

Research Article

Environmentally Friendly Antifouling Metabolites from Red Sea Organisms

Sultan Semran Al-Lihaibi ¹, Ahmed Abdel-Lateff,^{2,3} Waled Mohamed Alarif ¹,
Hajer Saeed Alorfi,⁴ Yasuyuki Nogata,⁵ and Tatsufumi Okino⁶

¹Department of Marine Chemistry, Faculty of Marine Sciences, King Abdulaziz University, P.O. Box 80207, Jeddah 21589, Saudi Arabia

²Department of Natural Products and Alternative Medicine, Faculty of Pharmacy, King Abdulaziz University, P.O. Box 80260, Jeddah 21589, Saudi Arabia

³Department of Pharmacognosy, Faculty of Pharmacy, Minia University, PO. Box 61511, Minia 61519, Egypt

⁴Department of Chemistry, Faculty of Science, King Abdulaziz University, PO. Box 80207, Jeddah 21589, Saudi Arabia

⁵Environmental Science Research Laboratory, Central Research Institute of Electric Power Industry, Abiko 270-1194, Japan

⁶Graduate School of Environmental Science and Faculty of Environmental Earth Science, Hokkaido University, Sapporo 060-0810, Japan

Correspondence should be addressed to Sultan Semran Al-Lihaibi; sallihaibi@kau.edu.sa

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Seventy-one marine organisms representing different classes of marine fauna and flora were collected from the Red Sea. They include sponges, hydrozoan, soft corals, sea cucumber, ascidian, cyanobacteria, and macroalgae. The methanolic extracts were evaluated for their toxicity and settlement inhibition effects by using cultured *Balanus amphitrite*. Thirty-three extracts displayed antifouling effects: four samples were highly potent at 1 $\mu\text{g}/\text{mL}$ with a percentage of settlement inhibition above 31%, twenty-two were potent at 10 $\mu\text{g}/\text{mL}$ with a percentage of settlement inhibition between 16 and 30%, and seven were active at 10 $\mu\text{g}/\text{mL}$ with a percentage of settlement inhibition between 0 and 15%. Two promising extracts were purified by employing several chromatographic techniques, leading to the isolation of 12 known compounds. The isolated compounds were evaluated for their antifouling activities and demonstrated potent antifouling effects with EC_{50} values of less than 10 $\mu\text{g}/\text{mL}$.

1. Introduction

Biofouling occurs due to the adhesion of barnacles, microbial slimes, and macroalgae on wetted marine surfaces [1–3]. These organisms produce adhesive organic materials which enhance the ability of microorganisms (microalgae, protozoa, and bacteria) to settle on the marine surfaces [4–6]. Such biofilms lead to several economic hazards, including increased fuel consumption, ship damage, and immigration of invasive macromolecules [6, 7]. The diverse fouled surfaces become vectors for transporting non-indigenous species when they are moved from one area to another. Then, they can attach to a new wetted surface and

form a core for new biofouling, which is responsible for increasing fuel consumption by \$56 million annually for the entire fleet of DDG-52 (midsize) crafts [8]. This process also increases the production costs of aquaculture operations by \$1.5 to \$3 billion annually [9]. Hence, biofouling is a real challenge for both chemistry and marine technology [10, 11].

Blocking marine-immersed surfaces to reduce the attachment of biofouling organisms can be performed by employing synthetic paint-added chemical compounds (biocides) [12, 13]. This has been done by using potent antifouling synthetic organotin compounds such as tributyltin self-polishing copolymer paints (TBT-SPC paints) [14]. Unfortunately, tributyltin-based paints have serious

adverse effects. They cause toxicity to humans, in the form of liver impairment, hypoglycemia, glycosuria, and respiratory disturbances. In experimental animals, they also showed immunosuppressive, endocrinopathic, neurotoxic, hepatotoxic, nephrotoxic, and skin & eye irritation effects. Gonadotoxic, embryotoxic, fetotoxic, and developmental effects were also observed. In addition, they have phytotoxic effects on marine organisms, even at low concentrations [15, 16]. Accordingly, the International Maritime Organization (IMO) stopped the application of organotin compounds as antifoulants, in January 2003. WHO has prohibited the use of organotin biocides since 1990 [17].

The safety profile of natural products has allowed researchers to apply a “back to nature” approach, aimed at developing environmentally friendly natural antifouling paints instead of using synthetics. Several successful examples have resulted from the application of this natural product strategy. Capsaicin, the major bioactive principle of chili pepper, effectively inhibited zebra mussel byssal attachment. It has been used in ship antifouling paints in China after approval from the National Environmental Protection Standard of the People’s Republic of China. In the last three decades, marine natural products have gained great attention, as an alternative solution after the ban of organotin derivatives [18].

Several studies have examined antifouling metabolites, which are isolated from marine invertebrates, particularly, sponges and soft corals [19–26]. Both organisms have established an impressive warehouse of chemical defense systems against biofouling. It is interesting that the majority of environmentally friendly antifouling metabolites, based on research done in the last 30 years, belong to the same natural classes: terpenoids (i.e., sesquiterpenes and cembranoid diterpenes), alkaloids, and steroids. On this basis, our sample collection was directed towards the marine organisms which can produce such metabolites [27–29].

The current manuscript focuses on discovering antifouling metabolites among selected Red Sea organisms. Seventy-one marine extracts were screened for their antifouling activity, including sponges, algae, tunicate, sea cucumbers, soft corals, and cyanobacteria. Two of the most promising extracts were purified, which led to the isolation of twelve known metabolites. The isolated compounds were evaluated for their antifouling effects.

2. Materials and Methods

2.1. Collection and Preparation of Samples. Both divers who collected the samples are marine biologists. The divers searched in different sites at depths ranging from 1 to 20 m. After collection of samples by proper methods, including scuba diving, the samples were sorted into separate boxes for each site. Underwater photographs were taken to help in systematic identification and herbarium formation. The collected samples were coded using the standard system indicating the date of sampling, the phylum and order number, etc. A simplified code was later used for the biological activity assays (SH-). Monthly cruises were organized by the team to explore the marine habitats of the north

and south coasts of Jeddah city. All diving sites were listed and the corresponding points localized using GPS techniques. Different selected samples were collected from different places along the Red Sea coast. These organisms were taxonomically identified, and their descriptions are listed in Table S1. Figure 1 illustrates the places of collection of the Red Sea samples.

2.1.1. Algal Samples. Specimens of marine algae, belonging to different genera, were collected from the water around the Jeddah coast (Table S1). After collection, the samples were washed with filtered seawater to remove associated debris and large epiphytes. A 10 min 5% ethanol wash was performed to clean the surface of microflorae. The cleaned material was lyophilized, and then the samples were extracted.

2.1.2. Sponge Samples. The sponge species (Table S1) were collected from different depths and bodies of water around Saudi Arabia, using different techniques such as skin and scuba diving. The sponges were washed with freshwater and transported to the laboratory for lyophilization and extraction.

2.1.3. Coral Samples. Soft corals (Table S1) were collected from different depths and different locations of the Jeddah coast, using different techniques such as skin and scuba diving. The samples were kept in ice boxes and transported to the laboratory for lyophilization and extraction.

2.1.4. Cyanobacteria Samples. Samples were obtained from coral reefs near Jeddah, Saudi Arabia, via scuba diving. They were collected by hand or using metal forceps and placed in nets or Ziploc bags (Table S1). As much as possible, seawater and foreign organisms were manually removed. The samples were immediately immersed in methanol for preservation and initial extraction. A portion of the samples was placed in Nalgene bottles containing a SWBG11 medium for culturing. A small part was also placed in Falcon tubes containing 10 mL RNA later for DNA extraction.

2.2. Identification of the Samples. The sample identification was done based on the morphological observation of the biologist during collection, followed by a deep investigation in the lab by different biologists employing high-resolution microscopes (Table S1). This was done through the cooperation of many marine biologists. There are special methods for identification of the cyanobacteria samples (see supplementary material available here).

2.3. Antifouling Assay. *B. amphitrite*, attached to bamboo poles, were procured from oyster farms in Lake Hamana, Shizuoka, Japan, and were maintained in an aquarium at 20°C by feeding them *Artemia salina* nauplii. Broods released I-II stage nauplii upon immersion in seawater after being dried overnight. The nauplii thus obtained

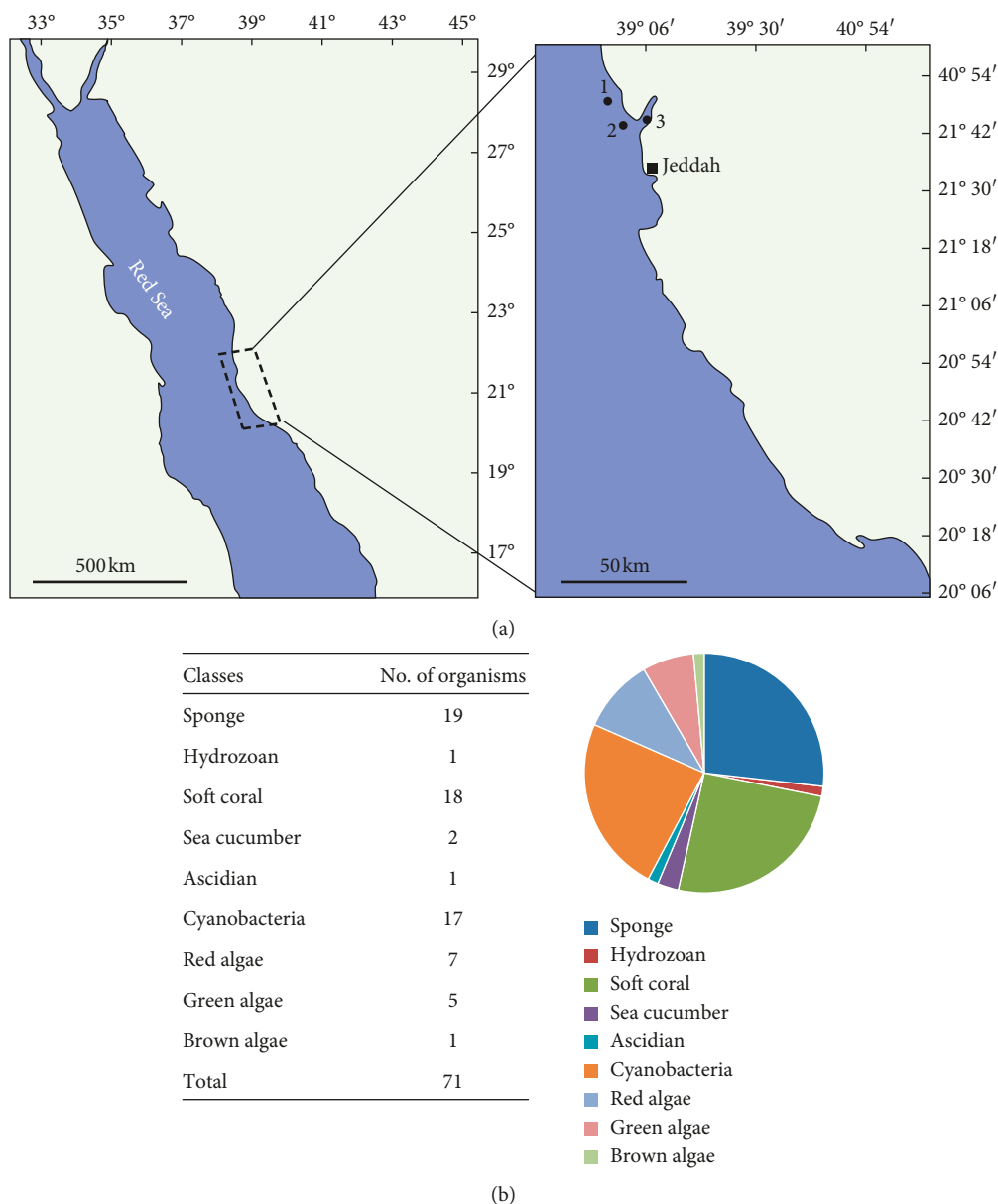


FIGURE 1: Places of collection of the marine samples from Saudi Red Sea territorial water. The map indicates the location of collected marine samples; different classes of the collected marine samples. (a) The map indicates the location of collected marine samples. (b) Different classes of the collected marine samples.

