ORIGINAL

Equalizing calculus in Trefftz method for solving two-dimensional temperature field of FC-72 flowing along the minichannel

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Abstract The paper focuses on the numerical solution to two-dimensional temperature field of boiling liquid flowing along a vertical, asymmetrically heated minichannel with a rectangular cross-section. One of the walls of a minichannel is DC supplied single-sided enhanced foil with mini-recesses distributed unevenly in the selected area. The parallel walls are made of glass panes for thermal insulation and they are intended for observation of the two-phase flow and the void fraction. The thin layer of thermosensitive liquid crystal paint on the outer side of the foil enabled to record two-dimensional temperature distribution of outer foil surface. The paper described computations based on Trefftz method for finding two-dimensional temperature field of boiling liquid flowing along the minichannel. The presented research is limited only to the liquid phase of the two-phase mixture observed in the minichannel. The velocity of liquid flowing through the minichannel is represented by a piecewise linear approximating function. To solve energy equation for liquid phase, Trefftz functions specially generated for this purpose were employed. Temperature field in the fluid was approximated by a linear combination of Trefftz functions. Equalizing calculus was applied to the Trefftz method to smooth temperature reduce measurements and measurement errors.

Temperature at the interface between working fluid and foil amounts to the saturation temperature. Temperature distribution in the foil and the glass pane was also computed using proper Trefftz functions.

List of symbols

A Section area (m^2)

b Approximation coefficients

c Specific heat $(J kg^{-1} K^{-1})$

hue Component from HSI system, based on hue, saturation, intensity

hue_i Hue value corresponding to the set foil temperature

J Functional

L Minichannel length (m)

m The order of polynomial approximating a calibration curve

n Derivative order

N Number of Trefftz functions

P Number of measurement points

 q_V Volumetric heat flux (W m⁻³)

Re Reynolds number

SEE Standard error estimation (K)

T Temperature (K)

u Velocity (m s⁻¹)

V Volume (m³)

v Trefftz function

x Distance along the minichannel length (m)

y Distance along the glass, foil and minichannel thickness (m)

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Greek symbols

- α Heat transfer coefficient (W m⁻² K⁻¹)
- Δ Absolute error (accuracy) (K)
- δ Thickness (m)



- ε Correction (K)
- Φ Likelihood function
- φ Void fraction
- σ Temperature error (K)
- λ Thermal conductivity (W m⁻¹ K⁻¹)
- ρ Density (kg m⁻³)
- Ω Lagrange function
- ω Lagrange multiplier

Subscripts

ave Average

G Glass

F Foil

i Measurement point

in Inlet

L Liquid

M Minichannel

max Maximal value

out Outlet

sat Saturation

v Vapour

Superscript

corr Corrected

1 Introduction

The development of the high-tech integrated circuits technology involves high power consumption and related heat generation; hence the cooling systems for microelectronic chips face the challenge of heat dissipation in confining spaces. Flow boiling in minichannels is an option because high heat transfer rate and minichannels are ideal for these applications due to their compactness and low pressure drop. Yet, these issues have not been studied enough. The research of the enhanced structures is increasingly attractive due to their theoretical enhancement potential for heat transfer. The summary of relevant literature focused on heat transfer in rectangular minichannels is presented in [1], and in addition, having enhanced surfaces—in [2, 3]. Trefftz method used for solving issues related to heat transfer in minichannels with new Trefftz functions is employed for smoothing measurement data and reducing measurement errors.

2 Experimental research

2.1 Experimental stand

The area of interest is the issue of heat transfer in cooling liquid flowing through a rectangular minichannel with one uniformly heated enhanced wall. The flow system was

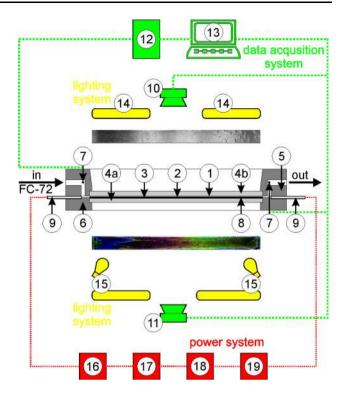


Fig. 1 The schematic diagrams of the most elements of the experimental stand: #1-minichannel, #2-heating foil, #3-liquid crystal layer, #4a, b-glass pane, #5-channel body, #6-front cover, #7-thermocouple, #8-enhanced side of the foil, #9-copper element #10-digital SLR camera, #11-digital camera, #12-data acquisition station, #13-a laptop, #14-halogen reflectors, #15-fluorescent and LED lamps, #16-inverter welder, #17-shunt, #18-ammeter, #19-voltmeter

