DR MICHELLE WATT (Orcid ID: 0000-0001-7843-0957)
DR BORJANA ARSOVA (Orcid ID: 0000-0002-0566-2009)
DR KYLIE JANE FOSTER (Orcid ID: 0000-0003-2651-3915)
PROF. MATTHEW GILLIHAM (Orcid ID: 0000-0003-0666-3078)
PROF. COLIN JENKINS (Orcid ID: 0000-0002-9347-8948)
DR NICOLAS L TAYLOR (Orcid ID: 0000-0003-2004-5217)

DR LARS H WEGNER (Orcid ID: 0000-0002-9263-8436)
PROF. STEPHEN TYERMAN (Orcid ID: 0000-0003-2455-1643)

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### **Energy costs of salt tolerance in crop plants**

Rana Munns<sup>\*1</sup>, David A. Day<sup>\*2</sup>, Wieland Fricke<sup>\*3</sup>, Michelle Watt<sup>\*4</sup>, Borjana Arsova<sup>4</sup>, Bronwyn J. Barkla<sup>5</sup>, Jayakumar Bose<sup>6</sup>, Caitlin S. Byrt<sup>6,7</sup>, Zhong-Hua Chen<sup>8</sup>, Kylie J. Foster<sup>9</sup>, Matthew Gilliham<sup>6</sup>, Sam W. Henderson<sup>10</sup>, Colin L.D. Jenkins<sup>2</sup>, Herbert J. Kronzucker<sup>11</sup>, Stanley J. Miklavcic<sup>9</sup>, Darren Plett<sup>11</sup>, Stuart J. Roy<sup>12</sup>, Sergey Shabala<sup>13</sup>, Megan C. Shelden<sup>6</sup>, Kathleen L. Soole<sup>2</sup>, Nicolas L. Taylor<sup>14</sup>, Mark Tester<sup>15</sup>, Stefanie Wege<sup>6</sup>, Lars H. Wegner<sup>16</sup>, Stephen D. Tyerman<sup>#6</sup>

<sup>1</sup> Australian Research Council (ARC) Centre of Excellence in Plant Energy Biology, and School of Agriculture and Environment, The University of Western Australia, Crawley, WA 6009, Australia; CSIRO Agriculture and Food, Canberra, ACT 2601, Australia <sup>2</sup>College of Science and Engineering, Flinders University, GPO Box 2100, Adelaide, South Australia 5001, Australia.

<sup>3</sup> School of Biology and Environmental Sciences, University College Dublin (UCD), , Dublin, Ireland

<sup>4</sup>Plant Sciences, Institute of Bio and Geosciences, Forschungszentrum Juelich, Helmholtz Association, 52425, Juelich, Germany

<sup>5</sup>Southern Cross Plant Science, Southern Cross University, Lismore NSW, 2481, Australia <sup>6</sup>Australian Research Council (ARC) Centre of Excellence in Plant Energy Biology, School of Agriculture, Food and Wine, University of Adelaide, Glen Osmond, SA 5064, Australia <sup>7</sup>Research School of Biology, Australian National University, ACT 2600, Australia

<sup>8</sup> School of Science and Health, Western Sydney University, Penrith, NSW 2751, Australia <sup>9</sup>Phenomics and Bioinformatics Research Centre, School of Information Technology and Mathematical Sciences, University of South Australia, Mawson Lakes, SA 5095, Australia <sup>10</sup> Commonwealth Scientific and Industrial Research Organisation, Agriculture and Food, Urrbrae, SA, 5064, Australia

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Rana Munns http://orcid.org/0000-0002-7519-2698

<sup>11</sup>School of Agriculture and Food, Faculty of Veterinary and Agricultural Sciences, The University of Melbourne, VIC, 3010, Australia

<sup>12</sup> Australian Research Council (ARC) Industrial Transformation Research Hub for Wheat in a Hot and Dry Climate, School of Agriculture, Food and Wine, University of Adelaide, Urrbrae, SA 5064, Australia

<sup>13</sup> Tasmanian Institute for Agriculture, University of Tasmania, Private Bag 54, Hobart, Tas. 7001, Australia and International Centre for Environmental Membrane Biology, Foshan University, Foshan, China.

<sup>14</sup> Australian Research Council (ARC) Centre of Excellence in Plant Energy Biology, School of Molecular Sciences and Institute of Agriculture, The University of Western Australia, Crawley, WA 6009, Australia

<sup>15</sup>King Abdullah University of Science and Technology (KAUST), Biological and Environmental Sciences & Engineering Division (BESE), Thuwal, 23955-6900, Saudi Arabia <sup>16</sup>Karlsruhe Institute of Technology, Institute for Pulsed Power and Microwave Technology (IHM), D-76344 Eggenstein-Leopoldshafen, Germany

<sup>#</sup>Author for Correspondence: Stephen D Tyerman

Phone (61) 83136663

Email: steve.tyerman@adelaide.edu.au

\*These authors contributed equally to this work

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### **ORCID**

David A. Day http://orcid.org/0000-0001-7967-2173 Wieland Fricke http://orcid.org/0000-0002-1514-1389 Michelle Watt http://orcid.org/0000-0001-7843-0957 Borjana Arsova http://orcid.org/0000-0002-0566-2009 Bronwyn J. Barkla http://orcid.org/0000-0002-4691-8023 Jayakumar Bose http://orcid.org/0000-0002-0565-2951 Caitlin S. Byrt http://orcid.org/0000-0001-8549-2873 Zhonghua Chen http://orcid.org/0000-0002-7531-320X Kylie J. Foster http://orcid.org/0000-0003-2651-3915 Matthew Gilliham http://orcid.org/0000-0003-0666-3078 Sam W. Henderson http://orcid.org/0000-0003-3019-1891 Colin L.D. Jenkins http://orcid.org/0000-0002-9347-8948 Herbert J. Kronzucker http://orcid.org/0000-0002-9358-0029 Stanley J. Miklavcic http://orcid.org/0000-0002-2361-246X Darren Plett http://orcid.org/0000-0002-9551-8755 Stuart J. Roy http://orcid.org/0000-0003-0411-9431 Sergey Shabala http://orcid.org/0000-0003-2345-8981 Megan C. Shelden http://orcid.org/0000-0002-7203-3763

II III V.

Kathleen L. Soole: http://orcid.org/0000-0002-8837-3404 Nicolas Taylor https://orcid.org/0000-0003-2004-5217 Mark Tester http://orcid.org/0000-0002-5085-8801 Stefanie Wege http://orcid.org/0000-0002-7232-5889 Lars Wegner http://orcid.org/0000-0003-4375-7508 Stephen D. Tyerman http://orcid.org/0000-0003-2455-1643

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### **Summary**

Agriculture is expanding into regions that are affected by salinity. This review considers energy costs of salinity tolerance in crop plants and provides a framework for a quantitative assessment of costs. Different sources of energy, and modifications of root system architecture that would maximise water versus ion uptake are addressed. Energy requirements for transport of NaCl to leaf vacuoles for osmotic adjustment could be small if there are no substantial leaks back across plasma membrane and tonoplast in root and leaf. The coupling ratio of H<sup>+</sup>-ATPase is also a critical component. One proposed leak, that of Na<sup>+</sup> influx across the plasma membrane through certain aquaporin channels, might be coupled to water flow, thus conserving energy. For the tonoplast, control of two types of cation channels is required for energy efficiency. Transporters controlling the Na<sup>+</sup> and Cl<sup>-</sup> concentrations in mitochondria and chloroplasts are largely unknown and could be a major energy cost. The complexity of the system will require a sophisticated modelling approach to identify critical transporters, apoplastic barriers and root structures. This modelling approach will inform experimentation and allow a quantitative assessment of the energy costs of salt tolerance to guide breeding and engineering of molecular components.

**Keywords:** barley and wheat, energy costs, membrane transport, photosynthesis, respiration, root anatomy, salt tolerance, sodium and chloride transport

### I. Introduction

Soil salinity in agricultural areas is increasing world-wide due to irrigation with brackish water and to seawater encroachment on low-lying coastal regions. Its impact on crop production is further increasing as the global demand for food means agriculture extends into naturally salt-affected lands (Fig. 1a).

Salt tolerance for crop plants means the ability to grow, although more slowly, and produce a harvestable yield. The level of salinity that affects crop yields depends on species, duration of exposure, and stage of crop development at which stress occurs. Saline soil (predominantly Na<sup>+</sup> and Cl<sup>-</sup> salts) is defined as having an EC<sub>e</sub> over 4 dS m<sup>-1</sup> equivalent to 40 mM NaCl (**Box 1**), and in a well-drained soil would be twice this, about 80 mM NaCl. This would reduce the growth of most crops by 15-20% (Munns and Tester, 2008). A higher soil salt concentration would increasingly reduce growth, but the extent of yield reduction is hard to predict as saline soils are never uniformly saline across the soil and at depth (Fig. **1b**).

The precise cause of the growth reduction remains elusive – is it a pre-emptive response to conserve resources via feed-forward root signals that reduce leaf expansion and stomatal conductance, or a result of the reduction in supply of photosynthate? Is the plant adapting to conserve energy and use it more efficiently, or is a reduced supply of energy from photosynthesis limiting its growth? These principles apply to a dry soil as much as to a saline soil (Munns, 2002).

Adaptions to saline soil are many. There is a delicate balance between excluding most of the salt to avoid it concentrating in leaves and take up enough for osmotic adjustment. Plants in saline soil must exclude almost all (about 98%) the salt otherwise leaf concentrations would quickly rise to toxic concentrations (Munns *et al.*, 2019). Osmotic adjustment using Na<sup>+</sup> and Cl<sup>-</sup> is 'cheaper' than using organic solutes, as long as the salt is sequestered in vacuoles while organic solutes provide the balancing osmotic pressure in the cytoplasm (Munns and Gilliham, 2015). Plants can avoid the high carbon cost of organic solutes for osmotic adjustment by using mainly Na<sup>+</sup> and Cl<sup>-</sup> but this also comes with a cost.

