# Few-Layer Nanoplates of Bi<sub>2</sub>Se<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> with Highly Tunable Chemical Potential

#### Desheng Kong,<sup>†</sup> Wenhui Dang,<sup>†</sup> Judy J. Cha,<sup>†</sup> Hui Li,<sup>†</sup> Stefan Meister,<sup>†</sup> Hailin Peng,<sup>\*,†</sup> Zhongfan Liu,<sup>†</sup> and Yi Cui<sup>\*,†</sup>

<sup>†</sup>Department of Materials Science and Engineering, Stanford University, Stanford, California 94305 and <sup>†</sup>Center for Nanochemistry, Beijing National Laboratory for Molecular Sciences (BNLMS), State Key Laboratory for Structural Chemistry of Unstable and Stable Species, College of Chemistry and Molecular Engineering, Peking University, Beijing 100871, People's Republic of China

**ABSTRACT** A topological insulator (TI) represents an unconventional quantum phase of matter with insulating bulk band gap and metallic surface states. Recent theoretical calculations and photoemission spectroscopy measurements show that group V–VI materials Bi<sub>2</sub>Se<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub>, and Sb<sub>2</sub>Te<sub>3</sub> are TIs with a single Dirac cone on the surface. These materials have anisotropic, layered structures, in which five atomic layers are covalently bonded to form a quintuple layer, and quintuple layers interact weakly through van der Waals interaction to form the crystal. A few quintuple layers of these materials are predicted to exhibit interesting surface properties. Different from our previous nanoribbon study, here we report the synthesis and characterizations of ultrathin Bi<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Se<sub>3</sub> nanoplates with thickness down to 3 nm (3 quintuple layers), via catalyst-free vapor—solid (VS) growth mechanism. Optical images reveal thickness-dependent color and contrast for nanoplates grown on oxidized silicon (300 nm SiO<sub>2</sub>/Si). As a new member of TI nanomaterials, ultrathin TI nanoplates have an extremely large surface-to-volume ratio and can be electrically gated more effectively than the bulk form, potentially enhancing surface state effects in transport measurements. Low-temperature transport measurements of a single nanoplate device, with a high-k dielectric top gate, show decrease in carrier concentration by several times and large tuning of chemical potential.

KEYWORDS Topological insulator, nanoplate, bismuth selenide, bismuth telluride

he growth and physical properties of  $A_2B_3$  (A = Bi, Sb; B = Se, Te) chalcogenide materials have been studied for more than half a century due to their interesting thermoelectric properties.<sup>1–3</sup> Recently, however, research interest in this family of compounds increased dramatically because Bi<sub>2</sub>Se<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub>, and Sb<sub>2</sub>Te<sub>3</sub> were discovered to be three-dimensional (3D) topological insulators (TIs), a new quantum matter with conductive massless Dirac fermions on their surface.<sup>4</sup> In TI, strong spin–orbit coupling induces nontrivial surface states that are robustly protected against any time reversal perturbations, such as crystal defects and nonmagnetic impurities.<sup>5–9</sup> The distinct properties of TI surface states are attractive for fundamental research as well as applications in spintronics, quantum information, and low-energy dissipation electronics. Experimentally, angle resolved photoemission spectroscopy (ARP-ES) measurements on bulk single crystals have verified the existence of the 3D TI and spin-momentum locked Dirac fermion nature of the surface states.<sup>10–13</sup>

 $Bi_2Se_3$ ,  $Bi_2Te_3$ , and  $Sb_2Te_3$  share the same layered rhombohedral crystal structure in space group  $D^5{}_{3d}$  (*R*3*m*). Each charge neutralized layer is formed by five covalently bonded atomic sheets, for example, Se–Bi–Se–Bi-Se in Bi<sub>2</sub>Se<sub>3</sub>,

\* To whom correspondence should be addressed: hlpeng@pku.edu.cn (H.P.) and yicui@stanford.edu (Y.C.). Received for review: 04/11/2010

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defined as a quintuple layer (QL) with thickness of  $\sim 1$  nm (Figure 1A). The quintuple layers are weakly bonded together by van der Waals interaction to form the crystal. These anisotropic crystallography characteristics are significant for controllable growth of TI materials.

