

EVALUATION OF ANTIOXIDANT AND ANTI-INFLAMMATORY PROPERTIES OF AQUEOUS EXTRACT OF WILD MUSHROOMS COLLECTED FROM HIMACHAL PRADESH

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ABSTRACT

Objective: The objective of this study is to elucidate phytochemical contents, antioxidant, and anti-inflammatory activity of wild mushrooms collected from Himachal Pradesh.

Methods: Quantitative analysis of phytochemical constituents was carried out using standard methods while 1,1 diphenylpicrylhydrazyl nitric oxide, hydrogen peroxide-free radical scavenging assay were used to evaluate the antioxidant properties of selected wild mushrooms. Anti-inflammatory capacity of samples was evaluated by human red blood cell membrane stabilization and albumin denaturation inhibition method.

Results: The results obtained revealed that *Pleurotus floridanus* showed higher total phenol, flavonoid carotenoid, and ascorbic acid contents of 61.13±2.3 mg/g, 15.2±1.13 mg/g, 12.42±0.42 µg/g, 17.36±0.40 µg/g, and 14.55±0.58 mg/g. All the species showed antioxidant potential, but *P. floridanus* proved to be more active while *Macrocybe* sp. proved to be least one.

Conclusion: The mushroom species analyzed have been showed to be good source of phytoconstituents, antioxidants, and also possess anti-inflammatory properties, thus can be used in the management of oxidative stress-induced disease.

Keywords: Wild mushrooms, Scavenging effect, Total phenol, Antioxidant, Anti-inflammatory.

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INTRODUCTION

Wild mushrooms since primords of human civilization have been extensively consumed mainly due to their delicate and unique flavor and taste they provide, having high amounts of water, protein fiber, and carbohydrates, and being rich in sources and minerals [1-4]. In mushrooms, lipids are present in low values which make them as valuable foods due to low-calorie and fat content food [2-4]. Due to these properties, their marketability, and economic contribution enhances by approximately two billion dollars [5]. Beyond these nutritional characteristics, mushrooms due to their richness in bioactive compounds that presented anticancer, antioxidant, and antimicrobial properties extensively studied for their medicinal properties [4,6,7]. The growing emergence of drug resistance bacterial strains is a serious threat to humankind for the effective treatment of infections; hence, to minimize the development of drug resistance strains it is important to choose the most appropriate antibiotics and to use them appropriately [8]. Stress on the body due to aging, obesity, and detrimental lifestyles are another significant health issue which often takes the form of oxidative tissue damage. In human body, during energy production in the mitochondrial respiratory chain, phagocytosis, fertilization, arachidonic acid metabolism, and xenobiotic metabolism-free radicals are constantly formed. Free radicals such as hydroxyl radicals (-OH), superoxide radicals and hydrogen peroxide radicals (H₂O₂) damage DNA, provoke uncontrolled chain reactions including lipid peroxidation which leads to the development of cancer, impair enzymes and structural proteins, neurological and cardiovascular diseases, cataracts, rheumatoid arthritis, and diabetes [9,10]. Reactive oxygen species activate nuclear factors; thus, inducing the synthesis of cytokines and therefore are responsible in the development of systemic inflammatory response syndrome. Adhesion molecules and inflammatory mediators are also formed later. Free radicals lead to loss of function and cell death by reacting with different

cell components at site of inflammation [11]. All organisms have defense systems which are not sufficient to protect against free radicals. Synthetic antioxidants including hydroxyanisole and butylated hydroxytoluene have side effects and thought to be responsible for liver damage and carcinogenesis [12]. As a result, natural antioxidants such as vitamin A, C, and E, carotenoids, flavonoids, and other simple phenolic compounds are preferred in food applications to prevent oxidative damage and to protect the human body [13]. Although many studies on nutrient compositions of different mushroom species have been done. The present study sought to investigate the antioxidant and anti-inflammatory activity of aqueous extract of mushrooms collected from Himachal Pradesh. The antioxidant activity of the mushrooms extract was assessed by measuring their free radical scavenging activity.

METHODS

Sample collection

Fruiting bodies of three mushrooms were collected from the forests of Solan and Baru Sahib and were labeled as sample no. 104, 105, and 127. The fruiting bodies were brought to laboratory in sterile plastic bags washed thoroughly with sterile distilled water and 0.001% mercuric chloride and with the aid of sterile blade; these were aseptically break lengthwise exposing the inner tissue. A small piece of tissue from each sample was aseptically transferred on to the plates of malt extract agar. Three wild isolates, namely, *Calocybe indica*, *Agrocybe aegerita*, and *Macrocybe* sp. were procured from Directorate of Mushroom Research Centre, Solan India.