This review considers the costs of salt tolerance at both the whole plant and cell levels. The costs of regulating ion and water uptake, and transport within the plant, are compared with the energy available from respiration to identify the most cost-efficient strategy, as a guide to breeding salt-tolerant crops.

#### Box 1

### Standard units and conversions

Electrical conductivity (EC): 1 dS m<sup>-1</sup> = 10 mM NaCl

 $EC_e$  is EC of a saturated soil paste extract. A saturated soil has about twice the water content of a well-drained soil (at field capacity), which would decrease further during periods without

 $EC_a$  is apparent EC, measured with a ground conductivity meter such as EM38. Usually given in mS m<sup>-1</sup> (100 \*dS m<sup>-1</sup>). Must be calibrated against soil samples.

Osmotic pressure  $(\pi) = cRT$  (van't Hoff equation) where c is the solute concentration in Osmol L<sup>-1</sup>, and RT is 2.48 at 25 °C (litre-MPa per mole). An ideal solution with osmotic potential of 0.1 MPa has 40 mOsmol L<sup>-1</sup> total solutes or 20 mmol L<sup>-1</sup> NaCl.

### Growth parameters – for wheat or barley in moderate salinity (150 mM NaCl or 15 dS m<sup>-1</sup>) in supported hydroponics

Growth rates (RGR) in saline soil: 0.1 g g<sup>-1</sup> d<sup>-1</sup> (Colmer *et al.*, 1995; Rivelli *et al.*, 2002) Photosynthesis rates: 25-28 µmol m<sup>-2</sup> s<sup>-1</sup> (James *et al.*, 2002; Fricke, 2017) Leaf respiration rates: 7 µmol g<sup>-1</sup>FW h<sup>-1</sup>; 0.5-0.8 µmol m<sup>-2</sup> s<sup>-1</sup> (Scafaro *et al.*, 2017)

Specific leaf area: 35 m<sup>2</sup> kg<sup>-1</sup> DW (Rawson, 1987)

Conversion of leaf respiration (R) from leaf area to FW basis (from µmol m<sup>-2</sup> s<sup>-1</sup> to

µmol g<sup>-1</sup>FW h<sup>-1</sup>) is therefore R\*SLA÷ FW/DW÷1000\*60\*60

FW/DW for leaves ranges from 4-10 depending on species.

Typical root respiration rate: 10-25 µmol g<sup>-1</sup>FW h<sup>-1</sup> (Alexova et al., 2014; Scafaro et al., 2017)

ATP produced by respiration (O<sub>2</sub> consumption) ATP:O<sub>2</sub> is 4.5 (Box 2)

Root surface area to fresh weight ratio: 70 cm<sup>2</sup> g<sup>-1</sup>FW (Fricke, pers. com.)

Typical FW/DW for roots grown in hydroponics: 15:1 (Husain et al., 2004; Chen et al., 2005); FW/DW for roots grown in soil or for woody perennials is about 10:1.

### II. Tissue respiration and energy supply

Mitochondrial ATP supply and the alternative pathway

The plant mitochondrial electron transport chain (mETC) contains two interconnected pathways with different terminal oxidases: cytochrome c oxidase (COX) as part of the phosphorylating, energy conserving classical mETC, and the alternative oxidase (AOX), which together with alternative NAD(P)H dehydrogenases forms a non-phosphorylating bypass of the classical mETC (Millar et al., 2011; Vanlerberghe, 2013) (Fig. 2). Electron transport entirely through the alternative pathway is not coupled to ATP synthesis and its operation can have a substantial impact on the efficiency of respiration by affecting the tissue ATP:O<sub>2</sub> ratio (Box 2; Supporting Information Notes S1). The activity of the alternative pathway in vivo is often stimulated by environmental stresses (Selinski et al., 2018), including salinity (Del-Saz et al., 2016). The role of the alternative pathway in plants remains uncertain but evidence suggests that it minimises production of reactive oxygen species (ROS) in mitochondria and plays an important role in stress tolerance. In a study on the effect of salinity on Arabidopsis, both external NADH oxidation and AOX increased in capacity because of an increase in the amount of the relevant proteins (Smith et al., 2009). Likewise, in an analysis of the mitochondrial proteome from salt-stressed wheat, a number of antioxidant defence enzymes were increased in abundance, with AOX among the most abundant (Jacoby et al., 2013). AOX may also play a more general role in the homeostasis of cell metabolism by modulating Tricarboxylic Acid (TCA) cycle operation under conditions where respiration is restricted by cellular energy status (Vanlerberghe 2013), and this may be important for the synthesis of secondary metabolites, including compatible solutes under saline conditions.

In many plants, AOX synthesis is stimulated by environmental and chemical stress, although usually low in the absence of stress (Selinski *et al.*, 2018). Inhibition of the mETC also triggers AOX synthesis which is pertinent since high salt concentrations can inhibit cytochrome path activity in isolated mitochondria (Jacoby *et al.*, 2011). In some other plants, particularly legumes, AOX protein is expressed constitutively and its activity can be substantial although subject to post-translational regulation by mitochondrial redox status and the presence of certain organic acids, especially pyruvate (Selinski *et al.*, 2018). In legumes, AOX contribution to respiration varies depending on tissue and developmental stage, and the

imposition of stresses such as drought and salinity. It can be very significant, leading to decreases in ATP levels *in vivo* (Millar *et al.*, 1998; Ribas-Carbo *et al.*, 2005; Del-Saz *et al.*, 2016). Mitochondria isolated from most species readily oxidise exogenous NADH, but we know little about the regulation of the alternative NAD(P)H dehydrogenases *in vivo*. Some of the alternative NADH dehydrogenases are transcriptionally responsive to salt stress (Smith *et al.*, 2009), and the oxidation of external NADH is stimulated by salt in isolated mitochondria. The oxidation of malate, pyruvate, succinate and glutamate, on the other hand, is inhibited, indicating a differential effect of salt on tissue respiration depending on the substrate respired (Jacoby *et al.*, 2016).

Relatively few accurate studies on the *in vivo* contribution of AOX to respiration have been made, mainly because of the difficulty involved in these studies. Accurate estimations require mass spectrometry to measure O<sup>18</sup> discrimination between AOX and COX (Del-Saz *et al.*, 2017). Since AOX activity has the potential to severely impact respiratory efficiency, it is important that more *in vivo* measurements of AOX engagement in a variety of species under saline conditions are made if we are to fully understand the energetics of salt tolerance.

### Sources of NADH for Tissue Respiration

Tissue respiration is at the centre of plant metabolic networks as the TCA cycle links it to both carbon and nitrogen metabolism and supplies much of the NADH required to maintain ATP supply (Nunes-Nesi *et al.*, 2013). Salt inhibition of the TCA cycle could activate alternative metabolic routes, including the GABA shunt pathway (Krasensky & Jonak, 2012; Nunes-Nesi *et al.*, 2013). The GABA shunt bypasses the oxoglutarate dehydrogenase complex (OGDC) catalysed step of the TCA cycle (Fig. 2), producing glutamate that enters the mitochondria for further catabolism (Che-Othman *et al.*, 2019) (Fig. 2). The GABA shunt is thought to be important in stress adaptation in plants by regulating cytosolic pH, limiting ROS production, regulating nitrogen metabolism and bypassing steps in the TCA cycle (Carillo, 2018). Links are suggested between salinity exposure and activity of the GABA shunt (Che-Othman *et al.*, 2017), but the exact mechanism of induction and its consequences for tissue respiration are only just starting to be revealed (Che-Othman *et al.*, 2019)

In addition to TCA cycle enzymes and Glycine Decarboxylase Complex (GDC), it is likely that alternative sources of NADH may become significant under salinity exposure. The induction of other dehydrogenases is observed in other stresses, where amino acids are

broken down and increases are seen in branched chain amino acid catabolism enzymes (Peng et al., 2015). Some of these dehydrogenases have been shown to contribute electrons directly to the ubiquinone pool of the mETC, effectively bypassing Complex I and consequently altering ATP synthesis. Improving our understanding of these different metabolic pathways will be required if we are to fully understand the implications of salinity exposure on NADH supply and mitochondrial ATP production.

The contribution of beta-oxidation of fatty acids to energy supply during salt stress is unknown, but we consider it not likely to play a major role in plant adaptation to this stress. The participation of mitochondria in this process in plants is controversial and it is generally assumed that most beta-oxidation occurs in the peroxisomes. The extent to which this contributes to ATP production is likely to depend on species and tissue and has not been investigated to our knowledge. More work is needed, especially under saline conditions.

Vacuolar proton-pumping pyrophosphatases provide additional energy sources for salinity tolerance.

High-energy phosphate containing molecules other than ATP can be used for energising processes. Vacuolar pyrophosphatases (Vacuolar H<sup>+</sup>-PPase, EC.3.6.1.1) pump protons across the tonoplast into vacuoles (Gaxiola *et al.*, 2016; Schilling *et al.*, 2017), using pyrophosphate (PPi) as an energy source (Fig. 3). They work together with vacuolar H<sup>+</sup>-ATPases to acidify the vacuole (Kriegel *et al.*, 2015; Schilling *et al.*, 2017). Plants with high expression of vacuolar H<sup>+</sup>-PPases have significant abiotic stress tolerance, including salinity tolerance (Gaxiola *et al.*, 2016; Schilling *et al.*, 2017).