Synthesizing low-dimensional structures out of this family of compounds is attractive for TI study. Compared with bulk samples, nanostructured TI materials are a compelling system for probing TI surface states due to their large

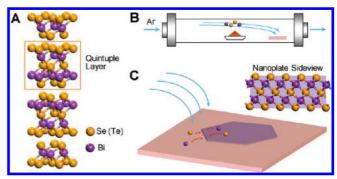


FIGURE 1. (A) Layered crystal structure of  $Bi_2Se_3$  and  $Bi_2Te_3$ , with each quintuple layer (QL) formed by five Bi and Se (or Te) atomic sheets. (B) A schematic drawing of a vapor—solid (VS) growth process for few-layer nanoplates (NPs) of  $Bi_2Se_3$  and  $Bi_2Te_3$  in a horizontal tube furnace. (C) A schematic drawing of NPs growth mechanism, involving gas phase atoms diffused and attached to the side surface. A side view of the NPs (1QL) shows the Se-terminated top and bottom surfaces with saturated bonds and side surfaces with dangling bonds ready to bind with incoming atoms.

surface-to-volume ratio, in favor of manifesting the surface states during transport measurements. For example, Aharonov–Bohm oscillation was recently observed in Bi<sub>2</sub>Se<sub>3</sub> nanoribbons by some of the present authors, providing direct transport evidence of the robust, conducting surface states.<sup>14,15</sup> The Kondo effect on magnetically doped Bi<sub>2</sub>Se<sub>3</sub> nanoribbons has also been observed.<sup>16</sup> In addition, theoretical calculations and ARPES measurements on ultrathin molecular beam epitaxy (MBE) films of Bi<sub>2</sub>Se<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> reveal the unconventional size effects: evolution from 2D TI to 3D TI phase depending on film thicknesses.<sup>17,18</sup> A thin film form of TI can be obtained by MBE growth, <sup>18,19</sup> but MBE is expensive and less accessible. Mechanical exfoliation of thin sheets from bulk crystals,<sup>20</sup> widely used for obtaining graphene from graphite, can also achieve thin TI layers, but with very low yield and irregular shapes. Nanoribbons and nanowires of chalcogenide materials are easily synthesized by a vapor-liquid-solid (VLS) process, <sup>14,16,21-24</sup> but only with moderate thickness range (~30-100 nm). A facile synthesis method to obtain ultrathin TI materials down to few QLs is desired. Here, we developed a catalyst-free vapor-solid (VS) synthesis process to grow ultrathin nanoplates (NPs) of  $Bi_2Se_3$  and  $Bi_2Te_3$  with thicknesses down to  $\sim$ 3 nm (3QLs). These NPs show a strong electrical gating response and highly tunable chemical potential in transport measurements, attractive for probing TI surface states and future device applications.

Synthesis of Bi<sub>2</sub>Se<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> NPs is carried out in a 12 in. horizontal tube furnace (Lindberg/Blue M) via catalystfree vapor transport and deposition process (Figure 1B), using a setup similar to our previous nanoribbon synthesis by VLS growth.<sup>14,16,24–26</sup> Powder of source materials (99.999 % Bi<sub>2</sub>Se<sub>3</sub> or Bi<sub>2</sub>Te<sub>3</sub>) is placed in the hot center of the tube furnace. Oxidized silicon wafers (300 nm SiO<sub>2</sub>/Si) are placed at the cold zone ( $\sim$ 12 cm away from the center) as growth substrates. For the synthesis, the tube is initially pumped down to a base pressure of about 100 mTorr and flushed with ultrapure Ar several times to remove oxygen residue. With Ar gas flow (50 sccm for  $Bi_2Se_3$  and 20 sccm for  $Bi_2Te_3$ ), the hot center of the furnace is heated to growth temperature (460-500 °C for Bi<sub>2</sub>Se<sub>3</sub> and 450-480 °C for Bi<sub>2</sub>Te<sub>3</sub>) and is held at the temperature and a certain pressure (100 Torr for Bi<sub>2</sub>Se<sub>3</sub> and 20 Torr for Bi<sub>2</sub>Te<sub>3</sub>) for about 5 min, followed by natural cooling down. The SiO<sub>2</sub>/Si substrate is in the temperature range of 320-360 °C.