Identification of fruiting bodies

The fruiting bodies of three unknown collected mushrooms were identified as *Pleurotus floridanus* (KU 925874), *P. ostreatus* (KU 892064), and *Trametes versicolor* (KU 892065) on the basis their morphological feature and molecular characterization.

Determination of bioactive compounds

Standards L- ascorbic acid, quercetin, gallic acid, Ferrous sulfate, Sodium salicylate, Folin - Cioalteu reagent, 1,1-Diphenyl-2 picrylhydrazyl (DPPH), and naphthylethylene diaminedihydrochloride was obtained from Sigma (St. Louis, MO, USA). All other chemicals and solvents were of analytic grade.

Phenolic compound estimation

Extraction solution (100 µl) was mixed with 1.5 ml of folin-cioalteu reagent and allowed to incubate at room temperature for one minute followed by the addition of sodium carbonate (60 g/l). The tubes were then vortex-mixed for about 15 seconds, and then, allowed to stand for 90 minutes in the dark at room temperature. Absorbance was then measured at 725 nm. Gallic acid was used as standard curve with concentration ranges from 1 to 100 µg/ml ($R^2=0.996$), and the results were expressed as mg gallic acid equivalent (GAE)/g DW. All experiments were performed in triplicates [14].

Flavonoid contents

A volume of 1.25 ml of distilled water was mixed with 250 µl extracts of mushrooms and 75 µl of 5% NaNO_2 solution. After 5 minutes, 150 µl of 10% $\text{AlCl}_3 \cdot \text{H}_2\text{O}$ was added, and then, 275 µl of distilled water and 500 µl of 1 M NaOH were added to the mixture after 6 minutes. The solution was then mixed well and intensity of pink color was measured at 510 nm. Quercetin with concentration range of 20-100 µg/ml ($R^2=0.9938$) was used for calibration curve. The results were expressed as milligram of (+) quercetin equivalent (CEs) per gram of extract [15].

Ascorbic acid determination

The dried mycelial extract (100 mg) was extracted with 10 ml of 1% metaphosphoric acid for 45 minutes at room temperature and filtered through Whatman no.4 filter paper. 1 ml of filtrate was then mixed with 9 ml of 2, 6 dichloroindophenol, and absorbance was measured within 30 minutes at 515 nm against a blank. The content of ascorbic acid would be calculated on the basis of calibration curve of authentic L-ascorbic with concentration range of 2-125 µg/ml ($R^2 0.9929$). All experiments were performed in triplicates [16].

β-carotene and lycopene determination

100 mg of dried extract was vigorously shaken with 10 ml of acetone-hexane mixture (4:6) for one minute and filtered through Whatman no. 4 filter paper. The absorbance of filtrate was then measured at 453, 505, and 663 nm [17].

Content of β-carotene and lycopene were calculated using following equations

$$\text{Lycopene (mg/100 ml)} = -0.0458 A_{663} + 0.372 A_{505} - 0.0806 A_{453}$$

$$\beta\text{-carotene (mg/100 ml)} = 0.216 A_{663} - 0.304 A_{505} + 0.452 A_{453}$$

Antioxidant activity determination

DPPH scavenging activity

Free radical scavenging activity of aqueous extract of all isolates of mushrooms was determined by DPPH method. 0.1 mM of DPPH solution was prepared in methanol, and 0.5 ml was added to 0.5 ml of extract. The mixture was then vortexed thoroughly and left for 45 minutes in the dark at room temperature. The absorbance was measured at 515 nm against blank. A lower absorbance represents higher DPPH scavenging activity. The capability of scavenging DPPH radicals was calculated using following equation

$$\text{DPPH scavenging effect (\%)} = (1 - A_s/A_c) \times 100$$

Where AC is the absorbance of control containing DPPH solution and AS is absorbance of extract solution containing DPPH [18].

Nitric oxide scavenging assay

Nitrite detection method was used to assess the interaction of extract of macro-fungi with nitric oxide. Sodium nitroprusside (10 mM) in 0.5 M

phosphate buffer pH 7.4 which spontaneously produced nitric oxide in aqueous solution was used as chemical source. Nitric oxide interacts with oxygen to produce stable products which lead to the production of nitrites.

A volume of 5 ml of riess reagent (α -naphthylethylenediamine 0.1% in water and sulfanilic acid 1% in H_3PO_4) was added after incubation for 5 hrs at 37°C. The absorbance of the solution was measured at 546 nm [19].

