

# EXTRACELLULAR BIOSYNTHESIS, OPTIMIZATION, CHARACTERIZATION AND ANTIMICROBIAL POTENTIAL OF ESCHERICHIA COLI D8 SILVER NANOPARTICLES

Mahmoud M. Nour El-Dein<sup>1</sup>, Zakaria A. M. Baka<sup>1</sup>, Mohamed I. Abou-Dobara<sup>1</sup>, Ahmed K.A. El-Sayed<sup>1</sup>, Mohamed M. El-Zahed<sup>\*1</sup>

Address(es): PhD student Mohamed El-Zahed,

<sup>1</sup> Damietta University, Faculty of Science, Department of Botany and Microbiology, New Damietta, Egypt.

\*Corresponding author: mohamed.marzouq90@gmail.com

ABSTRACT

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This study highlights the optimization of extracellular biosynthesis and antimicrobial efficiency of silver nanoparticles (AgNPs) using the crude metabolite of Escherichia coli D8 (MF06257) strain. The bacterial strain had been isolated from a sewage water stream located in Damietta City, Egypt. The optimum conditions for AgNPs production were at temperature 35°C, pH 7 and 1.5mM silver nitrate. The AgNPs biosynthesis was detected in culture filtrate within 1-2 minutes at room temperature (25±2°C) and sunlight. The characterization of AgNPs was studied by UV-Vis spectroscopy (maximum absorbance at 429 nm), X-ray diffraction (XRD) pattern (crystal planes were 110, 111, 200, 211, 220, and 311), transmission electron microscopy (TEM) (AgNPs were spherical in shape ranging from 6 to 17 nm), Fourier transform-infrared (FTIR) spectroscopy (the bands of symmetric and asymmetric amines were assigned at 3421.1 and 2962.13 cm<sup>-1</sup>, the stretching vibration band of aromatic and aliphatic (C-N) exist at 1392.35 and 1122.37 cm<sup>-1</sup> bands), Zeta potential analyser (AgNPs had a negative charge value; -33.6 mV) and size distribution by volume (the presence of capping agent enveloping the AgNPs with a mean size of 136.0-294.3 nm). Nitrate reductase (NR) was assayed as an important partner in the optimized production (the rate of NR reached to 2.18 U/ml). The study demonstrated that AgNPs are potent inhibitors of Staphylococcus aureus, E. coli, Pseudomonas aeruginosa, Alternaria alternata, Fusarium oxysporum and Aspergillus flavus. The antimicrobial activity of AgNPs was studied by TEM. TEM micrographs showed an inhibition of S. aureus cell multiplication. In case of F. oxysporum, a reduction in the size of treated cells, formation of a mucilage matrix connecting the hyphal cells together, the appearance of a big vacuole, lipid droplets an a severe leakage of cytoplasmic contents were detected. AgNPs exhibited MIC values of 6.25µg/ml and 50 µg/ml against S. aureus and Candida albicans, respectively. In addition, AgNPs showed synergy effects by their combination with fluconazole that increased fold areas especially against A. niger, A. flavus and F. oxysporum.

Keywords: Escherichia coli, silver, nanoparticles, optimization, antimicrobial

### INTRODUCTION

Among metallic nanoparticles, silver nanoparticles (AgNPs) have numerous applications in the field of nanobiotechnology due to their unique antimicrobial efficiency as growth inhibitors, killing agents or antibiotic carriers (**Hamidi** *et al.*, **2019**). AgNPs have widely attracted attention for the food, cosmetics and biomedical applications (Sondi and Salopek-Sondi, 2004).

In the last few years, different chemical and physical methods had been included in AgNPs synthesis. These incorporated methods produced contaminated, toxic AgNPs in low yields. So, scientific researchers went to the biological synthesis of AgNPs using microorganisms (Wang et al., 2019). Through microbial biosynthesis, numerous scientists used bacterial strains in AgNPs biosynthesis due to their rapid growth rate and highly efficient enzymatic system (Galvez et al., 2019). The use of bacterial crude metabolites was embedded as reducing agents of silver ions into safe and ecofriendly AgNPs that called extracellular biosynthesis (De Souza et al., 2019). The extracellular production is more prioritized than the intracellular which requires extraction and purification of AgNPs from the microbial growth. In addition, the extracellular production was confirmed to include high amounts of proteins which acted as capping agents (Annamalai and Nallamuthu, 2016). One of the mechanistic aspects for AgNPs biosynthesis is the secreted enzymes by bacteria that act as reducing agents for silver ions (Quinteros et al., 2016). The shape and size of the biosynthesized nanoparticles (NPs) could be handled throughout controlling the production parameters such as concentration of metal ions, temperature, incubation period, pH and effect of solar irradiation (Sumitha et al., 2019).