The synthesized products are characterized by scanning electron microscopy (SEM, FEI XL30 Sirion and Hitachi S-4800) and atomic force microscopy (AFM, Park Systems XE-70 and Digital Instruments Nanoscope IIIA). Figure 2 shows a SEM image of Bi<sub>2</sub>Te<sub>3</sub> NPs (SEM image of Bi<sub>2</sub>Se<sub>3</sub> NPs is provided in Supporting Information, Figure S1). NPs usually exhibit triangular or hexagonal morphologies with lateral dimensions extending from several micrometers up to ~20  $\mu$ m. On average, the lateral dimensions of Bi<sub>2</sub>Te<sub>3</sub> NPs. We observe 60° or 120°

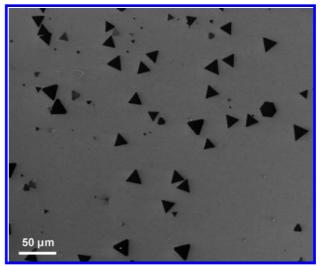


FIGURE 2. A SEM image of  $Bi_2Te_3$  NPs grown on oxidized silicon substrate (300 nm SiO<sub>2</sub>/Si). A SEM image of  $Bi_2Se_3$  NPs is provided in Supporting Information (Figure S1).

facets at the edges, as expected from their crystal structure. SEM images show that these NPs grow directly on oxidized Si surface via VS growth mechanism, without catalyst particles at the ends of NPs. They are strongly bonded to the substrate and hardly removed during sonication in organic solvents. The thickness of synthesized NPs is measured by AFM, shown in Figure 3A as an example. Most of the NPs have very flat surfaces with a uniform thickness across the lateral dimensions (Figure 3B for Bi<sub>2</sub>Te<sub>3</sub> and Figure 3C for  $Bi_2Se_3$ , respectively). The thickness distribution for  $Bi_2Te_3$  is obtained by scanning a substrate area of  $\sim 200 \,\mu\text{m}^2$  (Figure 3D). NPs with uniform thicknesses between 3QLs and 6QLs (3–6 nm) are major products. We also observe minority NP products whose thickness is not uniform but with 1-4 QL steps (Supporting Information, Figure S2, and panels B and C of Figure 4).

Contrast difference in optical images provides a facile way to identify ultrathin TI NPs of different thicknesses. Optical images are acquired by a Leitz ERGOPLAN microscope (10× eyepieces combined with  $50 \times$  and  $150 \times$  objective lens), under white light illumination conditions. As-grown ultrathin TI NPs are readily observable on oxidized silicon substrates  $(300 \text{ nm SiO}_2/\text{Si})$ , as shown in Figure 4A ( $50 \times$  objective lens). Here, we establish the relation between optical contrast and the thickness of TI NPs, by correlating optical images (Figure 4B, with  $150 \times$  objective lens) with AFM data (Figure 4C) for a NP with multiple single QL steps. As thickness increases from 4QLs to 13QLs (marked by black numbers in Figure 4C), the optical color of the NP in Figure 4B gradually changes from dark purple to yellow. For TI NPs less than 13QLs, the optical contrast variation allows differentiating thickness difference in a step of 1QL. Similar to identifying few-layer graphene deposited on oxidized silicon,<sup>27</sup> SiO<sub>2</sub> thickness is critical to enhance the optical contrast difference in ultrathin NPs of various thicknesses. Physical origin of the