Hydrogen peroxide-free radical scavenging activity

A solution of H_2O_2 (40 mM) was prepared in phosphate buffer (7.4 pH). The concentration of H_2O_2 was determined by absorption at 230 nm using spectrophotometer. Extracts (100-1000 µg/ml) were then added to H_2O_2 solution, and after 10 minutes of incubation, the absorbance was determined at 230 nm. Phosphate buffer without H_2O_2 was used as blank. Ascorbic acid was used as standard curve [20].

Anti-inflammatory activity

Human red blood cell (HRBC) membrane stabilization assay

Blood was collected from healthy volunteer who was not taken non-steroidal anti-inflammatory drugs (NSAID) for two weeks. The collected blood was mixed with equal volume of sterilized alsever solution (2% dextrose, 0.8% sodium citrate, 0.5% citric acid, and 0.42% sodium chloride in water). It was centrifuged at 3000 rpm for 15 minutes. Packed cell was washed with isosaline. Different concentration of extract (100, 500, and 1000 and 2000 µg/ml) was prepared in isosaline. The assay mixture contained 0.5 ml of extract 1ml phosphate buffer (0.15M, pH 7.4) 2 ml of hyposaline (0.36%), and 0.5 ml HRBC suspension and incubated at 3°C for 30 minutes and then centrifuged at 3000 rpm for 20 minutes. Diclofenac sodium was used as reference standard. Instead of hyposaline 2 ml of distilled water was used as control. The hemoglobin content in supernatant solution was estimated using spectrophotometer at 560 nm [21].

% hemolysis was calculated by

$$\% \text{ inhibition of hemolysis} = 100 (\text{OD1} - \text{OD2}) / \text{OD1}$$

% stabilization of HRBC membrane stabilization was calculated as:

$$\% \text{ age protection} = 100 - (\text{OD2} / \text{OD1}) \times 100$$

Where,

OD1 = Optical density of hypotonic buffer saline alone (control).

OD2 = Optical density of test sample in hypotonic buffer saline solution.

Albumin denaturation assay

A volume of 5 ml of reaction consist of 0.2 ml of egg albumin (from fresh hen's egg), 2.8 ml of phosphate buffer saline (pH 6.4), and 2 ml of varying concentration of aqueous extracts of each samples so that final concentrations become 100, 500, 1000, and 2000 µg/ml. The distilled water was used as control. The mixture was then incubated at 37°C in a BOD incubator for 15 minutes and then heated at 70°C for 5 minutes. After cooling, the absorbance was measured at 660 nm. Diclofenac sodium was used as reference drug [22].

The percentage inhibition of protein denaturation was calculated using following formula.

$$\% \text{ inhibition} = 100 \times (V_t/V_c - 1)$$

V_t = absorbance of test sample

V_c = absorbance of control

Statistical analysis

All the experimental analysis was carried out in triplicates. The results are expressed as mean values and standard deviation. The results were analyzed using one-way analysis of variance, (ANOVA) followed

by Turkey's HSD test using SAV v.9.1.3 program. Differences at $p < 0.05$ were considered to be statistically significant.

RESULTS

Quantitative phytochemical analysis

The determination of quantitative phytochemical content of aqueous extracts of selected mushrooms results as given in Table 1 revealed that total contents of phenols ranges from 22.46±1.87 to 53.3±0.93 mg/g, total flavonoid contents ranges from 2.97±0.11 to 12.91±0.11 mg/g, total ascorbic acid contents ranges from 2.77±0.03 to 12.01±0.89 mg/g, total β carotene contents ranges from 0.979±0.42 to 8.384±0.42 μ g/g, and total lycopene content ranges from 1.75±0.51 to 6.57±0.49. The extract of mushroom *P. floridanus* had the highest amount of phenol content (61.13±2.3 mg GAE/g) while *Macrocybe* sp. had least amount of phenol present (22.46±1.87 mg GAE/g). Flavonoids and ascorbic acid were also obtained but in small amount in comparison to phenolic compounds while β -carotene and lycopene were found in vestigial amount (>91 μ g/g) as shown in Table 1.

Evaluation of bioactivities of aqueous extracts of isolates

For *in-vitro* evaluation of antioxidant properties of mycelial aqueous extract from different isolates multiple assays based on different antioxidant mechanisms were carried out these were DPPH radicals, hydrogen peroxide, and nitric oxide-free radical scavenging activity assays.

