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Review Article

Engineered Nanomaterials: Knowledge Gaps in Fate, Exposure, Toxicity, and Future Directions

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The aim of this study is to identify current knowledge gaps in fate, exposure, and toxicity of engineered nanomaterials (ENMs), highlight research gaps, and suggest future research directions. Humans and other living organisms are exposed to ENMs during production or use of products containing them. To assess the hazards of ENMs, it is important to assess their physiochemical properties and try to relate them to any observed hazard. However, the full determination of these relationships is currently limited by the lack of empirical data. Moreover, most toxicity studies do not use realistic environmental exposure conditions for determining dose-response parameters, affecting the accurate estimation of health risks associated with the exposure to ENMs. Regulatory aspects of nanotechnology are still developing and are currently the subject of much debate. Synthesis of available studies suggests a number of open questions. These include (i) developing a combination of different analytical methods for determining ENM concentration, size, shape, surface properties, and morphology in different environmental media, (ii) conducting toxicity studies using environmentally relevant exposure conditions and obtaining data relevant to developing quantitative nanostructure-toxicity relationships (QNTR), and (iii) developing guidelines for regulating exposure of ENMs in the environment.

1. Introduction

Production and demand of engineered nanomaterials (ENMs) embedded in consumer products are growing significantly in the current years [1, 2]. ENMs can be directly released into the air, water, sediment, and soil media during their manufacturing, use, and disposal [3–9]. Nanosize particles are also sometimes unintentionally formed, such as those arising from the combustion of fossil fuels in motor vehicles and industries [3, 4]. In this review, ENMs refer to particulate materials having at least one of three dimensions in the 1–100 nm size range (nanoscale). In

contrast, nanoparticles are particulate materials with all three dimensions in the nanoscale.

ENMs are currently considered as an emerging class of environmental contaminants. Recent research advances have (i) improved analytical capabilities for detecting them in different environmental media, (ii) increased availability of related toxicity data, and (iii) increased public awareness. Nevertheless there are still many open questions that need to be answered in order to fully understand their origin, fate, and toxicity.

ENMs pose risk to human health and the environment via oral, inhalation, and dermal exposure routes [10–15].

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Table 1 summarises the key findings of recent studies related to their exposure assessment. These studies were selected to provide information about research trends in the area of risk assessment of different ENMs by means of different exposure routes. Most of these studies have focused on the framework development for (i) estimating exposure doses and risks from oral and inhalation exposure routes, (ii) prioritising nanomaterials for monitoring and risk estimating purposes, and (iii) identifying research strategies for intelligent testing of ENMs for nanosafety. Furthermore, results of toxicity studies suggest that ENMs can cross the cell membrane and induce toxic effects on different organs [16-19]. For example, TiO₂ nanoparticles have been observed to induce DNA and chromosomal damage in the liver [13, 20]. These effect-based studies also indicate that ENMs exert different toxicities in a particular target organ depending on their properties. This highlights the need for further investigations related to the exposure of ENMs.

To estimate the environmental and human health risks associated with ENM exposure, information about their size, shape, and the dose-response relation is required [37, 44]. The methodology suggested by the standard United States National Research Council [45, 46] proposes four steps for performing environmental and human health risk assessment (EHHRA) of chemicals and metals: (i) hazard identification, (ii) exposure assessment, (iii) dose-response assessment, and (iv) risk characterisation. A few studies have attempted to apply some of these steps for estimating risks of exposure to ENMs (Table 1). However, very few studies have carried out exhaustive risk assessment process for ENMs, probably due to the lack of required information on their fate, exposure, and toxicity.

The aims of this article are to highlight present knowledge gaps in environmental fate, exposure, and toxicity of ENMs. The suitability of currently available toxicity data in conducting EHHRA is also analysed. Issues identified in this study can help in developing efforts for addressing current data gaps for ENMs which can aid decision makers in the policy making process.

2. Scope

There are two main sources of nanomaterials in the environment: natural and anthropogenic. Natural sources include but are not limited to volcanoes, viruses, ocean spray, dust storms, bacteria, and bush fires. Anthropogenic sources fall into two categories: unintentionally produced nanomaterials such as those from combustion aerosols, particularly motor vehicle exhaust emission, coal fly ash, and wielding operations and intentionally produced nanomaterials with tailored properties. These may include nanowires, nanotubes, quantum dots, and fullerenes, mostly composed of metals and metal oxides [47]. In addition, nanoparticles can be incidentally formed in both of these categories [38]. This review focuses on four important aspects of the intentionally produced group of ENMs. These include (i) characterisation and application of ENMs, (ii) assessment of exposure to ENMs, (iii) hazard assessment, and (iv) environmental regulations. A literature review of published reports and journal articles is compiled

to map the existing knowledge and to identify current gaps. The reviewed studies include only key literature to highlight the covered topics and do not represent the complete list of all published studies so far due to the reasons of brevity. The concluding section presents issues on different aspects and future research needs.

3. Characterisation and Application of Nanomaterials

3.1. Characterisation of ENMs. ENMs can be characterised using different techniques. Microscopy, in particular, has been exceptionally useful in providing information about the size and morphology of nanomaterials [48]. Scanning and transmission electron microscopy (SEM and TEM, resp.), atomic force microscopy (AFM), scanning probe microscopy (SPM), and scanning tunnelling microscopy (STM) have all been extensively employed to observe morphological, compositional, and structural features of a wide range of nanomaterials. SEM is a versatile technique that can determine the morphology whilst TEM can be used for the observation of the finest details of the internal structure of materials [49]. High resolution images of the morphology and topography of a sample can be obtained and compositional analysis of a material can also be carried out by monitoring secondary X-rays produced by the electron-specimen interaction.

A number of other methods are also available for the characterisation of nanomaterials in air [50]. These include their characterisation in terms of particle concentration and size distribution by the condensation particle counter (CPC), differential mobility particle sizer (DMPS), differential mobility spectrometer (DMS), fast mobility particle sizer (FMPS), electrical low pressure impactor (ELPI), and scanning mobility particle sizer (SMPS) [50, 51], as discussed in Section 4. The Brunauer, Emmett, and Teller (BET) method is commonly used to evaluate the gas adsorption data and generate information about specific surface area and pore size of various types of nanoparticles [52]. Thermogravimetric analysis (TG) is conveniently used to determine physical changes in materials. This technique provides quantitative measurement of mass change associated with transition and thermal degradation in the materials [53]. Characteristic thermogravimetric curves are given for specific materials and chemical compounds due to unique sequence from physicochemical reactions occurring over specific temperature ranges and heating rates. Vibrational spectroscopy techniques, including the Fourier transform infrared (FTIR), the near-infrared, and the Raman spectroscopy, have also been used to characterise nanomaterials. For example, López-Lorente et al. [54] recently described the use of G-/D-band intensity ratios in the Raman spectroscopic measurements for the determination of the purity of carbon nanotube samples. In addition, X-ray diffraction is currently being used for determining crystallinity of nanomaterials [55]. This information is especially important in toxicity studies as mineralogical phase of nanomaterials can also influence toxic effects [56].