AgNPs have a strong bactericidal effect against a broad spectrum of bacteria such as *Pseudomonas* sp., *Acinetobacter* sp., *Escherichia* sp., *Vibrio* sp. and *Salmonella* sp. (Paul and Londhe, 2019). Furthermore, the biosynthesized AgNPs showed significance antifungal potential against *Aspergillus flavus*, *A. nomius* and *A. parasiticus*, *Alternaria alternata*, *Fusarium* sp., *Candida tropicalis*  and *C. albicans* was reported (**Bocate** *et al.*, **2019**). This makes the AgNPs a potential candidate as a new generation of antifungal agents.

The present work aimed to obtain a potent bioreductant bacterium possessing the ability to synthesize AgNPs extracellularly with efficient antimicrobial activity.

# MATERIAL AND METHODS

#### Chemicals

Culture media was purchased from Difco Laboratories, Detroit, Mich. Chemicals were purchased from Oxoid Ltd., England. Silver nitrate was purchased from Panreac Quimica S.L.U, Barcelona, Spain. Fluconazole (Diflucan) was purchased from Pfizer Inc., New York, NY.

### **Microbial strains**

The *E. coli* D8 strain was isolated from a sewage water stream located at Damietta City, Egypt). It was identified classically according to Bergey's Manual of Systematic Bacteriology (**Imhoff, 2005**). The 16S rRNA gene sequence was also performed in order to confirm its identification and deposited in the database under accession number MF06257.

The bacterial and fungal strains used for the antimicrobial activity were kindly provided by the culture collection of Microbiology Laboratory, Faculty of Science, Damietta University, Egypt (Table 1).

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Microorganism	Strains				
	Bacillus cereus ATCC6633				
	Staphylococcus aureus ATCC25923				
Bacteria	Escherichia coli ATCC25922				
	Klebsiella pneumoniae ATCC33495				
	Pseudomonas aeruginosa ATCC27853				
Yeast	Candida albicans ATCC10231				
	Aspergillus niger van Tiegh				
	A. flavus Link ex Fries group				
Fungi	A. fumigatus Fresenius				
	Alternaria alternata Fr. Keissler				
	Fusarium oxysporum f. sp. lycopersici Fol4287				

### **Experimental procedure**

### Extracellular biosynthesis of silver nanoparticles

*Escherichia coli* D8 MF06257 was grown in nutrient broth (NB) medium and incubated at 37°C for 48 hours at 150 rpm. After an incubation period, the bacterial crude metabolites were collected throughout centrifuging at 5000 rpm for 20 minutes aseptically. Two hundred  $\mu$ L of bacterial crude metabolites were added to 20 ml of an autoclaved aqueous solution of 1mM silver nitrate (1% v/v) in triplicates. The negative control was prepared by adding 200  $\mu$ L of the NB medium into 20 ml of the silver nitrate solution. All samples were incubated at 150 rpm for 5 days at 37°C in dark. After incubation, the appearance of brown colour was observed and measured spectrophotometrically as an indication of the production of AgNPs (Shahverdi *et al.*, 2007). The reaction mixtures were measured in the range of 370 to 750 nm at a resolution of 1 nm using a UV-Vis spectrophotometer (Beckman DU-40) against control test tube as the blank (silver nitrate solution and nutrient broth medium) (Krishnaraj *et al.*, 2012).

### Optimization of extracellular biosynthesized silver nanoparticles

The optimizing process included different parameters such as different concentrations of silver nitrate (0.5, 1, 1.5, 2, 2.5, 3, 3.5 and 4 mM), temperatures (25-45°C), pH values (5-10) and different incubation periods (12, 24, 36, 48, 60, 72 and 84 hours) (**Krishnaraj** *et al.*, **2012**). In addition, the synthesis of AgNPs was tested in the presence of solar irradiation at different periods of time (1, 2, 3, 4, 5 and 6 min) (**Sumitha** *et al.*, **2019**). All samples were measured spectrophotometrically in the range of 370 to 750 nm in order to detect the AgNPs biosynthesis.

### Characterization of biosynthesized silver nanoparticles

The X-ray diffraction (XRD) pattern of the AgNPs was recorded at  $2\theta$  values between  $10^{\circ}$  and  $80^{\circ}$  using a Cu X-ray tube at 40 kV and 30 mA with the X-ray diffractometer (model LabX XRD-6000, Shimadzu, Japan) at Nanotechnology Center, Kafrelsheikh University, Egypt).