Vacuolar H<sup>+</sup>-PPases may be particularly important when ATP supply is limited during abiotic stress. A significant portion of vacuolar acidification may be generated by non-ATPase pathways and used by Na<sup>+</sup>/H<sup>+</sup> and Cl<sup>-</sup>/H<sup>+</sup> antiporters to sequester Na<sup>+</sup> and Cl<sup>-</sup> in the vacuole as part of a tissue tolerance mechanism (Li *et al.* 2006; Kriegel *et al.*, 2015; Nguyen *et al.* 2016). H<sup>+</sup>-PPases are dependent on K<sup>+</sup>, thus K retention in the cytoplasm under salinity may be critical for their function (Shabala *et al.*, 2014). H<sup>+</sup>-PPases have been shown more recently to be involved with rapid mobilisation of sugars and carbohydrates from source to sink tissue (Pizzo *et al.*, 2015, Gaxiola *et al.*, 2016), and in faster metabolism of sugars in cells (Ferjani *et al.*, 2012). Both processes will contribute to enhancing a cell's energy budget.

### Energy supply from photosynthesis

Photosynthesis supplies energy in the form of reduced carbon to fuel growth and maintenance. In the majority of crop plants, salt stress decreases photosynthesis by (1) reducing CO<sub>2</sub> uptake through stomata (Rawson, 1986; Delfine *et al.*, 1999; James *et al.*, 2002), (2) causing ionic (Na<sup>+</sup>, K<sup>+</sup>, Cl<sup>-</sup>) imbalance within the chloroplasts, resulting in poor efficiency of light and dark reactions (Delfine *et al.*, 1999; Percey *et al.*, 2016; Bose *et al.*, 2017), and (3) causing oxidative damage to photosystems and membranes (Miller *et al.*, 2010; Bose *et al.*, 2014). However, the extent of the decrease in photosynthesis under salt varies widely among crop species and genotypes. In tolerant genotypes that are able to keep salt out of the leaf, photosynthetic supply is not affected. In such cases, the decrease in potential growth (compared to a non-saline control) must be directly proportional to, and a measure of, the energy cost associated with tolerance.

### How to estimate the energy cost of CO<sub>2</sub> fixation during salt stress

In a non-limiting environment (e.g. absence of photorespiration), 3 ATP and 2 NADPH are required per CO<sub>2</sub> assimilated (Wingler *et al.*, 2000). This demand can be met through absorption of four photons each by PSII and PSI and subsequent linear transport of four electrons from PSII to PSI (Kramer & Evans, 2011). Under salt stress there is an increase in energy demand for CO<sub>2</sub> assimilation due to an increase in photorespiration (Wingler *et al.*, 2000), protein turnover to repair and strengthen photosystem components (e.g. energy cost to import a protein into chloroplasts is about 650 ATP) (Miller *et al.*, 2010; Shi & Theg, 2013), and ion transport activity to maintain optimum ionic environment within the chloroplasts (Bose *et al.*, 2017). For example, during photorespiration, each oxygenation reaction consumes 3.25 ATP and 2 NADPH (Wingler *et al.*, 2000). Photorespiration can occur at ~25% the rate of net CO<sub>2</sub> assimilation (25°C, CO<sub>2</sub>=350 ppm) (Walker et al., 2016), which will increase the consumption of ATP and reducing equivalents per CO<sub>2</sub> fixed from 3 ATP and 2 NADPH to 5.375 ATP and 3.5 NADPH (Box 2, Supporting Information Notes S1).

To meet the high energy demand during salt stress, chloroplasts could increase ATP production by altering the H<sup>+</sup>/ATP ratio required by the ATP synthase or engaging with cyclic electron flow around PSI, using the water-water cycle, the malate valve, and

plastoquinol oxidase (Kramer & Evans, 2011). Among these mechanisms, enhanced cyclic electron flow around PSI has been shown to increase ATP production during salt stress, and the excess ATP generated has been suggested to fuel ion transport mechanisms that prevent salt over-accumulation into the chloroplasts (He *et al.*, 2015). Under control conditions cyclic electron flow amounts to 14%; whether it increases under salinity is important to know.

#### Box 2

A calculation of *in vivo* ATP : O<sub>2</sub> ratios (see Siedow and Day 2000 for details)

ATP:O<sub>2</sub> ratios can vary in vivo between 5.0 and 1.75 during sucrose oxidation in plant tissues, depending on the relative contributions of the cytochrome path and the AP, which in turn depends on environmental conditions. Oxidation of one molecule of sucrose via glycolysis yields 4 ATP directly as well as 4 pyruvate molecules and 4 NADH for further oxidation in the mitochondria. Complete oxidation of the pyruvate and NADH in the mitochondria consumes 12 O<sub>2</sub> and yields a number of ATP depending on the electron transport pathway engaged. Taking into account the magnitude of the proton motive force across the mitochondrial inner membrane (typically 240 mV) 3H<sup>+</sup> must move through the ATP synthase to generate an ATP molecule; the need to import Pi and ADP and export ATP consumes another H<sup>+</sup> equivalent, giving an H<sup>+</sup>:ATP of 4. Measurements with isolated mitochondria indicate that 10 H<sup>+</sup> are translocated out of the mitochondria for each NADH oxidised via complex I of the mETC, giving a maximal ATP:O ratio of 2.5, and 6 H<sup>+</sup> for each succinate or external NADH oxidised, giving an ATP:O<sub>2</sub> ratio of 1.5. From this it follows that when only the cytochrome pathway of the mETC operates, the yield of ATP from the complete oxidation of one sucrose molecule yields 60 ATP. If electron flow is via AOX, then this drops to 21 ATP (assuming that intra-mitochondrial NADH is oxidised only via Complex I rather than through the internal alternative NADH dehydrogenases). In reality, the *in vivo* ATP:O<sub>2</sub> ratio will be somewhere between the two extremes, depending on the relative contributions of the cytochrome path and the AP, which in turn depends on environmental conditions.

### B Calculation of energy fixed through photosynthesis

Net  $CO_2$  assimilation (A) can be calculated from the rates of Rubisco carboxylation (Vc) and oxygenation (Vo), and the  $CO_2$  release from mitochondrial day respiration (Rd) using the following formula (Rawson, 1986; Walker et al., 2016).

A = Vc - 0.5Vo - Rd

Each Rubisco oxygenation reaction consumes 3.25 ATP and 2 NADPH, and carboxylation reaction consumes 3 ATP and 2 NADPH (Wingler *et al.*, 2000; Miller *et al.*, 2010). Total energy requirement per  $CO_2$  assimilation can be calculated by summing the energy requirements oxygenation and carboxylation reactions. For Vc = 1.25, Vo = 0.5, A = 1 and Rd = 0, the energy requirement will be

1.25 (3 ATP + 2 NADPH) + 0.5 (3.25 ATP + 2 NADPH) = 5.375 ATP + 3.5 NADPH

As the energy partitioning during photosynthesis and photorespiration is predictable, the requirement of ATP and NADPH can be modelled during salt stress using pre-existing resources (Walker *et al.*, 2016)

(http://demonstrations.wolfram.com/FluxesAndEnergeticsOfPhotosynthesisAndPhotorespiration/).

### III. Water and ion transport

Membranes are inherently high energy barriers for water and ion transport. For energetics of salinity tolerance, any combined flux of Na<sup>+</sup> and Cl<sup>-</sup> across either plasma membrane or tonoplast and resultant feedbacks for maintenance of electroneutrality and negative membrane potential will entail energy costs to the cell. Elevated fluxes of Cl<sup>-</sup> and Na<sup>+</sup> under salinity stress will affect most other transport due to changes in membrane potential and pH (due to proton coupled transport). For water, control can be exerted and much higher permeability obtained via the operation of aquaporins (Maurel *et al.*, 2015). Aquaporins are linked to cell energetics (Chaumont & Tyerman, 2014), respond in a complex way to salinity (Boursiac *et al.*, 2005; Boursiac *et al.*, 2008; McGaughey *et al.*, 2018), and some may allow ion permeation (Byrt *et al.*, 2015; Kourghi *et al.*, 2017). Changes in water permeation will influence Na<sup>+</sup> and Cl<sup>-</sup> fluxes by virtue of their coupling by convection in the transpiration stream and in radial transport across the root (Foster & Miklavcic, 2016; Foster & Miklavcic, 2017).

In previous estimates of energy costs of ion fluxes in roots, the paradigm used is the number of membranes crossed and the number of protons consumed, which can then be converted to ATP demand based on stoichiometry of protons pumped per ATP hydrolysed (usually taken as 1:1) (Venema & Palmgren, 1995; Kurimoto *et al.*, 2004; Malagoli *et al.*, 2008). Some isotope flux calculations indicate that energy costs approach the ATP produced by respiration (e.g. Malagoli *et al.*, 2008 for rice), although subsequent publications have questioned these results (Britto & Kronzucker, 2015; Flam-Shepherd *et al.* 2018; Munns *et al.* this volume). These estimates are just for Na<sup>+</sup> in isolation and ignore the Cl<sup>-</sup> ion and feedback effects on other fluxes, e.g. loss of K<sup>+</sup> (Cuin *et al.*, 2008) and NO<sub>3</sub><sup>-</sup> (Teakle & Tyerman, 2010). Just the stoichiometry of the proton pump could have a profound effect on the energetics, and if less than 1H<sup>+</sup>/ATP (e.g. when the pump is not activated (Pedersen *et al.* 2017)) would reduce the energy efficiency substantially, and it is interesting that cytosolic K<sup>+</sup> and potentially Na<sup>+</sup> can have an effect on this (Buch-Pedersen *et al.*, 2006). The identification

of molecular components of energetically costly processes can be used to engineer plants for improved yield (Amthor *et al.*, 2018).

### Ways of calculating energy of transport

### Energy budgets

When considering a plant that tolerates salinity, we can ask many questions about how it manages in terms of energy (Amthor, 2000; Bijlsma *et al.*, 2000; Kooijman & Troost, 2007; Munns & Gilliham, 2015). Ideally, we want to know the total energy that is (A) captured, (B) expended on biochemical work including growth, maintenance of metabolism and respiration, and (C) invested in salt tolerance. Obtaining data on A-C is one thing; interpreting these data, is another. For example, if the energy drain is large, we may conclude that the responses come at high cost and, because they require such a large portion of energy, must be important. Yet, we could also argue that the responses are unlikely to operate in the long term, because they are so costly. It is desirable to include any component processes (Raven, 1985; Assmann & Zeiger, 1987; Bloom *et al.*, 1992; Scheurwater *et al.*, 2000), for example synthesis of transport proteins, in the energy budget of the overall process. However, in practice this is difficult to achieve, and it is easier to consider a well-defined start and endpoint of the process. We apply this rationale here to solutes in the first instance.