Table 1: Sample summary of ENM-related human effects and/or risk assessment studies.

Carempel [2] Inhalation ENMs considered 22] Inhalation — 22] Inhalation — 22] Inhalation and dermal — Inhalation and dermal — Inhalation Particles less than 2.5 μm in diameter [11] Inhalation Particles less than 2.5 μm in diameter [11] Inhalation Particles less than 2.5 μm in diameter [11] Inhalation Particles less than 2.5 μm in diameter [12] Inhalation Fullerenes [26] Inhalation Fullerenes [27] Inhalation Fullerenes [28] NA Fullerenes [29] NA Fullerenes [31] Inhalation TiO₂ and Ag [32] Inhalation Fullerenes [33] Inhalation Fullerenes [34] Oral TiO₂ and Ag [35] Inhalation ENMs [36] Inhalation ENMs [37] NA ENMs <th></th> <th></th> <th></th> <th></th>				
Inhalation	Reference	Exposure route	ENMs considered	Study focus
Inhalation	Morgan [21]	NA	I	Development of preliminary framework for risk analysis and risk management of ENMs using expert elicitation and mental modelling
NA	Maynard and Kuempel [2]	Inhalation	I	Review of airborne nanostructured particles and occupational health
Inhalation and dermal Inhalation Inhalation Inhalation Inhalation Inhalation NA Inhalation NA Inhalation NA Inhalation NA Inhalation Inhalation Inhalation Inhalation Oral Inhalation Oral Inhalation Inhalation Inhalation Inhalation Inhalation Oral Inhalation Oral Inhalation Inhalation Inhalation Oral Inhalation Inhalatio	Wiesner et al. [22]	NA	I	Assessment of risks of manufactured nanomaterials
Inhalation Carbon nanotubes Inhalation Particles less than 2.5 Inhalation Fine and ultrafine TiO ₂ NA Ag, TiO ₂ , and carbon nanotubes Inhalation Nah Fullerenes Inhalation Fullerenes, carbon nanotubes, CdSe, Ag, and Al NA CdSe, Ag, and Al Inhalation TiO ₂ and Ag Oral TiO ₂ and Ag Oral TiO ₂ and Ag NA ENMs	Tsuji et al. [23]	Inhalation and dermal	1	Review of risk assessment of ENMs due to inhalation and dermal routes
Inhalation Particles less than 2.5 \$\mum in diameter Inhalation Ag, TiO2, and carbon nanotubes Inhalation NA Fullerenes Inhalation Fine and ultrafine TiO2 NA Fullerenes Inhalation Fullerenes Inhalation Fullerenes Ag Inhalation TiO2 and Ag Oral TiO2 and Ag Oral TiO2 and Ag NA ENMs	Lam et al. [10]	Inhalation	Carbon nanotubes	Review of carbon nanotube toxicity and assessment of potential occupational and environmental health risks
Inhalation Fine and ultrafine TiO ₂ NA Ag, TiO ₂ , and carbon nanotubes Inhalation Nah Inhalation Fullerenes Inhalation Fullerenes Inhalation Fullerenes Inhalation Ag Oral TiO ₂ and Ag Oral TiO ₂ and Ag Inhalation TiO ₂ and Ag NA ENMs	Kandlikar et al. [24]	Inhalation	Particles less than 2.5 μ m in diameter	Health risk assessment of nanoparticles using expert judgment
Inhalation NA Inhalation NA Inhalation Inhalation NA Inhalation NA Inhalation Inhalation Inhalation Inhalation Inhalation Oral Inhalation Oral Inhalation NA ENMS	Dankovic et al. [11]	Inhalation	Fine and ultrafine TiO ₂	Calculation of risk of cancer using dose-response information for rats
Inhalation Nano-/fine TiO2 NA Inhalation Fullerenes Inhalation Fullerenes NA NA CdSe, Ag, and Al NA Inhalation TiO2 Inhalation TiO2 and Ag Oral TiO2 and Ag Oral TiO2 and Ag NA ENMS	Mueller and Nowack [3]	NA	Ag, TiO ₂ , and carbon nanotubes	Exposure modelling of ENMs in environment using life-cycle perspectives
Inhalation	Liao et al. [12]	Inhalation	Nano-/fine TiO ₂	Model-based assessment for human inhalation exposure risk assessment
Inhalation Fullerenes Inhalation Fullerenes NA CdSe, Ag, and Al NA — CdSe, Ag, and Al NA — — Inhalation TiO ₂ and Ag Oral TiO ₂ and Ag Oral TiO ₂ and Ag NA ENMs	Kroll et al. [25]	NA	1	Current in vitro methods in ENMs risk assessment: limitation and challenges
Inhalation Fullerenes NA CdSe, Ag, and Al NA — — Inhalation Ag Oral TiO ₂ and Ag Oral TiO ₂ and Ag NA ENMS	Shinohara et al. [26]	Inhalation	Fullerenes	Risk assessment
NA Fullerenes, carbon nanotubes, CdSe, Ag, and Al NA —— Inhalation Ag Oral TiO ₂ and Ag Oral TiO ₂ and Ag NA ENMS	Morimoto et al. [27]	Inhalation	Fullerenes	Inflammogenic effect of well-characterised fullerenes in inhalation and intratracheal instillation studies
Inhalation Ag Inhalation TiO ₂ and Ag Oral TiO ₂ and Ag Inhalation TiO ₂ and Ag Oral TiO ₂ and Ag NA ENMs	Tervonen et al. [28]	NA	Fullerenes, carbon nanotubes, CdSe, Ag, and Al	Risk-based classification systems of nanomaterials
80] NA — al. [31] Inhalation TiO2 al. [32] Inhalation TiO2 and Ag [33] Inhalation TiO2 and Ag (umar [34] Oral TiO2 and Ag [5] Oral TiO2 [6] Inhalation ENMs [8] NA ENMs ar [39] NA ENMs lar [39] NA ENMs lar [40] NA ENMs l. [42] NA ENMs l. [42] NA ENMs l. [42] NA ENMs l. [42] NA ENMs	O'Brien and Cummins [29]		NA	Development of a three-level risk assessment strategy for nanomaterials
al. [31] Inhalation TiO2 al. [32] Inhalation TiO2 and Ag [33] Inhalation TiO2 and Ag (umar [34] Oral TiO2 and Ag [35] Oral TiO2 [4] Inhalation ENMs [5] Inhalation ENMs [6] Inhalation ENMs [7] NA ENMs [8] NA ENMs Innar [40] NA ENMs Innar [40] NA ENMs Innar [41] NA ENMs Innar [42] NA ENMs Innar [42] NA ENMs Innar [42] NA ENMs	Grieger et al. [30]	NA	I	Study of current research priorities for nanomaterials and redefining of risk-based efforts
al. [32] Inhalation TiO2 [33] Inhalation TiO2 and Ag (umar [34] Oral TiO2 and Ag [35] Oral TiO2 [4] Inhalation ENMs [5] Inhalation ENMs [6] Inhalation ENMs [7] NA ENMs [8] NA ENMs [9] NA ENMs [1] NA ENMs [1] NA ENMs [1] NA ENMs [2] NA ENMs [3] NA ENMs [42] NA ENMs	Christensen et al. [31]	Inhalation	Ag	Investigation of feasibility and challenges associated with conducting HHRA for Ag ENMs and identification of related data gaps
[33] Inhalation TiO2 and Ag Kumar [34] Oral TiO2 and Ag 35] Oral TiO2 6] Inhalation ENMs 8] NA ENMs nar [39] NA ENMs lmar [40] NA ENMs lmar [40] NA ENMs l. [42] NA ENMs l. [42] NA ENMs l. [42] NA ENMs	Christensen et al. [32]	Inhalation	TiO ₂	Investigation of feasibility and challenges associated with conducting HHRA for TiO ₂ ENMs and identification of related data gaps
Kumar [34] Oral TiO2 and Ag 35] Oral TiO2 6] Inhalation ENMs 8] NA ENMs 1ar [39] NA ENMs 1mar [40] NA ENMs 1 NA ENMs 1 NA ENMs 1 Interpretable ENMs 1 NA ENMs 1 NA ENMs 1 ENMs ENMs	Gangwal et al. [33]	Inhalation	TiO ₂ and Ag	Method development for determining nanomaterials concentrations for ToxCast <i>in vitro</i> testing for occupational exposure potential
35] Oral TiO2 6] Inhalation ENMs NA ENMs 8] NA ENMs nar [39] NA ENMs lmar [40] NA ENMs l NA ENMs l. [42] NA ENMs l NA ENMs	Anandan and Kumar [34]	Oral	TiO ₂ and Ag	Estimation of risk to humans due to exposure of ENMs during inadvertent ingestion of stream water using liver cell line-based toxicity data
6] Inhalation ENMs NA ENMs 8] NA ENMs nar [39] NA ENMs Imar [40] NA ENMs] NA ENMs I. [42] NA ENMs I. [42] NA ENMs	Johnson et al. [35]	Oral	TiO ₂	Study on fate, behaviour, and environmental risk associated with sunscreen TiO ₂ ENMs; HHRA conducted for ingestion of TiO ₂ particles
8] NA ENMS ar [39] NA ENMS mar [40] NA ENMS] NA ENMS I. [42] NA ENMS I. [42] NA ENMS	Kumar et al. [36]	Inhalation	ENMs	Emissions estimates of nanomaterials from road vehicles in megacity Delhi and associated health impacts
8] NA ENMs nar [39] NA ENMs lmar [40] NA ENMs l NA ENMs l. [42] NA ENMs l NA ENMs l NA ENMs	Kumar [37]	NA	ENMs	Making a case for human health risk-based ranking nanomaterials in water for monitoring purposes
nar [39] NA ENMs 1 NA ENMs 3 NA ENMs 4. [42] NA ENMs 1. [42] NA ENMs 1. [42] NA ENMs	Kumar et al. [38]	NA	ENMs	Known, unknowns, and awareness related to nanomaterials in Indian environment
Imar [40] NA ENMs] NA ENMs I. [42] NA ENMs I. [42] NA ENMs	Singh and Kumar [39]	NA	ENMs	Identifying knowledge gaps in assessing health risks due to exposure of nanoparticles from contaminated edible plants
ENMs	Khanna and Kumar [40]	NA	ENMs	Including nanomaterials mixtures in human health risk assessment
I. [42] NA ENMs I NA ENMs	Frater et al. [41]	NA	ENMs	A multistakeholder perspective on the use of alternative test strategies for nanomaterial safety assessment
NA ENMs	Savolainen et al. [42]	NA	ENMs	Nanosafety in Europe 2015–2025: towards safe and sustainable nanomaterials and nanotechnology innovations
	Stone et al. [43]	NA	ENMs	Prioritising nanosafety research to develop a stakeholder-driven intelligent testing strategy