discussed in [1-5]. The essential part of the experimental stand is the test section with a minichannel (Fig. 1, #1). It is 1 mm deep, 40 mm wide and 360 mm long, oriented vertically. The heating element for the working fluid (FC-72) flowing along the minichannel is alloy foil (#3) stretched between the front cover (#6) and the channel body (#5). This thin foil (approx. 0.1 mm) is enhanced on the side of the fluid flowing in the minichannel. It is energized by copper elements (#9) with direct current of controlled intensity. It is possible to observe both surfaces of the minichannel through two openings covered with glass panes. One pane (#4a) allows observing changes in the temperature of the foil surface. It is a plain side of the heating foil (between the foil and the glass), covered with thermosensitive liquid crystal paint (#3). The opposite surface of the minichannel (from the enhanced side of the heating foil) is observed through the other glass plane (#4b), which helps recognize the two-phase flow patterns. K-type thermocouples and two pressure converters (#7) are installed in the inlet and outlet of the minichannel. The data sources from the experiment and image acquisition system include: #10-Canon Eos 550D digital SLR camera, #11-Canon G11 digital camera, #14-two 1 kW halogen reflectors with forced air cooling and heat resistant



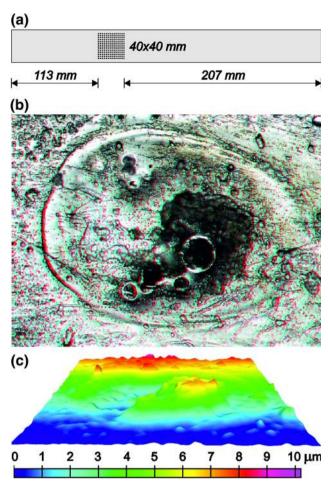


Fig. 2 The scheme of enhanced foil with mini-recesses **a** with photo **b** and 3-d topography **c** of the sample area

casing, and #15-fluorescent and LED lamps emitting "cool white light". Experimental data have been recorded with DaqBoard 2005 data acquisition station (#12) equipped with DASYLab software installed on a laptop (#13). The heating surface in the minichannel is provided with electric current by an inverter welder (#16) as current regulated DC power supply (up to 300 A). On the side of the foil contacting the fluid in the minichannel, mini-recesses were distributed unevenly in the selected area of the foil (40×40 mm) as shown in Fig. 2a. The recesses were obtained by spark erosion. The layer of melted metal of the foil and the electrode material, a few μ m high, reaching locally 5 μ m, accumulates around the cavities. The depth of the craters of cavities is usually below 1 μ m. The photographs of the plain and enhanced foil is shown in Fig. 2b, c.

2.2 Experimental methodology

Application of liquid crystals for the detection of twodimensional heating surface temperature distribution must be preceded by colour (*hue*) temperature calibration [1].

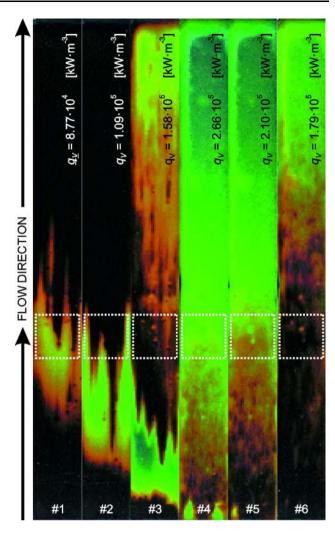


Fig. 3 Colour heating foil images while increasing and later decreasing heat flux supplied to the foil, experimental parameters: flow velocity 0.17 m/s, mass flux $285 \text{ kg/(m}^2 \text{ s})$, Reynolds number Re = 880, inlet pressure 125 kPa, inlet liquid subcooling 42 K, dotted lines show location of mini-recesses

FC-72 working fluid, at the temperature below its boiling point, flows laminarly into the minichannel (#1, Fig. 1). The gradual increase in the electric power supplied to the heating foil results in an increased heat flux transferred to the liquid in the minichannel. The current supplied via copper elements (#9) to the heating foil (#2) is controlled by an electrical system equipped with an inverter welder (#16). This leads to the incipience and next to the development of nucleate boiling. Then, the current supplied to the foil reduces gradually. Thanks to the liquid crystal layer located on its surface contacting the glass it is possible to measure the temperature distribution of the heating wall. Flow structure observation is carried out simultaneously at the opposite side of the minichannel.

