

NANO EXPRESS

Open Access

Efficiency enhancements in Ag nanoparticles-SiO₂-TiO₂ sandwiched structure via plasmonic effect-enhanced light capturing

Jinxia Xu^{1,4}, Xiangheng Xiao^{1,2*}, Andrey L Stepanov³, Fen Ren^{1,2}, Wei Wu¹, Guangxu Cai¹, Shaofeng Zhang¹, Zhigao Dai¹, Fei Mei⁴ and Changzhong Jiang^{1,2*}

Abstract

TiO₂-SiO₂-Ag composites are fabricated by depositing TiO₂ films on silica substrates embedded with Ag nanoparticles. Enhancement of light absorption of the nanostructural composites is observed. The light absorption enhancement of the synthesized structure in comparison to TiO₂ originated from the near-field enhancement caused by the plasmonic effect of Ag nanoparticles, which can be demonstrated by the optical absorption spectra, Raman scattering investigation, and the increase of the photocatalytic activity. The embedded Ag nanoparticles are formed by ion implantation, which effectively prevents Ag to be oxidized through direct contact with TiO₂. The suggested incorporation of plasmonic nanostructures shows a great potential application in a highly efficient photocatalyst and ultra-thin solar cell.

Keywords: Plasmonic effect, Ion implantation, Ag nanoparticles, Photocatalysis

Background

Titanium dioxide (TiO₂) has strong photocatalytic activity, high chemical stability, a long lifetime of photon-generated carriers, nontoxicity, and low cost, which make it one of the most widely used photocatalysts for hydrogen production and solar cells, as well as water and air remediation [1-3]. At modern times, TiO₂ becomes a hot research topic because of the potential applications in the field of environment and energy [4-6]. Unfortunately, owing to its wide band gap of 3.2 eV (at 390 nm), only approximately 4% solar spectrum can be utilized. During the last decades, great efforts have been made to modify the TiO₂ to enhance the visible light response. A considerable increase in the photocatalytic activity in the visible region has been observed by doping [7-10]. However, to date, the doping structure lacks reliable controllability. Recently, metallic nanostructures have been introduced into a semiconductor

film (e.g., ZnO, InGaN quantum wells) for enhancement of light emission, photocurrent solar cells [11-14], and photocatalysts [15-17] by a strong plasmonic effect of metallic nanostructures. In order to maximize the utilization rate of the UV region of the sunlight, in this letter, we design a new composite structure to enhance the light absorption efficiency by coupling TiO₂ to Ag nanoparticles (NPs) embedded in SiO₂ formed by low-energy Ag ion implantation. Ag NPs show a very intense localized surface plasmon resonance (SPR) in the near-UV region [18], which strongly enhances the electric field in the vicinity of the Ag NPs. This enhanced electric field at the near-UV region could increase the UV light absorption to boost the excitation of electron-hole pairs in TiO₂ and thus increase the photoelectric conversion efficiency. In this kind of structure, the Ag NPs embedded in SiO₂ serve two purposes. Firstly, SiO₂ as a protective layer prevents Ag to be oxidized through direct contact with TiO₂. Secondly, the size and depth distributions of the embedded Ag NPs can be controlled by choosing implantation parameters and post-implantation thermal treatment [19], which can tune the SPR spectrum of Ag NPs to match the absorption edge of TiO₂. Thus, it is possible to design nanostructures that

* Correspondence: xxh@whu.edu.cn; czjiang@whu.edu.cn

¹Department of Physics and Key Laboratory of Artificial Micro- and Nano-structures of Ministry of Education, Wuhan University, Wuhan 430072, People's Republic of China

²Center for Electron Microscopy and Hubei Nuclear Solid Physics Key Laboratory, Wuhan University, Wuhan 430072, People's Republic of China
Full list of author information is available at the end of the article

concentrate the light surrounding near Ag NPs, which enhance the light absorption of the TiO₂ film.

