

Ethanol Causes Acute Inhibition of Carbohydrate, Fat, and Protein Oxidation and Insulin Resistance

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Abstract

To study the mechanism of the diabetogenic action of ethanol, ethanol (0.75 g/kg over 30 min) and then glucose (0.5 g/kg over 5 min) were infused intravenously into six normal males. During the 4-h study, 21.8 ± 2.1 g of ethanol was metabolized and oxidized to CO_2 and H_2O . Ethanol decreased total body fat oxidation by 79% and protein oxidation by 39%, and almost completely abolished the 249% rise in carbohydrate (CHO) oxidation seen in controls after glucose infusion. Ethanol decreased the basal rate of glucose appearance (G_{Ra}) by 30% and the basal rate of glucose disappearance (G_{Rd}) by 38%, potentiated glucose-stimulated insulin release by 54%, and had no effect on glucose tolerance. In hyperinsulinemic-euglycemic clamp studies, ethanol caused a 36% decrease in glucose disposal. We conclude that ethanol was a preferred fuel preventing fat, and to lesser degrees, CHO and protein, from being oxidized. It also caused acute insulin resistance which was compensated for by hypersecretion of insulin.

Introduction

The effects of ethanol on carbohydrate (CHO)¹ metabolism are complex. When given to individuals whose glycogen stores have been depleted by fasting, ethanol can lead to severe hypoglycemia primarily by reducing hepatic glucose production through inhibition of gluconeogenesis (1, 2). The effects of ethanol in fed individuals are less well understood. In numerous studies, using many different experimental protocols, ethanol pretreatment has been associated with diminished (3), improved (4, 5), or unchanged (6, 7) glucose tolerance. Moreover, several patients have been reported in whom ethanol abuse led to overt diabetes mellitus which disappeared with abstinence from alcohol (8, 9). The mechanism for this diabetogenic effect of ethanol remains unknown. Because of the existing confusion in this area and the potential clinical importance of the problem, we have investigated, in normal male subjects, the acute effects of ethanol on the metabolism of an intravenous glucose load. To this end, we have measured total

body CHO, fat, and protein oxidation by indirect calorimetry, glucose appearance and disappearance from the extracellular space by [$3\text{-}^3\text{H}$]glucose dilution analysis and glucose disposal under hyperinsulinemic-euglycemic conditions with the clamp technique.

Methods

Subjects

Fourteen healthy male medical students and hospital employees, age 23.4 ± 0.8 yr (range 21–29 yr), body weight 81.0 ± 3.0 kg (range 70.5–108.0 kg), and relative ideal body weight $105 \pm 2\%$ (range 96–120%), participated in four study protocols. None had a family history of diabetes or alcoholism, admitted to drinking more than six drinks per week, or had consumed ethanol during the 72 h prior to any study and none were taking any medications. Informed written consent was obtained from all after explanation of the nature, purpose, and potential risks of each study. All study protocols were approved by the Human Research Committee of Temple University. The studies were performed in the General Clinical Research Center of Temple University Hospital and began between 7 and 8 a.m. after an overnight fast. The subjects were studied while reclining in hospital beds.

Study design

Four study protocols were used.

Protocol 1: glucose infusion studies. Six subjects were studied twice in random order with and without ethanol. A minimum of 7–10 d was allowed between ethanol and control (saline) studies. Catheters were placed in peripheral veins in each arm. One line was used to infuse ethanol, saline, glucose, and radiolabeled substances. A second line, in the contralateral arm, was used for blood sampling and was kept patent by a slow infusion of normal saline. Ethanol and control studies were identical except for the substitution of normal saline for ethanol in control studies. After a 2-h baseline period (–120 to 0 min), alcohol (0.75 g/kg, USP, Abbott Laboratories, N. Chicago, IL) diluted in normal saline to make a 12.5% solution, was infused intravenously for 30 min (0–30 min). Glucose (0.5 g/kg) was infused intravenously for 5 min starting at 60 min. Venous blood was collected for determination of plasma glucose concentration every 10 min for the next hour and at 30-min intervals thereafter. Venous blood samples for determination of insulin, glucagon, free fatty acids, norepinephrine, epinephrine, ethanol, acetate, and acetaldehyde concentrations were obtained at 15- or 30-min intervals for 4 h.

Ethanol was determined in aliquots of urine collected at ~ 30-min intervals. For estimation of breath ethanol excretion by gas chromatography, expired air was collected in glass flasks at 30-min intervals after ethanol administration as previously described for acetone (10). The gas chromatograph was equipped with a flame ionization detector and a $183 \times 2\text{-cm}$ glass column packed with Carbowax 20M/0.1% SP-10 (Supelco, Inc., Bellefonte, PA). Ethanol concentration was determined from peak areas by comparison with gaseous ethanol standards. Ethanol output was calculated from the breath ethanol concentration and the respiratory minute-volume measured with the Beckman Instruments Metabolic Cart (Palo Alto, CA). Respiratory gas exchange (indirect calorimetry) was determined at 30-min intervals with the Beckman Instruments Metabolic Measurement Cart as described (11). Col-

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1. Abbreviations used in this paper: a-v, arteriovenous; CHO, carbohydrate.

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lections of urine for nitrogen determinations, necessary for calculations of rates of protein oxidation, were made at 30-min intervals (when possible) throughout the study.

Protocol 2: glucose turnover studies. Protocol 2 was identical to protocol 1 except for a primed (40 μ Ci), continuous (0.4 μ Ci/min) [$3\text{-}^3\text{H}$]glucose infusion (New England Nuclear, Boston, MA) which was given for 6 h (-120 to +240 min). Four subjects were studied twice, in random order, with and without ethanol. 2 h (-120 to 0 min) were allowed for equilibration of [$3\text{-}^3\text{H}$]glucose. Plasma glucose was isolated from blood drawn at -30, -20, -10, 0, 30, 60, 70, 80, 90, 105, 120, 150, 180, 210, and 240 min for determination of [$3\text{-}^3\text{H}$]glucose specific activity (sp act) as previously described (12). Rates of glucose appearance (G_{Ra}) and glucose disappearance (G_{Rd}) were calculated for 30-min intervals using the equations of Steele et al. (13) for non-steady-state conditions.