#### Energising solute movement

The energy of a solute X, such as a mineral nutrient ion, can be quantified by the electrochemical potential of that solute ( $\mu_X$ ). The energy which is released by or required for the movement of X from soil (S) into a leaf cell (L) is  $\Delta\mu_X(SL) = \mu_X(L) - \mu_X(S)$ , with  $\Delta\mu_X(SL) < 0$  for a spontaneous movement of X down a gradient in energy, and  $\Delta\mu_X(SL) > 0$  for a movement of X energetically-uphill (for calculations see Figure 4). Option-1, we ignore all transport steps between S and L and only consider the start and end point of  $\mu_X$ . Option-2, we consider all the intermittent steps of transport. Option-3, we render the second option more complete and more complicated, by including transport across intracellular membranes, particularly the tonoplast. Option-4, we ignore the chemical nature of solute X and link its movement to the energisation of transport processes across the plasma membrane through the plasma membrane H<sup>+</sup>-ATPase (PM-H<sup>+</sup>-ATPase); we also ignore intermittent transport steps. One H<sup>+</sup> needs to be pumped at least for every solute moved from S to L (at least one

membrane crossing event) so we base our energy calculation on the proton motive force (pmf) (Palmgren, 2001; Fricke, 2017). Option-5, we include the energy which is required to synthesise/maintain the biomolecules (e.g. membrane transporters) and cellular processes (e.g. trafficking) that are required to support the transport steps (Raven, 1985). This approach is the most complete one, yet the difficulty is that all processes in a cell are somehow interconnected and should be considered. This is not possible to do as yet.

Option-1 provides the lowest estimate of energy for solute movement from S to L. This applies particularly to the situation under salt stress (high external Na<sup>+</sup> and Cl<sup>-</sup>). We deal with a cation and anion, each crossing plasma membranes with a significantly negative (inside) membrane electrical potential (Cuin *et al.*, 2003). This should favour uptake of Na<sup>+</sup>, even if it was as support of the co-transport of another solute (e.g. K<sup>+</sup>) into cells (Carden *et al.*, 2001; Carden *et al.*, 2003; Cuin *et al.*, 2003). By considering the chemical nature of each solute we 'save' energy (Fricke, 2019 – see current feature issue). Any intermediate transport between S and L causes 'frictional' loss of energy through heat. By omitting intermediate transport steps while taking the chemical nature of solute X into account, we obtain a lowest-cost estimate for solute accumulation.

#### Cost estimates

If the cost estimate for solute movement exceeds the energy available from carbon assimilation (day) or dark respiration (night), we have to question our assumptions on energy budgets; if the cost estimate is well within the energy available, it does not mean that our energy budgets and assumptions are correct, but we can conclude that this solute movement is theoretically possible. We can also compare our lowest-cost estimate with the minimum energy required to establish turgor pressure in cells, which results from accumulation of solutes and associated water uptake. Turgor pressure has many functions in cells (Beauzamy et al., 2014), and it also presents some energy, as 1 MPa of hydrostatic pressure corresponds to 10<sup>6</sup> J m<sup>-3</sup>. In Box 3 calculations of energy requirements are made for transport of Na<sup>+</sup> and Cl<sup>-</sup> from a saline soil solution to leaf vacuoles of a wheat plant that is accumulating 150 mM NaCl in order to osmotically adjust during growth. It is based on Option 4 (Fig. 4) but including minimal (not all) transport steps in the pathway and different possibilities that may occur at these steps with respect to leaks (Fig. 5). These are compared with rates of respiration in order to assess the proportion of energy available that is used in transport. This exercise serves to illustrate the magnitude of energy required as well as some of the parameters that are required, particularly the leaks across membranes in the pathway that can

substantial increase the energy required. These leaks are dealt with further in the following sections.

# Box 3 Calculation of minimum costs of NaCl accumulation in leaf and root vacuoles of a wheat plant growing in a saline soil of 150 mM NaCl and comparing effect of leaks.

NaCl concentration in vacuole = 150 mM

Growth rate =  $0.1 \text{ day}^{-1}$ 

Shoot/root ratio = 1.3 (Van Den Boogaard et al. 1996)

Leaf resp. =  $7 \mu mol gm_{fwt}^{-1} h^{-1} Root resp. = 10 \mu mol gm_{fwt}^{-1} h^{-1}$ 

Plant C used in root resp. = 16%; Plant C used in leaf resp. = 13% (Van Den Boogaard *et al.* 1996)

 $H^+/Na^+ = 1$ ,  $H^+/Cl^- = 2$ ,  $H^+/ATP = 1$ 

 $ATP/O_2 = 4.5$ 

If leak present =  $0.5 \times 10^{-5} \times$ 

Pathway: One way to leaf via xylem through roots (no bypass flow, Fig. 5)

Na<sup>+</sup> Root: 2 membranes, outside->symplast passive, symplast -> xylem active

Leaf: 2 membranes, xylem -> leaf symplast passive, symplast -> vacuole active

Cl Root: 2 membranes, outside -> root symplast active, symplast -> xylem passive Leaf: 2 membranes: xylem -> leaf symplast active, symplast -> vacuole passive

 $Total\ costs\ for\ Na^+ + Cl^-\ transport$ 

No leaks:

Leaf %ATP used = 7.1%

Root %ATP used (including for leaf supply) = 11.4%

% of root C used = 1.8%

% of leaf C used =0.96%

Total % plant C used = 2.76%

Leak (0.5 x net flux) at one membrane (root and leaf)

Leaf %ATP used = 14.2%

Root %ATP used (including for leaf supply) = 22.8%

% of root C used = 3.7%

% of leaf C used = 1.9%

Total % plant C used = 5.6%

*Leak* (0.5 x net flux) at two membranes (root and leaf)

Leaf %ATP used = 28.3%

Root %ATP used (including for leaf supply)= 45.6%

% of root C used = 7.4%

% of leaf C used = 3.8%

Total % plant C used = 11.2%

*Note*: These figures do not account for maintenance of membrane potential and do not include a small correction for respiration related to relative growth rate. They also do

### Energetics of Na<sup>+</sup> transport: a conundrum

Current dogma suggests acquisition of nutrient and toxic ions by plant root cells is controlled mainly by influx and efflux transporters residing in the plasma membranes. Bypass flow can occur via the apoplast in rice depending on the development of lateral roots and apoplastic barriers (Krishnamurthy *et al.*, 2011). This would be important to consider in energy calculations considering that it avoids "expensive" membrane steps, but control on fluxes is presumably more limited. The difference between unidirectional influx and efflux at the plasma membrane is the "net flux". In general, radiotracer experiments show that, as external ion concentrations increase, so do influx and efflux, while net fluxes remain low, indicating rapid cycling of ions (Britto & Kronzucker, 2006).

In the Rapid Transmembrane Sodium Cycling (RTSC) model of Na<sup>+</sup> transport in plants (Britto & Kronzuker, 2015), Na<sup>+</sup> initially enters the root cells passively down its electrochemical gradient due to a negative inward membrane potential (-80 to -120 mV), with an assumed cytoplasmic Na<sup>+</sup> concentration ([Na<sup>+</sup>]<sub>cyt</sub>) of between 10 and 30 mM (Munns and Tester, 2008), significantly less than [Na<sup>+</sup>]<sub>ext</sub> under saline conditions. Genetic variation for unidirectional influx exists, as salinity tolerant species such as *Suaeda maritima* and *Spergularia marina* have lower influxes than sensitive species like rice, wheat or Arabidopsis (Cheeseman *et al.*, 1985; Wang *et al.*, 2007). Na<sup>+</sup> then exits the cell, at nearly the influx rate, via active transport across the plasma membrane. Efflux measurements are notoriously difficult to perform accurately via radiotracer techniques (Volkov, 2015).

Several transporters that could allow Na<sup>+</sup> influx have been proposed, some only based on physiological characteristics, including voltage independent non-selective cation channels (vi-NSCCs) (Demidchik & Maathuis, 2007), AtPIP2;1 (an aquaporin as a candidate for vi-NSCCs; Byrt *et al.*, 2017), AKT1 (K<sup>+</sup> channel), LCT1 (low affinity cation transporter), HKT-type (Na<sup>+</sup> and/or K<sup>+</sup> transporters), KUP/HAK/KT (K<sup>+</sup> transporters) and CCC (cation-Cl cotransporter) dependent on location and co-transported ion gradients (Kronzucker & Britto, 2011). However, only SOS1 has been proposed as an apparent efflux transporter, although it may participate in internal Na<sup>+</sup> transport and perhaps only mediates Na<sup>+</sup> efflux from root tips where cell Na<sup>+</sup> concentrations remain low presumably to protect meristematic cells (Kronzucker & Britto, 2011). This may be genotype and Ca<sup>2+</sup> dependent (Wu *et al.*, 2015). The vi-NSCCs could allow passive efflux if the electrochemical gradients are appropriate

(McGaughey *et al.*, 2018). Future progress in development of crop cultivars with improved Na<sup>+</sup> transport traits will benefit from resolving the molecular identity of key Na<sup>+</sup> transporters.