Methods

High-purity silica slides were implanted by Ag ions at 20, 40, and 60 kV to a fluence of 5×10^{16} ions/cm² and at 40 kV to 1×10^{17} ions/cm² using a metal vapor vacuum arc ion source implanter, respectively. The TiO₂-SiO₂-Ag nanostructural composites were obtained by depositing TiO₂ films (100 nm thick) on the surface of the as-implanted silica substrates using a direct-current reactive magnetron sputtering system. For comparison, an un-implanted silica substrate was deposited with the TiO₂ film under the same growth condition. Subsequently, all deposited samples were annealed at 500°C in oxygen gas for 2 h to obtain an anatase-phase TiO₂ film. The TiO₂-covered silica substrates with embedded Ag NPs are named S1 to S4 as shown in Table 1. The optical absorption spectra of all the samples were measured using a UV-vis-NIR dual-beam spectrometer (Shimadzu UV 2550, Shimadzu Corporation, Kyoto, Japan) with wavelengths varying from 200 to 800 nm. Raman scattering spectra of all the samples were collected using a micro-Raman system (LabRAM HR800, HORIBA Jobin Yvon Inc., Edison, NJ, USA). An Ar laser (488.0 nm) was used as the excitation source, and the laser power was kept at 10 mW. The microstructure of the samples was investigated using JEOL JEM 2010 (HT) transmission electron microscopes (TEM; JEOL Ltd., Tokyo, Japan) operated at 200 kV.

The photocatalytic efficiencies of TiO₂ and TiO₂-SiO₂-Ag nanostructural composites with an area of 4 cm² were evaluated by measuring the degradation rates of 5 mg/L methylene blue (MB) solution under UV-vis irradiation. A mercury lamp (Osram 250 W (Osram GmbH, Munich, Germany) with a characteristic wavelength at 365 nm) was used as a light source. The TiO₂ and the TiO₂-SiO₂-Ag composite films were placed in 40 mL of MB solution with a concentration of 5 mg/L. Before irradiation, the samples were put in 40 mL of MB solution for 30 min in the darkness to reach absorption equilibrium. The decolorization of the MB solution was measured by an UV-vis spectrometer (Shimadzu UV 2550, Shimadzu Corporation) at the wavelength of 664.0 nm. The absorption spectrum of the MB solution was

Table 1 Ag ion implantation parameters for all samples

Sample	Fluence of ion implantation (ions/cm ²)	Energy of ion implantation (kV)
S1	5×10^{16}	20
S2	5×10^{16}	40
S3	1×10^{17}	40
S4	5×10^{16}	60

measured at a time interval of 30 min, and the total irradiation time was 4 h.

Results and discussion

Figure 1 shows the optical absorption spectra of S1 to S4 and the TiO₂ films. The absorption edge around 390 nm belongs to the intrinsic exciton absorption of TiO₂ [20]. The obvious absorption peaks at about 419 to 433 nm can be attributed to the SPR of Ag NPs formed by Ag ion implantation [21]. As seen, the SPR of Ag NPs is close to the exciton edge (around 390 nm) of anatase TiO₂. Therefore, it is expected that an efficient energy transfer from the Ag NPs to TiO₂ can occur. The position of the Ag SPR absorption peak of S2 is around 419 nm, which is a blue shift compared to that of the other three samples. The SPR peak of S2 is closest to the anatase TiO₂ exciton energy; therefore, the strongest resonant coupling effect between Ag SPR and the excitons of the TiO₂ films may be produced more effectively.