3.2. Applications of ENMs. The list of areas of human endeavour to which nanomaterials have been applied is growing rapidly. Many of these applications are a result of the special properties of nanomaterials such as their electrical, catalytic, mechanical, surface plasmon resonance and antimicrobial properties [5, 57]. Some applications include sensors, thermoelectric materials, photocatalysts, dye-sensitised solar cells, and products such as paints, plastics, sprays, cleaning agents, coatings, sunscreens, films, packing materials, nutraceuticals, and materials for drug delivery to name a few (see, e.g., Sadik et al. [48] and Goesmann and Fledmann [58]). Exposure to nanomaterials may occur during the fabrication and various application and disposal stages of the ENM-containing products, as discussed in Section 4.

4. Assessment of Exposure to ENMs

To assess human exposure to ENMs during the different stages of the life cycle of ENM-containing products, there is need to estimate their quantities when released to the environment together with toxicity-related parameters such as their size, morphology, and chemical composition. Although the field of ENM monitoring is yet not established, it draws from the existing detection and characterisation methods for monitoring particle pollution (e.g., measuring airborne nanoparticles in the air).

Inhalation of airborne ENMs is one the most important routes of entrance into the human body [59]. This is because these nanosized particles suspended in the air can spread over long distances from the point of their release, resulting in uncontrollable human exposure. ENMs can also be disposed to the environment in liquid suspensions (e.g., wastewater) and in solid media (i.e., wastewater sludge). Human exposure to ENMs from liquid and solid media is possible through swallowing or direct contact with the skin. Human exposure to the ENMs released into the terrestrial environment is less likely to occur, mainly due to their immobility.

4.1. Measuring ENMs in the Air. Nanosized particles can be released into the breathing air by both exhaust [50, 60–64] and nonexhaust sources [60, 62, 65–68]. These nanoscale particles undergo a number of physiochemical transformations [69, 70], resulting in both spatial and temporal variation in their concentrations in both indoor [71–74] and outdoor [69, 70, 75] ambient environments. Nevertheless, it is still challenging to apportion the fraction of the ENMs in the total nanoscale particulate material present in the atmospheric environment [67].