Protocol 3: [^{14}C]ethanol oxidation studies. Six subjects participated in these studies. The objective was to determine the extent to which intravenously administered ethanol was completely oxidized to CO_2 and H_2O during the 4-h study. Infusions of ethanol and glucose were as described in protocol 1 except for the addition of 50 μ Ci of [$1\text{-}^{14}\text{C}$]ethanol (New England Nuclear, sp act 10.4 mCi/mmol) to the ethanol (one subject received only 31 μ Ci of [$1\text{-}^{14}\text{C}$]ethanol by mistake). A transparent plastic hood, tightly fitted around the neck with a soft collar, was placed over the subject's head. The expired air, mixed with room air, which was continuously ventilated through the hood (at 35 liter/min), was drawn from the hood and collected in Douglas bags at 30, 50, 70, 90, 120, 150, 180, 210, and 240 min after ethanol was administered (between 0 and 30 min). Respiratory minute volumes and CO_2 concentrations of expired air were obtained at each sampling period from the Beckman Instruments Metabolic Measurement Cart. $^{14}\text{CO}_2$ sp act in the respiratory gas samples were determined within 6 h after collection using the method of Fredrickson and Ono (14). Preliminary studies with prepared air samples containing [^{14}C]ethanol showed no contamination of the CO_2 trap of the collection system. It has been shown that after infusion of [$1\text{-}^{14}\text{C}$]ethanol the respiratory $^{14}\text{CO}_2$ sp act and the $^{14}\text{CO}_2$ output rise in form of an exponential curve approaching with time an asymptotic value in subjects in whom the bicarbonate pool was not primed with $\text{NaH}^{14}\text{CO}_3$ (four of six in this study). This is due to the slow exchange between the labeled CO_2 originating from [$1\text{-}^{14}\text{C}$]ethanol oxidation and the unlabeled body bicarbonate pool. The asymptotic value of the sp act of $^{14}\text{CO}_2$ represents the ideal situation in which the sp act of $^{14}\text{CO}_2$ in exhaled air and that of $^{14}\text{CO}_2$ arising from tissue oxidation are identical. Issekutz et al. (15) have studied the time course of equilibration of the two pools by infusing $\text{NaH}^{14}\text{CO}_3$ and have shown that sp act of $^{14}\text{CO}_2$ at any time, t , can be corrected to the ideal sp act $^{14}\text{CO}_2$, $t = \infty$ (15). We have used their data to correct CO_2 sp act values of the four subjects. The corrected $^{14}\text{CO}_2$ sp act and the CO_2 production rate were used to calculate $^{14}\text{CO}_2$ output in subjects 1-4. In two additional subjects (5 and 6) the bicarbonate pool was primed with a bolus injection of $\text{NaH}^{14}\text{CO}_3$ (0.096 μ Ci/kg) prior to the start of the ethanol infusion. Venous blood samples during this study were obtained at 0, 30, 45, 60, 70, 90, and every 30 min thereafter for ethanol determinations which were used to calculate rates of ethanol disappearance from plasma.

Protocol 4: hyperinsulinemic-euglycemic clamp studies. Six subjects were studied twice, in random order, with and without ethanol. A minimum of 1 wk was allowed between studies. Regular human insulin (Humulin R, Eli Lilly & Co., Indianapolis, IN) was infused intravenously (1 mU/kg per min) for 5 h. Glucose concentrations were clamped at about 80 mg/dl by a feedback-controlled glucose infusion (Biostat, Ames Life Science Instruments, Elkhart, IN). Blood for the Biostat was continuously drawn from a superficial antecubital vein. 2 h (0-120 min) were allowed for attainment of steady-state hyperinsulinemia. To assess effects of different concentrations, ethanol infusion rates were calculated to produce plasma ethanol concentrations of ~ 60 and 120 mg/dl. At 120 min, ethanol (0.375 g/kg) or saline was infused intravenously over a 15-min period. Thereafter, an infusion of ethanol (100 mg/kg per h) or saline was delivered to maintain steady-

state ethanol concentrations. At 210 min a second ethanol load of 0.5 g/kg was administered over 30 min followed by the same maintenance infusion (100 mg/kg per h) until the completion of the study. Venous samples for determination of ethanol, insulin, glucagon, and free fatty acids were obtained at 0, 30, and every 15 min thereafter until completion of the studies. To determine whether it was necessary to obtain arterialized blood during euglycemic-hyperinsulinemic clamps we determined, in three separate studies, arteriovenous (a-v) glucose concentration differences during the basal and hyperinsulinemic phases of the clamps in blood collected simultaneously from a radial artery and a superficial antecubital vein. During euglycemic-euinsulinemic (insulin 9 μ U/ml) conditions glucose a-v differences were +3, +7, and 0 mg/dl, respectively (mean 3.3 mg/dl). During euglycemic-hyperinsulinemic (insulin 80 μ U/ml) conditions glucose a-v differences were +6, +5, and +6 mg/dl, respectively (mean 5.6 mg/dl). Thus, a-v glucose differences were small during euinsulinemic and hyperinsulinemic euglycemia, which eliminated the need for collection of arterialized blood.

In vitro insulin binding

In vitro insulin binding was determined with primary cultures of rat hepatocytes prepared as previously described (16). Monolayers of hepatocytes (~ 2×10^6 cells/flask) were preincubated in the absence or presence of ethanol in five different concentrations ranging from 100 to 1,000 mg/dl and with normal human serum (diluted 1:25) in MEM (final volume: 2.5 ml) at 37°C for 120 min. After preincubation, the medium was aspirated and the cells washed twice with ice-cold Krebs-Ringer bicarbonate buffer (pH 7.4). ^{125}I -insulin binding was then determined as described (16). Interference of ethanol with ^{125}I -insulin binding was assessed with cells not pre-incubated with ethanol but exposed to ethanol during the insulin binding assay.