Selected colour images of the heating foil in the measurement batch are presented in Fig. 3, the full batch was shown in [5]. When the current supplied to the



heating foil increases gradually (images from #1 up to #4), it causes the occurrence of boiling incipience. The "boiling front" is recognizable as the hue sequence pattern, which indicates gradual hue changes to the liquid crystals and then sharp hue changes in the liquid crystal layer. Out-of-sensitivity-range temperatures are shown in black. This phenomenon of the occurrence of nucleation hysteresis was discussed in [1-6]. When the heat flux continues to increase, a new hue sequence in the upper part of images appears. This occurs when developed nucleate boiling is in progress in the minichannel. Then the current supplied to the foil is gradually reduced (images #5 and #6). Under such conditions, mild hue changes in the direction opposite to the spectrum sequence are observed. As a result, heat transfer returns to forced single phase convection. Images taken when the heat flux had been supplied to the foil were used for further analysis.

2.3 Evaluation of the accuracy of heating foil temperature measurements with liquid crystal thermography

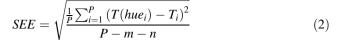
Error estimation is conducted in accordance with the principles of measurement accuracy analysis [7]. Root-mean-square error is adopted as a measure of error quantity. Root-mean-square errors are computed as the roots of the sum of the second powers of the products of function partial derivatives, with respect to an external parameter occurring in a direct measurement, by the mean error of a given parameter measurement.

The course of calculations used for the evaluation of the accuracy of heating foil temperature measurements with liquid crystals thermography is presented in [1]. Similarly to [1, 8], it is assumed that the mean temperature error for the single point, determined on the basis of *hue* in calibration experiment, amounts to:

$$\sigma = \frac{1}{P} \sum_{i=1}^{P} \sigma_{i}$$

$$= \frac{1}{P} \sum_{i=1}^{P} \sqrt{\left(\frac{\partial T(hue_{i})}{\partial hue_{i}} \Delta hue\right)^{2} + (2SEE)^{2} + (\Delta T_{L})^{2}}$$
(1)

where: i—measurement point; P—number of measurement points, T—temperature, Δhue —error in hue determination for a recorded and processed image; it is assumed it equals double standard deviation, determined on the basis of the data for sample surface image in calibration experiment; SEE—standard estimate of error in the fit of the calibration curve, determined with the use of the method of least squares, in accordance with formula:



where: m—the order of polynomial approximating a calibration curve, m=16; n—derivative order, n=1; ΔT_L —absolute error in the measurement of liquid temperature during calibration.

Detailed calculations are presented in [9]. Absolute error in the measurement of liquid temperature at the minichannel inlet and outlet during calibration ΔT_L consists of two errors: the one resulting from signal processing by acquisition cards included in the measurement data acquisition station (incorporating the errors in the thermocouple card and in the station with 16-bit analogue-todigital converter) and the thermocouple sensor errors. The absolute thermocouple sensor errors result from the error in thermocouple calibration and the error in the measurement with a thermocouple. After calculations, the error resulting from signal processing by acquisition cards included in the measurement data acquisition station amounting to 69.974 µV was obtained. In addition, the following results were calculated: the error in thermocouple calibration— 6.582 µV, the error in the measurement using the thermocouple—0.763 µV and then thermocouple sensor errors—6.626 µV. Subsequent calculations led $\Delta T_L = 70.287 \,\mu\text{V}$. On the basis of thermocouple temperature measurement range (type K thermocouple, measurement range up to 1,100 °C), and in the temperature range of the thermocouple card (from 0 to 100 mV) 1 K was obtained, corresponding to 90.91 μ V. Therefore, ΔT_L corresponds to the error in temperature measurement $\Delta T_L = 0.77 \text{ K}.$

Sample results of calculations in the error analysis for the conducted experiment are as follows: P = 115 calibration points: $\Delta hue = 1.45$; SEE = 0.13 K and $\Delta T_L = 0.77$ K [1]. After calculations according to formula (1), $\Delta \sigma = 0.86$ K.