The following characterizations of AgNPs were performed at TEM Unit at Mansoura University, Egypt. The shape and size of the optimized AgNPs were examined using TEM, a drop coating of nanocolloidal solution into carboncoated copper grid (Type G 200,  $3.05 \mu$ M diameter, TAAP, USA) was prepared and kept overnight under vacuum desiccation before loading them onto a specimen holder. TEM micrographs of samples were taken using TEM (JEOL, JEM-2100, Japan). Size distribution by volume and charge of AgNPs were measured by Zeta Potential Analyser (Malvern Zetasizer Nano-ZS90, Malvern, UK). A colloidal solution was used in this instrument by withdrawing 1 ml of solution into the instrumental cuvette for measuring (**Ruud** *et al.*, **1976; Hanaor** *et al.*, **2012**). The AgNPs capping agents were analysed by Fourier Transform Infrared Spectroscopy (FTIR) spectrum. It was done for the freeze-dried powder of AgNPs using FT/IR-4100 type A in the diffuse reflectance mode at a resolution of 16 cm<sup>-1</sup> at the range of 400-4000 cm<sup>-1</sup> (Siddique *et al.*, **2013**).

# Nitrate reductase assay

The assay of nitrate reductase (NR) was performed according to **Harley (1993)** depending on the reduction of nitrate into nitrite by nitrate reductase (NR). The NR activity was calculated pertain the amount of the produced nitrite during 60 minutes using 10 ml of sample. Production of one  $\mu$ mol nitrite/h/ml was defined as one unit of NR activity (U/ml).

### Antimicrobial activities of silver nanoparticles

### Agar well diffusion method

Agar well diffusion assay was performed according to the guidelines of the Clinical and Laboratory Standards Institute (Clinical and Laboratory Standards, 2006). The antimicrobial activities of the optimized biosynthesized AgNPs were investigated at concentrations of 50, 100, and 150  $\mu$ g/ml in dimethyl sulfoxide (DMSO). The antibacterial potential was tested against Gram-positive bacteria (Bacillus cereus ATCC6633 and S. aureus ATCC25923), Gram-negative bacteria (E. coli ATCC25922, Pseudomonas aeruginosa ATCC27853 and Klebsiella pneumoniae ATCC33495) on the Mueller-Hinton agar (MHA) plates. The antifungal potential was performed against fungal species (Aspergillus flavus Link ex Fries group, A. fumigatus Fresenius, A. niger van Tiegh, Fusarium oxysporum f. sp. lycopersici Fol4287 and A. alternata Fr. Keissler) with DOX agar plates in addition to C. albicans ATCC10231 using Bacto casitone agar plates. 100  $\mu$ L culture of each strain (0.5 McFarland standard (1-2 × 108 CFU/ml)) was inoculated separately into the agar plates. Wells (8 mm) were inoculated with 50 µl of AgNPs colloids. Penicillin G and Fluconazole were used as antibacterial and antifungal positive control standard, respectively. Plates were incubated at 37°C or 28°C for bacteria and fungi, respectively. After incubation, zones of inhibition (ZOI) were measured in terms of millimetres after 24 hours and 7 days for bacteria and fungi, respectively.

The synergistic effect of AgNPs combined with Penicillin G or Fluconazole was also determined against the pathogenic bacteria and fungi. 30  $\mu$ g Penicillin G or Fluconazole was loaded with 20  $\mu$ l (100  $\mu$ g/ml) of biosynthesized AgNPs colloid and tested by comparing to Penicillin G or Fluconazole alone (100  $\mu$ g/ml).

The increase in the fold area was calculated the mean of ZOI of each antimicrobial agent alone (Penicillin G or Fluconazole) and combined with AgNPs using the equation (B2 - A2)/A2, where A and B were ZOI for antimicrobial agent alone and combined with AgNPs, respectively (**Birla** *et al.*, **2009**). All the experiments were performed in triplicate.

### Minimal inhibitory concentration (MIC)

The MIC values for *S. aureus* ATCC25923 and *C. albicans* ATCC10231 were measured using broth microdilution method according to the guidelines of the National Committee for Clinical Laboratory Standards (NCCLS) (Clinical Laboratory Standards, 2008; 2017). A 0.5 McFarland standard of *S. aureus* ATCC25923 and *C. albicans* ATCC10231 were grown on Mueller-Hinton broth (MHB) and RPMI broth medium, respectively. Serial solutions of AgNPs, Penicillin G and Fluconazole (6.25-125  $\mu g/ml$  in water) were tested. Mixtures were incubated at 37 °C and 35 °C for *S. aureus* ATCC10231, respectively. After 48 hr, the growth turbidity was measured using a spectrophotometer against the growth control at 630 nm wavelength to determine n-values for each antimicrobial agent.