were cultured in 80% filtered seawater (filtered seawater diluted to 80% by deionized water) including penicillin G (20 $\mu\text{g}/\text{mL}$, ICN Biochemical) and streptomycin sulfate (30 $\mu\text{g}/\text{mL}$, Wako Pure Chemical Industries, Ltd.) at 25°C, by feeding them the diatom *Chaetoceros gracillis* (about 40×10^4 cells/mL). Larvae reached the cyprid stage in 5 days. The cyprids were collected and then stored at 4°C until use. The test samples were dissolved in ethanol. Aliquots of the solution were supplied to wells of 24-well polystyrene tissue culture plates and air-dried. 2 mL of 80% filtered seawater, and six 2-day-old cyprids were added to each well. Four wells were used for each concentration. The plates were kept in the dark for 48 h at

25°C, and the number of larvae that attached, metamorphosed, died, floated, or did not settle were counted under a microscope. The initial screening assays of the extracts were performed at 10 and 100 $\mu\text{g}/\text{mL}$ and then the promising extracts were re-evaluated at 1.0, 10, and 100 $\mu\text{g}/\text{mL}$. For isolated compounds, each concentration was repeated 3 times. The antifouling activity of the isolated compounds after 48 hours was expressed as an EC_{50} value, indicating the concentration that reduces the larval settlement to 50% of the control. The EC_{50} values were calculated by a probit analysis. When a probit analysis could not be adopted, then graphical methods were used to decide the EC_{50} values.

2.4. Extraction and Isolation

2.4.1. Extraction of Samples for Screening. The dried material (5–10 g of each organism) was macerated and exhaustively extracted by MeOH (200 mL × 3). The combined extracts were concentrated under a reduced vacuum with a rotary evaporator until dry.

2.4.2. Isolation of Compounds 1–6 from *Sarcophyton glaucum* (SH-21). The fresh soft coral *S. glaucum* (5.0 kg) was minced and exhaustively extracted with a mixture of CH₂Cl₂: MeOH (2:1 v/v, 24 hours for each batch, 22°C, 10 L × 3), and then the combined extracts were concentrated under vacuum and yielded a viscous blackish residue. The residue was partitioned between diethyl ether and water and then the organic layer was dried to give a yellowish material (30.0 g), which was fractionated on normal phase silica gel (NP-silica), eluted stepwise with *n*-hexane containing increasing amounts of diethyl ether, and then increased in polarity with EtOAc. One hundred fractions (F: 1–100) were collected. The fractions were investigated by a TLC pattern using a UV lamp and/or 50% sulfuric acid in methanol as a spraying reagent. The fraction F-3 eluted with *n*-hexane: diethyl ether (19:1, 300.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: diethyl ether (19:1). The band with $R_f = 0.60$ (a violet-red color with sulfuric acid-methanol) was purified to yield compound **5** as colorless oil (9.0 mg). The fraction F-13 eluted with *n*-hexane: diethyl ether (9:1, 120.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: diethyl ether (1:9), to give a band with $R_f = 0.8$ (a violet color with sulfuric acid-methanol) was purified to yield give compound **6** as colorless oil (10.0 mg). The fraction F-3 eluted with *n*-hexane: diethyl ether (19:1, 300.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: diethyl ether (19:1), leading to compound **3** (5.0 mg). The fraction F-13 eluted with *n*-hexane: diethyl ether (4:1, 125.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: diethyl ether (4:1), to give compound **4** (4.5 mg). The fraction F-41 eluted with *n*-hexane: EtOAc (9:1, 123.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: ethylacetate (4:1), leading to compound **2** (12.0 mg). The fraction F-50 eluted with *n*-hexane: EtOAc (4:1, 70.0 mg) was purified by RP-18 HPLC (MeOH/H₂O, 65:35) to yield compound **1** (3.0 mg).

2.4.3. Characterization of Compounds Isolated from *S. glaucum*

(1) 7*R*,8*S*-Dihydroxydepoxy-ent-sarcophine (1) [30]. Colorless crystals, $[\alpha]_D^{25} -125.0$ (*c* 0.18, MeOH); ¹H NMR (CDCl₃, 600 MHz): $\delta_H = 5.59$ (1H, dd, *J* = 10.2, 5.4 Hz, H-2), 4.95 (1H, d, *J* = 10.2 Hz, H-3), 2.19 (2H, m, H_a-5), 1.97 (1H, m, H_b-5), 2.32 (1H, m, H_b-6), 2.11 (1H, m, H_b-6), 3.50 (1H, dd, *J* = 10.8, 1.2 Hz, H-7), 1.82 (1H, m, H_a-9), 1.76 (1H, m, H_b-9), 1.77 (1H, m, H_a-10), 1.29 (1H, m, H_b-10), 5.01 (1H, dd, *J* = 9.6, 4.8 Hz, H-11), 2.05 (1H, m, H_a-13), 2.00 (1H, m, H_b-13), 2.78 (H, ddd, *J* = 13.2, 5.4, 2.4 Hz, H_a-14), 2.13 (1H, ddd, *J* = 18.0, 9.6, 7.8 Hz, H_b-14), 1.80 (3H, s, H-17), 1.87

(3H, s, H-18), 1.23 (3H, s, H-19), 1.61 (3H, s, H-20); ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 175.3$ (C, C-16), 163.2 (C, C-1), 144.4 (C, C-4), 134.9 (C, C-12), 125.2 (CH, C-11), 123.0 (CH, C-15), 121.4 (CH, C-3), 79.5 (CH, C-2), 73.0 (CH, C-7), 75.6 (C, C-8), 56.8 (CH₂, C-9), 37.3 (CH₂, C-13), 35.8 (CH₂, C-5), 35.7 (CH₂, C-14), 27.1 (CH₂, C-6), 24.5 (CH₃, C-19), 23.8 (CH₂, C-10), 16.4 (CH₃, C-20), 15.6 (CH₃, C-18), 9.1 (CH₃, C-17).

(2) ent-Sarcophine (2) [31]. White amorphous, mp 134–136°C; $[\alpha]_D^{25} -82.0$ (*c* 0.20, CHCl₃); ¹H NMR (CDCl₃, 600 MHz): $\delta_H = 5.57$ (1H, dd, *J* = 10.2, 1.2 Hz, H-2), 5.05 (1H, dd, *J* = 10.2, 1.2 Hz, H-3), 2.37 (2H, m, H₂-5), 1.90 (1H, m, H_a-6), 1.69 (1H, m, H_b-6), 2.68 (1H, dd, *J* = 4.8, 4.2 Hz, H-7), 2.10 (1H, m, H_a-9), 1.10 (1H, m, H_b-9), 2.26 (1H, m, H_a-10), 1.93 (1H, m, H_b-10), 5.17 (1H, dd, *J* = 9.6, 5.4 Hz, H-11), 2.18 (1H, ddd, *J* = 18.0, 12.6, 9.0 Hz, H_a-13), 2.01 (1H, ddd, *J* = 13.2, 10.8, 2.4 Hz, H_b-13), 2.78 (H, ddd, *J* = 13.2, 5.4, 2.4 Hz, H_a-14), 2.37 (1H, ddd, *J* = 18.0, 9.6, 7.8 Hz, H_b-14), 1.85 (3H, s, H-17), 1.89 (3H, s, H-18), 1.28 (3H, s, H-19), 1.61 (3H, s, H-20); ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 174.1$ (C, C-16), 162.3 (C, C-1), 144.0 (C, C-4), 135.5 (C, C-12), 124.9 (CH, C-11), 122.9 (CH, C-15), 120.6 (CH, C-3), 78.2 (CH, C-2), 61.4 (CH, C-7), 59.9 (C, C-8), 39.0 (CH₂, C-9), 37.4 (CH₂, C-5), 36.4 (CH₂, C-13), 27.6 (CH₂, C-14), 25.2 (CH₂, C-6), 23.3 (CH₂, C-10), 17.1 (CH₃, C-19), 16.1 (CH₃, C-18), 15.4 (CH₃, C-20), 9.0 (CH₃, C-17).

(3) Guaiacophene (Guai-5,7(11)-dien-8-one (3) [32]. Colorless oil, $[\alpha]_D^{25} -22.0$ (*c* 0.10, CHCl₃); ¹H NMR (CDCl₃, 600 MHz): $\delta_H = 2.38$ (1H, m, H-1), 1.78 (1H, m, H_a-2), 1.70 (2H, m, H_b-2), 1.87 (1H, m, H_a-3), 1.34 (1H, m, H_b-3), 2.54 (1H, sext., *J* = 7.1 Hz, H-4), 6.13 (1H, s, H-6), 2.50 (1H, dd, *J* = 11.4, 6.6 Hz, H_a-9), 2.45 (1H, dd, *J* = 11.4, 7.2 Hz, H_b-9), 2.35 (1H, br. Sep., *J* = 6.8 Hz, H-10), 1.85 (3H, s, H-12), 1.86 (3H, s, H-13), 1.17 (3H, d, *J* = 6.8 Hz, H-14), 0.93 (3H, d, *J* = 6.8 Hz, H-15); ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 207.6$ (C, C-8), 151.7 (C, C-5), 137.7 (C, C-7), 137.1 (C, C-11), 116.9 (CH, C-6), 51.1 (CH₂, C-9), 45.2 (CH, C-1), 40.2 (CH, C-4), 36.7 (CH, C-10), 34.0 (CH₂, C-3), 29.2 (CH₂, C-2), 22.7 (CH₃, C-13), 21.5 (CH₃, C-12), 19.6 (CH₃, C-14), 16.8 (CH₃, C-15).

(4) Gorgosterol (4) [33]. White amorphous powder; ¹H NMR (CDCl₃, 600 MHz): $\delta_H = 1.10$ –2.30 (28H, m), 3.53 (1H, dddd, *J* = 10.8, 10.8, 6.6, 4.8 Hz, H-3), 5.37 (1H, br d, *J* = 2.4 Hz, H-6), 0.68 (3H, s, H-18), 1.02 (3H, s, H-19), 0.93 (3H, d, *J* = 6.6 Hz, H-21), 0.19 (1H, ddd, *J* = 11.4, 9.0, 6.0 Hz, H-22), 0.25 (1H, m, H-24), 0.85 (3H, d, *J* = 6.6 Hz, H-26), 0.79 (3H, d, *J* = 6.6 Hz, H-27), 0.85 (3H, d, *J* = 6.6 Hz, H-28), 0.47 (1H, dd, *J* = 9.0, 4.2 Hz, H_a-29), 0.10 (1H, dd, *J* = 9.0, 4.8 Hz, H_b-29), 0.89 (3H, s, H-30); ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 141.0$ (C, C-5), 121.9 (CH, C-6), 72.0 (CH, C-3), 56.9 (CH, C-14), 53.6 (CH, C-17), 51.0 (CH, C-5), 50.9 (CH, C-9), 43.0 (C, C-13), 42.5 (C, C-4), 39.7 (CH₂, C-12), 37.3 (CH₂, C-1), 36.4 (C, C-10), 35.2 (CH, C-20), 32.2 (CH, C-22), 32.2 (CH, C-25), 31.9 (CH₂, C-7), 31.7 (CH, C-8), 28.8 (CH₂, C-16), 26.0 (C, C-23), 25.0 (CH₂, C-15), 22.4 (CH₃, C-21), 21.9 (CH₃, C-26), 21.8 (CH₃, C-30), 21.6 (CH₃, C-27),

21.1 (CH₂, C-11), 19.6 (CH₃, C-19), 16.0 (CH₃, C-28), 14.7 (CH₂, C-29), 12.0 (CH₃, C-18).