Analytic procedures

Plasma glucose was measured with a Beckman Instruments glucose analyzer. Plasma insulin (17) and glucagon (18) were determined by radioimmunoassay. Plasma free fatty acids were measured according to Lorch and Gey (19) after extraction according to Dole and Meinertz (20). Plasma catecholamines were measured radioenzymatically (21). Plasma and urine ethanol were determined with a kit (Sigma Chemical Co., St. Louis, MO) as were blood acetate and acetaldehyde (Boehringer Mannheim, Elkhart, IN). For the determination of acetate and acetaldehyde (22) by enzymatic analysis, blood samples were immediately deproteinized and neutralized with perchloric acid/KOH and metaphosphoric acid/KOH, respectively, prior to assay.

Statistical analysis and calculations

All data are expressed as the mean \pm SEM. Statistical significance was assessed using MANOVA or Student's two-tailed paired t test when appropriate.

Ethanol disappearance and metabolism

The rate of ethanol disappearance from the blood was calculated using the following formula: Ethanol disappearance = $B \times P \times \gamma$, where B = change in plasma ethanol concentration (mg/dl) obtained from the linear part of the plasma ethanol curve; P = body weight (kg) \times 1.06 to correct for differences in density between blood and water; and γ = coefficient of distribution for ethanol (and acetate) (23).

Ethanol metabolism was determined according to the following formula: Ethanol metabolized = ethanol disappeared from blood - ethanol excreted in urine and breath - ethanol present in plasma as acetate.

CHO, fat, and protein oxidation rates

Respiratory quotients (RQ) were obtained from the computerized analysis of respiratory gas exchange by the Metabolic Measurement Cart as described (24). Based on our finding ([^{14}C]ethanol oxidation studies, protocol 3) of complete oxidation of ethanol to CO_2 and H_2O during the 4-h study, these respiratory exchange data were corrected by subtracting the volumes of O_2 consumed and CO_2 produced during

ethanol oxidation (complete oxidation of 1 mol of ethanol consumes 3 mol of O₂ and produces 2 mol of CO₂). Protein oxidation rates were calculated from urinary nitrogen excretion and used to determine non-protein RQs. The ethanol-corrected non-protein RQs were then used to calculate CHO and fat oxidation rates as described previously (24).

Results

Effect of ethanol on glucose tolerance and insulin (Fig. 1). Infusion of ethanol (0.75 g/kg over 30 min) during the glucose infusion studies (protocol 1) resulted in a peak concentration of 140±8 mg/dl at 30 min. After equilibration (at ~ 75 min),

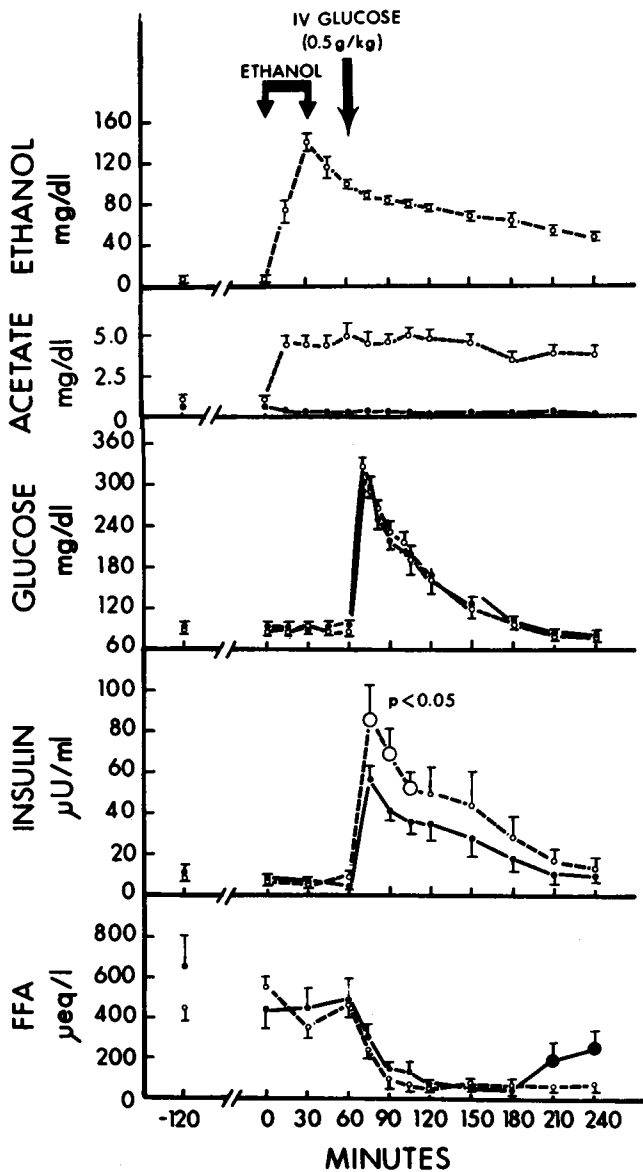


Figure 1. Effects of ethanol on glucose tolerance and insulin. Plasma concentrations of ethanol, acetate, glucose, insulin, and FFA after intravenous infusion of ethanol (0.75 g/kg from 0 to 30 min, broken line and open circles) or saline (solid line and solid circles) and glucose (0.5 g/kg from 60 to 65 min). Shown are mean±SEM. Large symbols indicate statistically significant differences between ethanol and saline studies. I, liter. $n = 6$.

ethanol concentrations declined at a constant rate and reached 48±5 mg/dl at 240 min. These ethanol concentrations resulted in mild motor hyperactivity at the beginning and sleepiness during the later phases of the study in most subjects.

Blood acetate concentrations rose quickly after ethanol infusion to a peak of 4.7±0.4 mg/dl and remained elevated at about this level throughout the study. Acetaldehyde (not shown) was undetectable (< 0.008 mg/ml) in peripheral venous blood in all subjects at the time of peak ethanol concentration (30 min) and remained undetectable in five of six subjects at the end of the study in accordance with previously reported data (25).

Ethanol infusion had no significant effects on plasma glucose concentrations nor the rate of glucose disappearance from plasma after intravenous glucose.

Ethanol had no effect on basal, but significantly enhanced glucose-stimulated insulin concentrations between 70 and 105 min. For instance, at 70 min, insulin concentrations with ethanol were 86±14 µU/ml compared with 56±8 µU/ml with saline ($P < 0.05$). The total 3-h glucose-stimulated insulin response (area under the curve between 60 and 240 min) was 54% greater with ethanol than with saline (7,153±1,710 vs. 4,650±905 µU/ml·min⁻¹, $P < 0.025$).