3 Void fraction determination

3.1 Visualisation of two phase flow structures

The analyses of the flow structure were based on the monochrome images of flow structures with an SLR camera of heating foil, obtained on the side contacting fluid flowing in a minichannel. They were processed using Corel graphics software. After the photos had been binarized, the analysis of phase volumes was developed using special software—Techsystem Globe. The software allowed the determination of the areas of the two phases and/or



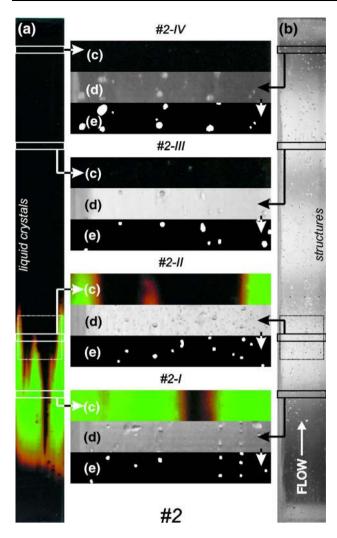


Fig. 4 Selected images for setting #2: (a) colour image of the heating foil, (b) greyscale image of flow structure image, (c, d, e) images of cross sections: colour image of the heating foil (c), two-phase flow structure images: real (d) and binarized (e); white colour refers to the vapour, and the black colour represents the liquid

percentage of the defined phase. Methodology of the conversion of greyscale image into monochrome image is presented in detail in [9]. In evaluation, the absolute error of the void fraction was assumed to be equal to the area (point) comprising 0.0064 mm²; it results from the resolution of the image taken by a digital camera [4, 9].

3.2 Void fraction determination methodology

Four initial settings (from #1 up to #4, Fig. 3) for increasing heat flux supplied to the heating surface of the foil were selected for analysis. Figure 4 shows a sample pictures for setting #2 with marked cross-sections (Fig. 4a, b), colour image of cross sections (Fig. 4c), two-phase flow structure images: real (Fig. 4d) and binarized (Fig. 4e). Four cross-sections of the selected images (size $5 \text{ mm} \times 40 \text{ mm}$), were placed at the distance of 90 mm (I),

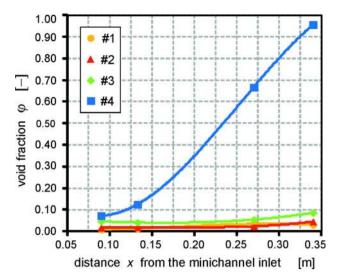


Fig. 5 Void fraction dependence along the minichannel length for selected cross sections for settings from #1 up to #4, parameters as for Fig. 3

133 mm (II), 270 mm (III) and 336 mm (IV) from the inlet to the minichannel. The void fraction was determined according to the following formula [1–5]:

$$\varphi = \frac{V_{v}}{V_{L} + V_{v}} = \frac{A_{v} \, \delta_{M}}{(A_{L} + A_{v}) \, \delta_{M}} = \frac{A_{v}}{A_{L} + A_{v}} \tag{3}$$

where: V—volume, A—cross section area, δ —depth.

In our notation, the subscripts referring to the glass pane, the heating foil, the liquid, the vapour and the minichannel are denoted with letters: G, F, L, v and M, respectively.

Liquid and vapour cross-section areas were obtained from phase image analysis performed by means of Techsystem Globe software. The results are presented in the Fig. 5 as void fraction dependence along the minichannel length.

4 Mathematical model

4.1 Problem formulation

In the mathematical model the investigations took into account the dimension along the flow direction (x), but the dimension perpendicular to it (y) was related to the test section composed of the glass pane, heating foil and minichannel with cooling liquid. Further considerations are focused on the central part of the minichannel (along its length) so that the physical phenomena occurring on outer sides of the minichannel have no influence on thermodynamical parameters in the region studied.

Subcooling liquid at a temperature lower than the saturation temperature enters the minichannel and it is heated by the heating foil during its flow. We assume for both



stationary and laminar flows (Re = 880) that velocity vector is parallel to the minichannel length so it has only one non-zero component u = u(y). The component u(y) is approximated by the following roof function:

$$u(y) = \frac{u_{max}}{\delta_M} (\delta_M - |2y - \delta_M|) \tag{4}$$

where u_{max} is equal to the doubled average velocity. The energy equation, exclusively for the liquid phase, has the following form:

$$\nabla^2 T_L = \frac{c_L \, \rho_L}{\lambda_L} \, u(y) \, \frac{\partial T_L}{\partial x} \tag{5}$$

where: $\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2}$, c—specific heat, ρ —density, λ —thermal conductivity.