# Transmission Electron Microscopy (TEM) of nanosilver treated microorganisms

The exponential-phase cultures of *S. aureus* ATCC25923 and *F. oxysporum* f. sp. *lycopersici* Fol4287 were subjected to silver nanocolloids (6.25, 50, 100 and 150 µg/ml and 50, 100 and 150 µg/ml, respectively) for 2 hours at 37°C and 30°C, respectively. Normal bacteria and fungi were included as controls. The cell cultures were centrifuged at 5000 rpm for 20 minutes, and then washed 3 times with distilled water. Fixative solution (2.5% glutaraldehyde in 0.1 M cacodylate buffer at pH 7) was added and left for 20 minutes at room temperature. The fixative was removed and then 0.1 M buffer was added for washing and postfixed with osmium tetroxide (2%, in the same buffer) for 90 minutes. The fixed cells were dehydrated using graded series of ethanol. The dehydrated cells were embedded in Epon-Araldite (1:1) mixture for 1 hour that polymerized at 65°C for 24 hours. The cells were cross section using an ultra-microtome (50 µm), double-stained with uranyl acetate and lead citrate and exposed to observation on carbon-coated copper grids (Type G 200, 3.05 µM diameter, TAAP, U.S.A.) using TEM (JEOL JEM-2100, Japan).

### Statistical analysis

The data were statistically analyzed using software system SPSS version 18. All values in the experiments were expressed as the mean  $\pm$  standard deviation (SD) and were analyzed with one-way Analysis of Variance (ANOVA). The significant level was set at p<0.05.

### RESULTS

### Optimization of biosynthesized AgNPs

*Escherichia coli* D8 MF06257 biosynthesized AgNPs within 72 hours in dark conditions. The production of AgNPs was demonstrated by the peak at 429 nm in the UV-Vis spectra. Using 1% of bacterial supernatants and 1.5 mM

concentration greatly enabled AgNPs synthesis (Figure 1a). According to pH value, the brown colour appeared at pH (5-6) and its intensity was increased with the increase in pH value (Figure 1b). Stable and monodispersed AgNPs were synthesized at pH7. It was found that 35oC was the optimal temperature for AgNPs synthesis (Figure 1c).

The brown colour appeared within 72 hours (Figure 1d) during incubation in dark conditions while biosynthesis occurred throughout a minute in case of the presence of solar irradiation (Figure 1e).





**Figure 1** Optimization of AgNPs using *E. coli* D8 MF062579. (a) Different concentrations (0.5-4 mM) of AgNO3. (b) Effect of pH value on AgNPs synthesis. (c) Effect of temperature. (d) Different incubation periods through nanosilver biosynthesis at dark conditions. (e) Effect of solar irradiation on nanosilver biosynthesis.

# Characterization of the biosynthesized AgNPs

The XRD pattern for *E. coli* D8 (MF062579) crude metabolite and AgNPs were shown in Figure 2. It showed six characteristic peaks of AgNPs that appeared at 31.7°, 37.6°, 45.7°, 57°, 64.26°, and 77.4°, corresponding to respective crystal planes (110), (111), (200), (211), (220), and (311) (Galvez *et al.*, 2019).



Figure 2 The XRD pattern of *E. coli* D8 (MF062579) crude metabolite; (a) and the produced AgNPs; (b).

TEM images (Figure 3) show the spherical shaped and well-dispersed AgNPs. The particle size distribution analysis in the present study showed a mean size of 6-17 nm.



Figure 3 Transmission electron micrograph of produced AgNPs (scale bar = 100 and 50 nm).

The Zeta potential study established the negative charge (-33.6) of the AgNPs (Figure 4a) and the size distribution by volume showed the presence of a capping agent surrounded AgNPs having mean size of 136.0-294.3 nm (Figure 4b).



Figure 4 (a) Zeta potential measurement analysis of AgNPs. (b) Size distribution by volume.

Figure 5 illustrates the FTIR spectrum of *E. coli* D8 (MF062579) crude metabolite and AgNPs which confirmed the presence of proteins in the AgNPs biosynthesis. The stretch, primary and secondary amines vibrations bands were noted at  $3421.1 \text{ cm}^{-1}$  and  $2962.13 \text{ cm}^{-1}$ , respectively. The stretching vibration of molecule appeared at  $1658.48 \text{ cm}^{-1}$  and  $1596.77 \text{ cm}^{-1}$  bands. The stretch C-N vibration of aromatic and aliphatic amines existed at  $1392.35 \text{ cm}^{-1}$  and  $1122.37 \text{ cm}^{-1}$  bands.