Plasma FFA concentrations were not significantly different in ethanol vs. control studies except during the last hour in which plasma FFA concentrations in the ethanol study remained suppressed, while those in the control study started to rise towards baseline levels. After glucose, FFA concentrations decreased precipitously from 480±121 and 470±61 µeq/liter at 60 min to 69±16 and 46±17 µeq/liter at 120 min in control and ethanol studies, respectively ($P < 0.01$).

Counterregulatory hormones (Fig. 2). Except for transient and isolated increases in norepinephrine at 75 min and epinephrine at 30 min, ethanol had no significant effects on plasma norepinephrine, epinephrine, or glucagon concentrations.

Oxidation of CHO, fat, and protein (Fig. 3). To assess possible effects of ethanol on intracellular substrate utilization, we measured CHO, fat, and protein oxidation rates by indirect calorimetry. The use of this method in the presence of ethanol, was validated in the [¹⁴C]ethanol oxidation studies (vide infra).

Ethanol infusions had no significant effect on mean resting metabolic rates (944±6 and 931±1 cal/min per 1.73 m² in ethanol and control studies, respectively).

Basal (-30 to 0 min) rates of CHO oxidation were 40±5 and 46±5 mg/min per 1.73 m² in control and ethanol studies, respectively. Saline or ethanol infusions had no statistically significant effects on CHO oxidation. Glucose infusion in controls caused a 249% increase in CHO oxidation at 120 min (from 47±12 to 117±11 mg/min per 1.73 m², $P < 0.025$). This glucose-stimulated increase was virtually abolished by ethanol and the total integrated above basal 3-h CHO oxidation was diminished by 80% (from 7.1±0.2 to 1.4±0.2 g per 1.73 m², $P < 0.05$).

Basal rates of fat oxidation in control and ethanol studies were similar (69±8 and 72±5 mg/min per 1.73 m²). Ethanol infusion suppressed fat oxidation by 87% from 72±3 (at 0 min) to 9±3 mg/min per 1.73 m² (at 60 min) ($P < 0.001$). Saline had no effect. After glucose infusion, fat oxidation decreased in controls in a reciprocal fashion to the increase in CHO oxidation (from 60±10 to 32±6 mg/min per 1.73 m²) and then rose toward basal values. In ethanol studies after glucose infusion,

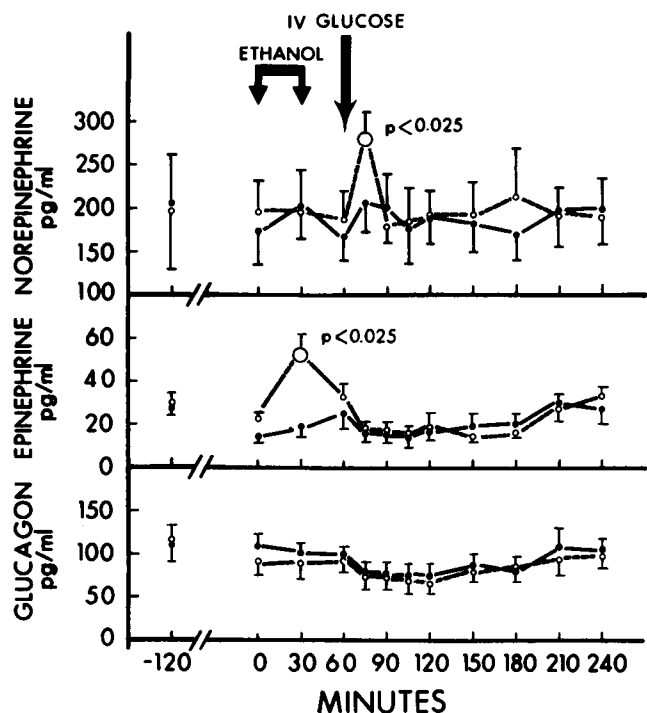


Figure 2. Effects of ethanol on counterregulatory hormones. Plasma concentrations of norepinephrine, epinephrine and glucagon after intravenous infusion of ethanol (or saline) and glucose. $n = 6$. Symbols are as in Fig. 1.

fat oxidation rates decreased further to almost 0 mg/min per 1.73 m² at 150 min and then rose slowly to 11±4 mg/min per 1.73 m² at 240 min. Ethanol decreased total integrated 4-h fat oxidation by 79% (from 12.0±1.8 to 2.5±0.2 g/1.73 m², $P < 0.0025$).

Basal protein oxidation rates were similar in control and ethanol studies (45±4 and 39±6 mg/min per 1.73 m²). In controls protein oxidation rates did not change during saline or glucose infusions. In response to ethanol it decreased from 41±5 (at 0 min) to 24±6 mg/min per 1.73 m² (at 90 min, $P < 0.005$). Ethanol decreased total integrated 4-h protein oxidation by 39% (from 11.0±0.7 to 6.7±1.2 g per 1.73 m², $P < 0.025$).

Ethanol metabolism and oxidation (Table 1 and Fig. 4). Ethanol (59.5±1.8 g) containing [1-¹⁴C]ethanol (46.8±2.9 μCi) was infused into six subjects and the exhaled CO₂ was collected quantitatively (protocol 3). During the 4-h study, 25.97±2.1 g of ethanol disappeared from the blood. Combined excretion of ethanol in urine and breath was 1.23±0.1 g, and 1.88±0.2 g of ethanol remained in the blood as acetate. Thus, the amount of ethanol that was metabolized (ethanol disappeared from blood - ethanol excreted - ethanol remaining in blood as acetate) was 21.8±2.1 g/4 h or 36.4±3.0% of the administered dose. Acetaldehyde was not considered in these calculations because of unmeasurably low blood concentrations in all subjects. This omission was inconsequential. Since acetaldehyde concentrations were < 0.008 mg/ml (the detection limit of our assay), < 0.44 g of ethanol was left in the blood as acetaldehyde, representing < 0.7% of the administered ethanol dose.