The following boundary conditions were specified for the Eq. (5):

temperature of the liquid in the inlet and outlet of the minichannel equals T_{in} and T_{out} , respectively, i.e.

$$T_L(0, y) = T_{in} \quad \text{for } 0 \le y \le \delta_M \tag{6}$$

$$T_L(L, y) = T_{out}$$
 for $0 \le y \le \delta_M$ (7)

temperature of the liquid and the foil at the interface are equal:

$$T_L(x,0) = \begin{cases} T_F(x,0), & \text{if} \quad T_F(x,0) < T_{sat}(x) \\ T_{sat}(x), & \text{if} \quad T_F(x,0) \ge T_{sat}(x) \end{cases}$$

$$\text{for } 0 \le x \le L$$
(8)

where T_{sat} is the saturation temperature dependent on the pressure, which changes linearly along the minichannel;

the two-phase mixture per unit volume in minichannel contains vapour phase and liquid phase in proportion φ and $(1-\varphi)$, respectively. We assumed that the same proportion of vapour and liquid phases refers to any cross-sectional area of the minichannel and then to the heat exchange surface [10]. Hence, all heat generated by the heating foil is transferred to liquid phase proportionally to the void fraction

$$\lambda_L \frac{\partial T_L}{\partial y} = \lambda_F (1 - \varphi(x)) \frac{\partial T_F}{\partial y} \quad \text{for} \quad y = 0 \quad \text{and} \quad 0 \le x \le L$$
(9)

Stationary heat transfer in the glass pane and heating foil can be described by the following differential equations:

(a) for glass
$$\nabla^2 T_G = 0$$
 (b) for foil $\nabla^2 T_F = -\frac{q_V}{\lambda_F}$ (10)

where q_V means volumetric heat flux. For these equations we assume following boundary conditions at the foil-glass interface:

$$T_F(x_i, -\delta_F) = T_G(x_i, -\delta_F) = T_i \tag{11}$$

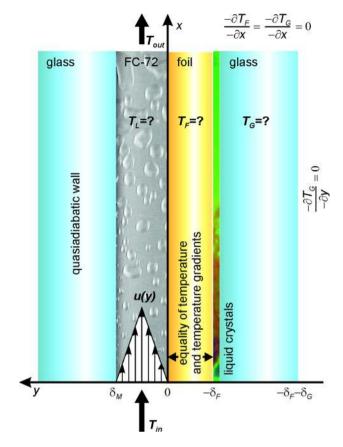


Fig. 6 Diagram of the boiling liquid flow in the minichannel, with the adapted boundary conditions (pictorial view, not to scale)

$$\lambda_F \frac{\partial T_F}{\partial y} = \lambda_G \frac{\partial T_G}{\partial y}$$
 for $y = -\delta_F$ and $0 \le x \le L$ (12)

where T_i means the measurements obtained with liquid crystal thermography at the foil-glass contact at discrete points $(x_i - \delta_F)$. To complete specification of the functions T_F and T_G we assume that the outer surfaces of the minichannel are thermally isolated. Figure 6 contains the geometry and prescribed conditions of the considered problem.

4.2 Trefftz method

Equation (5) subject to prescribed boundary conditions (6)—(9) will be solved by the Trefftz method [6, 10, 11]. In this method the unknown solution of a differential equation is approximated with a linear combination of the functions (so called Trefftz functions) which satisfy the governing equation. The unknown coefficients are determined by matching the boundary conditions. Before numerical calculations we have to generate Trefftz functions corresponding to Eq. (5) with roof-shaped velocity of the liquid, u(y). The unknown solution to Eq. (5) is expanded in Taylor series, around the point (0.0):



$$T_{L}(x,y) = T_{L}(0,0) + \frac{\partial T_{L}}{\partial x} x + \frac{\partial T_{L}}{\partial y} y + \frac{1}{2!} \frac{\partial^{2} T_{L}}{\partial x^{2}} x^{2}$$
$$+ \frac{2}{2!} \frac{\partial^{2} T_{L}}{\partial x \partial y} xy + \frac{1}{2!} \frac{\partial^{2} T_{L}}{\partial y^{2}} y^{2} + \frac{1}{3!} \frac{\partial^{3} T_{L}}{\partial x^{3}} x^{3} + \dots (13)$$

From Eq. (5) we get the dependence concerning the second partial derivative $\frac{\partial^2 T_L}{\partial v^2}$

$$\frac{\partial^2 T_L}{\partial y^2} = \frac{c_L \, \rho_L}{\lambda_L} u(y) \, \frac{\partial T_L}{\partial x} - \frac{\partial^2 T_L}{\partial x^2} \tag{14}$$

which we substitute into Eq. (13) and after some algebraic manipulations the unknown solution T_L has the following representation

$$T_L(x,y) = \sum_{n=1}^{\infty} b_n v_n(x,y)$$
(15)

where polynomials $v_n(x, y)$ are the desired Trefftz functions satisfying the energy Eq. (5) exactly.

