# Erratum: Viscosity of Water Substance–New International Formulation and Its Background

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# Errata

## Erratum: Viscosity of Water Substance—New International Formulation and Its Background

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On page 1158, equation (1) should read:

$$\mu = \mu_0 \exp\left[ \frac{\rho}{\rho^*} \sum_{i=0}^{5} \sum_{j=0}^{4} b_{ij} \left( \frac{T^*}{T} - 1 \right)^i \left( \frac{\rho}{\rho^*} - 1 \right)^j \right],$$

and the constant $a_1$ should read:

$$a_1 = 0.0177624.$$

## Erratum: JANAF Thermochemical Tables: 1978 Supplement

[J. Phys. Chem. Ref. Data 7, 793 (1978) ]

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On page 886 the tabulated values of $\Delta Hf_0^\circ$ and $\Delta Hf_{298.15}^\circ$ for $ZrCl_4$ (c) should be $-234.60 \pm 0.40$ and $-234.35 \pm 0.40$ kcal/mol, respectively. The heat of formation discussion and the numerical tabulation from 0-1000 K agrees with these values. On page 887, the first sentence in the heat of formation discussion should read "The heat of formation of gaseous $ZrCl_4$, $\Delta Hf_{298}^\circ = -207.93 \pm 0.5$ kcal/mol, is calculated from the heats of formation and sublimation of the crystal (l)."