Fig. 4 shows the observed and corrected sp act of ¹⁴CO₂, which were used to calculate ¹⁴CO₂ production from [1-¹⁴C]-

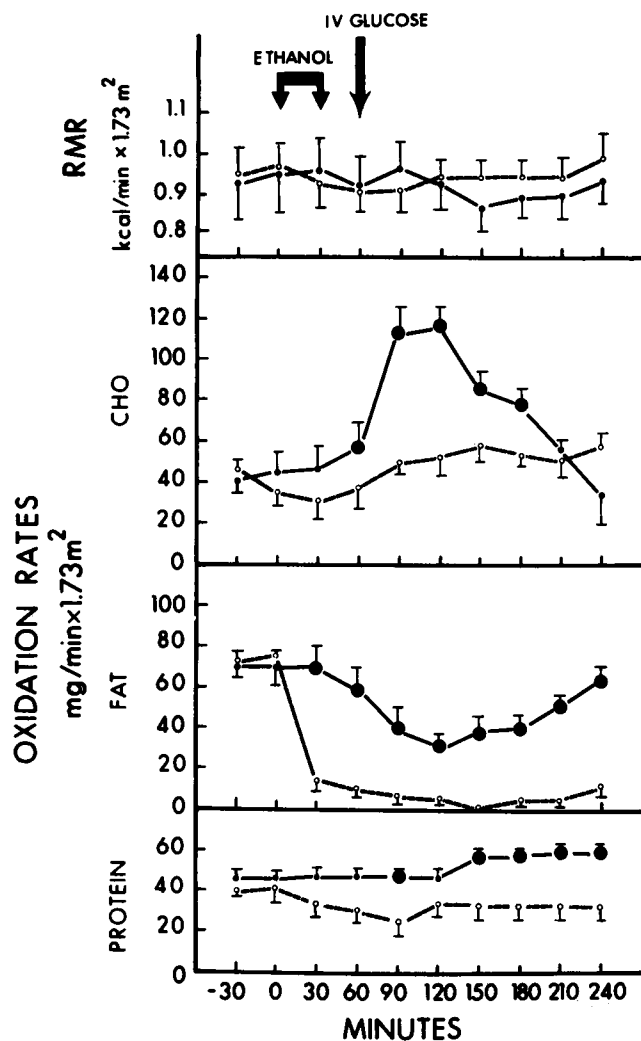


Figure 3. Effects of ethanol on resting metabolic rates (RMR) and rates of CHO, fat, and protein oxidation. Experimental design and symbols are as in Fig. 1. $n = 6$.

ethanol in four of six subjects. In the two subjects whose bicarbonate pool was primed with NaH¹⁴CO₃, ¹⁴CO₂ sp act were stable throughout the studies (6.3 and 3.9 nCi/mmol, respectively). Of the 46.8±2.9 μCi of [1-¹⁴C]ethanol infused, 17.5±1.7 μCi or 37.2±2.3% were recovered as ¹⁴CO₂ and thus had been completely oxidized. Since about the same amount of nonradioactive ethanol was metabolized (36.4±3.0%), essentially all of the ethanol which had been metabolized during the 4-h study was oxidized (100.7±3.7%). These results made it possible to correct the indirect calorimetry data for ethanol oxidation and to calculate rates of CHO and fat oxidation.

Glucose turnover (Fig. 5, protocol 2). Basal rates (0 min) of glucose appearance (G_{Ra}) were similar in ethanol and control studies. Infusion of ethanol caused a 30% decrease in G_{Ra} (from 2.4±0.2 to 1.7±0.1 mg/kg per min, $P < 0.05$) while saline had no effect. Because of the reduced basal G_{Ra} , peak total (endogenous + exogenous) G_{Ra} at 90 min was also less in ethanol than in control studies (8.7±0.7 vs. 9.7±0.6 mg/kg per min, $P < 0.005$) and remained lower than in controls until the end of the study. The total integrated 3-h glucose appearance

Table I. Ethanol Metabolism and Oxidation

Subjects	Ethanol infused	Ethanol* disappeared from blood	Ethanol excreted		Ethanol remaining in blood as acetate	Ethanol metabolized [‡]	[1- ¹⁴ C]ethanol			Ethanol Oxidized (% of infused)	
			Urine	Breath			Infused	Oxidized to ¹⁴ CO ₂	Metabolized (% of infused)		
											% of infused
1	64	35.54	0.916	0.193	2.67	31.76	49.6	50	22.5	45.0	0.907
2	60	27.60	0.628	0.215	2.07	24.69	41.1	50	20.0	40.0	0.973
3	54	22.92	1.174	0.148	1.95	19.65	36.4	50	21.2	42.5	1.168
4	62	22.92	0.843	0.221	2.05	19.81	31.9	50	15.3	30.6	0.959
5	58	20.90	0.917	0.195	0.794	18.99	32.7	50 [§]	15.5	31.0	0.948
6	59	19.57	1.692	0.238	1.743	15.89	26.9	31 [§]	10.5	33.8	1.092
Mean	59.5	25.97	1.028	0.202	1.88	21.80	36.43	46.83	17.5	37.15	1.007
SEM	1.8	2.1	0.138	0.012	0.23	2.10	2.98	2.89	1.7	2.3	0.037

* Amount that has disappeared from blood between 0 and 240 min (including ethanol metabolized, excreted, and present as acetate). [‡] Equals ethanol disappeared minus ethanol excreted in urine and breath minus ethanol equivalents present in plasma as acetate at end of study. [§] Bicarbonate pool was prelabeled with NaH¹⁴CO₃ (see text for details).

(60–240 min) was 24% less in ethanol than in control studies (515±25 vs. 677±31 mg/kg, *P* < 0.005).

Basal rates of glucose disappearance (*G_{Rd}*) were similar in ethanol and control studies. Ethanol infusion suppressed *G_{Rd}* by 38% (from 2.20±0.4 to 1.4±0.2 mg/kg per min, *P* < 0.05) while saline had no effect. *G_{Rd}* was lower in ethanol than in saline experiments during all except one time period (120 min). The total integrated 3 h (60–240 min) glucose disappearance in the presence of ethanol was 21% less compared with controls (533±25 vs. 677±30 mg/kg, *P* < 0.005).