The energy requirements for sustaining influx and efflux of Na<sup>+</sup> across the plasma membranes of root cells proposed by tracer experiments would result in a very large energy burden if it involves a Na<sup>+</sup>/H<sup>+</sup> antiporter (e.g. SOS1), because for every Na<sup>+</sup> effluxed across the plasma membrane, one H<sup>+</sup> must be extruded via the H<sup>+</sup>-ATPase. Further, for every H<sup>+</sup> extruded one ATP is hydrolysed, assuming optimal coupling to ATP (see above), and the ratio of ATP produced per O<sub>2</sub> consumed is 4.5 (Boxes 1 and 2). This would mean that, at most, 4.5 Na<sup>+</sup> can be extruded for every O<sub>2</sub> consumed in respiration. Estimates for maximal whole root respiration (expressed as O<sub>2</sub> consumption) are generally in the range of 10-25 μmol of O<sub>2</sub> g (root FW)<sup>-1</sup> h<sup>-1</sup>(Box 1). Consequently, sustaining many of the measured (or calculated) Na+ effluxes would require more ATP than produced by the entire maximal respiration of the root (Britto & Kronzucker, 2009). This problem was described initially for Cl fluxes in relation to respiration in frog epithelia and has been referred to as Ussing's conundrum (Ussing, 1947; Ussing, 1994; Britto & Kronzucker, 2009). Ussing (1994) suggested that the flux of one ion in the energetic 'uphill' direction may be coupled to the flux of another ion in the 'downhill' direction. This suggests that measurement of Na<sup>+</sup> fluxes in isolation will be prone to artefact as it does not consider the energetic effects of flux via other transporters (e.g. Cl<sup>-</sup>: Colmenero-Flores *et al.*, 2007).

Measurements of Na<sup>+</sup> influx via radiotracer techniques assume that the flux measurement is indeed trans-membrane, however, it is possible that apoplastic binding or exchange of the tracer creates an overestimate of the true Na<sup>+</sup> influx (Britto & Kronzucker, 2015; Munns et al., 2019). There are several techniques to measure Na<sup>+</sup> fluxes, each with advantages and disadvantages (Volkov, 2015). Gaining accurate measures of cytosolic ion concentrations is challenging, because of the very small volume of cytosol, yet this resolution is needed to model Na<sup>+</sup> fluxes (Foster & Miklavcic 2017). It is also clear that the contribution to Na<sup>+</sup> uptake of alternate pathways, such as 'bypass flow', and the influence of the epidermis, multiple cortical layers, and the root endodermal barriers on root radial transport, must be better understood (Yeo *et al.*, 1987; Flam-Shepherd *et al.*, 2018). Ultimately though, determination of the molecular identity of the transporters mediating Na<sup>+</sup> transport in plant roots will be required to dissect the system and provide targets for efforts to develop cultivars with superior Na<sup>+</sup> transport traits.

## Leakage of Na<sup>+</sup> across the tonoplast

Salinity tolerance of tissues is critically dependent on efficient vacuolar Na<sup>+</sup> sequestration. While the mechanisms of Na<sup>+</sup> transport into the vacuole are still a matter of debate (Bassil *et al.*, 2011), tonoplast-based NHX Na<sup>+</sup>/H<sup>+</sup> exchangers are considered likely candidates (Bassil *et al.*, 2019). However, active removal of Na<sup>+</sup> from cytosol to vacuole is only one component of vacuolar sequestration. Another often-neglected component is Na<sup>+</sup> retention in vacuoles. This retention is based on an efficient control of Na<sup>+</sup>-permeable vacuolar channels that mediate the back-leak of Na<sup>+</sup> into the cytosol and, if not regulated tightly, may result in a significant energy cost.

Two major types of Na<sup>+</sup>- permeable channels are present in the tonoplast (Isayenkov et al., 2010). The slow-activating (SV) channels are permeable to both mono- and divalent cations, and fast-activating (FV) channels that are permeable to monovalent cations only (Pottosin & Dobrovinskaya, 2018). While the molecular identity of FV channels remains unknown, SV channels are encoded by a TPC1 ('two-pore channel 1') protein (Hedrich et al., 2018). Both channels are ubiquitous and abundant in patch clamp experiments on vacuoles (Pottosin & Schnknecht, 2007; Demidchik et al., 2018). Being non-selective, these channels do not discriminate between Na<sup>+</sup> and other cations. As SV channels allow the release of K<sup>+</sup> from the vacuole, they are essential for maintaining cytoplasmic K<sup>+</sup> homeostasis (Hedrich et al., 2018). Their relative expression determines the vacuolar Ca<sup>2+</sup> storage capacity (Gilliham et al., 2011). Also, TPC1 channels play a key role in generating Ca<sup>2+</sup> waves in long-distance signalling in plants, including those under saline conditions (Choi et al., 2014). These channels may need to be open for normal cell metabolism and signalling, but salinity tolerance may require them to be closed most of the time. Model calculations (Shabala et al. this volume) show that if each cell has only one open SV channel at a given time, the backleak may exceed 305% of the influx. For in planta conditions (at resting cytosolic Ca2+ levels) the measured SV currents are ~10 pA per vacuole (Perez et al., 2008), which equates to approximately five open SV channels. This could mean that all available energy may be wasted in cycling of Na<sup>+</sup> in and out of the vacuole. Halophytes have developed an ability to reduce the number of open SV channels by several fold when grown under saline conditions (Bonales-Alatorre et al., 2013), by an unknown mechanism. This needs to be addressed in future studies to enable plant breeders to target reduced Na<sup>+</sup> cycling across the tonoplast.

### Transport of Na<sup>+</sup> and Cl<sup>-</sup> from soil to leaf

The cell-mediated pathway for Na<sup>+</sup> and Cl<sup>-</sup> transport from soil to shoot (i.e. neglecting bypass flow in the root) involves at least three plasma membrane components: 1) entry at the root symplast, 2) unloading towards the xylem apoplast, 3) re-entry at the shoot symplast; and finally crossing the tonoplast membrane for vacuolar sequestration (Fig. 4). For Na<sup>+</sup>, our present understanding of electrochemical gradients and transport mechanisms for the plasma membrane favours only a single energy-requiring step: efflux to the xylem apoplast, notwithstanding the secondary effects on other ions caused by passive Na<sup>+</sup> entry, such as depolarisation of the membrane potential and requirement for charge balance (Fig. 5a). For root-to-shoot Cl<sup>-</sup> transport, a minimum of two energy-requiring plasma membrane transport steps are required, at least under the normal paradigm that Cl entry is via proton cotransport, though it is possible for Cl to enter the symplasm passively at high external Cl concentrations depending on the membrane potential (Teakle and Tyerman 2010). These steps include: 1) uptake at the root symplast, and 2) re-uptake at the shoot symplast (Fig. 5a). In reality, it is possible that a multitude of Na<sup>+</sup> and Cl<sup>-</sup> pathways exist; e.g. Na<sup>+</sup> storage may occur in xylem parenchyma cells as a consequence of HKT mediated transport (Munns et al., 2012).

### Entry into the root symplast

Passive influx of Na<sup>+</sup> from the soil solution into the cytosol of root epidermal and cortical cells (Cheeseman, 1982) (Fig. **5a**) is likely achieved through non-selective cation channels (NSCC) (Tyerman *et al.*, 1997; Davenport & Tester, 2000; Demidchik & Tester, 2002). Subsets of plasma membrane water channels such as the Arabidopsis PIP2;1 and PIP2;2 aquaporins, which are abundant in root epidermal cell plasma membranes and are permeable to Na<sup>+</sup>, have been proposed as candidates for NSCCs (Byrt *et al.*, 2017; Kourghi *et al.*, 2017); these aquaporins qualitatively and semi-quantitatively match the NSCC properties as measured in isolated root protoplasts by patch-clamp (McGaughey *et al.*, 2018).

Chloride can enter the plant root symplast through secondary active transport either coupled to H<sup>+</sup> uptake or, at high salinity, passively due to an initial membrane depolarisation (Skerrett & Tyerman, 1994; Lorenzen, Inken *et al.*, 2004; Saleh & Plieth, 2013) (Fig. **5b**). Cl<sup>-</sup>:H<sup>+</sup> symport is likely to occur across only one membrane, which may be plasma membranes of root hairs, epidermal cells, or cortical cells (Fig. **5a**). Symport of 1Cl<sup>-</sup>:2H<sup>+</sup> has been

identified in plant root hairs (Felle, 1994), and the expression of *ZmNPF2.6* – which encodes a protein with 1Cl<sup>-</sup>:2H<sup>+</sup> activity – was detected in the plasma membranes of maize root epidermal and cortical cells (Wen *et al.*, 2017).

Once in the symplast, radial Na<sup>+</sup> and Cl<sup>-</sup> movement towards the xylem may occur through plasmodesmata, with no membrane steps required. However, cycling in and out of cells and vacuoles along the way may occur.

### Xylem loading and retrieval

Sodium transport to the xylem is an energy requiring step, most likely coupled to H<sup>+</sup> antiport, and could occur via a Na<sup>+</sup>-H<sup>+</sup> exchange with the acidic apoplast. Numerous candidate proteins might catalyse this Na<sup>+</sup> transport step in plants, for example SOS1 and CHXs (Shi *et al.*, 2003; Y. Qu *et al.*, unpublished; Zhu *et al.*, 2017) (Fig. 5a).

Chloride entry to the xylem apoplast is channel-mediated by SLAH1 and SLAH3 (Cubero-Font *et al.*, 2016). Expression of *SLAH1* and *SLAH3* is down regulated by salt and ABA, and a*slah1* mutant has reduced Cl<sup>-</sup> in shoots under NaCl treatment compared to wildtype (Cubero-Font *et al.*, 2016; Qiu *et al.*, 2017). While no energy *per se* is required for this step (Fig. **5a.b**), energy will be required for maintenance of membrane potential, because high efflux of Cl<sup>-</sup> will depolarise the membrane. Aluminium activated malate trasnporters (ALMT) that can function as Cl<sup>-</sup> channels may also release anions to the xylem (Li *et al.* 2017b).