To illustrate the strong near field induced by the SPR of Ag NPs, the Raman scattering spectra of S1 to S4 and TiO₂ are measured as presented in Figure 2. The observed Raman bands at 144, 199, 399, 516, and 640 cm⁻¹ can be assigned to the Eg, Eg, B1g, A1g, or B1g and Eg vibration modes of anatase phase, respectively, which are consistent with the characteristic patterns of pure anatase without any trace of a rutile or brookite phase [22]. It is found that the Raman intensity for S1 to S4 increases compared to that of TiO₂, and S2 shows the strongest Raman intensity. It is well known that Raman scattering intensity is proportional to the square of the electric field intensity [23]; thus, stronger Raman scattering attained from the TiO₂-SiO₂-Ag structure indicates that a stronger electric field is induced by Ag NPs embedded in the SiO₂ substrate.

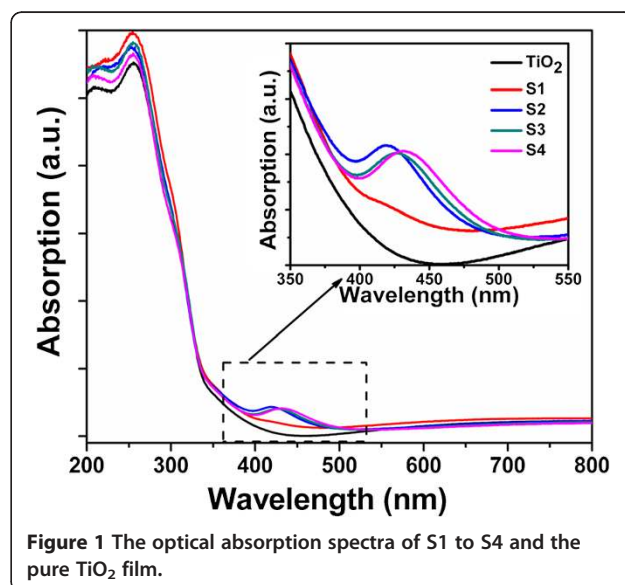
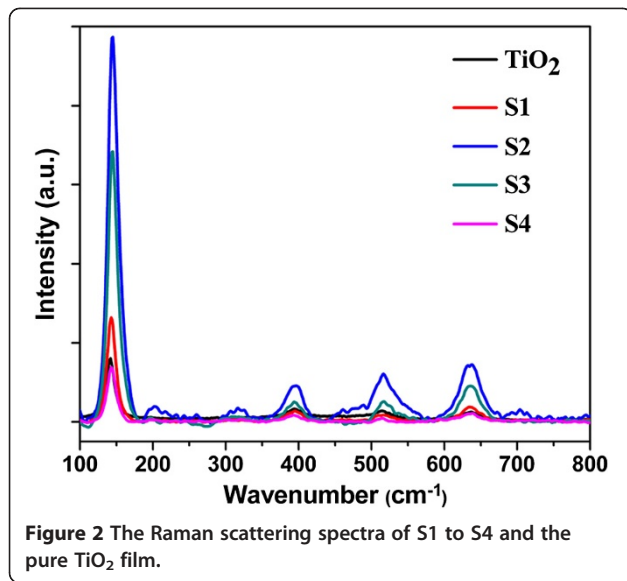


Figure 1 The optical absorption spectra of S1 to S4 and the pure TiO₂ film.



When the Ag NPs are irradiated by a laser in the spectral area of the particle absorption band's longer wavelength shoulder, a strong near field is produced due to the SPR, so Raman scattering is enhanced. As seen from Figure 2, the enhancement factors of Raman scattering of S1 to S4 are different because of various coupling field efficiencies. Thus, it is possible to conclude that the implantation energy and fluence have determined the Raman scattering enhancement factor.