Effect of ethanol on glucose disposal (Fig. 6). Intravenous administration of a glucose load after ethanol infusion was associated with increased release of insulin but unchanged glucose tolerance (Fig. 1), suggesting acute onset of insulin resistance. To study the ethanol-induced insulin resistance in more detail, we performed hyperinsulinemic-euglycemic

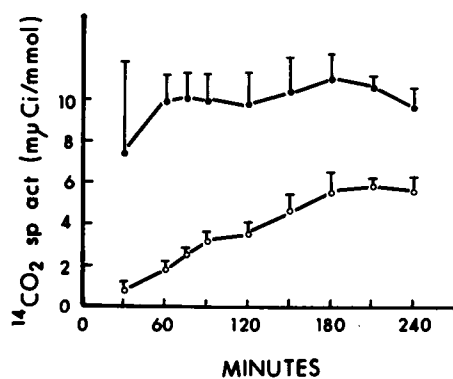


Figure 4. (Open circles) ¹⁴CO₂ sp act determined in four normal subjects after intravenous injection of 50 μCi of [1-¹⁴C]ethanol (protocol 3). (Solid circles) Corrected ¹⁴CO₂ sp act. The correction was done according to Issekutz et al. (15) and takes into account the slow exchange between the labeled CO₂ originating from [1-¹⁴C]ethanol oxidation and the unlabeled body bicarbonate pool as well as the empirical fact that maximal ¹⁴CO₂ sp act does not exceed 80% of its theoretical value.

clamp studies (protocol 4). Steady-state ethanol concentrations of 56±2 mg/dl were attained from 125 to 210 min. The second ethanol bolus and maintenance infusion raised steady-state ethanol concentrations to 117±4 mg/dl during the last

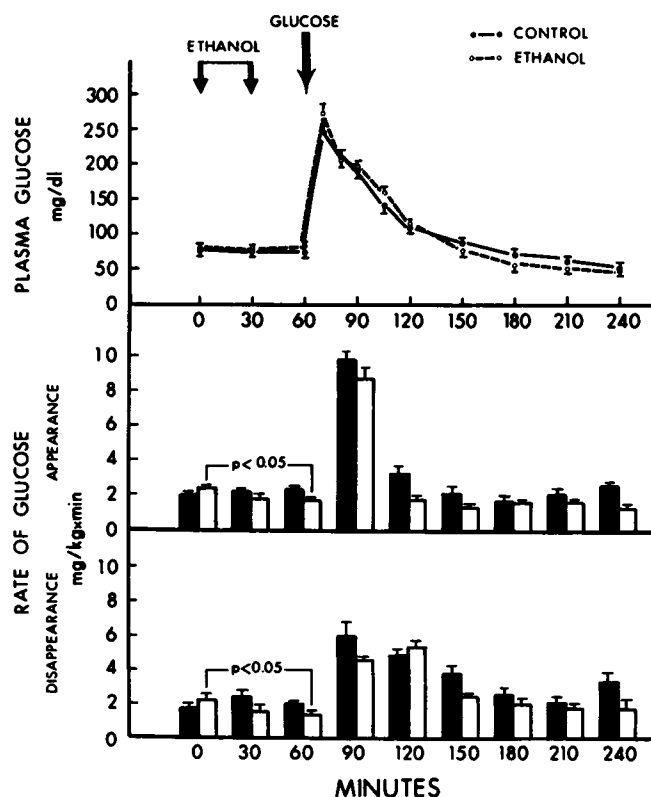


Figure 5. Effects of ethanol on glucose turnover. Plasma glucose and rates of glucose appearance and disappearance after intravenous infusion of ethanol (or saline) and glucose. Columns represent mean rates of glucose appearance or disappearance over 30-min periods. Symbols are as in Fig. 1. *n* = 4.

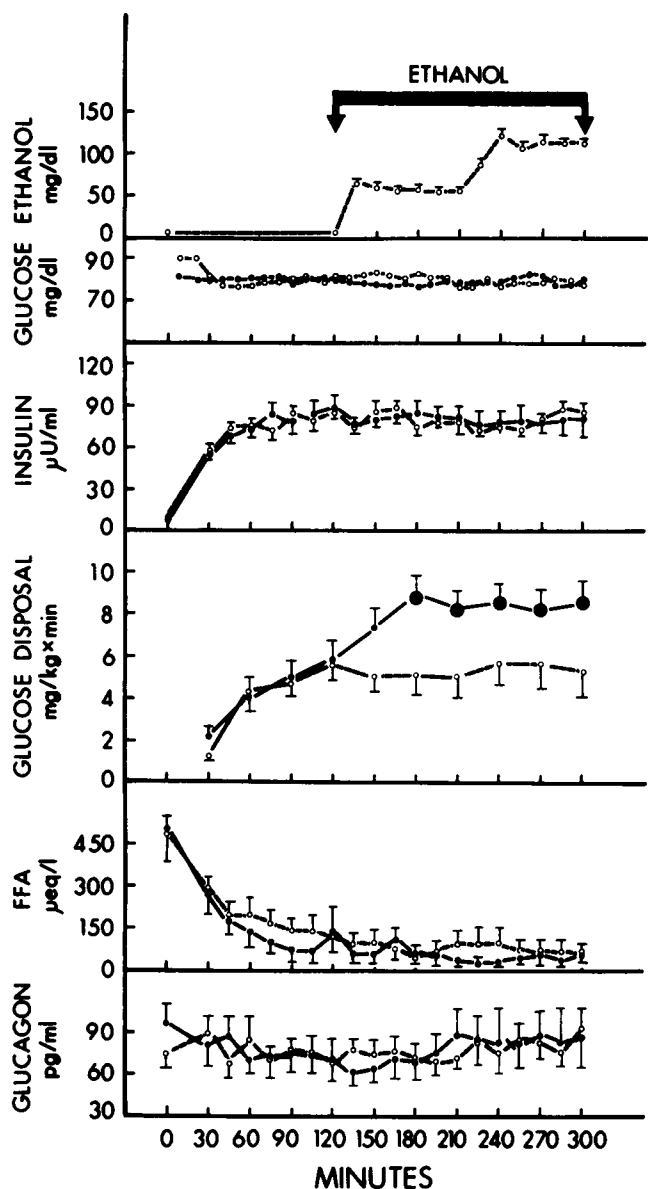


Figure 6. Effects of ethanol on glucose disposal during hyperinsulinemic-euglycemic clamps. Regular insulin (1 mU/kg per min) was infused from 0 to 300 min. Euglycemia was maintained by a variable, feedback-controlled glucose infusion. Glucose disposal represents the amount of glucose infused to maintain euglycemia. Ethanol was infused from 120 to 300 min (see text for details). Symbols are as in Fig. 1. I, liter. $n = 6$.