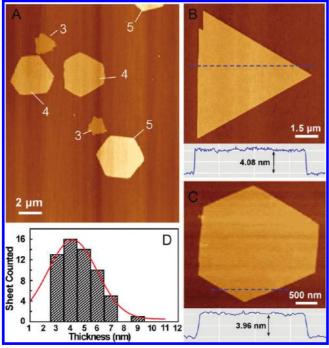


FIGURE 3. AFM characterizations of a few QL Bi<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Se<sub>3</sub> NPs. (A) Overview of several ultrathin Bi<sub>2</sub>Te<sub>3</sub> NPs. Numbers indicate the number of quintuple layers. (B) AFM image and a height profile (corresponding to the dashed line in the image) of a Bi<sub>2</sub>Te<sub>3</sub> NP with 4QLs. (C) AFM image and a height profile (corresponding to the dashed line in the image) of a Bi<sub>2</sub>Se<sub>3</sub> NP with 4QLs. (D) Histogram of Bi<sub>2</sub>Te<sub>3</sub> NP thickness obtained from substrate area of ~200  $\mu$ m<sup>2</sup>. The smooth curve is a Gaussian fit to the thickness distribution.

large optical contrast is ascribed to strong amplitude modulation of reflection at the air-2D TI crystal $-SiO_2$  interface.<sup>28</sup>

The reason that the VS growth mechanism can produce ultrathin NPs (3-6 nm) is due to the anisotropic bonding nature of Bi<sub>2</sub>Se<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub>. Nucleation takes place randomly on the SiO<sub>2</sub> surface and more probably on physical or chemical inhomogeneous sites of substrate, such as surface roughness and impurities. Nuclei have the c crystal axis perpendicular to the substrate surface. After nucleation, the growth process continues with a high anisotropic nature (Figure 1C). The atoms at the edge of the nuclei have dangling bonds ready to bind covalently with incoming atoms. The top surface is terminated with chemically saturated Se atoms. Atoms adsorbed onto the top surface from gas phase cannot form covalent bonds with the top Se atoms and, consequently, tend to diffuse around to find the edges. Therefore, the lateral dimension grows much faster than the vertical thickness dimension. The high growth anisotropy results from the bonding anisotropy. Using the ratio of the lateral to vertical dimension of NPs, we estimate that the growth anisotropy is several thousand. In fact, nucleation of additional quintuple layers on top of the existing NPs does take place slowly. Additional nucleations are believed to happen close to the edge of existing NPs, since we were able to catch some NPs with the edge thicker than the center (Figure 4 panels B and C and Supporting Information, Figure S2).

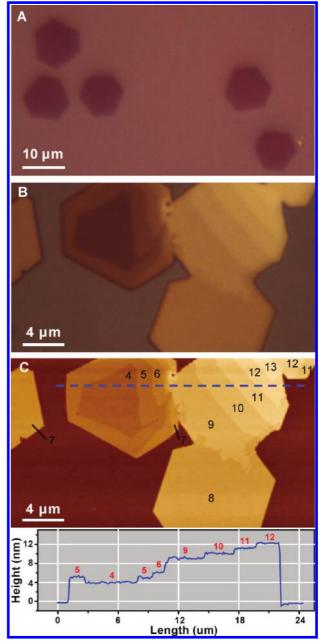


FIGURE 4. (A) A typical optical image of ultrathin NPs of  $Bi_2Te_3$  grown on an oxidized Si substrate (300 nm SiO<sub>2</sub>/Si) by  $50 \times$  objective lens. NPs have clear optical contrast from the substrate for easy identification. (B) High-magnification optical image (by  $150 \times$  objective lens) of NPs with multiple step edges. The color changes from dark purple to yellow color, as the thickness increases from 4QLs to 13QLs. (C) An AFM image and height profile (corresponding to the dashed line-cut) from the same NPs shown in (B). The number of QLs is marked in AFM (black) and height profile (red). These NPs have multiple single QL steps, confirming layer-by-layer growth modes. By comparing between (B) and (C), optical images indeed provide sufficient contrast for differentiating each QL from 4QLs to 13QLs.