Retrieval of Na<sup>+</sup> from the xylem is mediated by HKT1 type transporters (Davenport *et al.*, 2007; Munns *et al.*, 2012; B. Xu *et al.*, unpublished). Some HKT proteins have been shown to function as Na<sup>+</sup>/K<sup>+</sup> symporters at micromolar Na<sup>+</sup> concentrations. However, at millimolar Na<sup>+</sup> concentrations, as would be expected in the xylem apoplast under salt stress, HKT proteins display channel-like Na<sup>+</sup> uniporter activity (Munns *et al.*, 2012; Henderson *et al.*, 2018). Thus, xylem retrieval of Na<sup>+</sup> is likely to be passive under salt stress (Fig. **5b**). The Na<sup>+</sup> retrieved by HKT1 is effluxed from the upper part of the root (Davenport *et al.* 2005).

Re-entry to the shoot symplast and movement into the mesophyll cells

The pathway for Na<sup>+</sup> entering the shoot symplast from the xylem apoplast and then moving into mesophyll cells could be via NSCCs, perhaps Na<sup>+</sup> permeable aquaporins in the plasma membrane of bundle sheath cells (Shatil-Cohen *et al.*, 2011; Sade *et al.*, 2014) (Fig. **5a,b**). Chloride re-enters the shoot symplast from the xylem apoplast at the plasma membrane of bundle sheath cells via secondary active transport. The maize 1Cl<sup>-</sup>:2H<sup>+</sup> symporter *ZmNPF2.6* 

shows preferential expression in bundle sheath cells (Wen *et al.*, 2017), suggesting that a NPF protein family member catalyses this active Cl<sup>-</sup> influx in leaves. Once in the shoot symplast, Cl<sup>-</sup> can move into mesophyll cells through plasmodesmata where it can accumulate to high concentrations, yet accumulates, preferentially, to much higher concentrations in the epidermis in moderately salt-tolerant barley (Dietz *et al.*, 1992; Fricke *et al.*, 1994; Fricke *et al.*, 1996).

Although passive Na<sup>+</sup> and Cl<sup>-</sup> transport is considered to be energy-free, the electrochemical gradient that drives passive ion flux is generated by an energy dependent mechanism (the H<sup>+</sup>-ATPase). Hence changes in the functioning of the H<sup>+</sup>-ATPase under salinity would influence both passive and active Na<sup>+</sup> and Cl<sup>-</sup> transport. Osmotic stress, including from NaCl, increases the coupling ratio (that is more protons efflux per ATP hydrolysed) of the H<sup>+</sup>-ATPase (Kerkeb *et al.*, 2002; Janicka-Russak *et al.*, 2013). This would decrease the energy requirement for proton mediated secondary active transport without increasing the inward gradient for passive Na<sup>+</sup> entry.

Is transmembrane Na<sup>+</sup> cycling a key determinant of cell energy use?

Excessive cycling of Na<sup>+</sup> and that of other ions in root cells of glycophytes is one of the most puzzling observations in plant transport physiology. Even if part of the tracer Na<sup>+</sup> uptake (and release in washout studies) reflected exchange with binding sites in the root apoplast (see above discussion in Energetics of Na<sup>+</sup> transport: a conundrum), there is evidence that a considerable fraction of Na<sup>+</sup> is indeed taken up by root cortical cells and is subsequently released again, either at the same cell or further up the root after the Na<sup>+</sup> is withdrawn from the xylem (Davenport et al. 2005). The energetic costs of this cycling, if confirmed, calls for an investigation into the benefit the plant could potentially draw from such a process. It remains an intellectual challenge to understand why transport proteins allowing Na<sup>+</sup> entry into the root symplast exist in the plasma membrane, when salt exclusion is an important strategy to avoid saline stress in glycophytes. Earlier studies showed that salt starved roots ("low salt roots") readily take up Na<sup>+</sup> to generate turgor when there is essentially no other choice (Pitman et al., 1968), and obviously transport proteins are needed for this contingency. Since cells can express and alter their transporters to suit the conditions, the existence of a transport pathway for Na<sup>+</sup> entry does not mean that it is always operating. A contender for the NSCC, the aquaporin PIP2;1, is rapidly removed from the plasma membrane upon a salinity

stress (Boursiac *et al.*, 2008), perhaps because of its ability to transport Na<sup>+</sup>. However, this unavoidably compromises water transport and potentially K<sup>+</sup> transport, since PIP2;1 can also transport K<sup>+</sup> which is strongly linked in a broader sense to generation of turgor in plant cells (Qiu *et al.*, 2018).

It has been proposed that Na<sup>+</sup> influx via aquaporins may provide the driving force for simultaneous water uptake, via the same transport protein, against the water potential gradient (which may favour water efflux instead). This would require a close coupling of water and solute transport, allowing a transfer of free energy between the fluxes, which can be tested experimentally. Non-osmotic water transport may play a role in the generation of root pressure, repair of embolized xylem vessels, and control of cell elongation (Fricke, 2015; Wegner, 2015; Wegner, 2017). This mechanism implies large circular fluxes of the solute(s) driving the water transport, since solutes permeating together with a fixed number of water molecules need to be re-translocated back immediately at the expense of metabolic energy to maintain the (electro)chemical gradient. This implies that the seemingly futile cycling of Na<sup>+</sup> and K<sup>+</sup> has a physiological function rather than being just a waste of energy. Evidence in support of this hypothesis has been expounded previously (Wegner, 2017). It should be noted that the osmotic and non-osmotic water uptake mechanisms proposed by Wegner (2017) are mutually exclusive, at least at the cellular level, since water fluxes across the plasma membrane are facilitated by a high activity of aquaporins, which would short-circuit non-osmotic water uptake, rendering it unfeasible from an energetic point of view (Fricke 2017, Wegner 2015, 2017). These considerations may help to understand the complex regulation of aquaporin activity under salt stress (McGaughey et al. 2018).

### IV. Root anatomy and transport pathways

Pathways through root systems impact on the energy needed for plant salt tolerance. Plants must exclude nearly all the salt in the soil solution while taking up water, and maintain a low net rate of Na<sup>+</sup> and Cl<sup>-</sup> uptake (Munns et al., 2019 this issue). Here we assess three aspects that affect the ability of roots to exclude salt while taking up water, without exhausting the energy budget of the plant.

### Root systems and the anatomies of root types

The most important aspect of roots compared to shoots is that the stele tissues (xylem, phloem, pericycle and parenchyma cells) are internal, surrounded by a cortex (the innermost layer of which is known as the endodermis) and an epidermis (**Fig.** 6 a to d). Epidermal cells may differentiate into root hairs (Fig. 6e). Shoots have stele tissues distributed throughout parenchyma cells. Evidence has emerged that the root cortex must do the heavy lifting of excluding Na<sup>+</sup> from shoots (Munns *et al.*, Insight). Within this context, the large differences between cortex and stele anatomies are intriguing (Varney *et al.*, 1991; Watt *et al.*, 2008; Watt *et al.*, 2009). Large variation depends on root type (axile versus branch types), age, and soil (Figure **6a** to **d**).

The single root model underlies our current framework for salinity tolerance mechanisms (e.g. Fig. 5). This is generally an axile root from the embryo (seminal, Fig. 6a) or stem (nodal, Fig. 6d); branch roots are rarely considered (e.g. Faiyue *et al.*, 2012). The single root model greatly underestimates pathways for salts and water from soil to shoot, based on distances and cell sizes. In wheat, within 10-d of germination, the plant develops a system of different root types with branch roots (termed lateral or fine roots) that have emerged from an axile root (Fig. 6a or d). The complexity of the system increases with time. By the time of flowering, wheat roots below the topsoil can be 90% branch roots with the fine structure and anatomy shown in Fig. 6c (Watt *et al.*, 2008).

Soil conditions, including high salinity, strongly influence allocation between axile and branch roots (Rich and Watt, 2013). Durum wheat root systems were studied in a gradient of salinity, to mimic distribution under natural soil conditions (Rahnama *et al.*, 2011). Seminal axile root lengths in saline gradients were ca. 25% *less* than those of the control, while branch length was ca. 500% *greater*. The consequences of shifts to different root types could be large in terms of anatomy: in the saline gradient, ca. 26% of total root length shifted to finer branch roots (e.g. Figs. **6b, c**). Further the branch roots emerged much closer to the axile tip (ca. 3 cm in saline conditions compared with ca. 20 cm in non-saline).

Salinity inhibits cell division in the primary roots of species including wheat and barley (Rahnama *et al.* 2011; Shelden *et al.*, 2013). Branch root initiation and extension was uninhibited by external salt (Rahnama *et al.* 2011). Decreasing primary root length and allocating energy to the initiation of lateral roots may be linked to adaptation to salinity, a mechanism which was also seen by Zolla *et al.* (2010) in Arabidopsis. Branch roots arise

from pericycle cells and water for elongation may come from the phloem (Boyer *et al.* 2010). Salinity can promote differentiation of underlying xylem tissues in cotton (Reinhardt and Rost, 1995) and aging of roots in tomato (Snapp and Shennan, 1992). Taken together, these studies suggest that anatomical pathways across a root system may change upon exposure to saline soil compared to non-saline conditions (Fig. **6f**).

Importance of root system anatomical differences: estimating ATP costs of cortical cell layers

What could be the energy consequences of changing root anatomy on Na<sup>+</sup> transport costs? Here we compare scenarios of roots with single or double cortical cell layers to predict if root types may have different energy costs for Na<sup>+</sup>, K<sup>+</sup> and Cl<sup>-</sup> transport. An Arabidopsis model (Foster & Miklavcic, 2017) was applied to the simplest branch roots of wheat (Fig, **6c**). Wheat fine roots share similarities with Arabidopsis roots (image in Fig. **6c** inset), except the wheat root tested had two cortical cell layers, while the Arabidopsis root had one.