Figure 5 FTIR spectrum of *E. coli* D8 MF062579 crude metabolite with and without AgNPs.

### Nitrate reductase activity

The activity of NR was measured and calculated in the *E. coli* D8 (MF062579) crude metabolite, at the rate of 2.18 U/ml.

## **Antimicrobial Potential OF AgNPs**

# Agar well diffusion method

The AgNPs synthesized by *E. coli* D8 (MF062579) isolate crude metabolite were centrifuged and dried using freeze dryer (SIM international, USA, FD8-8T) and tested against different pathogenic bacterial and fungal strains. There are

significant differences in antimicrobial effects between the samples with AgNps, Penicillin G and AgNPs & Penicillin G treatment. Highly significant (P < 0.05) was observed between the microbial strains mainly *S. aureus* ATCC25923, *E. coli* ATCC25922, *P. aeruginosa* ATCC27853 (Table 2), *A. alternata* Fr. Keissler, *F. oxysporum* f. sp. *lycopersici* Fol4287 and *A. flavus* Link ex Fries group (Table 3) and the diameter of inhibition zone.

**Table 2** Antibacterial potential of AgNPs in comparison with benzylpenicillin (Penicillin G) as a standard drug in addition to the synergy action.

 (Highly significant = \*\*p < 0.05; n = 3).

		Zone of inhibition (mm, mean) *					
	Concentration, µg/mL	Gram-positive bacteria		Gram-negative bacteria			
Antibacterial agent		Bacillus cereus	Staphylococcus aureus	Eccharichia cali	Pseudomonas	Klebsiella	
					aeruginosa	pneumoniae	
		ATCC0055	ATCC25923	ATCC25922	ATCC27853	ATCC33495	
	50	12 ± 0.03**	20 ± 0.14**	18 ± 0.14**	17 ± 0**	$14 \pm 0.14^{**}$	
AgNPs	100	15 ± 0**	24 ± 0**	21 ± 0**	20 ± 0.06**	$18 \pm 0.14^{**}$	
	150	19 ± 0.06**	27 ± 0.14**	24 ± 0.14**	24 ± 0**	22 ± 0**	
	50	12 ± 0.03**	$10 \pm 0^{**}$	36 ± 0.06**	11 ± 0.06**	-ve	
Penicillin G	100	$14 \pm 0^{**}$	12 ± 0.03**	38 ± 0**	$14 \pm 0^{**}$	-ve	
	150	16 ± 0**	15 ± 0.14**	40 ± 0**	21 ± 0.03**	-ve	
	50	13 ± 0.03**	$18 \pm 0^{**}$	19 ± 0.06**	18 ± 0.03**	13 ± 0**	
AgNPs & Penicillin G	100	15 ± 0**	21 ± 0**	23 ± 0**	20 ± 0**	18 ± 0**	
	150	17 ± 0.03**	25 ± 0.14**	27 ± 0**	25 ± 0**	21 ± 0**	

\*Mean surface area of the inhibition zone was calculated for each from the mean diameter  $\pm$  SD.

Table 3 Antifungal potential of AgNPs in comparison with Fluconazole as a standard drug in addition to the synergy action.

Antifungal agent		-	Zone of innibition (mm, mean) *						
		Concentration, µg/mL	<i>Aspergillus niger</i> van Tiegh	A. flavus Link ex Fries group	A. fumigatus Fresenius	Alternaria alternata Fr. Keissler	<i>Fusarium</i> oxysporum f. sp. lycopersici Fol4287	Candida albicans ATCC10231	
		50	$44\pm0.06$	$31\pm0.03$	$35\pm0$	$22\pm0.14$	$11\pm0.06$	$11\pm0.03$	
AgNPs		100	$47\pm0.03$	$35\pm0.06$	$37\pm 0.03$	$25\pm0.03$	$15\pm0.03$	$13\pm0$	
		150	$52\pm0.14$	$39\pm0.06$	$40\pm0.03$	$27\pm0.06$	$17\pm0.03$	$15\pm0.03$	
		50	$11\pm0.06$	$12\pm0$	$13\pm03$	$15\pm0$	$10\pm0$	$11\pm0$	
Fluconazole		100	$13\pm0$	$14\pm0$	$15\pm0.03$	$19\pm0$	$12\pm0.03$	$13\pm0$	
		150	$18\pm0.14$	$17\pm0$	$19\pm0.03$	$22\pm0.06$	$15\pm0.06$	$16\pm0.14$	
AgNPs Fluconazole	&	50	$49\pm0.06$	$38\pm0$	$27\pm03$	$19\pm0$	$16 \pm 0$	$10\pm0.03$	
		100	53 ± 0.03♦	41 ± 0 <b>♦</b>	$33\pm0$	$22\pm0$	$18 \pm 0.06 \blacklozenge$	$12\pm0$	
		150	$55\pm0.06$	$44\pm0.03$	$36 \pm 0$	$24\pm0.06$	$22\pm0.06$	$15\pm0$	

\*Mean surface area of the inhibition zone was calculated for each from the mean diameter  $\pm$  SD.