To characterize the crystal structure and chemical compositions of synthesized products, NPs are studied by a 200 kV FEI Tecnai F20 transmission electron microscope (TEM) and energy-dispersive X-ray spectroscopy (EDX) equipped



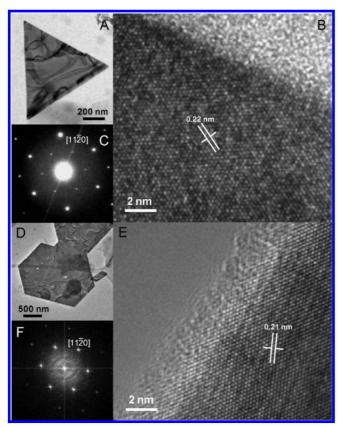


FIGURE 5. TEM characterizations of  $Bi_2Te_3$  and  $Bi_2Se_3$  NPs. (A) TEM image of a  $Bi_2Te_3$  NP grown on 100 nm  $SiN_x$  membrane. HRTEM image (B) and SAED pattern (C) acquired from the same  $Bi_2Te_3$  NP. (D) TEM image of a  $Bi_2Se_3$  NP transferred to a lacey carbon support film on a TEM copper grid. (E) HRTEM image obtained from the same  $Bi_2Se_3$  NP, with corresponding fast Fourier transformation (FFT) pattern shown in (F).

inside the TEM. TEM sample preparation details are provided in Supporting Information. Figure 5A shows a typical lowmagnification TEM image of Bi<sub>2</sub>Te<sub>3</sub> NPs grown on 100 nm SiN<sub>x</sub> membrane. Corresponding high-resolution TEM (HR-TEM) lattice fringes and spot pattern of selected area electron diffraction (SAED), shown in panels B and C of Figure 5, demonstrate the single crystalline nature of the NPs. A HRTEM image reveals expected hexagonal lattice fringes with a lattice spacing of 0.22 nm, consistent with the lattice spacing of (11-20) planes. Similar to VLS nanoribbons, NPs have (0001) facets as top and bottom surfaces, and (01-10)facets as side surfaces.<sup>14,16,24</sup> The TEM image of Bi<sub>2</sub>Se<sub>3</sub> NPs transferred to a lacey carbon support film on a TEM copper grid is shown in Figure 5D. The HRTEM image and corresponding FFT pattern, shown in panels E and F of Figure 5, prove it is also single crystal with similar structural characteristics to Bi<sub>2</sub>Te<sub>3</sub> NPs. EDX analysis out of single NPs confirms the correct stoichiometry, within the accuracy of the measurements (Supporting Information, Figure S3).

Electrical transport properties are studied in single NP devices with a top gate. Here, we use  $Bi_2Te_3$  NP devices to demonstrate the tunable chemical potentials and pronounced weak antilocalization effect. Ohmic contacts to NPs

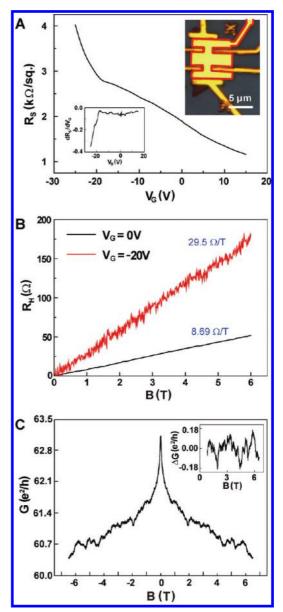


FIGURE 6. Electron transport measurements at 2 K. (A) Device sheet resistance  $R_{\rm S}$  vs top gate voltage  $V_{\rm G}$ .  $R_{\rm S}$  inceases by ~210% at negative gate voltage of -25 V. Differential resistance over gate voltage is shown in the inset on the left to highlight the enhancement of gating response for  $V_{\rm G} < -17.5$  V. A typical optical image of a NP device is shown in the inset on the right, in which the top gate is encircled by a red line to guide the eye. (B) Hall measurements at  $V_{\rm G} = 0$  V and  $V_{\rm G} = -20$  V. Areal carrier concentration decreases by a factor of 3 by applying -20 V negative gate voltage. (C) Magneto-conductance in the field range of  $\pm 6.5$  T. Conductance oscillations (inset) are obtained by subtracting the background.