DPPH-free radical scavenging activity

DPPH being stable-free radical has advantage of being unaffected by certain side reaction such as metal ion chelation and enzyme inhibition [23]. Freshly prepared DPPH exhibit purple deep color with absorption maximum at 517 nm, the deep purple of DPPH disappears due to quenching of its free radicals by antioxidants present in the medium, i.e., by providing hydrogen atom or by electron donation, through free radical attack on DPPH molecule, and convert them to a colorless bleached product, 2,2 diphenyl-1 hydrazine. As shown in Table 2 at 100-1000 μ g/ml, the scavenging ability of aqueous extract was highest for *P. floridanus* ranges from 44.4±0.06 to 64.16±0.21 followed by *P. ostreatus* ranges from 41.3±0.25 to 62.33±0.15, *T. versicolor* 38.5±0.01 to 60.56±0.12, *C. indica* 28.5±0.1 to 47.06±0.15, *Agrocybe aegerita* 24.86±0.15 to 45.36±0.06, *Macrocybe* sp. 21.06±0.2 to 38.76±0.15, respectively. Ascorbic acid was taken as standard which showed excellent scavenging activity of 82.96±0.2% as shown in Table 2.

Nitric oxide-free radical scavenging assay

Nitric oxide radical is associated with the development of inflammatory disorders [24]. Nitric oxide radicals generated from sodium nitroprusside were found to be inhibited by all aqueous extracts, as shown in Table 3. The aqueous extract of isolate *P. floridanus* showed maximum percentage inhibition effect on nitric oxide ranges which ranges from 28.23±0.05 to 62.83±0.11 followed by *P. ostreatus* having percentage inhibition effect ranges from 27.13±0.06 to 60.76, *T. versicolor* 24.7±0.12 to 56.37±0.06, *C. indica* 22±0.17 to 49.03±0.2, *Agrocybe aegerita* 20.06±0.2 to 36.1±0.17, and *Macrocybe* sp. having inhibition effect ranges from 18.42±0.04 to 33.63±0.05. Ascorbic acid was taken as standard showing excellent inhibition of nitric oxide radicals generated from sodium nitroprusside with percentage inhibition ranging from 30.45±0.08 to 83±0.17 as shown in Table 3.

Hydrogen peroxide-free radical scavenging assay

Hydrogen peroxide scavenging activity of selected mushrooms is shown in Table 4. All the mushrooms scavenge H_2O_2 in a dose-dependent manner. *P. floridanus* has highest H_2O_2 scavenging activity with value ranging from 39.46±0.15 to 62.21 while *Macrocybe* sp. has least scavenging activity with value ranging from 22.23±0.15 to 43.83±0.11. Ascorbic acid was taken as standard showing the excellent scavenging activity of H_2O_2 with percentage inhibition ranging from, 46.69±0.2 to 82.99±0.11.

Anti-inflammatory activity

The aqueous extracts were studied for their *in vitro* anti-inflammatory activity by HRBC membrane stabilization and albumin denaturation inhibition assay.

HRBC membrane stabilization assay

The inhibition of hypotonicity-induced HRBC membrane lysis, i.e., stabilization of HRBC membrane was taken as a measure of anti-inflammatory activity. The percentage membrane stabilization and hemolysis for aqueous extracts and diclofenac sodium were done at 100, 500, 1000, and 2000 μ g/ml. The aqueous extract of *P. floridanus* showed high percentage stabilization, i.e., 94.3±1.1, and low percentage hemolysis, i.e., 7.6±1.5, 9.66±1.5%, while extracts prepared from *Macrocybe* sp. showed low percentage stabilization, i.e., 77.7±1.3, and high percentage hemolysis, i.e., 36±1. With the increasing concentration, the membrane hemolysis is decreased as shown in Table 5 while membrane stabilization/protection increases as shown in Table 6. Diclofenac sodium was taken as standard showing excellent percentage stabilization, i.e., 96.28±0.8 and lowest percentage hemolysis, i.e., 5±1. Hence, anti-inflammatory activity of extracts was concentration dependent as shown in Table 5 and 6.

Albumin denaturation inhibition assay

Denaturation of proteins is well-documented cause of inflammation. The mechanism of the anti-inflammatory activity and the ability of extract protein denaturation were studied in the present investigation. All three aqueous extracts of mushrooms were effective in inhibiting heat-induced albumin denaturation. Maximum inhibition was observed from *P. floridanus* (86.5±0.7), followed by *P. ostreatus* (82.5±0.8), *T. versicolor* (80.4±1.5), *C. indica* (77.2±0.7), *Agrocybe aegerita* (74.6±0.9), and *Macrocybe* sp. (73±0.7) at concentration of 2000 μ g/ml. Diclofenac sodium, a standard anti-inflammation drug showed maximum inhibition 94.9±1.2 for all isolates at concentration of 2000 μ g/ml as shown in Table 7.