(5) *Guaia-5,11-dien* (5) [32]. Colorless oil, $[\alpha]_D^{25}$ -19.0 (c 0.13, CHCl₃); ¹H NMR (CDCl₃, 600 MHz): δ_H = 2.70 (1H, m, H-1), 1.52 (1H, m, Ha-2), 1.35 (2H, m, Hb-2), 1.50 (1H, m, Ha-3), 1.00 (1H, m, Hb-3), 2.30 (1H, dt, J = 13.2, 6.6 Hz, H-4), 5.19 (1H, dd, J = 7.2, 3.0 Hz, H-6), 2.62 (1H, m, H-7), 1.60 (1H, m, Ha-8), 1.50 (2H, m, Hb-8), 1.64 (1H, m, Ha-9), 1.20 (1H, m, Hb-9), 1.66 (H, m, H-10), 4.57 (H, brs, Ha-12), 4.56 (H, brs, Hb-12), 1.52 (3H, s, H-13), 0.81 (3H, d, J = 6.6 Hz, H-14), 0.64 (3H, d, J = 6.6 Hz, H-15); ¹³C NMR (CDCl₃, 150 MHz): δ_C = 150.8 (C, C-5), 147.2 (C, C-11), 122.4 (CH, C-6), 110.8 (CH₂, C-12), 47.4 (CH, C-7), 45.4 (CH, C-1), 41.0 (CH, C-4), 34.0 (CH₂, C-3), 30.2 (CH₂, C-2), 25.9 (CH₂, C-8), 21.6 (CH₃, C-13), 19.9 (CH₃, C-14), 19.6 (CH₃, C-15), 15.7 (CH₃, C-14).

(6) *Calamenene* (6) [34]. Yellow oil; ¹H NMR (CDCl₃, 600 MHz): δ_H = 2.26 (1H, m, H-1), 1.98 (1H, m, Ha-2), 1.36 (2H, m, Hb-2), 1.85 (1H, m, Ha-3), 1.65 (1H, m, Hb-3), 2.34 (1H, m, H-4), 7.05 (1H, s, H-5), 7.16 (1H, d, J = 7.8 Hz, H-7), 6.99 (1H, d, J = 7.8 Hz, H-8), 1.34 (2H, d, J = 7.2 Hz, H-11), 2.79 (1H, m, H-12), 0.75 (3H, d, J = 7.2 Hz, H-13), 1.03 (3H, d, J = 7.2 Hz, H-14), 2.33 (3H, s, H-15); ¹³C NMR (CDCl₃, 150 MHz): δ_C = 140.3 (C, C-9), 140.1 (C, C-10), 134.7 (C, C-6), 129.0 (CH, C-5), 127.0 (CH, C-8), 126.4 (CH, C-7), 43.0 (CH, C-4), 32.5 (CH, C-12), 30.8 (CH, C-1), 30.5 (CH₂, C-2), 22.3 (CH₃, C-11), 21.3 (CH₃, C-14), 21.1 (CH₃, C-15), 17.4 (CH₃, C-13).

2.4.4. Isolation of Compounds 7–12 from *Sarcophyton trocheliophorum* (SH-13). Soft coral material was washed with water and dried in the shade at room temperature. The dried material (79.0 g) was exhaustively extracted with equal volumes of CHCl₃/MeOH (2 × 6 L, 24 h for each batch) at room temperature. The residue (20.2 g) was partitioned between CH₂Cl₂ and water; the CH₂Cl₂ soluble material was dried to give a brownish oily material (9.2 g) and chromatographed on NP (Merck, 60G) column chromatography employing *n*-hexane/CH₂Cl₂, followed by EtOAc/MeOH mixtures with increasing polarity. Fractions of ~100 mL were collected. TLC was carried out by employing silica gel chromatoplates, appropriate solvent system, and 50% H₂SO₄ in MeOH as spraying reagent. Fractions containing a single compound were combined and further purified by preparative TLC of glass-supported silica gel plates (20 cm × 20 cm) of 250 μm thickness. The fraction eluted with *n*-hexane: methylene chloride (19:1, 37.0 mg) was purified by preparative TLC using solvent system *n*-hexane: methylene chloride (9:1). The fraction eluted with *n*-hexane (9:1, 36.0 mg) was purified by preparative TLC using solvent system *n*-hexane: methylene chloride (4:1). The band with R_f = 0.70 (Violet appearance under UV₂₅₄, and brown color with H₂SO₄-MeOH) was taken to give compound 7 as colorless oil (5.3 mg, 0.0067% dry weight). The fraction eluted with *n*-hexane: methylene chloride (7:3, 125.0 mg) was purified by preparative TLC using the solvent system *n*-

hexane: diethyl ether (8:2), to give two bands. The first band with R_f = 0.71 (violet color with sulfuric acid-methanol) was taken to give colorless oil (16.0 mg) of compound 10. The fraction eluted with *n*-hexane: EtOAc (9:1, 123.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: ethylacetate (8:2). The band with R_f = 0.50 (reddish color with sulfuric acid-methanol) was taken to give colorless oil (12.0 mg) of compound 11. The fraction eluted with *n*-hexane: EtOAc (8:2, 70.0 mg) was purified by RP-18 HPLC (MeOH/H₂O, 65:35) yielded compound 12 (3.0 mg). The fraction eluted with CH₂Cl₂:MeOH (9:1, 35.0 mg) was purified by preparative TLC using the solvent system *n*-hexane: EtOAc (1:4) to give two bands. The first band with R_f = 0.38 (brown color with H₂SO₄-MeOH) was taken to give colorless oil (3.5 mg) of compound 8. The second band with R_f = 0.35 (brown color with H₂SO₄-MeOH) was taken to give colorless oil (3.6 mg) of compound 9.

2.4.5. Characterization of Compounds Isolated from *S. trocheliophorum*

(1) *Trocheliene* (7) [35]. Gummy material; $[\alpha]_D^{20}$ -22.0 (c 0.02, C₆H₆); IR_{max} (film) cm⁻¹: 3050–2700 (CH), 1630, 1620 (C=C), 1510, 925, 740; ¹H NMR (CDCl₃, 600 MHz): δ_H = 2.01–2.05 (1H, m, H-2), 2.15–2.20 (1H, m, H-3a), 2.10–2.15 (1H, m, H-3b), 4.98 (1H, ddd, J = 12.0, 6.0, 1.2 Hz, H-4), 1.98–2.04 (1H, m, H-6a), 1.56–1.58 (1H, m, H-6b), 1.92–1.94 (1H, m, H-7a), 1.32–1.36 (1H, m, H-7b), 5.19 (1H, ddd, J = 15.0, 7.2, 1.2 Hz, H-8), 1.64–1.70 (1H, m, H-10a), 1.30–1.33 (1H, m, H-10b), 1.92–1.98 (1H, m, H-11a), 1.76–1.80 (1H, m, H-11b), 5.06 (1H, ddd, J = 12.6, 6.6, 1.2 Hz, H-13), 2.01–2.10 (1H, m, H-14a), 1.44–1.48 (1H, m, H-14b), 2.20–2.26 (1H, m, H-15), 0.71 (3H, d, J = 6.6 Hz, H-16), 0.99 (3H, d, J = 6.6 Hz, H-17), 1.57 (3H, s, H-18), 1.59 (3H, s, H-19), 1.00 (3H, s, H-20), 2.68–2.69 (1H, m, H-21), 1.81–1.86 (1H, m, H-22a), 1.57–1.61 (1H, m, H-22b), 2.12 (1H, d, J = 6.6 Hz, H-24a), 2.06 (1H, d, J = 6.6 Hz, H-24b), 7.02 (1H, s, H-26), 6.94 (1H, br d, J = 7.8 Hz, H-28), 7.11 (1H, br d, J = 7.8 Hz, H-29), 2.75 (1H, hex, H-31), 1.31–1.39 (1H, m, H-32a), 1.21–1.29 (1H, m, H-32b), 1.26–1.30 (1H, m, H-33a), 1.30–1.32 (1H, m, H-33b), 2.20–2.23 (1H, m, H-36a), 2.16–2.20 (1H, m, H-36b), 1.56 (3H, s, H-37), 4.65 (1H, d, J = 2.4 Hz, H-38a), 4.70 (1H, d, J = 2.4 Hz, H-38b), 2.32 (3H, s, H-39), 1.26 (3H, d, J = 6.6 Hz, H-40); ¹³C NMR (CDCl₃, 150 MHz): δ_C = 39.9 (C-1), 45.9 (C-2), 38.9 (C-3), 125.9 (C-4), 133.9 (C-5), 32.4 (C-6), 30.8 (C-7), 124.0 (C-8), 140.0 (C-9), 28.1 (C-10), 33.9 (C-11), 149.3 (C-12), 121.8 (C-13), 39.4 (C-14), 31.8 (C-15), 17.3 (C-16), 21.3 (C-17), 15.5 (C-18), 15.3 (C-19), 16.6 (C-20), 43.7 (C-21), 21.4 (C-22), 150.3 (C-23), 23.7 (C-24), 139.9 (C-25), 128.7 (C-26), 134.4 (C-27), 126.1 (C-28), 126.7 (C-29), 134.8 (C-30), 32.5 (C-31), 29.6 (C-32), 22.7 (C-33), 128.7 (C-34), 133.4 (C-35), 24.8 (C-36), 18.0 (C-37), 110.1 (C-38), 21.1 (C-39), 22.3 (C-40).