To understand the relationship between the size and depth distributions of the Ag NPs in silica glass and the Raman scattering enhancement factor of the TiO₂-SiO₂-Ag nanocomposites, the microstructural characterization

of S1 to S4 was investigated by TEM as shown in Figure 3. The TEM image of S1 (Figure 3a) shows that the size of the Ag NPs appears to have a wide distribution. However, increasing the implantation energy to 40 kV as shown in Figure 3b, the Ag NPs in S2 are quite uniform in size (with a size of 20 nm) and distribute at nearly the same depth of 7 nm from the surface. Under high energy ion implantation, more heat will be induced in the sample in a short time, which enhances the diffusion of Ag atoms. Therefore, the implanted Ag ions trend to aggregate to larger NPs around the projected range [24-26]. The near field induced by the SPR of the Ag NPs is very strong due to the presence of the formed Ag NPs with bigger size and the near-field dipolar interactions between adjacent particles [27]. On the other hand, the dipolar interactions between adjacent particles with nearly the same size can result in a blue shift of SPR [28]; thus, the blue shift in the SPR peak of the Ag NPs is observed in Figure 1, which may produce a strongest resonant coupling effect between the SPR of Ag NPs and TiO₂. It means that the stronger near field can be induced. In this case, S2 has the strongest Raman scattering enhancement factor. The size of the Ag NPs in S1 is smaller, and the distribution is wider than that in S2. It means that the near field induced by SPR of the Ag NPs in S1 is weaker than that in S2. Further increasing the implantation energy to 60 kV as presented in Figure 3d, the Ag NPs in S4 reside deeper below the surface than those in S2. Since the SP is an evanescent wave that exponentially decays with distance from the metal particles to the surface [29], the enhancement of Raman scattering decreases progressively with the increase of

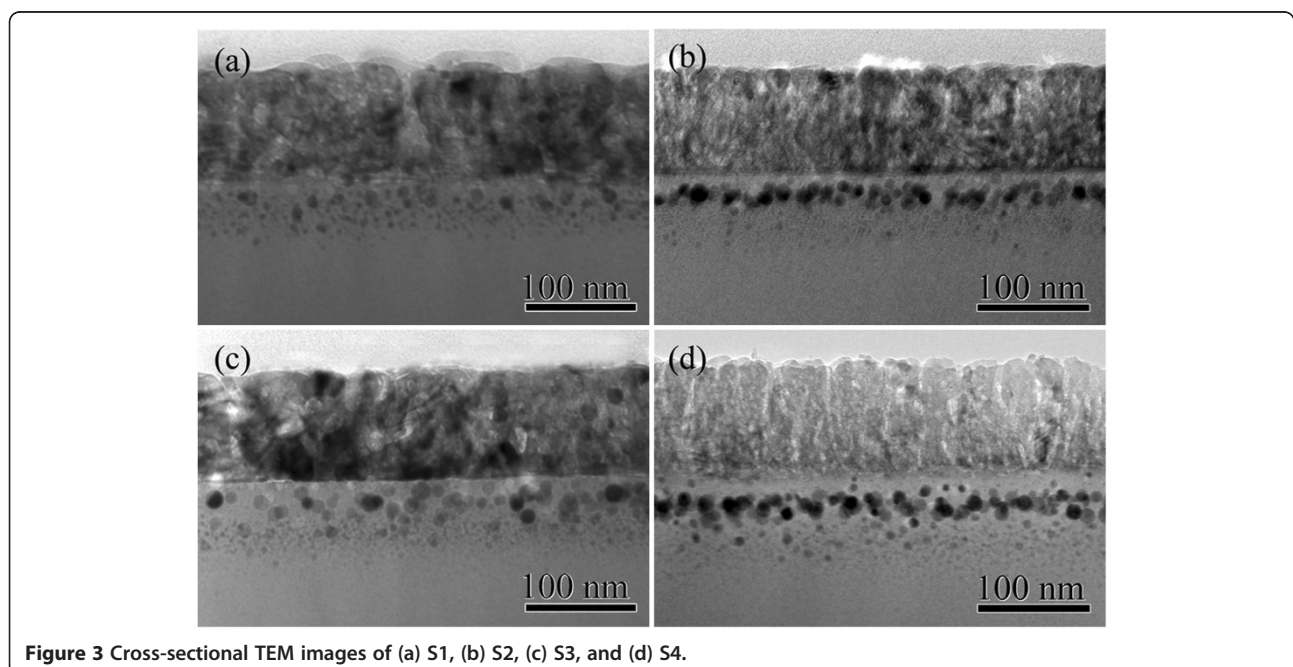


Figure 3 Cross-sectional TEM images of (a) S1, (b) S2, (c) S3, and (d) S4.