hour of the study. There was no difference in the effect of the two ethanol concentrations on glucose disposal nor on any of the other parameters measured. The most likely reason for this was that ethanol oxidation was maximal at the low ethanol concentration (~ 56 mg/dl) because of the low K_m of hepatic alcohol dehydrogenase (about 5 mg/dl). It also suggested that the observed effects were not caused by ethanol itself, but probably by one or more of its metabolites. Plasma glucose concentrations were maintained at 80 ± 0.4 mg/dl and 79 ± 0.4 mg/dl in ethanol and control studies, respectively.

Plasma insulin concentrations rose from 9 ± 1 to 80 ± 2 μ U/ml in ethanol studies and from 8 ± 1 to 81 ± 1 μ U/ml in

control studies as result of the insulin infusion (1 mU/kg per min). At these insulin concentrations, hepatic glucose production is known to be completely suppressed and the rate of glucose infused (by the Biostator) to maintain euglycemia equals the rate of glucose disposal, i.e., the amount of glucose leaving the extracellular glucose space (26). Since ethanol inhibited G_{Ra} , it seemed reasonable to assume that hepatic glucose production remained completely suppressed during ethanol plus insulin infusions. Glucose disposal rates prior to ethanol were not significantly different in ethanol or control studies (5.8 ± 0.8 vs. 5.9 ± 0.9 mg/kg per min at 120 min). In response to continued hyperinsulinemia, glucose disposal rates in the control studies rose from 5.8 ± 0.8 to a peak of 8.8 ± 1.1 mg/kg per min and remained at about this level. This increase was completely abolished by ethanol which kept glucose disposal rates at about 5.7 ± 1.1 mg/kg per min throughout. Integrated 3-h (120–300 min) glucose disposal with ethanol was 947 ± 132 mg/kg vs. $1,485 \pm 154$ mg/kg with saline (-36% , $P < 0.02$).

Ethanol had no effects on free fatty acid or glucagon concentrations.

In vitro insulin binding. Specific insulin binding to cultured rat hepatocytes was $7.2 \pm 0.3\%$ of the added trace ($n = 19$). No significant differences in specific binding were noted at any of the five ethanol concentrations (from 100 to 1,000 mg/dl) studied. This was true regardless of whether cell cultures were preincubated in the presence of ethanol or whether ethanol was added to the ^{125}I -insulin-binding assay mixture. Scatchard analysis of competitive insulin binding in the presence of 200 mg/dl of ethanol showed no effect of ethanol on binding affinity or receptor number. There was also no effect of ethanol on tracer stability as evidence by TCA precipitability nor on hepatocyte viability as judged by trypan blue exclusion.

Discussion

The major findings in this study, namely the profound inhibitory effect of ethanol on the oxidation of other fuels, were obtained by indirect calorimetry. In order to use this technic in the presence of ethanol, it was necessary to demonstrate that all the ethanol, which had disappeared from the blood during the glucose infusion study (24 ± 2.9 g) and was not excreted in urine or breath, nor remained in the body as acetate, had been completely oxidized to CO_2 and water. This was accomplished in six subjects in whom $^{14}\text{CO}_2$ was collected quantitatively for 4 h after they received ethanol containing trace amounts of $[1-^{14}\text{C}]$ ethanol followed by intravenous glucose (protocol 3). The results showed that virtually all the ethanol which was metabolized was oxidized to CO_2 and water (Table I). This allowed us to correct the respiratory quotients obtained by indirect calorimetry for the amounts of O_2 consumed and CO_2 generated from ethanol oxidation and to determine true CHO and fat oxidation rates.

To our knowledge, similar studies have not previously been performed in nonfasting human subjects. Bartlet and Barnett (27) have measured conversion of $[^{14}\text{C}]$ ethanol to $^{14}\text{CO}_2$ in pyruvate-treated rats. Similar to our results, they found that $\sim 40\%$ of the administered radioactivity was recovered in expired CO_2 within 4 h. Recently, it has been reported that ethanol can be esterified with long chain fatty acids in certain organs including pancreas, liver, and heart (28). The

quantities of ethanol metabolized through this non-oxidative pathway, however, appear to be too small (< 1% of the oxidative pathway) to affect our calculations.

The dose of ethanol given in the glucose infusion studies (protocol 1) resulted in plasma ethanol concentrations ranging from 140 ± 8 mg/dl (at the end of the ethanol infusion) to 48 ± 5 mg/dl (at the end of the study). These concentrations permitted ethanol oxidation to proceed at a constant rate of ~ 100 mg/min during the 4-h study (24 g/4 h, Table I) providing ~ 710 cal/min or 61% of total metabolic requirements ($1,157 \pm 7$ cal/min). During the basal (pre-glucose) phase this resulted in an 87% decrease in fat oxidation (from 72 to 9 mg/min per 1.73 m², $P < 0.001$) and a decline in its caloric contribution to the total metabolic rate from $\sim 70\%$ to $< 10\%$. It is noteworthy, that plasma FFA concentrations, the substrate for fat oxidation, did not change during this time (see Fig. 1). During the basal period (0–60 min), ethanol had no statistically significant effects on CHO or protein oxidation. Possible reasons may have been the small contributions of CHO and protein oxidation to the total metabolic rate at that time (each contributed $< 20\%$); another, the relatively small number of subjects, which may have prevented the observed decreases in mean CHO and protein oxidation rates from reaching statistical significance.