(2) *Sarcotrocheldiol A* (8) [35]. Colorless oil; $[\alpha]_D^{20}$ 62.4 (c 0.012, CHCl₃); IR_{max} (film) cm⁻¹: 3423 (OH), 3180 (OH), 2937 (C-H), 1645 (C=C), 1378, 1221, 1045; ¹H NMR (CDCl₃, 600 MHz): δ_H = 1.34–1.36 (1H, m, H-1), 4.62 (1H,

dd, $J = 10.8, 4.8$ Hz, H-2), 5.28 (1H, d, $J = 10.8$ Hz, H-3), 2.77 (1H, dd, $J = 12.0, 10.8$ Hz, H-5a), 2.56 (1H, dd, $J = 12.0, 4.2$ Hz, H-5b), 5.87 (1H, $J =$ ddd, 15.6, 10.8, 4.2 Hz, H-6), 5.41 (1H, br d, $J = 15.6$ Hz, H-7), 1.86–1.88 (1H, m, H-9a), 1.54–1.56 (1H, m, H-9b), 1.74–1.76 (1H, m, H-10a), 1.26–1.30 (1H, m, H-10b), 3.48 (1H, d, $J = 9.6$ Hz, H-11), 1.62–1.64 (1H, m, H-13a), 1.35–1.36 (1H, m, H-13b), 2.28–2.32 (1H, m, H-14a), 1.20–1.22 (1H, m, H-14b), 1.21–1.23 (1H, m, H-15), 0.88 (3H, d, $J = 6.6$ Hz, H-16), 0.76 (3H, d, $J = 6.6$ Hz, H-17), 1.83 (3H, s, H-18), 1.33 (3H, s, H-19), 1.04 (3H, s, H-20), ^{13}C NMR (CDCl_3 , 150 MHz): $\delta_{\text{C}} = 45.6$ (CH, C-1), 72.0 (CH, C-2), 124.4 (CH, C-3), 141.1 (C, C-4), 42.5 (CH₂, C-5), 128.2 (CH, C-6), 135.7 (C, C-7), 73.6 (C, C-8), 39.9 (CH₂, C-9), 22.8 (CH₂, C-10), 75.8 (CH₂, C-11), 74.9 (CH₂, C-12), 18.6 (C, C-13), 34.1 (CH, C-14), 29.4 (CH, C-15), 20.7 (CH₃, C-16), 20.5 (CH₃, C-17), 17.6 (CH₃, C-18), 29.3 (CH₃, C-19), 23.8 (CH₃, C-20).

(3) *Sarcotrocheldiol B* (**9**) [35]. Colorless oil [α]_D²⁰ 89.1 (c 0.010, CHCl_3); IR_{max} (film) cm^{-1} : 3383 (OH), 3180 (OH), 2937 (C–H), 1645 (C=C), 1378, 1221, 1045; ^1H NMR (CDCl_3 , 600 MHz): $\delta_{\text{H}} = 1.29$ – 1.32 (1H, m, H-1), 4.52 (1H, $J =$ dd, 10.8, 5.4 Hz, H-2), 5.23 (1H, d, $J = 10.8$ Hz, H-3), 2.13–2.15 (1H, m, H-5a), 2.17–2.19 (1H, m, H-5b), 1.88–1.90 (1H, m, H-6a), 2.12–2.15 (1H, m, H-6b), 3.88 (1H, dd, $J = 10.8, 1.2$ Hz, H-7), 2.48–2.51 (1H, m, H-9a), 2.28–2.32 (1H, m, H-9b), 1.95–1.97 (1H, m, H-10a), 1.28–1.30 (1H, m, H-10b), 3.76 (1H, d, $J = 9.6$ Hz, H-11), 1.62–1.64 (1H, m, H-13a), 1.38–1.40 (1H, m, H-13b), 2.26–2.28 (1H, m, H-14a), 1.26–1.28 (1H, m, H-14b), 1.18–1.20 (1H, m, H-15), 0.86 (3H, d, $J = 6.6$ Hz, H-16), 0.71 (3H, d, $J = 6.6$ Hz, H-17), 1.73 (3H, s, H-18), 5.07 (1H, br d, $J = 1.2$ Hz, H-19a), 5.06 (1H, br d, $J = 1.2$ Hz, H-19b), 1.05 (1H, s, H-20), ^{13}C NMR (CDCl_3 , 150 MHz): $\delta_{\text{C}} = 45.9$ (CH, C-1), 70.4 (CH, C-2), 125.3 (CH, C-3), 137.7 (C, C-4), 36.4 (CH₂, C-5), 30.1 (CH₂, C-6), 67.8 (CH, C-7), 147.5 (C-8), 30.9 (CH₂, C-9), 27.1 (CH₂, C-10), 70.6 (CH, C-11), 74.6 (C, C-12), 18.6 (CH₂, C-13), 33.7 (CH₂, C-14), 29.0 (CH₃, C-15), 20.7 (CH₃, C-16), 20.3 (CH₃, C-17), 14.2 (CH₃, C-18), 112.4 (CH₃, C-19), 23.5 (CH₃, C-20).

(4) *Deoxosarcophine* (**10**) [36]. Colorless oil; ^1H NMR (CDCl_3 , 600 MHz): $\delta_{\text{H}} = 5.54$ (1H, brd, $J = 9$ Hz, H-2), 5.23 (1H, brd, $J = 9$ Hz, H-3), 2.35 (1H, m, H-4), 1.90 (1H, m, Ha-5), 1.63 (1H, m, Hb-5), 2.7 (H, t, $J = 7.2$ Hz, H-6), 2.10 (1H, m, Ha-8), 1.00 (1H, m, Hb-8), 2.25 (1H, m, Ha-9), 1.9 (1H, m, Hb-9), 5.10 (1H, dd, $J = 6, 4.8$ Hz, H-10), 2.55 (1H, m, Ha-12), 1.91 (1H, m, Hb-12), 1.66 (1H, m, H-13), 4.50 (2H, m, H-16), 1.65 (3H, s, H3-17), 1.61 (3H, s, H3-18), 1.27 (3H, s, H3-19), 1.83 (3H, br s, H-20); ^{13}C NMR (CDCl_3 , 150 MHz): $\delta_{\text{C}} = 128.0$ (C, C-14), 83.7 (CH, C-1), 126.3 (CH, C-2), 139.5 (C, C-3), 38.0 (CH₂, C-4), 25.3 (CH₂, C-5), 62.0 (CH, C-7), 60.0 (C, C-7), 23.5 (CH₂, C-8), 39.9 (CH₂, C-9), 123.6 (CH, C-10), 136.8 (C, C-11), 36.9 (CH₂, C-13), 26.1 (CH₂, C-14), 131.4 (C, C-15), 78.3 (CH₂, C-16), 10.2 (CH₃, C-17), 15.1 (CH₃, C-18), 16.9 (CH₃, C-19), 15.6 (CH₃, C-20).

(5) *Sarcotrocheliol* (**11**) [37]. Colorless oil; ^1H NMR (CDCl_3 , 600 MHz): $\delta_{\text{H}} = 1.32$ (H, m, H-1), 4.53 (1H, dd, $J = 10.2, 5.4$ Hz, H-2), 5.27 (1H, d, $J = 10.2$ Hz, H-3), 2.19 (1H, m, Ha-

5), 1.97 (1H, m, Hb-5), 2.32 (1H, m, Ha-6), 2.11 (1H, m, Hb-6), 5.00 (1H, dd, $J = 10.2, 5.4$ Hz, H-7), 2.40 (1H, m, Ha-9), 1.97 (1H, m, Hb-9), 1.77 (1H, m, Ha-10), 1.29 (1H, m, Hb-10), 3.88 (1H, d, $J = 9.6$ Hz, H-11), 1.60 (1H, m, Ha-13), 1.40 (1H, m, Hb-13), 2.36 (1H, m, Ha-14), 1.22 (1H, m, Hb-14), 1.18 (1H, m, H-15), 0.88 (3H, d, $J = 6.6$ Hz, H-16), 0.73 (3H, d, $J = 6.6$ Hz, H-17), 1.65 (3H, s, H-18), 1.61 (3H, s, H-19), 1.02 (3H, s, H-20); ^{13}C NMR (CDCl_3 , 150 MHz): $\delta_{\text{C}} = 46.6$ (CH, C-1), 71.1 (CH, C-2), 125.4 (CH, C-3), 138.6 (C, C-4), 39.9 (CH₂, C-5), 25.2 (CH₂, C-6), 124.1 (CH, C-7), 136.0 (C, C-8), 33.7 (CH₂, C-9), 18.7 (CH₂, C-10), 31.9 (CH₂, C-11), 35.2 (CH₂, C-12), 75.0 (C, C-13), 71.9 (C, C-14), 29.0 (CH, C-15), 20.8 (CH₃, C-16), 20.3 (CH₃, C-17), 24.2 (CH₃, C-18), 17.3 (CH₃, C-19), 15.1 (CH₃, C-20).

(6) *Sarcotrocheliol Acetate* (**12**) [37]. Colorless oil; ^1H NMR (CDCl_3 , 600 MHz): $\delta_{\text{H}} = 1.26$ (1H, m, H-1), 4.52 (1H, dd, $J = 10.8, 5.4$ Hz, H-2), 5.50 (1H, d, $J = 10.8$ Hz, H-3), 2.15 (1H, m, H-5a), 2.10 (1H, m, H-5b), 2.30 (1H, m, H-6a), 2.07 (1H, m, H-6b), 5.05 (1H, dd, $J = 10.2, 4.8$ Hz, H-7), 1.97 (1H, m, H-9a), 1.63 (1H, m, H-9b), 1.66 (1H, m, H-10a), 1.50 (1H, m, H-10b), 5.37 (1H, d, $J = 10.2$ Hz, H-11), 1.51 (1H, m, H-13a), 1.23 (1H, m, H-13b), 1.77 (1H, m, H-14a), 1.28 (1H, m, H-14b), 1.17 (1H, m, H-15), 0.69 (3H, d, $J = 6.6$ Hz, H-16), 0.82 (3H, d, $J = 6.6$ Hz, H-17), 1.62 (3H, s, H-18), 1.56 (3H, s, H-19), 1.04 (3H, s, H-20), 2.06 (3H, s, CH_3CO); ^{13}C NMR (CDCl_3 , 150 MHz): $\delta_{\text{C}} = 171.0$ (C, C=O), 139.0 (C, C-4), 135.0 (C, C-8), 125.3 (CH, C-3), 124.5 (CH, C-7), 73.7 (C, C-12), 73.5 (CH, C-11), 71.4 (CH, C-2), 46.5 (CH, C-1), 39.8 (CH₂, C-5), 34.4 (CH₂, C-9), 34.3 (CH₂, C-14), 29.0 (CH₂, C-10), 29.0 (CH, C-15), 25.4 (CH₃, C-20), 25.3 (CH₂, C-6), 20.7 (CH₃, C-17), 20.3 (CH₃, C-16), 19.0 (CH₂, C-14), 17.0 (CH₃, C-19), 15.0 (CH₃, C-18), 21.3 (CH₃, CH_3CO).

3. Results and Discussion

Extracts of seventy-one marine organisms, representing seven different classes of marine fauna and flora, were assessed for their antifouling activity. These organisms were taxonomically identified, and their descriptions are listed in Table S1. Figure 1 illustrates the places of collection of the Red Sea samples. The samples were dried by standard methods and extracted with methanol. The extracts were divided into two batches and evaluated for their antifouling effects. The antifouling effects are listed in Tables S2, 1 and 2. The first screening batch is illustrated in Figure S1, and the second batch is illustrated in Figures S2–S5. Analysis of the antifouling activity indicated that the value settlement rates and mortality range from 0–100%.

Thirty-four samples are considered to be active (Figure S6). These samples consist of five classes: sponges, algae, tunicate, soft corals, and cyanobacteria. Four samples were highly potent at 1 $\mu\text{g}/\text{mL}$ (Figure S7) with a percentage of settlement inhibition lower than 31%; thirteen samples were potent at 10 $\mu\text{g}/\text{mL}$ with a percentage of settlement inhibition between 16 and 30%, and nineteen samples were potent at 10 $\mu\text{g}/\text{mL}$ with a percentage of settlement inhibition between 0 and 15%. Potent antifouling activities were exhibited by *Siphonochalina siphonella*, *Sarcophyton*

TABLE 1: Antifouling activity of the promising marine organisms.