Next, we approximate the solution T_L with finitely many $v_n(x, y)$'s which gives

$$T_L(x,y) \approx \sum_{n=1}^{N} b_n v_n(x,y)$$
 (16)

The unknown coefficients of this expansion, b_n will be selected to make the approximant T_L best match (in variational sense) the boundary conditions (6)–(9). In terms of computations, by proper choice of b_n 's one should minimize the functional J that describes mean square error between function T_L and prescribed boundary condition (6)–(9). The functional has the following form

$$J = \int_{0}^{\delta_{M}} \left(\sum_{n=1}^{N} b_{n} v_{n}(0, y) - T_{in} \right)^{2} dy + \int_{0}^{\delta_{M}} \left(\sum_{n=1}^{N} b_{n} v_{n}(L, y) - T_{out} \right)^{2} dy + \int_{0}^{L} \left(\sum_{n=1}^{N} \lambda_{L} b_{n} \frac{\partial v_{n}(x, 0)}{\partial y} - \lambda_{F} (1 - \varphi(x)) \frac{\partial T_{F}(x, 0)}{\partial y} \right)^{2} dx + \int_{0}^{L} \left(\sum_{n=1}^{N} b_{n} v_{n}(x, 0) - T_{L}(x, 0) \right)^{2} dx$$

$$(17)$$

Finally we get the solution T_L which satisfies the energy Eq. (5) exactly and the boundary conditions approximately. In the like manner we can obtain the solutions of Eqs. (10a and b) using Trefftz functions (harmonic polynomials) suitable for Laplace Eq. [11]. Trefftz method allows to determine two-dimensional distributions of temperature in the foil and in the glass as well as their gradients along the minichannel.

5 Equalizing calculus

Application of equalizing calculus means a choice of such corrections ε_i for measurements T_i that for the corrected measurements T_i^{corr} the following condition is satisfied

$$T_i^{corr} = T_i + \varepsilon_i \tag{18}$$

Corrections ε_i have normal distribution with the mean value equal to zero and finite variance equal to σ_i^2 , [12, 13], where errors σ_i are defined by the Eq. (1). For P of independent measurements the density function for corrected measurement results T_1^{corr} , T_2^{corr} ,..., T_P^{corr} (and whereby corrections $\varepsilon_1, \varepsilon_2, ..., \varepsilon_P$) is delineated as the product of the density function for normal distribution i.e.

$$\Phi(\varepsilon_{1}, \varepsilon_{2}, ..., \varepsilon_{P}) = \frac{1}{(2\pi)^{0.5P} (\sigma_{1}\sigma_{2}...\sigma_{P})} \exp\left[-\frac{1}{2} \sum_{i=1}^{P} \left(\frac{\varepsilon_{i}}{\sigma_{i}}\right)^{2}\right]$$
(19)

Function Φ is called the likelihood function. Determination of corrections ε_i leads to the determination of the maximum for likelihood function Φ . Corrections determined by this method are always compliant and asymptotically efficient, i.e. when P increases, the estimation result approaches the real value, whilst estimation variance will be as low as possible.

Function
$$\Phi$$
 is the highest when $\sum_{i=1}^{P} \left(\frac{\varepsilon_i}{\sigma_i}\right)^2 \to \min$. If this

is the case (when ε_i have assumed normal distribution) the maximum likehood function method leads to the same results as the method of least squares. When applying the equalizing calculus in Trefftz method, the corrected value of foil temperature T_F^{corr} should be determined to meet the following condition:

$$T_F^{corr}(x_i, \delta_G) - T_i^{corr} = 0 (20)$$

The determination of the maximum of the likehood function Φ , Eq. (19), assuming condition (20), leads to the minimalization of the Lagrange function:

$$\Omega = \sum_{i=1}^{P} \left(\frac{\varepsilon_i}{\sigma_i}\right)^2 + 2\sum_{i=1}^{P} \omega_i \left(T_F^{corr}(x_i, \delta_G) - T_i^{corr}\right) \to \min$$
(21)

where ω_i were Lagrange multipliers.

Having corrected temperature measurements, we recalculated measurement errors σ_i^{corr} according to the error propagation law [12, 13]. As expected, equalizing calculus smoothed the measurement data and reduced the errors, see Fig. 8. Owing to known values of T_i^{corr} , it was possible to recalculate approximated temperatures of glass, foil (as described in [6]) and liquid, denoted T_G^{corr} , T_F^{corr} and T_L^{corr} , respectively.