•Indicates significant larger inhibition zone than that of Fluconazole at p < 0.05 significant level.

The less inhibition effects of Fluconazole (15, 17 and 18 mm) were against the pathogenic fungi *F. oxysporum* f. sp. *lycopersici* Fol4287, *A. flavus* Link ex Fries group and *A. niger* van Tiegh, respectively. However, the biosynthesized AgNPs

revealed significant synergistic effects when companied with Fluconazole in addition to its antifungal activities showing higher fold areas (Table 4).

Table 4 Synergistic effect of AgNPs and Fluconazole.

Fungi	Zone of inhibition (m	Increase in fold area♦	
Fulgi	Fluconazole (A)	AgNPs & Fluconazole (B)	
Aspergillus niger van Tiegh	13	53	15.62
A. flavus Link ex Fries group	14	41	7.58
Fusarium oxysporum f. sp. lycopersici Fol4287	12	18	1.25

\*Mean surface area of the inhibition zone was calculated for each from the mean diameter.

•Increase in fold area was calculated as (B2 - A2)/A2, where A and B are the inhibition zones for Fluconazole and Fluconazole & AgNPs, respectively.

# Minimal inhibitory concentration

The biocidal action of *C. albicans* ATCC10231 growth was significantly higher at the concentrations of AgNPs 50,100 and 125  $\mu$ g/ml than lower concentrations. Fluconazole inhibited *C. albicans* ATCC10231 at 125  $\mu$ g/ml (Figure 6a). Both AgNPs and Penicillin G showed the same MIC values (6.25  $\mu$ g/ml) against *S. aureus* ATCC25923 in addition to complete inhibition at 25  $\mu$ g/ml (Figure 6b).





**Figure 6** (a) Effect of AgNPs and Fluconazole on the growth of *C. albicans* ATCC10231. (b) Effect of AgNPs and Penicillin G against *S. aureus* ATCC25923.

# TEM of nanosilver treated microorganisms

Antimicrobial activities of AgNPs against *S. aureus* ATCC25923 and *F. oxysporum* f. sp. *lycopersici* Fol4287 were easily demonstrated by TEM analysis as shown in Figures 7 and 8. TEM micrographs showed the morphological changes of the treated *S. aureus* ATCC25923 cells and inhibition of cell

multiplication. The treated *F. oxysporum* f. sp. *lycopersici* Fol4287. TEM micrographs showed many changes, including reduced size of treated cells, the formation of a mucilage matrix connecting the hyphal cells together, the appearance of big vacuole and lipid droplets with severe leakage of cytoplasmic contents.



Figure 7 The bactericidal effect of AgNPs on the ultrastructure of *S. aureus* ATCC25923. (a) is a negative control (without nanosilver). Note normal cell division (arrow) and DNA replication. (b), (c), (d) and (e) are treated samples, there is no cell division observed at 150 µg/mL of AgNPs. The amount of DNA appeared to be less than those of untreated ones (arrow).



Figure 8 The antifungal activity of AgNPs on the ultrastructure of *F. oxysporum* f. sp. *lycopersici* Fol4287. (a) is a negative control (without nanosilver). Note normal cell wall (W), plasma membrane (PM), Vacuole (V) and compact cytoplasm (Cy). (b), (c) and (d) are treated samples, note the formation of a mucilage substance (M) connecting the hyphae together. Note also big vacuole (V) and lipid droplets (L).