Sample no.	Classes	Name of species	Settlement rate ¹	Mortality ²
SH-03		<i>Siphonochalina siphonella</i>	0	0
SH-07		<i>Hyrtios erectus</i>	29	0
SH-26		<i>Halichlona</i> sp. 2	15	0
SH-29	Sponges	<i>Hyrtios</i> sp.	0	40
SH-36		<i>Dysideid</i> sponge	9	36
SH-40		<i>Callyspongia</i> sp.	12	10
SH-41		<i>Callyspongia</i> sp.	29	16
SH-04		<i>Sinularia polydactyla</i>	40	0
SH-11		<i>Cespitularia</i> sp.	0	0
SH-13		<i>Sarcophyton trocheliophorum</i>	0	0
SH-21		<i>Sarcophyton glaucum</i>	0	4
SH-23		<i>Dendronephytia</i> sp.	25	0
SH-24		<i>Sinularia leptoclades</i>	7	0
SH-54	Soft corals	<i>Sinularia</i> sp. 1	23	0
SH-55		<i>Sinularia</i> sp. 2	0	0
SH-56		<i>Lobophyton</i> sp.	19	0
SH-58		Soft coral K	22	0
SH-60		<i>Dendronephytia</i>	18	0
SH-61		Sponge N	16	0
SH-62		<i>Strenophyta</i>	24	0
SH-35	Sea firs	<i>Sertularia</i> sp.	9	2
SH-42	Tunicates	Ascidian sp.	28	0
SH-46		<i>Symploca</i> sp.	1	88
SH-47		Blue-green algae	9	0
SH-48		Blue-green algae	2	90
SH-49		Blue-green algae	0	49
SH-50	Cyanobacteria	Blue-green algae	20	67
SH-52		Blue-green algae	21	25
SH-65		Blue-green algae	6	0
SH-66		Blue-green algae	9	18
SH-67		Blue-green algae	19	79
SH-34	Algae	<i>Halimeda tuna</i>	12	15
SH-37		<i>Laurencia</i> sp.	25	17
Control ³	—	—	92	2

¹% settlement rates: 48 hr exposing the larvae to extracts at a concentration of 10 $\mu\text{g}/\text{mL}$. ²% mortality rates: 120 hr exposing the larvae to extracts at a concentration of 10 $\mu\text{g}/\text{mL}$. ³Control (no samples) of 4 times repeated.

TABLE 2: Statistical analysis of the bioactive marine organisms.

Marine classes	Marine classes	Tested sample	Active sample ¹
Algae	13	13	1
Sponges	19	19	3
Soft corals	18	18	8
Cyanobacteria	17	17	3
Others	4	4	1
Total	71	71	16

¹Those which showed under 20% larval settlement rates and low toxicity at 20 $\mu\text{g}/\text{well}$ (<20% lethality).

glaucum, *Sinularia leptoclades*, and *Hyrtios* species. They showed 90% settlement inhibition against barnacle cyprids' larvae at 10 $\mu\text{g}/\text{mL}$. In particular, *Hyrtios* sp. showed less toxicity to barnacle larva even at 100 $\mu\text{g}/\text{mL}$, which suggests environmentally benign and potent antifouling compounds (Table 1).

The chromatographic investigation of the extracts indicated that the diversity of the metabolites was distributed among the soft coral, sponges, and algal samples. Five samples were selected for further investigation: *Sarcophyton*

glaucum, *Sarcophyton trocheliophorum*, *Hyrtios* sp., *Laurencia obtuse*, and *Siphonochalina siphonella*. Based on the combination of their antifouling results and their historical productivity recorded in the literature, genus *Sarcophyton* are recognized as a rich source of macrocyclic cembrane-type diterpenoids and biscembranoids. To date, more than 300 natural cembranoid derivatives have been reported. Cembrane-type diterpenoids are a large family with diverse functions that are obtained from both terrestrial and marine organisms. They usually exhibit cyclic ether, lactone, or furan moieties linked to the cembrane framework. The cembrane derivatives play an important role from a biomedical perspective [38–44].

Hyrtios sp. showed promising antifouling activities, and their reported chemical profile indicated that they are a rich source of bioactive compounds such as sesterterpenes [45], sesquiterpenes [46], and macrolides. In addition to these metabolites, they produce indole and β -carboline alkaloids [47, 48]. Unfortunately, the quantity of the collected sample was not sufficient for chemical investigation, even after several attempts to collect the same *Hyrtios* sp. Finally, the

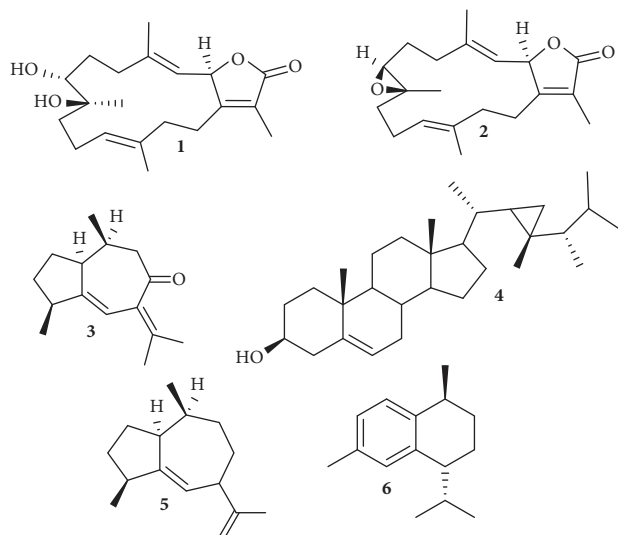


FIGURE 2: Compounds isolated from *Sarcophyton glaucum* 1–6.

antifouling activities of the remaining two samples (*Laurencia obtusa* and *Siphonochalina siphonella*) have been previously reported [49].

In the current study, processing of two soft coral samples (*Sarcophyton glaucum* (SH-21) and *Sarcophyton trocheliophorum* (SH-13)) was also presented. Investigation of *S. glaucum* (SH-21) has been done by employing different chromatographic techniques and led to isolation of 7R,8S-dihydroxydepoxy-ent-sarcophine (1), ent-sarcophine (2), guaia-5,7(11)-dien-8-one (3), gorgosten-5(E)-3 β -ol (4), guaia-5,11-dien (5), and calamenene (6) (Figure 2). *S. trocheliophorum* (SH-13) was fractionated and led to isolation of trocheliene (7), sarcotrocheldiol A (8) and sarcotrocheldiol B (9), deoxosarcophine (10), sarcotrocheliol (11), and sarcotrocheliol acetate (12) (Figure 3). All isolated compounds (1–12) [30–37] have been evaluated for their antifouling activities, and the results are shown in Figure 4.

It is important to emphasize that the isolated compounds showed antifouling effects which registered at 48 and 120 hours, which gives us a better idea of their toxicity. All the compounds, except 4, 6, and 10–12, showed EC₅₀ values more than 10 μ g/mL. The EC₅₀ values of the promising compounds (4, 6, and 10–12) were in the 1.1–2.9 μ g/mL range at 48 hours, while their EC₅₀ values were in the 1.24–10 μ g/mL range at 120 hours, as illustrated in Figure 4. Gorgosterol (4) showed EC₅₀ values of 1.69 μ g/mL at 48 hours and 10 μ g/mL at 120 hours, respectively. It is a C-30 steroid, which is isolated amongst the metabolites from *Sarcophyton glaucum*. Its antifouling activity is more potent than the similar reported compounds. It is important to mention here that an isolated steroid from the gorgonian coral *Subergorgia suberosa* inhibited the settlement of *Balanus neritina* larvae with EC₅₀ values of 6.25 and 7.8 μ g/mL, respectively, and LD₅₀ > 250 μ g/mL [50].

Calamenene (6) showed EC₅₀ values of 2.79 μ g/mL at 48 hours and 9.35 μ g/mL at 120 hours, respectively. It is sesquiterpenoidal in nature. The obtained antifouling results of compound 6 are in good agreement with the reported

activity of sesquiterpenes [23]. Compound (6) could be different in functionality from the reported antifouling sesquiterpenes while still keeping its terpenoidal scaffold. Chamigreene sesquiterpene is one of well-known potent antifouling metabolites, which was isolated from the red alga *Laurencia elata*; unfortunately, a leak in the supply prevented its development to biopaints. Moreover, avarol and avarone are sesquiterpenoidal derivatives which were isolated from *Dysidea avara*. They exhibited antifouling against cyprids of the barnacle *Balanus amphitrite*, with ED₅₀ values of 0.65 and 3.41 mg/mL, respectively, but their toxicity was low (LD₅₀ 13.3 and 27.2 mg/mL, respectively, against cyprids and 1.58 and 25.2 mg/mL, respectively, against nauplii) [23].

The compounds deoxosarcophine (10), sarcophytol (11), and sarcotrocheliol acetate (12) showed EC₅₀ values of 1.6, 1.12, and 2.9 μ g/mL at 48 hours and 1.8, 1.24, and 2.78 μ g/mL at 120 hours, respectively. These compounds are macrocyclic cembranoid in nature. Our results are closely similar to the previously published cembranoidal derivatives. For instance, pukalide, a furanocembranoid diterpene reported originally from the soft coral *Sinularia abrupta* and also from the gorgonian *Leptogorgia virgulata*, displayed potent inhibition towards the larval settlement of *Balanus amphitrite* larvae with EC₅₀ = 19 ng/mL [50–52]. The cembranoid alcohols, sinulariols J, P, and Y, isolated from the soft coral *Sinularia rigida* showed potent antifouling effects against the larval settlement of *B. amphitrite* and *Bugula neritina* larvae with EC₅₀ < 14.03 μ g/mL [53]. Pavidolides C and D, cembranoids with unusual C-5-C-9 and C-3-C-7 linkages, respectively, reported from the soft coral *Sinularia pavidata*, have been shown to inhibit settlement of *B. amphitrite* larvae with ED₅₀ values of 4.32 and 2.12 μ g/mL, respectively, and low cytotoxicity (LD₅₀ > 50 μ g/mL) [54]. Fortunately, our findings show that deoxosarcophine (10) and sarcophytol (11) are more potent than the aforementioned published compounds. The presence of some functionality such as epoxide, lactone, and a macrocyclic ring may play an important role in their effects.