The modelling predicts that a higher cortical cell layer number is associated with increased energy requirements for ion transport in salt-stressed roots (Fig. 6g). A modelled wheat root with only one cortical layer would use 20% less energy (per unit of root length) for ion transport than a root with two cortical cell layers. The plasma membrane surface area of the cells outside the endodermal apoplastic barrier correlated reasonably with the energy cost (Fig. 6h). The number of cells outside the outermost apoplastic barrier correlated with the energy cost better than the total root perimeter (or surface area).

The model used here presents one consequence of root system architecture differences under saline conditions (Fig. 6f). Anatomical changes to the root system to cope with salinity may lower energetics of ion movements across membranes that require ATP. The importance of living cell number to root energetics has been shown by Lynch and colleagues using modelling and measurements on plants exposed to low nutrient supply (e.g. Schneider *et al.*, 2017). Here we simulated fewer and smaller living cortical cells, not cortical cell senescence. Future studies are needed to test salinity stress tolerance implications of fewer cells with greater apoplasmic spaces.

Cell specialisations at the epidermis including hair growth and functions

The energy costs of the cell-to-cell pathway (plasma membrane and tonoplast transport of Na<sup>+</sup>, K<sup>+</sup> and Cl<sup>-</sup>) could be high, especially in a nodal axile root with 11 cortical cell layers and a seminal axile root with 6 layers (cp. Fig. **6d** with **a**). This prompts consideration of other possible mechanisms by which roots could prevent Na<sup>+</sup> reaching shoots.

The plasma membrane of epidermal cells contains Na<sup>+</sup> transporters such as HvHKT1;1 in barley (Han *et al.*, 2018), and SOS1 in Arabidopsis (Shi *et al.*, 2002). This suggests that epidermal cells sense and have a role in excluding salt from the roots. Confocal microscopy was used to monitor the distribution of Na<sup>+</sup> using an indicator dye, and fluorescence was higher in the vacuole of the salt-tolerant variety, a finding that was supported by a much lower cytosolic to vacuole ratio of Na<sup>+</sup> (Cuin *et al.*, 2011; Wu *et al.*, 2018). We estimated the effect of an apoplastic barrier in the epidermis on costs of ion transport in salt-stressed roots (Fig. **6g**). An apoplastic epidermal barrier reduced the energy costs by 18% and 10% for the two and one cortical cell layer roots, respectively. These results indicate that an epidermal barrier could lead to larger energy savings in roots with more cortical layers.

Root hairs (Fig. **6e**) have a larger surface to volume ratio than cortical cells and may capture most of the salt convected with water, retaining more salt from the soil solution than underlying cortical cells. Root hairs facilitate water uptake from soil at high transpiration rates (Carminati *et al.*, 2017). Root hairs were ca. 50% shorter in response to salinity in barley, which is typical of studies conducted with seedlings in hydroponics without soil (see Shabala *et al.*, 2003). Root hair cells can develop specialisations (Tyerman *et al.*, 1989), similar to modified trichomes on the aerial tissue of halophytes, known as epidermal bladder cells (EBC). EBC cells accumulate Na<sup>+</sup> in excess of that measured in mesophyll cells (Adams *et al.*, 1998; Barkla *et al.*, 2002). Interestingly, root hairs of the grass sorghum sequester compounds toxic to the cytoplasm in vacuoles and release them to soil by vesicular transport (Weston *et al.*, 2012). We speculate that root hairs may package salt into vesicles that are moved to the surface and extruded to the soil (Fig. **6e**). Future studies of root hair responses and mechanisms of sequenstering salt should be tested in soil and on different root types (Nestler *et al.*, 2016).

### Membrane dynamics avoid transmembrane crossings

Traditionally, the cellular view is static, with cells requiring mechanisms to keep salt out of the cell or sequester into the vacuole. Improved imaging techniques in recent years show constant, highly dynamic construction and de-construction of membranes in contact with the apoplast (walls). The vacuole appears as a flexible bubble, widening and narrowing the space between the tonoplast and plasma membrane (Picco *et al.*, 2015). This provides the opportunity for direct contact of the two membranes and the possibility of a direct "shunt" for ions to enter the vacuole either without crossing through the cytosol or crossing via a cytosolic micro domain with different conditions (Flowers *et al.*, 2018).

Such membrane contact sites have been identified between other organelles within a distance of 10 to 30 nm (Wu *et al.*, 2018). Evidence for direct channelling of ions between the endoplasmic reticulum (ER) and plasma membrane, and between ER and mitochondria (Prinz, 2014), opens the possibility for Na<sup>+</sup> ions to be transported in this manner. Once in the organellar lumen, the ions can be trafficked to the vacuole from the ER or pass through the Golgi and *trans*-Golgi network (TGN) to pre-vacuolar compartments and finally to the vacuole (Hönscher *et al.*, 2014). These routes exclude the energetically costly cytosolic import and export steps. The ER of two cells is usually connected through the plasmodesmata (Thomas *et al.*, 2008), providing an avenue for ion movement between cells in the lumen without transport across membranes, followed by sequestration in the vacuole. We speculate that the energy costs of this movement could be lower than crossing vacuolar and plasma membranes (4 crossings per cell).

In summary, root systems display multiple and flexible options to adjust energetics of salinity tolerance using anatomy and cell differentiations. An additional option to Fig. 4 is multiple pathways with different modes of transport (and energetics) in space and time within a plant.

### V. Where to from here?

Which of the many transport processes consume the most energy, and at what time and where in the plant? These questions are difficult to answer by experimental study alone. However, achieving a detailed, quantitative understanding of this complex whole-of-plant salt stress response is possible if physiological studies are accompanied by computational/biophysical

modelling (Foster & Miklavcic, 2017). Biophysical models could be adapted to incorporate energy fluxes and used to identify transporters that may be engineered to achieve improved energy efficiency in roots. The effect of endodermal barriers on ion uptake and energy costs of ion transport through the root has been quantified by models (Foster & Miklavcic, 2016; 2017). Models of salt and water transport in a single plant cell (Foster & Miklavcic, 2015), and OnGuard software which models guard cell transport, homeostasis and metabolism (Chen et al., 2012; Hills et al., 2012; Wang et al., 2017), could be adapted to calculate energy costs of salt tolerance. For example, analogous biophysical modelling would inform us of the transport mechanisms with the highest energy demands as well as allow us to identify transport steps as targets to minimise these energy costs while still maintaining conditions compatible with salt tolerance. In addition, biophysical modelling has the potential to answer questions on the quantity of ATP available. For example, existing models of transport in isolated plant cells combined with models of ATP production in animal cells (e.g. Bertram et al., 2006) could be adapted to quantitatively model ATP production in plant mitochondria. Modelling can also be used to investigate currently unproven hypotheses, such as the possibility of water co-transport with ions such as Na<sup>+</sup> (Wegner, 2017). Although existing models of ion and water transport in plants assume that water transport is passive, they could be adapted to explore the possibility of active water transport.

A potentially important component to include in models of energy use is the signalling that occurs in response to salt stress, including involvement of transporters, that leads to changes in gene expression, growth and remodelling of root systems (Julkowska & Testerink, 2015), which ultimately affects the energetics. Another aspect not dealt with here is the costs of altering the proteome since protein synthesis can consume a large amount of energy; in leaves it could consume 25% of the ATP produced by respiration (Li *et al.*, 2017a). Collaboration between theoreticians and experimentalists is needed to answer the questions raised here. This may address many of the gaps in our knowledge of salt tolerance mechanisms in plants. Accurate estimation of energy costs associated with salinity tolerance may provide a new approach to long standing problems - by the coordinated measurements of transport activities and root respiration on the same tissue under the same conditions. This, together with identification of the molecular components that can be modified to increase the amount of energy available for harvestable yield (Amthor *et al.*, 2019), may provide a new approach to crop breeding for saline soils.

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## **Supporting Information**

**Notes S1**: Net stoichiometry and ATP yield of sucrose oxidation with and without alternative oxidase, and energy budget for photosynthesis.

**Fig. 1** Field observations of salinity. **(a)** Barley growing in naturally salt-affected field near Corrigin, Western Australia. Photo courtesy Stuart Roy. **(b)** Typical variation in soil salinity spatially and at depth. Measurements with EM38 ground conductivity meter show the apparent electrical conductivity (EC<sub>a</sub>; mS m<sup>-1</sup>) in a farmer's paddock at two soil depths (0-50 cm and 0-100 cm) at Whitwarta, South Australia. The area shown is 0.65 ha (about 80 m<sup>2</sup>). Red = low EC<sub>a</sub> and blue = high EC<sub>a</sub>. Figure modified from Asif *et al.* (2018) under the terms of the Creative Commons Attribution 4.0 International License.

**Fig. 2** The reactions and enzymes of the plant mitochondrial tricarboxylic acid (TCA) cycle, GABA shunt and mitochondrial electron transfer chain (mETC) linking matrix NAD<sup>+</sup> and NADH pools and ATP production. The GABA (γ-amino butyric acid) shunt bypasses the oxoglutarate dehydrogenase complex catalysed step of the TCA cycle. The enzymes and complexes involved cytochrome pathway (Cyt Path) respiration (red), alternative pathway (Alt Path) respiration (green) and Complex V (CV) ATP production (light blue) and the import and export of its reactants and products (dark blue) by the adenine nucleotide translocator (ANT) and phosphate carrier (PiC) located on the inner mitochondrial membrane. This links to the pools of reductants (R) such as NADH and oxidants (O) such as NAD<sup>+</sup> that are recycled by the matrix TCA cycle or other alternative dehydrogenases (Alt DH). The GABA shunt (purple) includes transport steps across the mitochondrial membranes and enzyme reactions in both cytosol and matrix,, which can play a part in bypassing parts of the TCA cycle that become inhibited under stress (Che-Othman *et al.*, 2019).