The optical and electrical mobility detection techniques used to measure the natural and combustion-derived nanoparticles in ambient air are also used to measure the concentration of ENMs [50]. Optical techniques rely on the growth of nanomaterials by means of condensation and their subsequent counting by measuring the light they can scatter. The condensation particle counters (CPCs [76]) are widely used for monitoring and research purposes. Although CPCs can measure extremely low concentrations, their detection efficiency decreases markedly for materials having diameters below ~3 nm [77].

An alternative to optical counting is to measure the current produced by particles having a known charge distribution using an aerosol electrometer (AE [78]). The number concentration of the sampled nanomaterials can be estimated using the current inducted by the collection of the charged particles on an electrometer filter, the flow rate through the AE, and the charge distribution on the particles. The charge distribution of particles can be controlled by using an aerosol particle charger [79] operating upstream of the AE. Compared to the CPCs, the AEs can detect particles of any size. However, the threshold concentration for producing a signal above the signal-to-noise ratio is of the order of a few tens to hundreds of particles per cm³ for typical flow rates used through the AEs.

Sizing of particles in the gas phase can be achieved by a number of techniques depending on the desired range [80]. The most efficient instrument for determining the size of particles in the nanosize regime is the differential mobility analyzer (DMA [81]). DMAs classify airborne particles based on their electrical mobility, which by knowing the charge distribution and the morphology of the particles can be used to estimate their size. Although a wide range of DMA designs have been described in the literature [82], the cylindrical DMA is routinely used in combination with CPCs for measuring aerosol particle size distribution. To do so, the operating conditions (mainly the electric field strength) of the DMAs are increased in steps or continuously scanned to select particles of different sizes before being counted by the CPC. Typical scanning times of these systems take from a few seconds to a few minutes, depending on the desired resolution of the measurements [83]. Recent advances in the field have led to DMAs [84-86] and mobility spectrometers [50, 79] that can simultaneously classify particles in different size ranges, thereby reducing the time required to measure the entire size distribution of the particles in subsecond time fractions.

The pulmonary toxic effects induced by inhalation of ENMs are best correlated with the surface area rather than the concentration of the particles [87]. Instruments that can directly measure the surface rather than the number or the mass of the particles are already available in the market. These instruments rely on the attachment of ions or radioactive species onto the surface of particles and measuring of their current or radioactivity, respectively [88, 89]. In accordance with recent guidelines defining benchmarks or reference threshold values, surface area monitors that rely on the attachment of ions on the particles are also used to determine the mean particle diameter and the number concentration under the assumption that those are spherical [90]. Given that many types of nanomaterials do not have a spherical shape, the estimation of the surface area by these instruments can be misleading.

One of the greatest challenges in monitoring the concentration of ENMs in the air is to distinguish them from already present background nanomaterials, which can also vary in concentration, size, morphology, and composition, depending on the nature of natural and anthropogenic sources. Online distinction between engineered and background nanomaterials can be made by probing their intrinsic

properties such as morphology, solubility, and volatility, provided that these are detectably different among the two species. Tandem differential mobility analysis (TDMA [91]) can also be used to measure the intrinsic properties of particles such as morphology, vapour uptake, and volatility [92, 93]. Information on the chemical composition or the structure of the airborne nanomaterials can also be obtained by offline techniques such as electron microscopy [94]. Electron microscopes can also be used for determining the elemental composition of nanoparticles, using techniques such as electron dispersive X-ray (EDX) spectrometry [95, 96]. These techniques, however, are time-consuming, expensive, of high complexity, and therefore inappropriate for systematic monitoring.

4.2. Measuring ENMs in the Aquatic Environment. Methods for characterising nanomaterials suspended in liquid media have been developed over the past decades in view of monitoring and controlling their synthesis by liquid-based techniques. Dynamic light scattering (DLS) is the most widely used technique for measuring the size and shape of nanoparticles in liquid suspensions [97]. In DLS, the solution sample is illuminated by a monochromatic light source, and the light scattered by the particles is continuously measured by a photodetector. The recorded scattering fluctuations are then processed in order to determine the translational and rotational diffusion coefficients of the particles, which in turn are used to determine their size and shape [98]. A drawback of DLS is that the larger particles in the samples can mask the signal from the nanoparticles during the measurement [99]. Another drawback is that they are not suitable for the assessment of nanofibres or nanorods.

Another way of measuring the size distribution of nanoparticles in liquid suspensions is by tracking and analysing their motion. Nanoparticle tracking analysis (NTA) measures the scattering generated from particles undergoing Brownian motion in order to estimate their diffusion coefficients [100]. Although the signal of the larger particles in NTA can also mask the signal produced by the smaller particles due to their stronger scattering, the effect is less important than in the DLS [101]. As a result NTA is preferred to DLS for measuring nanosized particles.

Another technique for measuring particles with diameters down to the nanometer size range is flow field-flow fractionation (flow FFF [102]). In Flow FFF, the particles are separated in a crossflow channel, where larger particles are immobilised in an expanding channel faster than the smaller ones due to differences in their diffusivity [103]. A variation of this method is the asymmetric flow FFF (AF4), which is capable of fractionating particles and macromolecules having diameters down to 2 nm [104]. AF4 can separate particles with high resolution, thereby giving unique characterisation possibilities for nanoparticles suspended in liquid media.

5. Hazard Assessment

5.1. Toxicity Data Availability. Many studies are currently focusing on understanding toxic effects of ENMs in *in vitro* and *in vivo* studies on different species (rats, fishes,

algae, daphnids, and bacteria, amongst others). Among these studies, rats as animal models are used for obtaining reference doses. Very few toxicity studies are currently available which provide complete information on experimental conditions (i.e., nanoparticles characteristics, animal- or cell line-related information, exposure duration and frequency, and exposure medium and endpoints observed). To illustrate this aspect, summary of toxicity studies on nano-TiO₂-based ENMs for oral and inhalation exposure routes is presented in Tables 2 and 3, respectively. Review of these studies indicates that different works have used varying experimental conditions and endpoints for observing effects, making it challenging to obtain data for systematic comparison and risk estimation purposes. For example, there is a lack of detailed dataset from toxicological and epidemiological studies on inhalation toxicity to humans. To circumvent this issue, animal data are used for determining toxicity parameters for human external and lung doses [44]. Although use of animal data for reference dose quantification is a standard methodology in the absence of toxicity data on humans, there is inherent uncertainty in extrapolating findings on animals to humans (i.e., interspecies uncertainty). Further, there exists a difficulty in finding out toxicity studies with detailed information on experimental descriptions for comparison purposes. For example, O'Brien and Cummins [105] screened many studies of TiO₂ particles to identify toxicity works that are suitable for providing ingestion and inhalation-related animal data. After a thorough review, they were able to finally use findings from the following studies related to (i) inhalation (endpoint: lactate dehydrogenase (LDH) test [106-108]) and (ii) ingestion (endpoint: liver kidney and spleen coefficient; blood biomarker assays and histopathological examination [109]). These observations clearly indicate a need for standardising toxicity conditions for observing a given endpoint and facilitating easy comparison of toxicity data [110]. During the use of animal toxicity data, there is also a need for the accounting of the effects of animal species or strain on toxic effects and subsequently on dose-response parameters. For example, Kuempel et al. [44] discussed the exposure of ultrafine and fine TiO₂ particles to rat strains (Sprague-Dawley, Wistar, Fischer 344, and Long Evans). They found a difference in observed lung burden doses, indicating a need to consider the effect of rat strain type on doseresponse parameter since this can influence benchmark dose limit values that are normally calculated using dose-response