6 Results

Numerical calculations were performed for experimental data and they concerned forced flow of the cooling liquid FC-72 through asymmetrically heated vertical

minichannel. Calculations were conducted for settings from #1 up to #4, with all parameters of the experiment given in Fig. 3. Settings #1 and #2 apply to boiling incipience, setting #3 covers the interim status; setting #4 applies to the developed boiling. The void fraction

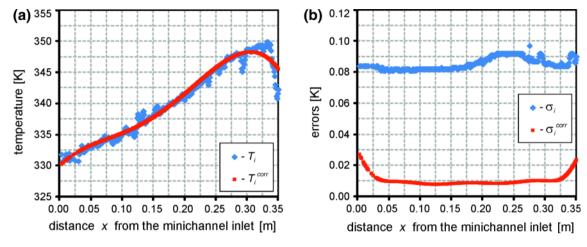


Fig. 7 a Temperature measurements T_i and corrected measurements T_i^{corr} ; **b** measurement errors σ_i and corrected measurement errors σ_i^{corr} , data for setting #4

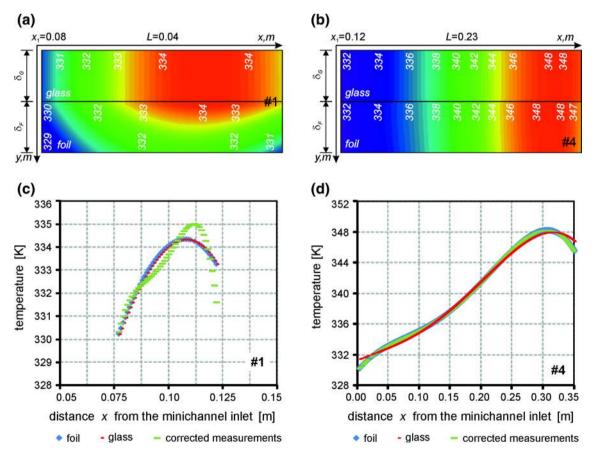
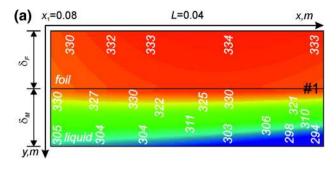


Fig. 8 a Two dimension temperature distributions of the glass pane and the heating foil, both obtained with Trefftz method for corrected measurements, **b** corrected temperatures of the corrected

measurements (green), glass pane (red) and foil (blue) at the glass-foil contact, data for settings #1 and #4 (colour figure online)





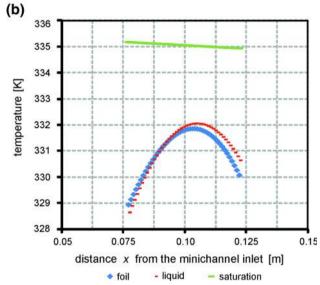
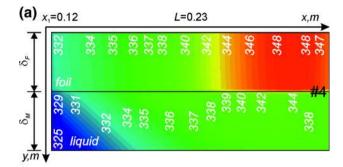


Fig. 9 a Two dimension temperature distributions of the heating foil and of the liquid, both obtained with Trefftz method, **b** corrected temperatures of the foil (*blue*), of the liquid (*red*) and saturation (*green*) at the foilliquid contact as a function of distance from the inlet to the channel. Setting #1 (colour figure online)

described in detail in 2.2 and refereed to the settings under consideration, was approximated with a third degree polynomial as shown in Fig. 5.

At first, corrected measurements T_i^{corr} were determined using equalizing calculus, Fig. 8b. Sample results obtained for setting #4 are presented in Fig. 7a. Using the equalizing calculus, on the basis of the error propagation, law corrected errors σ_i^{corr} were computed; their values are considerably lower than errors σ_i calculated from (1), Fig. 7b.

Then computations involved the use of Trefftz method for finding two-dimensional temperature distribution in the glass pane and the heating foil. In the next step temperatures were calculated in the liquid phase. We designed approximate temperature distributions T_G^{corr} , T_F^{corr} and T_L^{corr} to be polynomials of the same degree. Fig. 8a presents two-dimensional temperature distribution in the heating foil and the glass pane, both obtained by Trefftz method for corrected measurements T_i^{corr} . Either of the approximants satisfies the assumed conditions (11) and (12), with a relatively high accuracy, Fig. 8b. Similarly, Fig. 9a and Fig 10a shows



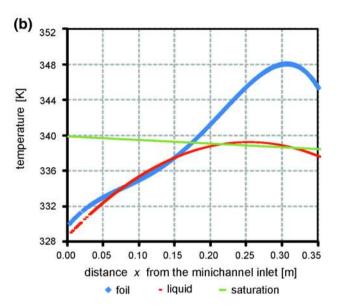


Fig. 10 a Two dimension temperature distributions of the heating foil and of the liquid, both obtained with Trefftz method, **b** corrected temperatures of the foil (*blue*), of the liquid (*red*) and saturation (*green*) at the foilliquid contact as a function of distance from the inlet to the channel. Setting #4 (colour figure online)