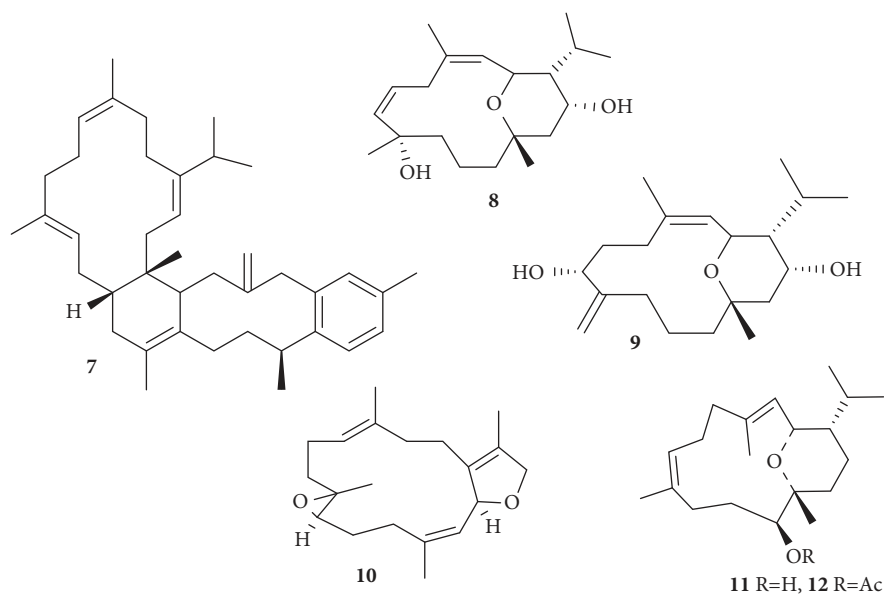
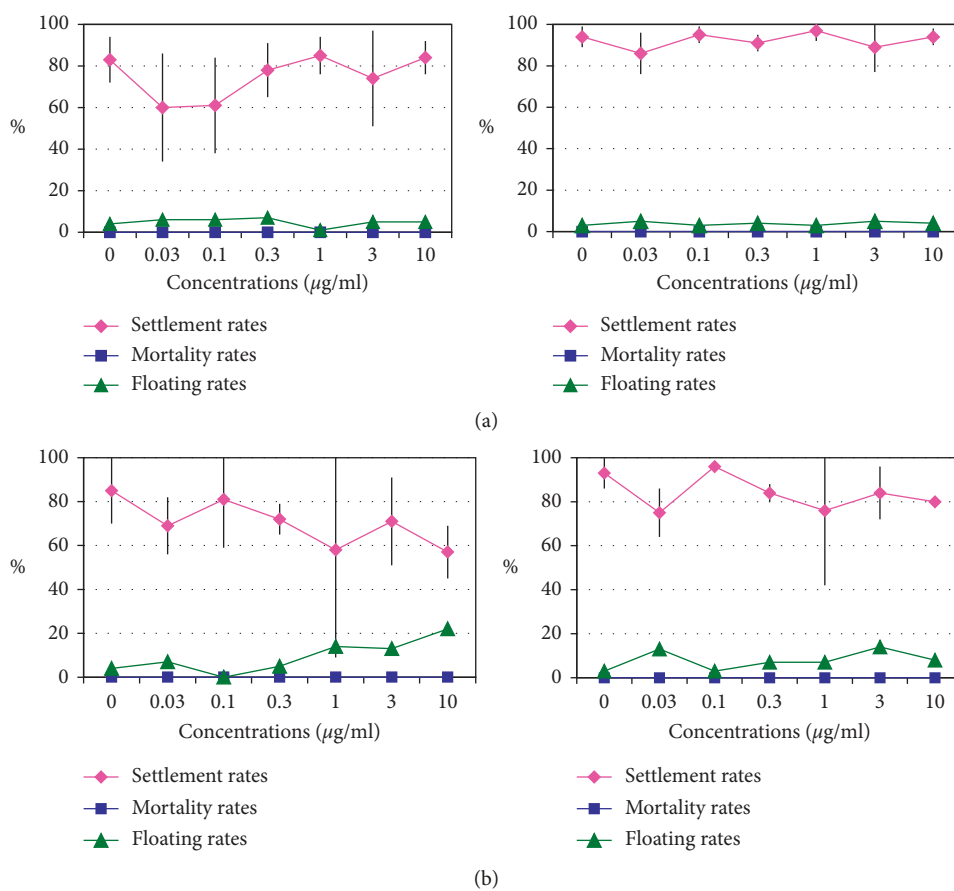
FIGURE 3: Compounds isolated from *Sarcophyton trocheliophorum* 7–12.

FIGURE 4: Continued.

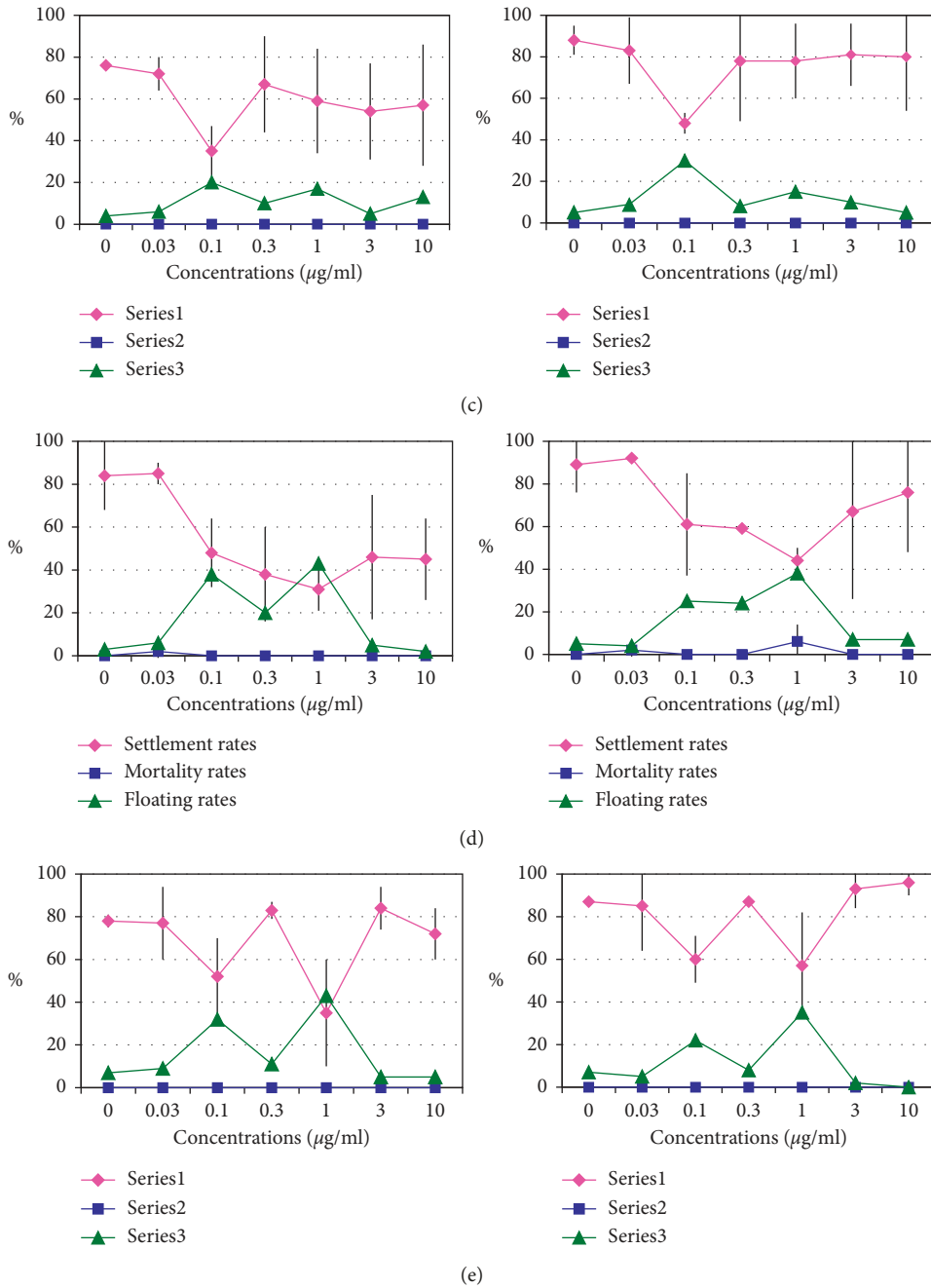


FIGURE 4: Continued.

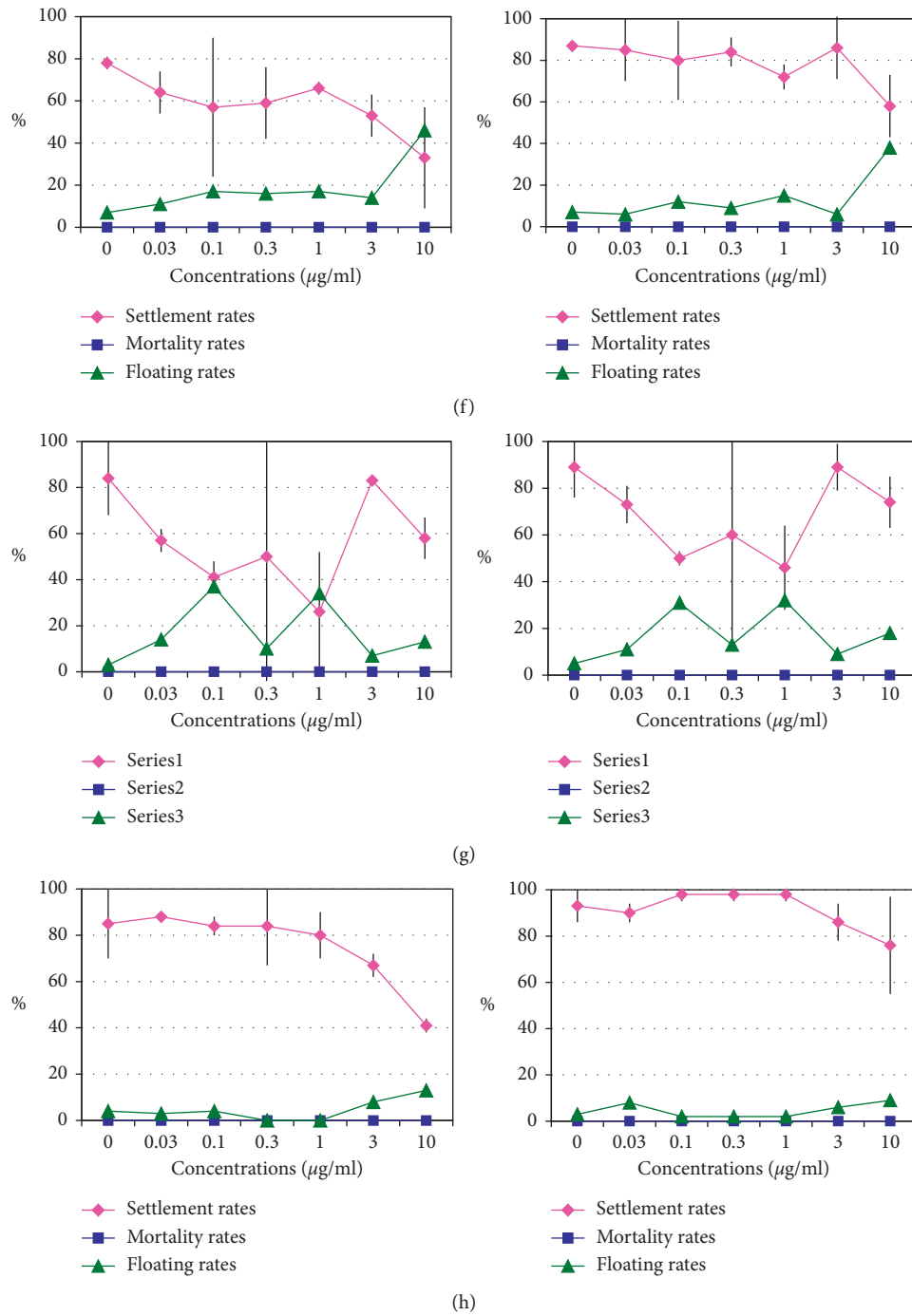
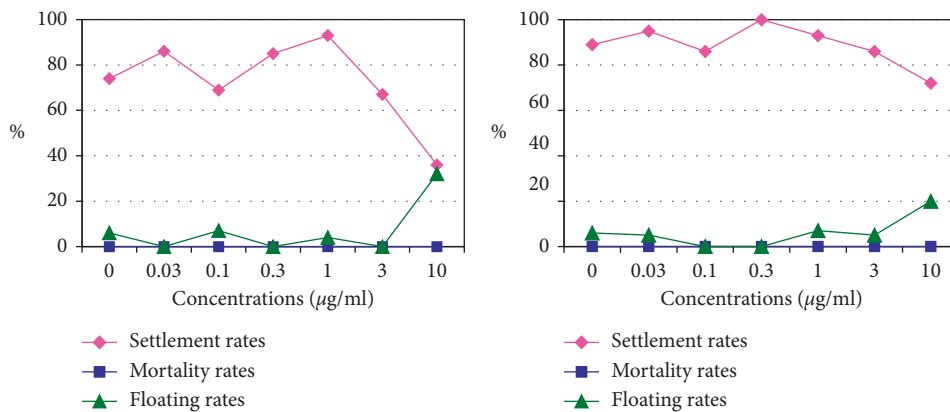
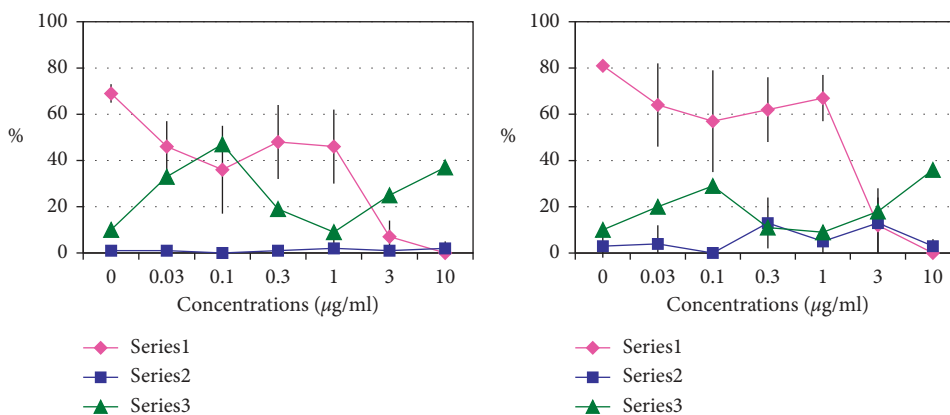