Furthermore, some researchers have studied ENM-based toxicity on cell lines (see Tables 2 and 3). These studies are easier to conduct than *in vivo* studies. However, findings of *in vitro* studies cannot be directly used to understand toxicity of ENMs to the target organs as toxic effects on cell lines cannot represent toxic effects on the whole target organ. These data are sometimes less useful than those obtained from animal models (e.g., rats) as findings of toxicity studies on cells cannot be easily extrapolated for determining toxicity to either human cell line or to human target organ. However, this step is not required if toxicity data is directly available for human cell line and/or human organs. There is a need for identifying mechanisms for extrapolating the findings of

Table 2: Oral toxicity of TiO₂ to rats and *in vitro* data on rat cell lines.

Reference	Exposed animals/cell line type	Experimental design	Endpoint
Hussain et al. [111]	BRL 3A rat liver cell line	Exposure of TiO ₂ particles for 24 hours <i>in vitro</i> to 40 nm TiO ₂ particles	Cytotoxicity; mitochondrial function
Trouiller et al. [13]	Rats	Exposure of TiO ₂ particles (21 nm size, 75% anatase, and 25% rutile) through drinking water	DNA single and double strand breakage; chromosomal damage <i>in vivo</i> ; DNA deletions in offspring
Kocbek et al. [112]	HEK cells	Effect of TiO ₂ and ZnO particles on HEK cells <i>in vitro</i>	Cytotoxicity

ENM: engineered nanomaterial; HEK cells: human embryonic kidney cells; ROS: reactive oxygen species.

Table 3: Inhalation toxicity of TiO₂ to rats and *in vitro* data on rat cell lines.

Reference	Exposed animals/cell line	Experimental design	Endpoint
Henrich et al. [113]	Wistar rats; inhalation	Chronic inhalation (24 months); 10 mg/m ³ TiO ₂ particles (15–40 nm size; mass-median aerodynamic diameter = 800 nm)	Carcinogenicity; histology; DNA addicts; alveolar lung clearance
Bermudez et al. [106]	Hamster and rat; inhalation	Subchronic inhalation (3.25 months); 10–250 mg/m³ pigmentary TiO ₂ particles (assumed diameter = 300 nm; mass-median aerodynamic diameter = 1400 nm)	Inflammatory response; cytotoxicity; lung cell proliferation; histopathology
Höhr et al. [108]	Rat; instillation	16 h instillation; 1–6 mg (diameter = 20–30 nm and 180 nm)	Acute inflammatory response; cell damage
Bermudez et al. [107]	Hamster and rat; inhalation	Subchronic inhalation (13 weeks); 0.5–10 mg/m ³ ; diameter = 21 nm; mass-median aerodynamic diameter = 1370 nm)	Inflammatory response; cytotoxicity; lung cell proliferation; histopathology
Renwick et al. [114]	Rat; instillation	24 h instillation; 0.125–0.5 mg (diameter = 29 nm and 250 nm)	Inflammatory response; epithelial injury; alveolar macrophage toxicity; lung clearance
Warheit et al. [115]	Rat; instillation	24 h, 1-week, 1-month, and 3-month instillation; 129.4 nm; 149.4 nm; 136 nm; 382 nm	Bronchoalveolar lavage (BAL) fluid inflammatory markers; cell proliferation; histopathology
Grassian et al. [116]	Rat; inhalation	4 h duration and 10-day duration; 3.5 nm	Total protein content; LDH activity; inflammatory cytokines; histopathology
Li et al. [117]	Rat; inhalation	Three-day inhalation; 3 nm; 20 nm	BAL fluid biochemical parameters; histopathology
Lui et al. [118]	Rat; instillation	One-week instillation; 5–50 nm	Histopathology; blood biochemical parameters; LDH, alkaline phosphatase, and acid phosphatase activity; alveolar macrophage phagocytotic ability
Falck et al. [56]	Bronchial cells (BEAS 2B)	Exposure to 8 doses (1–100 μ g/cm ²) of TiO ₂ particles: (1) nanorutile, (2) anatase with size <25 nm, and (3) fine rutile with size <5 μ m for 24, 28, and 72 hours of exposure	DNA damage
Bhattacharya et al. [20]	Bronchial cells (human BEAS 2B)	Toxicity of TiO ₂ particles in anatase crystal phase (size: <100 nm)	Reactive oxygen species (ROS) production; DNA adduct formation
Bhattacharya et al. [20]	Lung fibroblast cells (IMR-90)	Toxicity of ${\rm TiO_2}$ particles in anatase crystal phase (size: <100 nm)	Cytotoxicity

in vitro studies to *in vivo* studies. In this direction, some studies have used an endpoint that can be easily compared in *in vitro* and *in vivo* toxicity studies. For example, Faux et al. [119] used pulmonary inflammation as a response function for toxicity to epithelial cells in the centriacinar region of the

lungs and in a petri dish. They reported that *in vivo* response (i.e., an increase in polymorphonuclear leukocytes in bronchoalveolar lavage fluid from rats) and *in vitro* response (i.e., an increase in inflammatory cytokine interleukin-8 mRNA in human alveolar epithelial cell line A549) were found to

be comparable. More studies are required to explore this approach which can increase the usefulness of *in vitro* studies in the interest of time and cost.

On the other hand, recent studies are also considering the prospects of using alternative test methods to understand health risk and toxic effects of different chemicals. For example, the U.S. National Research Council's report [120] focuses on toxicity of chemicals to humans and suggests using alternative test methods to improve understanding. Studies are exploring suitability of alternative test methods (high throughput systems, in silico and in vitro models) for screening large number of samples containing ENMs and for reducing increased burden on animal testing [41, 43, 120]. For example, Frater et al. [41] applied predictive modelling and in vitro modelling for selecting carbon nanotubes in short-term inhalation or instillation studies. Similarly, Puzyn et al. [121] used experimental testing and quantum-mechanics-based computational modeling for studying cytotoxicity of metal oxide nanoparticles to E. coli. These recent efforts indicate the feasibility of using alternative test methods for toxicity studies in order to obtain data relevant to the dose-response step. These methods have also been mentioned in European Parliament and Council directives [122, 123]. However, more efforts are clearly needed to improve toxicity screening of increasing number of nanomaterials.