distance between the Ag NPs with the TiO₂ film; therefore, Raman scattering intensity of S4 has almost no enhancement. When the ion implantation fluence is increased to 1×10^{17} ions/cm² with an implantation energy of 40 kV (S3) as displayed in Figure 3c, large Ag NPs with a size of about 15 nm are formed near the surface and the small ones in the deeper SiO₂ matrix. The surface sputtering effect plays an important role for ion implantation at high fluence. The formed small Ag NPs near the surface are sputtered away by the subsequent implanted ions; as a result, the large Ag NPs are populated near the surface of S3 [24]. The Raman scattering enhancement factor is small with increasing implantation fluence. Therefore, the Raman scattering enhancement demonstrates that the strong near field is actually induced by introducing Ag NPs. The increased field could locally concentrate the light surrounding the Ag NPs and thus enhance the absorption of light.

In order to study the enhancement of light absorption in TiO₂-SiO₂-Ag nanostructural composites, the photocatalytic activities of S1 to S4 are investigated by the UV degradation of the MB solution at room temperature. For comparison, the TiO₂ film is carried out under the same experimental conditions. As shown in Figure 4a (inset), the concentration of MB is decreased upon the irradiation time, and the TiO₂ film can decompose 49% of MB after the UV irradiation for 4 h. However, the TiO₂-SiO₂-Ag nanostructural composite films obtained a higher photocatalytic efficiency than the pure TiO₂ film, and S2 has the highest photocatalytic efficiency compared to the other three samples and degraded 72% of MB. The enhancement ratio is as high as 47%. Meanwhile, the photodegradation of MB can be assumed to follow the classical Langmuir-Hinshelwood kinetics [30], and its kinetics can be expressed as follows:

$$\ln\left(\frac{A_0}{A}\right) = kt,$$

where k is the apparent first-order reaction rate constant (min⁻¹), and A_0 and A represent the absorbance before and after irradiation for time t , respectively. As displayed in Figure 4a, S2 shows the highest rate constant among all the samples. The k values of S2 are about two times than those of pure TiO₂. The kinetic rate constants follow the order S2 > S3 > S1 > S4 > TiO₂. This is consistent with the Raman scattering enhancement result.

The near-field enhancement in the TiO₂ layer due to the presence of the Ag NPs is also simulated using the finite-difference time-domain (FDTD) method as shown in Figure 4b. In our structure, we consider x as the light incident direction, the illuminating plane wave with a wavelength of 420 nm is y polarized, an Ag NP with a diameter of 20 nm is embedded in SiO₂, and the distance to the surface of the SiO₂ substrate is 7 nm. An

