneous DFO (2 g/day for 9–11 months) was shown to be effective in hereditary hemochromatosis patients where phlebotomy was not possible.⁸⁷ Monitoring of liver iron by repeated noninvasive liver iron-concentration measurements using superconducting quantum interference-device biosusceptometry has shown that DFO was as effective in liver-iron elimination (12 mg/day) as normal phlebotomy treatment (14.3 mg/day) of weekly 500 mL blood removals.⁸⁷

Successful iron-chelation treatments in TI and other IL-NTDT patients have also been reported using L_1 . For example, L_1 at low doses of 25–50 mg/kg/day was studied for 17 weeks and 86 weeks in seven HbE β -thalassemia and two TM iron-loaded nontransfused patients.⁸⁸ Substantial reductions in serum ferritin (from mean 2,168 to 418 μ g/L) and hepatic iron (from mean 20.3 to 11.7 mg/g dry weight) were observed in these patients.⁸⁸ Similar results were obtained in another study in 16 NTDT HbH disease (3/4 α -globin gene deletion) patients (29–76 years) using L_1 at 50 mg/kg/day initially for 4–8 months. The dose then increased in the majority of cases to 75 mg/kg/day.⁸⁹ In 18 months, mean serum ferritin

decreased from 1,492 to 519 $\mu\text{g/L}$, and in nine of the patients was below the normal cutoff point of 397 $\mu\text{g/L}$ before the 18-month monitoring period. Significant improvement in cardiac function and decrease in liver-iron content measured by MRI T_2 was also detected in 12 of the patients.⁸⁹ No reduction in iron load was observed in a control group of 16 patients with HbH disease who received no chelation treatment in the same study.⁸⁹

The use of higher doses of L_1 (75–100 mg/kg/day) in IL-NTDT can result in the rapid normalization of iron stores. For example, administration of L_1 at 75 mg/kg/day for 9 months in an iron-loaded, nontransfused TI patient with cardiac and other problems caused a substantial reduction to normal levels in serum ferritin (from 2,174 to 251 $\mu\text{g/L}$) and hepatic iron (from 14.6 to 1.9 mg/g dry weight).⁹⁰ An increase in cardiac and liver MRI T_2 relaxation time was also observed following the 9-month treatment, indicating the effective removal of iron from these organs by L_1 .⁹⁰

In a 5-year long-term randomized clinical trial of 88 occasionally transfused TI patients (median age 41 years, 48 females), the effectiveness of oral L_1 (50–75 mg/kg/day) on 47 patients with median serum ferritin 1,221 $\mu\text{g/L}$ was compared to a group of 41 patients with median serum ferritin 1,122 $\mu\text{g/L}$ treated with subcutaneous DFO (30–50 mg/kg/day, 5 days per week).⁹¹ Both chelators were found to be effective in reducing iron load, with median serum ferritin levels reaching 579 $\mu\text{g/L}$ in the L_1 group and 484 $\mu\text{g/L}$ in the DFO group. It was concluded in this >200 patient-year-long study that L_1 was associated with an efficacy and safety profile similar to that of DFO.⁹¹ Toxic side effects of L_1 were gastric intolerance, joint pains, neutropenia in six, and agranulocytosis in four patients. Compliance was 85% for patients on L_1 and 76% for DFO. The cost of chelation between L_1 and DF was not significantly different, and was estimated to be about ten times lower than that of DFX.^{56,57,91} A follow-up 5-year trial using L_1 and DFO continued with members of the same group of TI patients, with similar results.⁹¹

Rapid clearance of excess iron in iron-loaded nontransfused patients has also been observed using the L_1 –DFO combination, which appears to be easily achieved and also to be faster in comparison to TM patients.^{65–69} For example, a combination of intravenous DFO (30 mg/kg/day for 5 months) and oral L_1 (75 mg/kg/day for 2 months) was used in a juvenile hemochromatosis nontransfused iron-loaded patient (serum ferritin 3,773 $\mu\text{g/L}$) with severe cardiomegaly and congestive heart failure with an ejection fraction of 25%.⁹² The combination L_1 –DFO chelation treatment caused an improvement in the ejection fraction to 49% and blood

parameters, and also removal of antiarrhythmic drugs and normalization of serum ferritin.⁹² Increase in the ejection fraction has also been observed in different groups of TM patients using L_1 and the L_1 –DFO combination.^{24,65–69} Overall, effective and safe chelation therapy using DFO, L_1 , and their combination could be applied to TI and other IL-NTDT patients.