5.2. Effect of Physiochemical Characteristics of ENMs on Toxicity. Physical and chemical properties of the ENMs have been shown to influence their toxicity [124-127]. There is currently a need for additional data providing information on the effects of ENMs characteristics (i.e., size, shape, chemical composition, and degree of agglomeration) on their toxicity [44]. ENM size in the environment depends on media characteristics. For example, ENM size in water depends on water physicochemical characteristics, such as ionic strength, pH solution, and presence of other co-ions [128]. Therefore the size of the ENMs needs to be determined in the exposure media and this needs to be related to the results from the hazard studies. Some studies have assessed the effect of size on ENM-induced toxicities to determine size-dependent dose-response model of ENMs for a given endpoint. For example, Falck et al. [56] observed that the lowest dose inducing DNA damage on bronchial cells BEAS 2B is $1 \mu g/cm^2$ for fine rutile, $10 \mu g/cm^2$ for nanoanatase, and $80 \,\mu\text{g/cm}^2$ for nanorutile (see Table 2). Furthermore, particle size also affects extent of deposition in lungs during inhalation and its subsequent deposition in lungs and other organs as noted in Table 3 [44, 124]. For example, clearance of nanoparticles was found to be smaller than that of larger particles during inhalation exposure [124-126]. The finding of the Falck et al. [56] study also indicated the effect of crystallinity of nanoparticles on lowest dose inducing DNA damage. Further, chemical composition of ENMs also affects the extent of their toxicity [29]. For example, extent of toxicity of TiO2 and Ag nanoparticles differ for a given media. This information highlights the effects of size, shape, phase, and chemical composition on toxicity of ENMs and indicates a need for incorporating these factors into estimating the

ENM-based toxicity reference limits for animal and human exposure.

5.3. Nanomaterials Cooccurrence. There exists a possibility of cooccurrence of ENMs in the environment, indicating a chance of exposure to different types of ENMs simultaneously. To estimate human health risks due to exposure of more than one type of ENM in the environment, related toxicity data are required. Although data on exposure of individual ENMs to animals and human cell lines are available, data are still lacking on the effect of interaction with other chemicals and any possible subsequent synergistic or antagonistic effects on different target organs/organisms [40]. For example, Gou et al. [129] noted that synergism between the ENMs made of Ag and TiO₂ is plausible on bacteria. As this study was conducted on bacteria, the obtained information about interaction of Ag and TiO₂ particles cannot be used in EHHRA. To properly incorporate exposure of more than one type of ENM, information about cooccurrence of different ENMs in a given medium, chances of exposure of more than one type of ENM from a given exposure medium, and data on toxicity due to mixture of ENMs for a given target organ are required. For example, carbon nanotubes and fullerenes are emitted into the environment during combustion processes and coexist in air [130]. Both of these carbon-based ENMs affect alveoli region of lungs. A review by Aschberger et al. [131] indicates that toxicity studies are available only for carbon nanotubes or for fullerenes. However, these two types of carbon-based nanomaterials have not been studied together in toxicity studies. The information about effects of possible interactions of two or more than two ENMs on a given target organ need to be gathered to determine resultant dose-response curves.

Currently, the conventional approach is to develop doseresponse curves for a given endpoint using exposure of single type of contaminant. To account for toxicity due to cooccurring ENMs, one simplified approach could be to use reference values of exposure of single ENMs to target organ without assuming interaction with the other types of ENMs. It is similar to the USEPA's dose-addition-based risk assessment methodology [46, 132]. This methodology assumes that contaminants in mixture behaves similarly to how they would behave if they were present individually in water [45, 46]. However, this approach might not capture the realistic effect of action on site, and two or more ENMs might be affecting. Towards this, the first attempt could be made by calculating the USEPA's interaction-based hazard index (HI), as seen in (1) [46, 132], for a mixture of ENMs. The HI is generally used for chemicals. In (1), HQ_i is hazard quotient for *i*th chemical; f_{ij} is toxic hazard of the *j*th chemical relative to the total hazard from all chemicals potentially interacting with ith chemical. The magnitude of interaction between chemicals is represented by M_{ii} (an interaction magnitude parameter, default value = 5 [46]). The strength of evidence that ith chemical influences the toxicity of jth chemical is represented by B_{ij} (i.e., weight-ofevidence factor (2a)). Also, g_{ij} in (1) indicates a degree to which *i*th chemical and *j*th chemical are present in equitoxic amounts (2b). For the case of two chemicals, f value is

1 as $HQ_{12} = HQ_2/[HI_{add} - HQ_1] = HQ_2/[HQ_2]$. *B* can be assigned to default value of 0.5 based on guidelines from the USEPA [46] methodology, as also used by studies elsewhere [133, 134].

$$HI_{int} = \sum_{i=1}^{n} HQ_i \sum_{j \neq i}^{n} \left(f_{ij} \times M_{ij}^{B_{ij} \times g_{ij}} \right). \tag{1}$$

$$f_{ij} = \frac{HQ_j}{HI_{add} - HQ_i}$$
 (2a)

$$g_{ij} = \frac{\sqrt{\text{HQ}_i \times \text{HQ}_j}}{0.5 \times (\text{HQ}_i + \text{HQ}_j)}.$$
 (2b)

The USEPA's risk assessment methodology for chemicals for estimating risks can only be used for estimating risks due to exposure of mixture of ENMs provided this equation holds true for ENMs mixture as well. Currently, not much work has been conducted to understand applicability of (1) and ((2a) and (2b)) in this regard. The first step towards this approach is to gather information about all parameters used in these two equations. The following steps are encouraged to modify current ENM-related occurrence and toxicity studies: (i) calculate the chance of cooccurrence of different ENMs in a given medium, (ii) develop groups of cooccurring ENMs influencing a given endpoint, and (iii) obtain dose-response data during simultaneous exposure of two or more ENMs to target models under study if available or conduct studies to obtain these data to assess the effect of cooccurrence of ENMs on toxicity endpoints. Using this additional information, the USEPA methodology for mixture of chemicals ((1), (2a) and (2b)) may be applied to estimate risks due to exposure of ENMs, until detailed methodology for mixture of ENMs is available.