DISCUSSION

The antioxidant properties of fruits, vegetables, and mushrooms are due to the major contribution of polyphenols [25]. To study the importance of polyphenol compounds and their presence in the mushrooms, the estimation of these compounds in six isolates was done. The estimation of phenol was done by Folin-Ciocalteu's reagent by measuring the absorbance at 725 nm. Estimation of flavonoids was done by ferric chloride method, and estimation of ascorbic acid was done by metaphosphoric method. Estimation of β carotene and lycopene was done by Nagata and Yamashita [17].

Phenols are considered to be one of the major components of plants and mushrooms [26]. Due to the presence of hydroxyl group, these

Table 1: Bioactive compounds present in aqueous extract of isolates

Mushrooms	Phenols (mg/g)	Flavonoid (mg/g)	β -carotene (μ g/g)	Lycopene (μ g/g)	Ascorbic acid (mg/g)
<i>Pleurotus ostreatus</i>	53.3±0.93 ^b	13.02±0.12 ^b	8.384±0.42 ^b	6.57±0.49 ^b	12.01±0.89 ^b
<i>P. floridanus</i>	61.13±2.3 ^a	15.2±1.13 ^a	12.42±0.42 ^a	17.36±0.40 ^a	14.55±0.58 ^a
<i>T. versicolor</i>	44.57±2.87 ^c	11.44±0.12 ^c	7.17±1.17 ^{bc}	4.22±0.76 ^{cd}	8.68±0.59 ^c
<i>C. indica</i>	31.80±1.89 ^d	7.98±0.17 ^d	5.62±0.43 ^c	4.44±0.52 ^c	7.50±0.58 ^c
<i>A. aegerita</i>	25.89±1.42 ^e	7.37±0.11 ^{de}	1.66±0.61 ^d	3.03±0.31 ^{de}	4.17±0.89 ^d
<i>Macrocybe</i> sp.	22.46±1.87 ^e	6.24±0.11 ^e	0.979±0.42 ^d	1.75±0.51 ^e	2.77±0.03 ^d

A. aegerita: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*.

^{a, b, c, d, e}: different letters means significant difference at $P < 0.05$. Values are Mean±SD (n=3)

Table 2: Scavenging ability of aqueous extract from different isolates on 1, 1-diphenyl-2- picrylhydrazyl radicals

Isolates	Concentration of extract used ($\mu\text{g/ml}$) (Scavenging activity of free DPPH radicals) (%)			
	100	125	500	1000
<i>P. ostreatus</i>	41.3 \pm 0.25 ^c	46.9 \pm 0.2 ^c	53.46 \pm 0.15 ^c	62.33 \pm 0.15 ^c
<i>P. floridanus</i>	44.4 \pm 0.06 ^b	49.73 \pm 0.15 ^b	56.53 \pm 0.1 ^b	64.16 \pm 0.21 ^b
<i>T. versicolor</i>	38.5 \pm 0.1 ^d	43.6 \pm 0.1 ^d	53.13 \pm 0.07 ^d	60.56 \pm 0.12 ^d
<i>C. indica</i>	28.5 \pm 0.1 ^e	31.06 \pm 0.18 ^e	40.32 \pm 0.15 ^e	47.06 \pm 0.15 ^e
<i>A. aegerita</i>	24.86 \pm 0.15 ^f	29.63 \pm 0.15 ^f	36.23 \pm 0.2 ^f	45.36 \pm 0.06 ^f
<i>Macrocybe</i> sp.	21.06 \pm 0.2 ^g	27.43 \pm 0.15 ^g	30.63 \pm 0.1 ^g	38.76 \pm 0.15 ^g
Ascorbic acid	44.23 \pm 0.15 ^a	51.2 \pm 0.2 ^a	70.96 \pm 0.11 ^a	82.96 \pm 0.15 ^a

DPPH : Diphenylpicrylhydrazyl, *A. aegerita*: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*. ^{a, b, c, d, e, f, g}; Different letters means significant difference at $P < 0.05$. Values are Mean \pm SD ($n=3$)

Table 3: Scavenging ability of aqueous extract from different isolates on Nitric oxide free radicals