A number of clinical trials have been carried out to test the effect of DFX in TI and other IL-NTDT patients.²² The efficacy and toxicity of DFX appear to be dose-dependent. For example, in a randomized 1-year clinical trial involving 110 TI patients, some NTDT and others receiving occasional transfusions, DFX (5–10 mg/kg/day) caused no significant change in serum ferritin at 24 weeks and only significant change at 52 weeks at 10 mg/kg/day compared to baseline.⁹³ Similar changes were reported for liver-iron concentration. Adverse events were reported in 78% of the patients, and only 24% were thought to be drug-related, namely nausea, rash, and diarrhea.⁹³ In a continuation study for a further year, the results of the use of DFX (5–20 mg/kg/day) on efficacy and toxicity in a selected group of six not previously chelated of the placebo group and 18 of the 110 chelated patients (12 TI, six α -thalassemia, and six HbE β -thalassemia) were reported.⁹³ Median serum ferritin before the study was 825 (range: 393–2,169) $\mu\text{g/L}$, and by the end of the study had decreased to 427 (range: 102–1,010) $\mu\text{g/L}$. Similar changes were reported for liver-iron concentration.⁹³ Approximately 29% of the adverse events reported were similar to the previous study and thought to be drug-related.⁹³ In a different 1-year study involving a cohort of 50 TI patients, DFX (10–40 mg/kg/day) caused a decrease of serum ferritin from baseline mean 1,700 to 835 $\mu\text{g/L}$ in 8 months, but thereafter remained stable or slightly increased to 880 $\mu\text{g/L}$ on the 12th month.⁹⁴

In contrast to these studies, lower efficacy and higher toxicity was reported for DFX in TI and other IL-NTDT patients by investigators not related or funded by the manufacturers of DFX.⁹⁵ For example, in a retrospective study of 72 TI and TM patients of mean age 20 years, the use of DFX (20–40 mg/kg/day) for a mean 16.7 months resulted in a 25% dropout due to toxicity and low efficacy.⁹⁵ In particular, adverse effects of renal function were reported, including a >33% increase in serum creatinine in 43 patients. Persistent increase in serum creatinine in seven of the 43 patients caused discontinuation of DFX.⁹⁵ No significant change in serum ferritin was observed at the end of the study (initial $3,059 \pm 309$ versus final $2,965 \pm 309$ $\mu\text{g/L}$). Similarly, no change in cardiac iron load but a decrease in liver iron was observed in this

cohort of patients. The patients affected by the DFX toxicity reverted back to L_1 , DFO, or the L_1 -DFO combination.⁹⁵ High incidence of renal toxicity using DFX has also been reported in many other groups of patients, including young patients, by independent investigators not sponsored or without connections with the manufacturers of DFX.^{96–98}

The variation in the iron overload in patients of IL-NTDT and the restriction imposed on the safety for the use of the chelating drugs in different iron-overload levels, as well as the target iron-loaded organ(s), requires the design of specific chelation protocols for each patient case. Similar chelation protocols apply in the treatment of TM patients with equivalent iron overload to IL-NTDT patients. It appears that the selection of the most effective and safest chelation protocol together with the monitoring of the iron load using serum ferritin and MRI T_2 or T_2^* , as well as the use of prophylactic measures for chelating drug toxicity, could ensure the effective reduction and maintenance of iron load to normal levels and lead to an overall decrease in morbidity and mortality in IL-NTDT patients.

Future therapeutic approaches of iron-chelation therapy in iron-loaded nontransfused thalassemia patients caused by increased iron absorption

Patients with TI and other IL-NTDT have a diverse genotypic and clinical picture of variable severity, including different levels of iron overload and associated toxicity in the liver, heart, and other organs.^{10,11,17,19,27,32,73} Within this context, several approaches in relation to chelation therapy could be envisaged. The design and application of optimal chelation therapies will be required in each case based on a risk–benefit assessment of the drug(s) and other factors similar to TM patients.^{23,24} In particular, the cost of chelating drugs is very important, since most TI and other IL-NTDT patients live in developing countries and cannot afford expensive drug treatments.^{56,57,99,100}

DFO and L_1 appear to be effective in the removal of excess iron load in TI and other IL-NTDT patients, even at doses lower than those used in TM and other transfused patients. Selected combinations of DFO and L_1 , such as the International Committee on Chelation protocol, could be used as a first-line treatment not only for the rapid removal of excess cardiac iron in cases of iron-overload cardiomyopathy in TI and other IL-NTDT patients but also in the rapid clearance of liver-iron overload.^{23,24,65,66,86}