are fabricated by standard e-beam lithography and thermal evaporation of Cr/Au contacts (10 nm/160 nm). A thin high-k dielectric layer ( $\sim$ 30 nm) of Al<sub>2</sub>O<sub>3</sub> is deposited by atomic layer deposition (Cambridge NanoTech Inc., details in Supporting Information). An e-beam lithographically patterned Cr/Au (10 nm/160 nm) metal top gate is made by thermal evaporation afterward. An optical image of a typical device is shown as the right inset of Figure 6A. Transport measure-

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ments are performed at 2 K in a Quantum Design PPMS-7 instrument, combined with a digital lock-in amplifier (Stanford Research Systems) and a Keithley 2400 source meter (setup details are provided in Supporting Information). For resistance measurement, a four-point configuration is used to remove contribution from the contact resistance. The change in four-point sheet resistance R<sub>s</sub> with top gate voltage  $V_{\rm G}$  is shown in Figure 6A. The gating response indicates the NPs are a n-type doped semiconductor, similar to the case of Bi<sub>2</sub>Se<sub>3</sub> nanoribbons.<sup>14,16,24</sup> The n-type nature is usually ascribed to Te/Se vacancies and antisite defects formed during the growth process.<sup>11,13</sup> For a typical NP device, the resistance (Figure 6A) increases quasi-linearly with negative gate voltage up to -17.5 V, beyond which the resistance rises more rapidly when the gate voltage further increases toward negative direction. The inset of Figure 6A shows the slope of the resistance as a function of the gate voltage, demonstrating the sharper increase of the resistance beyond -17.5 V. This enhancement of the gating effects strongly suggests that the chemical potential is approaching the bottom of the conduction band at the gate voltage of -17.5V, beyond which a large portion of bulk electrons are being depleted by additional negative gate voltage. The Hall measurements (Figure 6B) show the areal carrier density of the NP is  $7.2 \times 10^{13}$  cm<sup>-2</sup> at zero gate voltage, and  $2.1 \times 10^{13}$  $cm^{-2}$  at -20 V gate voltage (change by a factor of 3). Hall measurements confirm as-grown NPs are a heavily doped n-type semiconductor. Nevertheless, NP devices show much stronger gate response than previously reported nanoribbons and mechanical exfoiliated samples.<sup>29,30</sup> The pronounced gate effects and tunable chemical potential are achieved by the combination of a strong electrical field from the top gate using a high-k dielectric layer and ultrathin thickness of NPs from VS growth. The magnetoconductance of the same NP device is shown in Figure 6C, where the applied magnetic field (B) is perpendicular to the NP plane. At low magnetic field |B| < 0.80 T, the weak antilocalization effect manifests as a sharp peak in the magnetoconductance trace, resulting from strong spin-orbit coupling in TI materials.14,31 The oscillations in conductance (inset of Figure 6C) show aperiodic field dependence, as universal conductance fluctuation (UCF) commonly observed in mesoscopic metallic and semiconducting structures.<sup>32</sup> It is also observed in our previous Bi<sub>2</sub>Se<sub>3</sub> nanoribbon measurement.<sup>14</sup> We expect that further lowering of the bulk carrier concentration may reveal interesting physics from TI surface states.

In summary, we develop a VS synthesis process to grow few-layer nanoplates of topological materials  $Bi_2Se_3$  and  $Bi_2Te_3$ , with thicknesses down to  $\sim 3$  nm, or three quintuple layers. Optical images reveal thickness-dependent color and contrast for NPs grown on top of an oxidized silicon surface (300 nm SiO<sub>2</sub>/Si), similar to the case of graphene. Single NP devices with high-k dielectric top gate configuration show a highly tunable chemical potential and a pronounced weak antilocalization effect.

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Supporting Information Available. Additional SEM image, AFM image, EDX spectra, TEM sample preparations, growth conditions for  $Al_2O_3$  dielectric films by atomic layer deposition (ALD), and transport measurement equipment setup. This material is available free of charge via the Internet at http://pubs.acs.org.

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