5.4. Fate of ENMs in Organisms and Translocation Assessment. Inhaled or ingested ENMs could be translocated to different parts of the body through systemic distribution [135]. As a result, the exposure of ENMs from one particular route might result in exposure of ENMs to different organs. Thus, exposure assessment and dose-response assessment stages need to consider this aspect for assessing overall toxicity from a given exposure route.

During inhalation exposure, ENMs may deposit in lungs where they may disaggregate into smaller particles in the alveolar lining fluid, resulting in exposure of both nanosize and micron-size particles to interior parts of the lungs [136]. Once particles are deposited in the respiratory tract, clearance mechanisms such as mucociliary clearance in airways or alveolar macrophage-mediated clearance in gas-exchange region govern deposition, clearance, and translocation of particles to other regions [124–126]. This translocation happens when particles enter in the lymph or blood circulation. Thus, the ENMs exposed through inhalation route reach to lungs as well as in other organs.

Similar observations have been found in the case of oral exposure of ENMs. For example, the biodistribution

experiment by Wang et al. [137] on adult mice showed that orally administered TiO2 was found to be retained mainly in the liver, spleen, kidneys, and lung tissues, indicating translocation of TiO2 particles to other tissues and organs after uptake by gastrointestinal tract. The work of Park et al. [138] reported that, when mice were orally treated with 1 mg/kg Ag particles for 14 days, small-sized Ag particles (22 nm, 42 nm, and 71 nm) were found to be distributed to brain, lung, liver, kidney, and testis. However, large-sized Ag particles (323 nm) were not detected in these organ tissues, indicating an effect of size on translocation. Furthermore, a recent study by van der Zande et al. [139] noticed in vivo formation of silver nanoparticles in rat organs during their oral exposure to silver salts over a study duration of 28 days. These findings indicate the possibility of *in vivo* formation of nanoparticles, which also needs to be considered in estimating exposure dose of nanoparticles in risk assessment process.

5.5. Toxicity Study Duration. Although data from acute- or subchronic studies on animal toxicity studies [13, 140] are available, they do not represent fully the toxicity of ENMs to target organs during long-term exposure. Thus, uncertainty on extrapolation from subchronic to chronic exposure data still exists. As chronic studies are expensive, there is a need for developing short-term tests to simulate chronic exposure scenario. In these studies, biological indictors whose response can be extended based on causal relationship for long-term exposure need to be explored in detail [44]. In addition, there could be differences in toxicodynamics and toxicokinetic response in short- and long-term studies during the use of low concentration of ENMs, which need to be considered in these short-term tests. Overall, there is a need for conducting more long-term toxicity studies or combining experimental data [29] and simulation-based hybrid approach to predict long-term toxicity effects.

Quantitative Nanostructure-Toxicity Relationship (QNTR). Quantitative structure-activity relationship (QSAR) has been widely used in pharmacology and related fields to predict the toxicity of drugs without the need for tedious, time-consuming, and expensive animal testing. An analogue of the model, QNTR, has been extended to nanomaterials. In this context, Winkler et al. [141] recently noted that structural properties of nanomaterials, which influence toxicity, may include size, shape, surface area, and degree of electrostatic interaction between nanomaterials and their cellular environments as well as other physicochemical properties of nanomaterials. These properties have been used as descriptors [121, 141, 142] for the prediction of the properties of new materials or for the explaining of their biological effects. In many instances, such predictions are facilitated by the use of multivariate data analysis techniques including regression models and artificial neural networks and verified by cross-validation, using training and validation data sets [143].

Emerging results from these types of studies indicate that the structural properties, which influence the toxicity of nanomaterials, can reside both in the core and on the

surface of the materials [141, 144]. However, to extend QSAR to QNTR effectively, a number of challenges need to be addressed. Winkler et al. [141] have mentioned some of these challenges as

- (i) lack of adequate definition and understanding of the biologically important entities that moderate the adverse effects of nanomaterials,
- (ii) need to choose the right assays that can be used to model and correlate the toxic effects of nanomaterials in vitro and in vivo,
- (iii) accurate modeling of the complex interactions between nanomaterials and biological systems,
- (iv) understanding of the relationship between QNTR methods and other computational approaches such as quantum chemistry and molecular dynamics.

These challenges arise due to following data gaps: (i) lack of sufficient empirical data on the composition of biocorona on the surfaces of nanomaterials; (ii) lack of in vitro data that can be used to predict in vivo effects of nanomaterials, and (iii) the paucity of descriptors that can specifically be used for nanomaterials [141]. Despite these challenges and data gaps, progress has been made in probing and understanding the interactions of nanomaterials with their biological environments. For example, Lynch and Dawson [145] described the use, strengths, and shortcomings of surface plasmon resonance, magnetic nanosensor arrays, surface adsorption index, isothermal titration calorimetry, shotgun proteomics analysis, fluorescence correlation spectroscopy, size exclusion chromatography, and liquid chromatography-tandem mass spectrometry in increasing the understanding of the interaction of nanomaterials with environmental molecules.

However, the gaps in knowledge must be fully identified and understood before an internationally acceptable approach to QNTR is proposed, which can appeal to all stakeholders. An important initiative taken towards achieving this goal was captured by a recent workshop (http://www.cost.eu/events/qntr) on the use of QNTR methods in modelling the biological effects of nanomaterials. This workshop highlighted the challenges envisaged and the proposed methods of overcoming these challenges. It also proposed short- to long-term steps that could be taken to address the challenges (Table 4). In a similar development, Stone et al. [43] reported that a European Commission funded a project, which was undertaken to identify the framework of future research priorities for the development of an intelligent testing strategy (ITS) for evaluating nanomaterials safety without the need for caseby-case assessment of human and environmental exposure. Some short-, medium-, and long-term research priorities emanating from the project are also included in Table 4. Readers interested in the full report are encouraged to visit http://www.nano.hw.ac.uk/research-projects/itsnano.html [43].