A critical question in relation to the treatment of iron overload in TI and other IL-NTDT is the timing of the initiation of the chelation therapy. Recent studies in TM patients suggest that the removal of iron by chelating drugs in TI and other IL-NTDT should be initiated as soon as the diagnosis is confirmed and with a major aim the complete removal of all excess iron and prevention of its accumulation in organs.^{23,24,86} In studies of TM patients with normal-range iron-store levels, lower overall doses of chelating drugs and in particular L_1 were used for achieving this aim.^{65,67} Furthermore, the advantage of this approach is that in patients with normal-range iron stores, the damage to organs is prevented and clinical management can continue without related organ complications, such as liver cirrhosis and cardiomyopathy.^{23,24}

L_1 should be considered as a first-line treatment for TI and other IL-NTDT patients with low or normal iron stores, since it would appear from studies in these categories that included cohorts of TM patients to be safer and more effective in comparison to DFO and DFX.^{74–77} The latter two chelating drugs are recommended by the manufacturers to be used in patients with serum ferritin levels higher than 500 $\mu\text{g/L}$.^{74–77} Similarly, the prophylactic use of low doses of L_1 (eg, 10 mg/kg/day) may be sufficient in preventing iron overload in young and older patients with TI and other IL-NTDT.

Many safety, efficacy, and financial questions are raised regarding the use of DFX in TI and IL-NTDT patients. Additional toxicity issues can be envisaged in patients with low iron stores, where DFX may cause an increase in the absorption of dietary iron and other toxic metals, eg, aluminum.^{23,48} DFX is also very expensive and cannot be afforded by the vast majority of patients, especially those living in developing countries.^{3,23,56,57,99,100} The introduction of combinations of DFX with DFO and L_1 increases the prospect of more effective and less toxic therapies for TM, TI, IL-NTDT, and other related conditions in comparison to DFX monotherapies.^{100–105} Pharmacological and toxicological data are important parameters for the construction of such combination therapies.^{101,106,107}

Although current interest in TI and other IL-NTDT patients is mainly related to the marketing efforts of the manufacturers of DFX and their academic partners, there are many other fields of investigation that need to be explored in addition to iron-chelation therapy. Some of the strategies for the treatment of the other clinical complications may also influence the rate of iron overload and chelation therapy. Within this context, the time of initiation and the rate of red blood-cell transfusions, as well as splenectomy, will have

a direct effect on iron loading and change in the chelation strategies.^{85,105} Similarly, the role of iron overload and chelation therapy on the activation of HbF production by hydroxyurea, which also has iron-chelation properties, has not yet been fully investigated.^{11,28,34–37,52,108} The effects of modulators of erythropoietin, hepcidin, ferroportin, and other regulatory molecules on iron metabolism and erythropoietic activity are also expected to influence future chelation-therapy strategies in TI and other IL-NTDT.^{38,109,110}

A major therapeutic strategy in TI, other IL-NTDT, and especially hereditary hemochromatosis is the prevention or decrease of gastrointestinal iron absorption. This can be accomplished by new, synthetic, or naturally occurring iron chelators that can bind iron in the gastrointestinal tract and inhibit or decrease the rate of its transfer and storage in the body. In particular, chelators and chelator–iron complexes that are not orally absorbed and chelators that form insoluble iron complexes could be considered for such treatments.^{33,111} Inhibitors of DMT1 activity and activators of hepcidin production could also be used for minimizing iron absorption. The combination of chelators and of chelators with regulators of DMT1, hepcidin, and ferroportin could also be considered for potential use in reducing increased iron absorption and associated iron overload in IL-NTDT.^{33,109–111}

The inhibition of iron absorption in IL-NTDT could be accompanied by relocation and reutilization of stored iron for erythropoiesis and other iron-metabolism activities. Within this context, L_1 has been shown to facilitate this process by the redistribution of iron stored in hepatocytes and macrophages to hemopoietic tissues via iron exchange with transferrin.^{112–119}

Overall, several treatments could become available for the future management of TI, other IL-NTDT, and hereditary hemochromatosis. These could involve the use of chelation therapies for increasing iron excretion, chelation therapies for inhibiting iron absorption, chelation therapies for iron redistribution, selected diets with lower and less absorbable forms of iron, the use of regulators of DMT1, hepcidin, and ferroportin and other proteins of iron metabolism involved in the transport of iron, activators of HbF production, and combinations of such treatments.

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Disclosure

The authors report no conflicts of interest in this work.

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