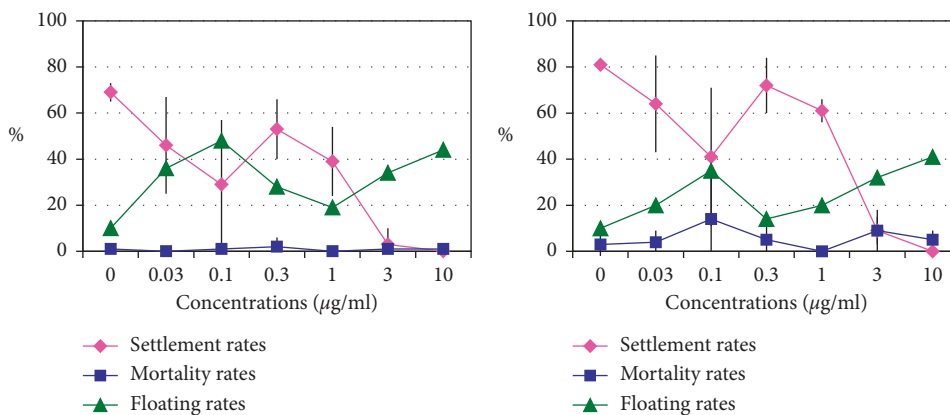
FIGURE 4: Continued.



(i)



(j)



(k)

FIGURE 4: Continued.

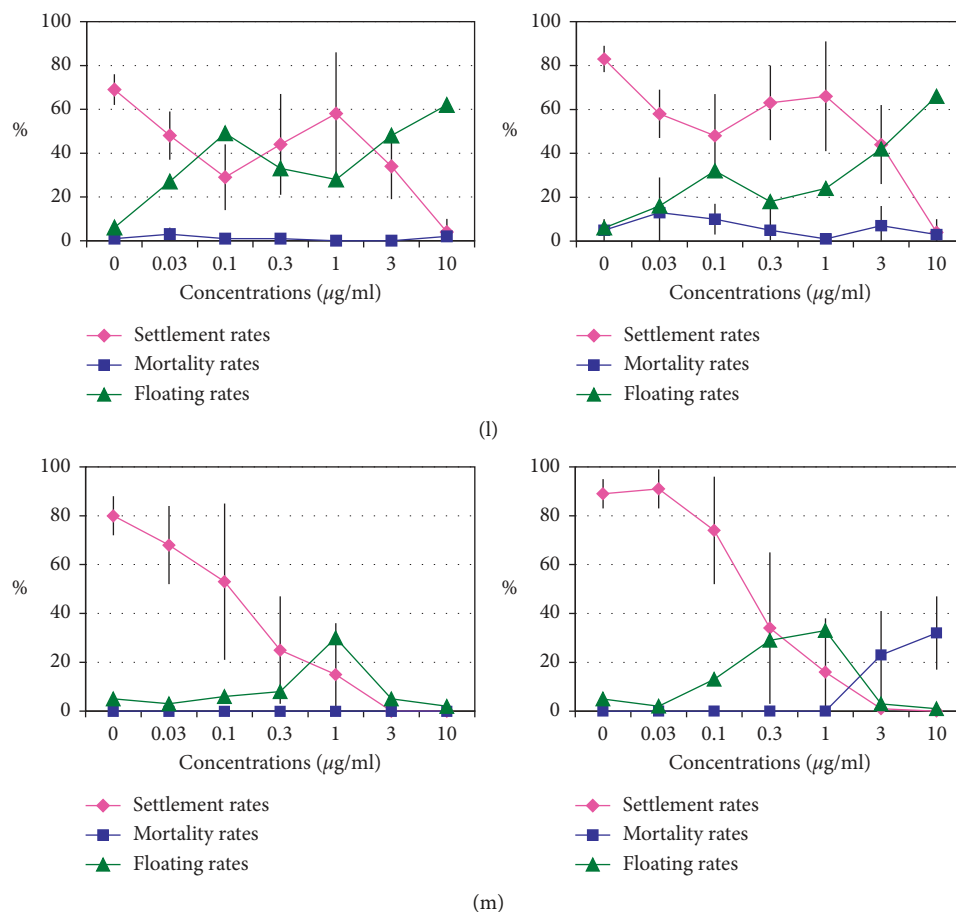


FIGURE 4: Antifouling activity of isolated compounds 1–12.

Moreover, their potency is comparable to that of 12-hydroxyisolaurene, which was reported in our previous paper on antifouling metabolites from Red Sea organisms [41]. A recent review listed ten additional antifouling cembranoids against barnacle larvae [55].

Several studies have investigated the antifouling metabolites isolated from marine invertebrates, particularly, sponges and soft corals. Both organisms have established an impressive warehouse of chemical defense systems against biofouling. It was interesting that the majority of friendly antifouling metabolites, identified in the last 30 years, belong to several natural classes: terpenoids (i.e., sesquiterpenes and cembranoidal diterpenes), alkaloids, and steroids [19, 27, 28]. Accordingly, our sample collection was directed towards the marine organisms which can produce such metabolites. It is wealthy that these samples, collected based on the previous reported data of the chemical and biological diversity, led to our obtained results.

4. Conclusion

Seventy-one marine samples, including sponges, algae, tunicate, sea cucumber, and soft corals, were collected and extracted with methanol. Thirty-three samples showed antifouling activity; four were highly potent at 1 µg/mL. Two

promising extracts were purified by employing several chromatographic techniques, which led to the isolation of 12 compounds. They were proven to possess potent antifouling activity with EC_{50} values less than 10 µg/mL.

Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.

Conflicts of Interest

The authors declare that there are no conflicts of interest.

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Supplementary Materials

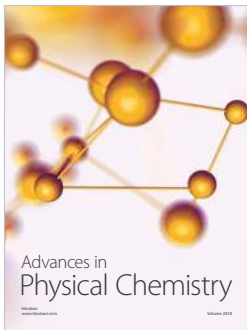
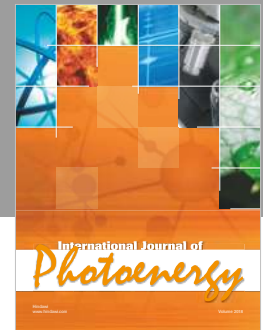
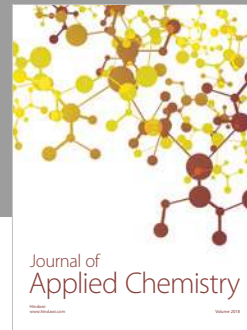
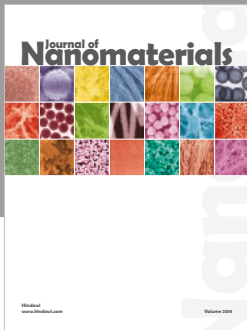
Taxonomical identification of cyanobacteria samples. The Red Sea cyanobacterial samples were collected from different

places by using scuba at a depth of 5–10 m near Jeddah, Saudi Arabia. Foreign particles were removed by hand, and seawater was squeezed out from the samples before storing in MeOH for transportation. A small portion of this material was also preserved in 10 mL of RNAlater (Ambion) solution for genetic analysis. The 16S rRNA gene sequence (GenBank KT380828) of this material is currently available online. (*Supplementary Materials*)