Fig. 3 A variety of roles for the vacuolar proton-pumping pyrophosphatase. A generic plant cell showing the variety of ways the vacuolar proton pumping pyrophosphatase (H<sup>+</sup> -PPase) can provide an alternative source of energy during salinity stress. (1) Vacuolar acidification. Localised to the tonoplast, the vacuolar H<sup>+</sup>-PPase (blue) will use energy released from the hydrolysis of PPi to orthophosphate (Pi) to pump protons (H<sup>+</sup>) into the vacuole. Along with vacuolar ATPases (purple), vacuolar H<sup>+</sup>-PPases establish an electrochemical potential for H<sup>+</sup> across the tonoplast, which is used by other vacuolar transporters (red & brown) to sequester Na<sup>+</sup> and Cl<sup>-</sup> into the vacuole. (2) Removal of inhibitory pyrophosphate (PPi). Vacuolar H<sup>+</sup>-PPases regulate PPi levels in the cytosol. Accumulation of PPi in the cytosol, particularly in younger tissues, can inhibit PPi dependent metabolic pathways, such as gluconeogenesis and the Smirnoff-Wheeler pathway. (3) Enhancing sucrose transport from source to sink tissues. In phloem companion cells, H<sup>+</sup>-PPases are shown to be localise to the plasma membrane, where it hypothesised they synthesises PPi from orthophosphate. This additional PPi is used to enhance sucrose metabolism in these cells, thereby generating more ATP to pump protons into the apoplast which can be used by sucrose transporters, ultimately enhancing sucrose transport into sieve elements. Not all of these processes will be occurring in all cells at all times, and some may be cell type specific. Figure modified from Khadilkar et al., (2016) and Schilling et al., (2017).

**Fig. 4** Five possible ways to calculate the energy associated with the movement of solutes from a soil location to target location in leaf. The options shown are not exhaustive. For details, see text. Explanation of symbols and abbreviations: red circle with X, solute species X;  $\mu_X S$ , chemical potential of solute X in compartment S (soil); other compartments are leaf (L) or cells (C) at the start (C1) or end (Cn) of transport path of solute; yellow symbol, energy associated with providing cellular infrastructure for transport, such as synthesis and regulation of transporter protein;  $H^+$ , proton pumping through the plasma-membrane-localised  $H^+$ -ATPase; blue circular lines, tonoplast depicting the large central vacuole. The picture shows a 16-d old barley plant.

Fig. 5 Active and passive membrane steps for the long distance transport of Na<sup>+</sup> and Cl<sup>-</sup> in plants under high and low salt concentrations. (a) Under low salt, Cl<sup>-</sup> enters the root and shoot symplast actively through a 1Cl-:2H+ symporter (e.g. ZmNPF2.6), and is loaded towards the xylem by channel-mediated passive transport (e.g. SLAH1/3 heteromer). Na<sup>+</sup> enters the symplast passively via non-selective cation channels (NSCC), and is effluxed via active proton exchange (e.g. through CHX or similar mechanism). (b) Under high salt, Cl<sup>-</sup> may enter the root symplast by passive transport, while passive Cl<sup>-</sup> efflux to the xylem (though SLAH1/3 heteromer) is downregulated at the transcriptional level. Na<sup>+</sup> exclusion from the root xylem into the symplast is likely to be mediated passively by HKT proteins. Apoplastic barriers in the root (not shown) are important to restrict by-pass flow of NaCl to the xylem. As ion exclusion is controlled mostly at the root level, it is expected that reuptake into the shoot symplast is similar under low and high salt. Cl will accumulate preferentially in leaf epidermal cells (Leigh & Tomos, 1993), but the mechanism for this remains unclear. A final transport step in to the vacuole of leaf cells would involve active transport of Na<sup>+</sup> and passive or active transport for Cl<sup>-</sup> (Fig. 3). RH = Root Hair; Epi = Epidermis; Cor = Cortex; XP = Xylem Parenchyma; BS = Bundle Sheath; Mes = Mesophyll; Vac = Vacuole. Grey rectangle represents the Casparian band.

Fig. 6 Changes in anatomy, root system and root cells that may be important in the energetics of salinity tolerance. (a-d) Cross sections of wheat roots, all at same magnification. Inset of (c), Arabidopsis primary root cross section, shown at same magnification as wheat 2nd order branch root (bar = 50 micrometres, from Sotta and Fujiwara, 2017). (e) Speculation about importance of root hairs as an epidermal barrier to Na<sup>+</sup> movement into the root. Left. Root hair of nodal root in (d) outlined to indicate surface area with soil. Right, Blow up of hair tip with hypothesized, drawn transport of vesicles (yellow) delivering Na<sup>+</sup> to the outside of the cells (bar = 25 micrometres). Speculation and drawing based on the root hairs of sorghum, which can transport sorgoleone to the surface and to the soil in vesicles (see the Cell specialisations at the epidermis including hair growth and functions subsection). (f) Schematic view of influence of salinity on a root system. Events: (1) shortening of primary root; (2) increased 1<sup>st</sup> order branch root length; (3) branch root and xylem maturity closer to the tip; (4) increased rate of root aging. See text for references to original research for these events. (g) Modelled effects of either one or two cortical cell layers, and presence of an epidermal barrier, on energy costs of transmembrane transport. (h) Positive relationship between plasma membrane surface area outside an apoplastic barrier and energy cost. The

Foster & Miklavcic (2017) model root geometry was adapted to simulate wheat roots with one or two cortical layers. For all simulations, the external medium contained 100 mM NaCl, and a hydraulic pressure of -0.3 MPa was assumed at the top boundary of the root. The remaining simulation conditions were as described in Foster & Miklavcic (2017) for the non-uniform transport scenario. c – cortex; PM – plasma membrane.

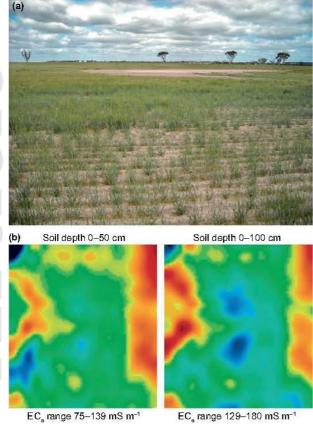


Figure 1
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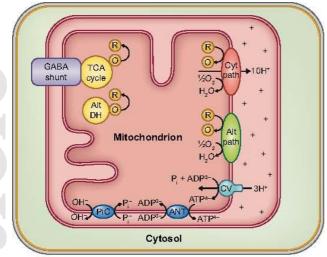


Figure 2

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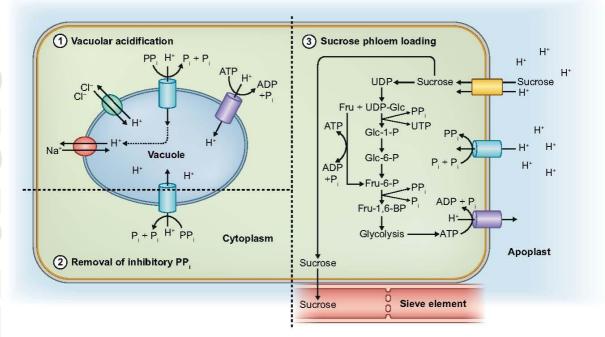


Figure 3
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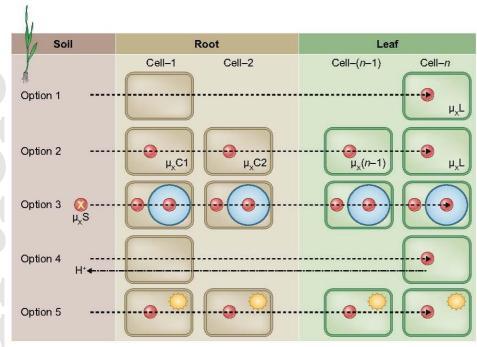


Figure 4
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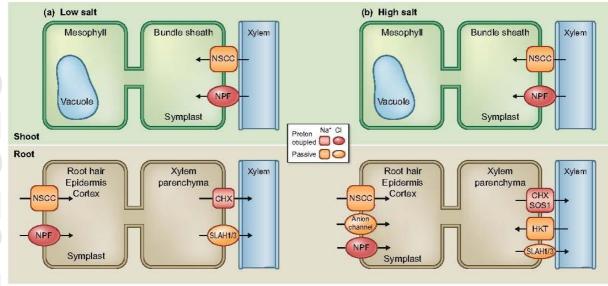


Figure 5

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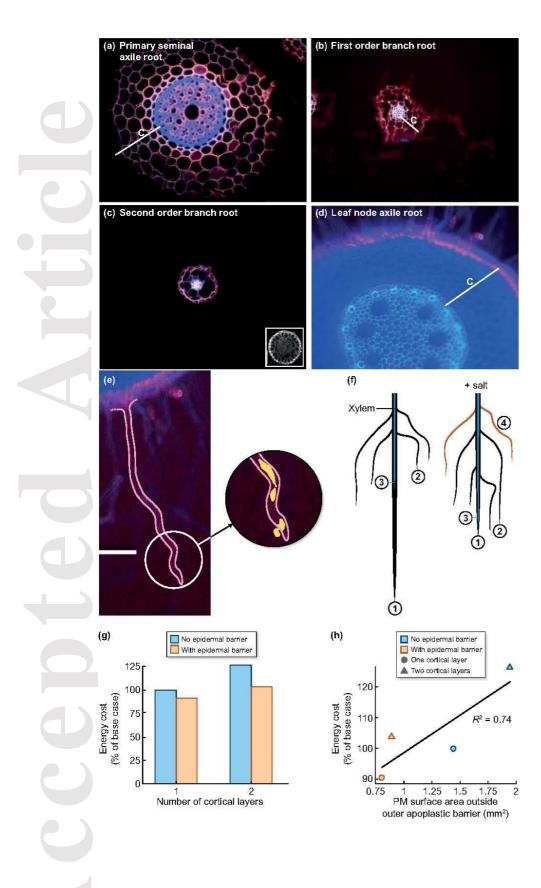


Figure 6
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