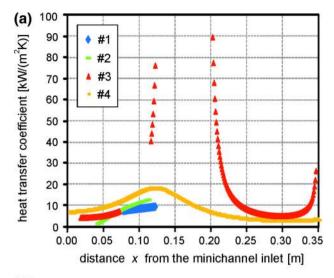
temperature field in the heating foil and the flowing liquid. Figure 9 applies to the case when boiling incipience occurs (data for setting #1), whilst Fig. 10 reflects the situation when developed boiling takes place (data for setting #4). One can notice the characteristic shape of isotherms in the liquid (see Figs. 9a, 10a) which is due to modelling the velocity vector as a roof function. In Figs. 9b and 10b temperatures of the heating foil and the liquid at the foil-liquid interface are compared to saturation temperature. In [10] similar research, assuming a parabolic liquid velocity profile, was conducted for bubbly and bubbly–slug flow.

The known liquid and foil temperature distributions allowed calculating heat transfer coefficient α at the foil-glass contact points from the Robin boundary condition [14].

$$-\lambda_F \frac{\partial T_F^{corr}(x,0)}{\partial y} = \alpha(x) \left(T_F^{corr}(x,0) - T_{ave}(x) \right). \tag{22}$$

Two options for computing liquid temperature were considered. In the first approach the averaged temperature





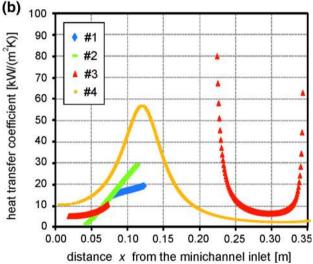


Fig. 11 Heat transfer coefficient as a function of distance from the inlet to the minichannel determined in the reference temperature: **a** from Eq. (23a), **b** from Eq. (23b)

across the whole width of the minichannel were assumed as reference temperature T_{ave} , Eq. (23a). Since the liquid reaches its maximal velocity along the axis of the minichannel, Eq. (2), in the second approach we took average temperature at points from x = 0 to $x = 0.5\delta_M$ as reference temperature T_{ave} , Eq. (23b).

(a)
$$T1_{ave}(x) = \frac{1}{\delta_M} \int_0^{\delta_M} T_L^{corr}(x, y) dy$$

(b) $T2_{ave}(x) = \frac{1}{0.5 \, \delta_M} \int_0^{0.5 \, \delta_M} T_L^{corr}(x, y) dy$ (23)

In the minichannel flow boiling, considerable heat transfer enhancement takes place at boiling incipience. It is observed as a sharp increase in the heat transfer coefficient at the foil-liquid contact, Fig. 11c. Similar results in flow boiling incipience were obtained in the study [4]. In the case of developed boiling, the heat transfer coefficient decreases as the void fraction increases. As far as the transition flow was concerned, the increase in the heat transfer coefficient was observed. For higher shares of the vapour phase in the two-phase mixture, heat transfer coefficient values will decrease.

7 Conclusions

The paper presents the approach to modelling laminar flow of the fluid during boiling incipience and developed boiling. When a mathematical model was formulated then the Trefftz method was used in order to solve it, however the validity of the solution is restricted to liquid phase only.

Special Trefftz functions had to be generated for the case of roof-shaped velocity of the liquid.

The proposed method based on Trefftz functions was successfully applied to the direct problem of finding temperature distribution in the glass pane. Moreover, this method turned to be suitable for the considered inverse problems: in the heating foil were we had a specified condition at one boundary and in the liquid where we assumed convective boundary condition in order to compute the heat transfer coefficient. The obtained numerical results show high compliance with physical assumptions of the model and have satisfactory accuracy.

Equilizing calculus applied to Trefftz method helped smooth the measurement data and reduce their errors.

In the minichannel flow boiling, considerable heat transfer enhancement takes place at boiling incipience. In the case of developed boiling, the heat transfer coefficient decreases as the void fraction increases. As far as the transition flow was concerned, the increase in the heat transfer coefficient was observed, but the data concerned a small range of the void fraction. For higher shares of the vapour phase in the two-phase mixture, heat transfer coefficient values will decrease. There are plans for indepth exploring transition and developed boiling area and future verification of the data.

Since the results based on Trefftz method turned to be promising, further research will be devoted to adapting the method to finding temperature distribution in the fluid which would include vapour phase as well as liquid phase.

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