### DISCUSSION

Bacteria are considered as an excellent source for the extracellular biosynthesis of nanomaterials. There is a bigwig whack to discover novel bacterial strains having motivated biological potential (Galvez et al., 2019). The crude metabolite of E. coli D8 MF06257 was used as a reducing agent, solvent typology and capping agent in the NPs extracellular biosynthesis. This type of synthesis is safe, renewable, simple, eco-friendly and cost-effective (Saifuddin et al., 2009). This biosynthesis was performed within 1-2 minutes at room temperature and sun light. The colour alteration into brown was due to the excitation of surface plasmon vibrations in the AgNPs (Baalousha et al., 2006). The reduction of silver ions may be resulting from the NADH dependent enzymes activity present in the crude metabolite and/or some redox agents such as sulfur-containing proteins (Krishnaraj et al., 2012). The present study reported the ability of E. coli D8 to produce NR (NADH dependent enzymes) with enzyme activity 2.18  $\mu$ mol/hr/ml while it was about 0.152  $\mu$ mol/ hr/ mL for B. subtilis as reported by Saifuddin et al. (2009). The deactivated NR of E. coli D8 metabolite (by heating) did not exhibit any synthesis of AgNPs in the dark condition, while it produced AgNPs in sunlight after 1-2 minutes indicating NR is not the only factor in the silver ion reduction. Duan et al. (2015) suggested quinones act as an electron shuttle compound in the presence of sunlight and reduced silver ions into AgNPs. Sharma et al. (2012) believed that the crude metabolite of E. coli contains three water soluble quinones; menaquinone, demethylmenaquinone and ubiquinone.

The Spectrophotometer analysis of the plasmon absorption of the produced AgNPs showed maximum peak at 429 nm indicating to the good dispersion of particles in the nanocolloids as reported by **Okumura** *et al.* (2016). Stability, size and shape of nanomaterials are essential characteristics for use in biomedical applications (**Pal** *et al.*, 2007). Variable conditions such as concentration of the metal ions, temperature, the incubation period, pH (**Krishnaraj** *et al.*, 2012) and effect of the solar irradiation (**Boopathi** *et al.*, 2012) were tested to form regular shaped, small sized, and stable NPs.

The optimization processes confirmed that the concentration of silver nitrate 1.5mM was the most suitable for AgNPs biosynthesis that was also reported by Rahimi et al. (2016) results. Studying the different levels of temperatures, 35°C were found to be optimum for the stable formation of AgNPs, which matched with Shahzad et al. (2019). The pH of the reaction mixture greatly influenced the AgNPs formation as well as stability (Xiu et al., 2012). The best conditional pH value was 7.0 which perform regular and stable AgNPs biosynthesis (Gurunathan et al., 2009). The synthesis of AgNPs in the presence of solar irradiation was observed through the first minute and became over-reading at UV-Vis spectroscopy after the sixth minute. Similarly, Streptomyces sp. was used to reduce AgNO3 in solution within few minutes using sunlight (Abou-Dobara et al., 2017). The XRD pattern for AgNPs diffraction peaks confirmed the incidence of the face centered cubic (FCC) crystal structure on the crystalline AgNPs (Hu et al., 2013) which matched with the produced AgNPs by Mukherji et al. (2018). These results confirmed the synthesis of AgNPs after the reduction of AgNO3 (Li and Liu, 2010). TEM micrographs showed that the biosynthesized NPs were found to be spherical in shape with a size in the range of 6-17 nm which lies in the best class (0-30 nm) which matched with those obtained by Balakumaran et al. (2016) and El-Naggar et al. (2016). Also, Gopinath and Velusamy (2013) used Bacillus sp. GP-23 as spherical NPs with size in the range of 7-21 nm producers. In addition, the produced AgNPs has a negative charge (-33.6) which matched with Bacillus sp. AgNPs, that reported by Elbeshehy et al. (2015).

The aggregation of NPs is considered as a common problem which decreases their biological potential. The outer capping agents determine the size and morphology of NPs by preventing their aggregation (**Duan et al., 2015**). The biosynthesized AgNPs from *E. coli* D8 crude metabolite were uniform and monodispersed in size as well as stable for more than 6 months without aggregation at room temperature compared to *Trichoderma longibrachiatum* AgNPs that produced by **Elamawi et al. (2018**). The FTIR spectrum confirmed the presence of proteins associated with *E. coli* D8 AgNPs, which might act as a capping and stabilizing agent. Moreover, the negative charge of AgNPs might increase the repulsion force between particles which minimize their aggregation (**Siddique et al., 2013**).

The biosynthesized AgNPs by the crude metabolite of *E. coli* D8 exhibited some potent inhibitory activities against all the pathogenic strains. The highest antibacterial activity of AgNPs by *E. coli* D8 crude metabolite was recorded against *S. aureus* ATCC25923 followed by *E. coli* ATCC25922, multi-drug resistant *P. aeruginosa* ATCC27853, *K. pneumoniae* ATCC33495 and *B. cereus* ATCC6633. Generally, the previous antimicrobial activities were more competitive to AgNPs produced by *B. licheniformis* (Gomaa, 2017) and matched with the antimicrobial activity of *S. viridodiastaticus* AgNPs (Mohamedin *et al.*, 2015).