6. Environmental Regulations of ENMs

A review of literature indicates that not much information is available on the regulation of ENM-related exposure [8, 67]. The scope of different available regulations and guidelines in the USA, European Union, and WHO is limited to organic compounds and toxic substances, but they do not specifically address ENM-related pollution. Air quality directives set by the US Environmental Protection Agency (USEPA) and the European Commission (EC) include concentration limits for particulate pollution [146, 147]. These limits, however, refer to mass concentration measurements (i.e., PM₁₀ and PM_{2.5}), which are not appropriate for airborne nanoparticles [75, 148]. Some studies have attempted to understand the suitability of current regulations in handling ENM pollution. For example, Beaudrie et al. [110] analysed U.S. Federal Environmental, Health, and Safety (EHS) regulations for their adequacy and application to ENMs using a life-cycle framework. They found that existing regulations for controlling occupational exposure and environmental pollution (e.g., the OSHA Act, the Clean Air Act, and the Clean Water Act) do not provide provisions for addressing regulation of ENM-related environmental pollution. Finding of this study indicated that there is a need for investigating suitability of current regulations for controlling environmental pollution of ENMs. As the development of environmental regulation for ENMs is still undergoing, regulatory bodies of some countries such as the US [95], UK [41, 120], and Australia [62, 63] are investigating if ENM-based products could be treated as new classes of existing substances, which are currently being regulated. Towards this, many countries have started collecting nanospecific data from each product and building a database so that it can be used in regulating ENMs. For example, the US is collecting substance-specific information for ENMs from different products for regulating ENMs [149]. It appears that regulatory bodies may continue to use existing regulations to address ENM pollution until ENM-related regulation is structured and approved. In this regard, some studies have suggested communication of nanospecific information of products to all stakeholders (manufacturers, users, and regulatory bodies) in an effective manner. For example, Nowack et al. [150] analysed release and behaviour of ENMs for the context of major accident prevention and suggested to include nanospecific data in material safety data sheets on a compulsory basis. Furthermore, a notable amount of current research is focusing on prioritising ENM-related research issues and exposure metrics to develop stakeholder-driven intelligent testing methods and strategies. For example, recent work of Stone et al. [43] used opinions of experts from government, academia, industry, funders, and NGOs for systematically identifying a need for information in the areas of physicochemical characterisation, exposure identification, and hazard identification and modelling approaches for short-, medium-, and long-term time scales. There is clearly a need for initiating regulatory efforts to develop guidelines and monitoring metrics for regulating the presence of ENMs in water, air, and soil media. Whether these limits should be

Table 4: Desirable (achievable) outcomes of quantitative nanostructure-toxicity relationship (QNTR) in the next ten years adapted from published studies (Winkler et al. [141]; Stone et al. [43]).

Short-term	Medium-term	Long-term
(i) Fully characterised nanomaterials that can be used in experiments	(i) Appreciable data on <i>in vivo</i> effects of nanomaterials	(i) Sufficient <i>in vitro</i> and <i>in vivo</i> information on nanomaterials to assist the development of regulatory measures
(ii) <i>In vitro</i> assays that are useful for the assessment of toxicologically relevant effects	(ii) Robust <i>in vivo</i> models for predicting the endpoints of nanomaterials' interaction with biological systems	(ii) Reliable models for prediction and risk-based classification of nanomaterials
(iii) Fast and highly efficient methods for measuring and modelling the interaction of nanomaterials with biological systems	(iii) In-depth understanding of the mechanism of nanomaterials toxicity	(iii) Integration of the data into legal framework, life-cycle assessment, and decision trees
(iv) Specific descriptors for modelling the relationships among the structures and toxicity properties of nanomaterials	(iv) Information on cohort and population; direct and indirect; short-/long-term reversible and irreversible effects	(iv) Implementation of risk assessment frameworks

based on number, mass or surface area concentrations are still contentious [8].

7. Summary, Conclusions, and Future Directions

This article provides current status of knowledge in areas of measurements, characterisation, and environmental risk assessment of the ENMs besides identifying the research gaps and future directions. The proposed future steps could help in (i) providing information about monitoring of ENMs and dose-response parameters reflecting environmental exposure conditions, and (ii) developing QNTR. The summary and key conclusions on the topic areas covered are listed below.

- (i) The field of ENM monitoring is yet fairly new and therefore the monitoring methods are generally adopted from the existing techniques for monitoring particle pollution in air and water environments. The greatest challenge in monitoring the concentration of ENMs in any environment is to distinguish them from the background nanomaterials.
- (ii) Methods are available for measuring particle number concentration and characteristics. Particle concentration and size distribution can be measured using instruments such as DLS, CPC, DMPS, FMPS, ELPI, and SMPS. Surface characterisation of nanomaterials is generally conducted using SEM, TEM, EDX, BET, TG, FTIR, near-infrared, and Raman's spectroscopy methods. These methods are usually complicated, time-consuming, and expensive, thereby making them inappropriate for routine monitoring of ENMs in different environmental media.
- (iii) Current toxicity studies do not use realistic exposure conditions during dose-response experiments and hence these are inappropriate for the EHHRA of ENMs. In addition, recent studies have started developing relationships between structural properties of nanomaterials and toxicity (i.e., QNTR) for use in the dose-response step. However, this approach

needs extensive empirical data on physicochemical characteristics and toxicity of ENMs. Such data are currently unavailable in abundance and need to be systematically obtained.

- (iv) In order to address mixture toxicity issue in the EHHRA of ENMs, the following steps are proposed. (i) Calculate chances of cooccurrence of different ENMs in a given medium. (ii) Develop groups of cooccurring ENMs influencing a given endpoint. (iii) Obtain dose-response data during simultaneous exposure of two or more ENMs to target cell line or target organ or animal used in toxicity studies. And (iv) use the USEPA methodology for mixture of chemicals for conducting risk estimation of exposure of ENMs until detailed methodology for mixture of ENMs is available. These steps need to be pursued for investigating their appropriateness in assessing health risks due to exposure to ENMs.
- (v) Despite increased awareness in public and recent reporting of ENMs in environment, not much information is yet available to regulate them due to the continuing technical difficulties.

Figure 1 presents the interlinking of identified factors affecting the EHHRA of ENMs. The following directions for future research works can assist in filling the research gaps on the topic areas covered:

- (i) developing a combination of different analytical methods for determining ENM mass concentration, particle concentration, and morphological information,
- (ii) conducting toxicity studies using conditions relevant to environmental exposure of ENMs for humans to obtain dose-response information useful in EHHRA of ENMs,
- (iii) obtaining relevant data in order to develop QNTR relationships,

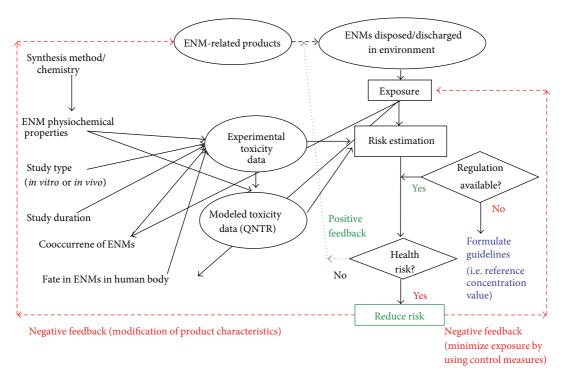


FIGURE 1: Schematic showing interlinkages of different factors for determining environmental and health risks due to ENM exposure.

(iv) initiating efforts for formulating guidelines for regulating presence of ENMs in different environmental media.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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12

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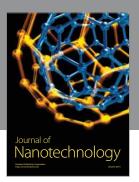
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