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## Erratum: fine and hyperfine excitation of $C_2H$ by collisions with He at low temperature

## by A. Spielfiedel,<sup>1</sup> N. Feautrier,<sup>1</sup>\* F. Najar,<sup>2</sup> D. Ben Abdallah,<sup>2</sup> F. Dayou,<sup>1</sup> M. L. Senent<sup>3</sup> and F. Lique<sup>1,4</sup>

<sup>1</sup>LERMA and UMR 8112 of CNRS, Observatoire de Paris-Meudon, F-92195 Meudon Cedex, France

<sup>2</sup>Laboratoire de Spectroscopie atomique, moléculaire et Application, Faculté des Sciences-Université Tunis el Manar, Tunis 1060, Tunisia <sup>3</sup>Departmento de Fisica y Quimica Teóricas, Instituto de Estructura de la Materia, CSIC, Calle Serrano 121, E-28006 Madrid, Spain

<sup>4</sup>LOMC - FRE 3102, CNRS-Université du Havre, 25 rue Philippe Lebon, BP 540, F-76058 Le Havre, France



**Figure 1.** Temperature variation of hyperfine resolved C<sub>2</sub>H-He rate coefficients for N = 3, J,  $F \rightarrow N' = 2$ , j', F' (upper panel), N = 4, J,  $F \rightarrow N' = 2$ , j', F' (lower panel) and  $\Delta j = \Delta N$  transitions.







**Figure 3.** Temperature variation of rate coefficients for  $N = 2, J = 2.5, F = 2 \rightarrow N = 2, j', F'$  transitions.

		N' = 3 j' = 2.5 F' = 2	N' = 3 j' = 2.5 F' = 3	N' = 3 j' = 3.5 F' = 3	N' = 3 j' = 3.5 F' = 4	N' = 2 j' = 1.5 F' = 1	N' = 2 j' = 1.5 F' = 2	N' = 2 j' = 2.5 F' = 2	N' = 2 j' = 2.5 F' = 3
<i>j</i> = 3.5	F = 3 $F = 4$	5.07 0.14	0.45 5.38	0.33 0.17	0.22 0.38	20.97 0.47	4.19 24.71	2.80 0.98	1.46 3.28
<i>j</i> = 4.5	F = 4 $F = 5$	0.11 0.11	0.15 0.16	5.49 0.12	0.31 5.68	0.37 0.41	0.68 0.64	25.76 0.43	2.63 27.96

**Table 1.** Hyperfine resolved C<sub>2</sub>H-He de-excitation rate coefficients (×10<sup>-12</sup> cm<sup>3</sup> s<sup>-1</sup>) for  $N = 4, j, F \rightarrow N' = 2, 3, j', F'$  transitions at 50 K.

The paper was published in MNRAS, 421, 1891 (2012). An error in the hyperfine energy levels which affected the results reported in Table 1 and Figs 4–6 was discovered. Wrong hyperfine energy levels of the  $C_2H$  molecule were used in the calculations of the hyperfine-resolved rate coefficients, resulting in an underestimation of those rate coefficients, especially at low temperature.

However, the comments (Section 4.2) and the conclusion (Section 5) do not change. The error is corrected here: Table 1 replaces the previous one and Figs 1, 2 and 3 replace previous Figs 4, 5 and 6, respectively. The complete set of corrected (de-)excitation

rates coefficients will be made available online in the LAMDA<sup>1</sup> and BASECOL<sup>2</sup> websites.

<sup>1</sup> http://www.strw.leidenuniv.nl/~moldata/ <sup>2</sup> http://basecol.obspm.fr/

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