The biosynthesized AgNPs by the crude metabolite of *E. coli* D8 exhibited some potent inhibitory activities against all the pathogenic strains. Furthermore, the antifungal activity of AgNPs exhibited a great interest as *E. coli* D8 AgNPs showed a significant antifungal activity against *C. albicans* ATCC10231. Balakumaran et al. (2016) AgNPs produced by *A. terreus* inhibited the growth

of *C. albicans* with a proximate activity. In addition, *E. coli* D8 AgNPs possessed a superior potent toxic effect against *A. niger* followed by *A. fumigatus, A. flavus, A. alternata* and *F. oxysporum* f. sp. *lycopersici.* Similarly, the biosynthesized AgNPs by *Streptomyces* sp. VITSTK7 showed anti-*Aspergillus* activity against *A. niger, A. flavus* and *A. fumigatus* with antifungal index in the range of 62-75% (**Thenmozhi et al., 2013**). The AgNPs of *E. coli* D8 showed a significant antifungal activity against *F. oxysporum* f. sp. *lycopersici* that matched with AgNPs, produced by *Cryphonectria* sp. (**Dar et al., 2013**). In contrast, the mycelial growth of plant pathogenic *A. alternata* was less inhibited by *E. coli* D8 AgNPs than AgNPs produced by *A. solani* F10 (**Abdel-Hafez et al., 2016**).

E. coli D8 AgNPs increased the antifungal activity of the Fluconazole and increased diameters of ZOI. This synergistic effect was revealed increases in the fold areas especially against A. niger, A. flavus and F. oxysporum f. sp. lvcopersici. Gajbhiye et al. (2009) reported that the combination between Fluconazole and AgNPs increased ZOI and fold areas against Phoma glomerata, P. herbarum, F. semitectum, Trichoderma sp. and C. albicans. It was thought that the synergistic effect may be due to formation of AgNP-Fluconazole complex by chelating that lead to more serious damage to microbe's cells (Fayaz et al., 2010). Antimicrobial potential depended on the dose and it increased by increasing the concentration. The MIC values for S. aureus ATCC25923 and C. albicans ATCC10231 were 6.25 µg/ml and 50 µg/ml, respectively. Paul et al. (2018) reported the MIC values for different Candida species against curcumin-AgNPs ranged from 31.2 µg/ml to 250 µg/ml. AgNPs of Pilimelia columellifera subsp. pallida SL19 strain inhibited C. albicans at concentration equal to 64 µg/ml as reported by Wypij et al. (2017). The MIC of the biosynthesized AgNPs against S. aureus was found to be 2 µg/ml as reported by Yuan et al. (2017) and Wady et al. (2014). Regarding antimicrobial activity, there are various hypothetical mechanisms of the action of AgNPs on microbes' cells, such as interacting with the cell wall leading to cell burst (Radzig et al., 2013). Both S. aureus ATCC25923 and F. oxysporum f. sp. lycopersici Fol4287 changed in response to their treatment with AgNPs which were easily trapped and absorbed through cell membranes. The changes included inhibition of S. aureus ATCC25923 multiplication in addition to having a low amount of DNA. The untreated hyphal cells of F. oxysporum f. sp. lycopersici Fol4287 showed a normal cell wall, compact cytoplasm, cell membrane and small vacuole. On the other hand, many changes were observed after the treatment by AgNPs such as the formation of a mucilage matrix connecting the hyphal cells together, the appearance of big vacuole and lipid droplets. The accumulating of AgNPs in cytoplasmic membrane and cytoplasm may be the main factor in the major morphological changes in addition to the accumulation in the cell nucleus (Abdel-Hafez et al., 2016). This accumulation may be indicated to the interaction of AgNPs with DNA (Vahdati and Sadeghi, 2013). In addition, the smallest AgNPs can penetrate the cell membranes and interact with their proteins (including thiol groups in enzymes) leading to blocking, inactivation and cell death (Radzig et al., 2013).

# CONCLUSIONS

*Escherichia coli* D8 (MF062579) crude metabolite was able to synthesize AgNPs within 1-2 minutes in a green and cost-effective method. The presence of protein was confirmed and could be acted as stabilizing and capping agents. This method provides AgNPs possessing competitive size, shape and antimicrobial action beside to synergy potential with Fluconazole against *A. niger* van Tiegh, *A. flavus* Link ex Fries group and *F. oxysporum* f. sp. *lycopersici* Fol4287. Thusly, *E. coli* could be developed as a nano-biofactory against pathogenic microbes having severe damage effects on their DNA structure.

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