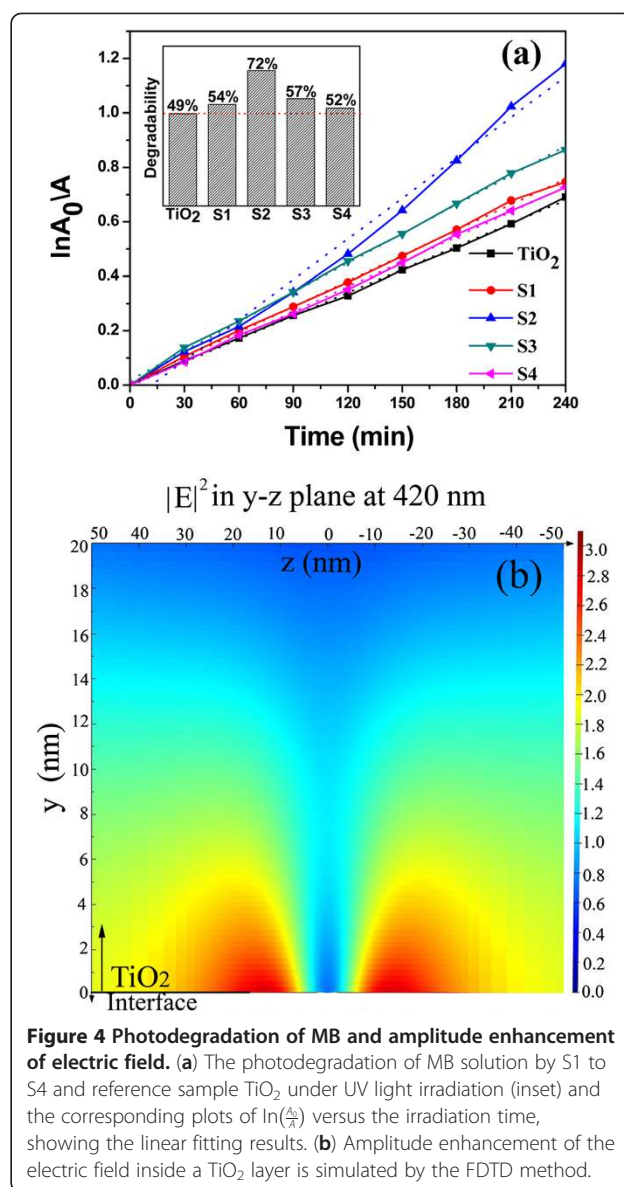


Figure 4 Photodegradation of MB and amplitude enhancement of electric field. **(a)** The photodegradation of MB solution by S1 to S4 and reference sample TiO₂ under UV light irradiation (inset) and the corresponding plots of $\ln(A_0/A)$ versus the irradiation time, showing the linear fitting results. **(b)** Amplitude enhancement of the electric field inside a TiO₂ layer is simulated by the FDTD method.

amplitude enhancement to 3 can be observed. Theoretical and experimental results show that an enhancement of the near field is induced by the SPR of Ag NPs. The SPR excitations cause a large increase in electromagnetic field in the vicinity of metal NPs. The localized amplification can increase the incident excitation field and boost the creation of hole–electron pairs, which results in the enhancement of the photocatalytic activity of TiO₂.

Conclusions

In conclusion, we have successfully demonstrated a plasmonic effect by simply incorporating Ag NPs with TiO₂ film. Optimum ion implantation conditions for Ag NPs synthesis in SiO₂ were experimentally estimated. The plasmonic effect occurring near the interface of TiO₂

and silica glass has effectively enhanced the light trapping. Both the experimental data and the simulations show that the enhancement effect is attained from the near-field enhancement induced by the SPR of Ag NPs. Our results have shown that the plasmonic effect has great potential in the application of increasing the UV light absorption in TiO₂ photocatalysts and opening up opportunities for highly efficient ultra-thin film solar cells.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

JX participated in the material preparation and data analysis and drafted the manuscript. XX conceived and co-wrote the paper. AS, FR, WW, GC, SZ, ZD, and FM participated in the sample characterization. CJ participated in its design and coordination. All authors read and approved the final manuscript.

Acknowledgments

The authors thank the National Basic Research Program of China (973 Program, 2009CB939704), the NSFC (10905043, 11005082, 91026014, 11175133, 51171132, 11004052, U1260102), the foundations from the Chinese Ministry of Education (311003, 20100141120042, 20110141130004), NCET, the Young Chenguang Project of Wuhan City (201050231055), the Fundamental Research Funds for the Central Universities, Hubei Provincial Natural Science Foundation (2011CDB270, 2012FFA042), and the Russian Foundation for Basic Research for the partial support.