Isolates	Concentration of extract used ($\mu\text{g/ml}$) (Scavenging activity of NO ₂ free radicals) (%)			
	100	125	500	1000
<i>P. ostreatus</i>	27.13 \pm 0.06 ^c	37.4 \pm 0.57 ^c	47.37 \pm 0.06 ^c	60.76 \pm 0.57 ^c
<i>P. floridanus</i>	28.23 \pm 0.05 ^b	39.1 \pm 0.086 ^b	51.1 \pm 0.09 ^b	62.83 \pm 0.11 ^b
<i>T. versicolor</i>	24.7 \pm 0.12 ^d	33.5 \pm 0.11 ^d	42.96 \pm 0.12 ^d	56.37 \pm 0.06 ^d
<i>C. indica</i>	22 \pm 0.17 ^e	31.3 \pm 0.1 ^e	37.3 \pm 0.06 ^e	49.03 \pm 0.2 ^e
<i>A. aegerita</i>	20.06 \pm 0.2 ^f	23.6 \pm 0.6 ^f	30.46 \pm 0.5 ^f	36.1 \pm 0.17 ^f
<i>Macrocybe</i> sp.	18.42 \pm 0.04 ^g	22.01 \pm 0.15 ^g	28.5 \pm 0.17 ^g	33.63 \pm 0.05 ^g
Ascorbic acid	30.45 \pm 0.08 ^a	42.23 \pm 0.12 ^a	61.06 \pm 0.06 ^a	83 \pm 0.17 ^a

A. aegerita: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*. ^{a, b, c, d, e, f, g}; different letters means significant difference at $P < 0.05$. Values are Mean \pm SD ($n=3$)

Table 4: Scavenging ability of aqueous extract from different isolates on hydrogen peroxide free radicals

Isolates	Concentration of extract used ($\mu\text{g/ml}$) (Scavenging activity of H ₂ O ₂ free radicals) (%)			
	100	125	500	1000
<i>P. ostreatus</i>	35.1 \pm 0.264 ^c	42.1 \pm 0.17 ^c	51.5 \pm 0.28 ^c	56.7 \pm 0.21 ^c
<i>P. floridanus</i>	39.46 \pm 0.15 ^b	46.06 \pm 0.15 ^b	54.3 \pm 0.31 ^b	62.21 \pm 0.16 ^b
<i>T. versicolor</i>	31.36 \pm 0.11 ^d	39.86 \pm 0.12 ^d	43.36 \pm 0.2 ^d	52.33 \pm 0.3 ^d
<i>C. indica</i>	28.6 \pm 0.2 ^e	34.03 \pm 0.25 ^e	42.3 \pm 0.2 ^e	49.4 \pm 0.3 ^e
<i>A. aegerita</i>	25.46 \pm 0.15 ^f	31.8 \pm 0.11 ^f	37.06 \pm 0.21 ^f	46.23 \pm 0.31 ^f
<i>Macrocybe</i> sp.	22.23 \pm 0.15 ^g	30.09 \pm 0.2 ^g	34.96 \pm 0.21 ^g	43.83 \pm 0.11 ^g
Ascorbic acid	46.69 \pm 0.2 ^a	51.53 \pm 0.15 ^a	60.5 \pm 0.2 ^a	75 \pm 0.17 ^a

A. aegerita: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*. ^{a, b, c, d, e, f, g}; Different letters means significant difference at $P < 0.05$. Values are Mean \pm SD ($n=3$)

are known to eliminate free radicals and also associated with the antioxidant effect of whole system [27]. In the earlier studies, it was observed that higher content of phenols due to their redox properties accounts for better free radical scavenging effects and can play an important role in absorbing in neutralizing quenching of free radicals singlet and triplet oxygen or decomposing peroxides [28]. Ascorbic acid is a compound that occurs naturally having antioxidant properties that are necessary for wound healing and strengthening of blood vessels. Flavonoids, on the other hand, are known to be most promising polyphenolic compound having a broad spectrum of chemical and biological activities that includes radical scavenging activity, antiviral, anti-allergic, anti-inflammatory, and vasodilating action. It has been also found in various studies that out of many pharmacological effects

of phenol and flavonoids are linked together and act as free radical scavengers and strong antioxidants. Carotenoids including β carotene and lycopene exert an antioxidant function that includes quenching of singlet oxygen and other electronically excited molecules and progression of degenerative diseases [29].

In the present study, it was revealed that the aqueous extract taken for quantitative analysis constitutes phenolic compounds in good amount. These compounds possess antioxidant functions; hence, the studied mushrooms can be harnessed in the management of oxidative stress induced disease.

In a wide variety of pathological manifestation free radicals are known to play a definite role, therefore, search of natural scavenger of reactive oxygen and nitrogen species seems to be highly desirable to protect against oxidative stress responsible for causing numerous diseases.

Being free radical DPPH has the advantage of being unaffected by certain side reaction that includes metal ion chelation and enzyme inhibition [23]. DPPH exhibit deep purple when prepared freshly with maximum absorption at 517 nm, due to quenching of its free radicals by antioxidants present in the medium the deep purple colour disappears, i.e., by electron donation or by providing hydrogen atom *via* free radical attack on DPPH molecule and convert them to colourless bleached product known as 2,2 diphenyl-1 hydrazine. This results in decrease in the absorption at 517 nm, hence, more rapidly the decrease in absorbance more potent the antioxidant activity of the extract [30]. The results of the present investigation suggest that the mycelial cultures of mushrooms contain phytochemicals constituents that are capable of donating hydrogen to a free radical to scavenge the potential damage.