References

- [1] M. Salta, J. A. Wharton, Y. Blache, K. R. Stokes, and J. F. Briand, "Marine biofilms on artificial surfaces: structure and dynamics," *Environmental Microbiology*, vol. 15, pp. 2879–2893, 2013.
- [2] J. Antunes, P. Leão, and V. Vasconcelos, "Marine biofilms: diversity of communities and of chemical cues," *Environmental Microbiology Reports*, vol. 11, no. 3, pp. 287–305, 2019.
- [3] J. Antunes, S. Pereira, T. Ribeiro et al., "A multi-bioassay integrated approach to assess the antifouling potential of the cyanobacterial metabolites portoamides," *Marine Drugs*, vol. 17, no. 2, p. 111, 2019.
- [4] H. Dang and C. R. Lovell, "Microbial surface colonization and biofilm development in marine environments," *Microbiology and Molecular Biology Reviews*, vol. 80, no. 1, pp. 91–138, 2016.
- [5] P.-Y. Qian, S. C. K. Lau, H.-U. Dahms, S. Dobretsov, and T. Harder, "Marine biofilms as mediators of colonization by marine macroorganisms: implications for antifouling and aquaculture," *Marine Biotechnology*, vol. 9, no. 4, pp. 399–410, 2007.
- [6] M. G. Hadfield, "Biofilms and marine invertebrate larvae: what bacteria produce that larvae use to choose settlement sites," *Annual Review of Marine Science*, vol. 3, no. 1, pp. 453–470, 2011.
- [7] E. Costas, R. Gonzalez, V. López-Rodas, and I. E. Huertas, "Mutation of microalgae from antifouling sensitivity to antifouling resistance allows phytoplankton dispersal through ships' biofouling," *Biological Invasions*, vol. 15, no. 8, pp. 1739–1750, 2013.
- [8] M. P. Schultz, "Effects of coating roughness and biofouling on ship resistance and powering," *Biofouling*, vol. 23, no. 5, pp. 331–341, 2007.
- [9] I. Fitridge, T. Dempster, J. Guenther, and R. de Nys, "The impact and control of biofouling in marine aquaculture: a review," *Biofouling*, vol. 28, no. 7, pp. 649–669, 2012.
- [10] M. P. Schultz and G. W. Swain, "The influence of biofilms on skin friction drag," *Biofouling*, vol. 15, no. 1–3, pp. 129–139, 2000.
- [11] Z. Ganwei, B. Renbi, S. Shusu, Z. Xiaoji, and G. Yongfu, "Hydrophilic and photo-crosslinkable diblock copolymers employed for robust antifouling membrane coatings," *Applied Surface Science*, vol. 464, pp. 429–439, 2019.
- [12] A. B. A. Boxall, S. D. Comber, A. U. Conrad, J. Howcroft, and N. Zaman, "Inputs, monitoring and fate modelling of anti-fouling biocides in UK estuaries," *Marine Pollution Bulletin*, vol. 40, no. 11, pp. 898–905, 2000.
- [13] A. Terlizzi, S. Frascchetti, P. Gianguzza, M. Faimali, and F. Boero, "Environmental impact of antifouling technologies: state of the art and perspectives," *Aquatic Conservation: Marine and Freshwater Ecosystems*, vol. 11, no. 4, pp. 311–317, 2001.
- [14] I. Imara, W. Miled, R. Ben Salma, and N. Ladhari, "Anti-fouling process and toxicity effects of antifouling paints on marine environment. a review," *Environmental Toxicology and Pharmacology*, vol. 57, pp. 115–130, 2018.
- [15] C. Alzieu, Y. Thibaud, M. Heral, and B. Boutier, "Evaluation des risques dus a l'emploides peintures anti-salissures dans-sales zones conchylicoles," *Revue des Travaux de l'Institut des Pêches Maritimes*, vol. 44, no. 4, pp. 305–348, 1980.
- [16] B. Antizar-Ladislao, "Environmental levels, toxicity and human exposure to tributyltin (TBT)-contaminated marine environment. A review," *Environment International*, vol. 34, no. 2, pp. 292–308, 2008.
- [17] S. Liu, J. Zhou, X. Ma, Y. Liu, Y. Ma, and C. Xia, "Ecotoxicity and preliminary risk assessment of nonivamide as a promising marine antifoulant," *Journal of Chemistry*, vol. 2016, Article ID 2870279, 4 pages, 2016.
- [18] I. Muñoz, M. J. Martínez Bueno, A. Agüera, and A. R. Fernández-Alba, "Environmental and human health risk assessment of organic micro-pollutants occurring in a Spanish marine fish farm," *Environmental Pollution*, vol. 158, no. 5, pp. 1809–1816, 2010.
- [19] J. R. Pawlik, "Marine invertebrate chemical defenses," *Chemical Reviews*, vol. 93, no. 5, pp. 1911–1922, 1993.
- [20] A. S. Clare, "Marine natural product antifoulants: status and potential," *Biofouling*, vol. 9, no. 3, pp. 211–229, 1996.
- [21] N. Fusetani, "Biofouling and antifouling," *Natural Product Reports*, vol. 21, no. 1, pp. 94–104, 2004.
- [22] P. Y. Qian, Y. Xu, and N. Fusetani, "Natural products as antifouling compounds: recent progress and future perspectives," *Biofouling*, vol. 26, pp. 223–234, 2010.
- [23] N. Fusetani, "Antifouling marine natural products," *Natural Product Reports*, vol. 28, no. 2, pp. 400–410, 2011.
- [24] P.-Y. Qian, Z. Li, Y. Xu, Y. Li, and N. Fusetani, "Mini-review: marine natural products and their synthetic analogs as antifouling compounds: 2009–2014," *Biofouling*, vol. 31, no. 1, pp. 101–122, 2015.
- [25] K. Kon-ya, N. Shimidzu, N. Otaki, A. Yokoyama, K. Adachi, and W. Miki, "Inhibitory effect of bacterial ubiquinones on the settling of barnacle, *Balanus amphitrite*," *Experientia*, vol. 51, no. 2, pp. 153–155, 1995.
- [26] K.-L. Wang, Z.-H. Wu, Y. Wang, C.-Y. Wang, and Y. Xu, "Mini-review: antifouling natural products from marine microorganisms and their synthetic analogs," *Marine Drugs*, vol. 15, no. 9, p. 266, 2017.
- [27] N. Fusetani and A. S. Clare, Eds., *Antifouling Compounds*, Springer, Berlin, Germany, 2006.
- [28] C. Hellio and D. Yebra, Eds., *Advances in Marine Antifouling Coatings and Technologies*, Woodhead Publishing, Cambridge, UK, 2010.
- [29] J. R. Pawlik, "Marine invertebrate chemical defences," *Chemical Reviews*, vol. 93, no. 5, pp. 1911–1922, 1993.
- [30] L.-G. Yao, H.-L. Liu, Y.-W. Guo, and E. Mollo, "New cem-branoids from the hainan soft coral *Sarcophyton glaucum*," *Helvetica Chimica Acta*, vol. 92, no. 6, pp. 1085–1091, 2009.
- [31] B. F. Bowden, J. C. Coll, A. Heaton et al., "The structures of four isomeric dihydrofuran-containing cem-branoid diterpenes from several species of soft coral," *Journal of Natural Products*, vol. 50, no. 4, pp. 650–659, 1987.
- [32] M. Feller, A. Rudi, N. Berer et al., "Isoprenoids of the soft coral *Sarcophyton glaucum*: nyalolide, a new biscebranoid, and other terpenoids†," *Journal of Natural Products*, vol. 67, no. 8, pp. 1303–1308, 2004.
- [33] R. L. Hale, J. Leclercq, B. Tursch et al., "Demonstration of a biogenetically unprecedented side chain in the marine sterol, gorgosterol," *Journal of the American Chemical Society*, vol. 92, no. 7, pp. 2179–2180, 1970.

- [34] A. H. Marcus, T. F. Molinski, E. Fahy, D. J. Faulkner, C. Xu, and J. Clardy, "5-Isothiocyanatopupukeanane from a sponge of the genus *Axinyssa*," *The Journal of Organic Chemistry*, vol. 54, no. 21, pp. 5184–5186, 1989.
- [35] M. Zubair, W. Alarif, K. Al-Footy et al., "New antimicrobial biscebranone hydrocarbon and cebranoid diterpenes from the soft coral *Sarcophyton trocheliophorum*," *Turkish Journal of Chemistry*, vol. 40, pp. 385–392, 2016.
- [36] Y. Kashman, E. Zadock, and L. Néeman, "Some new cebranone derivatives of marine origin," *Tetrahedron*, vol. 30, no. 19, pp. 3615–3620, 1974.
- [37] K. O. Al-Footy, W. M. Alarif, F. Asiri, M. M. Aly, and S.-E. N. Ayyad, "Rare pyrane-based cebranoids from the Red Sea soft coral *Sarcophyton trocheliophorum* as potential antimicrobial-antitumor agents," *Medicinal Chemistry Research*, vol. 24, no. 2, pp. 505–512, 2015.
- [38] Y.-B. Cheng, Y.-C. Shen, Y.-H. Kuo, and A. T. Khalil, "Cebranone diterpenoids from the Taiwanese soft Coral *Sarcophyton stolidotum*," *Journal of Natural Products*, vol. 71, no. 7, pp. 1141–1145, 2008.
- [39] R. Duffy, C. Wade, and R. Chang, "Discovery of anticancer drugs from antimalarial natural products: a MEDLINE literature review," *Drug Discovery Today*, vol. 17, no. 17–18, pp. 942–953, 2012.
- [40] M.-E. F. Hegazy, A. M. G. Eldeen, A. A. Shahat et al., "Bioactive hydroperoxyl cebranoids from the Red Sea soft coral *Sarcophyton glaucum*," *Marine Drugs*, vol. 10, no. 12, pp. 209–222, 2012.
- [41] Z. Xi, W. Bie, W. Chen et al., "Sarcophyolides B-E, new cebranoids from the soft coral *Sarcophyton elegans*," *Marine Drugs*, vol. 11, no. 9, pp. 3186–3196, 2013.
- [42] G.-H. Wang, H.-C. Huang, J.-H. Su et al., "Crassocolides N-P, three cebranoids from the Formosan soft coral *Sarcophyton crassocaule*," *Bioorganic & Medicinal Chemistry Letters*, vol. 21, no. 23, pp. 7201–7204, 2011.
- [43] J.-H. Su, A. F. Ahmed, P.-J. Sung et al., "Manaarenolides A–I, diterpenoids from the soft coral *Sinularia manaarensis*," *Journal of Natural Products*, vol. 69, no. 8, pp. 1134–1139, 2006.
- [44] Y. Lu, C.-Y. Huang, Y.-F. Lin et al., "Anti-inflammatory cebranoids from the soft corals *Sinularia querciformis* and *Sinularia granosa*," *Journal of Natural Products*, vol. 71, no. 10, pp. 1754–1759, 2008.
- [45] G. R. Pettit, R. Tan, and Z. A. Cichacz, "Antineoplastic agents. 542. Isolation and structure of sesterstatin 6 from the Indian ocean sponge *hyrtios erecta*1," *Journal of Natural Products*, vol. 68, no. 8, pp. 1253–1255, 2005.
- [46] M. Salmoun, C. Devijver, D. Daloze et al., "New sesquiterpene/quinones from two sponges of the Genus *Hyrtios*," *Journal of Natural Products*, vol. 63, no. 4, pp. 452–456, 2000.
- [47] P. Sauleau, M.-T. Martin, M.-E. T. H. Dau, D. T. A. Youssef, and M.-L. Bourguet-Kondracki, "Hyrtiazepine, an azepinoindole-type Alkaloid from the red sea marine sponge *Hyrtios erectus*1," *Journal of Natural Products*, vol. 69, no. 12, pp. 1676–1679, 2006.
- [48] Y. Sugiyama, Y. Ito, M. Suzuki, and A. Hirota, "Indole derivatives from a marine sponge-derived yeast as DPPH radical scavengers," *Journal of Natural Products*, vol. 72, no. 11, pp. 2069–2071, 2009.
- [49] S. S. Al-Lihaibi, A. Abdel-Lateff, W. M. Alarif, Y. Nogata, S.-E. N. Ayyad, and T. Okino, "Potent antifouling metabolites from Red Sea organisms," *Asian Journal of Chemistry*, vol. 27, no. 6, pp. 2252–2256, 2015.
- [50] S. H. Qi, C. H. Gao, P. Y. Qian, and S. Zhang, "Steroids from the south China sea gorgonian *Subergorgia suberosa*," *Natural Product Communications*, vol. 5, pp. 201–204, 2010.
- [51] D. J. Gerhart and J. C. Coll, "Pukalide, a widely distributed octocoral diterpenoid, induces vomiting in fish," *Journal of Chemical Ecology*, vol. 19, no. 11, pp. 2697–2704, 1993.
- [52] M. G. Missakian, B. J. Burreson, and P. J. Scheuer, "Pukalide, a furanocembranolid from the soft coral *Sinularia abrupta*," *Tetrahedron*, vol. 31, no. 20, pp. 2513–2515, 1975.
- [53] D. Lai, Y. Li, M. Xu et al., "Sinulariols A-S, 19-oxygenated cebranoids from the Chinese soft coral *Sinularia rigida*," *Tetrahedron*, vol. 67, no. 33, pp. 6018–6029, 2011.
- [54] S. Shen, H. Zhu, D. Chen et al., "Pavidolides A-E, new cebranoids from the soft coral *Sinularia pavida*," *Tetrahedron Letters*, vol. 53, no. 43, pp. 5759–5762, 2012.
- [55] S. H. Qi and X. Ma, "Antifouling compounds from marine invertebrates," *Marine Drugs*, vol. 15, p. 263, 2017.



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