Author details

¹Department of Physics and Key Laboratory of Artificial Micro- and Nano-structures of Ministry of Education, Wuhan University, Wuhan 430072, People's Republic of China. ²Center for Electron Microscopy and Hubei Nuclear Solid Physics Key Laboratory, Wuhan University, Wuhan 430072, People's Republic of China. ³Kazan Physical-Technical Institute, Russian Academy of Sciences, Kazan 420029, Russia. ⁴School of Electrical and Electronic Engineering, Hubei University of Technology, Wuhan 430068, People's Republic of China.

Received: 19 December 2012 Accepted: 18 January 2013

Published: 12 February 2013

References

1. Wang D, Zou Y, Wen S, Fan D: A passivated codoping approach to tailor the band edges of TiO₂ for efficient photocatalytic degradation of organic pollutants. *Appl Phys Lett* 2009, **95**:012106-1-3.
2. Han F, Kambala VSR, Srinivasan M, Rajarathnam D, Naidu R: Tailored titanium dioxide photocatalysts for the degradation of organic dyes in wastewater treatment: a review. *Appl Catal A-Gen* 2009, **359**:25-40.
3. Yang J, You J, Chen CC, Hsu WC, Tan HR, Zhang XW, Hong Z, Yang Y: Plasmonic polymer tandem solar cell. *ACS nano* 2011, **5**:6210-6217.
4. Min BK, Heo JE, Youn NK, Joo OS, Lee H, Kim JH, Kim HS: Tuning of the photocatalytic 1,4-dioxane degradation with surface plasmon resonance of gold nanoparticles on titania. *Catal Commun* 2009, **10**:712-715.
5. Kumar MK, Krishnamoorthy S, Tan LK, Chiam SY, Tripathy S, Gao H: Field effects in plasmonic photocatalyst by precise SiO₂ thickness control using atomic layer deposition. *ACS Catal* 2011, **1**:300-308.
6. Tong H, Quyang S, Bi Y, Umezawa N, Oshikiri M, Ye J: Nano-photocatalytic materials: possibilities and challenges. *Adv Mater* 2012, **24**:229-251.
7. Anpo M: Preparation, characterization, and reactivities of highly functional titanium oxide-based photocatalysts able to operate under UV-visible light. *Bull Chem Soc Jpn* 2004, **77**:1427-1442.
8. Asahi R, Morikawa T, Ohwaki T, Aoki K, Taga Y: Visible-light photocatalysis in nitrogen-doped titanium oxides. *Science* 2001, **293**:269-271.
9. Ghicov A, Macak JM, Tsuchiya H, Kunze J, Haeublein V, Frey L, Schmuki P: Ion implantation and annealing for an efficient N-doping of TiO₂ nanotubes. *Nano Lett* 2006, **6**(5):1080-1082.