Nitric oxide is a free radical present in mammalian cells responsible for the regulation of physiological processes. Excess production of NO is associated with several diseases [31]. Under aerobic conditions, it reacts with oxygen to produce stable products known as nitrate and nitrite. Nitric oxide can be toxic if present in higher concentration, therefore, its inhibition is an important goal.

Hydrogen peroxide occurs naturally at lower concentration levels in air, water plants, microorganisms, food, and human body [32]. It has strong oxidizing properties and can be formed *in vivo* by oxidizing enzymes such as superoxide dismutase. It can slowly oxidize a number of compounds by crossing the membrane. Hydrogen peroxide is rapidly composed into oxygen and water which results in the production of (-OH) that initiate lipid per-oxidation which leads to DNA damage [33]. Aqueous extract of mushroom's mycelial cultures effectively scavenged hydrogen peroxide and may attribute to the presence of phenolic groups that can donate electrons to hydrogen peroxide, thereby neutralizing it into water.

In the present study, free radical scavenging assay of aqueous extract of six mushrooms mycelial cultures was tested, and it was revealed that *P. floridanus* have great potential of antioxidant activity while *Macrocybe* sp. have least one.

For *in vitro* assessment of anti-inflammatory properties of extracts of all mushrooms isolates. Hypotonicity-induced HRBC membrane stabilization and albumin denaturation bioassay were selected. By inhibiting hypotonicity-induced lysis of erythrocyte membrane, the extract exhibits membrane stabilization effect. The erythrocytes stabilization implies that the extract may stabilize the lysosomal membrane because it is analogous to lysosomal membrane [34]. In limiting, the inflammatory response by preventing the release of lysosomal constituents of activated neutrophil such as bactericidal enzymes and proteases, which cause further tissue inflammation and damage on extracellular release lysosomal membrane stabilization is important [35]. It is believed that some of the NSAIDs are known to possess membrane stabilization properties which may contribute to the potency of their anti-inflammatory effect. The exact mechanism of the membrane stabilization by the extract is not known

Table 5: Effect of aqueous extract and standard on HRBC membrane hemolysis

Concentration ($\mu\text{g/ml}$)	% hemolysis of isolates						
	Diclofenac sodium	<i>P. ostreatus</i>	<i>P. floridanus</i>	<i>T. versicolor</i>	<i>C. indica</i>	<i>A. aegerita</i>	<i>Macrocybe sp.</i>
100	56 \pm 2 ^e	68.3 \pm 1.5 ^c	63 \pm 1 ^d	78.6 \pm 1.5 ^b	82.3 \pm 1.5 ^b	87.6 \pm 1.5 ^a	91.6 \pm 1.5 ^a
500	39 \pm 1 ^f	51 \pm 1 ^e	39.3 \pm 1.5 ^f	56.3 \pm 1.5 ^d	60 \pm 1 ^c	63.6 \pm 1.5 ^b	75 \pm 1 ^a
1000	11.6 \pm 1.5 ^e	18.6 \pm 1.5 ^{cd}	14.6 \pm 1.5 ^{de}	20.3 \pm 2.1 ^c	23 \pm 2 ^c	28.3 \pm 1.5 ^b	36 \pm 1 ^a
2000	5 \pm 1 ^d	14.3 \pm 1.5 ^c	7.6 \pm 1.5 ^d	16 \pm 1 ^c	20.6 \pm 1.5 ^b	24 \pm 1 ^b	30 \pm 2 ^a

A. aegerita: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*.
^{a, b, c, d, e}: Different letters means significant difference at P<0.05. Values are Mean \pm SD (n=3)