After glucose infusion, ethanol oxidation continued unchanged, the already severely depressed fat oxidation decreased further, approaching 0 mg/min at 150 min, and the 249% rise in CHO oxidation seen in controls (at 120 min) was mostly abolished. Thus, ethanol appeared to act as a preferred fuel, almost totally replacing fat as substrate for oxidation during the basal period and preventing the large glucose load, which entered the intracellular space after intravenous glucose administration, from being oxidized. Our results are compatible with earlier work by others showing that ethanol inhibited oxidation of FFA and of alanine in rat liver (29–31). They also support the proposition by others that the ethanol induced inhibition of fat oxidation may play a role in the fatty infiltration of the liver that accompanies prolonged alcohol intake (for review see Lieber [32]). The cause for the preferred oxidation of ethanol was not investigated. It is known, however, that ethanol is quantitatively oxidized in the liver to acetate (33). Acetate is a common product of the metabolism of CHO, fat, and some amino acids. It is readily activated to acetyl-coenzyme A and promptly oxidized in the citric acid cycle (34, 35).

A quantitative assessment of the effect of ethanol on CHO, fat, and protein oxidation has not been reported previously. Indirect calorimetry has been used to examine this question during the early part of this century (36–38). These investigators were unable to separate ethanol oxidation from the oxidation of other fuels and thus could not quantitate rates of CHO, fat, and protein oxidation. Nevertheless, they observed a decrease in the respiratory quotient under the influence of ethanol which they interpreted as an indication that ethanol had replaced other substrates as fuel for oxidation.

In addition to its profound effects on fuel oxidation, ethanol also altered the rates at which glucose entered and disappeared from its extracellular space. During the basal period (0–60 min) ethanol depressed G_{Ra} by 30% (from 2.4 ± 0.2 to 1.7 ± 0.1 mg/kg per min) and G_{Rd} by 38% (from 2.2 ± 0.4 to 1.4 ± 0.2 mg/kg per min). Since rates of appearance and disappearance were reduced similarly, there was no change in plasma glucose concentration.

After glucose infusion, ethanol caused a substantial increase in insulin release, as evidenced by an increase of $\sim 50\%$ in the area under the plasma insulin concentration curve (Fig. 1). This suggested that there was an acute increase in insulin resistance, requiring additional insulin to maintain normal glucose tolerance. It is also possible that ethanol and/or its metabolites may have enhanced the stimulatory action of other insulin secretagogues. Ethanol-induced insulin resistance and inhibition of glucose transport were demonstrated unequivocally in the hyperinsulinemic-euglycemic clamp studies which showed a 36% reduction in glucose disposal. By sampling venous instead of arterial blood, we may have slightly overestimated the decrease in glucose disposal, if a-v glucose differences during the euglycemic-hyperinsulinemic clamps were substantially larger than the 6 mg/dl determined in three separate experiments (see Methods). The observation that ethanol-related changes in CHO oxidation and insulin secretion occurred simultaneously suggested, but did not prove, that these two phenomena may have been causally related. Ethanol has been shown to inhibit uridine diphosphate glucose-glycogen-transferase in rat diaphragm (39) and to inhibit glycogen synthesis in rat liver (40). Thus, it may inhibit not only oxidative but nonoxidative pathways for glucose utilization and result in a decrease in glucose transport.

Our findings of a decrease in the rate of glucose disappearance (glucose turnover studies) were in agreement with those of others. Kreisberg et al. (41) in the only other glucose turnover study in humans, reported that ethanol given orally to 12-h fasted normal subjects, resulting in blood ethanol concentrations ranging from 35 to 75 mg/dl, did not change glucose tolerance but had a small ($\sim 10\%$) inhibitory effect on rates of glucose appearance and disappearance and markedly inhibited conversion of glucose to lactate. Lochner et al. (42) determined hepatic glucose output in 2–3-d fasted dogs by measuring arterial venous glucose concentration differences and found that ethanol (blood concentrations from 10 to 110 mg/dl) decreased glucose utilization by 25%. Yki-Jarvinen and Nikkila (43) using the hyperinsulinemic-euglycemic clamp technic, found that ethanol reduced glucose disposal by 23%. Clark and Evans (44) have shown that ethanol reduced glucose uptake by the isolated rat diaphragm. Lastly, potentiation of glucose mediated insulin release by ethanol has been reported by several investigators (4, 5, 7).

In our studies, changes in plasma levels of counterregulatory hormones could not be incriminated as a cause for the decreased glucose utilization. Glucagon concentrations did not change in response to ethanol. The peak plasma epinephrine concentration after ethanol administration (52 pg/ml) was too low to significantly affect glucose metabolism (45). The ethanol effect on plasma norepinephrine (peak concentration 285 pg/ml) may have reflected a physiologically significant activation of the sympathetic nervous system (46); however, whereas the ethanol effect on carbohydrate oxidation lasted for $2\frac{1}{2}$ h, the change in plasma norepinephrine was only transient (< 30 mm). Human growth hormone and cortisol were not measured because there is sufficient evidence in the literature to indicate that both hormones have no acute inhibitory effects on glucose utilization (47). It is therefore unlikely that human growth hormone or cortisol played a role in the development of insulin resistance which was seen within minutes after ethanol administration.

To investigate further the mechanisms by which ethanol

could have inhibited glucose disposal, we assessed its effect on insulin binding to cultured rat hepatocytes in vitro. Ethanol has been reported to change adipocyte surface membrane fluidity (48) which conceivably could alter insulin binding and action. We found, however, that ethanol, even at very high concentrations (from 100 to 1,000 mg/dl) did not influence insulin binding.

In conclusion, we have shown that under the conditions of this study, ethanol replaced fat as the predominant fuel for oxidation providing approximately two-thirds of basal caloric requirements. When an intravenous glucose load was given, ethanol prevented its oxidation. In addition, ethanol caused acute and comparable reductions in the rates of glucose appearance and disappearance. Ethanol did not alter glucose tolerance, because its diabetogenic effects on glucose disappearance and oxidation were counterbalanced by its potentiating action on glucose-stimulated insulin release and its reduction in hepatic glucose production (G_{RA}). These data suggest, that in individuals with impaired insulin secretory reserve, the net effect of ethanol may be a deterioration of glucose tolerance and may occasionally result in overt diabetes.

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