10. Xu JH, Li J, Dai WL, Cao Y, Li H, Fan K: Simple fabrication of twist-like helix N,S-codoped titania photocatalyst with visible-light response. *Appl Catal, B-Environ* 2008, **79**:72-80.
11. Xiao XH, Ren F, Zhou XD, Peng TC, Wu W, Peng XN, Yu XF, Jiang CZ: Surface plasmon-enhanced light emission using silver nanoparticles embedded in ZnO. *Appl Phys Lett* 2010, **97**:071909-1-3.
12. Zhou XD, Xiao XH, Xu JX, Cai GX, Ren F, Jiang CZ: Mechanism of the enhancement and quenching of ZnO photoluminescence by ZnO-Ag coupling. *Europhys Lett* 2011, **93**(57009):1-6.
13. Zhang SG, Zhang XW, Yin ZG, Wang JX, Dong JJ, Gao HL, Si FT, Sun SS, Tao Y: Localized surface plasmon-enhanced electroluminescence from ZnO-based heterojunction light-emitting diodes. *Appl Phys Lett* 2011, **99**(181116):1-3.
14. Okamoto K, Niki I, Shvartser A, Narukawa Y, Mukai T, Scherer A: Surface-plasmon-enhanced light emitters based on InGaN quantum wells. *Nature Mater* 2004, **3**:601-605.
15. Awazu K, Fujimaki M, Rockstuhl C, Tominaga J, Murakami H, Ohki Y, Yoshida N, Watanabe T: A Plasmonic photocatalyst consisting of silver nanoparticles embedded in titanium dioxide. *J Am Chem Soc* 2008, **130**:1676-1680.
16. Oh J-H, Lee H, Kim D, Seong TY: Effect of Ag nanoparticle size on the plasmonic photocatalytic properties of TiO₂ thin films. *Surf Coat Technol* 2011, **206**(1):185-189.
17. Subrahmanyam A, Biju KP, Rajesh P, Jagadeesh Kumar K, Raveendra Kiran M: Surface modification of sol gel TiO₂ surface with sputtered metallic silver for Sun light photocatalytic activity: initial studies. *Sol Energy Mater Sol Cells* 2012, **101**:241-248.
18. Kerker M: The optics of colloidal silver: something old and something new. *J Colloid Interface Sci* 1985, **105**:297-314.
19. Stepanov AL, Hole DE, Townsend PD: Modification of size distribution of ion implanted silver nanoparticles in sodium silicate glass using laser and thermal annealing. *Nucl Instr Meth Phys Res B* 1999, **149**:89-98.
20. Linsebigler AL, Lu GQ, Jr Yates JT: Photocatalysis on TiO₂ surfaces: principles, mechanisms, and selected results. *Chem Rev* 1995, **95**:735-758.
21. Ren F, Jiang CZ, Liu C, Fu DJ, Shi Y: Interface influence on the surface plasmon resonance of Ag nanocluster composite. *Solid State Commun* 2005, **135**:268-272.
22. Zhang WF, He YL, Zhang MS, Yin Zand Chen Q: Raman scattering study on anatase TiO₂ nanocrystals. *J Phys D Appl Phys* 2000, **33**:912-916.
23. Willets KA, Van Duyne RP: Localized surface plasmon resonance spectroscopy and sensing. *Annu Rev Phys Chem* 2007, **58**:267-297.
24. Ren F, Xiao XH, Cai GX, Wang JB, Jiang CZ: Engineering embedded metal nanoparticles with ion beam technology. *Appl. Phys. A* 2009, **96**:317-325.
25. Xiao XH, Ren F, Wang JB, Liu C, Jiang CZ: Formation of aligned silver nanoparticles by ion implantation. *Mater Lett* 2007, **61**:4435-4437.
26. Ren F, Jiang CZ, Liu C, Wang JB, Oku T: Controlling the morphology of Ag nanoclusters by ion implantation to different doses and subsequent annealing. *Phys Rev Lett* 2006, **97**(165501):1-4.
27. Biteen JS, Lewis NS, Atwater HA: Spectral tuning of plasmon-enhanced silicon quantum dot luminescence. *Appl Phys Lett* 2006, **88**(131109):1-3.
28. Maier SA, Atwater HA: Plasmonics: localization and guiding of electromagnetic energy in metal/dielectric structures. *J Appl Phys* 2005, **98**(011101):1-10.
29. Chen CW, Wang CH, Wei CM, Chen YF: Tunable emission based on the composite of Au nanoparticles and CdSe quantum dots deposited on elastomeric film. *Appl Phys Lett* 2009, **94**(071906):1-3.
30. Al-Ekabi H, Serpone N: Kinetic studies in heterogeneous photocatalysis. 1. Photocatalytic degradation of chlorinated phenols in aerated aqueous solutions over TiO₂ supported on a glass matrix. *J Phys Chem* 1988, **92**:5726-5731.

doi:10.1186/1556-276X-8-73

Cite this article as: Xu et al.: Efficiency enhancements in Ag nanoparticles-SiO₂-TiO₂ sandwiched structure via plasmonic effect-enhanced light capturing. *Nanoscale Research Letters* 2013 **8**:73.