Table 6: Effect of aqueous extract and standard on HRBC membrane stabilization

Concentration ($\mu\text{g/ml}$)	% stabilization of isolates						
	Diclofenac sodium	<i>P. ostreatus</i>	<i>P. floridanus</i>	<i>T. versicolor</i>	<i>C. indica</i>	<i>A. aegerita</i>	<i>Macrocybe sp.</i>
100	58.5 \pm 1.8 ^a	49.1 \pm 0.5 ^c	53.3 \pm 0.6 ^b	41.6 \pm 1.4 ^d	38.8 \pm 1.2 ^d	35.05 \pm 1 ^e	32.1 \pm 1.5 ^e
500	76.3 \pm 1.3 ^a	62.2 \pm 0.4 ^c	70.8 \pm 1.2 ^b	58.2 \pm 0.83 ^d	55.5 \pm 0.9 ^{de}	52.7 \pm 1.4 ^e	44.4 \pm 0.3 ^f
1000	91.4 \pm 1.1 ^a	86.2 \pm 1.2 ^{bc}	89.1 \pm 1.1 ^{ab}	84.7 \pm 1.6 ^c	82.9 \pm 1.4 ^c	78.9 \pm 1.3 ^d	73.3 \pm 0.9 ^e
2000	96.3 \pm 0.7 ^a	89.4 \pm 1.2 ^b	94.3 \pm 1.1 ^a	88.14 \pm 0.7 ^b	84.6 \pm 1.2 ^c	82.2 \pm 0.7 ^c	77.7 \pm 1.3 ^d

A. aegerita: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*.
^{a, b, c, d, e}: Different letters means significant difference at P<0.05. Values are Mean \pm SD (n=3)

Table 7: Effect of aqueous extract of mushrooms on albumin denaturation

Concentration ($\mu\text{g/ml}$)	% inhibition						
	Diclofenac sodium	<i>P. ostreatus</i>	<i>P. floridanus</i>	<i>T. versicolor</i>	<i>C. indica</i>	<i>A. aegerita</i>	<i>Macrocybe sp.</i>
100	43.1 \pm 1.3 ^a	31.54 \pm 2.3 ^b	35.2 \pm 1.6 ^b	26.2 \pm 1.4 ^c	24.6 \pm 0.2 ^{cd}	23 \pm 1.2 ^{cd}	21.9 \pm 1.2 ^d
500	60.04 \pm 1.3 ^a	44.4 \pm 0.66 ^{bc}	47.1 \pm 1.4 ^b	42.3 \pm 1.5 ^{cd}	41.3 \pm 1.1 ^{de}	38.9 \pm 0.3 ^e	34.9 \pm 0.3 ^f
1000	88.6 \pm 1.2 ^a	72.2 \pm 0.6 ^c	80.15 \pm 0.7 ^b	70.6 \pm 0.9 ^{cd}	69.1 \pm 0.7 ^d	64.3 \pm 0.9 ^e	57.1 \pm 0.7 ^f
2000	94.9 \pm 1.2 ^a	82.5 \pm 0.8 ^c	86.5 \pm 0.7 ^b	80.4 \pm 1.5 ^c	77.2 \pm 0.7 ^d	74.6 \pm 0.9 ^{de}	73 \pm 0.7 ^e

A. aegerita: *Agrocybe aegerita*, *C. indica*: *Calocybe indica*, *T. versicolor*: *Trametes versicolor*, *P. floridanus*: *Pleurotus floridanus*, *P. ostreatus*: *Pleurotus ostreatus*.
^{a, b, c, d, e}: Different letters means significant difference at P<0.05. Values are Mean \pm SD (n=3)

yet, and it is believed that hypotonicity-induced hemolysis may arise from shrinkage of the cells due to osmotic loss of intracellular electrolyte and fluid components.

Denaturation of tissue proteins was another cause of inflammatory disease [36]. For the anti-inflammatory drug development of agents that can prevent protein denaturation would be worthwhile. The stabilization of protein is indicated by the increments in absorbance of test samples with respect to control, i.e., inhibition of heat-induced protein (albumin) denaturation by extracts and reference drug diclofenac sodium [37].

The anti-denaturation effect was further supported by the change in viscosities. The viscosities of protein solutions increase on denaturation [38]. The relatively high viscosity of control dispersion substantiated this fact in the present study. The presence of all mushroom extracts taken in the study prevented this, implying inhibition of protein denaturation. The viscosities decreased when compared with control where no test extract/drug was added. However, the viscosities were found to decrease with concomitant decrease in concentration of test extract and reference drug as well. Although the viscosities of the test samples (extract/drug) of all concentrations were always less than that of control. This decrease in viscosities may be due to decrease in concentration of test extract/drug in reaction mixture, which resulted in decreased viscosity; and/or other uncertain physicochemical factors. Nevertheless, the viscosity data indicated inhibition of protein (albumin) denaturation. Further studies are required to study the effect of concentration of test agent on viscosity behavior of denatured protein dispersion.

CONCLUSION

The present study has been focused on antioxidant and anti-inflammatory potential of six mushrooms and determination of their

chemical composition in terms of phenolic compounds. Mushrooms consist of biologically active substances that act as remarkable products and can diversify everyday human diet as well as utilize for their medicinal properties. Hence, there is a great need of exploiting these mushrooms to new drug discovery.

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